Tailored-design Synthesis of Sulfapyrimidine Derivatives



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In this paper, we report an efficient and convenient approach for the synthesis of tailored-design target sulfapyrimidine derivatives expected to show remarkable antimicrobial activities. The approach is based on reacting arylsulfonyl guanidine with α , β -unsaturated carbonyl compounds to afford *N*-(4,6-diarylpyrimidin-2-yl)arylsulfonamide or with ylidene derivatives to afford *N*-(6-aryl-5-cyanopyrimidin-2-yl)arylsulfonamide, *N*-(4-amino-5-cyano-6-(methylthio)-pyrimidin-2-yl)-arylsulfonamide, and *N*-(5-cyanopyrimidin-2-yl)arylsulfonamide compounds through Michael addition reaction. The structure of the newly synthesized compounds was confirmed from spectral data and elemental analysis.

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INTRODUCTION

Sulfonamide-substituted heterocyclic compounds exhibit remarkable activities against both Gram-positive and Gram-negative bacteria [1]. Undoubtedly, the efficacy of the sulfonamide groups of this type of compounds, the ease of their oral administration, and its moderately low toxicity distinguish this class of compounds as one of the important antibacterial chemotherapeutic agents. Sulfapyrimidine [2,3], sulfapyridine [4], and sulfathiazole [5] were some of the early representatives of this group of sulfonamides. The unique antimicrobial properties of sulfapyrimidine inspired the further synthesis of a large number of pyrimidine derivatives of the sulfanilamide compounds, many of which with surprising activity and low toxicity characteristics [6]. Similar to other sulfonamides, sulfapyrimidine derivatives were identified as an inhibitor to dihydropteroate synthase enzyme, which some bacteria use to synthesize folic acid [7,8]. Sulfadiazine, a sulfapyrimidine drug (A) (Figure 1), is commonly used to treat urinary tract infections and burns. Combination of sulfadiazine and pyrimethamine can be used to treat toxoplasmosis, which is caused by Toxoplasma gondii [9,10]. Both sulfamerazine (**B**) and sulfadimidine (**C**) (Figure 1) are used in some acute bacterial infections

such as Escherichia coli enteritis in veterinary medicine [11], cerebrospinal meningitis, and allergic reaction to penicillins [12]. Sulfapyrimidine drugs with dimethyl groups such as sulfisomidine (D) (Figure 1) are used for treatment of urinary infections because of their remarkable solubility in urine and absence of kidney damage [13]. It is commonly used as a sulfa therapy in human serum hemolytic complement system [14]. Sulfadoxine (E) (Figure 1) is used in combination with pyrimethamine for treatment of malaria [15]. Sulfadimethoxine (F) (Figure 1), a sulfonamide antimicrobial agent, is regularly employed in veterinary medicine because of its enduring half-life of approximately 40 h. It is commonly used to treat several bacterial infections such as respiratory, soft tissue, and urinary tract infections [16]. It can also be administered as a standalone or in combination with some other ormetoprim to widen the spectrum of the target range [17] as well as treating coccidian parasites infections [18].

Different approaches have been reported in the synthesis of sulfapyrimidine derivatives. The condensation reaction of *B*-diketones or *B*-ketocarboxylic esters with arylsulfonyl guanidine is a well-known reaction for the preparation of sulfapyrimidine derivatives [19–21]. Another approach for the synthesis of sulfapyrimidine derivatives is by reacting different 2-aminopyrimidines to



Figure 1. Structure of sulfapyrimidine drugs.

arylsulfonyl chloride [22]. Several derivatives of diphenysulfapyrimidine acetates have also been synthesized by reacting chalcones with sulfaguanidine acetate [23,24]. Our research group has also been involved in synthesizing a variety of substituted heterocyclic derivatives with good biological activities [25-30]. In this paper, we report on a new approach for the synthesis of tailored-design target sulfapyrimidine derivatives expected to show remarkable antimicrobial activities. The approach is based on reacting arylsulfonyl guanidine with α,β -unsaturated carbonyl compounds or with ylidene derivatives of both malononitrile and ethyl cyanoacetate.

RESULTS AND DISCUSSION

Arylsulfonyl guanidine **3a**,**b** was first synthesized by reacting arylsulfonyl chloride with excess guanidine hydrochloride in the presence of potassium hydroxide in 1.4-dioxane at room temperature. To determine the appropriate reaction conditions for the synthesis of N-(4,6-diarylpyrimidin-2-yl)arylsulfonamide 5a-h (Scheme 1), the reaction of 3-(4-chlorophenyl)-1-(1.2)phenylprop-2-en-1-one mmol) 4a and benzenesulfonyl guanidine (1 mmol) 3a was chosen as a model experiment. Different reaction conditions were evaluated, and the results were all summarized in





Table 1. A preliminary screening of catalysts and solvents for the reaction showed that the use of potassium hydroxide and a nonpolar solvent such as 1.4-dioxane under reflux are crucial for this particular cyclic condensation reaction (Table 1, entry 6). Having established the optimal conditions for the condensation reaction, various α,β -unsaturated carbonyl compounds 4a-d (Scheme 1) were allowed to react with benzenesulfonyl 3a and tosyl guanidine 3b separately to form N-(4,6-diarylpyrimidin-2-yl)arylsulfonamide **5a**-d and 5e-h, respectively. The chemical structures of 5a-h were assigned on the basis of its analytical and spectral data. ¹H NMR spectrum of compound **5f**, for example, exhibited two singlet signals at δ 2.29 and 2.37 ppm assignable to the protons of two methyl groups as well as four doublet signals at 8 7.16, 7.25, 7.75, and 7.95 ppm for eight protons residing on the two 4-substituted benzene rings. A singlet signal at δ 7.47 ppm was assigned to the proton at C-5 of the pyrimidine ring.

We also attempted to react arylsulfonyl guanidine 3a,b with arvlidene-substituted malononitrile derivatives. 6ad. and arylidene-substituted ethyl cyanoacetate derivatives, 10a-d, through Michael addition reaction (Scheme 2). Different evaluated reaction conditions are summarized in Table 2. The optimal condition for the reaction of benzenesulfonyl guanidine 3a with 4chlorobenzylidene malononitrile 6b incorporated the use of sodium ethoxide in ethanol under reflux (Table 2, entry 4). Applying the same condition to the reaction of benzenesulfonyl guanidine 3a with 4-chlorobenzylidene ethyl cyanoacetate 10b to afford 5-cyano-6-oxo-1,6dihydropyrimidin-2-yl-benzenesulfonamide 13b resulted in a lower yield. However, employing potassium hydroxide in 1,4-dioxane (Table 2, entry 3) resulted in a higher yield. To explore the scope and limitations of this **Scheme 2.** Synthesis arylsulfonamide.

N-(6-aryl-5-cyanopyrimidin-2-yl)



of

reaction, we have extended the experimentations to include various substituted benzylidene of both malononitrile **6a–d** and ethyl cyanoacetate **10a–d** with both benzensulfonyl **3a** and tosyl guanidine **3b** to yield compounds **9a–f** and **13a–h**, respectively (Scheme 2). The structures of these products were deduced from their

	Ph	$\begin{array}{c} O \\ H_2 \\ S \\ O \\ N \\ A \\ A$	Ph N O N S Ph N O H O Ph H O Ph		
Entry	Catalyst	Solvent	Temp. (°C)	Time	Yield
1	Piperidine	Ethanol	Reflux	8	>10
2	K ₂ CO ₃	Ethanol	Reflux	8	>10
3	NaOEt	Ethanol	RT	24	_
4	NaOEt	Ethanol	Reflux	6	56
5	КОН	1,4-Dioxane	RT	24	25
6	KOH	1,4-Dioxane	Reflux	3	88

 $\label{eq:Table 1} Table \ 1$ Optimization of reaction condition of arylsulfonyl guanidine with \$\alpha,\beta\$-unsaturated carbonyl compounds.

RT, room temperature.

Table 2
Optimization of reaction condition of arylsulfonyl guanidine with arylidene derivatives of malononitrile and ethyl cyanoacetate

		Physics NH ₂	+CN					
		3a	6b: X=CN 10b: X=COOEt	9a: Y=NH ₂ 13b: Y=OH				
				9a		13b		
Entry	Catalyst	Solvent	Temp. (°C)	Time	Yield	Time	Yield	
1	Piperidine	Ethanol	Reflux	8	<20	16	<15	
2	K ₂ CO ₃	Ethanol	Reflux	8	30	10	27	
3	KOH	1,4-Dioxane	Reflux	5	50	3	88	
4	NaOEt	Ethanol	Reflux	3	60	3	71	

respective IR, ¹H NMR, ¹³C NMR, and elemental analysis. The reaction of arylsulfonyl guanidine **3a**,**b** with arylidene derivatives of both malononitrile and ethyl cyanoacetate is assumed to take place via the intermediacy of a nonisolable Michael adducts 7 and 11, respectively, followed by an intramolecular cyclization to afford the aforementioned compounds. In the case of arylidene malononitrile, this has occurred through the addition of the amino group to one of the cyano groups of the malononitrile, whereas in the case of arylidene ethyl cyanoacetate, the cyclization has occurred through the elimination of an ethanol molecule. The sulfapyrimidine compounds 9a-f and 13a-h were then resulted from the oxidation of non-isolable intermediates 8 and 12, respectively.

In order to extend the scope of the reaction of arylsulfonyl guanidine with ylidene compounds, we explored the reaction between arylsulfonyl guanidine 3a, b and ketene dithioacetal derivatives 16a,b. The reaction of 2-cyano-3,3-bis(methylthio)acrylonitrile 16a, prepared elsewhere [31], with arylsulfonyl guanidine in the presence of sodium ethoxide in ethanol under reflux gave *N*-(4-amino-5-cyano-6-(methylthio)pyrimidin-2-yl) arylsulfonamide 17a,b (Scheme 3). Moreover, the reaction of ethyl 2-cyano-3,3-bis(methylthio)acrylate 16b with arylsulfonyl guanidine 3a,b afforded the corresponding N-(5-cyano-4-(methylthio)-6-oxo-1,6dihydropyrimidin-2-yl)benzenesulfonamide 18a,b in good yield using potassium hydroxide in 1,4-dioxane (Scheme 3). The spectral and elemental analyses of compounds 17a,b and 18a,b were consistent with their structures. For example, IR spectrum of compound 17a displayed absorption bands at v 33430 and 3348 cm⁻¹ for NH and NH₂ groups, respectively, as well as a band at $v 2197 \text{ cm}^{-1}$ corresponding to the CN group. In the

Scheme 3. Synthesis of *N*-(5-cyano-6-(methylthio)pyrimidin-2-yl) arylsulfonamide.



case of compound **18b**, the IR spectra displayed absorption bands at v 3434, 2206, and 1644 cm⁻¹ for NH, CN, and C=O groups, respectively. The ¹H NMR spectrum of compound **17a** exhibited one singlet signals at δ 2.12 ppm due to SCH₃ protons in addition to one broad band at δ 6.57 ppm for NH₂ protons while for compound **18b**, it showed two singlet signals at δ 2.16 and 2.33 ppm for SCH₃ and CH₃ protons, respectively. The reaction proceeded *via* Michael addition of one of the amino groups in the arylsulfonyl guanidine to the C=C of the ketene dithioacetal compounds followed by the elimination of CH₃SH molecule and the subsequent intramolecular cyclization.

reaction of arylsulfonyl guanidine The with ethoxymethylene compounds to produce new derivatives of sulfapyrimidine was also investigated. Cyclization of compounds arylsulfonyl guanidine 3a,b with ethoxymethylenemalononitrile 19 in the presence of sodium ethoxide and ethanol under reflux afforded the corresponding products **20a**,**b** (Scheme 4). Additionally, the reaction of ethyl (ethoxymethylene)cyanoacetate 21 with arylsulfonyl guanidine 3a,b using KOH in 1,4dioxane under reflux resulted in the pyrimidine derivatives 22a,b in good yield. All products were characterized by spectral data and elemental analysis. The IR spectrum of compound 22a was characterized by the presence of absorption bands at v 3414, 2220, 1669 cm^{-1} corresponding to NH. CN. and C=O groups, respectively. Moreover, its ¹H NMR spectrum revealed the presence of a singlet signal at δ 7.95 ppm for the CH proton of pyrimidine ring and another singlet signal at δ 11.20 ppm for the NH proton of the sulfonamide group. To rationalize the aforementioned results, a possible reaction mechanism is proposed, in which the ethoxymethylene compounds underwent a Michael addition reaction with the arylsulfonyl guanidines that is followed by the

Scheme 4. Synthesis of *N*-(5-cyanopyrimidin-2-yl)arylsulfonamide.



Scheme 5. Reaction mechanism for the synthesis of *N*-(5-cyanopyrimidin-2-yl)arylsulfonamide.



elimination of ethanol, intramolecular cyclization, and an isomerization of the resultant molecule to afford the *N*-(5-cyanopyrimidin-2-yl)arylsulfonamide products (Scheme 5).

CONCLUSION

In conclusion, we have described an efficient approach for the preparation of 30 new derivatives of sulfapyrimidines by reacting arylsulfonyl guanidine with α . β -unsaturated carbonyl compounds or with vlidene derivatives such as arylidene-substituted malononitrile, arylidene-substituted ethyl cyanoacetate, 2-cyano-3,3bis(methylthio)acrylonitrile. ethvl 2-cvano-3.3bis(methylthio)acrylate, ethoxymethylenemalononitrile, and ethvl (ethoxymethylene)cyanoacetate through Michael addition reaction. The structure of newly synthesized compounds was confirmed spectroscopically and by elemental analysis.

EXPERIMENTAL

All melting points were uncorrected on a Gallenkamp melting point apparatus. IR spectra (KBr discs) were recorded on an FTIR plus 460 or Pye Unicam SP-1000 spectrophotometer. The ¹H and ¹³C NMR spectra were recorded on a Bruker Avance (III)-400 spectrometer (400 and 100 MHz, respectively) in DMSO- d_6 using Si (CH₃)₄ as an internal standard at the Ain Shams University, Cairo, Egypt. Elemental analyses were obtained from the Microanalytical Data Center at Cairo University, Egypt, and were performed on Vario El III Elemental CHNS analyzer. Progress of the reactions and the purity of compounds were monitored and checked by thin-layer chromatography (TLC) aluminum sheets coated with silica gel F254 (Merck), using a mixture of petroleum ether and ethyl acetates as an eluent.

General procedure for the synthesis of 5a-h. A solution of arylsulfonyl guanidine 3a,b (0.01 mol) and chalcone 4a-d (0.012 mmol) in dry dioxane (20 mL) containing potassium hydroxide (0.015 mol) was heated under reflux for 2–4 h as judged by TLC. The reaction mixture was cooled and poured into ice acidified with HCl. The solid product formed was filtered off, washed by ethyl acetate, and recrystallized from ethanol to give the respective products 5a-h.

N-(4-(4-Chlorophenyl)-6-phenylpyrimidin-2-yl)

benzenesulfonamide 5a. Beige crystals; yield 88%; mp 262–263°C; IR (KBr, cm⁻¹): υ 3433 (NH), 3061 (ArCH), 1562 (C=C), 1353, 1132 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.32–8.06 (m, 15H, 2C₆H₅, C₆H₄ & CH= pyrimidine). ¹³C NMR (100 MHz,

DMSO- d_6): δ 101.2 (CH= pyrimidine), 125.9, 127.1, 127.4, 128.0, 128.9, 129.1, 129.4, 135.0, 137.6, 138.6, 148.1, 148.4, 162.9, 164.5, 165.4 (Ar–C). *Anal.* Calcd. for C₂₂H₁₆ClN₃O₂S (421.90): C% 62.63; H% 3.82; N% 9.96. Found: C% 62.93; H% 3.91; N% 9.72.

N-(4-Phenyl-6-p-tolylpyrimidin-2-yl)benzenesulfonamide

5b. White crystals; yield 86%; mp 306–307°C; IR (KBr, cm⁻¹): v 3435 (NH), 3056 (ArCH), 1563 (C=C), 1356, 1136 (SO₂). ¹H NMR (400 MHz, DMSO- d_6): δ 2.36 (s, 3H, CH₃), 7.23–8.00 (m, 15H, 2C₆H₅, C₆H₄ & CH= pyrimidine). *Anal.* Calcd. for C₂₃H₁₉N₃O₂S (401.48): C% 68.81; H% 4.77; N% 10.47. Found: C% 68.98; H% 5.06; N% 10.12.

N-(4-(4-Methoxyphenyl)-6-phenylpyrimidin-2-yl)

benzenesulfonamide 5*c.* Beige crystals; yield 90%; mp 295–296°C; IR (KBr, cm⁻¹): υ 3436 (NH), 3059 (ArCH), 1564 (C=C), 1356, 1135 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.82 (s, 3H, OCH₃), 6.98 (d, J = 8 Hz, 2H, C₆H₄), 7.31–7.45 (m, 7H, 2C₆H₅ & CH= pyrimidine), 7.84–7.87 (m, 2H, C₆H₅), 7.99–8.01 (m, 4H, C₆H₅ & C₆H₄). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 55.9 (OCH₃), 100.5 (CH= pyrimidine), 114.2, 127.1, 127.3, 127.9, 128.7, 128.8, 129.3, 130.1, 139.2, 148.6, 161.1, 164.1 (Ar–C). *Anal.* Calcd. for C₂₃H₁₉N₃O₃S (417.48): C% 66.17; H% 4.59; N% 10.07. Found: C% 66.50; H% 4.42; N% 10.22.

N-(4,6-Bis(4-methoxyphenyl)pyrimidin-2-yl)

benzenesulfonamide 5d. White crystals; yield 86%; mp 236–237°C; IR (KBr, cm⁻¹): υ 3436 (NH), 3058 (ArCH), 1586 (C=C), 1363, 1164 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.85 (s, 6H, 2OCH₃), 7.08–7.10 (m, 2H, C₆H₄), 7.54–7.62 (m, 6H, 2C₆H₄, C₆H₅ & CH= pyrimidine), 8.07–8.19 (m, 7H, 2C₆H₄ & C₆H₅), 11.81 (s, 1H, NHSO₂). *Anal.* Calcd. for C₂₄H₂₁N₃O₄S (447.51): C% 64.41; H% 4.73; N% 9.39. Found: C% 66.53; H% 4.64; N% 10.21.

N-(4-(4-Chlorophenyl)-6-phenylpyrimidin-2-yl)-4-

methylbenzenesulfonamide 5e. White crystals; yield 83%; mp 292–293°C; IR (KBr, cm⁻¹): υ 3435 (NH), 2931 (ArCH), 1563 (C=C), 1357, 1135 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.29 (s, 3H, CH₃), 7.16 (d, *J* = 8 Hz, 2H, C₆H₄), 7.43–7.53 (m, 6H, 2C₆H₄, C₆H₅ & CH= pyrimidine), 7.62–7.65 (m, 2H, 2C₆H₄ & C₆H₅), 8.02–8.05 (m, 4H, 2C₆H₄ & C₆H₅). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 21.3 (CH₃), 101.2 (CH= pyrimidine), 127.2, 127.4, 128.4, 128.7, 129.2, 130.3, 135.0, 137.6, 138.6, 139.0, 141.4, 145.1, 158.8, 164.5, 165.3 (Ar–C). *Anal.* Calcd. for C₂₃H₁₈ClN₃O₂S (435.93): C% 63.37; H% 4.16; N% 9.64. Found: C% 63.57; H% 4.31; N% 9.52.

4-Methyl-N-(4-phenyl-6-p-tolylpyrimidin-2-yl)

benzenesulfonamide 5f. White crystals; yield 80%; mp 289–290°C; IR (KBr, cm⁻¹): υ 3433 (NH), 2918 (ArCH), 1567 (C=C), 1356, 1130 (SO₂). ¹H NMR (400 MHz, DMSO- d_6): δ 2.29 (s, 3H, CH₃), 2.37 (s, 3H,

CH₃), 7.16 (d, J = 8 Hz, 2H, C₆H₄), 7.25 (d, J = 8 Hz, 2H, C₆H₄), 7.43–7.44 (m, 3H, C₆H₅), 7.47 (s, 1H, CH= pyrimidine), 7.75 (d, J = 8 Hz, 2H, C₆H₄), 7.95 (d, J = 8 Hz, 2H, C₆H₄), 7.95 (d, J = 8 Hz, 2H, C₆H₄), 8.01–8.04 (m, 2H, C₆H₅). *Anal.* Calcd. for C₂₄H₂₁N₃O₂S (415.51): C% 69.37; H% 5.09; N% 10.11. Found: C% 69.67; H% 5.30; N% 9.88.

N-(4-(4-Methoxyphenyl)-6-phenylpyrimidin-2-yl)-4methylbenzenesulfonamide 5g. White crystals; yield 86%; mp 275–276°C; IR (KBr, cm⁻¹): υ 3436 (NH), 1561 (C=C), 1356, 1143 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.29 (s, 3H, CH₃), 3.83 (s, 3H, OCH₃), 7.00 (d, *J* = 8 Hz, 2H, C₆H₄), 7.17 (d, *J* = 8 Hz, 2H, C₆H₄), 7.43–7.44 (m, 4H, C₆H₅ & CH= pyrimidine), 7.75 (d, *J* = 8 Hz, 2H, C₆H₄), 8.02-8.53 (m, 4H, C₆H₄ & C₆H₅). *Anal.* Calcd. for C₂₄H₂₁N₃O₃S (431.51): C% 66.80; H% 4.91; N% 9.74. Found: C% 67.00; H% 4.80; N% 9.92.

N-(4,6-Bis(4-methoxyphenyl)pyrimidin-2-yl)-4-

methylbenzenesulfonamide 5h. White crystals; yield 89%; mp 302–303°C; IR (KBr, cm⁻¹): v 3438 (NH), 1593 (C=C), 1362, 1168 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.28 (s, 3H, CH₃), 3.83 (s, 6H, OCH₃), 6.98 (d, *J* = 8 Hz, 4H, 2C₆H₄), 7.15 (d, *J* = 8 Hz, 2H, C₆H₄), 7.38 (s, 1H, CH= pyrimidine), 7.73 (d, *J* = 8 Hz, 2H, C₆H₄), 8.00 (d, *J* = 8 Hz, 4H, 2C₆H₄). *Anal.* Calcd. for C₂₅H₂₃N₃O₄S (461.53): C% 65.06; H% 5.02; N% 9.10. Found: C% 65.33; H% 5.25; N% 9.20.

General procedure for the synthesis of 9a–f. To a solution of sodium ethoxide in ethanol, prepared from 0.013 mol of sodium added to 10 mL of absolute ethanol, and arylsulfonyl guanidine 3a,b (0.01 mol), arylidene malononitrile 6a-d (0.012 mol) was added. After heating the mixture under reflux for 2–4 h as judged by TLC, the reaction mixture was cooled and poured into ice acidified with HCl. The solid product was separated by filtration, washed ethanol, and dried under vacuum.

N-(4-Amino-6-(4-chlorophenyl)-5-cyanopyrimidin-2-yl)

benzenesulfonamide 9a. Beige crystals; yield 60%; mp >350°C; IR (KBr, cm⁻¹): v 3442, 3364 (NH, NH₂), 3100 (ArCH), 2209 (CN), 1611 (C=C), 1328, 1145 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.34–8.06 (m, 11H, C₆H₅, C₆H₄ & NH₂), 12.05 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 115.3 (CN), 90.9, 94.5, 128.0, 128.3, 130.2, 130.4, 131.3, 133.9, 135.8, 144.3, 155.1, 160.8 (Ar–C). *Anal.* Calcd. for C₁₇H₁₂ClN₅O₂S (385.83): C% 52.92; H% 3.13; N% 18.15. Found: C% 53.30; H% 3.23; N% 17.85.

N-(4-Amino-5-cyano-6-(4-methoxyphenyl)pyrimidin-2-yl) benzenesulfonamide 9b. White crystals; yield 55%; mp 278–279°C; IR (KBr, cm⁻¹): v 3445, 3357 (NH, NH₂), 3149 (ArCH), 2210 (CN), 1560 (C=C), 1318, 1158 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.83 (s, 3H, OCH₃), 7.06 (d, *J* = 8 Hz, 2H, C₆H₄), 7.49–7.70 (m, 7H, C₆H₅ & NH₂), 8.04 (d, *J* = 8 Hz, 2H, C₆H₄), 11.81 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₈H₁₅N₅O₃S (381.41): C% 56.68; H% 3.96; N% 18.36. Found: C% 56.95; H% 4.10; N% 18.19.

N-(4-Amino-5-cyano-6-phenylpyrimidin-2-yl)-4-

methylbenzenesulfonamide 9c. White crystals; yield 75%; mp 248–249°C; IR (KBr, cm⁻¹): υ 3400, 3340 (NH, NH₂), 3246 (ArCH), 2215 (CN), 1556 (C=C), 1327, 1159 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.38 (s, 3H, CH₃), 7.36 (d, *J* = 8 Hz, 2H, C₆H₄), 7.51–7.59 (m, 3H, C₆H₅), 7.69 (d, *J* = 8 Hz, 2H, C₆H₄), 7.95 (d, *J* = 8 Hz, 2H, C₆H₄), 11.74 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 21.5 (CH₃), 116.6 (CN), 81.4, 128.8, 128.9, 129.5, 131.5, 137.8, 143.8, 157.5, 164.8 (Ar–C). *Anal.* Calcd. for C₁₈H₁₅N₅O₂S (365.41): C% 59.16; H% 4.14; N% 19.17. Found: C% 59.39; H% 4.02; N% 19.32.

N-(*4*-*Amino*-6-(*4*-*chlorophenyl*)-5-*cyanopyrimidin*-2-*yl*)-4*methylbenzenesulfonamide* 9*d*. White crystals; yield 78%; mp 293–294°C; IR (KBr, cm⁻¹): v 3408, 3346 (NH, NH₂), 3161 (ArCH), 2215 (CN), 1554 (C=C), 1333, 1160 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.35 (s, 3H, CH₃), 7.36 (d, *J* = 11 Hz, 2H, C₆H₄), 7.62 (d, *J* = 11 Hz, 2H, C₆H₄), 7.72 (d, *J* = 11 Hz, 2H, C₆H₄), 7.93 (d, *J* = 11 Hz, 2H, C₆H₄), 11.75 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₈H₁₄ClN₅O₂S (399.85): C% 54.07; H% 3.53; N% 17.51. Found: C% 54.45; H% 3.25; N% 17.72.

N-(*4*-*Amino-5-cyano-6-(4-methoxyphenyl)pyrimidin-2-yl)-4-methylbenzenesulfonamide 9e.* Off white crystals; yield 80%; mp 290–291°C; IR (KBr, cm⁻¹): υ 3418, 3350 (NH, NH₂), 3149 (ArCH), 2214 (CN), 1567 (C=C), 1321, 1159 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.33 (s, 3H, CH₃), 7.08 (d, *J* = 8 Hz, 2H, C₆H₄), 7.35 (d, *J* = 8 Hz, 2H, C₆H₄), 7.73 (d, *J* = 8 Hz, 2H, C₆H₄), 7.94 (d, *J* = 8 Hz, 2H, C₆H₄), 7.73 (d, *J* = 8 Hz, 2H, C₆H₄), 7.94 (d, *J* = 8 Hz, 2H, C₆H₄), 11.67 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 21.4 (CH₃), 55.9 (OCH₃), 117.0 (CN), 66.8, 80.2, 128.4, 129.5, 130.7, 138.1, 143.6, 162.0, 165.0 (Ar–C). *Anal.* Calcd. for C₁₉H₁₇N₅O₃S (395.43): C% 57.71; H% 4.33; N% 17.71. Found: C% 57.95; H% 4.48; N% 17.92.

N-(4-Amino-5-cyano-6-p-tolylpyrimidin-2-yl)-4-

methylbenzenesulfonamide 9f. White crystals; yield 80%; mp 289–290°C; IR (KBr, cm⁻¹): v 3417, 3347 (NH, NH₂), 3163 (ArCH), 2214 (CN), 1567 (C=C), 1321, 1158 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.37 (s, 3H, CH₃), 2.39 (s, 3H, CH₃), 7.32–7.37 (m, 4H, 2 C₆H₄), 7.61 (d, *J* = 8 Hz, 2H, C₆H₄), 7.94 (d, *J* = 8 Hz, 2H, C₆H₄), 11.66 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₉H₁₇N₅O₂S (379.44): C% 60.14; H% 4.52; N% 18.46. Found: C% 60.47; H% 4.37; N% 18.25.

General procedure for the synthesis of 13a–h. A solution of arylsulfonyl guanidine **3a,b** (0.01 mol) and arylidene ethyl cyanoacetate **10a–d** (0.012 mmol) in dry dioxane (20 mL) containing potassium hydroxide (0.015 mol) was heated under reflux for 2–4 h as judged by TLC. The reaction mixture was cooled and poured into ice acidified with HCl. The solid product formed was

filtered off, washed by ethyl acetate, and recrystallized from ethanol to give the respective products 13a-h.

N-(5-*Cyano-6-oxo-4-phenyl-1,6-dihydropyrimidin-2-yl) benzenesulfonamide 13a.* Off white crystals; yield 85%; mp >350°C; IR (KBr, cm⁻¹): v 3431 (NH), 3053 (ArCH), 2210 (CN), 1650 (C=O), 1563 (C=C), 1350, 1142 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.57 (s, 1H, NH), 7.41–7.79 (m, 10H, 2C₆H₅), 11.32 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₇H₁₂N₄O₃S (352.37): C% 57.95; H% 3.43; N% 15.90. Found: C% 58.30; H% 3.58; N% 16.11.

N-(4-(4-Chlorophenyl)-5-cyano-6-oxo-1,6-dihydropyrimidin-2-yl)benzenesulfonamide 13b. Beige crystals; yield 88%; mp >350°C; IR (KBr, cm⁻¹): v 3438 (NH), 3057 (ArCH), 2207 (CN), 1650 (C=O), 1564 (C=C), 1348, 1143 (SO₂). ¹H NMR (400 MHz, DMSO-d₆): δ 3.57 (s, 1H, NH), 7.42–7.51 (m, 7H, C₆H₅ & C₆H₄), 7.76–7.78 (m, 2H, C₆H₅), 11.34 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-d₆): δ 118.7 (CN), 83.9, 127.3, 128.3, 128.4, 130.9, 130.7, 135.5, 136.5, 145.7, (Ar–C), 168.2 (CO). Anal. Calcd. for C₁₇H₁₁ClN₄O₃S (386.81): C% 52.79; H% 2.87; N% 14.48. Found: C% 53.08; H% 2.70; N% 14.73.

N-(5-Cyano-4-(4-methoxyphenyl)-6-oxo-1,6-

dihydropyrimidin-2-yl)benzenesulfonamide 13*c*. Beige crystals; yield 87%; mp >350°C; IR (KBr, cm⁻¹): υ 3432 (NH), 3053 (ArCH), 2212 (CN), 1645 (C=O), 1554 (C=C), 1346, 1142 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 3.57 (s, 1H, NH), 3.82 (s, 3H, OCH₃), 6.96 (d, *J* = 8 Hz, 2H, C₆H₄), 7.42–7.47 (m, 3H, C₆H₅), 7.52 (d, *J* = 8 Hz, 2H, C₆H₄), 7.78–7.80 (m, 2H, C₆H₅), 11.19 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₈H₁₄N₄O₄S (382.39): C% 56.54; H% 3.69; N% 14.65. Found: C% 56.90; H% 3.54; N% 14.94.

N-(5-Cyano-6-oxo-4-p-tolyl-1,6-dihydropyrimidin-2-yl)

benzenesulfonamide 13d. Beige crystals; yield 85%; mp >350°C; IR (KBr, cm⁻¹): υ 3431 (NH), 3052 (ArCH), 2206 (CN), 1651 (C=O), 1562 (C=C), 1347, 1140 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.35 (s, 3H, CH₃), 3.57 (s, 1H, NH), 7.21 (d, *J* = 8 Hz, 2H, C₆H₄), 7.39–7.46 (m, 5H, C₆H₅ & C₆H₄), 7.79 (d, *J* = 8 Hz, 2H, C₆H₅). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 21.3 (CH₃), 118.7 (CN), 83.9, 127.5, 128.3, 128.7, 128.8, 130.9, 137.6, 140.7, 142.7, 156.7, 163.7 (Ar–C), 169.7 (CO). *Anal.* Calcd. for C₁₈H₁₄N₄O₃S (366.39): C% 59.01; H% 3.85; N% 15.29. Found: C% 59.35; H% 4.05; N% 15.04.

N-(5-*Cyano-6-oxo-4-phenyl-1,6-dihydropyrimidin-2-yl)-4methylbenzenesulfonamide 13e.* White crystals; yield 81%; mp >350°C; IR (KBr, cm⁻¹): v 3434 (NH), 3051 (ArCH), 2209 (CN), 1651 (C=O), 1564 (C=C), 1349, 1142 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.35 (s, 3H, CH₃), 3.58 (s, 1H, NH), 7.24 (d, *J* = 13.6 Hz, 2H, C₆H₄), 7.42– 7.52 (m, 5H, C₆H₅), 7.68 (d, *J* = 11.2 Hz, 2H, C₆H₄), 11.24 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₈H₁₄N₄O₃S (366.39): C% 59.01; H% 3.85; N% 15.29. Found: C% 59.38; H% 3.98; N% 15.42. *N*-(*4*-(*4*-Chlorophenyl)-5-cyano-6-oxo-1,6-dihydropyrimidin-2-yl)-4-methylbenzenesulfonamide 13f. Beige crystals; yield 88%; mp >350°C; IR (KBr, cm⁻¹): υ 3436 (NH), 3047 (ArCH), 2206 (CN), 1649 (C=O), 1565 (C=C), 1348, 1141 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.34 (s, 3H, CH₃), 3.57 (s, 1H, NH), 7.24 (d, *J* = 8 Hz, 2H, C₆H₄), 7.50–7.55 (m, 4H, 2C₆H₄), 7.67 (d, *J* = 8 Hz, 2H, C₆H₄). *Anal.* Calcd. for C₁₈H₁₃ClN₄O₃S (400.84): C% 53.94; H% 3.27; N% 13.98. Found: C% 54.24; H% 3.03: N% 14.22.

N-(5-Cyano-4-(4-methoxyphenyl)-6-oxo-1,6dihydropyrimidin-2-yl)-4-methylbenzenesulfonamide 13g.

White crystals; yield 87%; mp >350°C; IR (KBr, cm⁻¹): v 3432 (NH), 3195 (ArCH), 2212 (CN), 1646 (C=O), 1557 (C=C), 1345, 1140 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.34 (s, 3H, CH₃), 3.83 (s, 3H, OCH₃), 6.98 (d, *J* = 8 Hz, 2H, C₆H₄), 7.24 (d, *J* = 8 Hz, 2H, C₆H₄), 7.57 (d, *J* = 8 Hz, 2H, C₆H₄), 7.68 (d, *J* = 8 Hz, 2H, 2H, C₆H₄), 11.19 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 21.3 (CH₃), 55.8 (OCH₃), 119.0 (CN), 83.0, 113.7, 127.5, 128.7, 129.8, 130.7, 140.7, 142.7, 156.5, 161.6, 163.9 (Ar–C), 168.9 (CO). *Anal.* Calcd. for C₁₉H₁₆N₄O₄S (396.42): C% 57.57; H% 4.07; N% 14.13. Found: C% 57.92; H% 4.28; N% 13.97.

N-(5-*Cyano*-6-*oxo*-4-p-tolyl-1,6-dihydropyrimidin-2-yl)-4methylbenzenesulfonamide 13h. White crystals; yield 82%; mp >350°C; IR (KBr, cm⁻¹): v 3438 (NH), 3027 (ArCH), 2206 (CN), 1651 (C=O), 1573 (C=C), 1346, 1146 (SO₂). ¹H NMR (400 MHz, DMSO-d₆): δ 2.34 (s, 3H, CH₃), 2.36 (s, 3H, CH₃), 7.21–7.30 (m, 4H, 2C₆H₄), 7.41 (d, *J* = 8 Hz, 2H, C₆H₄), 7.66 (d, *J* = 8 Hz, 2H, C₆H₄), 11.19 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-d₆): δ 21.3, 21.4 (2CH₃), 118.8 (CN), 83.6, 126.0, 127.5, 128.7, 129.4, 134.8, 140.6, 140.8, 142.7, 157.6, 163.7 (Ar–C), 169.6 (CO). Anal. Calcd. for C₁₉H₁₆N₄O₃S (380.42): C% 59.99; H% 4.24; N% 14.73. Found: C% 60.30; H% 4.03; N% 14.97.

General procedure for the synthesis of 17a,b. To a solution of sodium ethoxide in ethanol, prepared from 0.013 mol of sodium added to 10 mL of absolute ethanol, and arylsulfonyl guanidine 3a,b (0.01 mol), 16a (0.012 mol) was added. After heating the mixture under reflux for 2–4 h as judged by TLC, the reaction mixture was cooled and poured into ice acidified with HCl. The solid product was separated by filtration, washed ethanol, and dried under vacuum.

N-(4-Amino-5-cyano-6-(methylthio)pyrimidin-2-yl)

benzenesulfonamide 17a. Yellow crystals; yield 65%; mp 228–229°C; IR (KBr, cm⁻¹): υ 3430, 3348 (NH, NH₂), 3242 (ArCH), 2197 (CN), 1555 (C=C), 1403, 1132 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.12 (s, 3H, SCH₃), 6.57 (s, 2H, NH₂), 7.36–7.37 (m, 3H, C₆H₅), 7.77–7.79 (m, 2H, C₆H₅). *Anal.* Calcd. for $C_{12}H_{11}N_5O_2S_2$ (321.38): C% 44.85; H% 3.45; N% 21.79. Found: C% 45.10; H% 3.62; N% 22.00.

N-(4-Amino-5-cyano-6-(methylthio)pyrimidin-2-yl)-4-

methylbenzenesulfonamide 17b. Yellow crystals; yield 55%; mp 262–263°C; IR (KBr, cm⁻¹): υ 3435, 3340 (NH, NH₂), 3246 (ArCH), 2207 (CN), 1558 (C=C), 1409, 1136 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.17 (s, 3H, SCH₃), 2.30 (s, 3H, CH₃), 6.48 (s, 2H, NH₂), 7.22 (d, *J* = 8 Hz, 2H, C₆H₄), 7.65 (d, *J* = 8 Hz, 2H, C₆H₄). *Anal.* Calcd. for C₁₃H₁₃N₅O₂S₂ (335.40): C% 46.55; H% 3.91; N% 20.88. Found: C% 46.87; H% 4.18; N% 20.55.

General procedure for the synthesis of 18a,b. A solution of arylsulfonyl guanidine **3a,b** (0.01 mol) and **16b** (0.012 mmol) in dry dioxane (20 mL) containing potassium hydroxide (0.015 mol) was heated under reflux for 2–4 h as judged by TLC. The reaction mixture was cooled and poured into ice acidified with HCl. The solid product formed was filtered off, washed by ethyl acetate, and recrystallized from ethanol to give the respective products **18a,h**.

N-(5-*Cyano-4-(methylthio)-6-oxo-1,6-dihydropyrimidin-2-yl)* benzenesulfonamide 18a. Pale yellow crystals; yield 80%; mp >350°C; IR (KBr, cm⁻¹): v 3448 (NH), 3043 (ArCH), 2213 (CN), 1653 (C=O), 1556 (C=C), 1472, 1144 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.10 (s, 3H, SCH₃), 7.44–7.48 (m, 3H, C₆H₅), 7.74–7.76 (m, 2H, C₆H₅), 11.12 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 12.6 (SCH₃), 117.1 (CN), 81.5, 126.6, 128.6, 130.9, 145.5, 155.1, 174.3 (Ar–C), 162.8 (C=O). Anal. Calcd. for C₁₂H₁₀N₄O₃S₂ (322.36): C% 44.71; H% 3.13; N% 17.38. Found: C% 44.97; H% 3.00; N% 17.64.

N-(5-*Cyano-4-(methylthio)-6-oxo-1,6-dihydropyrimidin-2-yl)-4-methylbenzenesulfonamide* 18b. Yellow crystals; yield 87%; mp >350°C; IR (KBr, cm⁻¹): υ 3434 (NH), 3045 (ArCH), 2206 (CN), 1644 (C=O), 1558 (C=C), 1464, 1144 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.16 (s, 3H, SCH₃), 2.33 (s, 3H, CH₃), 7.24 (d, *J* = 8 Hz, 2H, C₆H₄), 7.62 (d, *J* = 8 Hz, 2H, C₆H₄), 11.03 (s, 1H, NHSO₂). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 12.6 (SCH₃), 21.3 (CH₃), 117.4 (CN), 81.3, 126.7, 129.0, 140.7, 142.8, 155.2, 174.1 (Ar–C), 161.7 (C=O). *Anal.* Calcd. for C₁₃H₁₂N₄O₃S₂ (336.39): C% 46.42; H% 3.60; N% 16.66. Found: C% 46.88; H% 3.45; N% 16.87.

General procedure for the synthesis of 20a,b. To a solution of sodium ethoxide in ethanol, prepared from 0.013 mol of sodium added to 10 mL of absolute ethanol, and arylsulfonyl guanidine **3a,b** (0.01 mol), **19** (0.012 mol) was added. After heating the mixture under reflux for 2–4 h as judged by TLC, the reaction mixture was cooled and poured into ice acidified with HCl. The solid product was separated by filtration, washed ethanol, and dried under vacuum.

N-(4-Amino-5-cyanopyrimidin-2-yl)benzenesulfonamide

20a. Off white crystals; yield 50%; mp 210–211°C; IR (KBr, cm⁻¹): v 3406, 3323 (NH, NH₂), 3219 (ArCH), 2218 (CN), 1555 (C=C), 1414, 1134 (SO₂). ¹H NMR

(400 MHz, DMSO- d_6): 7.43 (s, 2H, NH₂), 7.51–7.53 (m, 3H, C₆H₅), 7.82–7.83 (m, 2H, C₆H₅), 8.32 (s, 1H, CH= pyrimidine), 14.51 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₁H₉N₅O₂S (275.29): C% 47.99; H% 3.30; N% 25.44. Found: C% 48.33; H% 3.52; N% 25.17.

N-(4-Amino-5-cyanopyrimidin-2-yl)-4-

methylbenzenesulfonamide 20b. Off white crystals; yield 60%; mp 208–209°C; IR (KBr, cm⁻¹): υ 3438, 3351 (NH, NH₂), 3230 (ArCH), 2109 (CN), 1527 (C=C), 1409, 1132 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.35 (s, 3H, CH₃), 7.31 (d, *J* = 8 Hz, 2H, C₆H₄), 7.92 (d, *J* = 8 Hz, 2H, C₆H₄), 8.37 (s, 1H, CH= pyrimidine). *Anal.* Calcd. for C₁₂H₁₁N₅O₂S (289.31): C% 49.82; H% 3.83; N% 24.21. Found: C% 50.17; H% 3.60; N% 24.57.

General procedure for the synthesis of 22a,b. A solution of arylsulfonyl guanidine 3a,b (0.01 mol) and 21 (0.012 mmol) in dry dioxane (20 mL) containing potassium hydroxide (0.015 mol) was heated under reflux for 2–4 h as judged by TLC. The reaction mixture was cooled and poured into ice acidified with HCl. The solid product formed was filtered off, washed by ethyl acetate, and recrystallized from ethanol to give the respective products 22a,b.

N-(5-Cyano-6-oxo-1,6-dihydropyrimidin-2-yl)

benzenesulfonamide 22a. Off white crystals; yield 73%; mp >350°C; IR (KBr, cm⁻¹): υ 3414 (NH), 3065 (ArCH), 2220 (CN), 1669 (C=O), 1534 (C=C), 1361, 1144 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.39–7.46 (m, 3H, C₆H₅), 7.77–7.79 (m, 2H, C₆H₅), 7.95 (s, 1H, CH= pyrimidine), 11.20 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₁H₈N₄O₃S (276.27): C% 47.82; H% 2.92; N% 20.28. Found: C% 48.14; H% 2.71; N% 20.66.

N-(5-Cyano-6-oxo-1,6-dihydropyrimidin-2-yl)-4-

methylbenzenesulfonamide 22*b*. Beige crystals; yield 75%; mp >350°C; IR (KBr, cm⁻¹): v 3420 (NH), 3061 (ArCH), 2208 (CN), 1665 (C=O), 1544 (C=C), 1365, 1145 (SO₂). ¹H NMR (400 MHz, DMSO-*d*₆): δ 2.32 (s, 3H, CH₃), 7.21 (d, *J* = 8 Hz, 2H, C₆H₄), 7.67 (d, *J* = 8 Hz, 2H, C₆H₄), 7.94 (s, 1H, CH= pyrimidine), 11.09 (s, 1H, NHSO₂). *Anal.* Calcd. for C₁₂H₁₀N₄O₃S (290.30): C% 49.65; H% 3.47; N% 19.30. Found: C% 49.89; H% 3.54; N% 19.10.

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SUPPORTING INFORMATION

Additional supporting information may be found online in the Supporting Information section at the end of the article.