

Communication

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Metal-Free Approach to Biaryls from Phenols and Aryl Sulfoxides by Temporarily Sulfur-Tethered Regioselective C–H/C–H Coupling

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Supporting Information Placeholder

ABSTRACT: We have developed metal-free regiocontrolled dehydrogenative C–H/C–H cross-coupling of aryl sulfoxides with phenols by means of trifluoroacetic anhydride. Since the reaction would proceed through an interrupted Pummerer reaction followed by sulfonium-tethered [3,3]-sigmatropic rearrangement, the C–H/C–H coupling takes place exclusively between the *ortho* positions of both substrates. Various functional groups including carbonyl, halo, siloxy, and even boryl moieties are compatible. The biaryl products naturally possess hydroxy and sulfanyl groups, which allows the products to be useful synthetic intermediates, as evidenced by the syntheses of π -expanded heteroarenes such as unprecedented 7,12-dioxa[8]helicene.

Biarvl motifs represent a privileged substructure in bioactive compounds and organic functional materials. Therefore, a number of synthetic approaches to biaryl skeletons have been exploited. Over the past 40 years, transition metal-catalyzed cross-coupling reactions of aryl halides with aryl organometallic reagents have occupied a central place in biaryl synthesis.¹ For the last decade, direct aromatic C-H arylation has been emerging as a game-changing means to connect two aromatic rings.² The most straightforward and ideal method would be dehydrogenative C-H/C-H coupling reaction starting from two individual arenes. Since the pioneering discovery by Fagnou in 2007,³ dehydrogenative C-H/C-H cross-coupling reactions have been actively investigated by utilizing transition metal catalysts such as copper-catalyzed asymmetric BINOL synthesis.^{4,5} However, catalyst loadings to achieve efficient C-H/C-H coupling were usually high and most of transition metal complexes are expensive. In addition, heavy metal contaminations of products can be a fatal problem in research on bioactive agents⁶ as well as organic electronic devices.⁷ Consequently, it has been desired to establish new methodologies to realize efficient metal-free dehydrogenative C-H/C-H cross-coupling.^{4d,8} To this end, radical-mediated processes promoted by hypervalent iodine reagents⁹ or anodic oxidation¹⁰ have been developed. However, these strategies incur poor regioselectivity, mediocre functional group compatibility, and/or employment of large excess amounts of radicalaccepting arenes. Although several regioselective reactions have been reported including base-mediated arylation of heteroaromatic N-oxides,¹¹ Bartoli-type reaction of 2halonitroarenes with arylmagnesium reagents,12 chiral Brønsted acid-catalyzed atropselective synthesis of 2,2'diamino-1,1'-binaphthalenes through benzidine rearrangement,13 and Brønsted acid-mediated reaction of hydroxy

arenes with 2,5- or 2,6-substituted 1,4-benzoquinone analogues, 14 a restricted range of arenes could engage in these transformations.







(b) Dehydrogenative coupling of aryl sulfoxides with phenols (*This work*)



Recently, extended Pummerer reactions of unsaturated sulfoxides have been attracting increasing attention¹⁵ since they offer characteristic transformations utilizing the unique reactivity of sulfur. Some extended Pummerer reactions result in fascinating $C(sp^2)$ -H functionalization at the β position of the sulfinyl group.¹⁶ Indeed, we reported the synthesis of benzofurans by connecting phenols with peculiar vinyl sulfoxides, ketene dithioacetal monoxides (KDM), at each of their β $C(sp^2)$ -H bonds (Scheme 1a).¹⁷ A plausible mechanism includes 1) the activation of the sulfoxide with trifluoroacetic anhydride, 2) capture of the resulting sulfonium with phenol, 3) sulfonium-accelerated [3,3]-sigmatropic rearrangement¹⁸ with concomitant loss of the aromaticity of phenol, and 4) overall aromatization. During the course of the reaction, the cationic sulfonium moiety acts as a directing group to realize regioselective C-C bond formation.

 Table 1. Regiocontrolled dehydrogenative cross-coupling of aryl sulfoxides 1 with phenols 2



^{*a*}Performed at -40 °C for 2 h. ^{*b*}Combined yield of regioisomers. Isomeric ratios are in parentheses. ^{*c*}Performed on 19 mmol scale. ^{*d*}Performed at 40 °C. ^{*e*}Yield was determined by ¹H NMR analysis. ^{*f*}Performed on 5.0 mmol scale. ^{*g*}Reaction conditions: **1q** (0.20 mmol), 2-naphthol (**2t**, 3.5 equiv), (CF₃CO)₂O (3.0 equiv), CH₂Cl₂ (0.1 M), 25 °C, 1 h.

We envisioned that this extended Pummerer strategy with phenols as substrates would be applicable to the synthesis of biaryls. Our working hypothesis is depicted in Scheme 1b. Activation of the aryl sulfoxide 1 followed by nucleophilic substitution at the cationic sulfur center with phenol 2 would form intermediate **B**. Given that [3,3]-sigmatropic rearrangement would occur with the loss of the aromaticity of both aryl sulfonium and phenol, regiocontrolled C–C bond formation should result by a temporarily sulfonium-tethered intramolecular process. Finally, rearomatization of **C** furnishes biaryl product **3**. Biaryl **3** has synthetically useful hydroxy and sulfanyl moieties, further transformation being highly expected.¹⁹

Treatment of 2-(methylsulfinyl)-1-tosylindole (1a) and phenol (2a) with trifluoroacetic anhydride in CH₂Cl₂ at 25 °C afforded an 89% yield of 3-(2-hydroxyphenyl)-2methylsulfanyl-1-tosylindole (3aa) as the sole product (Table 1). As expected, the C-C bond formation took place exclusively between the *ortho* (or β) positions of 1a and 2a. Phenols having an electron-donating or -withdrawing group at the 4-position reacted with 1a to provide the corresponding biaryls **3ac-aj** in good yields. Carbonyl, halogen, and pinacolatoboryl moieties were well tolerated under the reaction conditions to yield **3ah-am**. Arylation of 3methoxyphenol (2n) or 3-isopropylphenol (20) with 1a occurred exclusively at the less hindered 6-position yielding **3an** or **3ao** as the single product, while that of 3-bromophenol (**2p**) gave a mixture of regioisomers **3ap** and **3ap'** with moderate selectivity favoring sterically less demanding **3ap**. This reaction also accommodates 2-substituted phenols to afford the corresponding sterically congested 2,6-disubstituted phenols **3aq** and **3ar** efficiently. Naphthols **2s** and **2t** were also converted to the corresponding arylated products **3as** and **3at**, respectively. Notably, 2-naphthol (**2t**) reacted at the C1 position exclusively due to the stronger double-bond character of the C1–C2 bond of the naphthalene core than that of the C2–C3 bond. These *ortho*-selectivities clearly indicate the reactions proceeded via the sulfonium-tethered process shown in Scheme 1b.

This biaryl synthesis was also applicable to a wide range of aryl sulfoxides. Indeed, 2-methylsulfinyl-substituted naphthalene **1b**, benzofuran **1c**, and benzothiophene **1d** afforded the desired products in good to excellent yields. Instead of the methylsulfinyl moiety, an aromatic p-tolylsulfinyl moiety served as an effective directing group. In the reaction of unsymmetrical diaryl sulfoxide **1e**, the potentially reactive p-tolyl group remained intact, and regioselective arylation occurred at the benzothiophene unit of the weaker aromaticity to yield **3ea**. C–C bond formation

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proceeded at the 2-position when 3-(methylsulfinyl)indole 1f was employed. Monocyclic pyrrolyl and thienyl sulfoxides 1g and **1h** also reacted at their 3-positions with perfect regioselectivity. Not only heteroaromatic sulfoxides but also phenyl sulfoxides 1i-l, 10, and 1p participated in the arylation.²⁰ The triisopropylsilyl ether in 1k survived under the reaction conditions as well as improved the regioselectivity of the arylation due to the steric hindrance (**3it** and **3jt** vs. **3kt**). Preferably, the phenyl groups of the sulfoxides should be electron-rich as parent methyl phenyl sulfoxide (11) afforded the product 3lt in 26% yield. We assume that the electrondonating group(s) at the 3- (and 5-) position(s) would stabilize the corresponding cationic intermediate C in Scheme 1b to facilitate the [3,3]-sigmatropic rearrangement of **B** to **C**. Indeed, 4-methoxyphenyl methyl sulfoxide (1m), which stabilizes not C but B, afforded a trace amount of product 3mt due to the elevated activation energy barrier for the pathway from **B** to **C**. Unfortunately, 4-trifluoromethylphenyl methyl sulfoxide (1n) afforded the product 3nt in very low yield. The coupling reaction of 2,7-di(dodecylsulfinyl)naphthalene (1q) with 2-naphthol (2t) occurred at the *peri* positions to afford 1,8-dinaphthylated naphthalene 3qt regioselectively in 71% yield. The formation of sterically congested 1,2,7,8-substituted **3qt** highlights the efficiency, robustness, and potentiality of the present methodology (vide infra, specifically in Scheme 3d). The reactions could be conducted on larger scales, as demonstrated in the synthesis of 3ca and 3ou, which clearly show the robustness of our system.





To further verify our mechanistic proposal in Scheme 1b, the reaction of 2,6-dimethylphenol (2v) was performed. If this reaction proceeded through a Friedel-Crafts pathway, 4arylated-2,6-dimethylphenol 3dv' should be the conceivable major product. However, C-C bond formation proceeded at the 3-position of 2v preferentially to furnish 3dv as the major product, along with 3dv' as the minor isomer (Scheme 2a). On the basis of this result, we are tempted to propose a mechanism for the meta-selective coupling. According to the scenario shown in Scheme 1b, cyclohexadienone intermediate 4 would be generated via an interrupted Pummerer reaction/[3,3]-sigmatropic rearrangement pathway. With the aid of trifluoroacetic acid generated in situ, 4 would then undergo protonation-induced 1,2-shift²¹ of the benzothiophene unit. Notably, intermediate 4 was indeed isolated in 58% yield when the coupling reaction was quenched by methanolic KOH

immediately in 5 s (Scheme 2b). Naturally, treatment of 4 with catalytic trifluoroacetic acid gave 3dv.²²



After the conversion of the phenolic hydroxy group into a triflate, global reduction by means of a hydrosilane/palladium catalyst system²³ gave 5ca (Scheme 3a), traceless biaryl coupling²⁴ hence being achieved. Instead of the reductive removal of the tethering units, we could take advantage of these functional groups for derivatizations of 3 into π -expanded heteroarenes. After removal of the propionate moiety on 3pu under basic conditions,²⁵ p-toluenesulfonic acid-mediated intramolecular condensation²⁶ furnished benzonaphthothiophene **6pu** in 73% yield over two steps (Scheme 3b). The sulfanyl moiety of **3** could be easily oxidized to a sulfonyl moiety by m-CPBA. As the methylsulfonyl moiety is a good leaving group, intramolecular S_NAr reactions resulted in the formations of highly fused dibenzofuran analogues 7ca, 7da, and 7ou in good yields (Scheme 3c). This S_NAr-based cyclization was also applicable to the synthesis of a polycyclic and nonplanar screw-shaped helicene analogue (Scheme 3d).²⁷ After oxidation of **3gt**, intramolecular S_NAr cyclization culminated in a rapid access to unprecedented 7,12-dioxa[8]helicene 8qt in 68% overall yield.

In summary, we have developed metal-free, highly regioselective dehydrogenative C–H/C–H cross-coupling of aryl sulfoxides with phenols by means of trifluoroacetic anhydride. The regiocontrolled $C(sp^2)$ – $C(sp^2)$ bond formation has been realized by the mechanism involving a sequence of interrupted Pummerer reaction and [3,3]-sigmatropic rearrangement directed by a temporary and removable tether between the sulfinyl group and the phenolic hydroxy group. Understandably, the products are useful synthetic intermediates, as clearly demonstrated by the synthesis of π -expanded heteroaromatic molecules including 7,12-dioxa[8]helicene. Further expansion of the scope of substrates and creation of more interesting aromatic cores as well as asymmetric biaryl synthesis through this strategy are now in progress.²⁸

ASSOCIATED CONTENT

Supporting Information

Experimental procedures, X-ray crystallographic analysis, photophysical property, and spectral data. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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