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## First Total Synthesis of (-)-8-epi-9-Deoxygoniopypyrone

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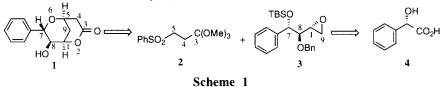
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**Abstract**: The structure and absolute configuration of natural 8-*epi*-9-deoxygoniopypyrone have been confirmed by an efficient and highly diastereoselective synthesis in 15 steps from (S)-mandelic acid with an overall yield of 43%. @ 1998 Elsevier Science Ltd. All rights reserved.

Keywords : bicyclic heterocyclic compounds, cleavage reactions, hydroxylation, styryllactones.

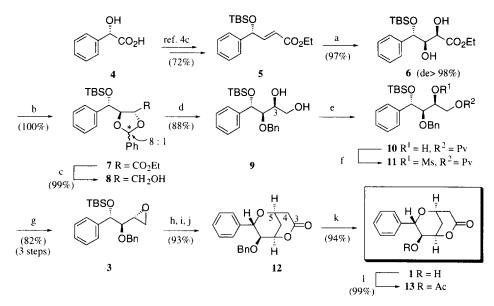
In 1995, (-)-8-*epi*-9-deoxygoniopypyrone  $1^1$  was isolated along with other antitumor styryllactones from the stem bark of *Goniothalamus dolichocarpus*.<sup>2</sup> The structure of 8-*epi*-9-deoxygoniopypyrone was determined as  $(1R^*, 5R^*, 7R^*, 8R^*)$ -8-hydroxy-7-phenyl-2,6-dioxabicyclo[3.3.1]nonan-3-one 1 based on two-dimensional NMR studies and by comparison of its NMR spectra with those of its epimer, (+)-9-deoxygoniopypyrone, the structure of which was established by X-ray crystallographic analysis.<sup>3</sup>

In the course of our program directed toward the stereoselective synthesis of styryllactones, we have recently completed the total synthesis of nine of them.<sup>4</sup> Herein, we described the first synthesis of 1 which is identical to the natural (-)-8-*epi*-9-deoxygoniopypyrone, thereby confirming its structure and absolute configuration. As outlined in Scheme 1, one of the key element of our synthetic strategy for 1 involves a coupling between the sulfone 2 and the epoxyde 3, readily available from (S)-mandelic acid 4 which authorizes the rapid construction of the 2,6-dioxabicyclo[3.3.1]nonan-3-one framework of 1.



As already described for its (R)-enantiomer, (S)-mandelic acid was readily transformed to the enantiopur (E)-a, β-unsaturated ester 5 in four steps and 72% overall yield (Scheme 2).<sup>4c</sup> Introduction of C-1-C-8 stereogenic centers of compound 1 was effected by the Sharpless asymmetric dihydroxylation (AD) using AD-mix- $\alpha$  in the presence of methanesulfonamide.<sup>5</sup> The dihydroxylation of **5** proceeded with a perfect matching double stereoselectivity to give exclusively the desired diol 6.6 Then, protection of the diol as a benzylidene acetal followed by reduction of the ester function by lithium aluminium hydride at 0°C furnished 7 in 99% yield. Hydroxy-directed regioselective reductive benzylidene cleavage in 8 with the BH<sub>3</sub>.Me<sub>2</sub>S-BF<sub>3</sub>.Et<sub>2</sub>O system gave the benzyl ether 9 in 88% yield.<sup>7</sup> The stage was now set up for the introduction of the epoxyde functionality with concomitant inversion of the stereogenic center at C-3. To this event, the primary hydroxyl group of the diol 9 was selectively protected as a pivaloate group and the C-3 alcohol function transformed to a leaving group by treatment with methanesulfonyl chloride in the presence of an excess of triethylamine. Subsequent oxirane ring formation mediated by sodium methoxide, via the saponification of the pivaloate, afforded the  $\alpha$ -epoxide 3 in 82% yield for the three-reaction sequence. The next task of the synthesis of 1, the introduction of the C-3-C-5 fragment bearing a lactone unit was realized using the Ghosez's methodology.<sup>8</sup> Thus, addition of the lithium salt of methyl 3-phenylsulfonyl orthopropionate 2 to the epoxide 3, in the presence of BF, Et<sub>2</sub>O, followed by acid treatment which effected cleavage of the silyl protecting group and lactone formation. Exposure of the crude mixture to an excess of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) led to the bicyclic lactone 12 via PhSO,H elimination and concomitant intramolecular Michael addition of the benzylic hydroxyl function to the resulting  $\alpha$ ,  $\beta$ -unsatured- $\gamma$ -lactone.

\*Corresponding author : Fax (33) 0472431214; e-mail : vatele@univ-lyon1.fr 0040-4039/98/\$ - see front matter © 1998 Elsevier Science Ltd. All rights reserved. *PII:* S0040-4039(98)02269-2



Scheme 2: (a) AD-mix-α, MeSO<sub>2</sub>NH<sub>2</sub>, *t*-BuOH-H<sub>2</sub>O (1:1), RT, 36h; (b) PhCH(OMe)<sub>2</sub>, CSA, benzene, reflux, 1h; (c) LiAlH<sub>4</sub>, Et<sub>2</sub>O, 0°C, 5min; (d) BH<sub>3</sub>.Me<sub>2</sub>S, CH<sub>2</sub>Cl<sub>2</sub>, 60 min then BF<sub>3</sub>.Et<sub>2</sub>O, 0°C, 15 min; (e) *t*-BuCOCl, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, RT, 15 min; (f) CH<sub>3</sub>SO<sub>2</sub>Cl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, RT, 30 min; (g) MeONa, Et<sub>2</sub>O-MeOH, RT, 5h; (h) PhSO<sub>2</sub>(CH<sub>2</sub>)<sub>2</sub>C(OMe)<sub>3</sub> 2, *n*BuLi, BF<sub>3</sub>.Et<sub>2</sub>O, THF, -78°C, 1h; (i) CF<sub>3</sub>CO<sub>2</sub>H-H<sub>2</sub>O (9:1), RT, 3h; (j) DBU, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 1 h; (k) TiCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, RT, 30 min; (l) Ac<sub>2</sub>O, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, RT, 1h.

Finally, quantitative debenzylation with TiCl<sub>4</sub><sup>9</sup> afforded (-)-8-*epi*-9-deoxygoniopypyrone 1 (mp 130-131°C (AcOEt-hexane),  $[\alpha]_{D}^{20}$  -90 (c 0.7, CHCl<sub>3</sub>)). The spectra and the physical data of the corresponding acetate of synthetic 1 (compound 13) are in accord with those of the natural material. <sup>2,10</sup> In conclusion, we have accomplished the first total synthesis of (-)-8-*epi*-9-deoxygoniopypyrone 1 in a highly stereocontrolled manner thereby establishing its relative and absolute configuration.

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## **References and notes**

1. This name, numbered in accordance with IUPAC rules, should be preferred to (-)-iso-5-deoxygoniopypyrone.<sup>2</sup>

2. Goh, S.H.; Ee, G.C.L.; Chuah, C.H. and Wei, C. Aust. J. Chem. 1995, 48, 199-205.

3. Fang, X.P.; Anderson, J.E.; Chang, C.J. and Mc Laughlin, J.L. J. Nat. Prod. 1991, 54, 1034-1043.

4. (a) Surivet, J.P.; Goré J. and Vatèle, J.M. Tetrahedron Lett. **1996**, 37, 371-374; (b) Surivet, J.P.; Goré J. and Vatèle, J.M. Tetrahedron **1996**, 37, 14877-14890; (c) Surivet, J.P.; Volle, J.N. and Vatèle, J.M. Tetrahedron : Asymmetry **1996**, 7, 3305-3311; (d) Surivet, J.P. and Vatèle, J.M. Tetrahedron Lett. **1996**, 37, 4373-4376; (e) Surivet, J.P. and Vatèle, J.M. Tetrahedron Lett. **1997**, 38, 819-820; (f) Surivet, J.P. and Vatèle, J.M. Tetrahedron Lett. **1997**, 38, 819-820; (f) Surivet, J.P. and Vatèle, J.M. Tetrahedron Lett. **1998**, 39, 7299-7300.

5. Sharpless, K.B.; Amberg, W.; Bennani, Y.L.; Crispino, G.A.; Hartung, J.; Jeong, K.S.; Kwong, H.L.; Morikawa, K.; Wang, Z.M.; Xu, D. and Zhang, X.L. J.Org.Chem. **1992**, *57*, 2768-2771.

6.Analytical and spectral data were obtained for all new compounds and are consistent with the structure assigned.

7. Saito, S.; Kuroda, A.; Tanaka, K. and Kimura, R. Synlett, 1996, 231-233.

8. Carretero, J.C. and Ghosez, L. Tetrahedron Lett. 1988, 29, 2059-2062.

9. Hori, H.; Nishida, Y.; Ohrui, H. and Meguro, H. J.Org. Chem. 1989, 54, 1346-1353.

10. Goh and Coll. did not report the optical rotation and melting point of  $1.^2$  Physical data of 13: synthetic (mp 138-139°C;  $[\alpha]_{D}^{20}$  -170 (c 1, CHCl<sub>3</sub>); natural (mp 140-142°C;  $[\alpha]_{D}^{20}$  -170.47 (c 1, CHCl<sub>3</sub>).<sup>2</sup>