

Accepted Manuscript

Synthesis of (Benzimidazol-2-yl)aniline derivatives as glycogen phosphorylase inhibitors

Shadia A. Galal, Muhammad Khattab, Fotini Andreadaki, Evangelia D. Chrysina, Jean-Pierre Praly, Fatma A.F. Ragab, Hoda I. El Diwani

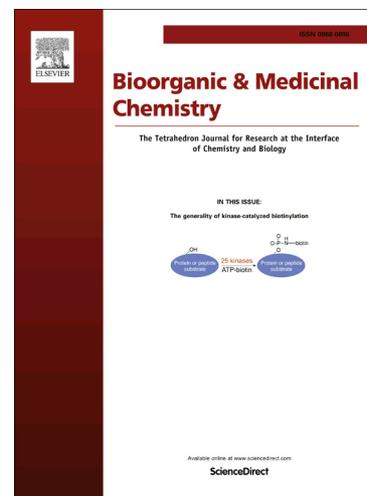
PII: S0968-0896(16)30686-1
DOI: <http://dx.doi.org/10.1016/j.bmc.2016.08.069>
Reference: BMC 13252

To appear in: *Bioorganic & Medicinal Chemistry*

Received Date: 20 May 2016
Revised Date: 28 August 2016
Accepted Date: 31 August 2016

Please cite this article as: Galal, S.A., Khattab, M., Andreadaki, F., Chrysina, E.D., Praly, J-P., Ragab, F.A.F., El Diwani, H.I., Synthesis of (Benzimidazol-2-yl)aniline derivatives as glycogen phosphorylase inhibitors, *Bioorganic & Medicinal Chemistry* (2016), doi: <http://dx.doi.org/10.1016/j.bmc.2016.08.069>

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



SYNTHESIS OF (BENZIMIDAZOL-2-YL)ANILINE DERIVATIVES AS GLYCOGEN PHOSPHORYLASE INHIBITORS

Shadia A. Galal,^a Muhammad Khattab,^a Fotini Andreadaki,^b Evangelia D. Chrysina,^b Jean-Pierre Praly,^c Fatma A.F. Ragab,^d Hoda I. El Diwani,^a

^a *Department of Chemistry of Natural and Microbial Products, Division of Pharmaceutical and Drug Industries, National Research Centre, Dokki, 12622, Cairo, Egypt.*

^b *Institute of Biology, Medicinal Chemistry & Biotechnology, National Hellenic Research Foundation, 48 Vassileos Constantinou Avenue, Athens, GR-11635, Greece.*

^c *Institut de Chimie et Biochimie Moléculaires et Supramoléculaires, Laboratoire de Chimie Organique 2 – Glycochimie, UMR 5246, CNRS, Université Claude Bernard Lyon 1, 43 Boulevard du 11 Novembre 1918, F-69622 Villeurbanne, France.*

^d *Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Cairo University, Cairo, Egypt.*

Abstract

A series of (benzimidazol-2-yl)-aniline (**1**) derivatives has been synthesized and evaluated as glycogen phosphorylase (GP) inhibitors. Kinetics studies revealed that compounds displaying a lateral heterocyclic residue with several heteroatoms (series 3 and 5) exhibited modest inhibitory properties with IC₅₀ values in the 400-600 μM range. Arylsulfonyl derivatives **7** (Ar: phenyl) and **9** (Ar: *o*-nitrophenyl) of **1** exhibited the highest activity (series 2) among the studied compounds (IC₅₀ 324 μM and 357 μM, respectively) with stronger effect than the *p*-tolyl analogue **8**.

Keywords: Benzimidazole ; Heterocycles ; glycogen phosphorylase ; enzyme inhibition

Corresponding authors:

E-mail: sh12galal@hotmail.com, sh12galal@yahoo.com

E-mail: echrysina@eie.gr, echry@tee.gr

E-mail: jean-pierre.praly@univ-lyon1.fr

1. Introduction

Alkaloids and generally nitrogen heterocycles are natural and synthetic compounds, displaying a wide range of properties and activities. They are frequently used as chemicals and for crop or health protection. This applies to indole¹ and benzimidazole derivatives,^{2,3} two sub-classes having natural representatives, as tryptophan and neurotransmitter serotonin or *N*-ribosyl-dimethylbenzimidazole the axial ligand for cobalt in vitamin B₁₂. Indole and benzimidazole scaffolds are important pharmacophores for drug discovery as they are good bioisosteres of adenine and guanine present in nucleosides and nucleotides. As revealed by academic and industrial researches, indole⁴ and benzimidazole derivatives,⁵⁻⁷ represent privileged substructures which may interact with proteins, enzymes, and biomolecules. We currently developed synthetic routes toward a number of benzimidazole derivatives exhibiting cytotoxic,⁸ antitumor⁹⁻¹² antiangiogenic,^{13,14} or analgesic properties,^{15,16} while others showed potent activity against HSV-1,¹⁷ or led to metal complexes studied as topoisomerase II inhibitors.¹⁷ Novel benzimidazole derivatives, found activators of AMP-protein kinase, have been patented for use in the treatment, prevention and suppression of diseases mediated by the AMPK-activated protein kinase.^{18,19} A collaborative work has offered the opportunity for evaluating new benzimidazole derivatives as potential antidiabetic drugs.

Diabetes, particularly its predominant form type 2 diabetes mellitus (T2DM) represents a global health problem, characterized by elevated circulating glucose (hyperglycemia). Although the etiology of T2DM is unclear, the overall metabolic dysfunction is attributed to both relative insensitivity of glucose-metabolizing tissues (muscles, liver, fat tissues) to insulin, and to deficient insulin production from the pancreas. Diabetes management involves diet control, exercise, and pharmacological treatments.²⁰ As they may fail in normalizing glycemia, thus increasing the risk of severe long-term complications, ongoing investigations from both academia and pharmaceutical companies are addressing the potential identified targets, particularly glycogen phosphorylase (GP). This enzyme catalyzes glycogen degradation to glucose-1-phosphate in the muscles, and in the liver with final release of glucose to the blood stream.²¹⁻²⁵ Therefore, GP inhibition is considered a validated pharmacological approach, as a means of limiting hyperglycemia.²⁶⁻²⁹ Due to extensive studies based mostly on kinetics and X-ray crystallography, of the muscles, and liver isoforms (a brain isoform also exists³⁰), GP (Fig. 1) is a well-known homodimeric enzyme³¹ which displays a phosphorylation site and various binding sites. They have been characterized as the catalytic site (referred to as the active), the inhibitor, the allosteric, the new allosteric or indole site and the glycogen storage site.³²⁻³⁴ A binding site, called benzimidazole binding site³⁵ was also

identified and recently, a quercetin-binding site.³⁶ However, the role of these two sites for the design of new potent GP inhibitors is yet to be elucidated.

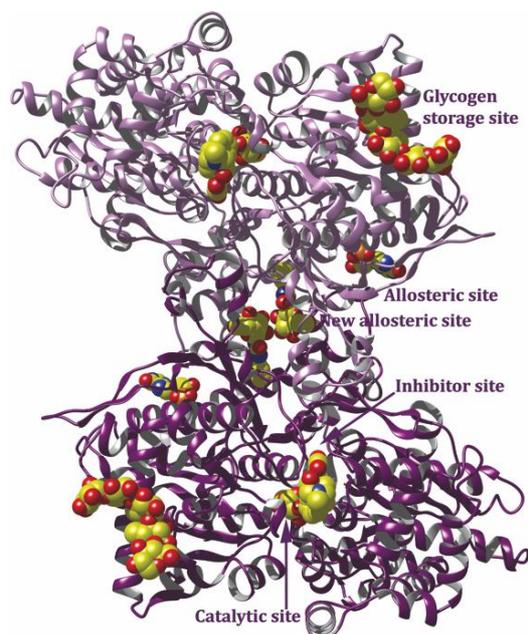
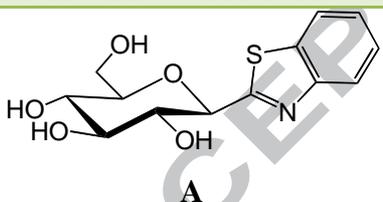
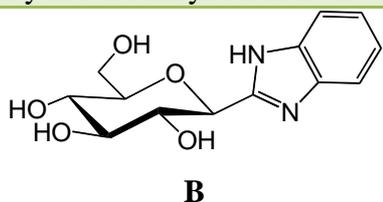
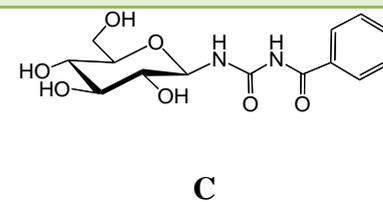
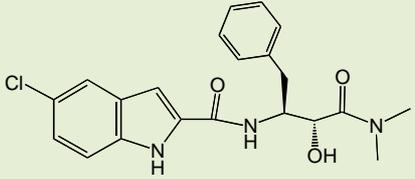
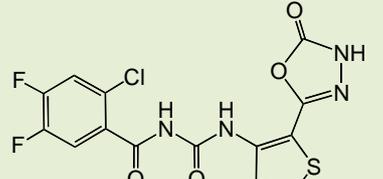


Fig. 1: The three dimensional structure of T-state rabbit muscle GPb shown where the distinct binding sites targeted for design of new potential antidiabetics are indicated. The figure was prepared using *Chimera*.³¹

The active site of GP accommodates the substrates glucose-1-phosphate and glycogen, and the inhibitors glucose and glucose analogues. Not surprisingly, glucose-based derivatives represent the most populated class of GP inhibitors. Among them, 2-(β -D-glucopyranosyl)-benzothiazole and benzimidazole (**A**, **B**),³⁷ proved to be moderate GP inhibitors (K_i against the unphosphorylated Rabbit Muscle GP, RMGPb are 229 μ M³⁸ or 76 μ M³⁵ and 11 μ M³⁸ or 9 μ M,³⁵ respectively). Both were found to bind at the enzyme catalytic site (as expected for glucose-based inhibitors) while promoting the inactive T-state³⁹ of the enzyme, thereby explaining the observed inhibitions. However, 2-(β -D-glucopyranosyl)-benzimidazole **B** was found to bind also at the new allosteric inhibitor site (Fig. 1), and additionally at benzimidazole binding site, located at the protein surface, far removed (~ 32 Å) from the other binding sites. Interestingly, *N*-benzoyl-*N*-glucosyl-urea **C** was found to bind to the catalytic site, and to the new allosteric inhibitor site, as proved by crystallographic studies.⁴⁰ Even for glucose-based inhibitors, crystal analyses of the enzyme-ligand complexes might reveal unusual binding modes as for benzimidazole, and acylurea-derived structures **B** and **C**.

Lead discovery by library screening revealed indole-carboxamide derivatives as **D** CP-91149,⁴¹ **E** CP320626 and others,⁴² as highly potent GP inhibitors (IC_{50} in the nanomolar range) showing synergism with glucose, e.g. lower IC_{50} in the presence of glucose (Table 1). Crystallographic studies showed their binding to the new allosteric site,^{43, 44} also referred to as the indole-binding site, as it accommodates many indole-2-carboxamide-derived potent inhibitors with stabilization of the inactive T-conformation of GP. Located at the interface of two GP subunits, it is a 30 Å long central cavity with an indole site on each subunit. Consequently, bis-indole derivatives are among the most potent known allosteric indole inhibitors,⁴⁵ but for other potent allosteric inhibitors (e.g. **F**), the binding site or mechanism of inhibition have not been established.⁴⁶ Compounds of the chloroindole series have reached phase II clinical trials, but their development has been discontinued, as *in vivo* tests have shown glycogen accumulation in both liver and muscle.^{24,45,48} While 2-nitrobenzimidazole has been recently reported to exhibit significant antihyperglycemic activity in alloxan-induced diabetic rats comparable to that of the antidiabetic sulfonylurea, glibenclamide,⁴⁸ the need for leads with an improved pharmacological profile still exists.⁴⁹ Although the pharmaceutical industries have continued their efforts in targeting glycogen phosphorylase it is more of the academic research groups that have a continuous interest in designing and developing new glycogen phosphorylase inhibitors.^{50,51}

Table 1: Selected GP inhibitors and their binding sites: glucose-based heterocycles, chloroindole-carboxamide derivatives and an analogue.

Glucose-derived inhibitors found to bind mostly at the catalytic site of GP		
 <p>A</p> <p>K_i (RMGPb) 229 μM,³⁸ 76 μM³⁵</p>	 <p>B</p> <p>K_i (RMGPb) 11 μM³⁸ 9 μM³⁵</p> <p>binding to GP was also observed at the indole-binding site and at the benzimidazole site³⁵</p>	 <p>C</p> <p>K_i (RMGPb) 4.6 μM³⁹</p> <p>binding to GP was also observed at the indole-binding site</p>
Potent heterocyclic GP's inhibitors and binding sites:		
new allosteric / indole site	new allosteric / indole site	binding mode unknown
		

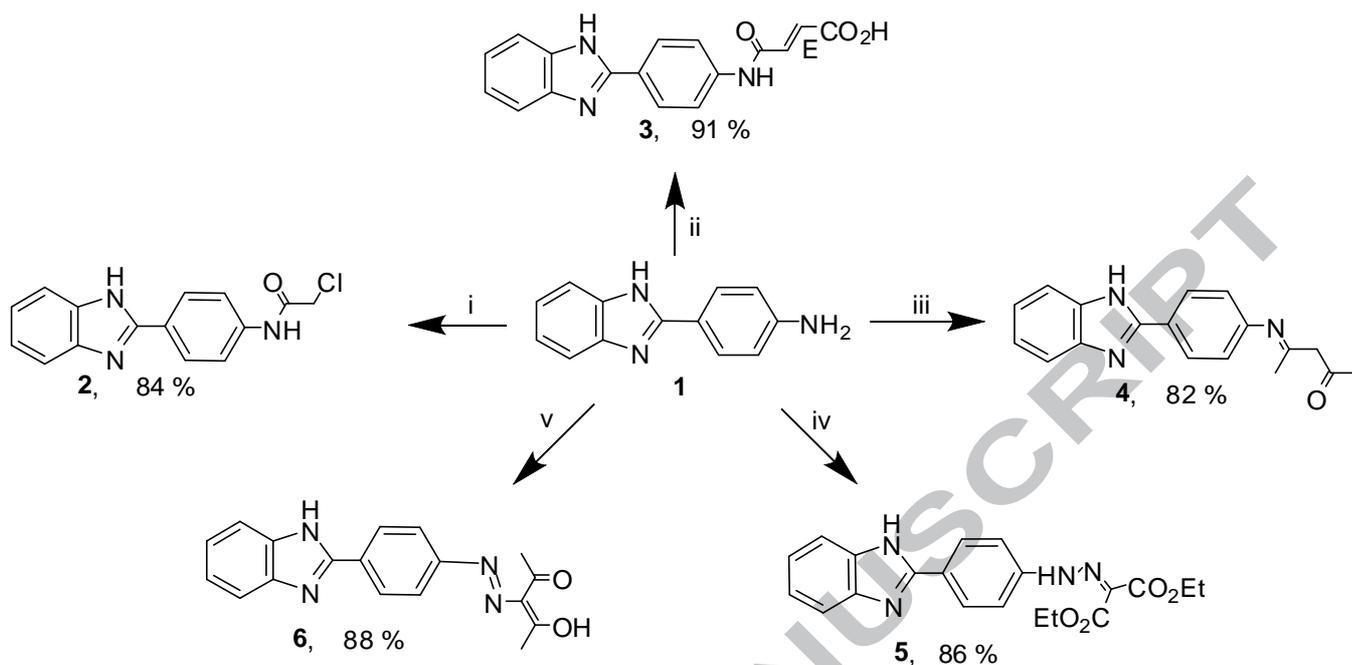
D CP-91149 Pfizer IC ₅₀ (rHLGPa) ⁴¹ 0.082 μM with [glucose] = 7.5 mM	E CP-320626 Pfizer IC ₅₀ (rHLGPa) ⁴² , 0.2 μM with [glucose] = 7.5 mM IC ₅₀ (RMGPb) ⁴³ 0.178 μM with [glucose] = 10 mM	F IC ₅₀ 30 μM (GPa) potent cpd displaying a 1,3,4-oxadiazol-5-oxo moiety similar to the polyazacycles found in inhibitors (series 3 & 5)
RMGPb: unphosphorylated rabbit muscle glycogen phosphorylase. rHLGPa : phosphorylated recombinant human liver glycogen phosphorylase		

Therefore, on the basis of the above information, a series of new (benzimidazol-2-yl)aniline derivatives were synthesized by grafting various acyclic, arylsulfonyl, and heterocyclic residues to the amino group, and they were evaluated as potential GP inhibitors. The results obtained are presented hereafter.

2. Results and discussion

2.1. Chemistry

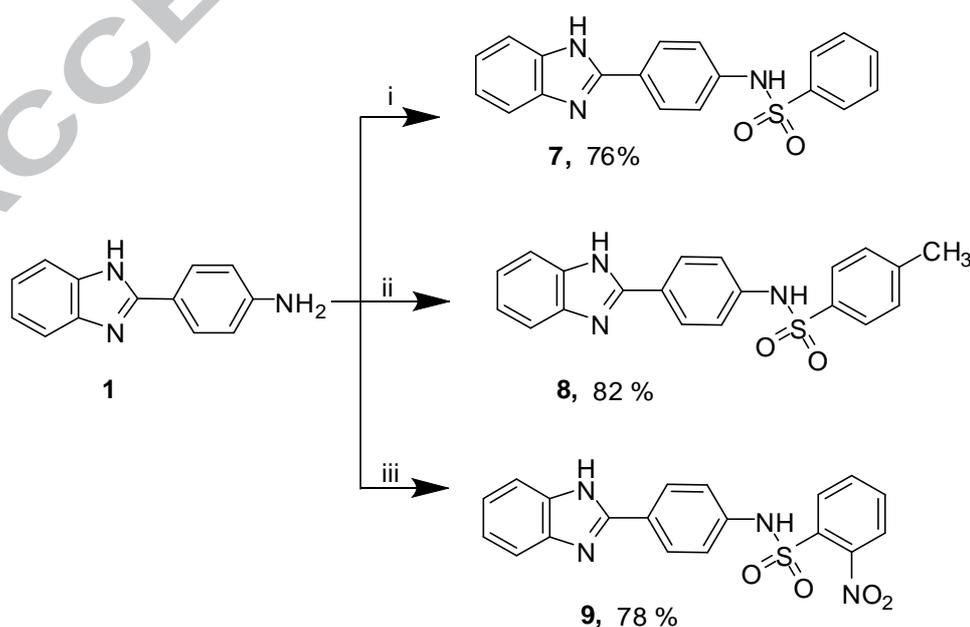
The starting material 4-(1*H*-benzo[*d*]imidazol-2-yl)aniline (**1**)⁵² and compounds **2**, **3**, **6**, **16**, **17**, **18** and **21**^{53, 54} (Schemes 1- 3) were known compounds which were synthesized according to described procedures. For the purpose of our study, we firstly prepared a series of derivatives of **1** by attaching a chain to its amino group (Scheme 1). Treatment of **1** with chloroacetyl chloride in DMF and using Et₃N, as a modification of the procedure described by Shahare *et al.*,⁵² led to 2-chloroacetamide derivative **2**,⁵² while its reaction with maleic anhydride afforded compound **3**.⁵³ Compound **4** was obtained by treating compound **1** with acetylacetone in glacial acetic acid. Compounds **5** and **6**⁵⁴ were produced via two steps, involving firstly diazotization of compound **1** achieved by using HCl and NaNO₂ at 0 °C. The second step was the reaction of the crude diazonium chloride with diethyl malonate or acetyl acetone to form compounds diethyl 2-(2-(4-(1*H*-benzo[*d*]imidazol-2-yl)phenyl)hydrazono)malonate (**5**) or 3-((4-(1*H*-benzo[*d*]imidazol-2-yl)phenyl)diazenyl)-4-hydroxypent-3-en-2-one (**6**), respectively.⁵³



Conditions: i- **1**, ClCOCH_2Cl , Et_3N , DMF; ii- **1**, maleic anhydride, toluene, reflux, 6h; iii- **1**, acetylacetone, glacial acetic acid, reflux, 10h; iv- **1** then a, 6N HCl, aq. NaNO_2 , 0°C ; b, diethyl malonate, ethanol, $5-10^\circ\text{C}$; c, aq. NaOAc , rt; v- **1** then a, 6N HCl, aq. NaNO_2 , 0°C ; b, acetyl acetone, acetone, $5-10^\circ\text{C}$; c, aq. NaOAc , rt.

Scheme 1: Synthetic routes to compounds 2-6 (Series 1)

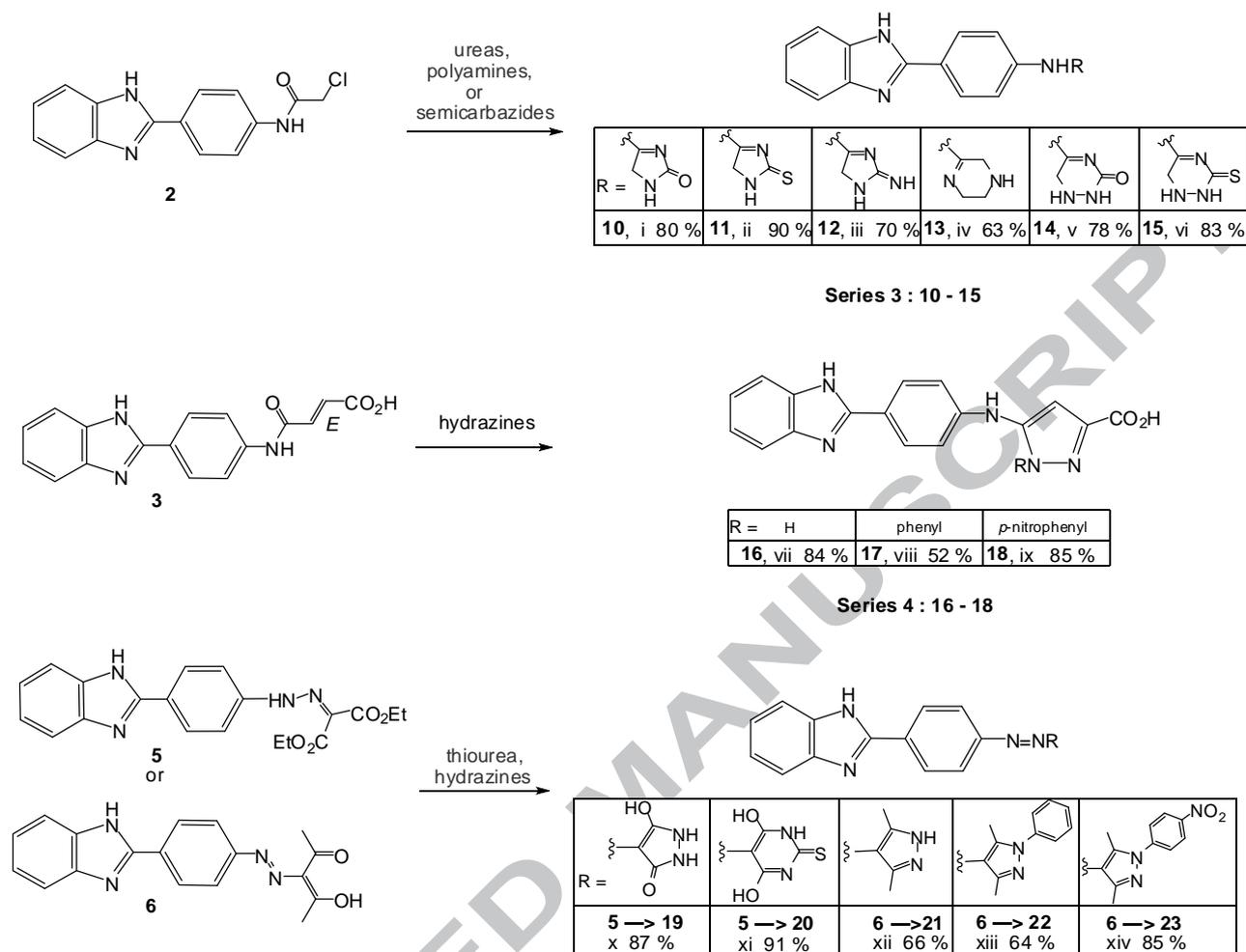
N-Sulfonyl derivatives **7-9** were synthesized by reacting compound **1** with benzene sulfonyl chloride, 4-tosyl chloride, 2-nitrobenzene sulfonyl chloride in the presence of Et_3N (Scheme 2).



Conditions: i- **1**, benzenesulfonyl chloride, acetone, Et₃N, 2 h; ii- **1**, 4-toluenesulfonyl chloride, acetone, Et₃N, 2 h; iii- **1**, 2-nitrobenzenesulfonyl chloride, acetone, Et₃N, 2 h.

Scheme 2: Synthetic routes to compounds **7-9** (Series 2)

The previously obtained derivatives **2**, **3**, **5** and **6** were subjected to nucleophilic reactions with ureas, polyamines, semicarbazides or hydrazines (Scheme 3) to afford compounds **10 - 23** (Series 3 – 5). The dihydroimidazoline derivatives **10-12** were prepared by cyclizing the chloroacetyl side chain of compound **2** with urea, thiourea or guanidine hydrochloride, respectively in the presence of anhydrous K₂CO₃. Reaction of **2**, under the same conditions, with ethylenediamine yielded the tetrahydropyrazine derivative **13**. Cyclization of the side chain in **2** using semicarbazide hydrochloride or thiosemicarbazide afforded compounds **14** and **15**, respectively. Reaction of **3** with hydrazine hydrate, phenyl hydrazine and 4-nitrophenyl hydrazine yielded compounds **16-18**.⁵³ Compound **5** was reacted with hydrazine hydrate to form compound 4-(2-(4-(1*H*-benzo[*d*]imidazol-2-yl)phenyl)hydrazono)pyrazolidine-3,5-dione (**19**). Reaction of compound **5** with thiourea yielded the pyrimidine derivative **20**. On the other hand, compound **6** was reacted with different hydrazines as hydrazine hydrate,⁵⁴ phenylhydrazine and 4-nitrophenylhydrazine to produce the pyrazoles **21**,⁵⁴ **22** and **23**, respectively.

**Series 5 : 19 - 23**

Conditions: i- **2**, urea, K_2CO_3 , DMF, reflux, 48 h; ii- **2**, thiourea, K_2CO_3 , DMF, reflux, 48 h; iii- **2**, guanidine hydrochloride, K_2CO_3 , DMF, reflux, 48 h; iv- **2**, ethylenediamine, K_2CO_3 , DMF, reflux, 48 h; v- **2**, semicarbazide hydrochloride, K_2CO_3 , DMF, reflux, 48 h; vi- **2**, thiosemicarbazide, K_2CO_3 , DMF, reflux, 48 h; vii- **3**, hydrazine hydrate, EtOH, reflux, 3 h; viii- **3**, phenylhydrazine, K_2CO_3 , DMF, reflux, 8 h; ix- **3**, 4-nitrophenylhydrazine, K_2CO_3 , DMF, reflux, 8 h; x- **5**, hydrazine hydrate 98%, EtOH, reflux, 3h; xi- **5**, thiourea, EtONa/EtOH, reflux, 26h; xii- **6**, hydrazine hydrate 98%, EtOH, reflux, 3h; xiii- **6**, phenylhydrazine, K_2CO_3 , DMF, reflux, 8h; xiv- **6**, 4-nitrophenyl hydrazine, K_2CO_3 , DMF, reflux, 8h.

Scheme 3: Synthetic routes to compounds **10-23** (Series 3-5)

2.2. Kinetic evaluation of the synthesized (benzimidazol-2-yl)aniline derivatives as RMGPb inhibitors.

The synthesized compounds were assayed for their inhibition potency against RMGPb, in the direction of glycogen synthesis. All compounds exhibited very poor solubility in water

therefore, they were initially in 100% DMSO and serial dilutions were then prepared to 10 or 20 % DMSO (1 to 2 % in the reaction). Compounds **2**, **4-6**, **7-9**, **15** were partially soluble in aq. DMSO (20%), while compound **16** was insoluble in DMSO and hence could not be tested. No inhibition was observed for compounds **1**, **2**, **5**, **6**, and **13**, while compounds **3**, **4**, **8**, **17-20** were only weak inhibitors (Table 2). Moderate inhibition was shown by compounds **10 - 12**, **14-15**, **21-23** that exhibited IC₅₀ values in range of 400-600 μ M. All these modest inhibitors belonged to series 3 and 5, which displayed a lateral heterocycle. The arylsulfonyl derivatives of **1**, **7** (Ar: phenyl) and **9** (Ar: *o*-nitrophenyl) exhibited the strongest inhibition with IC₅₀ of 324 μ M and 357 μ M, respectively. Considering that the solubility of the compounds was poor and the inhibitor solutions were saturated it could be assumed that the IC₅₀ values were even less than the ones stated. With the aim to explain the kinetic results obtained, structural studies were performed with the most potent ligands by soaking native preformed crystals of RMGPb in inhibitor solution. No binding was detected, though. Cocrystallization studies of (benzimidazol-2-yl)aniline derivatives are under investigation to shed light on their inhibitory effect and binding mode. Introduction of more polar substituents might have enhanced affinity and potency of these compounds.

Table 2: Kinetic data for compounds 1-23 tested as RMGPb inhibitors.

Compound	Inhibition (%) or IC ₅₀ (μ M)	Compound	Inhibition (%) or IC ₅₀ (μ M)	Compound	Inhibition (%) or IC ₅₀ (μ M)
Series 1		Series 3		Series 5	
1	NI	10	397.0 \pm 91.0	19	29.4 % inhibition at 500 μ M
2	NI	11	531.2 \pm 89.6	20	16.1 % inhibition at 200 μ M
3	45.0 % inhibition at 1000 μ M	12	515.9 \pm 16.9	21	621.9 \pm 99.3
4	7.0 % inhibition at 200 μ M	13	NI	22	402.2 \pm 19.7
5	NI	14	437.0 \pm 10.8	23	532.3 \pm 11.9
6	NI	15	508.3 \pm 61.0		
Series 2		Series 4			
7	324.2 \pm 47.9	16	-		
8	20.7 % inhibition at 500 μ M	17	25.0 % inhibition		

			at 300 μ M		
9	357.5 \pm 63.5	18	8.1 % inhibition at 250 μ M		
NI: No Inhibition					

2.3. Molecular physicochemical properties

The molecular physicochemical properties (ALOGPS 2.1) of arylsulfonyl derivatives **7** and **9** were evaluated to study how these compounds meet Lipinski rules (<http://www.scfbio-iitd.res.in/software/drugdesign/lipinski.jsp>). Table 3 displays the relevant parameters (molecular mass, hydrogen bond donor, hydrogen bond acceptor, logP, molar refractivity) calculated, and the acceptable range for applicability.

Compound	Molecular mass	H-bond donor	H-bond acceptor	logP	Molar refractivity
7	349	2	4	4.9	98.1
9	394	2	6	5.0	105.5
Applicability	< 500	< 5	< 10	< 5	40 - 130

It appears that both compounds comply with the Lipinsky rules, with a slight advantage for compound **7** (logP = 4.9), compared to **9**. Therefore, both compounds are suitable candidates for further developments aiming at identifying more potent glycogen phosphorylase inhibitors.

3. Conclusion

A series of (benzimidazol-2-yl)-aniline derivatives was investigated as potential RMGPb inhibitors, a thoroughly investigated target employed for the development of new antidiabetic pharmacological agents. Compounds **2-23** have been classified into five different series according to their lateral groups. (Benzimidazol-2-yl)-anilines with polyazacycles as in series 3 and 5 exhibited modest effect on RMGPb activity with IC₅₀ in the 400-600 μ M range. Reaction of **1** with arylsulfonyl chlorides produced (benzimidazol-2-yl)-benzene benzenesulfonamide derivatives **7-9**,

series 2, which guide us to compounds of the highest inhibitory effect among the studied derivatives with IC_{50} around 350 μM and could further be exploited as lead molecules for the design of specific inhibitors with increased affinity for the target.

4. Material and methods

4.1. Chemistry

Microanalyses and spectral data of the compounds were performed in National Research Centre, Cairo, Egypt. The IR spectra (4000-400 cm^{-1}) were recorded using KBr pellets in a Jasco FT/IR 300E Fourier transform infrared spectrophotometer on a Perkin Elmer FT-IR 1650 spectrophotometer. The $^1\text{H-NMR}$ spectra were recorded using 500 MHz NMR spectrometer. Chemical shifts are reported in parts per million (ppm) from the tetramethylsilane resonance in the indicated solvent. Coupling constants are reported in Hertz (Hz); spectral splitting patterns are designed as follow: singlet (s); doublet (d); triplet (t); multiplet (m). Column chromatography was performed on Merck silica gel 60 (200–400 mesh). The mass spectra were recorded using a Finnigan mat SSQ 7000 (Thermo. Inst. Sys. Inc., USA) spectrometer at 70 eV. All chemicals and solvents were purchased from Sigma Chemical Company. All chemicals used were of analytical grade. The petroleum ether had a boiling temperature in the 60-80 $^{\circ}\text{C}$ range.

4.1.1. 4-(4-(1H-Benzo[d]imidazol-2-yl)phenylimino)pentan-2-one (4).

To a solution of compound **1** (1.67 g, 8 mmol) in glacial acetic acid (10 mL), acetylacetone (0.78 mL, 7.6 mmol) was added. The reaction mixture was refluxed for 10 h. The excess solvent was evaporated under reduced pressure. The formed solid was washed with ethyl acetate and recrystallized from acetone (82% yield). M.p.: 302-304 $^{\circ}\text{C}$, $R_f = 0.52$ (ethyl acetate/pet. ether, 3:1). IR (KBr) $\nu_{\text{max}}/\text{cm}^{-1}$: 3430 (NH benzimidazole); 3067 (CH arom); 2992, 2852 (CH aliph); 1674 (C=O); 1626, 1596 (C=N (s)); 1546 (C=C arom). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 2.05 (s, 3H, $\text{CH}_3\text{-C=N}$); 2.41 (s, 3H, $\text{CH}_3\text{-C=O}$); 3.53 (s, 2H, CH_2); 7.14 (m, 2H, H_5 , H_6 benzimidazole); 7.56 (m, 2H, H_4 , H_7 benzimidazole); 7.70 (d, 2H, $J = 8.4$ Hz, H_2 , H_6 aminophenyl); 8.05 (d, 2H, $J = 8.4$ Hz, H_3 , H_5 aminophenyl); 12.84 (br., 1H, NH benzimidazole, D_2O exchangeable). ^{13}C NMR (125MHz, DMSO- d_6): 24.64, 31.02 and 45.46 (aliph. carbons), 115.08, 119.52, 122.46, 125.22, 127.65, 133.67, 135.21, 141.33, 151.77 (Ar-C), 152.87 (C=N), 169.22 (C=O). MS, m/z (%): 292 ($\text{M}^+ + 1$, 4%); 291 (M^+ , 20%); 248 (54%); 233 (100%); 209 (81%). Anal. Calcd for $\text{C}_{18}\text{H}_{17}\text{N}_3\text{O}$ (FW: 291.14): C, 74.20; H, 5.88; N, 14.42. Found: C, 74.28; H, 5.81; N, 14.37.

4.1.2. Diethyl 2-(2-(4-(1H-benzo[d]imidazol-2-yl)phenyl)hydrazono)malonate (5).

A cold solution of compound **1** (8.36 g, 0.04 mol) in 20 mL 6*N* HCl was prepared. In ice bath, a cold solution of NaNO₂ (2.21 g, 0.032 mol) in the minimum quantity of cold water was added portionwise with continuous stirring. The resultant diazonium salt was added to a solution of diethyl malonate (6.16 mL, 0.06 mol) in ethanol. The pH of the mixture was adjusted to 6.5 using sodium acetate solution. The yellow precipitate was collected by vacuum filtration, dried (86% yield) and recrystallized from ethyl acetate. M.p.: 230-232 °C, *R_f* = 0.68 (ethyl acetate/pet. ether, 1:1), IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3392 (NH benzimidazole); 3332 (NH aminophenyl); 3099, 3038 (CH arom); 2979, 2931 (CH aliph); 1666 (C=O); 1610 (C=N); 1526 (C=C arom). ¹H-NMR (DMSO-*d*₆, 500 MHz, δ ppm): 1.26 (m, 6H, 2 CH₃); 4.23 (q, 2H, CH₂); 4.30 (q, 2H, CH₂); 7.16 (d, 2H, *J* = 8.4 Hz, H₂, H₆ aminophenyl); 7.51 (m, 4H, H₄, H₅, H₆, H₇ benzimidazole); 8.14 (d, 2H, *J* = 8.4 Hz, H₃, H₅ diazenylphenyl); 12.00 (s, 1H, NH aminophenyl, D₂O exchangeable); 12.92 (br., 1H, NH benzimidazole, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 14.39, 14.63 (aliph. CH₃), 61.42, 61.90 (aliph. CH₂), 111.73, 115.91, 119.10, 122.16, 122.85, 123.16, 125.94, 128.27, 135.54, 143.95, 144.38 (Ar-C), 151.54 (C=N), 162.18, 162.87 (C=O). MS, *m/z* (%): 380 (M⁺, 46%); 335 (13%); 306 (17%); 261 (12%); 207 (100%). Anal. Calcd for C₂₀H₂₀N₄O₄ (FW: 380.15): C, 63.15; H, 5.30; N, 14.73. Found: C, 63.17; H, 5.36; N, 14.71.

4.1.3. General procedure for the preparation of compounds 7-9.

To a well-stirred solution of compound **1** (10 mmol) and triethylamine (0.5 mL) in acetone, benzenesulfonyl chloride, 4-toluenesulfonyl chloride or 2-nitrobenzenesulfonyl chloride (10 mmol) was added dropwise. The reaction mixture was stirred for 2h at room temperature and left overnight. The solvent was evaporated under reduced pressure. The solid was collected, washed with water, dried. Purification by column chromatography was achieved using ethyl acetate/pet. ether (3:1 ratio) as the mobile phase.

4.1.3.1. *N*-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)benzenesulfonamide (7).

Yield: 76%. M.p. 238°-240 C. TLC *R_f* = 0.69 (ethyl acetate/petroleum ether, 2:1). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3359, 3230 (NHs); 3096, 3016, 2985 (CH arom); 1611 (C=N); 1588 (C=C arom); 1495 (ν_{as} SO₂), 1476 (ν_{s} SO₂). ¹H-NMR (DMSO-*d*₆, 500 MHz, δ ppm): 7.173(m, 2H, H₃, H₅ benzenesulfonyl moiety); 7.27 (d, 2H, *J* = 7.65 Hz, H₂, H₆ aminophenyl moiety); 7.523-7.832 (m, 3H, H₄ benzenesulfonyl moiety, H₅, H₆ benzimidazole moiety); 7.836-7.853 (d, 2H, H₄, H₇ benzimidazole moiety); 8.022-8.044 (d, 2H, *J* = 7.65 Hz, H₃, H₅ aminophenyl moiety); 10.65(s, 1H, NH sulphonamide, D₂O exchangeable), 12.76 (s, 1H, NH benzimidazole, D₂O exchangeable) .

MS, m/z (%): 349(M⁺, 15%); 97 (100%). ¹³C-NMR (DMSO-*d*₆, 125 MHz, δ ppm): 112.11, 119.25, 119.77, 120.11, 122.47, 126.23, 126.86, 127.18, 128.03, 129.55, 129.81, 134.65, 135.31, 135.58, 138.93, 139.68, 139.91, 151.31. Anal. Calcd for C₁₉H₁₅N₃O₂S (FW: 349): C, 65.31; H, 4.33; N, 12.03; S, 9.18. Found: C, 65.66; H, 4.47; N, 12.18; S, 9.31.

4.1.3.2. *N*-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)-4-methylbenzenesulfonamide (8)

Yield: 82%. M.p.: 230-232 °C, R_f = 0.60 (ethylacetate/pet. ether, 2:1). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3373 (NH aminophenyl); 3061 (CH arom); 2922 (CH aliph); 1610 (C=N); 1596 (C=C arom); 1438 (ν_{as} SO₂), 1375 (ν_{s} SO₂). ¹H-NMR (DMSO-*d*₆, 500 MHz, δ ppm): 2.30 (s, 3H, CH₃); 7.07 (d, 2H, *J* = 9.2 Hz, H₂, H₆ aminophenyl); 7.34 (m, 2H, H₃ and H₅ tosyl); 7.43 (m, 2H, H₂ and H₆ tosyl); 7.71 (m, 2H, H₅, H₆ benzimidazole); 7.84 (m, 2H, H₄, H₇ benzimidazole); 8.01 (d, 2H, *J* = 9.2 Hz, H₃, H₅ aminophenyl); 10.91 (s, 1H, NH sulphonamide, D₂O exchangeable). 12.75, (s, 1H, NH, benzimidazole, D₂O exchangeable). MS, m/z (%): 363 (M⁺, 14%); 347 (50%); 208 (100%). Anal. Calcd for C₂₀H₁₇N₃O₂S (FW: 363): C, 66.10; H, 4.71; N, 11.56; S, 8.82. Found: C, 66.42; H, 4.95; N, 11.14; S, 8.33.

4.1.3.3. *N*-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)-4-nitrobenzenesulfonamide (9).

Yield: 78%. M.p.: 215-217 °C, R_f = 0.42 (EtAc/Pet. ether 2:1). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3374 (NH aminophenyl); 3029, 2974 (CH arom); 1611 (C=N); 1567 (C=C arom); 1513 (ν_{as} NO₂); 1466 (ν_{as} SO₂); 1437 (ν_{s} NO₂); 1394 (ν_{s} SO₂). ¹H-NMR (DMSO-*d*₆, 500 MHz, δ ppm): 7.39 (d, 2H, *J* = 8.4 Hz, H₂, H₆ aminophenyl moiety); 7.47 (m, 2H, H₅, H₆ benzimidazole moiety); 7.53 (m, 1H, H₄ nitrophenyl); 7.73 (m, 2H, H₄, H₇ benzimidazole moiety); 7.86 (m, 2H, H₅ and H₆ nitrophenyl moiety); 8.00 (m, 1H, H₃, nitrophenyl moiety); 8.06 (d, 2H, *J* = 8.4 Hz, H₃, H₅ aminophenyl moiety); 11.43 (br., 1H, NH aminophenyl, exchangeable). MS, m/z (%): (M⁺ - 1, 60%); 271 (28%); 208 (100%). Anal. Calcd for C₁₉H₁₄N₄O₄S (FW: 394): C, 57.86; H, 3.58; N, 14.21; S, 8.13. Found: C, 57.44; H, 3.41; N, 14.09; S, 8.41.

4.1.4. General procedure for the preparation of compounds 10-15:

Compound **2** (1.99 g, 7 mmol) was added to a solution of urea, thiourea, guanidine hydrochloride, ethylenediamine, semicarbazide hydrochloride or thiosemicarbazide (7 mmol) and K₂CO₃ (7 mmol) in DMF (20 mL) with gentle stirring at r.t. for 1 h. Then, the reaction mixture was refluxed for appropriate time. The products formation was monitored by TLC. After reaction completion, the reaction mixture was poured onto crushed ice with continuous stirring. The formed solids were collected by vacuum filtration. The crude solid was purified by column chromatography using ethyl acetate/pet. ether (2:1) as eluent.

4.1.4.1. 5-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenylamino)-1*H*-imidazol-2(5*H*)-one (10).

Yield: 80%. M.p.: 222-224 °C, crystallized from DMF, $R_f = 0.57$ (ethyl acetate/pet. ether/ EtOH, 3:1:0.5). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3419 (NH benzimidazole); 3340, 3279 (NH(s)); 3014 (CH arom); 2924 (CH aliph); 1667 (C=O); 1625, 1602, 1578 (C=N(s)); 1543 (C=C arom). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 5.46 (s, 2H, CH₂ imidazolone); 7.09 (s, 1H, NH, D₂O exchangeable); 7.50 (m, 2H, H₅, H₆ benzimidazole); 7.87 (m, 4H, H₄, H₇ benzimidazole and H₂, H₆ aminophenyl moieties); 8.29 (d, 2H, $J = 7.65$, H₃, H₅ aminophenyl); 11.41(br., NH, D₂O exchangeable) 12.51 (br., NH benzimidazole, D₂O exchangeable). MS, m/z (%): 291 (M^+ , 43%); 234 (73%); 209 (100%). Anal. Calcd for C₁₆H₁₃N₅O (FW: 291.11): C, 65.97; H, 4.50; N, 24.04. Found: C, 65.89; H, 4.44; N, 24.10.

4.1.4.2. 5-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenylamino)-1*H*-imidazole-2(5*H*)-thione (11).

90% Yield; M.p.: 308-310 °C, crystallized from DMF, $R_f = 0.6$ (ethyl acetate/pet. ether/ EtOH, 3:1:0.5). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3360 (NH benzimidazole); 3274, 3210 (NH (s)); 3053 (CH arom); 2888 (CH aliph); 1627, 1601 (C=N(s)); 1543 (C=C arom). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 3.60 (s, 2H, H₂ imidazolethione); 7.07 (s, 1H, NH, D₂O exchangeable); 7.22 (m, 2H, H₅, H₆ benzimidazole); 7.58 (m, 4H, H₄, H₇ benzimidazole and H₂, H₆ aminophenyl moieties); 8.11 (d, 2H, $J = 7.65$, H₃, H₅ aminophenyl); 10.36 (s, 1H, NH, D₂O exchangeable); 11.55 (br., 1H, NH benzimidazole, D₂O exchangeable). MS, m/z (%): 309 ($M+2$, 2%), 307 (M^+ , 59%); 275 (33%); 234 (76%); 209 (100%). Anal. Calcd for C₁₆H₁₃N₅S (FW: 307.09): C, 62.52; H, 4.26; N, 22.78; S, 10.43. Found: C, 62.57; H, 4.30; N, 22.75; S, 10.49.

4.1.4.3. N^5 -(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)-4*H*-imidazole-2,5-diamine (12).

70% Yield; M.p.: 228-231 °C, crystallized from DMF, $R_f = 0.64$ (ethyl acetate/pet. ether/ EtOH, 3:1:0.5). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3431 (NH benzimidazole); 3370, 3308, 3220 (NH(s) and NH₂ of enamine form); 3060 (CH arom); 2859 (CH aliph); 1620, 1604, 1589 (C=N(s)); 1541 (C=C arom). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 5.11 (s, 2H, H₂ imidazole amine); 5.57 (br., NH₂, D₂O exchangeable); 7.07 (m, 2H H₅, H₆ benzimidazole); 7.15 (m, 2H, H₄, H₇ benzimidazole), 7.53 (d, 2H, $J = 7.65$, H₂, H₆ aminophenyl); 8.11 (d, 2H, $J = 7.65$, H₃, H₅ aminophenyl); 9.95 (s, 1H, NH, D₂O exchangeable); 10.75 (s, 1H, NH, D₂O exchangeable); 12.81 (br., 1H, NH benzimidazole, D₂O exchangeable). MS, m/z (%): 290 (M^+ , 16%); 234 (47%); 210 (36%); 209 (100%). Anal. Calcd for C₁₆H₁₄N₆ (FW: 290.13): C, 66.19; H, 4.86; N, 28.95. Found: C, 66.23; H, 4.81; N, 28.90.

4.1.4.4. *N*-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)-3,4,5,6-tetrahydropyrazin-2-amine (13).

63% Yield; M.p.: 316-318 °C, crystallized from DMF, $R_f = 0.64$ (ethyl acetate/pet. ether/ EtOH, 3:1:0.5). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3445 (NH benzimidazole); 3312, 3207; (NH(s)); 3055 (CH arom); 2866 (CH aliph); 1628, 1604 (C=N(s)); 1506 (C=C arom.). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 3.41 (t, 2H, CH₂ pyrazinyl); 3.47 (t, 2H, CH₂ pyrazinyl); 3.54 (s, 2H, CH₂ pyrazinyl); 6.74 (m, 2H, H₅, H₆ benzimidazole); 7.44 (d, 2H, $J = 7.65$, H₂, H₆ aminophenyl); 7.68 (d, 2H, $J = 7.65$, H₃, H₅ aminophenyl); 8.01 (m, 2H, H₄, H₇ benzimidazole); 8.31 (s, 1H, NH, D₂O exchangeable); 10.47 (s, 1H, NH, D₂O exchangeable); 12.89 (br., 1H, NH benzimidazole, D₂O exchangeable). MS, m/z (%): 291 (M^+ , 46%); 234 (23%); 210 (100%). Anal. Calcd for C₁₇H₁₇N₅ (FW: 291.15): C, 70.08; H, 5.88; N, 24.04. Found: C, 70.11; H, 5.85; N, 24.00.

4.1.4.5. 5-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenylamino)-1,2-dihydro-1,2,4-triazin-3(6*H*)-one (14).

78% Yield; M.p. 270-272 °C, crystallized from DMF, $R_f = 0.76$ (ethyl acetate/pet. ether/ EtOH, 3:1:0.5). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3400 (NH benzimidazole); 3315- 3280, (NH(s)); 3059 (CH arom); 2928 (CH aliph); 1674 (C=O); 1626, 1606, 1587 (C=N(s)); 1537 (C=C arom.). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 5.55 (s, 2H, CH₂ triazinone); 6.81 (s, 1H, NH, D₂O exchangeable); 7.54 (m, 2H, H₅, H₆ benzimidazole); 7.90 (m, 4H, H₄, H₇ benzimidazole and H₂, H₆ aminophenyl moieties); 8.39 (d, 2H, $J = 7.65$, H₃, H₅ aminophenyl); 10.25 (s, 1H, NH, D₂O exchangeable); 11.01 (br., NH benzimidazole, D₂O exchangeable); 11.77 (s, 1H, NH, D₂O exchangeable). MS, m/z (%): 306 (M^+ , 36%); 234 (59%); 209 (100%). Anal. Calcd for C₁₆H₁₄N₆O (FW: 306.12): C, 62.74; H, 4.61; N, 27.44. Found: C, 62.77; H, 4.65; N, 27.42.

4.1.4.6. 5-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenylamino)-1,2-dihydro-1,2,4-triazin-3(6*H*)-thione (15).

83% Yield; M.p.: 202-204 °C crystallized from DMF, $R_f = 0.73$ (ethylacetate/pet. ether/ EtOH, 3:1:0.5). IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3429 (NH benzimidazole); 3340, 3260 (NH(s)); 3090, 2923 (CH arom); 2857 (CH aliph); 1627; 1609, 1576 (C=N(s)); 1552 (C=C arom.). $^1\text{H-NMR}$ (DMSO- d_6 , 500 MHz, δ ppm): 3.54 (s, 2H, CH₂ triazinethione); 7.15 (m, 2H, H₅, H₆ benzimidazole); 7.47 (m, 1H, H₄ benzimidazole); 7.59 (m, 1H, H₇ benzimidazole); 7.73 (d, 2H, $J = 7.65$, H₂, H₆ aminophenyl); 8.09 (d, 2H, $J = 7.65$, H₃, H₅ aminophenyl); 10.38 (s, 1H, NH, D₂O exchangeable); 10.61 (s, 1H, NH, D₂O exchangeable); 12.78 (br., 1H, NH benzimidazole, D₂O exchangeable). MS, m/z (%): 324 ($M^+ + 2$, 4%); 322 (M^+ , 89%); 290 (38%); 234 (69%); 209 (100%). Anal. Calcd for C₁₆H₁₄N₆S (FW: 322.10): C, 59.61; H, 4.38; N, 26.07; S, 9.95. Found: C, 59.56; H, 4.33; N, 26.12; S, 9.90.

4.1.4.7. 4-(2-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)diazenyl)-1,2-dihydro-3-hydroxypyrazol-5-one (19).

A mixture of compound **5** (1.33 g, 3.5 mmol) and hydrazine hydrate (0.49 mL, 10 mmol) in ethanol (40 mL) was refluxed for 3h. The excess solvent was evaporated under reduced pressure and then, the reaction mixture was poured onto crushed ice. The formed solid was collected by vacuum filtration, dried and recrystallized (87% yield) from EtOH. M.p.: 318-320 °C, $R_f = 0.66$ (CHCl₃/MeOH, 3:0.5), IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3650-2400 (H-bonded OH & NH); 3404 (NH benzimidazole); 3263, 3220 (NH pyrazole); 3060 (CH arom); 1656 (C=O); 1609 (C=N); 1536 (C=C arom). ¹H-NMR (DMSO-*d*₆, 500 MHz, δ ppm): 7.17 (m, 2H, H₅, H₆ benzimidazole); 7.54 (m, 2H, H₄, H₇ benzimidazole); 7.64 (d, 1H, $J = 8.4$ Hz, H₂ diazenylphenyl); 7.68 (d, 1H, $J = 8.4$ Hz, H₆ diazenylphenyl); 8.10 (d, 1H, $J = 8.4$ Hz, H₃ diazenylphenyl); 8.18 (d, 1H, $J = 8.4$ Hz, H₅ diazenylphenyl); 9.72 (s, 1H, NH pyrazole, D₂O exchangeable); 10.57 (br., 1H, NH pyrazole, D₂O exchangeable); 13.18 (br., 1H, NH benzimidazole, D₂O exchangeable); 13.93 (s, 1H, OH, D₂O exchangeable). MS, m/z (%): 320 (M⁺, 15%); 221 (19%); 208 (100%); 193 (48%); 117 (14%). Anal. Calcd for C₁₆H₁₂N₆O₂ (FW: 320.10): C, 60.00; H, 3.78; N, 26.24. Found: C, 60.04; H, 3.83; N, 26.23.

4.1.4.8. 5-(2-(4-(1*H*-Benzo[*d*]imidazol-2-yl)phenyl)diazenyl)-4,6-dihydroxypyrimidine-2(1*H*)-thione (20).

To a freshly prepared 0.64 molar solution of sodium ethoxide, thiourea (4 mmol) was added, and then the reaction was stirred at r.t. for one hour. Compound **5** (1.33 g, 3.5 mmol) was added, and then the reaction mixture was refluxed for 16h. The reaction mixture was left to cool, poured onto crushed ice with continuous stirring and the pH was adjusted to (7). The formed solid was collected by vacuum filtration, dried and recrystallized (91% yield) from ethanol. M.p.: 260-262 °C, $R_f = 0.36$ (ethyl acetate/pet. ether/C₂H₅OH, 1:1:0.5), IR (KBr) $\nu_{\max}/\text{cm}^{-1}$: 3427 (OH); 3278 (NH benzimidazole); 3211 (NH aminophenyl); 3044, 2994 (CH arom); 1606 (C=N); 1505 (C=C arom). ¹H-NMR (DMSO-*d*₆, 500 MHz, δ ppm): 7.14 (m, 2H, H₅, H₆ benzimidazole); 7.22 (d, 2H, $J = 8.4$ Hz, H₂, H₆ aminophenyl); 7.52 (m, 2H, H₄, H₇ benzimidazole); 8.06 (d, 2H, $J = 8.4$ Hz, H₃, H₅ aminophenyl); 9.48, (s, 1H, NH, D₂O exchangeable), 11.39 (br., 1H, NH benzimidazole, D₂O exchangeable); 12.47, 12.65 (s, 1H, OH, D₂O exchangeable). MS, m/z (%): 366 (M⁺+2, 6%); 364 (M⁺, 57%); 208 (100%); 193 (37%); 156 (24%). Anal. Calcd for C₁₇H₁₂N₆O₂S (FW: 364.07): C, 56.04; H, 3.32; N, 23.06; S, 8.80. Found: C, 56.09; H, 3.38; N, 23.11; S, 8.77.

4.1.5. General procedure for the preparation of compounds 22 and 23.

Compound **6** (1.12 g, 3.5 mmol) was added to a mixture of phenylhydrazine or 4-nitrophenyl hydrazine (3.5 mmol) and anh. K_2CO_3 (0.48 g, 3.5 mmol) in 10 mL DMF and the reaction mixture was refluxed for 8h. The excess solvent was evaporated under reduced pressure. The reaction mixture was poured onto ice. The formed solid was collected by vacuum filtration, dried and recrystallized from EtOH.

4.1.5.1. 2-(4-((3,5-Dimethyl-1-phenyl-1H-pyrazol-4-yl)diazenyl)phenyl)-1H-benzo[d]imidazole (22).

64% Yield; M.p.: 224-226 °C (EtOH), $R_f = 0.62$ (ethyl acetate/pet. ether/MeOH, 3:1:0.5), IR (KBr) ν_{max}/cm^{-1} : 3419 (NH benzimidazole); 3154, 2984 (CH arom); 2922 (CH aliph); 1612, 1595 (C=N); 1547 (C=C arom). 1H -NMR (DMSO- d_6 , 500 MHz, δ ppm): 2.48 (s, 3H, CH_3); 2.63 (s, 3H, CH_3); 7.22 (m, 2H, H_5 , H_6 benzimidazole); 7.48 (m, 1H, H_{4a} phenyl); 7.55 (m, 2H, H_4 , H_7 benzimidazole); 7.59 (m, 4H, H_{2a} , H_{3a} , H_{5a} , H_{6a} phenyl); 7.93 (d, 2H, $J = 7.6$ Hz, H_2 , H_6 diazenylphenyl); 8.31 (d, 2H, $J = 7.6$ Hz, H_3 , H_5 diazenylphenyl); 12.37 (br., 1H, NH benzimidazole, D_2O exchangeable). MS, m/z (%): 393 ($M^+ + 1$, 93%); 285 (37%); 208 (82%). Anal. Calcd for $C_{24}H_{20}N_6$ (FW: 392.17): C, 73.45; H, 5.14; N, 21.41. Found: C, 73.50; H, 5.19; N, 21.44.

4.1.5.2. 2-(4-((3,5-Dimethyl-1-(4-nitrophenyl)-1H-pyrazol-4-yl)diazenyl)phenyl)-1H-benzo[d]imidazole (23).

85% Yield; M.p.: 270-272 °C (EtOH), $R_f = 0.58$ (ethyl acetate/pet. ether/MeOH, 3:1:0.5), IR (KBr) ν_{max}/cm^{-1} : 3412 (NH benzimidazole); 3048 (CH arom); 2923 (CH aliph); 1606 (C=N); 1594, 1563 (C=C arom), 1509 (ν_{as} NO_2), 1417 (ν_s NO_2). 1H -NMR (DMSO- d_6 , 500 MHz, δ ppm): 2.40 (s, 3H, CH_3); 2.51 (s, 3H, CH_3); 7.19 (m, 2H, H_5 , H_6 benzimidazole); 7.51 (d, 2H, $J = 8.4$ Hz, H_2 , H_6 diazenylphenyl) 7.65 (m, 2H, H_4 , H_7 benzimidazole); 7.79 (d, 2H, $J = 8.4$ Hz, H_3 , H_5 diazenylphenyl); 7.85 (d, 2H, $J = 7.6$ Hz, H_{2a} , H_{6a} nitrophenyl); 8.28 (d, 2H, $J = 7.6$ Hz, H_{3a} , H_{5a} nitrophenyl); 12.89 (br., 1H, NH benzimidazole, D_2O exchangeable). MS, m/z (%): 438 ($M^+ + 1$, 24%); 361 (17%); 285 (48%); 210 (73%). Anal. Calcd for $C_{24}H_{19}N_7O_2$ (FW: 437.16): C, 65.89; H, 4.38; N, 22.41. Found: C, 65.96; H, 4.43; N, 22.40.

4.2. Kinetic evaluation of compound 1 – 23

RMGPb was isolated and purified with successive crystallization and recrystallization steps from rabbit skeletal muscle as described previously.⁵⁵ Evaluation of the inhibitory potency of the compounds on RMGPb was performed in the direction of glycogen synthesis at 30 °C, pH 6.8 in

the presence of 5 $\mu\text{g/mL}$ enzyme, 2 mM glucose-1-phosphate, 1 mM AMP and 1 % glycogen. The compounds tested exhibited very poor solubility in water, therefore they were dissolved in 100 % DMSO prior to kinetic experiments. Dilutions were prepared and their concentration in the reaction ranged from 5 μM to 1 mM in the presence of 1-2% DMSO. Enzyme activity was measured by the release of inorganic phosphate.^{56, 57}

Acknowledgments

Support from CNRS, University Claude-Bernard Lyon 1 and the French Agence Nationale de Recherche (project GPdia N° ANR-08-BLAN-0305) are gratefully acknowledged. The groups of the National Research Centre, Cairo, Egypt, are greatly thankful for the support of the Science and Technology Development Fund through the project STDF ID 1517. This work was supported by the FP7 Capacities coordination and support actions REGPOT-2009-1-No 245866 'ARCADE'.

References

1. Gribble, G. W. Heterocyclic Scaffolds II: Reactions and Applications of Indoles, Topics in Heterocyclic Chemistry; Springer: Science & Business Media, **2010**; Vol. 26.
2. Quaranta, L. Benzimidazole Fungicides: Bioactive Heterocyclic Compound Classes; Agrochemicals, **2012**, pp.103- 118.
3. Dattatri, Y.; Harika, R.; Meher, C. P.; Divyajyothi, P.; Sangeetha, R.; Santosh, P. *Asian J. Res. Chem.* **2013**, *6*, 588.
4. Minutolo, F.; Macchia, M.; Granchi, C.; Di Bussolo, V.; Giannaccini, G.; Lucacchini, A.; Hergenrother, P. J.; Calvaresi, E. C. W.O. Patent 2013,092,753 A1, **2013**.
5. Katritzky, A. R.; Dobchev, D. A.; Fara, D. C.; Karelson, M. *Bioorg. Med. Chem.* **2005**, *13*, 6598.
6. Velaparthy, U.; Liu, P.; Balasubramanian, B.; Carboni, J.; Attar, R.; Gottardis, M.; Li, A.; Greer, A.; Zoeckler, M.; Wittman, M. D.; Vyas, D. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 3072.
7. De Palma, A. M.; Heggermont, W. *Biochem. Biophys. Res. Commun.* **2007**, *353*, 628.
8. El-Seidy, A. M. A.; El-Zahany, E.; Barakat, A. S.; Youssef, N. S.; Galal, S. A.; Drweesh, S. A. *Synth. React. Inorg. Met.-Org. Nano-Metal Chem.* **2013**, *43*, 46.
9. Galal, S. A.; Abd El-All, A. S.; Abdallah, M. M.; El-Diwani, H. I. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 2420.
10. Galal, S. A.; Abdelsamie, A. S.; Rodriguez, M. L.; Kerwin, S. M.; El Diwani, H. I. *Eur. J. Chem.* **2010**, *1*, 67.
11. Abdel-Mohsen, H. T.; Ragab, F. A.; Ramla, M. M.; El Diwani, H. I. *Eur. J. Med. Chem.* **2010**, *45*, 2336.

12. Omar, M. A.; Shaker, Y. M.; Galal, S. A.; Ali, M. M.; Kerwin, S. M.; Li, J.; Tokuda, H.; Ramadan, R. A.; El Diwani, H. I. *Bioorg. Med. Chem.* **2012**, *20*, 6989.
13. Temirak, A.; Shaker, Y. M.; Ragab, F. A. F.; Ali, M. M.; Ali, H. I.; El Diwani, H. I., *Eur. J. Med. Chem.* **2014**, *87*, 868.
14. Temirak, A.; Shaker, Y. M.; Ragab, F. A.; Ali, M. M.; Soliman, S. M.; Mortier, J.; Wolber, G.; Ali, H. I.; El Diwani, H. I. *Arch. Pharm. (Weinheim)* **2014**, *347*, 291.
15. El-Nezhawy, A. O.; El-Naem, S. I.; Galal, S. A.; El-Diwani, H. I.; Abdel Salam, O. M.; Baiomy, A. R. *Pharm. Chem. J.* **2009**, *43*, 25.
16. Galal, S. A.; El-Naem, S. I.; El-Nezhawy, A. O.; Ali, M. A.; El-Diwani, H. I. *Arch. Pharm. (Weinheim)* **2011**, *344*, 255.
17. Galal, S. A.; Hegab, K. H.; Hashem, A. M.; Youssef, N. S. *Eur. J. Med. Chem.* **2010**, *45*, 5685.
18. Jiang, J.; Kassick, A. J.; Kekec, A.; Sebhat, L. K.; Sebhat, I. K. W.O. Patent 2011,106,273, A1, **2011**.
19. Bao, J.; Lan, P.; Lu, H.; Makara, G. M.; Romero, F. A.; Sebhat, I.; Wodka, D.; Dang, Q.; Chung, D. M.; Gibson, T. S. U.S. Patent 8,329,914 (B2), **2012**.
20. Jones, R. M.; Thurston, D. E.; Rotella, D.; Guccione, S.; Martinez, A. New therapeutic strategies for type 2 diabetes: Small molecule approaches; R. Soc. Chem.: Drug Discovery, **2012**; Vol. 27.
21. Somsák, L.; Czifrák, K.; Tóth, M.; Bokor, É.; Chrysin, E. D.; Alexacou, K. M.; Hayes, J. M.; Tiraidis, C.; Lazoura, E.; Leonidas, D. D.; Zographos, S. E.; Oikonomakos, N. G. *Curr. Med. Chem.* **2008**, *15*, 2933.
22. Oikonomakos, N. G. *Curr. Protein Pept. Sci.* **2002**, *3*, 561.
23. Oikonomakos, N. G.; Somsák, L. *Curr. Opin. Investig. Drugs* **2008**, *9*, 379.
24. Agius, L. *Mini Rev. Med. Chem.* **2010**, *10*, 1175.
25. Aguis, L. *Mol. Aspects Med.* **2015**, *46*, 34.
26. Docsa, T.; Czifrák, K.; Hüse, C.; Somsák, L.; Gergely, P. *Mol. Med. Rep.* **2011**, *4*, 477.
27. Nagy, L.; Docsa, T.; Szántó, M.; Brunyánszki, A.; Hegedüs, C.; Márton, J.; Kónya, B.; Virág, L.; Somsák, L.; Gergely, P.; Bai, P. *PLoS One* **2013**, *8*, e69420.
28. Goyard, D.; Konya, B.; Chajistamatiou, A. S.; Chrysin, E. D.; Leroy, J.; Balzarín, S.; Tournier, M.; Tusch, D.; Petit, P.; Duret, C.; Maurel, P.; Somsák, L.; Docsa, T.; Gergely, P.; Praly, J.-P.; Azay-Milhau, J.; Vidal, S. *Eur. J. Med. Chem.* **2016**, *108*, 444.
29. Docsa, T.; Marics, B.; Németh, J.; Hüse, C.; Somsák, L.; Gergely, P.; Peitl, B. *Curr. Top. Med. Chem.* **2015**, *15*, 2390.

30. Mathieu, C.; Li de la Sierra-Gallay, I.; Duval, R.; Xu, X.; Coccagn, A.; Léger, T.; Woffendin, G.; Camadro, J. M.; Etchebest, C.; Haouz, A.; Dupret, J. M.; Rodrigues-Lima, F. *J. Biol. Chem.*, "in press" <http://www.jbc.org/content/early/2016/07/08/jbc>. doi: 10.1074/jbc.M116.738898
31. Pettersen, E.F.; Goddard, T.D.; Huang, C.C.; Couch, G.S.; Greenblatt, D.M.; Meng, E.C.; Ferrin, T.E. *J. Comput. Chem.* **2004**, *25*, 1605.
32. Praly, J. -P.; Vidal, S. *Mini Rev. Med. Chem.* **2010**, *10*, 1102.
33. Gimisis, T. *Mini Rev. Med. Chem.* **2010**, *10*, 1127.
34. Somsák, L. *C.R. Chim.* **2011**, *14*, 211.
35. Chrysina, E. D.; Kosmopoulou, M. N.; Tiraidis, C.; Kardakaris, R.; Bischler, N.; Leonidas, D. D.; Hadady, Z.; Somsák, L.; Docsa, T.; Gergely, P. *Protein Sci.* **2005**, *14*, 873.
36. Kantsadi, A. L.; Apostolou, A.; Theofanous, S.; Stravodimos, G. A.; Kyriakis, E.; Gorgogietas, V. A.; Chatzileontiadou, D. S. M.; Pegiou, K.; Skamnaki, V. T.; Stagos, D. *Food Chem. Toxicol.* **2014**, *67*, 35.
37. Somsák, L.; Nagy, V.; Hadady, Z.; Docsa, T.; Gergely, P. *Curr. Pharm. Des.* **2003**, *9*, 1177.
38. Hadady, Z.; Tóth, M.; Somsák, L. *Arkivoc* **2004**, *7*, 140.
39. Chrysina, E. D. *Mini Rev. Med. Chem.* **2010**, *10*, 1093.
40. Oikonomakos, N. G.; Kosmopoulou, M.; Zographos, S. E.; Leonidas, D. D.; Chrysina, E. D.; Somsák, L.; Nagy, V.; Praly, J.-P.; Docsa, T.; Toth, B.; Gergely, P. *Eur. J. Biochem.* **2002**, *269*, 1684.
41. Martin, W. H.; Hoover, D. J.; Armento, S. J.; Stock, I. A.; McPherson, R. K.; Danley, D. E.; Stevenson, R. W.; Barrett, E. J.; Treadway, J. L. *Proc. Natl. Acad. Sci. U.S.A.* **1998**, *95*, 1776.
42. Hoover, D. J.; Lefkowitz-Snow, S.; Burgess-Henry, J.; Martin, W. H.; Armento, S. J.; Stock, I. A.; McPherson, R. K.; Genereux, P. E.; Gibbs, E. M.; Treadway, J. L. *J. Med. Chem.* **1998**, *41*, 2934.
43. Oikonomakos, N. G.; Skamnaki, V. T.; Tsitsanou, K. E.; Gavalas, N. G.; Johnson, L.N. *Structure* **2000**, *8*, 575.
44. Oikonomakos, N. G.; Chrysina, E. D.; Kosmopoulou, M. N.; Leonidas, D. D. *Biochim. Biophys. Acta* **2003**, *1647*, 325.
45. Loughlin, W. A. *Mini Rev. Med. Chem.* **2010**, *10*, 1139.
46. Schoenafinger, K.; Defossa, E.; Kadereit, D.; Von Roedern, E.; Klabunde, T.; Burger, H. J.; Herling, A.; Wendt, K. U. W.O. Patent 2,004,007,455, **2004**.
47. Barf, T. *Mini Rev. Med. Chem.* **2004**, *4*, 897.
48. Prapthi, B.; Lakshmi, K.; Vijaya, N. *Int. J. Basic Clin. Pharmacol.* **2013**, *2*, 814.

49. Hayes, J. M.; Leonidas, D. D. *Mini Rev. Med. Chem.* **2010**, *10*, 1156.
50. Gaboriaud-Kolar, N.; Skaltsounis, A.L. *Expert Opin. Ther. Pat.* **2013**, *23*, 1017.
51. Donnier-Maréchal, M.; Vidal, S. *Expert Opin. Ther. Pat.* **2016**, *26*, 199.
52. Shahare, M. B.; Kadam, V. J.; Jagdale, D. M.; Gandhi, P. S.; Gaikwad, P. L. *Intern. J. Res. Pharm. Chem.* **2012**, *2*, 132.
53. Khattab, M.; Ragab, F.; Galal, S.; Diwani, H. E. *Intern. J. Res. Pharm. Chem.* **2012**, *2*, 937.
54. Abdelgawad, M. A.; Kamel, G. M. *J. Appl. Sci. (Beni-Suef University)* **2012**, *1*, 80.
55. Oikonomakos, N. G.; Kontou, M.; Zographos, S. E.; Watson, K. A.; Johnson, L. N.; Bichard C. J. F.; Fleet G.W. J.; Acharya, K. R. *Protein Sci.* **1995**, *4*, 2469.
56. Fiske, C. H.; Subbarow, Y. *J. Biol. Chem.* **1925**, *66*, 375.
57. Saheki, S.; Takeda, A.; Shimazu, T. *Anal. Biochem.* **1985**, *148*, 277.

SYNTHESIS OF (BENZIMIDAZOL-2-YL)ANILINE DERIVATIVES AS GLYCOGEN PHOSPHORYLASE INHIBITORS

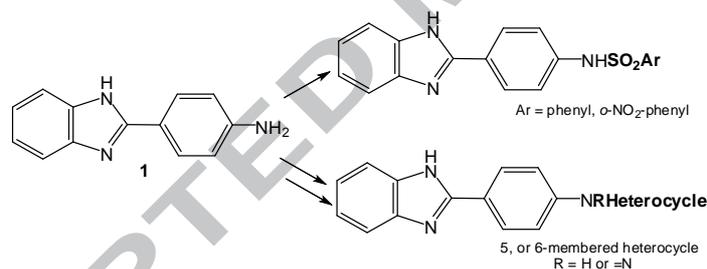
Shadia A. Galal,^{a*} Muhammad Khattab,^a Fotini Andreadaki,^b Evangelia D. Chrysina,^{b*} Jean-Pierre Praly,^{c*} Fatma A.F. Ragab,^d Hoda I. El Diwani,^a

^a Department of Chemistry of Natural and Microbial Products, Division of Pharmaceutical and Drug Industries, National Research Centre, Dokki, 12622, Cairo, Egypt

^b Institute of Biology, Medicinal Chemistry & Biotechnology, National Hellenic Research Foundation, 48 Vassileos Constantinou Avenue, Athens, GR-11635, Greece.

^c Institut de Chimie et Biochimie Moléculaires et Supramoléculaires, Laboratoire de Chimie Organique 2 – Glycochimie, UMR 5246, CNRS, Université Claude Bernard Lyon 1, 43 Boulevard du 11 Novembre 1918, F-69622 Villeurbanne, France.

^d Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Cairo University, Cairo, Egypt.



Arylsulfonyl derivatives of **1** showed higher inhibition (IC₅₀ ca 350 μM) against **glycogen phosphorylase** as compared to other amino or hydrazino heterocyclic analogues (IC₅₀ 400 - 600 μM).