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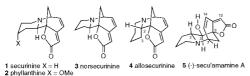
Total Synthesis of the Securinega Alkaloid (-)-Secu'amamine A

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The *Securinega* alkaloids are a family of approximately 25 tetracyclic compounds produced by several species of *Securinega* and *Phyllanthus* plants.¹ The most abundant *Securinega* alkaloid is securinine (1), isolated from the leaves of *Securinega suffruticosa*. Allosecurinine (4), which is the C2 epimer of 1, is also a common alkaloid of this group. Some additional congeneric alkaloids are (–)-norsecurinine (3) and phyllanthine (2). These alkaloids have a wide range of biological activities.^{1b} For example, securinine is a GABA receptor antagonist,^{2a} an antimalarial,^{2b} and has antibacterial activity.^{2c} Some *Securinega* alkaloids have been sporadically used in the clinic for diseases such as poliomyelitis, ALS, and chronic aplastic anemia.^{1b}

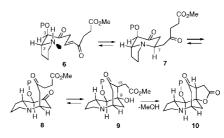


In 2003, Ohsaki and co-workers isolated (-)-secu'amamine A (5), a Securinega alkaloid having a new structural framework, from the leaves and twigs of Securinega suffruticosa var. amamiensis.³ The constitution and relative stereochemistry of this compound were determined primarily by NMR spectral analysis to be as depicted in 5. The absolute configuration of the alkaloid was established using the OMe-mandelate NMR method. Although the stereochemistry of the indolizidine moiety was not initially addressed, we have determined that the trans-fused invertomer shown in 5 is thermodynamically preferred (vide infra). Therefore, this metabolite has an intriguing tetracyclic bridged ring system containing four stereogenic centers, which differs in structure from all the other Securinega alkaloids. Recently, Magnus and Padilla have proposed that secu'amamine A is derived biogenetically from allosecurinine (4) via oxidation at C3 followed by a rearrangement.⁴ In this communication, we describe the first total synthesis of (-)secu'amamine A (5).

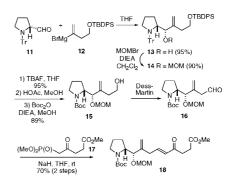
Our initial concept was to effect a multistep cascade cyclization sequence of the pyrrolidino enedione substrate **6** to afford the tetracyclic lactone **10** that possesses the complete secu'amamine A skeleton (Scheme 1). The first step of this transformation is conjugate addition of the amino group of **6** to the α,β -unsaturated ketone moiety to afford intermediate indolizidine **7**. Compound **7** would then undergo a ring flip and nitrogen inversion to conformer **8**. Subsequent ketone enolization, followed by intramolecular aldol condensation, would lead to keto hydroxy ester **9**. Finally, this compound would cyclize to tetracyclic γ -lactone **10**. We speculated that all of the steps from amino enone **6** to hydroxy ketone **9** would be reversible under basic protic conditions, and that the final lactonization to form tetracycle **10** might drive the overall process and produce the desired stereochemistry at the four requisite centers (C2, C3, C7, C9).

The synthesis commenced with *N*-tritylpyrrolidine aldehyde 11, which was prepared enantiomerically pure from D-proline.⁵ On the

Scheme 1

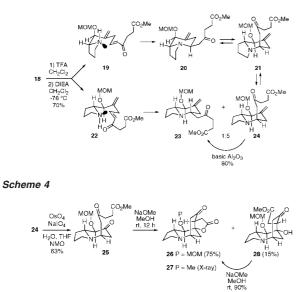


Scheme 2



basis of the work of Chemla et al.,⁵ aldehyde **11** was combined with known vinyl Grignard reagent **12**⁶ via a Felkin–Ahn addition to produce amino alcohol **13** as a single stereoisomer having the desired C2,3 configuration (Scheme 2). This alcohol was then protected as the MOM ether **14**. The silyl group of **14** was removed with fluoride, and the labile *N*-trityl group was replaced with a more stable Boc substituent to generate carbamate alcohol **15**. Dess–Martin oxidation of the primary alcohol functionality of **15** led to aldehyde **16**, which without purification reacted with the anion derived from known levulinate phosphonate **17**⁷ to yield (*E*)-enone **18**. Although a number of attempts were made to oxidatively cleave the *exo*methylene group of **18** to produce the *N*-Boc-protected precursor to the enedione cyclization substrate **6**, this transformation could not be effected. Therefore, we elected to explore a modified strategy using **18**.

After some experimentation, it was discovered that exposure of carbamate 18 to TFA in methylene chloride to remove the Boc group, followed by careful neutralization of the amine salt at low temperature using Hunig's base, led to a mixture of indolizidines 23 and 24, with the desired latter stereoisomer being formed as the major product with 5:1 selectivity in good combined yield (Scheme 3). The configuration and conformation of these compounds were determined to be as shown by 2D NMR NOE analysis. Although the cyclization products cannot be interconverted by exposure to various bases in solution, it was found that, when the major isomer 24 is adsorbed onto dry basic alumina and allowed to remain at room temperature overnight, it is transformed completely to the minor isomer 23. It seems reasonable that 24 arises from cyclization

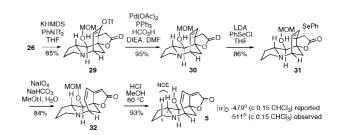


of the free amine derived from carbamate 18 via conformation 19, leading initially to the intermediate *cis*-indolizidine 20. This compound can undergo conformational isomerization to 21, followed by nitrogen lone pair inversion, to produce 24.

The isomeric cyclization product 23 could form directly via amino enone conformation 22. Our rationale for the conjugate addition to occur primarily via conformer 19 to afford 24 as the major kinetic product is based upon the known conformational preferences of acyclic allylic ethers.8 Thus, work by Gung and others has shown that the allylic hydrogen in such systems prefers to be eclipsed (in plane) with the double bond as is the case in conformation 19. An in plane C-O bond, as in the alternative conformer 22, is generally disfavored.

To continue the synthesis, the major *exo*-methylene indolizidine 24 was first oxidatively cleaved to ketone 25 (Scheme 4). We were pleased to find that exposure of this compound to sodium methoxide in methanol at room temperature afforded the desired tetracyclic lactone 26 in 75% yield along with a small amount of a polar compound which we believe is epimeric hydroxy ester 28. Support for this assignment is that treatment of 28 with NaOMe/MeOH produces lactone 26 in high yield. The structure of γ -lactone 26 was confirmed by an X-ray crystal structure analysis of the methylprotected analogue 27, prepared via the same route used for the MOM compound. Interesting features of the structure include a trans-fused indolizidine and a boat cyclohexanone ring.

Completion of the synthesis involved first selectively converting ketone 26 to the enol triflate 29,9 followed by palladium-mediated reduction to alkene **30** (Scheme 5).¹⁰ The γ -lactone moiety of **30** could be stereoselectively selenated to produce 31, which upon periodate oxidation underwent syn-elimination to give diene lactone 32.11 Finally, removal of the MOM protecting group with MeOH/ HCl yielded (-)-secu'amamine A (5) having proton and carbon NMR spectra identical to those of authentic material.^{3,12} Moreover, the observed optical rotation of synthetic 5 was in good accord with that of the natural alkaloid, thereby confirming the original assignment of absolute configuration. In addition, we have found



that, if the ¹H NMR spectrum of **5** is run in deuteriobenzene, peaks due to the protons at C2,6 are sufficiently dispersed to allow NOE analysis, which clearly demonstrated that the indolizidine is transfused. We have also been able to obtain an X-ray crystal structure of synthetic 5 which supports these conclusions.

In summary, we have devised a convergent enantioselective total synthesis of the novel Securinega alkaloid secu'amamine A (5). The synthesis requires 15 steps starting from D-proline-derived aldehyde 11 and proceeds in approximately 9% overall yield. Key steps include a stereoselective conjugate addition of amino enedione 19 to afford indolizidine 24 as the major product and cyclization of diketoester 25 to produce tetracyclic γ -lactone 26.

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Supporting Information Available: Experimental procedures for preparation of new compounds and copies of NMR spectra, as well as X-ray data for compounds 5 and 27. This material is available free of charge via the Internet at http://pubs.acs.org.

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Scheme 5

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