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Iodine-catalyzed chemoselective dehydrogenation and aromatization of Tetrahydro- β -carbolines: a short synthesis of Kumujian-C, Eudistomin-U, Norharmane, Harmane, Harmalan and Isoeudistomine-M

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ABSTRACT

Temperature controlled chemoselective dehydrogenation and aromatization of tetrahydro- β -carbolines, using molecular I_2 and H_2O_2 , in DMSO solvent affords a practical access to a series of corresponding 3, 4-dihydro- β -carbolines and β -carbolines respectively. This method has been successfully employed in the short synthesis of Kumujian-C, Eudistomin-U, Norharmane, Harmane, Harmalan and Isoeudistomine-M.

Keywords:

Dehydrogenation

Aromatization

Iodine

Carbolines

Kumujian-C

Eudistomin-U

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1. Introduction.

Aromatic β -carboline alkaloid scaffolds are observed in diverse natural sources¹ and show a broad range of biological activities such as neuroprotection,² antiviral,³ antitumor,⁴ anti-HIV,⁵ antimalarial,⁶ antiplasmodial,⁷ anti-leishmanial,⁸ anti-allergic⁹ etc. Diverse β -carbolines derivatives exhibit tyrosine-phosphorylation regulated kinase (DYRK) activity¹⁰ acetylcholinesterase inhibition.¹¹ The β -carboline moiety is a very important synthetic intermediate for a variety of pharmaceutically important natural products and drug molecules (fig. 1). The Significant biological properties of β -carboline makes it synthetically important unit. Enormous efforts have been devoted for inventing efficient and cost-effective catalysts for the dehydro-aromatization of heterocyclic compounds under mild conditions.¹² Several methods have been developed to synthesize β -carbolines. Pictet-Spengler reaction is one of the methods used to obtain tetrahydro- β -carbolines, which upon dehydrogenation offers β -carbolines. The reagents such as $K_2Cr_2O_7$,¹³ $KMnO_4$,¹⁴ copper salts,¹⁵ Palladium on carbon,^{16,35g} sulphur,¹⁷ SeO_2 ,¹⁸ MnO_2 ,¹⁹ NCS ,²⁰ DDQ ²¹ and NBS ²² have been used for the oxidative aromatization and dehydrogenation. Whereas K. C. Nicolaou *et al.*,²³ have reported IBX-mediated oxidative dehydrogenation of a secondary amine, Shannon S. Stahl *et al.*,^{23c} reported oxidation of secondary amines and nitrogen heterocycles using 1,10-phenanthroline-5,6-dione/ ZnI_2 , recently Subhabrata Sen *et al.*,²⁴ have used the

combination of IBX- $CeCl_3$ for the partial dehydrogenation of tetrahydro- β -carbolines to 3,4-dihydro- β -carbolines.

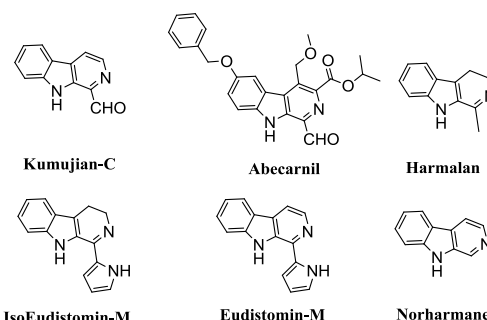


Fig. 1. Carboline alkaloids and drugs.

Recently Water *et al.*, have reported Dess-Martin periodinane (DMP) and 2-iodoxybenzoic acid (IBX) mediated oxidative aromatization, partial dehydrogenation of tetrahydro- β -carbolines at room temperature.^{25a} Kamal *et al.*, have reported $PhI(OAc)_2$ (diacetoxyiodobenzene) mediated oxidative decarboxylation of tetrahydro- β -carbolines to the corresponding β -carbolines^{25b}. However, most of the reported methods for the aromatization of tetrahydro- β -carbolines generally involve the following drawbacks. Pd mediated reactions need a higher temp. (up to 150 °C), long reaction hours, low yields, along with the possibility of C-N bond cleavage.²⁶ Selenium mediated reactions need huge

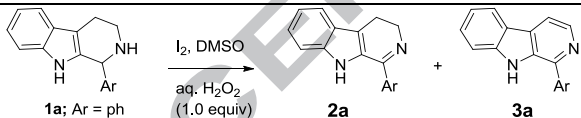
quantity of SeO₂ (10 equiv) offering very less yield (65 %).²⁷ Sulphur mediated reactions need 2 days to deliver aromatized product in low yields.²⁸ DDQ and chloranil were found useful in some cases but yields obtained were not satisfactory always.²¹ While hypervalent iodine reagent IBX is usually required in super stoichiometric amounts and/or often requires ester functionality for successful β -carboline formation.

We have been always interested in iodine mediated, significant organo-synthetic transformations. Some of the recent molecular I₂ mediated methods especially oxidative dehydrogenation reactions reported by our research group are, I₂/DMSO mediated facile dehydrogenation of dihydropyridazin-3(2H)one,²⁹ oxidative cyclizations of 2'-hydroxy and 2'-allyloxy chalcones directly to the corresponding flavones and iodoflavones³⁰ instead of flavanones. Regioselective oxidative dehydrogenation in anilinodihydrochalcones.³⁰ Aromatization and halogenation of 3,3a,4,5-tetrahydro-3-aryl-2-phenyl-2H-benzo[g]indazoles.³¹ Aromatization and in-situ iodination of tetrahydrocarbazoles.³² In pursuance of our efforts to develop new synthetic methods for the compounds of medicinal importance, we attempted the synthesis of 3,4-dihydro- β -carboline, and aromatic- β -carbolines using molecular iodine under easy reaction conditions. Additionally, this method has been employed for the synthesis of bioactive alkaloids such as Kumujian-C, Eudistomin-U, Isoeudistomin-U, Norharmane, Harmane, Harmalan, and Isoeudistomin-M.

1. Result and Discussion

Accordingly, when tetrahydro- β -carboline **1a** was stirred with 10 mol% molecular iodine at r.t. in DMSO, we got 3,4-dihydro- β -carboline **2a** with 50% yield (entry 1, Table 1). Increasing amount of molecular iodine to 25 mol% could not improve the yield significantly (55%) (Entry 2, Table 1). It is reported in the literature that peroxide promotes the dehydrogenation reaction.³³ Therefore, we employed TBHP and H₂O₂ (100 mol% each) along with 10 mol% I₂ in separate reactions, resulting in 60% yield of the selectively oxidized product **2a** (3, 4-dihydro- β -carboline) (Entry 3, Table 1). Increasing the I₂ to 25 mol% could not lead to better yield.

Table 1. Optimization of oxidative dehydrogenation and aromatization reaction

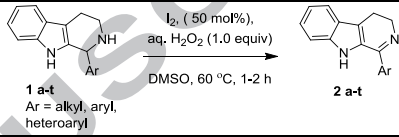
						
Entry	I ₂ (mol%)	Oxidant (100 mol%)	Temp (°C)	Time (h)	Yield ^a % (2a)	Yield ^b % (3a)
1	10	-	RT	4	50	-
2	25	-	RT	4	55	-
3	10	H ₂ O ₂	RT	3	60	-
4	25	H ₂ O ₂	RT	3	60	-
5	10	H ₂ O ₂	60	4	65	-
6	25	H ₂ O ₂	60	4	72	-
7	50	H ₂ O ₂	60	2	90	-
8	100	H ₂ O ₂	60	2	90	-
9	10	TBHP	RT	4	54	-
10	25	TBHP	RT	4	56	-
11	25	H ₂ O ₂	80	2	87	Trace
12	25	H ₂ O ₂	100	10	-	88
13	50	H ₂ O ₂	100	8	-	89
14	75	H ₂ O ₂	100	8	-	90
15	100	H ₂ O ₂	100	8	-	90

^{a,b} Clear TLC observed in long range wavelength.

Therefore we thought to increase the temperature, accordingly, when the reaction was carried out at 60 °C a slight increase in yield of the **2a** (65%) was observed. Use of 25 mol% of I₂, resulted in 72% yield at 60 °C. To our delight, we got 90 % yield of **2a** with 50 mol% I₂ in just 2 hours. However no further improvement was observed in terms of the yield with 100 mol%

of I₂. Changing the peroxide from H₂O₂ to *tert*-butylhydroperoxide (TBHP) with 10 and 25 mol% of I₂ at room temperature could lead to 54 and 56% of **2a**. Therefore we decided to continue with H₂O₂. Further, to check the effect of enhanced temperature. The reaction was carried out at 80 °C with 25 mol% I₂ surprisingly, resulted in 87% of **2a** and a trace quantity of aromatized product β -carboline **3a**. However at 20 °C increases in temperature offered exclusively the β -carboline **3a** in 88% yield over 10 hours. Increasing the iodine quantity to 50, 75 and 100 mol% could lead to 89-90% yield at the most (entry 13, 14, and 15, table 1). Therefore we preferred to continue with 25 mol% of I₂. Notably, an increase in the reaction temperature minimizes the use of molecular iodine and offers the aromatized product (Entry 12, Table 1). Here the controlled dehydrogenation and the aromatization of tetrahydro- β -carboline is temperature dependent.

Table 2. Chemoselective dehydrogenation of tetrahydro- β -carboline

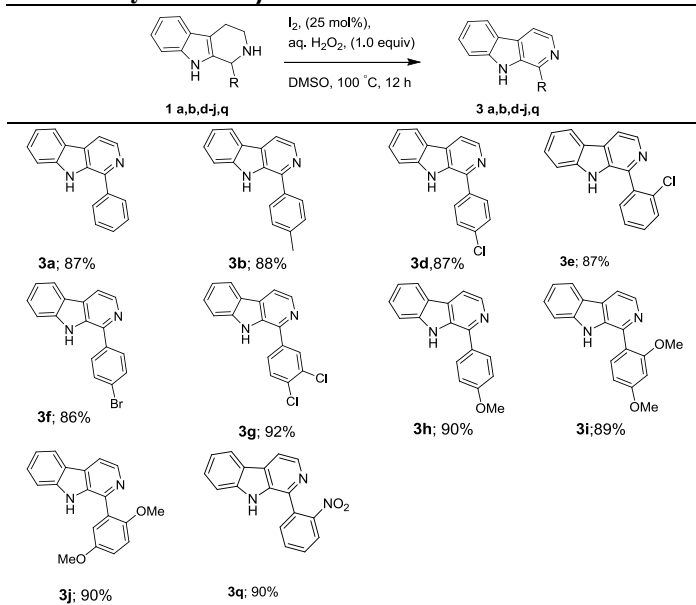
	
2a : 90%	2b : 90%
2c : 81%	2d : 88%
2e : 35%	2f : 88%
2g : 82%	2h : 90%
2i : 85%	2j : 86%
2k : 87%	2l : 86%
2m : 92%	2n : 81%
2o : 75%	2p : 80%
2q : 25%	2r : 78%
2s : 80%	2t : 80%
Isoeudistomin-M	
Harmalan	

^a Yields: 35% 3,4-dihydro- β -carboline; 50% β -carboline

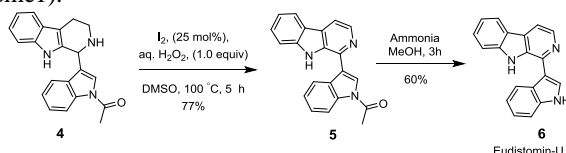
^b Yields: 25% 3,4-dihydro- β -carboline; 55% β -carboline

Having the selective oxidation and aromatization condition in hand, we planned to explore the selective oxidation condition first. Accordingly, a range of tetrahydro- β -carbolines was subjected under the selective oxidation condition (entry 7, Table 1), substrates bearing electron-donating substituents offered excellent yields (up to 92%) (Entry 2m, Table 2). In case of *ortho*-amino substitution, dihydro- β -carboline was obtained in 80% yield (entry 2p, table 2). However, in case of electron-withdrawing substituents like *ortho*-NO₂ and *ortho*-chloro the obtained yields of the dihydro- β -carbolines were 25% and 35% respectively, along with 55% and 50% corresponding β -carbolines. Slight low yields were observed in case of heterocyclic aryl rings (entry 2n, and 2o, Table 2).

A range of tetrahydro- β -carbolines was next examined to probe the substrate scope of I₂-H₂O₂ mediated aromatization process. Tetrahydro- β -carbolines with the substitutions such as Me, OMe and halogens on the phenyl ring offered β -carbolines in very good yields (table 3). Interestingly, tetrahydro- β -carbolines bearing an *o*-chlorophenyl, *o*-nitrophenyl got converted into the corresponding β -carboline in quite better yields, (Entry 3e and 3q, Table 3) contrary to the chemoselective dehydrogenation of the same substrates (Entry 2e and 2q, Table 2)

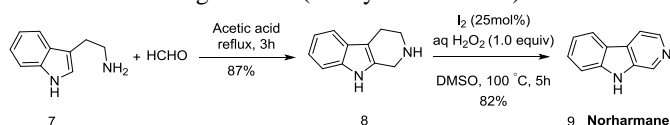
Table 3. Synthesis of β -carbolines.

The various tetrahydro- β -carbolines underwent the oxidative dehydrogenation prompted us to investigate the applicability of this method toward the synthesis of some important marine alkaloids such as Kumujian-C, Eudistomin-U, Norharmane, Harmane. Kumujian-C isolated from root wood of *Picrasma quassioides*^{34a}. Eudistomin-U was isolated from the marine ascidian *lissoclinum fragile* and was reported to possess anticancer activities with strong DNA binding potential, antimicrobial activity as well as exhibit antiviral activities against herpes simplex virus-1 and polio vaccine type-1 virus.^{34b} There are seven reports available for the synthesis of Eudistomin-U^{22,25,35}. Eudistomin-U was first reported by Molina *et al.*, using an iminophosphorane intermediate.^{35a} Queguiner *et al.*, prepared the natural product via sequential metalations^{35b}. While Yamaguchi & Itami reported a novel oxidative C-H/C-H coupling in their total synthesis.^{35c} Witulski Bernhard and Co-worker used a novel Ru or Rh-catalyzed [2+2+2] cyclization to generate the natural product^{35d}. More recently Stephen Water and co-worker reported aromatization of Pictet-Spengler intermediate to generate eudistomin-U²⁵. Later Kamal and co-worker used NCS & PhI(OAc)₂ reagent for the total synthesis of alkaloid^{22,25}. However, most of these syntheses need expensive metal catalysts,³⁵ critical reaction conditions prolonged reaction hours, multistep reaction sequence, resulting in low overall yield. We describe an efficient, three-step synthesis of Eudistomin-U. Synthesis of Eudistomin-U commenced with Pictet-Spengler reaction of tryptamine and N-acetyl-Indole-3-carbaldehyde to obtain the tetrahydro- β -carboline core **4**. Followed by oxidation of the obtained core **4** using I₂-DMSO under optimized conditions produced aromatized intermediates **5** (77% yields) subsequent deprotection gives eudistomin-U (60% yields, Scheme1).



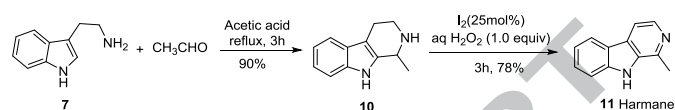
Scheme 1. Synthesis of eudistomin-U.

Synthesis of Norharman^{22,25b} **9** has been carried out by Pictet Spengler cyclization reaction of tryptamine & formaldehyde (37-41% W/V) to obtain 1,2,3,4-tetrahydro- β -carboline **8** (87%). The oxidation of tetrahydro- β -carboline **8** under the optimized reaction condition gives **9** in (82% yield Scheme2).



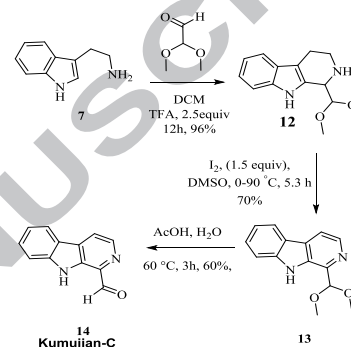
Scheme 2. Synthesis of norharmane.

Subsequently, Pictet-Spengler condensation of tryptamine & acetaldehyde provides 1-methyl-1,2,3,4-tetrahydro- β -carboline **10** 90% yield. Dehydrogenation of **10** under the optimized conditions proceeded smoothly, affording Harmane^{22,25b} **11** in (78% yield Scheme3).



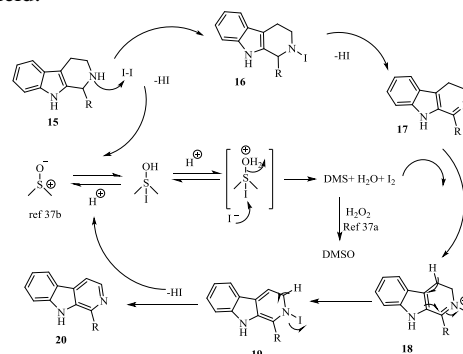
Scheme 3. Synthesis of harmane.

Finally, we also carried out the total synthesis of Kumujian-C **14**, which consists of formyl group at the C-1 position of ring C. We planned to apply aromatization reaction for the synthesis of Kumujian-C.



Scheme 4: Synthesis of Kumujian-C

Accordingly, the required tetrahydro- β -carboline **12** was achieved from the tryptamine **7** and dimethoxy glyoxal. The I₂ mediated aromatization of ring C under the optimized reaction condition (Entry 12, Table 1) offered intermediate **13** which upon stirring in AcOH at 60 °C delivered Kumujian-C³⁶ **14** in 40% overall yield.

Scheme 5: a Plausible mechanism for the I₂/DMSO/H₂O₂ mediated aromatization of tetrahydro- β -carbolines.

N-iodination of tetrahydro- β -carboline **15** takes place in the first step to give intermediate **16**. The dehydroiodination of N-iodotetrahydrocarboline results in the imine intermediate **17**. This imine intermediate gives ammonium intermediate **18** on further N-iodination. Subsequent deprotonation and dehydro-iodination offer the β -carboline. The generated HI further reacts with DMSO to regenerate the molecular iodine, along with dimethyl sulfide (DMS).^{37b} The regeneration of molecular I₂ helps to continue the cycle. The formed dimethylsulfide is oxidized to DMSO in presence of H₂O₂ to speed up the reaction time.^{37a}

Conclusion

In summary, we have developed a simple, efficient and metal free iodine mediated method, for the chemoselective dehydrogenation and aromatization of tetrahydro- β -carbolines to 3,4-dihydro- β -carbolines and β -carbolines. This protocol has been utilized for the synthesis of a library 3,4-dihydro- β -carboline derivatives as well as bioactive alkaloids Kumujian-C **14**, Eudistomin-U **5**, Isoeudistomin-M **20**, Norharmane **8**, Harmane **10** and Harmalan **2t** analogs.

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at.....

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Highlights

- ✓ 30 Examples.
- ✓ I₂ catalysed chemoselective dehydrogenation of tetrahydro- β -carboline to 3,4-dihydro- β -carboline.
- ✓ I₂ catalysed a new synthesis of synthesis Kumujian-C, Eudistomin-U, Harmalan and Isoeudistomine-M.
- ✓ Products synthesized in good to excellent yields.

Graphical Abstract

Iodine-catalyzed chemoselective dehydrogenation and aromatization of Tetrahydro- β -carbolines:

A short synthesis of Kumujian-C, Eudistomin-U, Norharmene, Harmene, Harmalan and Isoeudistomin-M

Sunil Gaikwad, Dayanand Kamble and Pradeep D. Lokhande*

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