

Organic Preparations and Procedures International: The New Journal for Organic Synthesis

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/uopp20>

SYNTHESIS OF N- (4-AMINOPHENYL) MONOAZACROWN ETHERS

Wei Zeng ^a, Ying Du ^a, Hongbo Li ^a, Xiaoxia Lu ^a & Shengying Qm ^a

^a Department of Chemistry, Sichuan University, Chengdu, 610064, P. R., CHINA E-mail:

Published online: 11 Feb 2009.

To cite this article: Wei Zeng, Ying Du, Hongbo Li, Xiaoxia Lu & Shengying Qm (2003) SYNTHESIS OF N- (4-AMINOPHENYL) MONOAZACROWN ETHERS, Organic Preparations and Procedures International: The New Journal for Organic Synthesis, 35:2, 228-231, DOI: [10.1080/00304940309355838](https://doi.org/10.1080/00304940309355838)

To link to this article: <http://dx.doi.org/10.1080/00304940309355838>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

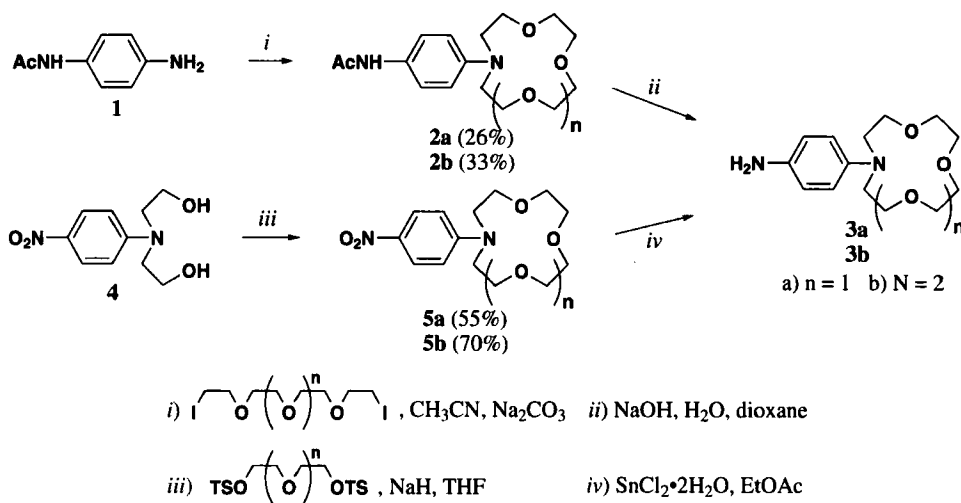
SYNTHESIS OF *N*-(4-AMINOPHENYL) MONOAZACROWN ETHERS

Submitted by
(05/07/02)

Wei Zeng, Ying Du, Hongbo Li, Xiaoxia Lu and Shengying Qin*

Department of Chemistry, Sichuan University
Chengdu, 610064, P. R. CHINA
e-mail: qin-shengying @ 263.net

Crown ethers containing an amino group promise to be advantageous intermediates for the synthesis of chromogenic crown ethers, liquid crystal crown ethers and molecule acceptors.¹⁻⁴ Whereas monoaza-18-crown-6, monoaza-15-crown-5, monoaza-12-crown-4, and their variously *N*-substituted derivatives have been known for many years,⁵⁻⁸ new methods for their preparation continue to appear.^{9,10} The preparation of *N*-(4-aminophenyl)monoazacrown ether **3b** has only been reported once *via* a three-step reaction from *N*-phenyldiethanolamine by cyclization, nitrosation, and reduction in an 9% overall yield.¹ We required a convenient synthesis of **3a** and **3b** for preparation of chromogenic monoazacrown ethers bearing picrylamino-type side arms; this paper herein describes two routes for the synthesis of *N*-(4-aminophenyl) monoaza-12-crown-4 (**3a**) and *N*-(4-aminophenyl)monoaza-15-crown-5 (**3b**) from readily available starting materials.



Compounds **3a** and **3b** were obtained from 4-aminoacetanilide in 21% and 28% overall yields, respectively. Arylamines **3a** and **3b** were also prepared in 49% and 60% yield, respectively, by aromatic nucleophilic substitution between 4-nitrochlorobenzene and diethanolamine. The second route exhibits many advantages such as higher overall yield, more convenient separation and purification compared to route 1.

EXPERIMENTAL SECTION

Mps were determined on a Yanaco MP-500 micro-melting point apparatus and are uncorrected. IR spectra were obtained on a Nicolet-1705X IR spectrometer. ^1H NMR spectra were recorded on a Bruker AC-200MHz in CDCl_3 with tetramethylsilane, as the internal standard. Mass spectra were obtained on a Finnigan MAT 4510 spectrometer. Elemental analyses were performed on a Carlo Erba-1160 elemental analyzer. Silica gel (60H for TLC, Qingdao, China) was used for flash column chromatography. 4-Aminoacetanilide (**1**),¹¹ 1,11-diiodo-3,6,9-trioxaundecane and 1,14-diiodo-3,6,9,12-tetraoxatetradecane,⁸ 2,2'-[(4-nitrophenyl)imino]bisethanol (**4**),¹² diethyleneglycol and triethyleneglycol ditosylates¹³ were synthesized according to published procedures. THF and acetonitrile were distilled and dried prior to use; all other reagents were of analytical grade and were used without further purification.

N-(4-Acetoamidophenyl)monoaza-12-crown-4 (2a).— To a stirred solution of anhydrous acetonitrile (500 mL), 4-aminoacetanilide (**1**) (4.02 g, 26.8 mmol), 1,11-diiodo-3,6,9-trioxaundecane (11.93 g, 27.0 mmol), and anhydrous sodium carbonate (16.8 g, 158.5 mmol) were added in one portion under an atmosphere of nitrogen. The mixture was stirred at reflux for 48 h; after cooling to 25° , the solution was filtered, concentrated to dryness *in vacuo* to give a residue, which was extracted with ethyl acetate (2 x 20 mL). The combined extract was dried (MgSO_4) and removal of the solvent gave a crude product, which was purified by column chromatography (silica gel, EtOAc) to afford 2.06 g (26%) of **2a**, as white crystal, mp. $67\text{--}69^\circ$. ^1H NMR: δ 7.77 (s, 1H, NH), 7.28–6.78 (5H, m, Ar-H), 3.80–3.65 (12H, m, 3 x CH_2OCH_2), 3.55 (4H, 2 x NCH_2), 2.09 (s, 3H, CH_3); IR (KBr, film): 3290, 2901, 2880, 1650, 1598, 1525, 1126 cm^{-1} ; MS (m/z): 308(M^+).

Anal. Calcd for $\text{C}_{16}\text{H}_{24}\text{N}_2\text{O}_4$: C, 62.33; H, 7.79; N, 9.09. Found: C, 62.43; H, 7.88; N, 9.25

N-(4-Acetoamidophenyl)monoaza-15-crown-5 (2b), mp. $84\text{--}86^\circ$, was prepared (33%) using a method similar to that for **2a**. ^1H NMR: δ 7.77 (s, 1H, NH), 7.30–6.80 (5H, m, Ar-H), 3.81–3.65 (16H, m, 4 x CH_2OCH_2), 3.55 (4H, 2 x NCH_2), 2.09 (s, 3H, CH_3); IR (KBr, film): 3300, 2900, 2880, 1650, 1598, 1525, 1126 cm^{-1} ; MS (m/z): 352(M^+).

Anal. Calcd for $\text{C}_{18}\text{H}_{28}\text{N}_2\text{O}_5$: C, 61.36; H, 7.95; N, 7.95. Found: C, 61.58; H, 7.72; N, 8.03

N-(4-Aminophenyl)monoaza-12-crown-4 (3a).— To a solution of **2a** (2.0 g, 6.5 mmol) in dioxane (40 mL), was added aqueous sodium hydroxide (40 mL, 20%). The mixture was stirred at reflux for 4 h. After cooling to 25° , most of the solvent was distilled off and the resulting residue was extracted with hot CHCl_3 (2 x 20 mL). The combined extract was concentrated to dryness to give a crude product, which was purified by column chromatography (silica gel, MeOH) to afford 1.4 g (81%) of **3a**, as a slightly reddish oil. ^1H NMR: δ 6.62–6.53 (4H, m, Ar-H), 5.28 (2H, s, NH_2), 3.80–3.65 (12H, m, 3 x CH_2OCH_2), 3.60 (4H, m, 2 x NCH_2); IR (neat): 3300, 1601, 1158 cm^{-1} ; MS (m/z): 266(M^+).

Anal. Calcd for $\text{C}_{14}\text{H}_{22}\text{N}_2\text{O}_3$: C, 63.16; H, 8.27; N, 10.53. Found: C, 62.94; H, 8.38; N, 10.69

***N*-(4-Aminophenyl)monoaza-15-crown-5 (3b)**, mp. 44–45°, *lit.*¹⁴ mp. 46°, was prepared (86%) using a method similar to that for **3a**. ¹H NMR: δ 6.62–6.53 (4H, m, Ar-H), 5.30 (2H, s, NH₂), 3.76–3.66 (16H, m, 4 x CH₂OCH₂), 3.61 (4H, m, 2 x NCH₂); MS (m/z): 310(M⁺).

***N*-(4-Nitrophenyl)monoaza-12-crown-4 (5a)**.- A 1-L three-necked flask was purged with N₂ and NaH (1.86 g, 77.5 mmol) was added to the reaction vessel and washed with hexanes (4 x 50 mL). THF (300 mL) was then added to flask. This suspension was heated to reflux with vigorous stirring. A solution of **4** (16.95 g, 75 mmol) and diethylene glycol ditosylates (31.05 g, 75 mmol) in THF (300 mL) was added dropwise fine. Reflux was continued for 20 h. The mixture was cooled to 0° and carefully quenched with H₂O, and the solvent was evaporated *in vacuo*. The residue was dissolved in H₂O (400 mL) and extracted with CH₂Cl₂ (3 x 200 mL). The combined organic layers were concentrated to dryness to give a crude product, which was purified by column chromatography (silica gel, EtOAc) to afford 12.3 g (55%) of **5a**, as yellow crystal, mp. 114–116°. ¹H NMR: δ 6.73–6.70 (4H, m, Ar-H), 3.80–3.65 (12H, m, 3 x CH₂OCH₂), 3.60 (4H, m, 2 x NCH₂); IR (KBr, film): 1600, 1340, 1126 cm⁻¹; MS (m/z): 296(M⁺).

Anal. Calcd for C₁₄H₂₀N₂O₅: C, 56.76; H, 6.76; N, 9.46. Found: C, 56.48; H, 6.85; N, 9.25

***N*-(4-Nitrophenyl)monoaza-15-crown-5 (5b)**, mp. 127–130°, *lit.*¹⁴ mp. 127–130°, was prepared (70%) from triethyleneglycol ditosylates using a method similar to that for **5a**.

***N*-(4-Aminophenyl)monoaza-12-crown-4 (3a)**.- A mixture of **5a** (2.79 g, 9 mmol), SnCl₂·2HO (2.5 g, 11.06 mmol), EtOAc (10 mL), and conc. HCl (2.95 mL) was stirred at 40° for 40 min; subsequently, H₂O (20 mL) was added to the mixture and stirred at 40° for 1 h. The pH of the solution was adjusted to 8–9 with 40% NaOH, filtered and the filtrate was extracted with CH₂Cl₂ (3 x 30 mL), dried (MgSO₄), and evaporated *in vacuo* to give a crude product, which was purified by column chromatography (silica gel, MeOH) to give 2.13 g (89%) of **3a**, as a slightly reddish oil. ¹H NMR: δ 6.64–6.55 (4H, m, Ar-H), 5.30 (2H, s, NH₂), 3.84–3.65 (12H, m, 3 x CH₂OCH₂), 3.58 (4H, m, 2 x NCH₂); IR (neat): 3300, 1603, 1155 cm⁻¹; MS (m/z): 266(M⁺).

Anal. Calcd for C₁₄H₂₂N₂O₃: C, 63.16; H, 8.27; N, 10.53. Found: C, 63.25; H, 8.11; N, 10.61

***N*-(4-Aminophenyl)monoaza-15-crown-5 (3b)**, mp. 44–46°, *lit.*¹⁴ mp. 46°, was prepared (85%) using a method similar to that for **3a**. ¹H NMR: δ 6.63–6.55 (4H, m, Ar-H), 5.30 (2H, s, NH₂), 3.74–3.63 (16H, m, 4 x CH₂OCH₂), 3.59 (4H, m, 2 x NCH₂); MS (m/z): 310(M⁺).

REFERENCE

1. J. P. Dix and F. Vögtle, *Chem. Ber.*, **113**, 457 (1980).
2. S. Y. Qin, Z. F. Lu, J. Z. Li and X. M. Liu, *Youji Huaxue.*, **15**, 424 (1995); *Chem. Abst.*, **123**, 340081x (1997).
3. M. G. Xie and Q. Jiang, *Liq. Cryst.*, **21**, 461 (1996).
4. R. A. Bartsch and M. D. Eley, *Tetrahedron.*, **52**, 8979 (1996).

5. R. N. Greane, *Tetrahedron Lett.*, 1793 (1972).
6. G. W. Gokel and B. J. Garcia, *Tetrahedron Lett.*, 317 (1977).
7. M. R. Johnson, I. O. Sutherland and F. F. Newton, *J. Chem. Soc. Perkin Trans I.*, 357 (1979).
8. M. J. Calrerley and J. Dale, *Acta Chem. Scand.*, **B36**, 241 (1982).
9. H. Maeda, Y. Nakatsuji and M. Okahara, *Chem. Commun.*, 471 (1981).
10. R. A. Schule, D. M. Dishong and G. W. Gokel, *Tetrahedron Lett.*, **22**, 2623 (1981).
11. C. M. Atkinson, C. W. Brown, J. McIntyre and J. C. E. Simpson, *J. Chem. Soc.*, 2023 (1954).
12. M. Bil and S. Brunner, W. H. Fr. 1506945, 1967; *Chem. Abst.*, **70**, 116216z (1969).
13. J. Dale and P. O. Kristiansew, *Acta Chem. Scand.*, **26**, 1471 (1972).
14. J. P. Dix and F. Vögtle, *Angew. Chem.*, **90**, 893 (1978).

SYNTHESIS OF α -HYDROXYTAMOXIFEN AND ITS 4-HYDROXY ANALOG

Submitted by M. R. Lashley,[†] C. W. Dicus,[†] K. Brown,^{††,‡} M. H. Nantz ^{*,†}
(07/16/02)

[†] *Department of Chemistry, University of California,
Davis CA 95616, USA*

^{††} *Biology and Biotechnology Research Program,
Lawrence Livermore National Laboratory,
Livermore, CA 94551, USA
e-mail: mhnantz@ucdavis.edu*

Tamoxifen is an anti-estrogen prescribed for the treatment of estrogen receptor-positive (ER+) breast cancer¹ and approved in the US for use as a chemopreventive agent for women who have an increased risk of developing cancer.² Although tamoxifen is a widely used adjuvant drug therapy, it is known to cause human endometrial cancer³ as well as liver cancer in rats.⁴ These observations have prompted many efforts to determine whether tamoxifen-induced endometrial carcinogenesis involves a genotoxic or hormonal mechanism.⁵ Recent studies on tamoxifen-