

A Selective Method for the Preparation of Primary Amides: Synthesis of Fmoc-L-4-Carboxamidophenylalanine and other Compounds

Wei Wang and John S. McMurray*

The University of Texas M.D. Anderson Cancer Center, Department of Neuro-Oncology,
Box 316, 1515 Holcombe Blvd., Houston, Texas, 77030, U.S.A.

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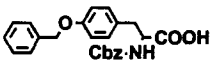
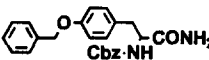
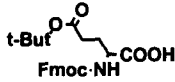
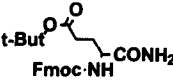
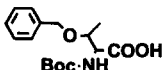
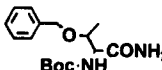
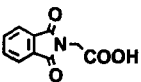
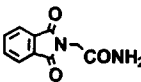
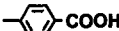
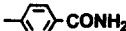
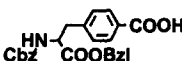
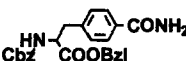
Abstract: A new method for the synthesis of primary amides was developed in which carboxylic acids were treated with ammonium chloride in the presence of peptide synthesis coupling agents and base at room temperature. These mild conditions allow the conversion of carboxyl groups to primary amides on substrates possessing sensitive functional groups such as esters and the base-labile Fmoc protecting group. © 1999 Elsevier Science Ltd. All rights reserved.

Traditional methods for the synthesis of unsubstituted carboxamides rely on the treatment of acyl halides, acid anhydrides, esters, or acids with ammonia.¹ However, there are difficulties associated with these procedures. Acylation of ammonia by acyl halides is highly exothermic and must be carefully controlled. Reaction of acid anhydrides with ammonia can be accompanied by the formation of undesired imides.² Additionally, since ammonia can react with carboxylic esters, acids, and other functional groups,³ these methods are sometimes not suitable for selective amide generation. Other modified methods have been reported: e.g. aminolysis of *N*-protected amino acid active esters with diaminomethane dihydrochloride,⁴ and treatment of carboxylic acids with 25% aqueous ammonia in presence of HOBt and DCC.⁵ For the latter, difficulties can arise from insolubility of starting materials and products or by competitive hydrolysis of the activated carboxyl group by hydroxide anion. We report a new, simple, and mild method for the conversion of carboxylic acids to primary amides using ammonium chloride, tertiary amine, and coupling agents typically used in peptide synthesis.

General procedure: Carboxylic acids were treated with PyBOP (1.5 eq.), HOBt (1.5 eq.), DIPEA (4 eq.), NH₄Cl (2 eq.) in DMF (4ml/mmol of acid) at room temperature to form the corresponding unsubstituted amides, followed by aqueous work-up. The results for the preparation of a series of amides of substrates possessing a variety of potentially labile groups are shown in Table 1. As monitored by reverse phase HPLC,⁶ starting materials were converted quantitatively to products after 0.5 h when PyBOP was used as the coupling agent.⁷ In contrast to ammonia, the base-labile Fmoc protecting group of Fmoc-Glu(OBzl)-OH (2) was stable to ammonium chloride. Under these conditions the carboxyl groups of 2 and Cbz-4-

carboxylphenylalanine-OBzl⁸ (6) underwent selective conversion to amides without undesired ammonolysis of either the *t*-butyl or benzyl esters of the starting materials.

Table 1

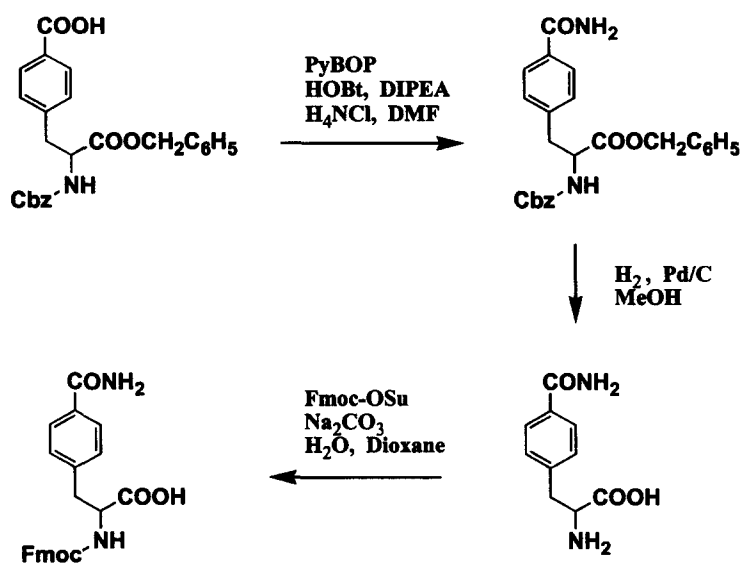
No.	Substrate	Product	Yield%
1			95
2			95
3			95
4			85
5			90
6			95

In addition to PyBOP, other peptide synthesis coupling agents such as HBTU, EDAC, and DCC were usable in this reaction (Table 2). Therefore, purification can be simplified by means of the different solubilities of the coupling agent and products, e.g. the organic soluble coupling agent DCC could be used to synthesize water-soluble amides. On the other hand, if the product is soluble in organic solvent, water-soluble coupling agents such as EDAC or PyBOP should be chosen, since the by-products can be removed by aqueous extraction. Reaction time varied with the coupling agent. As is generally the case in peptide synthesis, the carbodiimides DCC and EDAC took longer than the phosphonium and uronium agents. For derivatization of the α -carboxyl group of Cbz-Tyr(Bzl)-OH, reactions using PyBOP and HBTU were complete in 30 min whereas EDAC and DCC required 2 and 5 hr, respectively. Conversion of the side chain carboxyl group of Cbz-4-Cpa-OBzl using PyBOP was completed after 0.5h, while the reaction using EDAC required 3h. Thus PyBOP and HBTU were the most efficient reagents for this reaction.

This procedure was used to synthesize Fmoc-4-carboxamidophenylalanine from Cbz-4-Cpa-OBzl⁸ as shown in the Scheme.⁹ This new amino acid combines features of both phenylalanine and glutamine and may serve as a neutral replacement for 4-carboxyphenylalanine^{10,11} in peptide research and drug discovery efforts.

Table 2

Substrate	Coupling reagent	Reaction time (h)	Yield (%)
Cbz-Tyr(Bzl)-OH	PyBOP	0.5	100
	HBTU	0.5	100
	EDAC	2	100
	DCC	5	100
Cbz-4-Cpa(OBzl)-OH	PyBOP	0.5	100
	EDAC	3	100

**Scheme**

In summary, treatment of carboxylic acids with ammonium chloride and peptide synthesis coupling reagents in the presence of tertiary amines constitutes a straightforward method for the preparation of primary amides. Amides were prepared in excellent yields under very mild conditions which allowed selective aminolysis of activated carboxyl groups while susceptible functional groups such as esters and the base-labile Fmoc protecting group remained intact. Although our focus was on amino acids in this work, it is expected that this procedure will be useful in other areas of synthetic chemistry as well as solid phase organic synthesis.

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Abbreviations: Boc, *t*-butyloxycarbonyl; Bzl, benzyl; Cbz, benzyloxycarbonyl; Cpa, carboxyphenylalanine; DCC, *N,N*-dicyclohexylcarbodiimide; DIPEA, diisopropylethylamine; DMF, dimethylformamide; EDAC, 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride; FAB-MS, fast atom bombardment mass spectrometry; Fmoc, 9-fluorenylmethoxycarbonyl; Fmoc-OSu, *N*-(9-fluorenylmethoxycarbonyloxy)-succinimide; HBTU, 2-(1*H*-benzotriazole-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate; HOBt, 1-hydroxybenzotriazole; PyBOP, benzotriazole-1-yl-oxy-tris-pyrrolidino-phosphonium hexafluorophosphate.

References and Notes:

- Vogel A. "Practical Organic Chemistry" *Langman Scientific & Technical and Wiley: New York*, **1989**, pp 708-710.
- March J. "Advanced Organic Chemistry" *John Wiley & Sons: New York*, **1992**, pp 416-425.
- Bodanszky M., "The Practice of Peptide Synthesis", *Springer-Verlag: Berlin*, **1984**, pp 150-151, 173.
- Galaverra G., Corradini R., Cossena A. and Marchelli R. *Int. J. Peptide Protein Res.* **1993**, *42*, 53-57.
- Chen S. T., Wu S. H. and Wang K. T. *Synthesis* **1989**, 37-38.
- Analytical HPLC was performed on a Hewlett Packard 1090 liquid chromatograph using a Vydac C4 column (4.6 x 250 mm). The gradient was 15-75% B/25min with a flow rate of 1.5ml/min (A: 0.1% TFA in water; B: 0.1%TFA in acetonitrile). Elution was monitored with a diode array detector at both 230 nm and 274 nm.
- | Substrate | | | | Product | | | | |
|-----------|------|----------------|-----------------|---------|----------------|-----------------|----------------------|-------|
| No. | | HPLC*
(min) | IR (OH
band) | | HPLC*
(min) | IR (NH
band) | FAB-MS
calculated | found |
| 1 | acid | 17.396 | 3380-2500 | amide | 16.151 | 3375, 3321 | 405.18 | 405.2 |
| 2 | acid | 17.639 | 3380-2500 | amide | 16.399 | 3377, 3321 | 425.21 | 425.3 |
| 3 | acid | 13.768 | 3380-2500 | amide | 11.799 | 3391, 3334 | 309.18 | 309.2 |
| 4 | acid | 4.68 | 3350-2500 | amide | 3.26 | 3416 | 205.06 | 205.0 |
| 5 | acid | | 3170-2000 | amide | | 3391 | | |
| 6 | acid | 17.249 | 3380-2505 | amide | 15.993 | 3385, 3332 | 433.18 | 433.2 |
- Wang W., Obeysekere N.U. and McMurray J.S. *Tetrahedron Lett.* **1996**, *37*, 6661-6604.
- 0.43g (1mmol) of Cbz-4-Cpa-Bzl, 0.78g (1.5 mmol) of PyBOP, and 0.20g (1.5 mmol) of HOBt were dissolved in 4 ml of DMF. 0.58ml (4 mmol) of DIPEA and 0.01g (2 mmol) of NH₄Cl were added successively. The mixture was stirred for 0.5h at room temperature. Hydrogenolysis in the presence of 10% Pd/C removed the both benzyl based protecting groups, and was followed by treatment with Fmoc-OSu (1eq.) and Na₂CO₃ (3eq.) in water:dioxane (1:1) at room temperature for 4 h to afford 0.21 gm (0.5 mmol) of Fmoc-4-carboxyamidophenylalanine-OH.
- Matsoukas, J.M.; Agelis, G.; Wahhab, A.; Hondrelis, J.; Panagiotopoulos D.; Yamdagni R.; Wu, Q.; Mavromoustakos, T.; Maia, H.L.S.; Ganter, R.; Moore, G.J. *J. Med. Chem.* **1995**, *38*, 4660-4669.
- Lai, J.H., Marsilje, T.H., Choi, S., Nair, S.A., and Hangauer, D.G. *J. Peptide Res.* **1998**, *51*, 271-281.