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Viridicatumtoxins: Expanding on a rare tetracycline antibiotic scaffold

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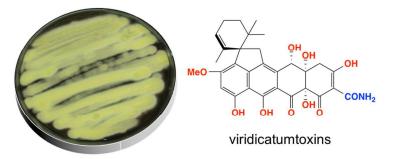
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KEYWORDS (viridicatumtoxins, tetracycline antibiotics, marine-derived fungus)

Abstract



Viridicatumtoxins, which belong to a rare class of fungal tetracycline-like mycotoxins, were subjected to comprehensive spectroscopic and chemical analysis, leading to reassignment/assignment of absolute configurations and characterisation of a remarkably acid-stable antibiotic scaffold. Structure activity relationship studies revealed exceptional growth inhibitory activity against vancomycin-resistant *Enterococci* (IC₅₀ 40 nM), >270-fold more potent than the commercial antibiotic oxytetracycline.

Introduction

Multidrug resistant (MDR) infectious diseases represent an extremely serious threat to modern healthcare, impacting the lives of people and communities across the world. As current and emerging mechanisms for MDR render existing antibiotics less effective, the need to develop new and improved antibiotics is becoming ever more urgent. One promising approach is to explore rare classes of microbial metabolites, first recognized but not developed as commercial antibiotics mid to late last century. Our investigations into the secondary metabolism of Australian marine-derived microbes provided an ideal platform from which to embark on such an exploration, delivering access to a wealth of bacterial and fungal strains rich in rare and novel metabolites, many with promising antibiotic properties.

In this report we describe a comprehensive exploration of the viridicatumtoxin scaffold, a rare class of fungal polyketides closely related to the tetracycline antibiotics. Tetracyclines have been at the forefront in treating infectious diseases for over 60 years, with first generation tetracycline antibiotics (*e.g.* chlortetracycline, tetracycline and oxytetracycline) inspiring second (*e.g.* minocycline and doxycycline) and third (*e.g.* tigecycline and eravacycline) generation variants, many in use today. Notwithstanding the considerable academic and industry investment in the tetracyclines over many decades, and the role they continue to play in modern healthcare, even this remarkably successful antibiotic pharmacophore is in need of renewal. We hypothesized that a better understanding of the chemical and antibiotic properties of their fungal-metabolite cousins, the viridicatumtoxins, would inform the future development of new and improved tetracycline antibiotics.

Results and Discussion

An agar plate (PYGA) cultivation of a *Paecilomyces* sp. (CMB-MF010), isolated from the inner tissues of an intertidal pulmonate mollusk (*Siphonaria* sp.) collected near Shorncliffe, Queensland, Australia, exhibited promising Gram-positive antibacterial properties. Fractionation of a scaled up cultivation yielded viridicatumtoxins A $(1)^1$ and B (2),^{2,3,4} and the new viridicatumtoxins D–F (**4**–**6**). Subsequent fractionation of a rice solid phase cultivation added to this diversity, yielding all other known members of this structure class, spirohexaline (7)⁵ and previridicatumtoxin (**8**),^{6,7} together with the new viridicatumtoxin C (**3**). In this report we present a comprehensive study of **1–8**, inclusive of detailed spectroscopic analysis leading to structure elucidation, assignment/re-assignment of absolute configurations, inclusive of commentary on biosynthetic relationships and chemical stability, as well as antibiotic and cytotoxic properties.

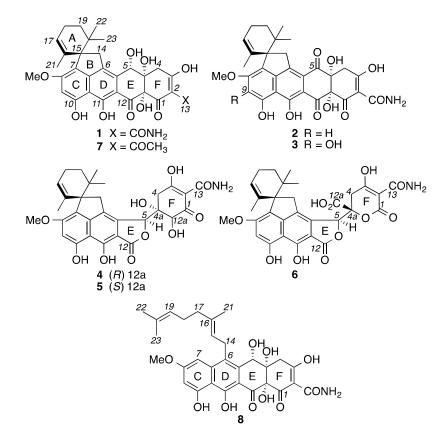


Figure 1. Paecilomyces sp. (CMB-MF010) viridicatumtoxins 1-8

HRESI(+)MS analysis of **1** and **2** revealed pseudo-molecular ions consistent with the molecular formula $C_{30}H_{31}NO_{10}$ (Δ mmu –0.4) and $C_{30}H_{29}NO_{10}$ (Δ mmu –0.2), which detailed analysis of the 1D and 2D NMR (CDCl₃) data attributed to the rare fungal polyketides viridicatumtoxin¹ and viridicatumtoxin B.^{2,3,4} Viridicatumtoxin was first described in 1973 by Hutchison *et al.* as a mycotoxin produced by maize meal cultures of *Penicillium viridicatum*.¹ Subsequent X-ray analyses permitted assignment of firstly its relative stereochemistry⁸ then subsequently its absolute structure.⁹ Further studies explored its biosynthesis.^{6-7,10-12} Viridicatumtoxin B was reported in 2008 by Kim *et al.* as a co-metabolite with viridicatumtoxin from a Korean soil *Penicillium* sp. (FR11), with both metabolites exhibiting anti-MRSA activity.² A 2013 synthesis by Nicolaou *et al.* revised the structure for viridicatumtoxin B (relative configuration only) including the crystal structure of its synthetic racemate, and saw viridicatumtoxin renamed viridicatumtoxin A.^{3,4} In 2013 Tomoda *et al.* described spirohexaline (7) as a co-metabolite with a purported (+)*-enantiomer* of viridicatumtoxin A from a rice-based cultivation of *Penicillium brasilianum* (FKI-3368).⁵

Although (+)-1 isolated from *Paecilomyces* sp. (CMB-MF010) correlated with the (+)enantiomer encountered by Tomoda *et al.*,⁵ this enantiomeric configuration was contradicted by our independent X-ray analysis which is essentially the same as that reported by Silverton *et al.*⁹ The crystal structure of 1 (isomorphous with the previously published structures of the dimethanol solvate^{8,9}) was found to be the di-methanol hemihydrate. The absolute structure was confirmed by statistical analysis of anomalous dispersion effects of 2045 Bijvoet pairs collected from an entire sphere of data with Cu-K α radiation (see ESI). The absolute configurations at all four chirotopic C-atoms (C-4a, C-5, C-12a and C-15) are *S*. To explain this contradiction we speculated that viridicatumtoxin A can exist as (+)-acid and (-)-salt forms with $[\alpha]_D$

measurements of opposite sign and comparable magnitude. Consistent with this hypothesis, the (+)-acid form of 1 ($[\alpha]_D$ +31.8, EtOH) isolated from CMB-MF010 in the presence of TFA, was treated with MgSO₄ to yield a Mg (–)-salt form ($[\alpha]_D$ –10.0, EtOH). As predicted, on exposure to HCl the (–)-salt reverted to the (+)-acid. Significantly, a commercial sample of viridicatumtoxin A was determined to be a (–)-salt, with HCl treatment leading to conversion to the (+)-acid. Comparable changes in the $[\alpha]_D$ of 1 and oxytetracycline (9) were also observed in a range of solvents, suggesting that this differential acid/salt optical rotation phenomenon is a characteristic of the tetracycline scaffold. Based on these observations, and on consideration of published isolation protocols, we propose that the (+)-enantiomers of viridicatumtoxins A and B reported by Tomoda *et al.*⁵ and Kim *et al.*² respectively, are in fact the free acids (+)-1 and (+)-2 respectively, as isolated and characterized herein.

Furthermore, with detailed spectroscopic analysis identifying 7 ($C_{31}H_{32}O_{10}$, Δ mmu +0.1) as spirohexaline,⁵ on biosynthetic grounds we take this opportunity to correct its absolute configuration, aligning it with the co-metabolite 1 (Figure 1). Likewise, spectroscopic analysis identified 8 ($C_{30}H_{33}NO_{10}$, Δ mmu -0.3) as previridicatumtoxin, first reported in 2012 as a late-stage biosynthetic precursor to 1 accumulated in the gene deletion strain *Penicillium aethiopicum* $\Delta gsfA$,^{6,7} we now acknowledge 8 as a natural product with an absolute configuration in common with its co-metabolite 1. Having fully characterized and assigned/corrected absolute configurations for all four known viridicatumtoxins (1–2, 7–8), we now describe the new viridicatumtoxins C–F (3–6).

HRESI(+)MS analysis of **3** returned a molecular formula ($C_{30}H_{29}NO_{11}$, $\Delta mmu - 0.1$) consistent with an oxygenated homologue of **2**. Comparison of the ¹H NMR (CDCl₃) data for **3** (Table 1) with those of **2** revealed the only significant differences as replacement of H-9 ($\delta_{\rm H} 6.80$) in **2** with a phenolic hydroxyl ($\delta_{\rm H}$ 5.97) in **3**, and a deshielding of 8-OMe ($\delta_{\rm H}$ 3.90 to $\delta_{\rm H}$ 4.05). Analysis of the ¹³C NMR data (Table 2) revealed substitution of the C-9 methine in **2** ($\delta_{\rm C}$ 102.6) with a phenolic moiety in **3** ($\delta_{\rm C}$ 140.3), the latter featuring an HMBC correlation to 10-OH ($\delta_{\rm H}$ 8.44). These observations, together with diagnostic 2D NMR correlations (Figure 2), permitted assignment of the structure for viridicatumtoxin C (**3**) as indicated, with the absolute configuration assigned on the basis of biosynthetic links to **1**. Supportive of this assignment, analytical scale treatment of **3** with acid or heat resulted in oxidative conversion to an *ortho*quinone (**3a**), as evidenced by conversion to an *ortho*-phenylenediamine adduct (**3b**) (Figures 2 and S3).

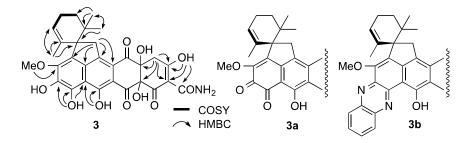


Figure 2. Diagnostic 2D NMR correlations for 3 and derivatives 3a-b

HRESI(+)MS analysis of 4 returned a molecular formula (C₃₀H₃₁NO₁₀, Δ mmu +0.2) isomeric with 1. Comparison of the 1D NMR (CDCl₃:MeOH-*d*₄ [4:1]) data for 4 (Tables 1–2) with 1, together with diagnostic 2D NMR correlations (Figure 3), confirmed common ring A-D and ring F (C-1 to C-4) substructures. Further comparisons revealed the oxymethine resonances for H-5/C-5 in 4 ($\delta_{\rm H}$ 5.65; $\delta_{\rm C}$ 80.2) were significantly deshielded compared to 1 ($\delta_{\rm H}$ 4.50; $\delta_{\rm C}$ 71.8), with 1 and 4 exhibiting common HMBC correlations from H-5 to C-4, C-4a, C-5a, C-6, C-11a and C-12a, and with the quaternary C-12a ($\delta_{\rm C}$ 80.3) and C-12 ($\delta_{\rm C}$ 195.3) in 1 replaced by oxymethine ($\delta_{\rm H}$ 4.53; $\delta_{\rm C}$ 74.3) and ester/lactone ($\delta_{\rm C}$ 155.0) resonances in 4. As further evidence of a C-12a

secondary OH moiety, a sample of **4** was converted to the tetraacetate **4a** (Figure S28) to reveal the predicted deshielding of the H-12a methine (**4** $\delta_{\rm H}$ 4.53; **4a** $\delta_{\rm H}$ 5.74).

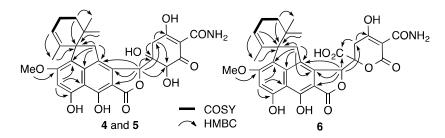


Figure 3. Diagnostic 2D NMR correlations for 4, 5 and 6

Based on these observations we propose that **4** is a *seco*-ring E analogue of **1**, formed by nucleophilic attack at C-12, leading to cleavage of the C-12 to C-12a bond and subsequent intramolecular lactonisation from 5-OH to C-12 (Figure 4). We anticipate that this sequence proceeds with retention of configuration about C-15, C-4a and C-5, and racemisation about C-12a. Consistent with this hypothesis, viridicatumtoxin E (**5**) was identified as the C-12a epimer of **4**, with individual C-12a configurations inferred from ROESY correlations (Table S11). Diagnostic ROESY correlations (Figures 5 and S41–S42) established *syn* relationships between $H_{4\alpha}/H_{12a}$ in **4**, and $H_{4\beta}/H_{12a}$ in **5**, with both **4** and **5** possessing a common H_5 to $H_{14\alpha}$ configuration. Significantly, a key ROESY correlation positioned H_{12a} and $H_{14\beta}$ in close proximity in **5** (C_{12a} *S*), necessitating an epimeric configuration for **4** (C_{12a} *R*).



52 53

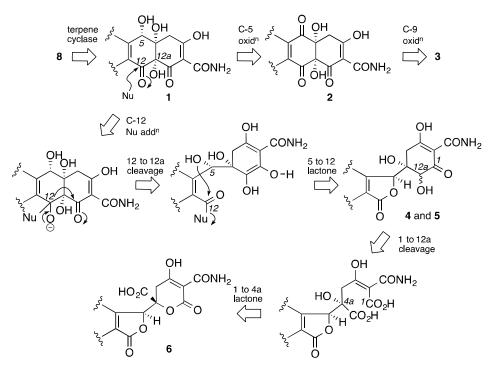


Figure 4. Plausible biosynthetic pathway linking 1 with 2–6 and 8

HRESI(+)MS analysis of **6** returned a molecular formula ($C_{30}H_{29}NO_{11}$, $\Delta mmu +0.2$) suggestive of an oxidized analogue of **4** (and **5**). Comparison of 1D and 2D NMR (CDCl₃:MeOH-*d*₄ [4:1]) (Tables 1–2) data for **6** with **4** revealed common rings A-E, and C-2 to C-4 ring F substructures (Figure 1), with 1D NMR differences focused around replacement of the C-12a/H-12a oxymethine (δ_C 74.3; δ_H 4.53) and C-1 ketone (δ_C 194.5) in **4** with a C-12a carboxylic acid (δ_C 170.3) and C-1 ester/lactone (δ_C 166.2) in **6**, and a deshielding of H₂-4 ($\Delta\delta_H$ 0.71 and 0.96) and C-4a ($\Delta\delta_C$ 5.3) in **6** compared to **4**. Diagnostic 2D HMBC correlations (Figure 3) positioned the carboxylic acid moiety at C-4a, necessitating closure of a C-1 to C-4a lactone, and permitting assignment of the planar structure as indicated. A plausible biosynthetic pathway (Figure 4) provides an alternate oxidative C-1 to C-12a cleavage of the precursor linking **1** with **4** and 5, to deliver viridicatum toxin F (6), with conservation of configuration about all chiral centers.

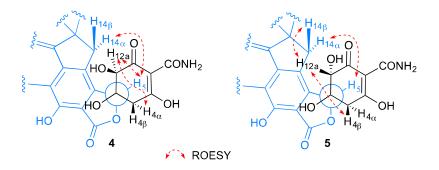


Figure 5. Diagnostic ROESY correlations for 4 and 5

Viridicatumtoxins D (4) and E (5) bear a striking structural similarity to α -apo-oxytetracycline (11) and β -apo-oxytetracycline (12), well-known but poorly characterized acid degradation products and impurities detected in the commercial antibiotic oxytetracycline (9).¹³ To support structure assignments for 4 and 5, we subjected 9 to two acid degradation protocols, (i) 1 h at 75°C in 0.1 M HCl_{aq} (pH 1), and (ii) 4 h at 37°C in 0.01 M HCl_{aq} (pH 2), to yield authentic standards of 11 and 12 together with the associated artifacts 10 and 13 (Figures 6 and S7–S8). Structures for 11–13 were confirmed by detailed spectroscopic analysis (Tables S17–S19). By contrast, extended treatment of 1 under these conditions (>10 h) resulted in no degradation (Figure S6), with the more forcing conditions of 20 h at 100 °C in 5% *p*-toluene sulfonic acid/toluene delivering very minor yields (~1.5%) of the quinone-methides 14 (C₂₈H₂₈O₇, Δ mmu –0.3) and 15 (C₂₇H₂₄O₆, Δ mmu –0.5) (Figures 7 and S9). Our studies reveal for the first time that viridicatumtoxin A is remarkably acid stable, in stark contrast to the highly acid labile tetracycline framework, confirming **4–6** as natural products not handling artifacts.

Table 1. ¹ H NM	R (600 MHz) data for viridicatu	mtoxins C-F	(3-6)
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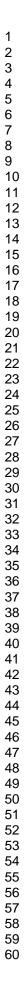
Pos.	3 ^a	4 ^b	5 ^b	6 ^b
4	α 2.83 (d, 18.6)	α 2.44 (d, 17.6)	α 2.39 (d, 18.3)	α 3.15 (d, 17.2)
	β 3.08 (d, 18.6)	β 2.02 (d, 17.6)	β 2.66 (d, 18.3)	β 2.98 (d, 17.2)
5	-	5.65 (s)	5.72 (s)	5.85 (s)
9	-	6.61 (s)	6.60 (s)	6.60 (s)
12a	-	4.53 (br s)	4.29 (br s)	_
14	α 4.00 (d, 19.8)	a 3.19 (d, 18.1)	α 3.25 (d, 17.6)	α 3.20 (d, 17.5)
	β 3.14 (d, 19.8)	β 3.09 (dd, 18.1, 1.6)	β 3.09 (d, 17.6)	β 2.92 (d, 17.5)
17	5.58 (br s)	5.37 (br s)	5.42 (br s)	5.39 (br s)
18	α 2.13 (m)	α 1.92 (br d, 18.0)	α 1.95 (br d, 18.4)	α 1.91 (br d, 18.6)
	β 2.23 (m)	β 2.14 (m)	β 2.17 (m)	β 2.15 (m)
19	1.71 (2H, m)	α 1.74 (m)	α 1.78 (m)	α 1.76 (m)
	-	β 1.25 (dd, 13.0, 5.9)	β 1.27 (dd, 13.3, 5.7)	β 1.25 (dd, 13.1, 5.8)
21	1.45 (s)	1.47 (s)	1.51 (s)	1.46 (s)
22	0.88 (s)	0.85 (s)	0.85 (s)	0.81 (s)
23	0.63 (s)	0.34 (s)	0.33 (s)	0.30 (s)
8-OCH ₃	4.05 (s)	3.80 (s)	3.80 (s)	3.78 (s)
3-ОН	17.91 (br s)	-	_	-
4a-OH	4.23 (br s)	_	_	_
9-OH	5.97 (br s) ^c	_	_	_
10-OH	8.44 (br s)	_	_	_
11-OH	14.14 (br s)	_	_	_
12a-OH	4.89 (br s)	_	-	_
13-NH ₂	a 9.21 (br s)	_	_	_
-	b 5.97 (br s) ^c	_	_	_

^a¹H NMR spectra were acquired in CDCl₃; ^b¹H NMR spectra were acquired in CDCl₃/CD₃OD (4:1); ^c Overlapping signals

Table 2. ¹³ C NMR (150 MHz) data for viridicatumtoxins A–F (1–6), spirohexaline (7) and	
previridicatumtoxin (8)	

Pos.	1 ^a	2 ^a	3 ^a	4 ^b	5 ^b	6 ^b	7 ^a	8 ^a
1	190.6	194.3	193.6	194.5	192.8	166.2	190.3	190.1
2	99.7	99.7	99.7	99.2	99.4	99.1	110.6	99.6
3	192.9	192.8	192.9	192.7	191.3	189.7	194.5	193.0
4	40.5	42.2	42.4	36.8	39.0	35.9	40.8	40.4
4a	71.6	77.8	77.3	76.0	75.9	81.3	71.3	71.3
5	71.8	188.8	188.7	80.2	81.2	80.5	71.7	71.3
5a	137.2	146.4°	145.6°	130.5	130.2	129.5	137.3	130.5
6	123.9	124.9 ^c	133.0 ^c	130.0	129.4	128.7	123.8	130.8
6a	147.3	144.8°	137.5°	148.7	148.1	148.5	147.3	140.9
7	122.8	127.3	114.9	121.9	121.3	121.7	122.8	99.9
8	160.9	161.3	150.4	157.8	157.6	157.8	160.9	164.8
9	100.0	102.6	140.3	98.9	99.0	99.1	100.0	102.2
10	158.1	158.4	141.6	155.8	155.3	155.9	158.1	161.1
10a	105.6	106.9	108.5	107.2	106.2	107.3	105.6	108.7
11	166.1	165.4	162.3	171.3	170.7	170.5	166.1	167.2
11a	105.2	107.3	108.6	103.6	102.4	103.4	105.2	104.2
12	195.3	195.2	196.6	155.0	153.5	154.8	195.4	195.7
12a	80.3	80.7	80.8	74.3	77.8	170.3	80.8	79.4
13	172.8	173.0	173.0	172.8	172.0	172.7	202.4	172.8
14	41.3	44.6	43.4	43.8	43.0	42.0	41.3	27.3
15	60.2	60.6	61.4	60.0	59.5	60.2	60.2	122.3
16	136.7	135.8	135.1	137.5	137.0	137.3	136.7	137.0
17	121.5	122.0	122.8	120.7	120.4	121.0	121.5	39.7
18	23.0	23.0	23.0	23.0	22.5	22.9	23.0	26.7
19	34.0	34.2	34.3	34.0	33.5	33.8	34.0	123.9
20	38.7	38.5	37.9	38.3	37.9	38.3	38.7	131.9
21	21.1	20.9	20.8	21.3	20.8	21.0	21.1	16.7
22	24.1	24.5	25.2	23.9	23.4	23.7	24.1	25.8
23	25.6	25.6	25.7	25.6	25.1	25.5	25.6	17.9
2-CO <u>CH</u> 3	_	_	-	_	_	_	27.9	-
8-OCH ₃	55.7	55.8	60.7	55.5	55.1	55.4	55.7	55.7

^{a 13}C NMR spectra were acquired in CDCl₃; ^{b 13}C NMR spectra were acquired in CDCl₃/CD₃OD (4:1); ^c The chemical shifts of carbons were tentatively assigned by ACD simulation



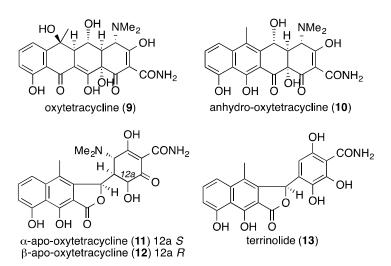


Figure 6. Oxytetracycline (9) and degradation products 10–13

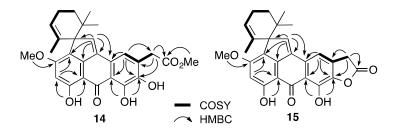


Figure 7. Viridicatumtoxin A (1) minor degradation products 14–15

Although originally described as a mycotoxin,¹ oral dosage of **1** has been judged non-toxic to mice (up to 350 mg/kg) and rats (up to 150 mg/kg).¹⁴ Armed with a comprehensive suite of all known and new viridicatumtoxins (**1–8**), we compared their growth inhibitory properties against a panel of drug-resistant Gram positive and negative bacteria, with those of the tetracycline antibiotic **9** and its acid degradation products **11–13** (Table S4). While **1–5** and **7–8** exhibited anti-MRSA properties comparable with **9**, viridicatumtoxin B (**2**) was especially effective against VRE, with an IC₅₀ >270-fold more potent than **9**, albeit with moderate levels of cytotoxicity against three human cancer cell lines (Table 3).

	MRSA-2 ^A	VRE ^B	NCI-H460 ^C	KB3-1 ^D	SW620 ^E
1	1.2	1.1	1.0	2.5	1.0
2	0.15	0.04	0.6	1.6	0.6
3	1.7	1.5	9.2	11.0	17.4
4	9.7	3.4	24.2	>30	20.7
5	3.3	1.5	>30	>30	>30
6	>30	>30	>30	>30	>30
7	2.3	3.1	11.3	6.3	16.7
8	4.4	4.8	5.3	4.1	6.0
9	0.5	11.0	15.9	21.3	>30
11	>30	>30	>30	>30	>30
12	>30	>30	>30	>30	>30
13	>30	>30	>30	>30	>30

Table 3. Antibacterial and cytotoxicity properties (IC₅₀ µM) of 1–9 and 11–13

^AAUS-RBWH-MRSA-02 (methicillin-resistant *Staphylococcus aureus*); ^BAUS-RBWH-VRE-01 (vancomycin-resistant *Enterococcus faecalis*), ^C NCI-H460 is a human lung carcinoma cell line, ^DKB3-1 is a human cervix carcinoma cell line, ^ESW620 is a human colon carcinoma cell line

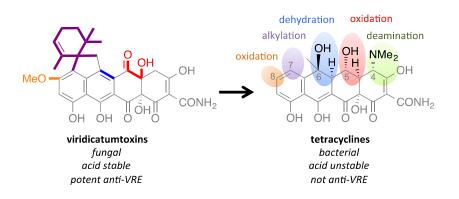


Figure 8. Viridicatumtoxin inspired modifications to tetracyclines

Knowledge of the acid-stable viridicatumtoxin scaffold informs our ability to increase the acid stability of the tetracycline antibiotics, which are easily degraded into the non-active analogues when taken orally. Possible modifications to the tetracycline framework includes deamination at C-4, oxidation at C-4a and C-5, dehydration across C-5a and C-6, alkylation at C-6 and C-7, and oxidation at C-8 (Figure 8).

In conclusion, our investigations into the chemistry of the marine mollusk-derived *Paecilomyces* sp. (CMB-MF010) have greatly expanded knowledge of the viridicatumtoxins, a rare family of mycotoxins closely related to the tetracycline class of antibiotics. In the course of our studies we documented the exceptional acid stability of the viridicatumtoxins, with co-metabolite inspired structure activity relationship investigations revealing the 5-oxo analogue, viridicatumtoxin B (2), as a potent vancomycin-resistant *Enterococci* (VRE) antibiotic. Knowledge acquired during our investigations has the potential to inform the redesign of the tetracycline framework, enhancing acid stability and oral bioavailability, while simultaneously optimizing for and extending antibiotic efficacy to include therapeutically challenging pathogens such as VRE.

Experimental Section

General Experimental Details. Specific optical rotations ($[\alpha]_D$) were acquired on a polarimeter in a 100 × 2 mm cell at room temperature. UV-visible spectra were obtained on a UV-visible spectrophotometer with 1 cm quartz cells. Circular dichroism (CD) spectra were recorded on a spectropolarimeter in 1 mm quartz cells at room temperature with scanning speed of 50 nm/min and 0.5 nm step scan over the wavelength from 200–600 nm. Nuclear magnetic resonance (NMR) spectra were acquired on a 600 MHz spectrometer with either a 5 mm PASEL 1H/D-13C Z-Gradient probe or 5 mm CPTCI 1H/19F-13C/15N/DZ-Gradient cryoprobe. In all cases spectra were acquired at 25 °C (unless otherwise specified) in solvents as specified in the text, with referencing to residual ¹H or ¹³C signals in the deuterated solvents. Electrospray ionization mass spectrometry (ESIMS) experiments were carried out on a LC/MSD (quadrupole) instrument in both positive and negative modes. High-resolution ESIMS spectra were obtained on a micrOTOF mass spectrometer either by direct injection in MeCN at 3 µL/min using sodium

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formate clusters as an internal calibrant or by hyphenating with LC system comprising of corresponding pump, column compartment, autosampler, photodiode array (PDA) detector.

Collection and Isolation of *Paecilomyces* **sp. (CMB-MF010).** The fungus CMB-MF010 was isolated in 2012 from the inner tissue of a marine pulmonate false limpet *Siphonaria* **s**p. collected at the rocky intertidal zone of Moora Park, Shorncliffe, Queensland. The fresh *Siphonaria* sample was transported in a sterile tube (50 mL) on ice to the laboratory, where it was rinsed in sterile natural seawater for 1 min and subjected to surface sterilization in 70% ethanol (ν/ν) for 30 s, after which it was washed with sterile seawater to remove traces of EtOH. Subsequently, the sample was dissected under aseptic conditions and the inner tissue placed on PYG agar plates (comprising 2% glucose, 1% peptone, 0.5% yeast extract, 0.02% chloramphenicol and 1.5% agar in 50% artificial seawater (1.7% salinity)). The plates were wrapped in parafilm and incubated at 26.5 °C for 3–4 weeks. A pure culture of fungus CMB-MF010 was obtained by single-colony serial transfer on agar plates and then cryopreserved at – 80 °C in 15% aqueous glycerol. Taxonomic analysis identified CMB-MF010 as a *Paecilomyces* sp. (see ESI).

Analytical Cultivation and Chemical Profiling of *Paecilomyces* sp. (CMB-MF010). *Paecilomyces* sp. (CMB-MF010) was screened in PSB broth (1% sucrose, 1% mannitol, 0.5% peptone, 0.3% yeast extract and 20% potato juice in water; 200 rpm at 26.5 °C for 7 d in microbioreactor), on PYG agar plates (2% glucose, 1% peptone, 0.5% yeast extract and 1.5% agar; 26.5 °C for 25 d) with and without natural sea salts and on rice solid medium (70 g rice, 0.3% peptone, 0.3% yeast extract, 0.1% monosodium glutamate and 1.7 g natural sea salt prepared in 100 mL distilled water; 26 °C for 25 d). After cultivation, the broth and solid matrix were

extracted exhaustedly with EtOAc and the organic phase was dried *in vacuo* to yield crude extracts. Analytes of crude extracts (5 mg/mL) were prepared in MeOH for HPLC-DAD-ESI(\pm)MS analysis by standard gradient elution (Zorbax SB-C₈ column, 150 × 4.6 mm, 5 µm, 1 mL/min gradient elution from 90% H₂O/MeCN to 100% MeCN with 0.05% formic acid in MeCN as modifier over 15 min). The fungus CMB-MF010 produced different secondary metabolites when grown on PYG agar and rice solid media (Figure S1).

Preparative Cultivation and Fractionation of *Paecilomyces* sp. (CMB-MF010). A single colony of *Paecilomyces* sp. (CMB-MF010) was sub-sampled on PYG agar plates (×100) (2% glucose, 1% peptone and 0.5% yeast extract, prepared in distilled water) and incubated at 26.5°C for 25 days. The resulting agar plates were harvested, diced (~1 cm³) and extracted with EtOAc until the organic phase was almost colorless. The combined organic phase was concentrated in vacuo to yield a crude extract (1173 mg) which was sequentially triturated to afford hexane (153.9 mg), CH₂Cl₂ (700.0 mg) and MeOH (144.0 mg) soluble partitions. The CH₂Cl₂ solubles were subjected to SPE chromatography (GracePureTM C₁₈-Max with a 10% stepwise gradient elution from 70% H₂O/MeOH to MeOH) and the peaks of interest (HPLC-DAD-MS) concentrated in two fractions. The first fraction (38.7 mg) was fractionated with HPLC (Zorbax SB-C₈ column, 250×9.4 mm, 5 µm, 3 mL/min gradient elution from 65% to 48% H₂O/MeCN over 5 min, followed by 48% to 43% H₂O/MeCN over 10 min, with a constant 0.01% TFA/MeCN modifier) to yield viridicatumtoxin D (4) ($t_R = 12.4$ min; 8.2 mg, 0.7%), viridicatumtoxin E (5) ($t_R = 11.8 \text{ min}$; 7.5 mg, 0.6%) and viridicatumtoxin F (6) ($t_R = 11.2 \text{ min}$; 2.6 mg, 0.2%). The second fraction (96.3 mg) was fractionated by HPLC (Zorbax SB-C₈ column, 250×9.4 mm, 5 µm, 3 mL/min gradient elution from 65% to 48% H₂O/MeCN over 5 min, followed by 48% to 43% H₂O/MeCN over 10 min, and from 43% to 30% H₂O/MeCN over 5

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min, with a constant 0.01% TFA/MeCN modifier) to afford viridicatumtoxin A (1) ($t_R = 18.9$ min; 14.1 mg, 1.2%) and viridicatumtoxin B (2) ($t_R = 20.3$ min; 0.6 mg, 0.05%).

For cultivation on rice solid media, a small sample (~1 cm³) of CMB-MF010 colony on agar was used to inoculate a 1 L Erlenmeyer flask containing sterile rice medium, and the mixture was incubated at 26 °C for 25 days. The fungal mycelia as well as the rice media were exhaustively extracted with EtOAc and concentrated in vacuo to obtain a combined EtOAc extract (673.0 mg), which was sequentially triturated to yield hexane (375.9 mg), CH₂Cl₂ (161.1 mg) and MeOH (22.1 mg) solubles, respectively. The CH₂Cl₂ solubles (161.1 mg) were subjected to C₁₈ SPE fractionation (90% H₂O/MeOH to MeOH) to yield 10 fractions. After HPLC-DAD-MS analysis the 50%, 40% and 30% H₂O/MeOH fractions were combined (18.4 mg) and further fractionated by HPLC (Zorbax SB-C₈ column, 250×9.4 mm, 5 μ m, 3 mL/min isocratic elution at 45% H₂O/MeCN over 20 min with a constant 0.01% TFA/MeCN modifier) to afford viridicatumtoxin A (1) (t_R = 13.9 min; 7.4 mg, 1.1%) and viridicatumtoxin C (3) (t_R = 10.5 min; 4.0 mg, 0.6%). After freeze-drying the slightly impure viridicatum toxin C (3) sample was re-purified by HPLC (same elution condition as above) to yield pure viridicatum toxin C (3) (t_R = 10.5 min; 2.9 mg, 0.4%) and pure *ortho*-quinone viridicatumtoxin C (**3a**) ($t_R = 9.8$ min; 0.6 mg). The 10% H₂O/MeOH fraction (53.0 mg) was further fractionated by HPLC (Zorbax SB-C₃) column, 250×9.4 mm, 5 µm, 3 mL/min gradient elution from 50% to 35% H₂O/MeCN over 15 min followed by a 5 min hold at 100% MeCN, with a constant 0.01% TFA/MeCN modifier) to yield spirohexaline (7) ($t_R = 15.4 \text{ min}$; 0.3 mg, 0.04%) and previridicatumtoxin (8) ($t_R = 15.9 \text{ min}$) min; 0.4 mg, 0.06%).

Note: % yields for compounds 1–8 are calculated as weight-to-weight estimate against the crude extract (1172.7 mg for agar plates and 673.0 mg for rice solid media).

Characterization of *Paecilomyces* sp. (CMB-MF010) Metabolites. *Viridicatumtoxin A* (1). Bright yellow powder; $[\alpha]_D^{22}$ +31.8 (*c* 0.2, EtOH); UV (MeOH) λ_{max} (log ε) 236 (4.44), 282 (4.53), 424 (3.94) nm; CD (MeOH) λ_{max} ($\Delta \varepsilon$) 237 (-14.4), 287 (+29.4), 346.5 (-3.0), 361.5 (-0.1), 422.5 (-2.7) nm; NMR (600 MHz, CDCl₃) see Tables 2 and S6; ESI(+)MS *m/z* 548 [M - H₂O + H]⁺, 588 [M + Na]⁺, ESI(-)MS *m/z* 564 [M - H]⁻; HRESI(+)MS *m/z* 588.1844 [M + Na]⁺ (calcd for C₃₀H₃₁NO₁₀Na⁺, 588.1840).

Viridicatumtoxin B (2). Bright yellow powder; $[\alpha]_D^{22}$ +43.3 (c 0.05, EtOH); UV (MeOH) λ_{max} (log ε) 255 (4.40), 285 (4.35, sh), 431 (3.94), 455 (3.99) nm; CD (MeOH) λ_{max} ($\Delta \varepsilon$) 206.5 (-5.4), 227.5 (+4.7), 256 (-1.8), 276.5 (+17.9), 328 (-1.4), 351 (-0.3), 385 (-1.5), 421 (+0.9), 434 (-0.7), 444.5 (-0.1), 459.5 (-4.2) nm; NMR (600 MHz, CDCl₃) see Tables 2 and S7; ESI(+)MS *m*/*z* 564 [M + H]⁺, 586 [M + Na]⁺, ESI(-)MS *m*/*z* 562 [M – H]⁻; HRESI(-)MS *m*/*z* 562.1721 [M – H]⁻ (calcd for C₃₀H₂₈NO₁₀, 562.1719).

Viridicatumtoxin C (3). Yellow powder; $[\alpha]_D^{22}$ +41.1 (c 0.08, EtOH); UV (MeOH) λ_{max} (log ε) 257 (4.42), 303 (4.40), 448 (4.04) nm; CD (MeOH) λ_{max} ($\Delta \varepsilon$) 226 (+6.1), 251 (-6.3), 271 (+7.9), 298 (+0.2), 321.5 (+2.5) nm; NMR (600 MHz, CDCl₃) see Tables 1–2 and S8; ESI(+)MS *m/z* 580 [M + H]⁺, 602 [M + Na]⁺, ESI(-)MS *m/z* 578 [M – H]⁻; HRESI(+)MS *m/z* 602.1634 [M + Na]⁺ (calcd for C₃₀H₂₉NO₁₁Na⁺, 602.1633).

Viridicatumtoxin D (4). Brown amorphous powder; $[\alpha]_D^{22}$ +21.7 (*c* 0.06, EtOH); UV (MeOH) λ_{max} (log ε) 263 (4.64), 344 (3.89), 396 (3.98) nm; CD (MeOH) λ_{max} ($\Delta \varepsilon$) 206 (-2.0), 225.5 (-10.7), 260 (+38.4), 279 (-6.7), 308 (-0.9), 378.5 (-2.7) nm; NMR (600 MHz, CDCl₃/CD₃OD = 4:1) see Tables 1–2 and S9; ESI(+)MS *m*/*z* 566 [M + H]⁺, ESI(-)MS *m*/*z* 564 [M – H]⁻;

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HRESI(+)MS m/z 588.1838 [M + Na]⁺ (calcd for C₃₀H₃₁NO₁₀Na⁺, 588.1840).

Viridicatumtoxin D tetratacetate (4a). A solution of **4** (2 mg, 3.54 µmol) in pyridine (250 µL) and acetic anhydride (200 µL) was stirred at room temperature for 30 h, after which it was dried under N₂ and the crude product subjected to HPLC fractionation (Zorbax SB-C₈ column, 250 × 9.4 mm, 5 µm, 3 mL/min gradient elution from 35% to 20% H₂O/MeCN over 15 min, with an isocratic 0.01% TFA/MeCN modifier) to yield the tetraacetate **4a** (t_R = 9.92 min; 1.5 mg, 75%) as light brown powder; UV (MeOH) λ_{max} (log ε) 268 (4.75), 347 (3.91) nm; ¹H NMR (600 MHz, CDCl₃/CD₃OD = 4:1) δ_H 7.07 (1H, s, H-9), 5.74 (1H, s, H-12a), 5.58 (1H, s, H-5), 5.46 (1H, br s, H-17), 3.86 (3H, s, 8-OCH₃), 3.43 (1H, d, *J* = 18.6 Hz, H-14α), 3.25 (1H, d, *J* = 18.6 Hz, H-14β), 2.58 (1H, d, *J* = 17.0 Hz, H-4α), 2.45 (3H, s, COCH₃), 2.41 (3H, s, COCH₃), 2.33 (3H, s, COCH₃), 2.30 (3H, s, COCH₃), 2.19 (1H, m, H-18β), 2.05 (1H, d, *J* = 17.0 Hz, H-4β), 1.97 (1H, br d, *J* = 18.2 Hz, H-18α), 1.80 (1H, ddd, *J* = 18.4, 12.6, 6.2 Hz, H-19α), 1.52 (3H, s, H-21), 1.31 (1H, dd, *J* = 13.4, 6.2 Hz, H-19β), 0.89 (3H, s, H-22), 0.33 (3H, s, H-23); ESI(+)MS *m/z* 734 [M + H]⁺, 756 [M + Na]⁺, ESI(-)MS *m/z* 732 [M - H]⁻; HRESI(+)MS m/z 756.2263 [M + Na]⁺ (calcd for C₃₈H₃₉NO₁₄Na⁺, 756.2263).

Viridicatumtoxin E (5). Brown amorphous powder; $[\alpha]_D^{22}$ +272.2 (*c* 0.05, EtOH); UV (MeOH) λ_{max} (log ε) 263 (4.64), 344 (3.91), 383 (3.99) nm; CD (MeOH) λ_{max} ($\Delta \varepsilon$) 211.5 (+0.9), 225.5 (-7.5), 266 (+40.6), 326 (-2.7), 360.5 (+2.1) nm; NMR (600 MHz, CDCl₃/CD₃OD = 4:1) see Tables 1–2 and S10; ESI(+)MS *m/z* 566 [M + H]⁺, 588 [M + Na]⁺, ESI(-)MS *m/z* 564 [M – H]⁻; HRESI(+)MS *m/z* 588.1840 [M + Na]⁺ (calcd for C₃₀H₃₁NO₁₀Na⁺, 588.1840).

Viridicatumtoxin F (6). Brown amorphous powder; $[\alpha]_D^{22}$ +146.2 (*c* 0.06, EtOH); UV (MeOH)

 λ_{max} (log ε) 263 (4.55), 347 (3.85), 376 (3.90) nm; CD (MeOH) λ_{max} ($\Delta \varepsilon$) 213.5 (+5.9), 247 (-5.4), 265 (+43.9), 328.5 (-1.3), 359 (+0.7) nm; NMR (600 MHz, CDCl₃/CD₃OD = 4:1) see Tables 1–2 and S12; ESI(+)MS *m/z* 580 [M + H]⁺, ESI(-)MS *m/z* 578 [M – H]⁻, 534 [M – CO₂ – H]⁺; HRESI(+)MS *m/z* 602.1631 [M + Na]⁺ (calcd for C₃₀H₂₉NO₁₁Na⁺, 602.1633).

Spirohexaline (7). Bright yellow powder; $[\alpha]_D^{22}$ +15.5 (c 0.03, EtOH); UV (MeOH) λ_{max} (log ϵ) 236 (4.15), 283 (4.34), 426 (3.69) nm; NMR (600 MHz, CDCl₃) see Tables 2 and S13; ESI(+)MS *m*/*z* 547 [M – H₂O + H]⁺, 587 [M + Na]⁺, ESI(–)MS *m*/*z* 563 [M – H]⁻; HRESI(–)MS *m*/*z* 563.1922 [M – H]⁻ (calcd for C₃₁H₃₁O₁₀, 563.1923).

Previridicatumtoxin (8). Bright yellow powder; $[\alpha]_D^{22}$ +98.8 (c 0.04, EtOH); UV (MeOH) λ_{max} (log ε) 241 (4.50), 273 (4.65), 415 (4.15) nm; NMR (600 MHz, CDCl₃) see Tables 2 and S14; ESI(+)MS *m*/*z* 550 [M – H₂O + H]⁺, 568 [M + H]⁺, 590 [M + Na]⁺, ESI(-)MS *m*/*z* 566 [M – H]⁻; HRESI(-)MS *m*/*z* 566.2035 [M – H]⁻ (calcd for C₃₀H₃₂NO₁₀, 566.2032).

Oxidation and Derivatization of Viridicatumtoxin C (3). A mixture sample of 3 (65%, *w/w*) and 3a (35%, *w/w*) in MeOH (40 µL) was heated at 40 °C overnight, and the product dried under N₂, re-dissolved in MeOH (20 µL) and subjected to HPLC-DAD-MS analysis (Zorbax SB-C₈ column, 150 × 4.6 mm, 5 µm, 1 mL/min gradient elution from 90% H₂O/MeCN to 100% MeCN with 0.05% formic acid as modifier over 15 min) to detect the significant transformation of viridicatumtoxin C (3) to *ortho*-quinone viridicatumtoxin C (3a) (90%, *w/w*); UV (MeOH) λ_{max} (log ε) 265 (4.48), 468 (3.65), 591 (3.72) nm; ESI(+)MS *m/z* 578 [M + H]⁺, ESI(-)MS *m/z* 576 [M - H]⁻; HRESI(+)MS *m/z* 600.1481 [M + Na]⁺ (calcd for C₃₀H₂₇NO₁₁Na⁺, 600.1476). To confirm the structure, 3a was treated with 1,2-phenylenediamine (PLDA) and analyzed by

 HPLC-DAD-HRMS (Gemini-NX C₁₈ column, 150 × 2.0 mm, 5 μ m, 250 μ L/min gradient elution from 90% H₂O/MeCN to 100% MeCN (with isocratic 0.05% formic acid as modifier) over 10 min at 35°C, then hold for 5 min and equilibrate for 4 min; 100 μ L/h injection of sodium formate clusters as an internal calibrant) to detect 95% conversion to the adduct **3b**; C₃₆H₃₁N₃O₉ *m/z* 650.2098 [M + H]⁺ (calcd for C₃₆H₃₂N₃O₉⁺, 650.2133) (Figure S3).

Acid Stability of Viridicatumtoxin A (1) and Oxytetracycline (9). Experiment 1: Individual aliquots of 1 (100 µg) and 9 (100 µg) were treated with 0.1 M HCl (200 µL, pH = 1) at 75 °C, with sampling (6 µL) at 1, 2, 3, 4, 7 and 10 h intervals. Individual analyte samples were analysed by HPLC-DAD-ESI(\pm)MS (Zorbax SB-C₈ column, 150 × 4.6 mm, 5 μ m, 1 mL/min gradient elution from 90% H₂O/MeCN to 100% MeCN with 0.05% formic acid as modifier over 15 min) (Figures S6 and S7). Experiment 2: For testing the stability of oxytetracycline (9) in simulated gastric acid condition an aliquot of 9 (100 μ g) was treated with 0.01 M HCl (200 μ L, pH = 2) at 37 °C, with sampling (6 μL) at 1, 2, 4, 6, 10 and 20 h intervals. Individual analyte samples were analysed by HPLC-DAD-ESI(\pm)MS (Zorbax SB-C₈ column, 150 × 4.6 mm, 5 µm, 1 mL/min gradient elution from 90% H₂O/MeCN to 100% MeCN with 0.05% formic acid as modifier over 15 min) (Figure S8). Experiment 3: For testing the stability of viridicatumtoxin A (1) under harsher acidic conditions an aliquot of 1 (0.8 mg) dissolved in 5% (w/v) p-TsOH toluene (200 µL) was treated with molecular sieves to remove residual water. The filtered solution was then heated to 100 °C with sampling (5 µL) at 1, 2, 3, 4, 6, 8, 10, 12, 22, 27, 32 and 48 h intervals. Individual analyte samples were dried, redissolved in MeOH and subjected to HPLC-DAD-ESI(\pm)MS analysis (Zorbax SB-C8 column, 150 × 4.6 mm column, 5 μ m, 1 mL/min gradient elution from 90% H₂O/MeCN to 100% MeCN over 15 min, with constant 0.05% formic acid modifier) (Figure S9).

Purification of Oxytetracycline and Viridicatumtoxin A Acid Degradation Products (11-

15). A sample of **9** (150 mg) was dissolved in 0.1 M HCl (1.5 mL) and stirred at 75 °C for 2 h, after which the solution was dried *in vacuo* to yield a crude product (157.2 mg) that was fractionated by preparative HPLC (Luna C₁₈ column, 250 × 21.2 mm, 10 µm, 20 mL/min gradient elution from 90% to 15% H₂O/MeCN over 20 min, with an isocratic 0.01% TFA modifier) to afford anhydro-oxytetracycline (**10**) ($t_R = 9.9$ min; 8.5 mg, 5.4%), α-apo-oxytetracycline (**11**) ($t_R = 7.5$ min; 36.4 mg, 23.2%), β-apo-oxytetracycline (**12**) ($t_R = 10.9$ min; 12.1 mg, 7.7%) and terrinolide (**13**) ($t_R = 12.8$ min; 52.1 mg, 33.1%). Note: anhydro-oxytetracycline (**10**) could not be obtained in pure form as it was unstable to the HPLC acidified mobile phase (0.01% TFA), undergoing facile conversion to **11–13**.

A sample of **1** (45 mg) dissolved in 5% (*w/v*) *p*-TsOH toluene (3 mL) was treated with molecular sieves to remove residual water, and the filtered reaction mixture was heated at 100 °C for 20 h, after which it was dried *in vacuo* and fractionated by preparative HPLC (Luna C₁₈ column, 250 × 21.2 mm, 10 μ m, 20 mL/min gradient elution from 35% H₂O/MeCN to 100% MeCN over 20 min, with an isocratic 0.01% TFA modifier) to afford **14** (*t*_R = 16.7 min; 0.8 mg, 1.8%) and **15** (*t*_R = 17.7 min; 0.6 mg, 1.3%).

Characterization of Oxytetracycline and Viridicatumtoxin A Acid Degradation Products

(11–15). *Oxytetracycline (9)*. Pale yellow powder; $[\alpha]_D^{22}$ –119.5 (*c* 0.2, MeOH); UV (MeOH) λ_{max} (log ε) 266 (3.99), 360 (3.86) nm; NMR (600 MHz, DMSO-*d*₆) see Table S16; ESI(+)MS *m*/*z* 461 [M + H]⁺, ESI(–)MS *m*/*z* 459 [M – H]⁻; HRESI(+)MS *m*/*z* 461.1560 [M + H]⁺ (calcd for C₂₂H₂₅N₂O₉⁺, 461.1555).

 α -Apo-oxytetracycline (11). Pale yellow powder; $[\alpha]_D^{22}$ +61.6 (c 0.2, MeOH); UV (MeOH)

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 λ_{max} (log ε) 250 (4.53), 332 (3.78), 381 (3.92), 397 (3.93) nm; NMR (600 MHz, DMSO-*d*₆) see Table S17; ESI(+)MS *m/z* 443 [M + H]⁺, ESI(-)MS *m/z* 441 [M - H]⁻; HRESI(+)MS *m/z* 443.1454 [M + H]⁺ (calcd for C₂₂H₂₃N₂O₈⁺, 443.1449).

β-Apo-oxytetracycline (12). Pale yellow powder; $[\alpha]_D^{22}$ –70.0 (*c* 0.2, MeOH); UV (MeOH) λ_{max} (log ε) 244 (4.46), 332 (3.74), 394 (3.85) nm; NMR (600 MHz, DMSO-*d*₆) see Table S18; ESI(+)MS *m/z* 443 [M + H]⁺, ESI(-)MS *m/z* 441 [M – H]⁻; HRESI(+)MS *m/z* 443.1454 [M + H]⁺ (calcd for C₂₂H₂₃N₂O₈⁺, 443.1449).

Terrinolide (13). Pale yellow powder; $[\alpha]_D^{22} - 29.3$ (*c* 0.2, MeOH); UV (MeOH) λ_{max} (log ε) 249 (4.53), 332 (3.91), 380 (3.98), 397 (3.98) nm; NMR (600 MHz, DMSO-*d*₆) see Table S19; ESI(+)MS *m/z* 398 [M + H]⁺, ESI(-)MS *m/z* 396 [M - H]⁻; HRESI(-)MS *m/z* 396.0721 [M - H]⁻ (calcd for C₂₀H₁₄NO₈⁻, 396.0719).

14. Yellow amorphous powder; $[α]_D^{22}$ +108.0 (*c* 0.04, EtOH); UV (MeOH) $λ_{max}$ (log ε) 251 (4.69), 291 (4.16), 303 (4.22), 394 (4.12) nm; ¹H NMR (600 MHz, CDCl₃) $δ_H$ 13.03 (1H, s, 12-OH), 10.96 (1H, s, 10-OH), 7.33 (1H, s, H-5), 7.05 (1H, s, H-14), 6.38 (1H, s, H-9), 6.05 (1H, br s, 12a-OH), 5.73 (1H, m, H-17), 3.87 (3H, s, 8-OCH₃), 3.80 (2H, d, *J* = 3.5, H-4), 3.75 (3H, s, 4-COOCH₃), 2.26 (2H, m, H-18), 1.91 (1H, dt, *J* = 13.2, 6.2 Hz, H-19α), 1.63 (1H, dt, *J* = 13.2, 6.8 Hz, H-19β), 1.11 (3H, d, *J* = 1.5 Hz, H-21), 0.87 (3H, s, H-22), 0.85 (3H, s, H-23); ¹³C NMR (150 MHz, CDCl₃) $δ_C$ 191.1 (C-11), 171.5 (C-3), 163.0 (C-10), 162.0 (C-8), 149.4 (C-12), 149.3 (C-6), 144.3 (C-14), 143.4 (C-12a), 132.8 (C-5a), 131.0 (C-16), 126.2 (C-4a), 124.6 (C-7), 124.5 (C-17), 124.4 (C-6a), 117.5 (C-5), 114.7 (C-11a), 106.2 (C-10a), 97.2 (C-9), 68.3 (C-15), 55.8 (8-OCH₃), 52.5 (4-COO<u>CH₃</u>), 37.0 (C-20), 35.9 (C-19), 35.9 (C-4), 27.0 (C-23), 26.9 (C-22), 23.2 (C-18), 19.7 (C-21); ESI(+)MS *m/z* 477 [M + H]⁺, ESI(-)MS *m/z* 475 [M - H]⁻;

HRESI(+)MS m/z 477.1911 [M + H]⁺ (calcd for C₂₈H₂₉O₇⁺, 477.1908).

15. Yellow amorphous powder; $[α]_D^{22}$ +125.3 (*c* 0.05, EtOH); UV (MeOH) $λ_{max}$ (log ε) 251 (4.74), 292 (4.13), 302 (4.16), 387 (4.15) nm; ¹H NMR (600 MHz, CDCl₃) $δ_H$ 13.11 (1H, s, 12-OH), 10.92 (1H, s, 10-OH), 7.42 (1H, s, H-5), 7.16 (1H, s, H-14), 6.41 (1H, s, H-9), 5.76 (1H, m, H-17), 3.89 (3H, s, 8-OCH₃), 3.86 (2H, br s, H-4), 2.27 (2H, m, H-18), 1.93 (1H, dt, *J* = 13.3, 6.2 Hz, H-19α), 1.62 (1H, dt, *J* = 13.3, 6.7, H-19β), 1.12 (3H, d, *J* = 1.5 Hz, H-21), 0.88 (3H, s, H-22), 0.86 (3H, s, H-23); ¹³C NMR (150 MHz, CDCl₃) $δ_C$ 191.2 (C-11), 172.5 (C-3), 163.2 (C-10), 162.2 (C-8), 148.5 (C-6), 147.5 (C-12), 146.5 (C-14), 142.0 (C-12a), 132.7 (C-5a), 130.5 (C-16), 130.2 (C-4a), 129.6 (C-6a), 124.9 (C-17), 124.5 (C-7), 116.6 (C-11a), 111.0 (C-5), 106.1 (C-10a), 97.5 (C-9), 68.5 (C-15), 55.8 (8-OCH₃), 37.1 (C-20), 35.9 (C-19), 33.9 (C-4), 27.0 (C-23), 26.9 (C-22), 23.1 (C-18), 19.7 (C-21); ESI(+)MS *m/z* 445 [M + H]⁺, ESI(-)MS *m/z* 443 [M – H]⁻; HRESI(-)MS *m/z* 443.1505 [M – H]⁻ (calcd for C₂₇H₂₃O₆⁻, 443.1500).

X-ray Crystallographic Analysis of $1 \cdot 2 \text{MeOH} \cdot 0.5 \text{H}_2 \text{O}$. The single crystal of $1 \cdot 2 \text{MeOH} \cdot 0.5 \text{H}_2 \text{O}$ was obtained from MeOH by slow evaporation at r.t. Data were collected at 190 K using an Oxford Diffraction Gemini CCD diffractometer with Cu Ka radiation and the crystal was cooled with an Oxford Cryosystems Desktop Cooler. Data reduction was performed with the CrysAllisPro program (Oxford Diffraction vers. 171.34.40). The structure was solved by direct methods with SHELXS86 and refined with SHELX97.¹⁵ The thermal ellipsoid diagram was produced with ORTEP3¹⁶ and all calculations were performed within the WinGX package.¹⁷

 $C_{30}H_{31}NO_{10} \bullet 2(CH_3OH) \bullet 0.5H_2O$, M = 638.65, monoclinic, a = 30.1333(7), b = 7.7644(2), c = 12.9297(3) Å, V = 2912.26(12) Å³, T = 190(2) K, space group C2, Z = 4, 16514 reflections measured, 4539 unique ($R_{int} = 0.0413$) which were used in all calculations. The final R(obs. data)

was 0.0466, goodness of fit 1.069. CCDC number 1429664. The absolute structure was confirmed by the methodology of Hooft *et al.*¹⁸

Associated Content

Supporting Information

¹H and ¹³C NMR spectra and tabulated 1D and 2D NMR data for 1–9, 11–15, taxonomy of *Paecilomyces* sp., optical rotation study of 1, acid and heat stability of 1, 3 and 9, HPLC chromatograms, bioassay data and X-ray crystallography of 1. This material is available free of charge via the Internet at <u>http://pubs.acs.org/</u>.

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Notes

The authors declare no competing financial interest.

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