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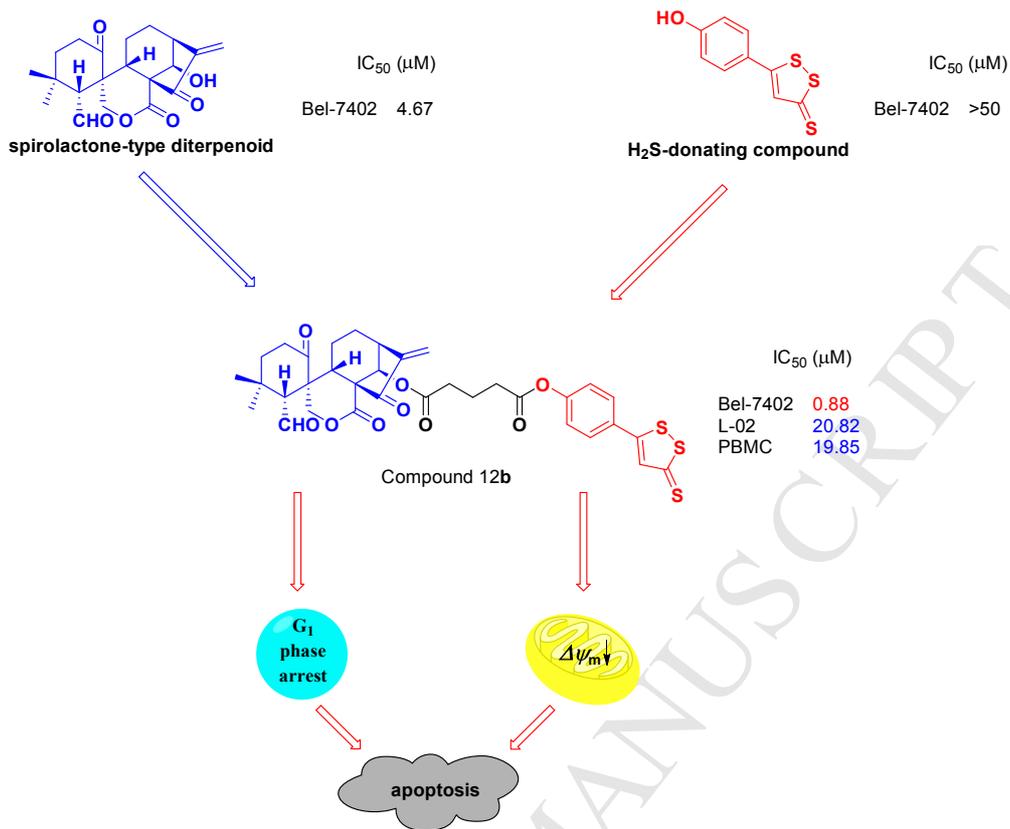
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1 Hydrogen sulfide donating *ent*-kaurane and spiro lactone-type
2 6,7-*seco-ent*-kaurane derivatives: Design, synthesis and antiproliferative
3 properties

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1

2 **Abstract:**

3 Motivated by our interest in hydrogen sulfide bio-chemistry and *ent*-kaurane diterpenoid
4 chemistry, 14 hydrogen sulfide donating derivatives (**9**, **11a-c**, **12a-c**, **13**, **14**, **16a-c** and **17a-b**) of
5 *ent*-kaurane and spirolactone-type 6,7-*seco-ent*-kaurane were designed and synthesized. Four
6 human cancer cell lines (K562, Bel-7402, SGC-7901 and A549) and two normal cell lines (L-02
7 and PBMC) were selected for antiproliferative assay. Most derivatives showed more potent
8 activities than the lead *ent*-kaurane oridonin. Among them, compound **12b** exhibited the most
9 potent antiproliferative activities, with IC₅₀ values of 1.01, 0.88, 4.36 and 5.21 μM against above
10 human cancer cell lines, respectively. Further apoptosis-related mechanism study indicated that
11 **12b** could arrest Bel-7402 cell cycle at G1 phase and induce apoptosis through mitochondria
12 related pathway. Through western blot assay, **12b** was shown to influence the intrinsic pathway by
13 increasing the expression of Bax, cleaved caspase-3, cytochrome *c* and cleaved PARP, meanwhile
14 suppressing procaspase-3, Bcl-2, Bcl-xL and PARP.

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20 **Key words:** diterpenoid, hydrogen sulfide, antiproliferative activity, apoptosis

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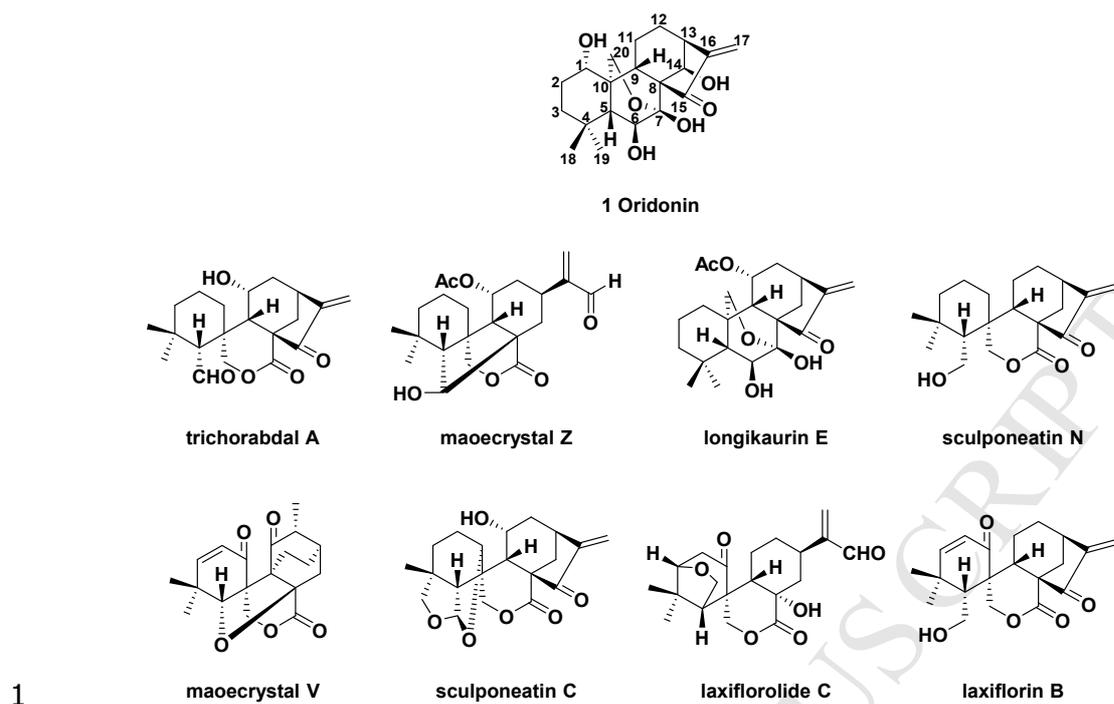
2 **1. Introduction**

3 Nature has long been an invaluable source, providing molecular skeletons for drug
4 discovery [1]. During the 1940s to the end of 2014, almost 50% of approved anticancer small
5 molecules were nature derived molecules [2]. The *Rabdosia*, an important and cosmopolitan genus
6 of the Labiatae family, has aroused considerable interests due to the abundance of *ent*-kaurane
7 diterpenoids [3-10]. Oridonin (**1**, Fig. 1), the main active component of *Rabdosia rubescens*, has
8 been the most extensively studied natural product of this genus [11-20]. Its structure and absolute
9 configuration were determined in 1970 [21]. It exhibited anticancer effects on leukemia [22],
10 pancreatic cancer [23], osteosarcoma [24], colorectal cancer [25], renal cell carcinoma [26] and
11 diffuse large B cell lymphoma [27]. One sub-type of 6,7-*seco-ent*-kauranes, spiro lactone-type
12 (7,20-lactone) diterpenoids, contains multiple stereogenic centers and complex ring systems, such
13 as trichorabdal A, maoecrystal Z, longikaurin E, sculponeatin N, maoecrystal V, sculponeatin C,
14 laxiflorolide C and laxiflorin B (Fig. 1) [28-34]. In particular, some of them exhibited promising
15 antiproliferative effects [35-37] with IC₅₀ values of 0.21, 0.29, 0.60 and 0.80 against K562,
16 HepG2, SW-480 and HL-60 human cancer cell lines, respectively.

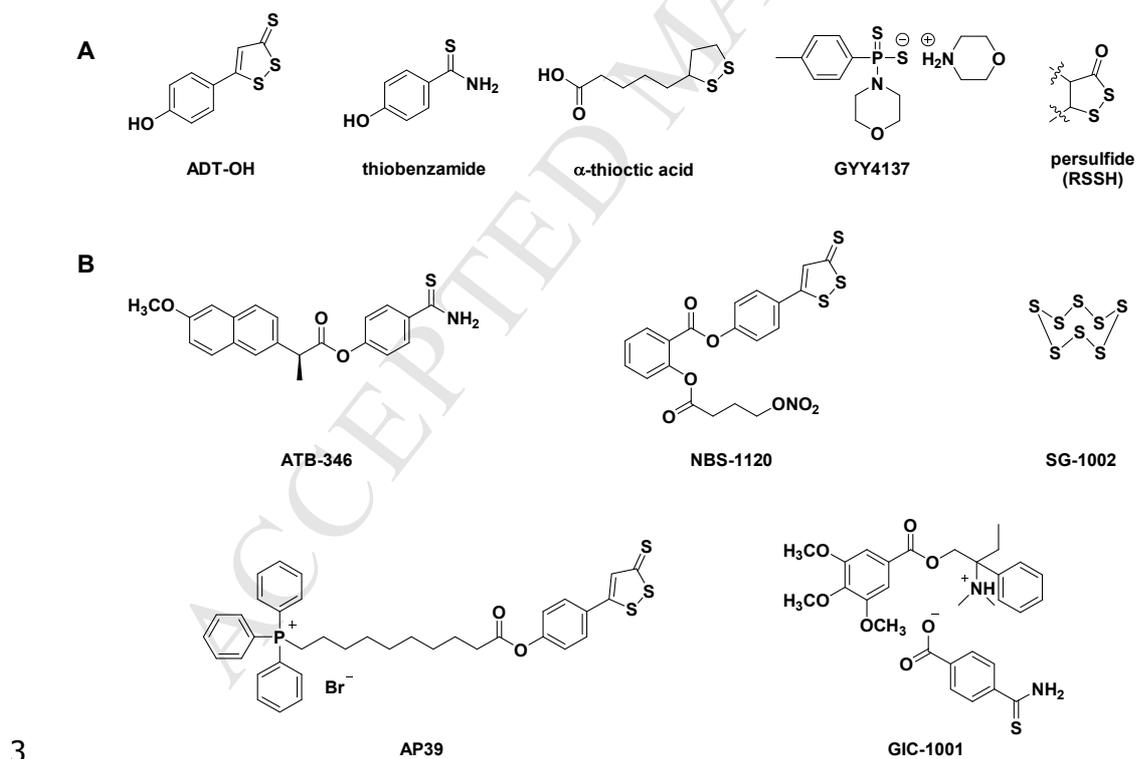
17 Although spiro lactone-type diterpenoids demonstrated sub-micromolar level
18 antiproliferative activities, their low isolation yields from *Isodon* plants can not meet the need for
19 further studies, which hindered their wide application as potential candidates in drug development.
20 Moreover, the complex stereogenic centers make it quite challenging to achieve via total synthesis
21 [38,39]. In general, through the analysis of active products [40-42], structural modification [43,44],
22 biological evaluation [45,46] and mechanism study [47-49], the deigned derivatives would show
23 better properties than the original forms and some of them would serve as drug candidates. In our
24 previous work, a convenient method for the conversion from oridonin to spiro lactone-type
25 skeleton in quantitative yield was achieved by oxidation with lead tetraacetate, which could
26 provide enough starting materials for the medicinal study of spiro lactone-type derivatives [50-53].

1 Small molecule donors have been widely used in drug discovery [54,55]. Hydrogen
2 sulfide (H₂S), like nitric oxide (NO) and carbon monoxide (CO), has been recognized as a
3 gasotransmitter related to health and disease [56,57]. The generation of endogenous H₂S is caused
4 by three important enzymes, cystathionine-γ-lyase (CSE), cystathionine-β-synthase (CBS) and
5 3-mercaptopyruvate sulfurtransferase (MST), with cysteine, homocysteine and cystathionine as
6 substrate [58,59]. Besides bioactivities in vascular tone, blood pressure, ischemia-reperfusion
7 injury, atherosclerosis, angiogenesis and neuroinflammation [60-65], recent studies elucidated that
8 H₂S showed moderate anticancer activity by inducing cell apoptosis via EGFR/ERK/MMP-2,
9 PI3K/Akt/mTOR and p38 MAPK/ERK1/2-COX-2 pathways [66-70]. Although H₂S plays
10 important roles in physiology and pathology, it can not be directly used in clinical treatment due to
11 uncontrollable dose and high toxicity. While the H₂S donors (including ADT-OH, thiobenzamide,
12 α-thioctic acid, GYY4137, persulfides etc. Fig. 2A) could partially solve the problems which
13 release H₂S at relative slow rates [71-74]. H₂S-releasing compounds also showed promising
14 properties and were instrumental for the exploration of drug candidates [75-79, Fig.2B].

15 Based on the above, we used two *ent*-kaurane derivatives (**2**, **6**) linked with two kinds of
16 H₂S donors (α-thioctic acid and ADT-OH) by ether bond and then converted them into
17 corresponding spirolactone-type derivatives. The antiproliferative effects of target compounds
18 were evaluated against chronic myelogenous leukemia K562, liver cancer Bel-7402, gastric cancer
19 SGC-7901, lung cancer A549, human normal liver L-02 and normal peripheral blood mononuclear
20 PBMC cell lines. Furthermore, apoptosis related mechanism of the most potent compound **12b**
21 was also explored, including cell cycle progression, apoptotic induction, mitochondria membrane
22 potentials decline and the expression of apoptosis-related proteins.



2 **Fig. 1.** The structures of oridonin and natural spirolactone-type 6,7-*seco-ent*-kaurane diterpenoids.



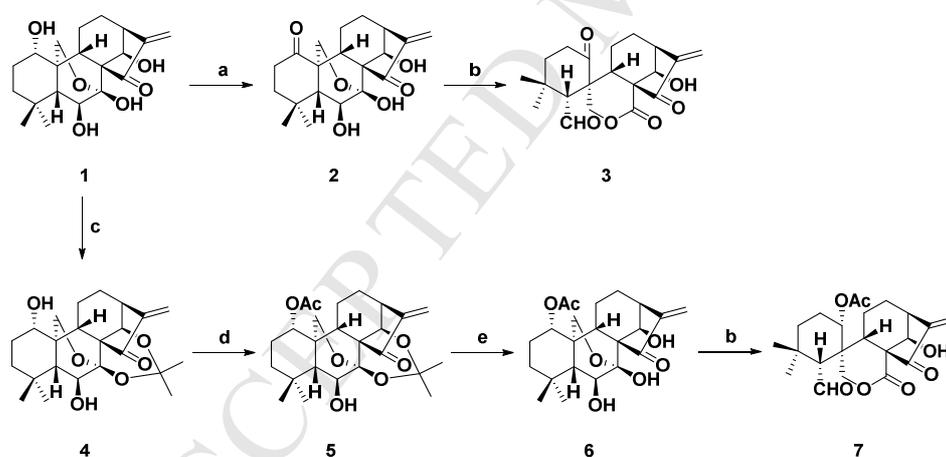
4 **Fig. 2.** The structures of selected H₂S donating molecules (A) and H₂S releasing drug candidates

5 (B).

1 **2. Result and discussion**

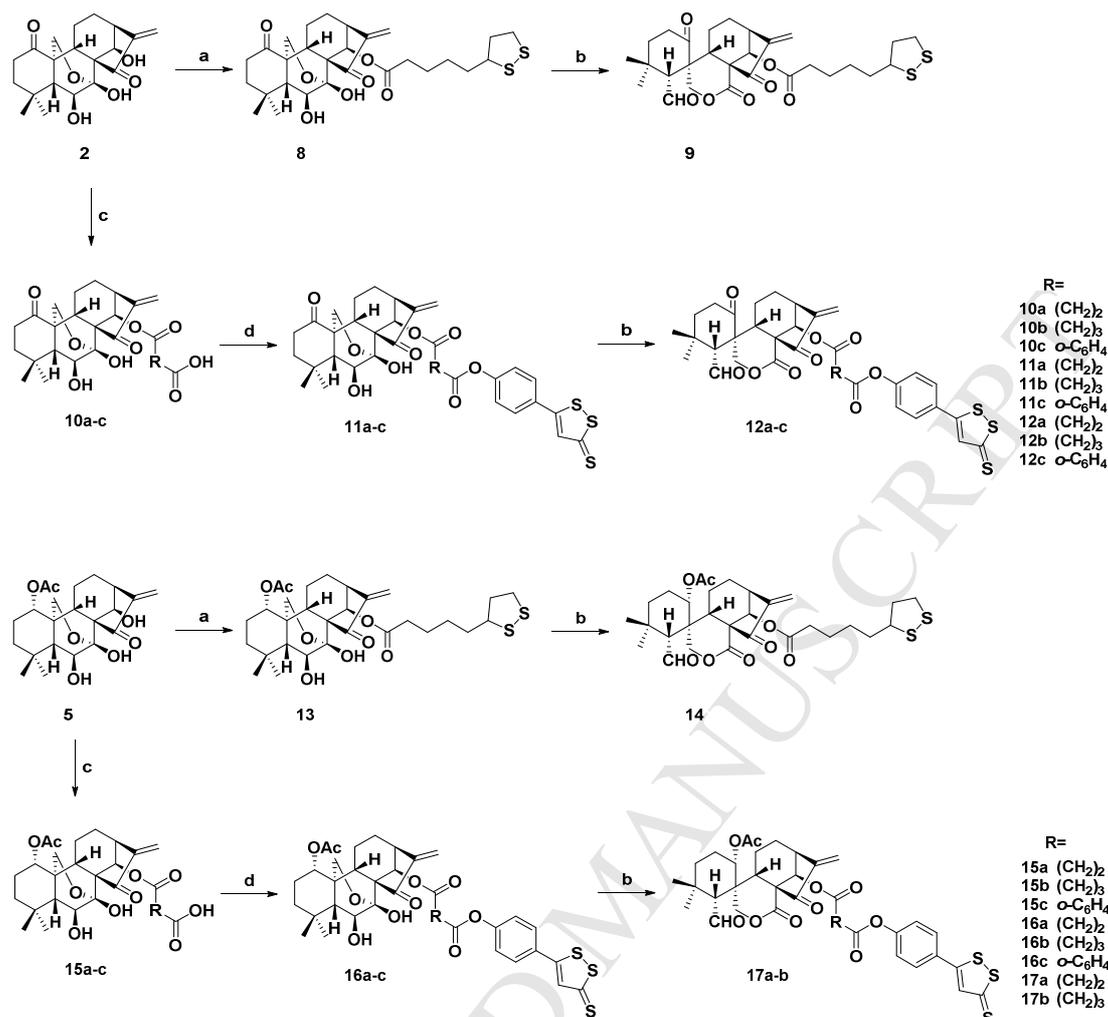
2 **2.1. Chemistry**

3 The whole synthetic route of H₂S-releasing derivatives is outlined in Scheme 1 and Scheme 2.
 4 Intermediate **2** was obtained through addition of Jones reagent to selectively oxidize the 1-OH of **1**.
 5 Treatment of **1** with 2,2-dimethoxypropane and TsOH in anhydrous acetone afforded derivative **4**,
 6 then reaction with Ac₂O/Et₃N in DCM acquired derivative **5**. And intermediate **6** was afforded via
 7 deprotection in the presence of 10% HCl. The derivatives **3** and **7** were prepared by lead
 8 tetraacetate and sodium carbonate (Na₂CO₃) in THF. The generated **2** and **5** were reacted with
 9 corresponding anhydride under the condition of 4-dimethylaminopyridine (DMAP) and
 10 triethylamine (TEA) in DCM at room temperature, which gave derivatives **10a-c** and **15a-c** in
 11 quantitative yields. Then H₂S-releasing derivatives **8**, **11a-c**, **13** and **16a-c** were obtained via
 12 reaction of corresponding diterpenoid derivatives with ADT-OH or α-thioctic acid in the presence
 13 of DMAP and carbodiimide hydrochloride (EDCI). Finally, spiro lactone-type derivatives **9**, **12a-c**,
 14 **14** and **17a-b** were achieved by oxidation using lead tetraacetate and Na₂CO₃ in THF.



15

16 **Scheme 1.** The synthetic route of intermediates. Reagents and conditions: (a) Jones reagent,
 17 acetone, 0 °C, 0.5 h; (b) Pb(OAc)₄, Na₂CO₃, THF, rt, 2 h; (c) 2,2-dimethoxypropane, acetone,
 18 TsOH, 56 °C, 1.5 h; (d) Ac₂O, TEA, DMAP, DCM, rt, 5 h; (e) 10% HCl, THF, rt, 4 h.



1

2 **Scheme 2.** Synthesis of H₂S-releasing derivatives **9**, **11a-c**, **12a-c**, **13**, **14**, **16a-c** and **17a-b**.3 Reagents and conditions: (a) α -thioctic acid, EDCI, DMAP, DCM, rt, 12 h; (b) Pb(OAc)₄, Na₂CO₃,

4 THF, rt, 2 h; (c) corresponding anhydride, TEA, DMAP, DCM, rt, 5 h; (d) ADT-OH, EDCI,

5 DMAP, DCM, rt, 8-12 h.

6 **2.2. Biological evaluation**7 **2.2.1 Antiproliferative activities and preliminary SAR**

8 The antiproliferative activities of 14 target derivatives (**9**, **11a-c**, **12a-c**, **13**, **14**, **16a-c** and

9 **17a-b**), lead compound oridonin (**1**), parental diterpenoids (**2**, **3**, **6** and **7**) and H₂S donors

10 (ADT-OH and α -thioctic acid) were tested by MTT assay (5-Fu was used as positive control).

11 Four human cancer and two normal cell lines were selected to test their cytotoxicity. As shown in

12 Table 1, all derivatives displayed inhibitory activities against selected human cancer cell lines.

1 Most derivatives exhibited more potent antiproliferative activities than oridonin ($IC_{50} = 4.79-18.86$
 2 μM), and even than positive control ($IC_{50} = 2.33-17.59 \mu M$). Besides promising antiproliferative
 3 activities, low cytotoxicity to normal cells was also observed, which is a key character in
 4 anticancer drug exploration. When the substitution R were alkyl groups (**11a-b**, **12a-b** and **16a-b**),
 5 their antiproliferative activities were more potent than corresponding ones with aromatic groups
 6 (**11c**, **12c** and **16c**) and improved IC_{50} values were observed with the extension of R group in
 7 Bel-7402 cells. It demonstrated that linear linker rather than benzene ring was advantageous for
 8 potency, and longer carbon bridge might contribute to superior inhibitory effects. Compounds **9**,
 9 **13** and **14** whose 14-OH directly incorporated with α -thioctic acid showed similarly
 10 antiproliferative activities to corresponding derivatives (**11a-b**, **12a-b**, **16a-b** and **17a-b**) which
 11 coupled with ADT-OH through a carbon bridge. Comparing 1-oxo *ent*-kaurane derivatives (**11a-c**)
 12 with corresponding spirolactone-type derivatives (**12a-c**), stronger antiproliferative activities were
 13 observed. But this phenomenon did not work in 1-acetylated derivatives (**16a-c** and **17a-b**).
 14 Furthermore, among spirolactone-type derivatives, 1-oxo derivatives (**12a-c**) showed more potent
 15 activities than 1-acetylated derivatives (**17a-b**). Among them, compound **12b** showed the
 16 strongest antiproliferative activities, with IC_{50} values of 1.01, 0.88, 4.36 and 5.21 μM against
 17 K562, Bel-7402, SGC-7901 and A549 human cancer cell lines, respectively. Furthermore,
 18 compound **12b** exhibited almost 24-fold less potent antiproliferative activities against L-02 ($IC_{50} =$
 19 $20.82 \mu M$) and PBMC ($IC_{50} = 19.85 \mu M$) normal cell lines, demonstrating selectivity between
 20 cancer and normal cell lines. Hence, compound **12b** was selected for further mechanism study.

21 **Table 1**

22 Antiproliferative activities ($IC_{50}^a \mu M$) of 14 target derivatives, parental diterpenoids and H_2S
 23 donors against four human cancer and two normal cell lines.

Compound	K562	Bel-7402	SGC-7901	A549	L-02	PBMC
1	4.79±0.12	8.31±0.29	7.87±0.36	17.80±0.75	18.68±1.28	>50
2	4.02±0.18	3.88±0.18	16.54±0.30	9.86±0.45	23.41±1.39	>50
3	3.26±0.16	4.67±0.25	19.48±0.94	5.31±0.40	29.70±1.07	>50

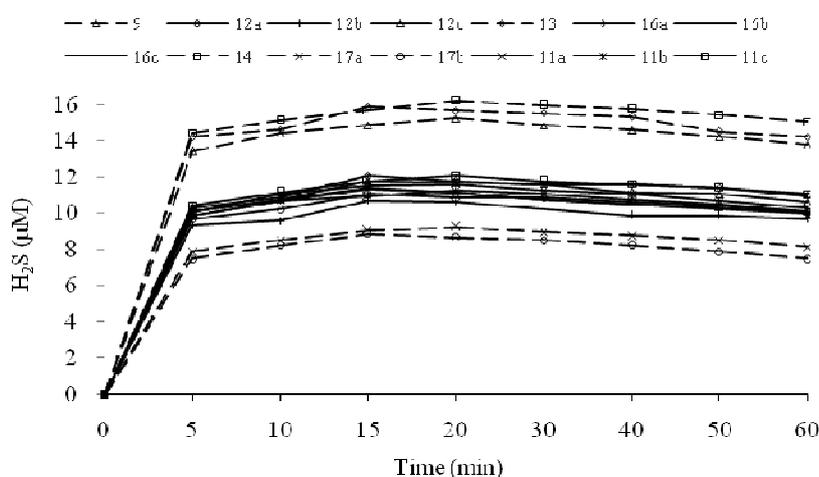
6	4.92±0.24	3.59±0.18	6.45±0.37	13.47±0.53	22.36±1.25	>50
7	8.96±0.49	>50	49.84±2.51	27.44±1.52	>50	>50
9	2.59±0.22	2.56±0.19	6.43±0.37	7.92±0.77	19.85±1.37	>50
11a	3.66±0.30	3.52±0.31	12.50±0.44	9.78±0.75	>50	29.62±1.31
11b	3.84±0.28	3.04±0.23	6.59±0.26	4.38±0.17	21.02±1.36	17.23±1.05
11c	4.80±0.10	5.34±0.27	14.87±0.53	7.69±0.40	42.64±2.63	>50
12a	1.82±0.07	1.38±0.12	4.24±0.25	5.54±0.42	18.19±1.44	16.58±1.26
12b	1.01±0.07	0.88±0.05	4.36±0.20	5.21±0.25	20.82±1.28	19.85±1.37
12c	4.25±0.23	3.26±0.14	5.93±0.33	7.45±0.38	26.21±1.55	31.45±1.61
13	2.84±0.17	2.25±0.16	6.02±0.20	8.65±0.36	41.10±2.94	28.83±1.45
14	6.75±0.31	4.58±0.24	10.04±0.39	19.89±0.81	36.47±1.32	>50
16a	2.54±0.18	2.87±0.08	5.49±0.14	6.34±0.41	23.24±1.54	44.15±2.26
16b	2.02±0.06	1.94±0.12	5.36±0.28	9.76±0.48	14.70±1.14	>50
16c	11.43±0.74	7.68±0.35	12.37±0.85	16.66±0.70	>50	39.53±1.69
17a	15.26±1.08	9.73±0.52	17.95±0.76	27.53±1.21	>50	25.86±1.35
17b	3.27±0.21	4.59±0.37	7.66±0.37	14.82±1.14	>50	30.65±1.41
ADT-OH	>50	>50	>50	>50	>50	>50
α -thioctic acid	>50	>50	>50	>50	>50	>50
5-Fu	4.02±0.22	17.59±0.86	6.38±0.32	2.33±0.12	>50	>50

1 ^aIC₅₀: Half inhibitory concentrations measured by the MTT assay, the cells were incubated
2 for 72 h. The values are expressed as averages \pm standard deviations of three independent
3 experiments.

4 2.2.2. H₂S release ability

5 H₂S-releasing capability of all derivatives was tested by the methylene blue (MB⁺) method.
6 As shown in Fig. 3, all derivatives could produce H₂S with a peak time around 15-20 min. It was
7 obvious that those derivatives with α -thioctic acid (**9**, **13** and **14**) showed the highest H₂S
8 generating ability, while releasing characteristics of the derivatives with ADT moisties were not
9 varied significantly from each other, except for **17a** and **17b**. Interestingly, 1-oxo derivatives
10 usually produced more H₂S (0.2-1 μ M) than 1-acetylated derivatives. Two spirolactone-type

1 derivatives with acetyl group at C-1 (**17a-b**) were found to release the lowest H₂S, which were
 2 consistent with their tempered antiproliferative activities. Among other derivatives with ADT-OH,
 3 spiro lactone-type derivatives **12a-c** showed moderate H₂S releasing ability while exhibiting
 4 superior antiproliferative activities. Furthermore, in contrast to the low antiproliferative effects,
 5 those with phthalic anhydride linker (**11c**, **12c** and **16c**) generated more H₂S. Moreover, with the
 6 increased length of R groups, stronger antiproliferative effects could be found but with a small
 7 reduction in H₂S generation.



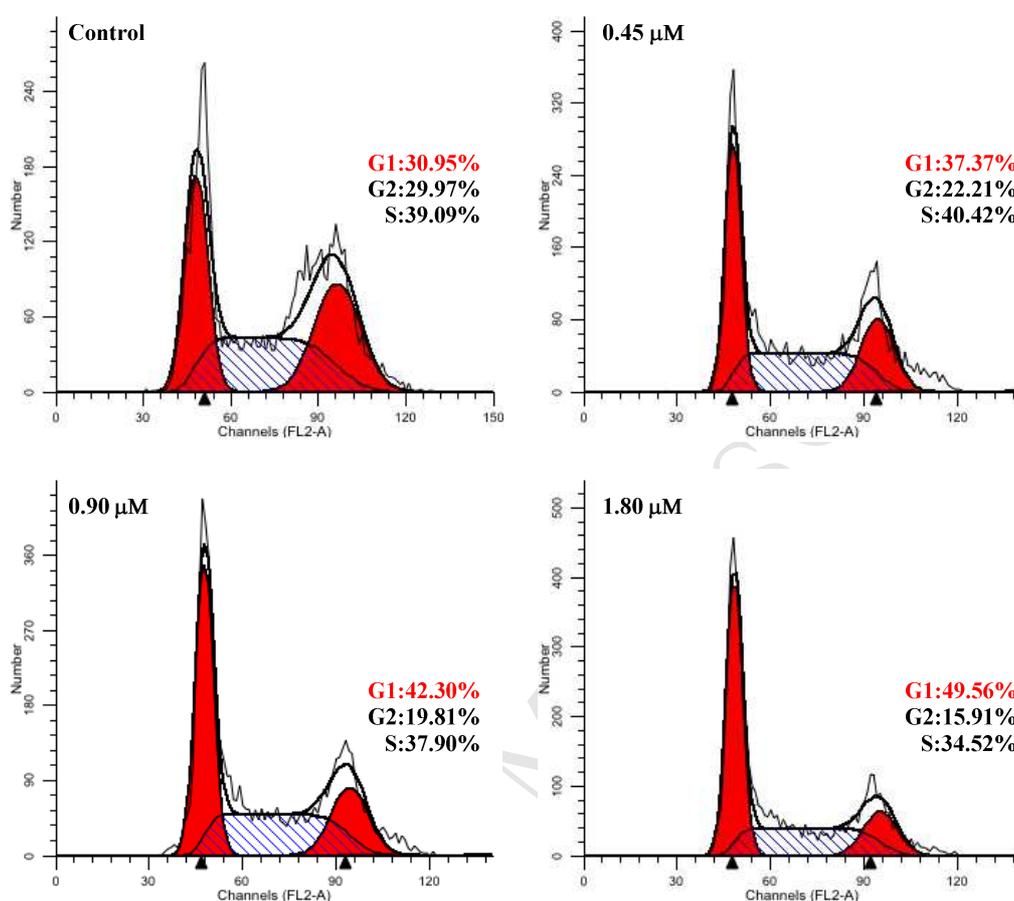
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9 **Fig. 3.** H₂S-releasing ability of target compounds. The values are expressed as averages of three
 10 independent experiments.

11 2.2.3. Cell cycle analysis

12 Cell cycle is the basic process of cell life activities. An interruption of certain phase of cell
 13 cycle leads to the inhibition of cell division and suppresses tumor growth. In order to demonstrate
 14 the antiproliferative mechanism, the influence on cell cycle of **12b** was tested and DNA content of
 15 cell nuclei was measured via flow cytometry. In this assay, Bel-7402 cells were treated with
 16 different concentrations of **12b** (0, 0.45, 0.90 and 1.80 µM) for 48 h. Then stained with propidium
 17 iodide (PI) for subsequent cytometry analysis. Cells treated with DMSO were used as control. As
 18 shown in Fig. 4, after treatment with 0, 0.45, 0.90 and 1.80 µM **12b**, the percentage of Bel-7402
 19 cells at G1 phase achieved a substantial increase from 30.95% to 37.37%, 42.30% and 49.56%,

1 respectively. According to the results, **12b** led to a dose-dependent accumulation of Bel-7402 cells
 2 at G1 phase.



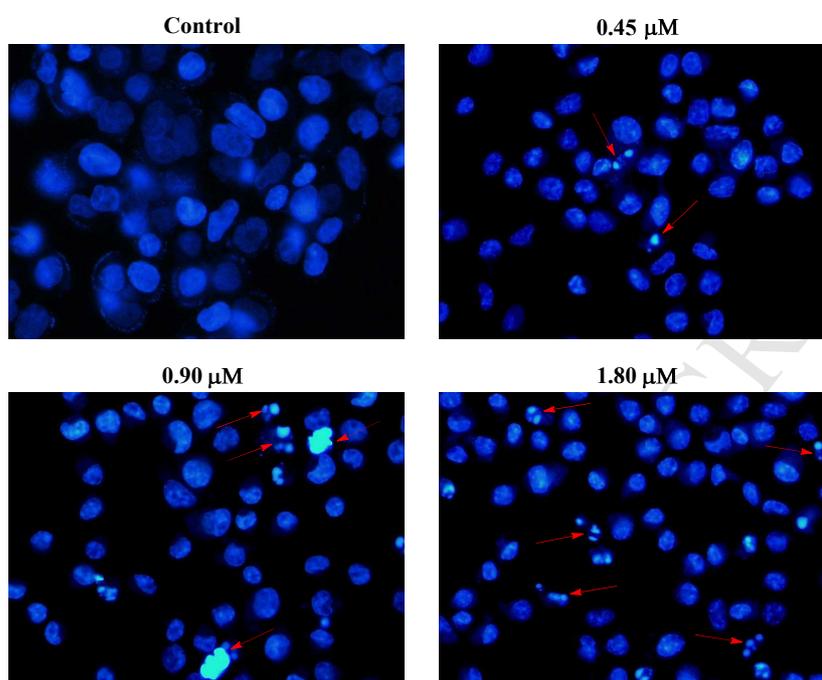
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4 **Fig. 4.** Influence on Bel-7402 cell cycle of **12b**. After incubated with **12b** (0.45, 0.90 and 1.80 μ M)
 5 or DMSO (control) for 48 h, Bel-7402 cells were stained with PI and then cell cycle distribution
 6 was analyzed by a flow cytometer.

7 2.2.4. The morphological analysis by Hoechst 33258 staining

8 The deviant morphological changes in chromatin, including cell shrinkage, chromatin
 9 condensation, the rupture of cell membrane and the nuclear fragmentation, indicate the occurrence
 10 of apoptosis. Hoechst 33258 staining assay was used to indicate apoptosis by fluorescence
 11 microscope [80]. As shown in Fig. 5, contrast to vehicle control, Bel-7402 cells presented bright
 12 blue chromatin shrink and nuclear fragmentation after treatment with **12b** (0.45, 0.90 and 1.80

1 μM). While control cells remained homogeneous nuclei with deep blue color. The Hoechst 33258
2 staining assay revealed that **12b** could induce apoptosis in Bel-7402 cells.

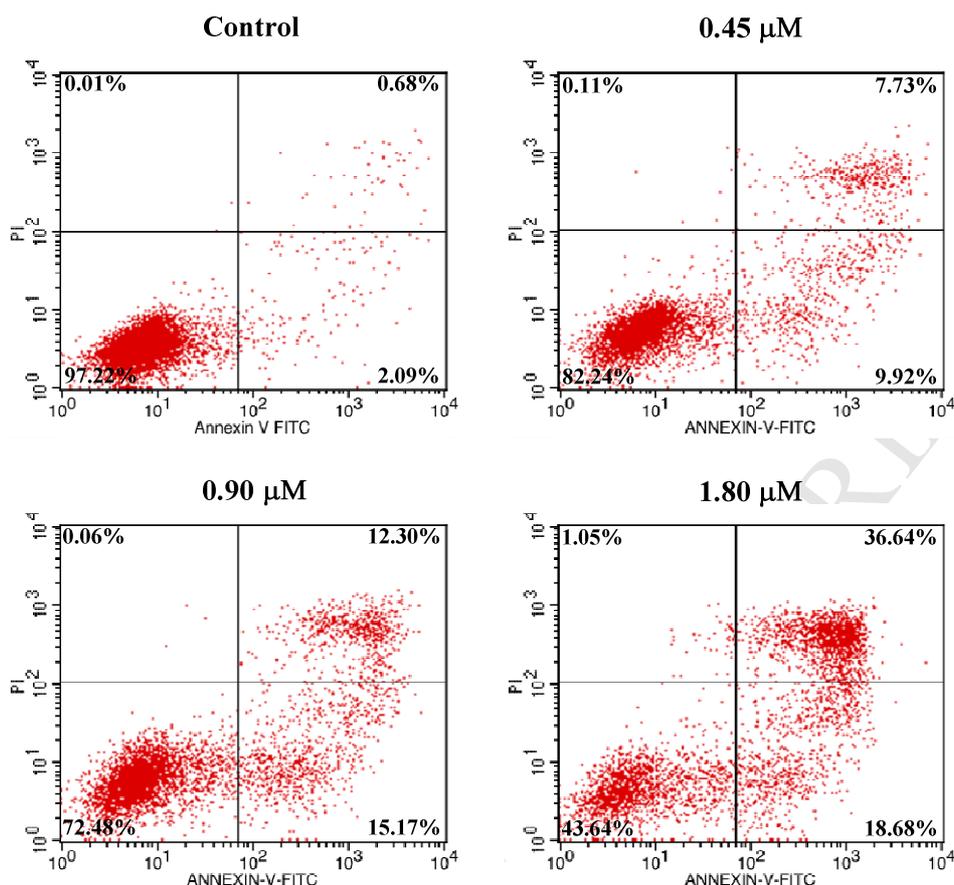


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4 **Fig. 5.** Hoechst staining of **12b** treated Bel-7402 cells. After treatment with **12b** (0.45, 0.90 and
5 1.80 μM) or vehicle for 48 h, Bel-7402 cells were stained in blue color and probed by a
6 fluorescent microscope.

7 2.2.5. Cell apoptosis assay

8 Apoptosis is an important biological process of programmed cell death, which maintain
9 regular functions and activities of cells. But cancer cells could grow incessantly by suppressing
10 apoptosis. Thus, utilizing chemical agents induced apoptosis in cancer cells is a potent therapeutic
11 method. In order to further confirm **12b** induced apoptosis in Bel-7402 cells, an annexin
12 V-FITC/PI binding assay was performed. Bel-7402 cells were treated with **12b** at indicated
13 concentrations (0, 0.45, 0.90 and 1.80 μM), then the percentage of apoptotic cells was measured
14 by a flow cytometry. As shown in Fig. 6, after incubation with **12b** for 48 h, apoptotic ratios
15 substantially rose from 2.77% of vehicle control to 17.65%, 27.47% and 55.32% for different
16 concentrations applied, respectively, in a dose-dependent manner. The results indicated that **12b**
17 possessed the ability to induce apoptosis in Bel-7402 cells.



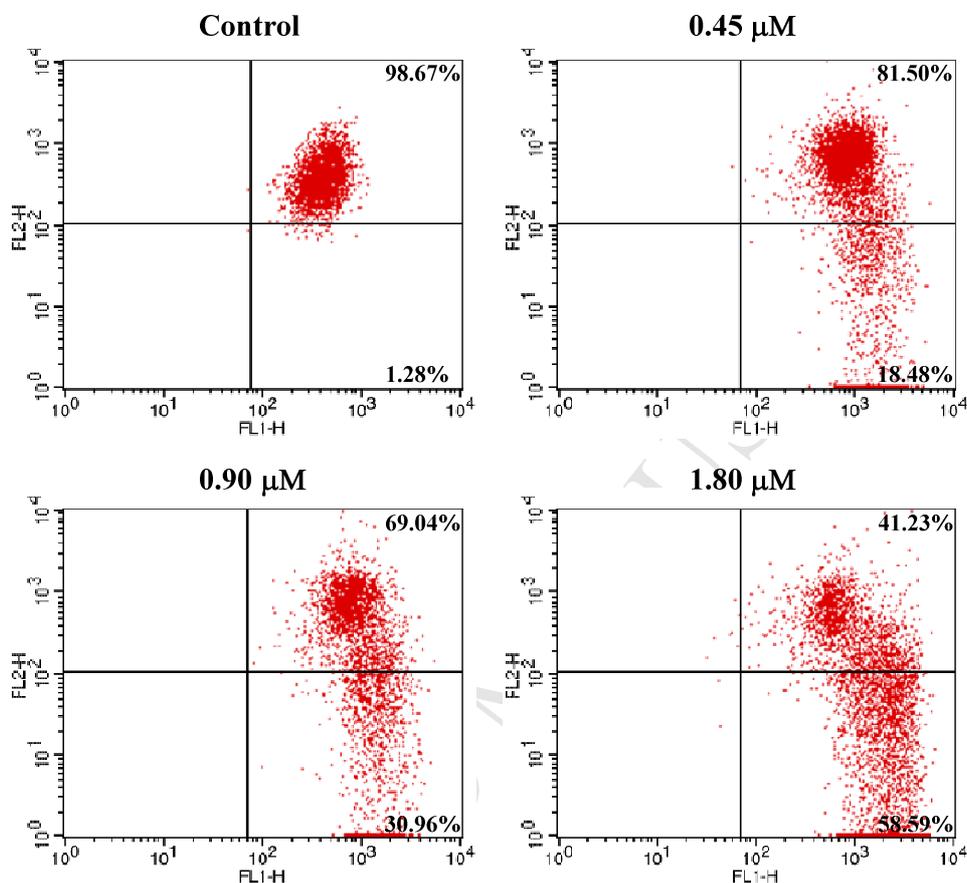
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2 **Fig. 6.** Apoptosis effects of **12b** on Bel-7402 cells. Bel-7402 cells were treated with **12b** for 48 h
 3 and analyzed via an annexin V-FITC/PI binding assay.

4 2.2.6. Mitochondria membrane potential analysis

5 The maintenance of mitochondrial membrane potentials plays an important role in
 6 mitochondrial integrity and biological function. Mitochondrial changes, especially losing
 7 mitochondrial membrane potentials, were significant events taking place during drug-induced
 8 apoptosis. Additionally, endogenous H₂S was recognized to induce apoptosis by activating the
 9 intrinsic mitochondrion-mediated pathways according to recent literatures [81]. The changes of
 10 mitochondrial membrane potentials were measured via flow cytometry, in order to demonstrate
 11 the influence of **12b** on mitochondria. Bel-7402 cells were incubated for 48 h with different
 12 concentrations of **12b** (0.45, 0.90 and 1.80 μM) and vehicle control DMSO. Then cells were
 13 treated with the lipophilic cationic fluorescent probe
 14 5,5',6,6'-tetrachloro-1,1',3,3'-tetraethylbenzimidazol-carbocyanine iodide (JC-1). As shown in Fig.

1 7, the percentage of cells with depolarized mitochondria ranged from 1.28% of control to 18.48%,
 2 30.96% and 58.59% in a dose-dependent manner. This indicated that **12b** induced Bel-7402 cells
 3 apoptosis via mitochondria related pathway.



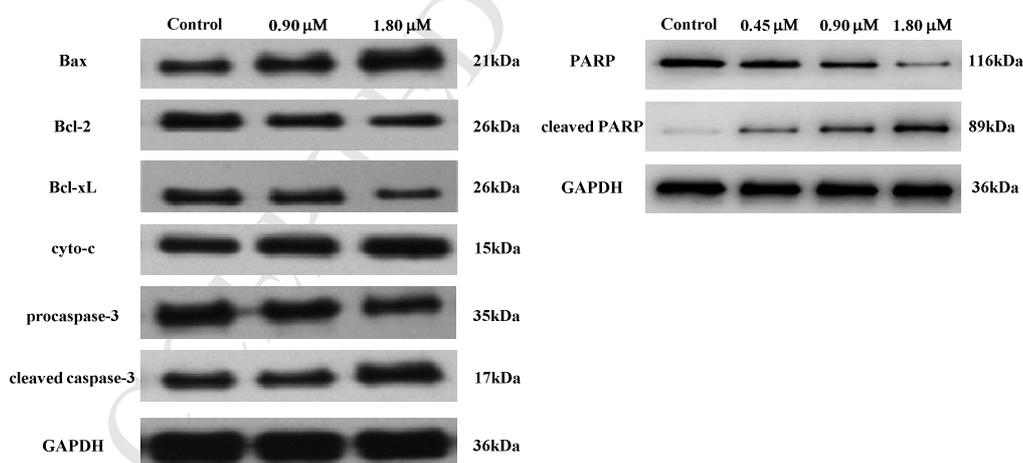
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5 **Fig. 7.** Influence on mitochondrial membrane potentials of **12b**. Bel-7402 cells were treated with
 6 **12b** for 48 h and then stained with JC-1. The changes of mitochondrial membrane potentials were
 7 analyzed by a flow cytometry.

8 2.2.7. Effects on apoptosis-related proteins

9 In general, apoptosis can be divided into two main pathways, intrinsic (the mitochondrial)
 10 pathway and extrinsic (the death receptor) pathway. Apoptosis through intrinsic pathway is
 11 usually mediated by a group of apoptosis related proteins. Bcl-2 family, an important protein
 12 family including Bcl-2, Bcl-xL and Bax, adjusts the mitochondrial membrane permeabilization and
 13 eventually leads to apoptosis. Bcl-2 protein is an antiapoptotic protein and exerts inhibitory effects

1 on proapoptotic proteins through mediating permeability transition pore. Bcl-xL protein exhibits
 2 antiapoptotic effect by blocking the destruction of mitochondrial outer membrane caused by Bax.
 3 In addition, it also plays an important role in cell necrosis. While Bax protein (proapoptotic
 4 protein) can insert itself into mitochondrial outer membrane, uplift the permeabilization and
 5 induce the mitochondria to release cytochrome *c*. Caspase activation is also a key event in the
 6 initiation and execution of apoptosis in cells. With the releasing of cytochrome *c*, downstream
 7 apoptotic protein caspase-9 is activated. Upon activation, caspase-9 cleaves procaspase-3 and
 8 generates activated caspase-3 (cleaved caspase-3) directly, and then cleaves poly(ADP-ribose)
 9 polymerase (PARP), eventually causing the execution of apoptosis. In order to find out the
 10 mediative mechanism of aforementioned apoptosis related proteins behind **12b** induced apoptosis
 11 in Bel-7402 cells, corresponding Western blot analysis was conducted. As shown in Fig. 8,
 12 treatment with **12b** caused the increased expression of proapoptotic Bax, cleaved caspase-3,
 13 cytochrome *c* and cleaved PARP and suppressed procaspase-3, PARP, Bcl-2 and Bcl-xL. The
 14 results revealed that **12b** induced Bel-7402 cells apoptosis via mitochondria mechanism mediated
 15 by apoptotic related proteins.



16

17 **Fig. 8.** Effects of **12b** on apoptosis-related proteins in Bel-7402 cells.

18 3. Conclusion

19 In this study, we designed and synthesized 14 H₂S-releasing derivatives. The antiproliferative
 20 activities of all derivatives were evaluated against four tumor and two normal cell lines. Most

1 derivatives displayed more potent antiproliferative activities than oridonin. Among them,
2 compound **12b** showed the strongest antiproliferative effects, with IC₅₀ values of 1.01, 0.88, 4.36
3 and 5.21 μ M against K562, Bel-7402, SGC-7901 and A549 human cancer cell lines, respectively.
4 Additionally, **12b** exhibited almost 24-fold less cytotoxicity against L-02 and PBMC cell lines,
5 indicating good selectivity between cancer and normal cells. Hence, compound **12b** was selected
6 for further mechanism study. The results demonstrated that the strong apoptotic effects of **12b**
7 were attributed to G1 phase cell cycle arrest, collapse of mitochondrial membrane potential and
8 mediation between proapoptotic and antiapoptotic proteins through mitochondrial pathway. In
9 conclusion, **12b** merits further investigation as a potential candidate for anticancer therapeutics
10 due to high efficiency and selectivity.

11 **4. Experimental**

12 *4.1. Chemistry*

13 Chemicals and solvents were commercially available. If required, further purification was
14 abided by standard methods. Oridonin was purchased from Nanjing Zelang Biology Technology
15 Co., Ltd with HPLC purity >98%. ¹H NMR and ¹³C NMR spectra were measured on a Bruker 400
16 or 600 MHz spectrometer in the indicated solvents (TMS as internal standard): the values of the
17 chemical shifts were expressed in δ (ppm) and the coupling constants (*J*) in Hz. Mass spectra were
18 recorded on an Agilent 1100 LC-MSD-Trap-SL. High resolution mass spectra (HR-MS) were
19 analyzed on Agilent Q-TOF B.05.01 (B5125.2).

20 *4.1.1 General procedures to synthesize target derivatives*

21 The preparation of ADT-OH was according to previous literature [82]. Oridonin (1 g, 2.76
22 mmol) was added in 20 mL of acetone and cooled to 0 °C. Jones reagent (3.2 mL, 8.28 mmol) was
23 added dropwise and stirred at 0 °C for 0.5 h. The mixture was quenched by dropwise addition of
24 isopropyl alcohol (1 mL) and extracted with DCM (3×30 mL). The combined organic layer was
25 washed with brine, dried over anhydrous Na₂SO₄ and evaporated under reduced pressure to
26 produce the crude compound **2**. Oridonin (182 mg, 0.50 mmol) was added into 10 mL acetone,
27 then mixed with TsOH (43.1 mg, 0.25 mmol), DMAP (30.5 mg, 0.25 mmol) and
28 2,2-dimethoxypropane (1.5 mL). The mixture was refluxed for 1.5 h and then concentrated in

1 vacuo to get compound **4**. **4** was dissolved in 5 mL DCM and mixed with 0.5 mL TEA, DMAP
2 (catalytic amount) and 0.75 mL Ac₂O. After stirring at room temperature for 6h, extracted with
3 DCM (3×30 mL), washed with brine, dried over anhydrous Na₂SO₄, and evaporated to give
4 compound **5**. **5** was added to 10 mL of 10% HCl/THF (1:1) and the solution was stirred at room
5 temperature for 4 h. Through post-processed as described above, compound **6** was got.

6 **2** (108.6 mg, 0.3 mmol) was added in 5 mL anhydrous DCM, then reacted with α-thioctic
7 acid (92.8 mg, 0.45 mmol) in the presence of EDCI (172.5 mg, 0.9mmol) and DMAP (7.4 mg,
8 0.06 mmol). The mixture was stirred at room temperature for 6 h and then extracted with DCM
9 (3×30 mL), washed with brine, dried over anhydrous Na₂SO₄ and evaporated in vacuo. Compound
10 **8** was obtained by purification with flash column chromatography (MeOH/DCM 1:200, v/v).
11 Following the procedures described for **8**, compound **13** was prepared from **5**.

12 Different *ent*-kaurane diterpenoid derivatives (1.5 mmol) were dissolved in 5 mL anhydrous
13 DCM and reacted with the corresponding anhydride (3.0 mmol), under the condition of TEA (1.04
14 mL, 7.5 mmol) and DMAP (91.6 mg, 0.75 mmol) for 5 h at room temperature. Then extracted
15 with DCM (3×30 mL), washed with brine, dried over anhydrous Na₂SO₄ and evaporated in vacuo
16 to get intermediates **10a-c** and **15a-c** without further purification. Intermediates **10a-c** and **15a-c**
17 (0.6 mmol) were dissolved in 5 mL anhydrous DCM, reacted with ADT-OH (135.6 mg, 0.6 mmol)
18 in the presence of EDCI (345.1 mg, 1.8 mmol) and DMAP (36.6 mg, 0.3 mmol). The mixture was
19 stirred at room temperature for 6 h and then extracted with DCM (3×30 mL), washed with brine,
20 dried over anhydrous Na₂SO₄, and evaporated in vacuo. Compounds **11a-c** and **16a-c** were
21 acquired by purification with flash column chromatography (MeOH/DCM 1:300, v/v).

22 H₂S-releasing *ent*-kaurane diterpenoid derivatives **8**, **11a-c**, **13** and **16a-b** (0.2 mmol) were
23 dissolved in 5 mL anhydrous THF, reacted with Pb(OAc)₄ (0.7 mmol) and Na₂CO₃ (1.6 mmol) at
24 room temperature for 3 h. The mixture was extracted with DCM (3×30 mL), washed with brine,
25 dried over anhydrous Na₂SO₄, and evaporated in vacuo to give corresponding crude products.
26 Spirolactone-type 6,7-*seco-ent*-kaurane derivatives **9**, **12a-c**, **14** and **16a-b** were attained by
27 purification with flash column chromatography (MeOH/DCM 1:400, v/v).

28 *4.1.1.1. Compound 9*. Pale yellow solid, yield: 42.1%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm): 9.84

1 (1H, d, $J = 3.4$ Hz, -CHO), 6.23 (1H, s, 17-CH₂), 5.95 (1H, s, 14-CH), 5.58 (1H, s, 17-CH₂), 4.82,
2 4.44 (each 1H, d, $J_A = J_B = 12.5$ Hz, 20-CH₂), 3.14 (1H, s, 13-CH), 2.56 (2H, m, 8'-CH₂), 2.44
3 (2H, m, 2-CH₂), 2.26 (2H, m, 2'-CH₂), 1.82 (1H, m, -CH₂), 1.72-1.52 (7H, m, -CH₂), 1.48-1.36
4 (4H, m, -CH₂), 1.22 (3H, s, 18-CH₃), 1.20 (3H, s, 19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm):
5 209.10, 201.91, 197.82, 172.87, 166.49, 146.64, 120.85, 73.71, 67.80, 61.50, 59.57, 52.43, 44.79,
6 41.63, 37.16, 36.02, 34.11, 33.54, 30.48, 29.68, 29.47, 29.30, 28.60, 27.73, 26.37, 25.81, 23.99,
7 18.83; C₂₈H₃₈O₇S₂ MS (ESI) m/z calcd for [M+K]⁺ 587.2, found 587.3.

8 4.1.1.2. *Compound 11a*. Orange red solid, yield: 26.2%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm):
9 7.67, 7.24 (each 2H, d, $J_A = J_B = 8.7$ Hz, Ar-H), 7.41 (1H, s, 8''-CH), 6.24 (1H, s, 17-CH₂), 6.00
10 (1H, s, 17-CH₂), 5.61 (1H, s, 7-OH), 5.42 (1H, d, $J = 11.8$ Hz, 6-OH), 4.31, 4.02 (each 1H, d, $J =$
11 10.7 Hz, 20-CH₂), 3.76 (1H, dd, $J = 11.7, 8.9$ Hz, 6-CH), 3.10 (1H, d, $J = 9.2$ Hz, 13-CH), 2.88
12 (2H, m, 2'-CH₂), 2.69 (2H, m, 3'-CH₂), 2.59-2.42 (2H, m, -CH₂), 2.38-2.29 (1H, m, -CH₂), 2.21
13 (1H, dd, $J = 13.7, 4.7$ Hz, -CH), 1.94 (2H, m, -CH₂), 1.77-1.58 (2H, m, -CH₂), 1.31 (1H, m, -CH₂),
14 1.19 (3H, s, 18-CH₃), 0.99 (3H, s, 19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm): 215.46,
15 211.51, 204.87, 170.71, 170.37, 153.49, 149.44, 136.01, 129.21, 128.18 (×2), 122.88 (×2), 122.20,
16 96.85, 74.25, 73.46, 64.93, 61.38, 59.61, 50.69, 48.51, 41.50, 38.31, 35.71, 33.37, 32.84, 30.50,
17 30.06, 29.37, 29.07, 23.16, 19.19; HRMS (ESI) m/z calcd for C₃₃H₃₄O₉S₃ [M+Na]⁺ 693.1257,
18 found 693.1289.

19 4.1.1.3. *Compound 11b*. Orange red solid, yield: 29.5%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm):
20 7.68, 7.22 (each 2H, d, $J_A = J_B = 8.68$ Hz, Ar-H), 7.41 (1H, s, 8''-H), 6.26 (1H, s, 17-CH₂), 5.97
21 (1H, s, 17-CH₂), 5.64 (1H, s, 7-OH), 5.41 (1H, d, $J = 11.76$ Hz, 6-OH), 4.31, 4.03 (each 1H, d, $J =$
22 10.64 Hz, 20-CH₂), 3.77 (1H, dd, $J = 11.5, 8.8$ Hz, 6-CH), 3.10 (1H, d, $J = 9.3$ Hz, 13-CH), 2.65
23 (2H, td, $J = 3.56, 7.4$ Hz, 3'-CH₂), 2.43 (2H, t, $J = 7.16$ Hz, 4'-CH₂), 2.22 (1H, dd, $J = 13.5, 4.4$ Hz,
24 -CH), 2.03 (2H, t, $J = 6.84$ Hz, 2'-CH₂), 1.94 (2H, m, -CH₂), 1.78-1.60 (2H, m, -CH₂), 1.32 (1H, m,
25 -CH₂), 1.19 (3H, s, 18-CH₃), 1.00 (3H, s, 19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm): 215.47,
26 211.54, 204.85, 171.33, 170.89, 153.49, 149.48, 136.01, 129.17, 128.20 (×2), 122.92 (×2), 122.14,
27 96.90, 74.15, 73.38, 64.95, 61.35, 59.68, 53.47, 50.68, 48.53, 41.55, 38.34, 35.73, 33.28, 33.14,

1 32.85, 30.50, 30.05, 23.18, 19.55, 19.18; HRMS (ESI) m/z calcd for $C_{34}H_{36}O_9S_3$ $[M+Na]^+$
2 707.1414, found 707.1451.

3 4.1.1.4. *Compound 11c*. Orange red solid, yield: 31.6%. 1H NMR ($CDCl_3$, 400 MHz), δ (ppm):
4 7.98 (1H, dd, $J = 7.4, 1.7$ Hz, 6'-H), 7.79 (1H, dd, $J = 7.4, 1.7$ Hz, 3'-H), 7.75 (2H, d, $J = 8.7$ Hz,
5 Ar-H), 7.58 (2H, m, 4', 5'-H), 7.45 (2H, d, $J = 8.7$ Hz, Ar-H), 7.44 (1H, s, 8''-H), 6.08 (1H, s,
6 17- CH_2), 6.08 (1H, s, 14-CH), 5.42 (1H, s, 17- CH_2), 4.70 (1H, d, $J = 10.8$ Hz, 6-OH), 4.63 (1H, s,
7 14-CH), 3.85, 3.27 (each 1H, d, $J = 9.6$ Hz, 20- CH_2), 2.62 (1H, m, - CH_2), 2.50 (1H, m, - CH_2),
8 2.28 (1H, m, - CH_2), 1.95-1.59 (5H, m, - CH_2), 1.51 (3H, s, 18- CH_3), 1.28 (3H, s, 19- CH_3); ^{13}C
9 NMR ($CDCl_3$, 100 MHz), δ (ppm): 215.51, 208.20, 202.90, 200.66, 171.77, 165.44, 153.90,
10 148.55, 136.06, 131.76, 131.72, 131.68, 131.01, 130.45, 129.74, 129.40, 128.62, 128.28 ($\times 2$),
11 123.05 ($\times 2$), 120.08, 93.20, 74.23, 67.62, 60.97, 58.46, 49.52, 41.04, 40.86, 36.22, 33.85, 33.40,
12 29.64, 22.70, 18.19; HRMS (ESI) m/z calcd for $C_{37}H_{34}O_9S_3$ $[M+H]^+$ 717.1286, found 717.1288.

13 4.1.1.5. *Compound 12a*. Orange red solid, yield: 21.8%. 1H NMR ($CDCl_3$, 600 MHz), δ (ppm):
14 9.85 (1H, d, $J = 3.4$ Hz, -CHO), 7.66, 7.23 (each 2H, d, $J_A = J_B = 8.6$ Hz, Ar-H), 7.40 (1H, s,
15 8''-CH), 6.24 (1H, s, 17- CH_2), 6.04 (1H, s, 14-CH), 5.59 (1H, s, 17- CH_2), 4.82, 4.48 (each 1H, d,
16 $J_A = J_B = 12.4$ Hz, 20- CH_2), 3.16 (1H, d, $J = 7.6$ Hz, 13-CH), 2.94 (1H, m, 2'- CH_2), 2.81 (1H, m,
17 2'- CH_2), 2.72 (2H, m, 3'- CH_2), 2.60 (2H, m, 2- CH_2), 2.48-2.35 (4H, m, - CH_2), 1.86 (2H, m, - CH_2),
18 1.73-1.62 (4H, m, - CH_2), 1.42 (2H, m, - CH_2), 1.25 (3H, s, 18- CH_3), 1.22 (3H, s, 19- CH_3); ^{13}C
19 NMR ($CDCl_3$, 150 MHz), δ (ppm): 215.53, 208.90, 201.89, 197.81, 171.81, 171.63, 170.41,
20 166.62, 153.55, 146.63, 136.03, 129.19, 128.86, 128.18 ($\times 2$), 122.95 ($\times 2$), 121.31, 74.13, 67.86,
21 61.26, 59.59, 52.52, 44.88, 41.53, 37.05, 36.25, 34.18, 30.63, 29.71, 29.53, 29.04, 28.99, 25.78,
22 18.81; HRMS (ESI) m/z calcd for $C_{33}H_{32}O_9S_3$ $[M+H]^+$ 669.1286, found 669.1299.

23 4.1.1.6. *Compound 12b*. Orange red solid, yield: 24.6%. 1H NMR ($CDCl_3$, 600 MHz), δ (ppm):
24 9.86 (1H, d, $J = 3.4$ Hz, -CHO), 7.67, 7.23 (each 2H, d, $J_A = J_B = 8.6$ Hz, Ar-H), 7.41 (1H, s,
25 8''-CH), 6.25 (1H, s, 17- CH_2), 6.03 (1H, s, 14-CH), 5.59 (1H, s, 17- CH_2), 4.86, 4.48 (each 1H, d,
26 $J_A = J_B = 12.4$ Hz, 20- CH_2), 3.16 (1H, d, $J = 7.3$ Hz, 13-CH), 2.65 (2H, m, 3'- CH_2), 2.60 (1H, m,
27 2'- CH_2), 2.45 (2H, m, 4'- CH_2), 2.38 (1H, m, 2'- CH_2), 2.01 (2H, m, - CH_2), 1.85 (1H, m, - CH_2),

1 1.69 (4H, m, -CH₂), 1.43 (1H, m, -CH₂), 1.25 (3H, s, 18-CH₃), 1.22 (3H, s, 19-CH₃); ¹³C NMR
2 (CDCl₃, 150 MHz), δ (ppm): 215.51, 209.17, 201.95, 197.81, 172.48, 171.91, 171.07, 166.64,
3 153.62, 146.60, 136.00, 129.09, 128.86, 128.20 (×2), 123.03 (×2), 121.07, 73.93, 67.85, 61.46,
4 59.65, 52.46, 44.85, 41.61, 37.15, 36.13, 34.14, 32.98, 32.73, 30.55, 29.49, 19.40, 18.83; HRMS
5 (ESI) *m/z* calcd for C₃₄H₃₄O₉S₃ [M+H]⁺ 683.1443, found 683.1436.

6 4.1.1.7. *Compound 12c*. Orange red solid, yield: 14.8%. ¹H NMR (CDCl₃, 600 MHz), δ (ppm):
7 9.86 (1H, d, *J* = 3.4 Hz, -CHO), 7.67, 7.23 (each 2H, d, *J*_A = *J*_B = 8.6 Hz, Ar-H), 7.41 (1H, s,
8 8"-CH), 6.25 (1H, s, 17-CH₂), 6.03 (1H, s, 14-CH), 5.59 (1H, s, 17-CH₂), 4.86, 4.48 (each 1H, d,
9 *J*_A = *J*_B = 12.4 Hz, 20-CH₂), 3.16 (1H, d, *J* = 7.3 Hz, 13-CH), 2.65 (2H, m, 3'-CH₂), 2.60 (1H, m,
10 2'-CH₂), 2.45 (2H, m, 4'-CH₂), 2.38 (1H, m, 2'-CH₂), 1.86 (2H, m, -CH₂), 1.80-1.60 (5H, m, -CH₂),
11 1.44 (1H, m, -CH₂), 1.25 (3H, s, 18-CH₃), 1.22 (3H, s, 19-CH₃); ¹³C NMR (CDCl₃, 150 MHz), δ
12 (ppm): 215.54, 208.89, 201.81, 197.76, 171.74, 166.42, 166.16, 153.81, 146.70, 136.09, 132.37,
13 131.80, 131.49, 130.28, 129.97, 129.74, 129.43, 129.16, 128.28 (×2), 123.02 (×2), 121.37, 74.87,
14 67.86, 61.24, 59.64, 52.69, 45.16, 41.50, 37.02, 36.36, 34.18, 30.69, 29.61, 25.73, 18.80; HRMS
15 (ESI) *m/z* calcd for C₃₇H₃₂O₉S₃ [M+Na]⁺ 739.1106, found 739.1110.

16 4.1.1.8. *Compound 13*. Pale yellow solid, yield: 19.8%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm):
17 6.15 (1H, s, 17-CH₂), 5.80 (1H, s, 14-CH), 5.51 (1H, s, 17-CH₂), 4.63 (1H, q, *J* = 10.6 Hz, 1-CH),
18 4.26, 4.18 (each 1H, d, *J* = 10.6 Hz, 20-CH₂), 3.78 (1H, brs, -OH), 3.17 (1H, d, *J* = 9.7 Hz,
19 13-CH), 2.53 (1H, dq, *J* = 6.4, 12.6 Hz, 8'-CH), 2.44 (1H, dt, *J* = 6.4, 12.6 Hz, 8'-CH), 2.30 (3H, t,
20 *J* = 7.5 Hz, 2', 7'-CH, -CH₂), 1.99 (3H, s, -CH₃), 1.93-1.23 (20H, m, -CH₂), 1.12 (6H, s, 18,
21 19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm): 206.21, 171.81, 169.87, 149.37, 120.47, 96.10,
22 76.40, 75.37, 73.77, 63.53, 61.48, 60.24, 56.21, 53.68, 41.33, 41.04, 40.21, 39.75, 38.47, 38.13,
23 34.45, 34.35, 33.58, 32.36, 30.20, 28.54, 25.17, 24.39, 21.54, 17.96; HRMS (ESI) *m/z* calcd for
24 C₃₀H₄₂O₈S₂ [M+Na]⁺ 617.2219, found 617.2204.

25 4.1.1.9. *Compound 14*. Pale yellow solid, yield: 39.1%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm):
26 9.79 (1H, d, *J* = 4.7 Hz, -CHO), 6.23 (1H, s, 17-CH₂), 5.79 (1H, s, 14-CH), 5.59 (1H, s, 17-CH₂),
27 5.10, 4.76 (each 1H, d, *J*_A = *J*_B = 12.5 Hz, 20-CH₂), 4.63 (1H, m, 1-CH), 2.98 (1H, d, *J* = 7.4 Hz,

1 13-CH), 2.31-2.24 (2H, m, -CH₂), 2.23 (2H, t, $J = 4.4$ Hz, 2'-CH₂), 2.10 (3H, s, -CH₃), 2.06 (2H,
2 m, -CH₂), 1.96-1.76 (4H, m, -CH₂), 1.71-1.31 (10H, m, -CH₂), 1.21 (3H, s, 18-CH₃), 0.99 (3H, s,
3 19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm): 202.41, 196.99, 172.60, 170.37, 166.59, 146.77,
4 121.45, 74.44, 72.23, 66.34, 62.68, 62.60, 61.62, 59.89, 55.42, 53.44, 45.63, 43.57, 42.16, 39.31,
5 35.46, 34.01, 32.86, 29.96, 28.69, 27.75, 23.93, 23.87, 21.44, 17.45; C₃₀H₄₀O₈S₂ MS (ESI) m/z
6 calcd for [M+K]⁺ 631.2, found 631.3.

7 4.1.1.10. *Compound 16a*. Orange red solid, yield: 18.8%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm):
8 7.68, 7.24 (each 2H, d, $J_A = J_B = 8.6$ Hz, Ar-H), 7.40 (1H, s, 8''-CH), 6.14 (1H, s, 17-CH₂), 6.08
9 (1H, d, $J = 10.8$ Hz, 6-OH), 5.93 (1H, s, 14-CH), 5.50 (1H, s, 17-CH₂), 4.63 (1H, dd, $J = 11.1, 5.5$
10 Hz, 1-CH), 4.30, 4.20 (each 1H, d, $J_A = J_B = 10.6$ Hz, 20-CH₂), 3.91 (1H, brs, -OH), 3.77 (1H, dd,
11 $J = 10.3, 6.9$ Hz, 6-CH), 3.14 (1H, d, $J = 10.0$ Hz, 13-CH), 2.90 (2H, m, 3'-CH₂), 2.72 (2H, m,
12 2'-CH₂), 2.52 (1H, m, -CH₂), 1.78 (1H, m, -CH₂), 1.48 (2H, m, -CH₂), 1.37-1.29 (4H, m, -CH₂),
13 1.99 (3H, s, -CH₃), 1.13 (6H, s, 18,19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm): 215.53,
14 206.10, 171.57, 170.72, 170.19, 169.84, 153.45, 149.42, 136.06, 129.30, 128.17($\times 2$), 122.84($\times 2$),
15 120.74, 95.91, 75.78, 75.36, 74.37, 63.59, 61.71, 59.89, 53.87, 41.18, 39.79, 38.16, 33.57, 32.46,
16 30.27, 29.70, 29.41, 29.07, 25.15, 21.65, 21.52, 18.13; HRMS (ESI) m/z calcd for C₃₅H₃₈O₁₀S₃
17 [M+Na]⁺ 737.1525, found 737.1534.

18 4.1.1.11. *Compound 16b*. Orange red solid, yield: 23.4%. ¹H NMR (CDCl₃, 400 MHz), δ (ppm):
19 7.68, 7.22 (each 2H, d, $J_A = J_B = 8.7$ Hz, Ar-H), 7.40 (1H, s, 8''-CH), 6.17 (1H, s, 17-CH₂), 6.08
20 (1H, d, $J = 10.6$ Hz, 6-OH), 5.89 (1H, s, 14-CH), 5.52 (1H, s, 17-CH₂), 4.63 (1H, dd, $J = 11.2, 5.6$
21 Hz, 1-CH), 4.30, 4.20 (each 1H, d, $J_A = J_B = 10.6$ Hz, 20-CH₂), 4.09 (1H, brs, -OH), 3.78 (1H, t, J
22 = 7.7 Hz, 6-CH), 3.16 (1H, d, $J = 9.9$ Hz, 13-CH), 2.65 (2H, td, $J = 7.3, 2.5$ Hz, 3'-CH₂), 2.44 (2H,
23 t, 2'-CH₂), 2.01 (3H, s, -CH₃), 1.78 (1H, m, -CH₂), 1.48 (2H, m, -CH₂), 1.38-1.28 (4H, m, -CH₂),
24 1.13 (6H, s, 18,19-CH₃); ¹³C NMR (CDCl₃, 100 MHz), δ (ppm): 215.52, 206.11, 171.64, 171.32,
25 170.72, 169.86, 153.44, 149.43, 136.04, 129.23, 128.20($\times 2$), 122.89($\times 2$), 120.64, 95.98, 75.85,
26 75.36, 74.13, 63.58, 61.62, 59.97, 53.78, 41.18, 39.76, 38.13, 32.43, 30.25, 25.16, 21.61, 21.55,
27 19.59, 18.08; HRMS (ESI) m/z calcd for C₃₆H₄₀O₁₀S₃ [M+Na]⁺ 751.1682, found 751.1649.

1 4.1.1.12. **Compound 16c**. Orange red solid, yield: 13.1%. ^1H NMR (CDCl_3 , 400 MHz), δ (ppm):
2 7.95 (1H, m, 6'-H), 7.75, 7.43 (each 2H, d, $J_A = J_B = 8.6$ Hz, Ar-H), 7.72 (1H, m, 3'-H), 7.62 (2H,
3 m, 4'-H,5'-H), 7.40 (1H, s, 8''-H), 6.07 (1H, s, 7-OH), 6.07, 5.39 (2H, s, 17- CH_2), 4.65 (1H, m,
4 1-CH), 4.32, 4.21 (each 1H, d, $J = 10.6$ Hz, 20-H), 4.00 (1H, brs, -OH), 3.78 (1H, m, 6-CH), 3.37
5 (1H, d, $J = 9.7$ Hz, 13-CH), 2.01 (3H, s, $-\text{CH}_3$), 1.78 (1H, m, $-\text{CH}_2$), 1.49 (2H, m, $-\text{CH}_2$), 1.39-1.30
6 (4H, m, $-\text{CH}_2$), 1.13 (6H, s, 18,19- CH_3); ^{13}C NMR (CDCl_3 , 100 MHz), δ (ppm): 215.52, 206.12,
7 171.51, 169.85, 166.02, 164.77, 153.65, 149.41, 136.13, 132.60, 132.45, 131.53, 129.82, 129.60,
8 129.54, 129.04, 128.22 ($\times 2$), 123.03 ($\times 2$), 120.76, 96.02, 75.36, 74.22, 63.62, 61.74, 60.04, 53.94,
9 40.69, 39.82, 38.16, 33.58, 32.44, 30.26, 29.70, 25.17, 21.63, 21.57, 18.10; HRMS (ESI) m/z calcd
10 for $\text{C}_{39}\text{H}_{38}\text{O}_{10}\text{S}_3$ $[\text{M}+\text{Na}]^+$ 785.1525, found 785.1526.

11 4.1.1.13. **Compound 17a**. Orange red solid, yield: 7.5%. ^1H NMR (CDCl_3 , 400 MHz), δ (ppm):
12 9.85 (1H, d, $J = 3.4$ Hz, -CHO), 9.79 (1H, d, $J = 4.6$ Hz, -CHO), 7.66, 7.23 (each 2H, d, $J_A = J_B =$
13 8.7 Hz, Ar-H), 7.39 (1H, s, 8''-CH), 6.23 (1H, s, 17- CH_2), 5.81 (1H, s, 14-CH), 5.59 (1H, s,
14 17- CH_2), 5.11, 4.76 (each 1H, d, $J_A = J_B = 12.4$ Hz, 20- CH_2), 4.63 (1H, m, 1-CH), 3.02 (1H, d, $J =$
15 7.7 Hz, 13-CH), 2.93 (1H, m, 2'- CH_2), 2.82 (1H, m, 2'- CH_2), 2.73 (2H, m, 3'- CH_2), 2.33-2.23 (2H,
16 m, $-\text{CH}_2$), 2.10 (3H, s, $-\text{CH}_3$), 1.90 (2H, m, $-\text{CH}_2$), 1.71-1.49 (6H, m, $-\text{CH}_2$), 1.21 (3H, s, 18- CH_3),
17 1.01 (3H, s, 19- CH_3); ^{13}C NMR (CDCl_3 , 100 MHz), δ (ppm): 215.50, 202.31, 196.82, 171.71,
18 171.22, 170.40, 166.68, 153.57, 146.77, 136.01, 129.18, 128.15 ($\times 2$), 122.87 ($\times 2$), 121.87, 74.55,
19 72.59, 66.44, 62.35, 59.80, 45.69, 43.58, 41.99, 39.44, 34.04, 32.90, 29.94, 29.21 ($\times 2$), 24.09 ($\times 2$),
20 21.41, 17.53; HRMS (ESI) m/z calcd for $\text{C}_{35}\text{H}_{36}\text{O}_{10}\text{S}_3$ $[\text{M}+\text{H}]^+$ 713.1549, found 713.1546.

21 4.1.1.14. **Compound 17b**. Orange red solid, yield: 10.1%. ^1H NMR (CDCl_3 , 400 MHz), δ (ppm):
22 9.80 (1H, d, $J = 4.7$ Hz, -CHO), 7.67, 7.22 (each 2H, d, $J_A = J_B = 8.6$ Hz, Ar-H), 7.41 (1H, s,
23 8''-CH), 6.26 (1H, s, 17- CH_2), 5.83 (1H, s, 14-CH), 5.60 (1H, s, 17- CH_2), 5.12, 4.77 (each 1H, d,
24 $J_A = J_B = 12.4$ Hz, 20- CH_2), 4.64 (1H, m, 1-CH), 3.01 (1H, d, $J = 7.7$ Hz, 13-CH), 2.65 (2H, m,
25 4'- CH_2), 2.45 (2H, m, 2'- CH_2), 2.28-2.22 (2H, m, $-\text{CH}_2$), 2.13 (1H, s, $-\text{CH}_3$), 2.10 (1H, d, $J = 4.8$
26 Hz, -CH), 2.03 (2H, m, 3'- CH_2), 1.22 (3H, s, 18- CH_3), 1.01 (3H, s, 19- CH_3); ^{13}C NMR (CDCl_3 ,
27 100 MHz), δ (ppm): 215.54, 202.34, 172.13, 171.04, 170.46, 166.73, 153.62, 146.80, 136.01,

1 128.17 ($\times 2$), 123.00 ($\times 2$), 121.70, 74.53, 72.26, 66.40, 62.49, 59.89, 45.69, 43.59, 42.09, 39.43,
2 34.04, 32.97 ($\times 2$), 29.96, 29.71, 23.93 ($\times 2$), 21.44, 19.45, 17.52; HRMS (ESI) m/z calcd for
3 $C_{36}H_{38}O_{10}S_3$ $[M+H]^+$ 727.1705, found 727.1703.

4 4.2. MTT assay

5 The antiproliferative activities were examined by the MTT assay following the method
6 described previously. All selected cell lines were incubated with target compounds for different
7 concentrations (64, 16, 4, 1, 0.25, 0.0625 and 0.015625 μM) through a standard 96-well plates.
8 After 72 h, MTT solution (20 mL, 5 mg/mL) and DMSO (150 mL/well) were added, then
9 incubated for 10 min at room temperature. After that, the absorbance (OD) data of each well at
10 490 nm wavelength were measured by a Microplate Reader (BIO-RAD). Four human cancer and
11 two normal cell lines were selected for the assay, and half inhibition rates (IC_{50}) were calculated.

12 4.3. H_2S release experiment

13 Sodium phosphate buffer was used to prepare the stock solution of Na_2S (20 mM) in 100 mL
14 volumetric flask. Aliquots of Na_2S stock solution were transferred into a 50 mL volumetric flask
15 to obtain standard solutions of 5, 10, 20, 40, 60, 80, 100 and 150 μM , respectively. 1 mL of each
16 standard solutions were added to react with the methylene blue (MB^+) cocktail (200 μL of 30 mM
17 $FeCl_3$ in 1.2M HCl, 200 μL of 20 mM *N,N*-dimethyl-1,4-phenylenediamine sulfate in 7.2M HCl
18 and 100 μL of 1% w/v of $Zn(OAc)_2$ in H_2O) at room temperature for 20 min in a triplicate manner
19 [83]. The mixture was measured at 670 nm in UV-Vis spectrophotometer and then the Na_2S
20 calibration curve was obtained. In order to promote compounds to release H_2S , TCEP or
21 L-cysteine was used as an accelerator. All compounds were dissolved in THF solution (40 mM)
22 and added into phosphate buffer in the presence of TCEP or L-cysteine (1 mM). Then 2 mL
23 mixture was transferred to colorimetric cuvette containing methylene blue (MB^+) cocktail in
24 designated time. After 20 min incubation, the absorbance of each compound was analyzed by
25 UV-Vis spectrophotometer at 670 nm. The H_2S concentration of each derivative was calculated
26 through standard curve.

27 4.4. Cell cycle analysis

28 Effects of **12b** on cell cycle arrest in Bel-7402 cells were tested and DNA content of cell

1 nuclei was measured via flow cytometry [84]. Bel-7402 cells were incubated in 6-well plates at
2 37 °C for 24 h. Then **12b** was dissolved in DMSO and joined cell culture with designated
3 concentrations in triplicate manner. While DMSO was used as control. After 48 h incubation, all
4 cells were centrifuged and fixed with 70% ethanol at 4 °C overnight and suspended again in PBS
5 mixed with 100 mL RNase A and 400 mL PI. Cell cycle distribution of DNA content was
6 evaluated via flow cytometer (FACS Calibur Becton-Dickinson, Franklin Lake, NJ, USA).

7 *4.5. Hoechst 333258 staining*

8 Hoechst 33258 staining assay was used to observe the morphology change of nuclei. Bel-7402
9 cells were cultured in 6-well plates at 37 °C with 2 mL medium. After 24 h of incubation, **12b** of
10 designated concentrations (0, 0.45, 0.90 and 1.80 µM) was added for another 48 h incubation.
11 Cells were harvested by mild trypsinization, collected by centrifugation and washed twice with
12 PBS. After that, 500 µL of Hoechst mixture (2 mg/mL) in PBS was added for 15 min at room
13 temperature in darkness in order to stain aforementioned cells. Finally, cells were washed by PBS
14 and assessed via a fluorescence microscope.

15 *4.6. Cell apoptosis assay*

16 Bel-7402 cells were treated with designated concentrations of **12b** (0, 0.45, 0.90 and 1.80 µM)
17 for 24 h incubation in 6-well plates. Then, cells were washed twice with PBS and suspended in
18 annexin V binding buffer. The mixture was incubated with V-FITC and PI at room temperature
19 for 15 min in dark. After that, cells with double-staining were analyzed by flow cytometry to
20 detect apoptotic cells.

21 *4.7. Mitochondrial membrane potential assay*

22 Bel-7402 cells were cultured in 6-well plates with different concentrations of **12b** (0, 0.45,
23 0.90 and 1.80 µM) for 48 h, then washed with PBS and stained with JC-1 at room temperature in
24 darkness. A flow cytometry was used to measure the number of cells with collapsed mitochondrial
25 membrane potentials.

26 *4.8. Western blot assay*

27 Bel-7402 cells were treated with **12b** (0, 0.45, 0.90 and 1.80 µM) in triplicate for 48 h.
28 Following the measurement of protein level, sodium dodecyl sulfate polyacrylamide gel

1 electrophoresis (10% gel, SDS-PAGE) was applied to split each cell lysates, and then transferred
2 onto nitrocellulose membranes. Afterwards membranes were blocked with 5% of nonfat milk,
3 incubated with monoclonal antibodies for 12 h at 4 °C. Then washed with TBST, incubated with
4 appropriate second antibodies and followed by chemiluminescence detection [85]. Protein
5 visualization was analyzed by Keygen ECL system (KeyGEN Biotech, Nanjing, China) while
6 resulting images were obtained by Clix ChemiScope chemiluminescence imaging system.
7 ChemiScope analysis program was used to gain relative optical densities for individual protein.

8

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Highlights

- H₂S donating *ent*-kaurane and spiro lactone-type derivatives were synthesized.
- 12b** showed potent antiproliferative effect against Bel-7402 cells.
- 12b** arrested Bel-7402 cell cycle at G₁ phase and induced mitochondria dysfunction.
- 12b** regulated the expression of apoptosis-related proteins.