

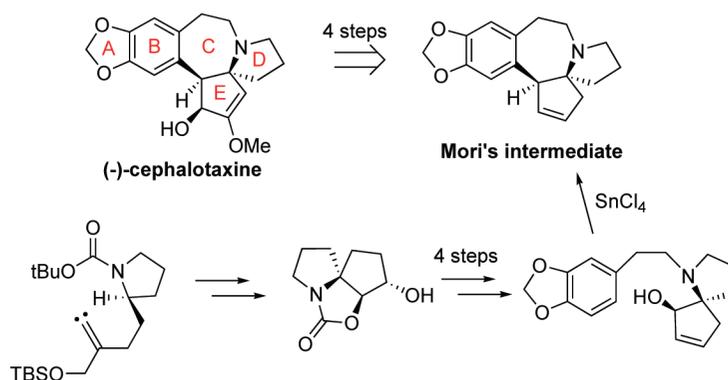
A Second Generation Formal Synthesis of (–)-Cephalotaxine

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A second generation formal synthesis of the alkaloid (–)-cephalotaxine has been achieved using an alkylidene carbene 1,5-CH insertion reaction to establish a key quaternary stereocenter. The carbene precursor was readily derived from L-proline, and the 1,5-CH insertion reaction was performed under Ohira's conditions using lithiotrimethylsilyldiazomethane (LTDM), which gave the desired spirocyclic product in 74% yield. The hydroxymethyl group was then oxidized and then decarbonylated (93%), and this material was easily transformed into the desired Friedel–Crafts cyclization precursor. Exposure of this material to SnCl₄ then gave the desired pentacyclic product, which was identical to that previously prepared by Mori and thus represents a formal total synthesis of (–)-cephalotaxine.

Introduction

The polycyclic alkaloid (–)-cephalotaxine¹ (**1**) is a fascinating target for total chemical synthesis (Figure 1) as its structure contains a number of interesting features, including an embedded 1-azaspiro[4.4]nonane ring system.

While cephalotaxine (**1**) has little known biological activity, it does represent the parent alkaloid core of a larger family of related natural products (e.g., homoharringtonine **2**) that show clinically useful anticancer activity,² and as a result, **1** has been the subject of much synthetic interest.^{3,4} As part of our ongoing interest in the enantioselective synthesis of **1**,^{5a} we have recently reported an enantioselective formal total synthesis,^{5b} and this is summarized in Scheme 1.

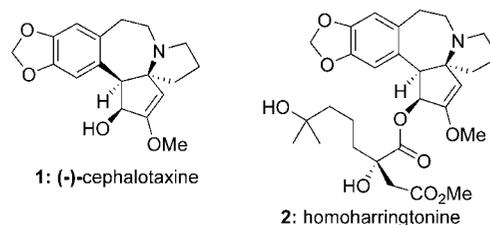
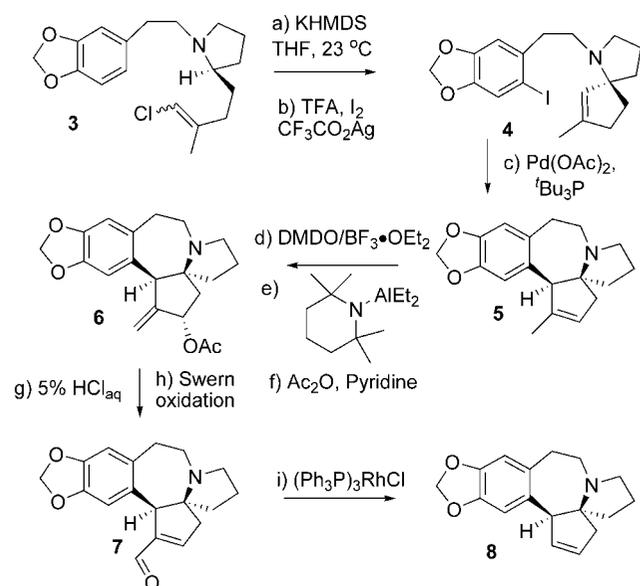


FIGURE 1. Cephalotaxine (**1**) and homoharringtonine (**2**).

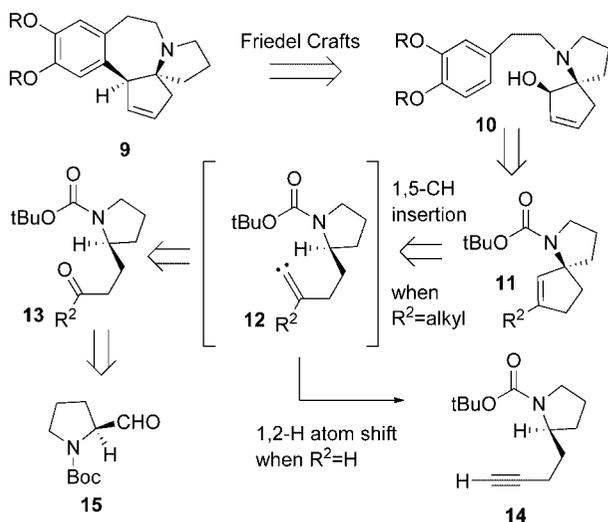
All of the steps up to the pentacycle **5** proceeded in acceptable yield (60–80%), but there were some problems associated with the conversion of **5** into “Mori's intermediate” **8**.^{4b} Effectively, we needed to remove the methyl group from the cyclopentene ring **5**, and this was done in a six-step sequence (Scheme 1). The first three steps (**5**→**6**) proceeded cleanly,⁶ but the subsequent two steps (**6**→**7**) could only be achieved in 31% combined yield. The problems surrounding this low yielding sequence were compounded by the inefficient final decarbonylation reaction (**7**→**8**) which only produced the desired compound **8** in modest yield (20–40%). As a result of these issues, we felt it necessary to explore an alternative synthetic approach (Scheme 2), and we now wish to report the

(1) Isolation: (a) Paudler, W. W.; Kerley, G. I.; McKay, J. *J. Org. Chem.* **1963**, *28*, 2194. Structure elucidation: (b) Abraham, D. J.; Rosenstein, R. D.; McGandy, E. L. *Tetrahedron Lett.* **1969**, 4085. (c) Arora, S. K.; Bates, R. B.; Grady, R. A.; Powell, R. G. *J. Org. Chem.* **1974**, *39*, 1269.

(2) (a) Levy, V.; Zohar, S.; Bardin, C.; Vekhoff, A.; Chaoui, D.; Rio, B.; Legrand, O.; Sentenac, S.; Rousselot, P.; Raffoux, E.; Chast, F.; Chevret, S.; Marie, J. P. *Br. J. Cancer* **2006**, *95*, 253. (b) Kantarjian, H. M.; Talpaz, M.; Santini, V.; Murgu, A.; Cheson, B.; O'Brien, S. M. *Cancer* **2001**, *92*, 1591.

SCHEME 1. Summary of the First Generation Synthesis of **1**

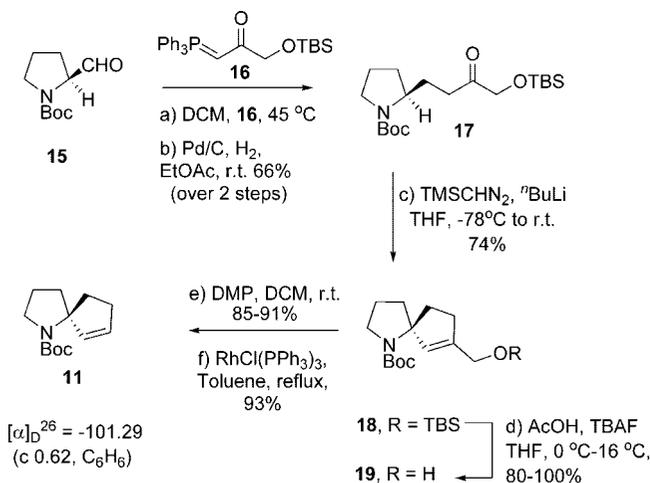
SCHEME 2. Revised Retrosynthetic Analysis



results of this work which have culminated in a second generation formal synthesis of **1**.

Results and Discussion

As the alkylidene carbene 1,5-CH insertion reaction performed well at constructing the key quaternary stereocenter (viz. **3**→**4**),⁷ we wanted to retain this in our revised route and we hoped that many of the issues surrounding the removal of the methyl group from **5** could be avoided by changing the order of some of the other key steps. Thus, the pentacycle **9** was disconnected via an intramolecular Friedel–Crafts alkylation in a similar manner to Mori^{4b} to reveal the allylic alcohol **10**, with **10** being accessed in only a few steps from the Boc-protected spirocycle **11** (R² = H). Further disconnection of **11**, via the alkylidene carbene **12**, revealed the proline-derived ketone **13** to be a suitable precursor.⁸ It is well-known that alkylidene carbenes such as **12** undergo rearrangement to the corresponding acetylenes⁹ (e.g., **14**) when R² = H, so we decided to work with an R² group that could act as surrogate for

SCHEME 3. Preparation of Spirocycle **11**

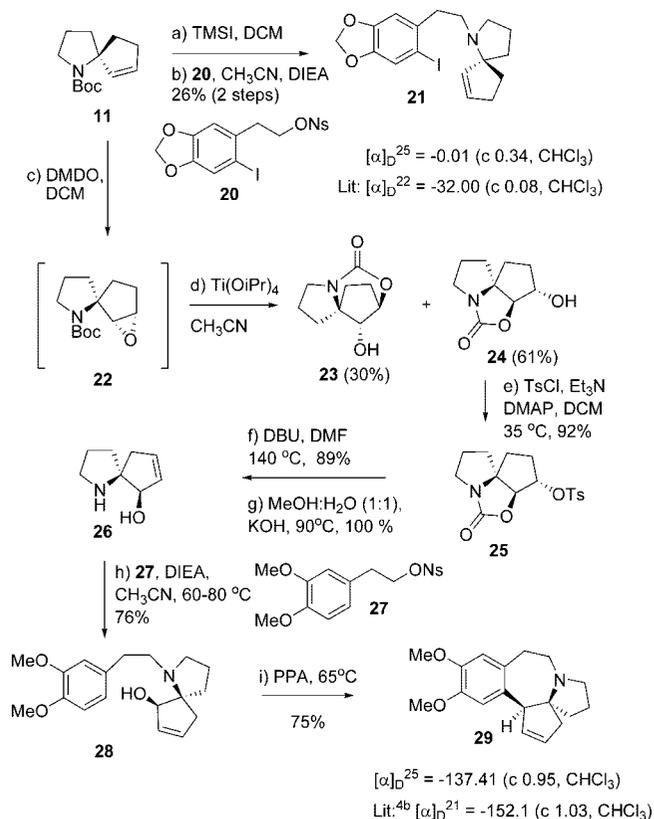
a proton until after the key CH insertion reaction. After considering several options, we decided to prepare the ketone **17** with R² as a protected hydroxymethyl group via Wittig reaction of the proline-derived aldehyde **15** and the known ylide **16**¹⁰ (Scheme 3). We felt that the TBS-protected hydroxymethyl moiety would be well suited to the CH insertion step and that rearrangement to the corresponding acetylene would be minimal.¹⁰ Fortunately, this proved to be the case and the desired spirocycle **18** could be isolated in good yield (74%) following exposure of **17** to lithio-TMS-diazomethane.⁸ The hydroxymethyl group was then removed in an efficient manner via oxidation to the aldehyde with Dess–Martin periodinane¹¹ (85–91%) and decarbonylation with Wilkinson's catalyst¹² (93%) to afford the desired spirocycle **11** (where R² = H).

Pleased by the successful formation of **11**, we next turned our attention to alkylation of the nitrogen with a suitably functionalized electrophile. As a short diversion from our originally planned retrosynthesis (Scheme 2), we briefly explored the possibility of accessing the aryl iodide **21**, which is

(3) Racemic total syntheses: (a) Auerbach, J.; Weinreb, S. M. *J. Am. Chem. Soc.* **1972**, *94*, 7172. (b) Semmelhack, M. F.; Chong, B. P.; Jones, L. D. *J. Am. Chem. Soc.* **1972**, *94*, 8629. (c) Semmelhack, M. F.; Chong, B. P.; Stauffer, R. D.; Rogerson, T. D.; Chong, A.; Jones, L. D. *J. Am. Chem. Soc.* **1975**, *97*, 2507. (d) Weinreb, S. M.; Auerbach, J. *J. Am. Chem. Soc.* **1975**, *97*, 2503. (e) Burkholder, T. P.; Fuchs, P. L. *J. Am. Chem. Soc.* **1988**, *110*, 2341. (f) Kuehne, M. E.; Bornmann, W. G.; Parsons, W. H.; Spitzer, T. D.; Blount, J. F.; Zubieta, J. *J. Org. Chem.* **1988**, *53*, 3439. (g) Burkholder, T. P.; Fuchs, P. L. *J. Am. Chem. Soc.* **1990**, *112*, 9601. (h) Ishibashi, H.; Okano, M.; Tamaki, H.; Maruyama, K.; Yakura, T.; Ikeda, M. *J. Chem. Soc., Chem. Commun.* **1990**, 1436. (i) Ikeda, M.; Okano, M.; Kosaka, K.; Kido, M.; Ishibashi, H. *Chem. Pharm. Bull.* **1993**, *41*, 276. (j) Lin, X.; Kavash, R. W.; Mariano, P. S. *J. Am. Chem. Soc.* **1994**, *116*, 9791. (k) Lin, X.; Kavash, R. W.; Mariano, P. S. *J. Org. Chem.* **1996**, *61*, 7335. (l) Tietze, L. F.; Schirok, H. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 1124. (m) Koseki, Y.; Sato, H.; Watanabe, Y.; Nagasaka, T. *Org. Lett.* **2002**, *4*, 885. (n) Suga, S.; Watanabe, M.; Yoshida, J. *J. Am. Chem. Soc.* **2002**, *124*, 14824. (o) Li, W.-D. Z.; Wang, Y.-Q. *Org. Lett.* **2003**, *5*, 2931. (p) Li, W.-D. Z.; Ma, B.-C. *J. Org. Chem.* **2005**, *70*, 3277. (q) Ma, B.-C.; Wang, Y.-Q.; Li, W.-D. Z. *J. Org. Chem.* **2005**, *70*, 4528. (r) Li, W.-D. Z.; Wang, X.-W. *Org. Lett.* **2007**, *9*, 1211.

(4) Enantioselective total syntheses: (a) Zhong, S.; Liu, W.; Ling, Y.; Li, R. *Zhongguo Yaowu Huaxue Zazhi* **1994**, *4*, 84. (b) Isono, N.; Mori, M. *J. Org. Chem.* **1995**, *60*, 115. (c) Nagasaka, T.; Sato, H.; Saeki, S.-I. *Tetrahedron: Asymmetry* **1997**, *8*, 191. (d) Ikeda, M.; El Bialy, S. A. A.; Hirose, K.-I.; Kotake, M.; Sato, T.; Bayomi, S. M. M.; Shehata, I. A.; Abdelal, A. M.; Gad, L. M.; Yakura, T. *Chem. Pharm. Bull.* **1999**, *47*, 983. (e) Tietze, L. F.; Schirok, H. *J. Am. Chem. Soc.* **1999**, *121*, 10264. (f) El Bialy, S. A. A.; Ismail, M. A.; Gad, L. M.; Abdelal, A. M. *Med. Chem. Res.* **2002**, *11*, 293. (g) Planas, L.; Perard-Viret, J.; Royer, J. *J. Org. Chem.* **2004**, *69*, 3087. (h) Eckelbarger, J. D.; Wilmut, J. T.; Gin, D. Y. *J. Am. Chem. Soc.* **2006**, *128*, 10370. (i) Zhao, Z.; Mariano, P. S. *Tetrahedron* **2006**, *62*, 7266. (j) Liu, Q.; Ferreira, E. M.; Stoltz, B. M. *J. Org. Chem.* **2007**, *72*, 7352.

SCHEME 4. Completion of Synthesis of Tetracycle 29

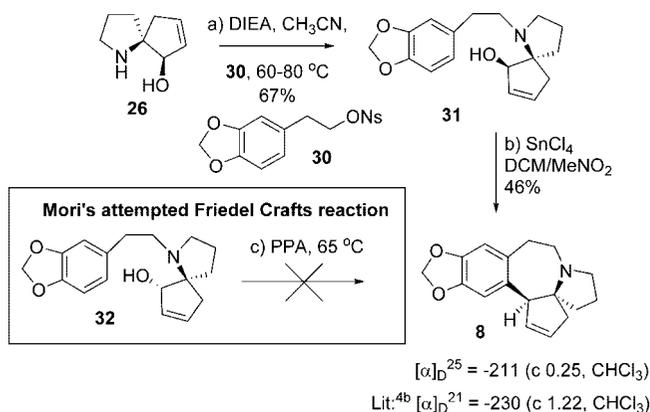


a known precursor to **8**,¹³ from the protected spirocycle **11** (Scheme 4). We quickly found that racemization occurs at the quaternary stereocenter upon deprotection of **11** under a variety of acidic conditions.¹⁴ Furthermore, only low yields of the racemic alkylated product **21** could be obtained, and despite much effort, we were not able to find acceptable conditions for this transformation. We therefore returned to our original synthetic approach to **8**.

Epoxidation of **11** with DMDO gave the epoxide **22** as a single diastereoisomer, with delivery of the oxygen occurring on the alkene face opposite to the bulky Boc group. Purification of this material by column chromatography (SiO_2) gave the desired epoxide **22** (48%), along with smaller amounts of two new compounds that were not present in the crude reaction mixture. These were later identified as the isomeric carbamates **23** (4%) and **24** (9%), which are obviously products of acid-mediated intramolecular epoxide opening. Rather than being problematic, we decided to exploit this intramolecular epoxide opening by immediately treating the DMDO epoxidation product **22** with a Lewis acid ($\text{Ti}(\text{O}^i\text{Pr})_4$, MeCN) to provide **23** (30%) and **24** (61%) in much improved yields. The stereochemistry of **24** was confirmed by X-ray crystallography (see Supporting Information).

Although both materials are potentially useful for the synthesis of cephalotaxine, we decided to advance the most abundant isomer **24** through the remaining steps. Thus, **24** was converted into its tosylate **25**, and this material was then treated with DBU to induce elimination to provide the corresponding cyclopentene **35** (structure

SCHEME 5. Shorter End-Game for the Formal Synthesis of (–)-Cephalotaxine



shown in Supporting Information). Hydrolysis of the carbamate (KOH, MeOH/ H_2O) then gave the amino alcohol **26** in excellent yield. Alkylation of **26** with the known nosylate **27** produced the desired Friedel–Crafts cyclization precursor **28**, which is the diastereoisomer of a compound previously used by Mori.^{4b} Upon treatment with neat polyphosphoric acid (PPA, 65 °C), **28** cyclized to give the desired tetrahydroazepine **29**, whose spectroscopic and analytical data were identical to those previously reported by Mori.^{4b}

Although we were pleased to have completed a formal synthesis of (–)-cephalotaxine (**1**) by accessing the known intermediate **29**, we felt that the synthesis could be improved further if we were able to access **8** directly via Friedel–Crafts cyclization of allylic alcohol **31** (Scheme 5). Mori prepared **8** from the dimethoxy aromatic derivative **29** in a two-step process on the way to his total synthesis of **1**,^{4b} so if this reaction was possible (viz. **31** → **8**, Scheme 5), then our formal synthesis would be shortened by two steps.

The reason for not trying this transformation (**31** → **8**) earlier was based on the fact that Mori reported that **32**, the diastereomer of **31**, did not undergo the desired Friedel–Crafts cyclization and instead had to resort to using the dimethoxy aromatic derivative (i.e., the diastereoisomer of **28**). We were slightly puzzled by this observation as Kuehne^{3f} had previously shown that Friedel–Crafts cyclization of a very closely related amide derivative of **32** was possible.¹⁵ Furthermore, we successfully performed an electrophilic

(5) (a) Worden, S. M.; Mapitise, R.; Hayes, C. J. *Tetrahedron Lett.* **2002**, *43*, 6011. (b) Esmieu, W. R.; Worden, S. M.; Catterick, D.; Wilson, C.; Hayes, C. J. *Org. Lett.* **2008**, *10*, 3045.

(6) For regioselective rearrangement of epoxides to allylic alcohols, see: Yasuda, A.; Tanaka, S.; Oshima, K.; Yamamoto, H.; Nozaki, H. *J. Am. Chem. Soc.* **1974**, *96*, 6513.

(7) For recent applications of alkylidene carbene 1,5-CH insertion reactions in synthesis, see: (a) Sakai, A.; Aoyama, T.; Shioiri, T. *Tetrahedron Lett.* **2000**, *41*, 6859. (b) Taber, D. F.; Neubert, T. D. *J. Org. Chem.* **2001**, *66*, 143. (c) Taber, D. F.; Neubert, T. D.; Rheingold, A. L. *J. Am. Chem. Soc.* **2002**, *124*, 12416. (d) Taber, D. F.; Storck, P. H. *J. Org. Chem.* **2003**, *68*, 7768. (e) Feldman, K. S.; Perkins, A. L.; Masters, K. M. *J. Org. Chem.* **2004**, *69*, 7928. (f) Auty, J. M. A.; Churcher, I.; Hayes, C. J. *Synlett* **2004**, 1443. (g) Grainger, R. S.; Owoare, R. B. *Org. Lett.* **2004**, *6*, 2961. (h) Wardrop, D. J.; Bowen, E. G. *Chem. Commun.* **2005**, 5106. (i) Taber, D. F.; Liang, J.-L.; Chen, B.; Cai, L. *J. Org. Chem.* **2005**, *70*, 8739. (j) Hayes, C. J.; Sherlock, A. E.; Selby, M. D. *Org. Biol. Chem.* **2006**, *4*, 193. (k) Hayes, C. J.; Sherlock, A. E.; Green, M. P.; Wilson, C.; Blake, A. J.; Selby, M. D.; Prodder, J. C. *J. Org. Chem.* **2008**, *73*, 2041.

(8) Ohira, S.; Okai, K.; Moritani, T. *J. Chem. Soc., Chem. Commun.* **1992**, 721.

(9) For a comprehensive review, please see: Knorr, R. *Chem. Rev.* **2004**, *104*, 3795.

(10) Bradley, D. M.; Mapitise, R.; Thomson, N. M.; Hayes, C. J. *J. Org. Chem.* **2002**, *67*, 7613.

(11) (a) Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* **1991**, *113*, 7277. (b) Ireland, R. E.; Liu, L. *J. Org. Chem.* **1993**, *58*, 2899.

(12) Ohno, K.; Tsuji, J. *J. Am. Chem. Soc.* **1968**, *90*, 99.

(13) Yoshida has prepared racemic **21** (see ref 3n) and has shown that it will undergo Heck cyclization under Tietze's conditions (see ref 4e). Mariano has recently reported an enantioselective synthesis of **21** (see ref 4i) but does not report the use of this material in the Heck cyclization.

aromatic iodination on a similar methylenedioxy-substituted aromatic ring to afford the iodide **20**,³ⁿ so, in principle, we felt that the aromatic ring present in **31** should be nucleophilic enough to react with the putative allylic cation electrophile that is produced during the Friedel–Crafts reaction.

We therefore prepared **31** by alkylation of **26** with the known nosylate **30**, and we exposed this material to Kuehne's conditions (SnCl₄, DCM/MeNO₂).^{3f} We avoided using polyphosphoric acid as we felt that the dioxolane ring present in **31** could decompose under these harshly acidic conditions. Pleasingly, we were able to isolate the desired pentacycle **8** as the major new product from the reaction, and its spectroscopic data were identical to those previously reported.^{4b} Since it has been shown that **8** can be converted into (–)-cephalotaxine (**1**), we have successfully completed a second generation formal synthesis of the natural product **1**.

Experimental Section

(R)-N-Boc-7-[(tert-butyl)dimethylsilyloxy]methyl-1-azaspiro[4.4]non-6-ene 18. *n*-Butyl lithium (11.1 mL, 2.28 M solution in hexanes, 25.4 mmol) was added dropwise to a cold (–78 °C) stirring solution of trimethylsilyldiazomethane (14.6 mL, 2 M solution in hexanes, 29.2 mmol) in THF (39 mL), and the resulting mixture was stirred for 1 h and 40 min at –78 °C. The ketone **17** (7.1 g, 19.1 mmol) in THF (25 mL) was added dropwise, and the resulting mixture was stirred for a further 80 min. The reaction mixture was warmed to 0 °C and stirred for 20 min. The reaction was quenched with NH₄Cl (24 mL of a saturated solution diluted with water (20 mL)), extracted with ether (100 and 50 mL), dried with MgSO₄, filtered, and concentrated in vacuo to give the crude product. Purification by column chromatography (petrol/Et₂O; 10/1–5/1) afforded the product **18** (5.16 g, 74%) as a colorless oil: [α]_D²⁵ –49.56 (*c* 1.36, CHCl₃); ν_{max}/cm^{–1} (CHCl₃) 2954, 2929, 2857, 1731, 1668, 1457; δ_H (500 MHz, C₆D₆, 343 K) 5.35 (1H, s), 4.12 (1H, d, *J* = 13.5 Hz), 4.08 (1H, d, *J* = 13.5 Hz), 3.38 (1H, br s), 3.27 (1H, app q, *J* = 8 Hz), 2.52–2.30 (2H, m), 2.16 (1H, app quint, *J* = 8 Hz), 1.70 (1H, m), 1.59 (2H, m), 1.49–1.62 (2H, m), 1.41 (9H, s), 0.93 (9H, s), 0.04 (6H, s); δ_C (125 MHz, 298 K) (2.9:1 mixture of rotamers; signals from major rotamer marked with *) 154.6* (C), 153.1 (C), 143.5 (C), 142.8* (C), 129.8* (CH), 129.2 (CH), 78.6* (C), 78.3 (C), 74.0* (C), 62.5 (CH₂), 62.2* (CH₂), 47.7 (CH₂), 47.5* (CH₂), 40.9* (CH₂), 39.5 (CH₂), 36.3* (CH₂), 35.2 (CH₂), 30.9* (CH₂), 28.5 (3 × CH₃), 28.4* (3 × CH₃), 25.9 (3 × CH₃), 22.8 (CH₂), 22.5* (CH₂), 18.3* (C), –5.4* (2 × CH₃); *m/z* (ES⁺) 390.2430 (M + Na, C₂₀H₃₇NNaO₃Si requires 390.2435); Anal. Calcd for C₂₀H₃₇NO₃: C, 65.3; H, 10.2; N, 3.8. Found: C, 64.9; H, 10.1; N, 3.9.

(R)-N-Boc-1-azaspiro[4.4]non-6-ene 11. To a stirring solution of **33** (260 mg, 1.03 mmol) in toluene (5.20 mL) at room temperature was added Rh(PPh₃)₃Cl (953 mg, 1.03 mmol), and the mixture was heated at reflux for 14 h. The reaction mixture was cooled to room temperature and diluted with EtOAc (23 mL). Then, 2 N NaOH (40 mL) was added, and the layers were separated. The aqueous layer was extracted with EtOAc (50 mL × 2). The combined organic layers were washed with brine (50 mL), dried with MgSO₄, filtered, and concentrated in vacuo to give a brown oil, which was purified by column chromatography (ether/pentane, 1/5–1/4) to give **11** as a yellow oil (213 mg, 93%): [α]_D²⁶ –101.3 (*c* 0.62, C₆H₆); ν_{max}/cm^{–1} (C₆H₆) 2928, 2870, 1698, 1685, 1408; δ_H (270 MHz, C₆D₆, 343 K) 5.61 (1H, br s), 5.47 (1H, br s), 3.45 (1H, br s), 3.37–3.26 (1H, m), 2.52 (1H, m), 2.36 (1H, m), 2.15 (1H, m), 1.63 (1H, m), 1.57 (2H, m), 1.46 (9H, s), 1.43–1.35 (2H, m); δ_C (125 MHz, 298 K) (1.0:0.8 mixture of rotamers) 154.2 (C),

153.1 (C), 137.3 (CH), 135.1 (CH), 130.9 (CH), 129.0 (CH), 78.4 (C), 78.1 (C), 74.7 (C), 74.3 (C), 48.1 (CH₂), 4.9 (CH₂), 40.9 (CH₂), 40.1 (CH₂), 36.1 (CH₂), 35.2 (CH₂), 32.0 (CH₂), 31.5 (CH₂), 28.7 (3 × CH₃), 28.6 (3 × CH₃), 23.2 (CH₂), 22.6 (CH₂); *m/z* (ES⁺) 246.1457 (M + Na, C₁₃H₂₁NNaO₂ requires 246.1465).

(3aS,13bS)-3,5,6,8,9,13b-Hexahydro-11,12-dimethoxy-4H-cyclopenta[*a*]pyrrolo[2,1-*b*]-[3]benzazepine 29. A mixture of **28** (32 mg, 0.105 mmol) and polyphosphoric acid (0.85 g) was stirred at 65 °C for 43 h. The reaction mixture was diluted with DCM (1.5 mL) quenched with saturated NaHCO₃ (15 mL), and the solution was made basic (pH = 14) by 15% NaOH (10 mL) solution. The mixture was extracted with DCM (10 mL × 2), CHCl₃ (10 mL), and finally with CHCl₃/IPA (2/1, 15 mL × 2), and the combined organic layers were dried with Na₂SO₄, filtered, and concentrated in vacuo to give **29** as a light yellow solid (22 mg, 73%). The crude material was judged >90% pure by ¹H and ¹³C NMR, and purification on neutral alumina with eluents (EtOAc/hexane, 1:3–3:1 and 1% Et₃N in EtOAc) gave purified **29** (11 mg, 36%) for analytical purposes: [α]_D²⁵ –137.4 (*c* 0.95, CHCl₃), Lit. [α]_D²¹ –152.1 (*c* 1.03, CHCl₃); ν_{max}/cm^{–1} (CHCl₃) 2930, 2808, 1747, 1693, 1608, 1488; δ_H (400 MHz, CDCl₃, 298 K) 6.68 (1H, s), 6.62 (1H, s), 5.79 (1H, app dq, *J* = 6, 2.2 Hz), 5.54 (1H, app dq, *J* = 6, 2.4 Hz), 3.89 (1H, m), 3.86 (3H, s), 3.84 (3H, s), 3.20 (1H, ddd, *J* = 14, 12.4, 7.6 Hz), 3.08 (1H, app td, *J* = 9.2, 4.8 Hz), 2.94 (1H, app td, *J* = 12, 6.8 Hz), 2.76 (1H, app dq, *J* = 17.6, 2.4 Hz), 2.56 (1H, dd, *J* = 11.2, 7.6 Hz), 2.45–2.59 (2H, m), 2.04–1.90 (3H, m), 1.83–1.61 (2H, m); δ_C (100 MHz, CDCl₃, 298 K) 147.7 (C), 147.2 (C), 132.6 (CH), 131.2 (C), 130.9 (C), 128.7 (CH), 144.2 (CH), 113.1 (CH), 68.2 (C), 62.4 (CH), 56.2 (CH₃), 56.8 (CH₃), 53.8 (CH₂), 49.2 (CH₂), 43.5 (CH₂), 34.8 (CH₂), 30.6 (CH₂), 20.0 (CH₂); *m/z* (ES⁺) 286.1794 (M + H, C₁₈H₂₄NO₂ requires 286.1802). These data are identical to those previously reported.^{4b}

(3aS,14bS)-3,5,6,8,9,14b-Hexahydro-4H-cyclopenta[*S*]-[1,3]dioxolo[4,5-*h*]pyrrolo[2,1-*b*]-[3]benzazepine 8. To the solution of **31** (20 mg, 0.067 mmol) in DCM/MeNO₂ (1:1, 2 mL) was added SnCl₄ (80 μL, 0.68 mmol) dropwise at –78 °C, and the mixture was then warmed to 50 °C and stirred at this temperature for 14 h. The reaction mixture was cooled to room temperature and diluted with EtOAc (5 mL). Then, 1 N HCl (1 mL) was added, and the pH was adjusted to 14 by addition of 2 N NaOH. The layers were separated, and the aqueous solution was extracted with EtOAc (10 mL), DCM (10 mL × 2), and finally with CHCl₃/MeOH (9:1, 10 mL). The combined organic layers were dried with MgSO₄ and concentrated in vacuo to give a brown oil, which was purified by column chromatography [1% Et₃N in EtOAc/pentane (1:1)] to yield **8** (9.2 mg, 46%) as a light yellow oil: [α]_D²⁴ –210.5 (*c* 0.25, CHCl₃), Lit. [α]_D²² –230.8 (*c* 1.22, CHCl₃); δ_H (400 MHz, CDCl₃, 298 K) 6.65 (1H, s), 6.59 (1H, s), 5.89 (1H, d, *J* = 1.6 Hz), 5.88 (1H, d, *J* = 1.6 Hz), 5.79 (1H, app dq, *J* = 5.9, 2.4 Hz), 5.52 (1H, app dq, *J* = 5.9, 2.2 Hz), 3.86 (1H, m), 3.19 (1H, ddd, *J* = 13.9, 12.5, 7.4 Hz), 3.06 (1H, app td, *J* = 9.0, 4.4 Hz), 2.90 (1H, app td, *J* = 11.7, 6.8 Hz), 2.74 (1H, app dq, *J* = 18, 2.4 Hz), 2.54 (1H, dd, *J* = 11.2, 7.5 Hz), 2.40 (1H, app td, *J* = 10, 6.1 Hz), 2.33 (1H, dd, *J* = 14.4, 6.4 Hz), 2.02–1.90 (3H, m), 1.83–1.64 (2H, m); δ_C (100 MHz, CDCl₃, 298 K) 146.4 (C), 146.0 (C), 132.5 (CH), 131.9 (C), 128.8 (CH), 110.9 (CH), 109.9 (CH), 100.9 (CH₂), 68.1 (C), 62.5 (CH), 53.7 (CH₂), 49.1 (CH₂), 43.4 (CH₂), 34.8 (CH₂), 30.7 (CH₂), 20.0 (CH₂); *m/z* (ES⁺) 270.1489 (M + H, C₁₇H₂₀NO₂ requires 270.1498), 271 (19, M + 2H). These data are identical to those previously reported.^{4b}

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Supporting Information Available: Experimental procedure for preparation of **17**, **19**, **21**, **22**, **24**, **25**, **26**, **28**, **31**, **34**, and **35**, X-ray structure of **24**, and copies of the ¹H and ¹³C NMR spectra of all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

(14) We have recently observed epimerization of a similar spirocycle under acidic conditions: Asari, A.; Angelov, P.; Auty, J. M.; Hayes, C. *J Tetrahedron Lett.* **2007**, *48*, 2631.

(15) Royer has independently confirmed the success of the Kuehne Friedel–Crafts reaction. See ref 4g.