



Accepted Article

Title: Palladium-Catalyzed Regioselective Domino Spirocyclization of Carbamoyl Chlorides with Alkynes and Benzynes

Authors: Chenchen Wang, Wenyu Zhao, Xianqing Wu, Jingping Qu, and Yifeng Chen

This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: Adv. Synth. Catal. 10.1002/adsc.202000537

Link to VoR: https://doi.org/10.1002/adsc.202000537

Palladium-Catalyzed Regioselective Domino Spirocyclization of Carbamoyl Chlorides with Alkynes and Benzynes

Chenchen Wang, Wenyu Zhao, Xianqing Wu, Jingping Qu* and Yifeng Chen*a

^a Key Laboratory for Advanced Materials and Joint International Research Laboratory of Precision Chemistry and Molecular Engineering, Feringa Nobel Prize Scientist Joint Research Center, Frontiers Science Center for Materiobiology and Dynamic Chemistry, School of Chemistry and Molecular Engineering, East China University of

Science and Technology, 130 Meilong Road, Shanghai, 200237, P. R. of China

E-mail: qujp@dlut.edu.cn; yifengchen@ecust.edu.cn

Received: ((will be filled in by the editorial staff))

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adsc.201#######.((Please delete if not appropriate))

Abstract: A palladium-catalyzed domino spirocyclization of carbamoyl chlorides with alkynes and benzynes, involving intramolecular C–H activation to afford valuable oxindole scaffolds bearing spiro quaternary stereocenters, has been developed. This onestep synthesis of spirooxindole is both step- and atomeconomic, proceeding with high regioselectivity in moderate to excellent yields.

Keywords: Palladium; Domino reaction; Spirooxindoles; Carbamoyl chlorides; C–H activation

Spirooxindoles are privileged structural units, with diverse applications in numerous bioactive natural products and pharmaceuticals (Figure 1).^[1] Their promising performance in the relevant areas and complex skeleton have encouraged the development of various synthetic strategies.^[2] The key structural characteristic of these compounds is the spirocyclic scaffold with quaternary carbon center; it is still of great of interest to develop new strategies to construct this scaffold. Among many accessible approaches based on the stepwise elaboration of two rings, onestep, direct formation of spirocycles with a quaternary center is undoubtedly an efficient and attractive approach.^[3] Alternatively, domino reactions are costand time-effective and can streamline the construction of complex molecules via formation of two or more bonds in a single batch without



Figure 1. Selected examples of natural products and pharmaceuticals.

the addition of extra catalysts and reagents, and are broadly applied in the pharmaceutical and other bulk chemical industry.^[4] Furthermore, transition-metalcatalyzed domino transformations have made notable achievements in the last few decades.^[5-7]

Henin and co-workers demonstrated that alkenetethered carbamoyl chloride could undergo a Pdcatalyzed intramolecular Heck reaction to afford α methylene γ -lactam.^[8-9] Since then, the intermolecular reaction pattern of the transition metal-catalyzed cascade cyclization/anion capture sequences of alkene tethered carbamoyl chlorides have been widely explored for the synthesis of valuable oxindole employing a variety of coupling components, including organometallic reagents,^[10] B₂pin₂,^[11] hexamethyldisilane,^[12] aldehyde,^[13] iodide^[14] and carbene^[15] generated *in situ* (Scheme 1A). Domino Heck transformation is an astounding development and the discovery of the ability of transition metals to interact with intermolecular alkynes in the domino spirocyclization would be an elegant progress in organometallic chemistry.^[16]

Recently, we disclosed a Pd-catalyzed domino-Heck reaction of carbamoyl chlorides to achieve the regiodivergent synthesis of oxindoles, obviating the use of prefunctionalized vinyl organometallics reagents (Scheme 1B). ^[17] Notably, when the substitution group on the styrene moiety is an aromatic group, the alkyl palladium intermediate, generated *via* the initial Heck type insertion, undergoes a subsequent 1,4-palladium shift^[18] to provide the aryl palladium intermediate, and affords the desired product *via* a second Heck reaction process. We envisaged capturing the palladacycle intermediate with suitable reaction component to tune the 1,4-Pd migration process. Herein, we report a Pd-catalyzed domino process using carbamoyl chlorides as a relay switch and a highly regioselective insertion of unsymmetrical external alkyne to afford spirooxindole scaffolds (Scheme 1C), exhibiting distinctive selectivity and reactivity.



Scheme 1. Synthesis of spirooxindoles from carbamoyl chlorides.

To investigate the feasibility of our hypothesis, we initially performed the reaction of carbamoyl chlorides (1a) and methyl 3-phenylpropiolate (2a) as model substrates under the following conditions: 1a (0.10 mmol, 1.0 equiv.), 2a (0.20 mmol, 2.0 equiv.), Pd₂dba₃ (0.0025 mmol, 2.5 mol%), IPr·HCl (0.01 mol, 10 mol%) and Cs_2CO_3 (0.15 mmol, 1.5 equiv.) in toluene (0.1 M) at 110 °C for 4 h, delivering the expected product 3a in 55% yield (Table 1, entry 1). It should be noted that a small amount of decarbonylation byproduct 7 formation was detected (Scheme 2). The regioselectivity of 3a to 3a' is more than 20:1. Amongst various ligand screening, it was found that the commercially available Pd(PPh₃)₂Cl₂ provided the highest yield (Table 1, entries 2-6). Under the standard conditions, the desired product 3a was obtained with 72% isolated yield. Next, we investigated the base the yield of product **3a** significantly screening: dropped down when K₃PO₄ was employed as the inorganic base (Table 1, entry 7); When CsOPiv, a common base for C-H activation was used, no desired product formation was observed, and the same scenario applied to NaOMe (Table 1, entries 8 and 9), indicating the nucleophilicity of alkoxy group was too strong, causing decomposition of the starting material **1a**. The solvent screening revealed that toluene is the optimal solvent for this domino transformation (Table 1, entries 10-12). Finally, lowering the reaction temperature to 80 °C resulted in sluggish conversion of **1a** and 66% yield could be obtained by extending reaction time to 20 h (Table 1, entry 13).

With the optimized conditions in hand, we investigated the applicability of domino spirocyclization for spirooxindole synthesis (Table 2). Unless otherwise stated, the ratio of regioisomers in all spirooxindoles are greater than 20:1. The substituent

Table 1. Optimization for the reaction conditions.^[a]

^[a] Reaction conditions: **1a** (1.0 equiv.), **2a** (2.0 equiv.), base

| Ph N-Me CI O 1a | PhCO2 2a (2.0 equiv base (1.5 equiv [Pd] (5 mol%) ligand (10 mol% solvent, T, 4 -2 | Me I :.) /.) //) //) //) | MeO ₂ C Ph N O Me 3a major | > 20:1 rr | Ph CO ₂ Me Ne 3a' minor |
|--------------------------|---|---|---|-----------|---|
| Entry | [Pd]/Ligand | Base | Solvent | T [°C] | Yield [%] ^[b] |
| 1 ^[c] | Pd ₂ dba ₃ /IPr+HCl | Cs ₂ CO ₃ | toluene | 110 | 55 |
| 2 ^[c] | Pd ₂ dba ₃ /PPh ₃ | Cs ₂ CO ₃ | toluene | 110 | 28 |
| 3 | Pd(OAc) ₂ /PPh ₃ | Cs_2CO_3 | toluene | 110 | 48 |
| 4 | Pd(PPh ₃) ₄ | Cs_2CO_3 | toluene | 110 | 61 |
| 5 | Pd(dppf)Cl ₂ | Cs ₂ CO ₃ | toluene | 110 | 25 |
| 6 | Pd(PPh ₃) ₂ Cl ₂ | Cs_2CO_3 | toluene | 110 | 78 (72 ^[d]) |
| 7 | Pd(PPh ₃) ₂ Cl ₂ | K ₃ PO ₄ | toluene | 110 | 28 |
| 8 | Pd(PPh ₃) ₂ Cl ₂ | CsOPiv | toluene | 110 | N.D. |
| 9 | Pd(PPh ₃) ₂ Cl ₂ | NaOMe | toluene | 110 | N.D. |
| 10 | Pd(PPh ₃) ₂ Cl ₂ | Cs_2CO_3 | CH ₃ CN | 110 | trace |
| 11 | Pd(PPh ₃) ₂ Cl ₂ | Cs ₂ CO ₃ | xylene | 110 | 61 |
| 12 | Pd(PPh ₃) ₂ Cl ₂ | Cs_2CO_3 | benzene | 110 | 52 |
| 13 ^[e] | Pd(PPh ₃) ₂ Cl ₂ | Cs_2CO_3 | toluene | 80 | 66 |

(1.5 equiv.), [Pd] catalyst (5 mol%), ligand (10 mol%), solvent (0.1 M), 110 °C, 4 h.

^[b] Corrected GC yield.

^[c] Pd₂(dba)₃ (2.5 mol%), ligand (10 mol%).

^[d] Isolated yield.

^[e] 20 h.

effect on the internal alkynes was first examined. Generally, the optimized reaction conditions were found to be suitable for various alkynes possessing different electronic properties. For instance, the substituted methyl 3-arylpropiolate containing both electron donating (-OMe) and electron-withdrawing groups (-Cl) on the aryl rings could be tolerated and provided **3b** and **3c** in good isolated yields. 1,3-Diphenylprop-2-yn-1-one was incorporated to give 3d in 63% yield. The reaction proceeded well with alkylsubstituted acetylate delivering the corresponding products 3e and 3f in good to excellent isolated yields (82–93%). Notably, it was found that the conjugated enyne 2g, selective insertion of alkyne with the maintenance of alkene functionality, to provide 3g in 50% yield, albeit with longer reaction time. The less reactive diarylacetylene cannot participate this domino reaction, and only low yield of the desired product (13% crude ¹H NMR yield) was observed when diethyl acetylenedicarboxylate was employed under the standard conditions. The reaction worked efficiently on the oxindole containing methoxy (3h) or chloride (3i) functionality. Interestingly, the substrate bearing bromide was well-tolerated and gave moderate reaction yield (3j). Furthermore, access to the orthofluoro substituted spirooxindole (3k) indicated the steric effect is inconspicuous. Product with paramethyl group at the phenyl ring (31) was formed in good yields. Moreover, electron-withdrawing ester and nitro groups (3m, 3n) were tolerated under the standard conditions with 80%, 44% isolated yields. The change of substituent on the nitrogen atom to Nsubstituted benzyl substrate produced the desired products (30) in 66% yield. The para-substitute effect

on the aromatic ring did not affect the selective 1,4-Pd migration, affording the desired product **3p** and **3q** in good yield. It is notable that the heterocycle thiophene substitution could smoothly undergo the domino 1,4-Pd shift as well to provide an expedient access of hetero-substituted spirooxindole **3r** in 72% isolated yield.





^[a] Standard conditions: 1 (1.0 equiv.), 2 (2.0 equiv.), Cs₂CO₃ (1.5 equiv.), Pd(PPh₃)₂Cl₂ (5 mol%), toluene (0.1 M), 110 °C, 4-20 h.

^[b] 1.0 mmol scale reaction.

The 2-(trimethylsilyl)phenyl trifluoromethanesulfonate, initially developed by Kobayashi in 1983,^[19] has become an powerful aryne precursor and was widely applied in organic

synthesis.^[20] However, these highly reaction aryne intermediate is easy to undergo the undesired cyclotrimerization with the transition metal catalyst to furnish triphenylene byproduct.^[21] To overcome this challenge, a series solvent and palladium source screenings were performed, and the yield of triphenylenes byproduct was formed less than 5% under the optimized conditions (see the Supporting Information). It was found that the combination of $C_{S_2}CO_3$ and C_{SF} was the crucial for this domino reaction, 5a was obtained in 63% isolated yield. The substrate scope with arynes were then carried out (Table 3). The reaction with substrates bearing methyl (5b) or methoxy group (5c) worked well, affording the desired products in moderate yields. To our delight, thiophene substitution (5d) could also undergo the domino migration and insertion reactions to generate the desired products in 62% yield. Next, the functionality of the carbamoyl chloride wan investigated. It was found that various functional group could be tolerated with this cascade benzyne insertion, including chloride (5e), methoxy (5f), methyl (5g) and ester (5h) group, to generate the products 5e-h in 53-69% isolated yield.

Table 3. Reaction scope with arynes.^[a]



^[a] Standard conditions: 1 (1.0 equiv.), 4 (2.0 equiv.), Cs₂CO₃ (1.5 equiv.), CsF (2.5 equiv.), Pd(PPh₃)₂Cl₂ (5 mol%) toluene/CH₃CN (v/v = 1:1, 0.1 M), 110 °C, 4 h

A plausible mechanism is shown in Scheme 2. The domino spirocyclization commences with oxidative addition of carbamoyl chlorides (1a) to Pd⁰ to afford the carbamoyl-Pd intermediate 6. Accompanied by decarbonylation to give the side product 7, 6 undergoes the intramolecular Heck reaction with tethered alkene to generate the alkyl-palladium(II) intermediate 8. The alkyl-Pd intermediate 8 subsequently performs an intramolecular C–H activation to afford a five-membered palladacycle (9), which undergoes a selective scission of $C(sp^2)$ -Pd bond under kinetic control *via* the regioselective migratory insertion to the activated alkynes,^[16e] followed by a final reductive elimination to provide the desired product **3a**.



Scheme 2. Plausible mechanism.

In conclusion, we have demonstrated a mild and efficient palladium-catalyzed domino spirocyclization of carbamoyl chlorides with alkynes and arynes. Distinguished by remarkable regioselectivity and diverse set of substitution patterns, this protocol is a novel strategy for synthesizing spirooxindoles and may provide further application value in complex agrochemicals and pharmaceuticals.

Experimental Section

General Experimental Procedure for the Synthesis of Spirooxindoles 3

An oven-dried vial charged with carbamoyl chlorides **1** (1.0 equiv.), $Pd(PPh_3)_2Cl_2$ (5 mol%), and Cs_2CO_3 (1.5 equiv.) was evacuated and backfilled with N_2 (This process was repeated three times). Internal alkynes **2** (2.0 equiv.) and toluene (0.1 M) were subsequently added to the reaction mixture. The tube was sealed with Parafilm, and allowed to stir for 4–20 h at 110 °C. The mixture was quenched with sat. aq. NH₄Cl and extracted with EtOAc. The separated organic layer was washed twice with water and once with brine, then dried over anhydrous Na₂SO₄, the resulting organic layer was purified by chromatography on silica gel column (PE/EtOAc) to afford the desired spirooxindoles **3**.

General Experimental Procedure for the Synthesis of Spirooxindoles 5

An oven-dried vial charged with carbamoyl chlorides 1 (1.0 equiv.), Pd(PPh_3)₂Cl₂ (5 mol%), and Cs₂CO₃ (1.5 equiv.)

CsF (2.5 equiv.) was evacuated and backfilled with N_2 (This process was repeated for three times). Aryne precursor **4** (2.0 equiv.) and toluene/CH₃CN (v/v = 1/1, 0.1 M) were subsequently added into the reaction mixture. The tube was sealed with Parafilm, and allowed to stir for 4 h at 110 °C. The mixture was quenched with sat. aq. NH₄Cl and extracted with EtOAc. The separated organic layer was washed twice with water and once with brine, then dried over anhydrous Na₂SO₄, the resulting organic layer was concentrated under reduced pressure, and the crude product was purified by chromatography on silica gel column (PE/EtOAc) to afford the desired spirooxindoles **5**.

Acknowledgements

This work was supported by NSFC/China (21421004, 21702060), Shanghai Rising-Star Program, Shanghai Municipal Science and Technology Major Project (Grant No.2018SHZDZX03) and the Program of Introducing Talents of Discipline to Universities (B16017), and the Fundamental Research Funds for the Centra Universities. Y.C. thank Prof. Feng Sha (ECUST) for generous donation of benzyne precursors for product 5b and 5c. The authors thank Research Center of Analysis and Test of East China University of Science and Technology for the help on NMR analysis.

References

- [1] For selected reviews on spirooxindole, see: a) C. V. Galliford, K. A. Scheidt, Angew. Chem. Int. Ed. 2007, 46, 8748-8758; Angew. Chem. 2007, 119, 8902-8912; b) T. L. Pavlovska, R. G. Redkin, V. V. Lipson, D. V. Atamanuk, Mol. Divers. 2016, 20, 299-344; c) L. Hon, R. Wang, Adv. Synth. Catal. 2013, 355, 1023-1052; d) N. R. Ball-Jones, J. J. Badillo, A. K. Franz, Org Biomol. Chem. 2012, 10, 5165-5181; e) A. P. Antonchick, C. Gerding-Reimers, M. Catarinella, M. Schürmann, H. Preut, S. Ziegler, D. Rauh, H. Waldmann, Nat. Chem. 2010, 2, 735-740; f) R. J. Cox, R. M. Williams, Acc. Chem. Res. 2003, 36, 127-139; g) N. Ye, H. Chen, E. A. Wold, P.-Y. Shi, J. Zhou, ACS Infect. Dis. 2016, 2, 382-392; h) A. Barakat, M. S. Islam, H. M. Ghawas, A. M. Al-Majid, F. F. El-Senduny, F. A. Badria, Y. A. Elshaier, H. A.Ghabbour, A. Barakat, RSC Adv. 2018, 8, 14335-14346.
- [2] a) H. Lin, S. J. Danishefsky, Angew. Chem. Int. Ed. 2003, 42, 36-51; Angew. Chem. 2003, 115, 38-53; b)
 C. Marti, E. M. Carreira, J. Am. Chem. Soc. 2005, 127, 11505-11515; c) M. Liu, X. Zhang, X. Huang, G. Dhawan, J. Evans, M. Kaur, J. P. Jasinski, W. Zhang, Eur. J. Org. Chem. 2019, 150-155; d) A. Yen, M. Lautens, Org. Lett. 2018, 20, 4323-4327; e) C. M. Le, X. Hou, T. Sperger, F. Schoenebeck, M. Lautens, Angew. Chem. Int. Ed. 2015, 54, 15897-15900; Angew. Chem. 2015, 127, 16127-16131.
- [3] a) J. E. Aho, P. M. Pihko, T. K. Rissa, *Chem. Rev.* 2005, 105, 4406-4440; b) Z. Yang, *Acc. Chem. Res.* 2019, 52, 480-491; (c) V. A. D'yakonov, O. A. Trapeznikova, A. Meijere, U. M. Dzhemilev, *Chem. Rev.* 2014, 114, 5775-5814; d) R. Rios, *Chem. Soc. Rev.* 2012, 41, 1060-1074.

- [4] For selected reviews on domino reaction, see: a) L. F. Tietze, *Chem. Rev.* 1996, 96, 115-136; b) L. F. Tietze, G. Brasche, K. M. Gericke, *Domino Reactions in Organic Synthesis*, Wiley-VCH, Weinheim, 2006, pp. 359-493; c) K. C. Nicolaou, J. S. Chen, *Chem. Soc. Rev.* 2009, 38, 2993-3009; d) H. Pellissier, *Chem. Rev.* 2013, 113, 442-524. For selected examples: e) W. E. Russey, P. R. Montellano, E. J. Corey, J. Am. Chem. Soc. 1966, 88, 4750-4751; f) D. J. Tao, Y. Slutskyy, M. Muuronen, A. Le, P. Kohler, L. E. Overman, J. Am. Chem. Soc. 2018, 140, 3091-3102.
- [5] For selected reviews on transition metal-catalyzed domino reaction, see: a) Y. Ping, Y. Li, J. Zhu, W. Kong, Angew. Chem. Int. Ed. 2019, 58, 1562-1573; Angew. Chem. 2019, 131, 1576-1587; b) W. Oppolzer, Pure Appl. Chem. 1990, 62, 1941-1948; c) V. Sridharan, R. Grigg, J. Organomet. Chem. 1999, 576, 65-87; d) M. Pérez-Gómez, L. Navarro, I. Saura-Llamas, D. Bautista, M. Lautens, J. García-López, Organometallics, 2017, 36, 4465-4476; e) H. A. Döndaş, M. D. G. Retamosa, J. M. Sansano, Organometallics, 2019, 38, 1828-1867; f) X. Yang, S. J. Kalita, S. Maheshuni, Y.-Y. Huang, Coord. Chem. Rev. 2019, 392, 35-48; g) Y. Yasui, Y. Takemoto, Chem. Rec. 2008, 8, 386-394.
- [6] a) X.-X. Wu, W.-L. Chen, Y. Shen, S. Chen, P.-F. Xu, Y.-M. Liang, Org. Lett. 2016, 18, 1784-1787; b) H. Hu, F. Teng, J. Liu, W. Hu, S. Luo, Q. Zhu, Angew. Chem. Int. Ed. 2019, 58, 9225-9229; Angew. Chem. 2019, 131, 1576-1587; c) H. Cong, G. C. Fu, J. Am. Chem. Soc. 2014, 136, 3788-3791; d) W. You, M. K. Brown, J. Am. Chem. Soc. 2015, 137, 14578-14581; e) Z.-X. Tian, J.-B. Qiao, G.-L. Xu, X. Pang, L. Qi, W.-Y. Ma, Z.-Z. Zhao, J. Duan, Y.-F. Du, P. Su, X.-Y. Liu, X.-Z. Shu, J. Am. Chem. Soc. 2019, 141, 7637-7643; f) Y. Jin, C. Wang, Angew. Chem. Int. Ed. 2019, 58, 6722-6726; Angew. Chem. 2019, 131, 6794-6798; g) L. Zhou, S. Li, B. Xu, D. Ji, L. Wu, Y. Liu, Z.-M. Zhang, J. Zhang, Angew. Chem. Int. Ed. 2020, 59, 2769-2775; Angew. Chem. 2020, 132, 2791-2797.
- [7] a) J.-K. Qiu, B. Jiang, Y.-L.Zhu, W.-J. Hao, D. Wang, J. Sun, P. Wei, S.-J. Tu, G. Li, *J. Am. Chem. Soc.* 2015, *137*, 8928-8931; b) Y. Zhang, G.-Z. Wu, G. Agnel, E.-I. Negishi, *J. Am. Chem. Soc.* 1990, *112*, 8590-8592; c) A. Whyte, B. Mirabi, A. Torelli, L. Prieto, J. Bajohr, M. Lautens, *ACS Catal.* 2019, *9*, 9253-9258.
- [8] a) T. Iwai, T. Fujihara, J. Terao, Y. Tsuji, J. Am. Chem. Soc. 2010, 132, 9602-9603; b) C. Tsukano, Y. Takemoto, Heterocycles, 2014, 89, 2271-2302; c) X. Li, J. Pan, S. Song, N. Jiao, Chem. Sci. 2016, 7, 5384-5389; d) W. Dong, G. Xu, W. Tang, Tetrahedron, 2019, 75, 3239-3247; e) Y. Yasui, I. Kakinokihara, H. Takeda, Y. Takemoto, Synthesis, 2009, 23, 3989-3993; f) C. Tsukano, M. Okuno, Y. Takemoto, Chem. Lett. 2013, 42, 753-755; g) C. Tsukano, M. Okuno, Y. Takemoto, Angew. Chem. Int. Ed. 2012, 51, 2763-2766; Angew. Chem. 2012, 124, 2817-2820.
- [9] F. Henin, J. Muzart, J.-P. Pete, *Tetrahedron Lett.* 1986, 52, 6339-6340.

- [10] a) M. Fielding, C. J. Urch, R. Grigg, *Chem. Commun.* 2000, 2239-2240; b) U. Anwar, M. R. Fielding, V. Sridharan, C. J. Urch, R. Grigg, *J. Organomet. Chem.* 2006, 691, 1476-1487; c) W. Sun, X. Shi, C. Chen, Y.-P. Zhu, Z. Liu, B. Zhu, *Asian J. Org. Chem.* 2020, 9, 575-578. d) C. Zhang, X. Wu, C. Wang, C. Zhang, J. Qu, Y. Chen, *Org. Lett.* 2020, 22, 6376–6381.
- [11] A. Whyte, K. I. Burton, J. Zhang, M. Lautens, Angew. Chem. Int. Ed. 2018, 57, 13927-13930; Angew. Chem. 2018, 130, 14123-14126.
- [12] a) H. Kamisaki, Y. Yasui, Y. Takemoto, *Tetrahedron Lett.* 2009, 50, 2589-2592; b) S. M. Hande, M. Nakajima, H. Kamisaki, C. Tsukano, Y. Takemoto, *Org. Lett.* 2011, 13, 1828-1831.
- [13] P. Fan, Y. Lan, C. Zhang, C. Wang, J. Am. Chem. Soc. 2020, 142, 2180-2186.
- [14] a) C. Chen, J. Hu, J. Su, X. Tong, *Tetrahedron Let*.
 2014, 55, 3229-3231; b) A. D. Marchese, M. Wollenburg, B. Mirabi, X. Abel-Snape, A. Whyte, F. Glorius, M. Lautens, *ACS Catal.* **2020**, *10*, 4780-4785.
- [15] W. Sun, C. Chen, Y. Qi, J. Zhao, Y. Bao, B. Zhu, Org. Biomol. Chem. 2019, 17, 8358-8363.
- [16] a) M. Pérez-Gómez, J.-A. García-López, Angew. Chem. Int. Ed. 2016, 55, 14389-14393; Angew. Chem.
 2016, 128, 14601-14605; b) H. Yoon, A. Lossouarn, F. Landau, M. Lautens, Org. Lett. 2016, 18, 6324-6327; c) H. Yoon, M. Rölz, F. Landau, M. Lautens, Angew. Chem. Int. Ed. 2017, 56, 10920-10923; Angew. Chem.
 2017, 129, 11060-11063; d) J. F. Rodríguez, A. D. Marchese M. Lautens, Org. Lett. 2018, 20, 4367-4370.
 e) I. Franzoni, H. Yoon, J.-A. García-López, A. I. Poblador-Bahamonde, M. Lautens, Chem. Sci. 2018, 9, 1496-1509.
- [17] X. Wu, Z. Tang, C. Zhang, C. Wang, L. Wu, J. Qu, Y. Chen, Org. Lett. 2020, 22, 3915-3921.
- [18] For selective reviews and examples on 1,4-Pd shift. see: a) A. Rahim, J. Feng, Z. Gu, Chin. J. Chem. 2019, 37, 929-945; b) S. Ma; Z. Gu, Angew. Chem. Int. Ed. 2005, 44, 7512-7517; Angew. Chem. 2005, 117, 7680-7685; c) R. Rocaboy, I. Anastasiou, O. Baudoin, Angew. Chem. Int. Ed. 2019, 58, 14625-14628; Angew. Chem. 2019, 131, 14767-14770; d) J. Pan, M. Su, S. L. Buchwald, Angew. Chem. Int. Ed. 2011, 50, 8647-8651; Angew. Chem. 2011, 123, 8806-8810; e) T. E. Barder, S. D. Walker, J. R. Martinelli, S. L. Buchwald, J. Am. Chem. Soc. 2005, 127, 4685-4696; f) T.-J. Hu M.-Y. Li, Q. Zhao, C.-G. Feng, G.-Q. Lin, Angew. Chem. Int. Ed. 2018, 57, 5871-5875; Angew. Chem. 2018, 130, 5973-5977; g) M. A. Campo, R. C. Larock, J. Am. Chem. Soc. 2002, 124, 14326-14327; h) M. A. Campo, Q. Huang, T. Yao, Q. Tian, R. C. Larock, J. Am. Chem. Soc. 2003, 125, 11506-11507; i) Q. Huang, A. Fazio, G. Dai, M. A. Campo, R. C. Larock, J. Am. Chem. Soc. 2004, 126, 7460-7461; j) M. A. Campo, H. Zhang, T. Yao, A. Ibdah, R. D. McCulla, Q. Huang, J. Zhao, W. S. Jenks, R. C. Larock, J. Am. Chem. Soc. 2007, 129, 6298-6307; k) G. Karig, M.-T. Moon, N. Thasana, T. Gallagher, Org. Lett. 2002, 4, 3115-3118.

- [19] Y. Himeshima, T. Sonoda, H. Kobayashi, *Chem. Lett.* 1983, 12, 1211-1214.
- [20] a) A. V. Dubrovskiy, N. A. Markina, R. C. Larock, Org. Biomol. Chem. 2013, 11, 191-218; b) P. M. Tadross, B. M. Stoltz, Chem. Rev. 2012, 112, 3550-3577; c) A.

Bhunia, S. R. Yetra, A. T. Biju, *Chem. Soc. Rev.* **2012**, *41*, 3140-3152.

[21] D. Peña, S. Escudero, D. Pérez, E. Guitián, L. Castedo, Angew. Chem. Int. Ed. 1998, 37, 2659-2661; Angew. Chem. 1998, 110, 2804-2806.

COMMUNICATION

Palladium-Catalyzed Regioselective Domino Spirocyclization of Carbamoyl Chlorides with Alkynes and Benzynes

Adv. Synth. Catal. Year, Volume, Page – Page

Chenchen Wang, Wenyu Zhao, Xianqing Wu, Jingping Qu* and Yifeng Chen*

