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## Spectroscopic peculiarities of CsCaI<sub>3</sub>:Tm<sup>2+</sup> single crystals examined through one-photon and excited state excitation spectroscopy

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### ABSTRACT

The CsCaI<sub>3</sub> single crystals doped with Tm<sup>2+</sup> were grown by Bridgman technique. Divalent thulium luminescence related to the 4f<sup>13</sup>-4f<sup>13</sup> intra-configurational and 4f<sup>12</sup>5d<sup>1</sup>-4f<sup>13</sup> inter-configurational transitions were analyzed as a function of temperature. Excited state excitation spectroscopy has been applied within a wide spectral range to examine peculiarities of transitions between different 4f<sup>12</sup>5d<sup>1</sup> excited states of Tm<sup>2+</sup>. This two-color excitation experiment was useful to assess d-d photon-absorption processes that are relevant to observed up-conversion phenomena attributed to effective anti-Stokes green luminescence around 550 nm. Furthermore, activation energy of 4f<sup>12</sup>5d<sup>1</sup> → 4f<sup>13</sup> luminescence quenching was estimated and high value of <sup>2</sup>F<sub>5/2</sub> → <sup>2</sup>F<sub>7/2</sub> emission cross-section was determined in CsCaI<sub>3</sub>:Tm<sup>2+</sup>. Our spectroscopic study was performed to complement and verify the previous investigation in this field.

#### **Keywords:**

Rare earth alloys and compounds  
Optical absorption spectroscopy  
Divalent thulium  
CsCaI<sub>3</sub>  
Excited state excitation

## 1. Introduction

In recent years there has been a steady increase in the theoretical and experimental studies of divalent lanthanide ions ( $\text{Ln}^{2+}$ ) doped in inorganic materials. Although the most abundant data exist for divalent samarium, europium and ytterbium, which are most easily reducible from the trivalent to divalent state [1], interconfigurational  $4f^N - 4f^{N-1}5d^1$  transitions of other  $\text{Ln}^{2+}$  ions have also received appreciable attention owing to their technological applications [2]. The intense and broad emission of divalent europium is desirable in phosphors [3,4,5,6]. The  $\text{Sm}^{2+}$ -doped materials are found as effective light conversion optical systems [7,8] and highly sensitive optical temperature sensors [9]. The  $\text{Tm}^{2+}$  doped halides exhibit promising properties for applications as efficient luminescent solar concentrators [10], whereas  $\text{SrB}_4\text{O}_7$  host activated with divalent thulium ions can be considered as a potential laser material and optical temperature sensor [11]. Possible laser applications were recognized for  $4f^{12}5d^1 - 4f^{13}$  transitions of  $\text{Yb}^{2+}$  [12]. It is well known that the critical issue for stabilization of lanthanide ions in their divalent state is the choice of appropriate host matrix [13]. In this regard a ternary halides  $\text{CsCaX}_3$  ( $X = \text{Cl}, \text{Br}$  or  $\text{I}$ ) that crystallize in a perovskite structure and belong to orthorhombic crystalline system with  $\text{Ca}^{2+}$  ions in the distorted octahedral coordination appear as very suitable hosts for stabilization of divalent lanthanides.  $\text{CsCaX}_3$  crystals contain divalent sites, hence the  $\text{Ln}^{2+}$  ions can be conveniently incorporated into host lattice. Several papers have been devoted to investigation of spectroscopic properties of divalent lanthanide ions in halidoperovskites. The spectroscopic properties of  $\text{Eu}^{2+}$  ions doped in the perovskites  $\text{CsCaX}_3$  and  $\text{CsSrX}_3$  ( $X = \text{Cl}, \text{Br}$ , and  $\text{I}$ ), which are considered as novel promising scintillators [14,15,16], were studied in [17]. The same compounds doped with  $\text{Yb}^{2+}$  ions also exhibit properties suitable for their applications in high-energy detection devices [18]. The comparative studies of photoluminescence properties of  $\text{Yb}^{2+}$  ions doped in the halidoperovskites  $\text{CsMX}_3$  ( $M = \text{Ca}, \text{Sr}; X = \text{Cl}, \text{Br}, \text{I}$ ) were presented [19] followed by systematic studies of the decay times and temperature dependence of the spin-enabled and spin-forbidden transitions of  $\text{Yb}^{2+}$  ions in these hosts [18, 20].

Halidoperovskites doped with  $\text{Tm}^{2+}$  ions appear as particularly interesting systems since different types of light emission were recognized in  $\text{Tm}^{2+}$ -doped  $\text{CsCaCl}_3$ ,  $\text{CsCaBr}_3$  and  $\text{CsCaI}_3$  crystals. The occurrence of multiple emitting  $4f^{12}5d^1$  states in  $\text{Tm}^{2+}$  was recently explained by

using *ab initio* calculations [21]. Radiative and nonradiative relaxation processes were studied and discussed for these systems [22,23,24]. The same authors reported on detailed analysis of Tm-doped halide crystals absorption spectra in the region of 4f-4f and 4f-5d excitations [25]. It has been shown that  $\text{Tm}^{2+}$  ions offer excellent advantage for the study of 4f-5d excitation processes. In subsequent studies the upconversion (UC) phenomena have been observed in  $\text{CsCaI}_3:\text{Tm}^{2+}$  crystals under excitation at 810 nm [26,27]. It is unusual type of upconversion phenomena involving the 4f-5d states of  $\text{Tm}^{2+}$ . Understanding of  $4f^{12}5d^1$  excited states are highly essential to explore peculiarities of multiple emission and UC processes in materials doped with divalent thulium ions. It has been recently proved that two-photon experiment is an useful method to characterize lanthanide excited states including transitions between different  $4f^{N-1}5d^1$  multiplets [28]. In particular it is especially suitable for probing the equilibrium distance offset between different  $4f^n-4f^{n-1}5d^1$  excited states [28,29] Since for  $\text{Tm}^{2+}$  in  $\text{CsCaI}_3$  emission originates not only from the lowest but also from the higher  $4f^{12}5d^1$  excited state it is a very suitable system for two-photon excited state excitation spectroscopy (ESE). UC process involving exclusively  $4f^{12}5d^1$  excited states has already been demonstrated in  $\text{CsCaI}_3:\text{Tm}^{2+}$  system [26,27], but only using excitation at a single wavenumber. The ESE experiment has not been reported.

The main aim of this study is to gain deeper insights into processes behind UC involving  $4f^{12}5d^1$  excited states of  $\text{Tm}^{2+}$  ion in  $\text{CsCaI}_3$ . Accordingly, in this work one photon excitation spectroscopy and especially two-photon excited state excitation spectroscopy (ESE) are employed to investigate and analyze  $4f^{13}$  and  $4f^{12}5d^1$  states responsible for activation of the observed down-converted and up-converted  $\text{Tm}^{2+}$  luminescence. A significant decrease of the  $4f^{12}5d^1 \rightarrow 4f^{13}$  luminescence lifetime was observed above  $T = 190$  K and effective value of  ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$  emission cross-section was estimated for  $\text{CsCaI}_3:\text{Tm}^{2+}$  crystal. Our original results are discussed in relation to previously obtained data for  $\text{Tm}^{2+}$  in  $\text{CsCaI}_3$ . Moreover, the noticeable differences between spectroscopic properties of investigated divalent thulium doped iodide crystal and those reported for  $\text{CsCaCl}_3:\text{Tm}^{2+}$  and  $\text{CsCaBr}_3:\text{Tm}^{2+}$  are observed and discussed.

## 2. Experimental section

CsCaI<sub>3</sub> single crystals doped with 0.14 at.% and 0.68 at.% of Tm<sup>2+</sup> (as determined with ICP-OES spectroscopy) were grown by Bridgman technique. Pure CsCaI<sub>3</sub> single crystal, prepared from stoichiometric amounts of CsI and CaI<sub>2</sub>, was crushed and mixed with appropriate amount of TmI<sub>2</sub>, then placed in a vitreous carbon crucible, which was put into silica ampoule and heated for several hours at 720 K under high dynamic vacuum. After sealing under vacuum the ampoule was lowered through the vertical furnace at 1015 K at a rate of 5 mm/h. TmI<sub>2</sub> used in this procedure was earlier synthesized via metallothermic route [30] in reaction of anhydrous TmI<sub>3</sub> with Tm powder (99.9%, Sigma-Aldrich) at 1125 K in a sealed niobium tube. TmI<sub>3</sub> was prepared from Tm<sub>2</sub>O<sub>3</sub> (99.99%, Stanford Materials) by the ammonium halide route [31]. The obtained crystals were of good optical quality (Fig. S1). The crystals were cut and polished in the argon glove-box and placed in airtight copper cuvette with quartz windows for absorption measurements or sealed in a quartz tube under low-pressure of helium gas for emission measurements.

The emission spectra were excited by a MDL-N-808-10W diode laser at 808 nm and MXL-F-445-3W diode laser at 445 nm, respectively. The luminescence was dispersed by an Optron Dongwoo monochromator with 750 mm focal length and detected by a Hamamatsu R955 photomultiplier or InGaAs detector depending on the spectral range. The excitation spectrum was recorded with a Dongwoo Model Scanning System consisting of LPS-200X Xenon Lamp, an excitation monochromator with 150 mm focal length, an emission monochromator having 750 mm focal length and a Hamamatsu R955 photomultiplier. Excited State Excitation spectra were recorded using MDL-N-808-10W diode laser as first excitation source. Up-converted green luminescence was monitored on Dongwoo DM711 monochromator and detected by a Hamamatsu R3896 photomultiplier. The second excitation pulses fluently tuned in the required spectral regions were delivered from Oportek Opolette 355 LD laser system. To record luminescence decay curves a femtosecond laser (Coherent Model “Libra”) that delivers a train of 89 fs pulses at a centre wavelength of 800 nm and a pulse energy of 1 mJ with a repetition rate regulated up to 1 kHz was used as an excitation source. To obtain light pulses at different wavelengths (230-2800 nm) the laser is coupled to an optical parametric amplifier (Light Conversion Model “OPerA”). Luminescence decay curves were recorded with a grating spectrograph (Princeton Instr. Model Acton 2500i) coupled to a streak camera (Hamamatsu Model C5680). For spectroscopic measurement at low temperature the samples were placed in an

Oxford Model CF 1204 continuous flow liquid helium cryostat equipped with a temperature controller.

### 3. Results and discussion

CsCaI<sub>3</sub> crystallizes in the orthorhombic Pbnm space group [32] and in doped crystals Tm<sup>2+</sup> ions substitute for Ca<sup>2+</sup> ions on a slightly distorted octahedral site (the actual site symmetry for Tm<sup>2+</sup> ions is C<sub>1</sub>).

Fig. 1 reveals the excitation spectrum of CsCaI<sub>3</sub>:0.14% Tm<sup>2+</sup> visible luminescence monitored at 547 nm and measured at T = 5K within 250 - 540 nm spectral region. The excitation spectrum is quite complex and consists of numerous broad bands, while some of them exhibit additional fine structure. The bands observed in the spectrum are due to transitions from the ground <sup>2</sup>F<sub>7/2</sub> multiplet of the 4f<sup>13</sup> configuration to the states of the 4f<sup>12</sup>5d<sup>1</sup> configuration [33]. Octahedral ligand-field interaction in CsCaI<sub>3</sub>:Tm<sup>2+</sup> results in splitting of 5d orbitals into t<sub>2g</sub> and e<sub>g</sub> sets of levels. These levels are coupled to <sup>2S+1</sup>L<sub>J</sub> multiplets of 4f<sup>12</sup> core originating from spin-orbit and Coulomb interactions. The resulting states will be labeled as (<sup>2S+1</sup>L<sub>J</sub>, t<sub>2g</sub>) and (<sup>2S+1</sup>L<sub>J</sub>, e<sub>g</sub>), respectively. The Coulomb interactions between the 5d electron and the 4f<sup>12</sup> core electrons additionally split these states into high-spin (S = 3/2) and low-spin (S = 1/2) states. The energy-level diagram of Tm<sup>2+</sup> in CsCaI<sub>3</sub> as well as the radiative and nonradiative processes relevant to the discussion presented in this study are shown in the inset in Fig. 1.

According to energy level diagram in Fig. 1 the first intense band observed in the excitation spectrum with maximum at ~19375 cm<sup>-1</sup> (516 nm) should be assigned to <sup>2</sup>F<sub>7/2</sub> → (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) transitions, whereas the group of bands observed in the 20500 - 25000 cm<sup>-1</sup> range, with maxima at about 21680 cm<sup>-1</sup> (461 nm), 22730 cm<sup>-1</sup> (440 nm) and 23830 cm<sup>-1</sup> (420 nm) are attributed to <sup>2</sup>F<sub>7/2</sub> → (<sup>3</sup>H<sub>6</sub>, e<sub>g</sub>) transitions.

Near-infrared emission spectrum of Tm<sup>2+</sup> was acquired in 1100-1200 nm spectral range after excitation of CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup> crystal at 445 nm. Fig. 2 (a) compares luminescence spectra of Tm<sup>2+</sup> in the near-infrared region measured at several different temperatures between 5 and 300 K. The spin-orbit interaction splits the <sup>2</sup>F term of the ground 4f<sup>13</sup> configuration of Tm<sup>2+</sup> into the ground <sup>2</sup>F<sub>7/2</sub> and the excited <sup>2</sup>F<sub>5/2</sub> multiplet. Accordingly, the observed emission band is related to f-f intra-configurational <sup>2</sup>F<sub>5/2</sub> → <sup>2</sup>F<sub>7/2</sub> transitions of Tm<sup>2+</sup>. It can be discerned that influence of temperature on the infrared luminescence excited at 445 nm in CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup>

is significant. Integrated luminescence intensity for the spectrum measured at  $T = 300$  K is smaller by a factor of six as compared to that recorded at  $T = 200$  K. In the room-temperature NIR emission spectrum of  $\text{Tm}^{2+}$  one pronounced peak with maximum at  $8767 \text{ cm}^{-1}$  is observed and its FWHM value was determined to be  $144 \text{ cm}^{-1}$ . The Füchtbauer-Ladenburg method was used to estimate the emission cross-section. The dependence between the emission cross-section  $\sigma_{em}$  and emission spectrum is described as:

$$\sigma_{em}(\lambda) = \frac{\beta \lambda^5 I(\lambda)}{8\pi n^2 c \tau_{rad} \int \lambda I(\lambda) d\lambda} \quad (1)$$

where  $I(\lambda)$  represents the experimental emission intensity at the wavelength  $\lambda$ ,  $c$  is the light velocity,  $n$ ,  $\beta$  and  $\tau_{rad}$  are the refractive index of material, branching ratio and radiative lifetime of the  ${}^2F_{5/2}$  multiplet, respectively. It was assumed that luminescence lifetime of  $0.65 \text{ ms}$  measured for diluted sample is not significantly different than  ${}^2F_{5/2}$  radiative lifetime. Inset of Fig. 2 presents simulated emission cross-section spectrum derived from the  ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$  luminescence of  $\text{Tm}^{2+}$  excited at  $445 \text{ nm}$ . The prominent line at  $1141 \text{ nm}$  dominates emission cross-section spectrum with advantageously high peak value which amounts to  $\sigma_{em} = 6.07 \times 10^{-20} \text{ cm}^2$ . For a comparison, significantly lower value of emission cross section  $\sigma_{em} = 3.50 \times 10^{-20} \text{ cm}^2$  was found for  $\text{SrB}_4\text{O}_7:\text{Tm}^{2+}$  [34].

Fig. 2 (b) presents the  $\text{Tm}^{2+}$  NIR emission spectrum measured at  $T = 5 \text{ K}$ . At this temperature only the lowest-energy crystal field sublevel of the emitting state is substantially populated and four emission lines are expected for  ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$  transitions having in mind that divalent thulium ions are coordinated by six  $\Gamma$  ligands in a distorted octahedral sites of  $C_1$  symmetry in  $\text{CsCaI}_3$  lattice. In accordance with this expectation the emission spectrum presented in Fig. 2 is dominated by four sharp lines located at  $8634$ ,  $8721$ ,  $8773$  and  $8854 \text{ cm}^{-1}$ . It should be mentioned that only two  ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$  emission lines identified at  $8782 \text{ cm}^{-1}$  and  $8790 \text{ cm}^{-1}$  by J. Grimm et al. [22], were assigned by these authors as electronic transitions. The remaining emission lines placed at longer wavelengths were found as vibronic sidebands. In a short-wavelength part of our spectrum the additional structureless band appears at  $8140 \text{ cm}^{-1}$ . In view of a fast decay of this emission ( $\tau \approx 1 \mu\text{s}$ ) it may be associated with transitions between levels of initial ( ${}^3F_4, t_{2g}$ ) and terminal ( ${}^3H_4, t_{2g}$ ) manifolds. The decay curves of observed luminescence are presented in Fig. 2 (c,d). The 10 Dq crystal-field splitting of the 5d part of the  $4f^{12}5d^1$



electron configuration in CsCaI<sub>3</sub>:Tm<sup>2+</sup> was estimated to be 8500 cm<sup>-1</sup> and more than fifteen cut-off phonons are needed to bridge the (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) - (<sup>3</sup>H<sub>4</sub>, t<sub>2g</sub>) energy gap. It is quite enough for (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) components to become emissive levels and consequently, the excitation energy may be down-converted to obtain NIR photons as result of d-d transitions.

Aside from f-f NIR emission of Tm<sup>2+</sup> the visible luminescence of CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup> crystal was detected at T = 5 K upon excitation at 445 nm and 808 nm as it is displayed in the Fig. 3. In fact, these spectra consist of one prominent structureless emission band centered in green spectral region at 547 nm. When the (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) Tm<sup>2+</sup> multiplet is excited at λ = 445 nm down-converted emission occurs. The energy gap between the (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) ↔ (<sup>3</sup>H<sub>6</sub>, t<sub>2g</sub>) multiplets was found to be 2700 cm<sup>-1</sup> whereas a maximum energy of phonons in CsCaI<sub>3</sub> is ħω<sub>max</sub> = 170 cm<sup>-1</sup> [14]. Accordingly, efficient radiative transitions from the (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) multiplet take place. The green Tm<sup>2+</sup> emission is associated with a transition from the lowest excited state of the (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) multiplet to the <sup>2</sup>F<sub>7/2</sub> Tm<sup>2+</sup> ground state. The inset in Fig. 3 presents the magnified fragment of emission spectrum recorded in 11500-13500 cm<sup>-1</sup> range when excited at 445 nm. This emission band peaked at 12100 cm<sup>-1</sup> is associated with (<sup>3</sup>H<sub>6</sub>, t<sub>2g</sub>) → <sup>2</sup>F<sub>7/2</sub> transitions.

When the CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup> crystal is excited at 808 nm the up-converted luminescence is found (Fig. 3). In this case the (<sup>3</sup>H<sub>6</sub>, t<sub>2g</sub>) S = 3/2 multiplet is excited as a result of the spin-forbidden transitions. The observed upconversion phenomena is atypical because the adequate 4f<sup>12</sup>5d<sup>1</sup> Tm<sup>2+</sup> excited states are involved to populate the divalent thulium luminescent levels. Efficiency of the green up-converted emission in CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup> is relatively high. The up-converted emission with a maximum located at λ = 546 nm when excited at 12350 cm<sup>-1</sup> (809.7 nm) has been reported by J. Grimm et. al. [24,27]. In contrast to the Tm<sup>2+</sup> luminescence presented in [24] our emission spectrum is characterized by a more intense band located at 485 nm (20618 cm<sup>-1</sup>) that can be attributed to transitions originating in the higher-energy levels of (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) multiplet. It has been concluded that two-photon excitation process is responsible for population of Tm<sup>2+</sup> luminescent level in upconversion phenomena. Furthermore it has been assumed that the Tm<sup>2+</sup> up-converted luminescence in CsCaI<sub>3</sub> mainly results from GSA (Ground State Absorption) /ESA (Excited State Absorption) process but contribution of GSA/ETU (Energy Transfer Upconversion) process is considered as well [27]. Upconversion phenomena in Tm<sup>2+</sup>-doped iodide can be effective because excited state absorption and energy transfer



upconversion processes involve  $5d(t_{2g}) \rightarrow 5d(e_g)$  excitation transitions [26] which are characterized by quite high values of oscillator strength.

Up-conversion phenomena observed in  $\text{CsCaI}_3:\text{Tm}^{2+}$  are activated by transitions between  $4f^{12}5d^1$  excited states. In order to get more detailed insight into these transitions a two-color excitation experiment was performed. The set-up consisting of two laser sources, namely a CW AlGaAs- diode laser and a tunable pulsed OPO laser was used to record excitation spectra in wide spectral regions. When CW  $\lambda = 808$  nm excitation wavelength is employed, a ground-state absorption occurs and the ( ${}^3\text{H}_6, t_{2g}$ ) multiplet is continuously populated. An additional pulsed excitation at varied wavelengths stimulate the already excited  $\text{Tm}^{2+}$  ions to populate the higher energy  $4f^{12}5d^1$  states. A second photon is resonant with the energy separation between involved  $\text{Tm}^{2+}$  excited states. Eventually, excited state excitation spectra of  $\text{CsCaI}_3:0.68\% \text{Tm}^{2+}$  monitoring anti-Stokes green emission at 547 nm were measured and are presented in Fig. 4. When the combination of these both aforementioned excitation sources was used the green UC emission signal was certainly enhanced comparing to CW single – laser excitation. This may indicate that ESA ( ${}^3\text{H}_6, t_{2g}$ )  $\rightarrow$  ( ${}^3\text{F}_4, t_{2g}$ ) contribution in UC process is highly relevant in  $\text{CsCaI}_3:\text{Tm}^{2+}$  crystal.

Two-color excited state excitation experiment can be utilized to examine equilibrium distance offset the involved  $4f^{12}5d^1$  states. It was assumed that electronic transitions between two excited states with comparable equilibrium bond lengths are related to ESE sharp spectral line while transitions between contracted and expanded excited states give rise to broad bands [28]. Our excited state excitation spectrum displayed in Fig. 4 consists of a few broad bands (labeled A-D) placed within  $6830\text{-}8330\text{ cm}^{-1}$ ,  $8665\text{-}9340\text{ cm}^{-1}$ ,  $11095\text{-}13825\text{ cm}^{-1}$  and  $14100\text{-}16815\text{ cm}^{-1}$  spectral ranges. The lower-energy ESE (A-B) bands can be attributed to ( ${}^3\text{H}_6, t_{2g}$ )  $\rightarrow$  ( ${}^3\text{F}_4, t_{2g}$ ) transitions and the remaining broadened bands (C-D) located at higher energy may be associated with two-photon absorption transitions to  $4f^{12}5d(t_{2g}/e_g)^1$  mixed states. At lower energy sides of A and B ESE bands sharp and narrow lines can be discerned, which begin to appear at  $6987\text{ cm}^{-1}$  and  $8766\text{ cm}^{-1}$ , respectively. These sharp components are related to transitions characterized by no change in potential energy surfaces equilibrium position of the various  $4f^{12}5d(t_{2g})^1$  excited states. In accordance with that for low-energy f-d states of  $\text{Tm}^{2+}$  in  $\text{CsCaI}_3$  there is rather similar Tm-ligand equilibrium distance in their potential energy surface. The adequate set of ESE bands attained during a two-photon excitation experiment was recently observed by M. de Jong. et. al.

[28] for CsCaCl<sub>3</sub>:Tm<sup>2+</sup> and CsCaBr<sub>3</sub>:Tm<sup>2+</sup> crystals. The onset of two observed groups of lines has been then established at 7100 cm<sup>-1</sup> and 9000 cm<sup>-1</sup>. It means that (<sup>3</sup>H<sub>6</sub>, t<sub>2g</sub>) → (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) ESE bands observed for CsCaI<sub>3</sub>:Tm<sup>2+</sup> are slightly shifted toward lower energy in relation to the chloride and bromide counterparts. It stems from more covalent character of iodide host as compared to chloride or bromide hosts as well as from differences in a crystal-field strength. Both these factors exert strong effect on barycenter of 4f<sup>12</sup>5d<sup>1</sup> configuration and energies of d-d transitions. The ESE band measured for CsCaI<sub>3</sub>:Tm<sup>2+</sup> contain more narrow and sharp lines than it is observed in CsCaCl<sub>3</sub>:Tm<sup>2+</sup>. This finding may be a consequence of a higher site symmetry of Tm<sup>2+</sup> ion in chloride (C<sub>4h</sub>) than in iodide (C<sub>1</sub>) crystal. As compared to previous results [28] our excited state excitation experiment was performed using higher-energy second excitation photons and ESE spectra were acquired up to 16800 cm<sup>-1</sup>. Actually, above 11000 cm<sup>-1</sup> broad and structureless bands appear, centered at 12037, 13492, 14804 and 15782 cm<sup>-1</sup>. These intense ESE bands may correspond to two-photon absorption from (<sup>3</sup>H<sub>6</sub>, t<sub>2g</sub>) state to higher-energy 4f<sup>12</sup>5d<sup>1</sup> states. The high intensity of ESE bands results from a large value of d-d two-photon absorption cross-section [35]. The spectral character of these broad bands implies that larger difference between equilibrium distance in potential energy surface of lower- and higher-energy 4f<sup>12</sup>5d<sup>1</sup> states occurs in CsCaI<sub>3</sub>:Tm<sup>2+</sup>.

Fig. 5 displays results of an experiment assessing the impact of temperature on thulium excited state relaxation dynamics. The visible luminescence decay curves were measured in a region 5 – 300 K at λ<sub>det.</sub>=547 nm under femtosecond pulse excitation at λ<sub>exc.</sub>= 450 nm. The relaxation of f-d (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) states in CsCaI<sub>3</sub>:Tm<sup>2+</sup> may result from radiative transitions and nonradiative multiphonon transitions. The energy separation between (<sup>3</sup>F<sub>4</sub>, t<sub>2g</sub>) and (<sup>3</sup>H<sub>6</sub>, t<sub>2g</sub>) states of Tm<sup>2+</sup> was found to be 2700 cm<sup>-1</sup> and roughly sixteen cut-off phonons are needed to bridge this energy gap. Accordingly, in this case the nonradiative relaxation should be effectively suppressed. Estimated experimental lifetime of green Tm<sup>2+</sup> emission (τ ≈ 1.3 μs) is practically constant from T = 5 K up to 190 K. Subsequently, at the higher temperature the lifetime is significantly shortened and amounts to 7 ns at room temperature. It is worth noticing that a single exponential time dependence was recognized for all analyzed decay curves. The luminescence decay curves measured at T=5K, 77 K and 298 K are presented in a lower part of Fig 5. The exponential decay function I=I<sub>0</sub>+Aexp(-t/τ<sub>exp.</sub>) was used to determine the adequate experimental lifetimes.

Effect of thermal energy on  $\text{Tm}^{2+}$  transition rates has been assessed with the Arrhenius plots defined as

$$W = A \exp\left(\frac{-E_q}{kT}\right) \quad (2)$$

equation (1) can be transformed to

$$\ln(W) = \ln(A) - \frac{E_q}{kT} \quad (3)$$

where  $W$  is the transition rate and  $A$  ( $s^{-1}$ ) denotes a transition rate given by the intercept of the line at  $1/T \rightarrow 0$  and is attributed to pre-exponential factor,  $k$  is Boltzmann constant and  $T$  is temperature. The parameter  $E_q$  (quenching energy,  $\text{cm}^{-1}$ ) is obtained from the fit of the (2) line. The lowest energy states of the  $4f^{12}5d^1 \text{Tm}^{2+}$  configuration are thermally populated with temperature increasing and dynamics of  $4f^{12}5d^1-4f^{13}$  vibronic transitions is more effective. Consequently, an increase of relaxation rate is observed as it is presented in inset of Fig. 5. The activation energy was estimated to be  $2697 \pm 15 \text{ cm}^{-1}$ . It means that our finding is highly consistent with the reported  $\Delta E = 2700 \text{ cm}^{-1}$  for energy gap between ( ${}^3\text{F}_4, t_{2g}$ ) and ( ${}^3\text{H}_6, t_{2g}$ ) states in  $\text{CsCaI}_3:\text{Tm}^{2+}$ . The comparable experimental lifetime of  $1 \mu\text{s}$  at  $T = 10 \text{ K}$  was determined for  $\text{CsCaI}_3:1 \text{ at.}\% \text{ Tm}^{2+}$  by J. Grimm et. al. [24]. Furthermore, these authors have observed that green radiative transitions are effective up to about  $250 \text{ K}$  and above this temperature the up-converted emission rapidly drops. This finding is consistent with our results revealing substantial decrease of measured lifetime already around  $T=200 \text{ K}$ . It was documented that radiative rate constant  $R$  of ( ${}^3\text{F}_4, t_{2g}$ )  $\rightarrow$   ${}^2\text{F}_{7/2}$  emission is the most prominent relaxation path in the iodide up to  $220 \text{ K}$ . Above this temperature nonradiative relaxation to ( ${}^3\text{H}_6, t_{2g}$ ) state is a dominant process [24]. It is worth noticing that the presented spectroscopic results for  $\text{CsCaI}_3:0.68\% \text{ Tm}^{2+}$  crystal are comparable to those obtained for  $\text{CsCaI}_3:0.14\% \text{ Tm}^{2+}$ , hence no significant differences were found for the sample containing lower divalent thulium concentration.

#### 4. Conclusions

The spectroscopic properties of  $\text{Tm}^{2+}$  - doped  $\text{CsCaI}_3$  single crystals were investigated in relation to previously reported data. Novel and valuable results were achieved particularly pertinent to excited state excitation experiment performed in a wide spectral range. Acquired excited excitation spectra are useful for analysis of d-d photon-absorption processes relevant especially in observed up-conversion phenomena. The  $4f^{12}5d \rightarrow 4f^{13}$  luminescence was analyzed as a function of temperature. The calculated activation energy was found to be fully in compliance to  $(^3F_4, t_{2g}) \leftrightarrow (^3H_6, t_{2g})$  excited states energy gap. Moreover Fuchtbauer–Ladenburg method was applied to estimate high value of emission cross section for NIR  $^2F_{5/2} \rightarrow ^2F_{7/2}$  transition in  $\text{CsCaI}_3:0.68\% \text{Tm}^{2+}$ . The noticeable differences between spectroscopic properties of investigated divalent thulium doped iodide crystal and those reported for  $\text{CsCaCl}_3:\text{Tm}^{2+}$  and  $\text{CsCaBr}_3:\text{Tm}^{2+}$  can be underlined. Emission bands measured for  $\text{CsCaI}_3:\text{Tm}^{2+}$  are located at lower energies in relation to chloride and bromide counterparts. We observed that the  $^2F_{5/2} \rightarrow ^2F_{7/2}$  emission intensity decreases significantly at  $T=250 \text{ K}$  in  $\text{CsCaI}_3:\text{Tm}^{2+}$  whereas this efficient quenching process occurs at  $T=100 \text{ K}$  for  $\text{CsCaCl}_3:\text{Tm}^{2+}$  and at  $T=150 \text{ K}$  for  $\text{CsCaBr}_3:\text{Tm}^{2+}$ . The green up-converted emission is observed for all halides but this luminescence is especially efficient in  $\text{CsCaI}_3:\text{Tm}^{2+}$  since a large  $(^3F_4, t_{2g}) - (^3H_6, t_{2g})$  energy gap is high enough for effective radiative transitions. Our excited state excitation experiment reveals some differences in the (ESE) spectra measured for iodide crystals and these studied for chloride and bromide. ESE bands related to different excited  $4f^{12}5d^1$  states in  $\text{CsCaI}_3:\text{Tm}^{2+}$  are shifted toward lower energy as compared to those recorded for chloride and bromide counterparts. These variations in energy level structure of the f–d states in  $\text{Tm}^{2+}$ -doped halides are relevant for efficiency of processes involved in upconversion phenomena as well.

#### Appendix A. Supplementary data

Supplementary data related to this article can be found at ...

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## Figures

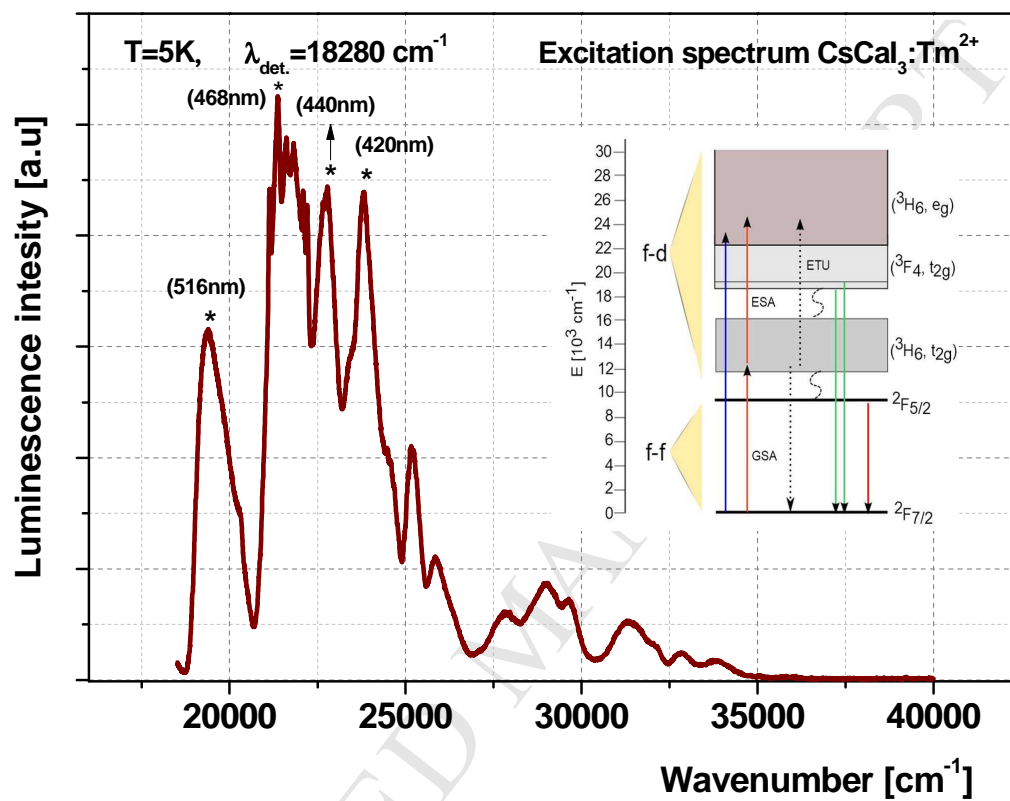


Fig. 1 One-photon excitation spectrum of CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup> recorded at 5 K while monitoring emission at λ = 547 nm. Inset presents energy-level diagram of Tm<sup>2+</sup> in CsCaI<sub>3</sub>.



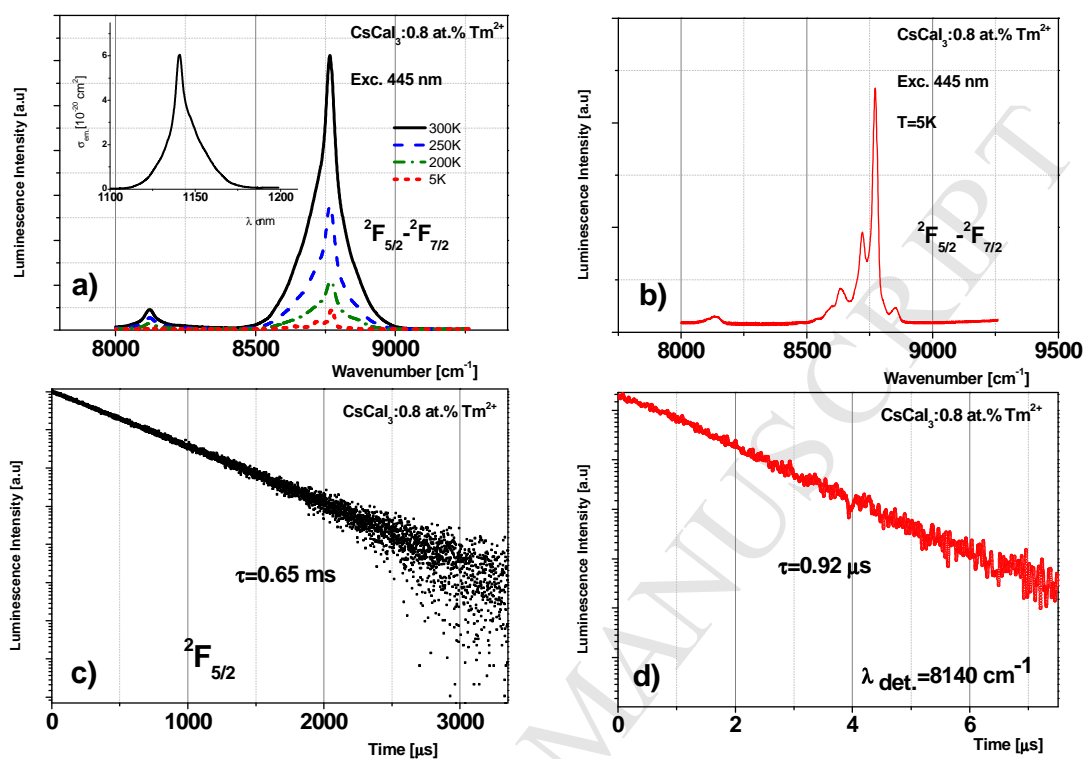


Fig. 2. a) Impact of temperature on <sup>2</sup>F<sub>5/2</sub>-<sup>2</sup>F<sub>7/2</sub> Tm<sup>2+</sup> emission spectra excited at 445 nm - inset shows the emission cross-section spectrum. b) Emission spectrum measured at T=5K. c) Decay curve of luminescence detected at  $\lambda=8767$  cm<sup>-1</sup>. d) Decay curve of luminescence detected at  $\lambda=8140$  cm<sup>-1</sup>.

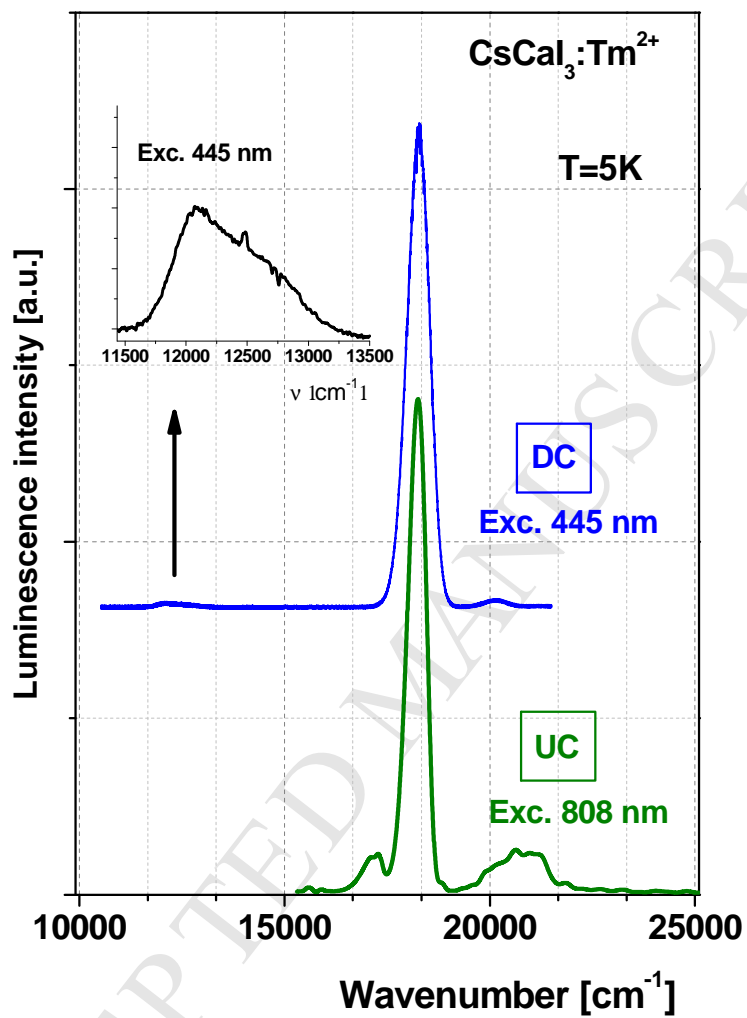


Fig. 3. Low-temperature ( $T=5\text{K}$ ) emission spectra of  $\text{CsCaI}_3:0.68\% \text{Tm}^{2+}$  crystal excited at 445 nm and 808 nm.

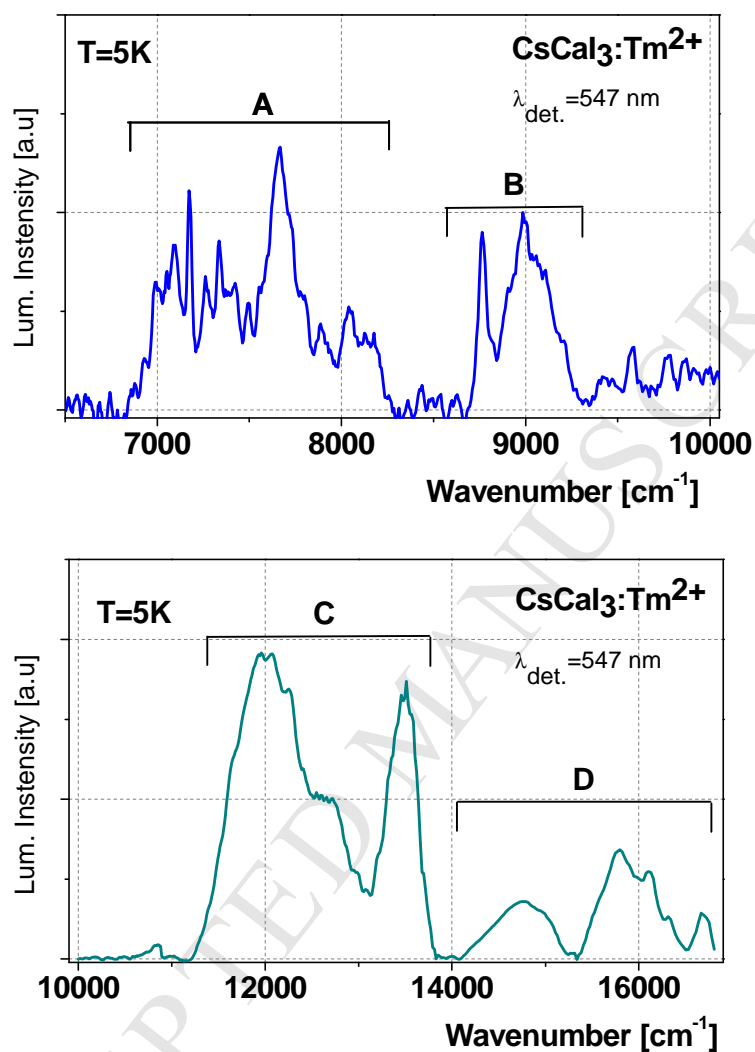


Fig. 4. Excited state excitation spectra of CsCaI<sub>3</sub>:0.68% Tm<sup>2+</sup> measured at T = 5K within 6000-17000 cm<sup>-1</sup> spectral range. The luminescence was monitored at λ=547 nm.

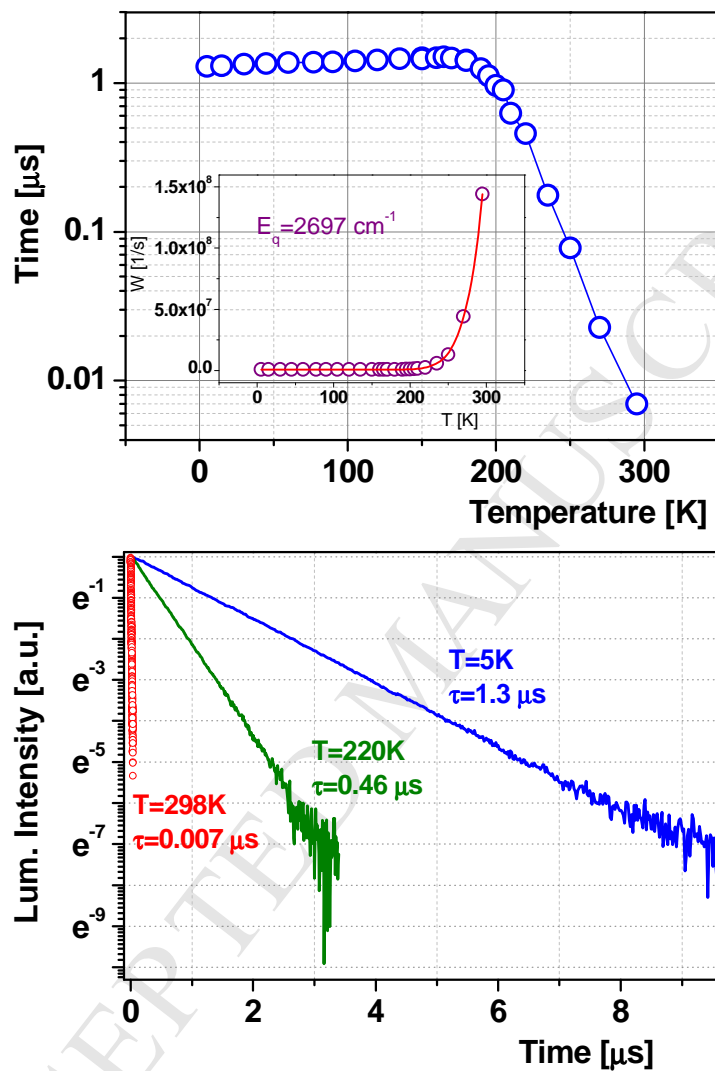


Fig. 5. Upper- Lifetime of the  $\text{CsCaI}_3:0.68\% \text{Tm}^{2+}$  luminescence monitored at  $18280 \text{ cm}^{-1}$  and plotted versus temperature. The inset shows the activation energy  $E_g = 2697 \text{ cm}^{-1} (\pm 15 \text{ cm}^{-1})$ . Lower – Decay curves of thulium luminescence measured at different temperatures.

- Excited state excitation spectra are presented for  $\text{Tm}^{2+}$  in  $\text{CsCaI}_3$ .
- The  $4f^{12}5d^1 \rightarrow 4f^{13}$  luminescence was analyzed in the function of temperature.
- Activation energy of  $4f^{12}5d^1 \rightarrow 4f^{13}$  luminescence quenching was estimated.
- A high value of  ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$  emission cross-section was determined.

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