Eur J Med Chem (1992) 27, 67–71 © Elsevier, Paris

Short communication

Synthesis and biological evaluation of 2-[substituted acetyl]amino-5-alkyl-1,3,4-thiadiazoles

AK Shakya¹, GK Patnaik², P Mishra^{1*}

¹Medicinal chemistry division, Department of Pharmaceutical Sciences, Dr Harisingh Gour Vishwavidyalaya, Sagar 470003 (MP); ²Division of Pharmacology, Central Drug Research Institute, Lucknow, 226001 (UP), India

(Received 17 October 1990; accepted 28 May 1991)

Summary — Some new N-substituted acetyl derivatives of 2-amino-5-alkyl-1,3,4-thiadiazoles have been prepared and investigated for antihistaminic and spasmolytic activity on guinea pig ileum. A few of the compounds showed encouraging effects. In addition, some of the compounds also showed anti-inflammatory activity.

1,3,4-thiadiazoles / spasmolytic activity / anti-inflammatory activity / anti-histaminic activity

Introduction

Among pharmacologically active compounds the arrangement of atoms >N-C-C-N<, -O-C-C-N< and >N-C-C-O-CO- are of importance, as far as inhibitors of histamine, serotonin (5-HT) and acetyl choline are concerned [1]. Our interest in 2-substituted amino-5-alkyl-1,3,4-thiadiazoles as biological active compounds [2–4] prompted us to prepare a few title compounds having -NH-CO-CH₂-N< arrangement and to undertake their biological evaluation for anti-histaminic, spasmolytic activity together with anti-inflammatory activity.

Chemistry

The reaction of 2-amino-5-alkyl-1,3,4-thiadiazoles 1, 2 with chloroacetyl chloride gave the 2-chloro acetyl amino-5-alkyl-1,3,4-thiadiazoles 3, 4. Condensation of 3, 4 with an excess of secondary amines gave the desired products (5-16) (fig 1).

The structures of derivatives were confirmed by elemental analysis and spectral data. The IR and ¹H NMR spectra were in agreement with the proposed structure.

Pharmacological results and discussion

Pharmacological results of the synthesized compounds are reported in table I. The ALD_{50} of the synthesized compounds ranged between 681–1000 mg/kg (ip) in mice except for compound 14 which was found to be 215 mg/kg (ip).

In the studies of the effect of the compounds on isolated organs all the compounds showed antihistaminic activity as expected, and all showed competitive antagonism towards histamine except for compound **12**, which showed non-competitive antagonism. (Figure 2 represents log dose *versus* response curve for compound **8**.) The 50% inhibitory concentration (IC_{so}) of the compounds ranged

$$\begin{array}{cccccccc} N & - N \\ R & S & NH_2 \end{array} \xrightarrow{C1 CH_2 COC1} & R & N & - N \\ R & CH_3 & ;1 \\ R & C_2H_5 & ;2 \end{array} \xrightarrow{R & C_1CH_2 COC1} & R & S & NHC CH_2CI \\ \hline \begin{array}{c} R & R & C_1CH_2 & C_1CH_2 \\ R & S & NHC & CH_2CI \\ R & R & C_2H_5 & ;4 \end{array} \\ \hline \begin{array}{c} 3,4 & - & HN & O \\ \hline \\ C_6H_6 & R & S & NHC & CH_2 & N \\ \hline \end{array} \xrightarrow{S & -16} \end{array}$$

Fig 1.

^{*}Correspondence and reprints

Table I. Pharmacological data of synthesized compounds.

Product	ALD ₅₀ mg/kg ip	Anti-inflammatory activity ^a % inhibition	IC	pA_2			
			Hist	Ach	Nicotine	5-HT	value Drug-hist
5	681	27.6	2.47	1.25	2.89	2.68	6.0
6	1000	13.6	1.56	2.34	2.40	2.34	6.2
7	681	41.6	1.20	_	_	_	7.5
8	681	-	0.568	_	-	8.71	8.0
9	681	-	1.89	_	_	-	7.1
10	681	_	1.26	-	-	_	7.4
11	> 1000	_	1.96	_	_	_	6.1
12	1000	20.2	2.74	1.96	1.92	8.50	_
13	> 1000	_	4.15	_	_	-	6.0
14	215	22.1	2.06	1.89	_	_	6.2
15	681	24.0	2.15	_	_		6.3
16	681	-	2.60	_	-	-	6.2
PB	-	46.5		_		_	_
MP	_	-	0.08	1.86	-	_	9.3

^aDose 1/10th of ALD₅₀ po, PB: phenylbutazone, MP: mepyramine.

between 1.20 x 10^{-7} to 4.15 x 10^{-7} mol/l except for **8** which was found to be very potent (IC₅₀ 5.68 x 10^{-8} mol/l). The pA₂ values of compounds ranged between 6.0–8.0. However, compounds **5**, **6** and **12** were found to show spasmolytic activity as well.

The introduction of 1,3,4-thiadiazoles nucleus in structure -NH-CO-CH₂-N< showed anti-inflammatory activity in addition to antihistaminic and spasmolytic activity in a few compounds *ie* 5, 6, 7, 12, 14 and 15 determined by carrageenan-induced paw oedema method. In particular the effect showed by 7 was comparable to that of the test drug *ie*, phenylbutazone (P < 0.01).

The compounds were also screened on the cardiovascular system. None of the compounds antagonised the effects of adrenaline, acetylcholine, isoprenaline and histamine *in vivo*. Compound **12** produced a concentration dependent inhibition of the blood pressure. The maximum fall in BP (\downarrow 80 mm for 50 min) was observed at a dose of 5 mg/kg bw (iv) (fig 3).



Fig 2. Log dose-response curves of histamine in the absence and in the presence of compound 8. Histamine alone • • •; in the presence of compound 8, \bigcirc 3.5 x 10⁻⁸ mol/l; • • 7.0 x 10⁻⁸ mol/l.



Fig 3. Effect of intravenously administered compound 12, 5 mg/kg on blood pressure, respiration and nictitating membrane of anaesthetized cat.

Experimental protocols

Chemistry

Melting points were taken on a Toshniwal melting point determination apparatus and are uncorrected. ¹H NMR spectra were recorded on a Varian EM-360 (90 MHz) spectrometer using tetramethyl silane (TMS) as the internal standard. IR spectra were recorded on a Perkin-Elmer 157 spectrophotometer. Microanalysis was performed using Carlo-Erba 1106 instruments.

2-Amino-5-alkyl-1,3,4-thiadiazoles 1,2

These were prepared by treating thiosemicarbazide, aliphatic carboxylic acid and concentrated sulphuric acid according to the reported method [5]. 5-Methyl: yield 92%, mp 221–223°C (reported 223°C). 5-Ethyl: 86%, 192–194°C (194°C).

2-Chloroacetyl amino-5-alkyl-1,3,4-thiadiazoles 3, 4

Compound 1, 2 (0.10 mol) and chloroacetyl chloride (0.11 mol) were refluxed in benzene (250 ml) for 3 h on a water bath. Benzene was then distilled off and the crude product obtained was recrystallized from ethanol (95% v/v). 5-Methyl: yield 90%, mp 235–36°C. 5-Ethyl: 92%, 224–225°C.

2-[Substituted acetyl]amino-5-alkyl-1,3,4-thiadiazoles 5-16

To a stirred suspension of 0.05 mol of compound **3**, **4** in 100 ml of benzene, appropriate amine (0.125 mol) was added dropwise. The reaction mixture was stirred and refluxed for 2--3 h. After separation of the hydrochloride of amine the organic layer was extracted several times with 1 N HCl. On neutralization with 1 N NH₄OH, the crude product separated out. Recrystallization was done using appropriate solvents. Yields, melting points, recrystallization solvents and elemental analysis are given in table II; IR and ¹H NMR spectral data are reported in table III.

Pharmacology

For the pharmacological studies, adult cats (2-4 kg), guinea pigs (300-400 g), albino rats (100-120 g) and albino mice

(20–25 g) of either sex were used. Hydrochloride salts of the compounds were utilised. Depending on the requirement compounds were dissolved in either distilled water or normal saline. The control group received vehicle only.

Approximate lethal dose (ALD₅₀)

The albino mice were divided into different groups of 4 animals each. These were then administered with the graded doses of the compounds intraperitoneally (ip). Observation for mortality was then made. The ALD_{50} was taken from Horn's table [6].

Smooth muscle relaxant activity in vitro

A 2–3-cm long piece of ileum from a freshly killed guinea pig was suspended in an organ bath containing aerated Tyrode solution (pH 7.4) at 34°C [7]. Contractions were recorded on a kymograph through a frontal writing lever. Spasmolytic activity of the compounds was assessed by its ability to prevent the contraction induced by a submaximal concentration (g/ml) of acetyl choline (2.5×10^{-8}) histamine (3.0×10^{-8}), serotonin (5-HT, 5.0 $\times 10^{-7}$) or nicotine (2.5×10^{-6}). In all the isolated preparations, graded doses of the spasmolytic compounds were tested against the submaximal concentration of the spasmogen and the IC₅₀ was calculated graphically in each experiment.

The organ bath was then connected to 2 bottles so that it could be filled either with the one containing Tyrode solution or that containing the antagonist (compounds **5–16**). The submaximal contractions were first obtained with histamine (3 x $10^{-7.8} \times 10^{-4}$) in absence and in presence of antagonist. The dose ratio (DR) was determined by plotting log dose versus response curve. The pA_2 value was then determined by plotting log (DR-1) versus log molar concentration graph.

Anti-inflammatory activity

Carrageenan-induced paw oedema method [8] was adopted. The albino rats were divided into groups of 5 animals each. The paw oedema was determined plethysmographically. One percent suspension of carrageenan in normal saline was injected (0.1 ml) into the subplantar surface of the right hind paw through a 26-gauge needle. The compounds were administered 1 h prior to carrageenan injection (dose 1/10th of ALD₅₀ po). After 3 h the paw volume was again determined. The results were compared with that of standard drug phenylbutazone (dose 30 mg/kg po).

Effect on the cardiovascular system

Cats of either sex were anaesthetised by injecting pentobarbitone sodium (40 mg/kg ip) [9]. The femoral vein was exposed, cannulated and connected with rubber tubing to a burette filled with normal saline. A midline incision was made on the neck to expose the trachea and one end of the tracheal cannula was inserted into it. The other end of the cannula was connected to Mayer's tambour by rubber tubing for recording respiratory changes. The common carotid artery was then exposed and clipped at the lower end. An arterial cannula was inserted and tied. The other end of the cannula was connected to a manometer with rubber tubing. A 10% w/v sodium citrate solution ip was filled in the manometer and the rubber tubing. The clip on the carotid artery was released, the blood pressure was then recorded on a slowly moving kymograph with a pointer floating on the mercury column. During the experiment the sympathetic nerve was also stimulated electrically. The drugs were then administered through the cannulated femoral vein and the effects recorded.

$R \xrightarrow{N} N$ NH CO CH ₂ X									
Compd	R	X	Formula	Elem foun C	ental analy d (calculate H	vsis ed) N	Yield	$\stackrel{mp}{^{\circ}C^a}$	Recryst solvent
5	-сн _з	-N/C4H9 CH3	C ₁₀ H ₁₈ N ₄ OS	49.05 (49.58)	7.20 (7.44)	23.32 (23.14)	72	97	E LOH/ACOE L
6	- c ₂ H5	-N ^{/C4H9} -N ^{/CH3}	^c 11 ^H 20 ^N 4 ^{OS}	51.32 (51.56)	7.65 (7.81)	21.74 (21.87)	70	75	EtOH
7	- CH3	- N	C ₁₀ H ₁₆ N ₄ OS	49.89 (50.00)	6.52 (6.67)	23.11 (23.34)	52	161	EtOH/AcOEt
8	- ^C 2 ^H 5	- N	^C 11 ^H 18 ^N 4 ^{OS}	51.67 (51.96)	7.01 (7.08)	21.89 (22.04)	56	148	MeoH/AcOEt
9	- C H 3		^C 11 ^H 18 ^N 4 ^{OS}	51.81 (51.96)	6.94 (7.08)	21.76 (22.04)	65	114	EtOH/H20
10	- ^C 2 ^H 5		°12 ^H 20 ^N 4 ^{OS}	53.57 (53.73)	7.39 (7.46)	20.77 (20.89)	55	145	EtOH/H20
11	- C H 3	-N_N-CH3	^C 10 ^H 17 ^N 5 ^{OS}	46.91 (47.05)	6.54 (6.67)	27.39 (27.45)	36	98	EtOH
12	- ^c 2 ^H 5	-N N-CH3	^C 11 ^H 19 ^N 5 ^{OS}	49.00 (49.07)	6.96 (7.06)	25.85 (26.02)	45	141	EtOH
13	-сн ₃	-N CH ₂ -	^C 13 ^H 16 ^N 4 ^{OS}	56.42 (56.52)	5.65 (5.79)	20.15 (20.28)	51	140	EtOII
14	- ^c 2 ^H 5	-N/CH3 CH2-	C ₁₄ H ₁₈ N ₄ OS	57.88 (57.93)	6.03 (6.20)	19.23 (19.31)	52	83	EtOH
15	- CH3	-N NH	^C 9 ^H 15 ^N 5 ^{OS}	44.55 (44.81)	6.15 (6.22)	29.00 (29.04)	53	120	EtOH
16	- C ₂ H5	-N _NH	^C 10 ^H 17 ^N 5 ^{OS}	46.89 (47.05)	6.59 (6.67)	27.37 (27.45)	50	109	EtOH

 Table II. Physico-chemical data of 2-[substituted acetyl]amino-5-alkyl-1,3,4-thiadiazoles.

^aMelting points were determined in open capillary and are uncorrected.

Table III. Spectral data of compounds 5–16.

Compd	$IR(KBr) v(cm^{-1})$	¹ H NMR CDCl ₃ /TMS
5	3200, 1645, 1400	0.90 (t, 3H, CH ₃); 1.35 (m, 6H, -(CH ₂) ₃ CH ₃); 2.05 (s, 3H, CH ₃ td); 2.70 (s, 3H, CH ₃ -N); 3.90 (s, 2H, -CO-CH ₂); 5.75 (br, 1H, NH, D ₂ O exchangeable)
6	3145, 1620, 1415	0.80 (t, 3H, CH ₃); 1.00 (t, 3H, CH ₃); 1.60 (m, 6H, (CH ₂) ₃ CH ₃); 2.60 (br, 3H, CH ₃ -N); 4.20 (s, 2H, -CO-CH ₂); 7.50 (br, 1H, NH)
7	3150, 1655, 1425	1.75 (br, 6H, CH ₂ , pipd); 2.25 (s, 3H, CH ₃ , td); 2.80 (m, 4H, CH ₂ -N-CH ₂ , pipd); 3.90 (s, 2H, -CO-CH ₂); 6.50 (br, 1H, NH)
8	3175, 1640, 1420	1.15 (t, 3H, CH ₃ -CH ₂ -); 1.60 (br, 6H, CH ₂ pipd); 2.65 (m, 4H, -CH ₂ -N-CH ₂ -pipd); 3.20 (q, 2H, CH ₃ -CH ₂ -); 3.75 (s, 2H, -CO -CH ₂); 5.80 (br, 1H, NH)
9	3210, 1650, 1445	1.10 (d, 3H, -CH ₃ , pipd); 1.95 (s, 3H, CH ₃ td); 2.4 (m, 9H pipd); 3.75 (s, 2H,-CO-CH ₂); 7.50 (br, 1H, NH)
10	1700, 1440	0.95 (d, 3H, -CH ₃ pipd); 1.25 (t, 3H, -CH ₂ -CH ₃); 2.25 (br, 9H, pipd); 2.90 (q, 2H, -CH ₂ -CH ₃); 4.15 (s, 2H, -CO-CH ₂); 8.50 (br, 1H, NH)
11	3300–3100, 1650, 1440	2.05 (s, 3H, CH ₃ td); 2.60 (s, 3H, CH ₃); 3.75 (s, 8H, pipz); 4.25 (s, 2H, -CO-CH ₂ -); 5.50 (br, 1H, NH)
12	3285–3100, 1645	1.25 (t, 3H, CH ₃ -CH ₂); 2.75 (s, 3H, CH ₃); 3.45 (s, 8H, pipz); 3.95 (q, 2H, CH ₃ -CH ₂ -) 4.15 (s, 2H, -CO-CH ₂ -); 7.25 (br, 1H, NH)
13	3150, 1680, 1440, 1350	2.20 (s, 3H, CH ₃ td); 2.55 (s, 3H, CH ₃ -N); 3.15 (s, 2H, -CO-CH ₂); 3.65 (s, 2H, CH ₂); 6.85 (br, 1H, NH); 7.15 (t, 5H, Ar)
14	3150, 1660, 1410, 1320	0.95 (t, 3H, CH ₃ -CH ₂); 2.70 (s, 3H, CH ₃ -N); 3.65 (q, 2H, CH ₃ -CH ₂ -); 4.05 (s, 2H, -CO-CH ₂); 4.20 (s, 2H, CH ₂); 7.30 (t, 5H, Ar); 8.15 (br, 1H, NH)
15	3180, 1650, 1420, 1350	1.95 (s, 3H, CH ₃ td); 3.45 (br, 8H, pipz); 3.90 (s, 2H, -CO-CH ₂); 8.05 (br, 2H, NH)
16	3100, 1625, 1435, 1325	1.10 (t, 3H, CH ₃ -CH ₂); 3.65 (br, 8H, pipz); 4.10 (q, 2H, CH ₃ -CH ₂); 4.20 (s, 2H, -CO-CH ₂ -); 5.85 (br, 2H, NH)

pipd = piperidine; pipz = piperazine; td = thiadiazole.

Acknowledgments

The authors wish to thank the head of the Department of Pharmaceutical Sciences, HG Vishwavidyalaya, Sagar (MP) and the head of the Division of Pharmacology, Central Drug Research Institute for providing the necessary facilities. Thanks are also due to the head, RSIC, CDRI. Lucknow (UP) for providing spectral and microanalysis data. One of us (AKS) is grateful to the Council of Scientific and Industrial Research, Ministry of Science and Technology, New Delhi for providing a senior research fellowship.

References

1 Witiak DT, Cavesrti RC (1980) In: Principles of Medicinal

Chemistry (Foye WO, ed) Lea and Febiger, Philadelphia, 473-494

- 2 Mishra P, Reddy UM, Agrawal RK (1989) J Inst Chem (India) 61, 31-32
- 3 Mishra P, Shakya AK, Agrawal RK, Patnaik GK (1990) Ind J Pharmacol 22, 113-116
- 4 Mishra P, Shakya AK, Agrawal RK, Patnaik GK (1990) J Ind Chem Soc 67, 520-521
- 5 Funatsukuri G, Ueda M (1966) Jpn Pat 20,944; Chem Abstr 66, 46430
- 6 Horn HJ (1956) Biometrics 22, 311–317
- 7 Ghosh MN (1984) Textbook of Experimental Pharmacology. J Shinha and Co, Calcutta, 88–92
- 8 Winter CA, Risley EA, Nuss GW (1962) Proc Soc Exp Biol Med 111, 534-537
- 9 Burn JH (1952) Practical Pharmacology. Blackwell Sci Publ, Oxford, 25–30