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One-Pot Synthesis of Phenylglyoxylic Acid from Racemic Mandelic Acids via Cascade Biocatalysis

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1 One-Pot Synthesis of Phenylglyoxylic Acid from Racemic

2 Mandelic Acids via Cascade Biocatalysis

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19 ABSTRACT

Phenylglyoxylic acid (PGA) are key building blocks and widely used to synthesize 20 pharmaceutical intermediates or food additives. However, the existing synthetic methods for 21 22 PGA generally involve toxic cyanide and complex processes. In order to explore an alternative method for PGA biosynthesis, we envisaged cascade biocatalysis for the one-pot 23 synthesis of PGA from racemic mandelic acid. A novel mandelate racemase named ArMR 24 showing higher expression level (216.9 U·mL⁻¹ fermentation liquor) was cloned from 25 Agrobacterium radiobacter and identified. And six recombinant E. coli strains were 26 engineered to co-express three enzymes of mandelate racemase, D-mandelate 27 28 dehydrogenase and L-lactate dehydrogenase, and transform racemic mandelic acid to PGA. 29 Among them, the recombinant E. coli TCD 04, engineered to co-express three enzymes of ArMR, LhDMDH and LhLDH, can transform racemic mandelic acid (100 mM) to PGA 30 31 with 98% conversion. Taken together, we provide a green approach for one-pot biosynthesis of PGA from racemic mandelic acid. 32

33 KEYWORDS: mandelate dehydrogenase, mandelate racemase, cascade reaction,
 34 phenylglyoxylic acid, whole cell catalysis

36 INTRODUCTION

Phenylglyoxylic acid (PGA) are key building blocks in the chemical synthesis process, and 37 widely used to synthesize pharmaceutical intermediates or food additives.¹⁻⁴ Compared with 38 39 chemical preparation generally involve toxic cyanide and complex processes,^{2, 5} microbial or enzymatic biosynthesis of PGA provides an eco-friendly approach to industrial-scale 40 production.^{2, 6-7} It is reported that many microorganisms can selectively transform mandelic 41 acid to PGA, including Gibberella fujikuroi, Alcaligenes bronchisepticus, Pseudomonas 42 polycolor, and Pseudomonas putida.⁸⁻⁹ D-mandelate dehydrogenase (DMDH) plays a key 43 role in the process of converting D-mandelic acid to PGA, and it is a member of 44 ketopantoate reductase family.^{5, 9-10} Heterologous expression of DMDH genes from many 45 microorganisms in E. coli was capable to oxidize mandelic acid into PGA.^{4, 10} In our 46 previous study, the PGA yield was about 40% from 50 mM racemic mandelic acid 47 catalyzed by co-culturing two engineered E. coli strains.⁴ Meanwhile, we have verified that 48 the NAD⁺ regeneration capacity could be enhanced by utilizing the pyruvic acid based on 49 co-culturing two engineered E. coli strains. However, the residual mandelic acid 50 significantly increases the cost of product purification. Therefore, enhancing the 51 transformation efficiency of substrates can not only improve the utilization rate of raw 52 materials, but also reduce the cost of purification. 53

54 Mandelate racemase (EC 5.1.2.2) is well-characterized racemases¹⁰⁻¹¹ and an ideal 55 candidate for the racemization of non-natural α -hydroxycarboxylic acids under 56 physiological reaction conditions, which had been applied in deracemization protocols in 57 connection with a kinetic resolution step.¹² So far, the *Pp*MR from *Pseudomonas putida*

ATCC 12633 (Uniprot No. P11444) is well-characterized and widely-applied mandelate 58 racemase, which k_{cat} values towards D- mandelic acid and L- mandelic acid were 822 and 59 756 s⁻¹. ¹²⁻¹³ And many cascade reactions had been designed and applied using the PpMR as 60 the core catalyst.^{10, 14-15} However, the industrial application of *Pp*MR was impeded due to 61 its poor soluble expression level. It had been reported that genome mining could be an 62 economical strategy in discovering novel enzyme with higher expression level,^{4, 16-18} 63 therefore it should also be used to discover novel mandelate racemase with higher 64 expression level. In addition, the mature DNA synthesis and sequencing techniques have 65 enabled the overproduction of a desired enzyme in microorganisms. 66

In this work, we tried to discover novel mandelate racemases with higher expression level, and achieve a eco-friendly synthesis of PGA from racemic mandelic acid via one-pot cascade biocatalysis (**Scheme 1**) combined with the previous research basis.³ As known to us, the PGA is more expensive than racemic mandelic acid, thus the process of converting racemic mandelic acid to PGA is highly value-added and attractive. Taken together, our results will build a solid basis for future studies, including chiral resolution of racemic mandelic acid, biosynthesis of PGA derivatives and so on.

74 MATERIALS AND METHODS

75 Reagents and Kits

PrimeSTAR[®] HS (Premix), DNA Marker and DNA Ligation Kit Ver.2.1 were purchased
from TaKaRa (Dalian,China). Restriction enzymes *Eco*R I, *Bam*H I, *Sal* I, *Hind* III, *Nde* I
and *Xho* I were purchased from New England Biolabs, Inc. (Ipswich, MA, England). *D*-mandelic acid, *L*-mandelic acid, racemic mandelic acid, *L*-lactic acid and PGA were

80 purchased from Aladdin (Shanghai, China). NAD⁺, NADP⁺, NADH and NADPH were purchased from Bontac-bio (Shenzhen, China). BCA-200 protein assay kit, IPTG and 81 protein markers were purchased from Sangon (Shanghai, China). Axygen[®] AxyPrep[™] PCR 82 83 Clean-Up Kit, Axygen[®] AxyPrep[™] DNA Gel Extraction Kit and Axygen[®] AxyPrep[™] Plasmid Miniprep Kit were purchased from Corning (New York, United States). EasyPure 84 Bacteria Genomic DNA Kit was purchased from TransGen Biotech (Beijing, China). All 85 86 other chemicals were of analytical grade and purchased from Solarbio (Beijing China). Strains, Plasmids and Culture Media 87

Escherichia coli BL21, E. coli DH5a/pACYCDuet-1, E. coli DH5a/pET28a, E. coli BL21/ 88 pET28a-LhDMDH⁴, E. coli BL21/ pET28a-PpMR¹⁰ and E. coli BL21/ pET28a-LbDMDH¹⁰ 89 90 preserved by our Lab were cultured in LB medium.⁴ E. coli BL21/ pET28a-LhLDH 91 contained an L-lactate dehydrogenase encoding gene from Lactobacillus helveticus, which 92 was artificially synthesized and constructed by Synbio Technologies Genes for Life after codon optimization. Agrobacterium radiobacter DSM 30147, Bradyrhizobium japonicum 93 CGMCC 1.2550, Clostridium ljungdahlii DSM 13528, Halomonas sp. S23321 and 94 Herbaspirillum rubrisubalbicans ATCC 19308 preserved by our group were used for 95 genomic DNA extraction after culturing in LB medium. 96

97 Genome Mining for Putative Mandelate Racemases

PpMR, a well-characterized mandelate racemase with high specific activity encoded by *Pseudomonas putida* ATCC 12633 (Uniprot No. P11444), was used as probe for the BLAST search for potential novel mandelate racemases, which were mainly from bacterial genome sequences and unidentified. In order to ensure the diversity of sequences, candidate sequences with the similarity of probe sequences between 20% and 65% were selected. And

103 the functions of these sequences were not commented and verified in detail.

104 Gene Cloning and Expression of Selected Mandelate Racemases

105 Based on the results of genome mining, many paired primers shown in Table S1 were designed and synthesized, which were used to amplify the gene fragments of potential novel 106 mandelate racemases. The cultivation, collection and genomic DNA extraction of original 107 strains were performed as reported before.⁴ Nextly, the gene fragments of potential 108 mandelate racemase were respectively amplified based on these genomic DNA. The 109 amplified genes were transformed into E. coli BL21 by the pET28a plasmid, and those 110 recombinant transformants were identified by DNA sequencing. The expressions of these 111 transformants were performed as reported before,⁴ and then the lysate supernatant was 112 purified to electrophoretic purity by Ni-chelating affinity chromatography.⁴ 113

114 **E**

Enzyme Activity and Protein Assays

The oxidation activity of *D*-mandelate dehydrogenase on *D*-mandelic acid and reduction 115 activity of L-lactate dehydrogenase on sodium pyruvate were assayed as reported before.⁴ 116 The racemization activity of mandelate racemase on L-mandelic acid was assayed by 117 determining the production of D-mandelic acid. 1 mL reaction mixture contained 100 mM 118 Tris-HCl buffer, pH 7.5, 3.3 mM Mg²⁺, proper quantity of enzyme and 2.0 mM L-mandelic 119 120 acid. The conversion reaction was maintained at 30 °C and 1000 r/min for 30 min. Then the reaction was stopped by adding 20 µL 2 M sulfuric acid, and extracted with equivoluminal 121 ethyl acetate, and the production of D-mandelic acid or residual quantity of L-mandelic acid 122 was determined by HPLC analysis in organic phase using a Shimadzu LC2010A HPLC 123

system with a UV detector (working wavelength 228 nm) at 30 °C, equipped with a CHIRALCEL® OJ-H (250 mm × 4.6 mm, 5 μ m, DAICEL INVESTMENT CO., LTD.). The mobile phase contained 92% hexane, 8% isopropanol and 0.1% TFA, and the flow rate of mobile phase was 1.0 mL min⁻¹. Retention time of *D*-mandelic acid was 19.9 min (**Figure S1**), and the one of *L*-mandelic acid was 23.0 min (**Figure S2**).

The protein concentration and SDS-PAGE were carried out as reported before.⁴ Then the apparent molecular weight of subunit was estimated with the Quantity One software based on the standard marker proteins.¹⁶ Meanwhile, the actual molecular weights of these recombinant enzymes were determined by mass spectrometric analysis at Xevo G2-XS Q-TOF. And the subunit numbers of these enzymes were calculated by comparing the apparent molecular weight and actual molecular weight.

Biochemical Characterization of Recombinant Mandelate Racemases

136 To estimate the temperature optima of the recombinant mandelate racemases, their activities were determined under the standard assay conditions as above, except temperatures ranging 137 from 20 to 60°C with an interval of 5 °C. To estimate the thermostability of these 138 recombinant mandelate racemases, they were incubated with 100 mM, pH 7.5 HEPES 139 buffer at 30, 40 and 50 °C, respectively, and the residual enzyme activity was determined by 140 141 sampling at different time. Then, the half-lives of these recombinant mandelate racemases 142 were calculated. The pH optimum of the mandelate racemase was assayed by the standard activity assay method with 100 mM glycine-NaOH buffer over the pH range of 8.0–10.5, 20 143 mM citric acid-sodium citrate buffer over the pH range of 5.0-6.0 and 100 mM Tris-HCl 144 buffer over the pH range of 6.5-7.5. To estimate influence of metal ions and EDTA on 145

mandelate racemase, the activity of mandelate racemase was assayed by the standard
activity assay method with various metal ions and EDTA at a final concentration of 1.0 mM.
The kinetic parameters of the mandelate racemases were determined by assaying the activity
on various substrate concentrations. The correlation curve was fitted in nonlinear fitting by
the Origin 9.0 software.^{4, 19}

151 Genetic Engineering of the Plasmids Containing DMDH and LDH Encoding Gene

152 Based on the analysis of multiple cloning sites of plasmid and restriction enzyme sites of the target gene, many paired primers shown in **Table S2** were designed and synthesized, which 153 were used to amplify the gene fragments of D-mandelate dehydrogenase and L-lactate 154 dehydrogenase. LhLDH was amplified from the strain of E. coli BL21/ pET28a-LhLDH 155 156 using primers LhLDH-F and LhLDH-R with PrimeSTAR[®] HS (Premix) Kit. The PCR product was double-digested with BamH I and Sal I, and then ligated to the BamH I and Sal 157 158 I digested MCS1 of pACYCDuet-1 plasmid with DNA Ligation Kit Ver.2.1. The ligation product was transformed into E. coli BL21 expression competent cells to give 159 pACYC-LhLDH. Then, the LbDMDH and LhDMDH was respectively amplified, 160 double-digested and then ligated to the Nde I and Xho I digested MCS2 of pACYC-LhLDH 161 plasmid to give pACYC-LhLDH-LbDMDH and pACYC-LhLDH-LhDMDH using similar 162 methods. 163

164 Engineering of *E. coli* (TCD01–06) Strains

165 Each of the *LDH-DMDH* plasmids (pACYC-*LhLDH-LbDMDH* and 166 pACYC-*LhLDH-LhDMDH*) and each of the *MR* plasmids (pET28a-*PpMR* and the 167 recombinant plasmid constructed as above) were co-transformed into competent cells of *E*.

168	coli BL21 (DE3) to give the six E. coli strains co-expressing DMDH, MR and LDH. The
169	details are provided in Table S3 of the Supporting Information.

170 'One-pot' Biosynthesis of PGA from Racemic Mandelic Acid

171 To evaluate application potential of the engineered *E. coli* strains in transforming racemic mandelic acid to PGA, these engineered E. coli strains were cultured and induced as above. 172 And then, 100 mM racemic mandelic acid was added and transformed for 48 h at 30°C. 173 174 During this transformation process, 100 µL samples were extracted for HPLC analysis every 4 hours.⁴ After the catalytic reaction was terminated, the purification of biotransformation 175 products was performed as reported before.⁴ And then, the product yield was determined by 176 HPLC analysis as above, and the product was characterized by ¹H NMR spectroscopy at 177 178 400 MHz.

179 Analysis of PGA

PGA was determined by HPLC analysis and ¹H NMR spectroscopy at 400 MHz as reported
before.⁴ Retention time of PGA was 3.1 min. The correlation curve was fitted in linear
fitting by the Origin 9.0.

183 RESULTS AND DISCUSSION

184 Gene Cloning and Expression of Mandelate Racemases

To search for an efficient mandelate racemase for deracemization of racemic mandelic acid, genome mining strategy was adopted. Ten gene fragments of potentially novel mandelate racemases were amplified and detected by agarose gel electrophoresis as shown in **Figure S3**, and then they have been cloned and over-expressed in *E. coli* BL21. To detect recombinant protein expression patterns, SDS-PAGE was carried out on a 12.5% gel, and its 190 results were shown in Figure 1. As shown in Figure 1, eight of them were successfully expressed except BiMR-B and BiMR-F. Interestingly, two of them displayed racemization 191 activity toward L-mandelic acid. Among them, a putative mandelate racemase (UNIPROT 192 193 accession no: B9JQ28) from Agrobacterium radiobacter, herein designated as ArMR, was confirmed with the highest racemization activity. It was examined and converted that the 194 ArMR had the racemization activity on L-mandelic acid of 216.9 U·mL⁻¹ fermentation 195 196 liquor, which was near to four times that of the probe. Meanwhile, another putative mandelate racemase (GenBank accession no: WP 058894565.1) from Herbaspirillum 197 rubrisubalbicans, herein designated as HrMR, was confirmed with racemization activity of 198 199 33.7 U·mL⁻¹ fermentation liquor, which was slightly lower than that of the probe. In 200 addition, as shown in Figure 1C, the soluble expression level of ArMR using pET28a plasmid was significantly higher than that of PpMR. Sequence alignment revealed a 201 202 relatively low amino acid sequence identity (about 60%) with the PpMR (Figure S4), indicating that they are novel mandelate racemases. 203

In order to characterize the enzymatic properties of the recombinant mandelate 204 racemases, they were firstly purified to homogeneity through Ni-NTA affinity 205 chromatography. As shown in Figure 1D, they were purified to homogeneity, and their 206 apparent molecular weights were about 43.0 kDa. Based on the results of high resolution 207 208 mass spectrometer analysis, the actual molecular weight of *PpMR*, *ArMR* and *HrMR* was 209 308460.8, 154811.2 and 157253.2 Da, respectively. Comparing with the apparent molecular 210 weight and actual molecular weight, the ArMR and HrMR were all comprised of four subunits, while the *Pp*MR was comprised of eight subunits as previously reported.²⁰ 211

The specific racemization activity of purified *Ar*MR and *Hr*MR was 107.4 and 21.5 U·mg⁻¹ for the *L*-mandelic acid, respectively. They were slightly lower than the one of the *Pp*MR determinated in this study, which specific racemization activity toward *L*-mandelic acid was 324.8 U·mg⁻¹. Nevertheless, the *Ar*MR can still reflect a higher level of enzyme activity in fermentation liquor owe to its higher proportion of soluble expression comparing with the probe, which further indicated that the racemases obtained, especially the *Ar*MR, has a great application prospect in deracemization of racemic mandelic acid.

219 Comparative Enzymatic Properties of Mandelate Racemases

220 In order to compare the enzymatic properties of these recombinant mandelate racemases, 221 they were investigated using L-mandelic acid as the model substrate. The temperature 222 properties of these mandelate racemases were shown in Figure 2. These purified mandelate racemases exhibited higher catalytic activity from 45 to 55°C, and the temperature optimum 223 224 of ArMR was 50°C, while the one of HrMR was 55°C. The half-life of these recombinant mandelate racemases at various temperatures was summarized in Table 1. Unfortunately, 225 the half-life of ArMR at 50 °C was 0.17 h, and its thermostability was too bad to be 226 impeded in enzymatic racemization of racemic mandelic acid. Hence, to eliminate the 227 influence of bad thermostability, whole cell catalysis was preferred at 30 °C in enzymatic 228 racemization of racemic mandelic acid. In addition, with the emergence of various 229 230 bioinformatical software, the computer-aided rational design is playing a more significant role in molecular modification of thermostability. Particularly, molecular dynamics 231 simulation has become more mature and reliable.^{3, 21} In next research, the molecular 232

233 modification of *Ar*MR for improving its thermostability will be carried out by *in silico*234 design.

235 The pH properties of these mandelate racemases were shown in Figure 3. The pH 236 value of buffer can significantly influence the activities of these enzymes as shown in Figure 3. These purified mandelate racemases had no racemization activity at pH value of 6 237 or below, which could be attributed to that mandelate racemase catalyzes deracemization of 238 substrate by a two-base mechanism.²² Proton gains and losses are catalyzed by two typical 239 alkaline amino acid residues, so it is speculated that the more acidic environment is not 240 241 conducive to proton exchange and significantly reduces the catalytic activity. In the alkaline 242 pH range, these enzymes activities were higher, and even in the glycine - NaOH buffer with 243 a pH value of 10, it still retained more than 10% of the relative activity. And the pH optimum of the ArMR was 8.0, while the one of the HrMR was 9.0. 244

245 The kinetic parameters of these mandelate racemases toward L-mandelic acid were summarized in **Table 2**. The K_m toward *D*-mandelic acid and *L*-mandelic acid of *Ar*MR was 246 1.44 and 0.81 mM, which was slightly higher than the one of the probe, ¹³ indicating that 247 248 ArMR displayed lower affinity toward D-mandelic acid and L-mandelic acid. Moreover, the k_{cat} values toward D-mandelic acid and L-mandelic acid of ArMR were 409.8 and 218.3 s⁻¹, 249 which were also distinctly lower than the ones (822 and 756 s⁻¹) of the probe. ^{13, 20} 250 251 Nevertheless, the ArMR can still reflect a higher level of enzyme activity in fermentation 252 liquor attribute to its higher proportion of soluble expression comparing with the probe, which further indicated that the ArMR could display further advantage in the 253 deracemization process of racemic mandelic acid by whole-cell catalysis. Meanwhile, it also 254

255 means that the catalytic activity of *Ar*MR still needs to be improved on the premise of 256 ensuring the proportion of soluble expression.

It is reported that the crystal structure of PpMR contained a Mg²⁺ in the catalytic center, 257 258 which is connected to the side chain of catalytic triad in the active center and played an important role in the catalytic process.²³ In this paper, we investigated the influence of Mg²⁺ 259 and other metal ions on the activity of mandelate racemase (Figure 4). As shown in Figure 260 261 4, there is no effect of the tested metal ions and EDTA on enzyme activity but the activity of racemase was improved by adding Mg^{2+} and Zn^{2+} . These results indicated that Mg^{2+} and 262 Zn^{2+} also played an important role in the catalytic processes of ArMR and HrMR, and they 263 might have the same catalytic mechanism as the PpMR, which would be further verified in 264 265 our subsequent studies.

266 Engineering and Expressing of *E. coli* (TCD01–06) Strains

267 Based on the techno-economic impact of reducing addition of coenzyme and cosubstrate, and enhancing regeneration of NAD⁺ and conversion efficiency of substrate, we 268 investigated the feasibility of co-expressing three relevant enzymes in an E. coli cell to 269 Scheme 270 achieve the purpose of 1. Each of the *LDH-DMDH* plasmids (pACYC-*LhLDH-LbDMDH* and pACYC-*LhLDH-LhDMDH*) and each of the *MR* plasmids 271 272 (pET28a-PpMR, pET28a-ArMR and pET28a-HrMR) were co-transformed into competent cells of E. coli BL21 (DE3) to give the six E. coli strains co-expressing DMDH, MR and 273 LDH. In the co-expression system, DMDH was dedicated for converting D-mandelic acid to 274 PGA, and LDH was designated to enhancing the regeneration of NAD⁺ by utilizing the 275 pyruvic acid generated by E. coli. Meanwhile, the MR was dedicated for converting 276

277 L-mandelic acid to D-mandelic acid, when the D-mandelic acid had been converted to PGA. After DNA sequencing identification, all of these E. coli strains were respectively named as 278 E. coli TCD 01 to E. coli TCD 06, the details of these E. coli strains are shown in Table S3 279 280 of the Supporting Information. To detect the expression level of co-expression system, catalytic activities of three enzymes were analyzed and summarized in Table 3. As shown 281 in Table 3, the three enzymes activities of recombinant E. coli TCD 04, engineered to 282 283 co-express three enzymes of ArMR, LhDMDH and LhLDH, were the highest and best proportion, which may be more conducive to the cascade reaction. 284

285 'One-pot' Biosynthesis of PGA from Racemic Mandelic Acid

286 To assess application potential of the engineered E. coli strains in transforming racemic 287 mandelic acid to give PGA, these engineered E. coli strains were cultured and induced as above. Then, racemic mandelic acid was added to a final concentration of 100 mM, and 288 289 transformed 48 h at 30°C. The racemic mandelic acid was bioconverted 48 h by engineered E. coli whole-cell, and the product was purified to chromatographically purity, which purity 290 was over 99%. And the retention time of the purified product was 3.1 min, which was in 291 accordance with the retention time of PGA. Meanwhile, combined with ¹H NMR spectra of 292 racemic mandelic acid (Figure S5) and the product (Figure S6), these results sufficiently 293 demonstrated that the racemic mandelic acid had been converted to PGA by one-pot cascade 294 295 biocatalysis. These results also indicated that the PGA had been successfully synthesized without adding coenzyme or cosubstrate, implying that the synthesis cost of PGA can be 296 significantly reduced by this strategy, which fully demonstrated the advantage of whole cell 297 catalysis. 24-26 298

299	In addition, the conversion curves of racemic mandelic acid catalyzed by these
300	engineered E. coli strains were measured and analyzed. As shown in Figure 5, the yield of
301	PGA assisted by the mandelate racemase was significantly higher than the one that was only
302	catalyzed by D-mandelate dehydrogenase and L-lactate dehydrogenase. ⁴ Among the strains,
303	the E. coli TCD04 containing pET28a-ArMR and pACYC-LhLDH-LhDMDH gave the
304	highest conversion to PGA, thus it was selected as the best catalyst for this cascade
305	biotransformation. Under the optimized conditions, 100 mM of racemic mandelic acid were
306	reacted with E. coli TCD04 to give PGA in 98% conversion in 48 h, and its space-time yield
307	was up to 7.36 g L ⁻¹ d ⁻¹ . Compared with the results we reported earlier, ⁴ the significantly
308	increased yields of PGA should mainly be attributed to that the co-expressed mandelate
309	racemase can continuously convert L-mandelic acid to D-mandelic acid, and maintain the
310	reaction so that the substrate can be transformed as much as possible. Moreover,
311	co-expression can avoid intercellular transmission of NAD ⁺ and NADH in regeneration
312	NAD^+ comparing with the co-culture mode, which can enhance the efficiency of NAD^+
313	regeneration. It was hardly surprising that these E. coli strains containing PpMR encoding
314	genes failed to show better productivity, although the $PpMR$ can display excellent kinetic
315	parameters. This result is consistent with that the $PpMR$ did not reflect a higher level of
316	enzyme activity in fermentation liquor due to its higher proportion of inclusion bodies in
317	recombinant E. coli cells, which further confirmed that the ArMR could display further
318	advantage in the racemization process of racemic mandelic acid by whole-cell catalysis.
319	Furthermore, these E. coli strains containing LbDMDH encoding genes showed lower
320	productivity than those E <i>coli</i> strains containing Lh DMDH which should be due to the

320 productivity than those *E. coli* strains containing *Lh*DMDH, which should be due to the

lower catalytic activity and soluble expression proportion of *Lb*DMDH expressed in *E. coli* cells. Meanwhile, this result also indicated that the mandelate dehydrogenase was the rate-limiting enzyme in the whole cascade reaction for the one-pot synthesis of PGA from racemic mandelic acid. All of these results showed that the engineered *E. coli* strain TCD 04 had tremendous application potential in chiral resolution of racemic mandelic acid and biosynthesis of PGA, which are also important intermediates for synthesis of pharmaceutical and natural products.^{4, 27-28}

In conclusion, a novel mandelate racemase (ArMR) from A. radiobacter was mined and 328 expressed in E. coli BL21. The ArMR displayed high catalytic activity and soluble 329 330 expression proportion. Interestingly, we achieved a green and efficient synthesis of PGA 331 from racemic mandelic acid via one-pot cascade biocatalysis using the engineered E. coli 332 strain coexpressing three enzymes of mandelate racemase, D-mandelate dehydrogenase and 333 L-lactate dehydrogenase. Among of these, the recombinant E. coli TCD 04, engineered to coexpress three enzymes of ArMR, LhDMDH and LhLDH, can transform 100 mM racemic 334 335 mandelic acid to give PGA with 98% conversion. The high-yielding synthetic methods use cheap and green reagents, and E. coli whole-cell catalysts, thus providing green and useful 336 alternative methods for manufacturing PGA. Taken together, we provide a green approach 337 for one-pot biosynthesis of PGA from racemic mandelic acid. 338

339 ASSOCIATED CONTENT

340 Supporting Information

Additional experimental results and other data are available free of charge via the Internet at
 http://pubs.acs.org. Sequences of the primers used for cloning and expression of novel

343	mandelate racemases (Table S1); Sequences of the primers used for genetic engineering of
344	the plasmids containing DMDH and LDH encoding gene (Table S2); List of the
345	recombinant strains and the plasmids contained (Table S3); The HPLC analysis spectrum of
346	the D- mandelic acid (Figure S1); The HPLC analysis spectrum of the L- mandelic acid
347	(Figure S2); The PCR amplification for the ten mandelate racemase genes (Figure S3);
348	Multi-sequence alignment of the representative mandelate racemases (Figure S4); The ¹ H
349	NMR analysis spectrum of the racemic mandelic acid (Figure S5); The 1H NMR analysis
350	spectrum of the product (Figure S6).

351 AUTHOR INFORMATION

352 Authors' Contributions

353 Cun-Duo Tang, Peng-Ju Ding and Hong-Ling Shi contributed equally to this work as the

354 first author.

355 Notes

356 The authors declare that they have no competing interests.

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366 REFERENCES

He, Y. C.; Pan, X. H.; Xu, X. F.; Wang, L. Q., Biosynthesis of benzoylformic acid from benzoyl cyanide
 with a new bacterial isolate of Brevibacterium sp. CCZU12-1. *Appl Biochem Biotechnol* 2014, *172* (6),
 3223-33.

- He, Y.-C.; Zhou, Q.; Ma, C.-L.; Cai, Z.-Q.; Wang, L.-Q.; Zhao, X.-Y.; Chen, Q.; Gao, D.-Z.; Zheng, M.;
 Wang, X.-D.; Sun, Q., Biosynthesis of benzoylformic acid from benzoyl cyanide by a newly isolated
 Rhodococcus sp. CCZU10-1 in toluene–water biphasic system. *Bioresource Technology* 2012, *115* (Supplement C), 88-95.
- Tang, C.; Shi, H.; He, Z.; Ding, P.; Jiao, Z.; Kan, Y.; Yao, L., Green biosynthesis of phenylglyoxylic acid
 by biotransformation using recombinant Escherichia coli whole cells. *CIESC Journal* 2018, 69 (6), 2627-2631.
- 376 4. Tang, C. D.; Shi, H. L.; Xu, J. H.; Jiao, Z. J.; Liu, F.; Ding, P. J.; Shi, H. F.; Yao, L. G.; Kan, Y. C.,

Biosynthesis of Phenylglyoxylic Acid by LhDMDH, a Novel d-Mandelate Dehydrogenase with High Catalytic
 Activity. *J Agric Food Chem* 2018, 66 (11), 2805-2811.

- 5. Li, D.; Zeng, Z.; Yang, J.; Wang, P.; Jiang, L.; Feng, J.; Yang, C., Mandelate racemase and mandelate
 dehydrogenase coexpressed recombinant Escherichia coli in the synthesis of benzoylformate. *Biosci Biotechnol Biochem* 2013, 77 (6), 1236-9.
- Bornscheuer, U. T.; Huisman, G. W.; Kazlauskas, R. J.; Lutz, S.; Moore, J. C.; Robins, K., Engineering
 the third wave of biocatalysis. *Nature* 2012, *485* (7397), 185-194.
- 7. Chen, X.; Yang, C.; Wang, P.; Zhang, X.; Bao, B.; Li, D.; Shi, R., Stereoselective biotransformation of
 racemic mandelic acid using immobilized laccase and (S)-mandelate dehydrogenase. *Bioresources and bioprocessing* 2017, 4 (1), 2-2.
- Miyamoto, K.; Ohta, H., Enantioselective oxidation of mandelic acid using a phenylmalonate
 metabolizing pathway of a soil bacteriumAlcaligenesbronchisepticus KU 1201. *Biotechnology Letters* 1992,
 14 (5), 363-366.
- 390 9. Wang, J.; Feng, J.; Li, W.; Yang, C.; Chen, X.; Bao, B.; Yang, J.; Wang, P.; Li, D.; Shi, R.,
- Characterization of a novel (R)-mandelate dehydrogenase from Pseudomonas putida NUST506. *Journal of Molecular Catalysis B: Enzymatic* 2015, *120*, 23-27.
- 393 10. Fan, C. W.; Xu, G. C.; Ma, B. D.; Bai, Y. P.; Zhang, J.; Xu, J. H., A novel D-mandelate dehydrogenase 394 used in three-enzyme cascade reaction for highly efficient synthesis of non-natural chiral amino acids. *J*
- *Biotechnol* **2015**, *195*, 67-71.
- 396 11. Gu, J.; Liu, M.; Guo, F.; Xie, W.; Lu, W.; Ye, L.; Chen, Z.; Yuan, S.; Yu, H., Virtual screening of
- mandelate racemase mutants with enhanced activity based on binding energy in the transition state. *Enzyme Microb Technol* 2014, 55, 121-7.
- 399 12. Felfer, U.; Goriup, M.; Koegl, M. F.; Wagner, U.; Larissegger-Schnell, B.; Faber, K.; Kroutil, W., The
- 400 Substrate Spectrum of Mandelate Racemase: Minimum Structural Requirements for Substrates and Substrate
- 401 Model. Advanced Synthesis & Catalysis 2005, 347 (7-8), 951-961.
- 402 13. Gu, J.; Yu, H., The role of residue S139 of mandelate racemase: synergistic effect of S139 and E317 on
- 403 transition state stabilization. Journal of biomolecular structure & dynamics **2012**, *30* (5), 585-93.
- 404 14. Resch, V.; Fabian, W. M. F.; Kroutil, W., Deracemisation of Mandelic Acid to Optically Pure

- 405 Non-Natural L-Phenylglycine via a Redox-Neutral Biocatalytic Cascade. *Advanced Synthesis & Catalysis*406 2010, 352 (6), 993-997.
- 407 15. Zhou, Y.; Wu, S.; Li, Z., One-Pot Enantioselective Synthesis of D-Phenylglycines from Racemic
- 408 Mandelic Acids, Styrenes, or Biobased L-Phenylalanine via Cascade Biocatalysis. Advanced Synthesis &
- 409 *Catalysis* **2017**, *359* (24), 4305-4316.
- Tang, C. D.; Shi, H. L.; Tang, Q. H.; Zhou, J. S.; Yao, L. G.; Jiao, Z. J.; Kan, Y. C., Genome mining and
 motif truncation of glycoside hydrolase family 5 endo-beta-1,4-mannanase encoded by Aspergillus oryzae
- 412 RIB40 for potential konjac flour hydrolysis or feed additive. *Enzyme Microb Technol* 2016, 93-94, 99-104.
- 413 17. Ma, B.-D.; Yu, H.-L.; Pan, J.; Liu, J.-Y.; Ju, X.; Xu, J.-H., A thermostable and organic-solvent tolerant
- 414 esterase from Pseudomonas putida ECU1011: Catalytic properties and performance in kinetic resolution of
- 415 α-hydroxy acids. *Bioresource Technology* **2013**, *133*, 354-360.
- Liang, C.; Nie, Y.; Mu, X.; Xu, Y., Gene mining-based identification of aldo-keto reductases for highly
 stereoselective reduction of bulky ketones. *Bioresources and bioprocessing* **2018**, *5* (1), 33.
- 418 19. Tang, C.-D.; Shi, H.-L.; Jiao, Z.-J.; Shi, H.-F.; Yao, L.-G.; Xu, J.-H.; Kan, Y.-C., Exploitation of
- 419 cold-active cephalosporin C acylase by computer-aided directed evolution and its potential application in
- 420 low-temperature biosynthesis of 7-aminocephalosporanic acid. Journal of Chemical Technology &
- 421 *Biotechnology* **2018**, *93* (10), 2925-2930.
- 422 20. Landro, J. A.; Gerlt, J. A.; Kozarich, J. W.; Koo, C. W.; Shah, V. J.; Kenyon, G. L.; Neidhart, D. J.;
- 423 Fujita, S.; Petsko, G. A., The role of lysine 166 in the mechanism of mandelate racemase from Pseudomonas
- 424 putida: Mechanistic and crystallographic evidence for stereospecific alkylation by (R)-.alpha.-phenylglycidate.
- 425 Biochemistry **1994**, *33* (3), 635-643.
- 426 21. Gao, S.-J.; Wang, J.-Q.; Wu, M.-C.; Zhang, H.-M.; Yin, X.; Li, J.-F., Engineering hyperthermostability
- 427 into a mesophilic family 11 xylanase from Aspergillus oryzae by in silico design of N-terminus substitution.
- 428 *Biotechnology and bioengineering* **2013**, *110* (4), 1028-1038.
- 429 22. Schafer, S. L.; Barrett, W. C.; Kallarakal, A. T.; Mitra, B.; Kozarich, J. W.; Gerlt, J. A.; Clifton, J. G.;
- 430 Petsko, G. A.; Kenyon, G. L., Mechanism of the Reaction Catalyzed by Mandelate Racemase: Structure and
- 431 Mechanistic Properties of the D270N Mutant. *Biochemistry* **1996**, *35* (18), 5662-5669.
- 432 23. Mitra, B.; Kallarakal, A. T.; Kozarich, J. W.; Gerlt, J. A.; Clifton, J. R.; Petsko, G. A.; Kenyon, G. L.,
- 433 Mechanism of the Reaction Catalyzed by Mandelate Racemase: Importance of Electrophilic Catalysis by 434 Glutamic Acid 317. *Biochemistry* **1995,** *34* (9), 2777-2787.
- 435 24. Cardillo, A. B.; Perassolo, M.; Sartuqui, M.; Rodríguez Talou, J.; Giulietti, A. M., Production of tropane
- 436 alkaloids by biotransformation using recombinant Escherichia coli whole cells. Biochemical Engineering
- 437 *Journal* **2017**, *125* (Supplement C), 180-189.
- 438 25. de Carvalho, C. C. C. R., Enzymatic and whole cell catalysis: Finding new strategies for old processes.
- 439 *Biotechnology Advances* **2011,** *29* (1), 75-83.
- 440 26. Lee, P. G.; Kim, J.; Kim, E. J.; Jung, E.; Pandey, B. P.; Kim, B. G., P212A Mutant of Dihydrodaidzein
- Reductase Enhances (S)-Equol Production and Enantioselectivity in a Recombinant Escherichia coli
 Whole-Cell Reaction System. *Appl Environ Microbiol* 2016, *82* (7), 1992-2002.
- 443 27. He, Y.-C.; Ma, C.-L.; Zhang, X.; Li, L.; Xu, J.-H.; Wu, M.-X., Highly enantioselective oxidation of
- 444 racemic phenyl-1,2-ethanediol to optically pure (R)-(-)-mandelic acid by a newly isolated Brevibacterium
- 445 lutescens CCZU12-1. Applied Microbiology and Biotechnology **2013**, 97 (16), 7185-7194.
- 446 28. Zhang, Z.-J.; Pan, J.; Li, C.-X.; Yu, H.-L.; Zheng, G.-W.; Ju, X.; Xu, J.-H., Efficient production of
- 447 (R)-(-)-mandelic acid using glutaraldehyde cross-linked Escherichia coli cells expressing Alcaligenes sp.
- 448 nitrilase. *Bioprocess and biosystems engineering* **2014**, *37* (7), 1241-1248.

449

451 Scheme and Figure Legends

- 452 Scheme 1. One-pot synthesis of PGA from racemic mandelic acids via cascade
 453 biocatalysis with recombinant *E. coli* cells containing the MR, DMDH and LDH.
- 454

455	Figure 1. The SDS-PAGE analysis for the potential recombinant mandelate
456	racemases. (A) Lane M, PageRuler Prestained Protein Ladder; lane 1, the bacteria lysate
457	supernatant of <i>E. coli</i> /pET28a-ArMR; lane 2, the bacteria lysate supernatant of <i>E</i> .
458	coli/pET28a-HrMR; lane 3, the bacteria lysate supernatant of E. coli/pET28a-BjMR-A; lane
459	4, the bacteria lysate supernatant of E. coli/pET28a-BjMR-C; lane 5, the bacteria lysate
460	sediment of <i>E. coli</i> /pET28a-ArMR; lane 6, the bacteria lysate sediment of <i>E</i> .
461	<i>coli</i> /pET28a- <i>HrMR</i> ; lane 7, the bacteria lysate sediment of <i>E. coli</i> /pET28a- <i>BjMR-A</i> ; lane 8,
462	the bacteria lysate sediment of E. coli/pET28a-BjMR-C. (B) Lane M, PageRuler Prestained
463	Protein Ladder; lane 1, the bacteria lysate supernatant of <i>E. coli</i> /pET28a- <i>BjMR-D</i> ; lane 2,
464	the bacteria lysate supernatant of E. coli/pET28a-BjMR-E; lane 3, the bacteria lysate
465	supernatant of <i>E. coli</i> /pET28a- <i>HhMR</i> ; lane 4, the bacteria lysate supernatant of <i>E</i> .
466	<i>coli</i> /pET28a- <i>ClMR</i> ; lane 5, the bacteria lysate sediment of <i>E. coli</i> /pET28a- <i>BjMR-D</i> ; lane 6,
467	the bacteria lysate sediment of E. coli/pET28a-BjMR-E; lane 7, the bacteria lysate sediment
468	of <i>E. coli</i> /pET28a- <i>HhMR</i> ; lane 8, the bacteria lysate sediment of <i>E. coli</i> /pET28a- <i>ClMR</i> . (C)
469	Lane M, PageRuler Prestained Protein Ladder; lane 1, the bacteria lysate supernatant of E.
470	<i>coli</i> /pET28a- <i>ArMR</i> ; lane 2, the bacteria lysate sediment of <i>E. coli</i> /pET28a- <i>ArMR</i> ; lane 3, the
471	bacteria lysate supernatant of E. coli/pET28a-PpMR; lane 4, the bacteria lysate sediment of
472	E. coli/pET28a-PpMR. (D) Lane M, PageRuler Prestained Protein Ladder; lane 1, the

bacteria lysate supernatant of *E. coli*/pET28a-*ArMR*; lane 2, the purified recombinant *Ar*MR;
lane 3, the bacteria lysate supernatant of *E. coli*/pET28a-*HrMR*; lane 4, the purified
recombinant *Hr*MR.

476

Figure 2. The temperature optima of the recombinant mandelate racemases. The
temperature optima were measured using the above method except the reaction temperatures
ranging from 20 to 60°C.

480

Figure 3. The pH optima of the recombinant mandelate racemases. The pH optima were assayed by the standard activity assay method as stated above with 100 mM glycine-NaOH buffer over the pH range of 8.0–10.5, 20 mM citric acid-sodium citrate buffer over the pH range of 5.0–6.0 and 100 mM Tris-HCl buffer over the pH range of 6.5–7.5.

486

Figure 4. The effect of the tested metal ions and EDTA on activity of the recombinant mandelate racemases. The activity of mandelate racemase was assayed by the standard activity assay method with various metal ions and EDTA at a final concentration of 1.0 mM.

Figure 5. The conversion curves of racemic mandelic acid catalyzed by these engineered *E. coli* strains. The reactions were carried out at 30°C and 200 rpm for 48 h in a 100-mL fermentation liquor containing 100 mM racemic mandelic acid without adding coenzyme or cosubstrate.

Tables

Table 1 The half-life of mandelate racemases at various temperatures.

	Half-life $t_{1/2}$ (h)			
Enzyme	30 °C	40 °C	50 °C	
ArMR	70.7	27.2	0.17	
HrMR	85.2	53.1	1.4	

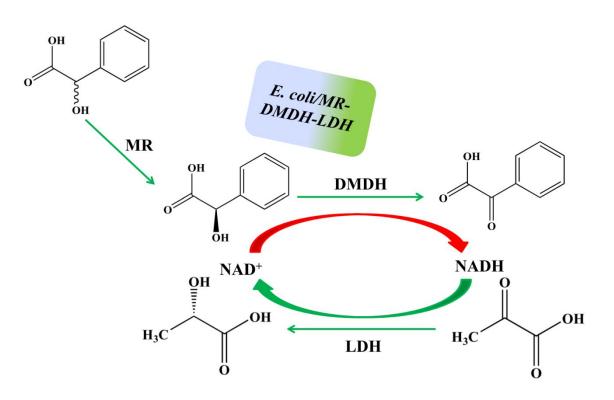
Enzyme	Substrates	K _M (mM)	k _{cat} (s ⁻¹)	V _{max} (μmol min ⁻¹ mg ⁻¹)	k _{cat} /K _M (mM ⁻¹ s ⁻¹)
<i>Ar</i> MR	D- mandelic acid	1.44 ± 0.2	409.8	582 ± 26.5	284.6
217 1011	L- mandelic acid	0.81 ± 0.1	218.3	310 ± 8.0	269.5
	D- mandelic acid	1.68 ± 0.3	80.2	102 ± 14	47.7
<i>Hr</i> MR	L- mandelic acid	1.37 ± 0.2	38.5	62 ± 8	28.1

Table 2 The kinetic parameters of mandelate racemases towards different substrates.

	Catalytic activity (U·mL ⁻¹ fermentation liquor)				
Strains	mandelate	D-mandelate	L-lactate		
	racemase	dehydrogenase	dehydrogenase		
TCD 01	40.7	25.8	194.2		
TCD 02	45.2	192.4	188.1		
TCD 03	190.4	23.2	213.8		
TCD 04	185.6	208.2	198.3		
TCD 05	36.1	26.9	210.4		
TCD 06	33.2	212.6	206.5		

Table 3 The catalytic activities of three enzymes by co-expression.

Scheme 1.



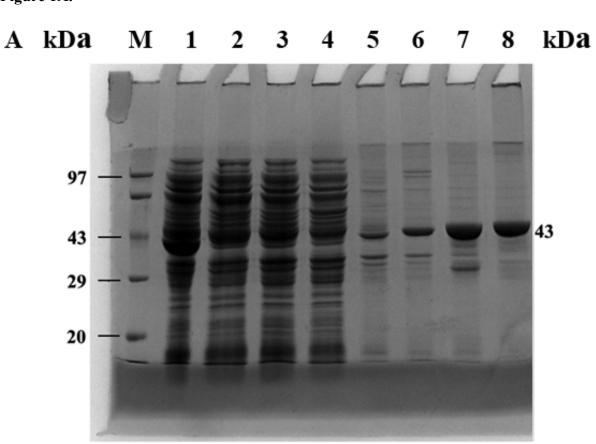
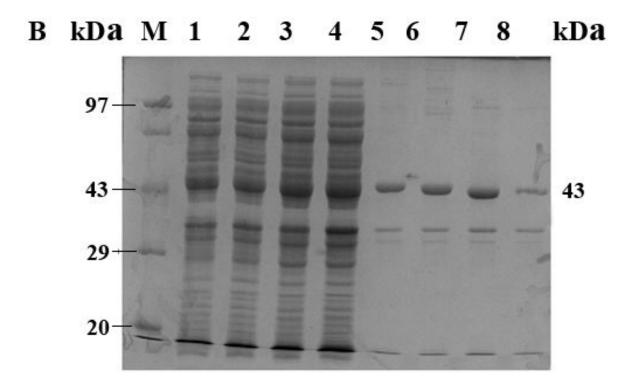


Figure 1A.





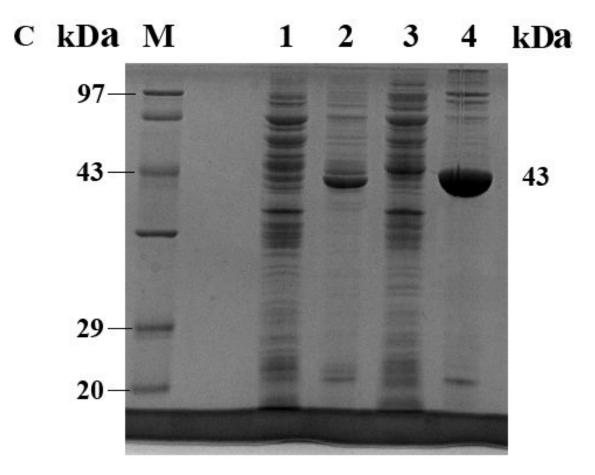
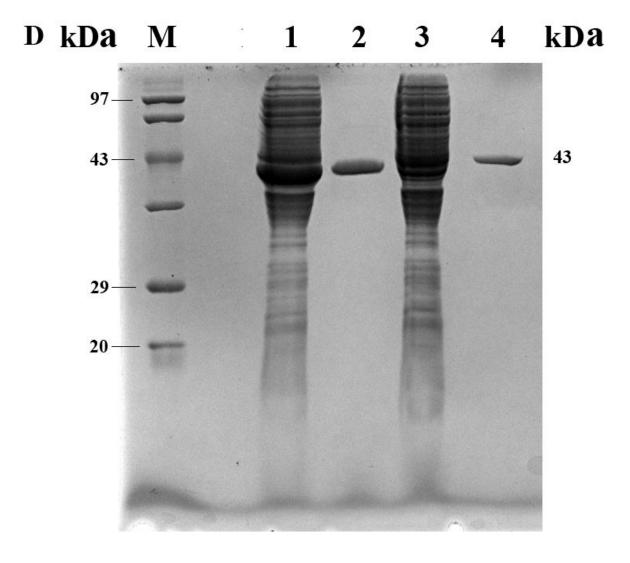


Figure 1C.







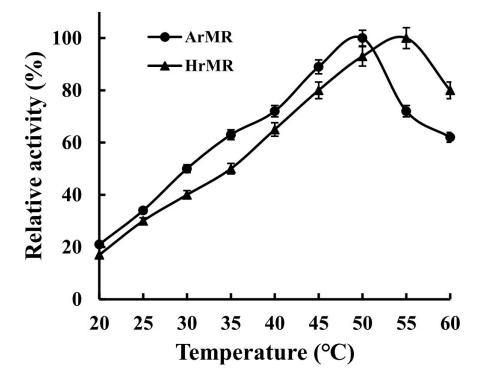
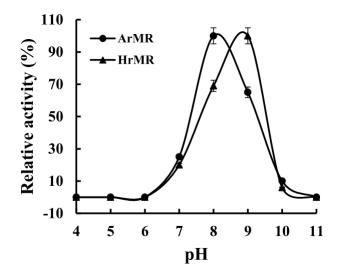
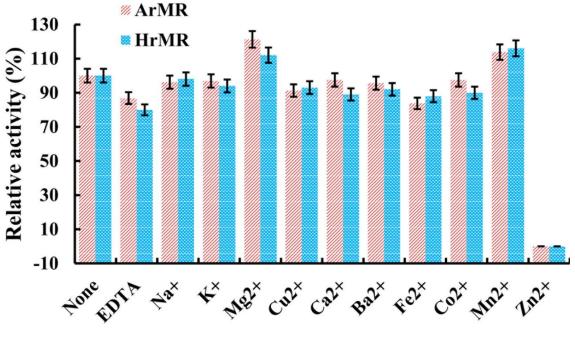


Figure 3.

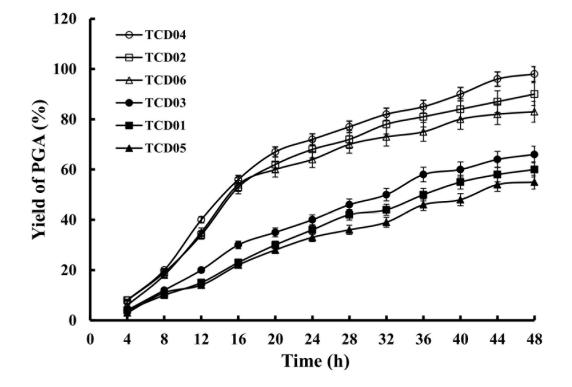






1 mM metal ions or EDTA





TOC graphic

