



Calcium depletion-mediated protease inhibition and apical-junctional-complex disassembly *via* an EGTA-conjugated carrier for oral insulin delivery

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ARTICLE INFO

Article history:

Received 30 September 2012

Accepted 20 November 2012

Available online 27 November 2012

Keywords:

Oral protein delivery

Chelating agent

Calcium depletion

Enzyme inhibition

Apical junctional complex

ABSTRACT

Calcium (Ca^{2+}) has a crucial role in maintaining the intestinal protease activity and in forming the apical junctional complex (AJC) that preserves epithelial barrier function. Ethylene glycol tetraacetic acid (EGTA) is a Ca^{2+} -specific chelating agent. To maintain the concentration of this chelator in areas where enzyme inhibition and paracellular permeation enhancement are needed, this study synthesized a poly(γ -glutamic acid)-EGTA conjugate (γ PGA-EGTA) to form nanoparticles (NPs) with chitosan (CS) for oral insulin delivery. The results of our molecular dynamic (MD) simulations indicate that Ca^{2+} ions could be specifically chelated to the nitrogen atoms, ether oxygen atoms, and carboxylate oxygen atoms in $[\text{Ca}(\text{EGTA})]^{2-}$ anions. By chelating Ca^{2+} , γ PGA-EGTA conferred a significant insulin protection effect against proteases in intestinal tracts isolated from rats. Additionally, calcium depletion by γ PGA-EGTA could stimulate the endocytosis of AJC components in Caco-2 cell monolayers, which led to a reversible opening of AJCs and thus increased their paracellular permeability. Single-photon emission computed tomography images performed in the biodistribution study clearly show the ^{125}I -insulin orally delivered by CS/ γ PGA-EGTA NPs in the heart, aorta, renal cortex, renal pelvis and liver, which ultimately produced a significant and prolonged hypoglycemic effect in diabetic rats. The above results confirm that this γ PGA-EGTA conjugate is a promising candidate for oral insulin delivery.

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1. Introduction

Ease of administration coupled with high patient compliance makes the oral route the preferred mode of drug administration. However, the drawbacks of orally administered hydrophilic macromolecules include enzymatic degradation in the gastrointestinal (GI) tract and barrier to physical absorption in the epithelium [1]. To address these issues, we recently reported a pH-responsive nanoparticle (NP) system composed of chitosan (CS) and poly(γ -glutamic acid) conjugated with diethylene triamine pentaacetic acid (γ PGA-DTPA) [2]. CS, a mucoadhesive polycation, enhances paracellular permeability by transiently opening tight junctions (TJs) between epithelial cells [3], whereas DTPA inhibits intestinal proteases and disrupts epithelial TJs by nonspecific chelation of divalent metal ions such as Ca^{2+} , Mg^{2+} and Zn^{2+} [4]. Our experimental results

indicated that this CS/ γ PGA-DTPA NP system effectively enhances insulin absorption throughout the entire small intestine [2].

Studies showed that calcium (Ca^{2+}) has crucial roles in chymotrypsin activity and in maintaining the thermodynamic stability of trypsin [5,6]. Trypsin and chymotrypsin are the major proteolytic enzymes present in the intestinal lumen, where cleavage of insulin molecules occurs [7]. Extracellular Ca^{2+} is also required to form the apical junctional complex (AJC) needed for epithelial barrier function. Therefore, depletion of Ca^{2+} in the intestinal lumen is advantageous for oral administrated insulin to prevent enzymatic degradation and enhance penetration through intestinal epithelial cells.

EGTA, a Ca^{2+} -specific chelator, has significantly greater Ca^{2+} affinity compared to other chelating agents [8]. Due to its Ca^{2+} -binding specificity, EGTA is more favorable for oral insulin delivery than DTPA. To take advantage of its Ca^{2+} -binding specificity, an amine-group containing EGTA derivative was first synthesized and then conjugated on γ PGA to form NPs with CS for oral insulin delivery. Structural changes in the as-prepared NPs were examined in different pH environments by dynamic light scattering (DLS), transmission electron microscopy (TEM), and small angle X-ray scattering (SAXS). Effects of γ PGA-EGTA on enzymatic inhibition and on paracellular permeability were

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evaluated *in vitro* in isolated rat intestinal segments and in Caco-2 cell monolayers, respectively. Furthermore, the mechanism of TJ opening induced by test NPs was studied by visualizing the ultra-structural changes in the epithelial AJC *via* TEM and confocal laser scanning microscopy (CLSM). The biodistribution of insulin orally delivered by the developed NPs was investigated by single-photon emission computed tomography (SPECT); their pharmacokinetic (PK) and pharmacodynamic (PD) profiles were then analyzed in a diabetic rat model.

2. Materials and methods

2.1. Materials

The 85%-deacetylated CS (MW 60 kDa) was purchased from Koyo Chemical Co. (Japan), whereas γ PGA (MW 100 kDa) was acquired from Vedan Co. (Taichung, Taiwan). Other chemicals and reagents used were of analytical grade and obtained from Sigma-Aldrich (St. Louis, MO, USA).

2.2. Synthesis and characterization of γ PGA-EGTA

An amine-group containing EGTA derivative was first prepared by incorporating an amino-functionalized aromatic moiety on the oxoethylenic bridge of EGTA [9]. Briefly, ethanolamine (0.9 g) was reacted with KHC_2O_3 (3.8 g) and tert-butyl bromoacetate ($\text{BrCH}_2\text{COO}^t\text{Bu}$, 7.4 g) in dimethylformamide (DMF, 25 mL) at room temperature for 24 h. The crude product in ethyl acetate ($\text{CH}_3\text{CO}_2\text{Et}$,

25 mL) was subsequently mixed with *N,N*-diisopropylethylamine (iPr_2NET , 2.2 g) and MsCl (1.9 g) and reacted for 24 h. The pure compound in DMF (15 mL) was then mixed with K_2CO_3 (1.6 g), NaI (1.8 g) and 4-nitrocatechol (0.8 g) and stirred for 24 h; the product was dissolved in methanol and hydrogenated with 10% Pd/C . The obtained EGTA derivative was purified by column chromatography and then conjugated on γ PGA (the γ PGA-EGTA conjugate, Fig. 1a) as described below.

The γ PGA-EGTA conjugate was prepared by reacting the acid form of γ PGA (0.1 g) with an equimolar amine-group of the EGTA derivative in the presence of 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC, 0.2 g) and *N*-hydroxysuccinimide (NHS, 0.1 g) in dimethyl sulfoxide (DMSO) at room temperature for 48 h. The product was purified by dialysis against deionized (DI) water for 3 days and then lyophilized. The Boc-deprotection reaction was performed by mixing the freeze-dried product with trifluoroacetic acid (TFA, 5 mL) at room temperature for 48 h. Finally, the synthesized γ PGA-EGTA conjugate was purified by dialysis against DI water for 3 days and lyophilized before further use.

The synthesized γ PGA-EGTA conjugate was characterized by ^1H NMR (Varian Unityionva 500 NMR Spectrometer, MO, USA) and FT-IR (Perkin-Elmer Spectrum RX 1 FT-IR System, Buckinghamshire, UK). The degree of substitution (DS) of EGTA on γ PGA was estimated by ^1H NMR spectroscopy. The Ca^{2+} binding efficiency of γ PGA-EGTA was determined by complexometric titration [10].

The conformation of γ PGA-EGTA in the presence of Ca^{2+} and Mg^{2+} was examined by molecular dynamic (MD) simulations [11].

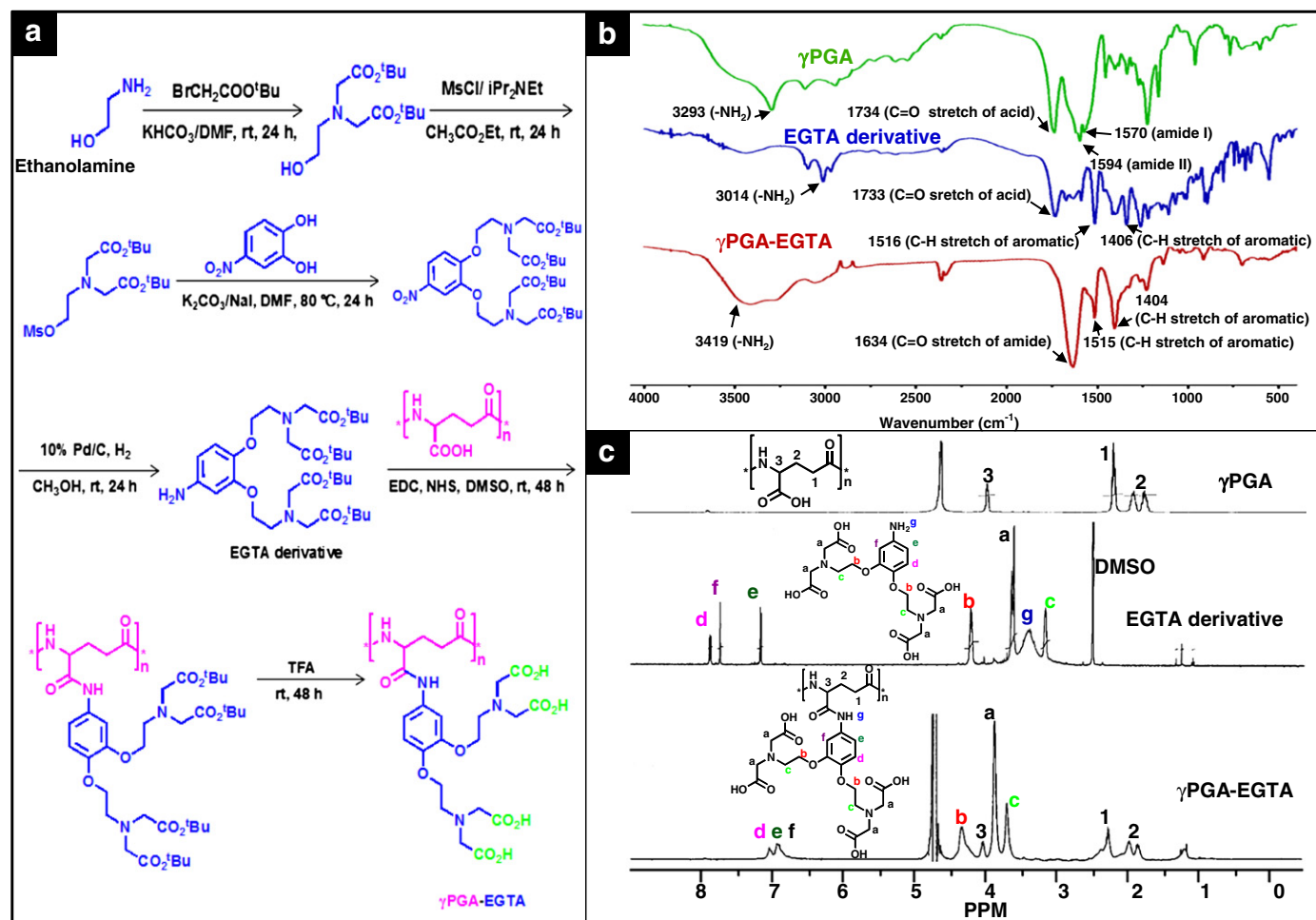


Fig. 1. Synthesis and characterization of the γ PGA-EGTA conjugate: (a) the reaction scheme showing the formation of an amino-functionalized EGTA derivative and its subsequent conjugation on the γ PGA backbone; (b) the FT-IR and (c) ^1H NMR spectra of the free-form γ PGA, EGTA derivative and synthesized γ PGA-EGTA conjugate. γ PGA: poly(γ -glutamic acid) and EGTA: ethylene glycol tetraacetic acid.

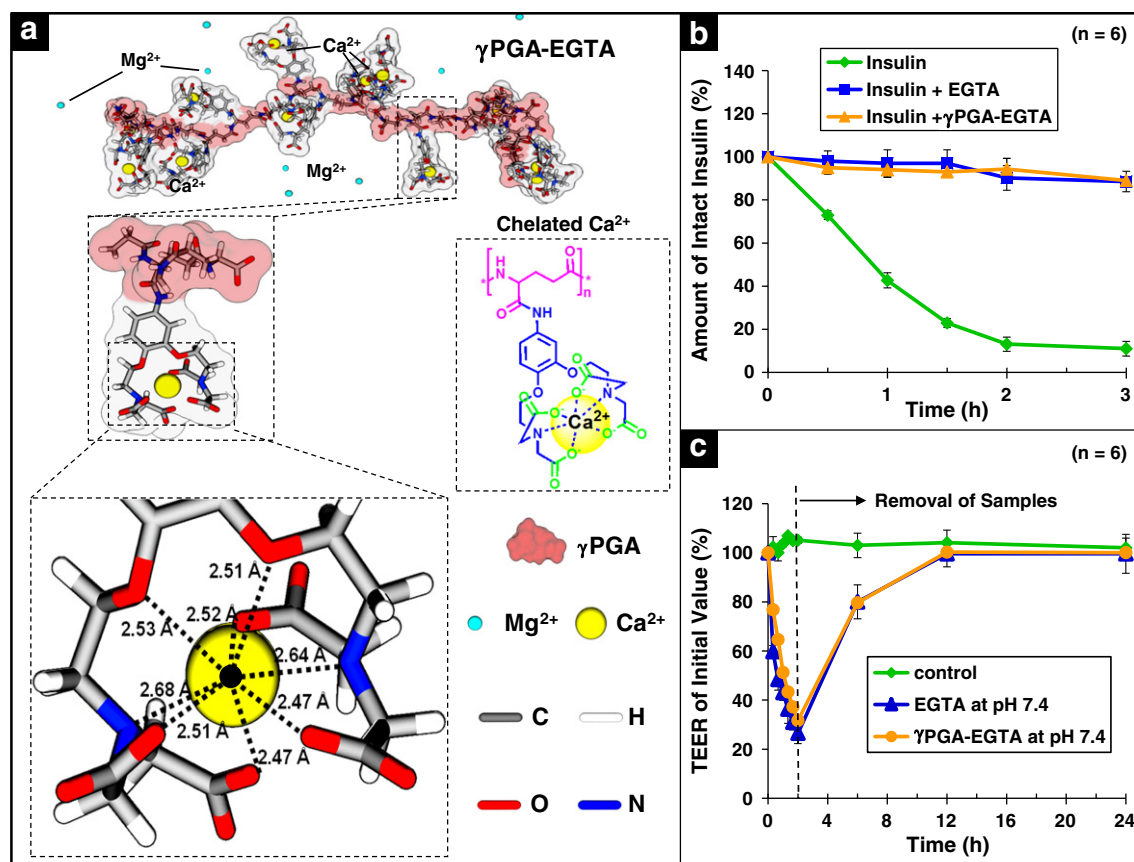


Fig. 2. (a) Results of molecular dynamic simulations showing the conformation of γ PGA-EGTA in the presence of Ca^{2+} and Mg^{2+} ; (b) effects of EGTA and γ PGA-EGTA on insulin protection against intestinal proteases and (c) transepithelial electrical resistance (TEER) in Caco-2 cell monolayers. Control: the group without any treatment.

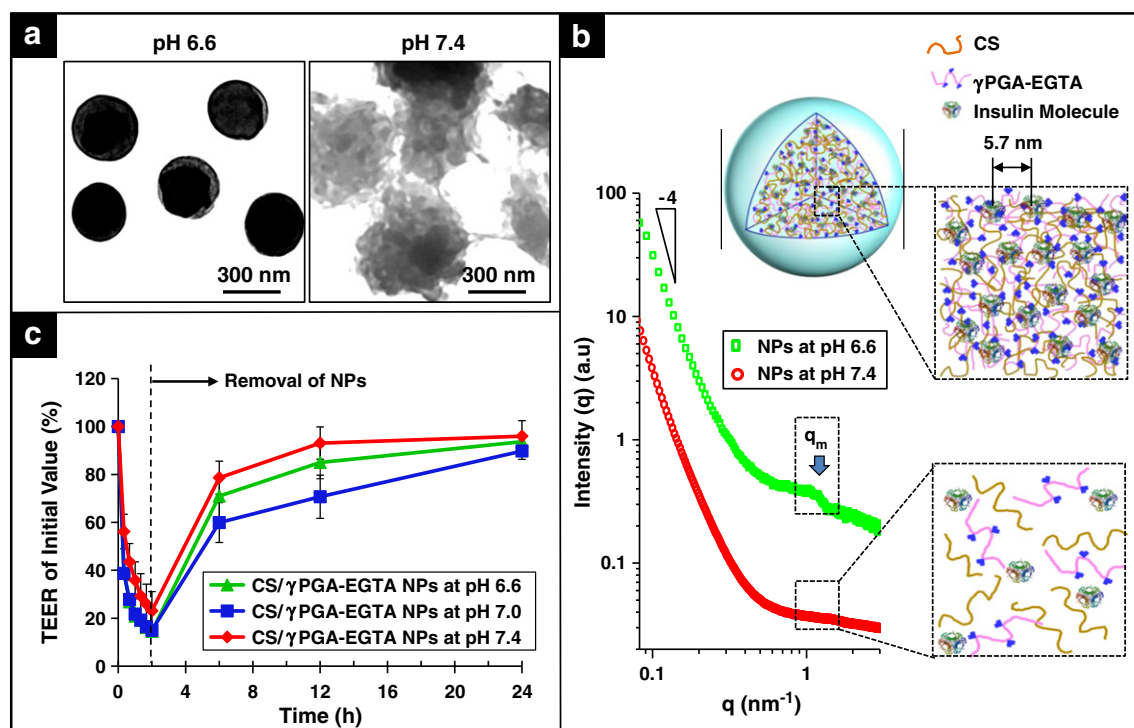


Fig. 3. (a) TEM micrographs and (b) small angle X-ray scattering (SAXS) profiles showing the pH-responsive characteristics of insulin-loaded CS/ γ PGA-EGTA nanoparticles (NPs); (c) effects of these NPs on the TEER of Caco-2 cell monolayers at distinct pH environments. CS: chitosan.

In body fluids, the predominant divalent metal ions are Ca^{2+} and Mg^{2+} [12], and Ca^{2+} and Mg^{2+} ion levels are higher in gastrointestinal (GI) mucosae than in blood [13,14]. Atomistic pictures of molecules were generated using UCSF Chimera.

2.3. Inhibition of intestinal proteolytic degradation

In vitro studies of intestinal protease inhibition by γ PGA-EGTA were performed using proximal sections of small intestine freshly isolated from rats (Wistar, 180–200 g) [2]. The intestinal tissues were incubated with a Krebs–Ringer buffer (KRB) solution containing insulin (1 mg/mL, 1 mL) and γ PGA-EGTA (5 mg/mL) at 37 °C; free-form EGTA (2 mg/mL) at 37 °C, which had a similar Ca^{2+} binding capacity, was used as a control. To quantify the amount of intact insulin remaining, test solutions (50 μ L) withdrawn at different time intervals were immediately analyzed by high-performance liquid chromatography (HPLC).

2.4. Permeability of epithelial AJs

The permeability of the epithelial AJs in Caco-2 cell monolayers after treatment with test samples was measured in terms of transepithelial electrical resistance (TEER). The cell monolayers were incubated individually with EGTA (4 mg/mL, 0.5 mL) and γ PGA-EGTA (10 mg/mL, 0.5 mL), which had equivalent Ca^{2+} binding capacities. The change of TEER for the tightness of cell monolayers was measured with a Millicell®-Electrical Resistance System (Millipore Corp., Bedford, MA, USA).

2.5. Preparation and characterization of CS/ γ PGA-EGTA NPs

A simple ionic-gelation method was used to prepare the insulin-loaded CS/ γ PGA-EGTA NPs [2]. The particle size and zeta potential of the prepared NPs were measured by DLS (3000 HS, Malvern Instruments Ltd., Worcestershire, UK), and morphology study was performed by TEM (JEOL 2010F, Tokyo, Japan). The loading efficiency (LE) and loading content (LC) of insulin in NPs were calculated using a method reported previously [15]. The internal structures of CS/ γ PGA-EGTA NPs were examined by SAXS [16] after resuspending the test NPs in phosphate buffered saline (PBS) of different pH values. Additionally, TEER measurements of Caco-2 cell monolayers incubated with CS/ γ PGA-EGTA NPs (0.02% w/v, 0.5 mL) were performed at pH values of 6.6, 7.0, or 7.4 (to simulate the pH environments in different portions of the small intestine) in the donor compartment and at pH 7.4 in the receiver compartment.

2.6. Ultra-structural changes in AJs

After treating test NPs with EGTA conjugate (CS/ γ PGA-EGTA NPs, 0.2 mg/mL) or without EGTA conjugate (CS/ γ PGA NPs), ultra-structural changes in epithelial AJs in Caco-2 cell monolayers were examined by TEM; paracellular permeability was then depicted by lanthanum nitrate staining. Lanthanum is an electron-dense rare-earth element widely used as a tracer for studying junctional barrier permeability [17]. The cell monolayers were fixed in 3.7% paraformaldehyde, washed with s-Collidine buffer, and incubated with 2% lanthanum for 2 h at room temperature. After rinsing in s-Collidine and PBS, samples were processed for TEM examination as detailed previously [18].

2.7. Translocation of AJC proteins

The Caco-2 cells were grown to confluence in 35-mm dishes with glass cover-slip bottoms and then treated with test NPs for 2 h; translocation of their AJC proteins was investigated by immunofluorescence staining. The treated cells were washed 3 times with pre-warmed PBS and then fixed and permeabilized in 100% acetone for 10 min at

–20 °C. Non-specific binding was blocked in 5% normal goat serum (Jackson ImmunoResearch Laboratories, West Grove, PA, USA) in PBS for 60 min at 37 °C. The cells were then incubated overnight in primary antibodies [mouse anti-claudin-4 (Invitrogen) and donkey anti-E-cadherin (Abcam, Cambridge, MA, USA)] at 4 °C. The results were visualized by applying appropriate Alexa-Fluor-conjugated secondary antibodies (Invitrogen); the cells were also counterstained with SYTOX blue (Invitrogen) for examination by inverted CLSM.

Extracellular calcium levels were measured by staining the treated cells with 40 mM Alizarin Red S (ARS), pH 4.0, for 10 min, and then washing with 70% ethanol and then with PBS to remove non-specific staining [19]. Samples were then visualized by an inverted phase-contrast light microscope (Nikon Eclipse-Ti, Nikon Instruments Inc., Melville, NY, USA).

2.8. Animal study

Animal studies were performed according to the “Guide for the Care and Use of Laboratory Animals” prepared by the Institute of Laboratory Animal Resources, National Research Council and published by the National Academy Press, revised in 1996.

The biodistribution of insulin orally delivered by CS/ γ PGA-EGTA NPs was studied in rats ($n = 3$) by using SPECT. In the study, ^{123}I -insulin was used to radiolabel insulin (^{123}I -insulin) using iodogen precoated tubes (Thermo Fisher Scientific, Rockford, IL, USA). Following the last SPECT scan, additional whole-body CT images of the heart, lungs, great vessels, liver and kidneys were acquired after intravenous administration of the Conray™ contrast agent (Mallinckrodt, MO, USA). To maximize the amount of contrast fluid stasis in the vasculature, the contrast agent was slowly injected via the tail vein (2 mL over 40 s). After receiving half of the Conray™ dose, each animal was sacrificed by co-injection of potassium chloride under deep anesthesia. The detailed protocols used in the image scanning and its quantification were previously described by our group [20,21]. To calculate the percentage of initial dose (% ID) within each area, the corresponding volumes of interest (VOIs) were manually drawn on the co-registered reference CT and dynamic scintigraphic images. Biodistribution data were expressed as % ID using the following formula:

$$\% \text{ ID} = \frac{\text{decay corrected radioactivity within the VOI at each time frame} \times 100\%}{\text{whole body radioactivity at the 1st image frame}}$$

Diabetes was induced in rats (Wistar, 250 g) via intraperitoneal injection of streptozotocin (STZ) as described previously [22]. The rats with induced diabetes were then fasted overnight and for the remainder of the experiment but with free access to water. The following formulations were administered individually to the diabetic rats: an enteric-coated capsule containing free-form insulin powder (30 IU/kg) and trehalose (oral); an enteric-coated capsule filled with CS/ γ PGA-EGTA NPs containing 30 IU/kg insulin (oral); and a subcutaneous (SC) injection of free-form insulin solution (5 IU/kg, $n = 6$ for each studied group). Blood sampling from the tail veins of rats was performed before administering the drug and at varying time intervals after dosing. The blood glucose levels were then measured with a glucose meter (LifeScan, Milpitas, CA, USA). Plasma

Table 1
Mean particle sizes (nm), polydisperse index, and zeta potential values (mV) of CS/ γ PGA-EGTA nanoparticles (NPs) at distinct pH environments ($n = 5$ batches). N/A: data were not available due to the observed disintegration of NPs.

pH value	Mean particle size	Polydisperse index	Zeta potential
6.0	328.6 \pm 2.3	0.23 \pm 0.02	38.7 \pm 0.2
6.6	354.3 \pm 3.1	0.25 \pm 0.05	12.1 \pm 0.4
7.0	1549.3 \pm 33.9	0.39 \pm 0.08	7.1 \pm 0.6
7.4	N/A	0.48 \pm 0.13	1.3 \pm 0.9

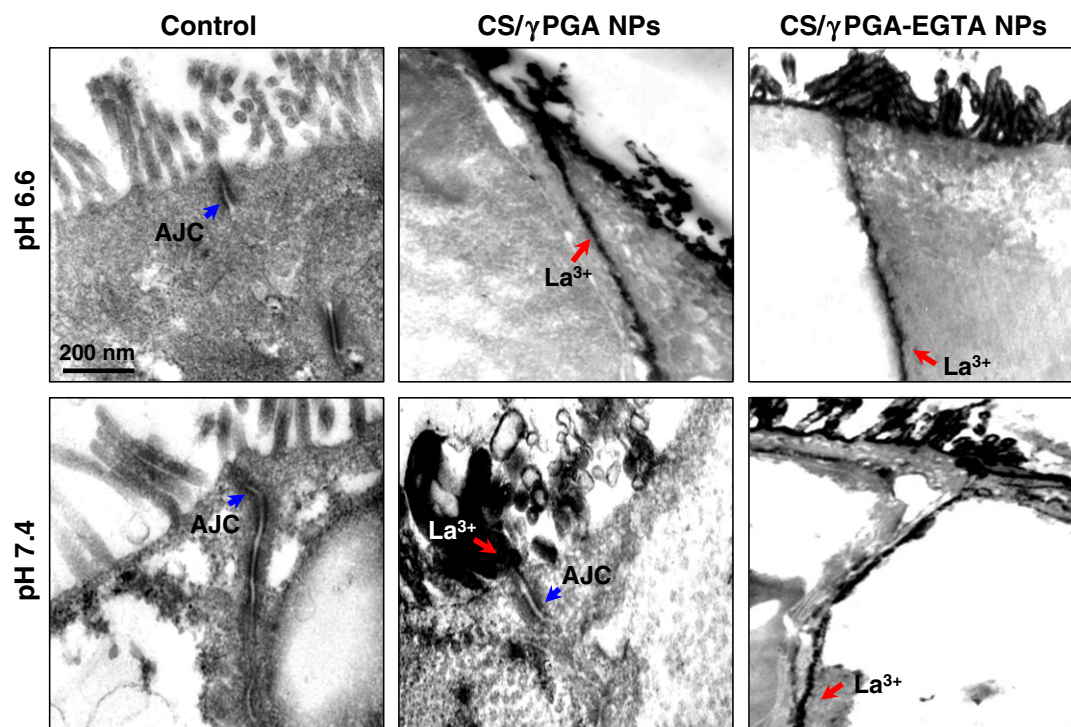


Fig. 4. TEM micrographs showing the effects of CS/γPGA NPs or CS/γPGA-EGTA NPs on the AJC opening and its subsequent paracellular permeability in the cell monolayers at pH 6.6 and 7.4. AJC: apical junctional complex; blue arrow: indicating an intact AJC, and red arrow: indicating the lanthanum (La^{3+}) permeability through the opened intercellular spaces.

insulin levels and the relative bioavailability (BA_R) of insulin after oral administration were calculated using the following formula [16].

$$\text{BA}_R = \frac{\text{AUC}_{(\text{oral})} \times \text{Dose}_{(\text{SC})}}{\text{AUC}_{(\text{SC})} \times \text{Dose}_{(\text{oral})}} \times 100\%$$

where AUC is the total area under the plasma insulin concentration versus time curve.

2.9. Statistical analysis

The two groups were compared by using statistical software (SPSS, Chicago, Ill) to perform one-tailed Student *t*-test. All data were expressed as means with standard deviations indicated (mean \pm SD). A difference of $P < 0.05$ was considered statistically significant.

3. Results and discussion

Calcium depletion by chelating agents such as EGTA is known to increase paracellular permeability [23] and has a crucial role in inhibiting intestinal enzyme activities [24]. The EGTA, a hexadentate ligand, is best known for its selectivity for Ca^{2+} over Mg^{2+} (ca. 10^6 -fold) [25]. In intestinal pH environments (pH 6.4–7.4), acidic groups of EGTA are deprotonated, which accelerates the association kinetics of Ca^{2+} [26].

To concentrate the chelating agent in areas where paracellular permeation enhancement and enzyme inhibition are needed, immobilization of this auxiliary agent on the drug-carrier matrix is crucial [27]. Our approach is to synthesize a γPGA-EGTA conjugate (Fig. 1a) to form NPs with CS for oral insulin delivery. Such a delivery system can concentrate the activities of permeation enhancement and enzyme inhibition in areas where drug release occurs in the intestinal mucus layer. This study shows that this CS/γPGA-EGTA NP system inhibits intestinal proteolytic degradation and promotes

insulin absorption by the small intestine, thus enhancing the bioavailability of insulin.

3.1. Synthesizing and characterizing γPGA-EGTA

Fig. 1b shows the FT-IR spectra of free-form γPGA, the synthesized EGTA derivative, and γPGA-EGTA. The γPGA spectra exhibited absorbance peaks at 1734 cm^{-1} for carboxyl groups, 1594 cm^{-1} for amide II, and 1570 cm^{-1} for amide I, while those of the EGTA derivative showed absorbance peaks characteristic of $-\text{COOH}$ (1733 cm^{-1}) and aromatic C–H stretching (1516 and 1406 cm^{-1}). For γPGA-EGTA, the presence of the peaks for amide bond (1634 cm^{-1}) and aromatic C–H stretching (1515 and 1404 cm^{-1}) confirms the successful conjugation of EGTA derivative on γPGA.

In further analysis of the synthesized γPGA-EGTA by ^1H NMR (Fig. 1c), the spectrum of the unmodified γPGA displayed chemical shifts characteristic of $\alpha\text{-CH}$ ($\delta = 4.01 \text{ ppm}$), $\beta\text{-CH}_2$ ($\delta = 1.96$ and 1.83 ppm) and $\gamma\text{-CH}_2$ ($\delta = 2.22 \text{ ppm}$) while that of the EGTA derivative showed shifts indicating aromatic phenyl (7.87 , 7.74 and 7.16 ppm) and methylene and primary amine protons (4.21 , 3.16 , 3.63 and 3.38 ppm). In the γPGA-EGTA conjugate, the characteristic peaks for aromatic protons (7.03 and 6.93 ppm) and methylene protons (4.31 , 3.68 and 3.85 ppm) were for EGTA derivative, and those for $\alpha\text{-CH}$ ($\delta = 4.02 \text{ ppm}$), $\beta\text{-CH}_2$ ($\delta = 1.95 \text{ ppm}$ and 1.83 ppm), and $\gamma\text{-CH}_2$ ($\delta = 2.25 \text{ ppm}$) were from γPGA. These ^1H NMR results confirm the successful conjugation of the EGTA derivative on the γPGA backbone. According to the ^1H NMR spectrum of γPGA-EGTA, the DS of EGTA derivative on γPGA was an estimated 50%, which indicated the presence of 0.5 mol of EGTA for each mole of γPGA monomer.

The complexometric titration results showed that the Ca^{2+} -binding capacity of γPGA-EGTA at neutral pH (simulating the intestinal pH environment) was $1666.7 \pm 8.3 \text{ } \mu\text{mol/g}$ conjugate ($n = 6$), which was equivalent to an EGTA-to- Ca^{2+} molar ratio of 1:1. The chelation of Ca^{2+} by γPGA-EGTA results from proton release from the EGTA structure at neutral pH [26,28].

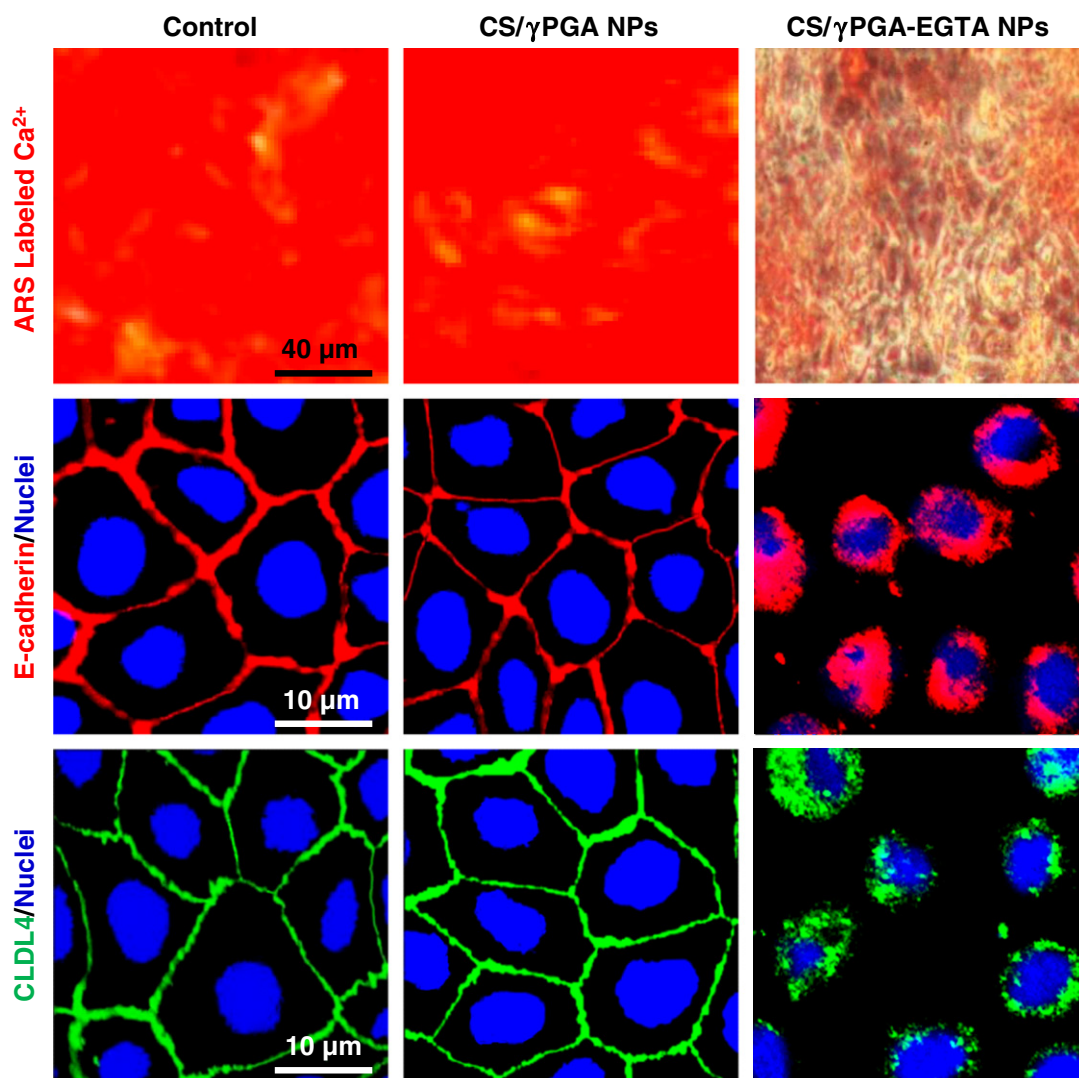


Fig. 5. Photomicrographs showing the extracellular calcium levels (as indicated by the red ARS-labeled Ca^{2+}) and the immunofluorescence staining of the cell monolayers before and after receiving CS/ γ PGA NPs or CS/ γ PGA-EGTA NPs at pH 7.4. ARS: Alizarin Red S.

The EGTA reportedly has a high binding affinity for Ca^{2+} (stability constant $\log K = 11.0$) and a relatively low affinity for Mg^{2+} ($\log K = 5.2$) [29]. The confirmation of the γ PGA-EGTA chelated with Ca^{2+} ions alone, obtained by MD simulations, is shown in Fig. 2a. The Ca^{2+} in $[\text{Ca}(\text{EGTA})]^{2-}$ anions binds to the nitrogen atoms (Ca–N), ether oxygen atoms (Ca–O), and carboxylate oxygen atoms (Ca–O) at average distances of 2.66 Å, 2.49 Å, and 2.52 Å, respectively. The $\text{Ca}^{2+}/\text{Mg}^{2+}$ discrimination of EGTA presumably stems from its binding cavity, which is the right size for Ca^{2+} but does not further envelop Mg^{2+} tightly because the carboxylate anions at each end of the chain begin butting into each other [30]. The DTPA has similar stability constants for Ca^{2+} ($\log K = 10.7$) and Mg^{2+} ($\log K = 9.3$) [29] and therefore can chelate both Ca^{2+} and Mg^{2+} ions [4].

3.2. Inhibition of intestinal enzyme degradation by γ PGA-EGTA

Insulin is very sensitive to trypsin and chymotrypsin, which are Ca^{2+} -dependent enzymes present in the intestinal fluid and mucus layer [31–33]. One strategy for inhibiting intestinal enzyme activities is removing essential metal ions such as Ca^{2+} from the enzyme structure [24]; the affinity for Ca^{2+} correlates with the strength of the enzyme inhibition. The inhibitory activity of γ PGA-EGTA against intestinal proteases was investigated in proximal intestinal tracts

freshly isolated from rats and placed in KRB, a simulated intestinal fluid [34].

Fig. 2b shows that the control group of free-form insulin was rapidly degraded by the intestinal proteolytic enzymes (approximately 90% insulin degradation within 3 h). Adding EGTA, or γ PGA-EGTA conferred a significant protective effect of insulin against proteases ($P < 0.05$). EGTA and γ PGA-EGTA conferred much larger insulin protection effects compared to their DTPA counterparts reported in our previous study (8% vs. 35% insulin degradation, respectively $P < 0.05$) [2]. These experimental results confirm that EGTA and γ PGA-EGTA are superior to their DTPA counterparts in terms of enzyme inhibition as they have specific capability to deprive Ca^{2+} from the structure of enzymes, which ultimately helps prevent insulin degradation.

3.3. Enhancement of the paracellular permeability

Calcium is critical for maintaining cell–cell junctions in various cell types [35]. The use of complexing agents for chelating extracellular Ca^{2+} can disrupt intercellular junctions and can increase paracellular permeability. The well-established *in vitro* Caco-2 cell model was used to evaluate absorption enhancement across the intestinal epithelium by assessing their TEER [36]. Fig. 2c shows that Caco-2 cell monolayers treated with free-form EGTA or γ PGA-EGTA produced a

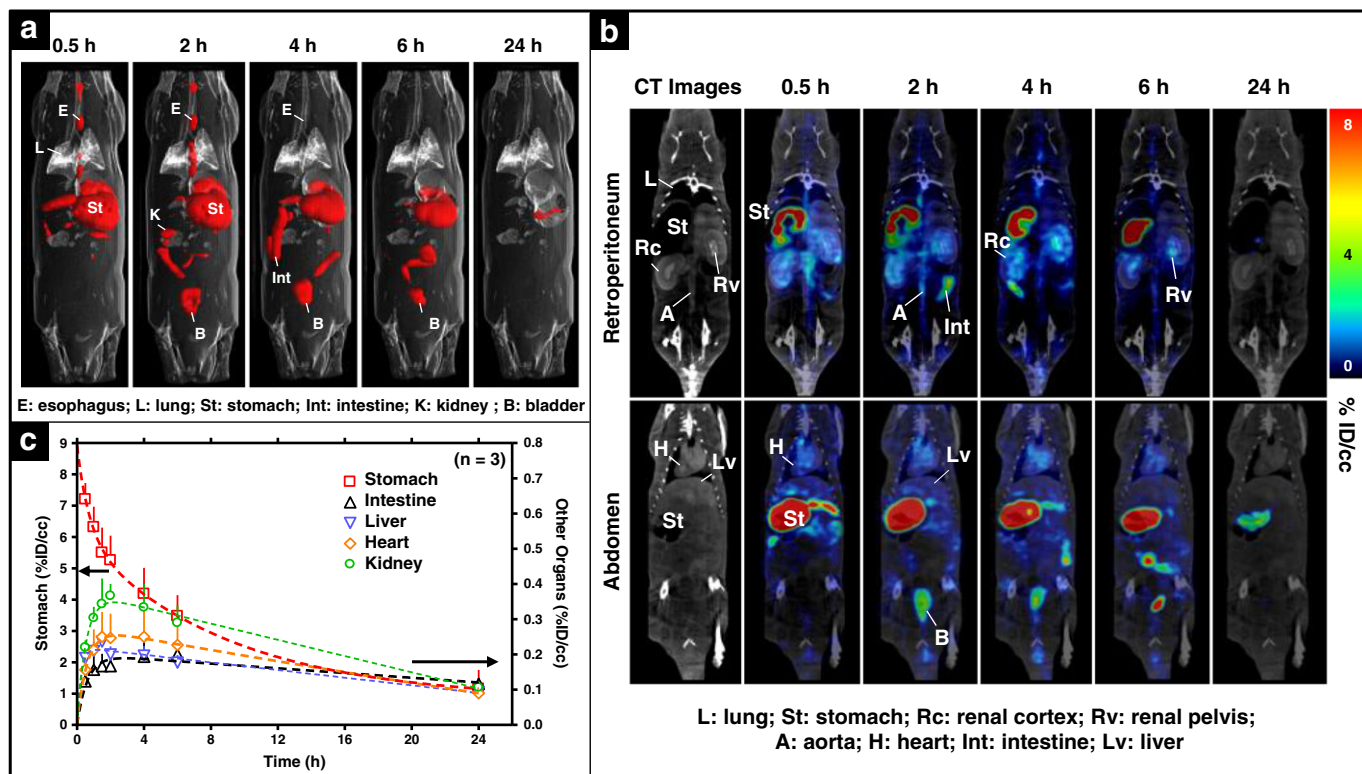


Fig. 6. (a) Soft-tissue contrast CT images (in gray) superimposed with 3D volume-rendering SPECT images (in red) showing the distribution of ¹²³I-insulin in various body compartments obtained at different time points; (b) biodistribution of ¹²³I-insulin illustrated by contrast-enhanced tomographic images. The concentrations of ¹²³I-insulin in different organs are shown in rainbow pseudo-color scale and (c) their percentages of ingested dose per volume (% ID/cm³) vs. time profiles of ¹²³I-insulin in the stomach (left y-axis) and other organs (right y-axis) are also demonstrated.

significantly larger reduction in TEER compared to the control cells ($P < 0.05$). The TEER reduction induced by EGTA and γ PGA-EGTA was larger than that induced by their DTPA counterparts [2]. Again, the larger reduction may have resulted from their binding specificity towards Ca^{2+} . After removing the test samples and replenishing the medium, TEER increased gradually and recovered almost completely within 24 h, indicating that the transient calcium depletion may allow a reversible modulation of paracellular permeability.

3.4. Characteristics of CS/ γ PGA-EGTA NPs

The CS/ γ PGA-EGTA NPs prepared in DI water (pH 6.0) had a particle size of 328.6 ± 2.3 nm and a zeta potential of 38.7 ± 0.2 mV; their insulin LE and LC were $78.7 \pm 0.4\%$ and $17.5 \pm 1.5\%$, respectively ($n = 5$ batches). The pH-sensitivity of test NPs was evaluated at distinct pH environments, representing the small intestine conditions. As pH was increased, the size of the test NPs significantly increased until they eventually disintegrated at pH 7.4 (Fig. 3a); however, their zeta potential values significantly decreased ($P < 0.05$, Table 1) due to CS deprotonation.

Fig. 3b shows the SAXS profiles in a log-log plot of CS/ γ PGA-EGTA NPs dispersed in different pH environments. Close study of the SAXS profiles of test NPs at pH 6.6 showed peak centering at about 1.1 nm^{-1} . Earlier, the authors showed that this peak is attributable to the characteristic correlation between the insulin molecules contained within the individual NP [18], where the relatively high concentration of insulin molecules in test NPs produces an interaction peak. The characteristic correlation distance between the insulin molecules calculated from the peak position via $d = 2\pi/q_m$ approximated 5.7 nm. No scattering peak occurred when the pH of the medium was increased to 7.4. At this pH value, the disintegrating test NPs released their originally embedded insulin molecules to the bulk media because CS deprotonation weakened their electrostatic attraction to

γ PGA-EGTA. Since most of the insulin molecules were not contained in the NPs, their spatial correlation peak was no longer observable in the SAXS profile.

Fig. 3c shows the TEER variations observed in the Caco-2 cell monolayers after receiving CS/ γ PGA-EGTA NPs at pH 6.6, 7.0 or 7.4. Regardless of pH exposure, all groups showed remarkably decreased TEER, indicating that the EGTA-conjugated NPs developed in the study may facilitate paracellular permeation in the pH environments from the proximal duodenum to the distal ileum (pH 6.4–7.4) along the intestinal tract. Conversely, the CS is well known to be inadequate for opening intercellular TJ's in neutral pH environments [37], which limits its potential use as a permeation enhancer only in the duodenum section.

3.5. Ultra-structural study of AJC opening and paracellular permeability

Ultra-structural study of the AJC opening and its subsequent paracellular permeability in the cell monolayers was performed by TEM after treatment with CS/ γ PGA NPs or CS/ γ PGA-EGTA NPs at pH 6.6 and 7.4. Lanthanum (La^{3+}) was used as the electron microscopic stain because its interaction with negatively charged cell-surface glycoproteins makes it effective for delineating intercellular spaces [17].

Fig. 4 shows that before treatment with test NPs (controls), the intact AJCs (as pointed by the blue arrow) appeared as highly electron-dense structures underneath the apical cell membrane and between adjacent cells. When treated with the test NPs without EGTA conjugate (CS/ γ PGA NPs) at pH 6.6, lanthanum staining of the paracellular spaces between adjacent cells was clearly observed (red arrow), suggesting the opening of intercellular junctions mediated by CS [18]. At pH 7.4, the AJCs remained intact, and limitation of the lanthanum staining to the brush-border surface suggested that CS/ γ PGA NPs do not effectively enhance permeation at neutral pH values. Whereas CS (pKa 6.5) facilitates paracellular transport only

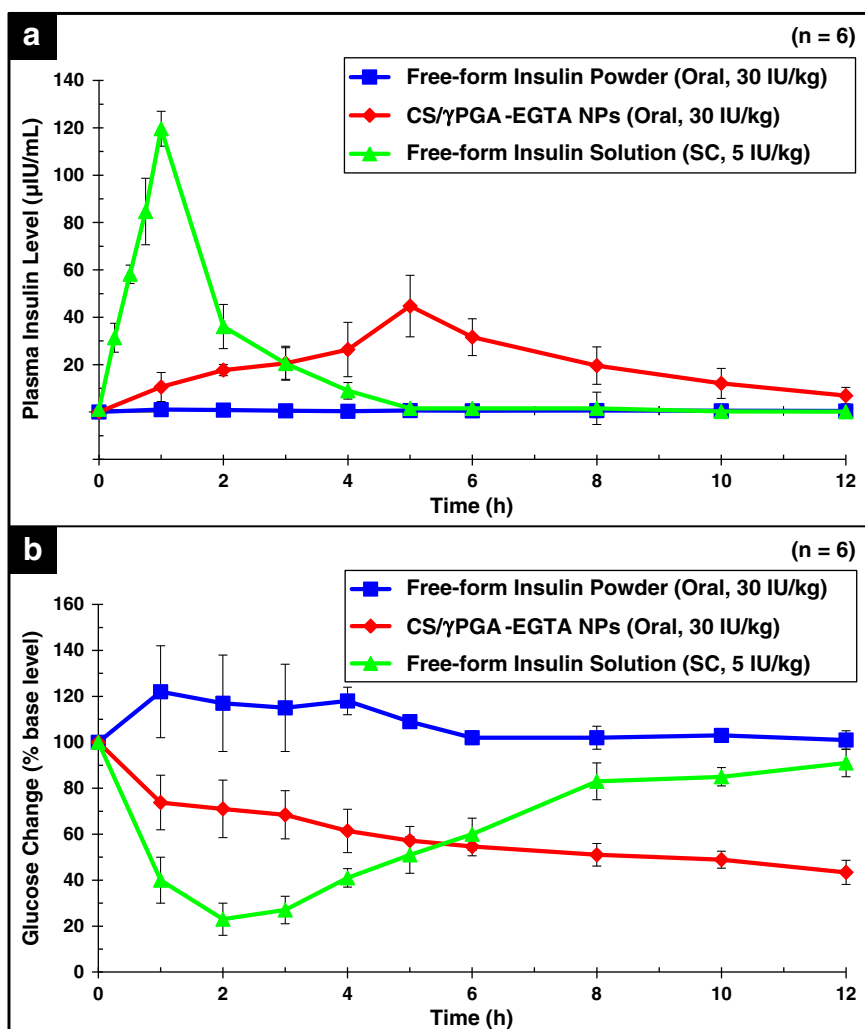


Fig. 7. (a) Plasma insulin level vs. time profiles and (b) blood glucose change vs. time profiles of diabetic rats treated with different formulations of insulin. Oral: oral administration and SC: subcutaneous injection.

after protonation [3], treatment with the EGTA-conjugated (CS/γPGA-EGTA) NPs caused the opening of AJs and significant penetration of lanthanum into the lateral intercellular spaces at both test pH environments, possibly due to the deprivation of extracellular Ca^{2+} by EGTA. Calcium depletion by EGTA reportedly increases paracellular permeability [29].

3.6. AJC disassembly induced by calcium depletion

The AJC, which is comprised of adherens junctions (AJs) and TJs, is an important regulator of cell structure and function [38]. AJs bring two plasma membranes together physically and enhance TJ assembly; thus, alteration of AJs modulates the TJ structure and the epithelial paracellular barrier function. *E-cadherin*, a Ca^{2+} -dependent adhesion molecule, is the major *trans-membrane protein* of AJs, and claudin-4 (CLDN4) is a TJ protein [3,39]. Neither AJs nor TJs have a static structure; they can be rapidly disassembled and reorganized in response to various extracellular stimuli.

Fig. 5 shows the extracellular calcium levels (as indicated by the red ARS-labeled Ca^{2+}) and the immunofluorescence staining of the cell monolayers before and after receiving CS/γPGA NPs or CS/γPGA-EGTA NPs at pH 7.4. Extracellular Ca^{2+} is essential for AJC formation and for maintaining cell–cell junctions [40]. Compared to the control group, treatment with test NPs without EGTA conjugate

(CS/γPGA NPs) did not significantly affect either extracellular calcium levels or translocation of E-cadherin and CLDN4.

In contrast, treatment with EGTA-conjugated NPs (CS/γPGA-EGTA NPs) substantially diminished extracellular calcium levels; additionally, internalization of both E-cadherin (red color) and CLDN4 (green color) to a cytosolic compartment surrounding the nucleus suggested that calcium depletion by EGTA stimulates endocytosis of AJC components. Internalization of AJs and TJs via a clathrin-dependent pathway appears to be a common mechanism to rapidly modulate the cell–cell sealing, which has been documented in the literature [40].

3.7. Biodistribution of insulin

Fig. 6a presents the radioactivity distribution of ^{123}I -insulin orally delivered by CS/γPGA-EGTA NPs over the 24-h period after ingestion in a rat model. The 3D volume-rendering SPECT images of ^{123}I -insulin (in red) were superimposed onto the soft-tissue contrast-volume-rendering CT images (in gray). Air in the hollow organs, including the bilateral lungs, stomach and intestine, was clearly visible in the volume-rendering CT images, whereas ^{123}I -insulin was identified in the GI track, kidneys and bladder via the SPECT images.

Additional details of the biodistribution of ^{123}I -insulin are visible in the coronal tomographic images (Fig. 6b), in which the radioactivity concentration of ^{123}I -insulin is presented in rainbow pseudo-color scale and superimposed on the CT images to distinguish anatomic

Table 2

Pharmacokinetic parameters of insulin in diabetic rats after subcutaneous (SC) administration of free-form insulin solution, oral administration of capsules containing free-form insulin powder or insulin-loaded CS/ γ PGA-EGTA NPs. C_{\max} : maximum plasma concentration; T_{\max} : time at which C_{\max} is attained; $AUC_{(0-12\text{ h})}$: area under the plasma concentration–time curve; BA_R : relative bioavailability ($n=6$).

	SC free-form insulin solution	Oral free-form insulin powder	Oral CS/ γ PGA-EGTA NPs
Dose (IU/kg)	5.0	30.0	30.0
C_{\max} ($\mu\text{U/mL}$)	119.63 ± 3.8	1.8 ± 0.7	44.7 ± 13.0
T_{\max} (h)	1.0	5.0	5.0
$AUC_{(0-12\text{ h})}$ ($\mu\text{U h/mL}$)	191.31 ± 30.7	6.5 ± 2.7	244.5 ± 33.7
BA_R (%)	100	0.5 ± 0.3	21.3 ± 1.5

localizations. The vascular structures, including the heart, aorta, renal cortex, renal pelvis and liver were visible in the contrast-enhanced CT images, while ^{123}I -insulin was clearly recognizable in the above-mentioned organs in the SPECT images. Fig. 6c shows the radioactivity concentrations of ^{123}I -insulin ($\% \text{ID}/\text{cm}^3$) observed in different organs. After ingestion, the radioactivity concentration of ^{123}I -insulin present in the stomach diminished gradually, while those in the intestine, liver, heart and kidneys increased with time, reaching their maximum values at 3–4 h following ingestion, revealing the absorption of insulin into the systemic circulation.

3.8. In vivo PK and PD profiles

The SC injection of free-form insulin solution resulted in a maximum plasma concentration at 1 h post administration. Blood glucose levels then sharply decreased within 1–3 h before returning to their basal levels (Fig. 7a and b). Subsequent oral administration of capsules

containing free-form insulin powder produced negligible insulin absorption and hypoglycemic effects. Conversely, administration of capsules containing insulin-loaded CS/ γ PGA-EGTA NPs produced a slow but prolonged reduction in blood glucose levels, indicating that the pharmacological activity of the absorbed insulin was still intact; their $AUC_{(0-12\text{ h})}$ was $244.5 \pm 33.7 \mu\text{U h/mL}$, and the corresponding BA_R was $21.3 \pm 1.5\%$ (Table 2).

3.9. Functionality of γ PGA-EGTA

Fig. 8 schematically depicts the functionality of γ PGA-EGTA in facilitating oral insulin delivery according to the above results. By specific deprivation of extracellular Ca^{2+} from the intestinal lumen (Figs. 2a and 5), γ PGA-EGTA can significantly reduce the enzymatic degradation of insulin (Fig. 2b). Additionally, *via* causing the endocytosis AJCs (Fig. 5), γ PGA-EGTA displays the ability to modulate the transepithelial resistance (Fig. 2c), enhance the paracellular permeation (Fig. 4), and promote the insulin absorption (Fig. 6a–c), thus increasing its oral bioavailability (Fig. 7a and b).

4. Conclusions

A calcium-specific depletion agent, γ PGA-EGTA, was successfully synthesized to formulate a carrier for oral insulin delivery. Our results indicate that γ PGA-EGTA plays a significant role in the prevention of enzymatic degradation of insulin and in the enhancement of its paracellular permeability through the modulation of epithelial AJC disassembly, ultimately producing a significant and prolonged hypoglycemic effect. These results confirm the efficacy of the proposed EGTA-conjugated carrier in promoting the bioavailability of orally administered insulin.

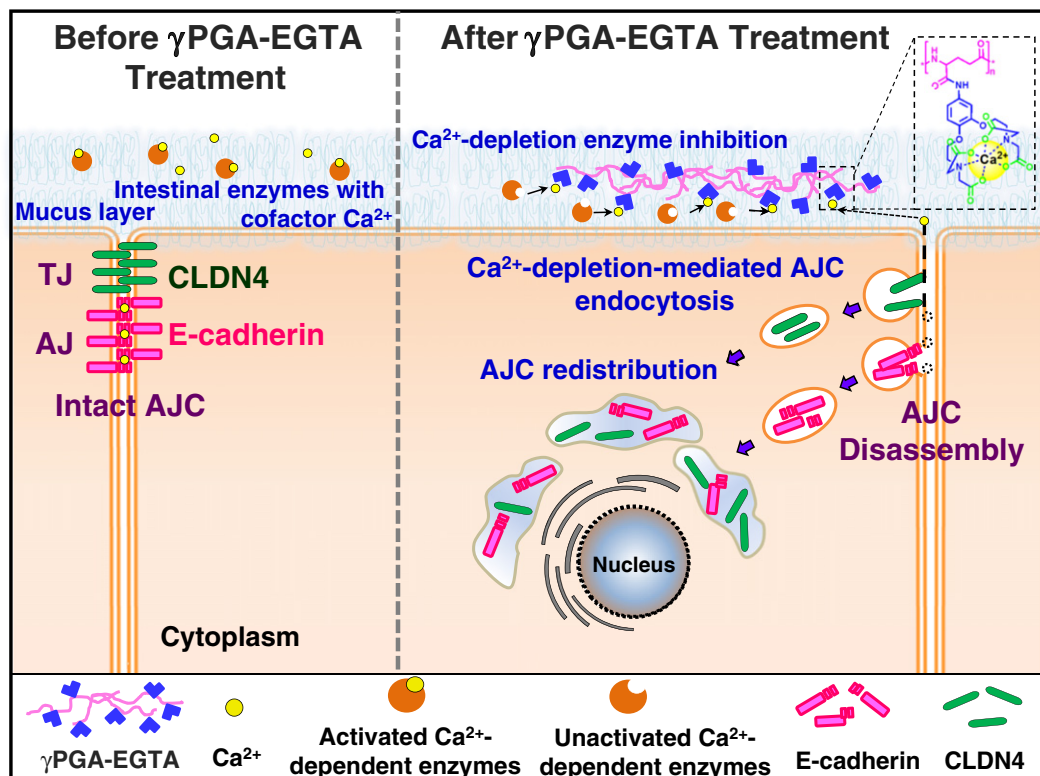


Fig. 8. Schematic illustrations displaying the specific deprivation of extracellular Ca^{2+} from the intestinal lumen by γ PGA-EGTA and its subsequent effects on the inhibition of intestinal proteases and the enhancement of paracellular permeation.

Acknowledgments

This work was supported by a grant from the National Science Council (NSC 100-2120-M-007-003), Taiwan, Republic of China. The molecular-imaging study was partially supported by a grant from Chang Gung Memorial Hospital (CMRPG391512).

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