

Asymmetric Functionalization of Bicycloalkenes by Catalytic Enantioselective Hydrosilylation

Yasuhiro Uozumi, Sang-Yong Lee, and Tamio Hayashi*

Catalysis Research Center and Graduate School of Pharmaceutical Sciences, Hokkaido University, Sapporo 060, Japan

Abstract: Hydrosilylation of norbornene with trichlorosilane in the presence of palladium catalyst (0.01–0.1 mol %) coordinated with (*R*)-MOP ligand gave a quantitative yield of *exo*-2-trichlorosilylnorbornane, which was oxidized with hydrogen peroxide to give (1*S*,2*S*,4*R*)-*exo*-2-norbornanol in 96% ee. The similar hydrosilylation and oxidation of *endo*-5,6-dicarbomethoxy-2-norbornene, bicyclo[2.2.2]octene, and norbornadiene gave the corresponding bicyclic alcohols of 94% ee, 92% ee, and 95% ee, respectively.

Asymmetric synthesis through a selective monofunctionalization of enantiotopic positions is one of the most attractive strategies for one-step construction of multiple chiral carbon centers.¹ In spite of the impressive development of enantioface selective asymmetric reactions catalyzed by transition metal complexes, the enantioselective approach still remains to be developed.² We have concentrated our studies on the catalytic asymmetric functionalization of meso bicyclo[2.2.1] system, because the optically active bicyclo[2.2.1]heptane derivatives represented by norbornanol are of great value as versatile chiral building blocks for the synthesis of a wide variety of important compounds.³ Those optically active bicyclo[2.2.1]heptanes have been mainly obtained by optical resolution of racemic compounds in either chemical or enzymatic procedures,⁴ or by asymmetric hydroboration⁵ and Diels-Alder reactions⁶ using a stoichiometric amount of chiral auxiliaries. Use of catalytic systems for the asymmetric reactions has not always been successful in terms of enantioselectivity or catalytic activity.^{2,7,8} We report here that the asymmetric functionalization with >96/4 enantioselectivity is realized through asymmetric hydrosilylation in the presence of not more than 0.1 mol % of palladium catalyst coordinated with (*R*)-2-methoxy-2'-diphenylphosphino-1,1'-binaphthyl ((*R*)-MOP).^{9,10}

Scheme 1

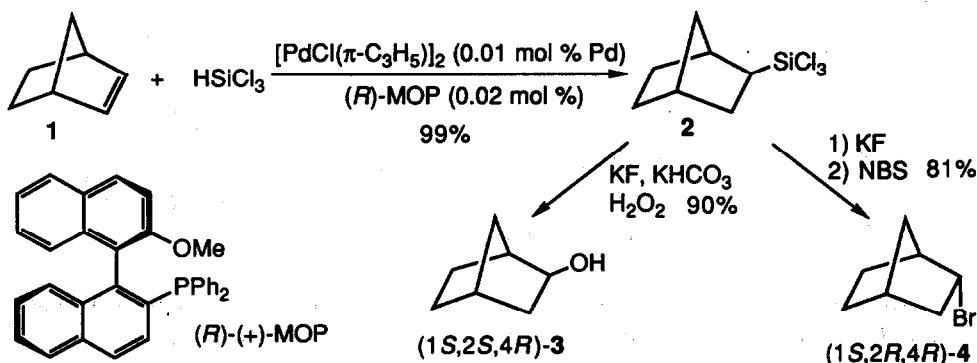


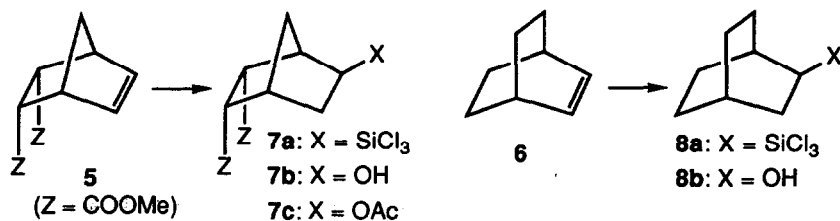
Table I. Asymmetric Hydrosilylation Catalyzed by Palladium-MOP^a

entry	olefin	conditions	product	yield ^b % (<i>exo</i> : <i>endo</i>) ^c	yield ^b % of alcohol	% ee
1	1	0 °C, 24 h	2	100 (100 : 0)	90 (3)	93 ^d
2	1	-20 °C, 3 d	2	99 (100 : 0)		96 ^d
3	5	0 °C, 24 h	7a	100 (100 : 0)	96 (7b)	94 ^e
4	6	0 °C, 24 h	8a	85 (—)	90 (8b)	92 ^d
5	9	0 °C, 24 h	10a	85 ^f (100 : 0)	89 (10b)	95 ^d

^a All reactions were run without solvent in the presence of palladium catalyst prepared in situ by mixing [PdCl(π -C₃H₅)]₂ (0.01–0.1 mol % Pd) and (*R*)-MOP (2 equiv to Pd). The ratio of olefin/HSiCl₃ is 1/1.20–1.25. ^b Isolated yield. ^c Determined by GLC and ¹H NMR analysis. ^d Determined by HPLC analysis of (3,5-dinitrophenyl)carbamate of the alcohol with Sumichiral OA-4500 (n-hexane/dichloroethane/ethanol = 50/10/1). ^e Determined by ¹H NMR analysis of acetate 7c using Eu(hfc)₃. ^f Nortricyclene 11 was also formed in 14%.

A typical procedure for the asymmetric synthesis of *exo*-2-norbornanol (3) from norbornene (1) (Scheme 1) is as follows: A mixture of norbornene (1, 15.0 g, 0.16 mol), trichlorosilane (20.0 mL, 0.20 mol), [PdCl(π -C₃H₅)]₂ (2.9 mg, 0.008 mmol, 0.01 mol % Pd) and (*R*)-MOP (14.8 mg, 0.032 mmol, 2 equiv to Pd) was stirred at 0 °C for 24 h. Removal of excess silane followed by distillation (65 °C/3.5 mm Hg) gave 100% yield (36.5 g) of *exo*-2-trichlorosilylnorbornane^{11,12} (2) as a single product. Oxidative conversion of 2 was performed with hydrogen peroxide by a modified Tamao's method^{13,14} to give *exo*-2-norbornanol (3) in over 90% yield. Sublimation in vacuo gave 13.3 g (74% yield) of analytically pure (1*S*,2*S*,4*R*)-3 with 93% ee. The absolute configuration was assigned on the basis of the optical rotation (3: [α]_D²⁵ -2.94° (c 10.55, CHCl₃), lit.¹⁵ [α]_D²⁵ -3.14° (c 3.1, CHCl₃)) and the enantiomeric excess was determined by HPLC analysis¹⁶ of the carbamate ester obtained by treatment with 3,5-dinitrophenyl isocyanate. The hydrosilylation carried out at -20 °C for 3 days (99% yield) raised the enantiomeric excess to 96% ee (entries 1 and 2 in Table I).

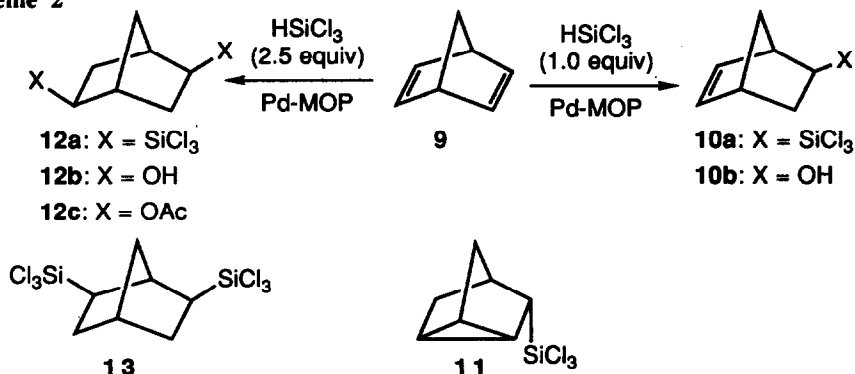
Trichlorosilane 2 can be converted into (1*S*,2*R*,4*R*)-*endo*-2-bromonorbornane (4)^{17,18} in 81% yield by treatment with excess potassium fluoride followed by bromination of the resulting pentafluorosilicate with *N*-bromosuccinimide.¹⁹ Dimethyl ester derivative 5 and bicyclo[2.2.2]octene 6 were also successfully subjected to the asymmetric hydrosilylation-oxidation under the similar reaction conditions to give the corresponding alcohols, (1*R*,2*S*,4*R*,5*S*,6*R*)-7b (94% ee)^{20,21} and (2*S*)-8b (92% ee),^{22,23} respectively (entries 3 and 4).



It is remarkable that the monofunctionalization of norbornadiene (9) forming *exo*-2-trichlorosilyl-5-norbornene (10a) is effected by the palladium-MOP catalyst with high chemo- and enantioselectivity (Scheme 2). It is in striking contrast to the reaction catalyzed by chloroplatinic acid²⁴ or palladium-triphenylphosphine²⁵

which gives nortricyclene **11** as a major product. Thus, the reaction of **9** with 1.0 equiv of trichlorosilane and the palladium-MOP catalyst (0.1 mol %) followed by the hydrogen peroxide oxidation gave (1*R*,2*S*,4*R*)-*exo*-2-hydroxy-5-norbornene²¹ (**10b**) with 95% ee (entry 5). The enantioselective hydrosilylation took place successively on the two double bonds of **9** in the reaction with 2.5 equiv of trichlorosilane, which gave 78% yield of chiral disilylnorbornane **12a** and meso isomer **13** in a ratio of 18 : 1. The oxidation of **12a** followed by acetylation of diol **12b** gave diacetate (1*R*,2*S*,4*R*,5*S*)-**12c**^{26,27} with >99% ee,²⁸ the high purity being as expected in the double stereoselection.

Scheme 2



Acknowledgment. We thank the Ministry of Education, Japan, for a Grant-in-Aid for Scientific Research and Asahi Glass Foundation for partial financial support of this work.

References and Notes

- For reviews: (a) Morrison, J. D., Ed. *Asymmetric Synthesis*; Academic Press: London, 1983–1985; Vol. 1–5. (b) N6grádi, M. *Stereoselective Synthesis*; Weinheim: New York, 1987.
- For reviews: (a) Noyori, R.; Kitamura, M. *Modern Synthetic Methods*; Scheffold, R., Ed.; Springer-Verlag: Berlin, 1989; Vol. 5, p 115. (b) Brunner, H. *Top Stereochem.* **1988**, *18*, 129. (c) Ojima, I.; Clos, N.; Bastos, C. *Tetrahedron* **1989**, *45*, 6901.
- They have been employed, for example, for the synthesis of prostanoids (ref 3a–b), thromboxanes (ref 3c–g), alkaloids (ref 3h), terpenes (ref 3i), insecticides (ref 3j), antibiotics (ref 3k), fragrances (ref 3l), and carbosugars (ref 3m–n). (a) Arndt, H. C.; Rajani, C. *Tetrahedron Lett.* **1982**, *23*, 2365. (b) Barraclough, K. *Tetrahedron Lett.* **1980**, *21*, 1897. (c) Narisada, M.; Ohtani, M.; Watanabe, F.; Uchida, K.; Arita, H.; Doteuchi, M.; Hanasaki, K.; Kakushi, H.; Otani, K.; Hara, S. *J. Med. Chem.* **1988**, *31*, 1847. (d) Martinelli, M. *J. Org. Chem.* **1990**, *55*, 5065. (e) Garland, R.; Miyano, M.; Pireh, D.; Clare, M.; Finnegan, P. M.; Swenton, L. *J. Org. Chem.* **1990**, *55*, 5854. (f) Ohtani, M.; Matsuura, T.; Watanabe, F.; Narisada, M. *J. Org. Chem.* **1991**, *56*, 2122. (g) Ohtani, M.; Matsuura, T.; Watanabe, F.; Narisada, M. *J. Org. Chem.* **1991**, *56*, 4120. (h) Takano, S.; Hatakeyama, S.; Ogasawara, K. *Tetrahedron Lett.* **1978**, 2519. (i) Van der Eycken, J.; Vandewalle, M.; Heinemann, G.; Laumen, K.; Schneider, M. P.; Kredel, J.; Sauer, J. *J. Chem. Soc., Chem. Commun.* **1989**, 306 and references cited therein. (j) Ley, S. V.; Santafianos, D.; Blaney, W. M.; Simmonds, M. S. *J. Tetrahedron Lett.* **1987**, *28*, 221. (k) Drian, C. L.; Greene, A. E. *J. Am. Chem. Soc.* **1982**, *104*, 5473. (l) Krotz, A.; Helmchen, G. *Tetrahedron Asymmetry* **1990**, *1*, 537. (m) Marschner, C.; Penn, G.; Griengl, H. *Tetrahedron Lett.* **1990**, *31*, 2873. (n) Baumgartner, H.; Marschner, C.; Pucher, R.; Griengl, H. *Tetrahedron Lett.* **1991**, *32*, 611.

- 4 (a) Klunder, A. J. H.; van Gastel, F. J. C.; Zwanenburg, B. *Tetrahedron Lett.* **1988**, *29*, 2697. (b) Metz, P. *Tetrahedron* **1989**, *45*, 7311. (c) Janssen, A. J. M.; Klunder, A. J. H.; Zwanenburg, B. *Tetrahedron Lett.* **1990**, *31*, 7219.
- 5 (a) Brown, H. C.; Desai, M. C.; Jadhav, P. K. *J. Org. Chem.* **1982**, *47*, 5065. (b) Joshi, N. N.; Pyun, C.; Mahindroo, V. K.; Singaram, B.; Brown, H. C. *J. Org. Chem.* **1992**, *57*, 504.
- 6 (a) Furuta, K.; Iwanaga, K.; Yamamoto, H. *Tetrahedron Lett.* **1986**, *27*, 4507. (b) Furuta, K.; Hayashi, S.; Miwa, Y.; Yamamoto, H. *Tetrahedron Lett.* **1987**, *28*, 5841. (c) Helmchen, G.; Karge, R.; Weetman, J. *Modern Synthetic Methods*; Scheffold, R., Ed.; Springer-Verlag: Berlin, **1986**; Vol. 4, p 262 and references cited therein. (d) Hartmann, H.; Hady, A. F. A.; Sartor, K.; Weetman, J.; Helmchen, G. *Angew. Chem., Int. Ed. Engl.* **1987**, *26*, 1143. (e) Oppolzer, W.; Chapuis, C.; Dupuis, D.; Guo, M. *Helv. Chem. Acta* **1985**, *68*, 2100 and references cited therein. (f) Mattay, J.; Mertes, J.; Maas, G. *Chem. Ber.* **1989**, *122*, 327.
- 7 (a) Burgess, K.; Ohlmeyer, M. J. *J. Org. Chem.* **1988**, *53*, 5178. (b) Burgess, K.; van der Donk, W. A.; Ohlmeyer, M. J. *Tetrahedron Asymmetry* **1991**, *2*, 613. (c) Sato, M.; Miyaura, N.; Suzuki, A. *Tetrahedron Lett.* **1990**, *31*, 231. (d) Corey, E. J.; Loh, T-P. *J. Am. Chem. Soc.* **1991**, *113*, 8966.
- 8 Catalytic systems so far reported (ref 7) usually require 1–10 mol % of catalyst. The present reaction proceeds smoothly with 0.01 mol % of catalyst.
- 9 Uozumi, Y.; Hayashi, T. *J. Am. Chem. Soc.* **1991**, *113*, 9887.
- 10 An enantioface selective hydrosilylation of 1-alkenes has been reported by the palladium-MOP catalyst (ref 9).
- 11 (a) Kuivila, H. G.; Warner, C. R. *J. Org. Chem.* **1964**, *29*, 2845. (b) Green, M.; Spencer, J. L.; Stone, F. G. A.; Tsipis, C. A. *J. Chem. Soc., Dalton* **1977**, 1159.
- 12 Previous work by use of a chiral ferrocenylphosphine ligand gave moderate chemical and optical yields: Hayashi, T.; Tamao, K.; Katsuro, Y.; Nakae, I.; Kumada, M. *Tetrahedron Lett.* **1980**, *21*, 1871.
- 13 (a) Tamao, K.; Ishida, N. *J. Organomet. Chem.* **1984**, *269*, C37. (b) Tamao, K.; Nakajo, E.; Ito, Y. *J. Org. Chem.* **1987**, *52*, 4412 and references cited therein.
- 14 To a suspension of KF (6 equiv) and KHCO₃ (9 equiv) in THF/MeOH (1/1) was added successively 2 (1 equiv) and 30% H₂O₂ (7 equiv) at 0 °C, and the mixture was stirred at ambient temperature for 15 h.
- 15 Irwin, A. J.; Jones, J. B. *J. Am. Chem. Soc.* **1976**, *98*, 8476.
- 16 Sumichiral OA-4500 (eluent: n-hexane/dichloroethane/ethanol = 50/10/1).
- 17 $[\alpha]^{25}_{\text{D}} +16.7^{\circ}$ (c 1.4, CHCl₃).
- 18 Bach, R. D.; Holubka, J. W.; Taaffee, T. H. *J. Org. Chem.* **1979**, *44*, 35.
- 19 Tamao, K.; Yoshida, J.; Murata, M.; Kumada, M. *J. Am. Chem. Soc.* **1980**, *102*, 3267.
- 20 Optical rotation of *exo*-2-trimethoxysilyl derivative which was prepared by methanolysis of 7a is $[\alpha]^{25}_{\text{D}} +14.7^{\circ}$ (c 1.3, benzene).
- 21 The absolute configuration was tentatively assigned by similarity in shifts using chiral shift reagent Eu(hfc)₃ or in elution order in the HPLC analysis.
- 22 $[\alpha]^{27}_{\text{D}} +30.2^{\circ}$ (c 1.0, CHCl₃).
- 23 Nakazaki, M.; Chikamatsu, H.; Naemura, K.; Asao, M. *J. Org. Chem.* **1980**, *45*, 4432.
- 24 The ratio of 10/11 was reported to be 1/2.1: Kuivila, H. G.; Warner, C. R. *J. Org. Chem.* **1964**, *29*, 351.
- 25 The reaction in the presence of [PdCl(π -C₃H₅)]₂ (1 mol % Pd) and PPh₃ (2 equiv to Pd) at 50 °C for 12 h gave 10 (37%) and 11 (56%).
- 26 $[\alpha]^{20}_{\text{D}} +7.60^{\circ}$ (c 0.16, CHCl₃).
- 27 Naemura, K.; Takahashi, N.; Ida, H.; Tanaka, S. *Chem. Lett.* **1991**, 657.
- 28 The other enantiomer was not detected by ¹H NMR analysis using chiral shift reagent Eu(hfc)₃.