ENANTIOSELECTIVITY IN THE HYDROLYSIS OF BICYCLIC MONOTERPENE ACETATES WITH THE CULTURED CELLS OF *NICOTIANA TABACUM*

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Abstract—The enantioselectivity in the hydrolysis of bornyl acetate, isobornyl acetate and isopinocampheyl acetate with the cultured cells of *Nicotiana tabacum* was investigated. The cultured cells were found to have the ability to hydrolyse enantioselectively the acetates, of which the configuration at the carbon atom bearing the acetoxyl group is *R*.

INTRODUCTION

The cultured cells of Nicotiana tabacum have the ability to hydrolyse the acetoxyl group of linaloyl acetate [1] and the hydroxyimino group of carvoxime [2]. It was observed that the acetoxyl group of (+)- and (-)- α -terpinyl acetates tends to experience enantioselective hydrolysis in the cultured cells [3]. We have now investigated the enantioselectivity in the hydrolysis of the acetates having a rigid skeleton using a suspension of the cultured cells of N. tabacum. The acetates used as a substrate were (1R,2S,4R)-(+) and (1S,2R,4S)-(-)-2-acetoxy-1,7,7trimethylbicyclo[2.2.1]heptanes [trivial names: (+)and (-)-bornyl acetates, 1a and 1b], (1S,2S,4S)-(-)-(1R,2R,4R)-(+)-2-acetoxy-1,7,7-trimethylbicycloand [2.2.1]heptanes [(-)- and (+)-isobornyl acetates, 2a and 2b], and (1S,2S,3S,5R)-(+)- and (1R,2R,3R,5S)-(-)-3-acetoxy-2, 6, 6-trimethylbicyclo [3.1.1] heptanes [(+)- and (-)-isopinocampheyl acetates, 3a and 3b].

RESULTS AND DISCUSSION

The acetates were incubated with a suspension of the cultured cells of *N. tabacum* in a similar manner as described in refs [1, 4]. Figure 1 shows the results of the time-course experiments in the biotransformation of bornyl acetates (1a and 1b), isobornyl acetates (2a and 2b) and isopinocampheyl acetates (3a and 3c). These acetates were hydrolysed to their corresponding alcohols by the cultured cells, but the extent of hydrolysis was quite different between their enantiomers. The hydrolysis of (1S,2R,4S)-(-)-bornyl acetate (1b) predominated over that of (1R,2S,4R)-(+)-enantiomer (1a), as given in Fig. 1, A and B. Similarly, the hydrolysis of (1R,2R,4R)-(-)-isobornyl acetate (2a) was more predominant rather than



OH **6b** $R^1 = H, R^2 = OH$

that of its enantiomer (2b) (C and D of Fig. 1). In the case of isopinocampheyl acetate, (1R,2R,3R,5S)-(-)enantiomer (3b) was hydrolysed in preference to its enantiomer (3a) (E and F of Fig. 1).

In the biotransformation of (+)-bornyl acetate (1a) and (-)-isobornyl acetate (2b) (A and C of Fig. 1), the yield of the respective hydrolysed products (4a and 5a) decreased gradually as an increase in the yield of camphor (7a) with the lapse of the incubation time. This indicates the formation of 7a is due to the successive oxidation of 4a and 5a with the cultured cells. No formation of camphor was found in the biotransformation of (-)-bornyl acetate (1b) and (+)-isobornyl acetate (2a). Such a difference in the formation of camphor might be due to the enantioselectivity in the oxidation of the respective hydro-

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Fig. 1. The time-courses in the biotransformation of (A) (1R,2S,4R)-(+)-bornyl acetate (1a), (B) (1S,2R,4S)-(-)bornyl acetate (1b), (C) (1R,2R,4R)-(-)-isobornyl acetate (2a), (D) (1S,2S,4S)-(+)-isobornyl acetate (2b), (E) (1S,2S,3S,5R)-(+)-isopinocampheyl acetate (3a) and (F) (1R,2R,3R,5S)-(-)-isopinocampheyl acetate (3b).

lysed products, because the enantioselectivity in the oxidations of borneol and isoborneol with the cultured cells of *N. tabacum* is in preference to (1R,2S,4R)- and (1R,2R,4R)-enantiomers (4a and 5a), respectively, as described in our previous paper [5].

Thus, it was established that a suspension of the cultured cells of N. tabacum have the ability to hydrolyse enantioselectively the acetoxyl groups of bornyl acetate, isobornyl acetate and isopinocampheyl acetate; the cultured cells prefer the acetates with the *R*-configuration at the carbon atom bearing an acetoxyl group.

EXPERIMENTAL

Preparative TLC: silica gel (0.5 mm) developed with EtOAc-hexane (3:7). GLC: FID, glass column (3 mm × 2 m) packed with 15% DEGS or 2% OV-17 on Chromosorb W (AW-DMCS; 80-100 mesh) at 120° or 80-150° (3°/min), respectively.

Substrates. (+)-Bornyl acetate (1a) $\{[\alpha]_{D}^{25} + 40.0^{\circ} (c \ 1.00; EtOH)\}$ (lit. [6] + 44.1°). (-)-Bornyl acetate (1b) $\{[\alpha]_{D}^{25} - 39.0^{\circ} (c \ 1.00; EtOH)\}$ (lit. [7] - 44.5°). (+)-Isobornyl acetate (2b)

 ${[\alpha]_D^{25} + 1.5^\circ (c \ 1.00; EtOH)}$ (lit. [8] + 1.7°) and (-)-isobornyl acetate (2a) ${[\alpha]_D^{25} - 2.0^\circ (c \ 1.00; EtOH)}$ (lit. [9] - 1.8°) were prepared from the corresponding alcohols by acetylation with Ac₂O in pyridine, followed by purification by preparative TLC. The alcohols, such as (+) and (-)-borneols (4a and 4b) and (-)- and (+)-isoborneols (5a and 5b), were prepared by reduction of (+)- and (-)-camphors (7a and 7b) (Tokyo Kasei Kogyo Co. Ltd.) with LiAlH₄.

(+)-Isopinocampheyl acetate (3a) $\{[\alpha]_{D}^{25} + 35.0^{\circ} (c \ 1.00; EtOH)\}$ (lit. [10] + 37.5°) and (-)-isopinocampheyl acetate (3b) $\{[\alpha]_{D}^{25} - 35.0^{\circ} (c \ 1.00; EtOH)\}$ were synthesized by the acetylation of the corresponding alcohols, which had been previously prepared in our laboratory [11] by the hydroboration-oxidation [12] of (-)- and (+)- α -pinene, respectively. All of the substrates used were > 99.7% pure on GLC.

Time-courses in the biotransformation of the acetates. The callus tissues [1] of N. tabacum was transplanted to freshly prepared Murashige and Skoog's medium [13] (100 ml per a 300-ml conical flask) containing 2 ppm of 2,4-dichlorophenoxyacetic acid and 3% sucrose and then grown for 3-4 weeks at 25° on a rotary shaker (70 rpm) in the dark. To the flask containing a suspension of the cells (about 50-70 g fr. wt/flask), the substrate (10 mg) was administered, and cultures were incubated under continuous shaking at 25° for 10 days in the dark. At a regular time interval, a part (10 ml) of the incubated mixture was pipetted out under sterile conditions and extracted with Et₂O. Each Et₂O extract was subjected to GLC. The products were identified by comparison (TLC, GLC and co-GLC) with authentic samples. The yields of the products were determined on the basis of the peak area on GLC and are expressed as a relative percentage to the total amount of the whole reaction mixture extracted.

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