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Design, synthesis and biological evaluations of quaternization harman analogues as potential antibacterial agents

Jiangkun Dai, Wenjia Dan, Siyu Ren, Congguo Shang, Junru Wang

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1 Design, synthesis and biological evaluations of quaternization

2 harman analogues as potential antibacterial agents

- 3 Jiangkun Dai, Wenjia Dan, Siyu Ren, Congguo Shang, and Junru Wang*
- 4 College of Chemistry & Pharmacy, Northwest A&F University, 22 Xinong Road,
- 5 Yangling 712100, Shaanxi, China.

6 *Corresponding author

7 E-mail: wangjunru@nwafu.edu.cn (J.R. Wang)

8 ABSTRACT

9 Thirty-three new quaternization harman analogues were synthesized and their 10 antibacterial activity against four Gram-positive and two Gram-negative bacteria were 11 evaluated. The structure-activity relationships were summarized and compounds 4f, 12 4i, 4l, 4u, 4w, 4x and 5c showed excellent antibacterial activity, low cytotoxicity, good thermal stability and "drug-like" properties. In particular, compound 4x 13 14 exhibited better bactericidal effect (4-fold superiority against methicillin-resistant 15 Staphylococcus aureus) than standard drugs fosfomycin sodium and ampicillin 16 sodium (minimum inhibitory concentration = 50 nmol/mL). Scanning electron 17 microscopy revealed morphological changes of the bacterial cell surface and the 18 docking evaluation provided a good total score (6.4952) for 4x which is close to the 19 score of ciprofloxacin (6.9723). The results indicated that the quaternization harman 20 analogues might exert their bactericidal effect by damaging bacterial cell membrane 21 and wall, and disrupting the function of type II topoisomerase. In addition, the *in vivo* 22 antibacterial assay with a protective efficacy of 81.3% further demonstrated the 23 potential of these derivatives as new bactericides and antibiotics.

- 24 Keywords: harman; quaternization; derivatization synthesis; bactericides; structure-
- 25 activity relationships; molecular docking.

26 1. Introduction

27 Infections caused by bacterial pathogens are a major cause of morbidity and 28 mortality worldwide. Although the successful treatment of such infections by 29 antibiotic drugs is widely regarded as a major medical breakthrough of the 20th 30 century, this achievement may not be sustainable in the future, as bacteria have 31 counteracted antibiotic pressure and developed or acquired resistances that render formerly efficacious drugs inactive [1,2]. In recent years, the emergence of the 32 33 antibiotic crisis has brought great challenges to normal production and life such as 34 healthcare systems, farming, and the food production industry [3–5]. For example, the 35 misuse and overuse of current antibiotics has resulted in a situation that higher dose is 36 required to successfully treat the bacterial infection [6]. Moreover, the intensive 37 antibacterial discovery effort has seen a dramatic decline in the large pharmacy industry in the last two decades [7]. "The cost in terms of lost global production 38 39 between now and 2050 would be an enormous 100 trillion USD if we do not take 40 action," as a UK Government report states [8]. Therefore, we need to develop new 41 drugs to replace the ones that no longer work.

42 Natural products are an important source of antibacterial agents [9]. The seeds of 43 Peganum harmala, containing about 2–6% pharmacologically active alkaloids which 44 are mostly β -carbolines such as harman L1 (Fig. 1), were used as antibacterial drugs 45 many years ago in North Africa and Middle East [10]. Canthin-6-one L4, a subclass 46 of β -carboline alkaloids with an additional D ring, isolated from Allium neapolitanum, 47 showed good antibacterial activity (MIC = $8-64 \mu g/mL$) against S. aureus [11]. 48 Eudistomin U L6, β -carboline with an additional indole ring at position 1, isolated 49 from several species of marine ascidians, also exhibited good antibacterial activity $(IC_{50} = 6.4 \ \mu g/mL)$ [12]. In our previous work, we also reported the antibacterial 50 51 activity of some modified β -carbolines L2–L5 [13,14].

52 With the aim of finding more promising antibacterial harman analogues, we 53 report here a strategy to generate the bioactive compounds. Our idea and design are 54 relatively simple (Fig. 1). Briefly, considering the antibacterial activity data of **L2** and

55 L3, we speculate that the substituents at position 1 are favorable. Simultaneously, the 56 activity data of L1 and L6 indicate that the aromatic substituents are more potent. So we deduce that an additional aromatic substituent at position 1 might improve the 57 bactericidal effect of harman. Our previous work showed that some N^3 -substituted 58 59 canthin-6-ones L5 exhibited approximate 32-fold advantage against S. aureus (MIC = 60 0.98 μ g/mL) over canthin-6-one L4 (MIC = 31.25 μ g/mL) [13]. The result indicates that quaternization derivatives might be promising. During the course of our initial 61 62 efforts to improve the antibacterial activity of harman L1, we confirmed the main 63 skeleton of the leading compound L7 based on the vital factors mentioned. In order to enrich the chemical diversity as well as to further study the structure-activity 64 65 relationships, the quaternization harman analogues incorporating the methoxy group 66 at position 6 were also synthesized. The synthesized compounds were characterized 67 and tested for their in vitro and in vivo antibacterial activity. In addition, their 68 structure-activity relationships, cytotoxicity, thermal stability, "drug-like" properties 69 and preliminary antibacterial mechanism were studied.

70 2. Results and Discussion

71 2.1. Chemistry

72 The initial synthetic aim was to obtain a convenient route for the preparation of 73 quaternization harman analogues. So, it was decided to begin with the preparation of 74 2a-2h (yield: 92%-96%) through the Pictet-Spengler reaction [15]. Compounds 3a-**3h** (yield: 62%–73%) were then easily obtained using a palladium catalyst [16]. 75 76 Subsequently, the quaternization reaction was carried out at room temperature [17]. 77 The synthesis of the target compounds 4a-4x (yield: 67%-96%) is shown in Fig. 2. In 78 order to clarify the influence of different substituted benzyl groups on activity, 79 compounds 5a-5i (yield: 72%-95%) were synthesized as shown in Fig. 3. Nuclear magnetic resonance (NMR) spectroscopy (¹H NMR, ¹³C NMR, DEPT 135, 80 81 heteronuclear multiple-bond correlation, heteronuclear single quantum correlation, and homonuclear correlation spectroscopy) of 4q confirmed that the benzylation 82 position is N^2 (Fig. S1). The successful synthesis of the intermediates **3a–3h** and the 83

quaternization harman derivatives (4a–4x and 5a–5i) was further confirmed by crystal
structures of the compounds 3e and 5f (Fig. 4). Moreover, these two crystal structures
further guided the structural confirmation of all the 41 compounds. The x-ray data
block (bond lengths, bond angles *etc.*) of compounds 3e and 5f has been presented in
Supplementary data.

89 All spectral and analytical data were consistent with the assigned structures. In 90 the NMR spectra of compound **3a**, the signals of the methyl group were detected around $\delta = 3.94$ ppm (¹H NMR) and $\delta = 52.2$ ppm (¹³C NMR). The signal of the 91 92 imino hydrogen atom was observed at $\delta = 8.81$ ppm (¹H NMR) and the signals of the 93 phenyl hydrogen atoms appeared in the aromatic region. Moreover, the signal of the carbonyl carbon atom was detected around $\delta = 166.8$ ppm (¹³C NMR). After 94 quaternization, the signals of the methylene group were detected around $\delta = 5.82$ ppm 95 (¹H NMR) and $\delta = 60.9$ ppm (¹³C NMR) in the NMR spectra of compound 4a, which 96 indicated that the quaternization harman derivatives were successfully synthesized. In 97 addition, the signals of the carbonyl could be found at 1718 and 1712 cm^{-1} in the IR 98 99 spectra of compounds 3a and 4a, respectively.

100 2.2. Antibacterial activity

101 Bacteria could be divided into two categories by Gram staining, Gram-positive 102 and Gram-negative bacteria. S. aureus and MRSA are the leading causes of bacterial 103 infections in humans with symptoms ranging from simple skin infections to severe 104 necrotizing fasciitis and pneumonia [18]. Bacillus cereus and Escherichia coli could 105 cause food poisoning, such as a diarrheal syndrome and an emetic syndrome, both 106 through the production of distinct toxins [19]. Bacillus subtilis and Ralstonia 107 solanacearum are major components of plant pathogens [20]. So four Gram-positive 108 bacteria (S. aureus CGMCC 1.8721, MRSA ATCC 43300, B. cereus CGMCC 1.1846, 109 and B. subtilis 769) and two Gram-negative bacteria (E. coli CGMCC 1.1636 and R. 110 solanacearum CGMCC 1.12711) were selected as the tested bacteria in this work. 111 Compounds (3a–3h, 4a–4x and 5a–5i) were evaluated for their *in vitro* antibacterial 112 activity through double dilution method, with fosfomycin sodium, ampicillin sodium,

113 and cefotaxime sodium as the positive controls.

114 The antibacterial results (Table 1) revealed that most of the quaternization 115 harman analogues displayed good in vitro biological activity against all of the tested 116 bacterial strains except for E. coli. Although most of the intermediates were inactive 117 under the tested concentrations, compound 3h exhibited weak activity (MIC = 200 118 nmol/mL) against MRSA and S. aureus. Compared with the positive control 119 fosfomycin sodium, compounds 4a–4l, 4n–4o and 4q–4x were found to be the most 120 potent compounds against S. aureus, with peak MICs of lower than 25 nmol/mL. Six 121 compounds (4f, 4i, 4l, 4u, 4w and 4x) displayed equal or superior activity against 122 MRSA compared with the positive controls. Fifteen compounds exhibited better 123 activity against *B. cereus* than ampicillin sodium. Specifically, the MIC of 4f and 4x 124 (25 nmol/mL) was equal to that of the positive controls fosfomycin sodium and 125 cefotaxime sodium. Nine compounds (4f, 4h, 4i, 4k, 4l, 4t, 4u, 4w and 4x) showed a 126 potential MIC (25 nmol/mL) against B. subtilis. As shown in Table 1, eight 127 compounds (4f, 4h, 4i, 4k, 4l, 4u, 4w and 4x) exhibited better activity (MIC = 25128 nmol/mL) against R. solanacearum than fosfomycin sodium (MIC = 50 nmol/mL). 129 The activity of different substituted benzyl derivatives was tested (Table 2), and 130 compound 5c was considered to be the compound with the most potential. 131 Interestingly, other quaternization harman analogues also showed similar effects on 132 different bacteria. For example, the MIC of the compound 133 2,6-dimethyl-9*H*-pyrido[3,4-*b*]indol-2-ium iodide against MRSA was 4 µg/mL [21].

134 2.3. Structure–activity relationships

Based on the antibacterial activity data, the structure–activity relationships were carefully investigated (Fig. 5). The derivatives containing a methoxy group at position 6 were practically the same or higher than that derivatives without this group, with the exception of pyridine. For example, the MIC of compound **4f** was 25 nmol/mL against *S. aureus*, which was better than **4c** (MIC = 50 nmol/mL). Compared with harman **L1** (MIC = 1000 μ g/mL, \approx 5494 nmol/mL), compound **3h** displayed better activity (MIC = 200 nmol/mL) against *S. aureus*, which indicates that an aromatic

142 group at position 1 is effective. Comprehensive analysis indicated that the 143 benzo[d][1,3]dioxole group is the most effective substituent at position 1, and the 144 furan group is better than the methyl benzoate group. For example, the MIC against 145 MRSA of compound 4x was 12.5 nmol/mL, which was better than compounds 4f (25 146 nmol/mL), 41 (25 nmol/mL) and 4r (100 nmol/mL). Compared with compound 4c 147 (MIC = 50 nmol/mL against R. solanacearum), with a methyl benzoate group substituent at position 1, the MIC of compound 4i (furan group at position 1, MIC = 148 149 25 nmol/mL against R. solanacearum) is better.

150 Quaternization is clearly beneficial for improving antibacterial potency. Various 151 substituted benzyl groups have a significant effect on the activity. The influence of different groups is easily seen in Table 1 and Table 2: $CF_3 > Br > Cl \approx CH_3 > F \approx NO_2$. 152 153 For example, the MIC of compound 4f was 25 nmol/mL against all the tested bacteria 154 except for E. coli. In addition, the MICs of compounds 5c (25 nmol/mL against S. aureus), **5f** (75 nmol/mL against S. aureus) and **5g** (50 nmol/mL against S. aureus) 155 156 indicate the activity sequence for substituents at different positions: para > meta > 157 ortho. Interestingly, similar findings have been reported by others, which supported 158 our SAR results (para substituents are better) and confirmed the advantages of 159 trifluoromethyl group [22].

160 2.4. Cell toxicity and molecular physicochemical properties

161 Many natural and synthetic agents with the ability to interact with DNA often 162 have relatively low therapeutic index, most likely due to the unspecific manner of 163 these agents in causing DNA damage both in lesions and in highly proliferative 164 normal tissues. Similarly, the frequency of utilization of β -carbolines as therapeutic 165 agents is greatly reduced because of their cytotoxic activity toward mammalian cells 166 (low selectivity) [23]. We determined the cytotoxic effects of the highly bioactive quaternization harman derivatives 4f, 4i, 4l, 4u, 4w, 4x and 5c on MARC 145 cells 167 168 (normal monkey kidney cells) and L02 cells (normal human hepatocyte) [24]. 169 Simultaneously, the activity under mass concentration was also tested. As shown in 170 Table 3, the cell viabilities of all the tested compounds were more than 87% on

MARC 145 cells and more than 71% on human hepatocyte L02 cells. The results
suggest that these molecules are of low toxicity. Interestingly, compound 4x exhibited
excellent viability (103%) on MARC 145 cells under an effective antibacterial dose (8
µg/mL).

175 In 1997, Lipinski et al. published what is widely regarded as the key paper 176 defining physicochemical and structural properties profiles for the optimal oral 177 availability of drugs [25]. In 2003, Clarke and Delaney reported that most herbicides 178 and fungicides also adhere to the Lipinski rule [26]. As we know, for good oral 179 bioavailability an ideal molecule should have good intestinal absorption and reduced 180 molecular flexibility [27]. The molecular properties of the preferred ionic derivatives 181 are shown in Table 4. For all the preferred compounds, the log P values were lower 182 than 5, the number of hydrogen bond acceptors were lower than 10, the number of 183 rotatable bonds were lower than 10, and the number of hydrogen bond donors were 184 lower than 5. Their physicochemical data therefore conform to the Lipinski rule 185 without considering the molecular weight, which demonstrate they might have good 186 "drug-like" properties. Further, the topological polar surface area (TPSA), a 187 parameter defined as the surface sum over all polar atoms and used extensively in 188 medicinal chemistry to predict absorption and optimize a compound's membrane 189 permeability, was calculated for the potent compounds. It is suggested that compounds with a TPSA greater than 140 Å^2 tend to be poor at permeating cell 190 191 membranes [27]. Based on the predicted data mentioned above, the calculated TPSA values for the highly bioactive quaternization harman derivatives would indicate a 192 193 possible good bioavailability.

194 2.5. Thermal stability of compound **4x** in vitro [28,29]

195 Considering the poor thermal stability of antibiotics, it was decided to further 196 investigate the bioactive compound **4x** using an *in vitro* assay, to consider whether its 197 thermal stability had improved compared with the control. As depicted in Fig. S2, 198 ampicillin sodium began to degrade after being kept at 37 °C for 6 h. After being kept 199 at 65 °C for 6 h, very little ampicillin sodium residue could be found. Compound **4x**

200 did not degrade any further under the same conditions.

201 2.6. Scanning electron microscopy analysis

202 SEM of MRSA and R. solanacearum revealed morphological changes in the 203 bacterial cell surface (Fig. 6). The surfaces of cells in the untreated group (Fig. 6A and 204 6C) were relatively smooth and regular, whereas when treated with compound 4x (Fig. 205 6B and 6D) there was shrinkage. Increased permeabilization of the membrane may 206 explain the leakage of cytoplasmic material. Additionally, bacteria exposed to the 207 compound 4x experienced cell wall disintegration. These results indicated that 208 compound 4x may exert its bactericidal effect by damaging bacterial cell membrane 209 and wall.

210 2.7. Molecular docking study

211 The Surflex-Dock scoring function is a weighted sum of non-linear functions based on the binding affinities of protein-ligand complexes coupled with their 212 213 crystallographically determined structures. Surflex-Dock scores are expressed in 214 $-\log_{10}(K_d)$ units to represent binding affinities [30,31]. Type II topoisomerase present 215 in bacteria is attractive target for antibacterial drug discovery, which is independently 216 essential for bacterial DNA replication [32]. Mechanism of topoisomerases inhibition 217 is known to occur in two ways: these inhibitors may bind with topoisomerase directly 218 or they may bind to DNA and alter its structure, so that it cannot be recognized by 219 topoisomerases [33]. A previous study demonstrated that large planar surface areas 220 could function at the DNA level via intercalation between base pairs. Further, it was 221 shown that DNA intercalation of molecules could disrupt the function of 222 topoisomerases, ultimately leading to cell death [18]. For quaternization harman 223 analogues, β -carboline core per se is a planar molecule. In some extent, their 224 structures are also similar to the commercial antibiotic ciprofloxacin which targets to 225 the type II topoisomerase. So we selected a crystal of bacterial type II topoisomerase 226 complex (PDB ID: 5IWM) as a possible target.

227

A docking investigation based on the Surflex-Dock program in the Sybyl-X 2.0

228 package was undertaken to explore the possible mechanism. Firstly, we performed a 229 re-docking of the extracted co-crystallized ligand present in the type II topoisomerase complex. The RMSD (root-mean-square deviation) value was 0.923Å (< 2.0 Å), 230 which indicated the docking protocol was feasible [34, 35]. The highly active 231 232 compound 4x gave a good total score (6.4952) which was close to the score of the 233 standard compound ciprofloxacin (6.9723). Interestingly, the MIC value of 234 ciprofloxacin against S. aureus (4 µg/mL) further supported the rationality of the 235 docking protocol. As shown in Fig. 7, compound 4x mainly bound to DNA area. 236 There are many $\pi - \pi$ stacked interactions between the plane structure of the small 237 molecule and the DNA base (DA10, DA11, DT10, and DT11). The hydrogen atom of 238 the methoxy group is adjacent to the DNA base (DC12), forming two carbonhydrogen bonds (2.65 and 2.69 Å). The trifluoromethyl fragment of compound 4x is 239 240 adjacent to the ASP83 residue of the target, forming a strong halogen bond with a 241 length of 2.73 Å. Simultaneously, two carbon-hydrogen bonds (2.32 and 2.50 Å) are 242 formed between the trifluoromethyl group and the SER84 residue. Interestingly, the interaction of the trifluoromethyl group confirms the activity and structure-activity 243 244 relationship finding that trifluoromethyl fragments are beneficial for improving 245 antibacterial activity.

As we all known, there are two types of topoisomerase, namely, type I 246 247 topoisomerase and type II topoisomerase [36]. In order to check the selectivity of the 248 titled compounds against type II topoisomerase over type I topoisomerase, a docking 249 investigation between compound 4x and type I topoisomerase complex (PDB ID: 250 4RUL) was also undertaken. The total score was 4.0326 which indicated that 251 compound 4x showed approximately 290-fold selectivity against type II 252 topoisomerase over type I topoisomerase (Total score = $-\log_{10}(K_d)$, $K_a \times K_d = 1$, K_d 253 represent dissociation constant, K_a represent binding constant, selectivity = K_a (type II 254 topoisomerase) / K_a (type I topoisomerase)).

255 Considering with the SEM results, we deduced that compound **4x** might attack 256 and cause damage to the bacterial cell membrane and wall. The consequent increased 257 permeability of the bacterial cell membrane will then allow compound **4x** to enter the

cells. The 4x in the cytoplasm will then disrupt the function of type II topoisomerase,which in turn caused cell death.

260 2.8. Protective effect in vivo[37,38]

The *in vivo* antibacterial activity (protective effect) of the highly bioactive compound **4x** was tested against *R. solanacearum* on eggplant leaf (Fig. 8). Compound **4x** was found to have a good preventative effect against *R. solanacearum*, with a protective efficacy of 81.3%. This demonstrates the potential of the quaternization harman derivatives as new bactericides.

266 **3.** Conclusions

267 In summary, 41 compounds, including 33 new quaternization harman derivatives, 268 were prepared via a convenient synthetic route and their antibacterial activity was 269 evaluated. The structure-activity relationships were summarized which provided 270 some important guidance for the development of antibacterial harman agents. The 271 highly bioactive compounds 4f, 4i, 4l, 4u, 4w, 4x and 5c, with a peak MIC of 4 272 µg/mL, showed low cytotoxicity, good thermal stability and "drug-like" properties. 273 Further, the SEM analysis and molecular docking evaluations suggested multiple 274 possible antibacterial targets of quaternization harman analogues, namely membrane, 275 wall and type II topoisomerase. As expected, the *in vivo* antibacterial assay 276 demonstrated the potential of the quaternization harman agents as new bactericides 277 and antibiotics. Overall, this work enriched the types of candidate antibiotics and 278 provided more options for solving the current antibiotic crisis.

279 Chirality of drugs is an eternal topic in medicinal chemistry. For some highly 280 active quaternization harman molecules, chiral axis is existed at position 1. In order to 281 separate the isomers, we have used twelve different separation conditions 282 (Supplementary data). Regrettably, we did not get the isomers. Here, we are to appeal 283 to researchers to further study this issue to promote the development and application 284 of quaternization harman antibiotics.

285 **4. Experimental**

11

286 4.1. Materials

¹H NMR and ¹³C NMR spectroscopy was carried out using an Avance 287 spectrometer (Bruker, Billerica, MA, USA) at 500 and 125 MHz. Chemical shifts 288 were measured relative to the residual solvent peaks of CD₃OD (¹H, $\delta = 3.31$ ppm; 289 ¹³C, $\delta = 49.00$ ppm) or dimethyl sulfoxide (DMSO- d_6) (¹H, $\delta = 2.50$ ppm; ¹³C, $\delta =$ 290 291 39.52 ppm) with tetramethylsilane as the internal standard. High-resolution mass 292 spectroscopy (HRMS) was undertaken using an AB SCIEX Triple TOF 5600⁺ 293 spectrometer. Liquid chromatography-electrospray ionization-tandem mass 294 spectrometry (LC-ESI-MS/MS) accomplished analysis using was 295 ultra-high-performance liquid chromatography (UHPLC) (Nexera UHPLC LC-30A) coupled to an AB SCIEX Triple TOF 5600⁺ spectrometer, with the system equipped 296 297 with a C18 trap column (Shim-pack XR-ODS, 2.0 mm i.d. \times 100 mm). Elemental 298 analyses were performed on a Heraeus CHN-O-Rapid instrument. Compounds 3e and 299 5f were collected at 100 K on a Rigaku Oxford Diffraction Supernova Dual Source, 300 Cu at Zero equipped with an AtlasS2 CCD using Cu Kα radiation. Data reduction was 301 carried out with the diffractometer's software Scanning electron microscopy (SEM) 302 analysis was carried out on Nova nano SEM450 instrument. Molecular docking 303 evaluation was conducted with the Sybyl-X 2.0 software and Discovery Studio 2017 304 client. Fourier transform infrared (FT-IR) spectra were obtained on a TENSOR37 305 spectrometer (Bruker) using KBr pellets, and the absorptions are reported in cm^{-1} . 306 The reaction progress was monitored by thin-layer chromatography (TLC) on silica gel GF254 (Qingdao Haiyang Chemical Co., Ltd., Qingdao, China) with ultraviolet 307 308 detection. Reagents were purchased from commercial sources (Bodi Chemical Co., 309 Ltd., Tianjin, China; Aladdin Industrial Co., Ltd., Shanghai, China), and used as 310 received.

311 4.2. Synthesis

312 4.2.1. Synthesis of intermediates 3a–3h

A solution of tryptamine (1.6 g, 10 mmol) or 5-methoxytryptamine (1.9 g, 10

314 mmol), aldehyde (methyl 4-formylbenzoate, furfural, pyridine-2-carboxaldehyde, or 315 piperonyl aldehyde) (11 mmol), and trifluoroacetic acid (100 µL) in dichloromethane 316 (50 mL) was stirred at room temperature for 36 h. Dichloromethane (30 mL) and 317 ammonia water (5%, 30 mL) were added, and the organic layer was separated. Next, 318 the water phase was extracted with dichloromethane, and the organic phases were 319 combined and dried over sodium sulfate and filtered. The solvent was then evaporated, 320 and the crude product purified by column chromatography (ethyl acetate as the eluent) 321 to give the products 2a-2h in 92-96% yield. Pd/C (10%, 0.1 eq) was then added to 322 the solution of 2a-2h in xylene (50 mL), and the solution was heated at 145 °C and 323 monitored by TLC. On completion, the resulting mixture was filtered and evaporated. 324 The residue was purified by silica gel column chromatography with petroleum ether 325 and ethyl acetate (v/v = 2:1) to produce **3a–3h** in 62–73% yield.

326 4.2.1.1. methyl 4-(9H-pyrido[3,4-b]indol-1-yl)benzoate (3a): white solid; yield, 65%; m.p. 147.4–148.3 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.81 (s, 1H), 8.58 (d, J = 2.5 Hz, 327 1H), 8.19–8.17 (m, 3H), 8.02 (d, J = 5 Hz, 2H), 7.97 (d, J = 5 Hz, 1H), 7.59–7.53 (m, 328 2H), 7.33 (t, J = 7.5 Hz, 1H), 3.94 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 166.8, 329 142.9, 142.8, 141.5, 140.5, 139.6, 133.6, 130.4, 130.2, 128.8, 128.1, 121.9, 121.8, 330 120.5, 114.4, 111.6, 52.2. IR (KBr, cm⁻¹) 3362, 3065, 2957, 1718, 1637, 1619, 1582, 331 1526, 1497, 1469, 1438, 1326, 1294, 1279, 1223, 1164, 1128, 1067, 1026, 1015, 825, 332 333 792, 769, 726. Elemental anal. calcd for $C_{19}H_{14}N_2O_2$: C, 75.48; H, 4.67; N, 9.27; 334 found C, 75.44; H, 4.69; N, 9.31. HRMS (ESI) m/z calcd for C₁₉H₁₄N₂O₂ [M+H]⁺ 303.1128, found 303.1127. 335

336 4.2.1.2. methyl 4-(6-methoxy-9H-pyrido[3,4-b]indol-1-yl)benzoate (**3b**): yellow solid; 337 yield, 62%; m.p. 124.1–125.3 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.84 (s, 1H), 8.57 (d, 338 J = 5 Hz, 1H), 8.19 (d, J = 10 Hz, 2H), 8.05 (d, J = 10 Hz, 2H), 7.96 (d, J = 2.5 Hz, 339 1H), 7.63 (s, 1H), 7.48 (d, J = 10 Hz, 1H), 7.26 (d, J = 10 Hz, 1H), 3.99 (s, 3H), 3.97 340 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 166.8, 154.5, 142.9, 141.7, 139.1, 135.5, 341 134.3, 130.3, 130.2, 130.1, 128.1, 122.1, 118.8, 114.3, 112.5, 103.6, 56.0, 52.2. IR 342 (KBr, cm⁻¹) 3064, 2998, 2951, 2832, 1722, 1609, 1584, 1495, 1468, 1436, 1400, 1286, 343 1218, 1177, 1132, 1116, 1037, 1014, 818, 768. Elemental anal. calcd for C₂₀H₁₆N₂O₃:

344 C, 72.28; H, 4.85; N, 8.43; found C, 72.26; H, 4.83; N, 8.47. HRMS (ESI) m/z calcd

345 for $C_{20}H_{16}N_2O_3$ [M+H]⁺ 333.1234, found 333.1230.

346 4.2.1.3. 1-(furan-2-yl)-9H-pyrido[3,4-b]indole (3c): brown solid; yield, 66%; m.p. 127.1–128.6 °C; ¹H NMR (500 MHz, CDCl₃) δ 9.54 (s, 1H), 8.50 (d, J = 2.5 Hz, 1H), 347 8.15 (d, J = 5 Hz, 1H), 7.89 (d, J = 2.5 Hz, 1H), 7.71 (s, 1H), 7.60 (s, 2H), 7.35 (s, 348 2H), 6.67 (s, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 154.4, 142.8, 140.5, 138.8, 133.4, 349 350 131.3, 130.3, 128.6, 121.7, 121.2, 120.1, 113.7, 112.3, 111.6, 108.8. IR (KBr, cm⁻¹) 351 3458, 1625, 1556, 1493, 1452, 1425, 1378, 1366, 1318, 1301, 1284, 1252, 1235, 1210, 1164, 998, 749, 458. Elemental anal. calcd for C₁₅H₁₀N₂O: C, 76.91; H, 4.30; N, 352 11.96; found C, 76.87; H, 4.35; N, 11.94. HRMS (ESI) m/z calcd for C₁₅H₁₀N₂O 353 354 $[M+H]^+$ 235.0866, found 235.0863.

4.2.1.4. 1-(furan-2-yl)-6-methoxy-9H-pyrido[3,4-b]indole (3d): brown solid; yield, 355 68%; m.p. 66.4–67.8 °C; ¹H NMR (500 MHz, CDCl₃) δ 9.30 (s, 1H), 8.41 (d, J = 5 356 Hz, 1H), 7.80 (d, J = 5 Hz, 1H), 7.67 (d, J = 2.5 Hz, 1H), 7.53 (d, J = 5 Hz, 1H), 7.46 357 358 (d, J = 10 Hz, 1H), 7.29 (d, J = 2.5 Hz, 1H), 7.22-7.20 (m, 1H), 6.64-6.63 (m, 1H), 7.22-7.20 (m, 100), 100 (m, 100), 100), 100 (m, 100), 100 (m, 100), 100), 100 (m, 100), 1003.93 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 154.5, 154.2, 142.7, 138.3, 135.3, 133.6, 359 360 131.9, 130.0, 121.6, 118.6, 113.5, 112.4, 112.3, 108.7, 103.4, 56.0. IR (KBr, cm⁻¹) 361 2923, 1606, 1585, 1556, 1489, 1462, 1437, 1379, 1287, 1253, 1215, 1163, 1125, 1020, 362 810, 772, 738, 622. Elemental anal. calcd for C₁₆H₁₂N₂O₂: C, 72.72; H, 4.58; N, 10.60; found C, 72.68; H, 4.60; N, 10.58. HRMS (ESI) m/z calcd for C₁₆H₁₂N₂O₂ [M+H]⁺ 363 265.0972, found 265.0970. 364

365 *4.2.1.5. I-(pyridin-2-yl)-9H-pyrido[3,4-b]indole (3e)*: white solid; yield, 70%; m.p. 366 139.4–141.2 °C; ¹H NMR (500 MHz, CDCl₃) δ 11.27 (s, 1H), 8.71–8.69 (m, 2H), 367 8.50 (d, *J* = 5 Hz, 1H), 8.09 (d, *J* = 5 Hz, 1H), 7.93 (d, *J* = 5 Hz, 1H), 7.83–7.80 (m, 368 1H), 7.56–7.50 (m, 2H), 7.24–7.22 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 158.0, 369 148.2, 140.6, 138.1, 138.1, 136.8, 134.8, 130.5, 128.4, 122.9, 121.7, 121.3, 121.1, 370 119.8, 115.4, 111.9. IR (KBr, cm⁻¹) 3395, 3291, 3057, 1625, 1588, 1564, 1484, 1446, 371 1414, 1362, 1316, 1280, 1234, 1146, 1091, 1065, 798, 743, 622, 597. Elemental anal.

- 372 calcd for C₁₆H₁₁N₃: C, 78.35; H, 4.52; N, 17.13; found C, 78.33; H, 4.55; N, 17.10.
- 373 HRMS (ESI) m/z calcd for $C_{16}H_{11}N_3$ [M+H]⁺ 246.1026, found 246.1025.

374 4.2.1.6. 6-methoxy-1-(pyridin-2-yl)-9H-pyrido[3,4-b]indole (3f): white solid; yield, 375 73%; m.p. 131.4–132.2 °C; ¹H NMR (500 MHz, CDCl₃) δ 11.09 (s, 1H), 8.67 (d, J = 10 Hz, 1H), 8.64 (d, J = 5 Hz, 1H), 8.44 (d, J = 5 Hz, 1H), 7.84 (d, J = 5 Hz, 1H), 376 377 7.80–7.76 (m, 1H), 7.48 (d, J = 2.5 Hz, 1H), 7.40 (d, J = 10 Hz, 1H), 7.23–7.19 (m, 1H), 7.16–7.14 (m, 1H), 3.86 (s, 3H); 13 C NMR (125 MHz, CDCl₃) δ 157.9, 154.0, 378 379 148.1, 138.1, 137.5, 136.7, 135.6, 135.3, 130.2, 122.8, 121.4, 121.2, 118.3, 115.3, 380 112.6, 103.5, 55.9. IR (KBr, cm⁻¹) 3347, 1583, 1564, 1490, 1455, 1439, 1414, 1362, 1292, 1280, 1221, 1176, 1139, 1118, 1033, 1020, 837, 823, 798, 742, 630. Elemental 381 382 anal. calcd for C₁₇H₁₃N₃O: C, 74.17; H, 4.76; N, 15.26; found C, 74.15; H, 4.79; N, 15.30. HRMS (ESI) m/z calcd for $C_{17}H_{13}N_3O [M+H]^+ 276.1131$, found 276.1128. 383 4.2.1.7. 1-(benzo[d][1,3]dioxol-5-yl)-9H-pyrido[3,4-b]indole (3g): white solid; yield, 384 66%; m.p. 167.5–168.9 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.63 (s, 1H), 8.57 (d, J = 5385 Hz, 1H), 8.20 (d, J = 10 Hz, 1H), 7.95 (d, J = 5 Hz, 1H), 7.60 (t, J = 7.5 Hz, 1H), 7.55 386

- 387 (d, J = 10 Hz, 1H), 7.51–7.49 (m, 2H), 7.36 (t, J = 7.5 Hz, 1H), 7.05–7.03 (m, 1H), 388 6.09 (s, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 148.5, 148.2, 142.6, 140.3, 139.5, 133.31, 389 132.8, 129.8, 128.5, 122.0, 121.8, 120.3, 113.5, 111.5, 108.8, 108.7, 103.1, 101.4. IR 390 (KBr, cm⁻¹) 3054, 2884, 1626, 1565, 1502, 1471, 1452, 1415, 1320, 1283, 1243, 1223, 391 1126, 1039, 935, 919, 825, 812, 747. Elemental anal. calcd for C₁₈H₁₂N₂O₂: C, 74.99; 392 H, 4.20; N, 9.72; found C, 74.95; H, 4.18; N, 9.76. HRMS (ESI) m/z calcd for
- 393 $C_{18}H_{12}N_2O_2 [M+H]^+$ 289.0972, found 289.0970.
- 394 *4.2.1.8. I-(benzo[d][1,3]dioxol-5-yl)-6-methoxy-9H-pyrido[3,4-b]indole (3h)*: yellow 395 solid; yield, 72%; m.p. 96.5–97.2 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.91 (s, 1H), 8.44 396 (d, *J* = 5 Hz, 1H), 7.82 (d, *J* = 5 Hz, 1H), 7.54 (d, *J* = 5 Hz, 1H), 7.40–7.39 (m, 2H), 397 7.35 (d, *J* = 10 Hz, 1H), 7.17–7.16 (m, 1H), 6.87–6.86 (m, 1H), 5.94 (s, 2H), 3.92 (s, 398 3H); ¹³C NMR (125 MHz, CDCl₃) δ 154.3, 148.4, 148.1, 142.9, 138.7, 135.5, 134.1, 399 132.7, 129.7, 122.3, 121.9, 118.5, 113.4, 112.5, 108.7,108.6, 103.5, 101.3, 56.0. IR 400 (KBr, cm⁻¹) 1565, 1501, 1470, 1447, 1285, 1248, 1209, 1158, 1113, 1038, 814.

401 Elemental anal. calcd for $C_{19}H_{14}N_2O_3$: C, 71.69; H, 4.43; N, 8.80; found C, 71.67; H, 402 4.41; N, 8.85. HRMS (ESI) m/z calcd for $C_{19}H_{14}N_2O_3$ [M+H]⁺ 319.1077, found 403 319.1074.

404 4.2.2. Synthesis of 4a-4x and 5a-5i

405 A solution of 3a-3h (1 mmol) and substituted benzyl bromide (5 mmol) in 406 acetonitrile (15 mL) was stirred at room temperature. The reaction was monitored by 407 TLC. On completion, the resulting mixture was directly evaporated, and the residue 408 was purified by silica gel column chromatography with dichloromethane and 409 methanol (v/v = 20:1) to produce **4a-4h** and **5a-5i** in 67–96% yield.

410 4.2.2.1. 2-benzyl-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido[3,4-b]indol-2-ium

411 bromide (4a): yellow solid; yield, 87%; m.p. 158.8–159.6 °C; ¹H NMR (500 MHz,

- 412 CD₃OD) δ 8.82 (dd, J = 15 HZ, 5 Hz, 2H), 8.49 (d, J = 10 Hz, 1H), 8.27 (d, J = 10 Hz,
- 413 2H), 7.82–7.78 (m, 1H), 7.69–7.65 (m, 3H), 7.52–7.49 (m, 1H), 7.32–7.27 (m, 3H),
- 414 6.95–6.93 (m, 2H), 5.82 (s, 2H), 4.00 (s, 3H); ¹³C NMR (125 MHz, CD₃OD) δ 165.9,
- 415 145.2, 139.2, 135.8, 135.6, 134.4, 134.1, 132.8, 132.5, 132.4, 130.2, 130.0, 128.8,
- 416 128.6, 127.0, 123.1, 122.1, 119.8, 117.0, 112.6, 60.9, 51.7. IR (KBr, cm⁻¹) 3397, 3058,
- 417 2948, 1712, 1629, 1580, 1528, 1496, 1454, 1434, 1403, 1341, 1329, 1281, 1107, 1014,
- 418 767, 757, 705. Elemental anal. calcd for $C_{26}H_{21}BrN_2O_2$: C, 65.97; H, 4.47; N, 5.92;
- 419 found C, 65.94; H, 4.51; N, 5.95. HRMS (ESI) m/z calcd for $C_{26}H_{21}BrN_2O_2$ [M–Br]⁺
- 420 393.1598, found 393.1592.

4.2.2.2. 1-(4-(methoxycarbonyl)phenyl)-2-(4-methylbenzyl)-9H-pyrido[3,4-b]indol-2-421 *ium bromide* (4b): yellow solid; yield, 96%; m.p. 156.7–157.9 °C; ¹H NMR (500 MHz, 422 CD₃OD) δ 8.82 (dd, J = 15 H_Z, 5 Hz, 2H), 8.47 (d, J = 10 Hz, 1H), 8.26 (d, J = 10 Hz, 423 424 2H), 7.81–7.77 (m, 1H), 7.67–7.63 (m, 3H), 7.51–7.48 (m, 1H), 7.31–7.27 (m, 2H), 6.95–6.93 (m, 2H), 5.82 (s, 2H), 4.00 (s, 3H), 2.27 (s, 3H); ¹³C NMR (125 MHz, 425 CD₃OD) δ 165.5, 143.3, 139.2, 135.7, 135.6, 134.4, 134.1, 132.6, 132.3, 132.1, 130.0, 426 129.7, 127.6, 125.9, 123.0, 121.7, 119.6, 118.8, 116.9, 112.6, 60.9, 50.9, 22.3. IR 427 (KBr, cm⁻¹) 3401, 3016, 2950, 1720, 1605, 1587, 1531, 1505, 1464, 1442, 1401, 1327, 428

- 429 1275, 1108, 1010, 769, 756. Elemental anal. calcd for C₂₇H₂₃BrN₂O₂: C, 66.54; H,
- 430 4.76; N, 5.75; found C, 66.52; H, 4.77; N, 5.78. HRMS (ESI) m/z calcd for
- 431 $C_{27}H_{23}BrN_2O_2 [M-Br]^+ 407.1754$, found 407.1754.
- $432 \qquad 4.2.2.3. \ 1-(4-(methoxycarbonyl)phenyl)-2-(4-(trifluoromethyl)benzyl)-9H-pyrido[3,4-2.2.3]{}$
- 433 *b]indol-2-ium bromide* (**4***c*): yellow solid; yield, 92%; m.p. 159.2–160.3 °C; ¹H NMR 434 (500 MHz, CD₃OD) δ 8.93 (d, J = 5 Hz, 1H), 8.87 (d, J = 10 Hz, 1H), 8.51 (d, J = 5435 Hz, 1H), 8.28 (d, J = 10 Hz, 2H), 7.82 (t, J = 5 Hz, 1H), 7.76–7.74 (m, 2H), 7.69 (d, J
- 436 = 10 Hz, 1H), 7.62 (d, J = 10 Hz, 2H), 7.52 (t, J = 7.5 Hz, 1H), 7.18 (d, J = 10 Hz, 437 2H), 5.99 (s, 2H), 4.02 (s, 3H); ¹³C NMR (125 MHz, CD₃OD) δ 165.9, 145.3, 139.2,
- 438 138.8, 135.7, 134.4, 134.4, 133.0, 132.5, 132.4, 130.5, 130.3, 130.0, 127.7, 125.7, 439 125.6, 125.6, 125.6, 124.9, 123.2, 122.8, 122.1, 119.8, 117.3, 112.7, 60.3, 51.8. IR 440 (KBr, cm⁻¹) 3347, 2360, 1581, 1565, 1490, 1455, 1439, 1414, 1362, 1291, 1280, 1221, 441 1176, 1139, 1119, 1090, 1034, 1020, 837, 824, 799, 768, 742, 701, 630, 543. 442 Elemental anal. calcd for $C_{27}H_{20}BrF_3N_2O_2$: C, 59.90; H, 3.72; N, 5.17; found C, 59.87; 443 H, 3.76; N, 5.12. HRMS (ESI) m/z calcd for $C_{27}H_{20}BrF_3N_2O_2$ [M–Br]⁺ 461.1471, 444 found 461.1469.
- 4.2.2.4. 2-benzyl-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido[3,4-b]indol-2 445 *-ium bromide* (4d): yellow solid; yield, 89%; m.p. 150.5–152.0 °C; ¹H NMR (500 446 MHz, DMSO-*d*6) δ 12.15 (s, 1H), 8.94 (t, *J* = 5 Hz, 2H), 8.18–8.16 (m, 2H), 8.11 (d, *J* 447 448 = 5 Hz, 1H), 7.78 (d, J = 5 Hz, 2H), 7.59 (d, J = 10 Hz, 1H), 7.45–7.43 (m, 1H), 7.31– 7.25 (m, 3H), 6.93 (d, \overline{J} = 5 Hz, 2H), 5.80 (s, 2H), 3.95 (s, 3H), 3.92 (s, 3H); ¹³C 449 450 NMR (125 MHz, DMSO-*d*6) δ 166.1, 155.3, 140.6, 139.6, 136.0, 135.3, 134.1, 133.1, 451 133.0, 132.4, 131.0, 130.3, 129.2, 128.9, 127.6, 124.0, 120.5, 118.1, 114.6, 104.2, 60.5, 56.3, 53.1. IR (KBr, cm⁻¹) 3401, 1720, 1612, 1578, 1499, 1437, 1278, 1222, 452 1113, 1024, 821, 767, 738, 704. Elemental anal. calcd for C₂₇H₂₃BrN₂O₃: C, 64.42; H, 453 454 4.61; N, 5.57; found C, 64.40; H, 4.64; N, 5.56. HRMS (ESI) m/z calcd for $C_{27}H_{23}BrN_2O_3 [M-Br]^+ 423.1703$, found 423.1698. 455
- 456 4.2.2.5. 6-methoxy-1-(4-(methoxycarbonyl)phenyl)-2-(4-methylbenzyl)-9H-pyrido[3,4
- 457 *-b]indol-2-ium bromide* (**4***e*): yellow solid; yield, 96%; m.p. 147.2–148.1 °C; ¹H NMR

458	(500 MHz, CD ₃ OD) δ 8.74 (dd, $J = 10$ Hz, 5 Hz, 2H), 8.27 (d, $J = 10$ Hz, 2H), 7.93 (d
459	<i>J</i> = 2.5 Hz, 1H), 7.68 (d, <i>J</i> = 10 Hz, 2H), 7.56 (d, <i>J</i> = 10 Hz, 1H), 7.45–7.42 (m, 1H),
460	7.11 (d, <i>J</i> = 5 Hz, 2H), 6.83 (d, <i>J</i> = 5 Hz, 2H), 5.74 (s, 2H), 4.01 (s, 3H), 3.97 (s, 3H),
461	2.30 (s, 3H); ¹³ C NMR (125 MHz, CD ₃ OD) δ 166.0, 155.8, 140.6, 139.3, 138.8, 135.7
462	133.4, 133.1, 132.8, 132.6, 131.4, 130.2, 129.9, 129.4, 127.0, 124.1, 120.2, 117.0,
463	113.6, 102.6, 60.6, 55.0, 51.7, 19.7. IR (KBr, cm ⁻¹) 3407, 3016, 2951, 1721, 1609,
464	1581, 1498, 1442, 1281, 1223, 1115, 1026, 826, 779, 707, 653. Elemental anal. calcd
465	for C ₂₈ H ₂₅ BrN ₂ O ₃ : C, 65.00; H, 4.87; N, 5.41; found C, 64.97; H, 4.86; N, 5.44.
466	HRMS (ESI) m/z calcd for $C_{28}H_{25}BrN_2O_3$ [M–Br] ⁺ 437.1860, found 437.1854.

4.2.2.6. 6-methoxy-1-(4-(methoxycarbonyl)phenyl)-2-(4-(trifluoromethyl)benzyl)-9H-467 pyrido[3,4-b]indol-2-ium bromide (4f): yellow solid; yield, 81%; m.p. 168.7–169.6 °C; 468 ¹H NMR (500 MHz, CD₃OD) δ 8.83 (s, 2H), 8.27 (d, J = 10 Hz, 2H), 7.94 (d, J = 2.5469 Hz, 1H), 7.72 (d, J = 10 Hz, 2H), 7.60 (d, J = 10 Hz, 2H), 7.56 (d, J = 10 Hz, 1H), 470 7.42–7.39 (m, 1H), 7.16 (d, J = 5 Hz, 2H), 5.96 (s, 2H), 4.00 (s, 3H), 3.96 (s, 3H); ¹³C 471 472 NMR (125 MHz, CD₃OD) δ167.3, 157.1, 142.0, 140.6, 140.3, 137.1, 135.0, 134.9, 473 134.3, 133.8, 131.9, 131.6, 131.4, 129.0, 127.0, 127.0, 126.9, 126.3, 125.5, 124.1, 121.6, 118.7, 115.0, 104.1, 61.5, 56.5, 53.1. IR (KBr, cm⁻¹) 3362, 3065, 2957, 1718, 474 1659, 1637, 1619, 1582, 1526, 1497, 1469, 1326, 1294, 1279, 1223, 1164, 1128, 1067, 475 1015, 870, 825, 792, 769, 726, 660, 645, 596. Elemental anal. calcd for 476 477 C₂₈H₂₂BrF₃N₂O₃: C, 58.86; H, 3.88; N, 4.90; found C, 58.90; H, 3.87; N, 4.85. HRMS 478 (ESI) m/z calcd for $C_{28}H_{22}BrF_3N_2O_3 [M-Br]^+ 491.1577$, found 491.1573.

479 4.2.2.7. 2-benzyl-1-(furan-2-yl)-9H-pyrido[3,4-b]indol-2-ium bromide (**4g**): red-brown solid; yield, 83%; m.p. 141.1–142.7 °C; ¹H NMR (500 MHz, CD₃OD) δ 480 481 8.75-8.70 (m, 2H), 8.43 (d, J = 10 Hz, 1H), 8.12 (d, J = 2.5 Hz, 1H), 7.83-7.77 (m, 2H), 7.50–7.47 (m, 1H), 7.36–7.34 (m, 3H), 7.27 (d, J = 2.5 Hz, 1H), 7.13–7.11 (m, 482 2H), 6.88 (dd, J = 5 Hz, 2.5 Hz, 1H), 6.09 (s, 2H); ¹³C NMR (125 MHz, CD₃OD) δ 483 148.7, 146.5, 141.8, 136.4, 136.1, 135.8, 135.7, 133.8, 130.8, 130.3, 130.0, 128.2, 484 124.3, 123.5, 121.1, 120.0, 118.2, 114.2, 113.9, 62.8. IR (KBr, cm⁻¹) 3399, 3055, 485 2924, 1627, 1580, 1500, 1453, 1335, 1270, 1249, 1201, 1153, 1018, 762, 742, 701. 486

487 Elemental anal. calcd for $C_{22}H_{17}BrN_2O$: C, 65.20; H, 4.23; N, 6.91; found C, 65.17; H, 488 4.25; N, 6.93. HRMS (ESI) m/z calcd for $C_{22}H_{17}BrN_2O$ [M–Br]⁺ 325.1335, found 489 325.1332.

490 4.2.2.8. 1-(furan-2-yl)-2-(4-methylbenzyl)-9H-pyrido[3,4-b]indol-2-ium bromide (4h): red-brown solid; yield, 85%; m.p. 156.6–157.8 °C; ¹H NMR (500 MHz, CD₃OD) δ 491 492 8.79–8.75 (m, 2H), 8.45 (d, J = 10 Hz, 1H), 8.08 (d, J = 2.5 Hz, 1H), 7.84–7.78 (m, 2H), 7.64 (d, J = 5 Hz, 2H), 7.50 (t, J = 5 Hz, 1H), 7.30–7.26 (m, 3H), 6.88–6.86 (m, 493 1H), 6.19 (s, 2H); 13 C NMR (125 MHz, CD₃OD) δ 149.1, 146.9, 141.9, 140.6, 136.7, 494 495 136.3, 134.3, 132.0, 131.1, 129.1, 127.4, 127.4, 124.8, 124.0, 121.5, 120.5, 118.7, 114.5, 114.3, 62.6, 31.0. IR (KBr, cm⁻¹) 3409, 3146, 3049, 2925, 2855, 1629, 1583, 496 497 1505, 1452, 1422, 1327, 1279, 1164, 1127, 1067, 1017, 753. Elemental anal. calcd for 498 C₂₃H₁₉BrN₂O: C, 65.88; H, 4.57; N, 6.68; found C, 65.85; H, 4.59; N, 6.67. HRMS 499 (ESI) m/z calcd for $C_{23}H_{19}BrN_2O[M-Br]^+$ 339.1492, found 339.1490.

500 4.2.2.9. 1-(furan-2-yl)-2-(4-(trifluoromethyl)benzyl)-9H-pyrido[3,4-b]indol-2-ium

bromide (4i): red-brown solid; yield, 78%; m.p. 185.6–187.2 °C; ¹H NMR (500 MHz. 501 502 CD₃OD) δ 8.80–8.75 (m, 2H), 8.45 (d, J = 2.5 Hz, 1H), 8.08 (s, 1H), 7.84–7.79 (m, 2H), 7.65 (d, J = 10 Hz, 2H), 7.50 (t, J = 5 Hz, 1H), 7.31–7.27 (m, 3H), 6.87 (dd, J = 503 12.5 Hz, 2.5 Hz, 1H), 6.2 (s, 2H); ¹³C NMR (125 MHz, CD₃OD) δ 148.8, 146.6, 504 141.6, 140.2, 136.5, 136.4, 136.0, 134.0, 131.9, 131.7, 130.8, 128.7, 127.1, 127.1, 505 506 127.1, 127.0, 126.3, 124.4, 124.2, 123.6, 122.0, 121.2, 120.2, 118.4, 114.2, 114.0, 62.2. IR (KBr, cm⁻¹) 3410, 3048, 1629, 1511, 1502, 1327, 1280, 1164, 1127, 1067, 507 508 1018, 753. Elemental anal. calcd for C₂₃H₁₆BrF₃N₂O: C, 58.37; H, 3.41; N, 5.92; 509 found C, 58.35; H, 3.43; N, 5.89. HRMS (ESI) m/z calcd for C₂₃H₁₆BrF₃N₂O [M–Br]⁺ 510 393.1209, found 393.1202.

511 4.2.2.10. 2-benzyl-1-(furan-2-yl)-6-methoxy-9H-pyrido[3,4-b]indol-2-ium

512 *bromide* (**4j**): red-brown solid; yield, 92%; m.p. 194.8–196.1 °C; ¹H NMR (500 MHz,

513 DMSO-*d*6) δ 12.43 (s, 1H), 8.88 (dd, J = 10 Hz, 5 Hz, 2H), 8.24 (d, J = 2.5 Hz, 1H),

514 8.07 (d, *J* = 5 Hz, 1H), 7.73 (d, *J* = 10 Hz, 1H), 7.49–7.47 (m, 1H), 7.35–7.32 (m, 4H),

515 7.08–7.06 (m, 2H), 6.95 (dd, J = 5 Hz, 2.5 Hz, 1H), 6.07 (s, 2H), 3.92 (s, 3H); ¹³C

516 NMR (125 MHz, DMSO-*d*6) δ 155.1, 148.7, 146.5, 141.8, 136.4, 136.1, 135.8, 133.8, 517 130.8, 130.3, 130.0, 128.2, 124.3, 123.5, 121.1, 120.0, 118.2, 114.2, 113.9, 62.8, 56.4. 518 IR (KBr, cm⁻¹) 3409, 3048, 2917, 1626, 1588, 1516, 1462, 1327, 1266, 1230, 1199, 519 1142, 1008, 767, 722. Elemental anal. calcd for C₂₃H₁₉BrN₂O₂: C, 63.46; H, 4.40; N, 520 6.44; found C, 63.45; H, 4.43; N, 6.40. HRMS (ESI) m/z calcd for C₂₃H₁₉BrN₂O₂ 521 [M–Br]⁺ 355.1441, found 355.1437.

1-(furan-2-yl)-6-methoxy-2-(4-methylbenzyl)-9H-pyrido[3,4-b]indol-2-iu 522 4.2.2.11. *m bromide* (4k): red-brown solid; yield, 81%; m.p. 191.6–193.2 °C; ¹H NMR (500 523 524 MHz, CD₃OD) δ 8.66 (dd, J = 10 Hz, 5 Hz, 2H), 8.10 (d, J = 5 Hz, 1H), 7.89 (d, J = 5Hz, 1H), 7.69 (d, J = 10 Hz, 1H), 7.47–7.45 (m, 1H), 7.26 (d, J = 5 Hz, 1H), 7.16 (d, J 525 = 5 Hz, 2H), 6.99 (d, J = 10 Hz, 2H), 6.88 (dd, J = 5 Hz, 2.5 Hz, 1H), 6.00 (s, 2H), 526 3.96 (s, 3H), 2.30 (s, 3H); 13 C NMR (125 MHz, CD₃OD) δ 157.3, 148.6, 141.9, 141.8, 527 140.2, 136.5, 135.1, 135.0, 132.8, 130.9, 128.2, 125.5, 121.6, 119.8, 118.1, 115.7, 528 115.2, 113.9, 104.0, 62.5, 56.4, 21.1. IR (KBr, cm⁻¹) 3424, 3047, 2998, 2953, 1632, 529 530 1605, 1582, 1512, 1498, 1469, 1434, 1351, 1311, 1295, 1221, 1184, 1126, 1023, 836, 531 807, 758. Elemental anal. calcd for C₂₄H₂₁BrN₂O₂: C, 64.15; H, 4.71; N, 6.23; found 532 C, 64.19; H, 4.70; N, 6.19. HRMS (ESI) m/z calcd for C₂₄H₂₁BrN₂O₂ [M-Br]⁺ 369.1598, found 369.1596. 533

1-(furan-2-yl)-6-methoxy-2-(4-(trifluoromethyl)benzyl)-9H-pyrido[3,4-b] 534 4.2.2.12. 535 indol-2-ium bromide (41): red-brown solid; yield, 76%; m.p. 198.5–199.6 °C; ¹H 536 NMR (500 MHz, CD₃OD) δ 8.72 (dd, J = 10 Hz, 5 Hz, 2H), 8.07 (d, J = 2.5 Hz, 1H), 537 7.92 (d, J = 2.5 Hz, 1H), 7.71–7.64 (m, 3H), 7.49–7.47 (m, 1H), 7.30–7.25 (m, 3H), 6.86 (dd, J = 5 Hz, 2.5 Hz, 1H), 6.17 (s, 2H), 3.97 (s, 3H); ¹³C NMR (125 MHz, 538 CD₃OD) δ 157.1, 148.5, 141.7, 141.4, 140.1, 136.3, 135.3, 135.0, 130.6, 128.4, 126.9, 539 540 126.8, 126.8, 126.8, 126.1, 125.4, 121.4, 120.7, 119.8, 118.2, 118.1, 115.0, 113.7, 103.7, 61.8, 56.2. IR (KBr, cm⁻¹) 3435, 3152, 3045, 1658, 1640, 1620, 1611, 1582, 541 542 1512, 1501, 1468, 1328, 1297, 1223, 1166, 1127, 1068, 1019. Elemental anal. calcd 543 for C₂₄H₁₈BrF₃N₂O₂: C, 57.27; H, 3.60; N, 5.57; found C, 57.22; H, 3.65; N, 5.55. HRMS (ESI) m/z calcd for $C_{24}H_{18}BrF_3N_2O_2$ [M–Br]⁺ 423.1315, found 423.1311. 544

545 2-benzyl-1-(pyridin-2-yl)-9H-pyrido[3,4-b]indol-2-ium bromide (4m): 4.2.2.13. yellow solid; yield, 67%; m.p. 137.6–138.9 °C; ¹H NMR (500 MHz, DMSO-d6) δ 546 547 12.42 (s, 1H), 9.05–8.92 (m, 2H), 8.93–8.92 (m, 1H), 8.58 (d, J = 5 Hz, 1H), 8.12 (td, 548 J = 5 Hz, 2.5Hz, 1H), 7.87 (d, J = 5 Hz, 1H), 7.82 (td, J = 5 Hz, 2.5Hz, 1H), 7.77– 7.75 (m, 1H), 7.71 (d, J = 5 Hz, 1H), 7.50 (td, J = 10 Hz, 5Hz, 1H), 7.28–7.22 (m, 549 3H), 6.94 (dd, J = 10 Hz, 2.5 Hz, 2H), 5.97 (s, 2H); ¹³C NMR (125 MHz, DMSO-*d*6) 550 δ 150.6, 147.0, 144.8, 138.2, 137.3, 135.0, 134.7, 134.7, 133.9, 132.4, 128.6, 128.4, 551 127.2, 127.0, 126.0, 123.7, 121.9, 119.4, 117.9, 113.1, 59.9. IR (KBr, cm⁻¹) 3414, 552 553 3056, 2780, 1629, 1574, 1459, 1330, 1270, 1243, 745. Elemental anal. calcd for C₂₃H₁₈BrN₃: C, 66.36; H, 4.36; N, 10.09; found C, 66.33; H, 4.39; N, 10.12. HRMS 554 555 (ESI) m/z calcd for $C_{23}H_{18}BrN_3 [M-Br]^+ 336.1495$, found 336.1494.

556 4.2.2.14. 2-(4-methylbenzyl)-1-(pyridin-2-yl)-9H-pyrido[3,4-b]indol-2-ium

bromide (4n): yellow solid; yield, 70%; m.p. 133.4-135.1 °C; ¹H NMR (500 MHz, 557 CD₃OD) δ 8.95 (dt, J = 10 Hz, 2.5 Hz, 1H), 8.78 (dd, J = 10 Hz, 5 Hz, 2H), 8.45 (d, J558 = 5 Hz, 1H), 8.12 (td, J = 10 Hz, 2.5 Hz, 1H), 7.81–7.75 (m, 3H), 7.67 (d, J = 10 Hz, 559 560 1H), 7.48 (td, J = 10 Hz, 5 Hz, 1H), 7.07 (d, J = 10 Hz, 2H), 6.87 (d, J = 10 Hz, 2H), 5.90 (s, 2H), 2.26 (s, 3H); 13 C NMR (125 MHz, CD₃OD) δ 152.2, 148.7, 146.8, 140.2, 561 562 139.6, 138.7, 136.8, 136.1, 135.5, 133.9, 132.6, 130.7, 128.7, 128.2, 127.5, 124.5, 123.4, 121.0, 118.7, 114.0, 61.9, 21.1. IR (KBr, cm⁻¹) 3404, 3050, 3008, 2779, 1628, 563 564 1573, 1518, 1461, 1430, 1332, 1267, 1143, 793, 753. Elemental anal. calcd for C₂₄H₂₀BrN₃: C, 66.98; H, 4.68; N, 9.76; found C, 66.94; H, 4.71; N, 9.79. HRMS 565 (ESI) m/z calcd for $C_{24}H_{20}BrN_3 [M-Br]^+$ 350.1652, found 350.1649. 566

567 4.2.2.15. 1-(pyridin-2-yl)-2-(4-(trifluoromethyl)benzyl)-9H-pyrido[3,4-b]indol-2-i 568 um bromide (**4o**): yellow solid; yield, 69%; m.p. 194.3–196.0 °C; ¹H NMR (500 MHz, 569 CD₃OD) δ 8.91 (d, J = 5 Hz, 1H), 8.86 (dd, J = 10 Hz, 5 Hz, 2H), 8.50 (d, J = 10 Hz, 570 1H), 8.09 (td, J = 10 Hz, 2.5 Hz, 1H), 7.84–7.79 (m, 2H), 7.75–7.73 (m, 1H), 7.69 (d, 571 J = 5 Hz, 1H), 7.56 (d, J = 5 Hz, 2H), 7.52 (t, J = 7.5 Hz, 1H), 7.18 (d, J = 10 Hz, 2H), 572 6.09 (s, 2H); ¹³C NMR (125 MHz, CD₃OD) δ 152.2, 148.5, 146.9, 140.1, 139.6, 138.8, 573 136.9, 136.5, 135.9, 134.1, 131.9, 129.2, 128.3, 127.6, 126.9, 126.9, 126.9, 126.8,

574 126.3, 124.6, 124.1, 123.6, 121.1, 118.9, 114.1, 61.4. IR (KBr, cm⁻¹) 3347, 3011, 575 1629, 1330, 1274, 1165, 1112, 1067, 756. Elemental anal. calcd for $C_{24}H_{17}BrF_3N_3$: C, 576 59.52; H, 3.54; N, 8.68; found C, 59.49; H, 3.53; N, 8.71. HRMS (ESI) m/z calcd for 577 $C_{24}H_{17}BrF_3N_3$ [M–Br]⁺ 404.1369, found 404.1367.

- 578 4.2.2.16. 2-benzyl-6-methoxy-1-(pyridin-2-yl)-9H-pyrido[3,4-b]indol-2-ium
- bromide (4p): yellow solid; yield, 79%; m.p. 157.2–158.6 °C; ¹H NMR (500 MHz, 579 DMSO-*d*6) δ 8.99 (dd, J = 15 Hz, 5 Hz, 2H), 8.91 (d, J = 5 Hz, 1H), 8.13–8.10 (m, 580 581 2H), 7.87 (d, J = 5 Hz, 1H), 7.77–7.74 (m, 1H), 7.63 (d, J = 10 Hz, 1H), 7.44–7.42 (m, 582 1H), 7.27–7.21 (m, 3H), 6.94 (d, J = 5 Hz, 2H), 5.96 (s, 2H), 3.91 (s, 3H); ¹³C NMR (125 MHz, DMSO-d6) δ 154.8, 150.6, 147.0, 140.3, 138.2, 137.4, 135.1, 134.7, 133.8, 583 133.2, 128.6, 128.4, 127.3, 127.0, 126.0, 123.6, 119.8, 117.8, 114.1, 103.8, 59.7, 55.8. 584 IR (KBr, cm⁻¹) 3409, 3039, 2958, 1608, 1571, 1515, 1499, 1472, 1454, 1435, 1358, 585 1348, 1296, 1274, 1147, 1132, 1017, 824, 791, 730. Elemental anal. calcd for 586 C₂₄H₂₀BrN₃O: C, 64.58; H, 4.52; N, 9.41; found C, 64.55; H, 4.54; N, 9.43. HRMS 587 (ESI) m/z calcd for $C_{24}H_{20}BrN_3O[M-Br]^+$ 366.1601, found 366.1593. 588
- 589 4.2.2.17. 6-methoxy-2-(4-methylbenzyl)-1-(pyridin-2-yl)-9H-pyrido[3,4-b]indol-2*ium bromide* (4q): yellow solid; yield, 75%; m.p. 164.8–165.6 °C; ¹H NMR (500 MHz, 590 591 DMSO-*d*6) δ 12.30 (s, 1H), 8.96–8.93 (m, 3H), 8.17–8.12 (m, 2H), 7.90 (d, J = 10 Hz, 2.5 Hz, 1H), 7.79–7.76 (m, 1H), 7.63 (d, J = 10 Hz, 1H), 7.46–7.44 (m, 1H), 7.05 (d, 592 J = 10 Hz, 2H), 6.85 (d, J = 10 Hz, 2H), 5.89 (s, 2H), 3.92 (s, 3H), 2.22 (s, 3H); ¹³C 593 594 NMR (125 MHz, DMSO-*d*6) δ 154.8, 150.6, 147.0, 140.1, 138.2, 137.9, 137.4, 135.0, 133.7, 133.1, 131.7, 129.2, 127.3, 127.0, 126.0, 123.6, 119.8, 117.8, 114.1, 103.8, 595 59.5, 55.8, 40.0, 39.8, 39.6, 39.5, 39.3, 39.1, 39.0, 20.6. IR (KBr, cm⁻¹) 3383, 3053, 596 3008, 1632, 1613, 1576, 1521, 1498, 1466, 1293, 1224, 1142, 1020, 825, 787. 597 598 Elemental anal. calcd for C₂₅H₂₂BrN₃O: C, 65.22; H, 4.82; N, 9.13; found C, 65.20; H, 599 4.80; N, 9.17. HRMS (ESI) m/z calcd for C₂₅H₂₂BrN₃O [M-Br]⁺ 380.1757, found 600 380.1752.
- 601 4.2.2.18. 6-methoxy-1-(pyridin-2-yl)-2-(4-(trifluoromethyl)benzyl)-9H-pyrido[3,4602 b]indol-2-ium bromide (4r): yellow solid; yield, 70%; m.p. 144.8–145.7 °C; ¹H NMR

603 $(500 \text{ MHz}, \text{CD}_3\text{OD}) \delta 8.89 \text{ (d, } J = 5 \text{ Hz}, 1\text{H}), 8.83 \text{ (dd, } J = 10 \text{ Hz}, 5 \text{ Hz}, 2\text{H}), 8.37 \text{ (d,}$ 604 J = 10 Hz, 1H), 8.17 (td, J = 10 Hz, 2.5 Hz, 1H), 7.89–7.81 (m, 1H), 7.72–7.69 (m, 1H), 7.61 (d, J = 5 Hz, 1H), 7.47 (d, J = 5 Hz, 2H), 7.39 (t, J = 7.5 Hz, 1H), 7.18 (d, J 605 = 10 Hz, 2H), 6.07 (s, 2H), 2.11 (s, 3H); 13 C NMR (125 MHz, CD₃OD) δ 153.4, 148.5, 606 146.8, 140.1, 139.6, 138.7, 137.0, 136.5, 135.9, 134.1, 131.9, 129.7, 128.0, 127.5, 607 608 127.1, 126.9, 126.9, 126.9, 126.6, 124.5, 124.0, 123.6, 121.0, 119.0, 114.2, 60.3, 20.1. 609 IR (KBr, cm⁻¹) 3409, 3006, 2958, 1611, 1578, 1516, 1499, 1472, 1438, 1296, 1224, 1139, 792. Elemental anal. calcd for C₂₅H₁₉BrF₃N₃O: C, 58.38; H, 3.72; N, 8.17; 610 611 found C, 58.34; H, 3.75; N, 8.20. HRMS (ESI) m/z calcd for C₂₅H₁₉BrF₃N₃O [M–Br]⁺ 612 434.1475, found 434.1470.

613 4.2.2.19. 1-(benzo[d][1,3]dioxol-5-yl)-2-benzyl-9H-pyrido[3,4-b]indol-2-ium

bromide (4s): yellow solid; yield, 83%; m.p. 152.8–153.4 °C; ¹H NMR (500 MHz, 614 CD₃OD) δ 8.76–8.72 (m, 2H), 8.40 (d, J = 10 Hz, 1H), 7.75 (dd, J = 15 Hz, 5 Hz, 1H), 615 7.67 (d, J = 5 Hz, 1H), 7.67 (t, J = 5 Hz, 1H), 7.32–7.31 (m, 3H), 7.12 (d, J = 5 Hz, 616 1H), 7.08–7.03 (m, 4H), 6.16 (dd, J = 10 Hz, 2.5 Hz, 2H), 5.85 (s, 2H); ¹³C NMR 617 618 (125 MHz, CD₃OD) δ 151.9, 150.3, 146.3, 141.6, 137.3, 136.0, 135.2, 134.9, 133.4, 619 130.1, 129.9, 128.5, 125.6, 124.4, 123.2, 122.4, 121.2, 118.1, 114.0, 110.8, 110.4, 620 103.6, 61.9. IR (KBr, cm⁻¹) 3401, 3022, 2930, 1605, 1492, 1430, 1331, 1229, 1140, 621 1109, 1032, 767. Elemental anal. calcd for C₂₅H₁₉BrN₂O₂: C, 65.37; H, 4.17; N, 6.10; 622 found C, 65.34; H, 4.15; N, 6.13. HRMS (ESI) m/z calcd for C₂₅H₁₉BrN₂O₂ [M–Br]⁺ 623 379.1441, found 379.1439.

1-(benzo[d][1,3]dioxol-5-yl)-2-(4-methylbenzyl)-9H-pyrido[3,4-b]indol-624 4.2.2.20. 2-ium bromide (4t): yellow solid; yield, 83%; m.p. 141.2–142.6 °C; ¹H NMR (500 625 626 MHz, CD₃OD) δ 8.73–8.67 (m, 2H), 8.40 (d, J = 5 Hz, 1H), 7.73 (t, J = 7.5 Hz, 1H), 627 7.65 (d, J = 5 Hz, 1H), 7.73 (t, J = 7.5 Hz, 1H), 7.14–7.12 (m, 3H), 7.09–7.07 (m, 1H), 7.04 (d, J = 2.5 Hz, 1H), 6.93 (d, J = 5 Hz, 2H), 6.18 (d, J = 10 Hz, 2H), 5.79 (s, 2H), 628 2.28 (s, 3H); ¹³C NMR (125 MHz, CD₃OD) δ 151.9, 150.3, 146.3, 141.5, 140.1, 137.2, 629 135.1, 134.7, 133.4, 133.0, 130.7, 128.7, 125.6, 124.3, 123.2, 122.5, 121.2, 118.0, 630 114.0, 110.8, 110.4, 103.6, 61.7, 21.1. IR (KBr, cm⁻¹) 3387, 3012, 2921, 1627, 1482, 631

- 632 1434, 1334, 1233, 1144, 1113, 1033, 753. Elemental anal. calcd for C₂₆H₂₁BrN₂O₂: C,
- 633 65.97; H, 4.47; N, 5.92; found C, 65.95; H, 4.49; N, 5.90. HRMS (ESI) m/z calcd for
- 634 $C_{26}H_{21}BrN_2O_2 [M-Br]^+$ 393.1598, found 393.1595.
- $635 \quad 4.2.2.21. \qquad 1-(benzo[d][1,3]dioxol-5-yl)-2-(4-(trifluoromethyl)benzyl)-9H-pyrido[3,$
- 4-b]indol-2-ium bromide (4u): yellow solid; yield, 89%; m.p. 156.8–157.2 °C; ¹H 636 NMR (500 MHz, CD₃OD) δ 8.82–8.79 (m, 2H), 8.44 (d, J = 10 Hz, 1H), 7.76 (t, J =637 7.5 Hz, 1H), 7.68 (d, J = 10 Hz, 1H), 7.61 (d, J = 10 Hz, 2H), 7.45 (t, J = 7.5 Hz, 1H), 638 639 7.23 (d, J = 10 Hz, 1H), 7.11–7.05 (m, 3H), 6.15 (dd, J = 15 Hz, 2.5 Hz, 2H), 6.00 (s, 640 2H); ¹³C NMR (125 MHz, CD₃OD) δ 157.1, 151.3, 150.2, 141.0, 139.5, 137.8, 135.0, 134.2, 131.4, 130.5, 129.1, 128.0, 127.8, 127.1, 126.8, 126.3, 124.9, 124.3, 123.9, 641 121.7, 121.7, 117.9, 114.8, 110.6, 110.2, 103.7, 102.6, 56.3. IR (KBr, cm⁻¹) 3402, 642 3015, 2780, 1628, 1484, 1435, 1325, 1236, 1166, 1118, 1065, 1036, 821, 753. 643 Elemental anal. calcd for C₂₆H₁₈BrF₃N₂O₂: C, 59.22; H, 3.44; N, 5.31; found C, 59.19; 644 H, 3.47; N, 5.34. HRMS (ESI) m/z calcd for $C_{26}H_{18}BrF_3N_2O_2 [M-Br]^+ 447.1315$, 645 646 found 447.1313.
- 647 4.2.2.22. 1-(benzo[d][1,3]dioxol-5-yl)-2-benzyl-6-methoxy-9H-pyrido[3,4-b]indol -2-ium bromide (4v): yellow solid; yield, 81%; m.p. 167.6–168.4 °C; ¹H NMR (500 648 649 MHz, CD₃OD) δ 8.68–8.63 (m, 2H), 7.90 (d, J = 2.5 Hz, 1H), 7.59 (d, J = 10 Hz, 1H), 7.45-7.42 (m, 1H), 7.34-7.32 (m, 3H), 7.10 (d, J = 5 Hz, 1H), 7.04-7.01 (m, 4H), 6.15650 (d, J = 5 Hz, 2H), 5.81 (s, 2H), 3.97 (s, 3H); ¹³C NMR (125 MHz, CD₃OD) δ 157.2, 651 152.0, 150.3, 141.9, 141.8, 137.5, 136.2, 134.3, 134.3, 130.2, 129.9, 128.4, 125.5, 652 653 125.2, 122.5, 121.7, 118.0, 115.0, 110.7, 110.4, 104.0, 103.7, 61.7, 56.4. IR (KBr, cm⁻ ¹) 3394, 3060, 3023, 2955, 2903, 2836, 2790, 1609, 1580, 1487, 1443, 1344, 1313, 654 1248, 1223, 1156, 1112, 1033, 928, 819. Elemental anal. calcd for C₂₆H₂₁BrN₂O₃: C, 655 656 63.81; H, 4.33; N, 5.72; found C, 63.82; H, 4.30; N, 5.74. HRMS (ESI) m/z calcd for 657 $C_{26}H_{21}BrN_2O_3[M-Br]^+ 409.1547$, found 409.1544.
- 4.2.2.23. 1-(benzo[d][1,3]dioxol-5-yl)-6-methoxy-2-(4-methylbenzyl)-9H-pyrido[3
 ,4-b]indol-2-ium bromide (4w): yellow solid; yield, 88%; m.p. 166.5–167.2 °C; ¹H
 NMR (500 MHz, CD₃OD) δ 8.68–8.62 (m, 2H), 7.89 (d, J = 2.5 Hz, 1H), 7.58 (d, J =

2.5 Hz, 1H), 7.42–7.40(m, 1H), 7.16–7.03 (m, 5H), 6.92 (d, J = 10 Hz, 2H), 6.16 (d, J 661 = 5 Hz, 2H), 5.76 (s, 2H), 3.95 (s, 3H), 2.30 (s, 3H); 13 C NMR (125 MHz, CD₃OD) δ 662 157.1, 151.9, 150.3, 141.8, 141.7, 140.1, 137.4, 134.2, 134.2, 133.1, 130.8, 128.5, 663 664 125.5, 125.1, 122.6, 121.7, 118.0, 115.0, 110.8, 110.4, 104.0, 103.7, 61.6, 56.4, 21.1. IR (KBr, cm⁻¹) 3442, 3058, 3025, 2963, 2892, 2835, 1612, 1578, 1523, 1485, 1439, 665 1344, 1317, 1299, 1224, 1188, 1113, 1038, 934, 822, 784. Elemental anal. calcd for 666 C₂₇H₂₃BrN₂O₃: C, 64.42; H, 4.61; N, 5.57; found C, 64.39; H, 4.57; N, 5.60. HRMS 667 (ESI) m/z calcd for $C_{27}H_{23}BrN_2O_3 [M-Br]^+ 423.1703$, found 423.1696. 668

669 4.2.2.24. 1-(benzo[d][1,3]dioxol-5-yl)-6-methoxy-2-(4-(trifluoromethyl)benzyl)-9 *H-pyrido*[3,4-*b*]*indo*]-2-*ium bromide* (4x): yellow solid; yield, 89%; m.p. 181.5–182.7 670 ^oC; ¹H NMR (500 MHz, CD₃OD) δ 8.72 (dd, J = 10 Hz, 5 Hz, 2H), 7.91 (d, J = 5 Hz, 671 672 1H), 7.64–7.58 (m, 3H), 7.43–7.41 (m, 1H), 7.20 (d, J = 5 Hz, 2H), 7.09–7.03 (m, 3H), 6.14 (d, *J* = 15 Hz, 2H), 5.95 (s, 2H), 3.96 (s, 3H); ¹³C NMR (125 MHz, CD₃OD) 673 674 δ 157.1, 152.0, 150.3, 141.9, 141.9, 140.6, 137.5, 134.6, 134.5, 131.8, 131.6, 129.0, 127.0, 127.0, 126.9, 126.9, 126.4, 125.5, 125.3, 124.2, 122.4, 121.7, 118.2, 115.0, 675 110.7, 110.4, 104.0, 103.7, 61.2, 56.4. IR (KBr, cm⁻¹) 3422, 3018, 2784, 1616, 1579, 676 677 1490, 1439, 1326, 1227, 1164, 1118, 1068, 1036, 936, 821. Elemental anal. calcd for 678 C₂₇H₂₀BrF₃N₂O₃: C, 58.18; H, 3.62; N, 5.03; found C, 58.16; H, 3.59; N, 5.06. HRMS 679 (ESI) m/z calcd for $C_{27}H_{20}BrF_3N_2O_3 [M-Br]^+ 477.1421$, found 477.1419.

680 4.2.2.25. 2-(4-fluorobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido 681 [3,4-b]indol-2-ium bromide (5a): yellow solid; yield, 73%; m.p. 158.9–156.3 °C; ¹H 682 NMR (500 MHz, DMSO-*d*6) δ 12.18 (s, 1H), 8.93 (dd, J = 10 Hz, 5 Hz, 2H), 8.18 (d, J = 5 Hz, 2H), 8.10 (d, J = 2.5 Hz, 1H), 7.78 (d, J = 10 Hz, 2H), 7.58 (d, J = 10 Hz, 683 684 1H), 7.45–7.43 (m, 1H), 7.10 (t, J = 10 Hz, 2H), 7.00–6.97 (m, 2H), 5.78 (s, 2H), 3.95 (s, 3H), 3.92 (s, 3H); 13 C NMR (125 MHz, DMSO-*d6*) δ 165.6, 162.8, 160.8, 154.8, 685 140.2, 139.1, 135.5, 133.5, 132.5, 131.9, 131.0, 130.4, 129.9, 129.5, 123.5, 119.9, 686 117.5, 115.7, 114.1, 103.7, 59.4, 55.7, 52.6. IR (KBr, cm⁻¹) 3407, 3063, 2953, 1716, 687 1636, 1607, 1580, 1510, 1498, 1468, 1436, 1351, 1296, 1278, 1221, 1163, 1120, 1042, 688 1029, 1013, 825, 786, 762. Elemental anal. calcd for C₂₇H₂₂BrFN₂O₃: C, 62.20; H, 689

690 4.25; N, 5.37; found C, 62.17; H, 4.29; N, 5.34. HRMS (ESI) m/z calcd for 691 $C_{27}H_{22}BrFN_2O_3 [M-Br]^+ 441.1609$, found 441.1605.

692 4.2.2.26. 2-(4-chlorobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido 693 [3,4-b]indol-2-ium chloride (5b): yellow solid; yield, 86%; m.p. 157.7–159.2 °C; ¹H 694 NMR (500 MHz, DMSO-*d*6) δ 12.13 (s, 1H), 8.89 (s, 2H), 8.18 (d, J = 10 Hz, 2H), 695 8.08 (d, J = 2.5 Hz, 1H), 7.77 (d, J = 5 Hz, 2H), 7.60 (d, J = 10 Hz, 1H), 7.46–7.44 (m, 1H), 7.33 (d, J = 10 Hz, 2H), 7.33 (d, J = 5 Hz, 2H), 5.77 (s, 2H), 3.96 (s, 3H), 3.93 (s, 696 3H). IR (KBr, cm⁻¹) 3383, 3062, 2951, 2837, 1715, 1633, 1610, 1580, 1496, 1468, 697 698 1453, 1436, 1405, 1350, 1295, 1279, 1223, 1188, 1120, 1089, 1029, 1015, 826, 811, 782. Elemental anal. calcd for C₂₇H₂₂Cl₂N₂O₃: C, 65.73; H, 4.49; N, 5.68; found C, 699 700 65.71; H, 4.47; N, 5.70. HRMS (ESI) m/z calcd for C₂₇H₂₂Cl₂N₂O₃ [M–Cl]⁺ 457.1313, 701 459.1284, found 457.1311, 459.1280.

702 4.2.2.27. 2-(4-bromobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido [3,4-b]indol-2-ium bromide (5c): yellow solid; yield, 91%; m.p. 160.9–162.4 °C; ¹H 703 NMR (500 MHz, DMSO-*d*6) δ 12.16 (s, 1H), 8.91 (dd, J = 10 Hz, 5 Hz, 2H), 8.19 (d, 704 705 J = 5 Hz, 2H), 8.19 (d, J = 2.5 Hz, 1H), 7.79 (d, J = 10 Hz, 2H), 7.59 (d, J = 10 Hz, 1H), 7.48–7.44 (m, 3H), 6.90 (d, J = 10 Hz, 2H), 5.75 (s, 2H), 3.95 (s, 3H), 3.92 (s, 706 3H). IR (KBr, cm⁻¹) 3374, 3148, 3062, 2951, 1716, 1635, 1610, 1580, 1527, 1497, 707 1469, 1453, 1436, 1405, 1294, 1280, 1223, 1186, 1111, 1071, 1029, 1013, 826, 807, 708 709 782, 770. Elemental anal. calcd for C₂₇H₂₂Br₂N₂O₃: C, 55.69; H, 3.81; N, 4.81; found C, 55.67; H, 3.83; N, 4.78. HRMS (ESI) m/z calcd for C₂₇H₂₂Br₂N₂O₃ [M-Br]⁺ 710 711 501.0808, 503.0788, found 501.0806, 503.0785.

7124.2.2.28.2-(2-fluorobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido713[3,4-b]indol-2-ium bromide (5d): yellow solid; yield, 72%; m.p. 155.8–156.9 °C; ¹H714NMR (500 MHz, DMSO-d6) δ 12.18 (s, 1H), 8.94 (dd, J = 10 Hz, 5 Hz, 2H), 8.16 (d,715J = 5 Hz, 2H), 8.11 (d, J = 2.5 Hz, 1H), 7.77 (d, J = 10 Hz, 2H), 7.59 (d, J = 10 Hz,7161H), 7.46–7.44 (m, 1H), 7.39–7.35 (m, 1H), 7.14–7.07 (m, 2H), 6.90 (t, J = 7.5 Hz,7171H), 5.88 (s, 2H), 3.95 (s, 3H), 3.92 (s, 3H); ¹³C NMR (125 MHz, DMSO-d6) δ 165.6,718160.5, 158.5, 154.8, 140.2, 139.1, 135.4, 133.8, 132.7, 132.4, 131.9, 130.9, 130.9,

719130.3, 129.8, 129.6, 129.6, 124.8, 123.6, 121.8, 121.7, 119.9, 117.4, 115.5, 115.3,720114.1, 103.7, 55.8, 54.9, 52.6. IR (KBr, cm⁻¹) 3409, 3046, 3015, 2954, 1723, 1608,7211580, 1497, 1454, 1439, 1295, 1278, 1225, 1183, 1119, 1102, 1028, 828, 764.722Elemental anal. calcd for $C_{27}H_{22}BrFN_2O_3$: C, 62.20; H, 4.25; N, 5.37; found C, 62.17;723H, 4.21; N, 5.41. HRMS (ESI) m/z calcd for $C_{27}H_{22}BrFN_2O_3$ [M–Br]⁺ 441.1609,724found 441.1597.

2-(3-fluorobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido 725 4.2.2.29. [3,4-b]indol-2-ium bromide (5e): yellow solid; yield, 79%; m.p. 147.2–149.0 °C; ¹H 726 727 NMR (500 MHz, DMSO-*d*6) δ 12.20 (s, 1H), 8.92 (t, J = 5 Hz, 2H), 8.17 (d, J = 10Hz, 2H), 8.10 (d, J = 5 Hz, 1H), 7.78 (d, J = 10 Hz, 2H), 7.59 (d, J = 5 Hz, 1H), 7.46– 728 729 7.43 (m, 1H), 7.33–7.29 (m, 1H), 7.13 (td, J = 10 Hz, 2.5 Hz, 1H), 6.83 (d, J = 10 Hz, 1H), 6.75 (d, J = 10 Hz, 1H), 5.80 (s, 2H), 3.95 (s, 3H), 3.92 (s, 3H). IR (KBr, cm⁻¹) 730 731 3397, 3006, 2952, 2835, 1722, 1635, 1611, 1579, 1526, 1498, 1467, 1452, 1437, 1403, 1348, 1279, 1223, 1184, 1131, 1118, 1030, 1013, 826, 779. Elemental anal. calcd for 732 C₂₇H₂₂BrFN₂O₃: C, 62.20; H, 4.25; N, 5.37; found C, 62.18; H, 4.23; N, 5.39. HRMS 733 734 (ESI) m/z calcd for $C_{27}H_{22}BrFN_2O_3 [M-Br]^+ 441.1609$, found 441.1607.

2-(2-bromobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido 735 4.2.2.30. [3,4-b]indol-2-ium bromide (5f): yellow solid; yield, 93%; m.p. 249.8–251.1 °C; ¹H 736 737 NMR (500 MHz, DMSO-*d*6) δ 12.21 (s, 1H), 8.94–8.81 (m, 2H), 8.15 (d, J = 10 Hz, 738 2H), 8.11 (d, J = 2.5 Hz, 1H), 7.76 (d, J = 10 Hz, 2H), 7.61–7.58 (m, 2H), 7.48–7.45 739 (m, 1H), 7.32-7.26 (m, 2H), 6.84 (d, J = 5 Hz, 1H), 5.81 (s, 2H), 3.93 (s, 3H), 3.93 (s, 3H). IR (KBr, cm⁻¹) 3009, 2986, 2951, 1720, 1608, 1577, 1496, 1437, 1297, 1268, 740 741 1229, 1133, 1118, 1104, 1029, 828, 765. Elemental anal. calcd for C₂₇H₂₂Br₂N₂O₃: C, 742 55.69; H, 3.81; N, 4.81; found C, 55.66; H, 3.84; N, 4.79. HRMS (ESI) m/z calcd for 743 $C_{27}H_{22}Br_2N_2O_3 [M-Br]^+$ 501.0808, 503.0788, found 501.0806, 503.0785.

7444.2.2.31.2-(3-bromobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-pyrido745[3,4-b]indol-2-ium bromide (**5g**): yellow solid; yield, 95%; m.p. 150.3–152.1 °C; ¹H746NMR (500 MHz, DMSO-d6) δ 12.19 (s, 1H), 8.93 (dd, J = 10 Hz, 5 Hz, 2H), 8.17 (d,747J = 10 Hz, 2H), 8.09 (d, J = 5 Hz, 1H), 7.76 (d, J = 10 Hz, 2H), 7.58 (d, J = 10 Hz,

748 1H), 7.49–7.43 (m, 2H), 7.21 (t, J = 7.5 Hz, 1H), 7.13 (t, J = 2.5 Hz, 1H), 6.91 (d, J =10 Hz, 1H), 5.79 (s, 2H), 3.95 (s, 3H), 3.92 (s, 3H). IR (KBr, cm⁻¹) 3409, 3046, 3015, 749 750 2954, 2837, 1723, 1608, 1580, 1497, 1454, 1439, 1349, 1295, 1279, 1225, 1183, 1151, 751 1133, 1119, 1102, 1028, 1012, 828, 764. Elemental anal. calcd for C₂₇H₂₂Br₂N₂O₃: C, 752 55.69; H, 3.81; N, 4.81; found C, 55.67; H, 3.84; N, 4.83. HRMS (ESI) m/z calcd for 753 $C_{27}H_{22}Br_2N_2O_3$ [M–Br]⁺ 501.0808, 503.0788, found 501.0807, 503.0786. 754 4.2.2.32. 2-(2,4-difluorobenzyl)-6-methoxy-1-(4-(methoxycarbonyl)phenyl)-9H-py 755 *rido*[3,4-b]*indo*l-2-*ium bromide* (5h): yellow solid; yield, 87%; m.p. 142.3–144.3 °C; 756 ¹H NMR (500 MHz, DMSO-*d*6) δ 12.14 (s, 1H), 8.93 (dd, J = 10 Hz, 5 Hz, 2H), 8.19 (d, J = 5 Hz, 2H), 8.10 (d, J = 2.5 Hz, 1H), 7.78 (d, J = 10 Hz, 2H), 7.58 (d, J = 5 Hz, 757 1H), 7.46–7.44 (m, 1H), 7.23–7.19 (m, 1H), 7.02–6.96 (m, 2H), 5.84 (s, 2H), 3.96 (s, 758 3H), 3.92 (s, 3H). IR (KBr, cm⁻¹) 3069, 3034, 3002, 2960, 1711, 1609, 1577, 1532, 759 760 1501, 1472, 1435, 1409, 1318, 1286, 1227, 1184, 1136, 1110, 1086, 1035, 1014, 960,

- 761 827, 767. Elemental anal. calcd for $C_{27}H_{21}BrF_2N_2O_3$: C, 60.12; H, 3.92; N, 5.19; 762 found C, 60.15; H, 3.90; N, 5.15. HRMS (ESI) m/z calcd for $C_{27}H_{21}BrF_2N_2O_3$ [M– 763 Br]⁺ 459.1515, found 459.1513.
- 6-methoxy-1-(4-(methoxycarbonyl)phenyl)-2-(4-nitrobenzyl)-9H-pyrido 764 4.2.2.33. 3,4-b]indol-2-ium chloride (5i): yellow solid; yield, 83%; m.p. 162.1–162.8 °C; ¹H 765 766 NMR (500 MHz, DMSO-*d*6) δ 12.34 (s, 1H), 9.01 (dd, J = 10 Hz, 5 Hz, 2H), 8.15– 8.12 (m, 5H), 7.78 (d, J = 5 Hz, 2H), 7.64 (d, J = 10 Hz, 1H), 7.50–7.47 (m, 1H), 7.23 767 (d, J = 10 Hz, 2H), 6.00 (s, 2H), 3.97 (s, 3H), 3.96 (s, 3H); ¹³C NMR (125 MHz, 768 769 DMSO-d6) δ 165.5, 154.8, 147.1, 142.3, 139.3, 137.0, 135.7, 133.7, 133.6, 132.9, 770 132.7, 132.4, 131.9, 130.4, 129.8, 128.1, 123.6, 120.0, 117.6, 103.6, 59.3, 55.7, 52.5. IR (KBr, cm⁻¹) 2998, 2952, 1715, 1610, 1580, 1522, 1500, 1434, 1405, 1345, 1318, 771 772 1285, 1229, 1135, 1110, 829. Elemental anal. calcd for C₂₇H₂₂ClN₃O₅: C, 64.35; H, 4.40; N, 8.34; found C, 64.31; H, 4.42; N, 8.33. HRMS (ESI) m/z calcd for 773 774 $C_{27}H_{22}ClN_{3}O_{5}[M-Cl]^{+}$ 468.1554, found 468.1551.
- 775 4.2.3. X-ray structures of compounds 3e and 5f

776X-ray quality crystal of compound **3e** was obtained from the dichloromethane:777petroleum ether = 4:1 solution after 3 days. The crystal structure of compound **3e** (Fig.7784) was determined on a Rigaku Oxford Diffraction Supernova Dual Source at 100 K,779Cu at Zero equipped with an AtlasS2 CCD using Cu Kα radiation. The atomic780coordinates have been deposited at the Cambridge Crystallographic Data Center781(CCDC) with CCDC numbers 1819364. It is worth mentioning that this crystal782structure has been published earlier [39].

783X-ray quality crystal of compound **5f** was obtained from the dichloromethane:784methanol = 2:1 solution after 3 days. The crystal structure of compound **5f** (Fig. 4)785was determined on a Rigaku Oxford Diffraction Supernova Dual Source at 100 K, Cu786at Zero equipped with an AtlasS2 CCD using Cu Kα radiation. The atomic787coordinates have been deposited at the Cambridge Crystallographic Data Center788(CCDC) with CCDC numbers 1819367.

789 4.3. Antibacterial assay

790 MICs were determined as described by the National Committee for Clinical 791 Laboratory Standards [40]. Four Gram-positive bacteria (S. aureus, MRSA, B. cereus, 792 and B. subtilis) and two Gram-negative bacteria (E. coli and R. solanacearum) were 793 selected as the tested bacteria. S. aureus 1.8721, B. cereus 1.1846 and and Escherichia 794 coli 1.1636 were purchased from the China General Microbiological Culture 795 Collection Center. MRSA (S. aureus ATCC 43300) and B. subtillis 769 were provided 796 by College of Chemistry & Pharmacy, Northwest A&F University. R. solanacearum 797 (CGMCC 1.12711) was provided by the College of Plant Protection, Northwest A&F 798 University. The MIC was defined as the minimum inhibitory concentration, each 799 compound resulting in visible inhibition on bacteria growth (incubation at 37 °C for 800 12–14 h). Each bacterial suspension was adjusted to a concentration of 1×10^6 801 CFU/mL. All compounds were thoroughly dried before weighing. Initially, the 802 compounds were dissolved in dimethyl sulfoxide (DMSO) to prepare the stock 803 solutions. The tested compounds and reference drugs were then prepared in liquid 804 Luria-Bertan media. The required concentrations were 200, 100, 50, 25, 12.5, 6.25

and 3.125 nmol/mL, respectively (DMSO < 0.5%). Their MICs of the preferred compounds **4f**, **4i**, **4u**, **4w**, **4x**, **5c** and reference drugs were also tested at concentrations of 256, 128, 64, 32, 16, 8, 4, 2 and 1 µg/mL.

808 4.4. Cell toxicity

809 The cytotoxicity of compounds 4f, 4i, 4l, 4u, 4w, 4x and 5c was assessed using 810 the 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-H-tetrazolium bromide (MTT) assay, 811 as described previously [41]. MARC 145 cells and L02 cells were treated with 812 compounds 4f, 4i, 4l, 4u, 4w, 4x and 5c at concentrations of 8 µg/mL. After 813 incubation, 10 μ L (5 mg/mL) of MTT in phosphate-buffered saline was added to each well, and the cells were incubated in 5% CO₂ at 37 °C for 4 h. The medium was 814 815 removed. Then the DMSO (150 µL) was added to each well. The wells were shocked 816 for 15 min in a flat shaker to dissolve purple formazan crystals. Finally, the plate was tested at 540 nm. The untreated group was considered as the control. The data were 817 818 analyzed by GraphPad Prism 6.0. The formula used for cell viability was (absorbance of compound-treated cells/absorbance of untreated cells) \times 100%. 819

820 4.5. Calculation of molecular physicochemical properties

Molecular properties, including the octanol–water partition coefficient, the topological polar surface area, the number of hydrogen bond acceptors, the number of hydrogen bond donors, the number of rotatable bonds, and the molecular volume, were calculated using the Molinspiration tool [42].

825 4.6. Thermal stability assay in vitro

The thermal stability assay *in vitro* used the UHPLC-ESI-MS/MS method [43]. A Nexera UHPLC LC-30A instrument was coupled to an AB SCIEX Triple TOF 5600⁺ spectrometer. During experiments, a 5 μ L (2 μ g/mL for ampicillin sodium, 1 μ g/mL for **4x**) sample was loaded along with mobile phase A (10% acetonitrile in double-distilled water for ampicillin sodium, 35% acetonitrile in double-distilled water for **4x**) at a flow rate of 0.4 mL/min. A sweeping collision energy setting of 35 ± 15 eV was applied to all precursor ions for collision-induced dissociation. The mass

spectrometer settings used varied, depending on the optimization required, but typical values were GS 1 = 50, GS 2 = 50, CUR = 35, and TEM = 500. The negative ion mode (-4500 V) was used for ampicillin sodium, and the positive ion mode (5500 V) was used for compound **4x**. The stability study was tested under three conditions (25 °C, 0 h; 37 °C, 6 h; 65 °C, 6 h).

838 4.7. SEM assay

839 A standard procedure was followed for SEM analysis [44]. Briefly, bacterial broth cultures in the log phase (30 mL) at $\sim 10^8$ CFU/mL were treated with freshly 840 prepared phosphate-buffered saline (0.1 M, pH 7.2) in combination with 1 h of 841 842 incubation at 37 °C. All samples were centrifuged at $4000 \times g$ for 10 min, washed, 843 and resuspended in phosphate-buffered saline. Bacterial cells were fixed with 2.5% glutaraldehyde, and then incubated at 37 °C on a glass slide (22×22 mm) for 20 min. 844 Samples were dehydrated by passing through gradient ethanol (10%, 30%, 50%, 70%, 845 846 90%, and 100%) for 5 min at each concentration. Slides were then dried with an 847 automatic critical point drying instrument (Leica EM CPD300), and a small amount 848 of gold was sputtered onto the samples using a sputter coater system to avoid charging in the microscope. The treated group, with compound 4x (0.5 MIC) added, was 849 incubated at 37 °C for 1 h. 850

851 4.8. Molecular docking study

Molecular docking evaluation was conducted using the Surflex-Dock program in 852 853 the Sybyl-X 2.0 package and Discovery Studio 2017 client. The crystal structures of 854 bacterial type II topoisomerase (PDB ID: 5IWM) and type I topoisomerase (PDB ID: 4RUL) were obtained from the Protein Data Bank (http://www.pdb.org). The crystal 855 856 structures of topoisomerases were prepared with all hydrogen atoms added, and charge added by the AMBER7 FF99 method. The structures of 4x and ciprofloxacin 857 858 were drawn in the Sybyl-X 2.0 package. Polar hydrogen atoms were then added, and 859 the energy optimized with the Tripos force field and by the Gasteiger-Hückel method.

860 4.9. Antibacterial assay in vivo[37,38]

861 The *in vivo* biological assay against R. *solanacearum* for the compound 4x was 862 carried out on an eggplant leaf. The concentration of compound 4x was 8 μ g/mL (5%) 863 DMSO in water as the solvent). The blank control was 5‰ aqueous DMSO. Each 864 suspension (V = 100 μ L) was dropped onto the leaf of eggplant, which was washed 865 and treated with water and 75% aqueous ethyl alcohol in advance. After the solvent 866 was evaporated in an ambient environment, the epidermis (Ø 4.5 mm) of the leaf was punctured with an inoculating needle, and the pathogen then inoculated. All the 867 868 treated leaves were placed in an illumination incubator (37 °C, 100% relative humidity, light/dark = 10 h/14 h) for 3 days, and the experiments were repeated three 869 870 times.

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877 Appendix A. Supplementary data

- 878 Supplementary data related to this article can be found.
- 879 **Conflict of interest**
- 880 The authors state no conflict of interest.
- 881 References

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Table Captions

- **Table 1.** *In vitro* antibacterial activity of compounds **3a–3h** and **4a–4x** (MIC, nmol/mL).
- **Table 2.** *In vitro* antibacterial activity of compounds **5a–5i** (MIC, nmol/mL).
- **Table 3.** *In vitro* antibacterial and cytotoxicity activity of some potential compounds.
- **Table 4.** Calculated molecular properties of some potential compounds.
- **Figure Captions**
- **Fig. 1.** Design of the quaternization harman analogues.
- 1008 Fig. 2. Synthesis of compounds 3a–3h and 4a–4x.
- **Fig. 3.** Synthesis of compounds **5a–5i**.
- **Fig. 4.** Crystal structures of compounds **3e** and **5f**.
- **Fig. 5.** Structure–activity relationships of the quaternization harman analogues against bacteria.
- 1012 Fig. 6. SEM of MRSA and *R. solanacearum*. (A) and (C) are untreated. (B) and (D) are treated
- 1013 with compound **4x**.
- 1014 Fig. 7. Three-dimensional conformations of compound 4x docked in bacterial type II
 1015 topoisomerase complex.
- 1016 Fig. 8. Representative photographs of compound 4x in the *in vivo* antibacterial assay: (left)
- 1017 untreated and (right) treated.

Const		Gram-posi	tive bacteria		Gram-n	egative bacteria
Compa	S. aureus	MRSA	B. cereus	B. subtilis	E. coli	R. solanacearum
3a	>200	>200	>200	>200	>200	>200
3b	>200	>200	>200	>200	>200	>200
3 c	>200	>200	>200	>200	>200	>200
3d	>200	>200	>200	>200	>200	>200
3e	>200	>200	>200	>200	>200	>200
3f	>200	>200	>200	>200	>200	>200
3g	>200	>200	>200	>200	>200	>200
3h	200	200	>200	>200	>200	>200
4a	100	200	200	100	>200	100
4 b	50	50	100	50	200	50
4 c	50	50	100	50	200	50
4d	100	200	200	100	>200	100
4e	50	100	50	50	>200	50
4f	25	25	25	25	200	25
4 g	100	100	>200	100	>200	100
4h	50	50	100	25	100	25
4i	25	25	50	25	50	25
4j	100	100	200	50	>200	100
4k	50	50	100	25	100	25
41	25	25	50	25	50	25
4 m	>200	>200	>200	>200	>200	>200
4n	25	50	100	100	>200	50
4 0	25	50	100	50	>200	50
4 p	>200	>200	>200	>200	>200	>200
4 q	100	100	200	100	>200	100
4r	100	100	200	100	>200	100
4 s	100	100	100	50	>200	100
4t	50	50	100	25	>200	50
4u	25	25	50	25	100	25
4 v	100	50	200	50	>200	50
4 w	25	25	50	25	200	25
4x	25	12.5	25	25	50	25
F.S. ^{<i>b</i>}	100	50	25	12.5	6.25	50
A.S. ^{<i>b</i>}	3.125	50	200	1.56	3.125	3.125
C.S. ^{<i>b</i>}	1.56	25	25	3.125	1.56	3.125

Table 1. In vitro antibacterial activities of compounds 3a–3h and 4a–4x (MIC, nmol/mL).^a

^{*a*}MIC = Minimum inhibitory concentration, MRSA = methicillin-resistant *S. aureus*.

^bPositive controls, F.S. = Fosfomycin sodium, A.S. = Ampicillin sodium, C.S. = Cefotaxime sodium.

Compd 5a		Gram-pos	itive bacteri	a	Gram-negative bacteria
5a	S. aureus	MRSA	B. cereus	B. subtilis	R. solanacearum
	100	200	200	75	150
5b	50	100	50	50	37.5
5c	25	37.5	37.5	25	25
5d	200	200	200	100	150
5e	200	200	200	75	150
5 f	75	75	100	50	50
5g	50	50	75	25	25
5h	100	>200	200	100	100
5i	100	200	200	200	100

Fable 2. In vitro antibacteria	l activity of compounds	5a–5i (MIC, nmol/mL).
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Comnd			MIC (µ	lg/mL)		Cell Via	ability $(\%)^a$
Compa	S. aureus	MRSA	B. cereus	B. subtilis	R. solanacearum	L02	MARC 145
4f	8	8	8	16	16	71.6±1.2	93.2±0.7
4i	8	8	16	8	8	76.2±1.8	87.1±1.6
41	8	8	16	8	8	74.3±0.8	88.5±1.5
4 u	8	8	16	8	8	85.7±1.4	96.3±0.9
4w	8	8	16	8	8	82.2±2.1	98.5±0.5
4x	8	4	8	16	8	84.6±1.3	103.2±2.4
5c	8	16	16	16	8	73.5±0.9	94.8±0.6
F.S.	16	8	4	2	8	-	-
A.S.	2	16	128	1	2	-	-
C.S.	1	16	16	2	2	-	-

Table 3. In vitro antibacterial and cytotoxicity activity of some potential compounds.

^{*a*}Cytotoxicity assay with potential compounds (8 μ g/mL), "-" = No test.

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Compd	LOP^{a}	$TPSA^b$	nON ^c	$nONNH^{d}$	nrotb ^e
4f	2.56	55.22	5	1	7
4i	1.50	32.82	3	1	4
41	1.54	42.05	4	1	5
4 u	2.25	38.14	4	1	4
4w	1.84	47.38	5	1	4
4 x	2.28	47.38	5	1	5
5c	2.48	55.22	5	1	6

Tuble in Culculated Indiceation properties of Some potential compounds

^{*a*}LOP = octanol–water partition coefficient, ^{*b*}TPSA = topologic polar surface area, ^{*c*}nON = number of hydrogen bond acceptors, ^{*d*}nONNH= number of hydrogen bond donors, ^{*e*}nrotb = number of rotatable bonds.



Fig. 1. Design of the quaternization harman analogues.



Fig. 2. Synthesis of compounds 3a–3h and 4a–4x.





Fig. 4. Crystal structures of compounds 3e and 5f.

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Fig. 5. Structure-activity relationships of the quaternization harman analogues against bacteria.

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Fig. 6. SEM of MRSA and *R. solanacearum*. (A) and (C) are untreated. (B) and (D) are treated with compound **4x**.



Fig. 7. Three-dimensional conformations of compound 4x docked in bacterial type II topoisomerase complex.





Fig. 8. Representative photographs of compound 4x in the *in vivo* antibacterial assay: (left) untreated and (right) treated.

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Highlights

- Thirty-three new quaternization harman analogues were synthesized and the peak MIC was 4 µg/mL.
- The structure–activity relationships were summarized.
- The compound **4x** showed low cytotoxicity, good thermal stability and "drug-like" properties.
- The compound **4x** could damage the bacterial cell membrane and wall, and disrupt the function of type II topoisomerase.