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# Trapping of trifluoroacetonitrile imines with mercaptoacetaldehyde and mercaptocarboxylic acids: an access to fluorinated 1,3,4-thiadiazine derivatives via (3+3)-annulation

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Dedicated to Professor Hans-Ulrich Reissig on the occasion of his 70th birthday



#### **Graphical abstract**

- nitrile imines derived from trifluoroacetonitrile can easily be trapped by bifunctional Snucleophiles
- adducts obtained with mercaptoethanal undergo a spontaneous (3+3)-cyclisation to give 6membered products

- oxidation of 1,3,4-thiadiazin-5-ols with PCC leads to the expected 1,3,4-thiadiazin-5-one derivatives
- 1:1 adducts obtained with mercaptocarboxylic acids undergo cyclization only in the presence of EDC methiodide

#### Abstract

The *in situ* generated highly electrophilic nitrile imines derived from trifluoroacetonitrile are efficiently trapped by monomeric mercaptoacetaldehyde at room temperature in THF solution. The initially formed 1:1 adducts undergo spontaneous cyclization leading to 6-membered 5,6-dihydro-1,3,4-thiadiazin-5-ols bearing the CF<sub>3</sub> group at the C(2) atom in good yields. Subsequent oxidation of the latter products using C<sub>5</sub>H<sub>5</sub>N·HCl·CrO<sub>3</sub> (PCC) led smoothly to the expected 1,3,4-thiadiazin-5-ones as final products. Alternatively, analogous 2-trifluoromethylated heterocycles were obtained via trapping of the same nitrile imines with  $\alpha$ -mercaptocarboxylic acids followed by 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide methiodide (EDC methiodide) induced cyclisation of the initially formed 1:1 adducts. In general, efficiency of the second approach was lower and depended on the type of substituent attached to phenyl ring in the starting 1,3-dipole.

**Keywords**: fluorinated nitrile imines, mercaptoacetaldehyde,  $\alpha$ -mercaptocarboxylic acids, (3+3)-annulation, trifluoromethylated 1,3,4-thiadiazines.

#### 1. Introduction

According to Huisgen classification nitrile imines belong to the class of so-called propargyl-allenyl 1,3dipoles [1]. Over decades they have been widely applied in the (3+2)-cycloaddition reactions leading to diverse nitrogen-containing five-membered heterocycles [2]. However, along with their tendency to react as 1,3-dipoles, they are also known as electrophilic reagents, and the reactions with O-, N-, and S-nucleophiles are reported [3]. In our continuing studies on (3+2)-cycloadditions with fluorinated nitrogen-centered 1,3-dipoles [4-6], much attention was paid to strongly electron-deficient nitrile imines **1** derived from trifluoroacetonitrile. For example, we found that they react easily not only with aromatic and aliphatic thioketones but also with monomeric thiochalcones yielding the 1,3,4-thiadiazole derivatives of type **2** in a fully regioselective manner [5]. In addition, they underwent cycloadditions with less reactive electron-rich C=C dipolarophiles such as vinyl ethers and alkoxyallenes leading to pyrazoles **3** and more complex spirobipyrazolines **4**, respectively [6]; the latter products being formed in a highly diastereoselective fashion via formal double (3+2)-cycloaddition (Scheme 1) [6b].

Mercaptoacetaldehyde (5), easily available from its stable dimer 1,4-dithiane-2,5-diol (6), is well known as a versatile building block for the preparation of the S-containing organic products [7], and its reactions with nitrile imines afforded 1,3,4-thiadiazine derivatives formed via spontaneous cyclisation of the initial 1:1 adducts [8].

The goal of the present work was to examine the course of the reactions between the in situ generated nitrile imines **1** and mercaptoacetaldehyde under mild conditions. In addition, similar reactions with other bifunctional substrates, e.g.  $\alpha$ -mercaptocarboxylic acids **7** were also of interest. Both procedures were expected to open up an access to hitherto unknown trifluoromethylated 1,3,4-thiadiazin-6-ones **8** in a complementary approach (Scheme 1).



Scheme 1. Reactions of fluorinated nitrile imines 1 with thiocarbonyls (thioketones and thiochalcones), vinyl ethers, and alkoxyallenes leading to (3+2)-cycloadducts 2-4, and (3+3)-annulations of 1 with bifunctional substrates 5 and 7 described herein.

#### 2. Results and Discussion

A series of recent papers demonstrate that the title trifluoroacetonitrile imines 1 are readily available via base-mediated dehydrohalogenation of the respective hydrazonoyl bromides of type 9 [5-6], and the requisite precursors were prepared in a two-step protocol comprising condensation of fluoral hydrate with arylhydrazines [9] followed by bromination of the resulting hydrazones with NBS [5a]. In a test experiment the reaction of N'tolyl bromide 9a (X = CH<sub>3</sub>) and 1,4-dithiane-2,5-diol (6, 0.55 equiv.) was carried out in dry THF at room temperature, in the presence of excess of Et<sub>3</sub>N (5.0 equiv.) (Scheme 2). The reaction progress was monitored by TLC until the starting bromide 9a was fully consumed (2 hrs). The resulting mixture was examined by <sup>1</sup>H NMR, and only one set of signals attributed to the sole intermolecular product could be found in this spectrum. Subsequent purification by flash column chromatography provided analytically pure material isolated as a fairly stable colourless solid (78% yield), and the structure of the expected 5,6-dihydro-1,3,4-thiadiazin-5-ol derivative 10a was confirmed by spectroscopic methods. For example, in the <sup>1</sup>H NMR spectrum, the diagnostic absorptions of the diastereotopic protons at C(6) were found at 2.93 (dd, J = 1.7, 12.9 Hz) and 3.18 (dd, J = 3.4, 12.9 Hz) ppm. In addition, along with the characteristic pattern attributed to the p-tolyl group, the signals located at 3.06 (d, J =10.7 Hz) and 5.64 (ddd, J = 1.7, 3.4, 10.7 Hz) ppm clearly evidenced the presence of the C(5)H(OH) group. In the  $^{13}$ C NMR spectrum two diagnostic quartets found at 119.7 ( $^{1}J_{C-F} = 273.0 \text{ Hz}$ ) and 120.7 ( $^{2}J_{C-F} = 38.9 \text{ Hz}$ ) ppm, and a singlet located at -67.6 ppm in the <sup>19</sup>F NMR spectrum confirmed the presence of the CF<sub>3</sub>-C= unit. Moreover, the (-)-ESI-MS spectrum showed the molecular peak of  $[M-H]^-$  at m/z = 273.5, which corresponded to the formula C11H11F3N2OS.

In continuation of the study, a series of trifluoromethylated 5,6-dihydro-1,3,4-thiadiazinols of type 10 was prepared starting with differently substituted hydrazonoyl bromides 9, including representatives functionalized with either strongly electron-donating (9c) or electron-withdrawing (9f,g) groups located at the *para* position of the phenyl ring. In all the studied cases, the expected heterocyclic products 10 were isolated in fine yields (64-92%), irrespective of the electronic properties of substituent X, although longer reaction times (up to 4hr) were noticed for the transformations of nitrile imines 1 functionalized with EWGs attached to the phenyl ring. The decrease in reaction rates can be explained by reduced nucleophilicity of NH group present in the initially formed S-adduct. For that reason, the subsequent spontaneous cyclisation is expected to be a limiting step. Notably, neither nitrile nor nitro and ester moieties interfered with the outcome of the developed (3+3)-annulation reaction.



Scheme 2. Synthesis of 1,3,4-thiadiazin-5-ols 10a-10h via (3+3)-annulation of mercaptoacetaldehyde (5) with the *in situ* generated trifluoroacetonitrile imines 1a-1h.

The presence of the hemiaminal group in the newly synthesized 1,3,4-thiadiazine derivatives **10** offers several possibilities for further functionalization of the core heterocycle. Particularly, taking into account the well documented significance of the 1,3,4-thiadiazinone derivatives as biologically active compounds [10], we paid some attention to oxidation protocols. Thus, the behavior of the model N-tolyl substituted 1,3,4-thiadiazin-5-ol **10a** was briefly checked in reactions with selected mild oxidants. Unfortunately, neither treatment with activated MnO<sub>2</sub> nor with IBX under standard reaction conditions (CH<sub>2</sub>Cl<sub>2</sub>, 24 hr) provided the desired product. However, the use of excess of pyridinium chlorochromate (PCC, 1.5 equiv.) as an oxidizing agent in CH<sub>2</sub>Cl<sub>2</sub> at room temperature, afforded after 2h the expected thiadiazinone derivative **8a**, which was isolated by flash column chromatography in high yield of 83% (Scheme 3). Analogous transformations were also performed starting with **10a,b,e-g** to give the expected 1,3,4-thiadiazin-5-one derivatives **8** in high yields of 63-91%.



Scheme 3. Oxidation of selected 5,6-dihydro-1,3,4-thiadiazin-5-ols 10 with PCC leading to 1,3,4-thiadiazin-5-ones 8.

In extension of the study, we turned our attention to  $\alpha$ -mercaptocarboxylic acids of type **7** as the alternative bifunctional HS-nucleophiles for the preparation of target fluorinated 1,3,4-thiadiazine derivatives [11]. As shown in Scheme 4, the test experiment was performed using model N-tolylhydrazonoyl bromide **9a** and thioglycolic acid (**7a**), in dry THF, in the presence of excess of Et<sub>3</sub>N. The reaction progress was monitored by TLC, and after the nitrile imine precursor **9a** was fully consumed, the resulting mixture was acidified with diluted HCl followed by standard aqueous work-up. <sup>1</sup>H NMR spectrum of mother liquor showed the presence two diagnostic signals (singlets) located at 3.57 (2 H, CH<sub>2</sub>) and 9.60 (1 H, NH) ppm along with the characteristic absorptions attributed to the *p*-tolyl group. That can serve as confirmation of the synthesis of amides and other carboxylic acid derivatives [12], 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide methiodide (EDC

methiodide) was examined as the coupling reagent for the subsequent cyclisation step. As expected, treatment of crude **11a** with EDC methiodide under standard reaction conditions (dry THF, room temperature) provided after 16hr the desired 1,3,4-thiadiazin-5-one derivative **8a** in 27% yield (for 2 steps). Brief optimization of the final annulation step with respect *e.g.* to the type of the solvent, and stoichiometry of substrates indicated that running the reaction in dry  $CH_2Cl_2$  in the presence of slight excess EDC methiodide leads to product **8a** in a satisfactory overall yield of 50%. Next, a series of nitrile imines **1c-1g** was tested in the reaction with thioglycolic acid (**7a**) under analogous reaction conditions to provide the expected formal (3+3)-annulation products **8c-8g** in moderate yields.



Scheme 4. Synthesis of 1,3,4-thiadiazin-5-ones 8 via two-step (3+3)-annulation of nitrile imines 1 and  $\alpha$ -mercaptocarboxylic acid 7a.

Finally, another three HS-nucleophiles functionalized with carboxylic groups, namely 2mercaptopropionic acid (**7b**), mercaptosuccinic acid (**7c**), and 3-mercaptopropionic acid (**12**) were involved into the study, and their reactions with hydrazonoyl bromide **9f** ( $X = NO_2$ ) provided solely the expected adducts of type **11** (Scheme 5). Subsequent treatment of compound **11h** with EDC methiodide afforded 6-methylated derivative **8h** in fair overall yield of 55%. In contrast, the attempted cyclisation of **11i** under analogous reaction conditions led mainly to decomposition products formed, very likely, via competitive rearrangement of the first formed adduct [13]. Expected 7-membered 1,3,4-thiadiazepine derivative could not be found in the crude reaction mixture. Similarly, the reaction of thiohydrazonate derivative **11j** with the same coupling agent provided a complex mixture of highly polar components, presumably the corresponding EDC-adducts of **11j**, along with unidentified decomposition products, although in this case, the formation of 6-membered product of type **8** in trace amounts only was observed.



Scheme 5. Trapping of *in situ* generated nitrile imine 1f with 2-mercaptopropionic acid (7b), mercaptosuccinic acid (7c), and 3-mercaptopropionic acid (12), and selected transformations of the initial adducts 11h-11j.

In order to test an alternative base-induced cyclisation approach [3d,11b], the first formed carboxylic acids **11i** and **11j** were converted into their methyl esters **13i** and **13j** in overall 24% and 25% yield, respectively (Scheme 5). Unfortunately, attempted activation of the NH group in compounds **13** with  $K_2CO_3$ , LiOH, MeONa, LDA, and NaH gave no satisfactory result, and either starting materials were recovered or decomposition products were formed depending on the reaction conditions. Apparently, limited stability of **13** in the presence of hydroxide and alkoxide reflect remarkably electrophilic character of esters **13**, enhanced by the presence of the strongly electron-withdrawing CF<sub>3</sub> group, and the observed decomposition is presumably initiated by nucleophilic attack onto the C atom of the thiohydrazonate group. Also, attempted cyclisation of **13i** and **13j** by heating either in solution (refluxing EtOH or toluene) or neat (120 °C) was in vain.

#### **3.** Conclusions

The presented study showed, that *in situ* generated electron-deficient fluorinated nitrile imines **1** can be smoothly trapped by bifunctional HS-nucleophiles, and in the case of mercaptoacetaldehyde (**5**) spontaneous cyclisation of the initial adducts leads to 1,3,4-thiadiazin-5-ols **10** as formal (3+3)-annulation products. The latter compounds could be easily oxidized to 1,3,4-thidiazin-5-ones of type **8** by treatment with PCC. In contrast, the analogous S-adducts derived from  $\alpha$ -mercaptocarboxylic acids **7** cyclized to a desired fluorinated 1,3,4-thidiazin-5-one **8** only upon treatment with the appropriate carbodiimide used as a coupling agent (EDC methiodide). The presented complementary approaches leading to a novel fluorinated 1,3,4-thiadiazines **10** and **8** nicely supplement protocols for the preparation of 1,3,4-thiadiazinones described in the literature [14], and other synthetic methods

towards fluoroalkylated N-heterocycles of practical applications [15]. In addition, the presented study emphasizes the importance of nitrile imines as useful, versatile building blocks for the efficient preparation of not only 5-membered heterocycles by (3+2)-cycloadditions with appropriate dipolarophiles but also larger, 6-membered ring products formed via two-step (3+3)-annulation approach using bifunctional nucleophiles.

#### 4. Experimental Part

#### 4.1. General information

Solvents and reagents were purchased (Sigma-Aldrich, Acros) and used as received without further purification. THF was dried over sodium-benzophenone and freshly distilled before usage. Dichloromethane was dried over CaH<sub>2</sub> prior to use. If not stated otherwise, reactions were carried out under argon in flame-dried flasks with addition of the reactants by using syringes; subsequent manipulations were conducted in air. Products were purified by column chromatography on silica gel (230-400 mesh). Reported yields refer to analytically pure samples. NMR spectra were measured on a Bruker AVIII 600 MHz (<sup>1</sup>H NMR [600 MHz]; <sup>13</sup>C NMR [151 MHz]; <sup>19</sup>F NMR [565 MHz]) or with a Varian Gemini 2000BB 200 MHz (<sup>19</sup>F NMR [188 MHz]) instruments. Chemical shifts are reported relative to solvent residual peaks (<sup>1</sup>H NMR:  $\delta = 7.26$  ppm [CDCl<sub>3</sub>]; <sup>13</sup>C NMR:  $\delta = 77.0$  ppm [CDCl<sub>3</sub>]) or to CFCl<sub>3</sub> ( $\delta = 0.00$  ppm) used as an external standard in <sup>19</sup>F NMR measurements. For detailed peak assignments in <sup>13</sup>C NMR 2D spectra were taken (COSY, HMQC, HMBC). IR spectra were measured with a FTIR NEXUS spectrometer (as thin film or KBr pellets). MS (ESI) were performed with a Varian 500-MS LC Ion Trap. Elemental analyses were obtained with a Vario EL III (Elementar Analysensysteme GmbH) instrument. Melting points were determined in capillaries with a MEL-TEMP II apparatus (Aldrich), and are uncorrected. Hydrazonoyl bromides 9a-9h were obtained by treatment of the corresponding arylhydrazones with N-bromosuccinimide (NBS), in dry DMF, following the literature protocol [5a]. The latter trifluoroacetaldehyde hydrazones were prepared by heating methanolic solutions of the appropriate hydrazine and excess fluoral hydrate in a closed ampoule at 75 °C overnight, in the presence of molecular sieves 4Å according to our earlier report [9].

#### 4.2. Synthesis of 1,3,4-thiadiazin-5-ones 8

*Method A:* To a solution of 5,6-dihydro-1,3,4-thiadiazin-5-ol **10** (1.0 mmol) in dichloromethane (10 mL) was added PCC (323 mg, 1.5 mmol) and the resulting mixture was stirred at room temperature until the starting **10** was fully consumed (TLC monitoring, 2-10 h). The mixture was filtered through a short pad of Celite, and the solvent was removed under reduced pressure. The crude product **8** was purified by flash column chromatography (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>).

*Method B:* To a solution of hydrazonoyl bromide **9** (1.0 mmol) and  $\alpha$ -mercaptocarboxylic acid **7** (1.1 mmol) in dry THF (10 mL) was added dropwise Et<sub>3</sub>N (2 mmol, 0.3 mL), and the resulting mixture was stirred at room temperature overnight. Then, reaction was acidified with aq. HCl (5%) and extracted with Et<sub>2</sub>O (3 × 15 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and the solvents were removed under reduced pressure. The residue was dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL), solid EDC methiodide was added (171 mg, 1.1 mmol) and the resulting mixture was stirred at room temperature overnight. The precipitate was filtered off, the solvents were

removed in vacuo, and the residue was purified by column chromatography (CC) to give 1,3,4-thiadiazin-5-one derivatives **8**.

#### 4.2.1. 4-(p-Tolyl)-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8a)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1), *Method* A: 227 mg (83%), *Method* B: 137 mg (50%), colourless solid, mp 45–47 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.38 (s, 3 H, CH<sub>3</sub>), 3.64 (s, 2 H, 6-H<sub>2</sub>), 7.24, 7.30 (2 d<sub>br</sub>,  $J \approx$  8.4 Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  21.1 (q, CH<sub>3</sub>), 26.4 (t, C-6), 119.2 (q, <sup>1</sup>J<sub>C-F</sub> = 275.0 Hz, CF<sub>3</sub>), 125.1, 129.5 (2 d, 4 CH), 135.9 (q, <sup>2</sup>J<sub>C-F</sub> = 39.5 Hz, C-2), 137.7, 138.1 (2 s), 156.6 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$ -68.1 (s, CF<sub>3</sub>) ppm; IR (KBr):  $\nu$  1694 (C=O), 1514, 1327, 1304, 1213, 1196, 1141, 1123, 1014 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 297.1 (100, [M+Na]<sup>+</sup>), 275.1 (28, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>9</sub>F<sub>3</sub>N<sub>2</sub>OS (274.3): C 48.17, H 3.31, N 10.21, S 11.69; found: C 48.21, H 3.44, N 10.21, S 11.66.

#### 4.2.2. 4-Phenyl-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8b)

CC (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>), *Method A*: 213 mg (82%), colourless oil. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.65 (s, 2 H, 6-H<sub>2</sub>), 7.34-7.37, 7.43-7.46 (2 m, 1 H, 4 H) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  26.4 (t, C-6), 119.1 (q, <sup>1</sup>*J*<sub>C-F</sub> = 275.2 Hz, CF<sub>3</sub>), 125.1, 128.0, 128.9 (3 d, 5 CH), 136.2 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.5 Hz, C-2), 140.0 (s), 156.6 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  –68.1 (s, CF<sub>3</sub>) ppm; IR (KBr):  $\nu$  2930, 1690 (C=O), 1597, 1494, 1327, 1198, 1124, 1016, 691 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 261.2 (100, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>10</sub>H<sub>7</sub>F<sub>3</sub>N<sub>2</sub>OS (260.2): C 46.15, H 2.71, N 10.76, S 12.32; found C 46.41, H 2.87, N 10.88, S 12.21.

#### 4.2.3. 4-(4-Methoxyphenyl)-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8c)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1 gradient CH<sub>2</sub>Cl<sub>2</sub>), *Method B*: 122 mg (42%), colourless solid, mp 85–86 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.63 (s, 2 H, 6-H<sub>2</sub>), 3.83 (s, 3 H, OCH<sub>3</sub>), 6.95, 7.33 (2 d<sub>br</sub>,  $J \approx$  9.1 Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  26.4 (t, C-6), 55.5 (q, OCH<sub>3</sub>), 114.2 (d, 2 CH), 119.2 (q, <sup>1</sup>*J*<sub>C-F</sub> = 275.1 Hz, CF<sub>3</sub>), 126.6 (d, 2 CH), 133.1 (s), 135.8 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.3 Hz, C-2), 156.6 (s, C=O), 159.2 (s, COMe) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$ -68.1 (s, CF<sub>3</sub>) ppm; IR (KBr):  $\nu$  1704 (C=O), 1513, 1333, 1254, 1193, 1143, 1020, 1009 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 313.1 (100, [M+Na]<sup>+</sup>), 291.1 (89, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>9</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub>S (290.3): C 45.52, H 3.13, N 9.65, S 11.05; found: C 45.55, H 3.17, N 9.62, S 11.04.

#### 4.2.4. 4-(4-Fluorophenyl)-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8d)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 2:3), *Method B*: 69 mg (25%), thick colourless oil. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.64 (s, 2 H, 6-H<sub>2</sub>), 7.10-7.14, 7.40-7.43 (2 m, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  26.3 (t, C-6), 115.8 (d, <sup>2</sup>*J*<sub>C-F</sub> = 23.0 Hz, 2 CH), 119.1 (q, <sup>1</sup>*J*<sub>C-F</sub> = 275.2 Hz, CF<sub>3</sub>), 127.0 (d, <sup>3</sup>*J*<sub>C-F</sub> = 8.7 Hz, 2 CH), 136.1 (d, <sup>4</sup>*J*<sub>C-F</sub> = 3.2 Hz, *i*-C), 136.5 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.5 Hz, C-2), 156.7 (s, C=O), 161.8 (d, <sup>1</sup>*J*<sub>C-F</sub> = 248.2 Hz, *i*-C) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$ -68.1 (s, CF<sub>3</sub>), -113.2 (m, Ar-F) ppm; IR (KBr):  $\nu$  1697 (C=O), 1508, 1331, 1238, 1199, 1124, 1020 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 279.2 (100, [M+H]<sup>+</sup>); elemental analysis (%) for C<sub>10</sub>H<sub>6</sub>F<sub>4</sub>N<sub>2</sub>OS (278.2): C 43.17, H 2.17, N 10.07, S 11.52; found: C 43.13, H 2.24, N 10.01, S 11.51.

#### 4.2.5. 4-(4-Bromophenyl)-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8e)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1), *Method A*: 213 mg (63%), *Method B*: 98 mg (29%), colourless solid, mp 54–56 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.64 (s, 2 H, 6-H<sub>2</sub>), 7.35, 7.56 (2 d<sub>br</sub>,  $J \approx 8.8$  Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  26.4 (t, C-6), 119.1 (q, <sup>1</sup>*J*<sub>C-F</sub> = 275.3 Hz, CF<sub>3</sub>), 121.6 (s), 126.5, 132.0 (2 d, 4 CH), 136.8 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.5 Hz, C-2), 139.0 (s), 156.6 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$ -68.1 (s, CF<sub>3</sub>) ppm; IR (KBr): v 1698 (C=O), 1489, 1329, 1200, 1147, 1130, 1011 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 341.0 (97, [M{<sup>81</sup>Br}+Na]<sup>+</sup>), 339.0 (100, [M{<sup>79</sup>Br}+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>10</sub>H<sub>6</sub>BrF<sub>3</sub>N<sub>2</sub>OS (339.1): C 35.42, H 1.78, N 8.26, S 9.45; found: C 35.42, H 1.69, N 8.30, S 9.58.

#### 4.2.6. 4-(4-Nitrophenyl)-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8f)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:3), *Method* A: 279 mg (91%), *Method* B: 220 mg (72%), colourless solid, mp 98–100 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.70 (s, 2 H, 6-H<sub>2</sub>), 7.75, 8.29 (2 d<sub>br</sub>,  $J \approx$  9.2 Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  26.6 (t, C-6), 119.0 (q, <sup>1</sup>J<sub>C-F</sub> = 275.6 Hz, CF<sub>3</sub>), 124.2, 124.7 (2 d, 4 CH), 138.4 (q, <sup>2</sup>J<sub>C-F</sub> = 39.8 Hz, C-2), 144.8, 146.3 (2 s), 156.8 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$ –68.2 (s, CF<sub>3</sub>) ppm; IR (KBr):  $\nu$  1701 (C=O), 1516, 1354, 1333, 1314, 1297, 1200, 1146, 1128 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 306.1 (100, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>10</sub>H<sub>6</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>S (305.2): C 39.35, H 1.98, N 13.77, S 10.50; found: C 39.32, H 2.03, N 13.63, S 10.43.

#### 4.2.7. 4-(5-Oxo-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-4-yl)benzonitrile (8g)

CC (SiO<sub>2</sub>, petroleum ether/EtOAc 4:1), *Method* A: 248 mg (87%), *Method* B: 160 mg (56%), pale yellow solid, mp 111–112 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.68 (s, 2 H, 6-H<sub>2</sub>), 7.67, 7.72 (2 d<sub>br</sub>,  $J \approx 8.1$  Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  26.5 (t, C-6), 111.1, 118.0 (2 s), 118.9 (q, <sup>1</sup>J<sub>C-F</sub> = 275.6 Hz, CF<sub>3</sub>), 124.8, 132.7 (2 d, 4 CH), 138.1 (q, <sup>2</sup>J<sub>C-F</sub> = 39.8 Hz, C-2), 143.4 (s), 156.7 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$  –68.2 (s, CF<sub>3</sub>) ppm; IR (KBr):  $\nu$  2229 (C=N), 1707 (C=O), 1604, 1332, 1303, 1182, 1146, 1134, 1107 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 308.0 (100, [M+Na]<sup>+</sup>), 286.1 (32, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>6</sub>F<sub>3</sub>N<sub>3</sub>OS (285.2): C 46.32, H 2.12, N 14.73, S 11.24; found: C 46.42, H 2.15, N 14.68, S 11.11.

#### 4.2.8. 6-Methyl-4-(4-nitrophenyl)-2-(trifluoromethyl)-4H-1,3,4-thiadiazin-5(6H)-one (8h)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:4), *Method B*: 175 mg (55%), light orange solid, mp 69–71 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  1.62 (d, *J* = 7.1 Hz, 3 H, CH<sub>3</sub>), 3.81 (q, *J* = 7.1 Hz, 1 H, 6-H), 7.73, 8.29 (2 d<sub>br</sub>, *J* ≈ 9.2 Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  15.9 (q, CH<sub>3</sub>), 34.3 (d, C-6), 119.1 (q, <sup>1</sup>*J*<sub>C-F</sub> = 275.3 Hz, CF<sub>3</sub>), 124.2, 124.7 (2 d, 4 CH), 137.5 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.8 Hz, C-2), 145.2, 146.2 (2 s), 160.6 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$  –68.4 (s, CF<sub>3</sub>) ppm; IR (KBr): *v* 1708 (C=O), 1602, 1516, 1353, 1329, 1115 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 342.0 (100, [M+Na]<sup>+</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>8</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>S (319.3): C 41.38, H 2.53, N 13.16, S 10.04; found: C 41.56, H 2.73, N 13.29, S 10.10.

#### 4.3. Synthesis of 1,3,4-thiadiazin-5-ols 10; general procedure

To a solution of the respective hydrazonoyl bromide 9 (1.0 mmol) and 1,4-dithiane-2,5-diol (6, 84 mg, 0.55 mmol) in dry THF (10 mL) was added dropwise Et<sub>3</sub>N (0.7 mL) and the resulting mixture was stirred at ambient temperature until the starting bromide was fully consumed (TLC monitoring). The precipitate

triethylammonium salt was filtered off, and the solvents were removed in vacuo. The resulting material was purified by column chromatography (CC) to afford 1,3,4-thiadiazin-5-ols **10**.

#### 4.3.1. 4-(p-Tolyl)-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-5-ol (10a):

Reaction time 2h; CC (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>), 215 mg (78%), colourless solid, mp 83–84 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.33 (s, 3 H, Me), 2.93 (dd, J = 1.7, 12.9 Hz, 1 H, 6-H), 3.06 (d, J = 10.7 Hz, 1 H, OH), 3.18 (dd, J = 3.4, 12.9 Hz, 1 H, 6-H), 5.64 (ddd, J = 1.7, 3.4, 10.7 Hz, 1 H, 5-H), 7.16, 7.29 (2 d, J = 8.4 Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  20.6 (q, Me), 28.5 (t, C-6), 71.1 (d, C-5), 116.6 (d, 2 CH), 119.7 (q, <sup>1</sup> $_{C-F} = 273.0$  Hz, CF<sub>3</sub>), 120.7 (q, <sup>2</sup> $_{J-F} = 38.9$  Hz, C-2), 129.7 (d, 2 CH), 133.2, 143.0 (2 s) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  –67.6 (s, CF<sub>3</sub>) ppm; IR (KBr): v 3416 (OH), 1578, 1511, 1331, 1263, 1188, 1137, 999, 813 cm<sup>-1</sup>; (-)-ESI-MS (m/z): 275.3 (100, [M–H]<sup>–</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>11</sub>F<sub>3</sub>N<sub>2</sub>OS (276.3): C 47.82, H 4.01, N 10.14, S 11.60; found: C 47.80, H 4.15, N 10.47, S 11.56.

#### 4.3.2. 4-Phenyl-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-5-ol (10b):

Reaction time 2h; CC (petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1 gradient CH<sub>2</sub>Cl<sub>2</sub>), 199 mg (76%), colourless solid, mp 45–47 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.92 (dd, J = 1.7, 12.9 Hz, 1 H, 6-H), 3.16 (dd, J = 3.4, 12.9 Hz, 1 H, 6-H), 3.21 (d, J = 10.4 Hz, 1 H, OH), 5.66 (ddd, J = 1.7, 3.4, 10.4 Hz, 1 H, 5-H), 7.10-7.13, 7.35-7.42 (2 m, 1 H, 4 H) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  28.5 (t, C-6), 70.8 (d, C-5), 116.4 (d, 2 CH), 119.6 (q, <sup>1</sup> $_{J_{C-F}}$  = 273.1 Hz, CF<sub>3</sub>), 121.6 (q, <sup>2</sup> $_{J_{C-F}}$  = 38.9 Hz, C-2), 123.5 (d, CH), 129.2 (d, 2 CH), 145.0 (s) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  -67.7 (s, CF<sub>3</sub>) ppm; IR (KBr):  $\nu$  3408 (OH), 2359, 1599, 1493, 1327, 1263, 1190, 1134, 1014, 758 cm<sup>-1</sup>; ESI-MS (m/z): 263.1 (100, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>10</sub>H<sub>9</sub>F<sub>3</sub>N<sub>2</sub>OS (262.3): C 45.80, H 3.46, N 10.68, S 12.22; found: C 45.71, H 3.52, N 10.64, S 12.07.

#### 4.3.3. 4-(4-Methoxyphenyl)-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-5-ol (10c):

Reaction time 2h; CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1 gradient CH<sub>2</sub>Cl<sub>2</sub>), 187 mg (64%), red oil. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.91 (dd, J = 1.7, 12.8 Hz, 1 H, 6-H), 3.14 (dd, J = 3.4, 12.8 Hz, 1 H, 6-H), 3.26 (s<sub>br</sub>, 1 H, OH), 3.78 (s, 3 H, OMe), 5.55 (dd<sub>br</sub>,  $J \approx 1.7$ , 3.4 Hz, 1 H, 5-H), 6.86-6.89, 7.28-7.31 (2 m, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  = 28.4 (t, C-6), 55.5 (q, OMe), 71.4 (d, C-5), 114.4, 118.5 (2 d, 2 CH each), 119.7 (q, <sup>1</sup>*J*<sub>C-F</sub> = 272.9 Hz, CF<sub>3</sub>), 120.5 (q, <sup>2</sup>*J*<sub>C-F</sub> = 38.7 Hz, C-2), 139.3, 156.2 (2 s) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  –67.6 (s, CF<sub>3</sub>) ppm; IR (film): v 3463 (OH), 2973, 1593, 1575, 1515, 1324, 1247, 1143, 1014, 822 cm<sup>-1</sup>; ESI-MS (m/z): 293.1 (100, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>11</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub>S (292.3): C 45.20, H 3.79, N 9.58, S 10.97; found; C 45.35, H 3.97, N 9.67, S 10.80.

#### 4.3.4. 4-(4-Fluorophenyl)-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-5-ol (10d):

Reaction time 3h; CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1), 185 mg (66%), yellow oil. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.96 (dd, J = 1.7, 12.9 Hz, 1 H, 6-H), 3.08 (s<sub>br</sub>, 1 H, OH), 3.20 (dd, J = 3.4, 12.9 Hz, 1 H, 6-H), 5.61 (dd<sub>br</sub>,  $J \approx 1.7$ , 3.4 Hz, 1 H, 5-H), 7.03-7.07, 7.34-7.37 (2 m, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  28.7 (t, C-6), 71.4 (d, C-5), 115.8 (d, <sup>2</sup>*J*<sub>C-F</sub> = 22.7 Hz, 2 CH), 118.3 (d, <sup>3</sup>*J*<sub>C-F</sub> = 8.0 Hz, 2 CH), 119.6 (q, <sup>1</sup>*J*<sub>C-F</sub> = 273.1 Hz, CF<sub>3</sub>), 121.5 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.0 Hz, C-2), 141.6 (d, <sup>4</sup>*J*<sub>C-F</sub> = 2.5 Hz, *i*-C), 159.4 (d, <sup>1</sup>*J*<sub>C-F</sub> = 242.9 Hz, *i*-C) ppm; <sup>19</sup>F NMR

(CDCl<sub>3</sub>, 565 MHz):  $\delta$  –67.7 (s, CF<sub>3</sub>), –120.0 (m, Ar-F) ppm; IR (film): *v* 3421 (OH), 2362, 1508, 1331, 1267, 1230, 1190, 1134, 1014, 831 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 281.0 (37, [M+H]<sup>+</sup>), 280 (100, M<sup>+</sup>), 263 (28). elemental analysis calcd (%) for C<sub>10</sub>H<sub>8</sub>F<sub>4</sub>N<sub>2</sub>OS (280.2): C 42.86, H 2.88, N 10.00, S 11.44; found: C 43.03, H 2.96, N 10.21, S 11.52.

#### 4.3.5. 4-(4-Bromophenyl)-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-5-ol (10e):

Reaction time 3h; CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1), 239 mg (70%), colourless solid, mp 74–75 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.96 (dd, J = 1.7, 12.9 Hz, 1 H, 6-H), 3.05 (d<sub>br</sub>,  $J \approx 11.1$  Hz, 1 H, OH), 3.22 (dd, J= 3.4, 12.9 Hz, 1 H, 6-H), 5.63 (ddd<sub>br</sub>,  $J \approx 1.7$ , 3.4, 11.1 Hz, 1 H, 5-H), 7.27-7.30, 7.44-7.47 (2 m, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  28.7 (t, C-6), 70.9 (d, C-5), 116.4 (s, CBr), 117.9 (d, 2 CH), 119.5 (q, <sup>1</sup> $J_{C-F}$  = 273.8 Hz, CF<sub>3</sub>), 122.3 (q, <sup>2</sup> $J_{C-F}$  = 39.0 Hz, C-2), 132.1 (d, 2 CH), 144.1 (s) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  -67.8 (s, CF<sub>3</sub>) ppm; IR (KBr): v 3493 (OH), 2924, 2853, 1571, 1493, 1329, 1264, 1139, 1116, 1014, 824 cm<sup>-1</sup>; (-)-ESI-MS (m/z): 339.3 (100, [M{<sup>79</sup>Br}-H]<sup>-</sup>), 341.3 (94, [M{<sup>81</sup>Br}-H]<sup>-</sup>); elemental analysis calcd (%) for C<sub>10</sub>H<sub>8</sub>BrF<sub>3</sub>N<sub>2</sub>OS (341.1): C 35.21, H 2.36, N 8.21, S 9.40; found: C 35.37, H 2.54, N 8.48, S 9.60.

#### 4.3.6. 4-(4-Nitrophenyl)-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-5-ol (10f):

Reaction time 3h; CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> 1:1), 218 mg (71%), brown solid, mp 135–136 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.08 (dd, J = 1.7, 13.1 Hz, 1 H, 6-H), 3.09 (d, J = 11.3 Hz, 1 H, OH), 3.32 (dd, J= 3.4, 13.1 Hz, 1 H, 6-H), 5.79 (ddd, J = 1.7, 3.4, 11.3 Hz, 1 H, 5-H), 7.52, 8.24 (2 d, J = 9.3 Hz, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  29.2 (t, C-6), 70.7 (d, C-5), 115.3 (d, 2 CH), 119.2 (q, <sup>1</sup> $J_{C-F}$  = 274.0 Hz, CF<sub>3</sub>), 125.4 (d, 2 CH), 126.0 (q, <sup>2</sup> $J_{C-F}$  = 39.4 Hz, C-2), 143.2, 149.3 (2 s) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  -68.1 (s, CF<sub>3</sub>) ppm; IR (KBr): v 3428 (OH), 1591 (NO<sub>2</sub>), 1511, 1431, 1343, 1311, 1260, 1195, 1140, 1113, 1014, 823 cm<sup>-1</sup>; ESI-MS (m/z): 308.1 (100, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>10</sub>H<sub>8</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>S (307.2): C 39.09, H 2.62, N 13.68, S 10.43; found: C 39.21, H 2.85, N 13.69, S 10.41.

#### 4.3.7. 4-(5-Hydroxy-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-4-yl)benzonitrile (10g):

Reaction time 3h; CC (SiO<sub>2</sub>, petroleum ether/AcOEt 1:1), 264 mg (92%), colourless solid, mp 155–156 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  3.04 (dd, J = 1.7, 13.1 Hz, 1 H, 6-H), 3.15 (s<sub>br</sub>, 1 H, OH), 3.29 (dd, J = 3.4, 13.1 Hz, 1 H, 6-H), 5.74 (m<sub>c</sub>, 1 H, 5-H), 7.47-7.52, 7.61-7.66 (2 m, 2 H each) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  29.0 (t, C-6), 70.6 (d, C-5), 106.2 (s, CN), 115.9 (d, 2 CH), 118.9 (s), 119.3 (q, <sup>1</sup>*J*<sub>C-F</sub> = 273.9 Hz, CF<sub>3</sub>), 125.2 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.3 Hz, C-2), 133.5 (d, 2 CH), 147.8 (s) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  –68.1 (s, CF<sub>3</sub>) ppm; IR (KBr): v = 3371 (OH), 2234 (CN), 1605, 1506, 1340, 1265, 1122, 1068, 1008, 829 cm<sup>-1</sup>; (-)-ESI-MS (*m*/*z*): 286.0 (100, [M–H]<sup>-</sup>); elemental analysis calcd (%) for C<sub>11</sub>H<sub>8</sub>F<sub>3</sub>N<sub>3</sub>OS (287.3): C 45.99, H 2.81, N 14.63, S 11.16; found: C 46.15, H 3.04, N 14.67, S 11.30.

#### 4.3.8. 4-(5-Hydroxy-2-(trifluoromethyl)-5,6-dihydro-4H-1,3,4-thiadiazin-4-yl)phenyl benzoate (10h):

Reaction time 4h; CC (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>/AcOEt 95:5), 271 mg (71%), colourless solid, mp 131–133 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.97 (dd, J = 1.6, 12.8 Hz, 1 H, 6-H), 3.10 (d, J = 11.0 Hz, 1 H, OH), 3.23 (dd, J = 3.5, 12.8 Hz, 1 H, 6-H), 5.68 (ddd, J = 1.6, 3.5, 11.0 Hz, 1 H, 5-H), 7.19-7.23, 7.44-7.48, 7.49-7.54, 7.62-7.66, 8.19-8.22 (5 m, 3 × 2 H, 1 H, 2 H) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  28.7 (t, C-6), 71.3 (d, C-5), 117.5 (d, 2

CH), 119.6 (q,  ${}^{1}J_{C-F} = 273.1$  Hz, CF<sub>3</sub>), 121.8 (q,  ${}^{2}J_{C-F} = 39.0$  Hz, C-2), 122.4, 128.6 (2 d, 2 CH each), 129.4 (s), 130.2 (d, 2 CH), 133.6 (d, CH), 143.0, 147.0 (2 s), 165.4 (s, C=O) ppm;  ${}^{19}F$  NMR (CDCl<sub>3</sub>, 565 MHz):  $\delta$  –67.7 (s, CF<sub>3</sub>) ppm; IR (KBr): *v* 3416 (OH), 2829, 1702 (C=O), 1452, 1504, 1320, 1265, 1126, 1008, 710 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 405.1 (100, [M+Na]<sup>+</sup>), 383.1 (7, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>17</sub>H<sub>13</sub>F<sub>3</sub>N<sub>2</sub>O<sub>3</sub>S (382.4): C 53.40, H 3.43, N 7.33, S 8.38; found: C 53.45, H 3.56, N 7.42, S 8.38.

#### 4.4. Synthesis of thiohydrazonates 13i and 13j

To a solution of hydrazonoyl bromide **9f** (312 mg, 1.0 mmol) and 3-mercaptopropionic acid (**12**, 117 mg, 1.1 mmol) or mercaptosuccinic acid (**7c**, 165 mg, 1.1 mmol) in dry THF (10 mL) was added dropwise excess Et<sub>3</sub>N (2 mmol, 0.3 mL) at room temperature, and the stirring was continued overnight. The resulting mixture was acidified with aq. HCl (5%), then extracted with  $Et_2O$  (3 × 15 mL), and the combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub>. After the solvents were removed under reduced pressure, the crude carboxylic acid **11i** or **11j** was dissolved in dry MeOH (10 mL), two drops of conc. H<sub>2</sub>SO<sub>4</sub> were added, and the solution was refluxed overnight. The mixture was cooled to room temperature, quenched with NaHCO<sub>3</sub> (5%, 10 mL) and extracted with Et<sub>2</sub>O (3 × 15 mL). Combined organics were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, the solvents were removed in vacuo, and the crude product **13i** or **13j** was purified by chromatography column.

#### 4.4.1. Methyl 3-((2,2,2-trifluoro-1-(2-(4-nitrophenyl)hydrazono)ethyl)sulfanyl)propanoate (13i)

CC (SiO<sub>2</sub>, petroleum ether/CH<sub>2</sub>Cl<sub>2</sub> gradient CH<sub>2</sub>Cl<sub>2</sub>), 86 mg (24%), pale orange solid, mp 87–89 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.68 (t, *J* = 6.1 Hz, 2 H, CH<sub>2</sub>), 3.08 (t, *J* = 6.1 Hz, 2 H, S-CH<sub>2</sub>), 3.82 (s, 3 H, OMe), 7.39, 8.23 (2 d<sub>br</sub>, *J* ≈ 9.1 Hz, 2 H each), 10.12 (s<sub>br</sub>, 1 H, NH) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  27.4, 33.1 (2 t, -CH<sub>2</sub>CH<sub>2</sub>-), 52.5 (q, OMe), 114.0 (d, 2 arom. CH), 120.8 (q, <sup>1</sup>*J*<sub>C-F</sub> = 272.5 Hz, CF<sub>3</sub>), 124.7 (q, <sup>2</sup>*J*<sub>C-F</sub> = 38.6 Hz, *C*-CF<sub>3</sub>), 125.8 (d, 2 arom. CH), 142.6, 147.5 (2 s), 172.8 (s, C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$  –66.4 (s, CF<sub>3</sub>) ppm; ESI-MS (*m*/*z*): 374 (23, [M+Na]<sup>+</sup>), 352.3 (100, [M+H]<sup>+</sup>); elemental analysis calcd (%) for C<sub>12</sub>H<sub>12</sub>F<sub>3</sub>N<sub>3</sub>O<sub>4</sub>S (351.3): C 41.03, H 3.44, N 11.96, S 9.13; found: C 41.19, H 3.51, N 11.87, S 9.31.

#### 4.4.2. Dimethyl 2-((2,2,2-trifluoro-1-(2-(4-nitrophenyl)hydrazono)ethyl)sulfanyl)succinate (13j)

CC (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>), 102 mg (25%), pale yellow solid, mp 113–114 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  2.91 (dd, J = 9.9, 18.3 Hz, 1 H, -CH<sub>2</sub>-), 3.06 (dd, J = 4.3, 18.3 Hz, 1 H, -CH<sub>2</sub>-), 3.77, 3.84 (2 s, 2 × 3 H, 2 OMe), 4.00 (dd, J = 4.3, 9.9 Hz, 1 H, S-CH), 7.43, 8.24 (2 d<sub>br</sub>,  $J \approx 9.1$  Hz, 2 H each), 10.48 (s<sub>br</sub>, 1 H, NH) ppm; <sup>13</sup>C NMR (CDCl<sub>3</sub>, 151 MHz):  $\delta$  35.1 (t, -CH<sub>2</sub>-), 42.3 (d, S-CH), 52.8, 53.2 (2 q, 2 OMe), 114.2 (d, 2 arom. CH), 120.5 (q, <sup>1</sup>*J*<sub>C-F</sub> = 272.6 Hz, CF<sub>3</sub>), 122.4 (q, <sup>2</sup>*J*<sub>C-F</sub> = 39.0 Hz, *C*-CF<sub>3</sub>), 125.8 (d, 2 arom. CH), 142.9, 147.4 (2 s), 170.8, 171.8 (2 s, 2 C=O) ppm; <sup>19</sup>F NMR (CDCl<sub>3</sub>, 188 MHz):  $\delta$ –66.4 (s, CF<sub>3</sub>) ppm; IR (KBr): v 3210 (N-H), 1737 (C=O), 1719 (C=O), 1603, 1552, 1339, 1252, 1148, 1125, 1108, 988 cm<sup>-1</sup>; ESI-MS (*m*/*z*): 432.0 (100, [M+Na]<sup>+</sup>); elemental analysis calcd (%) for C<sub>14</sub>H<sub>14</sub>F<sub>3</sub>N<sub>3</sub>O<sub>6</sub>S (409.3): C 41.08, H 3.45, N 10.27, S 7.83; found: C 41.30, H 3.40, N 10.46, S 7.73.

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#### References

- [1] (a) R. Huisgen, 1,3-Dipolar Cycloadditions. Past and Future, Angew. Chem. Int. Ed. Engl. 2 (1963) 565-598; 1,3-Dipolare Cycloadditionen, Rückschau und Ausblick, Angew. Chem. 75 (1963) 604-637; (b) R. Huisgen, 1,3-Dipolar Cycloadditions - Introduction, Survey, Mechanism, in: 1,3-Dipolar Cycloaddition Chemistry, A. Padwa (Ed.), Wiley-Interscience, New York, 1984, Volume 1, Chapter 1, pp. 1–176.
- [2] (a) J.T. Sharp, Nitrile Ylides and Nitrile Imines, in: Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry Toward Heterocycles and Natural Products, A. Padwa, W.H. Pearson (Eds.), John Wiley & Sons, New York, 2002, Chapter 7, pp. 473-538;

(b) S. Kanemasa, Nitrile Imines, in: Houben-Weyl Science of Synthesis, S.-I. Murahashi (Ed.), Georg Thieme Verlag KG, Stuttgart, 2004, pp. 41–52.

(a) A.F. Hegarty, J.A. Kearney, F.L. Scott, A free-radical analogue of the Chapman rearrangement; [3] Conversion of arylhydrazonates into N',N'-diarylhydrazides, J. Chem. Soc., Perkin Trans 2 (1973) 1422-1430;

(b) A.S. Shawali, B.E. Elenadouli, H.A. Albar, Cycloaddition of diphenylnitrilimine to coumarins. The synthesis of 3a,9b-dihydro-4-oxo-1*H*-benzopyrano[4,3-*c*]pyrazole derivatives, Tetrahedron 41 (1985) 1877-1884;

(c) K. Paulvannan, T. Chen, R. Hale, An improved synthesis of 1,2,4-triazoles using Ag<sub>2</sub>CO<sub>3</sub>, Tetrahedron 56 (2000) 8071-8076;

(d) H.M. Dalloul, E.-H.A.R. Mohamed, A.Z. El-Shorafa, Heterocyclic synthesis using nitrilimines: Part 9. Synthesis of new 1,3,4-thiadiazin-5-one derivatives, Z. Naturforsch. 63b (2008) 585-590.

(a) G. Mlostoń, E. Obijalska, M. Celeda, V. Mittermeier, A. Linden, H. Heimgartner, 1,3-Dipolar cycloadditions of fluorinated nitrones with thioketones, J. Fluorine Chem. 165 (2014) 27-32;

(b) M.K. Kowalski, G. Mlostoń, E. Obijalska, A. Linden, H. Heimgartner, First application of fluorinated nitrones for the synthesis of fluoroalkylated β-lactams via the Kinugasa reaction, Tetrahedron 72 (2016) 5305-5313;

(c) G. Mlostoń, M.K. Kowalski, E. Obijalska, H. Heimgartner, Efficient synthesis of fluoroalkylated 1,4,2-oxathiazoles via regioselective [3+2]-cycloaddition of fluorinated nitrile oxides with thioketones, J. Fluorine Chem. 199 (2017) 92-96.

[4]

[5] (a) G. Mlostoń, K. Urbaniak, G. Utecht, D. Lentz, M. Jasiński, Trifluoromethylated 2,3-dihydro-1,3,4-thiadiazoles via the regioselective [3+2]-cycloadditions of fluorinated nitrile imines with aryl, hetaryl, and ferrocenyl thioketones, J. Fluorine Chem. 192 (2016) 147–154;

(b) G. Utecht, J. Sioma, M. Jasiński, G. Mlostoń, Expected and unexpected results in reactions of fluorinated nitrile imines with (cyclo)aliphatic thioketones, J. Fluorine Chem. 201 (2017) 68–75;

(c) P. Grzelak, G. Utecht, M. Jasiński, G. Mlostoń, First (3+2)-cycloadditions of thiochalcones as C=S dipolarophiles: Efficient synthesis of 1,3,4-thiadiazoles via reactions with fluorinated nitrile imines, Synthesis 49 (2017) 2129–2137.

(a) G. Utecht, A. Fruziński, M. Jasiński, Polysubstituted 3-trifluoromethylpyrazoles: regioselective (3+2)-cycloaddition of trifluoroacetonitrile imines with enol ethers and functional group transformations, Org. Biomol. Chem. 16 (2018) 1252–1257;

(b) G. Utecht, G. Mlostoń, M. Jasiński A straightforward access to trifluoromethylated spirobipyrazolines through a double (3+2)-cycloaddition of fluorinated nitrile imines with alkoxyallenes, Synlett 29 (2018) 1753–1758.

[7] (a) F. Zamberlan, A. Fantinati, C. Trapella, 1,4-Dithiane-2,5-diol: An attractive platform for the synthesis of sulfur-containing functionalized heterocycles, Eur. J. Org. Chem. (2018) 3248–3264;

(b) I.S. Luna, Rayssa M.D. da Cruz, Ryldene M.D. da Cruz, R.S.A. de Araújo, F.J.B. Mendonça-Junior, 1,4-Dithiane-2,5-diol: A versatile synthem for the synthesis of sulfur-containing heterocycles, Curr. Org. Synth. 15 (2018) 1026–1042.

- [8] G. Xiao, L. Shen, C. Qisuan, Y. Yongping, Polysubstituted 1,3,4-thiadiazine compound and preparation method thereof, Patent No. B (2017) CN104844538.
- [9] A. Wojciechowska, M. Jasiński, P. Kaszyński, Tautomeric equilibrium in trifluoroacetaldehyde arylhydrazones, Tetrahedron 71 (2015) 2349–2356.
- [10] (a) T.M. Abdel-Rahman, Synthesis, Reactions, and Anticancer Activity of Some 1,3,4-Thiadiazole/Thiadiazine Derivatives of Carbazole, Phosphorus, Sulfur, Silicon Relat. Elem. 181 (2006) 1737–1754;

(b) N.A. Zigangirova, E.A. Kost, L.V. Didenko, L.N. Kapotina, E.S. Zayakin, S.I. Luyksaar, E.Y. Morgunova, E.D. Fedina, O.A. Artyukhova, A.V. Samorodov, N.V. Kobets, A small-molecule compound belonging to a class of 2,4-disubstituted 1,3,4-thiadiazine-5-ones inhibits intracellular growth and persistence of Chlamydia trachomatis, Journal of Medical Microbiology 65 (2016) 91–98;

(c) R.E. Khidre, B.F. Abdel-Wahab, G.E.A. Awad, Multi-component one-pot synthesis of novel (1,3,4-thiadiazin-2-ylamino)isoindoline-1,3-diones as antimicrobial agents, Heterocycles 94 (2017) 314–325;

(d) I.A.M. Radini, Design, Synthesis, and antimicrobial evaluation of novel pyrazoles and pyrazolyl 1,3,4-thiadiazine derivatives, Molecules 23 (2018) 2092–2103;

(e) H.M. Dalloul, K.A. El-Nwairy, A.Z. Shorafa, A.S.A. Samaha, Synthesis and antimicrobial activities evaluation of some new thiadiazinone and thiadiazepinone derivatives bearing sulfonamide moiety, Phosphorus, Sulfur, Silicon Relat. Elem. 193 (2018) 288–293.

- [11] (a) B.A. Thaher, H.-H. Otto, On the synthesis of 2-acetyl-4-aryl-6*H*-1,3,4-thiadiazin-5-ones by reaction of nitrilimines with α-mercapto alkanoic acids, Monatsh. Chem. 133 (2002) 1011–1016;
  (b) H.M. Dalloul, A.Z. El-Shorafa, On reactions of thiadiazinones: Synthesis of new 6-arylidene-1,3,4-thiadiazin-5-ones, Chem. Heterocycl. Comp. 45 (2009) 735–741;
  (c) N.H. Metwally, F.M. Abdelrazek, A. Jaeger, Facile synthesis of new (*Z*)-6-arylmethylidene-1,3,4-thiadiazin-5(6*H*)-one derivatives, Chem. Heterocycl. Comp. 48 (2013) 1696–1700.
- [12] E. Valeur, M. Bradley, Amide bond formation: beyond the myth of coupling reagents, Chem. Soc. Rev. 38 (2009) 606–631.
- [13] J.A. Zahra, B.A.A. Thaher, M.M. El-Abadelah, R. Boese, A convenient synthesis of some new 1,3,4benzothiadiazepin-5-ones, Heterocycles 63 (2004) 1153–1163.
- (a) W. Reeve, W.R. Coley III, Reactions of phenyl(trichloromethyl)carbinol with substituted thioureas, thiobenzhydrazide, and amino thiols to form heterocyclic compounds, Can. J. Chem. 57 (1979) 444–449;
  (b) Y. Matsubara, S. Yamada, M. Yoshihara, T. Maeshima, Synthesis of 4-(*p*-substituted phenyl)-4,5-dihydro-5-oxo-1,3,4-thiadiazines, Chem. Pharm. Bull. 32 (1984), 1590–1592;

(c) W. Hanefeld, M. Schlitzer, Neuartige Umwandlung von 3-(2-Chlor-2-phenylacetylamino)-rhodanin in 1,3,4-Thiadiazinederivate, Arch. Pharm. (Weinheim) 326 (1993) 249–250;

(d) N.N. Kuz'mich, B.Y. Lalaev, I.P. Yakovlev, T.L. Semakova, V.E. Zakhs, Thiobenzohydrazides and dithiocarbazates in the synthesis of new 1,3,4-thiadiazine and 1,3,4-thiadiazole derivatives, Russ. J. Gen. Chem. 79 (2009) 1583–1584;

(e) H. M. Dalloul, Synthesis of spiroheterocycles containing thiadiazole thiadiazine and triazine moieties from nitrilimines, Phosphorus, Sulfur, Silicon Relat. Elem. 186 (2011) 1876–1884;

(f) A. Kudelko, Reactions of  $\alpha$ -mercaptocarboxylic acid hydrazides with triethyl orthoesters: synthesis of 1,3,4-thiadiazin-5(*6H*)-ones and 1,3,4-oxadiazoles, Tetrahedron 68 (2012) 3616–3625.

[15] (a) P. Kirsch, Modern Fluoroorganic Chemistry: Synthesis, Reactivity, Applications, 2nd, Completely Revised and Enlarged Edition, Wiley-VCH, Weinheim, 2013.

(b) J.-P. Bégué, Bioorganic and Medicinal Chemistry of Fluorine, John Wiley & Sons: Hoboken, NJ, 2008;

(c) X.-G. Hu, L. Hunter, Stereoselectively fluorinated N-heterocycles: a brief survey, Beilstein J. Org. Chem. 9 (2013) 2696–2708;

(d) J. Wang, M. Sánchez-Roselló, J.L. Acena, C. del Pozo, A.E. Sorochinsky, S. Fustero, V.A. Soloshonok, H. Liu, Fluorine in pharmaceutical industry: Fluorine-containing drugs introduced to the market in the last decade (2001–2011), Chem. Rev. 114 (2014) 2432–2506;

(e) A.A. Gakh, Y. Shermolovich, Trifluoromethylated heterocycles, Curr. Top. Med. Chem. 14 (2014) 952–965.