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A copper-catalyzed reaction between terminal alkynes, oxiranes, and malonitrile has been described. In this transformation, copper acetylide was attacked on oxiranes to form homopropargyl alkoxy-copper intermediate that was further transferred to 2*H*-pyrane skeletons by reaction with malonitrile. We found that the reaction was not productive without hexafluoroisopropanol.

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INTRODUCTION

Terminal alkynes exhibited good reactivity both as nucleophiles and as electrophiles using appropriate catalytic systems [1–3]. Moreover, conversion of terminal alkynes to the corresponding acetylides and subsequent reactions with C-electrophiles is an important carboncarbon bond-forming reaction. Accordingly, there are many reports in literature featuring the additions of metalalkynylides to Michael acceptors [4-9]. In most of these transformations, metal acetylides formation occurred using equimolar amounts of pre-catalyst sources. During the last decade, in situ formations of metal acetylides offered an opportunity for the catalytic reactions of metal acetylides in C-C bond formations [10-12]. Additionally, there are many reports in literature featuring synthesis of heterocycle compounds catalysis in coinage metals [13]. Due to the high activity of Cu, metal acetylides synthesis has been extensively studied on Cu-based catalysts [14-17]. Since the pioneering work of Knöpfel and Carreira in catalytic conjugate addition reaction of terminal alkynes [18], the additions of metal acetylides on carbonyl, imine, and α -imino ester moieties for the construction of C-C bonds have been well explored [19-21]. Importantly, alkynylation of oxiranes with ring opening with metal acetylides is among the most versatile methods for the C-C bond formation [22–24]. As far as we could ascertain, the regioselectivity in the ring opening reactions of three-membered heterocycles depends on electronic and steric parameters on the substrates structures [25]. When the activated three-membered heterocycle is stable enough, metal

acetylides reacted with the intermediate to give the ringopened product. Whereas when the activated threemembered heterocycle is relatively unstable, Meinwald rearrangement occurred instantaneously to give the corresponding product, which reacted with the metal acetylides to give the product. While the importance of such reports cannot be overstated, the transformation suffered from the limitations such as low yields and regioselectivity. Recently, regioselective ring opening of oxiranes with lithium alkynyl ate complex has been reported [26]. Based on these finding, we decided to examine the feasibility of the oxiranes alkynylation strategy for the synthesis of 2H-pyrane motif.

RESULTS AND DISCUSSION

To examine the efficiency of the proposed reaction, phenyl acetylene (1a), 2-methyloxirane (2a), and malonitrile (3) were selected as reaction partners using AgOAc and $(i\text{-Pr})_2\text{EtN}$ in hexafluoroisopropanol (HFIP) at 85°C. Our optimization results are shown in Table 1. The targeted product 4a was achieved only in 25% yield together with the direct coupling by-product 5 in 42% yield (Table 1, entry 1). This compound must be generated by the direct attack of the silver acetylide to 2a. It could be deduced that the reaction did not proceed through the S_N1-like pathway as no products arising from the Meinwald rearrangement are detected in crude reaction mixture analysis. The examination of other silver precatalysts showed that silver salts were inefficient in these transformations (Table 1, entries 2–7). Pd(OAc)₂ resulted

 Table 1

 Optimization of the reaction conditions^a.

Ph===	+	+ NC NC	<u>Conditions</u>	NC Ph H ₂ N O	l/or P
1a	2a	3		4a	5

Entry	Catalyst	Base	Solvent	Yield (%)
1	AgOAc	(<i>i</i> -Pr) ₂ EtN	HFIP	25 (42) ^b
2	AgI	(i-Pr) ₂ EtN	HFIP	19 (53)
3	AgCl	(<i>i</i> -Pr) ₂ EtN	HFIP	31 (29)
4	AgNO ₃	(<i>i</i> -Pr) ₂ EtN	HFIP	38 (25)
5	AgF	(<i>i</i> -Pr) ₂ EtN	HFIP	50 (34)
6	$AgBF_4$	(<i>i</i> -Pr) ₂ EtN	HFIP	13 (22)
7	Ag_2CO_3	(<i>i</i> -Pr) ₂ EtN	HFIP	24 (16)
8	$Pd(OAc)_2$	(<i>i</i> -Pr) ₂ EtN	HFIP	46 (25)
9	AuCl ₃	(<i>i</i> -Pr) ₂ EtN	HFIP	70
10	(IPr)CuCl	(<i>i</i> -Pr) ₂ EtN	HFIP	68
11	CuOTf	(i-Pr) ₂ EtN	HFIP	80
12	Cu ₂ O	(<i>i</i> -Pr) ₂ EtN	HFIP	85
13	CuBr	(<i>i</i> -Pr) ₂ EtN	HFIP	65
14	CuCl	(<i>i</i> -Pr) ₂ EtN	HFIP	73
15	CuI	(<i>i</i> -Pr) ₂ EtN	HFIP	80
16	Cu(OTf) ₂	(<i>i</i> -Pr) ₂ EtN	HFIP	55
17	Cu ₂ O	Et ₃ N	HFIP	68
18	Cu ₂ O	K_2CO_3	HFIP	16
19	Cu ₂ O	Cs ₂ CO ₃	HFIP	38
20	Cu ₂ O	LiO ^t Bu	HFIP	13
21	Cu ₂ O	(<i>i</i> -Pr) ₂ EtN	MeCN	21
22	Cu ₂ O	(<i>i</i> -Pr) ₂ EtN	DMF	26
23	Cu ₂ O	(<i>i</i> -Pr) ₂ EtN	PEG 200	40
24	Cu ₂ O	(<i>i</i> -Pr) ₂ EtN	PEG 400	70
25	Cu ₂ O	—	HFIP	31

DMF, dimethylformamide; HFIP, hexafluoroisopropanol; PEG, polyethylene glycol.

^aReaction conditions: **1a** (1.2 mmol), **2** (1.0 mL), **3a** (1.0 mmol), catalyst (0.1 mmol), base (1.2 mmol), and Linde-type 3-Å molecular sieves (250 mg) in solvent (3 mL) at 85°C for 14 h.

^bThe digit in parentheses refer to the yield of 5.

in a moderate yield of the targeted product, which might arise from the protonolysis of palladium acetylide, while the reaction conducted with AuCl₃ proceeded with good successes (Table 1, entries 8 and 9). It was delightedly found that 10 mol % loading of N-heterocyclic carbene copper(I) catalyst (IPr)CuCl (IPr = 1.3-bis(2.6diisopropylphenyl)imidazol-2-ylidene) was turned out to effectively suppress the direct coupling by-product formation (Table 1, entry 10). The yield was highly improved upon using of CuOTf as the catalyst source (Table 1, entry 11). A copper pre-catalyst screen showed that Cu₂O gave superior yield in comparison with other copper catalysts (Table 1, entries 11-16). The superior catalytic activity of Cu2O is likely due to its good solubility in HFIP. Importantly, no direct coupling byproduct 5 was formed when the reaction was conducted with copper salts. The reactions conducted with Et₃N or inorganic bases formed lower yields than (i-Pr)2EtN (Table 1, entries 17-20). Replacing HFIP with MeCN or dimethylformamide resulted in inferior yields of **4a** (Table 1, entries 21 and 22). Polyethylene glycol 200 and 400 (known solvents for activating oxiranes and aziridines) resulted lower productivity (Table 1, entries 23 and 24). The control experiment showed that extremely low yield of **4a** was obtained in the absence of $(i-Pr)_2$ EtN, testifying the vital roles of base in this catalytic system (Table 1, entry 25).

Afterwards, we examined the scope of the transformation with different terminal alkynes and oxiranes (Table 2). Methyl oxirane **3a** afforded the corresponding 2*H*-pyrane motif **4a** with good success (Table 2, entry 1). It was delightedly found that oxiranes with longer chains gave the products in diminished yields, which might arise from the steric hindrance of alkyl branches (Table 2, entries 2 and 3). The reaction proceeded smoothly with phenoxy methyl-substituted oxirane (Table 2, entry 4). Oxirane possessing an ester motif was also tolerated (Table 2, entries 5 and 6). The transformation is sensitive

 Table 2

 Scope of terminal alkynes and oxiranes^a.



Entry	Alkyne	R^1	Epoxide	R^2 , R^3 , R^4	4, Yield (%)
1	1a	Ph	2a	CH ₃ , H, H	4a , 85
2	1a	Ph	2b	<i>n</i> -Pr, H, H	4b , 80
3	1a	Ph	2c	<i>n</i> -Pr, H, CH ₃	4c , 71
4	1a	Ph	2d	PhOCH ₂ , H, H	4d , 84
5	1a	Ph	2e	CH ₃ , H, CH ₃ OCO	4e , 69
6	1a	Ph	2f	CH ₂ CMeCO ₂ CH ₂ , H, H	4f, 88
7	1a	Ph	2g	(CH ₃) ₃ , H, H	4g , 60
8	1a	Ph	2h	$-(CH_2)_4-, H$	4h , 95
9	1a	Ph	2i	$-(CH_2)_5-, H$	4i , 47
10	1a	Ph	2j	Me ₂ CHOCH ₂ , H, H	4 j, 78
11	1a	Ph	2k	H, Ph, H	4k , 90 ^b
12	1a	Ph	21	Ph, Ph, H (trans 2S, 3S)	41 , 94
13	1a	Ph	2m	4-NO ₂ -C ₆ H ₄ , H, H	4m , 58
14	1b	CH ₃ OCH ₂	2d	H, Ph, H	4n , 63 ^b
15	1c	TMS	2d	H, Ph, H	40 , 60 ^b

^aFor all entries except stated otherwise: 1 (1.2 mmol), 2 (1.0 mmol), 3 (1.0 mmol), Cu_2O (0.1 mmol), (*i*-Pr)₂EtN (1.2 mmol), and 250 mg of Linde-type 3-Å molecular sieves, in hexafluoroisopropanol (HFIP) (3 mL) at 85°C for 14 h.

^bThe yield of internal-attacked product.

to steric hindrance of substrate as 2-(*tert*-butyl)oxirane (2g) yielded the targeted product in 60% yield (Table 2, entry 7). The meso oxirane derived from cyclohexene furnished the *trans*-configuration product **4h** in excellent yield (Table 2, entry 8). Cycloheptene oxide (2i) was less reactive in this developed transformation (Table 2, entry 9). This coupling reaction was also extended to phenyl-substituted oxiranes. As expected, 2-phenyloxirane (2k) was attacked by silver acetylide at the internal carbon (Table 2, entry 11). 2,3-Diphenyloxirane (21) also afforded a good yield (Table 2, entry 12). Oxirane bearing 4-nitrophenyl 2m underwent the transformation in different manners and give the terminal-attacked product exclusively (Table 2, entry 13). An obvious decrease in reaction outcome was observed when aliphatic terminal alkvne (1b) was used as the substrate, likely due to the lower acidity of aliphatic alkyne than that of phenyl acetylene (Table 2, entry 14). Trimethylsilylacetylene provided the 2H-pyrane skeleton in 60% yield, which could participate in organosilicon chemistry (Table 2, entry 15).

Mechanistically, copper(I) coordinates with the phenyl acetylene activating the alkyne carbon–hydrogen bond. The abstraction of the terminal proton easily by $(i-Pr)_2$ EtN leads to the formation of copper acetylide nucleophile **6**. Copper acetylide was attacked on oxiranes to form homopropargyl alkoxy-copper intermediates **7**, which were further attacked by dicyanocarbanion **8**, generated from malinitrile and DIEPA, followed by cyclization and

tautomerization, affording 2H-pyrane derivatives **4**. The investigation indicated that HFIP is necessary to furnish the reaction in acceptable yields (Scheme 1).

In this report, we attempt to develop a novel catalytic one-pot reaction involving *in situ* generated copper acetylides, oxiranes, and malonitrile. The control experiments showed that the choice of catalyst and the reaction medium had great impact in furnishing the desired transformation. Copper salts exhibited unique reactivity and selectivity in this transformation. We found that the efficiency and regioselectivity of the ring opening reaction of oxiranes depend on electronic and steric properties of substrates. Alkyl terminal alkynes required longer reaction times to furnish the reaction with good success. Gratifyingly, the reaction was proved to be general under the optimized conditions, performing well in most of the cases examined.

EXPERIMENTAL

Substrates and metal salts were purchased from commercial sources. All the solvents were purchased from PALACHEM Chemical Industries. Solvents were dried over molecular sieves before the use. Molecular sieves were activated at 260° C for 6 h and stored in desiccator in the presence of P₂O₅. The copper(I) salts were typically weighted in a glovebox and were stored in



Scheme 1. Proposed reaction progress path for the synthesis 2H-pyranes.

a desiccator while being utilized. All reactions were carried out in Schlenk tube (25 mL) using oven-dried glassware under a dry Ar atmosphere. All reactions were monitored by TLC analyses that were performed on silica gel 60 (Merck, item number 116835). Column chromatography was performed on silica gel 60 (particle size 63–200 µm) (Merck, item number 7734-3) on a glass column. Melting points (mp) were recorded with Electrothermal-9100 apparatus and are uncorrected. ¹H NMR spectra were recorded with Bruker DRX-500 AVANCE instrument, at 500 MHz (using TMS, as a reference), and ¹³C NMR spectra were recorded at 125 MHz (using the CDCl₃ triplet as reference) in CDCl₃ as solvent at ambient temperature. Chemical shifts were reported as δ in parts per million (ppm), and coupling constant (J) values are given in Hertz. Mass data were recorded with EIMS (70 eV): Finnigan-MAT-8430 mass spectrometer, in m/z. IR spectra were recorded with Shimadzu IR-460 FTIR spectrometer as KBr pellets. Elemental analyses (C, H, N) were performed with a Heraeus CHN-O-Rapid analyzer. The results agreed favorably with the calculated values.

General procedure for the preparation of 4. A Schlenk tube (25 mL) was charged with Cu_2O (0.1 mmol, 14 mg), (*i*-Pr)₂EtN (1.2 mmol), alkyne (1.2 mmol), and HFIP (3.0 mL). The mixture was stirred for 30 min at ambient conditions and then warmed up to 85°C. Oxirane (1.0 mmol) was then injected to the reaction vessel. The reaction mixture was then stirred for an additional 30 min at 85°C, followed by addition of a solution of malonitrile (1.0 mmol, 66 mg) in HFIP (1.0 mL). After consuming the starting material (TLC monitoring, 14 h), the resulting slurry was quenched with saturated aqueous NH₄Cl (5 mL) and extracted with EtOAc (3 × 10 mL). The organic layers were dried over MgSO₄, filtered, and concentrated *in vacuo*. The residue was subjected to flash chromatography (silica gel, hexane:EtOAc) to give the pure targeted product **4**.

6-Amino-4-benzylidene-2-methyl-3,4-dihydro-2H-pyran-5-

carbonitrile (4a, $C_{14}H_{14}N_2O$). The crude product was by column chromatography purified $(SiO_2;$ hexane/EtOAc 3/1, R_f: 0.52) affording 0.19 g (85%) of **4a**. Colorless oil. IR (KBr): \overline{V} = 3441, 3427, 3074, 2934, 2236, 1615, 1440, 1120. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 1.23$ (3H, d, ${}^{3}J = 5.9$ Hz, Me), 2.31 (1H, dd, ${}^{2}J = 11.4$, ${}^{3}J = 8.8$ Hz, CH), 2.49 (1H, dd, ${}^{2}J = 11.4$, ${}^{3}J = 5.1$ Hz, CH), 4.01–4.14 (1H, m, CH), 6.44 (1H, t, ${}^{4}J = 2.0$ Hz, CH), 7.01 (2H, br s, NH₂), 7.29 (1H, t, ${}^{3}J = 6.5$ Hz, CH), 7.42 (2H, t, ${}^{3}J = 6.5$ Hz, 2CH), 7.65 (2H, d, ${}^{3}J = 6.6$ Hz, 2CH). ${}^{13}C$ NMR (125.7 MHz, CDCl₃): $\delta_{\rm C} = 19.8$ (Me), 46.8 (CH₂), 76.5 (C), 79.6 (CH), 114.2 (CN), 125.1 (CH), 126.8 (CH), 128.2 (2CH), 130.0 (2CH), 137.6 (C), 143.4 (C), 184.1 (C). EI-MS $(70 \text{ eV}): m/z \ (\%) = 226 \ (M^+, 2), 210 \ (15), 136 \ (34), 121$ (56), 77 (100). Anal. Calcd (%) for (226.28): C, 74.31; H, 6.24; N, 12.38. Found: C, 74.66; H, 6.45; N, 12.61.

6-Amino-4-benzylidene-2-propyl-3,4-dihydro-2H-pyran-5-The crude product was carbonitrile (4b, $C_{16}H_{18}N_2O$). purified by column chromatography $(SiO_2;$ hexane/EtOAc 4/1, R_f: 0.34) affording 0.20 g (80%) 4b. Colorless oil. IR (KBr): $\overline{V} = 3422, 3411, 3043, 2966,$ 2273, 1611, 1376, 1052. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 0.92$ (3H, t, ${}^{3}J = 6.2$ Hz, Me), 1.36–1.57 (4H, m, 2CH₂), 2.24 (1H, dd, ${}^{2}J = 12.0$, ${}^{3}J = 4.9$ Hz, CH), 2.47 (1H, dd, ${}^{2}J = 12.0$, ${}^{3}J = 10.1$ Hz, CH), 3.96–4.08 (1H, m, CH), 6.48 (1H, t, ${}^{4}J$ = 1.8 Hz, CH), 7.03 (2H, br s, NH₂), 7.28 (1H, t, ${}^{3}J$ = 6.6 Hz, CH), 7.41 (2H, t, ${}^{3}J$ = 6.6 Hz, 2CH), 7.63 (2H, d, ${}^{3}J = 6.6$ Hz, 2CH). ${}^{13}C$ NMR

(125.7 MHz, CDCl₃): 13.1 (Me), 23.4 (CH₂), 40.14 (CH₂), 43.8 (CH₂), 76.1 (CH), 79.5 (C), 114.1 (CN), 126.3 (CH), 128.1 (CH), 129.5 (2CH), 130.8 (2CH), 136.1 (C), 144.8 (C), 182.2 (C). EI-MS (70 eV): m/z (%) = 254 (M⁺, 7), 211 (12), 164 (40), 121 (67), 91 (38), 77 (100). *Anal.* Calcd (%) for (254.33): C, 75.56; H, 7.13; N, 11.01. Found: C, 75.88; H, 7.45; N, 11.14.

6-Amino-4-benzylidene-2-methyl-2-propyl-3,4-dihydro-2H-The crude product pyran-5-carbonitrile (4c, $C_{17}H_{20}N_2O$). was purified by column chromatography (SiO₂; hexane/EtOAc 4/1, R_f: 0.49) affording 0.19 g (71%) 4c. Yellow oil. IR (KBr): $\overline{V} = 3420, 3411, 2936, 2182, 1609,$ 1448, 1222. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 0.95$ $(3H, t, {}^{3}J = 6.6 \text{ Hz}, \text{ Me}), 1.30-1.48 (7H, m, 2CH₂, Me),$ 2.26 (1H, dd, ${}^{2}J = 11.7$, ${}^{3}J = 9.2$ Hz, CH), 2.45 (1H, dd, ${}^{2}J = 11.7$, ${}^{3}J = 5.1$ Hz, CH), 6.40 (1H, t, ${}^{4}J = 1.8$ Hz, CH), 6.98 (2H, br s, NH₂), 7.31 (1H, t, ${}^{3}J$ = 6.9 Hz, CH), 7.45 (2H, t, ${}^{3}J$ = 6.7 Hz, 2CH), 7.66 (2H, d, ${}^{3}J$ = 6.7 Hz, 2CH). ¹³C NMR (125.7 MHz, CDCl₃): 13.93 (Me), 19.7 (CH₂), 27.2 (CH₃), 45.7 (CH₂), 49.1 (CH₂), 78.2 (C), 89.7 (C), 114.3 (CN), 125.3 (CH), 128.2 (CH), 130.0 (2CH), 132.6 (2CH), 136.8 (C), 145.02 (C), 183.6 (C). EI-MS (70 eV): m/z (%) = 268 (M⁺, 4), 225 (11), 134 (38), 135 (68), 77 (100). Anal. Calcd (%) for (268.36): C, 76.09; H, 7.51; N, 10.44. Found: C, 76.36; H, 7.83; N, 10.54.

6-Amino-4-benzylidene-2-(phenoxymethyl)-3,4-dihydro-2Hpyran-5-carbonitrile (4d, $C_{20}H_{18}N_2O_2$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 4/1, $R_{\rm f}$: 0.31) affording 0.27 g (84%) 4d. Pale yellow oil. IR (KBr): $\overline{V} = 3384, 3362, 2949, 2172,$ 1617, 1466, 1046. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 2.22$ (1H, dd, ²J = 10.5, ³J = 8.9 Hz, CH), 2.45 (1H, dd, ${}^{2}J = 10.5$, ${}^{3}J = 5.0$ Hz, CH), 3.93–4.12 (2H, m, 2CH), 4.56–4.67 (1H, m, CH), 6.51 (1H, t, ${}^{4}J = 1.9$ Hz, CH), 6.85 (2H, br s, NH₂), 6.90-7.05 (3H, m, 3CH), 7.29 (2H, t, ${}^{3}J$ = 6.3 Hz, 2CH), 7.35 (1H, t, ${}^{3}J$ = 6.4 Hz, CH), 7.43 (2H, t, ${}^{3}J$ = 6.4 Hz, 2CH), 7.65 (2H, d, ${}^{3}J$ = 6.3 Hz, 2CH). ¹³C NMR (125.7 MHz, CDCl₃): 40.1 (CH₂), 76.0 (CH₂), 79.2 (C), 85.2 (CH), 114.1 (CN), 114.9 (2CH), 121.3 (CH), 125.2 (CH), 128.3 (CH), 129.2 (2CH), 130.2 (2CH), 132.73 (2CH), 137.2 (C), 144.1 (C), 159.2 (C), 184.2 (C). EI-MS (70 eV): m/z (%) = 318 (M⁺, 6), 211 (14), 217 (17), 121 (82), 107 (68), 77 (100), 54 (46). Anal. Calcd (%) for (318.38): C, 75.45; H, 5.70; N, 8.80. Found: C, 75.71; H, 75.98; N, 8.93.

Methyl-6-amino-4-benzylidene-5-cyano-2-methyl-3,4dihydro-2H-pyran-2-carboxylate (4e, $C_{16}H_{16}N_2O_3$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 6/1, $R_{\rm f}$: 0.30) affording 0.20 g (69%) **4e**. Yellow oil. IR (KBr): \overline{V} = 3432, 3421, 3006, 2968, 2244, 1731, 1611, 1415, 1016. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H}$ = 1.53 (3H, s, CH₃), 2.70 (1H, dd, ²J = 10.9 Hz, ⁴J = 1.9 Hz, CH), 2.96 (1H, dd, ²J = 10.9 Hz, ⁴J = 1.9 Hz, CH), 3.71 (3H, s, OCH₃), 6.70 (1H, t, ⁴J = 1.9 Hz, CH), 6.94 (2H, br s, NH₂), 7.35 (1H, t, ³J = 6.8 Hz, CH), 7.44 (2H, t, ³J = 6.8 Hz, 2CH), 7.67 (2H, d, ³J = 6.7 Hz, 2CH). ¹³C NMR (125.7 MHz, CDCl₃): 23.5 (CH₃), 44.1 (CH₂), 57.2 (OCH₃), 79.5 (C), 84.1 (C), 114.1 (CN), 126.1 (CH), 127.1 (CH), 128.8 (2CH), 130.4 (2CH), 136.5 (C), 143.2 (C), 174.5 (C), 183.1 (C). EI-MS (70 eV): m/z (%) = 284 (M⁺, 4), 269 (14), 210 (33), 121 (76), 91 (44), 77 (100) 54 (38). *Anal*. Calcd (%) for (284.32): C, 67.59; H, 5.67; N, 9.85. Found: C, 67.93; H, 5.96; N, 9.94.

6-Amino-4-benzylidene-5-cyano-3,4-dihydro-2H-pyran-2-yl*methyl methacrylate (4f, C_{18}H_{18}N_2O_3).* The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 4/1, R_f: 0.30) affording 0.25 g (81%) 4f. Yellow oil. IR (KBr): \overline{V} = 3408, 3385, 2934, 2181, 1713, 1415, 1167. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 2.17$ $(3H, t, {}^{3}J = 1.7 \text{ Hz}, \text{CH}_{3}), 2.39 (1H, d, {}^{2}J = 11.1 \text{ Hz},$ ${}^{3}J = 9.5$ Hz, CH), 2.62 (1H, d, ${}^{2}J = 11.1$ Hz, ${}^{3}J = 4.9$ Hz, CH), 4.35-4.46 (1H, m, CH), 4.59-4.76 (2H, m, 2CH), 6.40 (1H, t, ${}^{4}J$ = 1.7 Hz, CH), 6.52 (1H, d, ${}^{2}J$ = 4.5 Hz, CH), 6.68 (1H, d, ${}^{2}J = 4.5$ Hz, CH), 6.89 (2H, br s, NH₂), 7.38 (1H, t, ${}^{3}J$ = 6.7 Hz, CH), 7.42 (2H, t, ${}^{2}J = 6.6$ Hz, 2CH), 7.68 (2H, d, ${}^{2}J = 6.7$ Hz, 2CH). ${}^{13}C$ NMR (125.7 MHz, CDCl₃): 18.6 (CH₃), 38.3 (CH₂), 73.3 (CH₂), 79.2 (C), 86.1 (CH), 114.3 (CN), 125.9 (CH), 126.9 (CH₂), 126.5 (CH), 129.2 (2CH), 131.5 (2CH), 136.2 (C), 138.0 (C), 143.1 (C), 169.2 (C), 182.8 (C). EI-MS (70 eV): m/z (%) = 310 (M⁺, 8), 269 (14), 212 (34), 121 (69), 91 (45), 77 (100). Anal. Calcd (%) for (310.35): C, 69.66; H, 5.85; N, 9.03. Found: C, 69.87; H. 6.07; N. 9.24.

6-Amino-4-benzylidene-2-(tert-butyl)-3,4-dihydro-2H-pyran-The crude product was 5-carbonitrile (4g, $C_{17}H_{20}N_2O$). purified by column chromatography (SiO₂; hexane/EtOAc 4/1, R_f: 0.49) affording 0.16 g (60%) 4g. Yellow oil. IR (KBr): $\overline{V} = 3415, 3398, 2975, 2186, 1610, 1427, 1118.$ ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 0.92$ (9H, s, 3CH₃), 2.42-2.60 (2H, m, 2CH), 3.78-3.89 (1H, m, CH), 6.56 $(1H, t, {}^{4}J = 1.7 \text{ Hz}, \text{CH}), 6.89 (2H, \text{ br s}, \text{NH}_{2}), 7.35 (1H, t, t)$ ${}^{3}J = 6.7$ Hz, CH), 7.46 (2H, t, ${}^{3}J = 6.8$ Hz, 2CH), 7.62 $(2H, d, {}^{3}J = 6.7 \text{ Hz}, 2CH).$ ${}^{13}C$ NMR (125.7 MHz,CDCl3): 27.3 (3CH₃), 36.9 (CH₂), 38.1 (C), 78.1 (C), 84.9 (CH), 114.5 (CN), 126.7 (CH), 127.6 (CH), 129.9 (2CH), 131.2 (2CH), 136.1 (C), 145.2 (C), 181.9 (C). EI-MS $(70 \text{ eV}): m/z \ (\%) = 268 \ (M^+, 4), \ 211 \ (17), \ 196 \ (34), \ 121$ (67), 77 (100), 57 (89). Anal. Calcd (%) for (268.36): C, 76.09; H, 7.51; N, 10.44. Found: C, 76.31; H, 67.83; N, 10.62.

2-Amino-4-benzylidene-4a,5,6,7,8,8a-hexahydro-4H-

chromene-3-carbonitrile (4h, $C_{17}H_{18}N_2O$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 3/1, $R_{\rm f}$: 0.44) affording 0.25 g (95%) **4h**. Yellow oil. IR (KBr): \overline{V} = 3433, 3346, 2896, 2181, 1622,

1460, 1115. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 1.31$ – 1.97 (8H, m, 4CH₂), 2.62–2.74 (1H, m, CH), 3.48–3.60 (1H, m, CH), 6.72 (1H, d, ⁴*J* = 1.9 Hz, CH), 7.04 (2H, br s, NH₂), 7.33 (1H, t, ³*J* = 6.8 Hz, CH), 7.42 (2H, t, ³*J* = 6.8 Hz, 2CH), 7.65 (2H, d, ³*J* = 6.8 Hz, 2CH). ¹³C NMR (125.7 MHz, CDCl₃): 26.7 (CH₂), 28.1 (CH₂), 32.6 (CH₂), 35.4 (CH₂), 47.6 (CH), 79.5 (C), 83.4 (CH), 114.1 (CN), 125.9 (CH), 126.4 (CH), 128.9 (2CH), 131.2 (2CH), 136.1 (C), 144.3 (C), 182.7 (C). EI-MS (70 eV): *m*/*z* (%) = 266 (M⁺, 7), 210 (14), 184 (35), 96 (80), 91 (44), 77 (100). *Anal.* Calcd (%) for (266.34): C, 76.66; H, 6.81; N, 10.52. Found: C, 76.95; H, 7.04; N, 10.73.

2-Amino-4-benzylidene-4,4a,5,6,7,8,9,9a-

octahvdrocvcloheptalblpvran-3-carbonitrile (4i. $C_{18}H_{20}N_2O$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 4/1, R_f: 0.21) affording 0.13 g (47%) 4i. Pale yellow solid, mp: 57-59°C. IR (KBr): \overline{V} = 3419, 3403, 2930, 2216, 1611, 1445, 1029. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 1.31 -$ 1.97 (10H, m, 5CH₂), 2.45-2.56 (1H, m, CH), 3.37-3.48 (1H, m, CH), 6.68 (1H, d, ${}^{4}J$ = 1.8 Hz, CH), 7.01 (2H, br s, NH₂), 7.32 (1H, t, ${}^{3}J$ = 6.8 Hz, CH), 7.40 (2H, t, ${}^{3}J = 6.8$ Hz, 2CH), 7.51 (2H, d, ${}^{3}J = 6.8$ Hz, 2CH). ${}^{13}C$ NMR (125.7 MHz, CDCl₃): 26.7 (CH₂), 27.5 (CH₂), 29.2 (CH₂), 31.4 (CH₂), 35.2 (CH₂), 37.1 (CH), 71.1 (CH), 79.4 (C), 114.2 (CN), 125.2 (CH), 127.1 (CH), 129.4 (2CH), 130.8 (2CH), 136.5 (C), 143.7 (C), 181.6 (C). EI-MS (70 eV): m/z (%) = 280 (M⁺, 12), 210 (15), 195 (29), 170 (44), 81 (76), 91 (43), 77 (100). Anal. Calcd (%) for (280.37): C, 77.11; H, 7.19; N, 9.99. Found: C. 77.32; H. 7.38; N. 10.16.

6-Amino-4-benzylidene-2-(isopropoxymethyl)-3,4-dihydro-2H-pyran-5-carbonitrile (4j, $C_{17}H_{20}N_2O_2$). The crude product was purified by column chromatography (SiOa:

product was purified by column chromatography (SiO₂; hexane/EtOAc 4/1, R_f: 0.53) affording 0.22 g (78%) 4j. Colorless solid, mp: 82–84°C. IR (KBr): \overline{V} = 3362, 3339, 3001, 2206, 1603, 1428, 1123. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 1.23$ (6H, d, ${}^{3}J = 6.1$ Hz, 2CH₃), 2.37 (1H, dd, ${}^{2}J = 11.2$ Hz, ${}^{3}J = 9.5$ Hz, CH), 2.62 (1H, dd, ${}^{2}J = 11.2$ Hz, ${}^{3}J = 5.1$ Hz, CH), 4.11–4.66 (4H, m, 4CH), 6.47 (1H, ${}^{4}J$ = 1.8 Hz, CH), 6.89 (2H, br s, NH₂), 7.35 (1H, t, ${}^{3}J$ = 6.5 Hz, CH), 7.44 (2H, t, ${}^{3}J$ = 6.6 Hz, 2CH), 7.64 (2H, d, ${}^{3}J$ = 6.6 Hz, 2CH). ${}^{13}C$ NMR (125.7 MHz, CDCl₃): 24.7 (2CH₃), 39.1 (CH₂), 69.1 (CH₂), 74.6 (CH), 79.1 (C), 86.1 (CH), 115.1 (CN), 126.1 (CH), 127.1 (CH), 129.3 (2CH), 130.8 (2CH), 137.1 (C), 143.5 (C), 181.9 (C). EI-MS (70 eV): m/z $(\%) = 284 \ (M^+, 5), \ 211 \ (15), \ 196 \ (41), \ 121 \ (78), \ 77$ (100), 73 (64). Anal. Calcd (%) for (284.36): C, 71.81; H, 7.09; N, 9.85. Found: C, 72.06; H, 7.38; N, 9.98.

6-Amino-4-benzylidene-3-phenyl-3,4-dihydro-2H-pyran-5carbonitrile (4k, C_{19}H_{16}N_2O). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 3/1, $R_{\rm f}$: 0.27) affording 0.26 g (90%) **4k**. Pale yellow solid, mp: 76–78°C. IR (KBr): $\overline{V} = 3412, 3400, 2948, 2189, 1613, 1382, 1022. {}^{1}H NMR (500.1 MHz, CDCl_3): <math>\delta_{\rm H} = 3.81 (1H, dd, {}^{3}J = 10.0 \text{ Hz}, {}^{3}J = 7.1 \text{ Hz}, CH), 4.40 (1H, dd, {}^{2}J = 12.0 \text{ Hz}, {}^{3}J = 10.0 \text{ Hz}, CH), 4.69 (1H, dd, {}^{2}J = 12.0 \text{ Hz}, {}^{3}J = 7.1 \text{ Hz}, CH), 4.69 (1H, dd, {}^{2}J = 12.0 \text{ Hz}, {}^{3}J = 7.1 \text{ Hz}, CH), 6.59 (1H, t, {}^{3}J = 1.8 \text{ Hz}, CH), 6.89 (2H, br s, NH_2), 7.32–7.50 (8H, m, 8CH), 7.65 (2H, d, {}^{3}J = 6.5 \text{ Hz}, 2CH). {}^{13}C NMR (125.7 \text{ MHz}, CDCl_3): 47.1 (CH), 71.3 (CH_2), 78.2 (C), 114.3 (CN), 125.1 (CH), 126.1 (CH), 127.4 (CH), 127.9 (2CH), 129.8 (2CH), 130.7 (2CH), 132.5 (2CH), 135.1 (C), 140.1 (C), 142.7 (C), 181.9 (C). EI-MS (70 eV):$ *m/z*(%) = 288 (M⁺, 3), 211 (24), 196 (36), 171 (50), 121 (74), 77 (100), 54 (36). Anal. Calcd (%) for (288.35): C, 79.14; H, 5.59; N, 9.72. Found: C, 79.42; H, 5.85; N, 9.94.

6-Amino-4-benzylidene-2,3-diphenyl-3,4-dihydro-2H-pyran-The crude product was 5-carbonitrile (4l, $C_{25}H_{20}N_2O$). purified bv column chromatography (SiO_2) : hexane/EtOAc 2/1, $R_{\rm f}$: 0.44) affording 0.34 g (94%) 4I. Yellow solid, mp: 127–129°C. IR (KBr): \overline{V} = 3376, 3348, 2962, 2235, 1643, 1432, 1176, 1007. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 4.49$ (1H, d, ${}^{3}J = 9.1$ Hz, CH), 5.23 (1H, d, ${}^{3}J = 9.1$ Hz, CH), 6.55 (1H, d, ${}^{3}J = 1.8$ Hz, CH), 6.88 (2H, br s, NH₂), 7.21–7.49 (13H, m, 13CH), 7.63 (2H, d, ${}^{3}J = 6.8$ Hz, 2CH). ${}^{13}C$ NMR (125.7 MHz, CDCl₃): 49.1 (CH), 78.1 (C), 83.9 (CH), 114.0 (CN), 126.1 (CH), 126.1 (CH), 127.0 (CH), 127.8 (2CH), 128.2 (2CH), 128.7 (CH), 129.0 (2CH), 129.4 (2CH), 130.1 (2CH), 132.6 (2CH), 136.1 (C), 139.1 (C), 142.5 (C), 144.1 (C), 183.2 (C). EI-MS (70 eV): m/z $(\%) = 364 (M^+, 5), 287 (14), 210 (34), 185 (41), 170$ (43), 121 (83), 77 (100). Anal. Calcd (%) for (364.45): C, 82.39; H, 5.53; N, 7.69. Found: C, 82.64; H, 5.86; N, 7.81.

6-Amino-4-benzylidene-2-(4-nitrophenyl)-3,4-dihydro-2Hpyran-5-carbonitrile (4m, $C_{19}H_{15}N_3O_3$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 1/1, R_f: 0.58) affording 0.19 g (58%) 4m. Yellow solid, mp: 153–155°C. IR (KBr): $\overline{V} = 3447$, 3411, 2963, 2191, 1605, 1594, 1368, 1221, 1080. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 2.650-2.85$ (2H, m, 2CH), 4.78–4.89 (21H, m, CH), 6.62 (1H, t, ${}^{4}J$ = 1.8 Hz, CH), 6.81 (2H, br s, NH₂), 7.31-7.43 (3H, m, 3CH), 7.61 $(2H, d, {}^{3}J = 6.7 \text{ Hz}, 2CH), 7.70 (2H, d, {}^{3}J = 6.9 \text{ Hz},$ 2CH), 8.21 (2H, d, ${}^{3}J = 6.9$ Hz, 2CH). ${}^{13}C$ NMR (125.7 MHz, CDCl₃): 45.1 (CH₂), 79.1 (C), 89.2 (CH), 114.5 (CN), 126.0 (CH), 126.7 (2CH), 127.1 (CH), 128.8 (2CH), 129.2 (2CH), 129.9 (2CH), 136.3 (C), 140.5 (C), 146.1 (C), 147.3 (C), 183.7 (C). EI-MS (70 eV): m/z $(\%) = 333 (M^+, 11), 211 (18), 196 (25), 171 (34), 121$ (78), 77 (100). Anal. Calcd (%) for (333.35): C, 68.46; H, 4.54; N, 12.61. Found: C, 68.69; H, 4.82; N, 12.80.

6-Amino-4-(2-methoxyethylidene)-3-phenyl-3,4-dihydro-2Hpyran-5-carbonitrile (4n, $C_{15}H_{16}N_2O_2$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 4/1, $R_{\rm f}$: 0.39) affording 0.16 g (63%) 4**n**. Colorless solid, mp: 78–80°C. IR (KBr): \overline{V} = 3413, 3400, 2924, 2211, 1611, 1459, 1381, 1250, 1103. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H}$ = 3.48 (3H, s, OMe), 4.03–4.25 (3H, m, 3CH), 4.45–4.61 (2H, m, 2CH), 5.81 (1H, t, ³J = 6.7 Hz, CH), 6.87 (2H, br s, NH₂), 7.310 (1H, t, ³J = 6.7 Hz, CH), 7.35–7.42 (4H, m, 4CH). ¹³C NMR (125.7 MHz, CDCl₃): 40.3 (CH), 57.4 (OMe), 69.1 (CH₂), 76.4 (CH₂), 79.1 (C), 114.2 (CN), 126.5 (CH), 128.5 (2CH), 129.1 (CH), 130.5 (2CH), 141.3 (C), 144.7 (C), 182.5 (C). EI-MS (70 eV): *m*/*z* (%) = 256 (M⁺, 2), 179 (15), 164 (28), 139 (40), 121 (79), 77 (100). Anal. Calcd (%) for (256.21): C, 70.29; H, 6.29; N, 10.93. Found: C, 70.48; H, 6.46; N, 11.16.

6-Amino-3-phenyl-4-((trimethylsilyl)methylene)-3,4-dihydro-2*H*-pyran-5-carbonitrile (40, $C_{16}H_{20}N_2OSi$). The crude product was purified by column chromatography (SiO₂; hexane/EtOAc 5/1, R_f: 0.28) affording 0.17 g (60%) 40. Colorless solid, mp: 60–62°C. IR (KBr): \overline{V} = 3345, 3303, 3066, 2924, 2260, 1613, 1539, 1447, 1331, 1083. ¹H NMR (500.1 MHz, CDCl₃): $\delta_{\rm H} = 0.11$ (9H, s, 3 Me), 3.90–3.98 (1H, m, CH), 4.41 (1H, dd, ${}^{2}J = 12.0$ Hz, ${}^{3}J$ = 5.9 Hz, CH), 4.63 (1H, dd, ${}^{2}J$ = 12.0 Hz, ${}^{3}J = 9.1$ Hz, CH), 5.95 (1H, d, ${}^{4}J = 1.7$ Hz, CH), 6.87 (2H, br s, NH₂), 7.30 (1H, t, ${}^{3}J$ = 6.4 Hz, CH), 7.35–7.44 (4H, m, 4CH). ¹³C NMR (125.7 MHz, CDCl₃): 1.9 (3CH₃), 49.5 (CH), 74.7 (C), 76.19 (CH₂), 114.6 (CN), 125.1 (CH), 127.9 (2CH), 128.9 (2CH), 129.8 (CH), 142.7 (C), 160.1 (C), 183.7 (C). EI-MS (70 eV): m/z $(\%) = 284 (M^+, 7), 207 (15), 192 (27), 167 (41), 121$ (73), 77 (100). Anal. Calcd (%) for (284.43): C, 67.56; H, 7.09; N, 9.85. Found: C, 67.82; H, 7.33; N, 10.04.

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