

2-Aminomethyl piperidines as novel urotensin-II receptor antagonists

Jian Jin,^{a,*} Yonghui Wang,^b Feng Wang,^c Dongchuan Shi,^a Karl F. Erhard,^b
Zining Wu,^c Brian F. Guida,^c Sarah K. Lawrence,^c David J. Behm,^a Jyoti Disa,^a
Kalindi S. Vaidya,^c Christopher Evans,^a Lynette J. McMillan,^c
Ralph A. Rivero,^b Michael J. Neeb^a and Stephen A. Douglas^a

^aCardiovascular and Urogenital Center of Excellence for Drug Discovery, GlaxoSmithKline,
709 Swedeland Road, King of Prussia, PA 19406, USA

^bOncology Center of Excellence for Drug Discovery, GlaxoSmithKline, 1250 South Collegeville Road, Collegeville, PA 19426, USA

^cMolecular Discovery Research, GlaxoSmithKline, 1250 South Collegeville Road, Collegeville, PA 19426, USA

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Abstract—A series of 2-aminomethyl piperidines has been discovered as novel urotensin-II receptor antagonists. The synthesis, initial structure-activity relationships, and optimization of the initial hit that resulted in the identification of potent, cross-species active, and functional urotensin-II receptor antagonists such as **1a** and **11a** are described.

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Human urotensin-II (hU-II), the most potent mammalian vasoconstrictor identified to date,¹ and its cognate receptor hUT (formerly known as the GPR-14 receptor) are proposed to be involved in the (dys)regulation of cardiorenal function,² and have been implicated in the etiology of numerous cardiorenal and metabolic diseases including hypertension,³ heart failure,^{4,5} atherosclerosis,⁶ renal failure,⁷ and diabetes.⁸ The impressive pharmacological activity of U-II has stimulated a great deal of interest in developing small molecule UT modulators. A number of non-peptidic UT ligands have recently been reported.⁹ Herein we describe the identification, synthesis, and initial structure-activity relationships (SAR) of a novel 2-aminomethyl piperidine series. Optimization of the series led to the identification of potent, competitive, and reversible UT antagonists such as **1a** and **11a** with excellent and broad cross-species functional activity.

High throughput screening (HTS) of the corporate compound collection using a fluorometric imaging plate reader (FLIPR) assay (measuring inhibition of hU-II-mediated $[Ca^{2+}]_i$ -mobilization in HEK293 cells expressing human recombinant UT receptor)¹⁰ led to the identification of **2**¹¹ as an antagonist with a pIC_{50} of 6.2 (Fig. 1). The compound also showed moderate hUT binding affinity with a pK_i of 6.4 in a [¹²⁵I]hU-II radioligand binding assay using HEK293 cell membranes stably expressing human recombinant UT receptors.¹⁰ Subsequent early exploration of the left-hand side (LHS) of this hit quickly resulted in the identification of an α -aryl acetamide sub-series exemplified by **3a** (pK_i 6.3).¹² Despite the modest binding affinity, compounds

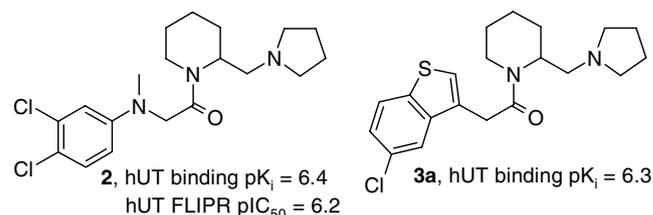
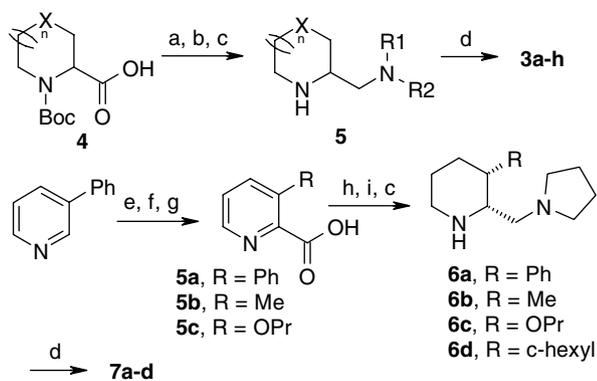


Figure 1. Structures of HTS hit **2** and α -aryl acetamide sub-series hit **3a**.

Keywords: Urotensin-II receptor antagonist; UT antagonist; 2-amino-methyl piperidine; Broad cross-species activity; Competitive, reversible, and functional antagonist.

* Corresponding author. Tel.: +1 610 270 4881; fax: +1 610 270 4490; e-mail: jian.jin@gsk.com



Scheme 1. Reagents and conditions: (a) amine, EDC, HOAt, CH₂Cl₂, rt; (b) 4 M HCl in dioxane, MeOH, rt; (c) LiAlH₄, THF, 0 °C–rt; (d) 2-(5-chlorobenzothiophen-3-yl)acetic acid, EDC, HOAt, CH₂Cl₂, rt; (e) *m*-CPBA, CH₂Cl₂, 0 °C–rt; (f) (CH₃)₂NCOCl, TMSCN, CH₂Cl₂, rt; (g) concd HCl and H₂SO₄, reflux; (h) pyrrolidine, EDC, HOAt, CH₂Cl₂, rt; (i) H₂, PtO₂, HOAc, rt.

2 and **3a** were considered as reasonable starting points for our hit-to-lead chemistry optimization aimed at improving potency via SAR exploration.

We first investigated the 2-aminomethyl piperidine region, also referred to as the diamine region. Custom diamines **5** were prepared from Boc-protected amino acids **4** via amide formation, deprotection, and LiAlH₄ reduction (**Scheme 1**). Subsequent coupling of diamines **5** with commercially available 2-(5-chlorobenzothiophen-3-yl)acetic acid produced the desired compounds **3a–h**.¹³ To explore the effect of 3-substituents on the central piperidine ring, 3-phenylpyridine was first converted to the corresponding 2-carboxylic acid **5a** via *N*-oxide formation, installation of the 2-CN group,¹⁴ and hydrolysis. Acid **5a** and commercially available **5b–c** were then converted to racemic *cis*-3-substituted-2-pyrrolidinylmethyl piperidines **6a–d** via amide formation, reduction of the pyridine ring, and subsequent amide reduction.¹⁵ Standard amide coupling of **6a–d** with commercially available acids produced the desired compounds **7a–d** and **8a–j**.

For the 2-aminomethyl moiety, pyrrolidine (**3a**) was greater than 10-fold more potent compared to piperidine (**3b**) and acyclic analog (**3e**) (**Table 1**). Interestingly, unlike piperidine (**3b**), morpholine (**3c**) showed moderate binding affinity while *N*-methyl piperazine (**3e**) had no appreciable affinity. The SAR indicated that the size and the basicity of the 2-aminomethyl moiety were critical to UT binding. As for the central piperidine ring moiety, 6-membered ring (**3a**) was preferred compared to 5- and 7-membered rings (**3g** and **h**) while morpholine (**3f**) was tolerated. Additional substituents on the central piperidine ring were then explored. We were pleased to find that 3-substituents (**7a–d**) improved affinity with 3-phenyl (**7a**) being optimal—resulting in close to 100-fold affinity improvement compared to **3a** (**Table 2**). The 4- and 5-phenyl analogs (**7e** and **f**)¹⁶ also had higher binding affinity compared to **3a**, but were less potent compared to 3-phenyl compound **7a**.

Table 1. SAR of the diamine region

Compound	<i>n</i>	X	NR ¹ R ²	hUT binding (pK _i) ^a
3a	1	CH ₂	Pyrrolidin-1-yl	6.3
3b	1	CH ₂	Piperidin-1-yl	5.1
3c	1	CH ₂	Morpholin-4-yl	5.9
3d	1	CH ₂	<i>N</i> -Methylpiperazin-1-yl	<5.1
3e	1	CH ₂	<i>N,N</i> -Diethylamino	<5.1
3f	1	O	Pyrrolidin-1-yl	6.1
3g	0	CH ₂	Pyrrolidin-1-yl	<5.1
3h	2	CH ₂	Pyrrolidin-1-yl	5.7

^a Mean of at least 3 determinations with standard deviation of <±0.3.

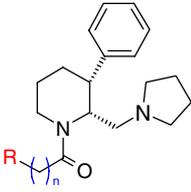
Table 2. Substitution on the central piperidine ring

Compound	R	Relative stereochemistry	hUT binding (pK _i) ^a
3a	H	—	6.3
7a	3-Phenyl	<i>cis</i>	8.1
7b	3-Methyl	<i>cis</i>	6.7
7c	3-Propoxy	<i>cis</i>	7.5
7d	3-Cyclohexyl	<i>cis</i>	7.6
7e	4-Phenyl	<i>cis</i>	6.7
7f	5-Phenyl	<i>trans</i>	7.5

^a Mean of at least 3 determinations with standard deviation of <±0.3.

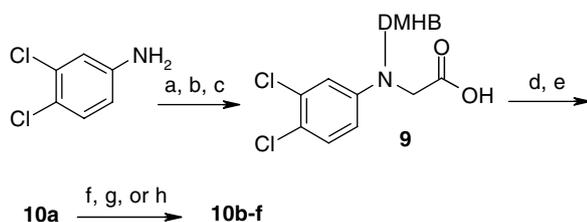
After identifying **6a** as the optimal diamine, we next turned our attention to optimizing the left-hand side (LHS) moiety of the α -aryl acetamide sub-series. Compared to unsubstituted 3-benzothiophene (**8c**), 5-substituted-3-benzothiophenes (**7a**, **8a–b**) were 15- to 30-fold more potent, with 5-Cl (**7a**) and 5-Br (**8a**) being optimal (**Table 3**). Indole is less preferred (**8d** vs **8a**) and 1-naphthyl (**8e**) was 100-fold more potent than 2-naphthyl (**8f**). Monocyclic aromatic groups such as 3-thiophene (**8g**) and 3,4-dichlorophenyl (**8h**), the preferred LHS group in the α -amino acetamide sub-series (*vide infra*), had much lower binding affinity. As for the length of the carbon linker, one carbon was preferred over no or two carbons (**8e** vs **8i**, **8b** vs **8j**).

To efficiently explore the α -amino acetamide sub-series, a solid-phase synthetic route outlined in **Scheme 2** was developed. Resin-bound α -amino acetic acid **9** was prepared via loading of 3,4-dichloroaniline onto commercially available 2,6-dimethoxy-4-polystyrene-benzyl-

Table 3. SAR of the LHS moiety of the α -aryl acetamide sub-series


Compound	R	n	hUT binding (p <i>K</i> _i) ^a
7a	5-Chlorobenzothiophen-3-yl	1	8.1
8a	5-Bromobenzothiophen-3-yl	1	8.1
8b	5-Methylbenzothiophen-3-yl	1	7.8
8c	Benzothiophen-3-yl	1	6.6
8d	5-Bromoindol-3-yl	1	6.2
8e	1-Naphthyl	1	7.1
8f	2-Naphthyl	1	5.1
8g	Thiophen-3-yl	1	<5.1
8h	3,4-Dichlorophenyl	1	5.6
8i	1-Naphthyl	0	<5.1
8j	5-Methylbenzothiophen-3-yl	2	5.5

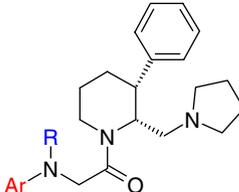
^a Mean of at least 3 determinations with standard deviation of $<\pm 0.3$.



Scheme 2. Reagents and conditions: (a) 2,6-dimethoxy-4-polystyrenebenzyloxy-benzaldehyde, Na(OAc)₃BH, DIEA, 1% of HOAc in NMP, rt; (b) methyl bromoacetate, DIEA, NMP, 80 °C; (c) potassium trimethylsilylanolate, THF, rt; (d) **6a**, PyBOP, DIEA, NMP, rt; (e) 50% of TFA in DCE, rt; (f) aldehyde, Na(OAc)₃BH, HOAc, DIEA, DCE, rt; (g) acyl chloride, TEA, CH₂Cl₂, rt; (h) MeSO₂Cl, pyridine, CH₂Cl₂, rt.

oxybenzaldehyde resin (DMHB resin), followed by alkylation and hydrolysis. Amide coupling of resin-bound acid **9** with diamine compound **6a** and subsequent resin cleavage provided compound **10a**, which was further elaborated into **10b–f** via reduction amination, acylation, or sulfonylation. Compounds **10g–r** were prepared in a similar manner as **10b** by replacing 3,4-dichloroaniline with various commercially available amines.

As shown in **Table 4**, the optimal R group is methyl (**10b**). While hydrogen (**10a**) and alkyl groups (**10b–d**) showed good binding affinity, acetyl (**10e**) had very poor affinity. Interestingly, methylsulfonyl (**10f**) was quite potent—indicating that the weakly basic center in this region was unnecessary for UT binding. 3,4-Dichlorophenyl (**10b**) was the optimal aryl (Ar) group. Bioisosteres such as 3,4-dimethylphenyl (**10g**) and 2-naphthyl (**10h**) were 10-fold less potent. *Meta*-substituted phenyl (**10i**) was preferred over *para*- (**10j**) and *ortho*- (**10k**) substituted ones. Comparing substituents on the phenyl ring, 3-chloro (**10i**) and 3-trifluoromethyl (**10m**) were preferred over 3-methyl (**10n**) and 3-methoxy (**10p**),

Table 4. SAR of the LHS moiety of the α -amino acetamide sub-series


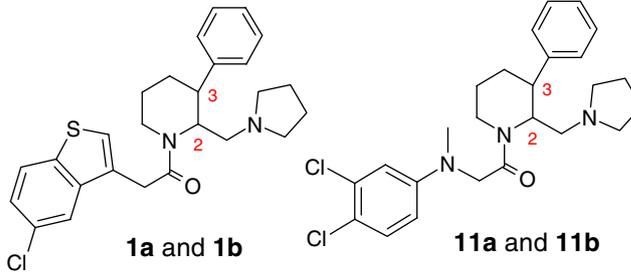
Compound	Ar	R	hUT binding (p <i>K</i> _i) ^a
10a	3,4-Dichlorophenyl	Hydrogen	7.2
10b	3,4-Dichlorophenyl	Methyl	7.8
10c	3,4-Dichlorophenyl	Ethyl	7.5
10d	3,4-Dichlorophenyl	Cyclopropyl-methyl	7.3
10e	3,4-Dichlorophenyl	Acetyl	5.8
10f	3,4-Dichlorophenyl	Methylsulfonyl	7.1
10g	3,4-Dimethylphenyl	Methyl	6.7
10h	2-Naphthyl	Methyl	6.8
10i	3-Chlorophenyl	Methyl	7.1
10j	4-Chlorophenyl	Methyl	6.3
10k	2-Chlorophenyl	Methyl	6.1
10m	3-Trifluoromethyl-phenyl	Methyl	7.0
10n	3-Methylphenyl	Methyl	6.4
10p	3-Methoxyphenyl	Methyl	6.1
10q	Phenyl	Methyl	5.5
10r	3,4-Dichlorobenzyl	Methyl	5.4

^a Mean of at least 3 determinations with standard deviation of $<\pm 0.3$.

and the unsubstituted phenyl (**10q**) was least preferred. Extending the phenyl to benzyl such as **10r** resulted in significant affinity loss.

We then investigated the preferred stereochemistry and found that (2*R*, 3*R*) was preferred as illustrated in **Table 5**. The (2*R*, 3*R*) enantiomers **1a** and **11a** were greater than 100-fold more potent than the (2*S*, 3*S*) enantiomers **1b** and **11b**.¹⁷

In addition to the high affinity to the hUT receptor, the preferred enantiomers **1a** and **11a** showed excellent

Table 5. Preferred stereochemistry


Compound	Stereochemistry and ee	hUT binding (p <i>K</i> _i) ^a
1a	2 <i>R</i> , 3 <i>R</i> (98% ee)	8.4
1b	2 <i>S</i> , 3 <i>S</i> (>99% ee)	6.0
11a	2 <i>R</i> , 3 <i>R</i> (98% ee)	8.2
11b	2 <i>S</i> , 3 <i>S</i> (>99% ee)	5.8

^a Mean of at least 3 determinations with standard deviation of $<\pm 0.3$.

Table 6. Binding affinity versus ortholog receptors and functional activity in rat and cat isolated arteries

Compound	UT binding (pK _i) ^a			Functional activity (pK _B) ^a	
	Human	Cat	Rat	Rat aorta	Cat femoral artery
1a	8.4	8.0	8.5	7.7	7.2
11a	8.2	8.1	8.2	7.9	7.2

^a Mean of at least 3 determinations with standard deviation of $<\pm 0.3$.

affinity to cat and rat UT receptors,¹⁸ demonstrating broad cross-species activity (Table 6). In the rat isolated aorta,¹⁸ compounds **1a** and **11a** blocked hU-II induced contraction with respective pK_{BS} of 7.7 and 7.9, and **1a** was found to be a competitive and reversible antagonist with a pA₂ of 8.0, which was comparable to the rat receptor binding affinity. In the cat isolated femoral artery,¹⁹ **1a** and **11a** also blocked hU-II induced contraction with pK_{BS} of 7.2 and 7.2, respectively. The competitiveness and broad cross-species activity of **1a** distinguish it from early UT antagonists such as palosuran.^{9c} However, several early compounds in the series had poor cytochrome P450 (CYP450) and in vivo pharmacokinetic (PK) parameters. For example, compound **10b** had 2D6 (IC₅₀ 0.8 μM) and 3A4 (IC₅₀ 1.4 μM) liability, and high clearance and low oral bioavailability in rat PK studies (CL 97 mL/min/kg, F 0–3%, T_{1/2} 3.8 h, V_{dss} 23 L/kg, 1.2 mg/kg iv and 2.2 mg/kg po).

In summary, SAR exploration of a novel 2-aminomethyl piperidine series identified via HTS led to the discovery of competitive and reversible UT receptor antagonists such as **1a** and **11a** with excellent affinity and functional activity, and broad cross-species activity. The further optimization of this series to improve PK and CYP450 properties will be the subject of future publications.

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- The chiral centers in the structures shown in the paper are racemic except the ones in **1a**, **1b**, **11a** and **11b**.
- The radioligand binding assay was used as the primary assay to generate SAR.
- All new compounds in this paper were characterized via LC/MS and ¹H NMR.
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- Diamine **6d** was produced as an over reduction by-product in the hydrogenation step from conversion of **5a** to **6a**.

16. Compounds **7e** and **7f** were prepared from the corresponding commercially available amino acids in the same way as the preparation of **3a** from **4** in Scheme 1.
17. (a) Intermediate **6a** was separated via chiral HPLC into two enantiomers, which were converted to **1a**, **1b**, **11a**, and **11b**. Chiral separation conditions: Chiralpak AD (77 × 240 mm, 20 μM) eluting at 300 mL/min with 0.1% isopropylamine in MeOH gave two components at R_T 3.3 and 5.1 min; (b) Stereochemistry was assigned by Vibrational Circular Dichroism (VCD) analysis of both enantiomers.
18. For cat and rat UT receptor radioligand binding and rat isolated aorta contractile assay details, see Ref. **9e**.
19. For cat isolated femoral artery assay details, see: Behm, D. J.; Stankus, G.; Doe, C. P.; Willette, R. N.; Sarau, H. M.; Foley, J. J.; Schmidt, D. B.; Nuthulaganti, P.; Fornwald, J. A.; Ames, R. S.; Lambert, D. G.; Calo', G.; Camarda, V.; Aiyar, N. V.; Douglas, S. A. *Br. J. Pharmacol.* **2006**, *148*, 173.