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Asymmetric Multicomponent Reactions: Diastereoselective Synthesis of Substituted Pyrrolidines and Prolines

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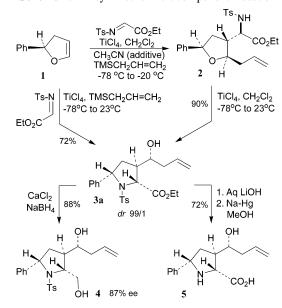
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ABSTRACT

A novel diastereoselective synthesis of substituted pyrrolidines has been developed. Asymmetric multicomponent reactions of optically active phenyldihydrofuran, *N*-tosyl imino ester, and silane reagents in a one-pot operation afforded highly substituted pyrrolidine derivatives diastereoselectively. The reaction is quite efficient and constructed up to three stereogenic centers in a single operation.

Multicomponent reactions (MCRs) that provide functionalized heterocyclic scaffolds in a single operation and in a stereodefined manner are of enormous importance in synthetic organic and medicinal chemistry. Despite the significance and potential of MCRs, there exist only a few versatile multicomponent reactions that truly generate diverse molecules with multiple stereocenters.¹ Our development of multicomponent reactions led to syntheses of a variety of substituted tetrahydrofurans and tetrahydropyrans containing multiple stereocenters.² Recently, we described multicomponent reactions with N-tosylimino ester that provided rapid access to a range of functionalized novel α-amino acids containing cyclic ether templates.³ As depicted in Scheme 1, multicomponent reactions of N-tosylimino ester with optically active phenyldihydrofuran (1) at -78 to -20 °C in the presence of allyltrimethylsilane as the nucleophile and CH₃CN as the additive provided a single diastereomer 2 in

very good yield. Interestingly, in the absence of CH₃CN



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Scheme 1. Asymmetric Multicomponent Reaction

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additive, this multicomponent reaction typically afforded tetrahydrofuran derivative 2 along with varying amounts (15-30%) of pyrrolidine derivative **3a** as the byproduct. Presumably, a Lewis acid catalyzed intramolecular rearrangement led to the formation of pyrrolidine derivative 3. The presence of CH₃CN additive completely prevented the formation of pyrrolidine derivative 3. Pyrrolidine rings are inherent to numerous bioactive natural products and medicinal agents.⁴ The biological significance of functionalized pyrrolidines and prolines continues to stimulate interest in their design and synthesis. In this context, a number of practical synthetic methodologies have been developed recently.⁵ In our continuing interest in probing enzyme-active sites with designed ligands containing heterocyclic templates,⁶ we sought to optimize the above multicomponent reaction conditions so as to synthesize functionalized pyrrolidine heterocycles in a stereopredictable manner. Herein we report asymmetric multicomponent reactions of optically active phenyldihydrofuran, N-tosylimino ester, and silane reagents in a one-pot operation to afford functionalized pyrrolidine and proline derivatives diastereoselectively.

As mentioned above, the multicomponent reaction of *N*-tosylimino ester, ⁷ 5-phenyldihydrofuran (1), ^{8,9} and allyltrimethylsilane in the absence of CH₃CN additive provided phenyltetrahydrofuran 2 along with pyrrolidine derivative 3a as the byproduct. We anticipated that the formation of pyrrolidine byproduct 3a evolved from phenyltetrahydrofuran 2 by a TiCl₄-promoted formation of benzylic carbocation followed by intramolecular ring closure with the sulfonamide. To examine this presumption, phenyltetrahydrofuran 2 was treated with 1.2 equiv of TiCl₄ in CH₂Cl₂ at -78 °C, and the resulting mixture was warmed to 23 °C for 2 h. Indeed, phenyltetrahydrofuran 2 smoothly converted to pyrrolidine derivative 3a as a single diastereomer in 90% yield. We then optimized the multicomponent reaction conditions to provide pyrrolidine derivative 3a. Thus, asymmetric multicomponent reactions leading to effective synthesis of various functionalized pyrrolidines were carried out as follows. Optically active phenyldihydrofuran (1, 1.2 equiv) and N-tosylimino

ester (1 equiv) in CH₂Cl₂ were treated with TiCl₄ (1 M solution in CH₂Cl₂, 1.2 equiv) at -78 °C for 1 h. Allyltrimethylsilane (3 equiv) was added, and the resulting mixture was allowed to warm to 23 °C and stirred for 1 h. After this period, the reaction was quenched with saturated aqueous NaHCO₃ solution. Standard workup and flash chromatography over silica provided pyrrolidine derivative **3a** in 72% yield as a single diastereomer (by ¹H and ¹³C NMR analysis). Reduction of **3a** with NaBH₄ in the presence of CaCl₂ in a mixture of EtOH and THF afforded diol **4** in 88% yield. The assignment of stereochemistry of the three new chiral centers of **3a** was made on the basis of the X-ray structure of **4** in Figure 1 as well as extensive NOESY experiments. ¹⁰

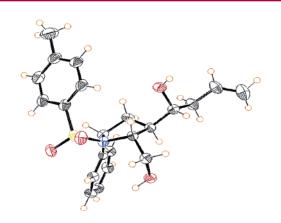


Figure 1. ORTEP drawing of X-ray structure of 4.

The optical purity of compound **4** was determined by its conversion to the corresponding Mosher ester. ¹¹ The ¹⁹F NMR analysis of the Mosher esters established that the optical purity was 87% ee. Compound **3a** was also converted to proline derivative **5** by saponification using aqueous LiOH followed by exposure of the resulting acid to Na–Hg in methanol at reflux. ¹² Proline derivative **5** was obtained in 72% yield in a two-step sequence.

We investigated the feasibility of this reaction protocol with a number of nucleophiles, and the results are summarized in Table 1. Multicomponent reactions with allyltributylstannane in the presence of 1.2 equiv of $TiCl_4$ proceeded with excellent diastereoselectivity (dr = 99/1, Table 1, entry 2) and good yield. When triethylsilane was used as a nucleophile, pyrrolidine derivatives were obtained

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⁽⁹⁾ Optical purity of 5-phenyl dihydrofuran was 88% ee.

⁽¹⁰⁾ Crystal data for 4: $C_{22}H_{27}NO_4S$; MW = 401.53; colorless crystal; crystal system, block; space group, P21; cell parameters, a=13.99719-(11) Å, b=10.4573(4) Å, c=14.4138(11) Å, $\alpha=96.230(3)^\circ$, V=2097.3-(2) Å³, Z=4; Mo K α radiation ($\lambda=0.71073$ Å, T=150 K), R1 = 0.045, WR2 = 0.076 ($I>2\sigma(I)$); R1 = 0.081, WR2 = 0.089 (all data). Crystallographic data has been deposited with the Cambridge Crystallographic Data Center (deposition no. 611303). These data can be obtained free of charge via the Internet at www.ccdc.cam.ac.uk/data_request/cif, by email to data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB2 1EZ, U.K. (11) Dale, J. A.; Dull, D. L.; Mosher, H. S. J. Org. Chem. 1969, 34, 2543.

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Table 1. Structures and Diastereoselectivities of Pyrrolidine Derivatives^a

entry	nucleophile	equiv of TiCl ₄	major Product(s)	yield (%) ^b	dr ^c
1	Me ₃ Si	1.2 F	OH H H H H H N 3a Ts	72	99/1
2	n-Bu₃Sn <u></u>	1.2	3a OH	68	99/1
3	Et ₃ SiH	1.2 F	H H H N CO ₂ Et	69	90/10
4	<i>n</i> -Bu₃SnH	1.2	3b	53	85/15
5	Ph OTBS 6a	4.2 F	OH O H H H Ph N CO ₂ Et	63	99/1
6	TBSO #Bu	4.2 I	H H OH O H H H OH O A But A	₁ 83	99/1
7	TBSO OPI	1 4.2 I	H H OH OF	'h 76	99/1
8	OMe OTMS	4.2 F	OH O H H OH ON 3f Ts CO ₂ Et	1e 61	99/1

^a All reactions were carried out as described in the text. ^b Yields refer to the isolated product. ^c Diastereomeric ratio determined by ¹H NMR and ¹³C NMR.

as a mixture of diastereomers (dr = 90/10, Table 1, entry 3). The corresponding reactions with tributyltin hydride also provided pyrrolidine derivative 3b. However, diastereoselectivity and yield were further reduced (Table 1, entry 4). We then examined a number of enolsilanes and ketene acetals as the nucleophiles. These reactions were considerably sluggish in the presence of 1.2 equiv of TiCl₄. Multicomponent reactions with enolsilane 6a in the presence of 1.2 equiv of TiCl₄ at -78 to +23 °C provided a mixture (\sim 1:1) of tetrahydrofuran and pyrrolidine derivatives. The use of 4.2 equiv of TiCl₄, however, afforded only pyrrolidine derivative 3c as a single diastereomer in 63% yield (Table 1, entry 5). The corresponding reaction with tert-butyl enol ether **6b** also gave a single diastereomer **3d** in excellent yield (Table 1, entry 6). These reactions with ketene acetals 6c and **6d** also proceeded with excellent diastereoselectivity.

Lewis acid catalyzed formation of pyrrolidines as well as a high degree of diastereoselectivity associated with these reactions can be rationalized on the basis of the proposed models in Scheme 2. As described previously,³ we presume that reaction of phenyldihydrofuran and N-tosylimino ester in the presence of TiCl₄ at -78 °C would furnish oxocar-

Scheme 2. Stereochemical Models

benium ion 7 which minimizes nonbonding interactions in the transition state. Reaction of 7 with allytrimethylsilane at -78 to -20 °C in the presence of CH₃CN would provide tetrahydrofuran 2 as described previously.³ However in the absence of CH₃CN, Lewis acid activation of tetrahydrofuran would result in a new oxonium ion 8. The S_N2 nucleophilic attack of sulfonamide nucleophile (NTs) to the LA-activated oxonium ion 8 may account for the observed complete inversion for 3a and 3c-f (Table 1, entries 1, 2, and 5-8). Alternatively, oxonium ion 8 may subsequently lead to a benzylic carbonium ion 9. Further rotation of the carboncarbon bond around the benzylic carbonium ion could provide an alternate carbonium ion 10. An intramolecular S_N1 attack by the NTs is likely to proceed through 9 over 10 to provide 3a predominantly because of the absence of developing nonbonding interactions between the phenyl ring and the bulky metal alkoxide. The proposed models account for the observed diastereoselectivity with triethylsilane and tributyltinhydride (Table 1, entries 3 and 4) where steric bulk of metal alkoxide is considerably reduced.

In summary, we have developed highly diastereoselective TiCl₄-catalyzed multicomponent coupling reactions to provide functionalized pyrrolidines with multiple stereocenters. The overall process is quite efficient and the protocol has constructed up to three contiguous asymmetric centers in a single operation. Further studies are under investigation in our laboratory.

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Supporting Information Available: Experimental procedures, spectral data for compounds **3–5**, and ¹H NMR and ¹³C NMR spectra for compounds **3–5**. This material is available free of charge via the Internet at http://pubs.acs.org.

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