

CHEMISTRY

AN ASIAN JOURNAL

www.chemasianj.org

Accepted Article

Title: Rhodium-catalyzed Oxidative Benzannulation of N-Pivaloylanilines with Internal Alkynes via Dual C-H Bond Activation: Synthesis of Highly Substituted Naphthalenes

Authors: Xuan Zhang; Xiaoqiang Yu; Yoshinori Yamamoto; Abdulrahman I. Almansour; Natarajan Arumugam; Raju Suresh Kumar; Ming Bao

This manuscript has been accepted after peer review and the authors have elected to post their Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: Chem. Asian J. 10.1002/asia.201601131

Link to VoR: <http://dx.doi.org/10.1002/asia.201601131>

A Journal of



A sister journal of *Angewandte Chemie*
and *Chemistry – A European Journal*

WILEY-VCH

Rhodium-catalyzed Oxidative Benzannulation of *N*-Pivaloylanilines with Internal Alkynes via Dual C–H Bond Activation: Synthesis of Highly Substituted Naphthalenes

Xuan Zhang^[a], Xiaoqiang Yu^{*[a]}, Yoshinori Yamamoto^[a,b], Abdulrahman I. Almansour^[c], Natarajan Arumugam^[c], Raju Suresh Kumar^[c], and Ming Bao^{*[a]}

^aState Key Laboratory of Fine Chemicals, Dalian University of Technology, Dalian 116023, China

^bWPI-AIMR (WPI-Advanced Institute for Materials Research), Tohoku University, Sendai 980-8577, Japan

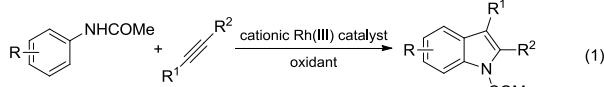
^cDepartment of Chemistry, College of Sciences, King Saud University, P.O. Box 2455, Riyadh 11451, Saudi Arabia

Abstract: An efficient method was developed for the synthesis of highly substituted naphthalenes through rhodium-catalyzed oxidative benzannulation of *N*-pivaloylanilines with internal alkynes. The benzannulation reaction proceeded smoothly via dual C–H bond activation to produce the corresponding highly substituted naphthalene products in satisfactory to good yields.

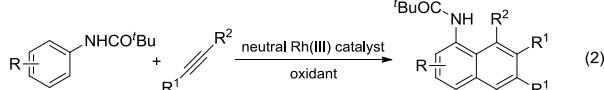
Introduction

The transition-metal-catalyzed direct functionalization of C–H bonds recently emerged as an extremely powerful tool for the synthesis of complex organic molecules.¹ To control site selectivity in molecules that contain diverse C–H bonds, this method required appropriate directing groups such as heterocycles,² aminocarbonyl groups,³ and acylamino groups⁴ linked on the substrates. Among the directing groups employed, acylamino groups are frequently utilized because they can be easily installed into the substrates and converted to other functional groups.⁵ For example, the acylamino groups linked on aromatic rings were successfully used as the directing groups for new carbon–carbon (including *C*_{aryl}–C_{alkenyl}⁶, *C*_{aryl}–C_{aryl}⁷, *C*_{aryl}–C_{alkyl}⁸, and *C*_{aryl}–C_{carbonyl}⁹) and carbon–hetero atom (including *C*_{aryl}–N¹⁰, *C*_{aryl}–O¹¹, *C*_{aryl}–B¹², *C*_{aryl}–Cl¹³, *C*_{aryl}–Br¹⁴, and *C*_{aryl}–I¹⁵) bond formation as well as for the carbocycle and heterocycle construction^{16–18}. It was demonstrated that highly functionalized indoles can be efficiently synthesized through direct C–H bond functionalization in the presence of a cationic rhodium(III) catalyst using acetyl amino group as the directing group (Eq. 1)^{17a,17c,17e}. In the course of our recent studies on the transition-metal-catalyzed direct functionalization of C–H bonds, the regioselectivity in the reaction of *N*-aryl amides with internal alkynes is found to be easily turned by using a neutral rhodium(III) catalyst and a sterically hindered directing group (Eq. 2). The results are reported in the current work.

Previous work



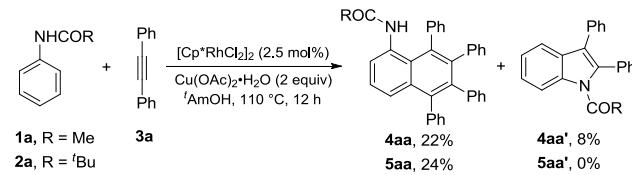
This work



Results and Discussion

In the initial study, the reactions of *N*-phenylacetamide (**1a**) and *N*-phenylpivalamide (**2a**) were performed in the presence of [Cp*RhCl₂]₂ as a precatalyst and Cu(OAc)₂ as an oxidant in *tert*-amyl alcohol (^tAmOH) at 110 °C for 12 h to investigate the effect of the steric hindrance of the directing group on the reaction regioselectivity (Scheme 1). The benzannulation product **4aa** was obtained in 22% yield along with an indole derivative **4aa'** in 8% yield from the reaction of **1a** with 1,2-diphenylethyne (**3a**). These results indicated that the benzannulation reaction could proceed in the presence of a neutral rhodium(III) catalyst. The subsequent investigation showed that the undesired indole forming reaction could be completely inhibited by installing a sterically hindered *tert*-butyl into the *N*-acyl group. The benzannulation product **5aa** was obtained in 24% yield as the sole product from the reaction of **2a** with **3a**. This result motivated us to optimize the reaction conditions for benzannulation using the neutral rhodium(III) catalyst and sterically hindered pivaloylamino group as the directing group.

Scheme 1. Effect of Steric Hindrance of Directing Group on Reaction Regioselectivity.



The benzannulation reaction of **2a** with **3a** was selected as a model to optimize the reaction conditions. Results are shown in Table 1. The use of RhCl₃ and RhCl(PPh₃)₃ as rhodium precatalysts instead of [Cp*RhCl₂]₂ led to no reaction (entry 1 vs. entries 2 and 3). The solvents were then screened using polar [^tAmOH, acetone, 1,4-dioxane, and *N,N*-dimethyl formamide (DMF)] and nonpolar [toluene and 1,2-dichloroethane (DCE)] solvents (entries 1 and 4–8). DMF proved to be the best solvent among the solvents examined (entry 8, 58% yield).¹⁹ The oxidants were finally screened using [Cp*RhCl₂]₂ as a precatalyst and DMF as a solvent. Cu(OAc)₂ proved to be the best oxidant among the tested oxidants Cu(OAc)₂, Cu(OTf)₂, Ag₂CO₃, and AgOAc (entry 8 vs. entries 9–11). The yield of benzannulation product **5aa** was found to increase with increased [Cp*RhCl₂]₂ loading (entry 12, 81%). Therefore, the subsequent benzannulation reactions of *N*-pivaloylanilines **2a**–

2p with internal alkynes **3a–3q** were performed in the presence of $[Cp^*RhCl_2]_2$ as a precatalyst and $Cu(OAc)_2$ as an oxidant in DMF at 110 °C for 12 h.

Table 1. Reaction Condition Screening^a

Entry	Catalyst	Oxidant	Solvent	Yield (%) ^b	2a + 3a → 5aa	
					Ph	NHCOC_2Bu
1	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	$^t\text{AmOH}$	24		
2	$RhCl_3$	$Cu(OAc)_2 \cdot H_2O$	$^t\text{AmOH}$	NR ^c		
3	$RhCl(PPh_3)_3$	$Cu(OAc)_2 \cdot H_2O$	$^t\text{AmOH}$	NR ^c		
4	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	Toluene	NR ^c		
5	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	DCE	23		
6	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	Acetone	40		
7	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	1,4-dioxane	18		
8	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	DMF	58		
9	$[Cp^*RhCl_2]_2$	$Cu(OTf)_2$	DMF	NR ^c		
10	$[Cp^*RhCl_2]_2$	Ag_2CO_3	DMF	NR ^c		
11	$[Cp^*RhCl_2]_2$	$AgOAc$	DMF	3		
12 ^d	$[Cp^*RhCl_2]_2$	$Cu(OAc)_2 \cdot H_2O$	DMF	81		

^aReaction conditions: **2a** (0.25 mmol, 44.3 mg), **3a** (0.5 mmol, 89.1 mg), catalyst (2.5 mol%), oxidant (0.5 mmol, 2 equiv) in solvent (2.0 mL) at 110 °C for 12 h under N_2 atmosphere. ^bIsolated yield. ^cNo reaction was observed; the starting materials were recovered. ^d5 mol% of $[Cp^*RhCl_2]_2$ was used.

The reactions of **2a** with various internal alkynes **3a–3q** were conducted under optimum conditions to explore the scope and limitation of the alkyne substrate, and the results are summarized in Table 2. The reactions of **2a** with diarylalkynes **3b–3f** bearing electron-withdrawing groups (F, Cl, and Br) on *para*- or *meta*-positions of benzene rings proceeded smoothly to provide the corresponding benzannulation products **5ab–5af** in moderate to good yields (entries 2–6, 63%–81% yields). The desired benzannulation product **5ag** was isolated in a satisfactory yield even when the diarylalkyne substrate **3g** bearing a strong electron-withdrawing group (CF_3) on *para*-positions of benzene rings (entry 7, 70% yield). The benzannulation products **5ah**, **5ai**, and **5ak–5am** were obtained in moderate yields (60%–67%) from the reactions of **2a** with diarylalkynes **3h**, **3i**, and **3k–3m** bearing electron-donating groups (Me and MeO) on *para*- or *meta*-positions of benzene rings (entries 8, 9, and 11–13). These abovementioned results indicated that the electronic property of the substituent linked on the benzene ring did not influence the reactivity of diarylalkyne. No reaction was observed when the diarylalkyne substrate, 1,2-di-*o*-tolylethyne (**3j**), was examined (entry 10). The non-reactivity of **3j** was considered to be due to the steric hindrance of the *ortho*-methyl substituent. The heterocycle-containing alkyne substrate, 1,2-di(thiophen-2-yl)ethyne (**3n**), was then tested, and the corresponding benzannulation product **5an** was isolated in 29% yield (entry 14). Surprisingly, when the unsymmetrical alkynes, 1-phenyl-1-propyne (**3o**), 1-phenyl-1-butyne (**3p**), and 1-phenyl-1-pentyne (**3q**), were finally examined under optimum conditions, the benzannulation products **5ao–5aq** were selectively obtained in moderate yields (57%–64%), and the generation of regioisomer was not observed at all. It was reported that no reaction was observed when this type of benzannulation was examined using *N*-acetylanilines and mono

aryl- and alkyl-substituted internal alkynes as starting materials in the presence of palladium catalyst.¹⁶

Table 2. Substrate Scope of Internal Alkynes^a

Entry	Internal Alkyne 3	Product 5	Yield (%) ^b	2a + 3a–3q → 5aa–5aq	
				$R^1 = R^2 = Ph$	$R^1 = R^2 = 4-FC_6H_4$
1	3a	5aa	81		
2	3b	5ab	63		
3	3c	5ac	76		
4	3d	5ad	81		
5	3e	5ae	77		
6	3f	5af	75		
7	3g	5ag	70		
8	3h	5ah	67		
9	3i	5ai	67		
10	3j	5aj	NR ^c		
11	3k	5ak	60		
12	3l	5al	67		
13	3m	5am	62		
14	3n	5an	29 ^d		
15	3o	5ao	57		
16	3p	5ap	64		
17	3q	5aq	60		

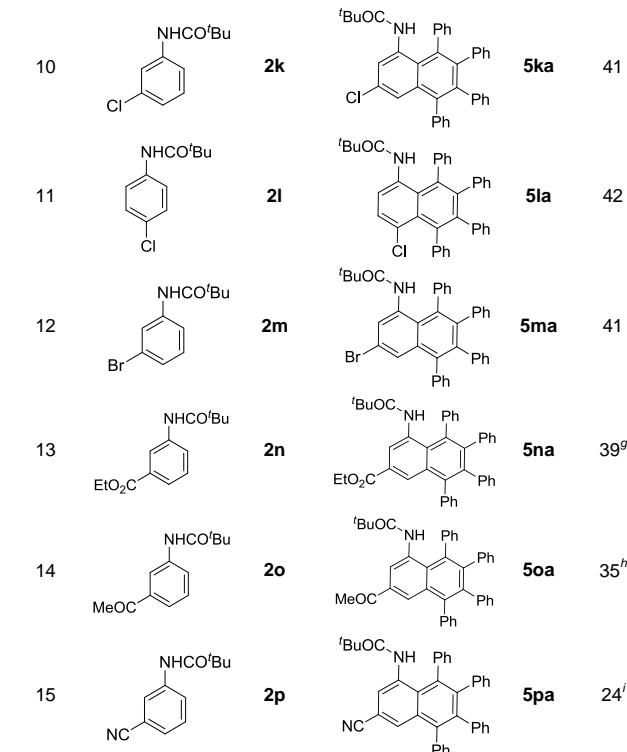
^aReaction conditions: **2a** (0.25 mmol, 44.3 mg), **3** (0.5 mmol, 2 equiv), $[Cp^*RhCl_2]_2$ (5 mol%, 7.8 mg), $Cu(OAc)_2 \cdot H_2O$ (0.50 mmol, 99.8 mg) in DMF (2.0 mL) at 110 °C for 12 h under N_2 atmosphere. ^bIsolated yield. ^cNo reaction was observed, and the starting materials were recovered. ^dThe starting material **2a** was recovered in 62% yield.

A series of *N*-pivaloylanilines were then examined with diphenylacetylene (**3a**) under the optimized reaction conditions to explore the scope and limitation of the *N*-pivaloylaniline substrate. The results are shown in Table 2. *N*-(*o*-tolyl)pivalamide (**2b**) and *N*-(*m*-tolyl)pivalamide (**2c**) presented different reactivity in the benzannulation reaction to give the corresponding products **5ba** and **5ca** in 50% and 79% yields, respectively (entries 1 and 2). These results indicated that the reactivity of the *N*-pivaloylaniline substrate is significantly influenced by the steric effect of the substituent linked on the benzene ring. Surprisingly, when methyl (Me) or normal butyl (^tBu) were substituted at the *para*-positions of *N*-pivaloylaniline substrates, the desired benzannulation reaction could not take place. However, the hydroarylation reaction of diphenylacetylene occurred to furnish byproducts **6a–6c** (entries 3–5), which was considered to be produced from the steric hindrance of *para*-substituents Me and ^tBu. Similarly, the reaction of the *N*-pivaloylaniline substrate **2g** that contains an *ortho*-methoxy group gives the benzannulation product **5ga** in a relatively low yield (60%) as the reaction of **2b** (entry 6). Good yields were observed in the reactions of *N*-pivaloylaniline substrates **2h** and **2i**, even when the methoxy group was linked on the *para*-position (entries 7 and 8). *N*-Pivaloylaniline substrates **2j**, **2k**, and **2l**, which bear a chloro atom on the *ortho*-, *meta*-, or *para*-positions, showed low reactivities in this type of benzannulation reaction to afford the corresponding products **5ja–5la** in low yields (entries 9–11, 25%–42%). Low yields were also observed in the reactions of *N*-pivaloylaniline substrates **2m**, **2n**, **2o**, and **2l** bearing electron-withdrawing groups (Br, CO₂Et, COMe, and

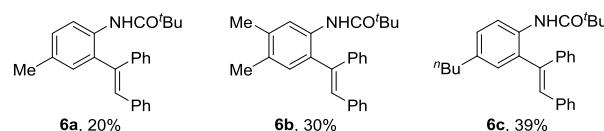
CN) on *meta*-positions (entries 12–15, 24%–41%). These abovementioned results indicated that the electronic property of the substituents strongly influenced the reactivities of *N*-pivaloylaniline substrates. All new products were identified through their NMR and HRMS data, as well as IR spectra. Product **5ao** was further identified by determining its X-ray structure.

Table 3. Substrate Scope of *N*-pivaloylanilines^a

Entry	<i>N</i> -pivaloylaniline 2	Product 5	Yield (%) ^b		
				2b	3a
1			50	2b	3a
2			79	2c	3a
3			0 ^c	2d	3a
4			0 ^d	2e	3a
5			0 ^e	2f	3a
6			60	2g	3a
7			85	2h	3a
8			88	2i	3a
9			25 ^f	2j	3a

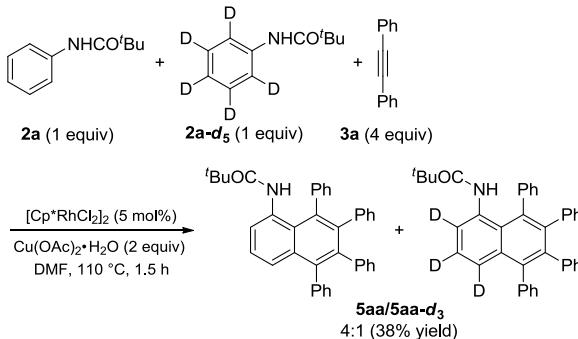
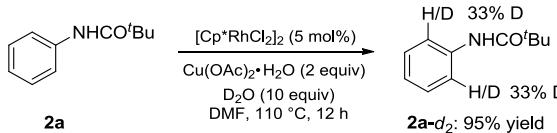


^aReaction conditions: **2** (0.25 mmol), **3a** (0.5 mmol, 89.1 mg), [Cp*RhCl₂] (5 mol%, 7.8 mg), Cu(OAc)₂·H₂O (0.5 mmol, 99.8 mg) in DMF (2.0 mL) at 110 °C for 12 h under N₂ atmosphere. ^bIsolated yield. ^cBy-product **6a** was separated in 20% yield, and the starting material **2d** was recovered in 63% yield. ^dBy-product **6b** was separated in 30% yield, and the starting material **2e** was recovered in 56% yield. ^eBy-product **6c** was separated in 39% yield, and the starting material **2f** was recovered in 49% yield. ^fThe starting material **2j** was recovered in 70% yield. ^gThe starting material **2n** was recovered in 55% yield. ^hThe starting material **2o** was recovered in 60% yield. ⁱThe starting material **2p** was recovered in 70% yield.



To gain insight into the mechanism of this type of benzanulation reaction, a deuterium labeling experiment was conducted to investigate whether the reaction proceeded via the C–H bond activation pathway (Scheme 2). A significant amount of deuterium-labeled **2a–d₂** was observed when the *N*-pivaloylaniline substrate **2a** was treated without alkyne substrate under optimized reaction conditions in the presence of D₂O. This observation reveals that the target reactions indeed proceeded via the C–H bond activation pathway. Further study on the deuterium kinetic isotope effect (DKIE) by conducting an intermolecular competition reaction between **2a** and **2a–d₅** demonstrated that the cleavage of the C–H bond in **2a** is involved in the rate-determining step (Scheme 3).

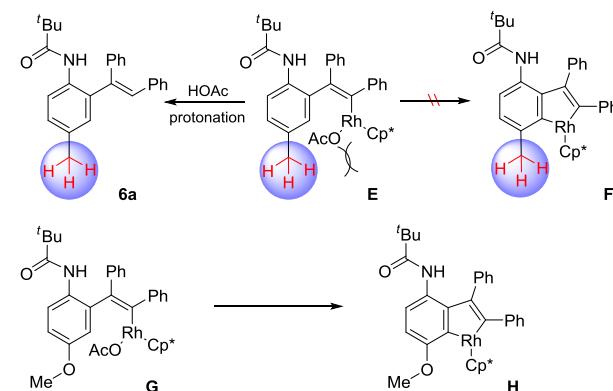
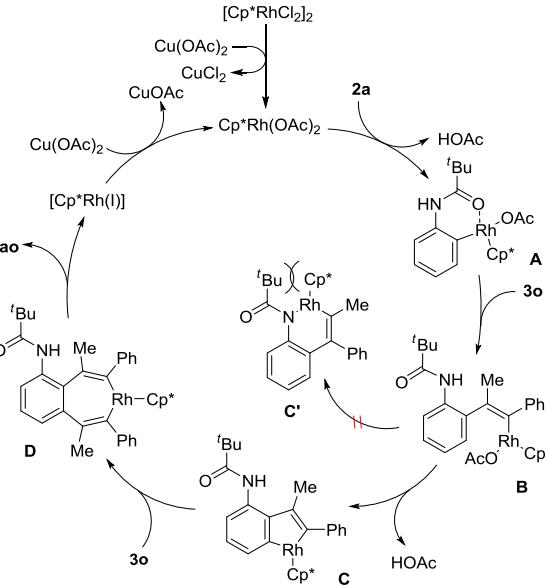
Scheme 2. H–D Exchange Experiment on Amide Substrate **2a** in the Presence of D₂O.



On the basis of our experimental outcomes and previous reports¹⁸, a plausible catalytic cycle is proposed to account for the present catalytic benzannulation reaction (Scheme 4). The catalytic cycle starts from $\text{Cp}^*\text{Rh}(\text{OAc})_2$, which is generated in situ from the ligand exchange reaction between $[\text{Cp}^*\text{RhCl}_2]_2$ and $\text{Cu}(\text{OAc})_2$. The coordination of the oxygen atom of **2a** to rhodium catalyst species and subsequent *ortho* C–H bond activation would generate a six-membered rhodacyclic intermediate **A** with the liberation of an acetic acid molecule. The insertion of **3o** into the Rh–C bond would regioselectively occur to produce intermediate **B**; the regioselectivity might be due to the different electron densities and steric effects of alkyne substituents (Me and Ph)²⁰. The intermediate **B** would subsequently undergo a second C–H bond activation to afford intermediate **C**. A second insertion of alkyne **3o** into the Rh–C bond in the intermediate **C** would also regioselectively occur to produce intermediate **D**, which would subsequently undergo reductive elimination to generate a highly substituted naphthalene product **5ao** and a Rh(I) species. The Rh(I) species would then be re-oxidized to active catalytic species $\text{Cp}^*\text{Rh}(\text{OAc})_2$ by $\text{Cu}(\text{OAc})_2$. The formation of intermediate **C'** is considered to be inhibited by the steric hindrance of the *tert*-butyl group. Consequently, the formation of the benzoindole product was inhibited.

The intermediate **E** derived from *N*-pivaloylaniline substrate **2d** could not proceed cyclorhodation to produce intermediate **F**, which was considered to be due to the strong steric hindrance between Me and Cp^* groups. The protonation reaction of intermediate **E** with HOAc generated in situ would take place to provide by-product **6a**. For the case of the *N*-pivaloylaniline substrate **2i**, the rhodacyclic intermediate **H** would be generated due to the weak steric hindrance caused in intermediate **G**; thereby the desired benzannulation product **5ia** could be obtained in a good yield.

Scheme 4. Proposed Reaction Mechanism.



Conclusions

In summary, the regioselectivity in the reaction of *N*-aryl amides with internal alkynes was successfully controlled by using a neutral rhodium(III) catalyst and a sterically hindered directing group. The rhodium-catalyzed regioselective oxidative benzannulation reaction proceeded smoothly via a dual C–H bond cleavage under mild conditions with affordable and simple starting materials to produce highly-substituted naphthalenes in moderate-to-good yields. This new protocol shows good functional group tolerance and broad substrate scope.

Experimental Section

1.1. General information

All reactions were carried out under a nitrogen atmosphere unless otherwise noted. Solvents were purified by standard techniques without special instructions. ¹H and ¹³C NMR spectra were recorded on a Bruker

For internal use, please do not delete. Submitted_Manuscript

Avance II-400 spectrometer (400 MHz for ^1H , 100 MHz for ^{13}C); CDCl_3 and TMS were used as a solvent and an internal standard, respectively. The chemical shifts are reported in ppm downfield (δ) from TMS, the coupling constants J are given in Hz. The peak patterns are indicated as follows: s, singlet; d, doublet; t, triplet; m, multiplet; q, quartet. IR spectra were recorded on a NEXUS FT-IR spectrometer. High resolution mass spectra were recorded on a GC-TOF mass spectrometry. TLC was carried out on SiO_2 (silica gel 60 F₂₅₄, Merck), and the spots were located with UV light, iodoplatinate reagent or 1% aqueous KMnO_4 . Flash chromatography was carried out on silica gel (SiO_2 60, 200–300 mesh). Melting points were determined using a micro-melting point apparatus and are uncorrected. $[\text{Cp}^*\text{RhCl}_2]_2^{21}$ and $\text{RhCl}(\text{PPh}_3)_3^{22}$ were prepared from $\text{RhCl}_3 \cdot \text{xH}_2\text{O}$ following a literature procedure. Unless otherwise noted below, all other compounds have been reported in the literature or are commercially available without any further purification. The starting materials **2a–2p**²³ and **3b–3n**²⁴ were prepared according to the known procedure.

1.2. Characterization of starting materials **2a–2p**

N-pivaloylaniline (2a)²⁵. White solid (3.3 g, 93% yield); mp: 132–133 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.53 (d, $J = 7.6$ Hz, 2H), 7.36 (bs, 1H), 7.31 (dd, $J = 8.0$, 7.6 Hz, 2H), 7.10 (dd, $J = 7.6$, 7.2 Hz, 1H), 1.32 (s, 9H).

N-(o-tolyl)pivalamide (2b)²⁶. White solid (0.86 g, 90% yield); mp: 110–112 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.86 (d, $J = 8.0$ Hz, 1H), 7.23–7.17 (m, 3H), 7.06 (dd, $J = 7.6$, 7.2 Hz, 1H), 2.26 (s, 3H), 1.34 (s, 9H).

N-(m-tolyl)pivalamide (2c)²⁷. White solid (0.87 g, 91% yield); mp: 124–125 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.43 (s, 1H), 7.29–7.27 (m, 2H), 7.20 (dd, $J = 7.6$, 7.6 Hz, 1H), 6.92 (d, $J = 7.2$ Hz, 1H), 2.33 (s, 3H), 1.31 (s, 9H).

N-(p-tolyl)pivalamide (2d)²⁶. White solid (0.87 g, 91% yield); mp: 119–120 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.40 (d, $J = 8.0$ Hz, 2H), 7.29 (bs, 1H), 7.11 (d, $J = 8.0$ Hz, 2H), 2.31 (s, 3H), 1.30 (s, 9H).

N-(3,4-dimethylphenyl)pivalamide (2e)²⁵. Light yellow solid (0.92 g, 90% yield); mp: 129–130 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.36 (s, 1H), 7.24–7.21 (m, 2H), 7.06 (d, $J = 8.0$ Hz, 1H), 2.24 (s, 3H), 2.21 (s, 3H), 1.30 (s, 9H).

N-(4-butylphenyl)pivalamide (2f). White solid (1.04 g, 89% yield); mp: 72–74 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.43 (d, $J = 6.8$ Hz, 2H), 7.32 (bs, 1H), 7.12 (d, $J = 7.6$ Hz, 2H), 2.57 (t, $J = 7.6$ Hz, 2H), 1.60–1.53 (m, 2H), 1.36–1.30 (m, 11H), 0.91 (t, $J = 7.6$ Hz, 3H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 176.6, 139.0, 135.7, 128.9, 120.1, 39.7, 35.2, 33.8, 27.8, 22.4, 14.1; IR (KBr) ν (cm⁻¹) 3315, 2956, 2928, 1653, 1597, 1519, 1408, 1317, 825; HRMS (ESI) Calcd for $\text{C}_{15}\text{H}_{23}\text{NONa}$ 256.1677 [M+Na]⁺, found 256.1679.

N-(2-methoxyphenyl)pivalamide (2g)²⁷. White solid (0.93 g, 90% yield); mp: 39–40 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.41 (d, $J = 8.0$ Hz, 1H), 8.13 (bs, 1H), 7.02 (dd, $J = 8.0$, 8.0 Hz, 1H), 6.96 (dd, $J = 8.0$, 8.0 Hz, 1H), 6.87 (d, $J = 8.0$ Hz, 1H), 3.90 (s, 3H), 1.32 (s, 9H).

N-(3-methoxyphenyl)pivalamide (2h)²⁷. White solid (0.93 g, 90% yield); mp: 130–131 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.39 (s, 1H), 7.35 (bs, 1H), 7.20 (dd, $J = 8.0$, 8.0 Hz, 1H), 6.94 (d, $J = 7.2$ Hz, 1H), 6.66 (d, $J = 8.4$ Hz, 1H), 3.80 (s, 3H), 1.32 (s, 9H).

N-(4-methoxyphenyl)pivalamide (2i). White solid (0.95 g, 92% yield); mp: 102–103 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.41 (d, $J = 8.8$ Hz, 2H), 7.36 (bs, 1H), 6.82 (d, $J = 8.8$ Hz, 2H), 3.77 (s, 3H), 1.28 (s, 9H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 176.6, 156.4, 131.2, 122.1, 114.1, 55.6, 39.5,

27.7; IR (KBr) ν (cm⁻¹) 3306, 2967, 1647, 1601, 1536, 1513, 1478, 1412, 1314, 1233, 1170, 1035, 828; HRMS (ESI) Calcd for $\text{C}_{12}\text{H}_{17}\text{NO}_2\text{Na}$ 230.1157 [M+Na]⁺, found 230.1166.

N-(2-chlorophenyl)pivalamide (2j)²⁸. White solid (0.88 g, 83% yield); mp: 75–76 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.42 (d, $J = 8.4$ Hz, 1H), 8.02 (bs, 1H), 7.36 (d, $J = 8.0$ Hz, 1H), 7.27 (dd, $J = 8.0$, 7.6 Hz, 1H), 7.03 (dd, $J = 7.6$, 7.6 Hz, 1H), 1.35 (s, 9H).

N-(3-chlorophenyl)pivalamide (2k)²⁶. White solid (0.90 g, 85% yield); mp: 124–125 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.68 (s, 1H), 7.36 (d, $J = 8.0$ Hz, 2H), 7.23 (dd, $J = 8.0$, 8.0 Hz, 1H), 7.27 (dd, $J = 8.0$, 7.6 Hz, 1H), 7.07 (d, $J = 8.0$ Hz, 1H), 1.31 (s, 9H).

N-(4-chlorophenyl)pivalamide (2l)²⁹. White solid (0.90 g, 85% yield); mp: 152–153 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.48 (d, $J = 8.8$ Hz, 2H), 7.33 (bs, 1H), 7.27 (d, $J = 8.4$ Hz, 2H), 1.31 (s, 9H).

N-(3-bromophenyl)pivalamide (2m)²⁷. White solid (1.08 g, 84% yield); mp: 137–138 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.81 (dd, $J = 2.0$, 1.6 Hz, 1H), 7.42 (d, $J = 8.0$ Hz, 1H), 7.36 (bs, 1H), 7.22 (d, $J = 8.0$ Hz, 1H), 7.16 (dd, $J = 8.0$, 8.0 Hz, 1H), 1.31 (s, 9H).

Ethyl 3-pivalamidobenzoate (2n)²⁶. White solid (1.01 g, 81% yield); mp: 59–60 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.01 (s, 1H), 7.95 (d, $J = 8.0$ Hz, 1H), 7.77 (d, $J = 8.0$ Hz, 1H), 7.55 (bs, 1H), 7.39 (dd, $J = 8.0$, 8.0 Hz, 1H), 4.37 (q, $J = 7.2$ Hz, 2H), 1.39 (t, $J = 7.2$ Hz, 3H), 1.33 (s, 9H).

N-(3-acetylphenyl)pivalamide (2o)²⁹. White solid (0.88 g, 80% yield); mp: 124–126 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.05 (s, 1H), 7.90 (d, $J = 7.6$ Hz, 1H), 7.69 (d, $J = 7.6$ Hz, 1H), 7.48 (bs, 1H), 7.43 (dd, $J = 8.0$, 7.6 Hz, 1H), 2.62 (s, 3H), 1.34 (s, 9H).

N-(3-cyanophenyl)pivalamide (2p). White solid (0.84 g, 83% yield); mp: 91–92 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.98 (s, 1H), 7.74 (d, $J = 8.0$ Hz, 1H), 7.57 (bs, 1H), 7.43–7.36 (m, 2H), 1.33 (s, 9H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 177.2, 139.1, 129.9, 127.7, 124.4, 123.3, 118.7, 113.0, 39.9, 27.6; IR (KBr) ν (cm⁻¹) 3356, 2969, 2231, 1665, 1587, 1535, 1480, 1423, 1167, 791, 682; HRMS (ESI) Calcd for $\text{C}_{12}\text{H}_{14}\text{N}_2\text{O}_2\text{Na}$ 225.1004 [M+Na]⁺, found 225.1001.

1.3. General Procedure for Synthesis of highly substituted naphthalenes **5** from *N*-pivaloylanilines **2a–2p** and internal alkynes **3a–3q**

A mixture of *N*-pivaloylaniline (**2**, 0.25 mmol, 1 equiv.), internal alkyne (**3**, 0.50 mmol, 2.0 equiv.), $[\text{Cp}^*\text{RhCl}_2]$ (3.9 mg, 0.00625 mmol, 2.5 mol%), and $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ (99.8 mg, 0.50 mmol, 2.0 equiv.) were charged in a Schlenk tube equipped with a stir bar. Dry DMF (2.0 mL) was added and the mixture was stirred at 110 °C for 12 h under N_2 atmosphere. The mixture was then washed with H_2O (3 × 20 mL) and extracted with CHCl_3 (3 × 20 mL). The combined organic phase were dried with Na_2SO_4 . All volatiles were evaporated under reduced pressure. The purification was performed by flash column chromatography on silica gel (eluent: petroleum ether : ethyl acetate = 10 : 1) to afford the corresponding highly substituted naphthalene product.

1.4. Characterization of highly substituted naphthalenes **4**, **5** and by-products **6**

N-(5,6,7,8-tetraphenylnaphthalen-1-yl)acetamide (4aa)¹⁶.

White solid (27.0 mg, 22% yield); mp: 146–148 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.95 (d, $J = 5.9$ Hz, 1H), 7.50 (d, $J = 6.8$ Hz, 1H), 7.40 (dd, $J =$

For internal use, please do not delete. Submitted_Manuscript

6.2, 6.7 Hz, 1H), 7.26–7.16 (m, 10H), 6.99 (bs, 1H), 6.85–6.73 (m, 10H), 1.40(s, 3H).

N-(2,3-diphenyl-1H-indol-1-yl)ethanone (4aa)^{17a}.

White solid (6.3 mg, 8% yield); mp: 139–141 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.44 (d, J = 8.4 Hz, 1H), 7.54 (d, J = 7.8 Hz, 1H), 7.41–7.37 (m, 1H), 7.35–7.19(m, 11H), 1.99 (s, 3H).

N-(5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5aa).

White solid (107.7 mg, 81% yield); mp: 159–160 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.88 (d, J = 7.6 Hz, 1H), 7.49 (d, J = 8.2 Hz, 1H), 7.40 (dd, J = 8.2, 7.6 Hz, 1H), 7.24–7.14 (m, 11H), 6.82–6.74 (m, 8H), 6.64–6.62 (m, 2H), 0.80 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.0, 142.3, 141.3, 140.5, 140.3, 139.9, 139.5, 139.0, 134.4, 134.2, 133.6, 131.3, 131.1, 128.5, 127.7, 127.2, 126.7, 126.5, 125.8, 125.6, 125.5, 125.3, 124.9, 39.4, 27.0; IR (KBr) ν (cm⁻¹) 3425, 3337, 3050, 3024, 1670, 1601, 1493, 1475, 1441, 1382, 1157, 910, 778, 736, 730, 698; HRMS (ESI) m/z Calcd for C₃₉H₂₉NONaCl₄ 690.0901 [M+Na]⁺, found 690.0915.

N-(5,6,7,8-tetrakis(4-fluorophenyl)naphthalen-1-yl)pivalamide (5ab).

White solid (95.1 mg, 63% yield); mp: 182–184 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.82 (d, J = 6.2 Hz, 1H), 7.48–7.41(m, 2H), 7.14–7.05(m, 5H), 6.96–6.89 (m, 4H), 6.68–6.65 (m, 2H), 6.59–6.56 (m, 6H), 0.87 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 176.9, 161.8 (d, ¹J_{C-F} = 246.7 Hz), 161.6 (d, ¹J_{C-F} = 244.7 Hz), 160.7 (d, ¹J_{C-F} = 243.9 Hz), 160.6 (d, ¹J_{C-F} = 244.2 Hz), 140.4, 138.9, 138.2, 137.80 (d, ⁴J_{C-F} = 3.0 Hz), 135.97 (d, ⁴J_{C-F} = 2.9 Hz), 135.81 (⁴J_{C-F} = 2.6 Hz), 135.29 (⁴J_{C-F} = 3.0 Hz), 134.1 (²J_{C-F} = 20.5 Hz), 132.53 (³J_{C-F} = 7.8 Hz), 132.23 (³J_{C-F} = 7.6 Hz), 126.12 (²J_{C-F} = 19.1 Hz), 125.7, 125.66, 125.64, 125.56, 115.35(d, ²J_{C-F} = 21.0 Hz), 114.84 (d, ²J_{C-F} = 21.1 Hz), 114.1, 114.0, 113.9, 113.8, 39.2, 26.9; IR (KBr) ν (cm⁻¹) 3444, 3321, 2956, 2917, 2848, 1659, 1604, 1509, 1478, 1380, 1224, 1157, 1093, 1015, 909, 830, 816, 777, 758, 734; HRMS (ESI) m/z Calcd for C₃₉H₂₉NONaF₄ 626.2083 [M+Na]⁺, found 626.2088.

N-(5,6,7,8-tetrakis(4-chlorophenyl)naphthalen-1-yl)pivalamide (5ac).

White solid (127.2 mg, 76% yield); mp: 268–270 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.76 (dd, J = 5.0, 3.7 Hz, 1H), 7.44–7.43 (m, 2H), 7.26–7.18 (m, 4H), 7.07 (dd, J = 10.0, 8.4 Hz, 4H), 6.92 (s, 1H), 6.89–6.86 (m, 4H), 6.64 (d, J = 8.4 Hz, 2H), 6.55 (d, J = 8.4 Hz, 2H), 0.88 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.0, 140.1, 139.7, 138.7, 138.2, 138.0, 137.7, 137.6, 134.1, 134.0, 133.5, 133.4, 133.0, 132.3, 132.2, 132.0, 131.9, 128.6, 128.2, 127.4, 127.3, 126.5, 126.33, 126.31, 125.7, 39.3, 26.9; IR (KBr) ν (cm⁻¹) 3439, 3326, 2960, 2921, 2865, 1653, 1492, 1395, 1378, 1092, 1015, 908, 837, 771, 758, 734; HRMS (ESI) m/z Calcd for C₃₉H₂₉NONaCl₄ 690.0901 [M+Na]⁺, found 690.0890.

N-(5,6,7,8-tetrakis(4-bromophenyl)naphthalen-1-yl)pivalamide (5ad).

White solid (163.1 mg, 77% yield); mp: 272–274 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.75 (dd, J = 3.6, 3.6Hz, 1H), 7.43 (d, J = 3.6Hz, 2H), 7.39 (d, J = 6.4Hz, 2H), 7.34 (d, J = 6.8Hz, 2H), 7.06–6.98 (m, 8H), 6.89 (s, 1H), 6.59 (d, J = 6.4Hz, 2H), 6.49 (d, J = 6.4Hz, 2H), 0.88 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.1, 140.6, 139.5, 138.7, 138.6, 138.5, 138.2, 137.4, 134.1, 134.0, 133.6, 132.6, 132.4, 132.3, 131.5, 131.2, 130.4, 130.3, 126.6, 126.5, 126.3, 125.8, 121.6, 121.3, 120.4, 120.2, 39.3, 26.9; IR (KBr) ν (cm⁻¹) 3438, 3329, 2956, 2924, 2854, 1654, 1489, 1391, 1101, 1071, 1011, 908, 827, 774, 764, 732; HRMS (ESI) m/z Calcd for C₃₉H₂₉NONaBr₄ 865.8880 [M+Na]⁺, found 865.8846.

N-(5,6,7,8-tetrakis(3-chlorophenyl)naphthalen-1-yl)pivalamide (5ae).

White solid (135.6 mg, 81% yield); mp: 120–122 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.81 (dd, J = 3.9, 3.8 Hz, 1H), 7.48–7.46 (m, 2H), 7.22–7.02 (m, 9H), 6.89–6.47 (m, 8H), 0.87 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.2, 143.6, 141.3, 141.2, 140.9, 139.6, 138.7, 137.3, 134.8, 134.6, 134.0,

133.9, 133.7, 133.30, 133.27, 133.2, 133.13, 133.10, 133.03, 132.96, 132.93, 132.88, 132.8, 131.1, 131.0, 130.84, 130.78, 129.72, 129.68, 129.3, 129.2, 129.1, 128.9, 128.8, 128.7, 128.6, 128.4, 128.3, 127.7, 127.5, 126.8, 126.5, 126.3, 126.1, 125.9, 39.4, 27.1; IR (KBr) ν (cm⁻¹) 3441, 3318, 2963, 1656, 1594, 1564, 1479, 1407, 1373, 1091, 1077, 908, 884, 778, 758, 732, 699; HRMS (ESI) m/z Calcd for C₃₉H₂₉NONaCl₄ 690.0901 [M+Na]⁺, found 690.0915.

N-(5,6,7,8-tetrakis(3-bromophenyl)naphthalen-1-yl)pivalamide (5af).

White solid (158.9 mg, 75% yield); mp: 117–119 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.82 (dd, J = 4.4, 4.3 Hz, 1H), 7.48 (d, J = 4.6 Hz, 2H), 7.40–7.31 (m, 4H), 7.20–6.49 (m, 13H), 0.87 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.1, 143.8, 141.4, 141.3, 141.1, 139.5, 138.5, 137.2, 134.1, 133.9, 133.81, 133.78, 133.6, 133.5, 130.51, 130.47, 130.3, 129.90, 129.85, 129.7, 129.5, 129.3, 129.2, 129.1, 128.8, 128.7, 128.5, 128.4, 126.8, 126.4, 125.9, 125.8, 122.8, 122.7, 122.18, 122.15, 122.10, 122.05, 122.01, 121.96, 121.38, 121.35, 121.2, 121.1, 120.99, 120.96, 39.3, 27.0; IR (KBr) ν (cm⁻¹) 3440, 3321, 3060, 2958, 2925, 2868, 1661, 1592, 1559, 1477, 1403, 1374, 1215, 1161, 1070, 996, 884, 817, 804, 756, 721, 695; HRMS (ESI) m/z Calcd for C₃₉H₂₉NO₂Br₄ 865.8880 [M+Na]⁺, found 865.8918.

N-(5,6,7,8-tetrakis(4-(trifluoromethyl)phenyl)naphthalen-1-yl)pivalamide (5ag).

White solid (140.6 mg, 70% yield); mp: 271–273 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.73 (d, J = 7.2 Hz, 1H), 7.54–7.44 (m, 6H), 7.29 (d, J = 7.9 Hz, 4H), 7.16–7.10 (m, 4H), 6.87 (d, J = 7.8 Hz, 2H), 6.83 (s, 1H), 6.75 (d, J = 7.8 Hz, 2H), 0.81 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.3, 145.5, 143.1, 142.7, 139.4, 138.9, 137.2, 134.4, 134.0, 133.7, 131.4, 131.3, 131.1, 131.0, 129.9–127.9 (m), 127.5, 127.2, 126.8, 126.1, 125.5, 125.2, 125.1, 125.1, 124.3 (q, J_{C-F} = 3.0 Hz), 124.0 (q, J_{C-F} = 3.1 Hz), 122.7, 122.5, 39.3, 26.9; IR (KBr) ν (cm⁻¹) 3357, 2955, 2918, 2849, 1654, 1616, 1457, 1324, 1165, 1122, 1105, 1065, 1019, 846, 756; HRMS (ESI) m/z Calcd for C₄₃H₂₉NONaF₁₂ 826.1955 [M+Na]⁺, found 826.1960.

N-(5,6,7,8-tetra-p-tolyl)naphthalen-1-yl)pivalamide (5ah).

White solid (98.5 mg, 67% yield); mp: 194–196 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.88 (d, J = 6.7 Hz, 1H), 7.44 (d, J = 8.4 Hz, 1H), 7.34 (dd, J = 8.2, 7.8 Hz, 1H), 7.06–6.96 (m, 9H), 6.62–6.59 (m, 6H), 6.49 (d, J = 7.9 Hz, 2H), 2.29 (s, 3H), 2.26 (s, 3H), 2.09 (s, 3H), 2.07 (s, 3H), 0.81 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.0, 141.5, 139.4, 139.20, 139.17, 137.7, 137.4, 137.1, 136.6, 135.8, 134.44, 134.37, 134.30, 134.25, 133.5, 131.2, 131.1, 131.0, 129.1, 128.3, 127.3, 127.2, 125.43, 125.38, 124.2, 39.3, 26.9, 21.4, 21.3, 21.2; IR (KBr) ν (cm⁻¹) 3421, 3022, 2956, 2923, 2868, 1670, 1510, 1466, 1378, 1157, 1110, 1021, 817, 756, 744; HRMS (ESI) m/z Calcd for C₄₃H₄₁NONa 610.3086 [M+Na]⁺, found 610.3094.

N-(5,6,7,8-tetra-m-tolyl)naphthalen-1-yl)pivalamide (5ai).

White solid (98.5 mg, 67% yield); mp: 92–94 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.93 (d, J = 7.5 Hz, 1H), 7.49 (d, J = 8.4 Hz, 1H), 7.41 (s, 1H), 7.38 (dd, J = 8.1, 8.0 Hz, 1H), 7.10–6.92 (m, 8H), 6.72–6.38 (m, 8H), 2.25–2.14 (m, 6H), 2.01–1.94 (m, 6H), 0.80 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 176.9, 142.2, 142.1, 141.5, 141.4, 140.3, 140.2, 139.9, 139.4, 139.33, 139.29, 139.0, 137.9, 137.8, 136.87, 136.85, 136.8, 135.7, 135.6, 135.49, 135.45, 135.4, 134.19, 134.15, 134.1, 133.5, 132.2, 132.1, 131.9, 128.4, 128.3, 128.24, 128.17, 128.0, 127.81, 127.75, 127.4, 127.3, 127.2, 126.3, 126.2, 126.1, 126.0, 125.7, 125.5, 125.4, 125.0, 124.1, 39.4, 26.9, 21.5, 21.2, 21.1; IR (KBr) ν (cm⁻¹) 3420, 3345, 3015, 2957, 2921, 2867, 1672, 1603, 1584, 1509, 1487, 1378, 1157, 1090, 779, 755, 733, 700; HRMS (ESI) m/z Calcd for C₄₃H₄₁NONa 610.3086 [M+Na]⁺, found 610.3088.

For internal use, please do not delete. Submitted_Manuscript

N-(5,6,7,8-tetrakis(4-methoxyphenyl)naphthalen-1-yl)pivalamide (5ak).

White solid (97.8 mg, 60% yield); mp: 102–104 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.92 (d, $J = 7.3$ Hz, 1H), 7.47 (d, $J = 8.4$ Hz, 1H), 7.36 (dd, $J = 8.2$, 7.9 Hz, 1H), 7.32 (s, 1H), 7.08–7.03 (m, 4H), 6.78–6.72 (m, 4H), 6.62 (d, $J = 8.6$ Hz, 2H), 6.51 (d, $J = 8.6$ Hz, 2H), 6.40–6.37 (m, 4H), 3.78 (s, 3H), 3.75 (s, 3H), 3.63 (s, 3H), 3.61 (s, 3H), 0.84 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.8, 158.5, 158.0, 156.9, 141.5, 139.2, 134.5, 134.3, 134.1, 133.5, 133.2, 132.9, 132.4, 132.3, 132.12, 132.05, 132.0, 125.4, 125.2, 123.9, 113.9, 113.1, 112.1, 55.4, 55.2, 55.0, 54.9, 39.3, 27.0; IR (KBr) ν (cm $^{-1}$) 3416, 2956, 2835, 1675, 1609, 1511, 1463, 1285, 1244, 1176, 1107, 1033, 830, 801, 755, 733; HRMS (ESI) m/z Calcd for $\text{C}_{43}\text{H}_{41}\text{NO}_5\text{Na}$ 674.2882 [M+Na] $^+$, found 674.2897.

N-(5,6,7,8-tetrakis(3-methoxyphenyl)naphthalen-1-yl)pivalamide (5al).

White solid (109.2 mg, 67% yield); mp: 101–103 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.85 (d, $J = 7.5$ Hz, 1H), 7.57–7.54 (m, 1H), 7.41 (dd, $J = 8.1$, 7.9 Hz, 1H), 7.32 (s, 1H), 7.19–7.10 (m, 2H), 6.86–6.69 (m, 8H), 6.41–6.18 (m, 6H), 3.69–3.66 (m, 3H), 3.62–3.59 (m, 3H), 3.50–3.46 (m, 6H), 0.84 (s, 9H); ^{13}C NMR (400 MHz, CDCl_3) δ 177.0, 159.51, 159.45, 158.94, 158.89, 158.12, 158.06, 143.4, 141.6, 141.5, 141.1, 140.7, 140.61, 140.57, 139.2, 138.3, 134.2, 134.0, 133.5, 129.5, 129.4, 128.63, 128.57, 127.6, 125.8, 125.6, 125.5, 125.1, 124.0, 123.9, 123.3, 116.5, 116.4, 116.1, 114.14, 114.08, 112.7, 112.0, 55.31, 55.26, 55.19, 55.15, 39.4, 27.0; IR (KBr) ν (cm $^{-1}$) 3420, 3000, 2956, 2834, 1672, 1600, 1578, 1486, 1465, 1427, 1376, 1284, 1246, 1214, 1159, 1045, 877, 756, 740, 697; HRMS (ESI) m/z Calcd for $\text{C}_{43}\text{H}_{41}\text{NO}_5\text{Na}$ 674.2882 [M+Na] $^+$, found 674.2870.

N-(5,6,7,8-tetrakis(3,5-dimethylphenyl)naphthalen-1-yl)pivalamide (5am).

White solid (99.8 mg, 62% yield); mp: 206–208 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.99 (d, $J = 6.9$ Hz, 1H), 7.62 (s, 1H), 7.48 (d, $J = 7.6$ Hz, 1H), 7.35 (dd, $J = 8.2$, 7.8 Hz, 1H), 6.80–6.77 (m, 5H), 6.72 (s, 1H), 6.39 (d, $J = 11.5$ Hz, 4H), 6.25 (s, 2H), 2.19 (s, 6H), 2.13 (s, 6H), 1.95 (s, 6H), 1.94 (s, 6H), 0.80 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.8, 141.9, 141.6, 140.1, 139.8, 139.1, 138.9, 137.5, 136.4, 135.01, 134.96, 134.9, 133.9, 133.4, 129.0, 128.42, 128.35, 127.84, 127.75, 126.6, 126.5, 126.2, 125.1, 124.4, 123.1, 39.3, 26.7, 21.3, 21.2, 21.0; IR (KBr) ν (cm $^{-1}$) 3415, 3002, 2956, 2917, 2864, 1673, 1601, 1513, 1479, 1388, 1375, 1355, 1260, 1205, 1157, 907, 843, 773, 755, 735, 704; HRMS (ESI) m/z Calcd for $\text{C}_{47}\text{H}_{49}\text{NONa}$ 666.3712 [M+Na] $^+$, found 666.3729.

N-(5,6,7,8-tetra(thiophen-2-yl)naphthalen-1-yl)pivalamide (5an).

Yellowish solid (40.3 mg, 29% yield); mp: 150–152 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.97 (d, $J = 7.5$ Hz, 1H), 7.71 (d, $J = 8.4$ Hz, 1H), 7.59 (s, 1H), 7.48 (dd, $J = 8.1$, 8.0 Hz, 1H), 7.34–7.31 (m, 2H), 7.05–6.91 (m, 6H), 6.66–6.63 (m, 2H), 6.54 (d, $J = 3.0$ Hz, 1H), 6.48 (d, $J = 3.0$ Hz, 1H), 0.97 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 177.2, 142.6, 140.63, 140.56, 139.7, 136.4, 135.3, 135.1, 134.5, 133.6, 130.1, 129.9, 129.7, 129.4, 129.30, 129.26, 129.0, 127.8, 127.4, 126.99, 126.96, 126.7, 126.5, 126.1, 125.9, 125.6, 125.4, 124.3, 120.1, 39.6, 27.2; IR (KBr) ν (cm $^{-1}$) 3416, 3343, 2960, 2926, 2868, 1669, 1501, 1473, 1368, 1224, 1158, 909, 850, 834, 756, 732, 693; HRMS (ESI) m/z Calcd for $\text{C}_{31}\text{H}_{25}\text{NONaS}_4$ 578.0717 [M+Na] $^+$, found 578.0729.

N-(5,8-dimethyl-6,7-diphenyl)naphthalen-1-yl)pivalamide (5ao).

White solid (58.1 mg, 57% yield); mp: 198–200 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.04 (d, $J = 8.4$ Hz, 1H), 7.73 (s, 1H), 7.69 (d, $J = 7.4$ Hz, 1H), 7.52 (dd, $J = 8.1$, 7.8 Hz, 1H), 7.14–7.04 (m, 6H), 6.96–6.93 (m, 4H), 2.52 (s, 3H), 2.42 (s, 3H), 1.34 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.8, 142.0, 141.9, 141.7, 139.6, 134.4, 134.2, 130.5, 130.2, 129.1,

129.0, 127.6, 127.4, 126.3, 126.1, 126.0, 125.5, 124.1, 39.5, 27.7, 22.2, 17.9; IR (KBr) ν (cm $^{-1}$) 3284, 2954, 2923, 2852, 1645, 1508, 1457, 1377, 1206, 748, 699; HRMS (ESI) m/z Calcd for $\text{C}_{29}\text{H}_{29}\text{NONa}$ 430.2147 [M+Na] $^+$, found 430.2139.

N-(5,8-diethyl-6,7-diphenyl)naphthalen-1-yl)pivalamide (5ap).

White solid (69.7 mg, 64% yield); mp: 91–93 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.14 (d, $J = 7.0$ Hz, 1H), 7.78 (s, 1H), 7.54–7.53 (m, 2H), 7.12–7.04 (m, 6H), 6.98–6.93 (m, 4H), 3.01 (q, $J = 7.3$ Hz, 2H), 2.83 (q, $J = 7.4$ Hz, 2H), 1.35 (s, 9H), 1.14 (t, $J = 7.4$ Hz, 3H), 1.06 (t, $J = 7.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 177.3, 141.7, 141.5, 141.4, 139.4, 136.6, 134.5, 133.73, 133.71, 130.2, 130.1, 128.4, 128.1, 127.28, 127.25, 126.0, 125.3, 125.0, 39.5, 27.6, 25.6, 24.1, 16.9, 15.6; IR (KBr) ν (cm $^{-1}$) 3302, 2965, 2931, 2872, 1651, 1602, 1496, 1442, 1366, 1238, 1185, 1167, 909, 760, 739, 700; HRMS (ESI) m/z Calcd for $\text{C}_{31}\text{H}_{33}\text{NONa}$ 458.2460 [M+Na] $^+$, found 458.2465.

N-(6,7-diphenyl-5,8-dipropyl)naphthalen-1-yl)pivalamide (5aq).

White solid (69.5 mg, 60% yield); mp: 139–141 °C. ^1H NMR (400 MHz, CDCl_3) δ 8.08 (d, $J = 8.3$ Hz, 1H), 7.74 (s, 1H), 7.55 (d, $J = 7.0$ Hz, 1H), 7.50 (dd, $J = 8.1$, 7.5 Hz, 1H), 7.11–7.03 (m, 6H), 6.97–6.93 (m, 4H), 2.95 (t, $J = 8.0$ Hz, 2H), 2.75 (t, $J = 8.1$ Hz, 2H), 1.61–1.55 (m, 2H), 1.46–1.41 (m, 2H), 1.35 (s, 9H), 0.82 (t, $J = 7.2$ Hz, 3H), 0.64 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 177.3, 142.0, 141.9, 141.7, 139.6, 135.4, 134.02, 133.98, 133.5, 130.5, 130.3, 128.4, 127.3, 126.0, 125.3, 124.8, 39.7, 34.6, 33.3, 27.8, 25.5, 24.7, 14.9, 14.4; IR (KBr) ν (cm $^{-1}$) 3299, 2959, 2929, 2870, 1652, 1601, 1496, 1466, 1442, 1367, 1239, 1255, 1183, 1167, 909, 753, 733, 700; HRMS (ESI) m/z Calcd for $\text{C}_{33}\text{H}_{37}\text{NONa}$ 486.2773 [M+Na] $^+$, found 486.2764.

N-(2-methyl-5,6,7,8-tetraphenyl)naphthalen-1-yl)pivalamide (5ba).

White solid (68.2 mg, 50% yield); mp: 230–232 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.51 (d, $J = 8.7$ Hz, 1H), 7.30–7.08 (m, 11H), 6.78–6.67 (m, 10H), 6.47 (s, 1H), 2.26 (s, 3H), 0.84 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.2, 143.8, 141.5, 140.59, 140.56, 139.9, 139.1, 138.4, 135.9, 134.8, 132.9, 131.7, 131.4, 131.3, 131.1, 130.7, 129.7, 129.6, 129.2, 129.1, 128.4, 128.1, 127.6, 127.0, 126.54, 126.49, 126.4, 125.4, 125.1, 39.2, 27.1, 19.5; IR (KBr) ν (cm $^{-1}$) 3435, 3341, 2956, 2924, 2854, 1649, 1492, 1459, 1441, 1383, 1157, 1071, 1027, 829, 759, 737, 699; HRMS (ESI) m/z Calcd for $\text{C}_{40}\text{H}_{35}\text{NONa}$ 568.2616 [M+Na] $^+$, found 568.2635.

N-(3-methyl-5,6,7,8-tetraphenyl)naphthalen-1-yl)pivalamide (5ca).

White solid (107.8 mg, 79% yield); mp: 244–246 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.81 (s, 1H), 7.25–7.13 (m, 12H), 6.81–6.73 (m, 8H), 6.63–6.61 (m, 2H), 2.38 (s, 3H), 0.80 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.9, 142.2, 140.6, 140.4, 140.0, 139.0, 138.8, 135.6, 134.24, 134.17, 133.3, 131.3, 131.1, 128.3, 127.6, 127.1, 126.5, 126.4, 125.34, 125.29, 125.14, 125.06, 124.6, 124.5, 123.6, 39.3, 26.9, 21.7; IR (KBr) ν (cm $^{-1}$) 3421, 3343, 2955, 2923, 1669, 1507, 1494, 1457, 1441, 1400, 1165, 1071, 1027, 858, 757, 735, 698; HRMS (ESI) m/z Calcd for $\text{C}_{40}\text{H}_{35}\text{NONa}$ 568.2616 [M+Na] $^+$, found 568.2615.

N-(2-methoxy-5,6,7,8-tetraphenyl)naphthalen-1-yl)pivalamide (5ga).

White solid (84.3 mg, 60% yield); mp: 259–261 °C. ^1H NMR (400 MHz, CDCl_3) δ 7.61 (d, $J = 9.4$ Hz, 1H), 7.23–7.00 (m, 10H), 6.77–6.69 (m, 10H), 6.53 (s, 1H), 6.36 (s, 1H), 3.87 (s, 3H), 0.86 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 177.3, 154.0, 143.3, 141.9, 140.6, 140.5, 139.9, 138.9, 137.2, 134.7, 131.6, 131.3, 131.2, 131.0, 130.9, 130.0, 129.8, 128.8, 128.5, 127.8, 127.6, 127.4, 126.5, 126.3, 126.1, 125.3, 125.0, 120.1, 113.4, 56.7, 39.0, 27.2; IR (KBr) ν (cm $^{-1}$) 3445, 3328, 3056, 3023, 2958, 2925, 2868, 1673, 1602, 1505, 1482, 1441, 1353, 1280, 1118, 756, 698;

For internal use, please do not delete. Submitted_Manuscript

HRMS (ESI) *m/z* Calcd for C₄₀H₃₅NO₂Na 584.2565 [M+Na]⁺, found 584.2569.

N-(3-methoxy-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5ha).

White solid (119.4 mg, 85% yield); mp: 184–186 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.86 (d, *J* = 2.4 Hz, 1H), 7.35 (s, 1H), 7.22–7.12 (m, 10H), 6.82–6.74 (m, 9H), 6.63–6.61 (m, 2H), 3.70 (s, 3H), 0.78 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 176.9, 157.0, 141.9, 140.6, 140.4, 140.1, 139.3, 139.0, 138.3, 135.2, 134.3, 131.4, 131.24, 131.19, 131.0, 128.5, 127.7, 127.3, 126.6, 126.5, 125.4, 125.1, 120.5, 114.7, 105.0, 55.3, 39.4, 27.0; IR (KBr) *ν* (cm⁻¹) 3420, 3056, 3022, 2960, 1674, 1618, 1513, 1493, 1442, 1401, 1356, 1261, 1204, 1178, 1166, 1146, 1069, 910, 756, 734, 699; HRMS (ESI) *m/z* Calcd for C₄₀H₃₅NO₂Na 584.2565 [M+Na]⁺, found 584.2587.

N-(4-methoxy-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5ia).

White solid (123.6 mg, 88% yield); mp: 100–102 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.57 (d, *J* = 8.3 Hz, 1H), 7.13 (s, 5H), 7.06–6.98 (m, 5H), 6.87 (s, 1H), 6.83 (d, *J* = 8.4 Hz, 1H), 6.76 (s, 6H), 6.65 (d, *J* = 5.9 Hz, 2H), 6.58–6.57 (m, 2H), 3.34 (s, 3H), 0.81 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.1, 155.6, 144.0, 142.6, 141.5, 140.39, 140.36, 139.8, 137.3, 134.4, 131.4, 131.2, 130.9, 129.6, 128.7, 128.3, 127.0, 126.8, 126.4, 126.33, 126.28, 125.5, 125.1, 124.9, 107.0, 56.0, 39.2, 27.0; IR (KBr) *ν* (cm⁻¹) 3437, 3327, 3056, 3022, 2956, 2926, 2868, 1663, 1601, 1573, 1493, 1460, 1441, 1366, 1260, 1129, 1071, 757, 736, 697; HRMS (ESI) *m/z* Calcd for C₄₀H₃₅NO₂Na 584.2565 [M+Na]⁺, found 584.2545.

N-(2-chloro-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5ja).

White solid (35.4 mg, 25% yield); mp: 244–246 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.55 (d, *J* = 9.1 Hz, 1H), 7.44 (d, *J* = 9.2 Hz, 1H), 7.30–7.26 (m, 3H), 7.20–6.98 (m, 8H), 6.89–6.71 (m, 9H), 6.69 (s, 1H), 6.64 (d, *J* = 6.8 Hz, 1H), 6.45 (d, *J* = 7.4 Hz, 1H), 0.87 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 176.8, 142.9, 142.4, 140.2, 140.0, 139.6, 139.3, 139.2, 135.5, 133.1, 132.7, 131.6, 131.1, 130.9, 130.8, 130.7, 130.42, 130.36, 129.8, 128.4, 128.2, 127.8, 127.74, 127.66, 127.1, 126.81, 126.75, 126.6, 126.5, 126.4, 125.6, 125.3, 39.3, 27.1; IR (KBr) *ν* (cm⁻¹) 3441, 3322, 3057, 3023, 2958, 2925, 2869, 1672, 1599, 1491, 1442, 1214, 1160, 1072, 1023, 926, 793, 756, 737, 698; HRMS (ESI) *m/z* Calcd for C₃₉H₃₂NONaCl 588.2070 [M+Na]⁺, found 588.2086.

N-(3-chloro-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5ka).

White solid (58.0 mg, 41% yield); mp: 213–215 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.10 (d, *J* = 2.0 Hz, 1H), 7.43 (d, *J* = 2.1 Hz, 1H), 7.38 (s, 1H), 7.24–7.13 (m, 10H), 6.82–6.78 (m, 6H), 6.74–6.72 (m, 2H), 6.62–6.60 (m, 2H), 0.78 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.0, 141.6, 141.5, 140.2, 140.1, 140.0, 139.2, 138.9, 135.2, 134.8, 134.5, 131.8, 131.3, 131.2, 131.0, 128.7, 127.9, 127.6, 127.0, 126.8, 126.6, 125.7, 125.4, 123.9, 123.7, 123.4, 39.5, 27.0; IR (KBr) *ν* (cm⁻¹) 3417, 3335, 3057, 3025, 2963, 2928, 2869, 1673, 1598, 1493, 1455, 1441, 1376, 1347, 1245, 1158, 1027, 909, 861, 733, 698; HRMS (ESI) *m/z* Calcd for C₃₉H₃₃NOCl 566.2251 [M+H]⁺, found 566.2300.

N-(4-chloro-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5la).

White solid (59.4 mg, 42% yield); mp: 205–207 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.80 (d, *J* = 6.4Hz, 1H), 7.51 (d, *J* = 6.4Hz, 1H), 7.18–7.12 (m, 6H), 7.08–7.03 (m, 5H), 6.81–6.78 (m, 6H), 6.63–6.62 (m, 2H), 6.55 (d, *J* = 4.8Hz, 2H), 0.81 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 176.9, 141.6, 141.4, 141.31, 141.25, 140.0, 139.9, 137.8, 134.8, 133.0, 131.6, 131.2, 131.1, 130.8, 129.8, 129.2, 128.5, 127.4, 126.9, 126.6, 126.5, 126.3, 125.4, 124.3, 39.4, 27.0; IR (KBr) *ν* (cm⁻¹) 3426, 3332, 3056, 3021, 2956, 2924, 2853, 1669, 1600, 1492, 1458, 1442, 1363, 1326, 1215, 1028, 814,

758, 732, 696; HRMS (ESI) *m/z* Calcd for C₃₉H₃₂NONaCl 588.2070 [M+Na]⁺, found 588.2056.

N-(3-bromo-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5ma).

White solid (62.6 mg, 41% yield); mp: 252–254 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.21 (d, *J* = 1.7 Hz, 1H), 7.60 (d, *J* = 1.8 Hz, 1H), 7.36 (s, 1H), 7.24–7.12 (m, 10H), 6.81–6.71 (m, 8H), 6.62–6.60 (m, 2H), 0.78 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.0, 141.6, 141.4, 140.1, 140.0, 139.9, 139.1, 138.8, 135.04, 134.96, 134.5, 131.2, 131.14, 131.10, 130.9, 128.6, 127.8, 127.6, 127.0, 126.7, 126.6, 125.6, 125.4, 123.6, 120.0, 39.5, 26.9; IR (KBr) *ν* (cm⁻¹) 3416, 3056, 3023, 2962, 1671, 1590, 1492, 1441, 1370, 1345, 1156, 1073, 1027, 911, 889, 859, 753, 733, 697; HRMS (ESI) *m/z* Calcd for C₃₉H₃₂NONaBr 632.1565 and 634.1545 [M+Na]⁺, found 632.1574 and 634.1562.

Ethyl 5,6,7,8-tetraphenyl-4-pivalamido-2-naphthoate (5na).

White solid (58.9 mg, 39% yield); mp: 138–140 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.41 (s, 1H), 8.28 (s, 1H), 7.22–7.15 (m, 11H), 6.82–6.73 (m, 8H), 6.63–6.62 (m, 2H), 4.32 (q, *J* = 7.2 Hz, 2H), 1.32 (t, *J* = 7.2 Hz, 3H), 0.81 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.1, 166.5, 143.5, 141.7, 141.1, 140.1, 139.94, 139.86, 139.0, 134.6, 134.0, 133.7, 131.3, 131.0, 128.5, 128.2, 127.9, 127.8, 127.7, 127.4, 127.0, 126.8, 126.6, 125.7, 125.5, 124.0, 61.2, 39.4, 27.0, 14.4; IR (KBr) *ν* (cm⁻¹) 3283, 3057, 3024, 2960, 2928, 2869, 1719, 1649, 1493, 1463, 1441, 1403, 1273, 1235, 1173, 1102, 1028, 910, 769, 732, 697; HRMS (ESI) *m/z* Calcd for C₄₂H₃₇NO₃Na 626.2671 [M+Na]⁺, found 626.2662.

N-(3-acetyl-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5oa).

Yellowish solid (50.2 mg, 35% yield); mp: 176–178 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.40 (s, 1H), 8.12 (s, 1H), 7.28–7.16 (m, 11H), 6.83–6.74 (m, 8H), 6.63–6.62 (m, 2H), 2.49 (s, 3H), 0.81 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 197.9, 177.2, 143.7, 141.5, 141.2, 140.0, 139.8, 139.0, 134.6, 134.4, 134.1, 133.7, 131.2, 131.02, 130.97, 128.5, 127.9, 127.8, 127.5, 127.1, 127.0, 126.8, 126.6, 125.7, 125.5, 122.6, 39.4, 27.0, 26.7; IR (KBr) *ν* (cm⁻¹) 3420, 3056, 3024, 2958, 2925, 2865, 1684, 1601, 1493, 1456, 1441, 1402, 1274, 1220, 1171, 910, 734, 698; HRMS (ESI) *m/z* Calcd for C₄₁H₃₅NO₂Na 596.2565 [M+Na]⁺, found 596.2565.

N-(3-cyano-5,6,7,8-tetraphenylnaphthalen-1-yl)pivalamide (5pa).

Yellow solid (33.4 mg, 24% yield); mp: 110–112 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.27 (s, 1H), 7.83 (s, 1H), 7.43 (s, 1H), 7.25–7.20 (m, 8H), 7.12 (d, *J* = 6.4 Hz, 2H), 6.84–6.80 (m, 6H), 6.73–6.71 (m, 2H), 6.61–6.59 (m, 2H), 0.79 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 177.2, 144.4, 140.85, 140.77, 140.3, 139.6, 139.5, 138.4, 135.1, 134.6, 133.6, 131.13, 131.08, 130.9, 130.8, 130.6, 128.8, 128.1, 127.9, 127.4, 126.9, 126.7, 126.4, 125.9, 125.7, 123.2, 119.1, 109.4, 39.6, 26.9; IR (KBr) *ν* (cm⁻¹) 3416, 3057, 3024, 2958, 2925, 2227, 1684, 1493, 1441, 1396, 1158, 1071, 1027, 735, 698; HRMS (ESI) *m/z* Calcd for C₄₀H₃₂N₂O₂Na 579.2412 [M+Na]⁺, found 579.2410.

(E)-N-(2-(1,2-diphenylvinyl)-4-methylphenyl)pivalamide (6a).

Colorless oil (18.5 mg, 20% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.12 (d, *J* = 8.3 Hz, 1H), 7.41 (bs, 1H), 7.24–7.15 (m, 11H), 7.11 (s, 1H), 6.75 (s, 1H), 2.32 (s, 3H), 0.99 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 176.5, 139.6, 138.8, 137.0, 134.5, 133.6, 132.2, 131.6, 129.9, 129.5, 129.3, 128.9, 128.4, 128.3, 127.5, 121.9, 39.7, 27.5, 21.0; IR (neat) *ν* (cm⁻¹) 3429, 3055, 3022, 2958, 2925, 2867, 1684, 1588, 1513, 1492, 1445, 1301, 1157, 921, 822, 759, 697; HRMS (ESI) *m/z* Calcd for C₂₆H₂₇NO₂Na 392.1990 [M+Na]⁺, found 392.2005.

(E)-N-(2-(1,2-diphenylvinyl)-4,5-dimethylphenyl)pivalamide (6b).

Colorless oil (28.8 mg, 30% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.06 (s, 1H), 7.38 (bs, 1H), 7.25–7.13 (m, 10H), 7.06 (s, 1H), 6.73 (s, 1H), 2.28 (s,

For internal use, please do not delete. Submitted_Manuscript

3H), 2.23 (s, 3H), 0.99 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.5, 139.5, 139.0, 137.2, 137.1, 133.3, 132.3, 132.1, 131.9, 130.0, 129.5, 128.9, 128.4, 128.2, 127.3, 123.0, 39.7, 27.5, 19.9, 19.3; IR (neat) ν (cm^{-1}) 3429, 3055, 3021, 2960, 2925, 2867, 1684, 1574, 1516, 1491, 1479, 1448, 1400, 1200, 1166, 921, 881, 756, 710, 695; HRMS (ESI) m/z Calcd for $\text{C}_{27}\text{H}_{29}\text{NONa}$ 406.2147 [M+Na] $^+$, found 406.2156.

(E)-N-(4-butyl-2-(1,2-diphenylvinyl)phenyl)pivalamide (6c).

Colorless oil (40.1 mg, 39% yield). ^1H NMR (400 MHz, CDCl_3) δ 8.13 (d, $J = 8.4$ Hz, 1H), 7.39 (bs, 1H), 7.23–7.15 (m, 12H), 7.10 (d, $J = 2.4$ Hz, 1H), 6.76 (s, 1H), 2.58 (t, $J = 7.7$ Hz, 2H), 1.63–1.56 (m, 2H), 1.40–1.31 (m, 2H), 0.99 (s, 9H), 0.92 (t, $J = 7.3$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 176.4, 139.7, 138.7, 137.0, 134.4, 133.3, 132.1, 131.0, 130.0, 129.5, 128.9, 128.7, 128.4, 128.3, 127.4, 121.9, 39.7, 35.2, 33.8, 27.5, 22.5, 14.1; IR (neat) ν (cm^{-1}) 3430, 3056, 3022, 2956, 2929, 2869, 1686, 1586, 1513, 1445, 1412, 1300, 1158, 922, 759, 697; HRMS (ESI) m/z Calcd for $\text{C}_{29}\text{H}_{33}\text{NONa}$ 434.2460 [M+Na] $^+$, found 434.2451.

Acknowledgements

We are grateful to the National Natural Science Foundation of China [Nos. 21372035, 21573032, and 21361140375 (NSFC-IUPAC program)] for their financial support. This work was also supported by the Fundamental Research Funds for the Central Universities (DUT15LK37). The authors extend their appreciation to the International Scientific Partnership Program ISPP at King Saud University for funding this research work through ISPP#0048.

Keywords: Rhodium catalysis • Benzannulation • *N*-Pivaloylaniline • Naphthalene

- [1] (a) Dyker, G. *Handbook of C-H Transformations: Applications in Organic Synthesis*; Wiley-VCH: Weinheim, **2005**. (b) Yu, J.-Q.; Shi, Z.-J. *C–H Activation*; Springer: Berlin, Germany, **2010**. For selected reviews, see: (c) Colby, D. A.; Bergman, R. G.; Ellman, J. A. *Chem. Rev.* **2010**, 110, 624. (d) Wencel-Delord, J.; Dröge, T.; Liu, F.; Glorius, F. *Chem. Soc. Rev.* **2011**, 40, 4740. (e) Cho, S. H.; Kim, J. Y.; Kwak, J.; Chang, S. *Chem. Soc. Rev.* **2011**, 40, 5068. (f) Yeung, C. S.; Dong, V. M. *Chem. Rev.* **2011**, 111, 1215. (g) Ackermann, L. *Chem. Rev.* **2011**, 111, 1315. (h) Li, B.-J.; Shi, Z.-J. *Chem. Soc. Rev.* **2012**, 41, 5588. (i) Engle, K. M.; Mei, T.-S.; Wasa, M.; Yu, J.-Q. *Acc. Chem. Res.* **2012**, 45, 788. (j) Neufeldt, S. R.; Sanford, M. S. *Acc. Chem. Res.* **2012**, 45, 936. (k) Jeffrey, J. L.; Sarpong, R. *Chem. Sci.* **2013**, 4, 4092. (l) Segawa, Y.; Maekawa, T.; Itami, K. *Angew. Chem., Int. Ed.* **2015**, 54, 66. For selected examples in total synthesis, see: (m) Feng, Y.; Chen, G. *Angew. Chem., Int. Ed.* **2010**, 49, 958. (n) Gutekunst, W. R.; Baran, P. S. *J. Am. Chem. Soc.* **2011**, 133, 19076. (o) Weinstabl, H.; Suhartono, M.; Qureshi, Z.; Lautens, M. *Angew. Chem., Int. Ed.* **2013**, 52, 5305. (p) Ueda, K.; Amaike, K.; Maceiczyk, R. M.; Itami, K.; Yamaguchi, J. *J. Am. Chem. Soc.* **2014**, 136, 13226. (q) Feng, Y.; Holte, D.; Zoller, J.; Umemiya, S.; Simke, L. R.; Baran, P. S. *J. Am. Chem. Soc.* **2015**, 137, 10160.
- [2] For selected examples, see: (a) Li, Y.; Li, B.-J.; Wang, W.-H.; Huang, W.-P.; Zhang, X.-S.; Chen, K.; Shi, Z.-J. *Angew. Chem., Int. Ed.* **2011**, 50, 2115. (b) Gao, K.; Yoshikai, N. *Chem. Commun.* **2012**, 48, 4305. (c) Xie, F.; Qi, Z.; Li, X. *Angew. Chem., Int. Ed.* **2013**, 52, 11862. (d) Zheng, J.; You, S.-L. *Angew. Chem., Int. Ed.* **2014**, 53, 13244. (e) Zhou, B.; Hu, Y.; Wang, C. *Angew. Chem., Int. Ed.* **2015**, 54, 13659.
- [3] For selected examples, see: (a) Karthikeyan, J.; Cheng, C.-H. *Angew. Chem., Int. Ed.* **2011**, 50, 9880. (b) Wang, H.; Grohmann, C.; Nimphius, C.; Glorius, F. *J. Am. Chem. Soc.* **2012**, 134, 19592. (c) Wang, H.; Beiring, B.; Yu, D.-G.; Collins, K. D.; Glorius, F. *Angew. Chem., Int. Ed.* **2013**, 52, 12430. (d) Zhang, S.-Y.; Li, Q.; He, G.; Nack, W. A.; Chen, G. *J. Am. Chem. Soc.* **2015**, 137, 531. (e) Zhang, L.-B.; Hao, X.-Q.; Liu, Z.-J.; Zheng, X.-X.; Zhang, S.-K.; Niu, J.-L.; Song, M.-P. *Angew. Chem., Int. Ed.* **2015**, 54, 10012. (f) Hermann, G. N.; Becker, P.; Bolm, C. *Angew. Chem., Int. Ed.* **2016**, 55, 3781.
- [4] For selected examples, see: (a) Boele, M. D. K.; Van Strijdonck, G. P. F.; De Vries, A. H. M.; Kamer, P. C. J.; De Vries, J. G.; Van Leeuwen, P. W. N. M. *J. Am. Chem. Soc.* **2002**, 124, 1586. (b) Brasche, G.; García-Fortanet, J.; Buchwald, S. L. *Org. Lett.* **2008**, 10, 2207. (c) Li, C.; Wang, L.; Li, P.; Zhou, W. *Chem. Eur. J.* **2011**, 17, 10208. (d) Neufeldt, S. R.; Seigerman, C. K.; Sanford, M. S. *Org. Lett.* **2013**, 15, 2302. (e) Hubrich, J.; Himmler, T.; Rodefeld, L.; Ackermann, L. *Adv. Synth. Catal.* **2015**, 357, 474.
- [5] Zhou, J.; Li, B.; Hu, F.; Shi, B.-F. *Org. Lett.* **2013**, 15, 3460.
- [6] Selected examples for $\text{C}_\text{aryl}-\text{C}_\text{alkenyl}$ bond formation via C–H bond activation, see: (a) Nishikata, T.; Lipshutz, B. H. *Org. Lett.* **2010**, 12, 1972. (b) Patureau, F. W.; Glorius, F. *J. Am. Chem. Soc.* **2010**, 132, 9982. (c) Ackermann, L.; Wang, L.; Wolfram, R.; Lygin, A. V. *Org. Lett.* **2012**, 14, 728. (d) Manikandan, R.; Jeganmohan, M. *Org. Lett.* **2014**, 16, 912. (e) Hermann, G. N.; Becker, P.; Bolm, C. *Angew. Chem., Int. Ed.* **2015**, 54, 7414. (f) Takahama, Y.; Shibata, Y.; Tanaka, K. *Chem. Eur. J.* **2015**, 21, 9053. (g) Manikandan, R.; Madasamy, P.; Jeganmohan, M. *ACS Catal.* **2016**, 6, 230.
- [7] Selected examples for $\text{C}_\text{aryl}-\text{C}_\text{aryl}$ bond formation via C–H bond activation, see: (a) Daugulis, O.; Zaitsev, V. G. *Angew. Chem., Int. Ed.* **2005**, 44, 4046. (b) Shi, Z.-J.; Li, B.; Wan, X.; Cheng, J.; Fang, Z.; Cao, B.; Qin, C.; Wang, Y. *Angew. Chem., Int. Ed.* **2007**, 46, 5554. (c) Yang, S.; Li, B.; Wan, X.; Shi, Z.-J. *J. Am. Chem. Soc.* **2007**, 129, 60667. (d) Li, B.-J.; Tian, S.-L.; Fang, Z.; Shi, Z.-J. *Angew. Chem., Int. Ed.* **2008**, 47, 1115. (e) Scarborough, C. C.; McDonald, R. I.; Hartmann, C.; Sazama, G. T.; Bergant, A.; Stahl, S. S. *J. Org. Chem.* **2009**, 74, 2613. (f) Yeung, C. S.; Zhao, X.; Borduas, N.; Dong, V. M. *Chem. Sci.* **2010**, 1, 331. (g) Neufeldt, S. R.; Sanford, M. S. *Adv. Synth. Catal.* **2012**, 354, 3517. (h) Huang, L.; Li, Q.; Wang, C.; Qi, C. *Org. Lett.* **2013**, 15, 3030. (i) Chinnagolla, R. K.; Jeganmohan, M. *Chem. Commun.* **2014**, 50, 2442. (j) Haridharan, R.; Muralirajan, K.; Cheng, C.-H. *Adv. Synth. Catal.* **2015**, 357, 366.
- [8] Selected examples for $\text{C}_\text{aryl}-\text{C}_\text{alkyl}$ bond formation via C–H bond activation, see: (a) Zhang, L.-Z.; Chen, K.; Chen, G.; Li, B.-J.; Luo, S.; Guo, Q.-Y.; Wei, J.-B.; Shi, Z.-J. *Org. Lett.* **2013**, 15, 10. (b) Chan, W.-W.; Zhou, Z.; Yu, W.-Y. *Chem. Commun.* **2013**, 49, 8214. (c) Graczyk, K.; Haven, T.; Ackermann, L. *Chem. Eur. J.* **2015**, 21, 8812. (d) Crisenza, G. E. M.; Sokolova, O. O.; Bower, J. F. *Angew. Chem., Int. Ed.* **2015**, 54, 14886.
- [9] Selected examples for $\text{C}_\text{aryl}-\text{C}_\text{carbonyl}$ bond formation via C–H bond activation, see: (a) Giri, R.; Lam, J. K.; Yu, J.-Q. *J. Am. Chem. Soc.* **2010**, 132, 686. (b) Fang, P.; Li, M.; Ge, H.; *J. Am. Chem. Soc.* **2010**, 132, 11898. (c) Wu, Y.; Li, B.; Mao, F.; Li, X.; Kwong, F. Y. *Org. Lett.* **2011**, 13, 3258. (d) Hesp, K. D.; Bergman, R. G.; Ellman, J. A. *J. Am. Chem. Soc.* **2011**, 133, 11430. (e) Yin, Z.; Sun, P. *J. Org. Chem.* **2012**, 77, 11339. (f) Xie, Y.; Yang, Y.; Huang, L.; Zhang, X.; Zhang, Y. *Org. Lett.* **2012**, 14, 1238. (g) Wang, S.; Yang, Z.; Liu, J.; Xie, K.; Wang, A.; Chen, X.; Tan, Z. *Chem. Commun.* **2012**, 48, 9924. (h) Szabó, F.; Daru, J.; Simkó, D.; Zs. Nagy, T.; Stirling, A.; Novák, Z. *Adv. Synth. Catal.* **2013**, 355, 685. (i) Kim, M.; Mishra, N. K.; Park, J.; Han, S.; Shin, Y.; Sharma, S.; Lee, Y.; Lee, E.-K.; Kwak, J. H.; Kim, I. S. *Chem. Commun.* **2014**, 50, 14249. (j) Zhou, C.; Li, P.; Zhu, X.; Wang, L. *Org. Lett.* **2015**, 17, 6198. (k) Han, L.; Wang, Y.; Song, H.; Han, H.; Wang, L.; Chu, W.; Sun, Z. *RSC Adv.* **2016**, 6, 20637.
- [10] Selected examples for $\text{C}_\text{aryl}-\text{Nbond}$ formation via C–H bond activation, see: (a) Tsang, W. C. P.; Zheng, N.; Buchwald, S. L. *J. Am. Chem. Soc.*

For internal use, please do not delete. Submitted_Manuscript

- 2005, 127, 14560. (b) Tsang, W. C. P.; Munday, R. H.; Brasche, G.; Zheng, N.; Buchwald, S. L. *J. Org. Chem.* **2008**, 73, 7603. (c) Ng, K.-H.; Chan, A. S. C.; Yu, W.-Y. *J. Am. Chem. Soc.* **2010**, 132, 12862. (d) Youn, S. W.; Bihn, J. H.; Kim, B. S. *Org. Lett.* **2011**, 13, 3738. (e) Sun, K.; Li, Y.; Xiong, T.; Zhang, J.; Zhang, Q. *J. Am. Chem. Soc.* **2011**, 133, 1694. (f) Martínez, Á. M.; Rodríguez, N.; Arrayás, R. G.; Carretero, J. C. *Chem. Commun.* **2014**, 50, 2801. (g) Li, Q.; Zhang, S.-Y.; He, G.; Ai, Z.; Nack, W. A.; Chen, G. *Org. Lett.* **2014**, 16, 1764.
- [11] Selected examples for C_{aryl} -Obond formation via C–H bond activation, see: (a) Wang, G.-W.; Yuan, T.-T.; Wu, X.-L. *J. Org. Chem.* **2008**, 73, 4717. (b) Jiang, T.-S.; Wang, G.-W. *J. Org. Chem.* **2012**, 77, 9504. (c) Shan, G.; Yang, X.; Ma, L.; Rao, Y. *Angew. Chem., Int. Ed.* **2012**, 51, 13070. (d) Yang, X.; Shan, G.; Rao, Y. *Org. Lett.* **2013**, 15, 2334. (e) Yang, F.; Song, F.; Li, W.; Lan, J.; You, J. *RSC Adv.* **2013**, 3, 9649. (f) Seth, K.; Nautiyal, M.; Purohit, P.; Parish, N.; Chakraborti, A. K. *Chem. Commun.* **2015**, 51, 191.
- [12] Selected example for C_{aryl} –B formation via C–H bond activation, see: Xiao, B.; Li, Y.-M.; Liu, Z.-J.; Yang, H.-Y.; Fu, Y. *Chem. Commun.* **2012**, 48, 4854.
- [13] Selected examples for C_{aryl} –Cl bond formation via C–H bond activation, see: (a) Wan, X.; Ma, Z.; Li, B.; Zhang, K.; Cao, S.; Zhang, S.; Shi, Z. *J. Am. Chem. Soc.* **2006**, 128, 7416. (b) Bedford, R. B.; Haddow, M. F.; Mitchell, C. J.; Webster, R. L. *Angew. Chem., Int. Ed.* **2011**, 50, 5524. (c) Sun, X.; Shan, G.; Sun, Y.; Rao, Y. *Angew. Chem., Int. Ed.* **2013**, 52, 4440. (d) Urones, B.; Martínez, Á. M.; Rodriguez, N.; Arrayás, R. G.; Carretero, J. C. *Chem. Commun.* **2013**, 49, 11044.
- [14] Selected examples for C_{aryl} –Br bond formation via C–H bond activation, see: (a) See Ref. 8b. (b) Schröder, N.; Wencel-Delord, J.; Glorius, F.; *J. Am. Chem. Soc.* **2012**, 134, 8298. (c) See Ref. 8c. (d) See Ref. 8d.
- [15] Selected example for C_{aryl} –I bond formation via C–H bond activation, see: Ref. 9b.
- [16] Selected example for carbocycle formation via C–H bond activation, see: Wu, J.; Cui, X.; Mi, X.; Li, Y.; Wu, Y. *Chem. Commun.* **2010**, 46, 6771.
- [17] Selected examples for heterocycle formation via C–H bond activation, see: (a) Stuart, D. R.; Bertrand-Laperle, M.; Burgess, K. M. N.; Fagnou, K. *J. Am. Chem. Soc.* **2008**, 130, 16474. (b) Xiong, T.; Li, Y.; Bi, X.; Lv, Y.; Zhang, Q. *Angew. Chem., Int. Ed.* **2011**, 50, 7140. (c) Cajaraville, A.; López, S.; Varela, J. A.; Saá, C. *Org. Lett.* **2013**, 15, 4576. (d) Liu, C.; Liu, D.; Zhang, W.; Zhou, L.; Lei, A. *Org. Lett.* **2013**, 15, 6166. (e) Hoshino, Y.; Shibata, Y.; Tanaka, K. *Adv. Synth. Catal.* **2014**, 356, 1577. (f) Gao, Y.; Huang, Y.; Wu, W.; Huang, K.; Jiang, H. *Chem. Commun.* **2014**, 50, 8370. (g) Lin, H.; Li, S.-S.; Dong, L. *Org. Biomol. Chem.* **2015**, 13, 11228.
- [18] (a) Umeda, N.; Hirano, K.; Satoh, T.; Shibata, N.; Sato, H.; Miura, M. *J. Org. Chem.* **2011**, 76, 13. (b) Song, G.; Gong, X.; Li, X. *J. Org. Chem.* **2011**, 76, 7583. (c) Shi, Z.; Tang, C.; Jiao, N. *Adv. Synth. Catal.* **2012**, 354, 2695.
- [19] A benzannulation product was obtained in relatively low yield (39%) when *N*-(naphthalen-1-yl)pivalamide was treated under the same reaction conditions. See: Zhang, X.; Yu, X.; Ji, D.; Yamamoto, Y.; Almansour, A. I.; Arumugam, N.; Kumar, R. S.; Bao, M. *Org. Lett.* **2016**, 18, 4246.
- [20] (a) Stuart, D. R.; Alsabeh, P.; Kuhn, M.; Fagnou, K. *J. Am. Chem. Soc.* **2010**, 132, 18326. (b) Larock, R. C.; Yum, E. K.; Refvik, M. D. *J. Org. Chem.* **1998**, 63, 7652.
- [21] Fujita, K.; Takahashi, Y.; Owaki, M. *Org. Lett.* **2004**, 6, 2785.
- [22] Wang, J.; Tong, X.; Xie, X. *Org. Lett.* **2010**, 12, 5370.
- [23] Bellamy, E.; Bayh, O.; Hoarau, C.; Trécourt, F.; Quéguiner, G.; Marsais, F. *Chem. Commun.* **2010**, 46, 7043.
- [24] Chuentragool, P.; Vongnam, K.; Rashatasakhon, P.; Sukwattansinitt, M.; Wacharasindhu, S. *Tetrahedron* **2011**, 67, 8177.
- [25] Chan, W.-W.; Zhou, Z.; Yu, W.-Y. *Chem. Commun.* **2013**, 49, 8214.
- [26] Phipps, R. J.; Gaunt, M. J. *Science* **2009**, 323, 1593.
- [27] Ng, K.-H.; Chan, A. S. C.; Yu, W.-Y. *J. Am. Chem. Soc.* **2010**, 132, 12862.
- [28] Bedford, R. B.; Haddow, M. F.; Mitchell, C. J.; Webster, R. L. *Angew. Chem. Int. Ed.* **2011**, 50, 5524.
- [29] Ueda, S.; Nagasawa, H. *J. Org. Chem.* **2009**, 74, 4272.

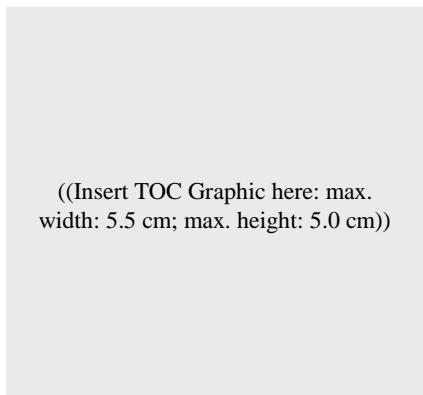
For internal use, please do not delete. Submitted_Manuscript

FULL PAPER**WILEY-VCH****Entry for the Table of Contents (Please choose one layout)**

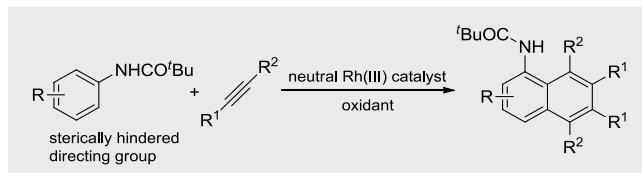
Layout 1:

FULL PAPER

Text for Table of Contents

*Author(s), Corresponding Author(s)***Page No. – Page No.**Title*

Layout 2:

FULL PAPER

The regioselectivity in the reaction of *N*-aryl amides with internal alkynes could be easily turned by using a neutral rhodium(III) catalyst and sterically hindered directing group to prepare highly substituted naphthalenes.

Xuan Zhang, Xiaoqiang Yu*, Xiujuan Feng, Yoshinori Yamamoto, Abdulrahman I. Almansour, Natarajan Arumugam, and Raju Suresh Kumar, Ming Bao*

Page No. – Page No.

Rhodium-catalyzed Oxidative Benzannulation of *N*-Pivaloylanilines with Internal Alkynes via Dual C–H Bond Activation: Synthesis of Highly Substituted Naphthalenes

For internal use, please do not delete. Submitted_Manuscript