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Characterization of Conformationally Constrained Benzanilide Scaffolds for Potent and Selective HDAC8 Targeting

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ABSTRACT: HDACs are an attractive therapeutic target for a variety of human diseases. Currently, all four FDA-approved HDACtargeting drugs are non-selective, pan-HDAC inhibitors, exhibiting adverse side-effects at therapeutic doses. While selective HDACinhibition has been proposed to mitigate toxicity, the targeted catalytic domains are highly conserved. Herein, we describe a series of rationally designed, conformationally constrained, benzanilide foldamers which selectively bind the catalytic tunnel of HDAC8. The series includes benzanilides, **MMH371**, **MMH409**, and **MMH410**, which exhibit potent *in vitro* HDAC8 activity (IC₅₀ = 66, 23, and 66 nM, respectively), and up to 410-fold selectivity for HDAC8 over the next targeted HDAC. Experimental and computational analysis of the benzanilide structure docked with human HDAC8 enzyme showed the adoption of a low energy L-shape conformer that favours HDAC8 selectivity. Conformationally constrained HDAC8-inhibitors present an alternative biological probe for further determining the clinical utility and safety of pharmacological knockdown of HDAC8 in diseased cells.

INTRODUCTION

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Histone Deacetylases (HDACs) constitute a family of eraser proteins that are canonically known to remove acetyl groups from specific Lys side-chains of histones in order to regulate transcription. Deacetylation of Lys affords the positively charged amino group (pKa \approx 10.3), which strengthens the electrostatic interactions between histones and the DNA backbone, reducing access to DNA by transcription factors. HDACs are known to be involved in gene transcription and other cellular processes.¹ Due to their role in gene transcription, they have the ability to increase the expression of growth promoting factors and decrease the expression of tumour suppressors.² Recently, HDACs have been found to have other non-histone substrates.³⁻⁵ For instance, knockdown of overexpressed HDAC2 in colorectal cells⁶ resulted in upregulation of the cyclin-dependent kinase inhibitor, p21, independent of p53, supporting the hypothesis that HDAC2 is involved in carcinogenesis.7 Moreover, numerous studies have demonstrated increased HDAC levels in several cancers compared to normal tissues.^{6,8–12}

Due to their biological roles in the progression of multiple cancers, including peripheral T-cell lymphoma (PTCL), multiple myeloma, and neuroblastoma, there is considerable interest in targeting HDACs. The United States Food and Drug Administration (FDA) has approved four pan-HDAC inhibitors to date: Vorinostat (SAHA), Romidepsin (FK228), Belinostat (PXD101) and Panobinostat (LBH-589).10 Pan-HDAC inhibition profiles are associated with adverse side-effects such diarrhea, thrombocytophenia, anemia and cardiac as toxicity.^{2,4,9,13,14} Mouse models have highlighted that selective knockdown of HDAC enzymes can improve therapeutic outcomes whereas the knockdown of other HDACs result in lethal side-effects. For instance, knockdown studies of HDAC6 and HDAC8 have resulted in impediment of tumor growth,^{15,16} whereas embryonic disruption of specific HDACs (including 1, 2, 3, 7) in mouse models is associated with lethal outcomes.¹¹ The toxic side-effects observed in clinical trials of approved pan-HDAC inhibitors correlate with side effects observed in certain HDAC knockouts (e.g. HDAC2, 3, 5, 9).¹¹ Selective HDAC inhibitors (HDACi) therefore stand to have wider therapeutic windows. As a result, significant efforts towards the development of small molecule HDAC-selective inhibitors

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have been made in the past decade. One of the main challenges with targeting HDACs selectively arises from the highly conserved amino acid sequence in the catalytic domain.^{17–20} Although the catalytic tunnel is highly conserved, the surface topology and tunnel shape vary between HDAC family members, thus providing an opportunity for selective small molecule intervention.

HDAC8 has been implicated in multiple neurological and Tcell malignancies as well as Schistosomiasis, caused by the human flat-worm *Schistosoma mansoni*. Previous studies involving small molecule inhibition of HDAC8 have revealed unique structural features that can be exploited for selective HDAC8 ablation.²¹ More specifically, soaking experiments of apo *Schistosoma mansoni* HDAC8 with HDAC8-selective inhibitors **PCI34051** (Figure 2b), NCC-149, as well as pan-HDAC inhibitor Quisinostat, indicated that HDAC8 selective inhibitors adopt an L-shape that allows them to interact with the 'HDAC8-specific pocket' formed by the catalytic tyrosine between the L1 and L6 loops (**Figure 1a**).²¹ Quisinostat adopts a linear conformation that does not interact with the L1-L6 HDAC8 selectivity filter, thereby exhibiting non-selective HDAC inhibition. Furthermore, the lack of rigidity in clinically approved HDAC inhibitors (Vorinostat, Panobinstat, Belinostat etc.) afforded by their multiple rotatable centers, allows them to adapt to the various HDAC tunnel shapes, and results in pan-HDAC inhibition.

We were interested in the design of rigid L-shaped scaffolds that interact favourably with the HDAC8 selectivity pocket and are poorly accommodated by other HDAC8 such as HDAC6 (**Figure 1b**). The traditional pharmacophoric model of HDACi consists of three groups – the cap group, the linker, and the zincbinding group (**Figure 2a**), with a majority of inhibitors containing flexible motifs with rotatable centers. Herein, conformationally constrained benzanilide foldamers, adopting an L-shaped conformation, are shown to afford HDAC8 tunnel selectivity (**Figure 1b, 2c**).



Figure 1. (a) Birds-eye view of hHDAC8 (PDB 1W22) compared to hHDAC6 (PDB 5EDU) surface topology. HDAC8 contains an L1-L6 basin whereas the L1 and L6 loops of HDAC6 are locked creating a linear Lys tunnel. (b) A 'tetris'-based model for determining HDAC8 selectivity. L-shaped inhibitors are accommodated by the HDAC8 tunnel but not in other HDACs.

RESULTS AND DISCUSSION

Structure Activity Relationship Studies. The HDAC inhibitor, 4-tert-butyl-N-[4-(hydroxycarbamoyl)phenyl]benzamide, previously published as **AES-350** (HDAC8 IC₅₀ = 245 nM, HDAC6 IC₅₀ = 24.4 nM)²², was the starting point for our structure-activity relationship (SAR). Whilst more selective for HDAC6, with potency of 245 nM against HDAC8, it was hoped that a medicinal chemistry program could improve HDAC8 activity whilst reducing other HDAC potency. As part of the SAR, the potency and selectivity of each small molecule was assessed in an enzymatic-activity based Electrophoretic Mobility Shift Assay (EMSA) against HDACs 3, 6, 8 and 11 (*Nanosyn*, USA). First, moving the *t*Bu moiety on the cap-group of **AES-350** from the *para*- to *meta*- position gave **1**, and resulted in increased HDAC8 activity, while retaining HDAC6 potency

(HDAC8 IC₅₀ = 34 nM, HDAC6 IC₅₀ = 71 nM). To improve selectivity, it was hypothesized that N-amide substitution might hinder the hydroxamic acid moiety from entering the linear HDAC6 tunnel and coordinating the Zn²⁺ atom at the base of the tunnel. The hypothesis was explored first with an N-Et (2) substituent, which exhibited HDAC8 activity ($IC_{50} = 83.5 \text{ nM}$), and >12-fold selectivity over HDACs 3, 6, and 11 (IC_{50} 's > 1 μ M). Exploration of *N*-alkylated benzamides for deriving HDAC8 selective inhibitors was then conducted. Interestingly, structural studies of benzanilide-scaffolds have shown the adoption of either an E- or Z conformation (or cis and trans)depending on the selection of substitution at N. For example, a proton substituent primarily favours a Z conformation, whereas N-methylated homologs favour the E conformation.²³ In the case of 2, it was hypothesized that the Et group favoured an E-(or 'L-shaped') conformation, which was proposed to be complimentary to the topology of HDAC8's catalytic pocket (Figure 1).

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Figure 2. (a) Pharmacophoric model of HDACi. (b) Structure of PCI-34051, an HDAC8 selective inhibitor. (c) Controlling *E*–*Z* equilibrium *via N*-substitution of benzanilide based hydroxamic acids. *N*-H benzanilides favor the *Z* conformation whereas, *N*-alkylation (R_1 = alkyl group) favors the *E* conformation.



The identification of 2 prompted a thorough and systematic SAR, guided by Hirano and Marek's structural observations, to identify an HDAC8 selective pharmacophore. First, to investigate the sp^2 amide bond, compound 3, containing a more flexible sp³ sulfonylamide, was prepared. Compound 3 exhibited pan-HDAC activity against representative HDACs, 3, 6, and 11. This data supports the hypothesis that conformational rigidity is important for HDAC8 selectivity with benzanilide scaffolds. To better interact with hydrophobic Phe residues in the tunnel, a second tBu substituent (3,5-tBu) was introduced to the cap group (7). Despite its size, the potency and selectivity of 7 were retained, with an $IC_{50} = 80$ nM against HDAC8 and a > 12-fold selectivity for the nearest HDAC enzyme. Next, a 3- CF_3 substituted analog, 4, was prepared, which marginally increased HDAC8 potency ($IC_{50} = 62 \text{ nM}$) and selectivity (16fold). Replacing the amide group (4) with a sulfonamide (5)resulted in a drop in potency (HDAC6; $IC_{50} = 129$ nM) and selectivity (>12 to 5-fold). Transposed-amide homologues, 13, 14, and 23, which are unable to adopt the L-shaped conformer, showed reduced HDAC8 potency and selectivity. Additionally, an N-alkylated transposed amide analogue, 24, also showed diminished HDAC8 potency and selectivity. This data supported the hypothesis that positioning of the amide, in addition to adopting the preferred L-shape, is critical to maintaining HDAC8 potency and selectivity.

Given the role of the N-substitution in promoting the $K_{E/Z}$ (Figure 2c), we probed the importance of the side group (R_1) by preparing a series of bulky N-aryl homologues of 4 (8-10). All benzanilides with bulky aryl side-groups, including those that incorporated the 3,5-tBu benzyl (8) and $3-CF_3$ benzyl (9), resulted in a loss of selectivity and potency against HDAC8. Further side-group optimization efforts included incorporation of an iPr (16) and methylene cPr on each of the HDAC8 selective leads, 2, 4, and 7. The inclusion of iPr (16) had minimal effects on both potency and selectivity relative to the N-Et analog (4), whereas the inclusion of the methylene cPr group (MMH371, 15, and 17) resulted in a decrease in both potency and selectivity with the exception of MMH371. MMH371, which includes a 3-tBu cap group with a methylene cPr R₁ side group, showed an increase in both potency and selectivity (HDAC8 IC₅₀ = 66 nM and >15-fold selectivity). MMH371 was also tested against the remaining class I HDAC family members in the same EMSA assay, and showed no activity against HDACs 1 and 2, IC₅₀'s >10 µM (table S2). Due to the potency and selectivity profiles of N-iPr derivatives, the N-iPr moiety was incorporated in a variety of subsequent derivatives. It is known that iPr and cPr groups are less-likely to be recognized and metabolized in biological systems.²⁴

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Table 1 $-IC_{50}$ values for inhibitors against HDACs 3, 6, 8, and 11 in an activity-based Electrophoretic Mobility Shift Assay (EMSA) (treatment range, 0.000244 - 1 μ M). Equilibrium dissociation constants (K_D) and IC₅₀ independently determined using Surface Plasmon Resonance (SPR) and Fluorescence Polarization (FP) assays, respectively, against HDAC8.

3					EN	ASA HDA	C IC ₅₀ (µN	1)			ED
4 5	Code	R ₁	X	\mathbf{R}_2					SI	SPR K _D (µM)	FP IC ₅₀ (µ M)
6 7					3	6	8	11			(µ111)
8 9 10	1	<u></u> ⊢H	° ///	3- <i>t</i> Bu	>1	0.066	0.0337	>1	1.96	_	_
11 12 13	2	\sim	° V	3- <i>t</i> Bu	>1	>1	0.0835	>1	>12.0	0.0261	0.203
14 15 16	3	\sim	o, o ∕s∕∕	3- <i>t</i> Bu	0.377	0.866	0.0553	0.802	6.82	0.0239	0.117
17 18 19	4	\sim	° V	3-CF ₃	>1	>1	0.0623	>1	>16.1	0.0375	0.144
20 21 22	5	\sim	o, _o ≻ ^s ∕∕	3-CF ₃	0.661	0.712	0.129	0.695	5.12	0.0459	0.340
23 24 25	6	\sim	oू_o ∕` ^s ∕∕	3,5-CF ₃	>1	>1	0.308	>1	>3.25	0.0947	0.547
20 27 28 29	7	\sim	° V	3,5- <i>t</i> Bu	>1	>1	0.0798	>1	>12.5	0.0951	0.108
30 31 32 33 34	8ª	Y CHA	° V	3-CF ₃	>2	0.752	>2	0.673	0.337	-	3.030
35 36 37 38	9ª	FFF	° V	3-CF ₃	>2	0.225	0.648	0.526	0.347	_	0.727
39 40 41 42	10ª	YO	° V	3-CF ₃	>2	0.175	0.264	>2	0.663	_	0.0596
43 44 45	11	<u></u> ⊢H	° V	3-CF ₃	0.254	0.0596	0.478	>1	0.124	-	0.266
46 47 48	12	\sim	° V	2-CF ₃	0.904	0.821	0.823	>1	0.998	-	0.368
49 50 51	13		\checkmark	3-CF ₃	0.312	0.118	>1	>1	0.118	_	_
52 53 54 55	14		\checkmark	3,5-CF ₃	>1	>1	>1	>1	0.528	-	_
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1 2	15	$\bigvee \bigtriangledown$	° V	3-CF ₃	>1	>1	0.116	>1	>8.62	0.00981	0.0328
3 4 5	16	\checkmark	° ↓ ↓	3-CF ₃	>1	>1	0.0651	>1	>15.4	0.00770	0.0625
6 7 8	MMH371 ^b	$\bigvee \bigtriangledown$	° V	3 <i>-t</i> Bu	>1	>1	0.0660	>1	>15.2	0.0108	0.0587
9 10 11 12	17	$\bigvee \bigtriangledown$	° V	3,5- <i>t</i> Bu	>1	>1	0.225	>1	>4.44	_	Below limit
13 14 15	18	}—H	° V	3,5- <i>t</i> Bu	0.0996	0.19	>1	>1	0.0996	_	_
16 17 18	MMH409°	\checkmark	° V	3,5-CF ₃	>1	>1	0.0234	>1	410	0.00350	0.113
19 20 21	MMH410 ^b	\checkmark	° V	3-OCF ₃	>1	>1	0.0655	>1	>15.3	0.00848	0.0728
22 23 24 25	19	\checkmark	° V	2-F, 3- CF ₃	>1	>1	0.0885	>1	>11.3	0.00840	0.0418
25 26 27 28	20	\checkmark	° V	3-CF ₃ , 4-F	>1	>1	0.119	>1	>8.40	0.00880	Below limit
29 30 31	21	\checkmark	° ↓ ↓	3,5- <i>t</i> Bu	>1	>1	0.107	>1	>9.35	0.0260	0.0119
32 33 34	22	ŀН	° ↓ ↓	3,5-CF ₃	0.672	0.112	0.682	-	0.137	_	_
35 36 37	23 ^d	}—H	° ↓ ✓	3-tBu	0.345	0.414	>1	>1	0.345	-	1.04
30 39 40 41	24ª	\sim	° ↓ ✓	3-tBu	>1	>1	0.732	>1	1.37	-	0.358
42 43 44	PCI34051°	_	_	-	>1	>1	0.0040 3	0.48 2	120	0.00310	0.134

Note: All compounds are based on the Markush structure shown on the left except for 25 and 26 and PCI34051. PCI34051 is a known HDAC8 selective inhibitor. The Selectivity Index (SI) is calculated as a ratio between the lowest non-HDAC8 IC₅₀ value and HDAC8 IC₅₀ value. ^aCompounds tested at a top concentration of 2 μ M. HDAC2, and 4 IC₅₀ values for 8-10 were >2 μ M. ^bMMH371, and MMH410 were both additionally tested against HDACs 1 and 2 yielding IC₅₀ values >10 µM. •MMH409 and PCI34051 were also re-tested against HDACs (1-7, 9-11) at a top concentration of 10 μ M to obtain a complete selectivity profile (SI = 410). IC₅₀ values for HDAC 1, 2, 4, 5, 7, 9, and 10 were >10 μ M and the IC₅₀ value against HDAC11 was 9.6 µM. ^dCompounds based on the Markush structure to the right. A colour gradient is applied for visualization of potency.

A 3,5-CF₃ substitution on the *N*-iPr benzanilide structure 16, resulted in a potent and most selective HDAC8 inhibitor, **MMH409** (IC₅₀ = 23 nM). A complete HDAC inhibition panel revealed a 410-fold selectivity for HDAC8 (Table 1, S2).

Alternative di-substituted analogs assessed included 2-F, 3-CF₃ (19), 3-CF₃, 4-F (20), 3,5-tBu (7), and 3-OCF₃ (MMH410). In addition to the 3-CF₃ substitution, fluorine substitution in the 2and 4-positions positions (19 and 20, respectively) both resulted

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in lower potency of 119 and 89 nM, respectively. A change in the CF₃ position from the 3-CF₃-substituted analog (4) to 2-CF₃ (12) also resulted in a loss of HDAC8 potency and selectivity. (4; IC₅₀ = 62 nM, 12; IC₅₀ = 823 nM, 4; SI >16, 12; SI ~ 1). The SAR strongly supports the hypothesis that either a mono- (3-) or di-(3,5-) substituted cap-group is a favourable chemotype for improved potency and selectivity against HDAC8. Notably, previous studies have also demonstrated the use of *meta*substituted hydroxamic acid zinc-binding groups to target HDAC8 selectively.²⁵⁻²⁷

Next, to confirm our hypothesis that *N*-alkylation facilitates Lshaped benzanilides and HDAC8 selectivity, we synthesized the unsubstituted *N*-amide derivatives of HDAC8 selective compounds **4**, **7**, and **MMH409**. All analogs lost HDAC8 potency (**11** (~8-fold), **18** (~13-fold) and **22** (29-fold)), and selectivity (**11** (>133-fold), **18** (>127-fold) and **22** (2994-fold)). All analogs showed increased potency against HDAC6, suggesting that *N*-H benzanilides preferentially adopt a linear conformation, more suitable for the HDAC6 tunnel.

Collectively, the benzanilide scaffold achieves HDAC8 potency and selectivity by: 1) incorporation of either a mono-(3-) or (3,5-) di-substituted (R_2) cap group; 2) a conformationally rigid amide motif (*i.e.* sp² carbonyl preferred rather than sp³ sulfonyl, or methylene carbon); 3) restriction of the size of *N*-alkyl group (R_1) on the benzanilide.

In vitro Target Validation and Determining the Kinetic Binding Profile. In order to corroborate the SAR trends in the EMSA activity assay, we assessed the library with an HDAC8 fluorescence polarization (FP) assay using an FP-probe, FITC-M344 (Table S5).²⁸ HDAC8 potency in the FP assay was consistent with the trends observed in the EMSA activity assays, with MMH371, MMH409 and MMH410 all showing nanomolar IC₅₀ values comparable to PCI34051 (PCI34051, FP; IC₅₀ = 133 nM; **MMH371**, FP; IC₅₀ = 59 nM; **MMH409**, FP; IC₅₀ = 110 nM; **MMH410**, FP; IC₅₀ = 72 nM; **Figure 3b**, Table 1, S5). In addition to the FP assessment, a Thermal Shift Assay (TSA) was performed to corroborate HDAC8-target engagement. A shift in HDAC8 melting temperature (T_m) was observed upon dosing with most HDAC8-targeting compounds (Table S6, Figure S4, S5). For example, 2 showed a 9 °C shift in stabilization relative to apo-HDAC8 ($T_m = 37 \text{ °C}$) while other HDAC8 selective inhibitors only showed a 1 °C shift. HDAC8

is known to be one of the most dynamic of HDAC proteins, which may permit binding to different HDAC8 conformations and result in varying degrees of stabilization.^{29,30}

The kinetic binding profile of the benzanilides was investigated using surface plasmon resonance (SPR) with fulllength HDAC8 (Figure 3c). All of the sulfonamide derivatives showed potent equilibrium dissociation constants (K_D) of between 10 and 100 nM. N-alkylated benzanilides, MMH371, MMH409, MMH410, and 19 achieved a K_D lower than 10 nM. Notably, MMH409 demonstrated a K_D of 3.5 nM, comparable to PCI34051 ($K_D = 3.1 \text{ nM}$; Table 1, S3,S4). The entire subset of SAR derivatives displayed similar association rate constants k_a (~10⁻⁵ M⁻¹s⁻¹, **Table S3**). The SPR results suggested that increased binding affinity was primarily due to the decrease in the kinetic dissociation rate, k_d. For instance, the N-Et sulfonamide, 6, had a 30-fold lower k_d than that of the N-iPrbenzanilide, MMH409, despite both compounds containing the same 3,5-CF₃ cap-group (Table S3, S4). All N-iPr-substituted derivatives demonstrated a lower k_d when compared to their N-Et counterparts; 21 and 16 had a 1.8-fold and 2.4-fold lower k_d compared to counterparts 2 and 4, respectively. The Nmethylene cPr derivative, MMH371 also had a marginally lower k_d (2-fold) compared to its ethylated counterpart, 2.

Based on literature precedent, it can be assumed that the difference in *N*-alkyl group bulk (Et, iPr, and methylene cPr) is responsible for inducing a more favorable L-shape and corresponding slower dissociation rate. This also implies that the *N*-iPr and *N*-methylene cPr benzanilide scaffolds tend to have longer drug-target residence times (MRT = $1/k_{off}$) compared to their *N*-Et counterparts (**Table S3**).^{31,32} Another explanation for the decrease in dissociation rates may be due to a conformational change in the binary complex³² after inhibitor binding, leading to a longer residence time. This is in agreement with the statistical fitting of SPR response curves using a two-state binding model³³ of *N*-iPr benzanilides **MMH409**, **MMH410**, and **21** with HDAC8 (**Table S4**, Chi² (RU²) ~ 0.045-0.062).



Figure 3. (a) Structures of **MMH371**, **MMH409**, and **MMH410**; (b) HDAC8 IC₅₀ for **PCI34051**, **MMH371**, **MMH409**, and **MMH410** using **FITC-M344** in a competition-based Fluorescence Polarization (FP) assay²⁸; (c) Logarithmic isoaffinity plot generated from rates of association (k_a) and dissociation (k_d) as determined by Surface Plasmon Resonance (SPR). The plot is divided into regions defined by their dissociation constant ($K_D = k_d/k_a$) (d) HDAC3, 6, 8, and 11 IC₅₀ plots for the three compounds determined by Nanosyn using an activity-based Electrophoretic Mobility Shift Assay (EMSA).

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Structural Studies of Benzanilide Scaffolds. Thus far, our hypothesis was that incorporation of small alkyl groups such as iPr (16, MMH409, MMH410, etc.) and methylene cPr (15, 17, MMH371) impart HDAC8 selectivity via an E conformation (Figure 1b, 2c). X-ray crystal structures of MMH371 and MMH409, confirmed that both molecules had non-linear conformations in the crystallized state (Figure 4b). The structures revealed the close proximity of the aryl groups of the cap- and linker moiety for both MMH371 and MMH409, with the distance between the ortho-hydrogens of the cap-group and linker, being 3.105 Å and 2.995 Å, respectively. The two phenyl motifs of the benzanilide scaffold were approximately perpendicular to each other. In MMH371, the distance between the ortho-protons of one phenyl ring to the other (3.206 Å) was larger than in MMH409, with the bulky tBu group oriented away from the hydroxamic acid substituted phenyl ring. 2D-NOESY experiments of *N*-alkylated HDAC8-selective inhibitors, 16, 17, 20 and MMH409 all showed ¹H-¹H NOE correlations between the protons from the two phenyl rings (Figure 4c, S10) confirming the E conformation is retained in solution (deuterated methanol). For instance, 17 (Figure 4c) showed a strong NOE correlation consistent with the L-shaped configuration. As expected, the N-H benzanilides, 11, 18 and 22, showed no NOE correlations in solution (Figure S11). DFT analysis was performed on the HDAC8 library to correlate the dependence of HDAC8-selectivity on L-shaped conformation. The E-Zequilibrium constant $K_{E/Z}$ (Figure 2c, 4a) was predicted for the complete SAR using Gaussian at the ω B97XD/6-31++G(d,p) level of theory with Polarizable Continuum Model (PCM) water solvent correction by determining the Gibbs free energy of the E (G_E) and Z (G_Z) foldamers (**Table S11**). The E-Z equilibrium constant (K_{*E/Z*}) was calculated by using $\Delta G_{E/Z}$ (Equation 1).

$$G_E - G_Z = -RT ln(K_{E/Z}) \tag{1}$$

A plot of log $K_{E/Z}$ versus the selectivity of HDAC8 molecules (**Figure 5d**) was used to computationally determine the dependence of HDAC8 selectivity on the conformational bias of the HDAC8 SAR library. A log $K_{E/Z} > 1$ represents an *E* conformer population of > 90%. Interestingly, all of the HDAC8 selective molecules with an SI > 8 displayed a log $K_{E/Z} > 1$ indicating a high-preference for the *E* (L-shaped) conformation. The *N*-H benzanilides (1, 11, 22) had >90% *Z* population with the exception of 20, a 3,5-*t*Bu cap group benzanilide (~72% *Z*, **Table S11**).

Although, some of the sulfonamides (3, 5) displayed some selectivity due to their flexible sp³ sulfur atom, their *E-Z*

equilibrium was much lower, while still predominantly preferring the *E* conformation (% E = 56 and 70 %, respectively). Some *N*aryl benzanilides however (9, 10), do not display HDAC8 potency despite their preference for the E conformation as predicted by DFT (~96% and ~95% E, respectively, Table S11). We presume that the greater steric bulk of 9 and 10 may prevent entry into the HDAC8 tunnel, thereby reducing its activity. Similarly, one of the *N*-alkylated transposed amides, 24, although preferring the L-shape (log $K_{E/Z}$ = 1.74, % E = 98%), is not potent against HDAC8 (IC₅₀ = 732 nM). It is hypothesized that transposition of the amide may result in a steric clash between the *N*-alkyl group and the tunnel entrance residues, or, result in a loss of favourable H-bonding interactions with the tunnel entrance residues. X-ray and NMR studies of our benzanilide-based hydroxamic acid pharmacophores confirmed previous conformational studies conducted on the effect of N-alkylation on conformation.^{23,34–39} While HDAC8 selectivity is proposed to be due to the adoption of the L-shape, there are additional factors, such as sterics and intermolecular bonding, that can also contribute to HDAC8 selectivity, or lack thereof.

In Silico Docking Studies. In order to validate the mode of binding of HDAC8 selective inhibitors, in silico computational studies were performed using Schrödinger Maestro 11.9.011 software with Glide within the Small-Molecule Drug Discovery Suite 2019-1.40-42 The HDAC8 inhibitor library was docked with hHDAC8 (PDB: 1W22) and docking scores were reported as a free energy of binding (ΔG_B) in the supplementary section (**Table** S12). All of the N-H benzanilides (1, 18, and 22), with the exception of 11, showed a higher $\Delta G_{\rm B}$ than their N-iPr and Nmethylene cPr HDAC8 selective counterparts, in agreement with the activity assays. Compound 1 had a ΔG_B of -8.097 kcal/mol as compared to its N-methylene cPr HDAC8-selective analogue, **MMH371**, which was calculated to have an $\Delta G_{\rm B}$ of -8.563 kcal/mol. Similarly, 18 and 22, had docking scores of -4.680 kcal/mol and -4.696 kcal/mol, whereas their N-iPr HDAC8 selective counterparts, 21 and MMH409 had a ΔG_B of -4.725 kcal/mol and -7.873 kcal/mol, respectively. The three HDAC8 selective leads MMH371, MMH409 and MMH410 were chosen for studying the docking poses in comparison with the N-H benzanilides 1, 11, 18, and 22. All three inhibitors showed binding to hHDAC8 in their L-shaped conformation, with the cap-groups bound in the HDAC8 L1-L6 basin (Figure 5). This was consistent with previous HDAC8-inhibitor structural studies that indicated that HDAC8-selective inhibitors bind to a unique HDAC8-selective pocket formed by a catalytic tyrosine and L1-L6 loops.²¹ The N-H benzanilides were accommodated within the lysine tunnel in a linear fashion as predicted by DFT analysis without any interaction with the L1-L6 basin.



Figure 4. *N*-alkylation of benzanilide scaffolds facilitate adoption of an *E* or L-shaped conformation. (a) The *E*-*Z* equilibrium of the benzanilide **MMH410**. **MMH410** is 93% *E* as predicted by DFT analysis. The *E*-*Z* equilibrium of **MMH410** depends on R_1 composition; (b) X-ray crystal structures of **MMH371** and **MMH409**; (c) Two-dimensional NOESY experiment on **17**, a potent and selective HDAC8 inhibitor, showing NOE

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correlations consistent with the L-shape; (d) A log $K_{E/Z}$ versus HDAC8 selectivity plot, showing the dependence of HDAC8 selectivity on the *E-Z* benzanilide equilibrium. ΔG of *E-Z* equilibrium was calculated using Density Functional Theory (DFT) analysis of the HDAC8 SAR and used to determine Log $K_{E/Z}$. All values for Selectivity Index (SI), were plotted as the maximum reported value and are likely underestimates of the true HDAC8 selectivity. For example, **17** displayed SI >4.44, and the value 4.44 was used for the plot.

Determining Biological Applicability. Lead compounds, **MMH371, MMH409**, and **MMH410**, were evaluated for cellular permeability by conducting a Parallel Artificial Membrane Permeability Assay (PAMPA) (**Table 2**).⁴³ In the PAMPA assay, methotrexate and testosterone were used as negative and positive controls, respectively. A permeability coefficient ($-Log P_e$) lower than 6 generally defines a compound as having a high permeability profile. All compounds tested displayed a $-Log P_e < 6$ predicting a high permeability profile (**Table 2, S7**). To gauge structural stability, compounds were assessed for stability in human plasma. **MMH371** and **MMH-410** demonstrated good stability in human plasma ($t_{1/2} > 4$ h) with **MMH409** being less stable (~1.5 h, **Table 2**).

Table 2 – Properties of top three HDAC8 selective molecules. In vitro permeability (-Log P_e and % recovery) and stability profiles $(t_{1/2})$ in human plasma.

	PA	АМРА			
Compound ID	-Log P _e	Recovery%	Plasma Half-life t _{1//2} (min) ^a		
MMH371	5.70	102.5	278		
MMH409	5.42	95.4	93		
MMH410	5.80	101.8	266		

^aHalf-life determined from an average of three runs.

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While not fully understood, the disease relevance of aberrant HDAC8 activity has been implicated in advancedstage neuroblastoma.^{3,9,44} Neuroblastoma is a childhood extracranial solid-tumour associated with a lack of differentiation and maturation of neural crest cells, and <50% long-term survival probability of high-risk patients.³ Additionally, patients that undergo chemotherapy suffer from immediate and long-term toxicities, which makes newer therapies with minimal toxicities, such as those targeting cell maturation, highly desirable.^{3,45} The three lead HDAC8 inhibitors were assessed in a neuroblastoma cell (BE(2)-C colony assay and the potency was determined. It was found that **MMH371** (IC₅₀ = 2.1 μ M) was essentially equipotent with known HDAC8 inhibitor, **PCI34051** (IC₅₀ = 3.1 μ M) (**Figure 6a**).

Immunofluorescence imaging in neuroblastoma cells (BE(2)-C) was employed to determine the acetylation levels of the HDAC8 substrate, Structural Maintenance of Chromosomes protein 3 (SMC3), after being dosed with MMH410.3,26,44,46,47 Inhibition of HDAC enzymes typically leads to increased acetylation of their substrates. Immunofluorescence imaging revealed a dose-dependent increase in acetylation levels similar to that of PCI34051 (Figure 6c, S15). The neuroblastoma cells were found to form visible clusters upon addition of MMH410, which may be indicative of a stress response.⁴⁸ Furthermore, Western blot analysis of Ac-SMC3 and Ac-tubulin (HDAC6 substrate) demonstrated in cellulo HDAC8 selectivity over HDAC6 for MMH410 whereas MMH371 did not. Ac-SMC3 showed a dose-dependent increase for MMH371 and MMH410, whereas Ac-tubulin remained constant even at a concentration of 10 µM for MMH410 but showed a dosedependent increase for MMH371 (Figure 6d). MMH410 displayed a greater selectivity for HDAC8 in cells than MMH371.

MMH371 and MMH410 were selected for further stability profiling to determine their application as in vivo probes. CD-1 mice were given HDAC8 inhibitors interperitoneally (IP), at a single dose (100 mg/kg), and blood was collected at multiple time points over a 12 h period post-administration. MMH371 exhibited a modest increase in $t_{1/2}$ as compared to PCI34051 (4.1 h vs. 3.3 h, respectively; Figure 6b). While the maximal concentration, Cmax, (2893 ng/mL, 7.88 µM) was significantly lower than that of **PCI34051** ($C_{max} = 16967 \text{ ng/mL}, 57.6 \mu \text{M}$), their overall exposures were comparable (MMH371, AUClast = 5851 h*ng/mL; PCI34051, AUC_{last} = 6946 h*ng/mL; Table S9, S10). MMH410 was shown to exhibit a $t_{1/2}$ comparable to PCI34051, but exhibited a significantly lower maximal concentration and exposure ($t_{1/2} = 2.7$ h; $C_{max} = 440$ ng/mL, 1.14 μ M; AUC_{last} = 1653 h*ng/mL; Figure 6b, table S8). MMH371 and MMH410 both demonstrated a greater in vivo mean residence time (MRT) over PCI34051 (MMH371, MRT = 2.77 h; MMH410, MRT = 3.00 h; PCI23051, MRT = 0.45 h; Table S8, S9, S10).

Conclusion and Future Remarks. HDAC selective inhibitors are hypothesized to prevent or lower acute therapy related toxicities associated with pan-HDAC inhibitors. One approach to achieving selectivity has been to exploit the topological differences between various HDACs.



Figure 5. Computational docking studies to determine mode of action of *N*-alkylated versus *N*-H benzanilide scaffolds against hHDAC8 (PDB 1W22). (a) Three HDAC8 selective inhibitors (**MMH371**, **MMH409**, and **MMH410**) overlaid and docked onto hHDAC8. The *N*-alkylated benzanilide scaffolds adopt an L-shape where the cap-groups dock in the L1-L6 basin. (b) The *N*-H benzanilide scaffolds (**1**, **11**, **18**, **22**) preferentially adopt a linear conformation that is unable to interact with the L1-L6 basin.



Figure 6. (a) Colony assay of neuroblastoma (BE(2)-C) cells with three HDAC8 selective inhibitors (**MMH371**, **MMH409**, **MMH410**). (b) *In vivo* half-life of **MMH371** ($t_{1/2} = 4.1$ h), **MMH410** ($t_{1/2} = 2.7$ h) and **PCI34051** ($t_{1/2} = 3.3$ h) in CD-1 mice dosed at 100 mg/kg, measured as a function of time (h) versus concentration (ng/mL) of MMH410 in blood. (c) Immunofluorescence images of DAPI (blue) nuclear staining and Ac-SMC3 (red) showing a dose dependent increase of Ac-SMC3 upon addition of MMH410 and PCI34051. (d) Western blot analyses of Ac-SMC3 and Ac-tubulin in BE(2)-C cells treated with **MMH410** and **MMH371** (DMS0, 2.5, 5, 7.5, and 10 μ M).

In this study, we have developed a series of potent and selective HDAC8 inhibitors based on the conformational restrictions afforded by *N*-alkylated benzanilides. We have shown the following: (1) A thorough SAR demonstrating the structural determinants required for HDAC8 selectivity; (2) *N*-alkylation of benzanilides affords an L-shaped conformation required for HDAC8 selectivity; (3) the biological utility of L-shaped

benzanilides as potential chemotherapeutics and biological probes. The current FDA approved HDAC inhibitors have paninhibition profiles due to their predominantly linear conformation and/or conformational degrees of freedom which permit interaction with a variety of HDAC catalytic Lys tunnels. Utilizing conformationally rigid scaffolds for exploiting tunnel shapes and specific surface topologies is a promising strategy for attaining HDAC protein-selectivity. The conformationally constrained benzanilide foldamer-based hydroxamic acid pharmacophore presents an interesting pharmacophore for achieving selective targeting of HDAC8.

EXPERIMENTAL DETAILS

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General Methods. All solvents and chemicals were purchased from Sigma-Aldrich, Combi-Blocks, Alfa-Aesar, Caledon Laboratory Chemicals, and Oakwood Chemical. Anhydrous solvents such as methanol, acetonitrile, and DMF were used from anhydrous-grade Sure-Seal bottles without further drying treatments. THF, DCM, DCE, and chloroform were dispensed using the MBraun-Solvent Dispensary System (MB-SPS). All reactions under non-aqueous conditions were conducted in ovendried glassware. Reaction progress was monitored by Thin Layer Chromatography (TLC) and visualized by Ultraviolet light (UV, 254/365 nm). Microwave irradiation was carried out in an Initiator Sixty Biotage apparatus at a high-absorption setting at 2.54 GHz.

18 A 400 MHz Bruker, 600 MHz Agilent DD2, and 700 MHz 19 Agilent DD2 NMR spectrometer equipped with an HCN cryoprobe, were used to collect ¹H, ¹³C, and ¹⁹F NMR 20 spectra. Two dimensional 1H-1H Nuclear Overhauser 21 Spectroscopy (NOESY) studies were conducted on the 700 MHz 22 Varian at room temperature in deuterated methanol. All 23 deuterated solvents for NMR were purchased from either Sigma-24 Aldrich or Cambridge Isotope Laboratories. Chemical shifts (δ) 25 are reported in parts per million after calibration to the residual 26 solvent peak from the deuterated solvent. Coupling constants (J) 27 are reported in Hertz (Hz) and multiplicities are reported as 28 follows: singlet (s), doublet (d), triplet (t), quartet (q), pentet (p), 29 sextet (sex), septet (sep), multiplet (m), broad (br), and the 30 combination of these is represented as a combination of their abbreviations (eg. doublet of triplet (dt)). Low-resolution mass 31 spectrometry (LRMS) using Waters LC-MS equipped with a 32 Micromass ZQ MS in both positive and negative ESI modes was 33 used to confirm the success of the reaction along with NMR. 34 High-resolution mass spectrometry was performed for all final 35 compounds at the Advanced Instrumentation for Molecular 36 Structure (AIMS) at the University of Toronto using the Agilent 37 6538 Ultra High Definition (UHD) Accurate-Mass Q-TOF 38 LC/MS System. 39

Column Chromatography was conducted using a gradient of ethyl acetate and hexane on manually packed SNAP 10, 25, 50, 100 g cartridges with 40-60 µm silica gel on a Biotage system. Compounds were wet-loaded by dissolving in minimal DCM or dry-loaded by adsorbing onto silica gel. Flow rates 12, 25, and 50 mL/min were used for cartridges SNAP 10, 25, 50 and 100 respectively. Compounds were detected by measuring absorbance at 254 nm and 280 nm. Preparatory High-Performance Liquid Chromatography (PrepHPLC) was carried out on all final compounds to ensure a final purity >95%. The PrepHPLC was equipped with an XSelect CSH Phenyl-Hexyl C18 column (30 x 250mm), a Waters 2489 Dual Absorbance UV-Visible Detector, a Waters 2545 Quaternary Gradient Module, and a Waters 2707 Autosampler. All compounds were detected using 254/280 nm UV-light and purified using a linear gradient of a mixture of H₂O and 0.1 % formic acid (A), and acetonitrile and 0.1% formic acid (B) using two different methods as indicated. The methods are illustrated below as the linear gradient of B depicted by its % (v/v) composition and its time duration at a flow rate of 50 mL/min.

Method 1: $2\% \rightarrow 100\%$ over 50 minutes then 100% for 10 minutes

Method 2: 5% \rightarrow 100% over 50 minutes then 100% for 10 minutes

Analytical HPLC was used to determine the purity of final compounds after PrepHPLC. It was carried out using a Phenomenex Luna 5.0 μ m C18 4.6 mm x 150 mm column fitted on a Hewlett Packard Series 1100 analytical HPLC at a flow rate of 1.2 mL/min using method 1 and 2 as indicated. Compound purity is reported as the relative area of the compound peak to all other peaks on the chromatogram. The retention time (t_R) for each peak is reported in minutes. The purity of all compounds reported is >95% as determined by the analytical HPLC.

X-ray diffraction data were collected on a Bruker Kappa APEX-DUO diffractometer using a Copper ImuS (microsource) tube with multi-layer optics and were measured using a combination of f scans and w scans. The data were processed using APEX3 3 (Bruker, 2018). Absorption corrections were carried out using SADABS.⁴⁹ The structures were solved using SHELXT⁵⁰ and refined using SHELXL-2018⁵¹ for full-matrix least-squares refinement that was based on F^2 .

Synthesis of 2-9, 11 (skipped STAB step), 12, 15-17, 19-21, MMH371, MMH409, and MMH410



Synthesis of 13, and 14

Scheme 4



Note: First step was skipped for 23.

Esterification of 4-aminobenzoic acid

To a solution of 0.1 mmol/mL solution of 4-aminobenzoic acid in ethanol was added sulfuric acid to make a sulfuric acid:ethanol ratio of 1:20. The reaction was heated to 90 °C overnight then diluted with ethyl acetate and extracted with 0.5 M NaOH followed by a brine wash. The solution was concentrated *in vacuo* to give a peachy-flaky solid.

Reductive amination of aryl amines

Aryl amines from schemes 1,4 and 5 were dissolved in DCE to make a 0.1 mmol/mL solution. Subsequently, the respective aldehyde (1.5 eq.), acetic acid (1.5 eq.), sodium triacetoxyborohydride (STAB, 3 eq.) and one scoop of $MgSO_4$

were added concurrently. The reaction was stirred for 4 hours and concentrated *in vacuo* and adsorbed onto silica gel. The crude mixture with silica was dry-loaded for column chromatography which afforded the desired intermediates.

Acylation/sulfonylation of aryl amines

To a solution of the respective sulfonyl or acyl chloride (1.5 to 2 eq.) in dichloromethane (0.1 mmol of aryl amine/mL) was added diisopropyl amine (DIPEA, 4 eq.) which produced a smoke suspended above the solution. Then aryl amine was added, and the solution was left to stir overnight. The solution was dried *in vacuo* with silica gel for dry-loading on a column for column chromatography which yielded the desired intermediates.

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Deprotection of methyl/ethyl esters

To a solution of THF, methanol and MilliQ H₂O in a ratio of 3:1:1 containing the methyl/ethyl ester (0.1 mmol/mL) was added lithium hydroxide (3 eq.). The mixture was left to stir overnight. The reaction was complete in approximately 16 hours as indicated by TLC and LRMS. The solution was diluted with ethyl acetate and washed twice with 1 M HCl followed by a brine wash and drying with magnesium sulfate. Drying *in vacuo* yielded relative pure intermediates that were used directly for the subsequent step without further purification.

10 Synthesis of THP-protected hydroxamic acids

The aryl carboxylic acid intermediates were converted to their acyl chloride intermediates by the use of the Vilsmeier-Haack reagent generated in situ. The aryl carboxylic acid intermediate was added to a solution of THF (0.1 mmol/mL), followed by the addition of no more than two drops of DMF, and oxalyl chloride (5 eq.) which caused immediate effervescence. The solution was purged with nitrogen gas until the effervescence halted. The reaction was then stirred for 1 hour under inert atmosphere. The solution containing the acyl chloride was dried immediately in vacuo (15 min rotary evaporator followed by 15 min highvacuum) with minimal exposure to air and moisture. The crude mixture was re-dissolved in THF, and DIPEA (4 eq.) was added, resulting in a layer of smoke above the solution. Subsequently, O-(Tetrahydro-2H-pyran-2-yl)hydroxylamine (2 eq.) was added and the mixture was left to stir overnight. Drying in vacuo followed by column chromatography yielded the desired THPprotected hydroxamic acid.

Deprotection of the THP-protected hydroxamic acids

The THP-protected hydroxamic acid was dissolved in 4 M hydrochloric acid in dioxanes (0.1 mmol/mL dioxane) and left to stir for 4 hours. The reaction was monitored by TLC and LRMS which indicated a complete conversion. The solution was dried *in vacuo* and the crude mixture was purified by Preparative HPLC.

Benzyl protection of 4-aminobenzoic acid

4-aminobenzoic acid was dissolved in DMF (2 mmol/mL) and stirred with Cs_2CO_3 (1.1 eq.) for 20 minutes. Benzyl bromide (1 eq.) was added subsequently, and the solution was stirred overnight. The mixture was diluted with ethyl acetate and washed thrice with water followed by a brine wash. Drying by MgSO₄ and concentrating *in vacuo*, followed by column chromatography afforded two products: a peachy mono-Obenzylated product, and another di-N,O-benzylated product (used for scheme 3).

Benzyl deprotection of benzyl ester intermediates

To a solution of benzyl ester in THF (0.1mmol/mL) was added methanol to make a THF:methanol ratio of 2:1. The solution was purged with nitrogen for 15 minutes before the addition of 10% palladium on carbon (0.1 eq.) followed by another 5 minute nitrogen purge. Hydrogen gas was bubbled through the solution for 1 hour using a long-needle submerged in the solution. TLC indicated the reaction was complete and the reaction was diluted with ethyl acetate and filtered through celite.

3-(tert-butyl)-N-(4-(hydroxycarbamoyl)phenyl)benzamide

(1). ¹H NMR (400 MHz, DMSO- d_6) δ 10.77 – 10.71 (br, 1H), 8.70 – 8.61 (m, 1H), 7.51 – 7.19 (m, 6H), 6.65 (d, J = 8.4 Hz, 2H), 1.19 (s, 9H). ¹³C NMR (101 MHz, Methanol- d_4) δ 151.2, 144.1, 131.2, 128.0, 127.2, 126.8, 126.2, 125.4, 125.2, 124.6, 119.5, 116.7, 34.1, 30.1. HRMS (ESI+) m/z calcd for $[C_{18}H_{20}N_2O_3]^+$: 312.37, found: 312.20. HPLC $t_R = 11.586 \min (97.6\% \text{ purity})$.

3-(tert-butyl)-N-ethyl-N-(4-

(hydroxycarbamoyl)phenyl)benzamide (2).¹H NMR (700 MHz, Methanol- d_4) δ 7.64 (d, J = 8.5 Hz, 2H), 7.31 (dt, J = 7.5, 1.8 Hz, 1H), 7.22 – 7.14 (m, 5H), 4.00 (q, J = 7.1 Hz, 2H), 1.20 (t, J = 7.1 Hz, 3H), 1.10 (s, 9H). ¹³C NMR (176 MHz, Methanol- d_4) δ 174.1, 168.1, 153.2, 148.7, 137.9, 133.1, 130.5, 130.3, 130.3, 129.3, 128.3, 128.2, 47.7, 36.7, 32.7, 14.3. HRMS (ESI+) m/z calcd for [C₂₀H₂₅N₂O₃]+: 341.18652, found: 341.18632, HPLC $t_R = 19.045$ min (>99% purity).

4-((3-(tert-butyl)-N-ethylphenyl)sulfonamido)-N-

hydroxybenzamide (3). ¹H NMR (400 MHz, Acetonitrile-*d*₃) δ 7.75 – 7.70 (m, 3H), 7.54 – 7.47 (m, 2H), 7.41 (s, 1H), 7.15 (d, *J* = 8.3 Hz, 2H), 3.63 (q, *J* = 7.1 Hz, 2H), 1.26 (s, 9H), 1.03 (t, *J* = 7.1 Hz, 3H). ¹³C NMR (101 MHz, Acetonitrile-*d*₃) δ 164.6, 152.5, 142.1, 137.4, 130.2, 129.0, 128.5, 127.6, 124.5, 124.2, 117.4, 45.0, 34.6, 30.2, 13.3. HRMS (ESI+) *m/z* calcd for $[C_{19}H_{25}N_2O_4S]$ +: 377.1530, found: 377.1534. HPLC *t*_R = 21.260 min (99.4%).

N-ethyl-N-(4-(hydroxycarbamoyl)phenyl)-3-

(trifluoromethyl)benzamide (4). ¹H NMR (400 MHz, Acetonitrile- d_3) δ 7.67 – 7.58 (m, 4H), 7.53 (t, J = 7.4 Hz, 1H), 7.45 – 7.38 (m, 1H), 7.23, (d, J = 8.5 Hz, 2H), 4.77 (br, 2H), 3.97 (q, J = 7.1 Hz, 2H), 1.19 (t, J = 7.1 Hz, 3H). ¹³C NMR (101 MHz, Acetonitrile- d_3) δ 168.2, 145.8, 137.5, 132.0, 128.9, 128.3, 128.2, 127.9, 126.8, 126.1, 125.3, 117.3, 45.0, 12.2. ¹⁹F NMR (376 MHz, Acetonitrile- d_3) δ -63.42 (s, 3F). HRMS (ESI+) m/z calcd for [C17H16F3N2O3]+: 353.1108, found: 353.1109. HPLC t_R = 18.184 min (>99% purity).

4-((N-ethyl-3-(trifluoromethyl)phenyl)sulfonamido)-N-

hydroxybenzamide (5). ¹H NMR (400 MHz, Acetone-*d*₆) δ 10.87 (br, 1H), 8.07 (d, *J* = 7.2 Hz, 1H), 7.94 – 7.81 (m, 5H), 7.26 (d, *J* = 8.6 Hz, 2H), 3.76 (q, *J* = 7.0 Hz, 2H), 1.09 (t, *J* = 7.1 Hz, 3H). ¹³C NMR (101 MHz, Acetone-*d*₆) δ 205.3, 141.4, 139.6, 131.8, 131.3, 131.0, 130.7, 129.6, 128.6, 127.8, 126.1, 124.1, 45.3, 13.4. ¹⁹F NMR (376 MHz, Acetone-*d*₆) δ -63.43 (s, 3F). HRMS (ESI+) *m/z* calcd for [C₁₆H₁₆F₃N₂O₄S]+: 389.0775, found: 389.0770. HPLC $t_{\rm R}$ = 20.241 min (>99% purity).

4-((N-ethyl-3,5-bis(trifluoromethyl)phenyl)sulfonamido)-N-

hydroxybenzamide (6). ¹H NMR (400 MHz, Acetone-*d*₆) δ 10.90 (br, 1H), 8.43 (s, 1H), 8.14 – 8.09 (m, 2H), 7.90 (d, *J* = 8.6, 2H), 7.32 (d, *J* = 8.5, 2H), 3.81 (q, *J* = 7.1 Hz, 2H), 1.11 (t, *J* = 7.1 Hz, 3H). ¹³C NMR (101 MHz, Acetone-*d*₆) δ 205.3, 141.0, 132.3, 128.9, 128.8, 127.9, 126.7, 126.3, 124.1, 121.4, 45.7, 13.3. ¹⁹F NMR (376 MHz, Acetone-*d*₆) δ -63.52 (s, 6F). HRMS (ESI+) *m/z* calcd for [C₁₇H₁₅F₆N₂O₄S]+: 457.0649, found: 457.0644. HPLC *t*_R = 22.030 min (>99% purity).

3,5-di-tert-butyl-N-ethyl-N-(4-

(hydroxycarbamoyl)phenyl)benzamide (7). ¹H NMR (400 MHz, Acetone- d_6) δ 10.56 (s, 1H), 7.70 (d, J = 8.5 Hz, 2H), 7.37 (t, J = 1.9 Hz, 1H), 7.20 (d, J = 8.5 Hz, 2H), 7.18 (d, J = 1.8 Hz, 2H), 3.99 (q, J = 7.1 Hz, 2H), 1.22 – 1.14 (m, 21H). ¹³C NMR (101 MHz, Acetone- d_6) δ 205.9, 170.0, 150.1, 146.9, 135.5, 127.9, 127.6, 123.4, 123.2, 117.0, 44.7, 34.4, 30.5, 12.4. HRMS (ESI+) m/z calcd for [C₂₄H₃₃N₂O₃]+: 397.2486, found: 397.2487. HPLC $t_{\rm R} = 22.814$ min (>99% purity).

N-(3,5-di-tert-butylbenzyl)-N-(4-

(hydroxy carbamoyl) phenyl) - 3 - (trifluoromethyl) benzamide

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(8). ¹H NMR (400 MHz, Acetone- d_6) δ 7.76 – 7.58 (m, 5H), 7.52 - 7.43 (m, 1H), 7.37 (s, 1H), 7.22 - 7.20, 7.12 (m, 4H), 5.24 (s, 2H), 1.28 (s, 18H). ¹³C NMR (101 MHz, Acetone- d_6) δ 205.3, 168.3, 150.7, 145.9, 137.5, 136.4, 132.3, 130.2, 129.9, 129.5, 129.0, 127.9, 127.7, 126.3, 125.4, 122.4, 121.0, 53.5, 34.4, 30.8. ¹⁹F NMR (376 MHz, Acetone- d_6) δ -63.35 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{30}H_{34}F_{3}N_{2}O_{3}]+: 527.2516$, found: 527.2518. HPLC $t_{\rm R} = 28.483 \text{ min}$ (>99% purity).

N-(4-(hydroxycarbamoyl)phenyl)-3-(trifluoromethyl)-N-(3-

(trifluoromethyl)benzyl)benzamide (9). ¹H NMR (700 MHz, Methanol- d_4) δ 7.66 (s, 1H), 7.60 – 7.56 (m, 6H), 7.54 (d, J = 7.7Hz, 1H), 7.48 (t, J = 7.7 Hz, 1H), 7.41 (t, J = 7.8 Hz, 1H), 7.14 $(d, J = 8.6 \text{ Hz}, 2\text{H}), 5.28 (s, 2\text{H}).^{13}\text{C NMR} (176 \text{ MHz}, \text{Methanol-}$ d_4) δ 172.3, 168.0, 147.7, 140.6, 139.0, 134.6, 133.7, 132.9, 131.8, 131.4, 130.6, 130.5, 129.1, 129.0, 127.9, 127.6, 127.0, 126.8, 126.0, 125.5, 55.3. ¹⁹F NMR (658 MHz, Methanol-d₄) δ -64.24 (s, 3F), -64.50 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{23}H_{17}F_6N_2O_3]$ +: 483.1147, found: 483.1144. HPLC t_R = 23.546 min (>99% purity).

N-benzyl-N-(4-(hydroxycarbamoyl)phenyl)-3-

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(trifluoromethyl)benzamide (10). ¹H NMR (700 MHz, Methanol- d_4) δ 7.66 (s, 1H), 7.60 – 7.50 (m, 4H), 7.36 (t, J = 7.8 Hz, 1H), 7.28 (d, J = 7.3 Hz, 2H), 7.24 (t, J = 7.3 Hz, 2H), 7.20 (t, J = 7.3 Hz, 1H), 7.10 (d, J = 8.4 Hz, 2H), 5.18 (s, 2H).NMR (176 MHz, Methanol-d₄) δ 172.2, 168.1, 147.9, 139.3, 139.2, 134.6, 133.5, 132.7, 131.4, 130.9, 130.8, 130.7, 130.4, 130.0, 129.0, 127.9, 126.3, 55.8. 19F NMR (658 MHz, Methanol d_4) δ -64.22 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{22}H_{18}F_{3}N_{2}O_{3}]$ +: 415.1273, found: 415.1275. HPLC t_{R} = 22.033 min (>99% purity).

N-(4-(hydroxycarbamoyl)phenyl)-3-

29 (trifluoromethyl)benzamide (11). ¹H NMR (700 MHz, 30 Methanol- d_4) δ 8.24 (s, 1H), 8.19 (d, J = 7.8 Hz, 1H), 7.87 (d, J =7.7 Hz, 1H), 7.84 (d, J = 8.7 Hz, 2H), 7.76 (d, J = 8.7 Hz, 2H), 31 7.71 (t, J = 7.8 Hz, 1H). ¹³C NMR (176 MHz, Methanol- d_4) δ 32 168.9, 168.4, 144.3, 138.3, 133.6, 133.3, 131.9, 130.7, 130.4, 33 130.2, 126.9, 126.6, 122.8. ¹⁹F NMR (658 MHz, Methanol-d₄) δ 34 -64.22 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{15}H_{12}F_{3}N_{2}O_{3}]+:$ 35 325.0795, found: 325.0794. HPLC $t_{\rm R} = 10.522$ min (>99%) 36 purity). 37

N-ethyl-N-(4-(hydroxycarbamoyl)phenyl)-2-

38 (trifluoromethyl)benzamide (12). ¹H NMR (700 MHz, 39 Methanol- d_4) δ 7.61 (d, J = 8.47 Hz, 2H), 7.59 (d, J = 7.7 Hz, 1H), 40 7.40 - 7.33 (m, 2H), 7.26 - 7.24 (m, 3H), 3.99 (br, 2H), 1.22 (t, J 41 = 7.2 Hz, 3H). ¹³C NMR (176 MHz, Methanol- d_4) δ 171.0, 168.1, 42 147.1, 137.1, 134.1, 134.0, 131.9, 131.4, 130.9, 130.5, 129.7, 43 129.0, 126.6, 47.1, 14.1. ¹⁹F NMR (658 MHz, Methanol- d_4) δ -44 59.94 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{17}H_{16}F_{3}N_{2}O_{3}]+$: 353.1108, found: 353.1107. HPLC $t_{\rm R} = 9.442 \min (>99\% \text{ purity})$. 45

N-hvdroxy-4-(N-(3-

(trifluoromethyl)benzyl)acetamido)benzamide (13). ¹H NMR 47 $(700 \text{ MHz}, \text{Methanol-}d_4) \delta 7.76 (d, J = 8.4 \text{ Hz}, 2\text{H}), 7.53 (d, J =$ 48 7.2 Hz, 1H), 7.49 (s, 1H), 7.48 – 7.43 (m, 2H), 7.23 (d, J = 8.449 Hz, 2H), 5.01 (s, 2H), 1.92 (s, 3H). ¹³C NMR (176 MHz, 50 Methanol- d_4) δ 174.0, 168.2, 147.6, 141.0, 134.7, 134.6, 133.1, 51 131.7, 130.9, 130.7, 127.5, 126.6, 126.0, 54.4, 23.9. ¹⁹F NMR 52 (658 MHz, Methanol- d_4) δ -64.21 (s, 3F). HRMS (ESI+) m/z53 calcd for $[C_{17}H_{16}F_{3}N_{2}O_{3}]$ +: 353.1108, found: 353.1108. HPLC t_{R} 54 = 18.616 min (98.3% purity). 55

4-(N-(3,5-bis(trifluoromethyl)benzyl)acetamido)-N-

hydroxybenzamide (14). ¹H NMR (700 MHz, Methanol- d_4) δ 7.84 (s, 1H), 7.84 – 7.77 (m, 4H), 7.26 (d, J = 8.5 Hz, 2H), 5.09 (s, 2H), 1.94 (s, 3H). ¹³C NMR (176 MHz, Methanol- d_4) δ 174.2, 168.1, 147.4, 143.1 134.9, 134.1, 131.4, 131.1, 130.6, 125.9, 123.6, 54.1, 23.8. ¹⁹F NMR (658 MHz, Methanol-d₄) δ -64.44 (s, 6F). HRMS (ESI+) m/z calcd for $[C_{18}H_{15}F_6N_2O_3]$ +: 421.0979, found: 421.0978. HPLC $t_{\rm R} = 20.187 \min(98.6\% \text{ purity})$.

N-(cyclopropylmethyl)-N-(4-(hydroxycarbamoyl)phenyl)-3-

(trifluoromethyl)benzamide (15). ¹H NMR (700 MHz, Methanol- d_4) δ 7.67 (d, J = 8.5 Hz, 2H), 7.61 (s, 1H), 7.56 (d, J =7.9 Hz, 1H), 7.52 (d, J = 7.8 Hz, 1H), 7.39 (t, J = 7.8 Hz, 1H), 7.29 (d, J = 8.5 Hz, 2H), 3.84 (d, J = 7.2 Hz, 2H), 1.11 - 1.04 (m, J = 7.2 Hz), 1.11 - 1.04 (m,1H), 0.48 - 0.41 (m, 2H), 0.18 - 0.12 (m, 2H). ¹³C NMR (176) MHz, Methanol-d₄) δ 172.1, 168.2, 148.5, 139.6, 134.4, 133.6, 132.7, 131.3, 131.0, 130.5, 128.7, 127.7, 126.3, 57.1, 11.9, 5.5. ¹⁹F NMR (658 MHz, Methanol-d₄) δ -64.38 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{19}H_{18}F_3N_2O_3]$ +: 379.1262, found: 379.1259. HPLC $t_{\rm R}$ = 21.253 min (>99% purity).

N-(4-(hydroxycarbamoyl)phenyl)-N-isopropyl-3-

(trifluoromethyl)benzamide (16). ¹H NMR (700 MHz, Methanol- d_4) δ 7.66 (d, J = 8.3 Hz, 2H), 7.57 (s, 1H), 7.50 (t, J =7.7 Hz, 2H), 7.36 (t, J = 7.9 Hz, 1H), 7.27 (d, J = 8.3 Hz, 2H), 4.99 (br, 1H), 1.23 (d, J = 6.9 Hz, 6H). ¹³C NMR (176 MHz, Methanol- d_4) δ 172.3, 168.1, 144.5, 140.5, 134.4, 133.9, 133.4, 132.6, 131.2, 130.1, 128.3, 127.2, 126.4, 51.3, 22.4. ¹⁹F NMR (658 MHz, Methanol-d₄) δ -64.37 (s, 3F). HRMS (ESI+) m/z calcd for $[C_{18}H_{18}F_{3}N_{2}O_{3}]$ +: 367.1264, found: 367.1263. HPLC t_{R} = 20.680 min (>99% purity).

3-(tert-butyl)-N-(cyclopropylmethyl)-N-(4-

(hydroxycarbamoyl)phenyl)benzamide (MMH371). ¹H NMR $(700 \text{ MHz}, \text{Methanol-}d_4) \delta 7.64 \text{ (d}, J = 8.6 \text{ Hz}, 2\text{H}), 7.30 \text{ (dt}, J =$ 7.6, 1.7 Hz, 1H), 7.23 (d, J = 8.6 Hz, 2H), 7.22 – 7.19 (m, 2H), 7.17 (t, J = 7.6 Hz, 1H), 3.84 (d, J = 7.1 Hz, 2H), 1.11 (s, 9H), 1.09 - 1.05 (m, 1H), 0.46 - 0.42 (m, 2H), 0.14 (q, J = 5.2 Hz, 2H).¹³C NMR (176 MHz, Methanol-*d*₄) δ 174.3, 168.1, 153.2, 149.3, 137.9, 133.1, 130.8, 130.28, 130.27, 129.2, 128.3, 128.2, 57.0, 36.7, 32.7, 12.0, 5.4. HRMS (ESI+) *m/z* calcd for [C₂₂H₂₇N₂O₃]+: 367.1943, found: 367.2016. HPLC $t_{\rm R} = 27.958$ min (>99%) purity).

3,5-di-tert-butyl-N-(cyclopropylmethyl)-N-(4-

(hydroxycarbamoyl)phenyl)benzamide (17). ¹H NMR (700 MHz, Methanol- d_4) δ 7.66 (d, J = 8.3 Hz, 2H), 7.33 (s, 1H), 7.22 (d, J = 8.1 Hz, 2H), 7.14 (s, 2H), 3.85 (d, J = 7.2 Hz, 2H), 1.15 (s, 18H), 0.43 (q, J = 7.9 Hz, 2H), 0.14 (q, J = 5.2 Hz, 2H). ¹³C NMR (176 MHz, Methanol-d₄) δ 174.6, 168.0, 153.0, 149.6, 137.3, 132.9, 130.7, 130.3, 126.1, 125.8, 57.0, 36.9, 32.9, 12.0, 5.5. HRMS (ESI+) m/z calcd for $[C_{26}H_{35}N_2O_3]$ +: 423.2642, found: 423.2636. HPLC $t_{\rm R}$ = 34.486 min (>99% purity).

3.5-di-tert-butyl-N-(cvclopropylmethyl)-N-(4-

(hydroxycarbamoyl)phenyl)benzamide (18). ¹H NMR (700 MHz, Methanol- d_4) δ 7.85-7.80 (m, 2H), 7.79 – 7.73 (m, 4H), 7.69 - 7.65 (m, 1H), 1.37 (s, 18H). ¹³C NMR (176 MHz, Methanol- d_4) δ 171.3, 169.0, 153.8, 144.6, 136.8, 130.1, 130.1, 128.6, 124.2, 122.9, 37.2, 33.0. HRMS (ESI+) m/z calcd for $[C_{22}H_{29}N_2O_3]$ +: 369.2173, found: 369.2172. HPLC $t_R = 35.032$ min (>99% purity).

N-(4-(hydroxycarbamoyl)phenyl)-N-isopropyl-3,5-

bis(trifluoromethyl)benzamide (MMH409). ¹H NMR (700 MHz, Methanol- d_4) δ 7.85 (s, 2H), 7.80 (s, 1H), 7.70 (d, J = 8.2

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Hz, 2H), 7.33 (d, J = 8.2 Hz, 2H), 5.02 (br, 1H), 1.24 (d, J = 7.1Hz, 6H). ¹³C NMR (176 MHz, Methanol- d_4) δ 170.3, 167.9, 144.2, 141.9, 134.7, 133.6, 133.5, 131.0, 130.3, 125.5, 125.1, 51.4, 22.3. ¹⁹F NMR (658 MHz, Methanol- d_4) δ -64.52 (s, 6F). HRMS (ESI+) m/z calcd for $[C_{19}H_{17}F_6N_2O_3]$ +: 435.1138, found: 435.1131. HPLC $t_{\rm R} = 28.875 \text{ min} (>99\% \text{ purity}).$

N-(4-(hydroxycarbamoyl)phenyl)-N-isopropyl-3-6

(trifluoromethoxy)benzamide (MMH410). ¹H NMR (700 MHz, Methanol- d_4) δ 7.66 (d, J = 8.4 Hz, 2H), 7.31 – 7.22 (m, 4H), 7.15 (s, 1H), 7.12 (d, J = 7.7 Hz, 1H), 4.97 (br, 1H), 1.22 (d, J = 6.9 Hz, 6H). ¹³C NMR (176 MHz, Methanol- d_4) δ 172.1, 10 168.1, 151.0, 144.5, 141.6, 134.4, 133.3, 132.2, 130.1, 129.1, 11 124.3, 123.0, 122.9, 51.3, 22.4. 19F NMR (658 MHz, Methanol-12 d_4) δ -59.61 (s, 3F). HRMS (ESI+) m/z calcd for 13 $[C_{18}H_{18}F_{3}N_{2}O_{4}]$ +: 383.1211, found: 383.1209. HPLC t_{R} = 18.415 14 min (>99% purity).

15 2-fluoro-N-(4-(hydroxycarbamoyl)phenyl)-N-isopropyl-3-

16 (trifluoromethyl)benzamide (19). ¹H NMR (700 MHz, 17 Methanol- d_4) δ 7.64 (d, J = 8.1 Hz, 2H), 7.52 (dt, J = 28.4, 7.1 Hz, 2H), 7.27 (d, J = 8.1 Hz, 2H), 7.17 (t, J = 7.8 Hz, 1H), 5.04 18 (h, J = 6.7 Hz, 1H), 1.22 (d, J = 7.0 Hz, 6H). ¹³C NMR (176 MHz, 19 Methanol- d_4) δ 168.0, 167.7, 157.9, 156.4, 143.2, 135.7, 135.0, 20 133.0, 130.2, 130.0, 126.9, 124.9, 120.5, 50.8, 22.3. ¹⁹F NMR 21 $(658 \text{ MHz}, \text{Methanol}-d_4) \delta - 63.05 \text{ (d}, J = 12.6 \text{ Hz}, 3\text{F}), -118.30 - 63.05 \text{ (d}, J = 12.6 \text{ Hz}, 3\text{F})$ 22 -118.44 (m, 1F). HRMS (ESI+) m/z calcd for $[C_{18}H_{17}F_4N_2O_3]$ +: 23 385.1170, found: 385.1171. HPLC $t_{\rm R} = 18.324$ min (>99%) 24 purity). 25

4-fluoro-N-(4-(hydroxycarbamoyl)phenyl)-N-isopropyl-3-

26 (trifluoromethyl)benzamide (20). ¹H NMR (700 MHz, 27 Methanol- d_4) δ 7.68 (d, J = 8.4 Hz, 2H), 7.62 (d, J = 6.3 Hz, 1H), 28 7.56 – 7.52 (br, 1H), 7.27 (d, J = 8.4 Hz, 2H), 7.13 (t, J = 9.5 Hz, 29 1H), 4.97 (br, 1H), 1.23 (d, J = 6.9 Hz, 6H). ¹³C NMR (176 MHz, 30 Methanol-d₄) δ 171.2, 168.1, 144.6, 136.9, 136.3, 134.5, 133.3, 130.2, 129.8, 124.8, 120.0, 119.1, 119.0, 51.5, 22.3. ¹⁹F NMR 31 $(658 \text{ MHz}, \text{Methanol}-d_4) \delta - 63.10 (d, J = 12.7 \text{ Hz}, 3\text{F}), -114.56 -$ 32 -114.67 (m, 1F). HRMS (ESI+) m/z calcd for $[C_{18}H_{17}F_4N_2O_3]$ +: 33 385.1167, found: 385.1166. HPLC $t_{\rm R} = 18.703$ min (>99%) 34 purity). 35

3,5-di-tert-butyl-N-(4-(hydroxycarbamoyl)phenyl)-N-

36 isopropylbenzamide (21). ¹H NMR (700 MHz, Methanol- d_4) δ 37 7.65 (d, J = 8.4 Hz, 2H), 7.28 (s, 1H), 7.20 (d, J = 8.4 Hz, 2H), 38 7.08 (s, 2H), 4.99 (br, 1H), 1.24 (d, J = 6.8 Hz, 6H), 1.16 (s, 18H). 39 ¹³C NMR (176 MHz, Methanol- d_4) δ 174.8, 168.0, 152.9, 145.6, 40 138.3, 133.8, 133.2, 129.9, 125.6, 125.1, 51.2, 36.9, 32.9, 22.5. 41 HRMS (ESI+) m/z calcd for $[C_{25}H_{35}N_2O_3]$ +: 411.2642, found: 42 411.2642. HPLC $t_{\rm R}$ = 23.806 min (>99% purity).

43 N-(4-(hydroxycarbamoyl)phenyl)-3,5-

44 bis(trifluoromethyl)benzamide (22). ¹H NMR (700 MHz, 45 Methanol- d_4) δ 8.55 (s, 2H), 8.20 (s, 1H), 7.86 (d, J = 8.7 Hz, 2H), 46 7.82 - 7.75 (d, J = 8.7 Hz, 2H). ¹³C NMR (176 MHz, Methanol-47 d_4) δ 168.9, 166.6, 144.1, 139.8, 134.4, 130.7, 130.2, 127.5, 125.8, 122.8. ¹⁹F NMR (564 MHz, Methanol-d₄) δ -64.38 (s, 6F). 48 HRMS (ESI+) m/z calcd for $[C_{16}H_{11}F_6N_2O_3]$ +: 393.0677, found: 49 393.0675. HPLC $t_{\rm R} = 20.396 \text{ min} (98.2\% \text{ purity}).$ 50

*N*¹-(3-(*tert*-butyl)phenyl)-*N*⁴-hydroxyterephthalamide (23). 51 ¹H NMR (700 MHz, Methanol- d_4) δ 7.99 (d, J = 8.3 Hz, 2H), 7.85 52 (d, J = 8.3 Hz, 2H), 7.74 (s, 1H), 7.52 (d, J = 7.0 Hz, 1H), 7.27 (t, 53 J = 7.9 Hz, 1H), 7.20 (ddd, J = 7.8, 1.9, 1.0 Hz, 1H), 1.32 (s, 9H). 54 ¹³C NMR (176 MHz, Methanol-*d*₄) δ 169.1, 168.4, 154.4, 140.6, 55 140.5, 137.7, 130.7, 130.1, 129.6, 124.1, 120.79, 120.77, 36.9, 56

33.0. HRMS (ESI+) m/z calcd for [C18H21N2O3]+: 313.1547, found: 313.1545. HPLC *t*_R = 27.776 min (>99% purity).

N¹-(3-(tert-butyl)phenyl)-N¹-ethyl-N⁴-

hydroxyterephthalamide (24). ¹H NMR (700 MHz, Methanol d_4) δ 7.55 (d, J = 7.7 Hz, 2H), 7.31 (d, J = 8.0 Hz, 2H), 7.23 – 7.16 (m, 2H), 7.05 – 6.93 (m, 2H), 4.00 – 3.96 (br, 2H), 1.21 (t, J = 7.2 Hz, 3H), 1.12 (s, 9H). ¹³C NMR (176 MHz, Methanol- d_4) δ 172.9, 168.2, 155.1, 144.6, 142.3, 135.5, 131.4, 130.8, 128.9, 128.7, 126.9, 126.4, 47.4, 36.6, 32.7, 14.3. HRMS (ESI+) m/z calcd for $[C_{20}H_{25}N_2O_3]$ +: 341.1860, found: 341.1856. HPLC t_R = 27.544 min (>99% purity).

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI:

Additional figures and tables illustrating the HDAC binding and inhibition profiles using FP, EMSA, SPR, and TSA; additional tables summarizing PAMPA, and in vivo pharmacokinetic (PK) results of selected compounds from the SAR; additional figures and tables illustrating the L-shape of inhibitors using twodimensional NOESY experiments, DFT, and docking experiments; additional figures indicating the existence of a conformational equilibrium using variable-temperature onedimensional NMR experiments; additional figures demonstrating cellular HDAC8-target engagement the using immunofluorescence western blotting; compound and characterization data (PDF)

X-ray crystallographic data (CIF)

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Notes

All animal experiments performed in the manuscript were in compliance with institutional guidelines.

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ABBREVIATIONS

HDAC, histone deacetylase; HDACi, HDAC inhibitor; PAMPA, parallel artificial membrane permeability assay; FP, fluorescence polarization; SPR, surface plasmon resonance; TLC, thin layer chromatography; LRMS, low-resolution mass spectrometry; HRMS, high-resolution mass spectrometry; NMR, nuclear magnetic resonance; NOE, nuclear Overhauser effect; NOESY, nuclear Overhauser effect spectroscopy; PK, pharmacokinetic; DFT, density functional theory; Ac, acetylated; SMC3, Structural

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57 58 Maintenance of Chromosomes protein 3; IC₅₀, half maximal inhibitory concentration.

REFERENCES

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- Ropero, S.; Esteller, M. The Role of Histone Deacetylases (HDACs) in Human Cancer. *Mol. Oncol.* 2007, 1, 19–25.
- (2) Seto, E.; Yoshida, M. Erasers of Histone Acetylation : The Histone Deacetylase Enzymes. *Cold Spring Harb Perspect Biol* **2014**, *6*, 1–26.
- (3) Rettig, I.; Koeneke, E.; Trippel, F.; Mueller, W. C.; Burhenne, J.; Fabian, J.; Schober, A.; Fernekorn, U.; Deimling, A. Von; Deubzer, H. E.; Milde, T.; Witt, O.; Oehme, I. Selective Inhibition of HDAC8 Decreases Neuroblastoma Growth in Vitro and in Vivo and Enhances Retinoic Acid-Mediated Differentiation. *Cell Death Dis.* 2015, *6*, e1657.
- Ramos, T. L.; Sánchez-abarca, L. I.; Redondo, A.; (4)Almeida, A. M.; Puig, N.; Ortega, R.; Preciado, S.; Rico, Muntión, S. HDAC8 Overexpression in A.; Mesenchymal Stromal Cells from JAK2+mveloproliferative Neoplasms: А New Therapeutic Target? Oncotarget 2017, 8, 28187-28202.
- (5) Hassan, M. M.; Olaoye, O. O. Recent Advances in Chemical Biology Using Benzophenones and Diazirines as Radical Precursors. *Molecules* 2020, 25, 2285.
- (6) Mariadason, J. M. Making Sense of HDAC2 Mutations in Colon Cancer. *Gastroenterology* 2008, 135, 1457– 1459.
- Huang, B. H.; Laban, M.; Leung, C.; Lee, L.; Lee, C. K.; Raju, G. C.; Hooi, S. C. Inhibition of Histone Deacetylase 2 Increases Apoptosis and P21 Cip1 / WAF1 Expression, Independent of Histone Deacetylase 1. *Cell Death Differ*. 2005, *12*, 395–404.
- (8) Barneda-Zahonero, B.; Parra, M. Histone Deacetylases and Cancer. *Mol. Oncol.* **2012**, *6*, 579–589.
- Oehme, I.; Deubzer, H. E.; Lodrini, M.; Milde, T.; Witt,
 O. Targeting of HDAC8 and Investigational Inhibitors in Neuroblastoma. *Expert Opin. Investig. Drugs* 2009, 18, 1605–1617.
- Yoon, S.; Eom, G. H. HDAC and HDAC Inhibitor : From Cancer to Cardiovascular Diseases. *Chonnam Med. J.* **2016**, *52*, 1–11.
- (11) Hyun-Jung, K.; Suk-Chul, B. Histone Deacetylase Inhibitors: Molecular Mechanisms of Action and Clinical Trials as Anti-Cancer Drugs. *Am J Transl Res* 2011, *3*, 166–179.
- (12) Joanna, F.; A, V. G. L.; Mathieu, V.; Sarah, S.; Sarah, D.; Karin, V.; Tamara, V.; Vera, R. Histone Deacetylase Inhibition and the Regulation of Cell Growth with Particular Reference to Liver Pathobiology. *J. Cell. Mol. Med.* 2009, *13*, 2990–3005.
- Li, T.; Zhang, C.; Hassan, S.; Liu, X.; Song, F.; Chen,
 K.; Zhang, W. Histone Deacetylase 6 in Cancer. 2018, *1*, 1–10.
- (14) Ceccacci, E.; Minucci, S. Inhibition of Histone Deacetylases in Cancer Therapy: Lessons from Leukaemia. *Br. J. Cancer* 2016, *114*, 605–611.
- (15) Zhang, S. L.; Zhu, H. Y.; Zhou, B. Y.; Chu, Y.; Huo, J. R.; Tan, Y. Y.; Liu, D. L. Histone Deacetylase 6 Is (28) Overexpressed and Promotes Tumor Growth of Colon

Cancer through Regulation of the MAPK/ERK Signal Pathway. Onco. Targets. Ther. 2019, 12, 2409–2419.

- (16) Song, S.; Wang, Y.; Xu, P.; Yang, R.; Ma, Z.; Liang, S.; Zhang, G. The Inhibition of Histone Deacetylase 8 Suppresses Proliferation and Inhibits Apoptosis in Gastric Adenocarcinoma. *Int. J. Oncol.* 2015, 47, 1819– 1828.
- (17) Thaler, F.; Mercurio, C. Towards Selective Inhibition of Histone Deacetylase Isoforms: What Has Been Achieved, Where We Are and What Will Be Next. *ChemMedChem* **2014**, *9*, 523–536.
- (18) Bieliauskas, A. V.; Pflum, M. K. H. Isoform-Selective Histone Deacetylase Inhibitors. *Chem. Soc. Rev.* 2008, 37, 1402–1413.
- (19) Hsu, K. C.; Liu, C. Y.; Lin, T. E.; Hsieh, J. H.; Sung, T. Y.; Tseng, H. J.; Yang, J. M.; Huang, W. J. Novel Class IIa-Selective Histone Deacetylase Inhibitors Discovered Using an in Silico Virtual Screening Approach. *Sci. Rep.* 2017, 7, 1–13.
- (20) Dallavalle, S.; Pisano, C.; Zunino, F. Development and Therapeutic Impact of HDAC6-Selective Inhibitors. *Biochem. Pharmacol.* **2012**, *84*, 756–765.
- Marek, M.; Shaik, T. B.; Heimburg, T.; Chakrabarti, A.; Lancelot, J.; Ramos-Morales, E.; Da Veiga, C.; Kalinin, D.; Melesina, J.; Robaa, D.; Schmidtkunz, K.; Suzuki, T.; Holl, R.; Ennifar, E.; Pierce, R. J.; Jung, M.; Sippl, W.; Romier, C. Characterization of Histone Deacetylase 8 (HDAC8) Selective Inhibition Reveals Specific Active Site Structural and Functional Determinants. *J. Med. Chem.* 2018, *61*, 10000–10016.
- (22) Shouksmith, A. E.; Gawel, J. M.; Nawar, N.; Sina, D.; Raouf, Y. S.; Bukhari, S.; He, L.; Johns, A. E.; Manaswiyoungkul, P.; Olaoye, O. O.; Cabral, A. D.; Sedighi, A.; de Araujo, E. D.; Gunning, P. T. Class I/IIb-Selective HDAC Inhibitor Exhibits Oral Bioavailability and Therapeutic Efficacy in Acute Myeloid Leukemia. ACS Med. Chem. Lett. 2020, 11, 56–64.
- (23) Hirano, T.; Osaki, T.; Fujii, S.; Komatsu, D.; Azumaya, I.; Tanatani, A. Fluorescent Visualization of the Conformational Change of Aromatic Amide or Urea Induced by N-Methylation. *Tetrahedron Lett.* 2009, 50, 488–491.
- (24) Talele, T. T. The "Cyclopropyl Fragment" Is a Versatile Player That Frequently Appears in Preclinical/Clinical Drug Molecules. *J. Med. Chem.* **2016**, *59*, 8712–8756.
- (25) De Vreese, R.; D'hooghe, M. Synthesis and Applications of Benzohydroxamic Acid-Based Histone Deacetylase Inhibitors. *Eur. J. Med. Chem.* **2017**, *135*, 174–195.
- (26) Zhao, C.; Zang, J.; Ding, Q.; Inks, E. S.; Xu, W.; Chou, C. J.; Zhang, Y. Discovery of Meta-Sulfamoyl N-Hydroxybenzamides as HDAC8 Selective Inhibitors. *Eur. J. Med. Chem.* 2018, *150*, 282–291.
- (27) Heimburg, T.; Chakrabarti, A.; Lancelot, J.; Marek, M.; Melesina, J.; Hauser, A. T.; Shaik, T. B.; Duclaud, S.; Robaa, D.; Erdmann, F.; Schmidt, M.; Romier, C.; Pierce, R. J.; Jung, M.; Sippl, W. Structure-Based Design and Synthesis of Novel Inhibitors Targeting HDAC8 from Schistosoma Mansoni for the Treatment of Schistosomiasis. J. Med. Chem. 2016, 59, 2423–2435.
 - Ralph Mazitschek, Vishal Patelb, Dyann F. Wirth, and J. C. Development of a Fluorescence Polarization Based Assay for Histone Deacetylase Ligand Discovery.

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Bioorg Med Chem Lett. 2008, 18, 2809-2812.

- (29) Brunsteiner, M.; Petukhov, P. A. Insights from Comprehensive Multiple Receptor Docking to HDAC8. J. Mol. Model. 2012, 18, 3927–3939.
- (30) Deschamps, N.; Simões-Pires, C. A.; Carrupt, P. A.; Nurisso, A. How the Flexibility of Human Histone Deacetylases Influences Ligand Binding: An Overview. *Drug Discov. Today* 2015, 20, 736–742.
- (31) Tiwary, P.; Limongelli, V.; Salvalaglio, M.; Parrinello, M. Kinetics of Protein-Ligand Unbinding: Predicting Pathways, Rates, and Rate-Limiting Steps. *Proc. Natl. Acad. Sci. U. S. A.* 2015, *112*, E386–E391.
- (32) Bernetti, M.; Masetti, M.; Rocchia, W.; Cavalli, A. Kinetics of Drug Binding and Residence Time. *Annu. Rev. Phys. Chem.* 2019, *70*, 143–171.
- (33) Futamura, M.; Dhanasekaran, P.; Handa, T.; Phillips, M. C.; Lund-Katz, S.; Saito, H. Two-Step Mechanism of Binding of Apolipoprotein E to Heparin: Implications for the Kinetics of Apolipoprotein E-Heparan Sulfate Proteoglycan Complex Formation on Cell Surfaces. J. Biol. Chem. 2005, 280, 5414–5422.
- (34) Chabaud, L.; Clayden, J.; Helliwell, M.; Page, A.; Raftery, J.; Vallverdú, L. Conformational Studies of Tertiary Oligo-m-Benzanilides and Oligo-p-Benzanilides in Solution. *Tetrahedron* 2010, 66, 6936– 6957.
- (35) Okamoto, I.; Nabeta, M.; Yamamoto, M.; Mikami, M.; Takeya, T.; Tamura, O. Solvent-Dependent Conformational Switching of the Aromatic N-Methyl Amides Depending upon the Acceptor Properties of Solvents. *Tetrahedron Lett.* 2006, 47, 7143–7146.
- (36) Azumaya, I.; Kagechika, H.; Fujiwara, Y.; Itoh, M.; Yamaguchi, K.; Shudo, K.; Fujiwara, Y.; Itoh, M.; Yamaguchi, K. Twisted Intramolecular Charge-Transfer Fluorescence of Aromatic Amides: Conformation of the Amide Bonds in the Excited States. J. Am. Chem. Soc. 1991, 113, 2833–2838.
- (37) Okamoto, I.; Nabeta, M.; Minami, T.; Nakashima, A.; Morita, N.; Takeya, T.; Masu, H.; Azumaya, I.; Tamura, O. Acid-Induced Conformational Switching of Aromatic N-Methyl-N-(2-Pyridyl)Amides. *Tetrahedron Lett.* 2007, 48, 573–577.
- (38) Itai, A.; Toriumi, Y.; Tomioka, N.; Kagechika, H.; Azumaya, I.; Shudo, K. Stereochemistry of N-Methylbenzanilide and Benzanilide. *Tetrahedron Lett.* 1989, 30, 6177–6180.
- (39) Okamoto, I.; Nabeta, M.; Hayakawa, Y.; Morita, N.; Takeya, T.; Masu, H.; Azumaya, I.; Tamura, O. Acid-Induced Molecular Folding and Unfolding of N-Methyl Aromatic Amide Bearing 2,6-Disubstituted Pyridines. J. Am. Chem. Soc. 2007, 129, 1892–1893.
- (40) Friesner, R. A.; Murphy, R. B.; Repasky, M. P.; Frye, L. L.; Greenwood, J. R.; Halgren, T. A.; Sanschagrin, P. C.; Mainz, D. T. Extra Precision Glide: Docking and Scoring Incorporating a Model of Hydrophobic Enclosure for Protein-Ligand Complexes. J. Med. Chem. 2006, 49, 6177–6196.
 - (41) Friesner, R. A.; Banks, J. L.; Murphy, R. B.; Halgren, T.

A.; Klicic, J. J.; Mainz, D. T.; Repasky, M. P.; Knoll, E. H.; Shelley, M.; Perry, J. K.; Shaw, D. E.; Francis, P.; Shenkin, P. S. Glide: A New Approach for Rapid, Accurate Docking and Scoring. 1. Method and Assessment of Docking Accuracy. *J. Med. Chem.* **2004**, *47*, 1739–1749.

- (42) Halgren, T. A.; Murphy, R. B.; Friesner, R. A.; Beard, H. S.; Frye, L. L.; Pollard, W. T.; Banks, J. L. Glide: A New Approach for Rapid, Accurate Docking and Scoring. 2. Enrichment Factors in Database Screening. *J. Med. Chem.* 2004, 47, 1750–1759.
- (43) Shouksmith, A. E.; Shah, F.; Grimard, M. L.; Gawel, J. M.; Raouf, Y. S.; Geletu, M.; Berger-Becvar, A.; De Araujo, E. D.; Luchman, H. A.; Heaton, W. L.; Bakhshinyan, D.; Adile, A. A.; Venugopal, C.; O'Hare, T.; Deininger, M. W.; Singh, S. K.; Konieczny, S. F.; Weiss, S.; Fishel, M. L.; Gunning, P. T. Identification and Characterization of AES-135, a Hydroxamic Acid-Based HDAC Inhibitor That Prolongs Survival in an Orthotopic Mouse Model of Pancreatic Cancer. J. Med. Chem. 2019, 62, 2651–2665.
- (44) Chakrabarti, A.; Oehme, I.; Witt, O.; Oliveira, G.; Sippl,
 W.; Romier, C.; Pierce, R. J.; Jung, M. HDAC8 : A Multifaceted Target for Therapeutic Interventions. *Trends Pharmacol. Sci.* 2015, *36*, 481–492.
- (45) Brodeur, G. M. Neuroblastoma: Biological Insights into a Clinical Enigma. *Nat. Rev. Cancer* **2003**, *3*, 203–216.
- (46) Dasgupta, T.; Antony, J.; Braithwaite, A. W.; Horsfield, J. A. HDAC8 Inhibition Blocks SMC3 Deacetylation and Delays Cell Cycle Progression without Affecting Cohesin-Dependent Transcription in MCF7 Cancer Cells. J. Biol. Chem. 2016, 291, 12761–12770.
- (47) Emmons, M. F.; Faião-Flores, F.; Sharma, R.; Thapa, R.; Messina, J. L.; Becker, J. C.; Schadendorf, D.; Seto, E.; Sondak, V. K.; Koomen, J. M.; Chen, Y. A.; Lau, E. K.; Wan, L.; Licht, J. D.; Smalley, K. S. M. HDAC8 Regulates a Stress Response Pathway in Melanoma to Mediate Escape from BRAF Inhibitor Therapy. *Cancer Res.* 2019, *79*, 2947–2961.
- (48) Giuliano, M.; Shaikh, A.; Lo, H. C.; Arpino, G.; De Placido, S.; Zhang, X. H.; Cristofanilli, M.; Schiff, R.; Trivedi, M. V. Perspective on Circulating Tumor Cell Clusters: Why It Takes a Village to Metastasize. *Cancer Res.* 2018, 78, 845–852.
- Krause, L.; Herbst-Irmer, R.; Sheldrick, G. M.; Stalke, D. Comparison of Silver and Molybdenum Microfocus X-Ray Sources for Single-Crystal Structure Determination. J. Appl. Crystallogr. 2015, 48, 3–10.
- (50) Sheldrick, G. M. SHELXT Integrated Space-Group and Crystal-Structure Determination. *Acta Crystallogr. Sect. A Found. Crystallogr.* **2015**, *71*, 3–8.
- (51) Sheldrick, G. M. Crystal Structure Refinement with SHELXL. Acta Crystallogr. Sect. C Struct. Chem. 2015, 71, 3–8.

