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4-Aryl pyrrolidines as a novel class of orally efficacious antimalarial agents. Part 1:
Evaluation of 4-aryl-*N*-benzylpyrrolidine-3-carboxamides

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Abstract

Identification of novel chemotypes with antimalarial efficacy is imperative to combat the rise of *Plasmodium* species resistant to current antimalarial drugs. We have used a hybrid target-phenotype approach to identify and evaluate novel chemotypes for malaria. In our search for drug-like aspartic protease inhibitors in publicly available phenotypic antimalarial databases, we identified GNF-Pf-4691, a 4-aryl-*N*-benzylpyrrolidine-3-carboxamide, as having a structure reminiscent of known inhibitors of aspartic proteases. Extensive profiling of the two terminal aryl rings revealed a structure-activity relationship in which relatively few substituents are tolerated at the benzylic position, but the 3-aryl position tolerates a range of hydrophobic groups and some heterocycles. Out of this effort, we identified (+)-**54b** (CWHM-1008) as a lead compound. **54b** has EC₅₀ values of 46 nM and 21 nM against drug sensitive *Plasmodium falciparum* 3D7 and drug resistant Dd2 strains, respectively. Furthermore, **54b** has a long half-life in mice (4.4 h) and is orally efficacious in a mouse model of malaria (q.d.; ED₉₉ ~ 30 mg/kg/day). Thus, the 4-aryl-*N*-benzylpyrrolidine-3-carboxamide chemotype is a promising novel chemotype for malaria drug discovery.

Introduction

Efforts at reducing incidents of malaria have stalled in recent years after a decade of progressive reduction in disease burden. The majority of malaria-related deaths occur in Africa and are caused by *Plasmodium falciparum*, the most lethal to man of the *Plasmodium* species. The introduction of artemisinin and artemisinin combination therapies (ACT) in 2005 led to a remarkable 60% reduction in mortality rates between 2000-2015.¹ However, according to the World Health

Organization, there were ~219 million new cases of malaria and 435,000 deaths in 2017, highlighting a leveling out of progress in reducing global malarial disease burden since 2015.² Furthermore, there are reports of emerging resistance to artemisinin in Southeast Asia.^{1, 3-4} As the artemisinins are the only fully effective class of antimalarial drugs available today, it is crucial that additional antimalarial drugs be developed with new mechanisms of action as the next line of defense to combat developing resistance to known drugs. These efforts, along with efforts to control malaria transmission will be needed if successful global eradication of malaria is to be achieved.

The *Plasmodium* parasite has a complex lifecycle including sexual replication in the mosquito stage and asexual replication in the human liver and blood stages. These lifecycle stages involve numerous potential opportunities for intervention. The challenge is to identify unexploited biological targets that will result in effective parasite killing. A broad class of proteases, the aspartic proteases, have been successfully exploited for the treatment of AIDS—more than 10 FDA approved drugs have been developed that inhibit the HIV aspartic protease. *Plasmodium* has multiple aspartic proteases that play key roles in the survival of the parasite in its human host.⁵⁻⁶ Identification of inhibitors of the *Plasmodium* aspartic proteases has the potential to provide a novel mechanism for anti-malarial therapies. However, a major challenge to developing protease inhibitors as drugs is identifying compounds with cellular activity commensurate with their enzyme inhibition potency. This is often due to poor physicochemical properties, efflux mechanisms, protein binding or poor target-related biochemical efficiencies.

Rather than focus on a substrate-based inhibitor design approach that may provide potent but non-drug-like peptidomimetics, we elected to identify known, drug-like aspartic protease inhibitors with inherent antimalarial cell activity for optimization. We termed this approach as a

“hybrid target-phenotype drug discovery”. Scientists at GlaxoSmithKine (GSK),⁷⁻⁸ Novartis,⁹ and St. Jude’s Children’s Hospital¹⁰ recently screened large chemical libraries for antimalarial activity in a standard *Plasmodium falciparum* 3D7-infected red blood cell assay. This effort identified ~20,000 compounds with antimalarial activity, and the data were made public. Since a number of aspartic proteases have been shown to play essential roles in the *Plasmodium* life cycle, we hypothesized that aspartic protease inhibitors with antimalarial activity may be acting through one or more *Plasmodium* aspartic proteases. Such inhibitors would provide a powerful starting point for drug discovery: phenotypic hits (cell active) with a limited but efficacious set of potential drug targets. By mining these databases for aspartic protease-inhibiting chemotypes, we and others have found multiple inhibitor classes with drug-like properties that may be acting on the parasite through an aspartic protease mechanism due to their structural similarity to known aspartic protease inhibitors.¹¹⁻¹³ Indeed, we have shown drug-like compounds from the aminohydantoin series inhibits plasmepsins II, IV, V, and X.^{6,11} The pyrrolidine class described herein also appears to be particularly drug-like and is ideal for lead optimization to identify novel clinical candidates for treatment of malaria.

In our previous work, we evaluated a set of spiropiperidine hydantoins such as CWHM-505 (**1**) (**Fig. 1**).¹² While we were able to improve the potency of this series to sub-100 nanomolar, they suffered from particularly poor metabolic stability, presumably due to the cleavage of the benzylic piperidine. A number of pyrrolidine and piperidine-based inhibitors of aspartic proteases BACE, renin and *Pf*plasmepsin 2 (PM-II) have been reported in the literature (**2-5**).¹⁴⁻¹⁸ These inhibitors have been shown to bind to the aspartic acid residues in the active site via the protonated nitrogen of the pyrrolidine (e.g., PDB ID: 3UFL).¹⁴

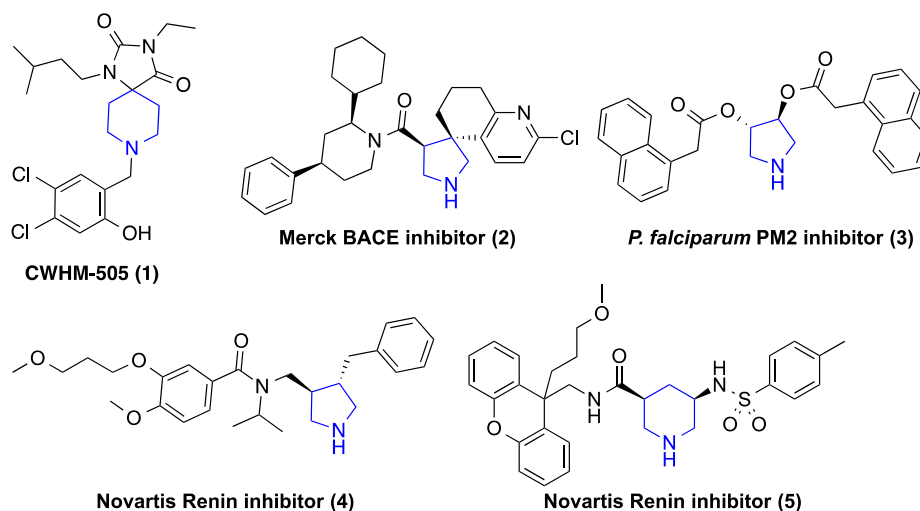


Figure 1. Antimalarial spiropiperidine hydantoin and examples of pyrrolidine- and piperidine-based aspartic protease inhibitors. Common basic group is highlighted in blue.

We were attracted to pyrrolidine as an antimalarial pharmacophore as it would avoid the primary metabolic liability in our previous work on benzyl piperidines (e.g., **1**) as the basic nitrogen is not substituted in these compounds. Indeed, renin-inhibiting pyrrolidine (**4**) had been reported to have a two-hour half-life in rats.¹⁵ Using the simple unsubstituted pyrrolidine core (**Fig. 2**), we searched the GSK TCAMS and Novartis malaria phenotypic screening databases and found 45 hits containing this pharmacophore. Many of these hits were aminoquinoline analogs of chloroquine and amodiaquine and were thus eliminated from our consideration for lack of novel mechanism of action. Six compounds, however, were members of the unique antimalarial pharmacophore represented by racemic Novartis compound GNF-Pf-4691 (**6**; **Fig. 2**). We resynthesized GNF-Pf-4691 (**6**) and assayed it in a SYBR Green *Pf* 3D7 assay¹⁹ and found it to have an IC_{50} of 385 nM, within 3-fold of the reported database value. To our delight, we found this compound to have a 100-fold cytotoxicity selectivity index and an improved metabolic stability profile in mouse liver microsomes. On the basis of this data, we elected to initiate

structure-activity relationship (SAR) studies to determine whether this pyrrolidine series should be pursued for full lead optimization. We independently varied the amide benzyl ring and the pyrrolidine aryl ring to develop some preliminary SAR and identified the preferred stereoisomers of the pyrrolidine core. Using this SAR, we combined preferred groups at each position to identify lead compounds suitable for in vivo efficacy studies.

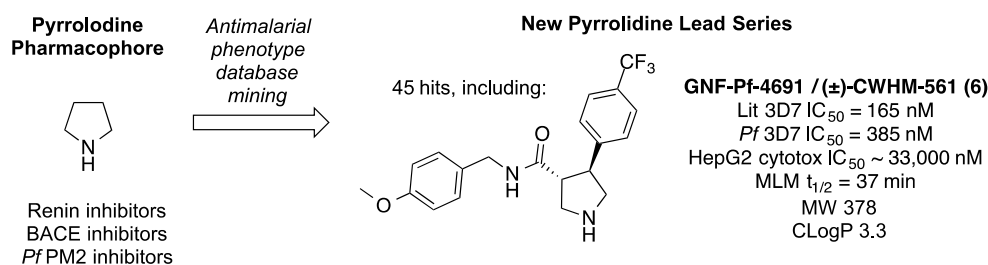
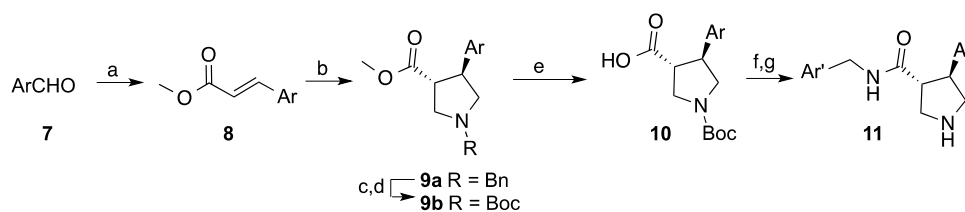


Figure 2. Strategy to identify phenotypic protease inhibitor-like hits with drug-like properties while eliminating liabilities of earlier series.

Results and Discussion

Synthesis. It is well established that *trans*-3,4-disubstituted pyrrolidines can be readily and stereo selectively synthesized through a 3+2 cycloaddition reaction using N-(methoxymethyl)-N-(trimethylsilylmethyl)benzylamine.²⁰⁻²¹ Accordingly, racemic *trans*- α,β -unsaturated esters (**8**) were prepared by the Wittig reaction and then converted to the corresponding pyrrolidines (**9a**) through the unstabilized azomethine ylide generated by TFA (Scheme 1). To simplify preparation of the final products, the benzyl group was replaced with Boc (**9b**). Ester hydrolysis followed by amide coupling and Boc-deprotection yielded the final *trans*-pyrrolidines (**11**) as racemates.



Scheme 1. Synthesis of racemic pyrrolidines. Reagents and conditions: (a) $\text{Ph}_3\text{P=CHCO}_2\text{Me}$; (b) TFA, $\text{Me}_3\text{SiCH}_2\text{N(Bn)CH}_2\text{OMe}$; (c) $\text{NH}_4\text{CO}_2\text{H}$, Pd/C; (d) Et_3N , Boc_2O ; (e) LiOH; (f) ArCH_2NH_2 , HATU; (g) HCl.

Amide benzyl ring SAR. Holding the right-hand side constant as 4-trifluoromethylphenyl, we systematically modified the terminal amide aryl ring. Initially new analogs were synthesized as racemates to more efficiently profile the SAR and develop our understanding of the pharmacophore. Representative analogs are shown in **Fig. 3**. Substitution on the phenyl ring is essential for potency. 4-Methoxy (**6**) and 4-methyl (**12**; GNF-Pf-3587) groups give 10-fold increases in potency relative to H (**13**). The most potent substituent in this position identified to date is dimethylamine (**14**), approximately two-fold more potent than the lead methoxy analog. Conversion of the dimethyl aniline into an indoline ring (**15**) was tolerated but extending it as a diethyl aniline (**16**) reduced potency by 2-3-fold.

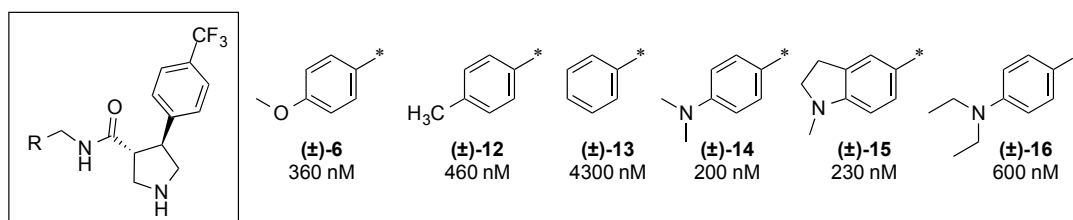


Figure 3. Initial amide benzyl ring SAR. IC_{50} values are given as average potency values in the Pf3D7 assay ($n \geq 3$). Standards chloroquine and artemesinin have IC_{50} values of 54 ± 0.4 nM and 33 ± 0.6 nM, respectively.

Pyrrolidine aryl ring SAR. Holding the amide aryl ring constant as the dimethyl aniline, we systematically explored SAR of the pyrrolidine aryl ring (**Fig. 4**). This position is also quite sensitive to substitution patterns. 4-CF₃ (**14**) is the preferred substituent with *t*-butyl (**19**) being an appropriate replacement, but replacement with methyl (**17**) or chloro (**18**) led to 3- to 4-fold reductions in potency. We also explored potential replacements for the trifluoromethyl group such as difluoromethyl (**20**), CF₂CH₃ (**21**), and pentafluorosulfide (**22-23**). These groups were suitable replacements with IC₅₀ values ranging from 130 to 470 nM.

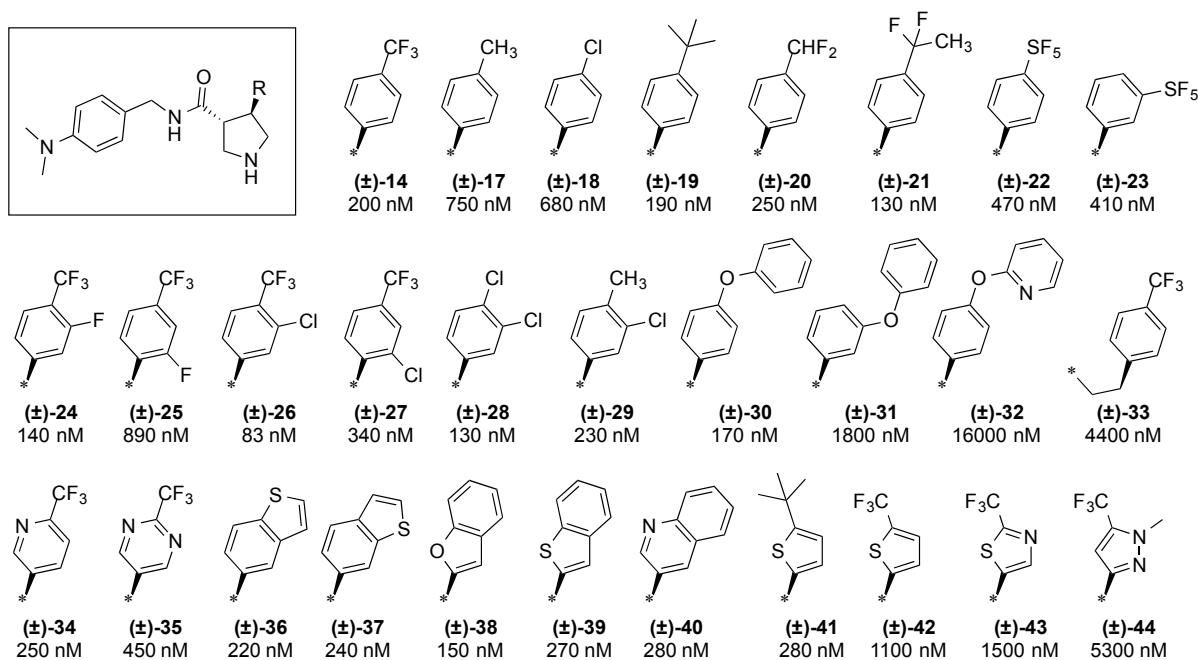


Figure 4. Initial pyrrolidine aryl ring SAR. IC₅₀ values are given as average potency values in the *Pf*3D7 assay (*n*≥3).

3,4-Disubstitution is tolerated and may even be favorable for enhancement of potency (**24-29**). For example, with an IC₅₀ of 83 nM, the 4-CF₃-3-Cl analog (**26**) is one of the most potent racemates

we have identified. 2,4-Disubstitution was less well tolerated leading to 6-8-fold reductions in potency (**25**, **27**).

Extension of the aryl ring with phenylether was tolerated in the 4-position (**30**) but not the 3-position (**31**) or as a pyridylether (**32**). Extending the aryl ring itself led to loss of potency (**33**).

The phenyl ring itself can be replaced with pyridine without loss in potency (**34**). Incorporation of a second ring nitrogen (e.g., pyrimidine **35**) leads to modest erosion of potency. A number of other heterocycles were explored (**36-44**). Benzothiophenes (**36-37**, **39**), benzofuran (**38**), and 2-quinoline (**40**) were all essentially equipotent with 4-trifluoromethylphenyl (**14**). Thiophene analogs designed to mimic the CF₃ and t-butyl phenyl group were tolerated as t-butyl (**41**) but not so as CF₃ (**42**). Other heterocycles such as thiazole (**43**) and pyrazole (**44**) led to >8-fold losses in potency.

Pyrrolidine Stereochemical SAR. The stereochemistry of the pyrrolidine was also investigated (**Fig. 5**). The cycloaddition chemistry leads to stereospecific formation of the pyrrolidine ring as a racemic mixture of trans isomers (**46a-b**). After replacement of the benzyl groups for Boc groups (**47a-b**) and hydrolysis of the methyl esters to the acids, the (*S*)-benzyloxazolidinone chiral auxiliary coupled to enable resolution of the resulting diastereomers **48a-b** (first eluent) and **49a-b** (second eluent) by silica gel chromatography. The isolated diastereomers were then hydrolyzed to furnish the carboxylic acids **50a-b** and **51a-b** as single enantiomers. These acids were then coupled to benzylic amines and deprotected to give the final products **52-55**. The absolute stereochemistry was determined by x-ray crystallography for **54b** to be the (3*R*,4*S*)-configuration (**Fig. 5B**).

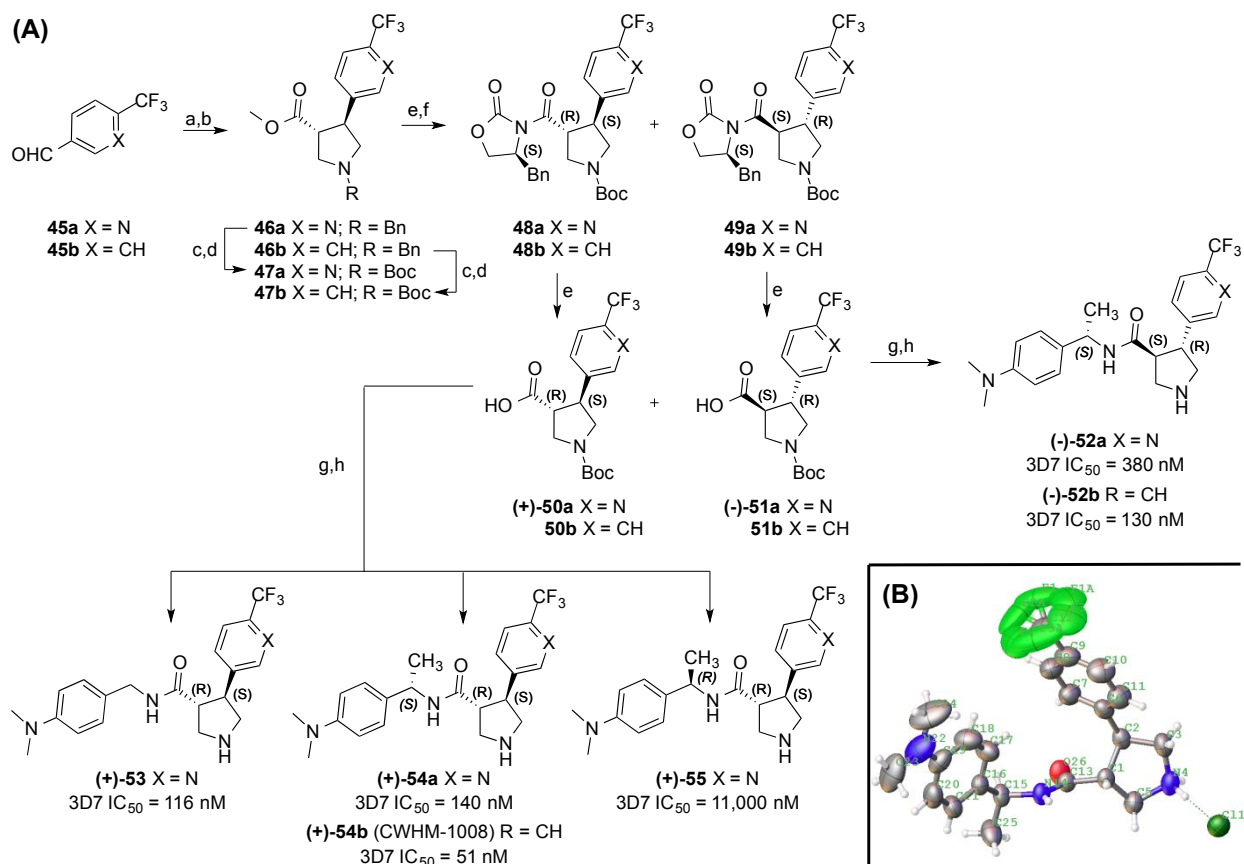


Figure 5. Synthesis and SAR of pyrrolidine enantiomers 52-55. (A) IC₅₀ values are given as average potency values in the *Pf* 3D7 assay ($n \geq 3$). Reagents and conditions: (a) $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Me}$; (b) TFA, $\text{Me}_3\text{SiCH}_2\text{N}(\text{Bn})\text{CH}_2\text{OMe}$; (c) $\text{NH}_4\text{CO}_2\text{H}$, Pd/C; (d) Et_3N , Boc_2O ; (e) LiOH; (f) (*S*)-4-benzyl-2-oxazolidinone, pivaloyl chloride, TEA, LiCl; (g) ArCH_2NH_2 , HATU; (h) HCl. (B) ORTEP representation of the crystal structure of **54b** as an HCl salt.

There is a modest three-fold difference in potency between (*3R,4S*)-**54a** and (*3S,4R*)-**52a** enantiomers with the (*3R,4S*)-configuration being preferred (**Fig. 5A**). In contrast, stereochemistry at the α -benzyl carbon has a dramatic effect on potency with the (*S*)-Me group being tolerated (**53**) and the (*R*)-Me group being detrimental to potency by >30-fold (**55**). Comparing pyridine analog

54a versus phenyl analog **54b** shows a nearly 3-fold preference for phenyl. A similar relationship holds for pyridine analog **52a** versus phenyl analog **52b**.

Revisiting the amide aryl ring SAR. Having identified the (3*R*,4*S*) stereochemistry as preferred, we revisited our studies on the amide aryl ring SAR using the 4-trifluoropyridine moiety as the pyrrolidine aryl ring substituent as this would give lower lipophilicity. As described in the synthesis section, we utilized enantiomer **50a** as a synthetic intermediate to derive a series of analogs. SAR with the 4-trifluoropyridine moiety is shown in **Fig. 6**. As with the 4-trifluoromethylphenyl series, the dimethylaniline is preferred (**34**). Modest changes to pyridine (**56**), pyrrolidine (**57**), and N-pyrazole (**58**) led to modest to dramatic losses in potency. Other substitutions (**59-61**) and extension of the dimethylamine (**62**) were not tolerated.

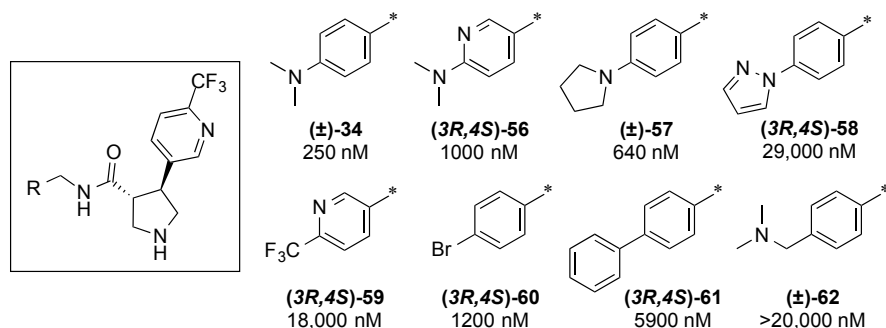


Figure 6. Amide aryl ring SAR with 4-trifluoromethylpyridine. IC_{50} values are given as average potency values in the *Pf*3D7 assay ($n \geq 3$).

Pyrrolidines are potent on the drug-resistant Dd2 strain of *P. falciparum*. With an IC_{50} value of 51 nM, compound **54b** was selected as our lead compound for further profiling. **54b** was profiled in extensive side-by-side studies with chloroquine (CQ) in the multi-drug resistant Dd2 strain of *P. falciparum*. **54b** has equivalent potency to CQ in the *Pf*3D7 strain ($IC_{50} = 46 \pm 6$ nM).

vs. CQ $IC_{50} = 38 \pm 2$ nM) and 10-fold greater potency than CQ against the multi-drug resistant Dd2 ($IC_{50} = 21 \pm 1$ nM vs. CQ $IC_{50} = 196 \pm 14$ nM).

Inhibition of Aspartic Proteases. Since our original hypothesis was that these pyrrolidines might be aspartic protease inhibitors, we profiled a select set of seven compounds for inhibition of human β -secretase (BACE1), *Pf* plasmepsin II (PM-II) and *Pf* plasmepsin IV (PM-IV) enzymes (Table 1; entries 1-7). However, none of the seven compounds inhibited these aspartic proteases. **6** was also tested for inhibition of *Pf* PM-V, PM-IX and PM-X in a knockdown assay (PM-V) and in western blots looking for PM-IX and X substrate processing in compound-treated parasites but was found to be inactive against these proteases (data not shown).^{6, 11, 22-23} Thus it appears that these pyrrolidines are not inhibitors of plasmepsins II, IV, V, IX or X. To date, we have not identified a biomolecular target for these pyrrolidines, and we cannot rule out inhibition of other *Plasmodium* aspartic proteases.

Table 1. In vitro Profiling Data

Entry	Cmpd	<i>Pf</i> 3D7 IC_{50} , nM	BACE1 IC_{50} , nM	<i>Pf</i> PM-II IC_{50} , nM	<i>Pf</i> PM-IV IC_{50} , nM	hERG IC_{50} , nM	hERG/3D7 Ratio	MLM $t_{1/2}$, min	MLM Cl_{int} , μ L/min/mg
1	12	460	>10,000	>10,000	>10,000	8,950	19	161	1.4
2	14	200	>10,000	>10,000	>10,000	9,410	47	35	6.2
3	16	600	>10,000	>10,000	>10,000	4,740	8	n.d.	n.d.
4	24	140	>10,000	>10,000	>10,000	8,540	61	34	6.4
5	26	83	>10,000	>10,000	>10,000	7,560	91	33	6.7
6	30	170	>10,000	>10,000	>10,000	2,110	12	31	7.2
7	34	230	>10,000	>10,000	>10,000	21,700	94	70	3.1
8	35	450	n.d.	n.d.	n.d.	>50,000	>111	n.d.	n.d.
9	54a	140	n.d.	n.d.	n.d.	>50,000	>357	n.d.	n.d.
10	54b	51	n.d.	n.d.	n.d.	32,000	627	38	n.d.

n.d. = not determined.

Affinity for the hERG channel. A potential concern with this series is its potential for binding the hERG channel given all of these compounds contain a basic amine pyrrolidine core.²⁴⁻²⁵ To address this, we evaluated ten compounds in this series for hERG binding in a competitive binding assay (Table 1). Compounds tested have binding affinities for hERG ranging from 2 μ M to >50 μ M. While some of the compounds have modest 8- to 20-fold hERG/3D7 ratios, our best compounds had hERG binding affinities of >20 μ M and selectivity ratios of 90- to 600-fold. Compounds with a pyrrolidine phenyl ring (entries 1-6) tended to have stronger affinity for the hERG channel versus pyridine (**34** and **54a**) and pyrimidine (**35**) (entries 7-9). The exception to this trend is **54b** (entry 10) which had a >600-fold selectivity vs. the hERG channel. It is possible that this is due to this compound being tested as a single enantiomer or the chiral α -methyl group may also reduce hERG potency.

In vitro and in vivo pharmacokinetics. Six compounds were profiled for metabolic stability in mouse liver microsomes (MLM; Table 1). All six compounds were moderately to very stable with MLM half-lives ranging from 31 to 161 min. **12** and **34** and **54b** were selected for mouse PK studies (Table 2). **12** has relatively high clearance and a modest half-life in mice. In contrast, and **34** and **54b** have good half-lives and low clearance in mice. **54b** was found to have suitable oral bioavailability for *in vivo* efficacy studies.

Table 2. Mouse Pharmacokinetic Data

Compound	Route	Animals	Dose, mg/kg	t _{1/2} , h	CL _z , mL/min/kg	V _z , L/kg	F, %
12	i.v.	4	1	1.7	52	7.3	-
34	i.v.	4	2	6.6	5.9	3.2	-
34	p.o.	4	5	-	-	-	13
54b	i.v.	4	2	4.4	3.3	1.3	-
54b	p.o.	3	5	-	-	-	32

CL_z = apparent rate of clearance; V_z = apparent volume of distribution.

54b is orally efficacious in a mouse model of malaria. Given the oral bioavailability of **54b**, we evaluated it as a tool compound to provide *in vivo* proof-of-concept in the murine Peters 4-day suppressive test using *P. chabaudi* ASCQ (a CQ-resistant strain).^{19, 26-27} NIH mice were inoculated with *P. chabaudi* ASCQ parasitized red blood cells. After 4 h, **54b** was dosed orally at 3, 10 and 30 mg/kg/day once daily for four days. Parasitemia levels were determined 24 h after the last treatment (**Table 3**). At 30 mg/kg/day, **54b** inhibits parasitemia at 98.7%. The dose-response data allow us to approximate an ED₉₀ of ~20 mg/kg/day and an ED₉₉ of ~30 mg/kg/day. Compound concentrations in the plasma were determined at 1 h, 6 h and 24 h post dose on the last day of treatment. Compound concentrations in the plasma (110 nM at 24 h) remained above the *Pf*3D7 IC₅₀ (46 nM) for the full 24 h period only for the most fully efficacious dose of 30 mg/kg/day.

Table 3. *In vivo* efficacy of pyrrolidine 54b in *P. chabaudi* ASCQ infected mice.

Oral Dose (qd)	% Inh. of Growth	Plasma Compound Concentration (nM)		
		1 h	6h	24 h
54b , 3 mg/kg/day	22.4 ± 11.6	640	140	0
54b , 10 mg/kg/day	75.2 ± 10.6	1,160	960	10
54b , 30 mg/kg/day	98.7 ± 0.2	2,150	1,630	110
CQ, 10 mg/kg/day	95.0 ± 1.1	-	-	-

Inhibition of parasitemia after 4 days of qd oral dosing and plasma compound concentrations on Day 4. n=6 animals/group.

Conclusions

Herein, we describe the profiling of aryl pyrrolidine carboxamides as a novel class of antimalarial agents. We have generated an extensive SAR for the terminal aryl rings. The SAR is rather narrow in that relatively few substituents are permissible without significant losses in

potency. On the carboxamide aryl ring, dimethylaniline is vastly superior to nearly every functional group evaluated. On the pyrrolidine aryl ring, para-trifluoromethyl can be replaced by a limited set of small lipophilic moieties, including *t*-butyl, CHF₂, and CF₂CH₃. Addition of a halogen in the meta-position is beneficial. Replacement of the phenyl ring with heterocycles such as pyridine, pyrimidine, benzofuran and thiophene, is also tolerated. (3*R*,4*S*) is the preferred stereochemistry on the pyrrolidine ring and (*S*)-Me is preferred in the benzyl position. Out of this effort, we identified **54b** as a lead compound for further profiling. **54b** is more potent on the drug resistant Dd2 strain, has suitable selectivity over the hERG channel, has a long half-life in mice, and is orally efficacious in a mouse model of malaria (q.d.; ED₉₉ ~ 30 mg/kg/day). Thus, compound **54b** (CWHM-1008) presents a promising lead for optimization as an antimalarial drug with a low molecular weight, modest lipophilicity, excellent antimalarial potency, and long half-life oral bioavailability in mice.

Experimental Section

General. Commercially obtained reagents were used without further purification. All reactions were monitored by TLC with silica gel-coated plates. Chemical structures and IUPAC names were generated using CambridgeSoft ChemDraw Ultra 10.0 or CDD Vault (www.collaborativedrug.com). Specific rotation value were recorded on AUTOPOL®μOJ-H, 4.6mm x 250mm 5 IV-T (λ= 589 nm, 50 mm cell, 20°C). MS analyses were performed on the API 2000 electrospray mass spectrometer in positive/negative ion mode. The scan range was 100–1000d. ¹H NMR spectra were recorded on a Bruker AV-400 or 500 MHz spectrometer. Chemical shifts (δ) are given in relative to tetramethylsilane (δ 0.00 ppm) in CDCl₃. Coupling constants, *J*, were reported in hertz unit (Hz). All compounds were ≥95% pure by HPLC

conducted on an Agilent 1260 system using a reverse phase C18 column with diode array detector and a methanol/water (0.1% NH₄OH) gradient unless stated otherwise. HRMS spectra were recorded on an ABSciex 5600+ instrument.

Compounds were filtered for PAINS substructures using the PAINS (Pan-assay interference compounds) filters available at <http://fafdrugs4.mti.univ-paris-diderot.fr>. 38 of the 47 final compounds described here have the dialkyl aniline substructure PAIN alert. However, this substructure is only considered an interference substructure for AlphaScreen technology as they can be quenchers of singlet oxygen and are not considered general PAINS.²⁸ The assays used herein are cell-based assays and include correlation of compound potency from cell based assays to in vivo efficacy models.

(±)-(3R,4S)-methyl 1-benzyl-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylate (46a). To a stirred solution of methyl (triphenylphosphoranylidene)acetate (20.1 g, 60.0 mmol) in dichloromethane (100 mL) was added 6-(trifluoromethyl)nicotinaldehyde (10.0g, 57.1 mmol) at 0 °C, and then the resulting mixture was stirred at room temperature for 4 h . The reaction was monitored by TLC (10% ethyl acetate in petroleum ether). The reaction was concentrated *in vacuo*. Flash chromatography (petroleum ether/ethyl acetate: 95/5) afforded (*E*)-methyl 3-(6-(trifluoromethyl)pyridin-3-yl)acrylate (13.2 g, 99 % yield) as a white solid.

To a stirred solution of (*E*)-methyl 3-(6-(trifluoromethyl)pyridin-3-yl)acrylate (13.2 g, 57.1 mmol) in dichloromethane (150 mL) was added N-(methoxymethyl)-N-(trimethylsilylmethyl)-benzylamine (17.8 g, 74.3 mmol). The resulting mixture was cooled to 0 °C and a solution of TFA (0.8 mL, 0.1 eq) in dichloromethane (2.0 mL) was added dropwise. The reaction mixture was allowed to warm to room temp and stirred for 16 h. The solvent was removed by evaporation in

vacuo and the resulting oil was purified by flash column chromatography (petroleum ether/ethyl acetate: 80/20) to give the title compound (16.8 g, 81% yield). MS: $m+1=365.3$.

(±)-(3R,4S)-1-tert-butyl-3-methyl-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-1,3-dicarboxylate (47a). To a flask was added **46a** (16.8 g, 46.2 mmol), ammonium formate (9.2 g, 146 mmol), Pd/C (1.7 g, 10%) and MeOH (200 mL). The reaction flask was flushed with argon three times and then heated at 70°C for 30 min. The reaction was cooled to room temp and filtered through Celite®, rinsing with MeOH (50 mL). To this mixture was added triethylamine (23.3 g, 231 mmol), then cooled to 0 °C and di-tert-butyl decarbonate (30.2 g, 139 mmol) was added dropwise. The reaction mixture was allowed to warm to room temp and stirred for 16 h. The solvent was removed by evaporation in vacuo and the resulting oil was purified by flash column chromatography (petroleum ether/ethyl acetate: 80/20) to give the title compound (15.6 g, 90% yield). MS: $m+1=375.2$.

(+)-(3R,4S)-1-(tert-butoxycarbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylic acid (51a) and (-)-(3S,4R)-1-(tert-butoxycarbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylic acid (51a). To a flask was added **47a** (15.6 g, 44.4 mmol) and methanol (150 mL). A solution of LiOH (4.7 g, 111 mmol) in water (100 mL) was added dropwise. The reaction mixture was stirred at room temperature for 4 h. The most of methanol was removed by evaporation, diluted with water and basified with hydrochloric acid (2M) to pH 4. The mixture was extracted with EtOAc (3 × 150 mL). The combined organic extracts were washed with water then brine, dried over sodium sulfate, filtered and concentrated to afford (±)-(3R,4S)-1-(tert-butoxycarbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylic acid (14.5 g, 91% yield).

To a suspension of (\pm) -(3R,4S)-1-(tert-butoxycarbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylic acid (14.5 g, 40.3 mmol) in anhydrous THF (300 mL) was added triethylamine (10.2 g, 101 mmol). The resulting mixture was cooled to -20 °C and a solution of pivaloyl chloride (5.8g, 48.4 mmol) in THF (20 mL) was added dropwise. The reaction mixture was stirred at -20 °C for 2 h and then added LiCl (1.8 g, 44.3 mmol) was added followed by benzyl-2-oxazolidinone (7.1 g, 40.3 mmol). The reaction mixture was allowed to warm to room temp and stirred for 4 h. The solvent was removed by evaporation in vacuo and the resulting oil was purified by flash column chromatography (petroleum ether/ethyl acetate: 90/10) to separate the mixture of **48a** and **49a**.

The first eluting compound was (3R,4S)-tert-butyl3-((S)-4-benzyl-2-oxooxazolidine-3-carbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-1-carboxylate (**48a**) obtained as a white solid (6.7 g, 32% yield). TLC R_f = 0.52 (petroleum ether/ethyl acetate: 3/1). To a flask was added **48a** (6.7 g, 12.9 mmol) and THF (100 mL). A solution of LiOH (1.1 g, 25.8 mmol) in water (25 mL) was added dropwise. The reaction mixture was stirred at room temperature for 4 h. The most of methanol was removed by evaporation, diluted with water and basified with hydrochloric acid (2M) to pH 4. The mixture was extracted with EtOAc (3 × 150 mL). The combined organic extracts were washed with water then brine, dried over sodium sulfate, filtered and concentrated to afforded (+)-(3R,4S)-1-(tert-butoxycarbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylic acid (**50a**) (3.8 g, 82% yield). MS: $m-1$ = 359.2. ^1H NMR (400 MHz, DMSO- d_6) δ ppm 8.74 (s, 1H), 8.09 (d, J = 8.0 Hz, 1H), 7.86 (d, J = 8.0 Hz, 1H), 3.81 (t, J = 8.8 Hz, 1H), 3.73 (td, J = 8.8, 1.6 Hz, 1H), 3.67 (m, 1H), 3.49-3.36 (m, 2H), 3.29 (t, J = 10.0 Hz, 1H), 1.41 (d, J = 8.4 Hz, 9H). $[\alpha]_D^{20}$ +30.1 (c 0.09, MeOH).

The second eluting compound was diastereomer (3S,4R)-tert-butyl3-((S)-4-benzyl-2-oxooxazolidine-3-carbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-1-carboxylate (**49a**) obtained as a white solid (7.1 g, 34% yield). TLC R_f = 0.41 (petroleum ether/ethyl acetate: 3/1). **49a** was hydrolyzed with LiOH as described above to give (-)-(3S,4R)-1-(tert-butoxycarbonyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxylic acid (**51a**) (3.7 g, 75% yield). MS: m/z = 359.2. ^1H NMR (400 MHz, DMSO- d_6) δ ppm 8.75 (s, 1H), 8.10 (d, J = 8.0 Hz, 1H), 7.87 (d, J = 8.0 Hz, 1H), 3.81 (t, J = 8.8 Hz, 1H), 3.77-3.66 (m, 2H), 3.44 (br, 2H), 3.30 (t, J = 9.6 Hz, 1H), 1.41 (d, J = 9.6 Hz, 9H). $[\alpha]_D^{20}$ -30.4 (c 0.13, MeOH).

(+)-(3R,4S)-N-(4-(dimethylamino)benzyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-3-carboxamide (**53**). To a suspension of **50a** (400 mg, 1.11 mmol), 4-(aminomethyl)-*N,N*-dimethylaniline (200 mg, 1.33 mmol), HATU (630 mg, 1.66 mmol) in dichloromethane (10 mL) was added triethylamine (0.31 mL, 2.22 mmol). The reaction mixture was stirred at room temp for 16 h. The mixture was diluted with dichloromethane (30 mL) and washed with saturated sodium carbonate solution then brine, dried over sodium sulfate, filtered and concentrated. The residue was purified by flash column chromatography (petroleum ether/ethyl acetate: 1/1) to give (330 mg, 61% yield) of (3R,4S)-tert-butyl3-(4-(dimethylamino)benzylcarbamoyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-1-carboxylate as a white solid.

To a suspension of (3R,4S)-tert-butyl3-(4-(dimethylamino)benzylcarbamoyl)-4-(6-(trifluoromethyl)pyridin-3-yl)pyrrolidine-1-carboxylate (330 mg, 0.67 mmol), in dichloromethane (6 mL) was added TFA (3.0 mL). The reaction mixture was stirred at room temp for 2 h. The solvent was removed by evaporation in vacuo and the resulting oil was diluted with dichloromethane (30 mL) and washed with saturated sodium carbonate solution then brine, dried over sodium sulfate, filtered and concentrated. The residue was purified by flash column chromatography (DCM/ NH_3

in MeOH: 20/1) to give the title compound 160 mg (61% yield). ^1H NMR (500 MHz, $\text{DMSO}-d_6$) δ ppm 8.67(s, 1H), 8.25(t, $J=5.5$, 1H), 7.96 (d, $J=8.0$, 1H), 7.83 (d, $J=8.0$, 1H), 6.86 (d, $J=8.5$, 2H), 6.58(d, $J=8.5$, 2H), 4.18(dd, $J=14.5$, 6.0, 1H), 4.01(dd, $J=14.5$, 6.0, 1H), 3.52(q, $J=10.5$, 1H), 3.26 (t, $J=10.0$, 1H), 2.99 (q, $J=8.0$, 1H), 2.97(q, $J=8.0$, 1H), 2.85 (t, $J=10.0$, 1H), 2.83 (s, 6H). $[\alpha]_{\text{D}}^{20} +73.2$ (c 0.11, MeOH). HRMS (ESI) m/z : $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{20}\text{H}_{24}\text{F}_3\text{N}_4\text{O}$ 393.1902; found 393.1889.

(-)-(3S,4R)-N-[(1S)-1-[4-(dimethylamino)phenyl]ethyl]-4-[6-(trifluoromethyl)pyridin-3-yl]pyrrolidine-3-carboxamide (52a). The title compound was synthesized from **51a** and (*S*)-4-(1-aminoethyl)-*N,N*-dimethylbenzenamine as according to the method described for **53**. Yellow solid (110 mg, 54% yield). ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ ppm 8.66 (s, 1H), 8.15 (d, $J = 8.0$ Hz, 1H), 7.97 (d, $J = 8.0$ Hz, 1H), 7.84 (d, $J = 8.0$ Hz, 1H), 7.06 (d, $J = 8.4$ Hz, 2H), 6.65 (d, $J = 8.4$ Hz, 2H), 4.79 (m, 1H), 3.36 (m, 1H), 3.30 (1H, overlapped with the peak of H_2O), 3.20 (m, 1H), 2.90-2.84 (m, 9H), 1.23 (s, 1H), 1.19 (t, $J = 6.8$ Hz, 3H). $[\alpha]_{\text{D}}^{20} -122.4$ (c 0.12, MeOH). HRMS (ESI) m/z : $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{21}\text{H}_{26}\text{F}_3\text{N}_4\text{O}$ 407.2058; found 407.2043.

(-)-(3S,4R)-N-[(1S)-1-[4-(dimethylamino)phenyl]ethyl]-4-[4-(trifluoromethyl)phenyl]pyrrolidine-3-carboxamide (52b). The title compound was synthesized from **51b** and (*S*)-4-(1-aminoethyl)-*N,N*-dimethylbenzenamine as according to the method described for **53**. White solid, 95 mg (56% yield). ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ ppm 8.10 (d, $J = 8.4$ Hz, 1H), 7.65 (d, $J = 8.0$ Hz, 2H), 7.47 (d, $J = 8.0$ Hz, 2H), 7.06 (d, $J = 8.4$ Hz, 2H), 6.66 (d, $J = 8.4$ Hz, 2H), 4.79 (m, 1H), 3.49 (q, $J = 16.0$ Hz, 1H), 3.30 (1H, overlapped with the peak of H_2O), 3.18 (t, $J = 8.0$ Hz, 1H), 2.90-2.80 (m, 7H), 2.75 (dd, $J = 14.4, 8.8$ Hz, 1H). $[\alpha]_{\text{D}}^{20} -156.7$ (c 0.10, MeOH). HRMS (ESI) m/z : $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{22}\text{H}_{27}\text{F}_3\text{N}_3\text{O}$ 406.2106; found 406.2102.

(+)-(3R,4S)-N-[(1S)-1-[4-(dimethylamino)phenyl]ethyl]-4-[6-(trifluoromethyl)pyridin-3-yl]pyrrolidine-3-carboxamide (54a). The title compound was synthesized from **50a** and (*S*)-4-(1-aminoethyl)-*N,N*-dimethylbenzenamine as according to the method described for **53**. White solid, 78 mg (48% yield). ¹H NMR (400 MHz, DMSO-*d*₆) δ ppm 8.62 (s, 1H), 8.15 (d, *J* = 8.0 Hz, 1H), 7.90 (dd, *J* = 8.0 Hz, 1H), 7.81 (d, *J* = 8.0 Hz, 1H), 6.78 (d, *J* = 8.4 Hz, 2H), 6.51 (d, *J* = 8.8 Hz, 2H), 4.79 (m, 1H), 3.41 (m, 1H), 3.30 (2H, overlapped with the peak of H₂O), 3.20 (t, *J* = 7.2 Hz, 1H), 2.98 (m, 2H), 2.85 (t, *J* = 6.4 Hz, 1H), 2.83 (s, 6H), 1.24 (d, *J* = 7.2 Hz, 3H). [α]_D²⁰ +38.3 (*c* 0.12, MeOH). HRMS (ESI) *m/z*: [M + H]⁺ Calcd for C₂₁H₂₆F₃N₄O 407.2058; found 407.2048.

(+)-(3R,4S)-N-[(1S)-1-[4-(dimethylamino)phenyl]ethyl]-4-[4-(trifluoromethyl)phenyl]pyrrolidine-3-carboxamide (54b). The title compound was synthesized from **50b** and (*S*)-4-(1-aminoethyl)-*N,N*-dimethylbenzenamine as according to the method described for **53**. White solid, 125 mg (63% yield). ¹H NMR (500 MHz, DMSO-*d*₆) δ ppm 8.09 (d, *J* = 8.0 Hz, 1H), 7.61 (d, *J* = 8.0 Hz, 2H), 7.41 (d, *J* = 8.0 Hz, 2H), 6.83 (d, *J* = 8.5 Hz, 2H), 6.53 (d, *J* = 8.5 Hz, 2H), 4.81 (m, 1H), 3.39 (q, *J* = 17.0 Hz, 1H), 3.28 (t, *J* = 9.5 Hz, 1H), 3.19 (t, *J* = 8.0 Hz, 1H), 2.97 (t, *J* = 8.0 Hz, 1H), 3.39 (q, *J* = 16.0 Hz, 1H), 2.82 (s, 6H), 2.77 (t, *J* = 10.0 Hz, 1H), 1.25 (d, *J* = 8.0 Hz, 3H). [α]_D²⁰ +39.1 (*c* 0.14, MeOH). HRMS (ESI) *m/z*: [M + H]⁺ Calcd for C₂₂H₂₇F₃N₃O 406.2106; found 406.2098.

(+)-(3R,4S)-N-[(1R)-1-[4-(dimethylamino)phenyl]ethyl]-4-[6-(trifluoromethyl)pyridin-3-yl]pyrrolidine-3-carboxamide (55). The title compound was synthesized from **50a** and (*R*)-4-(1-aminoethyl)-*N,N*-dimethylbenzenamine as according to the method described for **53**. White solid, 102 mg (58% yield). ¹H NMR (400 MHz, DMSO-*d*₆) δ ppm 8.66 (s, 1H), 8.16 (d, *J* = 8.0 Hz, 1H), 7.95 (d, *J* = 8.0 Hz, 1H), 7.84 (d, *J* = 8.0 Hz, 1H), 7.05 (d, *J* = 8.0 Hz, 2H), 6.65 (d, *J* = 8.4

Hz, 2H), 4.78 (m, 1H), 3.50 (q, $J = 15.2$ Hz, 1H), 3.30 (2H, overlapped with the peak of H₂O), 3.20 (t, $J = 8.0$ Hz, 1H), 2.87-2.67 (m, 8H), 1.18 (d, $J = 6.8$ Hz, 3H). $[\alpha]_{\text{D}}^{20} +149.6$ (c 0.12, MeOH). HRMS (ESI) m/z : $[M + H]^+$ Calcd for C₂₁H₂₆F₃N₄O 407.2058; found 407.2057.

Biological Assays. Aspartic protease enzyme assays, mouse liver microsome assays, in vivo PK studies, in vivo efficacy studies were performed as we have previously described.¹¹

In vitro Antimalarial Assays (3D7 and Dd2). In vitro antimalarial activity was determined by a malaria SYBR Green I-based fluorescence (MSF) method described previously by Smilkstein et al.²⁹ with slight modification.³⁰ Stock solutions of each test drug were prepared in DMSO at a concentration of 20 mM. The drug solutions were serially diluted with culture medium and distributed to asynchronous parasite cultures on 96-well plates in quadruplicate in a total volume of 100 μ l to achieve 0.5% parasitemia with a 2% hematocrit in a total volume of 100 μ l. The plates were then incubated for 72 h at 37°C. After incubation, 100 μ l of lysis buffer with 0.2 μ l/ml SYBR Green I was added to each well. The plates were incubated at 37°C for an hour in the dark and then placed in a 96-well fluorescence plate reader (Spectramax Gemini- EM; Molecular Diagnostics) with excitation and emission wavelengths at 497 nm and 520 nm, respectively, for measurement of fluorescence. The 50% inhibitory concentration (IC₅₀) was determined by nonlinear regression analysis of logistic dose-response curves (GraphPad Prism software). Antimalarial potency of compounds was determined by this technique for both *Plasmodium falciparum* 3D7 (CQ-sensitive) and Dd2 (multi-drug resistant) strains. 3D7 assay results are given as an average of at least three replicates.

In vivo Antimalarial Efficacy Suppressive Assay. In vivo antimalarial activity was determined for **54b** against the rodent CQ-resistant *Plasmodium chabaudi* ASCQ strain according to the 4-day suppressive test.^{19, 31} Briefly, 5-week old female NIH mice ($n = 6$ per group) were

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3 inoculated i.p. with 2×10^7 parasitized (*Plasmodium chabaudi* ASCQ) red blood cells. Thereafter,
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5 the compounds were administered as a suspension in 0.5% carboxymethyl cellulose orally to the
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7 animals once daily at 4 h, 24 h, 48 h and 72 h post inoculation. Groups including a vehicle control
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9 and chloroquine (CQ) as a reference drug were included. Parasitemia levels were determined on
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11 the day following the last treatment (Day 4). Plasma drug concentration was determined by LC-
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13 MS using 100 μ L orbital blood plasma sample from Day 3, which was collected at 1 h, 6 h, 24 h
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15 after dosing.
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19 The Institutional Animal Care and Use Committee at the Guangzhou Institutes of Biomedicine
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21 and Health, Chinese Academy of Sciences, reviewed and approved the animal use in these studies.
22
23 The animal care and use program is run entirely according to Association for Assessment and
24
25 Accreditation of Laboratory Animal Care, International (AAALACi) standards and is assured by
26
27 the Office of Laboratory Animal Welfare (OLAW identification number A5748-01).
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31 **Fluorescence polarization hERG assay.** The inhibition activity of compounds on hERG
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33 potassium channel was determined using Predictor® hERG Fluorescence Polarization Assay Kit
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35 (Life Technologies, Carlsbad, CA, USA). Experiments were performed according to the
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37 manufacturer's instructions. In brief, the reactions were carried out in 384-well plates including
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39 10 μ L of Predictor® hERG Membrane and 10 μ L Predictor® hERG Tracer Red with appropriate
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41 amount of compound or positive control. Reactions were incubated for 2 h at room temp and then
42
43 read on an EnVision Multilabel Reader (Perkin Elmer, Inc.) using polarized excitation and
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45 emission filters. Data were analyzed using Graphpad Prism 5 (GraphPad Software Inc., San Diego,
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Supporting Information. Synthetic procedures and characterization data for final products **6**, **12-44**, and **56-62**, ^1H and ^{13}C NMR spectra for **54b**, and x-ray crystallographic data for **54b**. Molecular Formula Strings are provided as a .csv file. This information is available free of charge on the ACS Publications Website at DOI:xxx.

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Author Contributions. MJM designed compounds. JL, JX, FL, JG, ZL, HM and SAM synthesized compounds. LQ, LD, DA, SZ, and X Li tested compounds for antimalarial activity. AJP, ASN and ZT tested compounds for inhibition of plamepsins. X Liu determined PK of compounds. YL determined absolute stereochemistry of compounds. ZT determined the hERG activity of compounds. MJM, DEG, XC and MDT designed and directed the experiments. MJM and JL wrote the manuscript. All authors have approved of the final version of the manuscript.

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Abbreviations. ACT, artemisinin combination therapy; Pf, *Plasmodium falciparum*; PM, plasmepsin; SI, Selectivity Index; TCAMS, Tres Cantos Antimalarial Set; MLM, mouse liver microsomes; CQ, chloroquine; mg/kg/day, milligrams per kilogram.

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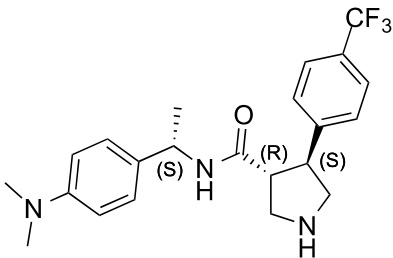
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Table of Contents Graphic



(+)-54b (CWHM-1008)

3D7 IC₅₀ = 46 ± 6 nM (n = 5)
Dd2 IC₅₀ = 21 ± 1 nM (n = 10)
murine PK: t_{1/2} = 4.4 h; F = 32%
murine *P. chabaudi* ED₉₀ ~20 mg/kg/day
ED₉₉ ~30 mg/kg/day