Iodine-Mediated Synthesis of 2-(Methylthio)-4H-chromen-4-ones and Study of Their Halogenation Reactions

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ABSTRACT: An efficient iodine-mediated method is developed for the synthesis of functionalized 2-(methylthio)-4H-chromen-4ones by intramolecular cyclization of easily accessible 1-(2-benzyloxy-aryl)-3,3-bis-methylsulfanyl-propenones. The synthesized chromen-4-ones turn out to be a key precursor for various kinds of chemical reactions. Mechanistically, we observed that iodinemediated intramolecular cyclization of ketene dithioacetal proceeded through a radical pathway. 3-Halo-2-(methylthio)-4Hchromen-4-ones were achieved via various two- or one-pot halogenation approaches.

INTRODUCTION

Chromone skeleton is present in many naturally occurring compounds,¹ and its derivatives exhibit a broad range of important biological activities such as anti-inflammatory, antiallergic, antitumor, anticancer, and antimicrobial.² Chromones are privileged structures and valuable intermediates in the synthesis of new heterocyclic compounds.^{2c,3} The most well-known synthetic routes enclose Claisen condensation,⁴ Vilsmeier-Haack reaction,⁵ Baker-Venkataraman rearrangement,⁶ Simonis and Ruhemann reactions,^{3,7} catalytic reactions mediated by palladium,⁸ and the oxidative cyclization of 2'hydroxychalcones,⁹ 2'-allyloxychalcones,¹⁰ or 2'-benzyloxy-6'-hydroxychalcone¹¹ using a catalytic amount of iodine in DMSO. Yang et al. reported a green approach for the synthesis of 3-thiocyanated chromones from o-hydroxyaryl enaminones under electrochemical thiocyanation conditions.¹² In recent years, C-H activation and annulation methodologies based on salicylaldehyde derivatives have been developed for the synthesis of chromones employing substrates such as alkynes, alkynoic acids, diazo compounds, and sulfoxonium ylides.¹³ Larock and co-workers described the successful use of iodine monochloride for the synthesis of chromones, benzo[b] furans, and α -pyrones via electrophilic cyclization reaction of oalkoxyalkynones, o-alkoxyalkynes, and alkynyl ester, respectively.

On the other hand, α -aroyl- α -bromoketene dithioacetals have been synthesized via direct bromination of the corresponding α -aroylketene dithioacetals using NBS in CCl₄.¹⁵ Besides, a study for the synthesis of α -bromo- α , β unsaturated ketones from α , β -unsaturated ketones (including 3-bromo-chromone from chromone) has been accomplished using a mixture of DMP and $Et_4N^+Br^-$ or $NBS-Et_3N\cdot 3HBr$ and K_2CO_3 in dry DCM.^{16,17} However, there is no particular study established for the halogenation of chromone derivatives.

So far, very limited literature studies for the synthesis of 2/3-(methylthio)-4H-chromen-4-ones are reported. In 2015, sulfenylation of chromones was first reported by Zheng and co-workers via C-H functionalization.¹⁸ Zhou and co-workers reported another sulfenylation method by using alkyl halides and Na₂S₂O₃.¹⁹ These methods provide 3-alkylthio- or 3arylthio-chromones, while no substitution occurs at the C-2 position even if the reaction site C-3 is blocked. Lee and Pak have reported the one-pot synthesis of 2-(methylthio)-4Hchromen-4-ones from 2-hydroxyacetophenones through a ketene dithioacetal intermediate.²⁰ Although this method provides direct preparation of 2-(methylthio)-4H-chromen-4ones, it requires very low reaction temperature $(-78 \ ^{\circ}C)$ and the use of strong and moisture sensitive base lithium bis(trimethylsilyl)amide (LiHMDS), and the generality of protocol was not tested. We have tried to obtain the desired product under the same reaction conditions, but we got low yield. Apart from this, there are no other specific methods

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known for the synthesis of 2-(methylthio)-4H-chromen-4-ones.

Functionalized ketene dithioacetals are versatile intermediates in the synthesis of various classes of heterocyclic compounds such as furans, thiophenes, pyrroles, pyridines, and lactones.²¹ Herein, we reported a synthetic route to 2-(methylthio)-4*H*-chromen-4-ones via iodine-mediated cyclization of α -aroylketene dithioacetals. In addition, a survey for the halogenation of chromones was accomplished.

RESULT AND DISCUSSION

The required precursors 1a-k were synthesized by reaction of functionalized acetophenones, carbon disulfide, and methyl iodide under basic conditions at low temperature (Scheme 1).²² This method failed to provide precursor 11 using 1,2-

Scheme 1. Synthesis of 1-(2-Benzyloxy-aryl)-3,3-bismethylsulfanyl-propenones $1a-k^a$ and 1-(2-

(Benzyloxy)phenyl)-2-(1,3-dithiolan-2-ylidene)ethan-1-one 11^b



^aPrecursors **1a–k** were synthesized by stirring functionalized acetophenones (2.0 mmol), carbon disulfide (2.40 mmol), methyl iodide (4.40 mmol), and NaH (8 mmol) in dry THF (10.0 mL) at 0 $^{\circ}$ C for 5–6 h. ^bPrecursor **11** was synthesized by stirring 1-(2-(benzyloxy)phenyl)ethan-1-one (2.0 mmol), carbon disulfide (2.40 mmol), 1,2-dibromoethane (4.40 mmol), and potassium *tert*-butoxide (8.0 mmol) in dry THF (8.0 mL) at 0 $^{\circ}$ C for 6 h.

dibromoethane in lieu of methyl iodide under the same reaction conditions. Therefore, we have tried another procedure,²³ and we successfully achieved the desired precursor 11 in 52% yield (Scheme 1).

To find optimal reaction conditions for intramolecular oxidative cyclization, 1-(2-(benzyloxy)phenyl)-3,3-bis-(methylthio)prop-2-en-1-one 1a was selected as a model substrate. The results are presented in Table 1. Treatment of 1a with CuI or Cu(OAc)₂ as a catalyst in DMSO provided no desired product (entries 1 and 2). The use of (diacetoxyiodo)-

Table 1. Exploration and Optimization Studies for the Synthesis of 2-(Methylthio)-4H-chromen-4-one^a

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^{*a*}Unless noted otherwise, reactions were performed on a 0.30 mmol scale using **1a** as a starting material at 80 °C. ^{*b*}Isolated yield. ^{*c*}Complex mixture formation. ^{*d*}The reaction was run at room temperature. ^{*c*}Decomposition of the starting material. ^{*f*}The reaction was run at 60 °C.

benzene as an oxidizing agent in DMSO or DCE did not produce the compound 2a (entries 3 and 4). It was also found that using tert-butyl hydroperoxide along with CuI in DMSO or CH₃CN gives a complex mixture and the use of DDQ in CH_3CN provided no desired product (entries 5–7). The use of trifluoroacetic acid in DCE at room temperature provides the desired chromone in 73% yield (entry 8), while increasing the temperature up to 80 °C does not provide better yield. The treatment of 1a with DMSO-I2 reagent commonly used for the oxidative cyclization of 2'-hydroxychalcones to flavones results in decomposition of the starting material (entry 9). Furthermore, the use of iodine catalyst in benzene and THF provides no product and only a trace amount of 2a, respectively (entries 10 and 11). When iodine catalyst was used in DCE solvent at 80 °C, the reaction proceeded well and we successfully achieved the desired chromone in 95% yield within 3 h (entry 12). Lowering the temperature from 80 to 60 °C dropped the yield to 81% (entry 13).

To study the effect of a protecting group at the ortho hydroxyl group of α -aroylketene dithioacetal, other protecting groups such as allyl, methoxymethyl, and methyl were used instead of the benzyl group (Scheme 2). When the ortho hydroxyl group was protected with allyl, methoxymethyl, and methyl group, 74%, 70%, and 0% yields of 2-(methylthio)-4*H*-chromen-4-one **2a** were obtained, respectively. Therefore, further synthesis was carried out with 1-(2-benzyloxy-aryl)-3,3-bis-methylsulfanyl-propenones as precursor following the optimal condition (entry 12).

Initially, we examined the scope of the oxidative cyclization reaction with respect to various 1-(2-(benzyloxy)aryl)-3,3-bis(methylthio)prop-2-en-1-ones. The reaction proceeded well and the desired chromones were achieved in good to excellent yields as illustrated in Scheme 3. The structure of **2g** was confirmed using single-crystal X-ray diffraction (see Figure S1, Supporting Information (SI)). It was noticed that 1-(2/1-benzyloxy-naphthalen-1/2-yl)-3,3-bis-methylsulfanyl-prope-

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Scheme 2. Examination of Various Protecting Groups for the Synthesis of 2-(Methylthio)-4H-chromen-4-one⁴



"All reactions were performed by stirring 1 (0.50 mmol) and I_2 (0.25 mmol) in DCE (1 mL) at 80 °C for 3 h.

Scheme 3. Synthesis of 2-(Methylthio)-4H-chromen-4-ones from 1-(2-Benzyloxy-aryl)-3,3-bis-methylsulfanyl-propenones^a



^{*a*}All reactions were performed by stirring 1 (1.0 mmol) and I_2 (0.50 mmol) in DCE (2 mL) at 80 °C for 3 h. ^{*b*}The procedure was scaled up to a gram scale.

nones 1g and 1h provided relatively lower yields of the desired products as compared to 1-(2-benzyloxyphenyl)-3,3-bismethylsulfanyl-propenones 1a-f and 1i, probably due to involvement of an additional electronic factor. Then, we have developed the practical synthesis of some chromone derivatives via conducting scale-up experiments. Thusly, chromone products 2a and 2f were obtained in yields of 89% (0.52 g) and 86% (1.06 g), respectively.

Next, we have synthesized 3-chloro/bromo-chromones in high yields using NCS/NBS at 80 °C. A trial using NIS also provides the corresponding iodinated product in a competitive yield. Since NIS is quite an expensive reagent as compared to NCS and NBS, a new method for the synthesis of 3-iodo-chromones was established by heating 2a-h with DDQ and iodine in DCE for 2 h at 80 °C (Scheme 4).

Further, we investigated the halogenation under a one-pot approach from 1-(2-benzyloxy-aryl)-3,3-bis-methylsulfanylpropenones. After monitoring the complete formation of 2-(methylthio)-4H-chromen-4-ones, NBS was added to the reaction mixture under the same reaction condition, and surprisingly, major 3-bromo-chromone along with minor 3iodo-chromone was observed. An attempt to quench the excess of iodine prior to addition of NBS using sodium thiosulfate was unsuccessful. Moreover, addition of NCS under similar reaction conditions provides 3-iodo-chromone along with a trace of 3-chloro-chromone. However, 3-iodo-chromones were selectively afforded by addition of iodine, followed by dropwise addition of a solution of NCS in DCE (Scheme 5).

In order to study the effect of iodine during one-pot halogenation, we independently carried out the reaction of 6methyl-2-(methylthio)-4*H*-chromen-4-one **2c** with 1 equiv of iodine and 0.5 equiv of NCS/NBS at room temperature and afforded 3-iodo-6-methyl-2-(methylthio)-4*H*-chromen-4-one **3i** selectively (Scheme 6).

Scheme 4. Scope of Synthesis of 3-Halo-2-(methylthio)-4H-chromen-4-ones^{a,b}



^aBromination and chlorination reactions were performed by stirring 2 (0.30 mmol) and (NBS or NCS) (0.36 mmol) in DCE (1 mL) at 80 °C for 30 min. ^bIodination reaction was performed by stirring 2 (0.30 mmol), DDQ (0.36 mmol), and I₂ (0.36 mmol) in DCE (1 mL) at 80 °C for 2 h.

Scheme 5. Scope of One-Pot Synthesis of 3-Iodo-2-(methylthio)-4H-chromen-4-ones^a



^{*a*}The reaction was performed by stirring 1 (0.30 mmol) and I₂ (0.15 mmol) in DCE (1 mL) at 80 $^{\circ}$ C for 3 h, and subsequent treatment with I₂ (0.15 mmol) and a solution of NCS (0.22 mmol) in DCE (0.5 mL) at 80 $^{\circ}$ C for 30 min.

Scheme 6. Iodination Using NCS/NBS and Iodine at Room Temperature^{*a*}(Synthesis of 3-Iodo-6-methyl-2-(methylthio)-4*H*-chromen-4-one)



^aIodination reaction was performed by stirring 2c (0.30 mmol), I_2 (0.30 mmol), and NXS (0.15 mmol) in DCE (1 mL) at room temperature for 1 h.

Scheme 7. Control Experiments

The mechanism of the oxidative cyclization was investigated and established by the control experiments (Scheme 7, eqs 1 and 2). Initially, we performed the cyclization reaction in the presence of a radical scavenger, (2,2,6,6-tetramethylpiperidin-1-yl)oxyl (TEMPO) or butylated hydroxytoluene (BHT), under standard conditions. The reaction was completely inhibited in the presence of TEMPO, whereas the product yield was significantly lowered in the presence of BHT (Scheme 7, eqs 1 and 2). In addition, we have identified the BHT-trapped chromone compound 4a, indicating that a radical pathway might be involved in this transformation. Therefore, the free radical mechanism starts with abstraction of one electron from the ethereal oxygen lone pair of compound 1 to form intermediate I, which undergoes cyclization to generate radical cation II, followed by deprotection of the benzyl group to provide radical intermediate III and benzyl iodide. The removal of the SMe radical from intermediate III results in the formation of chromone 2, and the SMe radical reacts with benzyl iodide to form benzyl(methyl)sulfane (Scheme 8).

To probe the mechanism, starting material 1l was used and we have isolated 2-(1,3-dithiolan-2-ylidene)-1-(2-hydroxyphenyl)ethan-1-one 5 in 79% yield within 2 h. Although, when the reaction was carried out for a longer time (12 h), 2-((2-(benzylthio)ethyl)thio)-4H-chromen-4-one 2l was isolated in 19% yield. This result confirms that the SMe group was eliminated as benzyl(methyl)sulfane.

The mechanism of the halogenation reaction was established by the control experiments (Scheme 7, eqs 3–5). Radical scavenger experiment of 2a with TEMPO or BHT using NXS in DCE at 80 °C provide the desired products 3a-b in high yields (Scheme 7, eq 3), indicating that radical species is not involved and the reaction proceeds via an electrophilic substitution pathway. The halogenation reactions of compound 2f using DDQ/I₂ at 80 °C or NXS/I₂ at room



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Scheme 8. Proposed Reaction Mechanism for Intramolecular Cyclization of 1-(2-Benzyloxy-aryl)-3,3-bis-methylsulfanylpropenones Using Iodine in DCE



Scheme 9. Proposed Reaction Mechanism for Iodination of 2-(Methylthio)-4H-chromen-4-ones Using I₂ and DDQ



Scheme 10. Synthetic Utility of 2-(Methylthio)-4H-chromen-4-ones



temperature were totally inhibited in the presence of TEMPO or BHT. We have also identified the BHT-trapped chromone compound in all cases (Scheme 7, eqs 4 and 5), indicating that a radical pathway might be involved in this reaction.

A plausible reaction mechanism for the iodination of 2-(methylthio)-4*H*-chromen-4-ones with DDQ and I_2 is shown in (Scheme 9). A single electron transfer from 2 to DDQ results in the formation of radical cation I, which reacts with iodine to generate the cation intermediate II, followed by proton transfer to provide product 3. A similar reaction mechanism is proposed for iodination using NXS and I_2 (see Scheme S1, SI).

To demonstrate the utility of 2-(methylthio)-4*H*-chromen-4-ones, various kinds of chemical reactions were conducted as shown in Scheme 10. Synthesis of halogenated chromones 3a - x is considered to be an example of an electrophilic substitution reaction at the C-3 position of chromones. The Liebeskind–Srogl coupling reaction was carried out using the previously reported literature²⁴ to provide 4'-ethylflavone 6 in 92% yield. Nucleophilic substitution reaction at the C-2 position was performed using primary and secondary amines to

provide the corresponding aminated chromones in very good yields. Deprotection of the benzyl group from compound **8b** and **2b** was carried out in the presence of an excess amount of TFA to give 3-chloro-7-hydroxy-2-(phenylamino)-4*H*-chromen-4-one **9** and 7-hydroxy-2-(methylthio)-4*H*-chromen-4-one **10** in 93% and 95% yields, respectively. On the basis of a previously reported procedure,²⁵ we have achieved reductive desulfuration of 2-(methylthio)-4*H*-chromen-4-one **2a** and 3-methyl-2-(methylthio)-4*H*-chromen-4-one **2i** in very good yields through refluxing with freshly prepared Raney nickel in ethanol for 12 h.

CONCLUSION

In summary, we have developed an efficient method for the synthesis of 2-(methylthio)-4H-chromen-4-ones via oxidative cyclization of easily accessible precursors 1-(2-benzyloxy-aryl)-3,3-bis-methylsulfanyl-propenones under mild conditions in high yields. Bromination and chlorination of 4H-chromen-4ones at the C-3 position proceeded through an electrophilic substitution reaction. Iodination was achieved through a free radical pathway. One- and two-pot approaches were developed to access iodinated chromones using I_2/NCS and I_2/DDQ , respectively, at 80 °C. Iodination using I2/NXS at room temperature was the most interesting finding, where no chlorination or bromination occurs as side reaction and selective iodination was achieved. Various kinds of chemical reactions were conducted to demonstrate the synthetic potential of 2-(methylthio)-4H-chromen-4-ones. During synthetic applicability, we have achieved 3-methyl-4H-chromen-4one (CAS No. 85-90-5) in good yield by simple reductive desulfuration, which is an expensive reagent.

EXPERIMENTAL SECTION

General Information. Commercially available reagent and solvent purchased by Sigma-Aldrich and Alfa Aesar were used directly without further purification. Reactions were monitored by analytical thin-layer chromatography (TLC). All reactions were conducted using dried glassware. IR spectra were recorded on an IR spectrophotometer, and stretching frequencies were reported in wavenumber (cm⁻¹). The ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded in CDCl₃ solution using chloroform peak (7.24 ppm for ¹H and 77.0 ppm for ¹³C) as reference. Only spectra of compounds 9 and 10 were recorded in DMSO-d₆ solution using the DMSO peak (2.50 ppm for ¹H and 39.52 ppm for ¹³C) as reference. Compounds $2a_{,}^{26}6_{,}^{27}7a_{,}b_{,}^{28}$ and $11a_{,}b^{29}$ are known; the remaining compounds are newly synthesized. The coupling constants J are reported in Hz, and signal patterns are reported as s, singlet; br, broad; d, doublet; t, triplet; q, quartet; m, multiplet; dd, doublet doublets. High-resolution mass spectra were recorded on a quadrupole-time-of-flight (Q-TOF) mass spectrometer equipped with an electrospray ion source (ESI-TOF).

General Procedure for the Synthesis of 1-(2-Benzyloxyaryl)-3,3-bis-methylsulfanyl-propenones (1a–k). To a stirred suspension of NaH (8.0 mmol) (previously washed with hexane in order to remove mineral oil) in THF (8.0 mL) was added a solution of functionalized acetophenones (2.0 mmol) in THF (2.0 mL) dropwise within 15 min at 0 °C. The reaction mixture was stirred for 20 min at 0 °C; then carbon disulfide (2.40 mmol) was added dropwise within 15 min at 0 °C. The reaction mixture was stirred for 30 min at 0 °C; then methyl iodide (4.40 mmol) was added dropwise during 15 min at 0 °C. The reaction mixture was stirred for 15 min at 0 °C, then warmed to room temperature, and stirred for 5–6 h. Completion of the reaction was monitored by TLC. After completion, THF was removed under vacuum, and ice-cold water (20.0 mL) was added to the mixture and stirred well. The formed solid was filtered

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and was washed with water (20.0 mL \times 2), dried, and recrystallized from hexane.

Synthetic Procedure for 1-(2-(Benzyloxy)phenyl)-2-(1,3dithiolan-2-ylidene)ethan-1-one 1l. To a stirred suspension of potassium tert-butoxide (898 mg, 8.0 mmol) in dry THF (3.0 mL) at 0 °C was dropwise added a dry THF solution (3.0 mL) of 1-(2-(benzyloxy)phenyl)ethan-1-one (453 mg, 2.0 mmol) and carbon disulfide (145 mg, 2.40 mmol). The reaction mixture was vigorously stirred at 0 °C for 90 min. To this suspension was added a dry THF solution (2 mL) of 1,2-dibromoethane (379 mg, 4.40 mmol) dropwise within 10 min at 0 °C. The reaction mixture was stirred for 6 h. After completion of the reaction, the mixture was poured into a saturated aqueous solution of ammonium chloride with crushed ice. Then, the aqueous solution was extracted with ethyl acetate (3×20) mL). The combined organic extracts were washed with brine (1×20) mL), dried, and then concentrated under vacuum to obtain the crude product. Pure product was obtained by silica gel column chromatography using 10% EtOAc/hexane.

General Procedure for the Synthesis of 2-(Methylthio)-4*H*-chromen-4-ones (2a–i). A 10 mL Schlenk tube equipped with a magnetic stirring bar was charged with 1-(2-benzyloxy-aryl)-3,3-bismethylsulfanyl-propenones 1a-i (1 mmol) and DCE (2 mL), followed by addition of iodine (0.50 mmol). The reaction mixture was stirred at 80 °C (oil bath) for 3 h. After completion, the reaction mixture was quenched with 3% Na₂S₂O₃ (5 mL), diluted with brine (10 mL), and extracted with DCM (15 mL × 3). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through silica gel column chromatography using 10% EtOAc/ hexane to afford the desired chromones.

Gram-Scale Procedure for the Synthesis of 2-(Methylthio)-4*H*-chromen-4-ones 2a. A 50 mL round-bottom flask equipped with a magnetic stirring bar was charged with 1-(2-(benzyloxy)phenyl)-3,3-bis(methylthio)prop-2-en-1-one 1a (1 g, 3.02 mmol) and DCE (6 mL), followed by addition of iodine (1.5 mmol). The reaction mixture was stirred at 80 °C (oil bath) for 4–5 h. After completion, the reaction mixture was quenched with 3% Na₂S₂O₃ (50 mL), diluted with brine (50 mL), and extracted with DCM (50 mL × 3). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through silica gel column chromatography using 10% EtOAc/hexane to afford the desired chromone 2a in 89% yield (0.52 g).

Gram-Scale Procedure for the Synthesis of 6-Methoxy-2-(methylthio)-4*H*-chromen-4-one 2f. A 50 mL round-bottom flask equipped with a magnetic stirring bar was charged with 1-(2-(benzyloxy)-5-methoxyphenyl)-3,3-bis(methylthio)prop-2-en-1-one If (2 g, 5.5 mmol) and DCE (10 mL), followed by addition of iodine (2.7 mmol). The reaction mixture was stirred at 80 °C (oil bath) for 4–5 h. After completion, the reaction mixture was quenched with 6% Na₂S₂O₃ (100 mL), diluted with brine (100 mL), and extracted with DCM (75 mL × 3). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through silica gel column chromatography using 10% EtOAc/hexane to afford the desired chromone 2f in 86% yield (1.06 g).

General Procedure for the Synthesis of 3-Chloro/bromo-2-(methylthio)-4*H*-chromen-4-ones (3a,b, 3d,e, 3g,h, 3j,k, 3m,n, 3p,q, 3s,t, and 3v,w). A solution of 2-(methylthio)-4*H*-chromen-4ones 2a-h (0.30 mmol) in DCE (1 mL) was treated with NBS or NCS (0.36 mmol) and then stirred at 80 °C (oil bath) for 30 min. Completion of the reaction was monitored by TLC. After completion, DCE was removed under vacuum, and the crude 3-chloro and 3bromo-chromones were purified by column chromatography using 5% EtOAc/hexane.

General Procedure for the Synthesis of 3-lodo-2-(methylthio)-4*H*-chromen-4-ones (3c, 3f, 3i, 3l, 3o, 3r, 3u, and 3x) Using I_2 /DDQ. A solution of 2-(methylthio)-4*H*-chromen-4-ones 2a-h (0.30 mmol) in DCE (1 mL) was treated with DDQ (0.36 mmol) and I_2 (0.36 mmol) in sequence and then stirred at 80 °C (oil bath) for 2 h. After completion, the reaction mixture was quenched with 5% Na₂S₂O₃ (5 mL), diluted with brine (10 mL), and extracted with DCM (15 mL × 3). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through silica gel

column chromatography using 5% EtOAc/hexane to afford 3-iodo-2-(methylthio)-4H-chromen-4-ones.

General Procedure for the One-Pot Synthesis of 3-lodo-2-(methylthio)-4H-chromen-4-ones (3c, 3f, 3i, 3l, 3u, and 3x). A solution of 1-(2-benzyloxy-aryl)-3,3-bis-methylsulfanyl-propenones 1 (0.30 mmol) in DCE (1 mL) was treated with iodine (0.15 mmol) and then stirred at 80 °C (oil bath) for 3 h. The complete formation of chromones 2 was monitored by TLC. Thereafter, I₂ (0.15 mmol) was added to the reaction mixture, followed by dropwise addition of a solution of NCS (0.22 mmol) in DCE (0.5 mL) at 80 °C (oil bath). The reaction mixture was stirred for 30 min. Upon completion, it was quenched with 3-5% Na₂S₂O₃ (5 mL), diluted with brine (10 mL), and extracted with DCM (15 mL × 3). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through silica gel column chromatography using 5% EtOAc/hexane to afford the desired products.

Synthetic Procedure for 3-lodo-6-methyl-2-(methylthio)-4H-chromen-4-one (3i) Using NXS and lodine at Room Temperature. A solution of 6-methyl-2-(methylthio)-4H-chromen-4-one 2c (62 mg, 0.30 mmol) in DCE (1 mL) was treated with I₂ (76 mg, 0.30 mmol), followed by addition of NCS or NBS (0.5 equiv, 0.15 mmol), and then stirred at room temperaturefor 1 h. After completion, the reaction mixture was quenched with 5% Na₂S₂O₃ (5 mL), diluted with brine (10 mL), and extracted with DCM (15 mL × 3). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through silica gel column chromatography using 5% EtOAc/hexane to afford compound 3i.

The presence of only the iodinated product was confirmed using a ¹³C NMR spectrum to prove the selectivity of this method.



General Procedure for the Synthesis of 2-sec-Amino-4Hchromen-4-ones (7a,b). A mixture of 2-(methylthio)-4H-chromen-4-one 2a (0.20 mmol) and pyrrolidine/piperidine (0.15 mL) was stirred at 60 °C (oil bath) for 2 h. Completion of the reaction was monitored by TLC. After completion, the excess of secondary amine was removed under vacuum and recrystallized from methanol to afford 2-sec-amino-4H-chromen-4-ones.

General Procedure for the Synthesis of 2-(Phenylamino)-4H-chromen-4-ones (8a,b). A mixture of 2-(methylthio)-4Hchromen-4-one/7-(benzyloxy)-3-chloro-2-(methylthio)-4H-chromen-4-one 2a/3d (0.20 mmol) and aniline (1 mL) was stirred at 90 °C (oil bath) for 24 h. Completion of the reaction was monitored by TLC. After completion, the excess of aniline was removed through silica gel column chromatography using DCM; then the desired product was collected using 20% MeOH/DCM.

General Procedure for the Synthesis of 7-Hydroxy-4H-chromen-4-ones (9 and 10). A mixture of 7-(benzyloxy)-2-(methylthio)-4H-chromen-4-one/7-(benzyloxy)-3-chloro-2-(phenylamino)-4H-chromen-4-one 2b/8b (0.20 mmol) and TFA (1 mL) was stirred at 90 °C (oil bath) for 6 h. Completion of the reaction was monitored by TLC. After completion, the excess of TFA was removed under vacuum and the obtained precipitate was washed with water, filtered, and dried to afford the desired products without further purification.

1-(2-(Benzyloxy)phenyl)-3,3-bis(methylthio)prop-2-en-1-one (1a). Yield: 97% (641 mg), yellow solid (recrystallized from hexane), mp: 126–128 °C; IR (KBr): 3030, 2918, 1585, 1476, 752 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.84 (d, 1H, J = 7.6 Hz), 7.42–7.29 (m, 6H), 7.04–6.99 (m, 2H), 6.91 (s, 1H), 5.06 (s, 2H), 2.44 (s, 3H), 1.88 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 185.4, 163.9, 156.7, 136.2, 132.5, 131.4, 129.5, 128.5, 128.2, 128.0, 121.1, 114.9, 112.5, 70.6, 16.3, 15.0; HRMS (ESI) m/z calcd for $C_{18}H_{19}O_2S_2$ [M + H]⁺ 331.0821, found 331.0825.

1-(2,4-Bis(benzyloxy)phenyl)-3,3-bis(methylthio)prop-2-en-1one (1b). Yield: 90% (812 mg), white solid (recrystallized from hexane), mp: 135–137 °C; IR (KBr): 3031, 2921, 1589, 1478, 757 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.94 (d, 1H, J = 8.8 Hz), 7.43–7.33 (m, 10H), 6.95 (s, 1H), 6.64 (d, 1H, J = 8.8 Hz), 6.62 (s, 1H), 5.08 (s, 2H), 5.01 (s, 2H), 2.43 (s, 3H), 1.78 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 184.1, 163.0, 162.6, 158.5, 136.3, 136.0, 133.5, 128.6, 128.5, 128.4, 128.1, 127.5, 122.6, 115.1, 106.4, 100.2, 70.8, 70.1, 16.2, 15.1; HRMS (ESI) *m*/*z* calcd for C₂₅H₂₅O₃S₂ [M + H]⁺ 437.1240, found 437.1237.

1-(2-(Benzyloxy)-5-methylphenyl)-3,3-bis(methylthio)prop-2-en-1-one (1c). Yield: 93% (640 mg), white solid (recrystallized from hexane), mp: 117–119 °C; IR (KBr): 3030, 1593, 1483, 762 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.65 (s, 1H), 7.42–7.31 (m, 5H), 7.19 (d, 1H, *J* = 7.6 Hz), 6.92–6.89 (m, 2H), 5.04 (s, 2H), 2.46 (s, 3H), 2.29 (s, 3H), 1.91 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 185.7, 163.8, 154.7, 136.5, 133.0, 131.8, 130.6, 129.3, 128.6, 128.2, 128.0, 115.0, 112.8, 70.9, 20.3, 16.4, 15.1; HRMS (ESI) *m/z* calcd for C₁₉H₂₁O₂S₂ [M + H]⁺ 345.0977, found 345.0968.

1-(2-(Benzyloxy)-5-chlorophenyl)-3,3-bis(methylthio)prop-2-en-1-one (1d). Yield: 95% (693 mg), yellow solid (recrystallized from hexane), mp: 112–113 °C; IR (KBr): 3032, 2918, 1606, 1485, 750 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (d, 1H, *J* = 2.4 Hz), 7.41–7.30 (m, 6H), 6.93 (d, 1H, *J* = 8.8 Hz), 6.84 (s, 1H), 5.04 (s, 2H), 2.45 (s, 3H), 1.88 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 183.7, 165.7, 155.2, 135.9, 131.9, 131.2, 130.9, 128.7, 128.4, 128.1, 126.5, 114.3, 114.2, 71.2, 16.4, 15.1; HRMS (ESI) *m/z* calcd for C₁₈H₁₈ClO₂S₂ [M + H]⁺ 365.0431, found 365.0435.

1-(2-(Benzyloxy)-5-bromophenyl)-3,3-bis(methylthio)prop-2-en-1-one (**1e**). Yield: 89% (729 mg), yellow solid (recrystallized from hexane), mp: 106–108 °C; IR (KBr): 3057, 2910, 1573, 1456, 746 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.94 (d, 1H, J = 2.0 Hz), 7.45 (dd, 1H, J = 8.4, 2.0 Hz), 7.40–7.31 (m, 5H), 6.88 (d, 1H, J = 8.4 Hz), 6.83 (s, 1H), 5.04 (s, 2H), 2.45 (s, 3H), 1.90 (s, 3H); $^{13}C{}^{1}H{}$ NMR (CDCl₃, 100 MHz): δ 183.6, 165.6, 155.7, 135.8, 134.8, 134.0, 131.3, 128.7, 128.4, 128.1, 114.7, 114.3, 113.8, 71.2, 16.5, 15.1; HRMS (ESI) m/z calcd for C₁₈H₁₈BrO₂S₂ [M + H]⁺ 408.9926, found 408.9920.

1-(2-(Benzyloxy)-5-methoxyphenyl)-3,3-bis(methylthio)prop-2en-1-one (**1f**). Yield: 90% (649 mg), yellow solid (recrystallized from hexane), mp: 96–98 °C; IR (KBr): 3032, 2914, 1597, 1479, 1278, 1037, 761 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.43–7.29 (m, 6H), 6.98 (s, 1H), 6.94 (s, 2H), 5.00 (s, 2H), 3.78 (s, 3H), 2.46 (s, 3H), 1.91 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 184.9, 164.4, 154.0, 151.2, 136.6, 130.3, 128.5, 128.2, 128.1, 119.3, 115.0, 114.9, 114.7, 71.9, 55.7, 16.5, 15.1; HRMS (ESI) *m*/*z* calcd for C₁₉H₂₁O₃S₂ [M + H]⁺ 361.0927, found 361.0933.

1-(2-(Benzyloxy)naphthalen-1-yl)-3,3-bis(methylthio)prop-2-en-1-one (**1g**). Yield: 96% (728 mg), white solid (recrystallized from hexane), mp: 113–115 °C; IR (KBr): 3028, 2922, 1623, 1469, 748 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.83 (d, 1H, *J* = 8.4 Hz), 7.72 (d, 1H, *J* = 8.4 Hz), 7.68 (d, 1H, *J* = 7.6 Hz), 7.38–7.17 (m, 8H), 6.35 (s, 1H), 5.13 (s, 2H), 2.47 (s, 3H), 2.15 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 189.0, 163.7, 152.8, 137.0, 131.5, 130.8, 129.2, 128.4, 127.8, 127.7, 127.1, 127.0, 126.8, 124.7, 124.1, 116.1, 115.1, 71.5, 16.9, 14.9; HRMS (ESI) *m*/*z* calcd for C₂₂H₂₁O₂S₂ [M + H]⁺ 381.0977, found 381.0987.

1-(1-(Benzyloxy)naphthalen-2-yl)-3,3-bis(methylthio)prop-2-en-1-one (1h). Yield: 89% (676 mg), brown viscous liquid (recrystallized from hexane); IR (KBr): 3060, 2922, 1613, 1476, 758 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.29 (m, 1H), 7.89–7.82 (m, 2H), 7.66 (d, 1H, *J* = 8.8 Hz), 7.55–7.48 (m, 4H), 7.42–7.33 (m, 3H), 7.13 (s, 1H), 4.96 (s, 2H), 2.53 (s, 3H), 2.11 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 186.0, 165.0, 154.0, 136.6, 136.1, 129.3, 128.4, 128.2, 128.1, 128.0, 127.8, 127.6, 126.5, 126.3, 124.2, 122.9, 114.0, 78.3, 16.6, 14.9; HRMS (ESI) *m*/*z* calcd for C₂₂H₂₁O₂S₂ [M + H]⁺ 381.0977, found 381.0962.

1-(2-(Benzyloxy)phenyl)-2-methyl-3,3-bis(methylthio)prop-2-en-1-one (1i). Yield: 51% (348 mg), yellow viscous liquid (recrystallized from hexane); IR (KBr): 3030, 2919, 1593, 1449, 752 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.75 (d, 1H, *J* = 7.6 Hz), 7.42–7.29 (m, 6H), 7.03–6.95 (m, 2H), 5.04 (s, 2H), 2.21 (s, 3H), 2.00 (s, 3H), 1.89 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 196.7, 157.7, 147.1, 136.0, 133.3, 132.3, 131.6, 128.7, 128.6, 128.2, 127.8, 120.8, 112.4, 70.7, 20.1, 16.9, 16.1; HRMS (ESI) *m*/*z* calcd for C₁₉H₂₁O₂S₂ [M + H]⁺ 345.0977, found 345.0972.

1-(2-(Allyloxy)phenyl)-3,3-bis(methylthio)prop-2-en-1-one (1j). Yield: 92% (516 mg), white solid (recrystallized from hexane), mp: 50–52 °C; IR (KBr): 3072, 2920, 1598, 1478, 760 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.73 (d, 1H, *J* = 8.0 Hz), 7.35 (t, 1H, *J* = 7.6 Hz), 6.98 (t, 1H, *J* = 7.6 Hz), 6.91 (s, 1H), 6.88 (d, 1H, *J* = 8.4 Hz), 6.08–5.98 (m, 1H), 5.39 (d, 1H, *J* = 17.2 Hz), 5.26 (d, 1H, *J* = 10.4 Hz), 4.55 (d, 2H, *J* = 5.2 Hz), 2.47 (s, 3H), 2.42 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 185.9, 163.5, 156.3, 132.8, 132.2, 131.1, 130.0, 121.0, 117.9, 114.9, 112.6, 69.2, 17.1, 15.0; HRMS (ESI) *m*/*z* calcd for $C_{14}H_{17}O_2S_2$ [M + H]⁺ 281.0664, found 281.0651.

1-(2-(Methoxymethoxy)phenyl)-3,3-bis(methylthio)prop-2-en-1one (1k). Yield: 86% (488 mg), white solid (recrystallized from hexane), mp: 55–57 °C; IR (KBr): 2921, 1597, 1479, 761 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.68 (d, 1H, J = 7.2 Hz), 7.35 (s, 1H), 7.13–7.02 (m, 2H), 6.82 (s, 1H), 5.20 (s, 2H), 3.46 (s, 3H), 2.49 (s, 3H), 2.46 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 186.1, 163.9, 154.9, 132.1, 130.9, 130.6, 122.1, 115.3, 114.5, 95.1, 56.3, 17.2, 15.0; HRMS (ESI) m/z calcd for C₁₃H₁₇O₃S₂ [M + H]⁺ 285.0614, found 285.0611.

1-(2-(Benzyloxy)phenyl)-2-(1,3-dithiolan-2-ylidene)ethanone (1)). Yield: 52% (342 mg), white solid, eluent (hexane/EtOAc = 9:1), mp: 101–102 °C; IR (KBr): 3065, 2924, 1595, 1483, 753 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.80 (dd, 1H, *J* = 7.8, 1.8 Hz), 7.49 (s, 1H), 7.44–7.29 (m, 6H), 7.03–6.96 (m, 2H), 5.15 (s, 2H), 3.42 (t, 2H, *J* = 6.2 Hz), 3.32 (t, 2H, *J* = 6.2 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 186.3, 165.5, 157.0, 136.4, 132.6, 131.2, 129.1, 128.6, 127.8, 127.1, 121.1, 113.8, 113.0, 70.6, 38.8, 35.3; HRMS (ESI) *m*/*z* calcd for C₁₈H₁₇O₂S₂ [M + H]⁺ 329.0664, found 329.0657.

2-(*Methylthio*)-4*H*-chromen-4-one (**2a**). Yield: 95% (182 mg), eluent (hexane/EtOAc = 9:1); ¹H NMR (CDCl₃, 400 MHz): δ 8.13 (d, 1H, *J* = 8.0 Hz), 7.60 (t, 1H, *J* = 7.8 Hz), 7.36 (t, 2H, *J* = 8.8 Hz), 6.19 (s, 1H), 2.52 (s, 3H); HRMS (ESI) *m*/*z* calcd for C₁₀H₉O₂S [M + H]⁺ 193.0318, found 193.0320.

7-(Benzyloxy)-2-(methylthio)-4H-chromen-4-one (**2b**). Yield: 96% (286 mg), white solid, eluent (hexane/EtOAc = 9:1), mp: 150–152 °C; IR (KBr): 2925, 1622, 1548, 1439, 1337, 1163, 757 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.03 (dd, 1H, *J* = 8.8 Hz), 7.42–7.32 (m, 5H), 6.97 (dd, 1H, *J* = 8.8, 1.6 Hz), 6.84 (d, 1H, *J* = 1.6 Hz), 6.10 (s, 1H), 5.10 (s, 2H), 2.47 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 175.2, 169.5, 162.9, 158.3, 135.6, 128.7, 128.3, 127.4, 127.1, 117.2, 114.7, 107.1, 101.0, 70.5, 13.8; HRMS (ESI) *m*/*z* calcd for C₁₇H₁₅O₃S [M + H]⁺ 299.0736, found 299.0747.

6-Methyl-2-(methylthio)-4H-chromen-4-one (**2c**). Yield: 91% (188 mg), white solid, eluent (hexane/EtOAc = 9:1), mp: 96–98 °C; IR (KBr): 2924, 1651, 1548, 1463, 1356, 1131, 753 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.92 (s, 1H), 7.40 (d, 1H, *J* = 8.4 Hz), 7.27 (d, 1H, *J* = 8.4 Hz), 6.17 (s, 1H), 2.51 (s, 3H), 2.41 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 175.8, 169.8, 155.0, 135.2, 134.5, 125.1, 123.0, 116.9, 107.1, 20.8, 13.7; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₁O₂S [M + H]⁺ 207.0474, found 207.0482.

6-*Chloro-2-(methylthio)-4H-chromen-4-one* (**2d**). Yield: 85% (193 mg), white solid, eluent (hexane/EtOAc = 9:1), mp: 126–128 °C; IR (KBr): 2920, 1647, 1541, 1429, 1323, 1143, 723 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.08 (d, 1H, *J* = 2.8 Hz), 7.53 (dd, 1H, *J* = 8.8, 2.8 Hz), 7.32 (d, 1H, *J* = 8.8 Hz), 6.16 (s, 1H), 2.51 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 174.3, 170.5, 155.0, 133.5, 131.2, 125.3, 124.5, 118.9, 107.2, 13.8; HRMS (ESI) *m*/*z* calcd for C₁₀H₈ClO₂S [M + H]⁺ 226.9928, found 226.9936.

6-Bromo-2-(methylthio)-4H-chromen-4-one (**2e**). Yield: 83% (225 mg), pale yellow solid, eluent (hexane/EtOAc = 9:1), mp: 113-115 °C; IR (KBr): 2926, 1656, 1544, 1460, 1327, 1141, 765

cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.29 (d, 1H, *J* = 2.0 Hz), 7.71 (dd, 1H, *J* = 8.8, 2.0 Hz), 7.30 (d, 1H, *J* = 8.8 Hz), 6.21 (s, 1H), 2.56 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 174.1, 170.4, 155.5,

136.2, 128.5, 124.9, 119.1, 118.6, 107.3, 13.8; HRMS (ESI) m/z calcd

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for C₁₀H₈BrO₂S [M + H]⁺ 270.9423, found 270.9425. 6-Methoxy-2-(methylthio)-4H-chromen-4-one (2f). Yield: 91% (202 mg), yellow solid, eluent (hexane/EtOAc = 9:1), mp: 100–101 °C; IR (KBr): 2943, 1612, 1544, 1427, 1338, 1134, 1244, 1070, 805 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.50 (d, 1H, *J* = 2.8 Hz), 7.29 (d, 1H, *J* = 9.2 Hz), 7.17 (dd, 1H, *J* = 9.2, 2.8 Hz), 6.16 (s, 1H), 3.86 (s, 3H), 2.51 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 175.4, 169.3, 156.8, 151.5, 124.0, 122.8, 118.5, 106.7, 105.2, 55.8, 13.7; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₁O₃S [M + H]⁺ 223.0423, found 223.0428.

3-(*Methylthio*)-1*H-benzo*[*f*]*chromen-1-one* (**2***g*). Yield: 79% (192 mg), white solid, eluent (hexane/EtOAc = 9:1), mp: 144–146 °C; IR (KBr): 2923, 1630, 1561, 1438, 1317, 1139, 955, 823, 755 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 9.98 (d, 1H, *J* = 8.8 Hz), 8.00 (d, 1H, *J* = 9.2 Hz), 7.84 (d, 1H, *J* = 7.6 Hz), 7.71 (t, 1H, *J* = 7.6 Hz), 7.57 (t, 1H, *J* = 7.6 Hz), 7.41 (d, 1H, *J* = 9.2 Hz), 6.32 (s, 1H), 2.53 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 177.8, 166.7, 157.9, 135.0, 130.6, 130.5, 129.1, 128.1, 127.1, 126.5, 116.8, 116.6, 110.5, 13.8; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₁O₂S [M + H]⁺ 243.0474, found 243.0482.

2-(*Methylthio*)-4*H*-benzo[*h*]chromen-4-one (**2h**). Yield: 63% (152 mg), yellow solid, eluent (hexane/EtOAc = 9:1), mp: 100–102 °C; IR (KBr): 2924, 1621, 1548, 1437, 1343, 1164, 914, 824, 757 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.31 (d, 1H, *J* = 7.2 Hz), 8.06 (d, 1H, *J* = 8.4 Hz), 7.85 (d, 1H, *J* = 7.2 Hz), 7.70–7.60 (m, 3H), 6.34 (s, 1H), 2.63 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 175.5, 168.6, 153.8, 135.5, 129.0, 127.9, 127.0, 125.1, 123.1, 121.7, 120.5, 119.4, 108.6, 13.8; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₁O₂S [M + H]⁺ 243.0474, found 243.0478.

3-Methyl-2-(methylthio)-4H-chromen-4-one (2i). Yield: 86% (178 mg), white solid, eluent (hexane/EtOAc = 9:1), mp: 100–102 °C; IR (KBr): 3012, 1617, 1551, 1466, 1366, 1216, 751 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.18 (dd, 1H, J = 7.6, 1.2 Hz), 7.57 (t, 1H, J = 7.8 Hz), 7.38–7.33 (m, 2H), 2.62 (s, 3H), 2.05 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 175.0, 162.7, 156.5, 132.6, 126.2, 125.0, 122.6, 116.8, 116.3, 13.2, 10.3; HRMS (ESI) m/z calcd for C₁₁H₁₁O₂S [M + H]⁺ 207.0474, found 207.0476.

2-((2-(Benzylthio)ethyl)thio)-4H-chromen-4-one (**2**I). Yield: 19% (62 mg), viscous liquid, eluent (hexane/EtOAc = 9:1); IR (KBr): 2923, 1649, 1550, 1461, 1348, 1130, 758 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.12 (dd, 1H, *J* = 8.4, 2.0 Hz), 7.49 (t, 1H, *J* = 8.0 Hz), 7.32–7.06 (m, 8H), 3.85 (s, 2H), 3.40 (t, 2H, *J* = 7.6 Hz), 2.93 (t, 2H, *J* = 7.6 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 174.8, 162.3, 156.3, 139.1, 133.1, 128.7, 128.3, 126.4, 126.2, 125.3, 122.9, 121.7, 116.7, 38.6, 31.0, 30.1; HRMS (ESI) *m*/*z* calcd for C₁₈H₁₇O₂S₂ [M + H]⁺ 329.0664, found 329.0671.

3-Chloro-2-(methylthio)-4H-chromen-4-one (**3a**). Yield: 80% (55 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 144–146 °C; IR (KBr): 2925, 2854, 1644, 1525, 1463, 1353, 1320, 1215, 960, 750 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.18 (d, 1H, *J* = 8.0 Hz), 7.63 (t, 1H, *J* = 7.8 Hz), 7.40 (t, 2H, *J* = 4.4 Hz), 2.64 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.5, 164.8, 156.0, 133.3, 126.4, 125.7, 122.3, 116.7, 115.0, 13.3; HRMS (ESI) *m*/*z* calcd for C₁₀H₈ClO₂S [M + H]⁺ 226.9928, found 226.9930.

3-Bromo-2-(methylthio)-4H-chromen-4-one (3b). Yield: 90% (73 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 175–177 °C; IR (KBr): 2924, 2854, 1646, 1516, 1463, 1352, 1320, 950, 756 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.18 (d, 1H, *J* = 7.2 Hz), 7.62 (t, 1H, *J* = 7.4 Hz), 7.39 (t, 2H, *J* = 7.2 Hz), 2.64 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.7, 166.0, 156.1, 133.3, 126.6, 125.9, 121.9, 116.8, 105.7, 14.1; HRMS (ESI) *m*/*z* calcd for C₁₀H₈BrO₂S [M + H]⁺ 270.9423, found 270.9426.

3-lodo-2-(methylthio)-4H-chromen-4-one (**3c**). Yield: 87% (83 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 200–202 °C; IR (KBr): 2924, 2852, 1637, 1504, 1463, 1347, 1317, 938, 757 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.20 (dd, 1H, J = 8.4, 1.6 Hz), 7.64

(t, 1H, *J* = 7.2 Hz), 7.43–7.40 (m, 2H), 2.64 (s, 3H); ${}^{13}C{}^{1}H$ NMR (CDCl₃, 100 MHz): δ 171.2, 168.2, 156.2, 133.3, 126.8, 125.9, 119.9, 116.6, 83.2, 15.7; HRMS (ESI) *m*/*z* calcd for C₁₀H₈IO₂S [M + H]⁺ 318.9284, found 318.9284.

7-(Benzyloxy)-3-chloro-2-(methylthio)-4H-chromen-4-one (**3d**). Yield: 76% (76 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 162–164 °C; IR (KBr): 2925, 2854, 1644, 1620, 1436, 1311, 1268, 1101, 763 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.05 (d, 1H, *J* = 8.8 Hz), 7.43–7.32 (m, 5H), 6.99 (dd, 1H, *J* = 8.8, 1.6 Hz), 6.85 (d, 1H, *J* = 1.6 Hz), 5.11 (s, 2H), 2.60 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.0, 163.9, 162.7, 157.4, 135.4, 128.7, 128.4, 127.5, 127.5, 116.1, 115.1, 114.7, 100.6, 70.6, 13.3; HRMS (ESI) *m/z* calcd for $C_{17}H_{14}ClO_3S$ [M + H]⁺ 333.0347, found 333.0350.

7-(*Benzyloxy*)-3-bromo-2-(methylthio)-4H-chromen-4-one (**3e**). Yield: 86% (97 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 143–144 °C; IR (KBr): 2925, 2855, 1642, 1618, 1436, 1308, 1271, 1099, 760 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.11 (d, 1H, *J* = 9.2 Hz), 7.43–7.33 (m, 5H), 7.04 (dd, 1H, *J* = 9.2, 2.4 Hz), 6.89 (d, 1H, *J* = 2.4 Hz), 5.13 (s, 2H), 2.62 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.2, 165.0, 162.7, 157.5, 135.4, 128.7, 128.4, 127.8, 127.5, 115.7, 115.1, 105.5, 100.6, 70.6, 14.1; HRMS (ESI) *m/z* calcd for C₁₇H₁₄BrO₃S [M + H]⁺ 376.9847, found 376.9858.

7-(Benzyloxy)-3-iodo-2-(methylthio)-4H-chromen-4-one (**3***f*). Yield: 73% (93 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 152–154 °C; IR (KBr): 2923, 2854, 1617, 1585, 1493, 1434, 1304, 1270, 1092, 754 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.09 (d, 1H, *J* = 9.2 Hz), 7.43–7.33 (m, 5H), 7.02 (dd, 1H, *J* = 9.2, 2.4 Hz), 6.88 (d, 1H, *J* = 2.4 Hz), 5.13 (s, 2H), 2.61 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 170.7, 167.3, 162.8, 157.9, 135.5, 128.8, 128.5, 128.2, 127.5, 115.1, 114.1, 100.5, 83.3, 70.8, 15.6; HRMS (ESI) *m/z* calcd for C₁₇H₁₄IO₃S [M + H]⁺ 424.9703, found 424.9696.

3-Chloro-6-methyl-2-(methylthio)-4H-chromen-4-one (**3g**). Yield: 79% (57 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 154–155 °C; IR (KBr): 2925, 2852, 1644, 1527, 1481, 1317, 1102, 963, 762 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.99 (s, 1H), 7.44 (dd, 1H, J = 8.8, 2.0 Hz), 7.32 (d, 1H, J = 8.8 Hz), 2.64 (s, 3H), 2.43 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.6, 164.7, 154.4, 135.9, 134.4, 125.7, 122.0, 116.7, 114.8, 20.9, 13.3; HRMS (ESI) m/z calcd for C₁₁H₁₀ClO₂S [M + H]⁺ 241.0085, found 241.0091.

3-Bromo-6-methyl-2-(methylthio)-4H-chromen-4-one (**3h**). Yield: 90% (78 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 167–169 °C; IR (KBr): 2924, 2855, 1642, 1520, 1478, 1312, 1095, 950, 764 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.98 (d, 1H, J = 2.0 Hz), 7.43 (dd, 1H, J = 8.8, 2.0 Hz), 7.31 (d, 1H, J = 8.8 Hz), 2.63 (s, 3H), 2.43 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.8, 166.0, 154.5, 136.2, 134.6, 125.9, 121.7, 116.6, 105.5, 20.9, 14.1; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₀BrO₂S [M + H]⁺ 284.9579, found 284.9586.

3-lodo-6-methyl-2-(methylthio)-4H-chromen-4-one (**3i**). Yield: 89% (89 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 174–175 °C; IR (KBr): 2924, 2854, 1640, 1512, 1467, 1307, 1086, 942, 760 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.95 (s, 1H), 7.43 (dd, 1H, *J* = 8.4, 2.0 Hz), 7.29 (d, 1H, *J* = 8.8 Hz), 2.62 (s, 3H), 2.42 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 171.2, 168.0, 154.7, 136.0, 134.5, 126.0, 119.8, 116.4, 83.1, 21.0, 15.6; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₀IO₂S [M + H]⁺ 332.9441, found 332.9431.

3,6-Dichloro-2-(methylthio)-4H-chromen-4-one (**3***j*). Yield: 92% (72 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 196–198 °C; IR (KBr): 2961, 2918, 1645, 1514, 1423, 1340, 1303, 1207, 956, 763 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.16 (d, 1H, *J* = 2.8 Hz), 7.57 (dd, 1H, *J* = 8.8, 2.8 Hz), 7.38 (d, 1H, *J* = 8.8 Hz), 2.65 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 168.3, 165.4, 154.3, 133.5, 131.8, 125.8, 123.5, 118.6, 115.1, 13.4; HRMS (ESI) *m*/*z* calcd for C₁₀H₇Cl₂O₂S [M + H]⁺ 260.9538, found 260.9545.

3-Bromo-6-chloro-2-(methylthio)-4H-chromen-4-one (**3k**). Yield: 96% (88 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 220– 222 °C; IR (KBr): 2927, 2848, 1641, 1510, 1417, 1311, 1307, 1298, 937, 767 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.17 (d, 1H, J = 2.4 Hz), 7.58 (dd, 1H, J = 9.2, 2.4 Hz), 7.38 (d, 1H, J = 9.2 Hz), 2.65 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 168.5, 166.4, 154.4, 133.5, 131.8, 126.0, 122.9, 118.6, 105.6, 14.2; HRMS (ESI) *m*/*z* calcd for C₁₀H₇BrClO₂S [M + H]⁺ 304.9033, found 304.9024.

6-Chloro-3-iodo-2-(methylthio)-4H-chromen-4-one (**3***I*). Yield: 95% (101 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 218–220 °C; IR (KBr): 2927, 2850, 1635, 1498, 1456, 1334, 1298, 1242, 937, 767 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.16 (d, 1H, *J* = 2.4 Hz), 7.58 (dd, 1H, *J* = 9.2, 2.4 Hz), 7.38 (d, 1H, *J* = 9.2 Hz), 2.64 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 170.0, 168.6, 154.7, 133.5, 131.8, 126.2, 121.1, 118.4, 83.0, 15.7; HRMS (ESI) *m*/*z* calcd for C₁₀H₇ClIO₂S [M + H]⁺ 352.8894, found 352.8899.

6-Bromo-3-chloro-2-(methylthio)-4H-chromen-4-one (**3***m*). Yield: 92% (84 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 195–197 °C; IR (KBr): 2951, 2920, 1649, 1510, 1429, 1340, 1301, 1205, 954, 765 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.32 (d, 1H, *J* = 2.4 Hz), 7.71 (dd, 1H, *J* = 9.2, 2.4 Hz), 7.32 (d, 1H, *J* = 9.2 Hz), 2.65 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 168.2, 165.4, 154.7, 136.2, 129.0, 123.8, 119.2, 118.9, 115.1, 13.4; HRMS (ESI) *m*/ *z* calcd for C₁₀H₇BrClO₂S [M + H]⁺ 304.9033, found 304.9039.

3,6-Dibromo-2-(methylthio)-4H-chromen-4-one (**3n**). Yield: 94% (99 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 224–225 °C; IR (KBr): 2929, 2854, 1645, 1504, 1417, 1336, 1296,1203, 948, 763 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.33 (d, 1H, *J* = 2.4 Hz), 7.72 (dd, 1H, *J* = 9.2, 2.4 Hz), 7.32 (d, 1H, *J* = 9.2 Hz), 2.65 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 168.4, 166.5, 154.9, 136.3, 129.1, 123.2, 119.3, 118.8, 105.6, 14.2; HRMS (ESI) *m*/*z* calcd for C₁₀H₇Br₂O₂S [M + H]⁺ 348.8528, found 348.8537.

6-Bromo-3-iodo-2-(methylthio)-4H-chromen-4-one (**3o**). Yield: 91% (108 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 200– 202 °C; IR (KBr): 3007, 2912, 1635, 1494, 1456, 1379, 1338, 1292, 931, 763 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.32 (d, 1H, *J* = 2.4 Hz), 7.72 (dd, 1H, *J* = 8.4, 2.4 Hz), 7.31 (d, 1H, *J* = 8.4 Hz), 2.64 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.9, 168.6, 155.2, 136.3, 129.3, 121.5, 119.2, 118.6, 83.0, 15.7; HRMS (ESI) *m*/*z* calcd for C₁₀H₇BrIO₂S [M + H]⁺ 396.8389, found 396.8382.

3-Chloro-6-methoxy-2-(methylthio)-4H-chromen-4-one (**3p**). Yield: 91% (70 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 196–170 °C; IR (KBr): 2956, 2929, 1633, 1514, 1438, 1333, 1307, 1240, 1095, 950, 744 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.55 (d, 1H, *J* = 3.2 Hz), 7.34 (d, 1H, *J* = 9.2 Hz), 7.20 (dd, 1H, *J* = 9.2, 3.2 Hz), 3.87 (s, 3H), 2.64 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.4, 164.5, 157.3, 151.0, 123.2, 123.0, 118.3, 114.5, 105.7, 56.0, 13.3; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₀ClO₃S [M + H]⁺ 257.0034, found 257.0040.

3-Bromo-6-methoxy-2-(methylthio)-4H-chromen-4-one (**3q**). Yield: 93% (84 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 164–166 °C; IR (KBr): 2999, 2933, 1624, 1506, 1471, 1328, 1301, 1290, 1193, 1020, 948, 736 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.46 (d, 1H, J = 3.2 Hz), 7.27 (d, 1H, J = 9.2 Hz), 7.14 (dd, 1H, J = 9.2, 3.2 Hz), 3.83 (s, 3H), 2.59 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.4, 165.5, 157.2, 150.9, 122.8, 122.4, 118.2, 105.7, 104.9, 55.9, 14.0; HRMS (ESI) m/z calcd for C₁₁H₁₀BrO₃S [M + H]⁺ 300.9529, found 300.9525.

3-lodo-6-methoxy-2-(methylthio)-4H-chromen-4-one (**3***r*). Yield: 90% (94 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 180– 182 °C; IR (KBr): 2999, 2924, 1610, 1502, 1467, 1325, 1294, 1259, 1020, 941, 732 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.51 (d, 1H, *J* = 3.2 Hz), 7.31 (d, 1H, *J* = 9.2 Hz), 7.17 (dd, 1H, *J* = 9.2, 3.2 Hz), 3.85 (s, 3H), 2.61 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 171.0, 167.8, 157.3, 151.3, 122.9, 120.8, 118.0, 106.1, 82.6, 55.9, 15.6; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₀IO₃S [M + H]⁺ 348.9390, found 348.9394.

2-Chloro-3-(methylthio)-1H-benzo[f]chromen-1-one (**3s**). Yield: 71% (59 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 158– 160 °C; IR (KBr): 2924, 2855, 1629, 1540, 1461, 1384, 1262, 1010, 757 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 10.01 (d, 1H, *J* = 8.8 Hz), 8.04 (d, 1H, *J* = 8.8 Hz), 7.87 (d, 1H, *J* = 8.4 Hz), 7.75 (t, 1H, *J* = 8.0 Hz), 7.61 (t, 1H, *J* = 7.6 Hz), 7.44 (d, 1H, *J* = 9.2 Hz), 2.68 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 171.2, 161.7, 157.2, 135.2, 130.8, 130.4, 129.5, 128.2, 127.2, 126.9, 117.7, 116.3, 115.6, 13.3;

HRMS (ESI) m/z calcd for $C_{14}H_{10}ClO_2S [M + H]^+$ 277.0085, found 277.0094.

2-Bromo-3-(methylthio)-1H-benzo[f]chromen-1-one (**3t**). Yield: 84% (81 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 196– 198 °C; IR (KBr): 2923, 2855, 1629, 1530, 1461, 1377, 1288, 990, 813, 759 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 10.01 (d, 1H, *J* = 8.8 Hz), 8.05 (d, 1H, *J* = 8.8 Hz), 7.87 (d, 1H, *J* = 7.6 Hz), 7.74 (t, 1H, *J* = 8.0 Hz), 7.61 (t, 1H, *J* = 7.6 Hz), 7.45 (d, 1H, *J* = 9.2 Hz), 2.68 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 171.2, 163.0, 157.2, 135.2, 130.7, 130.2, 129.3, 128.2, 127.2, 126.8, 116.1, 114.8, 108.8, 14.0; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₀BrO₂S [M + H]⁺ 320.9585, found 320.9594.

2-lodo-3-(methylthio)-1H-benzo[f]chromen-1-one (**3u**). Yield: 75% (83 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 209–210 °C; IR (KBr): 2925, 2855, 1626, 1518, 1438, 1377, 1288, 979, 810, 761 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 9.99 (d, 1H, *J* = 8.4 Hz), 8.02 (d, 1H, *J* = 8.8 Hz), 7.85 (d, 1H, *J* = 7.6 Hz), 7.72 (t, 1H, *J* = 7.6 Hz), 7.59 (t, 1H, *J* = 7.6 Hz), 7.42 (d, 1H, *J* = 9.2 Hz), 2.66 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 172.6, 165.5, 157.5, 135.2, 130.8, 130.1, 129.4, 128.2, 127.3, 126.8, 116.0, 113.3, 87.7, 15.6; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₀IO₂S [M + H]⁺ 368.9441, found 368.9439.

3-Chloro-2-(methylthio)-4H-benzo[h]chromen-4-one (**3**ν). Yield: 76% (63 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 208– 209 °C; IR (KBr): 2924, 2854, 1637, 1530, 1436, 1218, 979, 760 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.31 (d, 1H, *J* = 8.8 Hz), 8.14 (d, 1H, *J* = 8.8 Hz), 7.93 (d, 1H, *J* = 7.6 Hz), 7.79–7.62 (m, 3H), 2.85 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.7, 163.6, 153.3, 135.5, 129.4, 128.3, 127.5, 125.9, 123.2, 121.3, 121.1, 118.7, 116.4, 13.6; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₀ClO₂S [M + H]⁺ 277.0085, found 277.0087.

3-Bromo-2-(methylthio)-4H-benzo[h]chromen-4-one (**3**w). Yield: 75% (72 mg), yellow solid, eluent (hexane/EtOAc = 19:1), mp: 194–196 °C; IR (KBr): 2927, 2856, 1634, 1524, 1436, 1215, 978, 759 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.27 (d, 1H, *J* = 8.0 Hz), 8.11 (d, 1H, *J* = 8.4 Hz), 7.90 (d, 1H, *J* = 7.6 Hz), 7.76–7.61 (m, 3H), 2.82 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 169.7, 164.6, 153.2, 135.7, 129.3, 128.3, 127.5, 125.9, 123.2, 121.4, 121.3, 118.2, 107.1, 14.4; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₀BrO₂S [M + H]⁺ 320.9585, found 320.9581.

3-lodo-2-(methylthio)-4H-benzo[h]chromen-4-one (**3x**). Yield: 74% (82 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 204–205 °C; IR (KBr): 2928, 2856, 1631, 1518, 1429, 1211, 975, 760 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.23 (d, 1H, *J* = 7.6 Hz), 8.08 (d, 1H, *J* = 8.8 Hz), 7.90 (d, 1H, *J* = 8.0 Hz), 7.73–7.61 (m, 3H), 2.80 (s, 3H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 171.2, 167.1, 153.5, 135.6, 129.4, 128.4, 127.5, 126.0, 123.0, 121.6, 121.4, 116.4, 85.0, 16.0; HRMS (ESI) *m*/*z* calcd for C₁₄H₁₀IO₂S [M + H]⁺ 368.9441, found 368.9449.

3-(3,5-Di-tert-butyl-4-hydroxybenzyl)-2-(methylthio)-4H-chromen-4-one (4a). Yield: 13% (26 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 180–181 °C; IR (KBr): 3435, 2919, 2865, 1723, 1619, 1546, 1490, 1340, 1219, 1025, 768 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.18 (dd, 1H, J = 8.0, 1.6 Hz), 7.60–7.55 (m, 1H), 7.37–7.32 (m, 4H), 5.00 (s, 1H), 3.82 (s, 2H), 2.63 (s, 3H), 1.39 (s, 18H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 174.8, 163.6, 157.1, 152.4, 135.8, 133.3, 130.4, 126.0, 125.7, 124.0, 122.8, 120.9, 118.5, 34.6, 31.2, 30.6, 13.7; HRMS (ESI) m/z calcd for C₂₅H₃₁O₃S [M + H]⁺ 411.1988, found 411.1982.

3-(3,5-Di-tert-butyl-4-hydroxybenzyl)-6-methoxy-2-(methylthio)-4H-chromen-4-one (**4b**). Yield: 41% (54 mg), white solid, eluent (hexane/EtOAc = 19:1), mp: 175–177 °C; IR (KBr): 3426, 2926, 2872, 1718, 1612, 1552, 1487, 1355, 1232, 1029, 779 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.57 (d, 1H, *J* = 2.8 Hz), 7.30 (d, 1H, *J* = 9.2 Hz), 7.25 (s, 2H), 7.16 (dd, 1H, *J* = 9.2, 2.8 Hz), 5.01 (s, 1H), 3.87 (s, 3H), 3.83 (s, 2H), 2.62 (s, 3H), 1.40 (s, 18H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 174.5, 163.3, 156.7, 152.1, 151.4, 135.4, 130.1, 125.7, 123.7, 122.5, 120.5, 118.2, 105.6, 55.9, 34.2, 30.9, 30.3, 13.5; HRMS (ESI) *m*/*z* calcd for C₂₆H₃₃O₄S [M + H]⁺ 441.2094, found 441.2089. pubs.acs.org/joc

2-(1,3-Dithiolan-2-ylidene)-1-(2-hydroxyphenyl)ethanone (5). Yield: 79% (188 mg), yellow solid, eluent (hexane/EtOAc = 9:1), mp: 90–92 °C; IR (KBr): 2924, 2856, 1620, 1492, 1206, 1041, 759 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 12.79 (s, 1H), 7.68 (d, 1H, *J* = 7.6 Hz), 7.41–7.35 (m, 2H), 6.95 (d, 1H, *J* = 8.8 Hz), 6.84 (t, 1H, *J* = 7.2 Hz), 3.50 (t, 2H, *J* = 6.4 Hz), 3.39 (t, 2H, *J* = 6.4 Hz); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 189.5, 170.0, 163.0, 135.2, 128.4, 119.7, 118.6, 118.4, 107.1, 39.2, 35.5; HRMS (ESI) *m*/*z* calcd for C₁₁H₁₁O₂S₂ [M + H]⁺ 239.0195, found 239.0186.

2-(4-Ethylphenyl)-4H-chromen-4-one (6). Yield: 92% (46 mg), eluent (hexane/EtOAc = 19:1); ¹H NMR (CDCl₃, 400 MHz): δ 8.20 (dd, 1H, *J* = 8.0, 1.6 Hz), 7.82 (d, 2H, *J* = 8.0 Hz), 7.68–7.64 (m, 1H), 7.53 (d, 1H, *J* = 8.4 Hz), 7.38 (t, 1H, *J* = 7.2 Hz), 7.32 (d, 2H, *J* = 8.4 Hz), 6.77 (s, 1H), 2.70 (q, 2H, *J* = 7.6 Hz), 1.25 (t, 3H, *J* = 7.6 Hz); HRMS (ESI) *m*/*z* calcd for C₁₇H₁₅O₂ [M + H]⁺ 251.1067, found 251.1062.

2-(*Pyrrolidin-1-yl*)-4*H*-chromen-4-one (**7a**). Yield: 81% (35 mg) (recrystallized from methanol); ¹H NMR (CDCl₃, 400 MHz): δ 8.09 (dd, 1H, *J* = 7.6, 1.6 Hz), 7.46–7.42 (m, 1H), 7.26–7.19 (m, 2H), 5.24 (s, 1H), 3.44 (br, 4H), 1.98 (t, 4H, *J* = 6.8 Hz); HRMS (ESI) *m*/*z* calcd for C₁₃H₁₄NO₂ [M + H]⁺ 216.1019, found 216.1025.

2-(*Piperidin-1-yl*)-4*H*-chromen-4-one (**7b**). Yield: 85% (39 mg) (recrystallized from methanol); ¹H NMR (CDCl₃, 400 MHz): δ 8.05 (dd, 1H, *J* = 7.6, 1.2 Hz), 7.47–7.43 (m, 1H), 7.26–7.19 (m, 2H), 5.55 (s, 1H), 3.45 (br, 4H), 1.62 (br, 6H); HRMS (ESI) *m*/*z* calcd for C₁₄H₁₆NO₂ [M + H]⁺ 230.1176, found 230.1159.

2-(Phenylamino)-4H-chromen-4-one (**8a**). Yield: 84% (40 mg), white solid, eluent (DCM/MeOH = 4:1), mp: 205–207 °C; IR (KBr): 3165, 2926, 1615, 1553, 1240, 758 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.14 (d, 1H, *J* = 8.0 Hz), 7.56–7.51 (m, 2H), 7.38–7.27 (m, 6H), 7.19 (t, 1H, *J* = 7.2 Hz), 5.79 (s, 1H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 177.5, 161.4, 153.9, 136.3, 132.5, 129.6, 125.8, 125.7, 124.9, 123.3, 122.9, 116.3, 88.0; HRMS (ESI) *m/z* calcd for C₁₅H₁₂NO₂ [M + H]⁺ 238.0863, found 238.0871.

7-(*Benzyloxy*)-3-chloro-2-(phenylamino)-4H-chromen-4-one (**8b**). Yield: 82% (101 mg), white solid, eluent (DCM/MeOH = 4:1), mp: 179–181 °C; IR (KBr): 2925, 2857, 1610, 1554, 1423, 1275, 758 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.10 (d, 1H, *J* = 8.8 Hz), 7.44–7.26 (m, 11H), 6.99 (dd, 1H, *J* = 8.8, 2.0 Hz), 6.80 (d, 1H, *J* = 2.0 Hz), 5.08 (s, 2H); ¹³C{¹H} NMR (CDCl₃, 100 MHz): δ 170.3, 162.4, 156.4, 153.4, 136.0, 135.6, 129.4, 128.7, 128.4, 127.5, 127.3, 125.6, 122.3, 116.3, 114.1, 101.1, 95.2, 70.5; HRMS (ESI) *m/z* calcd for C₂₂H₁₇ClNO₃ [M + H]⁺ 378.0891, found 378.0890.

3-Chloro-7-hydroxy-2-(phenylamino)-4H-chromen-4-one (9). Yield: 93% (54 mg), white solid, mp: 238–240 °C; IR (KBr): 2925, 2857, 1609, 1556, 1432, 1250, 761 cm⁻¹; ¹H NMR (DMSO-d₆, 400 MHz): δ 10.66 (s, 1H), 9.70 (s, 1H), 7.82 (d, 1H, *J* = 8.4 Hz), 7.45–7.39 (m, 4H), 7.24–7.20 (m, 1H), 6.86 (dd, 1H, *J* = 8.4, 2.0 Hz), 6.61 (d, 1H, *J* = 2.0 Hz); ¹³C{¹H} NMR (DMSO-d₆, 100 MHz): δ 169.5, 161.8, 156.6, 153.4, 136.9, 129.0, 128.0, 126.6, 125.1, 123.4, 114.5, 101.8, 93.8; HRMS (ESI) *m*/*z* calcd for C₁₅H₁₁ClNO₃ [M + H]⁺ 288.0422, found 288.0426.

7-Hydroxy-2-(methylthio)-4H-chromen-4-one (**10**). Yield: 95% (40 mg), white solid, mp: 200–202 °C; IR (KBr): 2926, 2857, 1612, 1567, 1375, 764 cm⁻¹; ¹H NMR (DMSO-d₆, 400 MHz): δ 7.81 (d, 1H, *J* = 8.4 Hz), 6.88 (d, 1H, *J* = 8.4 Hz), 6.82 (s, 1H), 6.13 (s, 1H), 4.77 (br, 1H), 2.56 (s, 3H); ¹³C{¹H} NMR (DMSO-d₆, 100 MHz): δ 173.9, 169.1, 162.6, 158.0, 126.7, 115.5, 114.8, 106.4, 102.0, 13.4; HRMS (ESI) *m*/*z* calcd for C₁₀H₉O₃S [M + H]⁺ 209.0267, found 209.0278.

4H-Chromen-4-one (**11***a*). Yield: 92% (27 mg), eluent (hexane/ EtOAc = 9:1); ¹H NMR (CDCl₃, 400 MHz): δ 8.18 (d, 1H, *J* = 7.6 Hz), 7.83 (d, 1H, *J* = 6.0 Hz), 7.65 (t, 1H, *J* = 7.6 Hz), 7.44–7.36 (m, 2H), 6.32 (d, 1H, *J* = 6.0 Hz); HRMS (ESI) *m*/*z* calcd for C₉H₇O₂ [M + H]⁺ 147.0441, found 147.0436.

3-Methyl-4H-chromen-4-one (11b). Yield: 81% (26 mg), eluent (hexane/EtOAc = 9:1); ¹H NMR (CDCl₃, 400 MHz): δ 8.21 (d, 1H, J = 8.4 Hz), 7.77 (s, 1H), 7.61 (t, 1H, J = 6.8 Hz), 7.41–7.34 (m, 2H), 2.01 (s, 3H); HRMS (ESI) *m*/*z* calcd for C₁₀H₉O₂ [M + H]⁺ 161.0597, found 161.0602.

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Spectral data for all new compounds (PDF)

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Notes

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REFERENCES

(1) (a) Lichtenstein, A. H. Isoflavones and Cardiovascular Disease risk. J. Nutr. **1998**, *128*, 1589–1592. (b) Abou-Zaid, M. M.; Lombardo, D. A.; Kite, G. C.; Grayer, R. J.; Veitch, N. C. Acylated Flavone C-Glycosides from Cucumis Sativus. *Phytochemistry* **2001**, *58*, 167–172. (c) Beecher, G. R. Overview of Dietary Flavonoids: Nomenclature, Occurrence and Intake. J. Nutr. **2003**, *133*, 32488– 3254S.

(2) (a) Matin, A.; Gavande, N.; Kim, M. S.; Yang, N. X.; Salam, N. K.; Hanrahan, J. R.; Roubin, R. H.; Hibbs, D. E. 7-Hydroxybenzopyran-4-one Derivatives: A Novel Pharmacophore of Peroxisome Proliferator-Activated Receptor α and γ (PPAR α and γ) Dual Agonists. J. Med. Chem. 2009, 52, 6835–6850. (b) Verma, A. K.; Pratap, R. Chemistry of Biologically Important Flavones. Tetrahedron 2012, 68, 8523–8538. (c) Keri, R. S.; Budagumpi, S.; Pai, R. K.; Balakrishna, R. G. Chromones as a Privileged Scaffold in Drug Discovery: A Review. Eur. J. Med. Chem. 2014, 78, 340–374.

(3) Gaspar, A.; Matos, M. J.; Garrido, J.; Uriarte, E.; Borges, F. Chromone: A Valid Scaffold in Medicinal Chemistry. *Chem. Rev.* 2014, 114, 4960-4992.

(4) (a) Kaye, P. T.; Nchinda, A. T.; Gray, C. A. Chromone Studies. part 11. Synthesis and Electron Impact Mass Spectrometric Study of Granulosin and Side-Chain Analogues. *J. Chem. Res.* **2002**, 2002, 321–325. (b) Williams, D. A.; Smith, C.; Zhang, Y. An Efficient Procedure for the Preparation of Natural Products Bearing the 2-(2-Phenylethyl)chromone Skeleton. *Tetrahedron Lett.* **2013**, *54*, 4292– 4295.

(5) Nohara, A.; Umetani, T.; Sanno, Y. Studies on Antianaphylactic Agents-I; A Facile Synthesis of 4-Oxo-4*h*-1-benzopyran-3-carboxalde-hydes by Vilsmeier Reagents. *Tetrahedron* **1974**, *30*, 3553–3561.

(6) (a) Thasana, N.; Ruchirawat, S. The Application of the Baker-Venkataraman Rearrangement to the Synthesis of Benz[b]indeno[2,1-e]pyran-10,11-dione. *Tetrahedron Lett.* 2002, 43, 4515-4517.
(b) Santos, C. M. M.; Silva, A. M. S.; Cavaleiro, J. A. S. Synthesis of New Hydroxy-2-Styrylchromones. *Eur. J. Org. Chem.* 2003, 2003, 4575-4585.

(7) Ellis, G. P.; Barker, G. Chromones-2- and -3-carboxylic acids and Their Derivatives. *Prog. Med. Chem.* **1973**, *9*, 65-116.

(8) (a) Yue, Y.; Peng, J.; Wang, D.; Bian, Y.; Sun, P.; Chen, C. J. Synthesis of 4*H*-Chromen-4-one Derivatives by Intramolecular Palladium-Catalyzed Acylation of Alkenyl Bromides with Aldehydes. *J. Org. Chem.* **2017**, *82*, 5481–5486. (b) Zhao, X.; Zhou, J.; Lin, S.; Jin, X.; Liu, R. C-H Functionalization via Remote Hydride Elimination: Palladium Catalyzed Dehydrogenation of ortho-Acyl Phenols to Flavonoids. *Org. Lett.* **2017**, *19*, 976–979. (c) Yang, Q.; Alper, H. Synthesis of Chromones via Palladium-Catalyzed Ligand-Free Cyclocarbonylation of *o*-Iodophenols with Terminal Acetylenes in Phosphonium Salt Ionic Liquids. *J. Org. Chem.* **2010**, *75*, 948–950. (d) Liu, J.; Song, w.; Yue, Y.; Liu, R.; Yi, H.; Zhuo, K.; Lei, A. Pd(OAc)₂/S=PPh₃ Accelerated Activation of *gem*-Dichloroalkenes for the Construction of 3-arylchromones. *Chem. Commun.* **2015**, *51*, 17576–17579.

(9) Kausar, H.; Khan, M. U.; Begum, S. N.; Baig, M. M. V.; Baseer, M. A. Synthesis of Some New Halogen Substituted Flavone Derivatives. *Inter. J. Pharm. Sci. Res.* **2013**, *5*, 76–79.

(10) Lokhande, P. D.; Sakate, S. S.; Taksande, K. N.; Navghare, B. Dimethylsulfoxide-Iodine Catalysed Deprotection of 2'-Allyloxychalcones: Synthesis of Flavones. *Tetrahedron Lett.* **2005**, *46*, 1573.

(11) Pinto, D. C. G. A.; Silva, A. M. S.; Cavaleiro, J. A. S. Synthesis of 5-Hydroxy-2-(phenyl or styryl)chromones and of Some Halo Derivatives. *J. Heterocycl. Chem.* **1996**, *33*, 1887–1893.

(12) Yang, Z.; Wang, Y.; Hu, L.; Yu, J.; Li, A.; Li, L.; Yang, T.; Zhou, C. Electrochemically Induced Thiocyanation of Enaminones: Synthesis of Functionalized Alkenes and Chromones. *Synthesis* **2020**, *52*, 711–718.

(13) (a) Baruah, S.; Kaishap, P. P.; Gogoi, S. Ru(II)-Catalyzed C-H Activation and Annulation of Salicylaldehydes with Monosubstituted and Disubstituted alkynes. *Chem. Commun.* **2016**, *52*, 13004–13007. (b) Raja, G. C. E.; Ryu, J. Y.; Lee, J.; Lee, S. Ruthenium-Catalyzed C-H Activation of Salicylaldehyde and Decarboxylative Coupling of Alkynoic Acids for the Selective Synthesis of Homoisoflavonoids and Flavones. *Org. Lett.* **2017**, *19*, 6606–6609. (c) Sun, P.; Gao, S.; Yang, C.; Guo, S.; Lin, A.; Yao, H. Controllable Rh(III)-Catalyzed Annulation between Salicylaldehydes and Diazo Compounds: Divergent Synthesis of Chromones and Benzofurans. *Org. Lett.* **2016**, *18*, 6464–6467. (d) Cai, L.; Zhu, X.; Chen, J.; Lin, A.; Yao, H. Rh(III)-Catalyzed C-H Activation/Annulation of Salicylaldehydes with Sulfoxonium Ylides for the Synthesis of Chromones. *Org. Chem. Front.* **2019**, *6*, 3688–3692.

(14) (a) Zhou, C.; Dubrovsky, A. V.; Larock, R. C. Diversity-Oriented Synthesis of 3-Iodochromones and Heteroatom Analogues via ICI-Induced Cyclization. J. Org. Chem. **2006**, 71, 1626–1632. (b) Yue, D.; Yao, T.; Larock, R. C. Synthesis of 2,3-Disubstituted Benzo[b]furans by the Palladium-Catalyzed Coupling of o-Iodoanisoles and Terminal Alkynes, Followed by Electrophilic Cyclization. J. Org. Chem. **2005**, 70, 10292–10296. (c) Yao, T.; Larock, R. C. Synthesis of Isocoumarins and α -Pyrones via Electrophilic Cyclization. J. Org. Chem. **2003**, 68, 5936–5942.

Article

(15) (a) Singh, G.; Ila, H.; Junjappa, H. Bromination of α aroylketene S,S-acetals: synthesis of novel α -aroyl- α -bromoketene S,Sacetals and their further synthetic transformations. Synthesis **1985**, 1985, 165–169. (b) Kumar, S. V.; Acharya, A.; Ila, H. Synthesis of 2,4,5-Trisubstituted Oxazoles with Complementary Regioselectivity from α -Oxoketene Dithioacetals and β -(Methylthio)- β -(het)aryl-2propenones. J. Org. Chem. **2018**, 83, 6607–6622.

(16) Ramanarayanan, G. V.; Shukla, V. G.; Akamanchi, K. G. A Novel and One Step Procedure for Preparation of α -Bromo- $\alpha_{\beta}\beta$ -Unsaturated Carbonyl Compounds. Synlett **2002**, 12, 2059–2061.

(17) Jyothi, D.; Hari Prasad, S. A Remarkably Simple One-Step Procedure for the Preparation of α -Bromo- α , β -unsaturated Carbonyl Compounds. *Synlett* **2009**, 2009 (14), 2309–2311.

(18) Zhao, W.; Xie, P.; Bian, Z.; Zhou, A.; Ge, H.; Zhang, M.; Ding, Y.; Zheng, L. Ammonium Iodide Induced Nonradical Regioselective Sulfenylation of Flavones via a C-H Functionalization Process. J. Org. Chem. **2015**, 80, 9167–9175.

(19) Tang, Q.; Bian, Z.; Wu, W.; Wang, J.; Xie, P.; Pittman, C. U.; Zhou, A. Making Flavone Thioethers Using Halides and Powdered Sulfur or Na₂S₂O₃. *J. Org. Chem.* **2017**, *82*, 10617–10622.

(20) Lee, G. H.; Pak, C. S. Synthesis of 2-Methylthio-4H-4chromenones via Ketene dithioacetal in the One-pot Reaction. *Synth. Commun.* **1999**, *29*, 2539–2545.

(21) (a) Kolb, M. Ketene Dithioacetals in Organic Synthesis: Recent Developments. *Synthesis* **1990**, *1990* (3), *171–190*. (b) Pan, L.; Liu, Q. [5 + 1]-Annulation Strategy Based on Alkenoyl Ketene Dithioacetals and Analogues. *Synlett* **2011**, *2011* (8), *1073–1080*. (c) Pan, L.; Bi, X.; Liu, Q. Recent developments of ketene dithioacetal chemistry. *Chem. Soc. Rev.* **2013**, *42*, 1251–1286.

(22) (a) Singh, S.; Althagafi, I.; Yadav, P.; Panwar, R.; Kumar, A.; Pratap, R. Base Mediated Synthesis of α -Aminated aroyl/acetylnaphthalenes through [4+2] Annulations. *Tetrahedron* **2014**, 70, 8879– 8884. (b) Panwar, R.; Althagafi, I.; Shally; Shaw, R.; Elagamy, A.; Shah, C.; Yadav, P.; Pratap, R. Transition Metal Free Synthesis of Multifunctional Thiomethylated-benzenes from Aryl/Heteroaryl/ Cyclopropyl methyl ketones. *Tetrahedron* **2020**, 76, 131183. (c) Shally; Althagafi, I.; Shaw, R.; Elagamy, A.; Kumar, A.; Pratap, R. [5+ 1] Annulation strategy for the synthesis of multifunctional biaryls and p-teraryls from 1, 6-Michael acceptor ketene dithioacetals. *Org. Biomol. Chem.* **2020**, 18, 6407. (d) Singh, S.; Reddy, T. S.; Panwar, R.; Misra, R.; Pratap, R. 2-(2,2-Bis-benzylamino-1-cyanovinyl)-benzonitrile: A selective turn-off fluorescent Cu²⁺ sensor. *Chemistry Select* **2016**, 1, 2576–2580.

(23) Liu, C.; Taheri, A.; Lai, B.; Gu, Y. Synergistic Catalysis-Induced Ring-Opening Reactions of 2-Substituted 3,4-dihydropyrans with α -Oxoketene dithioacetals. *Catal. Sci. Technol.* **2015**, *5*, 234–245.

(24) Kusturin, C.; Liebeskind, L. S.; Rahman, H.; Sample, K.; Schweitzer, B.; Srogl, J.; Neumann, W. L. Switchable Catalysis: Modular Synthesis of Functionalized Pyrimidinones via Selective Sulfide and Halide Cross-Coupling Chemistry. *Org. Lett.* **2003**, *5*, 4349–4352.

(25) Wang, Q.; Yang, X.; Wu, P.; Yu, Z. Photoredox-Catalyzed C-H Arylation of Internal Alkenes to Tetrasubstituted Alkenes: Synthesis of Tamoxifen. *Org. Lett.* **201**7, *19*, 6248–6251.

(26) Lee, G. H.; Pak, C. S. Synthesis of 2-Methylthio-4H-4chromenones via Ketene dithioacetal in the One-pot Reaction. *Synth. Commun.* **1999**, *29*, 2539–2545.

(27) Awuah, E.; Capretta, A. Access to Flavones via a Microwave-Assisted, One-pot Sonogashira-Carbonylation-Annulation Reaction. *Org. Lett.* **2009**, *11*, 3210–3213.

(28) Kaye, P. T.; Ramaite, I. D. I. Chromone Studies. part 9.1 Dynamic NMR Analysis of Rotational Isomerism in 2-(*N*,*N*-Dialkylamino)chromones. *J. Chem. Res., Synop.* **1997**, 414–415.

(29) (a) Balakrishna, C.; Kandula, V.; Gudipati, R.; Yennam, S.; Devi, P. U.; Behera, M. An Efficient Microwave-Assisted Propylphosphonic Anhydride (T3P®)-Mediated One-pot Chromone Synthesis via Enaminones. *Synlett* **2018**, *29*, 1087–1091. (b) Santhosh, K. C.; Balasubramanian, K. K. A Facile Synthesis of 3-Alkylchromanones and Chromones. *Tetrahedron Lett.* **1991**, *32*, 7727–7730.