Chemistry Letters 1998

Total Synthesis of Sollasin a and Sollasin d via Photocycloaddition of Methyl 2.4-Dioxopentanoate to Methyl E-2-Methyl-2-butenoate

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(Received October 16, 1997; CL-970798)

Starting from photocycloaddition of methyl 2,4-dioxopentanoate to methyl tiglate, an electron-deficient olefin, methyl-migrated monocyclofarnesyl sesquiterpenoids, *d*,*l*-sollasin a and d, antibacterial compounds, were synthesized.

Methyl 2,4-dioxopentanoate (1) is a versatile C_5 photosynthon to obtain homologous isoprenoids; 1 e.g., iridanes from 1 and isoprene, 2 cyclonerodanes from 1 and myrcene. 3 Particularly, its usefulness has been verified by the reactions with conjugated olefins $^{2-5}$ or by easy furnishment of five-membered carbocycles 6,7 in high regioselectivity. However, the photocycloadditions examined have been limited to that with olefins having electron-donating substituents, such as cycloalkenes, 8 alkyl- and alkoxyethenes, 9 and arylethenes. 10 Therefore, it is worthy of investigating the reactivity and regioselectivity of the photochemical reaction of 1 with an electron-deficient olefin, methyl E-2-methyl-2-butenoate (methyl tiglate, 2a).

Irradiation of a benzene solution 1 and 2a in a quartz vessel

under nitrogen atmosphere for 3 h afforded, after silica-gel chromatography, five photoproducts (3, 43%, 4, 3.5%, 5, 8%, 6, 1.5%, and 7, 3%). Among them, 3, 4, and 5 were *retro*-aldolized [2+2] cycloadducts; 3 and 4 were diastereomers each other, and 5 was their regioisomer.

In the 1 H NMR spectra, **5** showed an AB-type pair of doublets, J=18 Hz, while **3** and **4** showed an each ABX-spin system. Therefore, their regiochemistry was assigned as depicted in Scheme 1. Assuming a retention of geometrical structure of **2a** during the reaction, the major product, **3**, was assigned as $(3R^*,4S^*)$ -derivative, and **4**, as $(3R^*,4R^*)$ -. This assignment was supported by NOE measurement. The structure of **3** suggested that the cycloaddition was controlled by not the electronic effect, but the stability of the delocalized diradical intermediate; *i.e.*, less-substituted site of the olefins formed a linkage to 3-position of **1**. 11

This means that the tetrasubstituted cyclohexenone from the major product 3 could be employed for synthesis of methyl-migrated monocyclofarnesyl derivatives, e.g., sollasin d, isolated from a marine sponge, *Poecillastra sollasi*, ¹², ¹³ as an antibacterial compounds against *Candida albicans* and *Cryptococcus neoformans*. Herein the total synthesis of sollasin d as well as sollasin a in the racemic form is described. Recently, Angers and Canonne synthesized sollasin a. ¹⁴

The cyclization of 3 gave a desired cyclohexenone (8). Removal of the oxygen functions was then achieved *via* diisobutylaluminum hydride reduction of 8 to a triol (9), acetylation to a triacetate (10), Birch reduction (with lithium in liquid ammonia in the presence of 2-propanol) to a (1,2,6-trimethylcyclohex-2-enyl)methanol (11), and PDC-oxidation to an aldehyde (12). The Grignard reaction with allylmagnesium chloride of 12 gave a homoallyl alcohol (13), and the Wacker reaction 15 with its acetate

$$\label{eq:conditional_condition} \begin{split} \text{Reagents: a) TsOH/ benzene; b) DIBAH/ toluene; c) } &\text{Ac}_2\text{O/ pyridine; d) Li/ liq. NH}_3, \textit{i-PrOH}; e) PDC/ CH}_2\text{Cl}_2; f) \\ &\text{CH}_2\text{=CHCH}_2\text{MgCI/ THF; g) } &\text{O}_2, \text{ CuCI, PdCI/ aq. DMF; h) TsOH/CH}_2\text{Cl}_2; j) } &\text{(C}_8\text{H}_{17})_3\text{NMeCI, NaHCO}_3, Na}_2\text{S}_2\text{O}_4. \end{split}$$

Scheme 2.

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(14) afforded a β -acetoxy ketone (15). Elimination and dithionite-reduction under phase-transfer conditions¹⁶ of 15 afforded a common precursor (16).

The final step of the synthesis was accomplished by Wittig-Horner reaction; 17 i.e., 16 was condensed at room temperature for 72 h with the anion prepared from methyl (diethoxyphosphoryl)acetate 17 to give sollasin a (18) 18 in 70% yield, and with the anion carrying α -pyrone chromophore (19) to give sollasin d (20) in 71% yield. The spectral data stated in literature 12 were identical within experimental error to our totally synthesized samples.

It is interesting to note that the irradiation of 1 with methyl Z-2-methyl-2-butenoate (methyl angelate, 2b), afforded two common major products, 3 and 4 in different ratio; the ratio of 3:4 was 3:7 after 15 h, and 4:6 after 30 h, and furthermore, 2a was detected in the recovered butenoate fractions. Irradiation of 2b without adding 1 caused no photoisomerization. Consequently, the isomerizations between 2a and 2b should occur via an exciplex formation $(1\cdots 2)^*$.

$$\begin{bmatrix} 2b & & & & & & & & & & & & \\ & 2b & & & & & & & & \\ & 2b & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ &$$

Finally, the present procedure will offer an applicability of 1 as a photosynthon to prepare sterically crowded alicyclics.

References and Notes

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- 18 The new compounds are fully characterized. Due to a limited space available, only the ¹H and ¹³C NMR spectra of **18** and **20** in CDCl₃ are compiled. The discrepancies of the ¹³C NMR chemical shifts from the figures cited in ref. 12 are shown in brackets.
 - **18**: δ (H)=0.86(3H, d, J=7.0 Hz), 0.87(3H, s), 1.46(2H, m), 1.52(2H, m), 1.60(3H, td, J=2.3, 1.1 Hz), 1.67(1H, m), 1.68(1H, m), 1.79(1H, m), 1.95(2H, m), 2.07(1H, m), 2.17(3H, d, J=1.1 Hz), 3.68(3H, s), 5.44(1H, br s), and 5.66(1H, qm, J=1.1 Hz); δ (C)=15.8(+0.1), 19.1(2C, ±0), 21.0(±0), 25.5(±0), 26.9(±0), 33.3(±0), 34.4(±0), 35.7(±0), 40.4(-0.1), 50.8(±0), 114.6(±0), 124.7(±0), 139.0(±0), 161.6(±0), and 167.3(±0).
 - **20**: $\delta(H)$ =0.88(3H, d, J=6.0 Hz), 0.89(3H, s), 1.45(2H, m), 1.54(2H, m), 1.61(3H, td, J=2.3, 1.1 Hz), 1.70(1H, m), 1.81(1H, m), 1.95(2H, m), 2.09(1H, m), 2.12(3H, brs), 3.80(3H, s), 5.41(1H, d, J=2.2 Hz), 5.45(1H, brs), 5.77(1H, brs), and 5.78(1H, d, J=2.2 Hz); $\delta(C)$ =15.8 (±0), 19.1(±0), 19.6(±0), 21.0(±0), 27.0 (+0.1), 25.5 (±0), 33.3 (±0), 34.7(±0), 36.5(±0), 40.5(±0), 55.7(±0), 87.4 (±0), 100.7(±0), 116.3(±0), 124.6(-0.1), 139.0(±0), 151.3(-0.1), 160.4(±0), 164.5(±0), and 171.4(±0).