

Atropselective Synthesis of *N*,*C*-Bis(diphenylphosphanes) from Bridged 2-Arylindoles Based on Effective Point-to-Axial Asymmetric Inductions after an Unusual Dilithiation^{\perp}

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Supporting Information



ABSTRACT: An asymmetric methanolysis of glutaric anhydride and 6 ensuing steps gave veratrol-annulated dimethylcycloheptenone diastereomers with 99% *ee*; ring closures occurred by Friedel–Crafts acylations of carboxylic acids obtained by stereospecific hydrogenolyses of a pair of diastereomeric δ -lactones. The mentioned cycloheptenones and Ph–NH–NH₂ underwent Fischer indole syntheses providing the tetracyclic indoles *cis*- and *trans*-14a, respectively. Double lithiations with BuLi and quenchings with ClPPh₂ furnished the diphosphanes *cis*- and *trans*-15 with perfect (*P*)- and (*M*)-atropselectivity, respectively.

E nantiomerically pure diphosphanes are abundantly used ligands in asymmetric catalysis.¹ An important subclass thereof contains a diisocyclic C_2 -symmetric 1,1'-biaryl-2,2'bis(diphosphane) scaffold. The most famous diphenylphosphane of this design is BINAP.² Many variations thereof are known, including H₈-BINAP,³ SEGPHOS,⁴ DIFLUOR-PHOS,⁵ SUNPHOS,⁶ SYNPHOS,⁷ SOLPHOS,⁸ a nameless aza analog thereof,9 and many others.1 Related diphosphane designs are BIPHEMP,¹⁰ MeO-BIPHEP,¹¹ C_nTunaPhos = TUNEPHOS,¹² Bu-PQ-PHOSPHOS,¹³ Pent-PQ-PHOS-PHOS,¹⁴ Hex-PQ-PHOSPHOS,¹⁴ and many more.¹ Almost all diphosphanes of this kind have been synthesized atropunselectively: by resolving the racemic phosphane, the corresponding racemic bis(phosphane oxide), or a racemic precursor. The only atropselective syntheses in this series used point-to-axial asymmetric inductions which allowed Ullmann couplings (inter alia) to afford the bis(phosphane oxide) precursors of one diphosphane with $ds = 78:22^8$ and of three other diphosphanes with $de \ge 98\%$.^{13,14}

Diheterocyclic C_2 -symmetric 1,1'-biaryl-2,2'-bis-(diphosphanes) have been reported as well.¹⁵ Figure 1 illustrates them by the biindolyldiphosphanes 1-4.¹⁶⁻¹⁹ Compounds $1-3b^{16-18}$ and all respective diphosphanes based on other heterocycles²⁰ stem from resolving racemic mixtures, wherein the biaryl axis had been established without atropcontrol. The only atropselective diphosphane syntheses in this series exploited point-to-axial asymmetric inductions in two dibrominations of a 2,2'-biindolyl (ds = 95:5 and 5:95).¹⁹



Figure 1. C_2 -symmetric 3,3'-biindolyl-2,2'-bis(diphosphane) 1,¹⁶ 2,2'-biindolyl-1,1'-bis(diphosphane) 2,¹⁷ and 2,2'-biindolyl-3,3'-bis-(diphosphanes) 3a,b,¹⁸ and 4.¹⁹

Refunctionalizations afforded the diphosphanes 4 [= (R,R,P)-4] and *dia*-4 [= (R,R,M)-4] atropisomerically pure.¹⁹

Monoheterocyclic and therefore only C_1 -symmetric biarylbis(diphenylphosphanes) exist, too. Those with an arylindole scaffold $(5-9,^{21-23}$ Figure 2) can be regarded as hybrids of the earlier mentioned diisocyclic C_2 -symmetric 1,1'-biaryl-2,2'-bis(diphosphane) scaffolds and the diphosphane motifs from Figure 1. None of them has been obtained atropisomerically pure yet. In contrast, the present study reveals perfectly atropselective syntheses of the arylindole diphosphanes *cis*-15, which is *P*-configure 4).

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Figure 2. Arylindole-based diphosphanes known to date.

Our approach was inspired by a 60 year old observation:²⁴ The UV absorption of the cycloheptadieno-phenylindole **10** (Figure 3) exhibits a hypsochromic shift relative to the



Figure 3. Cycloheptadieno-2-arylindoles: scaffolds with temporary, if not permanent twists.

analogous cyclohexadieno-phenylindole.²⁴ This was attributed to a decreased conjugation across the indole-phenyl bond due to "the polymethylene chain acting as a lever which turns the benzene moiety out of the indole plane".²⁴ The resulting twist about the phenyl-indole bond causes the formation of an (equilibrating) 50:50 mixture of an (M)- and a (P)atropisomer.²⁵ We wanted to design cycloheptadieno-phenylindoles which would twist unidirectionally (with or without equilibrating with each other) and thereby result as single atropisomers in a conceptually novel manner.

The atropisomerization of a few derivatives of the unsubstituted cycloheptadieno-phenylindole 10²⁶ and some related biaryls was studied by ¹H NMR spectroscopy: Compound 13²⁷ did not atropisomerize, but compounds 11²⁸ and 12²⁷ did. Biphenyl—rather than 2-phenylindole containing $a CH_2 - CH_2 - CH_2$ bridge $(\Delta G^{\ddagger}_{atropisomerization in the {}^{1}_{H NMR spectrometer} = 12.5 \text{ kcal/mol}^{29})$ and their CH₂-C(CO₂Et)₂-CH₂- or CH₂-O-CH₂-containing analogs atropisomerize with similar rates.³⁰ A biphenyl with a meso-configured C(Me)H-NH-C(Me)H bridge atropisomerizes more slowly ($\Delta G^{\ddagger} = 13.6 \text{ kcal/mol}^{31}$) and a biphenyl with a O–CH₂–O bridge much faster ($\Delta G^{\ddagger} \approx 1 \text{ kcal/mol}^{32}$). Related biphenyls contain the chiral bridges C(Me)H-NH- $C(Me)H^{31}$ or $CH_2-NH-C(Me)H^{33}$ and are single atropisomers. Their stereocenter(s) differentiate(s) the stabilities of the respective (diastereomorphic!) atropisomers so much that the more stable atropisomer does not cross the atropisomerization barrier—no matter how low it might be—because there is no driving force.

The preceding insights and our objective of modifying the (P)- and (M)-twisted scaffold **10** to exclusively (P)- or exclusively (M)-twisted derivatives let us screen the atropisomerical preferences of several type-**14** and type-*dia*-**14** cyloheptadieno-arylindoles (formulas: Figure 3). Proving

our concept, we synthesized their representatives *cis*- and *trans*- $14a^{34}$ and derived the arylindolediphosphanes *cis*- and *trans*-15 therefrom (formulas: Figure 4).



Figure 4. C_1 -symmetric biarylbis(diphosphanes) *cis*-15 and *trans*-15, their precursors *cis*-14a and *trans*-14a (cf. Figure 3), and their conceived origin from the enantiomerically pure δ -ketoester 18.

Our retrosynthetic analyses of the cyloheptadieno-arylindoles cis- and trans-14a proceeded via the equally configured cylohepteno-veratroles cis- and trans-17, respectively, to the veratrol-containing δ -ketoester 18 (Figure 4). Its aliphatic moiety should originate from an unsymmetric derivative 16 of 3-methylglutaric acid. Type-16 compounds can be obtained by desymmetrizing 3-methylglutaric anhydride (19) by an asymmetric methanolysis.³⁵ We started synthesizing compounds cis- and trans-14a using Bolm's variant³⁶ of such a reaction³⁷ (Scheme 1). Methanolysis of **19** in chlorobenzene rather than in $CCl_4^{36,38}$ provided 97% of the glutaric half-ester **16a** with 70% *ee;*³⁹ three low-temperature recrystallizations^{40a} of the salt(s)⁴⁰ formed with 1.0 equiv of (-)-cinchonidine increased the ee³⁹ of regenerated 16a to 99%. This compound was converted into the acid chloride 16b. Without purification, a CH₂Cl₂ solution acylated veratrol (20) at room temp after the addition of 1.0 equiv of either AlCl₃ or Me₂AlCl. To our surprise, the resulting δ -ketoester 18 was racemic.^{41,42} We surmise that racemization occurred by the mechanism suggested in Scheme 2.

We proceeded to the (*R*)-configured δ -ketoester **18** rather than to *rac*-**18** after activating the glutaric half-ester **16a** as the *S*-ethyl thioester **16c**, which we isolated in 70% yield, or as the mixed anhydride **16d** (accessed with MeO₂C-O-CO₂Me⁴³ rather than with MeO₂C-Cl), which we carried on without purification (Scheme 1). The thioester **16c** and an excess of veratrylzinc halide **21a** (prepared from 4-iodoveratrol by successive I \rightarrow Li and Li \rightarrow ZnCl exchanges) coupled in a Fukuyama ketone synthesis.⁴⁴ It delivered over 60% of the δ ketoester **18**. However, this compound was hard to separate from considerable amounts of accompanying 3,3',4,4'tetramethoxybiphenyl by flash chromatography on silica gel.⁴⁵ Coupling the mixed anhydride **16d** with the veratrolboronic acid **21b** [prepared from 4-bromoveratrol as described;⁴⁶ Scheme 1. Synthesis of δ -Ketoester 18^a



^aReaction conditions: (a) MeOH (3.0 equiv), (-)-quinine (1.1 equiv), chlorobenzene/toluene (1:1), -55 °C, 72 h; 97%, 70% ee; (b) (-)-cinchonidine (1.0 equiv), acetone/H₂O, 40 °C \rightarrow 4 °C, 24 h; aq. HCl (1 M); 20%, 99% ee; (c) (COCl)₂ (1.1 equiv), DMF (cat.), CH₂Cl₂, room temp, 3 h; (d) AlCl₃ (1.0 equiv), **20** (1.0 equiv), CH₂Cl₂, room temp, 20 h; 97% over the 2 steps; (e) same as (d) but Me₂AlCl (1.0 equiv), DMAP (20 mol %), CH₂Cl₂, room temp, 18 h; 70%; (g) MeO₂C-O-CO₂Me (3.0 equiv), **21b** (1.2 equiv), Pd(OAc)₂ (3.0 mol %), P(*p*-anisyl)₃ (7.0 mol %), H₂O (17 mol %), THF, 80 °C, 16 h; 79%; (h) **21a** (3.0 equiv), PdCl₂(PPh₃)₂ (5.0 mol %), THF, room temp, 16 h; 60% over the 2 steps.

Scheme 2. Mechanism of the Enantiomerization $16b \rightarrow ent-16b$ Which Causes the Racemization of the Acid Chloride 16b under both Friedel–Crafts Acylation Conditions of Scheme 1



yield: 56% (60%⁴⁶)] was a better synthesis^{43a} of the desired δ -ketoester **18** because no difficult to separate 3,3',4,4'-tetramethoxybiphenyl interfered.

Our cycloheptadieno-arylindole syntheses continued by adding MeMgCl to the carbonyl group of the δ -ketoester 18 (Scheme 3). Quenching the reaction mixture with satd. aq. NaHCO₃, extractive workup, and purification by flash chromatography on silica gel⁴⁵ furnished no δ -hydroxyesters but the derived valerolactones 29^{47a} and *dia*-29^{47b} with *ds* = 60:40. Next, the benzylic C–O bonds of lactones 29 and *dia*-29 were hydrogenolyzed in EtOH catalyzed by 15 mol % Pd/ C. Under this proviso, both hydrogenolyses exhibited \geq 95% inversion of configuration. This rendered the carboxylic acids 30 (whose follow-up product *cis*-17 was X-rayed,^{47c} revealing the steric course of the preceding hydrogenolysis) and *dia*-30, respectively. Less Pd/C, other catalysts [Pd-black, Pd(OH)₂], or other solvents (MeOH, AcOEt, and 2-Me-THF) gave inferior selectivities. Hydrogenolyses of tertiary C_{benzvlic}–O Scheme 3. Synthesis of the Cycloheptadieno(arylindolyls) cis- and trans-14^a



^{*a*}Reaction conditions: (a) MeMgCl (1.5 equiv), THF, −78 °C → room temp, 18 h; 37% **29**,^{47a} 25% *dia*-**29**,^{47b} 20% *de*; (b) H₂ (balloon), Pd/C (10 wt-%, 15 mol %), EtOH, room temp, 19 h; 90% **30** or 92% *dia*-**30**; (c) Polyphosphoric acid (10 times the mass of **30** or *dia*-**30**), sulfolane, 100 °C, 1 h; 90% *cis*-**17**^{47c} or 88% *trans*-**17**; (d) PhNHNH₂ (1.0 equiv), HCl (concd, 4.0 equiv), EtOH, reflux, 24 h; 72% *cis*-**14** or 68% *trans*-**17**.

bonds akin to our examples $29 \rightarrow 30$ and $dia-29 \rightarrow dia-30$ and proceeding with inversions of configurations, too, were first investigated by Mitsui et al.⁴⁸ One such hydrogenolysis of a tolyl-substituted δ -lactone was accomplished by Campagne et al.⁴⁹ during our own studies.

At 100 °C the carboxylic acids **30** and *dia*-**30** underwent Friedel–Crafts cyclizations in polyphosphoric acid⁵⁰ which we diluted with sulfolane⁵¹ to improve miscibilities and thereby increase yields (Scheme 3). This led to the benzosuberone *cis*-**17**^{47c} in 90% yield rather than only in 53% yield and to the benzosuberone *trans*-**17** in 88% yield. *Cis*-**17** crystallized from *n*-hexane with dr = 100:0 while *trans*-**17** remained an oil with dr = 95:5. The last steps of Scheme 3 are Fischer indole syntheses combining the Friedel–Crafts products *cis*-**17** or *trans*-**17** with phenylhydrazine in hot ethanolic HCl.⁵² A 72% yield of the bridged arylindole *cis*-**14** resulted from *cis*-**17**, and a 68% yield of *trans*-**14** resulted from *trans*-**17**. *Cis*-**14** was diastereopure immediately, whereas *trans*-**14** had to be crystallized to become diastereopure. The H-N- $C^2-C^{1'}-C^{2'}-H$ motifs of the arylindoles *cis*and *trans*-14 gave Li-N- $C^2-C^{1'}-C^{2'}-Li$ motifs upon exposure to 2 equiv of *n*BuLi provided that TMEDA was present (Scheme 4, not depicted). Treatment with ClPPh₂

Scheme 4. N-Directed Lithiation/Phosphorylation of the Cycloheptadieno(indolylaryls) *cis-* and *trans-*14; Lithiation Analogies and Differences^a



"Reaction conditions: (a) *n*BuLi (2.5 equiv), TMEDA (2.5 equiv), Et₂O, room temp, 3 h; ClPPh₂ (10 equiv), -78 °C \rightarrow room temp, 18 h; 72%;^{47d} (b) Same as (a); 63%;^{47e} (c) Reference 53; LiTMP (5.0 equiv), THF, -10 °C, 1 h; HCO₂Et (1.0 equiv), TMSCl (10.0 equiv), -78 °C \rightarrow room temp; 75%; (d) Same as (a) but using *s*BuLi instead of *n*BuLi; 40%.

established the $Ph_2P-N-C^2-C^{1'}-C^{2'}-PPh_2$ motifs of the diphosphane targets *cis*- and *trans*-15. Purification by flash chromatography on silica gel⁴⁵ rendered *cis*-15 in 72% yield and *trans*-15 in 63% yield.

The double C- and N-lithiation employed in the last step of our routes to the diphosphanes *cis*- and *trans*-15 was unprecedented. The closest-related dilithiation of that kind was the transformation $31 \rightarrow 33^{53}$ (Scheme 4; the NH group of 31 belongs to a vinylogous amide and thus should be more acidic than the NH group of the indoles *cis*- and *trans*-14). The feasibility of lithiating a 2-arylindole *twice* does not require the support of a MeO group: 3-Methyl-2-phenylindole (32) was both C- *and* N-lithiated with *s*BuLi (though not with *n*BuLi); this allowed progression to the previously described⁵⁴ diphosphane 8 in 40% yield.⁵⁵

In the solid state, the $N-C^2-C^{1'}-C^{2'}$ motifs of our diphosphanes are twisted *in opposite senses*. These twists allow the following dihedral angle quantifications: +62.8° in *cis*-15^{47d} and -67.9° in *trans*-15. ^{47e} The different signs of these angles are due to the biaryl axis being (*P*)-configured in diphosphane *cis*-15 yet (*M*)-configured in *trans*-15. This contrast arises in response to the changing configurations of the stereocenter attached to C^{6'} (*S*-configured in *cis*-15, *R*-configured in *trans*-15), no matter the invariant (*R*)-configuration of the stereocenter attached to C³. Differently expressed, the diphosphane syntheses of the present study

were not only atropselective but additionally atropdivergent. These findings underline the viability of a concept first laid out in the discussion of Figure 3.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.9b03896.

Experimental procedures, characterization data, NMR spectra, and details concerning the X-ray analyses. (PDF)

Accession Codes

CCDC 1533413, 1533414, 1549686, 1561798, and 1578270 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc. cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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DEDICATION

[⊥]Dedicated to Professor Rolf Huisgen at the occasion of his 99th birthday.

REFERENCES

(1) Reviews of enantiomerically pure diphosphane ligands: (a) Wencel-Delord, J.; Panossian, A.; Leroux, F. R.; Colobert, F. *Chem. Soc. Rev.* **2015**, *44*, 3418–3430. (b) Pereira, M. M.; Calvete, M. J. F.; Carrilho, R. M. B.; Abreu, A. R. *Chem. Soc. Rev.* **2013**, *42*, 6990– 7027. (c) Gillespie, J. A.; Zuidema, E.; van Leeuwen, P. W. N. M.; Kamer, P. C. J. Phosphorus Ligand Effects in Homogeneous Catalysis and Rational Catalyst Design. In *Phosphorus(III) Ligands in Homogeneous Catalysis: Design and Synthesis*; Kramer, P. C. J., van Leeuwen, P. W. N. M., Eds.; John Wiley & Sons: Chichester, 2012; pp 1–26. (d) Li, W.; Zhang, X. Chiral Phosphines and Diphosphines. In *Phosphorus(III) Ligands in Homogeneous Catalysis: Design and Synthesis*; Kramer, P. C. J., van Leeuwen, P. W. N. M., Eds.; John Wiley & Sons: Chichester, 2012; pp 27–80. (e) Ohkuma, T.; Kurono, N. BINAP. In *Privileged Chiral Ligands and Catalysts*; Zhou, Q.-L., Ed.; Wiley-VCH: 2011; pp 1–45.

(2) Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; Noyori, R. J. Am. Chem. Soc. **1980**, *102*, 7932–7934.

(3) Zhang, X.; Mashima, K.; Koyano, K.; Sayo, N.; Kumobayashi, H.; Akutagawa, S.; Takaya, H. *Tetrahedron Lett.* **1991**, *32*, 7283–7286.

(4) Saito, T.; Yokozawa, T.; Ishizaki, T.; Moroi, T.; Sayo, N.; Miura, T.; Kumobayashi, H. *Adv. Synth. Catal.* **2001**, *343*, 264–267.

(5) Jeulin, S.; Duprat de Paule, S.; Ratovelomanana-Vidal, V.; Genêt, J.-P. Angew. Chem. 2004, 116, 324–329; Angew. Chem., Int. Ed. 2004, 43, 320–325.

(6) (a) Sun, Y.; Wan, X.; Guo, M.; Wang, D.; Dong, X.; Pan, Y.; Zhang, Z. *Tetrahedron: Asymmetry* **2004**, *15*, 2185–2188. (b) Zhang, H.-W.; Meng, Q.-H.; Zhang, Z.-G *Chin. J. Chem.* **2008**, *26*, 2098– 2102.

(7) Duprat de Paule, S.; Jeulin, S.; Ratovelomanana-Vidal, V.; Genêt, J.-P.; Champion, N.; Dellis, P. *Tetrahedron Lett.* **2003**, *44*, 823–826.

(8) Kesselgruber, M.; Lotz, M.; Martin, P.; Melone, G.; Müller, M.; Pugin, B.; Naud, F.; Spindler, F.; Thommen, M.; Zbinden, P.; Blaser,

H.-U. Chem. - Asian J. 2008, 3, 1384–1389.
(9) Qiu, L.; Qi, J.; Pai, C.-C.; Chan, S.; Zhou, Z.; Choi, M. C. K.;

Chan, A. S. C. Org. Lett. 2002, 4, 4599–4602. (10) Schmid, R.; Cereghetti, M.; Heiser, B.; Schönholzer, P.; Hansen, H.-J. Helv. Chim. Acta 1988, 71, 897–929.

(11) Schmid, R.; Foricher, J.; Cereghetti, M.; Schönholzer, P. Helv. Chim. Acta 1991, 74, 370-380.

(12) As CnTunaPhos: (a) Zhang, Z.; Qian, H.; Longmire, J.; Zhang, X. J. Org. Chem. **2000**, 65, 6223–6226. As TUNEPHOS: (b) Sun, X.; Zhou, L.; Li, W.; Zhang, X. J. Org. Chem. **2008**, 73, 1143–1146.

(13) (a) Qiu, L.; Wu, J.; Chan, S.; Au-Yeung, T. T.-L.; Ji, J.-X.; Guo, R.; Pai, C.-C.; Zhou, Z.; Li, X.; Fan, Q.-H.; Chan, A. S. C. *Proc. Natl. Acad. Sci. U. S. A.* **2004**, *101*, 5815–5820. (b) Qiu, L.; Kwong, F. Y.; Wu, V.; Lam, W. H.; Chan, S.; Yu, W.; Li, Y.-M.; Guo, R.; Zhou, Z.; Chan, A. S. C. J. Am. Chem. Soc. **2006**, *128*, 5955–5965.

(14) n = 1-2: Reference 13b.

(15) (a) Arshad, N.; Kappe, C. O. Adv. Heterocycl. Chem. 2010, 99, 33–59. (b) Li, Y.-M.; Yu, W.-Y.; Chan, A. S. C. In Phosphorus Ligands in Asymmetric Catalysis: Synthesis and Applications, Vol. 1; Börner, A., Ed.; Wiley-VCH: Weinheim, 2008; pp 260–283. (c) Wu, J.; Chan, A. S. C. Acc. Chem. Res. 2006, 39, 711–720. (d) Au-Yeung, T. T.-L.; Chan, A. S. C. Coord. Chem. Rev. 2004, 248, 2151–2164. (e) Benincori, T.; Rizzo, S.; Sannicò, F. J. J. Heterocycl. Chem. 2002, 39, 471–485.

(16) Berens, U.; Brown, J. M.; Long, J.; Selke, R. Tetrahedron: Asymmetry **1996**, 7, 285–292.

(17) Benincori, T.; Brenna, E.; Sannicoló, F.; Trimarco, L.; Antognazza, P.; Cesarotti, E.; Zotti, G. J. Organomet. Chem. **1997**, 529, 445–453.

(18) Benincori, T.; Piccolo, O.; Rizzo, S.; Sannicoló, F. J. Org. Chem. 2000, 65, 8340–8347.

(19) Baumann, T.; Brückner, R. Angew. Chem. 2019, 131, 4762–4768; Angew. Chem., Int. Ed. 2019, 58, 4714–4719.

(20) (a) Benincori, T.; Brenna, E.; Sannicolò, F.; Trimarco, L.; Antognazza, P.; Cesarotti, E.; Demartin, F.; Pilati, T. J. Org. Chem. **1996**, 61, 6244–6251. (b) Benincori, T.; Brenna, E.; Sannicoló, F.; Trimarco, L.; Antognazza, P.; Cesarotti, E.; Zotti, G. J. Organomet. Chem. **1997**, 529, 445–453. (c) Benincori, T.; Cesarotti, E.; Piccolo, O.; Sannicolò, F. J. Org. Chem. **2000**, 65, 2043–2047. (d) Andersen, N. G.; Parvez, M.; McDonald, R.; Keay, B. A. Org. Lett. **2000**, 2, 2817–2820. (e) Pai, C.-C.; Lin, C.-W.; Lin, C.-C.; Chen, C.-C.; Chan, A. S. C. J. Am. Chem. Soc. **2000**, 122, 11513–11514.

(21) Artemova, N. V.; Chevykalova, Ma. N.; Luzikov, Y. N.; Nifant'ev, I. E.; Nifant'ev, E. E. *Tetrahedron* **2004**, *60*, 10365–10370.

(22) Kuang, F.; Su, Q.; Zhou, Y.; Yuan, A. Faming Zhuanli Shenqing **2017**, CN 107445989 A 20171208.

(23) Sannicolò, F.; Benincori, T.; Rizzo, S.; Gladiali, S.; Pulacchini, S.; Zotti, G. *Synthesis* **2001**, 2001, 2327–2336.

(24) Huisgen, R.; Ugi, I. Justus Liebigs Ann. Chem. **1957**, 610, 57–66. (25) In this letter, "atropisomer", "to atropisomerize", and "atropisomerization" refer to biaryls exchanging an (M)-conformation about their biaryl axis for a (P)-conformation or undergoing the reverse processes, by which "(P)-configured biaryl atropisomers" render "(M)-configured biaryl atropisomers".

(26) 100 ¹H NMR spectrum of compound **10**: Hong, B.-C.; Jiang, Y.-F.; Chang, Y.-L.; Lee, S.-J. *J. Chin. Chem. Soc.* **2006**, *53*, 647–662. (27) 100 MHz ¹H NMR spectra of compounds **12** and **13**: Fujimori, K.; Yamane, K. *Bull. Chem. Soc. Jpn.* **1978**, *51*, 3579–3581.

(28) The 300 MHz ¹H NMR spectrum of compound 11 displays each ring-CH₂ as a singlet: Campeau, L.-C.; Parisien, M.; Jean, A.; Fagnou, K. J. Am. Chem. Soc. **2006**, 128, 581–590.

(29) Rotzler, J.; Gsellinger, H.; Bihlmeier, A.; Gantenbein, M.; Vonlanthen, D.; Häussinger, D.; Klopper, W.; Mayor, M. Org. Biomol. Chem. **2013**, *11*, 110–119.

(30) 300 MHz ¹H NMR spectra in CDCl₃: Hennings, D. D.; Iwama, T.; Rawal, V. H. *Org. Lett.* **1999**, *1*, 1205–1208.

(31) Saudan, L. A.; Bernardinelli, G.; Kündig, E. P. Synlett 2000, 483–486.

(32) Masters, K. S.; Bihlmeier, A.; Klopper, W.; Bräse, S. Chem. Eur. J. 2013, 19, 17827-17835.

(33) Pira, S. L.; Wallace, T. W.; Graham, J. P. Org. Lett. 2009, 11, 1663-1666.

(34) First synthesis of the scaffold (all $R^i = H$) of a cycloheptadienoarylindole 14 containing the same phenyl-substituents ($R_n = 2,3$ dimethoxy) as our cycloheptadieno-arylindoles *cis*- and *trans*-14a: Reference 26.

(35) Early review: Chen, Y.; McDaid, P.; Deng, L. Chem. Rev. 2003, 103, 2965–2984.

(36) Cyclic anhydrides without 3-methylglutaric anhydride (ref 36c): (a) Short communication: Bolm, C.; Gerlach, A.; Dinter, C. L. *Synlett* **1999**, 195–196. (b) Full paper: Bolm, C.; Schiffers, I.; Dinter, C. L.; Gerlach, A. J. Org. Chem. **2000**, 65, 6984–6991. (c) Asymmetric methanolyses of this compound based on chinchona alkaloids—as in ref 36a, b—have not been reported, but other methanolysis conditions were (258 SciFinder hits; selected references: Supporting Information).

(37) Not using 3-methylglutaric anhydride: (a) Short communication: Hiratake, J.; Yamamoto, Y.; Oda, J.-i. J. Chem. Soc., Chem. Commun. 1985, 1717–1719. (b) Full paper: Hiratake, J.; Inagaki, M.; Yamamoto, Y.; Oda, J.-i. J. Chem. Soc., Perkin Trans. 1 1987, 1053– 1058.

(38) Methanolyzing the anhydride **19** in CCl_4^{36} furnished the halfester **16a** in 98% yield with 70% *ee.*

(39) The *ee*'s of the half-ester **16a** were determined by HPLC (details: Supporting Information) after esterification with *p*-bromophenol: Manzano, R.; Andrés, J. M.; Muruzábal, M.-D.; Pedrosa, R. J. Org. Chem. **2010**, 75, 5417–5420.

(40) (a) Low-temperature recrystallization: Lehr, K.; Fürstner, A. *Tetrahedron* **2012**, *68*, 7695–7700. (b) Original recrystallization: Poppe, L.; Novák, L.; Kolonits, P.; Bata, A.; Szántay, C. *Tetrahedron* **1988**, *44*, 1477–1487.

(41) The *ee* of the δ -ketoester **18** was determined by HPLC (details: Supporting Information).

(42) (a) The conditions of step (e) of Scheme 1 allowed 5-chloro-2-hexyl-1-methylindole to be acylated with the acid chloride **16b** (*ee* not reported) within 1 h which gave 90% of a δ -ketoester of 80% *ee*: Patel, P.; Reddy, C. N.; Gore, V.; Chourey, S.; Ye, Q.; Quedraogo, Y. P.; Gravely, S.; Pawell, W. S.; Rokack, J. ACS Med. Chem. Lett. **2014**, 5, 815–819. The indole is more nucleophilic than veratrol (**20**) and thus leaves less or no time for the acid chloride **16b** to racemize.

(43) Method: (a) Gooßen, L. J.; Ghosh, K. Angew. Chem. 2001, 113, 3566–3568; Angew. Chem., Int. Ed. 2001, 40, 3458–3460.
(b) Gooßen, L. G.; Winkel, L.; Döhring, A.; Ghosh, K.; Paetzold, J. Synlett 2002, 8, 1237–1240.

(44) Method: Tokuyama, H.; Yokoshima, S.; Yamashita, T.; Fukuyama, T. *Tetrahedron Lett.* **1998**, *39*, 3189–3192.

(45) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923–2925.

(46) Zhang, Q.-Q.; Xie, J. H.; Yang, X.-H.; Xie, J.-B.; Zhou, Q.-L. Org. Lett. **2012**, *14*, 6158–6161.

(47) (a) Racemic lactone **29** formed crystals whose structure was elucidated by X-ray analysis; the corresponding data are contained in CCDC 1578270. (b) Racemic lactone *dia*-**29** formed crystals whose structure was elucidated by X-ray analysis; the corresponding data are contained in CCDC 1533413. (c) Racemic benzosuberone *cis*-**17** (obtained by the Friedel–Crafts acylation of the hydrogenolysis product *dia*-**29**) formed crystals whose structure was determined by

Organic Letters

X-ray crystallography; the corresponding data are contained in CCDC 1533414. (d) Racemic diphosphane *cis*-15 formed crystals whose structure was elucidated by X-ray analysis; the corresponding data are contained in CCDC 1549686. (e) Racemic diphosphane *trans*-15 formed crystals whose structure was elucidated by X-ray analysis; the corresponding data are contained in CCDC 1561798.

(48) (a) Mitsui, S.; Iijima, K.; Masuko, T. Nippon Kagaku Zasshi 1963, 84, 833–838. (b) Mitsui, S.; Imaizumi, S.; Takamura, I.; Takamura, M. Nippon Kagaku Zasshi 1963, 84, 838–841. (c) Mitsui, S.; Iijima, K.; Masuko, T. J. Chem. Soc. Japan, Pure Chem. Sect. (Nippon Kagaku Zasshi) 1963, 84, 842–845. (d) Konno, K.; Mitsui, S. Nippon Kagaku Zasshi 1964, 85, 497–500. (e) Mitsui, S.; Iijima, K. J. Chem. Soc. Japan, Pure Chem. Sect. (Nippon Kagaku Zasshi) 1964, 86, 682– 686. (f) Mitsui, S.; Kudo, Y.; Kobayashi, M. Tetrahedron 1969, 25, 1921–1927. (g) Mechanistic studies: Garbisch, E. W., Jr.; Schreader, L.; Frankel, J. J. J. Am. Chem. Soc. 1967, 89, 4233–4235.

(49) Spielmann, K.; de Figueiredo, R. M.; Campagne, J.-M. J. Org. Chem. 2017, 82, 4737-4743.

(50) Method: Uhlig, F. Angew. Chem. 1954, 66, 435-436.

(51) Method: Nelson, P. H.; Untsch, K. G. U.S. Patent 4038299A 1977.

(52) Reaction conditions: Dufour, F.; Kirsch, G. Synlett 2006, 2006, 1021–1022.

(53) Scopton, A.; Kelly, T. R. J. J. Org. Chem. 2005, 70, 10004-10012.

(54) Reference 23 prepared the same diphosphane 8 via the identical dilithio intermediate but gained the latter by converting a N–H to a N–Li bond (monolithiation) and a C–Br to a C–Li bond (Li/Br exchange).

(55) 2-Phenylindole underwent such a dilithiation, too. Quenching with Ph_2PCl delivered a complex mixture but quenching with Me_3SiCl led to 3-(trimethylsilyl)-2-[2-(trimethylsilyl)phenyl]indole in 50% yield.