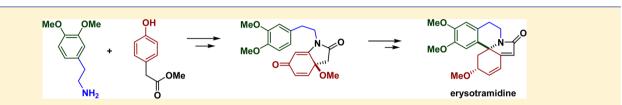
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Synthesis of the Erythrina Alkaloid Erysotramidine

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S Supporting Information



ABSTRACT: A concise synthesis of erysotramidine (an alkaloid belonging to the erythrina family) was achieved starting with an inexpensive phenol and amine derivative. The synthesis is based on oxidative phenol dearomatizations mediated by a hypervalent iodine reagent and includes a novel route to a key indolinone moiety.

E rythrina alkaloids¹ are natural tetracyclic compounds **1** isolated from wide variety of tropical plants (Figure 1). The

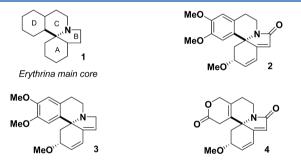
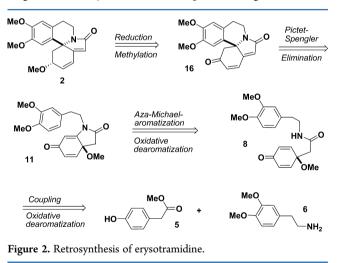


Figure 1. Erythrina alkaloid members.

structures contain an aza-spiran ring and exhibit hypotensive, sedative, and anticonvulsive properties as well as curare-like effects.² The compounds may be divided into aromatic structures such as erysotramidine³ **2** and erysotrine^{2c} **3** and nonaromatic structures such as β -erythroidine **4**.⁴ The publication of several elegant syntheses has generated widespread interest in these compounds in the scientific community.^{5,6}

In this note, we describe a concise preparation of erysotramidine, an alkaloid belonging to the aromatic erythrina family, in eight steps.

Our approach employed a hypervalent iodine reagent⁷ to promote dearomatization of phenol subunits into advanced functionalized intermediates. Hypervalent iodine reagents are often used in synthetic procedures because these environmentally benign and inexpensive reagents reduce the need for toxic heavy metals. Transformations involving hypervalent iodine have already demonstrated their remarkable synthetic utility, as described in the pioneering work of Kita and coworkers.⁸ Our synthesis involved two oxidative dearomatizations, a new tandem aza-Michael-rearomatization process to produce a key indolinone moiety, a Pictet–Spengler cyclization,⁹ a stereoselective reduction, and an etherification. One objective of this note is to illustrate the effectiveness of hypervalent iodine reagents in total synthesis of natural products, Figure 2.

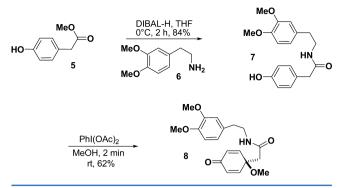


The synthesis began with commercially available phenol **5** and amine **6**, which were joined through an amide linkage to produce 7 in 84% yield using a reaction promoted by an aluminum salt.¹⁰ Amide 7 was treated in methanol with (diacetoxyiodo)-benzene (DIB) to induce the first oxidative dearomatization, resulting in the functionalized prochiral dienone **8** in 62% yield. An interesting aspect of this transformation is its ability to reconfigure the inert unsaturations of the phenol moiety 7 into the oxidized intermediate **8**, in which the unsaturated bonds are readily functionalized, Scheme 1.

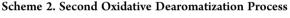
Treatment of dienone 8 with TMS-OTf followed by $BF_3 \cdot Et_2O$ led directly to indolinone 10. This transformation took place through silyl activation of the enone functionality, enabling a 1,4

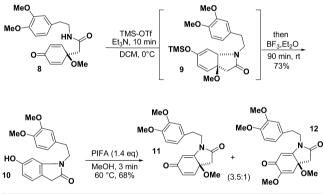
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Scheme 1. First Oxidative Dearomatization Process



attack via a Michael process to afford the bicyclic intermediate 9, which was subsequently transformed into the aromatic compound 10 through further Lewis acid activation of the methoxy group. The transformation of compound 7 into 10 represents a new two-step route to indolinone derivatives through the equivalent of a C-H activation mediated by a hypervalent iodine reagent,¹¹ Scheme 2. The first step of this





sequence is related to the noteworthy stereoselective process developed by Wipf and Kim to produce hydroindole cores.¹² It should be noted that the prochiral dienone 8 obtained during the first activation could be used in an enantioselective synthesis of the erythrina alkaloids if the first aza-Michael process was enantioselectively controlled.¹³ Unfortunately, in our hands, all asymmetric 1,4-addition attempts failed. At this stage, a second aromatic activation mediated by bis(trifluoroacetoxy)iodobenzene (PIFA) led to a separable mixture of the desired dienone 11 and a similar byproduct 12 bearing one more methoxy group in 68% yield, in a favorable ratio (3.5:1) of compound 11. Formation of compound 12 could be explained by a first methanol addition, during the oxidative activation, in ortho position instead of the required para position followed by a rearomatization process that would lead to an ortho-methoxy analogue of compound 10 that would be then oxidized into byproduct 12, Scheme 2.

The main tetracyclic core 13 was obtained under acidic conditions in 71% yield. The transformation resembled a Pictet–Spengler rearrangement, which is a popular method of producing the aza-spiran moiety of the erythrina alkaloids.⁹ The ketone functionality may be selectively reduced at this point using a hydride reagent. We had supposed that the bowl-shape generated by rings A and B would control the stereoselectivity, but unfortunately hydride reduction using LiBH₄, Red-Al, or L-

Selectride primarily produced the undesired diastereomer 14 (9:1 by NMR). Interestingly, and for reasons that are unclear, only Luche's conditions resulted in a favorable ratio (2.4:1 by NMR) of the desired diastereomer 15, ¹⁴ Scheme 3.

Scheme 3. Formation of the Main Tetracyclic System MeC MeC H₃PO₄ D С 80°C, 3 h. MeO 11 А 71% OMe ั๊กM_ MeC MeO Reduction MeO MeC 5 min HO HO OMe 15 ОМе 14 Solvent yield (%) Reagent Temp 14:15 -78°C LiBH₄ THF 9:1 63

0°C

1:2.4

82

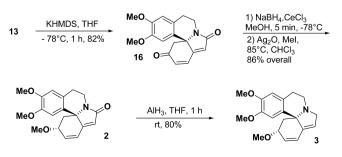
TFE

NaBH₄-CeCl₃

If compound 15 represented a potential precursor of erysotramidine 2, then the stereoselectivity observed during the reduction process was not good and required optimization. Consequently, the transformation of compound 13 into the desired target was improved by delaying the ketone reduction in order to produce the desired alcohol stereocenter with greater stereoselectivity at the end of the synthesis. Consequently, the polyconjugated system appearing in the structure of 2 was first obtained by treatment of compound 13 with KHMDS, leading to the flattened structure 16 through an E1cB mechanism in 82% yield. It should be noted that the conformation of the polyconjugated scaffold 16 is more conducive to the desired stereochemistry than its precursor 13, and treatment of 16 with a hydride provided the desired alcohol with a diastereoselectivity of 9:1 by NMR. The synthesis of erysotramidine 2 was concluded with a mild and quantitative methylation in the presence of Ag₂O and iodomethane in 86% yield overall from compound 16. The NMR and mass spectrometry data obtained were identical with the data reported in the literature.⁶ⁱ It should be noted that erysotrine $\overline{3}$, a relative alkaloid of the same family, may be obtained by further treatment of erysotramidine 2 with AlH₃ in 80% yield, as previously reported in the literature,¹⁵ Scheme 4.

A natural product belonging to the erythrina family was synthesized from inexpensive starting materials. The synthesis demonstrated the power and utility of hypervalent iodine chemistry in total synthesis and highlighted the development of a novel method for producing indolinone cores.

Scheme 4. Syntheses of Erysotramidine and Erysotrine



EXPERIMENTAL SECTION

Unless otherwise indicated, ¹H and ¹³C NMR spectra were recorded at 300 and 75 MHz, respectively, in CDCl₃ solutions. Chemical shifts are reported in ppm on the δ scale. Multiplicities are described as s (singlet), d (doublet), dd, ddd, etc. (doublet of doublets, doublet of doublets of doublets, etc.), t (triplet), q (quartet), quin (quintuplet), m (multiplet), and further qualified as app (apparent), br (broad). Coupling constants, *J*, are reported in Hz. HRMS were measured in the electrospray (ESI) mode on a LC-MSD TOF mass analyzer.

N-(3,4-Dimethoxyphenethyl)-2-(4-hydroxyphenyl)**acetamide (7).** To a mixture of 2-(3,4-dimethoxyphenyl)ethanamine 6 (6.16 mmol, 1.04 mL, 2.8 equiv) in anhydrous THF (17 mL, 0.1 M) at 0 °C was added dropwise DIBAL-H in hexane (1.0 M, 6.65 mmol, 6.65 mL, 3.0 equiv). The solution was stirred for 10 min at 0 °C and then was allowed to warm to room temperature. The mixture was stirred for 1.5 h, a solution of methyl 2-(4-hydroxyphenyl)acetate 5 (2.2 mmol, 368.2 mg, 1.0 equiv) in THF (5 mL) was added, and the mixture was stirred overnight. The reaction was quenched by addition of 1.0 M HCl, and the organic layer was washed with brine and ethyl acetate and dried over sodium sulfate. The crude was purified by flash silica gel chromatography (65 \rightarrow 90% ethyl acetate/hexane) to give 579.5 mg of product 7 (84% yield) as a white solid. mp 156 °C; ¹H NMR (300 MHz, DMSO d_6) δ 9.25 (br, 1H), 7.97 (br, 1H), 7.04 (d, J = 7.8 Hz, 2H), 6.86 (d, J =8.0 Hz, 1H), 6.80 (s, 1H), 6.70 (d, J = 8.0 Hz, 3H), 3.74 (s, 6H), 3.41 (s, 2H, rotamer), 3.27 (t, J = 6.9 Hz, 2H), 2.66 (t, J = 6.9 Hz, 2H); ¹³C NMR (75 MHz, DMSO-d₆) δ 170.4, 155.7, 148.4, 147.1, 131.8, 129.7, 126.4, 120.3, 114.8, 112.4, 111.7, 55.4, 55.2, 41.5, 40.3, 34.5; HRMS (ESI) *m/z*: calcd for C₁₈H₂₂NO₄ (M + H)⁺, 316.1543; found, 316.1536.

N-(3,4-Dimethoxyphenethyl)-2-(1-methoxy-4-oxocyclohexa-2,5-dien-1-yl)acetamide (8). To a solution of compound 7 (0.944 mmol, 297.7 mg, 1.0 equiv) in methanol (7.4 mL) at 0 °C was added a solution of DIB (1.322 mmol, 425.6 mg, 1.4 equiv) in methanol (2.0 mL). The reaction was stirred for 5 min and was filtered in silica gel with 20% MeOH in DCM. The crude was purified by flash silica gel chromatography (80% ethyl acetate/hexane) to provide a yellow oil (201.2 mg, 62% yield). ¹H NMR (300 MHz, CDCl₃) δ 6.78 (d, *J* = 8.6 Hz, 1H), 6.72 (c, 2H), 6.67 (d, *J* = 10.2 Hz, 2H), 6.40 (t, *J* = 6.6 Hz, 1H), 6.29 (d, *J* = 10.2 Hz, 2H), 3.84 (s, 3H), 3.82 (s, 3H), 3.52 (q, *J* = 6.6 Hz, 2H), 3.08 (s, 3H), 2.76 (t, *J* = 6.6 Hz, 1H), 2.45 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 184.3, 167.7, 148.9, 148.3, 147.6, 131.4, 131.0, 120.5, 111.7, 111.1, 72.9, 55.7, 52.9, 49.8, 46.4, 40.3, 34.9; HRMS (ESI) *m/z*: calcd for C₁₁9H₂₄NO₅ (M + H)⁺, 346.1649; found, 346.1650.

1-(3,4-Dimethoxyphenethyl)-6-hydroxyindolin-2-one (10). Dienone 8 (0.037 mmol, 12.8 mg, 1.0 equiv) and anhydrous triethylamine (0.11 mmol, 15.4 μ L, 3.0 equiv) were dissolved in anhydrous DCM (7.0 mL, 0.2 M). TMSOTf (0.092 mmol, 16.8 µL, 2.5 equiv) was added at 0 °C, and the reaction mixture was allowed to warm to room temperature. The reaction was followed by TLC until the starting material disappeared (10 min), and BF₃·Et₂O was then added (0.11 mmol, 14.0 μ L, 3.0 equiv) during 90 min. An aqueous solution of ammonium chloride was added, and the organic layer was washed with brine and dried over sodium sulfate. The product was purified by flash column chromatography (60% ethyl acetate/hexane) to provide a yellow oil 10 (7.6 mg, 68% yield). ¹H NMR (300 MHz, CDCl₃) δ 7.05 (d, J = 7.8 Hz, 1H), 6.78 (s, 1H), 6.76 (d, J = 14.6 Hz, 1H), 6.38 (dd, J = 8.0, 2.1 Hz, 1H), 6.47 (d, J = 2.1 Hz, 1H), 3.86 (t, J = 8.0 Hz, 1H), 3.85 (s, 3H), 3.83 (s, 3H), 3.43 (s, 3H), 2.89 (t, J = 8.0 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 175.8, 156.1, 148.9, 147.7, 145.6, 130.6, 125.0, 120.7, 116.0, 112.0, 111.3, 108.3, 97.1, 55.9, 55.8, 41.6, 35.1, 33.1; HRMS (ESI) m/z: calcd for C₁₈H₂₀NO₄ (M + H)⁺, 314.1387; found, 314.1379

1-(3,4-Dimethoxyphenethyl)-3a-methoxy-3,3a-dihydro-1*H***-indole-2,6-dione (11).** To a solution of compound **10** (0.056 mmol, 17.6 mg, 1.0 equiv) in methanol (0.4 mL) at 60 °C was added a solution of PIFA (0.078 mmol, 33.7 mg, 1.4 equiv) in methanol (0.4 mL). The reaction was stirred for 5 min and was filtered on silica gel with 20% MeOH in DCM. The crude was purified by flash silica gel chromatography (40% ethyl acetate/hexane) to provide a brown oil (11.2 mg, 58% yield) of compound **11** and a dark yellow oil (2.0 mg, 10% yield) of compound **12.** ¹H NMR (300 MHz, CDCl₃) δ 6.76 (d, *J* =

8.0 Hz, 4H), 6.70 (d, J = 10.0 Hz, 1H), 6.68 (s, 1H), 6.53 (d, J = 10.0 Hz, 1H), 6.34 (dd, J = 10.0, 1.4 Hz, 1H), 5.78 (d, J = 1.4 Hz, 1H), 3.90 (m, 1H), 3.82 (s, 3H), 3.80 (s, 3H), 3.52 (m, 1H), 3.35 (dd, J = 8.7, 5.1 Hz, 1H), 2.91 (s, 3H), 2.80 (t, 7.6 Hz, 2H), 2.72 (d, 16.2 Hz, 1H), 2.56 (d, 16.2 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 184.6, 173.1, 158.0, 148.8, 147.7, 136.9, 133.5, 129.5, 120.5, 111.6, 111.1, 103.9, 72.7, 55.7, 50.9, 42.1, 41.3, 32.6; HRMS (ESI) m/z: calcd for C₁₉H₂₂NO₅ (M + H)⁺:344.1492; found: 344.1484.

1-(3,4-Dimethoxyphenethyl)-3,3a-dihydro-3a,5-dimethoxy-1H-indole-2,6-dione (12). ¹H NMR (300 MHz, CDCl₃) δ 6.80 (d, J = 8.0 Hz, 4H), 6.74 (dd, J = 8.2, 2.0 Hz, 1H), 6.69 (d, J = 2.0 Hz, 1H), 6.37 (t, J = 2.1 Hz, 1H), 5.52 (s, 1H), 3.87 (s, 3H), 3.85 (s, 3H), 3.78 (t, J = 8.0 Hz, 2H), 3.39 (s, 6H), 3.30 (d, J = 2.1 Hz, 2H), 2.82 (t, 8.0 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 191.4, 172.7, 154.6, 149.0, 148.0, 129.6, 129.3, 128.5, 120.7, 111.8, 111.4, 97.2, 91.9, 55.9, 55.8, 50.2 (*2), 42.0, 32.4, 32.1,; HRMS (ESI) m/z: calcd for C₂₀H₂₃NNaO₆ (M + Na)⁺, 396.1418; found, 396.1419.

(4aS,13b*R*)-4a,11,12-Trimethoxy-4a,5,8,9-tetrahydro-1*H*-indolo[7a,1-a]isoquinoline-2,6-dione (13). The compound 11 (0.088 mmol, 30.0 mg, 1.0 equiv) and phosphoric acid 85% (11.1 mL) were stirred at reflux for 3 h. The mixture was poured into a 2 M NaOH solution, extracted with DCM, washed with water, and dried with Na₂SO₄. The residue was then purified by silica gel chromatography with 100% ethyl acetate to give enone 13 (21.2 mg, 71%) as a brown oil. ¹H NMR (300 MHz, CDCl₃) δ 7.12 (d, *J* = 10.5 Hz, 1H), 6.59 (d, *J* = 10.5 Hz, 1H), 6.27 (d, *J* = 10.6 Hz, 1H), 4.31 (ddd, *J* = 13.0, 8.0, 2.2 Hz, 1H), 3.82 (s, 3H), 3.68 (s, 3H), 3.12 (d, *J* = 16.2 Hz, 1H), 3.11 (m, 1H), 3.03 (s, 3H), 2.96 (d, *J* = 17.2 Hz, 1H), 2.91 (m, 1H), 2.81 (d, *J* = 17.2 Hz, 1H), 2.70 (d, *J* = 16.2 Hz, 1H), 2.69 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 195.1, 168.6, 148.2, 148.1, 147.3, 128.5, 126.8, 125.8, 111.7, 108.6, 78.8, 68.5, 55.6, 51.6, 51.4, 41.3, 35.1, 28.0; HRMS (ESI) *m*/*z*: calcd for C₁₉H₂₂NO₅ (M + H)*:344.1492; found: 344.1487.

(25,4a5,13bR)-2-Hydroxy-4a,11,12-trimethoxy-4a,5,8,9-tetrahydro-1H-indolo[7a,1-a]isoquinolin-6(2H)-one (14). To a solution of 13 (9 mg, 0.026 mmol, 1.0 equiv) in trifluoroethanol (0.26 mL) at 0 °C was added CeCl₃·7H₂O (0.11 mmol, 20 mg, 4.1 equiv) and then NaBH₄ (0.110 mmol, 2.1 mg, 4.1 equiv). The solution was allowed to room temperature, and the reaction was followed by TLC. The reaction was quenched by a solution of NH₄Cl, and the organic layer was washed with brine and DCM and dried over sodium sulfate. The residue was then purified by silica gel chromatography (2% MeOH in DCM) to give the compound 14 (5.2 mg, 58%). ¹H NMR (300 MHz, CDCl₃) δ 7.07 (s, 1H), 6.58 (s, 1H), 6.06 (s, 2H), 4.28 (t, J = 6.0 Hz, 1H), 4.13 (ddd, *J* = 13.1, 6.5, 3.3 Hz, 1H), 3.85 (s, 3H), 3.83 (s, 3H), 3.26 (m, 1H), 3.18 (s, 3H), 3.05 (m, 1H), 2.69 (d, J = 16.9 Hz, 1H), 2.68 (m, 1H), 2.49 (d, J = 16.9 Hz, 1H), 2.34 (dd, J = 14.3, 5.6 Hz, 1H), 2.19 (dd, J = 14.3, 5.6 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 172.2, 147.9, 146.7, 132.6, 130.0, 129.0, 126.5, 111.4, 77.8, 65.1, 64.2, 55.8, 55.6, 51.5, 43.0, 41.8, 35.9, 27.3; HRMS (ESI) m/z: calcd for $C_{19}H_{24}NO_5$ (M + H)⁺, 346.1649: found. 346.1642.

(2*R*,4aS,13b*R*)-2-Hydroxy-4a,11,12-trimethoxy-4a,5,8,9-tetrahydro-1*H*-indolo[7a,1-a]isoquinolin-6(2*H*)-one (15). Pale yellow oil: 2.2 mg, 24%; ¹H NMR (300 MHz, CDCl₃) δ 6.83 (s, 1H), 6.64 (s, 1H), 6.11 (dd, *J* = 10.4, 1.9 Hz, 1H), 6.02 (dt, *J* = 10.4, 1.9 Hz, 1H),4.39 (br, 1H), 4.13 (m, 1H), 3.86 (s, 3H), 3.78 (s, 3H), 3.40 (m, 1H), 2.98 (s, 3H), 2.90 (m, 2H), 2.85 (d, *J* = 16.9 Hz, 1H), 2.72 (dd, *J* = 16.9, 1.2 Hz, 1H), 2.57 (ddd, *J* = 12.4, 5.1, 1.2 Hz, 1H), 1.73 (dd, *J* = 12.4, 10.2 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 169.7, 148.1, 146.8, 132.1, 130.7, 127.1, 125.6, 111.5, 109.9, 77.7, 66.4, 65.2, 55.9, 55.7, 51.1, 45.1, 41.7, 34.9, 27.4; HRMS (ESI) *m*/*z*: calcd for C₁₉H₂₄NO₅ (M + H)⁺, 346.1649; found, 346.1653.

(S)-11,12-Dimethoxy-8,9-dihydro-1*H*-indolo[7a,1-a]isoquinoline-2,6-dione (16). To a solution of 15 (0.0273 mmol, 9.4 mg, 1.0 equiv) in anhydrous THF (0.4 mL) at -78 °C was added KHMDS (0.5 M in toluene, 0.06 mmol, 0.12 mL, 2.2 equiv) dropwise. The reaction was followed by TLC, and after completion, a solution of NH₄Cl was added. The organic layer was washed with brine and AcOEt and dried with Na₂SO₄. The product was purified by silica gel chromatography (5% MeOH in DCM) to give the compound 16 (7.0 mg, 82%) as a yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.75 (d, J = 10.1 Hz, 1H), 6.84 (s, 1H), 6.65 (s, 1H), 6.41 (d, J = 10.1 Hz, 1H), 6.38 (s, 1H), 4.21 (ddd, J = 11.6, 4.9, 1.7 Hz, 1H), 3.83 (s, 3H), 3.73 (s, 3H), 3.39 (m, 1H), 3.27 (d, J = 15.1 Hz, 1H), 3.02 (m, 1H), 2.82 (m, 1H), 2.79 (d, J = 15.1 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 195.2, 169.7, 154.7, 148.7, 147.2, 138.4, 131.8, 128.0, 125.9, 125.5, 112.4, 107.8, 67.7, 55.9, 55.8, 52.4, 36.9, 27.6; HRMS (ESI) m/z: calcd for C₁₈H₁₈NO₄ (M + H)⁺, 312.1230; found, 312.1235.

(2R,13bS)-2-Hydroxy-11,12-dimethoxy-8,9-dihydro-1Hindolo[7a,1-a]isoguinolin-6(2H)-one (17). To a solution of 16 (0.0112 mmol, 3.5 mg, 1.0 equiv) in methanol (0.12 mL) at $-78 \degree \text{C}$ was added CeCl₃·7H₂O (0.046 mmol, 17.1 mg, 4.1 equiv) and then NaBH₄ (0.045 mmol, 1.7 mg, 4.0 equiv). After 5 min, the reaction was quenched by aqueous NH₄Cl, and the organic layer was washed with brine and DCM and dried over sodium sulfate. The residue may be purified by silica gel chromatography (2% MeOH in chloroform) to afford compound 17 (3.1 mg, 90%). However, this intermediate was airsensitive and was rapidly used for the further transformation. ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 6.87 \text{ (dd, } J = 10.1, 2.4 \text{ Hz}, 1\text{H}), 6.79, (s, 1\text{H}), 6.71$ (s, 1H), 6.30 (d, J = 10.2 Hz, 1H), 6.03 (s, 1H), 4.30 (br, 1H), 3.97 (m, 1H), 3.85 (s, 3H), 3.75 (s, 3H), 3.60 (m, 1H), 3.07 (m, 1H), 2.96 (dd, J = 15.9, 1.9 Hz, 1H), 2.81 (dd, J = 11.2, 4.4 Hz, 1H), 1.70 (dd, J = 11.2, 10.4 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 170.9, 156.9, 148.5, 147.0, 138.9, 128.4, 126.4, 123.7, 120.3, 112.1, 108.0, 66.5, 55.9, 45.0, 37.4, 27.0; HRMS (ESI) m/z: calcd for C₁₈H₂₀NO₄ (M + H)⁺, 314.1387; found, 314.1392.

Erysotramidine (2). To a degassed solution of 17 (0.01 mmol, 3.1 mg, 1.0 equiv) in anhydrous acetonitrile (0.2 mL) were added Ag₂O (0.05 mmol, 11.6 mg, 5.0 equiv) and MeI (0.05 mmol, 3 μ L, 5.0 equiv). The solution was stirred in a sealed tube overnight at reflux in the dark. The reaction was quenched with water, and the organic layer was washed with brine and AcOEt and dried with Na₂SO₄. The residue was purified by flash column chromatography (5% MeOH in DCM) to provide erysotramidine **2** (3.4 mg, 95%). ¹H NMR (300 MHz, CDCl₃) δ 6.89 (dd, *J* = 10.1, 2.2 Hz, 1H), 6.80 (s, 1H), 6.72 (s, 1H), 6.33 (d, *J* = 10.1 Hz, 1H), 6.04, (s, 1H), 4.00 (dt, *J* = 13.2, 7.5 Hz, 1H), 3.86 (s + m, 4H), 3.76 (s, 3H), 3.62 (m, 1H), 3.34 (s, 3H), 3.04 (m, 2H), 2.81 (dd, *J* = 11.0, 5.0 Hz, 1H), 1.71 (dd, *J* = 11.0, 10.1 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 170.9, 157.0, 148.6, 147.0, 136.3, 128.7, 126.5, 124.1, 120.3, 112.2, 108.2, 74.9, 66.3, 56.4, 56.1, 55.9, 41.4, 37.3, 27.0; HRMS (ESI) *m*/*z*: calcd for C₁₉H₂₂NO₄ (M + H)⁺, 328.1541; found, 328.1543.

ASSOCIATED CONTENT

Supporting Information

General experimental procedures and ¹H and ¹³C NMR characterization data. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Tsuda, Y.; Sano, T. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: San Diego, 1996; Vol. 48, pp 249.

(2) (a) Boekelheide, V. In *The Alkaloids*; Manske, R. H. F., Ed.; Academic Press: New York, 1960; Vol. 7, pp 201. (b) Hill, R. K. In *The Alkaloids*; Manske, R. H. F., Ed.; Academic Press: New York, 1967; Vol. 9, pp 483. (c) Dyke, S. F.; Quessy, S. N. In *The Alkaloids*; Rodrigo, R. G. A., Ed.; Academic Press: New York, 1981; Vol. 18, pp 1. (d) Chawala, A. S.; Jackson, A. H. Nat. Prod. Rep. 1984, 1, 371.

(3) (a) Ito, K.; Furukawa, H.; Haruna, M. Yakugaku Zasshi 1973, 93, 1611. (b) Mantle, P. G.; Laws, I.; Widdowson, D. A. Phytochemistry 1984, 23, 1336.

(4) Aguilar, M. I.; Giral, F.; Espejo, O. Phytochemistry 1981, 20, 2061. (5) (a) Joo, J. M.; David, R. A.; Yuan, Y.; Lee, C. Org. Lett. 2010, 12, 5704. (b) Tuan, L. A.; Kim, G. Bull. Korean Chem. Soc. 2010, 31, 1800. (c) Liang, J.; Chen, J.; Liu, J.; Li, L.; Zhang, H. Chem. Commun. 2010, 46, 3666. (d) Onoda, T.; Takikawa, Y.; Fujimoto, T.; Yasui, Y.; Suzuki, K.; Matsumoto, T. Synlett 2009, 1041. (e) Yoshida, Y.; Mohri, K.; Isobe, K.; Itoh, T.; Yamamoto, K. J. Org. Chem. 2009, 74, 6010. (f) Stanislawski, P. C.; Willis, A. C.; Banwell, M. G. Chem.-Asian J. 2007, 2, 1127. (g) Shimizu, K.; Takimoto, M.; Sato, Y.; Mori, M. J. Organomet. Chem. 2006, 691, 5466. (h) Stanislawski, P. C.; Willis, A. C.; Banwell, M. G. Org. Lett. 2006, 8, 2143. (i) Kim, G.; Kim, J. H.; Lee, K. Y. J. Org. Chem. 2006, 71, 2185. (j) Yasui, Y.; Suzuki, K.; Matsumoto, T. Synlett 2004, 619. (k) Fukumoto, H.; Esumi, T.; Ishihara, J.; Hatakeyama, S. Tetrahedron Lett. 2003, 44, 8047. (1) Shimizu, K.; Takimoto, M.; Mori, M. Org. Lett. 2003, 5, 2323. (m) Kalaitzakis, D.; Montagnon, T.; Antonatou, E.; Vassilikogiannakis, G. Org. Lett. 2013, 15, 3714. (n) Isobe, K.; Mohri, K.; Takeda, N.; Suzuki, K.; Hosoi, S.; Tsuda, Y. Chem. Pharm. Bull. 1994, 42, 197. (o) Tsuda, Y.; Hosoi, S.; Katagiri, N.; Kaneko, C.; Sano, T. Chem. Pharm. Bull. 1993, 41, 2087. (p) Tsuda, Y.; Hosoi, S.; Sano, T.; Suzuki, H.; Toda, J. Heterocycles 1993, 36, 655. (q) Isobe, K.; Mohri, K.; Suzuki, K.; Haruna, M.; Ito, K.; Hosoi, S.; Tsuda, Y. Chem. Pharm. Bull. 1991, 32, 1195. (r) Tsuda, Y.; Hosoi, S.; Nakai, A.; Sakai, Y.; Abe, T.; Ishi, Y.; Kiuchi, F.; Sano, T. Chem. Pharm. Bull. 1991, 39, 1365. (s) Sano, T.; Toda, J.; Kashiwaba, N.; Ohshima, T.; Tsuda, Y. Chem. Pharm. Bull. 1987, 35, 479. (t) Ito, K.; Suzuki, F.; Haruna, M. J. Chem. Soc., Chem. Commun. 1978, 733.

(6) (a) Ogawa, S.; Iida, N.; Tokunaga, E.; Shiro, M.; Shibata, N. Chem.—Eur. J. 2010, 16, 7090. (b) Juma, B.; Adeel, M.; Villinger, A.; Reinke, H.; Spannenberg, A.; Fischer, C.; Langer, P. Adv. Synth. Catal. 2009, 351, 1073. (c) Tietze, L. F.; Tolle, N.; Kratzert, D.; Stalke, D. Org. Lett. 2009, 11, 5230. (d) Zhang, F.; Simpkins, N. S.; Blake, A. J. Org. Biomol. Chem. 2009, 7, 1963. (e) Padwa, A.; Wang, Q. J. Org. Chem. 2006, 71, 7391. (f) Wang, Q.; Padwa, A. Org. Lett. 2006, 8, 601. (g) Gao, S.; Tu, Y. Q.; Hu, X.; Wang, S.; Hua, R.; Jiang, Y.; Zhao, Y.; Fan, X.; Zhang, S. Org. Lett. 2006, 8, 2373. (h) El Bialy, S. A. A.; Braun, H.; Tietze, L. F. Angew. Chem., Int. Ed. 2004, 43, 5391. (i) Padwa, A.; Lee, H. I.; Rashatasakhon, P.; Rose, M. J. Org. Chem. 2004, 69, 8209. (j) Lee, H. I.; Cassidy, M. P.; Rashatasakhon, P.; Padwa, A. Org. Lett. 2003, 5, 5067. (k) Chikaoka, S.; Toyao, A.; Ogasawara, M.; Tamura, O.; Ishibashi, H. J. Org. Chem. 2003, 68, 312. (1) Padwa, A.; Hennig, R.; Kappe, C. O.; Reger, T. S. J. Org. Chem. 1998, 63, 1144. (m) Tsuda, Y.; Hosoi, S.; Katagiri, N.; Kaneko, C.; Sano, T. Heterocycles 1992, 33, 497. (n) Rigby, J. H.; Qabar, M. J. Am. Chem. Soc. 1991, 113, 8975. (o) Tsuda, Y.; Hosoi, S.; Nakai, A.; Ohshima, T.; Sakai, Y.; Kiuchi, F. J. Chem. Soc., Chem. Commun. 1984, 1216. (p) Sano, T.; Toda, J. Heterocycles 1982, 18, 229. (q) Mondon, A.; Nestler, H. J. Chem. Ber. 1979, 112, 1329. (r) Haruna, M.; Ito, K. J. Chem. Soc., Chem. Commun. 1976, 345. (s) Mondon, A.; Nestler, H. J. Angew. Chem. 1964, 76, 651.

(7) (a) Quideau, S.; Looney, M. A.; Pouységu, L. Org. Lett. 1999, 1, 1651. (b) Traoré, M.; Ahmed-Ali, S.; Peuchmaur, M.; Wong, Y. S. Tetrahedron 2010, 66, 5863. (c) Andrez, J. A.; Giroux, M. A.; Lucien, J.; Canesi, S. Org. Lett. 2010, 12, 4368. (d) Liang, H.; Ciufolini, M. A. Tetrahedron 2010, 66, 5884. (e) Pouységu, L.; Sylla, T.; Garnier, T.; Rojas, L. B.; Charris, J.; Deffieux, D.; Quideau, S. Tetrahedron 2010, 66, 5908. (f) Guérard, K. C.; Sabot, C.; Beaulieu, M. A.; Giroux, M. A.; Canesi, S. Tetrahedron 2010, 66, 5893. (g) Jen, T.; Mendelsohn, B. A.; Ciufolini, M. A. J. Org. Chem. 2011, 76, 728. (h) Beaulieu, M. A.; Sabot, C.; Achache, N.; Guérard, K. C.; Canesi, S. Chem.-Eur. J. 2010, 16, 11224. (i) Beaulieu, M. A.; Guérard, K. C.; Maertens, G.; Sabot, C.; Canesi, S. J. Org. Chem. 2011, 76, 9460. (j) Guérard, K. C.; Guerinot, A.; Bouchard-Aubin, C.; Ménard, M. A.; Lepage, M.; Beaulieu, M. A.; Canesi, S. J. Org. Chem. 2012, 77, 2121. (k) Pouységu, L.; Deffieux, D.; Quideau, S. Tetrahedron 2010, 66, 2235. (1) Traoré, M.; Ahmed-Ali, S.; Peuchmaur, M.; Wong, Y. S. J. Org. Chem. 2011, 76, 1409. (m) Pouységu,

The Journal of Organic Chemistry

L.; Marguerit, M.; Gagnepain, J.; Lyvinec, G.; Eatherton, A. J.; Quideau, S. Org. Lett. 2008, 10, 5211. (n) Jacquemot, G.; Menard, M. A.; L'Homme, C.; Canesi, S. Chem. Sci. 2013, 4, 1287. (o) Ozanne-Beaudenon, A.; Quideau, S. Angew. Chem., Int. Ed. 2005, 44, 7065. (p) Desjardins, S.; Andrez, J. A.; Canesi, S. Org. Lett. 2011, 13, 3406. (q) Jaquemot, G.; Canesi, S. J. Org. Chem. 2012, 77, 7588. (r) Yoshimura, A.; Middleton, K. R.; Luedtke, M. W.; Zhu, C.; Zhdankin, V. V. J. Org. Chem. 2012, 77, 11399. (s) Yoshimura, A.; Middleton, K. R.; Todora, A. D.; Kastern, B. J.; Koski, S. R.; Maskaev, A. V.; Zhdankin, V. V. Org. Lett. 2013, 15, 4010. (t) Beaulieu, M. A.; Ottenwaelder, X.; Canesi, S. Chem.—Eur, J. 2014, 20, 7581.

(8) (a) Tamura, Y.; Yakura, T.; Haruta, J.; Kita, Y. J. Org. Chem. **1987**, *52*, 3927. (b) Kita, Y.; Tohma, H.; Hatanaka, K.; Takada, T.; Fujita, S.; Mitoh, S.; Sakurai, H.; Oka, S. J. Am. Chem. Soc. **1994**, *116*, 3684. (c) Dohi, T.; Maruyama, A.; Takenaga, N.; Senami, K.; Minamitsuji, Y.; Fujioka, H.; Caemmerer, S. B.; Kita, Y. Angew. Chem., Int. Ed. **2008**, *47*, 3787. (d) Dohi, T.; Kita, Y. Chem. Commun. **2009**, 2073. (e) Dohi, T.; Itob, M.; Yamaokaa, N.; Morimotoa, K.; Fujiokab, H.; Kita, Y. Tetrahedron **2009**, *65*, 10797. (f) Dohi, T.; Yamaoka, N.; Kita, Y. Tetrahedron **2010**, *66*, 5775.

(9) Maryanoff, B. E.; Zhang, H.-C.; Cohen, J. H.; Turchi, I. J.; Maryanoff, C. A. Chem. Rev. 2004, 104, 1431.

(10) Huang, P.-Q.; Zheng, X.; Deng, X.-M. Tetrahedron Lett. 2001, 42, 9039.

(11) Giroux, M. A.; Guérard, K. C.; Beaulieu, M. A.; Sabot, C.; Canesi, S. *Eur. J. Org. Chem.* **2009**, 3871.

(12) (a) Wipf, P.; Kim, Y. Tetrahedron Lett. **1992**, 33, 5457. (b) Wipf, P.; Kim, Y.; Goldstein, D. J. Am. Chem. Soc. **1995**, 117, 11106. (c) Wipf, P.; Li, W. J. Org. Chem. **1999**, 64, 4576. (d) Wipf, P.; Mareska, D. A. Tetrahedron Lett. **2000**, 41, 4723. (e) Methot, J. L.; Wipf, P. Org. Lett. **2000**, 2, 4213.

(13) Maertens, G.; Menard, M. A.; Canesi, S. *Synthesis* **2014**, 1573 and articles cited therein.

(14) Structures of compounds 14 and 15 have been assigned by NMR NOE. In addition, these compounds have been treated in the presence of KHMDS and subsequently O-methylated (Ag₂O, CH₃I) to produce the known natural product erysotramidine 2, which was obtained from diastereoisomer 15.

(15) (a) Sano, T.; Toda, J.; Kashiwaba, N.; Ohshima, T.; Tsuda, Y. *Chem. Pharm. Bull.* **1987**, *35*, 479. (b) Sano, T.; Toda, J.; Maehara, N.; Tsuda, Y. *Can. J. Chem.* **1987**, *65*, 94.

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