

Available online at www.sciencedirect.com



CARBOHYDRATE RESEARCH

Carbohydrate Research 338 (2003) 303-306

www.elsevier.com/locate/carres

# Synthesis of 1,2,3-tri-O-acetyl-5-deoxy-D-ribofuranose from D-ribose

Pothukuchi Sairam,\* Ramachandra Puranik, Bhatraju Sreenivasa Rao, Ponnapalli Veerabhadra Swamy, Sharad Chandra

Critical Care Division, Dr Reddy's Laboratories Ltd., IDA Bollaram, Hyderabad, Andhra Pradesh 500 016, India

Received 31 May 2002; accepted 3 November 2002

### Abstract

A practical route towards the synthesis of 1,2,3-tri-O-acetyl-5-deoxy-D-ribofuranose from D-ribose is described. The key steps include deoxygenation of methyl 2,3-O-isopropylidene-5-O-sulfonyloxy- $\beta$ -D-ribofuranoside by reductive displacement employing hydride reagents. Subsequent total hydrolysis followed by acetylation led to the title compound in 56% overall yield from D-ribose. The sequence is simple, inexpensive, high yielding and clearly suitable for multi-gram preparations. © 2003 Elsevier Science Ltd. All rights reserved.

Keywords: 1,2,3-tri-O-acetyl-5-deoxy-D-ribofuranose; Lithium tri-t-butoxyaluminum hydride (LTTBA); Sodium bis-(methoxyethoxy) aluminium hydride (SMEAH)

# 1. Introduction

5-Deoxy-D-ribose is key constituent of several nucleosides, which exhibit a variety of biological functions. For example 5'-deoxy-iodotubercidin  $(1)^1$  is a nucleoside derivative of 5-deoxy-D-ribose. Compound 1 is a good inhibitor of adenosine kinase, which provides basis for the design of new therapeutic agents for the treatment of inflammatory disease. The prodrug strategy involving tumor-selective delivery of 5-fluorouracil by sequential conversion of a fluoropyrimidine carbamate, capecitabine  $(2)^2$  also requires the tri-O-acetyl derivative of 5-deoxy-D-ribose, namely 1,2,3-tri-O-acetyl-5-deoxy-D-ribofuranose, for its synthesis. Our involvement with this latter compound as a starting material, in a linear synthetic route, required examination and modification of existing chemistry to enable a practical bulk synthesis of this compound (Fig. 1).

# 2. Results and discussion

There are two major synthetic routes available in literature for the synthesis of this intermediate. The first one focuses on preparing a methyl 2,3-O-isopropylidene-5halo- $\beta$ -D-ribofuranoside<sup>3-5</sup> from methyl 2,3-O-isopropylidene- $\beta$ -D-ribofuranoside,<sup>6</sup> followed by reductive dehalogenation employing either palladium-on-charcoal, or Adams platinum catalyst, yielding 5-deoxy-Dribose.<sup>4,7</sup> The other route follows the direct deoxygenation of a methyl 2,3-O-isopropylidene-5-Osulfonyloxy- $\beta$ -D-ribofuranoside,<sup>8-10</sup> by reductive displacement using a hydride reagent to give 5-deoxy-D-ribose.<sup>11</sup> Other modification of this general





<sup>\*</sup> Corresponding author *E-mail address:* sairam@drreddys.com (P. Sairam).



Scheme 1. Reagents: (i)  $SnCl_2 2H_2O$ ,  $Conc.H_2SO_4$ , acetone, MeOH; (ii) Mesyl chloride, pyridine,  $CH_2Cl_2$  (Compound 5); **OR** tosyl chloride, pyridine,  $CH_2Cl_2$  (Compound 5); **OR** triffic anhydirde, pyridine,  $CH_2Cl_2$  (Compound 7); (iii) NaBH<sub>4</sub>, Me<sub>2</sub>SO (compound 5,6,7); **OR** SMEAH, THF (compound 5,6,7); **OR** LTTBA, THF (compound 5,6,7); (iv) 0.04N H<sub>2</sub>SO<sub>4</sub>; Ac<sub>2</sub>O, pyridine.

Table 1 Percentage yields of isolated products (4 and 8) obtained by reaction of compounds 5, 6, and 7 with hydride reagents

Hydride reagent	Compound (5)		Compound (6)		Compound (7)	
	4	8	4	8	4	8
NaBH₄	_	80	_	89	70	20
SMEAH	70	_	45	50	30	50
LTTBA	75	-	70	20	_	95

approach required additional steps, which render them unattractive candidates for a large-scale process.<sup>12,13</sup>

Opting the later method, in the present article, we report a comparative study of deoxygenation of three different sulfonyloxy-activated sugar derivatives using three different hydride reagents, two of which are being employed for the first time for such a conversion. Commercially available D-ribose (3) is converted into methyl 2.3-*O*-isopropylidene- $\beta$ -D-ribofuranoside (4) by a known method.<sup>6</sup> The hydroxyl group at C-5 carbon of compound 4 was activated using three different sufonyl protecting agents viz, methanesulfonyl chloride,8 p-toluenesulfonyl chloride,9 and trifluoromethanesulfonic anhydride<sup>10</sup> to give compounds 5, 6, and 7. Compounds 5, 6, and 7 were independently reduced by three different hydride reagents, namely, sodium borohydride (NaBH<sub>4</sub>),<sup>11</sup> lithium tri-t-butoxyaluminum hydride (LTTBA), and sodium bis(methoxyethoxy) aluminium hydride (SMEAH), to yield 2,3-O-isopropylidene-5-deoxy-β-D-ribofuramethyl noside (8). Further simplification of reported procedures<sup>4,5</sup> adopted for total hydrolysis and subsequent acetylation yielded the desired intermediate 1,2,3tri-O-acetyl-5-deoxy-D-ribofuranose (9) in 56% overall

yield from D-ribose (Scheme 1). The best yield reported<sup>14</sup> earlier for the synthesis of title compound was 52%, and involved hazardous operations like bromination and catalytic hydrogenation.

The main drawback of displacement of sulfonyloxy activated sugar hydroxyl groups by hydride reagents is that they often undergo O–S cleavage<sup>15</sup> to give back the starting compounds. The results of reduction by different hydride reagents of compounds **5**, **6**, and **7** yielded either the 5-deoxy sugar derivative **8** or the product of O–S cleavage, namely the 5-hydroxy sugar derivative **4**, and are summarized in Table 1.

#### 3. Conclusion

We have successfully developed an efficient four-step process for the synthesis of 1,2,3-tri-O-acetyl-5-deoxy-D-ribofuranose. The reaction conditions are operationally simple, robust, and amenable to multi-gram scale. The most favored method of activating the protected ribose 4 would be *p*-toluenesulfonyl chloride to give 6, as two other sulfonylating reagents gave either lower yields or were labile and difficult to handle. Subsequent deoxygenation of the activated sugar derivative 6 was best achieved by using  $NaBH_4$  which yields compound 8 in high yields in simple and cost-effective way. Total hydrolysis of 8 followed by acetylation yields the title compound.

### 4. Experimental

### 4.1. General methods

Melting points were determined with a Büchi 535 apparatus and are uncorrected. IR spectra were recorded on Perkin–Elmer 1650 FTIR. <sup>1</sup>H NMR (200 MHz) and <sup>13</sup>C NMR (50 MHz) spectra were taken with a Varian Gemini FT-NMR instrument in CDCl<sub>3</sub> and Me<sub>2</sub>SO-d<sub>6</sub> using Me<sub>4</sub>Si as internal standard. Mass spectra were recorded on a, Hewlett-Packard model-5989 instrument with a direct insertion probe at 20 eV. Gas chromatograms were taken with a Shimadzu GC 17A instrument. TLC was performed on Silica Gel 60 F<sub>254</sub> 230 mesh (E. Merck); detection was executed by spraying with a solution of Ce(NH<sub>4</sub>)<sub>2</sub>(NO<sub>3</sub>)<sub>6</sub>, (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub>, as well as H<sub>2</sub>SO<sub>4</sub> in water and subsequent heating on a hot plate.

# 4.2. Preparation of Methyl-2,3-O-isopropylidene- $\beta$ -D-ribofuranoside (4)

A mixture of D-ribose (3, 250 g, 1.66 mol) and  $SnCl_2 \cdot 2H_2O$  (375 g, 1.66 mol) were suspended in acetone (5 L) and methanol (1.3 L) with catalytic amount of conc.H<sub>2</sub>SO<sub>4</sub> (18.7 g, 186 mmol) and heated at 40–45 °C for 20 h. The mixture was filtered, and the filtrate was neutralized (pH 6–7) with NaHCO<sub>3</sub> solution. The resulting solution was once again filtered through a Celite bed and evaporated to remove acetone and MeOH. The aqueous solution thus obtained was extracted with EtOAc, washed with brine, dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated in vacuo to yield **4** (300 g, 88% yield). <sup>1</sup>H NMR data were in agreement with those reported in the literature.<sup>6</sup>

## 4.3. Preparation of methyl-2,3-*O*-isopropylidene-5-*O*mesyl-β-D-ribofuranoside (5)

Compound 4 (2 g, 9 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) and pyridine (6 mL, 73 mmol) was added to methanesulfonyl chloride (2 mL, 25 mmol) and stirred at 0-5 °C for 16 h. The reaction mixture was successively washed with 1N HCl, water, NaHCO<sub>3</sub>, brine. The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to yield a syrupy mass, which was further crystallized from Et<sub>2</sub>O to give **5** (1.8 g, 65% yield) as a white solid. <sup>1</sup>H NMR data was in agreement with those reported in the literature.<sup>8</sup>

# 4.4. Preparation of methyl-2,3-*O*-isopropylidene-5-*O*-tosyl-β-D-ribofuranoside (6)

Compound 4 (350 g, 1.71 mol) dissolved in  $CH_2Cl_2$  and pyridine on reaction with *p*-toluenesulfonyl chloride (630 g, 3.3 mol) yielded 6 (546 g, 89% yield) as a white solid. <sup>1</sup>H NMR data were in agreement with those reported in the literature.<sup>9</sup>

# 4.5. Preparation of methyl-2,3-*O*-isopropylidene-5-*O*trifluoromethanesulfonyl-β-D-ribofuranoside (7)

Following the procedure just mentioned, compound 4 (1 g, 4 mmol) in  $CH_2Cl_2$  (10 mL) containing pyridine (0.7 mL, 9 mmol) on reaction with trifluoromethanesulfonic anhydride (1.2 mL, 73 mmol) at 0 °C for 4 h yielded 7 (0.8 g, 50%) as an colourless oil. Due to thermal decomposition at room temperature, no spectral evidence could be obtained.<sup>10</sup>

## 4.6. Preparation of methyl-2,3-*O*-isopropylidene-5-deoxy-β-D-ribofuranoside (8)

Compound 5 (10.0 g, 35 mmol) in  $Me_2SO$  (50 mL), was reacted with NaBH<sub>4</sub> (6.1 g, 175 mmol) for 12 h at 80-85 °C. The mixture was cooled and 5% aqueous AcOH (100 mL) was added. The mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>, washed with water, dried (Na<sub>2</sub>SO<sub>4</sub>), concentrated, and purified by column chromatography (10% EtOAc in petroleum ether) to yield 8 (5.3 g, 80%). The NMR data were in agreement with those reported in the literature;<sup>13</sup>  $[\alpha]_{D}^{23} - 110^{\circ}$  (c 2, EtOH) [lit<sup>5</sup>  $[\alpha]_{D}^{23}$ -109° (c 2, EtOH)]; IR (neat): 2985, 2938, 1210, 1101, 1057, 870 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.25 (d, 3 H,  $J_{5,4}$ 7, H-5), 1.45 and 1.29 (2s, each 3 H, CMe<sub>2</sub>), 3.31 (s, 3 H, OMe), 4.32 (q, 1 H,  $J_{4,5}$  7, H-4), 4.49 (d, 1 H,  $J_{2,1}$  6, H-2), 4.61 (d, 1 H, J<sub>3,2</sub> 6, H-3), 4.92 (s, 1 H, H-1); <sup>13</sup>C NMR(CDCl<sub>3</sub>, 50 MHz): 110 (C-1), 108 (C, CMe<sub>2</sub>), 84 (C-2), 83 (C-3), 81 (C-4), 52 (CH<sub>3</sub> OCH<sub>3</sub>), 25 and 23  $(Me_2, CMe_2)$ , 19 (C-5); mass spectrum: m/z 173 (M<sup>+</sup> –  $CH_3$ ). The structure was also established by the absence of methylene carbon in DEPT. Similarly, 5 (5.0 g, 17 mmol) in THF (50 mL), was treated with SMEAH (70% solution in toluene; 26 mL, 85 mmol) and refluxed for 12 h. The mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> and worked up using 1N HCl to give exclusively compound 4 (2.5 g, 70%). Compound 5 in THF (50 mL) reacted with LTTBA (30% solution in THF, 50 mL, 9 h) and processed as just described also yielded compound 4 (0.54 g, 75%) exclusively.

Compound **6** (300.0 g, 0.83 mol) in Me<sub>2</sub>SO (2.2 L) was allowed to react with NaBH<sub>4</sub> (127 g, 3.4 mol; 80-85 °C; 3 h). Work up with 5% aqueous AcOH, yielded compound **8** (140 g, 89% yield) upon distillation (bp 65–70 °C, 0.2 mmHg). Similarly **6** (5.0 g, 13 mmol) in THF (50 mL) on reaction with SMEAH (70% solu-

tion in toluene, 20 mL, 69 mmol, 12 h, reflux) followed by column chromatography (10% EtOAc-petroleum ether) gave **8** (1.3 g, 50%) and **4** (1.3 g, 45%). However compound **6** (5.0 g, 13 mmol) in THF (100 mL) when treated with LTTBA (30% solution in THF, 55 mL, 65 mmol, reflux, 16 h) and work up using procedures mentioned earlier provided **8** (0.5 g, 20%) and **4** (2.0 g, 70%).

Compound 7 (1.0 g, 2.9 mmol) in Me<sub>2</sub>SO (10 mL) was treated with NaBH<sub>4</sub> (0.6 g, 14 mmol, rt, 2 h). Work up with 5% aq. AcOH yielded **8** (0.1g, 20% yield) and **4** (0.42 g, 70%). Similarly 7 (10.0 g, 30 mmol) in THF (100 mL) on reaction with SMEAH (70% solution in toluene, 50 mL, 148 mmol, -40 °C; 1 h) provided **8** (2.8 g, 50%) and **4** (1.8 g, 30%). Furthermore 7 (10.0 g, 30 mmol) in THF (100 mL), when treated with LTTBA (30% solution in THF, 125 mL, 147 mmol; 0 °C 1 h) using similar work up procedures mentioned earlier for LTTBA reactions, yielded **8** (5.35 g, 95%) exclusively.

# 4.7. Preparation of 1,2,3-tri-*O*-acetyl-5-deoxy-D-ribofuranose (9)

Compound 8 (50 g, 0.26 mol) in 0.04N H<sub>2</sub>SO<sub>4</sub> (500 mL) was heated to 80–90 °C for 3 h. The mixture was cooled to rt, neutralized with solid Na<sub>2</sub>CO<sub>3</sub>, and evaporated to dryness. The residue was dissolved in pyridine (800 mL), treated with Ac<sub>2</sub>O (210 mL, 2.2 mol), and stirred at rt for 16 h. Saturated NaHCO<sub>3</sub> (5 L) was poured into the reaction mixture, and extracted with CH<sub>2</sub>Cl<sub>2</sub> (1 × 3 L, 2 × 2 L). The combined organic phase was washed with water (2 × 1 L), dried (Na<sub>2</sub>SO<sub>4</sub>), and concentrated in vacuo to 600 mL. To this solution silica gel (65 g) and activated charcoal (6.5 g), were added and the solution was stirred for 1 h at rt. The solution was then filtered, and the filtrate evaporated in vacuo to provide a mixture of the two anomers of the title

compound **9** (56 g, 81%) as a syrupy mass. The anomeric mixture (1.0 g) partially crystallized from hexane to yield the crude  $\beta$  anomer, which was further crystallized from Et<sub>2</sub>O-hexane to yield the pure  $\beta$  anomer (0.6 g) as white solid, mp: 63–64 °C [lit<sup>4</sup> 64–65 °C];  $[\alpha]_{D}^{23} - 27.0^{\circ}$  (*c*, 2 in CHCl<sub>3</sub>) [lit<sup>4</sup>  $[\alpha]_{D}^{23} - 26.9^{\circ}$  (*c*, 2 in CHCl<sub>3</sub>)].

### References

- Cook, A. F.; Holman, M. J. Nucleosides Nucleotides 1984, 3, 401–411.
- Shimma, N.; Umeda, I.; Arasaki, M.; Murasaki, C.; Masubuchi, K.; Kohchi, Y.; Miwa, M.; Ura, M.; Sawada, N.; Tahara, H.; Kuruma, I.; Horii, I.; Ishitsuka, H. *Bioorg. Med. Chem.* 2000, *8*, 1697–1706.
- 3. (a) Folkers, K.A.; Shunk, C.H. U.S. Patent 2, 847, 413, 1958;
- (b) Folkers, K. A.; Shunk, C. H. Chem. Abstr. 1959, 53, 3252g.
- Kissman, H. M.; Baker, B. R. J. Am. Chem. Soc. 1957, 79, 5534–5540.
- 5. Lavene, P. A.; Stiller, E. T. J. Biol. Chem. 1934, 106, 421.
- Gosh, A. K.; Liu, W. J. Org. Chem. 1996, 61, 6175–6182.
   Shunk, C. H.; Lavigne, J. B.; Folkers, K. J. Am. Chem.
- Soc. 1955, 77, 2210–2212.
- Wartchow, C. A.; Wang, P.; Bednarski, M. D.; Callstrom, M. R. J. Org. Chem. 1995, 60, 2216–2226.
- Sarabio-Garcia, F.; Lopez-Herrera, F. J. *Tetrahedron* 1996, 53, 4757–4768.
- Binkley, R. W.; Ambrose, M. G.; Hehemann, D. G. J. Org. Chem. 1980, 45, 4387–4391.
- 11. Lerner, L. M. J. Org. Chem. 1978, 43, 161-162.
- 12. Hanessian, S.; Lablanc, Y.; Lavallee, P. *Tetrahedron Lett.* **1982**, *23*, 4411–4414.
- 13. Sano, H.; Takeda, T.; Migita, T. Synthesis 1988, 402–403.
- 14. (a) D'Sousa, R.; Kiss, J. Patent EP 0 021 231, 1981;
  (b) D'Sousa, R.; Kiss, J. Chem. Abstr. 1981, 95, 62602t.
- Shemid, H.; Karrer, P. Helv. Chem. Acta 1949, 32, 1371– 1378.