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Picolyl amides of betulinic acid as antitumor agents causing tumor cell apoptosis

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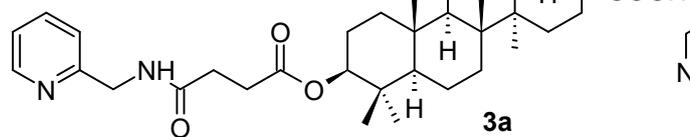
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ACCEPTED MANUSCRIPT
G-361, $IC_{50} = 2.4 \pm 0.0$ mM, TI = 19.3

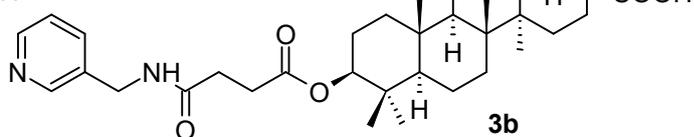
G-361, $IC_{50} = 0.5 \pm 0.1$ mM, TI = 100

MCF7, $IC_{50} = 2.2 \pm 0.2$ mM

MCF7, $IC_{50} = 1.4 \pm 0.1$ mM



3a



3b

ACCEPTED MANUSCRIPT

1 Picolyl amides of betulinic acid as antitumor agents causing tumor cell apoptosis

2

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26 Abstract

27 A series of picolyl amides of betulinic acid (**3a–3c** and **6a–6c**) was prepared and subjected to
28 the cytotoxicity screening tests. Structure-activity relationships studies resulted in finding
29 differences in biological activity in dependence on *o*-, *m*- and *p*-substitution of the pyridine
30 ring in the target amides, when cytotoxicity data of **3a–3c** and **6a–6c** were obtained and
31 compared. The amides **3b** and **3a** displayed cytotoxicity (given in the IC₅₀ values) in G-361
32 (0.5 ± 0.1 μM and 2.4 ± 0.0 μM, respectively), MCF7 (1.4 ± 0.1 μM and 2.2 ± 0.2 μM,
33 respectively), HeLa (2.4 ± 0.4 μM and 2.3 ± 0.5 μM, respectively) and CEM (6.5 ± 1.5 μM
34 and 6.9 ± 0.4 μM, respectively) tumor cell lines, and showed weak effect in the normal human
35 fibroblasts (BJ). Selectivity against all tested cancer cells was determined and compared to
36 normal cells with therapeutic index (TI) between 7 and 100 for compounds **3a** and **3b**. The
37 therapeutic index (TI = 100) was calculated for human malignant melanoma cell line (G-361)
38 *versus* normal human fibroblasts (BJ). The cytotoxicity of other target amides (**3c** and **6a–6c**)
39 revealed lower effects than **3a** and **3b** in the tested cancer cell lines.

40
41 **Key words:** betulinic acid; picolyl amine; amide; cytotoxicity; therapeutic index

42 43 1. Introduction

44 In the past years, we have published a paper dealing with the investigation of heteroaromatic
45 amides of selected steryl hemiesters [1]. Several of those compounds were based on *o*-, *m*-
46 and *p*-picolyl amines, displaying only moderate cytotoxicity. Those compounds suffered from
47 low polarity of the steryl skeleton. Nevertheless, even in that range of biological activity,
48 difference in cytotoxicity values could be seen, based on *o*-, *m*- or *p*-substitution in picolyl
49 amides. More recently, our attention has been focused at the investigation of triterpenoid acid
50 and their derivatives, which mostly showed considerable cytotoxicity and high and sometimes

51 selective antimicrobial activity. The compounds completed a majority of ADME parameters
52 required for prediction of potential drugs, and – in several cases – displayed important
53 supramolecular characteristics, among which formation of supramolecular hydrogels was one
54 of the most important features [2,3].

55 Betulinic acid, (3 β)-3-hydroxylup-20(29)-en-28-oic acid, is practically insoluble in water,
56 mainly due to the presence of large lipophilic backbone and inadequate number of hydrophilic
57 groups [4]. Even with this feature, which means, in fact, that this natural product does not
58 correspond to the Lipinski [5] and Ghose [6] rules describing potentially biologically active
59 compounds, it displays anti-HIV activity [7], antitumor [8,9] and antidiabetic activity [10,11].
60 Nature itself possesses natural mechanisms for increasing bioavailability of natural products
61 by forming their water-soluble conjugates (e.g. glycosides) [12]. To improve its
62 pharmacological characteristics in forms of the novel compounds has always been a challenge
63 for researchers, to find its new derivatives and related compounds [12]. Betulinic acid is a
64 plant product, to be found in genus *Betulla*, *Diospiros*, *Paeonia*, *Syzigium* or *Ziziphus* [13].
65 Concerning the antitumor effect, derivatives of betulinic acid have been used against a variety
66 of tumor cell lines: malignant brain tumor, primitive neuroectodermal tumor [14], human
67 chronic myelogenous leukemia, and against most of prevalent human cancer types, such as
68 cervical, prostate, breast, lung or colorectal cancer [12].

69 Many bioactive secondary metabolites of plants, including betulinic acid, induce apoptosis
70 pathway in cancer cells to exert their selective cytotoxic effects [15]. The cysteine-aspartic
71 acid protease (caspase) family proteins play a central role in the execution phase of apoptosis,
72 and the mechanism of initiation of apoptosis is mediated by the caspase cascade activation
73 [16]. Caspase-3 and caspase-7 are downstream factors activated by caspase-8 and caspase-9,
74 which – in turn – are activated predominantly by the extrinsic (death receptor) and intrinsic
75 (mitochondrial) pathways, respectively [17]. Caspase-3 is an executioner protease that results

76 in cleavage of poly-(ADP-ribose) polymerase (PARP), subsequent DNA degradation and
77 apoptotic death [18]. Previous studies showed that betulinic acid also displays anti-
78 inflammatory, anti-HIV, anti-helminthic, anti-nociceptive activities [19]. Several studies were
79 done to explain the molecular mechanisms of betulinic acid mediated antitumor activity. The
80 process seems to be largely dependent on the ability of betulinic acid to trigger the
81 mitochondrial pathway of apoptosis in cancer cells. Betulinic acid shows direct effect on
82 mitochondria. Successive treatment of betulinic acid on cancer cells disrupts the
83 mitochondrial transmembrane potential, which facilitates the cells to undergo apoptosis.
84 Involvement of excessive amount of reactive oxygen species was found to be the most
85 important factor for loss of mitochondrial membrane integrity of melanoma cells treated with
86 betulinic acid. It directly targets to mitochondria, which – in turn – regulates the downstream
87 caspase activation and side by side overcomes resistance property. Thus, betulinic acid has
88 not developed resistance in cancer cells, and, thereby, it became permissible agents for future
89 cancer therapy [15].

90 Aminomethylpyridines (picolyl amines), and related heteroaromatic amines, including
91 numbers of their derivatives, and also the derived *N*-oxides, had already been investigated for
92 their pharmacological activity [1,20-22], as markers of solid tumors [23] or agents in
93 supramolecular self-assembly [24]. Lipophilic derivation of picolyl amines have resulted in
94 obtaining conjugates, bearing ester and amide bond in each molecule; that type of study was
95 made in our previous work [1]. The importance of such compounds consists in enabling
96 transportation of potentially biologically active compounds through biomembrane and they
97 often form cationic immune-stimulating complexes [25,26]. The derivatives of picolyl amines
98 have also been found to be involved in activation of the caspase cascade and promote cell
99 death [27].

100 The objectives of the current research consist in (a) a synthesis of a series of picolyl amides of
101 betulinic acid both, at C(17)-COOH and at C(3)-OH (in the latter subseries using succinic
102 acid as a junction unit), and (b) investigation of cytotoxicity of the target compounds. A
103 comparison of the current series of compounds with those published earlier [1] will also be
104 discussed.

105

106 2. Experimental part

107 2.1. General

108 The ^1H NMR and the ^{13}C NMR spectra were recorded on a Bruker AVANCE 600 MHz
109 spectrometer at 600.13 MHz and 150.90 MHz in deuteriochloroform, using tetramethylsilane
110 ($\delta = 0.0$) as internal reference. ^1H NMR data are presented in the following order: chemical
111 shift (δ) expressed in ppm, multiplicity (s, singlet; d, doublet; t, triplet; q, quartet; m,
112 multiplet), coupling constants in Hertz, number of protons. For unambiguous assignment of
113 both ^1H and ^{13}C signals 2D NMR ^1H , ^{13}C gHSQC and gHMBC spectra were measured using
114 standard parameters sets and pulse programs delivered by producer of the spectrometer.
115 Copies of the NMR spectra are presented in the Supplementary material. Infrared spectra (IR)
116 were measured with a Nicolet 205 FT-IR spectrometer. Mass spectra (MS) were measured
117 with a Waters ZMD mass spectrometer in a positive ESI mode. Optical rotation was measured
118 on an Autopol IV instrument (Rudolph Research Analytical, USA) at 589 nm wavelength, and
119 the value was corrected to 20°C. The PE 2400 Series II CHNS/O Analyzer (Perkin Elmer,
120 USA) was used for simultaneous determination of C, H, and N (accuracy of CHN
121 determination better than 0.30 % abs.). TLC was carried out on silica gel plates (Merck
122 60F₂₅₄) and the visualization was performed by the UV detection and by spraying with the
123 methanolic solution of phosphomolybdic acid (5%) followed by heating. For column
124 chromatography, silica gel 60 (0.063-0.200 mm) from Merck was used. All chemicals and

125 solvents were purchased from regular commercial sources in analytical grade and the solvents
126 were purified by general methods before use.

127

128 2.2. (3 β)-3-[(3-Carboxypropanoyl)oxy]lup-20(29)-en-28-oic acid (**2**)

129 Succinic anhydride (1.62 g; 16.2 mmol; 3.5 eq) and DMAP (170 mg; 1.39 mmol, 0.30 eq)
130 were added to a solution of betulinic acid (**1**; 2.1 g; 4.6 mmol) in dry pyridine (20 mL). The
131 reaction mixture was stirred at r.t. for 5 days. After stopping the reaction, the resulting
132 mixture was poured onto ice, hydrochloric acid was added to adjust pH = 7, the organic layer
133 was extracted with chloroform, and the extract was dried over sodium sulfate. Evaporation of
134 the solvent afforded a solid that was purified by column chromatography, yielding 1.93 g
135 (75%) of **2**.

136 ^1H NMR: δ 0.79 (3H, s, H23), 0.79 (3H, s, H24), 0.80 (3H, s, H25), 0.87 (3H, s, H26), 0.94
137 (3H, s, H27), 1.64 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29), 2.09-2.13 (2H, m, H16), 2.22 (1H,
138 ddd, $J_1 = 3.6$ Hz, $J_2 = 11.6$ Hz, $J_3 = 12.9$ Hz, H13), 2.44-2.52 (4H, m, H2'-H3'), 2.95 (1H, dt, J_1
139 = 4.8 Hz, $J_2 = 11.0$ Hz, $J_3 = 11.0$ Hz, H19), 4.38 (1H, dd, $J_1 = 4.7$ Hz, $J_2 = 11.6$ Hz, H3), 4.56
140 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 = 1.4$ Hz, $J_4 = 2.4$ Hz, H30), 4.69 (1H, dq, $J_1 = 0.7$ Hz, $J_2 =$
141 0.7 Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz, H30). ^{13}C NMR: δ 14.31 (q, C27), 15.65 (q, C24), 15.81 (q,
142 C25), 16.35 (q, C26), 17.67 (t, C6), 18.90 (q, C29), 20.42 (t, C11), 23.27 (t, C2), 25.00 (t,
143 C12), 27.56 (q, C23), 28.77 (t, C2'), 29.15 (t, C21), 29.15 (t, C3'), 30.06 (t, C15), 31.64 (t,
144 C16), 33.68 (t, C22), 36.28 (t, C1), 36.60 (s, C10), 37.38 (s, C4), 37.54 (d, C13), 37.67 (t,
145 C7), 40.21 (s, C8), 41.99 (s, C14), 46.57 (d, C19), 48.50 (d, C18), 49.61 (d, C9), 54.62 (d,
146 C5), 55.36 (s, C17), 79.98 (d, C3), 109.58 (t, C30), 150.26 (s, C20), 171.56 (s, C1'), 173.32 (s,
147 C4'), 177.17 (s, C28). MS: m/z 555.2 [M-H] $^-$. IR (KBr; cm^{-1}): 2946 (C-H val.), 2872 (-CH₃
148 val.), 1734 (-C=O val.). For C₃₄H₅₂O₆ (556.77) calculated C (73.34), H (9.41), found C
149 (73.37), H (9.40). M.p. 244-246 °C.

150
151 2.3. (3 β)-3-({4-Oxo-4-[(pyridin-n-ylmethyl)amino]butanoyl}oxy)lup-20(29)-en-28-oic acid
152 [**3a** (n=2), **3b** (n=3) and **3c** (n=4)]
153 *o*-, *m*- or *p*-Picoyl amine (20 μ l; 0.198 mmol; 1.1 eq) and T3P (0.32 ml; 0.54 mmol; 3 eq)
154 were added to a solution of **2** (100 mg; 0.18 mmol) in dry pyridine (2 mL), and the reaction
155 mixture was stirred at r.t. for 1 day. After stopping the reaction, saturated solution of sodium
156 bicarbonate (4 mL) was added to reaction mixture, and stirring continued for several more
157 hours. The resulting mixture was extracted with chloroform, and dried over sodium sulfate.
158 Evaporation of the solvent afforded a solid, which was purified by column chromatography.
159 Yields: **3a** (85 %), **3b** (91 %), **3c** (89 %).
160 **3a**: ^1H NMR: δ 0.77 (3H, s, H23), 0.78 (3H, s, H24), 0.80 (3H, s, H25), 0.87 (3H, s, H26),
161 0.95 (3H, s, H27), 1.65 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.3$ Hz, H29), 2.09-2.14 (2H, m, H16), 2.23
162 (1H, ddd, $J_1 = 4.7$ Hz, $J_2 = 11.7$ Hz, $J_3 = 12.9$ Hz, H13), 2.44-2.55 (4H, m, H2'-H3'), 2.95 (1H,
163 dt, $J_1 = 5.3$ Hz, $J_2 = 11.0$ Hz, $J_3 = 11.0$ Hz, H19), 4.31 (1H, dd, $J_1 = 6.0$ Hz, $J_2 = 15.9$ Hz, H5'),
164 4.35 (1H, dd, $J_1 = 6.0$ Hz, $J_2 = 15.9$ Hz, H5'), 4.38 (1H, dd, $J_1 = 4.8$ Hz, $J_2 = 11.6$ Hz, H3),
165 4.56 (1H, dq, $J_1 = 1.3$ Hz, $J_2 = 1.3$ Hz, $J_3 = 1.3$ Hz, $J_4 = 2.4$ Hz, H30), 4.69 (1H, bd, $J = 2.4$ Hz,
166 H30), 7.24 (1H, dddd, $J_1 = 0.6$ Hz, $J_2 = 0.6$ Hz, $J_3 = 1.2$ Hz, $J_4 = 4.8$ Hz, $J_5 = 7.5$ Hz, H8'), 7.26
167 (1H, dtt, $J_1 = 0.6$ Hz, $J_2 = 0.6$ Hz, $J_3 = 1.1$ Hz, $J_4 = 1.1$ Hz, $J_5 = 7.8$ Hz, H10'), 7.72 (1H, dt, J_1
168 = 1.8 Hz, $J_2 = 7.7$ Hz, $J_3 = 7.7$ Hz, H9'), 8.45 (1H, t, $J = 6.0$ Hz, NH), 8.48 (1H, ddd, $J_1 = 0.9$
169 Hz, $J_2 = 1.8$ Hz, $J_3 = 4.8$ Hz, H7'). ^{13}C NMR: δ 14.32 (q, C27), 15.66 (q, C24), 15.82 (q, C25),
170 16.36 (q, C26), 17.67 (t, C6), 18.90 (q, C29), 20.43 (t, C11), 23.29 (t, C2), 25.01 (t, C12),
171 27.58 (q, C23), 29.16 (t, C21), 29.35 (t, C2'), 29.87 (t, C3'), 30.06 (t, C15), 31.66 (t, C16),
172 33.70 (t, C22), 36.29 (t, C1), 36.59 (s, C10), 37.38 (s, C4), 37.54 (d, C13), 37.69 (t, C7),
173 40.22 (s, C8), 41.99 (s, C14), 44.17 (t, C5'), 46.58 (d, C19), 48.50 (d, C18), 49.62 (d, C9),
174 54.63 (d, C5), 55.37 (s, C17), 79.87 (d, C3), 109.58 (t, C30), 120.88 (d, C10'), 121.97 (d, C8'),

175 136.56 (d, C9'), 148.71 (d, C7'), 150.28 (s, C20), 158.64 (s, C6'), 170.88 (s, C1'), 171.88 (s,
176 C4'), 177.18 (s, C28). MS: m/z 647.3 [M+H]⁺, 645.0 [M-H]⁻. IR (KBr; cm⁻¹): 3377 (N-H val.
177 amide), 2945 (C-H val.), 2871 (-CH₃ val.), 1728 (-C=O val. ester), 1659 (-C=O val. amide).
178 For C₄₀H₅₈N₂O₅ (646.90) calculated C (74.27), H (9.04), N (4.33), found C (74.29), H (9.03),
179 N (4.35). M.p. 121-124 °C. $[\alpha]_D^{20} = +15.5$ (*c* 0.323).

180 **3b**: ¹H NMR: δ 0.70-0.73 (1H, m, H5), 0.76 (3H, s, H24), 0.76 (3H, s, H23), 0.78 (3H, d, $J =$
181 0.6 Hz, H25), 0.88 (3H, s, H26), 0.92 (3H, d, $J = 0.7$ Hz, H27), 1.63 (3H, dd, $J_1 = 0.7$ Hz, $J_2 =$
182 1.3 Hz, H29), 2.17 (1H, ddd, $J_1 = 3.7$ Hz, $J_2 = 11.6$ Hz, $J_3 = 12.9$ Hz, H13), 2.22 (2H, ddd, $J_1 =$
183 2.9 Hz, $J_2 = 4.0$ Hz, $J_3 = 13.1$ Hz, H16), 2.46 (2H, t, $J = 6.6$ Hz, H3'), 2.58-2.66 (2H, m, H2'),
184 2.94 (1H, dt, $J_1 = 5.0$ Hz, $J_2 = 11.0$ Hz, $J_3 = 11.0$ Hz, H19), 4.39 (1H, dd, $J_1 = 5.9$ Hz, $J_2 = 14.9$
185 Hz, H5'), 4.41 (1H, dd, $J_1 = 5.7$ Hz, $J_2 = 14.9$ Hz, H5'), 4.42 (1H, dd, $J_1 = 6.2$ Hz, $J_2 = 10.1$ Hz,
186 H3), 4.55 (1H, dq, $J_1 = 1.3$ Hz, $J_2 = 1.3$ Hz, $J_3 = 1.3$ Hz, $J_4 = 2.4$ Hz, H30), 4.68 (1H, dq, $J_1 =$
187 0.7 Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.3$ Hz, H30), 6.20 (1H, bt, $J = 5.8$ Hz, NH), 7.23 (1H,
188 ddd, $J_1 = 0.8$ Hz, $J_2 = 4.8$ Hz, $J_3 = 7.8$ Hz, H10'), 7.63 (1H, ddt, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 =$
189 1.8 Hz, $J_4 = 2.1$ Hz, $J_5 = 7.8$ Hz, H11'), 8.48 (1H, bdd, $J_1 = 1.8$ Hz, $J_2 = 4.8$ Hz, H9'), 8.50 (1H,
190 bs, H7'). ¹³C NMR: δ 14.71 (q, C27), 16.13 (q, C24), 16.17 (q, C25), 16.51 (q, C26), 18.24 (t,
191 C6), 19.41 (q, C29), 20.99 (t, C11), 23.77 (t, C2), 25.60 (t, C12), 28.02 (q, C23), 29.78 (t,
192 C21), 29.78 (t, C2'), 30.72 (t, C3'), 31.25 (t, C15), 32.29 (t, C16), 34.38 (t, C22), 37.09 (t, C1),
193 37.23 (s, C10), 37.95 (s, C4), 38.49 (d, C13), 38.49 (t, C7), 40.85 (t, C5'), 41.09 (s, C8), 42.56
194 (s, C14), 47.01 (d, C19), 49.46 (d, C18), 50.57 (d, C9), 55.56 (d, C5), 56.38 (s, C17), 81.68
195 (d, C3), 109.64 (t, C30), 123.82 (d, C10'), 134.65 (s, C6'), 136.46 (d, C11'), 147.92 (d, C9'),
196 148.35 (d, C7'), 150.45 (s, C20), 171.74 (s, C1'), 172.73 (s, C4'), 179.99 (s, C28). MS: m/z
197 647.3 [M+H]⁺, 645.3 [M-H]⁻. IR (KBr; cm⁻¹): 3350 (N-H val. amide), 2946 (C-H val.), 2871
198 (-CH₃ val.), 1729 (-C=O val. ester), 1657 (-C=O val. amide). For C₄₀H₅₈N₂O₅ (646.90)

199 calculated C (74.27), H (9.04), N (4.33), found C (74.30), H (9.02), N (4.35). M.p. 131-134
200 °C. $[\alpha]_{\text{D}}^{20} = +12.8$ (*c* 0.383).

201 **3c**: ^1H NMR: δ 0.70-0.74 (1H, m, H5), 0.76 (3H, s, H23), 0.76 (3H, s, H24), 0.78 (3H, s,
202 H25), 0.87 (3H, s, H26), 0.91 (3H, bd, H27), 0.98 (2H, dq, $J_1 = 4.0$ Hz, $J_2 = 13.0$ Hz, $J_3 = 13.0$
203 Hz, $J_4 = 13.0$ Hz, H12), 1.62 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.3$ Hz, H29), 2.16 (1H, ddd, $J_1 = 3.6$
204 Hz, $J_2 = 11.5$ Hz, $J_3 = 12.8$ Hz, H13), 2.21 (2H, ddd, $J_1 = 3.2$ Hz, $J_2 = 3.8$ Hz, $J_3 = 13.0$ Hz,
205 H16), 2.48 (1H, dt, $J_1 = 6.5$ Hz, $J_2 = 6.5$ Hz, $J_3 = 15.0$ Hz, H3'), 2.50 (1H, dt, $J_1 = 5.7$ Hz, $J_2 =$
206 5.7 Hz, $J_3 = 15.0$ Hz, H3'), 2.62 (1H, dt, $J_1 = 5.7$ Hz, $J_2 = 5.7$ Hz, $J_3 = 17.3$ Hz, H2'), 2.66 (1H,
207 dt, $J_1 = 6.6$ Hz, $J_2 = 6.6$ Hz, $J_3 = 17.3$ Hz, H2'), 2.95 (1H, dt, $J_1 = 5.0$ Hz, $J_2 = 10.9$ Hz, $J_3 =$
208 10.9 Hz, H19), 4.38 (1H, bdd, $J_1 = 6.1$ Hz, $J_2 = 16.0$ Hz, H5'), 4.42 (1H, bdd, $J_1 = 6.1$ Hz, $J_2 =$
209 16.0 Hz, H5'), 4.43 (1H, dd, $J_1 = 5.5$ Hz, $J_2 = 11.0$ Hz, H3), 4.54 (1H, dq, $J_1 = 1.3$ Hz, $J_2 = 1.3$
210 Hz, $J_3 = 1.3$ Hz, $J_4 = 2.3$ Hz, H30), 4.67 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.3$
211 Hz, H30), 6.30 (1H, t, $J = 6.1$ Hz, NH), 7.16-7.18 (2H, m, H7'), 8.47-8.49 (2H, m, H8'). ^{13}C
212 NMR: δ 14.68 (q, C27), 16.06 (q, C24), 16.15 (q, C25), 16.49 (q, C26), 18.18 (t, C6), 19.37
213 (q, C29), 20.92 (t, C11), 23.75 (t, C2), 25.50 (t, C12), 28.01 (q, C23), 29.73 (t, C21), 29.94 (t,
214 C2'), 30.64 (t, C3'), 31.16 (t, C15), 32.26 (t, C16), 34.28 (t, C22), 37.08 (t, C1), 37.16 (s,
215 C10), 37.90 (s, C4), 38.38 (t, C7), 38.40 (d, C13), 40.75 (s, C8), 42.42 (s, C14), 42.48 (t, C5'),
216 46.97 (d, C19), 49.33 (d, C18), 50.47 (d, C9), 55.48 (d, C5), 56.34 (s, C17), 81.73 (d, C3),
217 109.63 (t, C30), 122.47 (d, C7'), 148.49 (s, C6'), 149.08 (d, C8'), 150.49 (s, C20), 171.98 (s,
218 C1'), 172.91 (s, C4'), 180.46 (s, C28). MS: m/z 647.2 $[\text{M}+\text{H}]^+$, 645.3 $[\text{M}-\text{H}]^-$. IR (KBr; cm^{-1}):
219 3358 (N-H val. amide), 2946 (C-H val.), 2870 ($-\text{CH}_3$ val.), 1729 ($-\text{C}=\text{O}$ val. ester), 1663 ($-\text{C}=\text{O}$
220 val. amide). For $\text{C}_{40}\text{H}_{58}\text{N}_2\text{O}_5$ (646.90) calculated C (74.27), H (9.04), N (4.33), found C
221 (74.26), H (9.05), N (4.31). M.p. 145-147 °C. $[\alpha]_{\text{D}}^{20} = +12.8$ (*c* 0.188).

222

223 2.4. (3 β)-3-(Acetyloxy)lup-20(29)-en-28-oic acid (**4**)

224 Acetic anhydride (0.45 mL; 4.7 mmol; 1.44 eq), DMAP (52 mg; 0.43 mmol, 0.13 eq) and
225 EDIPA (1 mL) were added to a solution of betulinic acid (**1**; 1.5 g; 3.3 mmol) in dry THF (10
226 mL). The reaction mixture was refluxed for 3 h. After stopping the reaction, water (10 mL)
227 was added and the resulting mixture was stirred for 1 hour. The resulting mixture was
228 extracted with chloroform, and the extract was dried over sodium sulfate. Evaporation of the
229 solvent afforded a solid, which was purified by column chromatography, resulting finally in
230 1.55 g (95 %) of **4**.

231 ^1H NMR: δ 0.83 (3H, s, H23), 0.84 (3H, s, H25), 0.85 (3H, bs, H24), 0.93 (3H, s, H26), 0.97
232 (3H, bs, H27), 1.05 (2H, dq, $J_1 = 4.9$ Hz, $J_2 = 13.0$ Hz, $J_3 = 13.0$ Hz, $J_4 = 13.0$ Hz, H12), 1.19
233 (2H, dt, $J_1 = 3.2$ Hz, $J_2 = 3.2$ Hz, $J_3 = 13.6$ Hz, H21), 1.70 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz,
234 H29), 2.04 (3H, s, OAc), 2.13 (2H, ddd, $J_1 = 3.7$ Hz, $J_2 = 11.6$ Hz, $J_3 = 12.8$ Hz, H1), 2.27
235 (2H, dt, $J_1 = 3.8$ Hz, $J_2 = 3.8$ Hz, $J_3 = 13.1$ Hz, H16), 3.01 (1H, dt, $J_1 = 5.0$ Hz, $J_2 = 10.9$ Hz, J_3
236 = 10.9 Hz, H19), 4.47 (1H, dd, $J_1 = 5.5$ Hz, $J_2 = 10.9$ Hz, H3), 4.61 (1H, dq, $J_1 = 1.4$ Hz, $J_2 =$
237 1.4 Hz, $J_3 = 1.4$ Hz, $J_4 = 2.3$ Hz, H30), 4.74 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 =$
238 2.3 Hz, H30). ^{13}C NMR: δ 14.64 (q, C27), 16.02 (q, C24), 16.17 (q, C25), 16.46 (q, C26),
239 18.15 (t, C6), 19.32 (q, OAc), 19.33 (q, C29), 20.84 (t, C11), 23.68 (t, C2), 25.43 (t, C12),
240 27.94 (q, C23), 29.68 (t, C21), 30.54 (t, C15), 32.13 (t, C16), 34.22 (t, C22), 37.03 (d, C13),
241 37.11 (s, C10), 37.79 (t, C7), 38.37 (s, C4), 38.39 (t, C1), 40.68 (s, C8), 42.41 (s, C14), 46.92
242 (d, C19), 49.25 (d, C18), 50.38 (d, C9), 55.40 (d, C5), 56.35 (s, C17), 80.94 (d, C3), 109.74 (t,
243 C30), 150.36 (s, C20), 171.06 (s, OAc), 181.47 (s, C28). MS: m/z 497.1 $[\text{M}-\text{H}]^-$. IR (KBr; cm^{-1})
244 1): 2945 (C-H val.), 2871 (-CH₃ val.), 1735 (-C=O val. ester), 1694 (-C=O val. conjugated
245 acid). For C₃₂H₅₀O₄ (498.74) calculated C (77.06), H (10.10), found C (77.09), H (10.08).
246 M.p. 263-266 °C.

247

248 2.5. (3 β)-28-Oxo-28-[(pyridin-n-ylmethyl)amino]lup-20(29)-en-3-yl acetate [**5a** (n=2), **5b**
249 (n=3) and **5c** (n=4)]

250 A solution of oxalyl chloride in dry dichloromethane (2 M; 1.4 mL; 2.8 mmol; 7 eq) was
251 added to a solution of **4** (200 mg; 0.4 mmol) in dry dichloromethane (4 mL), the reaction
252 mixture was stirred for 3 h, then evaporated and dissolved again in dry dichloromethane (4
253 mL). EDIPA (0.182 mL; 1.04 mmol; 2.6 eq) and *o*-, *m*- or *p*-picoyl amine (0.049 mL; 0.48
254 mmol; 1.2 eq) were added, and the reaction mixture was stirred at r.t. through night (for 12 h).
255 After stopping the reaction, evaporation of the solvent afforded a solid, which was purified by
256 column chromatography. Yields: **5a** (86 %), **5b** (99 %), **5c** (87 %).

257 **5a**: ¹H NMR: δ 0.72 (1H, dd, $J_1 = 2.1$ Hz, $J_2 = 11.7$ Hz, H5), 0.75 (3H, s, H24), 0.77 (3H, s,
258 H25), 0.77 (3H, s, H23), 0.78 (3H, s, H26), 0.90 (3H, d, $J = 0.6$ Hz, H27), 1.08 (2H, dt, $J_1 =$
259 3.2 Hz, $J_2 = 3.2$ Hz, $J_3 = 13.4$ Hz, H21), 1.64 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29), 1.99
260 (3H, s, H2'), 2.04 (2H, dt, $J_1 = 3.3$ Hz, $J_2 = 3.3$ Hz, $J_3 = 13.6$ Hz, H16), 2.38 (1H, ddd, $J_1 = 3.7$
261 Hz, $J_2 = 11.5$ Hz, $J_3 = 12.9$ Hz, H13), 3.10 (1H, dt, $J_1 = 4.6$ Hz, $J_2 = 11.1$ Hz, $J_3 = 11.1$ Hz,
262 H19), 4.41 (1H, dd, $J_1 = 5.5$ Hz, $J_2 = 10.0$ Hz, H3), 4.46 (1H, dd, $J_1 = 5.2$ Hz, $J_2 = 15.9$ Hz,
263 H3'), 4.52 (1H, dd, $J_1 = 5.2$ Hz, $J_2 = 15.9$ Hz, H3'), 4.54 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 =$
264 1.4 Hz, $J_4 = 2.4$ Hz, H30), 4.69 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz,
265 H30), 6.85 (1H, t, $J = 5.2$ Hz, NH), 7.17 (1H, m, $J_1 = 0.6$ Hz, $J_2 = 0.6$ Hz, $J_3 = 1.2$ Hz, $J_4 = 4.9$
266 Hz, $J_5 = 7.5$ Hz, H7'), 7.29 (1H, m, $J_1 = 0.6$ Hz, $J_2 = 0.6$ Hz, $J_3 = 1.2$ Hz, $J_4 = 7.8$ Hz, H9'),
267 7.64 (1H, dt, $J_1 = 1.8$ Hz, $J_2 = 7.6$ Hz, $J_3 = 7.6$ Hz, H8'), 8.49 (1H, ddd, $J_1 = 0.8$ Hz, $J_2 = 1.8$
268 Hz, $J_3 = 4.9$ Hz, H6'). ¹³C NMR: δ 14.58 (q, C27), 15.98 (q, C24), 16.17 (q, C25), 16.45 (q,
269 C26), 18.15 (t, C6), 19.42 (q, C29), 20.91 (t, C11), 21.31 (q, C2'), 23.67 (t, C2), 25.56 (t,
270 C12), 27.91 (q, C23), 29.41 (t, C21), 30.86 (t, C15), 33.65 (t, C16), 34.26 (t, C22), 37.08 (s,
271 C10), 37.71 (d, C13), 37.76 (s, C4), 38.33 (t, C1), 38.35 (t, C7), 40.70 (s, C8), 42.45 (s, C14),
272 44.17 (t, C3'), 46.78 (d, C19), 50.06 (d, C18), 50.49 (d, C9), 55.40 (d, C5), 55.76 (s, C17),

273 80.93 (d, C3), 109.35 (t, C30), 122.34 (d, C7'), 122.64 (d, C9'), 137.12 (d, C8'), 148.50 (d,
274 C6'), 150.98 (s, C20), 157.08 (s, C4'), 171.01 (s, C1'), 176.39 (s, C28). MS: m/z 589.2
275 $[M+H]^+$, 587.2 $[M-H]^-$. IR (KBr; cm^{-1}): 2946 (C–H val.), 2870 (–CH₃ val.), 1734 (–C=O val.
276 ester), 1651 (–C=O val. amide). For C₃₈H₅₆N₂O₃ (588.86) calculated C (77.51), H (9.59), N
277 (4.76), found C (77.53), H (9.58), N (4.75). M.p. 110-115 °C.

278 **5b**: ¹H NMR: δ 0.72 (1H, dd, $J_1 = 2.1$ Hz, $J_2 = 11.0$ Hz, H5), 0.77 (3H, s, H24), 0.78 (3H, d, J
279 = 0.7 Hz, H25), 0.78 (3H, s, H23), 0.79 (3H, s, H26), 0.90 (3H, s, H27), 1.07 (2H, dt, $J_1 = 3.3$
280 Hz, $J_2 = 3.3$ Hz, $J_3 = 13.5$ Hz, H21), 1.64 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29), 1.89 (2H, dt,
281 $J_1 = 3.4$ Hz, $J_2 = 3.4$ Hz, $J_3 = 13.7$ Hz, H16), 1.99 (3H, s, H2'), 2.39 (1H, ddd, $J_1 = 3.7$ Hz, $J_2 =$
282 11.5 Hz, $J_3 = 12.9$ Hz, H13), 3.09 (1H, dt, $J_1 = 4.5$ Hz, $J_2 = 11.2$ Hz, $J_3 = 11.2$ Hz, H19), 4.31
283 (1H, dd, $J_1 = 5.9$ Hz, $J_2 = 14.9$ Hz, H3'), 4.41 (1H, dd, $J_1 = 5.3$ Hz, $J_2 = 9.6$ Hz, H3), 4.48 (1H,
284 dd, $J_1 = 6.1$ Hz, $J_2 = 14.9$ Hz, H3'), 4.55 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 = 1.4$ Hz, $J_4 = 2.4$
285 Hz, H30), 4.69 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz, H30), 6.08 (1H, t, J
286 = 6.0 Hz, NH), 7.24 (1H, ddd, $J_1 = 0.9$ Hz, $J_2 = 4.9$ Hz, $J_3 = 7.8$ Hz, H8'), 7.66 (1H, bddd, $J_1 =$
287 1.8 Hz, $J_2 = 2.2$ Hz, $J_3 = 7.8$ Hz, H9'), 8.47 (1H, dd, $J_1 = 1.8$ Hz, $J_2 = 4.9$ Hz, H7'), 8.52 (1H,
288 bd, $J = 2.2$ Hz, H5'). ¹³C NMR: δ 14.57 (q, C27), 16.01 (q, C24), 16.21 (q, C25), 16.45 (q,
289 C26), 18.14 (t, C6), 19.45 (q, C29), 20.91 (t, C11), 21.31 (q, C2'), 23.67 (t, C2), 25.56 (t,
290 C12), 27.91 (q, C23), 29.39 (t, C21), 30.81 (t, C15), 33.68 (t, C16), 34.29 (t, C22), 37.09 (s,
291 C10), 37.68 (d, C13), 37.77 (s, C4), 38.32 (t, C1), 38.37 (t, C7), 40.71 (t, C3'), 40.73 (s, C8),
292 42.44 (s, C14), 46.65 (d, C19), 50.10 (d, C18), 50.51 (d, C9), 55.43 (d, C5), 55.66 (s, C17),
293 80.91 (d, C3), 109.48 (t, C30), 123.66 (d, C8'), 135.15 (s, C4'), 136.22 (d, C9'), 148.24 (d,
294 C7'), 148.73 (d, C5'), 150.74 (s, C20), 171.03 (s, C1'), 176.26 (s, C28). MS: m/z 589.2
295 $[M+H]^+$, 587.2 $[M-H]^-$. IR (KBr; cm^{-1}): 3270 (N–H val. amide), 2941 (C–H val.), 2871 (–CH₃
296 val.), 1725 (–C=O val. ester), 1659 (–C=O val. amide). For C₃₈H₅₆N₂O₃ (588.86) calculated C
297 (77.51), H (9.59), N (4.76), found C (77.50), H (9.60), N (4.78). M.p. 241-244 °C.

298 **5c**: ^1H NMR: δ 0.72 (1H, dd, $J_1 = 2.1$ Hz, $J_2 = 11.1$ Hz, H5), 0.78 (3H, s, H24), 0.79 (3H, s,
299 H25), 0.79 (3H, s, H23), 0.82 (3H, s, H26), 0.91 (3H, s, H27), 1.11 (2H, dt, $J_1 = 3.4$ Hz, $J_2 =$
300 3.4 Hz, $J_3 = 13.5$ Hz, H21), 1.64 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29), 1.91 (2H, dt, $J_1 = 3.3$
301 Hz, $J_2 = 3.3$ Hz, $J_3 = 13.9$ Hz, H16), 1.99 (3H, s, H2'), 2.39 (1H, ddd, $J_1 = 3.7$ Hz, $J_2 = 11.6$
302 Hz, $J_3 = 13.0$ Hz, H13), 3.08 (1H, dt, $J_1 = 4.5$ Hz, $J_2 = 11.1$ Hz, $J_3 = 11.1$ Hz, H19), 4.31 (1H,
303 dd, $J_1 = 6.0$ Hz, $J_2 = 15.6$ Hz, H3'), 4.42 (1H, dd, $J_1 = 5.9$ Hz, $J_2 = 10.7$ Hz, H3), 4.45 (1H, dd,
304 $J_1 = 6.1$ Hz, $J_2 = 15.6$ Hz, H3'), 4.55 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 = 1.4$ Hz, $J_4 = 2.4$ Hz,
305 H30), 4.69 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz, H30), 6.10 (1H, t, $J =$
306 6.1 Hz, NH), 7.18-7.21 (2H, m, H5'), 8.48-8.51 (2H, m, H6'). ^{13}C NMR: δ 14.58 (q, C27),
307 16.09 (q, C24), 16.22 (q, C25), 16.45 (q, C26), 18.15 (t, C6), 19.45 (q, C29), 20.91 (t, C11),
308 21.31 (q, C2'), 23.66 (t, C2), 25.55 (t, C12), 27.91 (q, C23), 29.46 (t, C21), 30.80 (t, C15),
309 33.72 (t, C16), 34.34 (t, C22), 37.09 (s, C10), 37.68 (d, C13), 37.77 (s, C4), 38.38 (t, C7),
310 38.38 (t, C1), 40.77 (s, C8), 42.24 (s, C14), 42.46 (t, C3'), 46.60 (d, C19), 50.10 (d, C18),
311 50.52 (d, C9), 55.44 (d, C5), 55.72 (s, C17), 80.91 (d, C3), 109.55 (t, C30), 122.57 (d, C5'),
312 148.98 (s, C4'), 149.41 (d, C6'), 150.65 (s, C20), 171.04 (s, C1'), 176.38 (s, C28). MS: m/z
313 589.2 $[\text{M}+\text{H}]^+$, 587.2 $[\text{M}-\text{H}]^-$. IR (KBr; cm^{-1}): 3378 (N-H val. amide), 2946 (C-H val.), 2869
314 ($-\text{CH}_3$ val.), 1734 ($-\text{C}=\text{O}$ val. ester), 1642 ($-\text{C}=\text{O}$ val. amide). For $\text{C}_{38}\text{H}_{56}\text{N}_2\text{O}_3$ (588.86)
315 calculated C (77.51), H (9.59), N (4.76), found C (77.49), H (9.57), N (4.78). M.p. 105-108
316 $^\circ\text{C}$.

317

318 2.6. (3 β)-3-Hydroxy-*N*-(pyridin-*n*-ylmethyl)lup-20(29)-en-28-amide [**6a** ($n=2$), **6b** ($n=3$) and
319 **6c** ($n=4$)]

320 LiOH \cdot H $_2$ O (54 mg; 1.3 mmol; 5 eq) was added to a solution of **5a**, **5b** or **5c** (151 mg; 0.26
321 mmol) in methanol (15 mL), and the reaction mixture was refluxed for 5 h. After stopping the
322 reaction, the solvent was evaporated, and the residue was dissolved in chloroform and filtered.

323 Evaporation of the solvent afforded a solid, which was purified by column chromatography.
324 Yields: **6a** (82 %), **6b** (84 %), **6c** (83 %).

325 **6a**: ^1H NMR: δ 0.62 (1H, dd, $J_1 = 2.3$ Hz, $J_2 = 11.5$ Hz, H5), 0.70 (3H, s, H24), 0.74 (3H, d, J
326 = 0.8 Hz, H25), 0.75 (3H, s, H23), 0.90 (3H, s, H26), 0.91 (3H, d, $J = 0.6$ Hz, H27), 1.08 (2H,
327 dt, $J_1 = 3.1$ Hz, $J_2 = 3.1$ Hz, $J_3 = 13.5$ Hz, H21), 1.64 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29),
328 2.04 (2H, dt, $J_1 = 3.3$ Hz, $J_2 = 3.3$ Hz, $J_3 = 13.5$ Hz, H16), 2.38 (1H, ddd, $J_1 = 3.6$ Hz, $J_2 = 11.6$
329 Hz, $J_3 = 13.1$ Hz, H13), 3.10 (1H, dt, $J_1 = 4.6$ Hz, $J_2 = 11.1$ Hz, $J_3 = 11.1$ Hz, H19), 3.13 (1H,
330 dd, $J_1 = 4.9$ Hz, $J_2 = 11.5$ Hz, H3), 4.46 (1H, dd, $J_1 = 5.1$ Hz, $J_2 = 12.0$ Hz, H1'), 4.53 (1H, dd,
331 $J_1 = 5.1$ Hz, $J_2 = 12.0$ Hz, H1'), 4.54 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 = 1.4$ Hz, $J_4 = 2.4$ Hz,
332 H30), 4.69 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz, H30), 6.85 (1H, bt, $J =$
333 5.1 Hz, NH), 7.17 (1H, ddd, $J_1 = 1.3$ Hz, $J_2 = 5.0$ Hz, $J_3 = 7.6$ Hz, H5'), 7.29 (1H, bdt, $J_1 = 1.0$
334 Hz, $J_2 = 1.0$ Hz, $J_3 = 7.7$ Hz, H7'), 7.64 (1H, dt, $J_1 = 1.8$ Hz, $J_2 = 7.6$ Hz, $J_3 = 7.6$ Hz, H6'),
335 8.49 (1H, ddd, $J_1 = 1.0$ Hz, $J_2 = 1.8$ Hz, $J_3 = 5.0$ Hz, H4'). ^{13}C NMR: δ 14.63 (q, C27), 15.34
336 (q, C24), 15.98 (q, C25), 16.12 (q, C26), 18.28 (t, C6), 19.46 (q, C29), 20.90 (t, C11), 25.61
337 (t, C12), 27.40 (t, C2), 27.99 (q, C23), 29.43 (t, C15), 33.67 (t, C16), 33.88 (t, C21), 34.34 (t,
338 C22), 37.17 (s, C10), 37.74 (d, C13), 38.34 (t, C7), 38.69 (t, C1), 38.83 (s, C4), 40.69 (s, C8),
339 42.46 (s, C14), 44.21 (t, C1'), 46.77 (d, C19), 50.07 (d, C18), 50.59 (d, C9), 55.34 (d, C5),
340 55.77 (s, C17), 78.95 (d, C3), 109.30 (t, C30), 122.33 (d, C4'), 122.61 (d, C7'), 137.05 (d,
341 C6'), 148.58 (d, C3'), 151.02 (s, C20), 157.12 (s, C2'), 176.36 (s, C28). MS: m/z
342 547.3[M+H]⁺, 545.3 [M-H]⁻. IR (KBr; cm⁻¹): 3368 (N-H val. amide), 2942 (C-H val.), 2868
343 (-CH₃ val.), 1649 (-C=O val. amide). For C₃₆H₅₄N₂O₂ (546.83) calculated C (79.07), H
344 (9.95), N (5.12), found C (79.09), H (9.93), N (5.14). M.p. 210-211 °C. $[\alpha]_{\text{D}}^{20} = +10.3$ (c
345 0.350).

346 **6b**: ^1H NMR: δ 0.61 (1H, dd, $J_1 = 2.3$ Hz, $J_2 = 11.0$ Hz, H5), 0.70 (3H, s, H24), 0.76 (3H, d, J
347 = 0.7 Hz, H25), 0.79 (3H, s, H23), 0.91 (3H, s, H26), 0.91 (3H, d, $J = 0.8$ Hz, H27), 1.07 (2H,

348 dt, $J_1 = 3.2$ Hz, $J_2 = 3.2$ Hz, $J_3 = 13.4$ Hz, H21), 1.63 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29),
349 1.88 (2H, dt, $J_1 = 3.3$ Hz, $J_2 = 3.3$ Hz, $J_3 = 13.5$ Hz, H16), 2.39 (1H, ddd, $J_1 = 3.7$ Hz, $J_2 = 11.5$
350 Hz, $J_3 = 13.0$ Hz, H13), 3.09 (1H, dt, $J_1 = 4.4$ Hz, $J_2 = 11.0$ Hz, $J_3 = 11.0$ Hz, H19), 3.12 (1H,
351 dd, $J_1 = 4.9$ Hz, $J_2 = 11.4$ Hz, H3), 4.31 (1H, dd, $J_1 = 5.8$ Hz, $J_2 = 15.0$ Hz, H1'), 4.47 (1H, dd,
352 $J_1 = 6.1$ Hz, $J_2 = 15.0$ Hz, H1'), 4.55 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 = 1.4$ Hz, $J_4 = 2.4$ Hz,
353 H30), 4.69 (1H, dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz, H30), 6.06 (1H, bt, $J =$
354 6.0 Hz, NH), 7.22 (1H, ddd, $J_1 = 0.9$ Hz, $J_2 = 4.9$ Hz, $J_3 = 7.9$ Hz, H6'), 7.64 (1H, bddd, $J_1 =$
355 1.7 Hz, $J_2 = 2.3$ Hz, $J_3 = 7.6$ Hz, H7'), 8.49 (1H, bdd, $J_1 = 1.7$ Hz, $J_2 = 4.9$ Hz, H5'), 8.50 (1H,
356 bdq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.3$ Hz, H3'). ^{13}C NMR: δ 14.62 (q, C27), 15.34
357 (q, C24), 16.02 (q, C25), 16.15 (q, C26), 18.26 (t, C6), 19.49 (q, C29), 20.90 (t, C11), 25.60
358 (t, C12), 27.39 (t, C2), 27.96 (q, C23), 29.41 (t, C15), 30.83 (t, C21), 33.70 (t, C16), 34.37 (t,
359 C22), 37.17 (s, C10), 37.71 (d, C13), 38.33 (t, C7), 38.70 (t, C1), 38.83 (s, C4), 40.42 (s, C8),
360 40.72 (t, C1'), 42.45 (s, C14), 46.63 (d, C19), 50.12 (d, C18), 50.61 (d, C9), 55.35 (d, C5),
361 55.67 (s, C17), 78.94 (d, C3), 109.42 (t, C30), 123.62 (d, C6'), 135.04 (s, C2'), 136.06 (d,
362 C7'), 148.41 (d, C5'), 148.83 (d, C3'), 150.77 (s, C20), 176.23 (s, C28). MS: m/z
363 547.3[M+H]⁺, 545.3 [M-H]⁻. IR (KBr; cm^{-1}): 3284 (N–H val. amide), 2933 (C–H val.), 2867
364 (–CH₃ val.), 1640 (–C=O val. amide). For C₃₆H₅₄N₂O₂ (546.83) calculated C (79.07), H
365 (9.95), N (5.12), found C (79.05), H (9.96), N (5.11). M.p. 254-256 °C. $[\alpha]_{\text{D}}^{20} = +11.1$ (c
366 0.307).

367 **6c**: ^1H NMR: δ 0.62 (1H, dd, $J_1 = 2.2$ Hz, $J_2 = 11.1$ Hz, H5), 0.71 (3H, s, H24), 0.77 (3H, bs,
368 H25), 0.82 (3H, s, H23), 0.92 (3H, bs, H27), 0.92 (3H, s, H26), 1.12 (2H, dt, $J_1 = 3.5$ Hz, $J_2 =$
369 3.5 Hz, $J_3 = 13.5$ Hz, H21), 1.64 (3H, dd, $J_1 = 0.7$ Hz, $J_2 = 1.4$ Hz, H29), 1.74 (2H, ddd, $J_1 =$
370 0.9 Hz, $J_2 = 8.1$ Hz, $J_3 = 12.3$ Hz, H7), 2.39 (1H, ddd, $J_1 = 3.6$ Hz, $J_2 = 11.6$ Hz, $J_3 = 12.8$ Hz,
371 H13), 3.09 (1H, dt, $J_1 = 4.6$ Hz, $J_2 = 11.1$ Hz, $J_3 = 11.1$ Hz, H19), 3.13 (1H, dd, $J_1 = 4.9$ Hz, $J_2 =$
372 11.4 Hz, H3), 4.32 (1H, dd, $J_1 = 6.0$ Hz, $J_2 = 15.6$ Hz, H1'), 4.45 (1H, dd, $J_1 = 6.1$ Hz, $J_2 =$

373 15.6 Hz, H1'), 4.55 (1H, dq, $J_1 = 1.4$ Hz, $J_2 = 1.4$ Hz, $J_3 = 1.4$ Hz, $J_4 = 2.4$ Hz, H30), 4.69 (1H,
374 dq, $J_1 = 0.7$ Hz, $J_2 = 0.7$ Hz, $J_3 = 0.7$ Hz, $J_4 = 2.4$ Hz, H30), 6.05 (1H, bt, $J = 6.0$ Hz, NH),
375 7.16-7.18 (2H, m, H4'), 8.49-8.51 (2H, m, H3'). ^{13}C NMR: δ 14.63 (q, C27), 15.33 (q, C24),
376 16.10 (q, C25), 16.16 (q, C26), 18.27 (t, C6), 19.49 (q, C29), 20.90 (t, C11), 25.60 (t, C12),
377 27.39 (t, C2), 27.96 (q, C23), 29.47 (t, C15), 30.82 (t, C21), 33.75 (t, C16), 34.42 (t, C22),
378 37.18 (s, C10), 37.71 (d, C13), 38.39 (t, C7), 38.71 (t, C1), 38.84 (s, C4), 40.76 (s, C8), 42.21
379 (t, C1'), 42.48 (s, C14), 46.59 (d, C19), 50.12 (d, C18), 50.61 (d, C9), 55.37 (d, C5), 55.73 (s,
380 C17), 78.96 (d, C3), 109.49 (t, C30), 122.47 (d, C3'), 148.49 (s, C2'), 149.78 (d, C4'), 150.70
381 (s, C20), 176.23 (s, C28). MS: m/z 547.3 $[\text{M}+\text{H}]^+$, 545.3 $[\text{M}-\text{H}]^-$. IR (KBr; cm^{-1}): 3352 (N-H
382 val. amide), 2942 (C-H val.), 2871 ($-\text{CH}_3$ val.), 1670 ($-\text{C}=\text{O}$ val. amide). For $\text{C}_{36}\text{H}_{54}\text{N}_2\text{O}_2$
383 (546.83) calculated C (79.07), H (9.95), N (5.12), found C (79.06), H (9.94), N (5.15). M.p.
384 246-248 $^\circ\text{C}$. $[\alpha]_{\text{D}}^{20} = +3.7$ (c 0.454).

385

386 2.7. Cell Cultures

387 The screening cell lines: T-lymphoblastic leukemia CEM; breast carcinoma MCF7; cervical
388 carcinoma HeLa; malignant melanoma G-361; and human foreskin fibroblasts BJ were
389 obtained from the American Type Culture Collection (Manassas, VA, USA). Cells were
390 cultured in DMEM (Dulbecco's Modified Eagle Medium, Sigma, MO, USA). Media used
391 were supplemented with 10% fetal bovine serum, 2 mM L-glutamine, and 1% penicillin-
392 streptomycin. The cell lines were maintained under standard cell culture conditions at 37 $^\circ\text{C}$
393 and 5% CO_2 in humid environment. Cells were subcultured twice or three times a week using
394 the standard trypsinization procedure.

395

396 2.8. Cytotoxicity screening tests

397 Description of the experimental procedure used in cytotoxicity test was already published

398 [1,2]. IC₅₀ values obtained with the compounds **3a–3c** and **6a–6c** are shown in Table 1.

399

400 2.9. Flow cytometry analysis

401 G-361 cancer cells were trypsinized, seeded in 100 mm culture dishes, and after 24 h of
402 stabilization incubated with the tested compounds. After additional 24 h, the cells were again
403 detached with trypsin, washed, fixed and stained in 0.1% [v/v] Triton X-100, 0.2 mg.mL⁻¹
404 RNase A and 10 µg.mL⁻¹ propidium iodide in PBS. Their DNA contents were then assessed
405 by a flow cytometer (FACS VerseTM, Becton Dickinson, NJ, USA), and the distribution of
406 cells in subG₁ (“apoptotic cells”), G₀/G₁, S and G₂/M peaks were quantified by histogram
407 analysis using BD FACSuite software (Becton Dickinson, NJ, USA). The experiments were
408 repeated three times. The differences between control and treated cells were analyzed by one-
409 way analysis of variance ANOVA using Microsoft Excel 2010. Signification was marked
410 with asterisk (p < 0.05).

411

412 2.10. SDS–polyacrylamide gel electrophoresis and immunoblotting

413 Cells (G-361) were seeded into culture medium in 100-mm culture dishes at a density of 1.5 ×
414 10⁶ cells.mL⁻¹ and after 24 h of stabilization treated with tested compounds. DMSO was used
415 as a vehicle for controls. After another 24 h of treatment, the cells were washed three times
416 with cold PBS (10 mM, pH 7.4) and lysed in ice-cold RIPA protein extraction buffer (20 mM
417 Tris-HCl, pH 7.4, 5 mM EDTA, 2 mM EGTA, 100 mM NaCl, 2 mM NaF, 0.2% Nonidet P-
418 40, 30 mM PMSF, 1 mM DTT, 10 mg.mL⁻¹ of aprotinin and leupeptin). The lysate was
419 collected into a microfuge tube and incubated on ice for 1 h. It was then cleared by
420 centrifugation at 18,000 × g for 30 min at 4 °C and the supernatant was collected. Proteins in
421 lysates were quantified by the Bradford method and diluted with Laemmli electrophoresis
422 buffer. The proteins were then separated on 10% or 12% SDS-polyacrylamide gels,

423 transferred to nitrocellulose membranes (Bio-Rad Laboratories, CA, USA) and stained with
424 Ponceau S (Sigma Aldrich, St. Louis, MO, USA) to check equal protein loading. The
425 membranes were blocked with 5% (w/v) non-fat dry milk and 0.1% Tween-20 in PBS for 2 h
426 and probed with specific primary antibodies (Cell Signaling Technology, Danvers, MA, USA;
427 Santa Cruz Biotechnology, CA, USA) overnight. Anti-caspase-3 and anti-caspase-7
428 recognized zymogen and also its fragment. After washing in PBS and PBS with 0.1% Tween-
429 20, the membranes were probed with horseradish peroxidase-conjugated secondary antibodies
430 and visualized with West Pico Supersignal chemiluminescent detection reagent (Thermo
431 Fisher Scientific, Rockford, USA). Signal was detected by a CCD camera (Fujifilm, Tokio,
432 Japan). To confirm equal protein loading, immunodetection was performed with anti- β -actin
433 monoclonal antibody (Santa Cruz Biotechnology, CA, USA). The experiments were repeated
434 three times. The protein expression in treated cells was compared to untreated controls.

435

436 2.11. Activities of caspase 3/7

437 Treated G-361 cells were harvested by centrifugation and homogenized in an extraction buffer
438 (10 mM KCl, 5 mM HEPES, 1 mM EDTA, 1 mM EGTA, 0.2% CHAPS, plus protease
439 inhibitors: aprotinin, leupeptin, PMSF; pH 7.4) on ice for 20 min. The resulting homogenates
440 were clarified by centrifugation at $18,000 \times g$ for 20 min at 4°C . The protein contents of the
441 samples were quantified by the Bradford method and they were diluted to equivalent protein
442 concentrations. Lysates were then incubated for 1 h with 100 mM Ac-DEVD-AMC as a
443 substrate (Sigma, MO, USA) in an assay buffer (25 mM PIPES, 2 mM EGTA, 2 mM MgCl_2 ,
444 5 mM DTT, pH 7.3). For negative controls, the lysates were supplemented with 100 mM Ac-
445 DEVD-CHO (Sigma, MO, USA) as a caspase-3/7 inhibitor. The fluorescence of the product
446 was measured using a Fluoroskan Ascent microplate reader (Labsystems, Vantaa, Finland) at
447 346/442 nm (ex/em). Experiments were repeated three times in triplicates. The differences

448 between control and treated cells were analyzed by one-way analysis of variance ANOVA
449 using Microsoft Excel 2010. Signification was marked with asterisk ($p < 0.05$).

450

451 **3. Results and Discussion**

452 3.1. Synthetic protocol

453 The synthesis of both series of the target picolyl amides (**3a–3c** and **6a–6c**) is efficient and
454 easy. To get **3a–3c** (Scheme 1), betulinic acid (**1**) was first esterified by succinic anhydride
455 [2,26] to prepare its hemiester **2**, which was then subjected to a formation of an amide bond
456 with *o*-, *m*- and *p*-picolyl amines, using T3P as condensation agent [2,27]. To get **6a–6c**
457 (Scheme 2), protecting of the C(3)-OH group as acetate ester was required, and achieved by
458 reacting betulinic acid (**1**) with acetic anhydride in dry THF, using EDIPA as a base and
459 DMAP as reaction promotor [2]. Free carboxylic group of **1** was then converted to betulinic
460 acyl chloride by means of oxalyl chloride in dichloromethane [2,28], and used subsequently
461 without isolation in a formation of amide with *o*-, *m*- and *p*-picolyl amines, using again
462 EDIPA as a base [27], affording **5a–5c**. Protecting acetate group was removed by LiOH.H₂O
463 in methanol [2], yielding the required amides **6a–6c**.

464 *Cytotoxicity*: The experimental data obtained (Table 1) revealed high importance of **3b** and
465 **3a**, while **3c** and **6a–6c** displayed only moderate cytotoxicity. The cytotoxicity data of **3a–3c**
466 clearly demonstrate the importance of substitution of pyridine ring in structure-activity
467 relationship studies: while *o*- and *m*-substitution of the pyridine ring resulted in highly
468 cytotoxic compounds **3a** and **3b**, *p*-substitution revealed its disadvantage in **3c**. Both cytotoxic
469 compounds, **3a** and **3b**, showed very low cytotoxicity in normal fibroblasts, and became
470 potential candidates for cytotoxic agents against G-361 ($2.4 \pm 0.0 \mu\text{M}$ and $0.5 \pm 0.1 \mu\text{M}$,
471 respectively), MCF7 ($2.2 \pm 0.2 \mu\text{M}$ and $1.4 \pm 0.1 \mu\text{M}$, respectively), HeLa ($2.3 \pm 0.5 \mu\text{M}$ and
472 $2.4 \pm 0.4 \mu\text{M}$, respectively) and CEM ($6.9 \pm 0.4 \mu\text{M}$ and $6.5 \pm 1.5 \mu\text{M}$, respectively) cancer

473 cell lines. Moreover, **3b** was the most active in melanoma cancer cell line G-361 (IC_{50} $0.5 \pm$
474 $0.1 \mu\text{M}$) compared to other tested cancer cell lines with weak activity in normal human cells
475 (BJ) that is important for wide therapeutic window. We determined selectivity against all
476 tested cancer cells compared to normal cells with therapeutic index between 7 and 100 for
477 compounds **3a** and **3b**. The highest therapeutic index ($TI = 100$) was calculated for human
478 malignant melanoma cell line (G-361) *versus* normal human fibroblasts (BJ) with **3b**.
479 Difluoroderivatives of dihydrobetulinic acid showed selectivity with therapeutic index
480 between 2 and 8 against cancer lines [29]. In another study, therapeutic index of 9.1 was
481 measured for conjugate of (3 β)-3-*O*-acetylbetulinic acid with a triazole ring further
482 substituted by benzaldehyde [30]. Based on the literature search, no derivative of betulinic
483 acid was as highly active in melanoma cancer cell line G-361 as **3b**. In a future, we would like
484 to design other compounds with higher selectivity, mainly based on the structure of **3b**.
485 Together with obtained results, these derivatives could be promising for *in vivo* experiments
486 for development new anticancer drugs.
487 Concerning **6a–6c**, their lower cytotoxicity has been expected due to the substitution of
488 carboxyl group in these derivatives of betulinic acid. However, the great disadvantage of **6a–**
489 **6c** is their cytotoxicity in normal fibroblasts, which is sometimes even higher than that in the
490 cancer cell lines. The cytotoxicity of **6a–6c** in CEM, MCF7 and HeLa cancer cell lines is
491 comparable with the cytotoxicity of the compounds investigated earlier [1]. The equal
492 moderate cytotoxicity with no selectivity as for **6a–6c** was also shown for saponins of
493 betulinic acid in different cancer cell lines [33].

494

495 3.2. Picolyl amides of betulinic acid caused apoptosis

496 We have observed whether compounds **3a–3c** influence the cell cycle and cause apoptosis in
497 human malignant melanoma cells G-361. Cells were treated with 0.5, 2.5, 12.5 or 25 μM of

498 three substances (**3a–3c**) for 24 h. Then the cells were harvested, stained and analyzed by
499 flow cytometry, western blotting or caspase activation assay. Flow cytometric analysis
500 showed an increase in the subG₁ phase of the cell cycle (apoptotic cells) in G-361 cell line
501 after treatment with **3a** and **3b**. In turn, **3c** did not enhance the number of subG₁ cells and
502 caused no changes in the cell cycle distribution. In these cells, treatment with **3a** and **3b**
503 increased the proportion of cells in the G₀/G₁ phase and decreased their proportion in the
504 concomitant phases, mainly G₂/M (Figure 1). Apoptotic markers were detected by western
505 blotting to examine the antiproliferative activities of tested compounds **3a–3c**. The initiation
506 of apoptosis is mediated by caspase cascade activation [34]. Caspase-3 and caspase-7 are
507 executioner proteases that result in cleavage of poly-(ADP-ribose) polymerase (PARP) and
508 subsequent DNA degradation and apoptotic death [35]. The western blots showed degradation
509 of caspase-7 into cleaved fragments (a part of the apoptotic cascade) after treatments with **3a**
510 and **3b** in G-361 cells (Figure 2). PARP cleavage, a final step of caspase activation, was
511 observed in human malignant melanoma cells treated by **3a** and **3b** (Figure 2). pRb was
512 downregulated by 12.5 μ M and 25 μ M of **3a** and **3b**, but total Rb also decreased in those
513 treatments. It is probably due to reaching the cytotoxicity towards human melanoma cells. No
514 changes in the expressions of anti-apoptotic proteins Bcl-2 and Mcl-1 and also p53 were
515 observed. Downregulation of phospho-p44/42 MAPK (Erk 1/2) (Thr202/Tyr204) was
516 detected after the treatment with 12.5 and 25 μ M of **3a** and **3b** for 24 h. Total level of p44/42
517 was not changed. This kinase is responsible for proliferation, differentiation, motility, and
518 death of cells. The compounds **3a** and **3b** inhibited those activities mediated by Erk 1/2. These
519 data were supplemented by estimation of caspase-3/7 activity in G-361 treated cells using the
520 fluorogenic substrate Ac-DEVD-AMC and/or caspase-3/7 inhibitor Ac-DEVD-CHO.
521 Compound **3a** increased activity of caspase-3/7 after the treatment with 25 μ M up to eighteen
522 fold. Compound **3b** induced the activity of caspase-3/7; after 24 h treatment with

523 concentration 25 μ M, a twelve fold increase in the effector caspases was observed compared
524 to the untreated control (Figure 3). Unlike **3a** and **3b**, compound **3c** did not affect the activity
525 of caspases-3/7 after 24 h. These results show that **3a** and **3b** can induce apoptosis by caspase-
526 3 activation, and target proliferation and differentiation in G-361 cells.

527

528 3.3. Physico-chemical parameters

529 To support rational design of the target compounds investigated in this work, their physico-
530 chemical parameters have been calculated using ACD/iLabs software and databases [36]. The
531 parameters calculated for **3a–3c** and **6a–6c** were compared with the Lipinski [5] rule of five
532 and with the Ghose [6] rules. The rules describe molecular properties important for a small
533 molecule drug pharmacokinetics in the human body, including their absorption, distribution,
534 metabolism and excretion (known as ADME parameters). However, the rules do not predict
535 displaying of the pharmacological activity. Lipinski [5] rule of five considers partition
536 coefficient ($\log P$, range -0.4 to +5.6), molar refractivity (range 40 to 130), molecular weight
537 (range 180 to 500), number of atoms in the molecule (20 to 70) and polar surface area (up to
538 14 nm). The data for comparison are presented in Table 2. However, the target compounds
539 **3a–3c** and **6a–6c** do not correspond to the range given for any of the above Lipinski and
540 Ghose rules [5,6], except of the parameter giving the number of atoms in the molecule.
541 Despite those facts, **3a–3c** and **6a–6c** display cytotoxicity. Exceptions are already known,
542 where pharmacologically active compounds do not correspond to all rules, we have observed
543 such result as well [1,2].

544 The most important molecular descriptors are the partition coefficient ($\log P$) and the
545 distribution coefficient ($\log D$) [37]. In chemical and pharmaceutical sciences, both, $\log P$ and
546 $\log D$ are measures of hydrophilicity or hydrophobicity of the studied compound, and are
547 useful for estimating distribution of a drug within the body, where $\log D$ shows the

548 dependence on the pH of the matrix. Hydrophobic drugs are then preferentially distributed to
549 hydrophobic compartments (e.g., lipid bilayer of cells), while hydrophilic drugs are
550 preferentially distributed to hydrophilic compartments (e.g., blood serum). The distribution
551 coefficient is a pH dependent value, and, therefore, the value at pH = 7.4 (the physiological
552 pH value of blood serum) is of particular importance (see also Table 2). Thus, $\log P$ expresses
553 a ratio of concentrations of non-ionized compound between two phases, non-polar (octanol)
554 and polar (water), while $\log D$ expresses the ratio of the sum of the concentrations of all forms
555 of the compound (ionized and non-ionized) in each of the two phases. In pharmacology, $\log P$
556 and $\log D$ indicate how easily the drug can reach its intended target in the body, how strong
557 its effect will be once it reaches its target and how long it will remain in the body in an active
558 form. The $\log P$ values calculated for either **3a–3c** or **6a–6c** are identical, and they exceed the
559 values given by the Lipinski and Ghose rules [5,6] (Table 2). Another supportive parameter is
560 the predicted aqueous solubility, $\log S$. S (in mol dm⁻³) is the concentration of the solute in a
561 saturated solution that is in equilibrium with the crystalline solid, and it is a pH dependent
562 parameter. Standard range for $\log S$ at pH 7.4 is $-6.5/+0.5$. Among the prepared compounds,
563 only **3a–3c** show values for $\log S$ in this range (Table 2).

564

565 3.4. ADME parameters:

566 The importance of some of the ADME parameters for evaluation of pharmacokinetic
567 properties of the prepared compounds is summarized in Table 2. The blood brain barrier
568 (BBB) and plasma protein binding are two of the important factors affecting distribution of
569 the compound in the human body. Several parameters assist in evaluation of each potential
570 drug for its BBB transport [38] The rate of brain penetration, $\log PS$, is a logarithm of the
571 permeability-surface area coefficient that measures the ability of a drug to cross the BBB and
572 to move into brain tissue over time. It is one of the relevant parameters for evaluation of the

573 rate of BBB penetration. The extent of brain penetration parameter, $\log PB$, indicates if the
574 drug might be active or inactive on the central nervous system (CNS). The typical values for
575 $\log PB$ are $-1.5/+1.5$. Most of the current drugs show the $\log PB$ value up to $+2$ (active on
576 CNS) or down to -2 (inactive on CNS). The $\log PB$ values calculated for **3a–3c** and **6a–6c**
577 indicate that these compounds will be inactive on CNS (Table 2). However, the complex of
578 the BBB parameters is completed by the brain/plasma equilibration rate, the parameter
579 expressed as $\log PS * f_{u, brain}$ that is a mathematical modeling parameter based on time
580 required for reaching brain equilibrium. It is dependent on the brain unbound fraction ($f_{u, brain}$).
581 This value indicates if the drug may potentially be active on CNS together with the $\log BB$
582 parameter, the predicted brain/blood partition coefficient (cf. Supplementary material). The
583 parameter $\log BB$ is a hybrid parameter determined by permeability, plasma and brain tissue
584 binding, and active transport mechanism, standard range $-3.0/+1.2$. The $\log BB$ values
585 calculated for **3a–3c** and **6a–6c** appear in the required range of this parameter. Nevertheless,
586 there are numerous exceptions already known. A supportive parameter, plasma protein
587 binding (PPB) gives calculated quantity of a drug bound to a protein.

588 Another important ADME parameter is bioavailability. Among the prepared compounds, the
589 calculated bioavailability for **3a–3c** is lower than that for **6a–6c**. However, the experimental
590 data show that **3a–3c** are generally more cytotoxic to the tested cancer cell lines than **6a–6c**.
591 The rules [5,6] describe molecular properties important for a small molecule drug
592 pharmacokinetics in the human body, including their absorption, distribution, metabolism and
593 excretion (ADME parameters). However, the rules do not predict whether a compound will
594 display pharmacological activity. Nevertheless, the importance of several ADME parameters,
595 namely blood-brain barrier and plasma protein binding, reflects the distribution of the
596 compound in the human body [39]. A comparison of the calculated physico-chemical
597 properties with the measured cytotoxicity shows that every time a new class of compound is

598 being investigated, no available experimental screening may be intentionally omitted (Tables
599 1 and 2).

600

601 3.5. A comparison of **3a–3c** and **6a–6c** with the compounds prepared earlier [1, 2]

602 Earlier prepared compound, bearing picolyl amine moiety in the molecule, derived either
603 from lanosterol, cholesterol or sitosterol, displayed cytotoxicity in the range of $IC_{50} = 25\text{--}45$
604 μM , or were inactive ($IC_{50} > 50 \mu\text{M}$). The earlier published data also show that the physico-
605 chemical and ADME parameters calculated for those compounds, corresponded to the
606 Lipinski and Ghose rules [5,6] partially only. The current compounds **3a–3c** and **6a–6c** show
607 generally higher cytotoxicity than the earlier series of compounds. Only **3a** and **3b** display
608 considerable cytotoxicity, and **3b** show the IC_{50} value on G361 in the range $IC_{50} = 0.5 \pm 0.1$
609 μM , with the therapeutic index $TI = 100$. This finding augments the importance of this
610 particular compound **3b**. In turn, even **3a** and **3b** corresponded to a limited number of
611 calculated physico-chemical and ADME parameters. Any calculation of those parameters is
612 useful and often desired in the drug design. However, the conclusions of the calculations
613 should always be supported by the experimental data for confirmation or correction.

614

615 4. Conclusion

616 Amides **3b** and **3a** represent the most important members of this series of compounds because
617 of their high cytotoxicity against human melanoma cell line G-361 (Table 1). The former of
618 this couple of amides (**3b**) is active in concentration $0.5 \pm 0.1 \mu\text{M}$, and, subsequently it does
619 not damage healthy cells. It also displays high cytotoxicity in other tested cell lines.

620 Investigating structure-activity relationship in dependence on substitution of pyridine ring, the
621 conclusions are different for each subseries. In the former subseries (**3a–3c**), the cytotoxicity
622 of amides bearing *m*- and *o*-substituted pyridine ring (**3b** and **3a**) showed high cytotoxicity in

623 all tested cell lines, while *p*-substitution of the pyridine ring (**3c**) resulted in weak cytotoxicity
624 (Table 1). Our hypothesis is that this result might be connected with different mechanism of
625 action of **3b** and **3a** in comparison with **3c** due to the modifications in their structures. This
626 hypothesis will be focused in a future investigation, because its confirmation may be
627 important for better understanding of structure-activity relationship. However, in the latter
628 subseries (**6a–6c**), amides bearing *p*- and *o*-substituted pyridine ring showed either
629 comparable cytotoxicity, or *p*-substituted amide was more active, and the *m*-substituted
630 derivative displayed the lowest cytotoxicity (Table 1). The most important finding with **6a–6c**
631 was their high activity in the normal human cells. Due to that finding, **6a–6c** are much less
632 suitable anti-cancer agents than **3a–3c** that are, in turn, potential anti-cancer candidate agents.
633
634 Supplementary material is available on the web page of the journal.

635
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771 Abbreviations: CEM, cells of human T-lymphoblastic leukemia; MCF7, cells of human breast
772 adenocarcinoma; HeLa, cells of human cervical cancer; G-361, cells of human malignant
773 melanoma; BJ, normal human fibroblasts; T3P, propylphosphonic anhydride; EDIPA, ethyl
774 diisopropyl amine.

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796 [Scheme and Figure Captions]

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798 Scheme 1

799 i: succinic anhydride, DMAP, dry pyridine; ii: 2-, 3- or 4-aminomethylpyridine, T3P, dry
800 pyridine

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802 Scheme 2

803 i: acetic anhydride, EDIPA, DMAP, THF; ii: oxalyl chloride, dichloromethane; iii: 2-, 3- or 4-
804 aminomethylpyridine, EDIPA, dichloromethane; iv: LiOH.H₂O, methanol

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806 Figure 1

807 Histograms with the distributions of G-361 cells in the G₀/G₁, S, and G₂/M cell cycle phases

808 (A), and the subG₁ fraction of cells (B) measured by flow cytometric analysis, after 24 h

809 treatment with **3a**, **3b** and **3c** relative to untreated controls. Data indicate the percentages (%)

810 of the number of cells in respective phases. Experiments were repeated three times in

811 triplicates. Error bars are omitted for clarity. (A) Analysis of variance (one-way ANOVA)

812 between control and treated cells failed due to high biological variability in replicates. (B)

813 Differences between control and treated cells among subG₁ phase were significant ($p < 0.05$)

814 *.

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816 Figure 2

817 Western blot analysis of apoptotic proteins (PARP, pRb, Rb, zymogen and fragment of

818 caspase-7, pAKT, AKT, pp44/42 MAPK (Erk 1/2) (Thr202/Tyr204), p44/42 MAPK (Erk 1/2)

819 in human malignant melanoma cells G-361 treated with amides. The expression of proteins in

820 cells treated with 0.5, 2.5, 12.5 and 25 μ M of compounds **3a**, **3b** and **3c** for 24 h were

821 compared with the protein expression of untreated control cells (0+ means control with
822 DMSO). The expression of β -actin was used as a protein loading marker.

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824 Figure 3

825 Activity of caspase-3/7 in malignant melanoma cells G-361 treated with 0.5, 2.5, 12.5 and 25
826 μ M amides **3a**, **3b** and **3c** compared with untreated control cells for 24 h. The data indicate
827 the relative increase in caspase-3/7 activity. Experiments were repeated three times in
828 triplicates. * Differences between control and treated cells were significant ($p < 0.05$).

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846 Table 1. IC₅₀ [μM] values in four cancer cell lines and normal human fibroblasts for **3a–3c**
 847 and **6a–6c** after 72 h

Compound	Cytotoxicity (IC ₅₀ [μM])					TI ^a
	CEM ^b	MCF7 ^c	HeLa ^d	G-361 ^e	BJ ^f	
3a	6.9 ± 0.4	2.2 ± 0.2	2.3 ± 0.5	2.4 ± 0.0	46.2 ± 2.8	19.3
3b	6.5 ± 1.5	1.4 ± 0.1	2.4 ± 0.4	0.5 ± 0.1	50.0 ± 0.0	100.0
3c	22.6 ± 5.9	>50	40.9 ± 4.7	32.1 ± 2.1	>50	ND ^g
6a	18.6 ± 1.0	27.0 ± 5.5	14.7 ± 0.2	16.4 ± 1.6	15.8 ± 1.4	1.0
6b	25.3 ± 6.9	38.7 ± 1.0	23.6 ± 2.3	18.1 ± 0.1	17.5 ± 2.7	1.0
6c	18.6 ± 3.7	21.1 ± 4.3	14.8 ± 0.2	11.6 ± 1.6	11.2 ± 1.4	1.0

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849 ^a Therapeutic index (TI) calculated for G-361 line *versus* fibroblasts BJ; ^b CEM, cells of
 850 human T-lymphoblastic leukemia; ^c MCF7, cells of human breast adenocarcinoma; ^d HeLa,
 851 cells of human cervical cancer; ^e G-361, human malignant melanoma cell line ^f BJ, normal
 852 human fibroblasts; ^g ND = not determined.

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863 Table 2. Physico-chemical and ADME parameters of the target compounds **3a–3c** and **6a–6c** calculated using the ACD/iLabs software [36]

Compd. or ref. No.	MW	Physico-chemical and ADME parameters ^a									
		log <i>P</i>	log <i>D</i> (pH 7.4)	log <i>S</i> (pH 7.4)	bioav. [%]	log <i>PS</i> * <i>f_{u, brain}</i>	log <i>PS</i>	log <i>PB</i>	log <i>BB</i>	PPB [%]	H _{acc} / H _{don} / n.m.b.
3a	646.90	7.56	4.78	-5.01	< 30	-5.8	-4.0	-0.89	-0.0	99.80	7/2/9
3b	646.90	7.56	5.26	-5.04	< 30	-5.8	-3.9	-0.89	-0.0	99.80	7/2/9
3c	646.90	7.56	5.07	-5.48	< 30	-5.8	-3.9	-0.88	-0.0	99.80	7/2/9
6a	546.83	7.00	7.00	-6.76	30-70	-4.1	-1.8	-0.30	-0.0	99.75	4/2/4
6b	546.83	7.00	7.00	-6.61	30-70	-4.1	-1.8	-0.30	-0.0	99.75	4/2/4
6c	546.83	7.00	6.99	-6.58	30-70	-4.1	-1.8	-0.30	-0.0	99.75	4/2/4
ref. [32]	max. 500	max. 5.0									max. 10 / max. 5 / -
ref. [33]	Max. 500	Max. 5.6									

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865 ^a $\log P$ – partition coefficient; $\log D$ – distribution coefficient; $\log S$ – predicted aqueous solubility; bioav. = bioavailability – the degree of
866 availability of a chemical by the target tissue; $\log PS * f_{u, brain}$ – the brain/plasma equilibration rate, the parameter that is a mathematical modeling
867 parameter based on time required for reaching brain equilibrium; $\log PS$ – logarithm of the permeability-surface area coefficient; $\log PB$ – the
868 extent of brain penetration parameter; $\log BB$ – a hybrid parameter determined by permeability, plasma and brain tissue binding, and active
869 transport mechanism; PPB – plasma protein binding; $H_{acc} / H_{don} / n.m.b.$ = number of H-bond acceptors / number of H-bond donors / number of
870 movable bonds.

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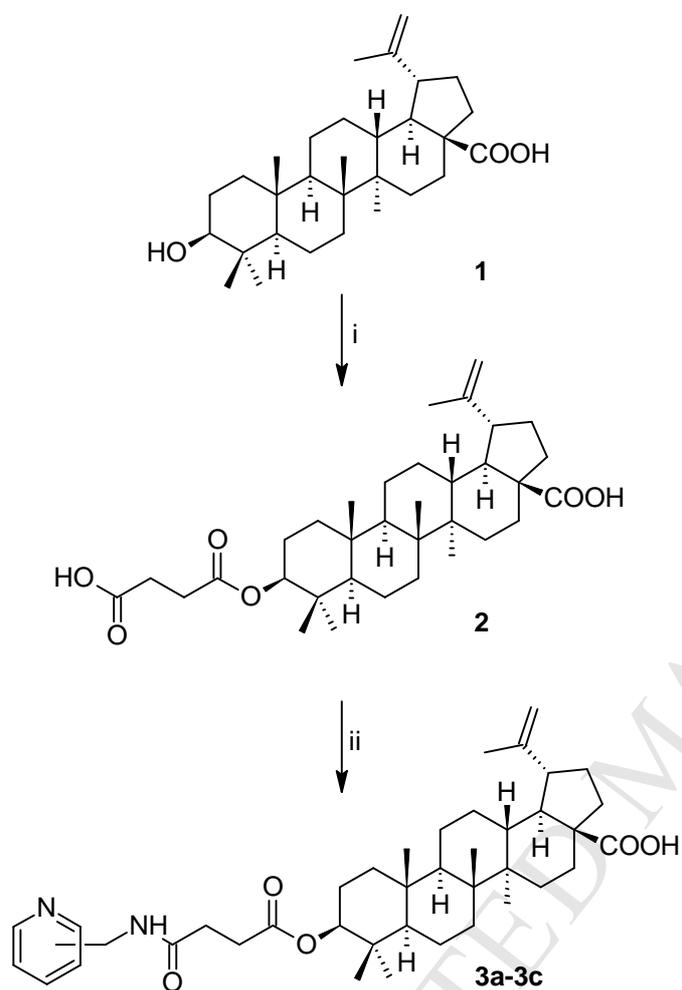
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881 Scheme 1

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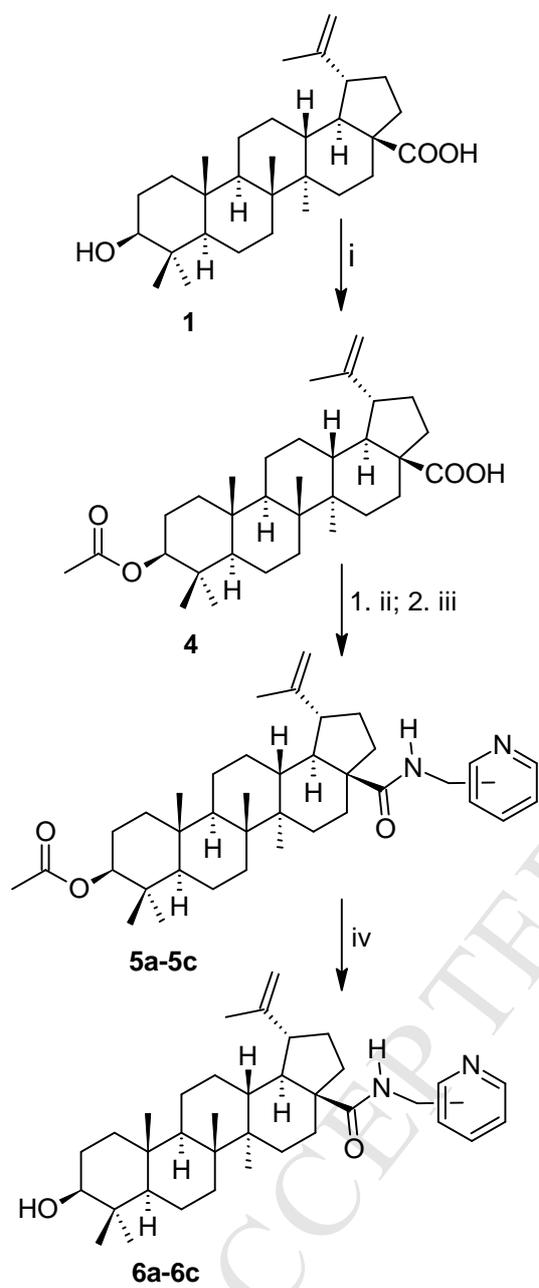
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893 Scheme 2

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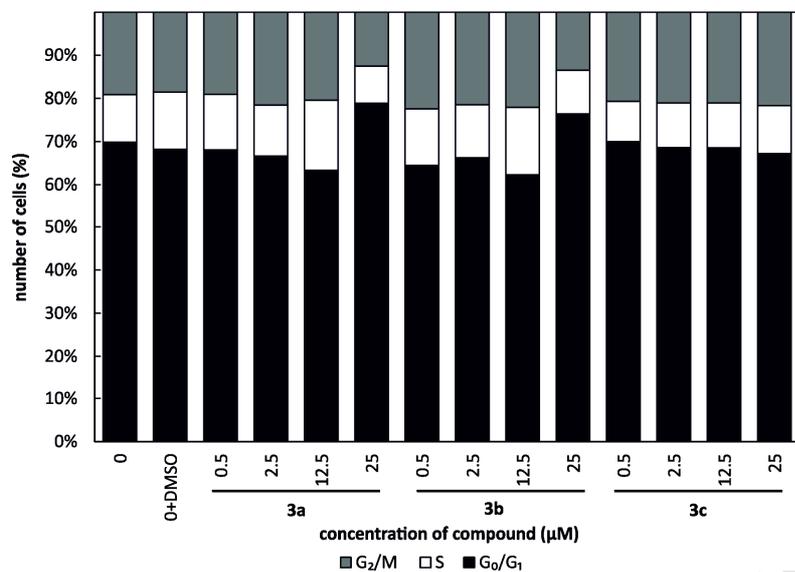
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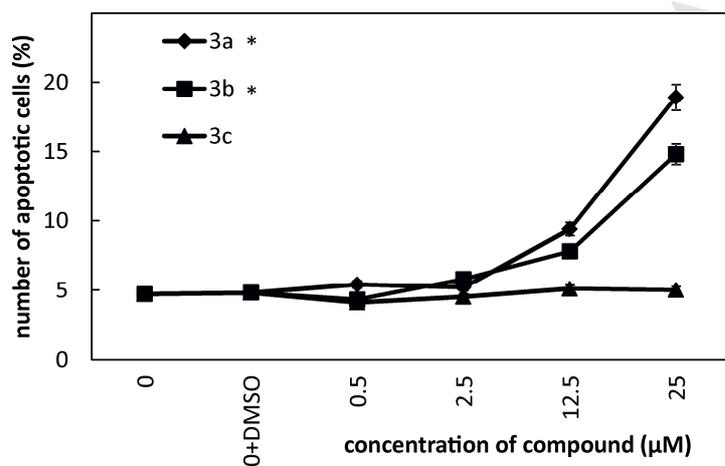
901 Figure 1

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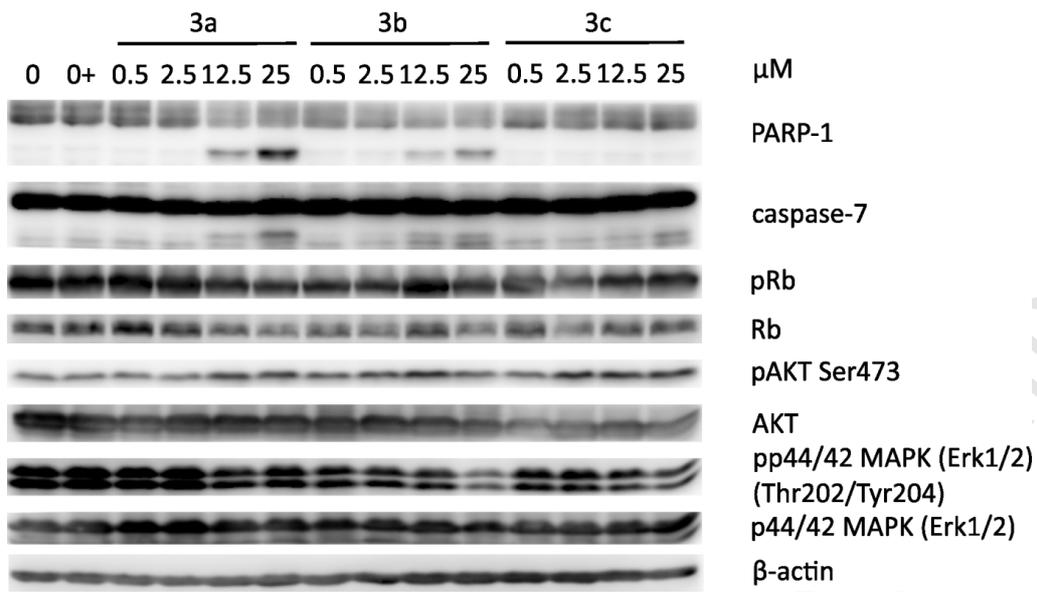
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914 Figure 2



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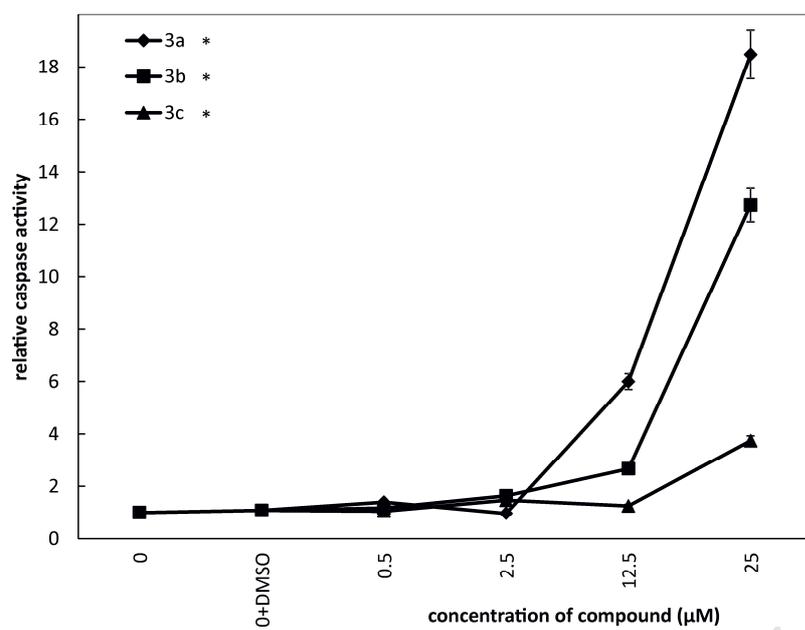
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932 Figure 3



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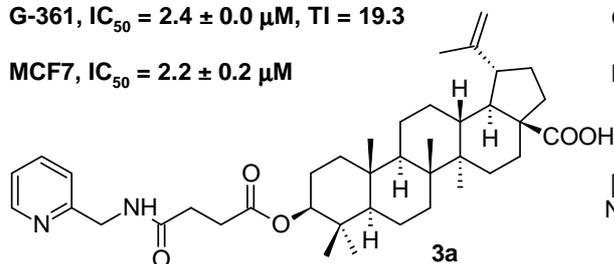
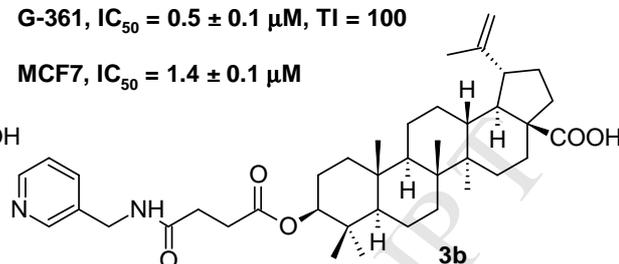
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950 Graphical Abstract

951

G-361, $IC_{50} = 2.4 \pm 0.0 \mu M$, TI = 19.3**MCF7, $IC_{50} = 2.2 \pm 0.2 \mu M$** **3a****G-361, $IC_{50} = 0.5 \pm 0.1 \mu M$, TI = 100****MCF7, $IC_{50} = 1.4 \pm 0.1 \mu M$** **3b**

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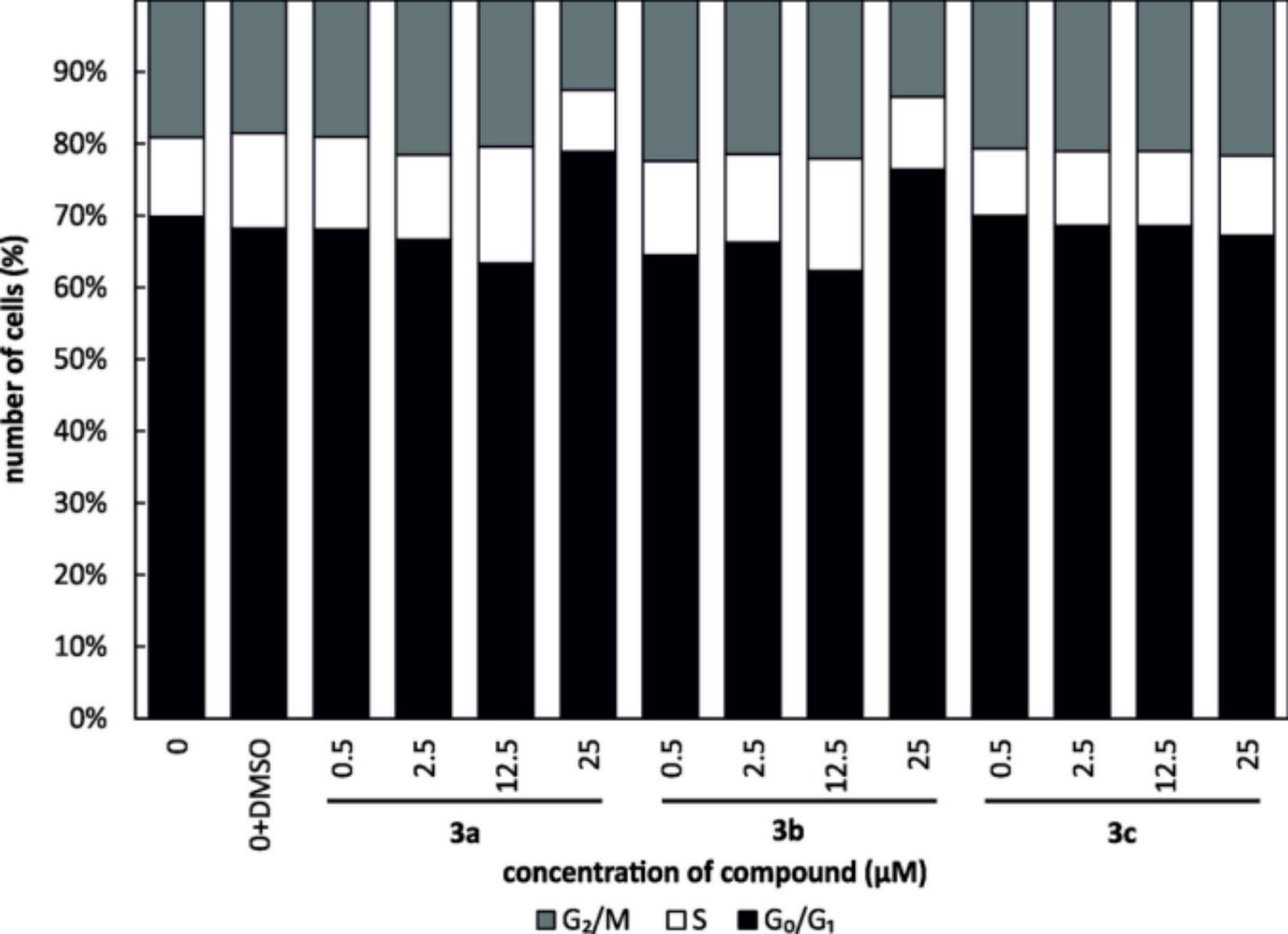
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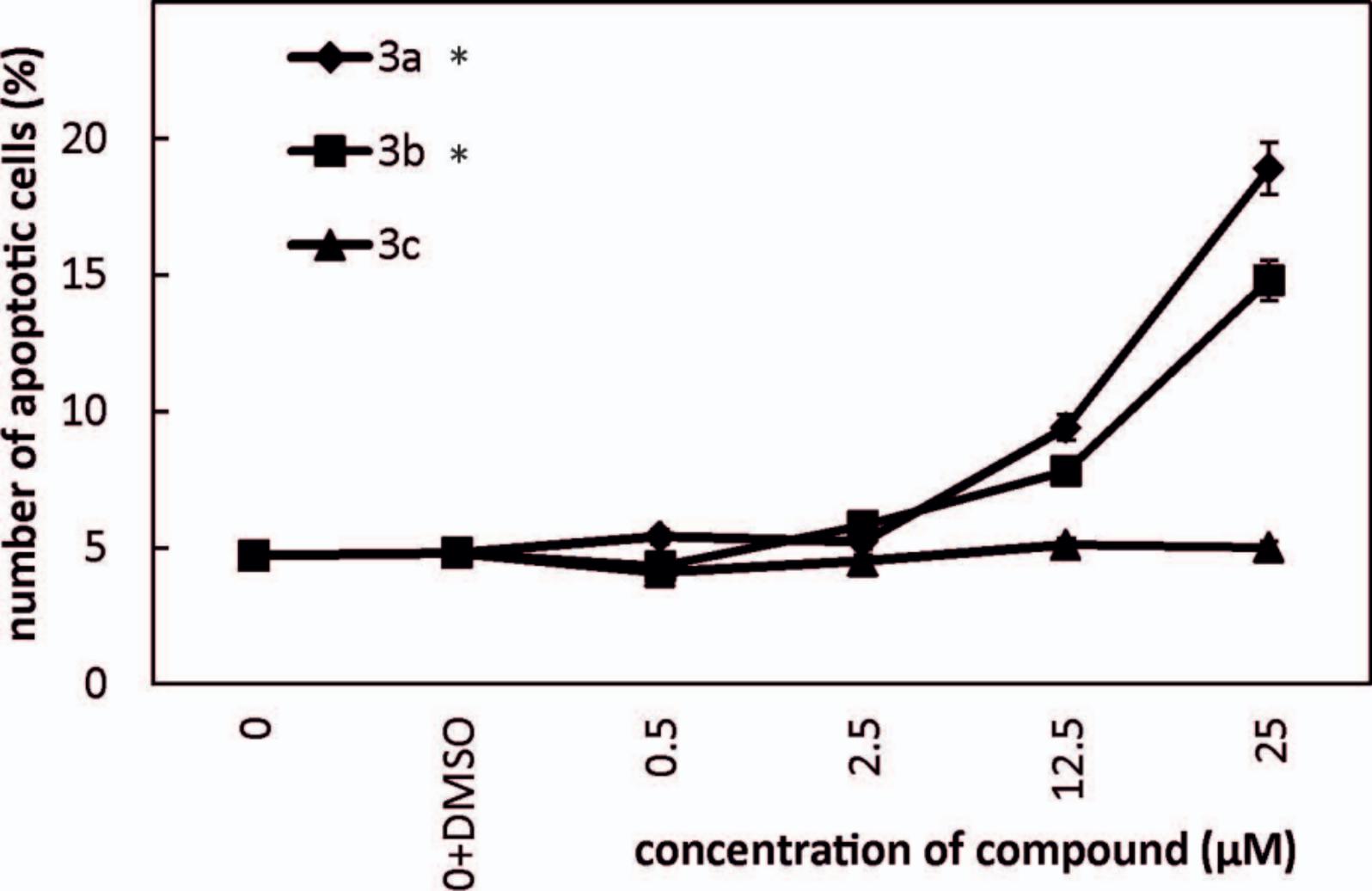
Table 1. IC₅₀ [μM] values in four cancer cell lines and normal human fibroblasts for **3a–3c** and **6a–6c** after 72 h

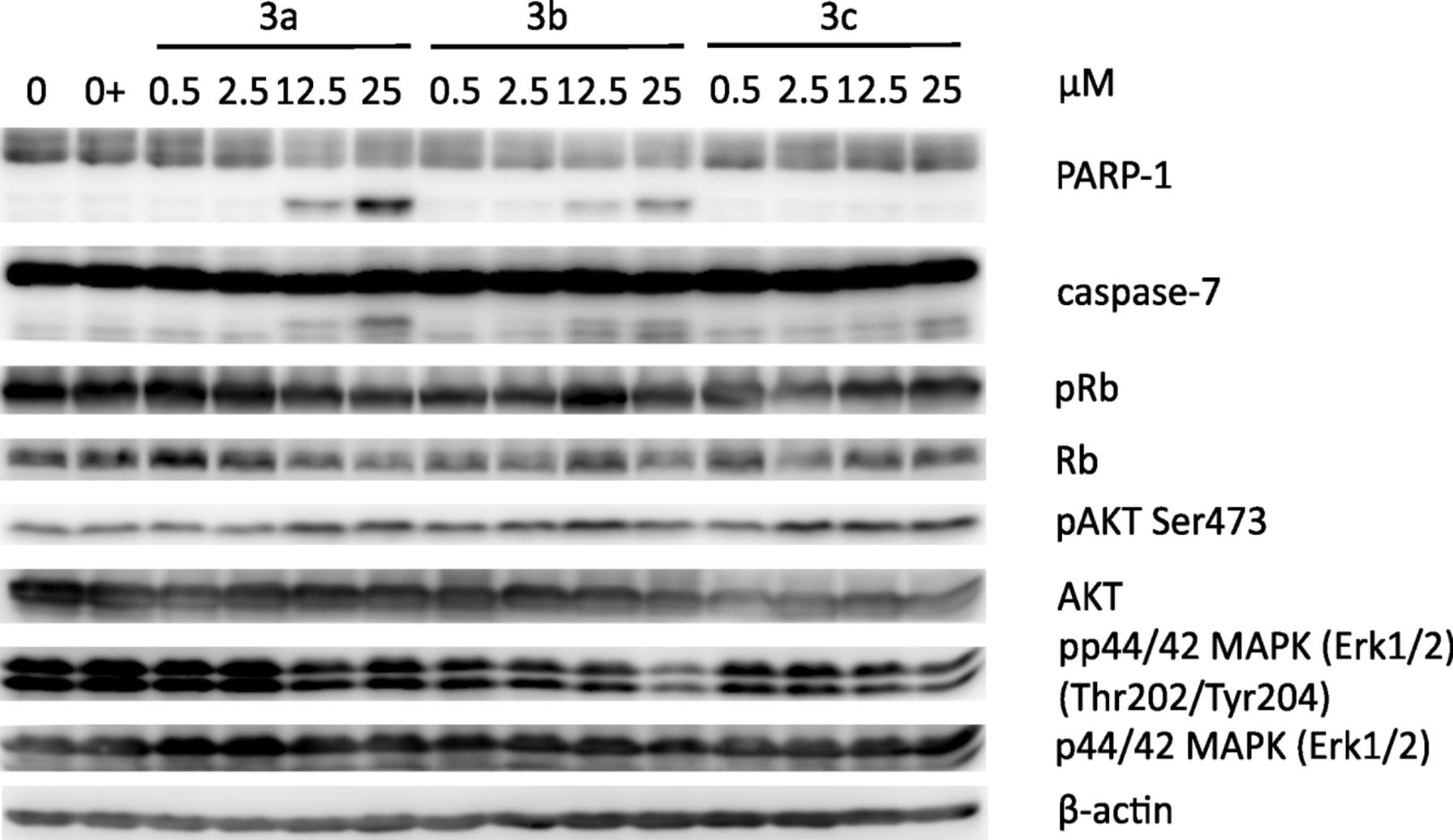
Compound	Cytotoxicity (IC ₅₀ [μM])					TI ^a
	CEM ^b	MCF7 ^c	HeLa ^d	G-361 ^e	BJ ^f	
3a	6.9 ± 0.4	2.2 ± 0.2	2.3 ± 0.5	2.4 ± 0.0	46.2 ± 2.8	19.3
3b	6.5 ± 1.5	1.4 ± 0.1	2.4 ± 0.4	0.5 ± 0.1	50.0 ± 0.0	100.0
3c	22.6 ± 5.9	>50	40.9 ± 4.7	32.1 ± 2.1	>50	ND ^g
6a	18.6 ± 1.0	27.0 ± 5.5	14.7 ± 0.2	16.4 ± 1.6	15.8 ± 1.4	1.0
6b	25.3 ± 6.9	38.7 ± 1.0	23.6 ± 2.3	18.1 ± 0.1	17.5 ± 2.7	1.0
6c	18.6 ± 3.7	21.1 ± 4.3	14.8 ± 0.2	11.6 ± 1.6	11.2 ± 1.4	1.0

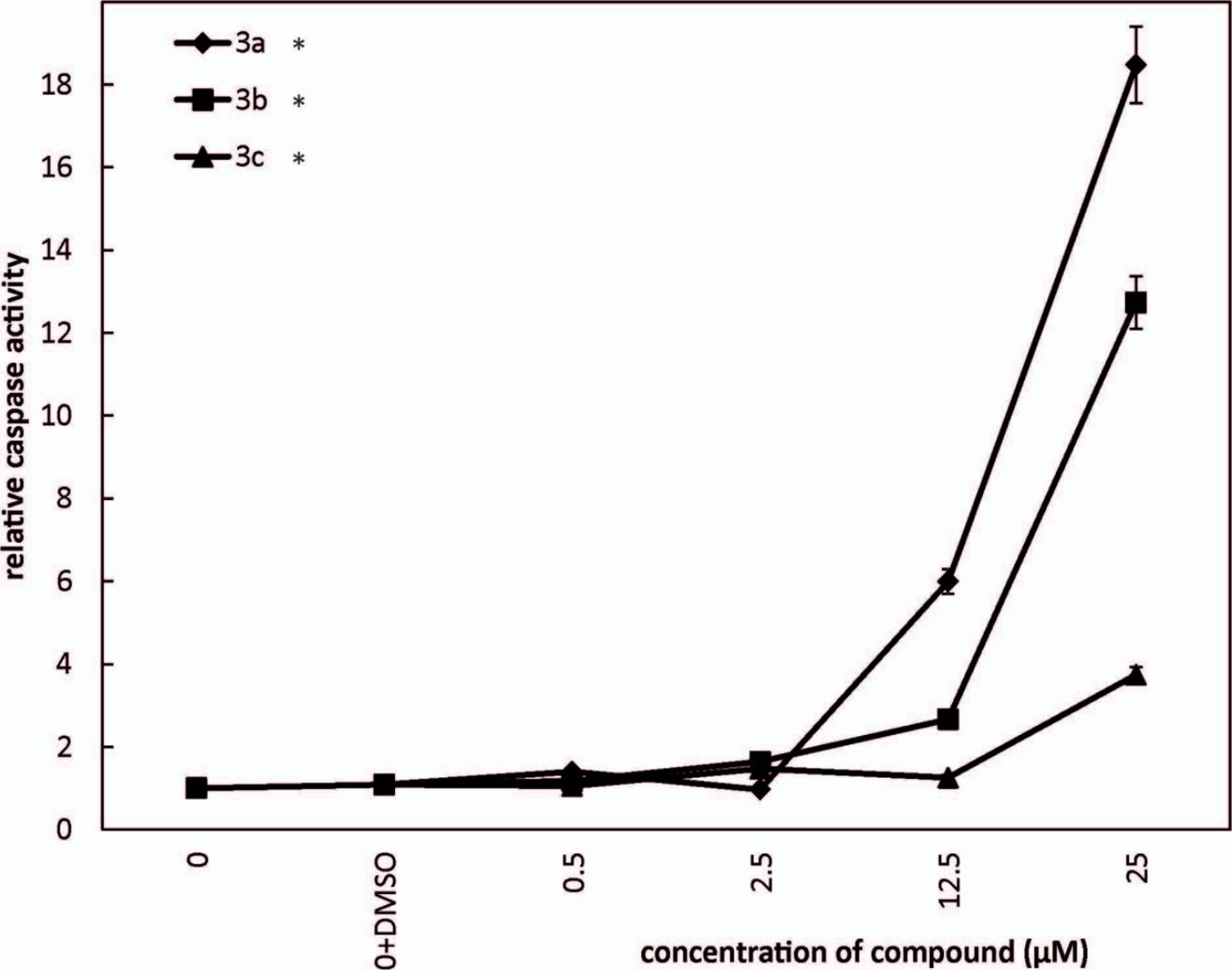
^a Therapeutic index (TI) calculated for G-361 line *versus* fibroblasts BJ; ^b CEM, cells of human T-lymphoblastic leukemia; ^c MCF7, cells of human breast adenocarcinoma; ^d HeLa, cells of human cervical cancer; ^e G-361, human malignant melanoma cell line ^f BJ, normal human fibroblasts; ^g ND = not determined

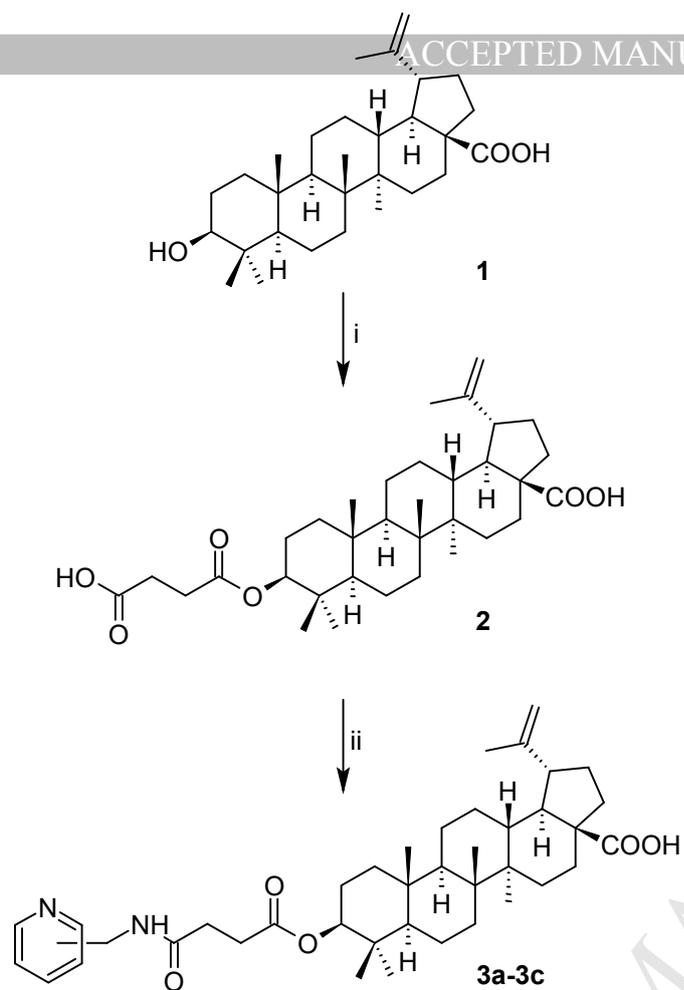
^a $\log P$ – partition coefficient; $\log D$ – distribution coefficient; $\log S$ – predicted aqueous solubility; bioav. = bioavailability – the degree of availability of a chemical by the target tissue; $\log PS * f_{u, brain}$ – the brain/plasma equilibration rate, the parameter that is a mathematical modeling parameter based on time required for reaching brain equilibrium; $\log PS$ – logarithm of the permeability-surface area coefficient; $\log PB$ – the extent of brain penetration parameter; $\log BB$ – a hybrid parameter determined by permeability, plasma and brain tissue binding, and active transport mechanism; PPB – plasma protein binding; $H_{acc} / H_{don} / n.m.b.$ = number of H-bond acceptors / number of H-bond donors / number of movable bonds.

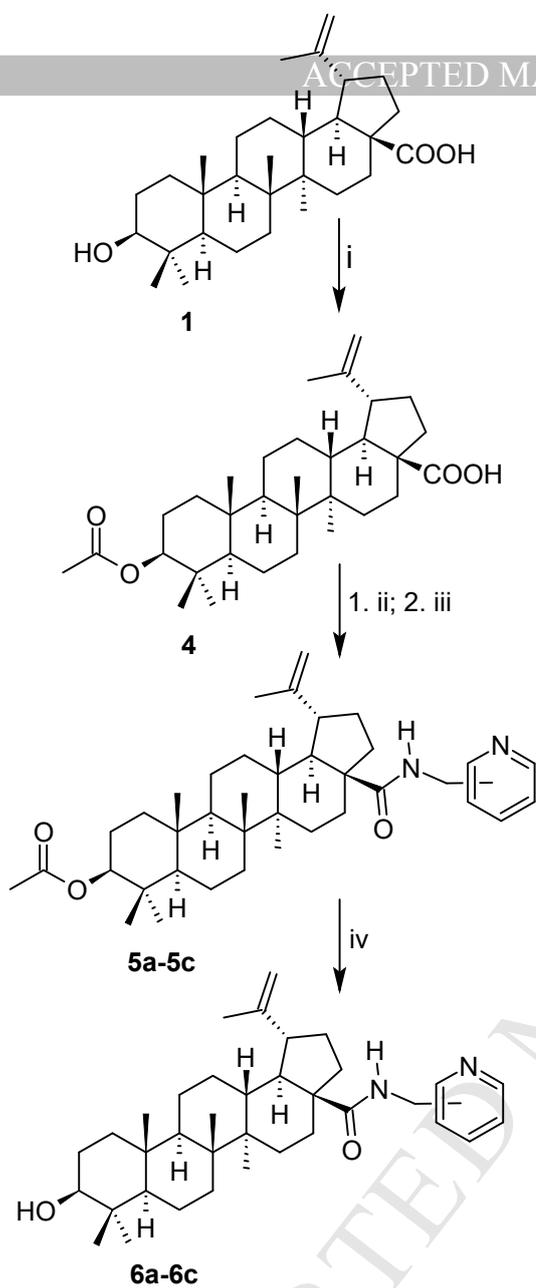












Highlights

Picolyl amides of betulinic acid as antitumor agents causing tumor cell apoptosis

Uladzimir Bildziukevich, Lucie Rárová, David Šaman, and Zdeněk Wimmer

- Picolyl amides of betulinic acid cause human malignant melanoma cell apoptosis;
- Comparison of substitution at the C(3)-OH and C(17)-COOH on cytotoxicity studied;
- Facilitating cellular uptake and enhancing antitumor properties;
- Cytotoxicity of one of the target compounds on G-361 found with TI = 100.