A Synthesis of Chiral 1,1,3-Trisubstituted 1,2,3,4-Tetrahydro- β -carbolines by the Pictet-Spengler Reaction of Tryptophan and Ketones: Conversion of (1R,3S)-Diastereomers into Their (1S,3S)-Counterparts by Scission of the C(1)-N(2) Bond

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The Pictet–Spengler cyclization of the imines (3) prepared by the condensation of L-tryptophan methyl ester (1) and aryl methyl ketones (2), using titanium(IV) isopropoxide as an iminating reagent, quantitatively proceeded, when treated with trifluoroacetic acid (TFA) or formic acid, to provide two diastereomers, that is (1S,3S)-1-aryl-3-isopropoxycarbonyl-1-methyl-1,2,3,4-tetrahydro- β -carbolines (4) and their (1R,3S)-diastereomers (5), of which the diastereomer ratios varied from 1 to 5 depending on the reaction conditions. The (1R,3S)-diastereomers (5) are thermodynamically more stable than their (1S,3S)-congeners (4), as shown by equilibration experiments in TFA. The conversion of 4 to 5 (also 5 to 4) should occur under acidic conditions by cleavage of the C(1)-N(2) bond with complete retention of configuration at the C-3 chiral center. The low diastereo-selectivity observed in the Pictet–Spengler reaction of 1 and 2 is concluded to be a stereochemical outcome under conditions of kinetic control (lower temperature, shorter reaction time), while the high diastereo selectivity with preferential formation of the more stable isomer (5) is the result of thermodynamically controlled experiments (higher temperature, longer reaction time).

Key words tetrahydro- β -carboline; Pictet–Spengler reaction; stereoselectivity; tryptophan; aryl methyl ketone

The Pictet–Spengler reaction¹⁾ has long been an important reaction for the syntheses of biologically important isoquinolines and β -carbolines.^{2,3)} Recently, we have developed a highly efficient method of synthesizing 1,1-disubstituted tetrahydroisoguinolines⁴⁾ and tetrahydro- β -carbolines⁵⁾ by the Pictet-Spengler reaction of arylethylamines and ketones using titanium(IV) isopropoxide as an iminating reagent. This method has an advantage in that the reaction can be carried out without the isolation of the acid-labile imines under a one-pot procedure. Although the Pictet–Spengler reactions using tryptophan and aldehydes which produce chiral 1,3disubstitued 1,2,3,4-tetrahydro- β -carbolines (TH β Cs) have been extensively investigated, 6-12) the reaction with ketone has not been hitherto investigated. Here we describe the Pictet-Spengler reaction of tryptophan methyl ester with aryl methyl ketones, which may provide a convenient method of synthesizing chiral 1,1,3-trisubstituted TH β Cs.

Results and Discussions

Pictet-Spengler Reaction of Tryptophan Methyl Ester (1) with Aryl Methyl Ketones (2) The condensation reaction of L-tryptophan methyl ester (1) and aryl methyl ketones (2a: acetophenone, 2b: 1-(4-chlorophenyl)-2-ethanone, 2c: 1-(4-methoxyphenyl)-2-ethanone) to the imine intermediates (3a-c) was carried out by heating in titanium(IV) isopropoxide at 70 °C for 3 h without using any solvent. To the mixture of the in situ formed imines (3) was added a large excess amount of trifluoroacetic acid (TFA) or formic acid (HCOOH). The reaction mixture was then allowed to react at 70 °C or at room temperature for appropriate times to complete the Pictet-Spengler cyclization. During the imination, the ester exchange reaction from the methyl ester to the isopropyl ester occurred. The reactions, in all cases, produced two diastereometric 1-aryl-3-isopropoxycarbonyl-1-methyl-TH β Cs (4) and (5) in good total yields (Chart 1, Table 1).

The reaction of 1 and 2a with TFA at 70 °C for 1 h provided 4a and 5a in yields of 10% and 45%, respectively (Table 1, Run 1). The same treatment with TFA at room temperature for 1 h improved the reaction to give higher yields of **4a** (31%) and **5a** (54%) (Table 1, Run 2). When this reaction mixture reacted at room temperature for a longer period of time (18h), the yields of products decreased slightly (4a: 14%, 5a: 50%) (Table 1, Run 3). The facts indicated that TFA rapidly induced the cyclization and at the same time slowly decomposed the products 4a and 5a by long contact with the acid. The reaction with HCOOH also induced the expected cyclization, although it was slow, to give results similar to those of TFA-induced cyclization. Thus, the reaction at 70 °C for 18 h gave 4a (13%) and 5a (58%) (Table 1, Run 4), while at room temperature for 18 h the reaction produced **4a** (39%) and **5a** (41%) (Table 1, Run 5).

The reaction of **1** with 1-(4-chlorophenyl)-2-ethanone (**2b**) and 1-(4-methoxyphenyl)-2-ethanone (**2c**) with TFA or HCOOH gave results similar to those of **2a** (**2b**: Table 1, Runs 6—10 and **2c**: Table 1, Runs 11—15).

The structures of products **4** and **5** were readily determined by elementary and spectral analyses (Mass, IR, 1 H-and 13 C-NMR, $[\alpha]_{D}$). In the 1 H-NMR spectrum of *cis*-diastereomer (**4a**), the signal of 4-H appeared as a double doublet at δ 2.87 (Hax, J=11, 15 Hz) and δ 3.19 (Heq, J=4, 15 Hz) and the signal of 3-H appeared as a double doublet at δ 4.00 (J=4, 11 Hz). The coupling constants (J=4, 15 Hz) between 3-H and 4-H showed that the 3-isopropoxycarbonyl group is in equatorial orientation when the tetrahydro- β -carboline ring has a half chair form. In the 1 H-NMR spectrum of *trans*-diasteromer (**5a**), a similar signal pattern appeared. The similar coupling constant value (J=5, 11 Hz) between 3-H and 4-H indicated that the 3-isopropoxycarbonyl group of **5a** had the same orientation (equatorial) as **4a**. The stereochemistries of the substituents at the C-1 and C-3 chiral cen-

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Table 1. Pictet-Spengler Reaction of L-Tryptophan Methyl Ester (1) with Aryl Methyl Ketones (2) Using Titanium(IV) Isopropoxide

Run -	Keto	nes (2)	Acids	Cond	itions		Yields (%)	of β -carbolin	es (THβCs)		Ratio of	
Kuii -		R	Acids	Temp (°C)	Time (h)	4 (1)	S,3 <i>S</i>)	5 (1)	R,3S)	4+5	5/4 ^{a)}	
1	2a	Н	TFA	70	1	4a	10	5a	45	55	4.5	
2	2a	Н	TFA	rt	1	4a	31	5a	54	85	1.7	
3	2a	Н	TFA	rt	18	4a	14	5a	50	64	3.6	
4	2a	Н	HCOOH	70	18	4a	13	5a	55	71	4.2	
5	2a	Н	HCOOH	rt	18	4a	39	5a	41	80	1.2	
6	2 b	C1	TFA	70	1	4b	15	5b	57	72	3.8	
7	2 b	C1	TFA	rt	1	4b	34	5b	46	80	1.4	
8	2 b	C1	TFA	rt	18	4b	14	5b	68	82	4.9	
9	2 b	C1	HCOOH	70	18	4b	9	5b	44	53	4.9	
10	2 b	C1	HCOOH	rt	18	4b	27	5b	37	64	1.4	
11	2c	OCH ₃	TFA	70	1	4c	14	5c	59	73	4.9	
12	2c	OCH ₃	TFA	rt	1	4c	13	5c	62	75	4.8	
13	2c	OCH ₃	TFA	rt	18	4c	14	5c	59	73	4.2	
14	2c	OCH ₃	НСООН	70	18	4c	15	5c	47	62	3.1	
15	2c	OCH ₃	НСООН	rt	18	4c	6	5c	13	19	2.2	

a) The ratios were calculated by the isolated yields of 4 and 5.

To compound 2-5

a: R=H, b: R=CI, c: R=OCH₃

Ti(O-iPr)₂

$$R^1 = CH_3$$
, $R^2 = CH_3$
 $R^1 = CH_3$, $R^2 = CH_3$
 $R^2 = CH_$

Table 2. $^{1}\text{H-}$ and $^{13}\text{C-NMR}$ Signals (C_{1} and C_{3}) and [α]_D Values of 4 and 5

TH <i>β</i> Cs	R	¹ H-N	$MR(\delta)$	¹³ C-NMR (ppm)		f-1 (): M OII
		C ₁ -CH ₃	C ₃ -H (<i>J</i> , Hz)	C_1	C ₃	$- [\alpha]_D(c)$ in MeO
4a	Н	1.92	4.00 (4, 11)	57.0	53.3	-53.3° (1.0)
4b	C1	1.90	3.99 (4, 11)	56.7	53.3	$-45.4^{\circ}(1.0)$
4c	OCH ₂	1.89	3.99 (4, 11)	56.5	53.4	$-50.1^{\circ}(1.0)$
5a	Н	1.86	3.50 (5, 11)	57.0	52.2	$-10.0^{\circ}(1.0)$
5b	C1	1.82	3.45 (5, 11)	56.6	52.2	$-14.4^{\circ}(0.5)$
5c	OCH ₂	1.84	3.50 (5, 11)	56.5	52.1	$-14.6^{\circ}(1.0)$

ters were determined on the basis of 2D-nuclear Overhauser and exchange spectroscopy (NOESY) and difference in nuclear Overhauser effect (DIF-NOE). In the NOESY spectrum of 4a, the signal of the C-1 methyl proton at δ 1.92 showed a correlation of the C-3 proton at δ 4.00. Irradiation of the C-1 methyl signal caused enhancement of the C-3 proton signal (14%), indicating that the stereochemistry of the C-1 methyl group and the C-3 proton of 4a is a cis-diaxial orientation. In the NOESY spectrum of 5a, the signal of the C-3 proton at δ 3.50 showed a correlation of the C-1 phenyl proton at δ 7.1—7.9. Irradiation of the C-3 proton signal caused en-

hancement of the C-1 phenyl signal (17%), indicating that the relative stereochemistry of the C-1 phenyl group and the C-3 proton of $\bf 5a$ is a *cis*-diaxial orientation. Thus, it was determined that the stereochemistry of the C-1 phenyl group and the C-3 isopropyl ester is *cis* for $\bf 4a$ (*cis*-diastereomer) and *trans* for $\bf 5a$ (*trans*-diastereomer). The $[\alpha]_D$ values of $\bf 4a$ and $\bf 5a$ obtained by the reaction of runs 1—5 were always consistent (Table 2). This fact strongly suggested that the Pictet–Spengler cyclization did not involve any racemization process. The structures of products $\bf 4b$, $\bf 4c$ and $\bf 5b$, $\bf 5c$ from 1-(4-chlorophenyl)- ($\bf 2b$) and 1-(4-methoxyphenyl)-2-

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ethanone (**2c**) were readily determined by their analogy in the ¹H- and ¹³C-NMR signals, and optical rotations with those for **4a** and **5a**, respectively (Table 2).

In order to determine whether the L-tryptophan chiral center had been partially racemized before inducing the Pictet–Spengler cyclization, we carried out following experiment. D-Tryptophan methyl ester (ent-1) on the Pictet–Spengler reaction with 2a using TFA at room temperature for 1 h yielded two products, ent-4a (24%) ($[\alpha]_D$ +52.7°) and ent-5a (38%) ($[\alpha]_D$ +9.9°). Their ¹H- and ¹³C-NMR signals were identical with those of 4a and 5a, but the specific optical rotations were opposite in sign (4a: $[\alpha]_D$ –53.3°) and (5a: $[\alpha]_D$ –10.0°).

The optical purities of 5a and ent-5a were shown to be 100% by the ¹H-NMR spectral inspection of the Mosher esters 7 and 8, which did not show any contamination of their corresponding diastereomers. This fact also supported that the epimerization at the C-3 chiral had not occurred before the Pictet–Spengler cyclization. Thus, the structures of 4a and 5a, including the stererocheimstry C-1 and C-3 chiral centers, were assigned to be (1S,3S)- and (1R,3S)-3-isopropoxycarbonyl-3-methyl-1-phenyl-TH β C, respectively (Chart 3).

Epimerization at C-1 Chiral Center The diastereomer ratio of 5/4 in the Pictet–Spengler reactions varied from 1 to 5 depending on such reaction conditions as the nature of acid, temperature, and time, as shown in Table 1. The reaction under more forced conditions showed high diastereo-selectivity, while the reaction under milder conditions gave low diastereo-selectivity. For example, the ratio of 5a/4a in the reaction of 2a with HCOOH at 70 °C for 18 h is 4.2 (Table 1, Run 4), while the ratio of 5a/4a in the reaction at room temperature for 18 h is 1.2 (Table 1, Run 5). The Pictet–Spengler reactions of 2b and 2c gave stereochemical outcomes similar to those of 2a. The facts clearly demonstrated that the acidinduced cyclization of the imine intermediate (3) occurred in a non-diastereoselective manner to yield the cis-diastereomer (4) and trans-diastereomer (5) in about a 1:1 ratio, then the cis-diastereomer (4) was converted into the trans isomer (5) by acid-induced isomerization.

Table 3. TFA-Induced Isomerization of TH β Cs (4) and (5)^{a)}

D	TH 00	D	Condi	Ratio of		
Run	TH β Cs	R	Temp (°C)	Time	5/4 ^{b)}	
1	4a	Н	rt	1 h	1.4	
2	4a	H	rt	3 h	3.0	
3	4b	C1	rt	1 h	0.8	
4	4b	C1	rt	3 h	3.0	
5	4c	OCH ₃	rt	20 min	3.0	
6	4c	OCH ₃	rt	1 h	3.0	
7	5a	Н	rt	3 h	4.0	
8	5b	C1	rt	3 h	10.0	
9	5b	C1	rt	18 h	4.5	
10	5c	OCH ₃	rt	20 min	4.0	

a) The products 4 and 5 were quantitatively recovered and no other products were detected by the TLC inspections. b) The ratios were measured by the intensity of the C_3 -H signal.

Fig. 1. Most Stable Comformer of 4a and 5a Optimized by AM1

The pure *cis*-diastereomer (4), when treated with TFA at room temperature for the appropriate times, caused the isomerization to form a 1:3 mixture of 4 and 5 in a quantitative yield. The *trans*-diastereomer (5) on similar reactions with TFA also quantitatively produced the mixture of 4 and 5 in about a 1:4 ratio, as shown in Table 3. These experiments revealed that the *trans*-diastereomer (5) is thermodynamically more stable than the *cis*-diastereomer (4). The 1 H-NMR spectra as described above showed that the *cis*- (4a) and *trans*-diastereomer (5a) adopts half-chair conformations with 3-equatorial isopropoxycarbonyl groups, as shown in Fig. 1. Using their conformation, we calculated the heat of formation of 4a and 5a by AM1 theory, which showed that 5a is more stable ($\Delta H_f = -9.71 \text{ kcal/mol}$) than 4a ($\Delta H_f = -9.71 \text{ kcal/mol}$) than 4a ($\Delta H_f = -9.71 \text{ kcal/mol}$)

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Chart 4

-6.17 kcal/mol), supporting the conclusion obtained from the isomerization experiments.

Furthermore, either the *cis*-isomer (4) or the *trans*-isomer (5) recovered after this isomerization reaction showed $[\alpha]_D$ values and signs identical with those of the starting materials, indicating that the racemization of either 4 or 5 did not occur.

The TFA-induced epimerization at the C-1 chiral center is able to occur *via* two processes; a) the recombination of the imines (3) generated by a retro-Pictet–Spengler reaction, and b) the C(1)–N(2) bond fission-recombination. The mechanism of the retro-Pictet–Spengler reaction can be discarded since in the product of the isomerization experiments described above, there is no indication of the formation of the acid-labile imines (3) that readily produce tryptophan isopropyl ester and aryl methyl ketone.

The facts described above, in turn, supported the C(1)–N(2) bond fission-recombination mechanism proposed by Zhang^{13,14)} and Cook¹²⁾ for the epimerization at the C-1 chiral center of 1,3-disubstituted TH β C. That is, the epimerization is able to occur *via* the carbocation 9 and its rotatory isomer 10 formed by the acid-induced cleavage of the C(1)–N(2) bond (Chart 4). This epimerization mechanism is consistent with the fact that the 4-methoxyphenyl derivatives (4c, 5c) isomerized more rapidly than phenyl (4a, 5a) or 4-chlorophenyl derivatives (4b, 5b). The methoxy moiety would help to stabilize the carbocation 9cA *via* the resonance form 9cB, and therefore to facilitate not only the cleavage of the C(1)–N(2) bond but also the epimerization of the C-1 chiral center.

It is concluded that the low diastereo-selectivity observed in the Pictet–Spengler reaction of 1 and 2 is the stereochemical outcome based on conditions of kinetic control, and that the higher stereoselectivity with the preferential formation of the *trans*-diastereomer (5) is the result of the thermodynamic control experiment.

Experimental

Unless otherwise noted, the following procedures were adopted. Melting points were taken on a Yanagimoto SP-M1 hot-stage melting point apparatus and are uncorrected. IR spectra were measured as films for oils and gums, and on KBr disks for solids with a HORIBA FT-710 spectrophotometer, and the values are given in cm $^{-1}$. NMR spectra were measured on a JEOL JNM-AL300 ($^{1}\text{H-NMR}$: 300 MHz, $^{13}\text{C-NMR}$: 75 MHz) or a JEOL JNM α 500 ($^{1}\text{H-NMR}$: 500 MHz, $^{13}\text{C-NMR}$: 125 MHz) NMR spectrometer in CDCl $_{3}$ with tetramethylsilane as an internal standard, and the chemical shifts are given in δ values. Low-resolution electron impact ionization mass spectra (LR-EI-MS) were taken on a JEOL JMS-AM20 mass spectrometer at 70 eV using a direct inlet probe. High-resolution electron impact ionization mass spectra (HR-EI-MS) were taken on a JEOL JMS-D300 mass spectrometer at 70 eV using a direct inlet probe. Elemental analyses were recorded on a

Yanaco CHN-corder MT-3. Optical rotations were determined using a JASCO DIP-1000 digital polarimeter in MeOH. TLC was performed on Merck precoated Silica gel 60 F₂₅₄ plates. Column chromatography was carried out with silica gel (Wakogel C-200). The organic extract from each reaction mixture was washed with brine, dried over anhydrous Na₂SO₄, and concentrated *in vacuo* to dryness.

The Pictet–Spengler Reaction of L-Tryptophan methyl ester with Acetophenone (2a): Typical procedure L-Tryptophan methyl ester hydrochloride (1) (1.2 g, 4.59 mmol) in H₂O (50 ml) was basified with 10% K₂CO₃ solution and extracted with AcOEt. After removal of the solvent *in vacuo*, the residue was mixed with 2a (0.46 g, 3.83 mmol) and Ti(O-*i*Pr)₄ (1.63 g, 5.75 mmol), and the mixture was heated at 70 °C for 3 h under an argon atmosphere. To the reaction mixture was added the mixture of TFA (43.6 g, 0.383 mol) and trifluoroacetic anhydride (0.8 g, 3.83 mmol) at 0 °C, then the mixture was heated at 70 °C for 1 h. The reaction mixture was diluted with MeOH (100 ml) and passed through a short SiO₂ column (MeOH) to remove TiO₂. The eluent was concentrated *in vacuo* (*ca.* 30 ml) and the residue was neutralized with 10% NaOH solution and extracted with CHCl₃. After removal of the solvent of the extract *in vacuo*, the residue was purified by column chromatography over SiO₂ (benzene–acetone=30:1) to give 4a (111 mg, 10%) and 5a (600 mg, 45%).

(1*S*,3*S*)-3-Isopropoxycarbonyl-1-methyl-1-phenyl-1,2,3,4-tetrahydro- β -carboline (**4a**): Colorless prisms recrystallized from Et₂O–hexane, mp 184—185 °C. IR: 3374, 3343, 2977, 1735, 1710. UV: 279.0 (8800). ¹H-NMR: 1.30 (3H, d, J=7 Hz, CH(C \underline{H}_3)₂), 1.32 (3H, d, J=7 Hz, CH(C \underline{H}_3)₂), 1.92 (3H, s, C \underline{H}_3), 2.87 (1H, dd, J=11, 15 Hz, 4-H), 3.19 (1H, dd, J=4, 15 Hz, 4-H), 4.00 (1H, dd, J=4, 11 Hz, 3-H), 5.11 (1H, m, C \underline{H} (CH₃)₂), 7.1—7.6 (10H, m, Ph-H, Ar-H). ¹³C-NMR: 21.8 (2×CH(C \underline{H}_3)₂), 25.7 (C \underline{H}_3), 26.4 (C4), 53.3 (C3), 57.0 (C1), 68.6 (C \underline{H} (CH(\underline{H}_3)₂), 108.1 (C4a), 110.9 (C8), 118.3 (C6), 119.5 (C5), 121.9 (C7), 126.9 (C4b), 127.0 (2×PhCH), 127.8 (PhCH), 128.6 (2×PhCH), 136.0 (PhC), 139.1 (C9a), 145.3 (C8a), 172.9 (COO). LR-EI-MS: m/z 348 (M $^+$), 245 (base peak). HR-EI-MS: Calcd for C₂₂H₂₄N₂O₂: 348.1835. Found: 348.1817. *Anal*. Calcd for C₂₂H₂₄N₂O₂: C, 75.83; H, 6.94; N, 8.04. Found: C, 75.94; H, 7.07; N, 8.08. [α]_D -53.3° (c=1.0 in MeOH).

(1*R*,3*S*)-3-Isopropoxycarbonyl-1-methyl-1-phenyl-1,2,3,4-tetrahydro- β -carboline (**5a**): Colorless needles recrystallized from Et₂O-hexane, mp 175—176 °C. IR: 3369, 2979, 1724, 1654. UV: 278.5 (8500). ¹H-NMR: 1.24 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.27 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.86 (3H, s, C $\underline{\text{H}}_3$), 2.82 (1H, dd, J=11, 15 Hz, 4-H), 3.01 (1H, dd, J=5, 15 Hz, 4-H), 3.50 (1H, dd, J=5, 11 Hz, 3-H), 5.06 (1H, sep, J=6 Hz, C $\underline{\text{H}}$ (CH₃)₂), 7.1—7.9 (10H, m, Ph-H, Ar-H). ¹³C-NMR: 21.7 (CH($\underline{\text{CH}}_3$)₂), 21.8 (CH($\underline{\text{CH}}_3$)₂), 25.9 (C4), 29.3 ($\underline{\text{CH}}_3$), 52.2 (C3), 57.0 (C1), 68.5 ($\underline{\text{CH}}$ (CH(CH₃)₂), 108.8 (C4a), 110.9 (C8), 118.4 (C6), 119.7 (C5), 122.0 (C7), 126.5 (2×PhCH), 127.1 (C4b), 127.2 (PhCH), 128.2 (2×PhCH), 135.9 (PhC), 137.4 (C9a), 145.9 (C8a), 173.1 ($\underline{\text{COO}}$). LR-EI-MS: m/z 348 (M⁺), 218 (base peak). HR-EI-MS: Calcd for C₂₂H₂₄N₂O₂: 348.1838. Found: 348.1863. *Anal.* Calcd for C₂₂H₂₄N₂O₂: C, 75.83; H, 6.94; N, 8.04. Found: C, 75.74; H, 6.98; N, 8.05. [α]_D -10.0° (c=1.0 in MeOH).

The Pictet–Spengler Reaction of 1 with 2b The reaction of **1** (4.59 mmol) and **2b** (590 mg, 3.83 mmol) under the condition described in Table 1 (Run 6) gave **4b** (490 mg, 34%) and **5b** (670 mg, 46%).

(1S,3S)-1-(4-Chlorophenyl)-3-isopropoxycarbonyl-1-methyl-1,2,3,4-tetrahydro-β-carboline (**4b**): Colorless needles recrystallized from Et₂O-hexane, mp 195—197 °C. IR: 3384, 3342, 2979, 2935, 1718. ¹H-NMR: 1.30 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.32 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.90 (3H, s, C $\underline{\text{H}}_3$), 2.86 (1H, dd, J=11, 15 Hz, 4-H), 3.19 (1H, dd, J=4, 15 Hz, 4-H), 3.99 (1H, dd, J=4, 11 Hz, 3-H), 5.12 (1H, sep, J=6 Hz, C $\underline{\text{H}}$ (CH₃)₂), 7.1—

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7.6 (9H, m, Ar-H, Ph-H). 13 C-NMR: 21.8 (2×CH($\underline{CH_3}$)₂), 25.6 ($\underline{CH_3}$), 26.2 (C4), 53.3 ($\underline{CH(CH_3)_2}$), 56.7 (C1), 68.7 ($\underline{CH(CH_3)_2}$), 108.3 (C4a), 110.9 (C5), 118.4 (C8), 119.7 (C6), 122.1 (C7), 126.8 (C4b), 128.6 (2×Ph \underline{CH}), 128.7 (2×Ph \underline{CH}), 133.7 (Ph \underline{C}), 136.1 (Ph \underline{C}), 138.4 (C9a), 144.0 (C8a), 172.8 (\underline{COO}). LR-EI-MS: m/z 382, 384 (M⁺), 367 (base peak). HR-EI-MS: Calcd for C₂₂H₂₃N₂O₂Cl: 382.1448, 384.1416. Found: 382.1469, 384.1411. Anal. Calcd for C₂₂H₂₃N₂O₂Cl: C, 69.01; H, 6.05; N, 7.32. Found: C, 68.84; H, 6.19; N, 7.30. [α]_D -45.4° (c=1.0 in MeOH).

(1*R*,3*S*)-1-(4-Chlorophenyl)-3-isopropoxycarbonyl-1-methyl-1,2,3,4-tetrahydro-β-carboline (**5b**): Colorless needles recrystallized from Et₂O–AcOEt, mp 245—247 °C. IR: 3417, 3349, 2975, 1720. ¹H-NMR: 1.25 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.28 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.28 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.82 (3H, s, C $\underline{\text{H}}_3$), 2.80 (1H, dd, J=11, 15 Hz, 4-H), 3.08 (1H, dd, J=5, 15 Hz, 4-H), 3.45 (1H, dd, J=5, 11 Hz, 3-H), 5.06 (1H, sep, J=6 Hz, C $\underline{\text{H}}$ (CH₃)₂), 7.1—7.6 (8H, m, Ph-H, Ar-H), 7.94 (1H, br s, Ar-N $\underline{\text{H}}$). ¹³C-NMR: 21.7 (CH(C $\underline{\text{H}}_3$)₂), 21.8 (CH(C $\underline{\text{H}}_3$)₂), 25.9 (C4), 29.2 (C $\underline{\text{H}}_3$), 52.2 (C3), 56.6 (C1), 68.6 (CH(CH₃)₂), 108.9 (C4a), 111.0 (C8), 118.5 (C6), 119.8 (C5), 122.3 (C7), 127.0 (C4b), 128.0 (2×PhC $\underline{\text{H}}$), 128.3 (2×PhC $\underline{\text{H}}$), 133.1 (PhC), 135.9 (PhC), 136.9 (C9a), 144.5 (C8a), 173.1 (COO). LR-EI-MS: m/z 382, 384 (M $^+$), 367 (base peak). HR-EI-MS: Calcd for C₂₂H₂₃N₂O₂Cl: 382.1446, 384.1418. Found: 382.1430, 384.1420. *Anal.* Calcd for C₂₂H₂₃N₂O₂Cl: C, 69.01; H, 6.05; N, 7.32. Found: C, 68.75; H, 6.10; N, 7.35. [α]_D - 14.4° (c=0.5 in MeOH).

The Pictet–Spengler Reaction of 1 with 2c The reaction of **1** (4.59 mmol) and **2c** (570 mg, 3.83 mmol) under the condition described in Table 1 (Run 11) gave **4c** (183 mg, 13%) and **5c** (892 mg, 62%).

(1*S*,3*S*)-3-Isopropoxycarbonyl-1-(4-methoxyphenyl)-1-methyl-1,2,3,4-tetrahydro-β-carboline (**4c**): Colorless needles recrystallized from Et₂O-hexane, mp 125—127 °C. IR: 3378, 2977, 1724, 1608. ¹H-NMR: 1.30 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.31 (3H, d, J=6 Hz, CH(C $\underline{\text{H}}_3$)₂), 1.89 (3H, s, C $\underline{\text{H}}_3$), 2.85 (1H, dd, J=11, 15 Hz, 4-H), 3.18 (1H, dd, J=4, 15 Hz, 4-H), 3.79 (3H, s, OC $\underline{\text{H}}_3$), 3.99 (1H, dd, J=4, 11 Hz, 3-H), 5.11 (1H, sep, J=6 Hz, C $\underline{\text{H}}$ (CH $_3$)₂), 6.8—6.9 (2H, m, Ph-H), 7.1—7.2, 7.5—7.6 (total 4H, m, Ar-H), 7.3—7.4 (2H, m, Ph), 7.44 (1H, br s, Ar-N $\underline{\text{H}}$)). ¹³C-NMR: 21.8 (2×CH(C $\underline{\text{H}}_3$)₂), 25.8 ($\underline{\text{CH}}_3$), 26.4 (C4), 53.4 (C3), 55.3 (OC $\underline{\text{H}}_3$), 56.5 (C1), 68.5 (C $\underline{\text{H}}$ (CH $_3$)₂), 108.0 (C4a), 110.9 (C8), 113.8 (2×PhCH), 118.2 (C6), 119.5 (C5), 121.9 (C7), 127.0 (C4b), 128.3 (2×PhCH), 136.0 (PhC), 137.5 (C9a), 139.3 (C8a), 159.0 (PhC), 172.9 (COO). LR-EI-MS: m/z 378 (M $^+$), 363 (base peak). HR-EI-MS: Calcd for C₂₃H₂₆N₂O₃: 378.1944. Found: 378.1979. *Anal.* Calcd for C₂₃H₂₆N₂O₃: C, 72.99; H, 6.92; N, 7.40. Found: C, 72.74; H, 7.02; N, 7.31. [α]_D –50.1° (c=1.0 in MeOH).

(1*R*,3*S*)-3-Isopropoxycarbonyl-1-(4-methoxyphenyl)-1-methyl-1,2,3,4-tetrahydro-β-carboline (**5c**): Colorless needles recrystallized from Et₂O-hexane, mp 178—180 °C. IR: 3369, 2979, 1724, 1608. ¹H-NMR: 1.24 (3H, d, *J*=6 Hz, CH(CH₃)₂), 1.28 (3H, d, *J*=6 Hz, CH(CH₃)₂), 1.84 (3H, s, CH₃), 2.81 (1H, dd, *J*=11, 15 Hz, 4-H), 3.09 (1H, dd, *J*=5, 15 Hz, 4-H), 3.50 (1H, dd, *J*=5, 11 Hz, 3-H), 3.76 (3H, s, OCH₃), 5.06 (1H, sep, *J*=6 Hz, CH(CH₃)₂), 6.8—6.9 (2H, m, Ph-H), 7.1—7.2 (4H, m, Ar-H), 7.3—7.6 (2H, m, Ph-H), 7.88 (1H, br s, Ar-NH). ¹³C-NMR: 21.7 (CH(CH₃)₂), 21.8 (CH(CH₃)₂), 25.9 (C4), 29.3 (CH₃), 52.1 (C3), 55.2 (OCH₃), 56.5 (C1), 68.4 (CH(CH₃)₂), 108.6 (C4a), 110.9 (C8), 113.4 (2×PhCH), 118.4 (C6), 119.6 (C5), 122.0 (C7), 127.1 (C4b), 127.7 (2×PhCH), 135.9 (PhC), 137.6 (C9a), 138.0 (C8a), 158.6 (PhC), 173.2 (COO). LR-EI-MS: *m/z* 378 (M⁺), 149 (base peak). HR-EI-MS: Calcd for C₂₃H₂₆N₂O₃: 378.1944. Found: 378.1964. *Anal.* Calcd for C₂₃H₂₆N₂O₃: C, 72.99; H, 6.92; N, 7.40. Found: C, 72.94; H, 7.00; N, 7.36. [α]_D –14.6° (*c*=1.0 in MeOH).

The Pictet–Spengler Reaction of p-Tryptophan Methyl Ester with 2a D-Tryptophan methyl ester hydrochloride (ent-1) (1.2 g, 4.59 mmol) in H₂O (50 ml) was basified with 10% K₂CO₃ solution and extracted with AcOEt. After removal of the solvent *in vacuo*, the residue was mixed with 2a (0.46 g, 3.83 mmol) and Ti(O-iPr)₄ (1.63 g, 5.75 mmol), then the mixture was heated at 70 °C for 3 h under an argon atmosphere. To the reaction mixture was added a mixture of TFA (43.6 g, 0.383 mol) and trifluoroacetic anhydride (0.8 g, 3.83 mmol) at 0 °C, then the mixture was stirred at room temperature for 1 h. The reaction mixture was diluted with MeOH (100 ml) and passed through a short SiO₂ column (MeOH) to remove TiO₂. The eluent was concentrated *in vacuo* (*ca.* 30 ml) and the residue was neutralized with 10% NaOH solution and extracted with CHCl₃. After removal of the solvent *in vacuo*, the residue was purified by column chromatography over SiO₂ (benzene–acetone=30:1) to give ent-4a (315 mg, 24%) and ent-5a (505 mg, 38%), respectively.

(1R,3R)-3-Isopropoxycarbonyl-1-methyl-1-phenyl-1,2,3,4-tetrahydro- β -carboline (**ent-4a**): Colorless prisms recrystallized from Et₂O–hexane, mp 186—187 °C. IR, ¹H- and ¹³C-NMR were identical with those of **4a**. LR-EI-MS: m/z 348 (M⁺), 333 (base peak). HR-EI-MS: Calcd for C₂₂H₂₄N₂O₂:

348.1838. Found: 348.1859. *Anal.* Calcd for $C_{22}H_{24}N_2O_2$: C, 75.83; H, 6.94; N, 8.04. Found: C, 75.77; H, 7.06; N, 8.06. $[\alpha]_D$ +52.7° (c=1.0 in MeOH).

(1*S*,3*R*)-3-Isopropoxycarbonyl-1-methyl-1-phenyl-1,2,3,4-tetrahydro-β-carboline (**ent-5a**): Colorless needles recrystallized from Et₂O–hexane, mp 170—172 °C. IR, ¹H- and ¹³C-NMR were identical with those of **5a**. LR-EI-MS: m/z 348 (M⁺), 333 (base peak). HR-EI-MS: Calcd for C₂₂H₂₄N₂O₂: 348.1835. Found: 348.1829. *Anal*. Calcd for C₂₂H₂₄N₂O₂: C, 75.83; H, 6.94; N, 8.04. Found: C, 75.75; H, 7.13; N, 7.90. [α]_D +9.9° (c=1.0 in MeOH).

Reduction of 5a with LiAlH₄: Typical Procedure LiAlH₄ (22 mg, 0.575 mmol) was added to a solution of 5a (200 mg, 0.575 mmol) in THF (10 ml) at 0 °C, then the mixture was stirred for 2.5 h at the same temperature. Water was added to the reaction mixture and the mixture was extracted with CHCl₃. After removal of the solvent, the residue was purified by column chromatography over SiO₂ (AcOEt-hexane=4:1) to give (1R,3S)-3hydoxymethyl-1-methyl-1-phenyl-1,2,3,4-tetrahydro- β -carboline (6) (154) mg, 92%) as a colorless gum. IR: 3397, 3291, 2969, 2925. ¹H-NMR: 1.80 $(3H, s, CH_3)$, 2.47 (1H, dd, J=11, 15 Hz, 4-H), 2.70 (1H, dd, J=4, 15 Hz, 4-H)H), 2.9—3.0 (1H, m, 3-H), 3.48 (1H, dd, J=9, 11 Hz, $C\underline{H}_2OH$), 3.69 (1H, dd, J=4, 11 Hz, C \underline{H}_2 OH), 7.1—7.5 (9H, m, Ar-H, Ph-H), 7.99 (1H, br s, Ar-NH). ¹³C-NMR: 24.7 (C4), 29.8 (CH₃), 51.2 (C3), 57.3 (C1), 66.0 (CH₂OH), 109.3 (C4a), 110.9 (C8), 118.4 (C6), 119.6 (C5), 122.0 (C7), 126.6 (2×PhCH), 127.1 (C4a), 128.2 (2×PhCH), 135.9 (PhC), 138.1 (C9a), 146.1 (C8a). LR-EI-MS: m/z 292 (M⁺), 277 (base peak). HR-EI-MS: Calcd for $C_{19}H_{20}N_2O$: 292.1573. Found: 292.1572. [α]_D +38.0° (c=0.5 in MeOH).

Reduction of ent-5a with LiAlH₄ Reduction of ent-5a (200 mg, 0.575 mmol) with LiAlH₄ (22 mg, 0.575 mmol) and purification by column chromatography over SiO₂ (benzene–acetone=3:1) gave (1R,3R)-3-hydoxymethyl-1-methyl-1-phenyl-1,2,3,4-tetrahydro- β -carboline (ent-6) (140 mg, 83%) as a colorless gum. IR, 1 H- and 13 C-NMR were identical with those of **6**. LR-EI-MS: m/z 292 (M⁺), 277 (base peak). HR-EI-MS: Calcd for C₁₉H₂₀N₂O: 292.1573. Found: 292.1558. [α]_D -39.7° (c=0.5 in MeOH).

Synthesis of (1R,3S)-(1-Methyl-1-phenyl-1,2,3,4-tetrahydro-\beta-carboline-3-yl)methyl (R)- α -Methoxy- α -trifluoromethylphenyl Acetate (7): Typical Procedure A solution of 6 (42 mg, 0.144 mmol), pyridine (1.7 ml), DMAP (7 mg) and (-)-MTPA-Cl (90 mg, 0.360 mmol) in CCl₄ (3.5 ml) was stirred at room temperature for 24 h under an argon atmosphere. The reaction mixture was diluted with water and the mixture was extracted with CHCl₃. After removal of the solvent in vacuo, the residue was purified by column chromatography over SiO₂ (AcOEt-hexane=4:1) to give 7 (48 mg, 66%) as a pale yellow gum. IR: 3058, 2965, 2921, 1745. ¹H-NMR: 1.69 (3H, s, CH_3), 2.57 (1H, dd, J=11, 15 Hz, 4-H), 2.73 (1H, dd, J=5, 15 Hz, 4-H), 3.1—3.2 (1H, m, 3-H), 3.46 (3H, d, J=1 Hz, OCH₃), 4.32 (1H, dd, J=7, 11 Hz, $C\underline{H}_2O$), 4.45 (1H, dd, J=4, 11 Hz, $C\underline{H}_2O$), 7.1—7.6 (14H, m, 2×Ph-H, Ar-H), 7.91 (1H, br s, Ar-NH). ¹³C-NMR: 24.8 (C4), 29.5 (CH₃), 48.2 (C3), 55.4 (C1), 57.2 (OCH₃), 69.5 (CH₂O), 84.7 (C-CF₃), 108.7 (C4b), 110.9 (C8), 118.3 (C6), 119.6 (C5), 122.0 (C7), 126.4 (2×PhCH), 127.0 (C4a), 127.1 (PhCH), 127.4 (2×PhCH), 128.1 (2×PhCH), 128.4 (2×PhCH), 129.7 (PhCH), 132.1 (PhC), 135.9 (C9a), 137.6 (C8a), 146.0 (PhC), 166.5 (s, O-CO). LR-EI-MS: m/z 508 (M⁺), 58 (base peak). HR-EI-MS: Calcd for $C_{29}H_{27}N_2O_3F_3$: 508.1972. Found: 508 1952

Synthesis of (1S,3R)- $(1-Methyl-1-phenyl-1,2,3,4-tetrahydro-<math>\beta$ -carboline-3-yl)methyl (R)- α -Methoxy- α -trifluoromethylphenyl Acetate (8) 8 (54 mg, 89%) was obtained from the reaction of ent-6 (35 mg, 0.120 mmol) and (-)-MTPA-Cl (90 mg, 0.360 mmol) after purification by column chromatography over SiO₂ (AcOEt-hexane=1:1) as a pale yellow gum. IR: 3407, 2962, 2919, 1751. ¹H-NMR: 1.73 (3H, s, $C\underline{H}_3$), 2.56 (1H, dd, J=11, 15 Hz, 4-H), 2.71 (1H, dd, J=4, 15 Hz, 4-H), 3.08 (1H, m, 3-H), 3.46 (3H, d, $J=1 \text{ Hz}, \text{ OC}_{\underline{H}_3}$), 4.33 (1H, dd, J=7, 11 Hz, C \underline{H}_2 O), 4.39 (1H, dd, J=4, 11 Hz, CH_2O), 7.1—7.5 (14H, m, 2×Ph-H, Ar-H), 7.92 (1H, br s, Ar-NH). ¹³C-NMR: 24.6 (C4), 29.4 (–<u>C</u>H₃), 48.3 (C3), 55.4 (C1), 57.2 (O<u>C</u>H₃), 69.4 (CH₂O), 84.6 (C-CF₃), 108.7 (C4b), 110.9 (C8), 118.3 (C6), 119.6 (C5), 122.0 (C7), 126.3 (2×PhCH), 127.0 (C4a), 127.1 (PhCH), 127.4 (2×PhCH), 128.1 (2×PhCH), 128.4 (2×PhCH), 129.7 (PhCH), 132.2 (PhC), 135.9 (PhC), 137.4 (C9a), 146.1 (C8a), 166.4 (O-CO). LR-EI-MS: m/z 508 (M⁺), 58 (base peak). HR-EI-MS: Calcd for $C_{29}H_{27}N_2O_3F_3$: 508.1974. Found: 508.2011.

Epimerization Reaction of 4 and 5 under Acidic Condition: General Procedure 4 or **5** (50 mg, 0.14 mmol) in TFA (5 ml) was stirred at room temperature at the appropriate time (see Table 3). The 1 H-NMR spectrum of the mixture was measured. The ratios of **5/4** were calculated from the intensities of the C-3 H signals of the products. Optical rotations of **4** and **5** were measured after column chromatographic purification. **4a**: $[\alpha]_D - 55.8^\circ$ (c=1.0 in MeOH). **4b**: $[\alpha]_D - 44.5^\circ$ (c=0.5 in MeOH). **4c**: $[\alpha]_D - 51.0^\circ$

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(c=0.5 in MeOH). **5a**: $[\alpha]_{\rm D}$ -12.5° (c=0.9 in MeOH). **5b**: $[\alpha]_{\rm D}$ -14.0° (c=1.0 in MeOH). **5c**: $[\alpha]_{\rm D}$ -14.9° (c=1.0 in MeOH).

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