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Regioisomeric synthesis of dihydrofuro[2,3-*d*]pyrimidines in a diastereoselective manner involving nitrogen ylides in one-pot three-component reaction

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Graphical Abstract

Leave this area blank for abstract info. Regioisomeric Synthesis of Dihydrofuro[2,3-d]pyrimidines in a Diastereoselective manner Involving Nitrogen Ylides in One-pot Three-component Reaction Leema Dutta, Meenakshi Sharma^a and Pulak J Bhuyan* Medicinal Chemistry Division, CSIR-North East Institute of Science and Technology, Jorhat-785006, Assam, India. ^aAcSIR, New Delhi Et_3N N CH₃ ArCHO + Ar CH₃CN ĊΗ₃ reflux CH 2 3 4 5

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Regioisomeric Synthesis of Dihydrofuro[2,3-d]pyrimidines in a Diastereoselective manner Involving Nitrogen Ylides in One-pot Three-component Reaction

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Synthesized some novel regionsomers of furo[2,3-d] pyrimidines in a diastereoselective manner *via* one-pot three-component reaction of barbituric acids, aryl aldehydes and pyridinium bromides in the presence of triethylamine as base. Both the isomers were obtained in comparable ratio with excellent overall yield.

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1. Introduction

Furopyrimidines, particularly furo[2,3-d]pyrimidines represent a very important class of compounds that possess wide range of biological activities such as antibacterial,¹ antifungal,² antitumor,³ antiviral,⁴ antifolate⁵ and anti-human cytomegalovirus activity.⁶ Compounds with this molecular motif are also active on blood circulatory system⁷ and stimulate the skin preparative regeneration.⁸ In view of the vast biological activities and importance to pharmaceutical discovery research, considerable attention has been diverted towards the synthesis and molecular manipulation of furo[2,3d]pyrimidines. Furo[2,3-d]pyrimidines are usually prepared either from suitably functionalized furan ring⁹ by generating the pyrimidine ring on the parent furan ring or from a pyrimidine derivative by constructing a furan ring on the parent pyrimidine ring.¹⁰ Very recently, an excellent review was published describing various synthetic routes for furopyrimidines, particularly furo[2,3-d]pyrimidines.¹¹

Nitrogen ylides represent an important class of nucleophilic or bifunctional reagents which can undergo cyclization, olefination, rearrangement reactions etc, and hence have wide application for the construction of diverse carbocyclic and heterocyclic compounds of synthetic and biological significance.¹² In this regard, pyridinium ylides have extensive

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application for the synthesis of dihydrofuran and its annelated derivatives which produced the desired compounds in a diastereosective manner. 13

In continuation of our work on pyrimidines,¹⁴ in the present paper, we report an unprecedented regioisomeric synthesis of some highly functionalized dihydrofuro[2,3-d]pyrimidines **4** and **5** in diastereoselective manner from one-pot threecomponents reaction of barbituric acids **1**, aryl aldehydes **2** and pyridinium bromides **3** in the presence of triethylamine as base (Scheme 1). During the reaction, nitrogen ylides were formed and involved in [4+1] annulation as well as [2+1] annulation processes which after intramolecular ring transformation produced the two regioisomeric products.

Synthesis of dihydrofuro[2,3-d]pyrimidines 4 and 5



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2. Results and Discussion

The study was initiated by performing the three-component reaction of N,N-dimethylbarbituric acid 1 (R¹ = Me), benzaldehyde 2 (Ar = Ph) and benzylpyridinium bromide 3 (R² = COPh) in various solvents and base to determine the optimal condition for the model reaction. Among the solvents viz water, ethanol, DMSO, dichloromethane and acetonitrile, the acetonitrile was found to be most effective. The reaction did not proceed in water as barbituric acid derivative 1 (R^1 = Me) was not soluble in the solvent. The model reaction was also investigated employing different bases viz. Et₃N, DBU and K₂CO₃. The results were summarized in Table 1. It was found that the reaction in acetonitrile and use of 1.2 mmol of Et₃N as base furnished the maximum yield of the two products 4a and 5a in less time duration in comparison to others (Entry 6). Therefore, all subsequent reactions were carried out under this reaction condition only. While, equimolar amount of the base is required just for the formation of the ylide from pyridinium salt, we observed that, the reaction proceeded in the presence of equimolar amount of the base providing very low yield of the products (Entry 5), and that indicated simultaneous formation of the ylide and the products in the reaction process.

Table 1^a: Optimisation of solvent and base

Entry Solvent Base Time Product Yield%***
Entry Solvent Base Time Product Yield%***

		(1.20 mmol)	(h)			
1	Water	Et_3N	10	NR	0	
2	DMSO	Et_3N	7	4a	22	2
				5a	18	
3	C ₂ H ₅ OH	Et ₃ N	8	4 a	14	
				5a	12	-
4	CH_2Cl_2	Et ₃ N	5	4a	13	-
				5 a	11	-
5	CH ₃ CN	Et ₃ N*	4) 4a	12	-
		Ċ		5a	10	1
6	CH ₃ CN	Et ₃ N	4	4 a	40	
				5a	38	1
7	CH ₃ CN	Et ₃ N**	4	4 a	36	
				5a	33	1
8	CH ₃ CN	K ₂ CO ₃	10	4 a	15	
				5a	12]
9	CH ₃ CN	DBU	8	4 a	23	
				5a	19	E a

Increase in the load of the catalyst beyond 1.2 mmol did not improve the yield of the products (Entry 7). The model reaction was initiated with the preparation of benzylpyridinium bromide **3** ($R^2 = COPh$) from the reaction of phenacyl bromide and pyridine in ethyl acetate at room temperature following the existing reaction procedure.¹⁵

Table 2^{*a*}: Synthesis of dihydrofuro[2,3-*d*]pyrimidines 4 and 5

C لار R ¹)	Et,	N R ¹	Ar		R ²
O	I ← O + Ar	CHO + CHO Br — CHO + CH3		$R^2 +$	o N) ∕'''Ar O
1	113	2 3 ^{R² reflu}	4 CH ₃		CH ₃	5
Ent.	\mathbf{R}^{1}	Ar	R ²	Prod.	Time	Yield ^b
				/	(h)	%
1	CH ₃	C_6H_5	COPh	4 a	4	40
				5a		38
2	CH ₃	p-CH ₃ -C ₆ H ₄	COPh	4b	4.5	36
-	~~~			5b		33
3	CH ₃	p-NO ₂ -C ₆ H ₄	COPh	4c	2.5	45
				5c		42
4	CH ₃	p-Cl-C ₆ H ₄	COPh	4d	3.5	43
_				5d		40
5	CH ₃	p-Br-C ₆ H ₄	COPh	4e	3.5	42
	ζ7	~ **		5e	_	40
6	н	C_6H_5	COPh	4f	5	35
			0005	5f	-	32
7	CH_3	C_6H_5	COOEt	4g	5	40
0	-		~~~~	5g	_	36
8	CH_3	p-OCH ₃ -C ₆ H ₄	COOEt	4h	6	32
0	<u>au</u>		COOL	5h	~	28
9	CH_3	p-CH ₃ -C ₆ H ₄	COOEt	41	6	36
10	CU		COOF	51	21	34
10	CH ₃	p-NO ₂ -C ₆ H ₄	COOEt	4j 	3h	44
11	CU		COOF	эј 41-	4	40
11	CH_3	p-CI-C ₆ H ₄	CODEI	4K	4	42
10	CU	n Dr C II	COOE	эк 41	15	39 41
12	СП3	<i>р</i> -ы-с ₆ п ₄	CODEI	41 51	4.3	41
12	TT	СЦ	COOE	51 4m	6	39 22
13	п	C ₆ ⊓ ₅	COUE	4111 5m	0	30
14	СЦ	СН	CN	Jili An	35	30
14	СП3	C ₆ 11 ₅	CN	-+11 5 m	5.5	34
15	СЦ		CN	311 40	3	38
15	C113	p-C1-C6114		-0 50	5	35

^aReaction conditions: *N*,*N*-dimethyl barbituric acid (1mmol), benzaldehyde (1mmol), benzyl pyridinium bromide (1mmol) *1 mmol; **1.5 mmol; *** isolated yield Ent. = Entry, Prod. = Product,

^aReaction conditions: Substituted Barbituric acid **1** (1mmol), Aryl aldehyde **2** (1mmol), Pyridinium ylide **3** (1mmol), Et₃N (1.20 mmol), CH₃CN (5ml), ^{*b*}isolated yield

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The benzylpyridinium bromide **3** ($R^2 = COPh$) so obtained was treated with *N*,*N*-dimethylbarbituric acid $\mathbf{1}$ ($\mathbf{R}^1 = \mathbf{M}\mathbf{e}$) and benzaldehyde 2 (Ar = Ph) in one-pot three-component reaction protocol in refluxing acetonitrle in the presence of triethylamine as base. The progress of the reaction was monitored by TLC which indicated the formation of two products. The reaction was allowed to reflux for 4 hours till the completion of the reaction. The solvent was removed under reduced pressure and the compounds were separated by chromatographic methods and isolated the expected dihydrofuro[2,3-d]pyrimidines 4a (Ar = Ph, $R^2 = COPh$) along with unexpected regeoisomeric compound 5a (Ar = Ph, R^2 = COPh) in 40% & 38 % yield respectively in a diastereoselective manner. The generality of the reaction was established by synthesizing the compounds **4a-o** and **5a-o** by utilizing various substituted barbituric acids 1, aryl aldehydes 2 and pyridinium salts 3 and characterizing them (Table 2). The structures of the compounds were ascertained from the spectroscopic data and elemental analysis. As for example, in the ¹H NMR spectra, the two protons at 2,3-positions of dihydrofuran ring of compound 4a (Entry 1) display two doublets at δ 4.04 and 4.19 with the vicinal coupling constant J = 9.5 Hz. It is well recognized that stereochemical relations among vicinal protons in fused dihydrofurans cannot be reliably determined by simply measuring coupling constants e.g. changing the size of the substituent at the dihydrofuran ring of benzodihydrofurans causes a reversal in the value of J_{cis} and J_{trans} .¹⁶ In view of that, the NOESY NMR spectra of the compound was studied properly where the two protons showed no corelation, and hence it was concluded that the product 4a was *trans* dihydrofuro[2,3-d]pyrimidines. In contrary, ¹H NMR spectra of the other compound (Entry 1) showed two protons of the dihydrofuran ring as two doublets at δ 5.16 (J = 5.5) and 6.04 (J = 5.5) ppm respectively which indicates the formation of isomeric dihydrofuro[2,3d]pyrimidine compound 5a. The study of the NOESY NMR spectra confirmed the formation of the cis form of the compound 5a as the two protons showed strong correlation in the spectra. The formation of these cis and trans forms of regioisomeric compounds are in contrary to many other reactions which produced exclusively one regioisomer of the fused dihydrofuran derivatives, and usually the trans form with coupling constants in the range of J = 4-6 Hz.¹⁸ In our present case, it might be because of the unique nonplanner structure of the substituted uracil moiety. In IR spectrum, stretching frequencies at 1687.2. 1557.6 cm⁻¹ of 4a and 1685.4, 1560.2 cm⁻¹ of **5a** confirmed the presence of the furan ring and amide C=O functional groups in both the The mass spectrum shows, a compounds. sharp distinguishable peak of compound 4a and 5a at 363.41 [M+H]⁺ and 363.32 [M+H]⁺ respectively. Interestingly, even though we considered the formation of the compound 5 as unexpected, a literature survey revealed one example which reported the formation of exclusively this type of isomeric compounds (two compounds only) by using sulfur ylides, and our product 5h was found to be comparable in all respect to one of that compounds.¹⁷ Again, although the compound 4a was the usual expected product from the reaction of nitrogen ylides, the formation of the unexpected product 5a in a comparable ratio might be because of the typical nonplanner structure of pyrimidinedione moiety.

Then we studied the effect of substituent of the substrates in the reaction process. It was observed that the reaction requires 2-6 h in refluxing conditions depending upon the substituent on aryl aldehyde and pyridinium ylide to afford the desired compounds. Since aromatic aldehydes are more reactive than aliphatic towards nucleophillic attack, hence more emphasis was given on aryl aldehyde. The reactions were smooth and easy in aldehydes having electron withdrawing substituent (NO₂, Cl, Br) in comparison to electron releasing substituent (H, CH₃) and also provide better yield of the products in less reaction time.

Mechanism for the formation of dihydrofuro[2,3-d]pyrimidines 4a and 5a



To expand the scope of the reaction, different pyridinium ylides were generated from various pyridinium salts *viz.* phenacyl pyridinium bromide, 1-ethoxy carbonyl methyl pyridinium iodide¹⁹ and 1-cyanomethyl pyridinium bromide,²⁰ and utilized during the course of the study. Benzylpyridinium ylide obtained from benzylpyridinium bromide was most effective which may be accounted from the mechanism that interprets the resonance stabilization of enolate ion by phenyl

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group. The reaction with 1-cyanomethyl pyridinium ylide was not smooth, and provided less yields even on increasing the reaction time, hence we have limited our experiment with this ylide.

The effect of substituents in barbituric acids was also studied properly and observed that N,N-dimethylbarbituric acid was more reactive than N-methylbarbituric acid and products were obtained in better yield in less reaction time. Moreover, in case of N-methylbarbituric acid, selectively the more favoured N-methyl amidic carbonyl group involved in the cyclization process to produce only the two regioisomeric dihydrofuro[2,3-d]pyrimidines (Entry 6 & 13, Table 2) as expected, and such selectivity is well precedented.²¹

A reasonable mechanism for the formation of the regioisomers is depicted in the scheme 2 taking the formation of 4a and 5a as example. The reaction occurred via initial formation of the knoevenagel condensed product [A] from the reaction of N,N-dimethylbarbituric acid 1 ($R^1 = Me$) and benzaldehyde 2 (Ar = Ph), and the ylide [B] from the benzylpyridinium bromide **3** ($R^2 = COPh$) in the presence of triethyl amine (Scheme 2). The intermediates [A] and [B] then reacted to produce the intermediate [C] which followed two different reaction path to afford the product 4a & 5a respectively. In path I, the intermediate [C] produced the compound 4a by a direct cyclization process by eliminating pyridine. But in path II, first, intermediate [D] with the cyclopropane ring was formed, which underwent an intramolecular ring transformation, and proceeded through the intermediates [E] and [F] in the presence of base to produce the product 5a. The mechanism for the formation of both types of compounds from ylides is well documented.^{13c,17}

In conclusion, we have reported an unprecedented regioisomeric synthesis of some highly functionalized dihydrofuro[2,3-*d*]pyrimidines by generating nitrogen ylides in an one-pot three-components reaction barbituric acids, aryl aldehydes and pyridinium bromides in the presence of triethylamine as base. The products were formed in a diastereoselective manner and overall good yields. A suitable mechanism is given for the reaction process which shows that nitrogen ylides involve in [4+1] annulations as well as [2+1] annulation followed by intramolecular ring transformation in the presence of base to afford two isomeric products.

4. Experimental Section

All chemicals were purchased from Merck and Aldrich chemical companies. The reagents and solvents were used without drying. The IR spectra were recorded on Perkin Elmer system 2000 FTIR spectrometer. ¹H NMR and ¹³C NMR Spectra were recorded on Bruker AV500 Avance – III 500 MHz and 125 MHz FT NMR in CDCl₃ using TMS as an internal standard. Chemical shifts (δ units) are given from TMS (0 ppm). Chemical shifts for CDCl₃ were reported at 7.26 ppm (δ units). Mass spectra were recorded in Bruker Daltonics ESQUIRE 3000 LC ESI ion trap mass spectrometer. Analytical thin layer chromatography (TLC) was performed using E-Merck aluminium-backed silica gel plates coated with 0.2 mm thickness of silica gel. Melting points (uncorrected) were determined in open capillary tubes on a Buchi B-540 apparatus. Elemental analyses were

performed on Perkin-Elmer 2400 spectrometer at the Analytical Chemistry Division, CSIR-NEIST, Jorhat.

General Procedure for preparation of dihydrofuro[2,3*d*]pyrimidines 4 & 5: A mixture of *N*,*N*-dimethylbarbituric acid 1 ($\mathbb{R}^1 = \mathbb{M}e$, 156 mg, 1mmol), benzaldehyde 2 (Ar = Ph, 106 mg, 1mmol) and benzylpyridinium 3 ($\mathbb{R}^2 = \mathbb{C}OPh$, 278 mg, 1mmol) in acetonitrile (5 mL) were taken in round bottom flask. Triethyl amine (121 mg, 1.20 mmol) were added to the reaction mixture and allowed to reflux for 4 h at 80 °C. The solvent was removed under reduced pressure, and the crude products were purified by column chromatography (silica gel 100-200#, petroleum ether/ethyl acetate, 9:1 ratio), which afforded the desired dihydrofuro[2,3-*d*]pyrimidine derivatives **4a** and **5a** in 40% (144 mg) and 38% (137 mg) respectively.

Similarly, compounds **4b-o** and **5b-o** were synthesized and characterized.

6-Benzoyl-1,3-dimethyl-5-phenyl-5,6-dihydrofuro[2,3-d]-

pyrimidine-2,4(1H,3H)-dione (4a): Yield: 144 mg (40%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.50; mp. 134.0-135.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.22 (s, 3H), 3.24 (s, 3H), 4.17 (d, J = 9.5 Hz, 1H), 4.26 (d, J = 9.5 Hz, 1H), 7.24-8.24 (m, 10H); ¹³C NMR (CDCl₃, 125 MHz) δ 28.8, 29.0, 47.9, 87.4, 90.6, 128.1 (2C), 128.3 (2C), 128.6 (2C), 128.8, 129.5 (2C), 131.0, 133.7, 136.0, 151.3, 163.4, 165.7, 191.2; IR (KBr, cm⁻¹) v_{max}:1180.7, 1451.9, 1679.1, 2960.4; MS (ESI): 363.4 [M+H]⁺; Anal. Cald. For C₂₁H₁₈N₂O₄: C, 69.60; H, 5.01; N, 7.73%. Found: C, 69.72; H, 5.02; N, 7.75%.

5-Benzoyl-1,3-dimethyl-6-phenyl-5,6-dihydrofuro-[2,3-d]-

pyrimidine-2,4(1H,3H)-dione (5a): Yield: 137 mg (38%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.32; mp. 131.0.-132.1 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.19 (s, 3H), 3.32 (s, 3H), 5.16 (d, J = 5.5 Hz, 1H), 6.04 (d, J = 5.5 Hz, 1H), 7.18-8.02 (m, 10H); ¹³C NMR (CDCl₃, 125 MHz) δ 27.9, 29.2, 53.9, 87.4, 89.7, 125.6 (2C), 128.9 (2C), 129.2 (2C), 129.2 (2C), 129.6, 133.1, 133.9, 135.7, 149.4, 159.4, 162.2, 191.5; IR (KBr, cm⁻¹) v_{max}:1180.5, 1451.4, 1678.3, 2960.2; MS (ESI): 363.3 [M+H]⁺; Anal. Cald. For C₂₁H₁₈N₂O₄: C, 69.60; H, 5.01; N, 7.73%. Found: C, 69.55; H, 5.03; N, 7.74%.

6-Benzoyl-1,3-dimethyl-5-(p-tolyl)-5,6-dihydrofuro-[2,3-d-

]pyrimidine-2,4(1H,3H)-dione (4b): Yield: 135 mg (36%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.52; mp. 131.1.-133.6 °C; ¹H NMR (CDCl₃, 500 MHz) δ 2.17 (s, 3H), 3.21 (s, 3H), 3.24 (s, 3H), 4.03 (d, *J* = 9.5 Hz, 1H), 4.20 (d, *J* = 9.5 Hz, 1H), 7.21-7.64 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.1, 28.8, 28.9, 46.7, 86.3, 90.1, 122.9, 128.1 (2C), 128.8 (2C),130.1, 131.1 (2C), 131.5 (2C), 133.8, 135.8, 151.1, 163.3, 165.4, 190.7; IR (KBr, cm⁻¹) v_{max}: 1180.9, 1452.2, 1678.8, 2960.5; MS (ESI): 377.1 [M+H]⁺; Anal. Cald. For C₂₂H₂₀N₂O₄: C, 70.20; H, 5.36; N, 7.44%. Found: C, 70.32; H, 5.37; N, 7.42%.

5-Benzoyl-1,3-dimethyl-6-(p-tolyl)-5,6-dihydrofuro-[2,3-d]-

pyrimidine-2,4(1H,3H)-dione (**5b**): Yield: 124 mg (33%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.33; mp. 130.2.-130.9 °C; ¹H NMR (CDCl₃, 500 MHz) δ 2.17 (s, 3H), 3.20 (s, 3H), 3.25 (s, 3H), 4.95 (d, J = 6 Hz, 1H), 5.82 (d, J = 6 Hz, 1H), 7.18-7.62 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 21.1, 28.7, 28.8, 45.6, 87.2, 90.3, 123.4, 128.2 (2C), 129.3 (2C),130.4, 131.0 (2C), 131.7 (2C), 133.8, 136.2, 150.4, 164.1, 165.2, 191.2; IR (KBr, cm⁻¹) v_{max}: 1181.6, 1451.8, 1678.7, 2960.9; MS (EI): 377.3 [M+H]⁺; Anal. Cald. For C₂₂H₂₀N₂O₄: C, 70.20; H, 5.36; N, 7.44%. Found: C, 70.18; H, 5.34; N, 7.45%.

6-Benzoyl-1,3-dimethyl-5-(p-nitro-phenyl)-5,6-dihydrofuro-

[2,3-*d*]*pyrimidine*-2,4(1H,3H)-*dione* (4c): Yield: 183.2 mg (45%); Orange solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.47; mp. 151.5-153.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.20 (s, 3H), 3.22 (s, 3H), 4.15 (d, *J* = 9.5 Hz, 1H), 4.25 (d, *J* = 9.5 Hz, 1H), 7.22-8.28 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 29.0, 29.2, 41.1, 87.4, 90.3, 123.5 (2C), 127.9 (2C), 128.9 (2C), 130.5 (2C), 134.0, 135.6, 138.6, 147.8, 150.9, 163.3, 164.9, 190.1; IR (KBr, cm⁻¹) v_{max}: 1180.6, 1451.4, 1678.1, 2960.3; MS (ESI): 408.3 [M+H]⁺; Anal. Cald. For C₂₁H₁₇N₃O₆: C, 61.91; H, 4.21; N, 10.31%. Found: C, 62.00; H, 4.22; N, 10.29%.

5-Benzoyl-1,3-dimethyl-6-(p-nitro-phenyl)-5,6-dihydrofuro-

[2,3-*d*]*pyrimidine*-2,4(1*H*,3*H*)-*dione* (**5***c*): Yield: 171 mg (42%); Orange solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.29; mp. 149.2-150.9 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.24 (s, 3H), 3.35 (s, 3H), 5.01 (d, *J* = 6 Hz, 1H), 6.01 (d, *J* = 6 Hz, 1H), 7.03-8.20 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 28.9, 29.5, 53.0, 80.3, 85.8, 123.5 (2C), 128.0 (2C), 128.9 (2C), 130.5 (2C), 133.9, 135.7, 138.6, 147.8, 151.5, 162.9, 164.9, 190.0; IR (KBr, cm⁻¹) v_{max}: 1180.7, 1451.3, 1678.0, 2960.2; MS (ESI): 408.5 [M+H]⁺; Anal. Cald. For C₂₁H₁₇N₃O₆: C, 61.91; H, 4.21; N, 10.31%. Found: C, 61.90; H, 4.20; N, 10.32%.

6-Benzoyl-1,3-dimethyl-5-(p-chloro-phenyl)-5,6-dihydrofuro-

[2,3-*d*]*pyrimidine*-2,4(1H,3H)-*dione* (4d): Yield: 170.3 mg (43%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.48; mp. 146.7-148.9 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.14 (s, 3H), 3.18 (s, 3H), 3.98 (d, *J* = 9.5 Hz, 1H), 4.15 (d, *J* = 9.5 Hz, 1H), 7.09-7.85 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 28.8, 28.9, 41.4, 86.7, 90.2, 128.1 (2C), 128.6 (2C), 128.8 (2C), 129.3. 131.2 (2C), 133.8, 134.6, 135.8, 151.2, 163.3, 165.4, 190.7; IR (KBr, cm⁻¹) v_{max}: 1181.0, 1451.8, 1680.1, 2960.7; MS (ESI): 397.1 [M+H]⁺; Anal. Cald. For C₂₁H₁₇ClN₂O₄: C, 63.56; H, 4.32; N, 7.06%. Found: C, 63.71; H, 4.31; N, 7.07%.

5-Benzoyl-1,3-dimethyl-6-(p-chloro-phenyl)-5,6-dihydrofuro-

[2,3-d]pyrimidine-2,4(1H,3H)-dione (5d): Yield: 158 mg (40%); Light Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.31; mp. 139.9-142.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.27 (s, 3H), 3.41 (s, 3H), 5.18 (d, *J* = 5.5 Hz, 1H), 6.11 (d, *J* = 5.5 Hz, 1H), 7.21-7.62 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 28.0, 29.7, 53.8, 87.3, 89.8, 127.2 (2C), 128.7 (2C), 129.3 (2C), 129.4 (2C), 134.1, 135.6, 135.6, 136.3, 151.3, 159.3, 162.0, 197.8 ; IR (KBr, cm⁻¹) v_{max}: 1180.5, 1450.7, 1679.8, 2960.5; MS (ESI): 397.3 [M+H]⁺; Anal. Cald. For C₂₁H₁₇ClN₂O₄: C, 63.56; H, 4.32; N, 7.06%. Found: C, 63.78; H, 4.33; N, 7.05%.

6-Benzoyl-1,3-dimethyl-5-(p-bromo-phenyl)-5,6-dihydrofuro-[2,3-d]pyrimidine-2,4(1H,3H)-dione (4e): Yield: 185 mg

(42%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.48; mp. 143.1-144.5 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.12

(s, 3H), 3.16 (s, 3H), 3.94 (d, J = 10.5 Hz, 1H), 4.15 (d, J = 10.5 Hz, 1H), 7.12-7.89 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 28.8, 28.9, 46.7, 87.3, 89.8, 122.9, 128.1 (2C), 128.8 (2C), 130.1, 131.1 (2C), 131.5 (2C), 133.8, 135.8, 151.2, 163.3, 165.4, 190.8; IR (KBr, cm⁻¹) v_{max}: 1180.4, 1451.9, 1678.9, 2960.7; MS (ESI): 442.2 [M+H]⁺; Anal. Cald. For C₂₁H₁₇BrN₂O₄: C, 57.16; H, 3.88; N, 6.35%. Found: C, 57.35; H, 3.89; N, 6.33%.

5-*Benzoyl-1,3-dimethyl-6-(p-bromo-phenyl)-5,6-dihydrofuro-*[2,3-*d*]*pyrimidine-2,4(1H,3H)-dione* (5*e*): Yield: 176 mg (40%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.32; mp. 137.2-138.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.12 (s, 3H), 3.28 (s, 3H), 4.72 (d, *J* = 6 Hz, 1H), 5.97 (d, *J* = 6 Hz, 1H), 7.08-7.88 (m, 9H); ¹³C NMR (CDCl₃, 125 MHz) δ 29.5, 29.7, 53.7, 87.2, 89.7, 123.6, 127.4 (2C), 128.3 (2C), 128.7 (2C), 132.5 (2C), 134.2, 135.7, 136.9, 151.3, 159.4, 162.2, 197.7; IR (KBr, cm⁻¹) v_{max}: 1180.2, 1452.0, 1678.4, 2960.8 ; MS (ESI): 442.5 [M+H]⁺; Anal. Cald. For C₂₁H₁₇BrN₂O₄: C, 57.16; H, 3.88; N, 6.35%. Found: C, 57.25; H, 3.87; N, 6.36%.

6-Benzoyl-3-methyl-5-phenyl-5,6-dihydrofuro[2,3-d]-

pyrimidine-2,4(1H,3H)-dione (4f): Yield: 122 mg (35%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.49; mp.137.5-138.1 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.13 (s, 3H), 4.03 (d, J = 9.5 Hz, 1H), 4.17 (d, J = 9.5 Hz, 1H), 7.12-7.90 (m, 10H), 10.33 (s, 1H, NH); ¹³C NMR (CDCl₃, 125 MHz) δ 28.7, 47.8, 87.2, 91.4, 128.3 (2C), 128.6 (2C), 129.1, 129.4 (2C), 129.6 (2C), 131.1, 133.4, 134.0, 152.4, 162.3, 165.8, 190.3; IR (KBr, cm⁻¹) v_{max}: 1191.0, 1681.2, 1744.3, 2924.9; MS (ESI): 349.4 [M+H]⁺; Anal. Cald. For C₂₀H₁₆N₂O₄: C, 68.96; H, 4.63; N, 8.04%. Found: C, 68.92; H, 4.64; N, 8.05%.

5-Benzoyl-3-methyl-6-phenyl-5,6-dihydrofuro[2,3-d]-

pyrimidine-2,4(1H,3H)-dione (5f) : Yield: 112 mg (32%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.30; mp. 135.2-135.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.18 (s, 3H), 5.19 (d, J = 5.5 Hz, 1H), 6.03 (d, J = 5.5 Hz, 1H), 7.15-8.01 (m, 10H), 10.21 (s, 1H, NH); ¹³C NMR (CDCl₃, 125 MHz) δ 27.8, 55.4, 87.3, 89.8, 125.6 (2C), 128.8 (2C), 129.0 (2C), 129.2 (2C), 129.7, 133.2, 133.9, 135.6, 149.5, 158.3, 162.3, 191.3; IR (KBr, cm⁻¹) v_{max}:1191.6, 1680.8, 1743.6, 2925.5; MS (ESI): 349.3 [M+H]⁺; Anal. Cald. For C₂₀H₁₆N₂O₄: C, 68.96; H, 4.63; N, 8.04%. Found: C, 68.99; H, 4.62; N, 8.03%.

Ethyl-1,3-dimethyl-2,4-dioxo-5-phenyl-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-6-carboxylate* (*4g*): Yield: 132 mg (40%); Dark yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.49; mp. 182.5-183.9 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.25 (t, 3H), 3.25 (s, 3H), 3.32 (s, 3H), 4.04 (d, *J* = 9.5 Hz, 1H), 4.18-4.25 (m, 3H), 7.18-7.47 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 28.0, 29.4, 52.5, 62.0, 88.7, 90.0, 125.6 (2C), 126.7, 128.7 (2C), 137.4, 151.4, 159.2, 161.9, 171.1; IR (KBr, cm⁻¹) v_{max} : 1192.0, 1681.3, 1734.9, 2925.0; MS (ESI): 331.2 [M+H]⁺; Anal. Cald. For C₁₇H₁₈N₂O₅: C, 61.81; H, 5.49; N, 8.48%. Found: C, 61.59; H, 5.50; N, 8.45%.

Ethyl-1,3-dimethyl-2,4-dioxo-6-phenyl-1,2,3,4,5,6hexahydrofuro[2,3-d]pyrimidine-5-carboxylate (5g): Yield: 119.1 mg (36%); Dark yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.26; mp. 178.7-179.2 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.18 (t, 3H), 3.21 (s, 3H), 3.32 (s, 3H), 4.15-4.22 (m, 3H), 5.96 (d, *J* = 6 Hz, 1H), 7.18-7.37 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 28.3, 29.5, 52.5, 62.0, 85.9, 90.0, 125.5 (2C), 128.5 (2C), 129.0, 137.5, 151.4, 159.3, 161.8, 171.4; IR (KBr, cm⁻¹) v_{max}: 1191.4, 1681.1, 1735.2, 2925.2; MS (ESI): 331.4 [M+H]⁺; Anal. Cald. For C₁₇H₁₈N₂O₅: C, 61.81; H, 5.49; N, 8.48%. Found: C, 61.93; H, 5.47; N, 8.50%.

Ethyl-5-(p-methoxyphenyl)-1,3-dimethyl-2,4-dioxo-

1,2,3,4,5,6-hexahydrofuro-[2,3-d]pyrimidine-6-carboxylate (**4**h): Yield: 115 mg (32%); Colourless solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.50; mp. 173.4-175.2 °C; ¹H NMR (CDCl₃, 500 MHz) δ 0.97 (t, 3H), 3.13 (s, 3H), 3.27 (s, 3H), 3.79(s, 3H,OMe), 3.94(d, *J* = 9.5 Hz, 1H), 4.08 (d, *J* = 9.5 Hz, 1H), 4.38 (m, 2H), 6.84-7.25 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 28.5, 29.6, 52.5, 60.2, 62.4, 87.7, 89.2, 123.6, 127.2 (2C), 132.0(2C), 151.4, 159.2, 161.8, 16,1.9 169.4; IR (KBr, cm⁻¹) v_{max}: 1190.1, 1683.1, 1734.0, 2923.7; MS (ESI): 361.2 [M+H]⁺; Anal. Cald. For C₁₈H₂₀N₂O₆: C, 59.99; H, 5.59; N, 7.77%. Found: C, 59.97; H, 5.58; N, 7.75%.

$\label{eq:expectation} Ethyl-6-(p-methoxyphenyl)-1, 3-dimethyl-2, 4-dioxo-$

1,2,3,4,5,6-hexahydrofuro-[2,3-d]pyrimidine-5-carboxylate (**5h**): Yield: 101 mg (28%); Colourless solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.28; mp. 170.1-172.3°C; ¹H NMR (CDCl₃, 500 MHz) δ 1.24 (t, 3H), 3.26 (s, 3H), 3.33 (s, 3H), 3.76 (s, 3H, OMe), 4.16-4.26 (m,3H), 5.98 (d, *J* = 6 Hz, 1H), 7.19-7.38 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 28.0, 29.5, 52.4, 56.0, 61.9, 86.0, 90.3, 114.2, 127.8 (2C),129.5 (2C), 151.4, 159.2, 161.7, 161.9, 171.5; IR (KBr, cm⁻¹) v_{max}: 1190.7, 1684.1, 1733.7, 2924.5; MS (ESI): 361.1 [M+H]⁺; Anal. Cald. For C₁₈H₂₀N₂O₆: C, 59.99; H, 5.59; N, 7.77%. Found: C, 60.01; H, 5.60; N, 7.79%.

Ethyl-1,3-dimethyl-2,4-dioxo-5-(p-tolyl)-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-6-carboxylate* (*4i*): Yield: 123 mg (36%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.51; mp. 175.4-177.6 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.19 (t, 3H), 2.24 (s, 3H), 3.08 (s, 3H), 3.11 (s, 3H), 3.67 (d, *J* = 9.5 Hz, 1H), 3.84 (d, *J* = 9.5 Hz, 1H), 4.19 (m, 2H),7.01-7.23 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 20.9, 28.8, 29.5, 46.0, 62.2, 86.0, 88.2, 126.6 (2C), 129.3 (2C), 134.4, 139.6, 151.5, 161.9, 162.8, 171.5; IR (KBr, cm⁻¹) v_{max}: 1190.9, 1680.1, 1735.1, 2925.5; MS (ESI): 345.5 [M+H]⁺; Anal. Cald. For C₁₈H₂₀N₂O₅: C, 62.78; H, 5.85; N, 8.13%. Found: C, 63.00; H, 5.87; N, 8.10%.

Ethyl-1,3-dimethyl-2,4-dioxo-6-(p-tolyl)-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-5-carboxylate* (5*i*): Yield: 116 mg (34%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.28; mp. 169.9-172.4 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.23 (t, 3H), 2.32 (s, 3H), 3.30 (s, 3H), 3.39 (s, 3H), 4.22-4.30 (m, 3H), 5.99 (d, *J* = 6 Hz, 1H),6.84-7.38 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 21.1, 28.0, 29.2, 52.4, 61.9, 86.0, 88.9, 125.8 (2C), 128.9, 129.6 (2C), 139.6, 151.4, 159.2, 161.7, 169.6; IR (KBr, cm⁻¹) v_{max}: 1190.8, 1680.0, 1734.8, 2925.6; MS (ESI): 345.1 [M+H]⁺; Anal. Cald. For C₁₈H₂₀N₂O₅: C, 62.78; H, 5.85; N, 8.13%. Found: C, 62.93; H, 5.86; N, 8.15%.

Ethyl-1,3-dimethyl-5-(p-nitrophenyl)-2,4-dioxo-1,2,3,4,5,6hexahydrofuro[2,3-d]pyrimidine-6-carboxylate (*4j*): Yield: 165 mg (44%); Dark orange solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.44; mp. 204.3- 207.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.21 (t, 3H), 3.13 (s, 3H), 3.31 (s, 3H), 3.65 (d, *J* = 9.5 Hz, 1H), 3.91 (d, *J* = 9.5 Hz, 1H), 4.22 (m, 2H), 7.38 (d, *J* = 11.5 Hz, 2H), 8.11 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (CDCl₃, 125 MHz) δ 13.9, 28.8, 28.9, 44.6, 62.1, 87.1, 89.1, 123.4 (2C), 124.2, 130.4 (2C), 138.5, 147.7, 150.9, 163.2, 165.4; IR (KBr, cm⁻¹) v_{max}: 1191.2, 1515.5, 1734.0, 2925.9; MS (ESI): 376.3 [M+H]⁺; Anal. Cald. For C₁₇H₁₇N₃O₇: C, 54.40; H, 4.57; N, 11.20%. Found: C, 54.59; H, 4.58; N, 11.18%.

Ethyl-1,3-dimethyl-6-(p-nitrophenyl)-2,4-dioxo-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-5-carboxylate* (5*j*): Yield: 150 mg (40%); Orange solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.22; mp. 200.9-202.1 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.41 (t, 3H), 3.21 (s, 3H), 3.40 (s, 3H), 4.09 (d, *J* = 6 Hz, 1H), 4.45 (m, 2H), 6.00 (d, *J* = 6 Hz, 1H), 7.20-8.20 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.1, 28.9, 29.0, 52.3, 62.2, 88.2, 89.0, 123.3 (2C), 125.3, 130.5 (2C), 138.4, 147.6, 151.0, 163.1, 166.3; IR (KBr, cm⁻¹) v_{max}: 1191.1, 1515.4, 1680.3, 1734.1, 2925.8 ; MS (ESI): 376.6 [M+H]⁺; Anal. Cald. For C₁₇H₁₇N₃O₇: C, 54.40; H, 4.57; N, 11.20%. Found: C, 54.49; H, 4.55; N, 11.17%.

Ethyl-1,3-dimethyl-5-(p-chlorophenyl)-2,4-dioxo-1,2,3,4,5,6-hexahydrofuro[*2,3-d*]*pyrimidine-6-carboxylate* (*4k*): Yield: 153 mg (42%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.46; mp. 198.1- 199.5 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.21 (t, 3H), 3.13 (s, 3H), 3.28 (s, 3H), 3.63 (d, *J* = 9.5 Hz, 1H), 3.81 (d, *J* = 9.5 Hz, 1H), 4.20 (m, 2H), 7.09-7.75 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 13.9, 28.8, 28.8, 46.2, 61.9, 87.4, 89.9, 128.4 (2C), 129.3, 130.7 (2C), 134.5, 151.2, 163.1, 165.8, 165.9; IR (KBr, cm⁻¹) v_{max}: 1192.5, 1680.7, 1735.1, 2925.4; MS (ESI): 365.3 [M+H]⁺; Anal. Cald. For C₁₇H₁₇ClN₂O₅: C, 55.97; H, 4.70; N, 7.68%. Found: C, 56.08; H, 4.69; N, 7.71%.

Ethyl-1,3-dimethyl-6-(p-chlorophenyl)-2,4-dioxo-1,2,3,4,5,6hexahydrofuro[*2,3-d*]*pyrimidine-5-carboxylate* (*5k*): Yield: 142.1 mg (39%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.23; mp. 195.7-197.4 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.29 (t, 3H), 3.25 (s, 3H), 3.30 (s, 3H), 4.10 (d, *J* = 6 Hz, 1H), 4.20 (m, 2H), 5.93 (d, *J* = 6 Hz, 1H), 7.13-7.34 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.1, 29.3, 29.7, 52.7, 62.2, 85.9, 88.4, 127.1 (2C), 128.3, 129.2 (2C), 130.6, 151.4, 159.2, 161.9, 171.3; IR (KBr, cm⁻¹) v_{max}: 1191.3, 1680.5, 1734.9, 2925.8; MS (ESI): 365.1 [M+H]⁺; Anal. Cald. For C₁₇H₁₇ClN₂O₅: C, 55.97; H, 4.70; N, 7.68%. Found: C, 56.10; H, 4.71; N, 7.70%.

Ehyl-1,3-dimethyl-5-(p-bromophenyl)-2,4-dioxo-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine*-6-*carboxylate* (4*l*): Yield: 167 mg (41%); Light orange solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.46; mp. 193.2-194.6.2 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.28 (t, 3H), 3.13 (s, 3H), 3.32 (s, 3H), 3.62 (d, *J* = 9.5 Hz, 1H), 3.79 (d, *J* = 9.5 Hz, 1H), 4.20 (m, 2H), 7.05-7.43 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 13.9, 28.8, 29.5, 46.1, 61.9, 85.0, 86.8, 122.7, 130.0, 131.0 (2C), 131.4 (2C), 151.2, 163.1, 165.7, 165.8; IR (KBr, cm⁻¹) v_{max} : 1190.8, 1681.1, 1734.7, 2925.6; MS (ESI): 410.1 [M+H]⁺; Anal. Cald. For C₁₇H₁₇BrN₂O₅: C, 49.89; H, 4.19; N, 6.85%. Found: C, 50.03; H, 4.20; N, 6.83%.

Ethyl-1,3-dimethyl-6-(p-bromo-phenyl)-2,4-dioxo-1,2,3,4,5,6-hexahydrofuro-[2,3-d]pyrimidine-5-carboxylate (51): Yield: 143 mg (35%); Light orange solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.24; mp. 190.7-192.3 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.32 (t, 3H), 3.32 (s, 3H), 3.39 (s, 3H), 4.17 (d, J = 6 Hz, 1H), 4.30 (m, 2H), 5.98 (d, J = 6 Hz, 1H), 7.14-7.61 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 14.0, 28.5, 29.5, 52.5, 62.4, 85.7, 87.7, 123.6, 127.2 (2C), 132.0 (2C), 136.4, 151.3, 159.1, 161.8, 171.0; IR (KBr, cm⁻¹) v_{max}: 1191.7, 1681.2, 1734.1, 2925.8 ; MS (ESI): 410.3 [M+H]⁺; Anal. Cald. For C₁₇H₁₇BrN₂O₅: C, 49.89; H, 4.19; N, 6.85%. Found: C, 50.05; H, 4.20; N, 6.86%.

Ethyl-3-methyl-2,4-dioxo-5-phenyl-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-6-carboxylate* (*4m*): Yield: 105 mg (33%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.47; mp. 186.6-188.40 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.24 (t, 3H), 3.25 (s, 3H), 4.01 (d, *J* = 9.5 Hz, 1H), 4.15-4.22 (m, 3H), 7.15-7.37 (m, 5H), 10.31 (s, 1H, NH); ¹³C NMR (CDCl₃, 125 MHz) δ 14.1, 28.0, 53.2, 62.1, 88.8, 91.2, 125.6 (2C), 126.8, 128.6 (2C), 137.5, 151.3, 158.3, 162.9, 172.3; IR (KBr, cm⁻¹) v_{max}:1191.2, 1680.5, 1743.4, 2925.3; MS (ESI): 317.3 [M+H]⁺; Anal. Cald. For C₁₆H₁₆N₂O₅: C, 60.75; H, 5.10; N, 8.86%. Found: C, 60.79; H, 5.09; N, 8.85%.

Ethyl-3-methyl-2,4-dioxo-6-phenyl-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-5-carboxylate* (*5m*): Yield: 95.35 mg (30%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.25; mp. 185.1-186.7 °C; ¹H NMR (CDCl₃, 500 MHz) δ 1.17 (t, 3H), 3.22 (s, 3H), 4.14-4.26 (m, 3H), 5.92 (d, *J* = 6 Hz, 1H), 7.17-7.36 (m, 5H), 10.33 (s, 1H, NH); ¹³C NMR (CDCl₃, 125 MHz) δ 14.1, 28.5, 53.6, 62.1, 86.9, 90.0, 125.6 (2C), 128.4 (2C), 129.1, 137.6, 151.3, 159.4, 162.3, 172.3; IR (KBr, cm⁻¹) v_{max}:1191.4, 1680.9, 1743,0, 2925.1; MS (ESI): 317.5 [M+H]⁺; Anal. Cald. For C₁₆H₁₆N₂O₅: C, 60.75; H, 5.10; N, 8.86%. Found: C, 60.77; H, 5.11; N, 8.87%.

1,3-Dimethyl-2,4-dioxo-5-phenyl-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-6-carbonitrile* (4*n*): Yield: 105 mg (37%); Dark yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.44; mp. 96.4-96.8 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.20 (s, 3H), 3.48 (s, 3H), 4.21 (d, *J* = 9 Hz, 1H), 5.30 (d, *J* = 9 Hz, 1H), 7.12-7.51 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 29.0, 29.2, 45.1, 80.1, 88.5, 114.0, 123.4, 128.1, 130.6 (2C), 131.8 (2C), 150.5, 162.0, 163.8; IR (KBr, cm⁻¹) v_{max}: 1194.4, 1681.2, 2251.5, 2925.4; MS (ESI): 284.5 [M+H]⁺; Anal. Cald. For C₁₅H₁₃N₃O₃: C, 63.60; H, 4.63; N, 14.83%. Found: C, 63.79; H, 4.64; N, 14.85%.

1,3-Dimethyl-2,4-dioxo-6-phenyl-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-5-carbonitrile* (5*n*): Yield: 96 mg (34%); Yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.20; mp. 94.2-95.1 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.19 (s, 3H), 3.27 (s, 3H), 4.82 (d, *J* = 6 Hz, 1H), 4.94 (d, *J* = 6 Hz, 1H), 7.10-7.39 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 29.1, 29.3, 39.0, 80.1, 87.5, 115.1, 123.3, 128.1, 130.5 (2C), 131.8 (2C), 150.6, 162.1, 164.7; IR (KBr, cm⁻¹) v_{max}: 1194.3, 1681.5, 2251.4, 2925.3; MS (ESI): 284.1 [M+H]⁺; Anal. Cald.

For $C_{15}H_{13}N_3O_3$: C, 63.60; H, 4.63; N, 14.83%. Found: C, 63.58; H, 4.62; N, 14.82%.

5-(p-Chlorophenyl)-1,3-dimethyl-2,4-dioxo-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-6-carbonitrile* (40): Yield: 120.2 mg (38%); Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.42; mp. 100.7-101.2 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.28 (s, 3H), 3.35 (s, 3H), 4.29 (d, *J* = 9.5 Hz, 1H), 5.60 (d, *J* = 9.5 Hz, 1H), 7.15-7.55 (m, 5H); ¹³C NMR (CDCl₃, 125 MHz) δ 28.1, 29.5, 45.1, 80.1, 87.0, 119.5, 128.0 (2C), 129.8 (2C), 134.0, 139.5, 151.0, 160.1, 161.7; IR (KBr, cm⁻¹) v_{max}: 1681.8, 2251.3, 2925.7, 1194.3; MS (ESI): 318.2 [M+H]⁺; Anal. Cald. For C₁₅H₁₂ClN₃O₃: C, 56.70; H, 3.81; N, 13.23%. Found: C, 56.73; H, 3.82; N, 13.25%.

6-(p-Chlorophenyl)-1,3-dimethyl-2,4-dioxo-1,2,3,4,5,6-

hexahydrofuro[2,3-*d*]*pyrimidine-5-carbonitrile* (5*o*): Yield: 111 mg (35%);Light yellow solid; Rf (Pet. Ether. 60-80/EtOAc, 8:2) 0.18; mp. 98.2-99.5 °C; ¹H NMR (CDCl₃, 500 MHz) δ 3.20 (s, 3H), 3.24 (s, 3H), 4.79 (d, *J* = 6 Hz, 1H)), 4.93 (d, *J* = 6 Hz, 1H), 7.09-7.39 (m, 4H); ¹³C NMR (CDCl₃, 125 MHz) δ 29.5, 29.7, 38.1, 82.4, 88.8, 116.6, 126.8 (2C), 129.7 (2C), 134.1, 136.5, 150.9, 158.3, 161.8; IR (KBr, cm⁻¹) v_{max} : 2253.1, 1680.2, 1194.6, 2925.3; MS (ESI): 318.5 [M+H]⁺ Anal. Cald. For C₁₅H₁₂ClN₃O₃: C, 56.70; H, 3.81; N, 13.23%. Found: C, 56.84; H, 3.79; N, 13.27%.

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5. References and notes

- 1. Dave, C. G.; Shah, R. D. Molecules 2002, 7, 554.
- Bhuiyan, M. M. H.; Rahman, K. M. M.; Hossain, M. K.; Rahim, M. A.; Hossain, M. I. Croat. *Chem Acta*. 2005, 78, 633.
- Gangjee, A.; Zeng, Y.; Ihnat, M.; Warnke, L. A.; Green, D. W.; Kisliuk, R. L.; Lin, F. T. *Bioorg. Med. Chem.* 2005, 13, 5475.
- Janeba, Z.; Balzarini, J.; Andrei, G.; Snoeck, R.; De Clercq, E.; Robins, M. J. J. Med. Chem. 2005, 48, 4690.
- (a) Gangjee, A.; Zeng, Y.; McGuire, J. J.; Kisliuk, R. L. J. Med. Chem. 2005, 48, 5329. (b) Gangjee, A.; Zeng, Y.; McGuire, J. J.; Mehraein, F.; Kisliuk, R. L. J. Med. Chem. 2004, 47, 6893.
- Robins, M. J.; Miranda, K.; Rajwanshi, V. K.; Peterson, M. A.; Andrei, G.; Snoeck, R.; De Clercq, E.; Balzarini, J. J. *J. Med. Chem.* 2006, 49, 391.
- 7. Hunger, A.; Hoffmann, K. Helv. Acta 1957, 40, 1319.
- 8. Tomioka, Y.; Ohkubo, Y. K.; Yamazaki, M. *Chem. Pharm. Bull.* **1985**, *33*, 1360.
- (a) Miyazaki, Y.; Matsunaga, S.; Tang, J.; Maeda, Y.; Nakano, M.; Philippe, R. J.; Shibahaa, M.; Liu, W.; Sato, H.; Wang, L.; Nolte, R. T. *Bioorg. Med. Lett.* **2005**, *15*, 2203. (b) Maeda, Y.; Nakano, M.; Sato, H.; Miyazaki, Y.; Schweiker, S. L.; Smith, J. L.; Truesdale, A. T. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3907. (c) *Prousek, J.; Jurãšek, A.; Kovãč, J. Coll. Czech. Chem.Commun.* **1980**, *45*, 1581. (d) Dang, Q.; Liu, Y. *Tetrahedron Lett.* **2009**, *50*, 6758.
- (a) Ramasamy, K.; Robins, R. K.; Revankar, G. R. J. Chem. Soc., Chem. Commun. 1989, 560. (b) Strekowski, L.; Wydra, R. L.; Janda, L.; Harden, D. B. J. Org. Chem. 1991, 56, 5610. (c) Gangjee, A.; Zeng, Y.; McGuire, J. J.; Kisliuk, R. L. J. Med. Chem. 2000, 43, 3125. (d) Floppe, N.; Fischer, L. M.; Howes, R.; Kierstan, P.; Potter, A.; Robertson, A. G. S.; Urgenor, A. E. J. Med. Chem. 2005, 48, 4332. (e) Gangjee, A.; Yang, J.;

McGuire, J. J.; Kisliuk, R. L. *Bioorg. Med.Chem.* **2006**, *14*, 8590. (f) Gangjee, A.; Li, W.; Lin, L.; Zeng, Y.; Ihnat, M.; Warnke, L. A.; Green, D. W.; Cody, V.; Pace, J.; Queener, S. F. *Bioorg. Med. Chem.* **2009**, *17*, 7324. (g) Cikotiene, I.; Buksnaitiene, R.; Rudys, S.; Morkunas, M.; Motiejaitis, D. *Tetrahedron* **2010**, *66*, 251. (h) Gangjee, A.; Jain, H. D.; Phan, J.; Guo, X..; Queener, S. F.; Kisliuk, R. L. *Bioorg. Med. Chem.* **2010**, *18*, 953.

- 11. Abdel-Hamid, M. A.; El-Kazak, A. M.; Seada1, M. H.; Farouk, O. M. J. Adv. Chem. **2013**, *13*, 229.
- (a) Vedejs, E.; Piotrowski, D. W.; Tucci, F. C. J. Org. Chem.
 2000, 65, 5498 (b) Vanecko, J. A.; West, F. G. Org. Lett.,
 2002, 4, 2813. (c) Villar, I. S.; Gradillas, A.; Domínguez, G.; Pérez-Castells, J. Tetrahedron Lett. 2010, 51, 3095. (d) Jiang,
 K.; Chen. Y. Tetrahedron Lett., 2014, 55, 2049. (c) Shi, Z.; Tan, B.; Leong, W. W. Y.; Zeng, X.; Lu, M.; Zhong, G. Org. Lett., 2010, 12, 5402.
- (a) Jacob, J.; Hende, E. V.; Claessens, S.; Kimpe, N. D. Curr. Org. Chem. 2011, 15, 1340. (b) Bull, J. A.; Mousseau, J. J.; Pelletier, G.; Charette, A. B. Chem. Rev. 2012, 112, 2642. (c) Osyanin, V. A.; Osipov, D. V.; Klimochkin, Y. N. J. Org. Chem. 2013, 78, 5505. (d) Wang, Q.; Hou, H.; Hui, L.; Yan, C. J. Org. Chem. 2009, 74, 7403. (e) Day, J.; Uroos, M.; Castledine, R. A.; Lewis, W.; McKeever-Abbasb, B.; Dowden, J. Org. Biomol. Chem. 2013, 11, 6502.
- (a) Sharma, M.; Bhuyan, P. J. *RSC Adv.*, **2014**, *4*, 15709. (b) Sharma, M.; Bhuyan, P.J. *RSC Adv.*, **2014**, *4*, 60640.
- 15. Wimalasena, K.; Haines, D. C. J. Org. Chem. 1994, 59, 6475.
- Chem 605 Structure Determination Using Spectroscopic Methods: 5.05 - Vicinal Proton-Proton Coupling ³J_{HH.}; Reich, H. J, © Copyright Hans J. Reich 2010, University of Wisconsin.
- 17. Appel, R.; Hartmann, N.; Mayr, H. J. Am. Chem. Soc. 2010, 132, 17894.
- (a) Mross, G.; Holtz, E.; Langer, P. J. Org. Chem. 2006, 71, 8049; (b) Ranu, B. C.; Adak, L.; Banerjee, S. Tetrahedron Lett. 2008, 49, 4617.
- Bayazit, M. K.; Coleman, K. S. J. Am. Chem. Soc. 2009, 131, 10670.
- Ding, X.; Taniguchi, K.; Hamamoto, Y.; Sada, K.; Fujinami, S.; Ukaji, Y.; Inomata, K. Bull. Chem. Soc. Jpn. 2006, 79, 1069.
- 21. Joshi, K. C.; Jain, R.; Nishith, S. Heterocycles, 1990, 31, 31.

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