

Article

Chemodivergent One-pot Multi-Component Synthesis of Pyrroles and Tetrahydropyridines under Solvent free, Catalyst free and Grinding method

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3 **Chemodivergent One-pot Multi-Component Synthesis of Pyrroles and**
4 **Tetrahydropyridines under Solvent free, Catalyst free and Grinding**
5 **method**
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20
21 **Abstract**
22

23 A highly efficient, chemoselective synthesis of a library of polysubstituted pyrroles
24 and tetrahydropyridines has been achieved through the one-pot, multi-component reactions of
25 ethyl (*E*)-3-(aryl/alkyl amino) acrylates, 2,2-dihydroxy-1-arylethan-1-ones and malononitrile
26 under solvent and catalyst free grinding conditions. The selective formation of pyrrole or
27 tetrahydropyridines relied on the substitution in the *N*-aryl of ethyl (*E*)-3-(4-arylamino)
28 acrylates. These reactions presumably occurred via a domino Knoevenagel condensation,
29 Michael addition followed by intramolecular cyclization sequence of reactions in a single
30 transformation.
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38 **Key words**

39 *Grinding, chemoselective, multi-component, tetrahydropyridines, solvent-free, catalyst-free*
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Introduction

Multicomponent reactions (MCRs) are versatile protocols employed for the synthesis of a variety of complex heterocycles and natural products.¹⁻³ These reactions involve the formation of multiple bonds in a single transformation in a one-pot process without the isolation of intermediates.⁴⁻⁸ Consequently, MCRs have been significantly used for the synthesis of wide range of heterocyclic compounds.⁹

Nitrogen containing heterocycles are wide spread in nature¹⁰ and play a significant role in chemical biology. Among these, pyrroles and pyridines are associated with unique biological activities.¹¹⁻¹³ For instance, arecoline, a naturally occurring tetrahydropyridine derivative exhibits stimulating effect as muscarinic acetylcholine receptor (Figure 1).¹⁴ In addition several synthetic tetrahydropyridines have been shown to display important biological activities. On the other hand 2,3,5-trisubstituted pyrroles have also been studied to exhibit significant biological activities.^{15,16} The tetrahydropyridines and pyrroles may be synthesised *via* a number of classical methods which also include the treatment of allylic alcohols with potassium bis(trimethylsilyl)amide,¹⁷⁻²¹ intermolecular amido carbonylation of carbomates²² and intramolecular condensations of aldehydes with enamides. Several methods such as transition metal catalysed cycloadditions, coupling of enamides with alkynes²³ and oxidative cyclization of *N*-allylamines are available for the syntheses of polysubstituted pyrroles.²⁴⁻²⁷ However most of these methods have drawbacks such as the use of expensive reagents or catalysts, lower yields and poor selectivity. In the present work, we have developed an efficient domino protocol for the synthesis of polysubstituted pyrroles **4** and tetrahydropyridines **5** from the one-pot three-component domino reactions of 2,2-dihydroxy-1-arylethan-1-ones **1**, malononitrile **2** and ethyl (*E*)-3-(4-arylamino) acrylates **3** under catalyst and solvent free conditions under grinding method (Scheme 1).

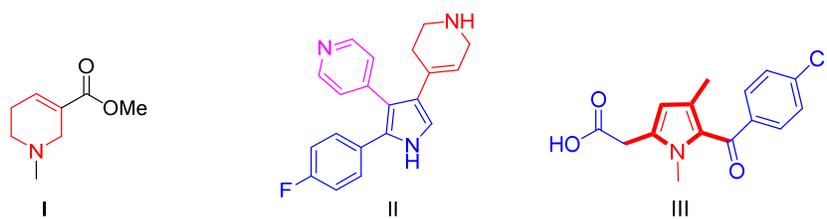
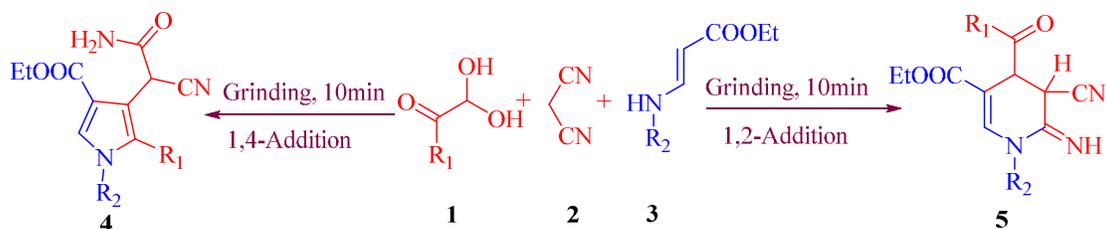


Figure 1 Selected drugs comprising tetrahydropyridine and pyrrole core

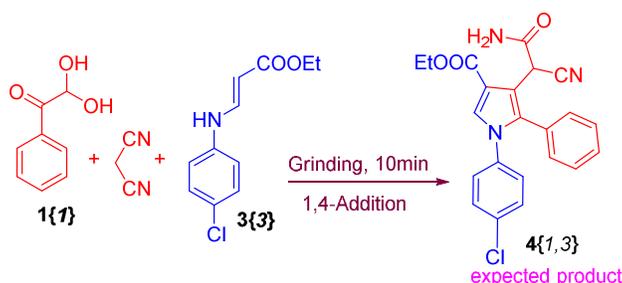


Scheme 1 Synthesis of pyrroles **4** and tetrahydropyridines **5**

Results and discussion

Initially, to optimize the reaction conditions a representative reaction of 2,2-dihydroxy-1-phenylethane-1-one **1**{**I**}, malononitrile **2**{**I**} and ethyl (*E*)-3-((4-bromophenyl)amino)acrylate **3**{**3**} that presumably affords the pyrrole **4**{**I,3**} was performed under various conditions (Table 1).

Table 1 Optimization of the reaction conditions



Entry	Solvent	Temperature (°C)	Time	Yield of 4{1,1,3} (%) ^b
1	EtOH	Reflux	30 min	80
2	EtOH	RT	2 h	50
3	MeOH	Reflux	2 h	65
4	MeOH	RT	30 min	62
5	MeOH	RT	2 h	50
6	MeCN	Reflux	4 h	60
7	THF	Reflux	6 h	35
8	CHCl ₃	RT	6 h	50
9	CH ₂ Cl ₂	RT	6 h	32
10	H ₂ O	Reflux	6 h	40
11	- ^a	Grinding	10 min	85
12	- ^a	Grinding	20 min	83
13	- ^a	Grinding	30 min	80

^aSolvent free condition ^bIsolated yield

To begin with, the above reactants were refluxed in ethanol and monitored by TLC intermittently. After 30 min of continuous reflux the reaction attained completion as evident from the TLC analysis. Then the reaction mixture was treated with cold ethanol (3 mL) and the resultant precipitate was filtered and dried to obtain **4{I,3}** in 80% yield (Table 1, entry 1) as shown by the ESI mass and NMR studies (Figure-2) disclosed the formation of pyrrole **4{I,3}**.

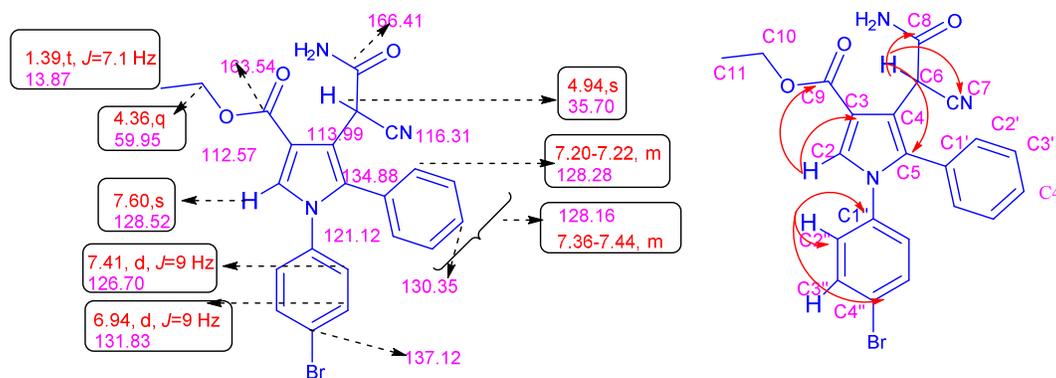


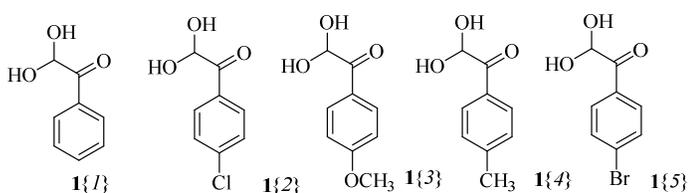
Figure 2 ^1H and ^{13}C chemical shifts and HMB correlations of **4{I,3}**

In contrast the above reaction took 2 h for 50 % completion at room temperature (Table 1, entry-2). Further exploration on the reaction conditions with a view to increase the yield of **4{I,3}** was done by employing solvents such as MeOH, MeCN, THF, CHCl_3 , CH_2Cl_2 and water under RT as well as reflux. From Table 1 it is evident that refluxing the reactants in ethanol was the optimum condition (Entry 1). However, inspired by solvent-free reactions, we investigated the reaction under the solvent-free grinding method. To our surprise the reaction afforded **4{I,3}** in 85% yield after 10 min (Table 1, entry 11). We found that the solvent-free grinding method gave superior yields in less time when compared to other conditions (Table 1).

Having optimized the reaction conditions, we then explored its applicability for library production employing various aryl 2,2-dihydroxy-1-arylethan-1-ones **1{I-5}** and ethyl (*E*)-3-(arylamino)acrylates **3{I-II}** (Figure 3 and Scheme 2). The reaction occurred well in all these cases affording good yields of the product (Table 2). Moreover, with the available variations we noted an interesting trend in the reaction with respect to the substitutions in the *N*-phenyl ring of enamine **3**. In general, the *N*-phenyl, *N*-heterocycle and the presence of electron withdrawing substituent in the *N*-phenyl ring in **3** afforded the expected pyrroles **4**

(Table 2, Entries 1–7), whereas the enamines **3** with electron donating substituents led to the unprecedented formation of tetrahydropyridines **5** under similar reaction conditions (Table 2, Entries 8–19). The formation of **5** was ascertained from the NMR spectroscopy. As a representative case, in the ^1H NMR of **5**{**3**,**7**} the characteristic H-4 and H-3 protons appeared as doublets at 5.21 and 4.39 ppm ($J = 7.2, 6.9$ Hz), respectively. This revealed the selective formation of tetrahydropyridines **5** and precluded the formation of pyrrole **4** (Figure 4). The structure of all the pyrroles **4** and the tetrahydropyridines **5** was elucidated with the help of NMR spectroscopy.

2,2-Dihydroxy-1-arylethan-1-ones



Enamines

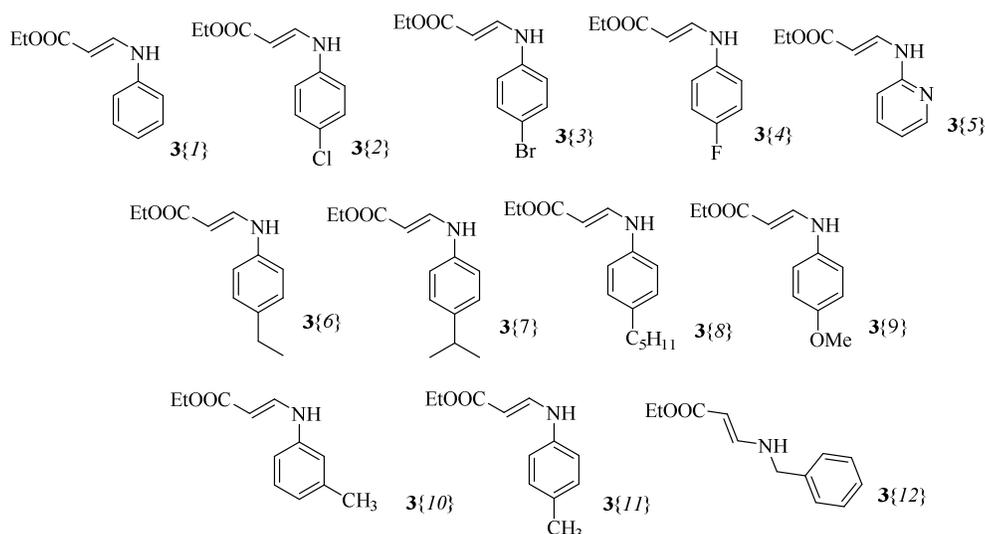
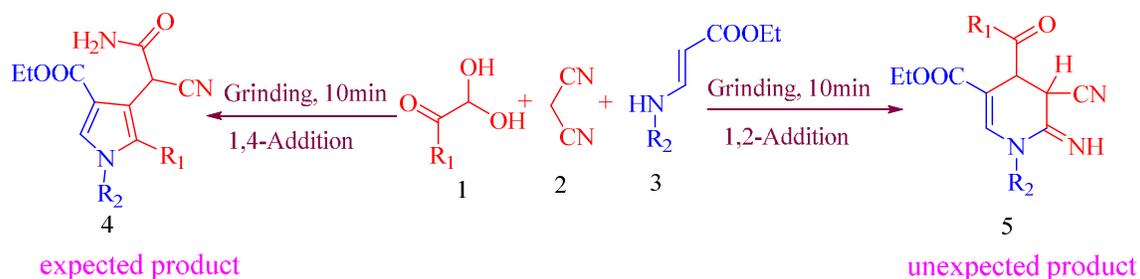


Figure 3 Diversity of reactants



Scheme 2 Synthesis of pyrroles 4 and tetrahydropyridines 5

Table-2 Yield of pyrrole 4 and tetrahydropyridine 5

Entry	Comp	R ₁	R ₂	Yield (%) ^a
1	4{1,1}	C ₆ H ₅	C ₆ H ₅	82
2	4{1,2}	C ₆ H ₅	4-ClC ₆ H ₄	80
3	4{1,3}	C ₆ H ₅	4-BrC ₆ H ₄	84
4	4{1,4}	C ₆ H ₅	4-FC ₆ H ₄	78
5	4{1,5}	C ₆ H ₅	2-Pyridyl	70
6	4{3,1}	4-OMeC ₆ H ₅	C ₆ H ₅	83
7	4{4,1}	4-MeC ₆ H ₄	C ₆ H ₅	82
8	5{1,6}	C ₆ H ₅	4-C ₂ H ₅ C ₆ H ₄	82
9	5{1,7}	C ₆ H ₅	4- ⁱ PrC ₆ H ₄	84
10	5{1,8}	C ₆ H ₅	4-C ₅ H ₁₁ C ₆ H ₄	79
11	5{1,9}	C ₆ H ₅	4-OMeC ₆ H ₄	70
12	5{1,10}	C ₆ H ₅	3-MeC ₆ H ₄	75
13	5{2,9}	4-ClC ₆ H ₄	4-OMeC ₆ H ₄	70
14	5{2,11}	4-ClC ₆ H ₄	4-MeC ₆ H ₄	76
15	5{3,6}	4-OMeC ₆ H ₄	4-C ₂ H ₅ C ₆ H ₄	83
16	5{3,7}	4-OMeC ₆ H ₄	4- ⁱ PrC ₆ H ₄	85
17	5{3,8}	4-OMeC ₆ H ₄	4-C ₅ H ₁₁ C ₆ H ₄	80
18	5{4,6}	4-MeC ₆ H ₄	4-C ₂ H ₅ C ₆ H ₄	75
19	5{5,11}	4-BrC ₆ H ₄	4-MeC ₆ H ₄	72
20	4{1,12}	C ₆ H ₅	-CH ₂ -C ₆ H ₅	79
21	4{2,12}	4-ClC ₆ H ₄	-CH ₂ -C ₆ H ₅	75
22	4{3,12}	4-OMeC ₆ H ₅	-CH ₂ -C ₆ H ₅	80

^aIsolated yields after filtration

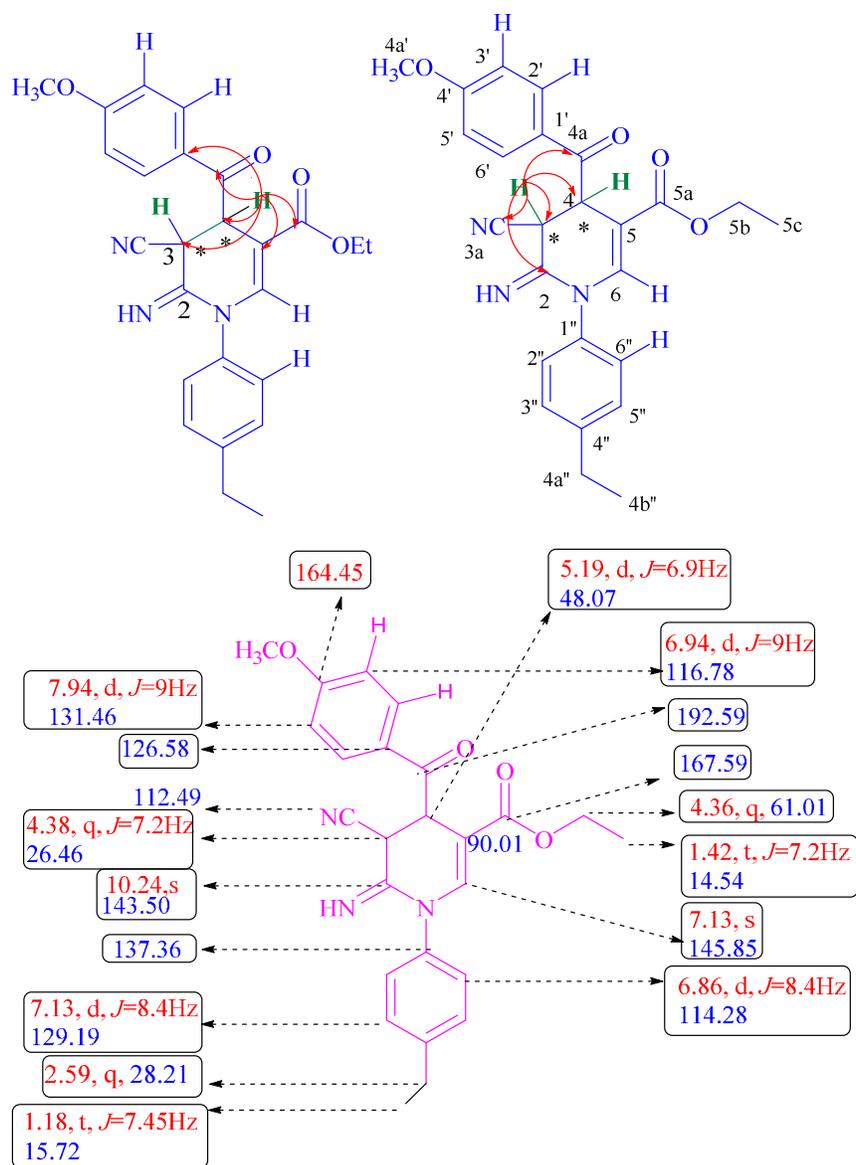
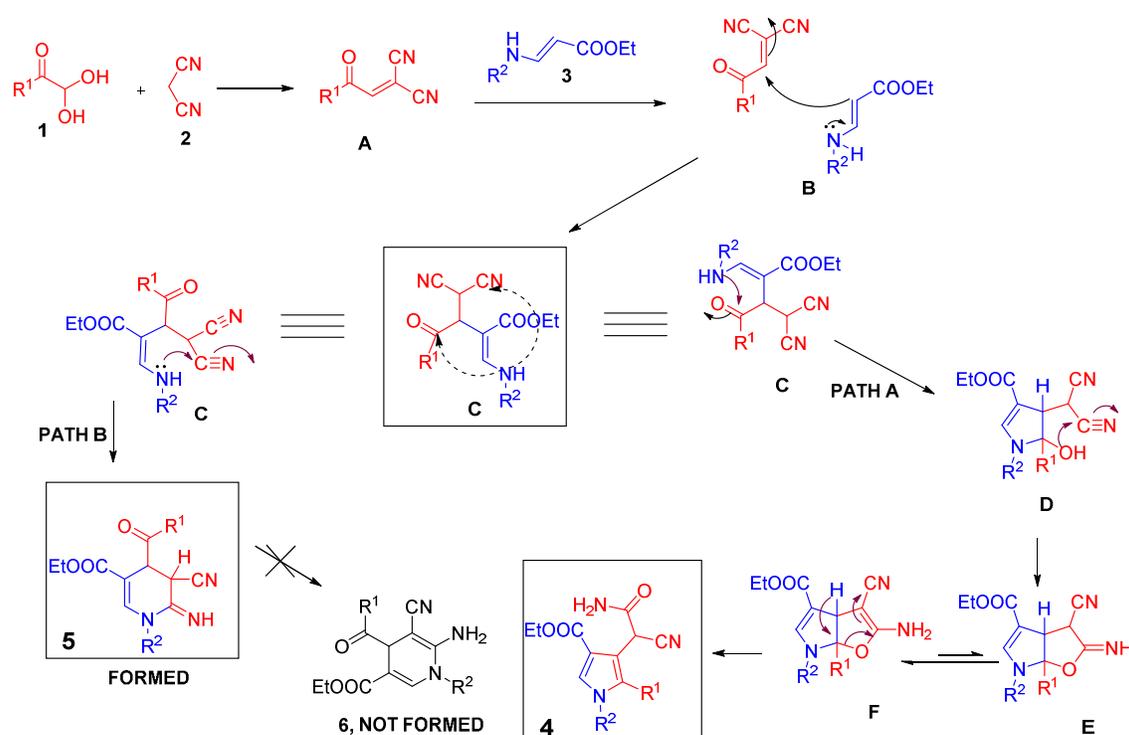


Figure 4 HMB correlations and ^1H and ^{13}C NMR chemical shifts of **5{3,6}**

A plausible mechanism for the formation of **5** is depicted in Scheme 3. Initially, the Knoevenagel condensation of 2,2-dihydroxy-1-arylethan-1-one **1** and malononitrile **2** furnishes 2-(2-oxo-2-arylethylidene)malononitrile **A**. Further, the Michael addition of **A** with enamine **3** affords intermediate **C**. The subsequent intramolecular *N*-cyclization of **C** may presumably involve the secondary amine group and the carbonyl or the nitrile functions *via* path A or B, respectively as shown in Scheme 3. Interestingly, in the cases of *N*-phenyl, *N*-heterocycle and the presence of electron withdrawing substituent in the *N*-phenyl ring in **C**,

cyclization involving the secondary amine group and the carbonyl is favoured and proceeds through path A leading to the formation of intermediate **D**, which then undergoes cyclization to afford **E**. The bicyclic intermediate **E** further undergoes ring opening via intermediate **F** to afford the polysubstituted pyrroles **4**.²⁷ On the other hand, the presence of electron donating substituents in the *N*-phenyl ring of the intermediate **C** favours the cyclization involving the secondary amine group and the nitrile function to afford the tetrahydropyridine **5** via path B. These tetrahydropyridine **5** were reluctant towards tautomerization to afford the dihydropyridine isomer **6**. Furthermore, it is to be noted that albeit the formation of two contiguous stereocentres, the reaction afforded a single diastereomer of the product **5**. The above two factors were evident from the NMR studies.



Scheme 3 Plausible mechanism for formation of tetrahydropyridines and pyrroles

With the above interesting results in hand we then studied the outcome of the above domino reaction in the presence of ethyl (*E*)-3-(benzylamino)acrylate **3**{**12**} with a view to investigate the selectivity involved in the reaction. However, the reaction afforded the expected pyrrole **4** as a sole product under similar conditions (Scheme 2 and Table 2, Entries 20–22). The structure of these pyrroles was elucidated with NMR spectroscopy and in one

representative case *viz.* **4{1,12}**, the structure was further confirmed from the single crystal X-ray studies (Figure 5).

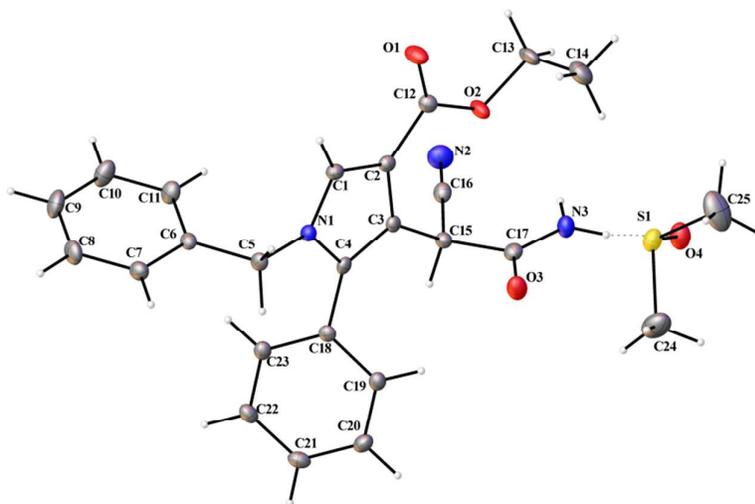
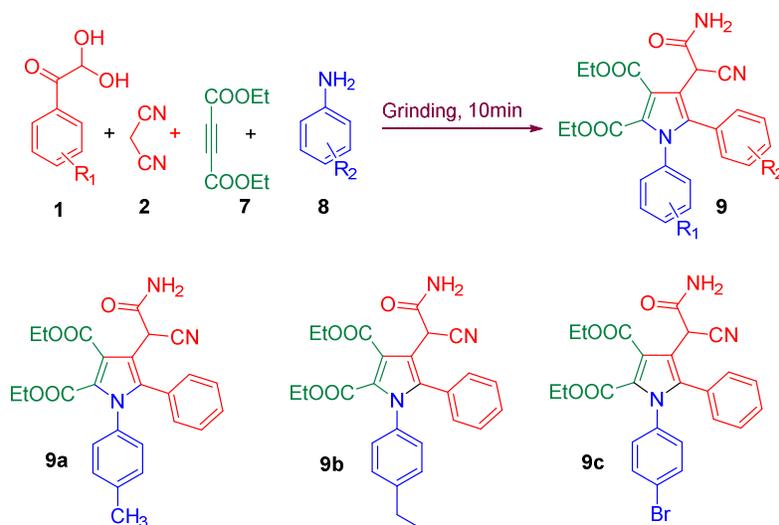


Figure 5 ORTEP diagram of **4{1,12}** (CCDC No.1428613)

It is to be noted that the four-component reactions of 2,2-dihydroxy-1-phenylethan-1-one **1**, malononitrile **2**, diethyl acetylene carboxylate **7** and aromatic amines **8** have been reported by the Feng *et al.*²⁷ This reaction in ethanol under reflux for 30 min led to the formation of pyrroles **9** up to 80% yields. We investigated this reaction under our optimized conditions with a view to study the selectivity involved, *viz.*, whether the reaction affords pyrrole or tetrahydropyridine. Accordingly a mixture of 2,2-dihydroxy-1-phenylethan-1-one **1**, malononitrile **2**, diethyl acetylene carboxylate **7**, aromatic amines **8** was ground well for 10 min (Scheme 4). The reaction afforded pyrroles **9a-c** as the sole products in up to 85% yield. Substrate scope of these reaction conditions were explored with aromatic amines comprising electron withdrawing or electron releasing substituents. However, in all these cases the reaction afforded pyrrole **9** as the sole product.



Scheme 4 Synthesis of fully substituted pyrroles **9a-c**

Conclusions

In conclusion, we have developed a facile method for synthesis of a library of polysubstituted pyrroles and tetrahydropyridines chemoselectively in excellent yields *via* a one-pot, multi-component domino reactions of ethyl (*E*)-3-(4-arylamino) acrylates, 2,2-dihydroxy-1-arylethan-1-ones and malononitrile under solvent and catalyst-free conditions under grinding. The chemoselectivity depended on the substitution in the *N*-phenyl ring of ethyl (*E*)-3-(4-arylamino) acrylates.

Experimental Procedures

General procedure for synthesis of 4 and 5

A mixture of 2,2-dihydroxy-1-arylethan-1-one **1** (1 mmol), malononitrile **2** (1 mmol) and ethyl (*E*)-3-(4-arylamino)acrylate **3** (1 mmol) was taken in a mortar and ground continuously for 10 min. The reaction was monitored by TLC until completion. The syrupy formed was washed with cold ethanol and filtered through the filtration flask to afford the pure product.

Ethyl 4-(2-amino-1-cyano-2-oxoethyl)-1-(4-bromophenyl)-5-phenyl-1H-pyrrole-3-carboxylate **4**{*I*,*3*}

1
2
3 White solid, yield: 84% (394mg), mp: 190–193 °C ^1H NMR (300 MHz, CDCl_3) δ 7.60 (s,
4 1H), 7.42 (d, $J = 8.2$ Hz, 2H), 7.38 – 7.31 (m, 3H), 7.24 – 7.18 (m, 2H), 6.95 (d, $J = 8.2$ Hz,
5 2H), 5.61 (s, 2H), 4.94 (s, 1H), 4.37 (q, $J = 7.1$ Hz, 2H), 1.39 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR
6 (75 MHz, CDCl_3) δ 166.21, 163.54, 137.13, 134.89, 131.84, 130.35, 128.52, 128.28, 128.12,
7 126.71, 121.12, 116.32, 113.99, 112.57, 59.95, 35.70, 13.87 ppm; LC-MS. calcd. m/z 451.05
8 found 452.25 (M+H) $^+$. Anal. Calcd. For $\text{C}_{22}\text{H}_{18}\text{BrN}_3\text{O}_3$ calcd. C, 58.42; H, 4.01; N, 9.29,
9 found C, 58.43; H, 4.03; N, 9.28%.

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17 Ethyl 5-cyano-1-(4-ethylphenyl)-6-imino-4-(4-methoxybenzoyl)-1,4,5,6-tetrahydropyridine-
18 3-carboxylate **5{3,6}**

19
20
21 White solid, yield: 83% (373mg), mp: 125–126 °C ^1H NMR (300 MHz, CDCl_3) δ 10.27 (s,
22 1H), 7.95 (d, $J = 9$ Hz, 2H), 7.21 (s, 1H), 7.13 (d, $J = 8.4$ Hz, 2H), 6.94 (d, $J = 9$ Hz, 2H),
23 6.86 (d, $J = 8.4$ Hz, 2H), 5.21 (d, $J = 6.3$ Hz, 1H), 4.41 (d, $J = 7.2$ Hz, 1H), 4.36 (q, $J = 7.4$
24 Hz, 2H), 2.63 (q, 2H), 1.42 (t, $J = 7.2$ Hz 3H), 1.18 (t, $J = 7.5$ Hz, 3H); ^{13}C NMR (75 MHz,
25 CDCl_3) δ 192.59, 167.59, 164.45, 145.34, 140.50, 137.35, 131.46, 129.18, 129.04, 126.57,
26 116.77, 112.68, 112.49, 90.65, 61.01, 55.67, 48.78, 28.21, 26.42, 15.71, 14.54 ppm; LC-MS.
27 calcd. m/z 431.18, found 432.22 (M+H) $^+$. Anal. Calcd. For. $\text{C}_{25}\text{H}_{25}\text{N}_3\text{O}_4$ calcd. C, 69.59; H,
28 5.84; N, 9.74, found C, 69.57; H, 5.82; N, 9.75%.

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35 Diethyl 4-(2-amino-1-cyano-2-oxoethyl)-5-phenyl-1-(p-tolyl)-1H-pyrrole-2,3-dicarboxylate
36 **9a**

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38
39 White solid, yield: 86% (232mg) mp: 189–190 °C ^1H NMR (300 MHz, CDCl_3) δ 7.29 (dd, $J =$
40 8.3, 2.7 Hz, 4H), 7.18 – 7.14 (m, 2H), 7.04 (d, $J = 7.4$ Hz, 4H), 6.77 (s, 1H), 5.67 (s, 2H),
41 4.89 (s, 1H), 4.37 (dd, $J = 14.2, 7.1$ Hz, 2H), 4.15 (dd, $J = 14.3, 7.1$ Hz, 2H), 2.30 (s, 3H),
42 1.37 (t, $J = 7.1$ Hz, 3H), 1.14 (d, $J = 7.1$ Hz, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 166.79,
43 163.83, 161.53, 138.98, 137.30, 134.08, 131.07, 130.70, 129.46, 129.20, 128.59, 128.15,
44 127.57, 116.32, 114.24, 111.71, 61.95, 61.30, 36.25, 21.18, 14.12, 13.80 ppm; LC-MS. calcd.
45 m/z 459.18 found, 458.07 (M-H) $^-$. Anal. Calcd. For $\text{C}_{26}\text{H}_{25}\text{N}_3\text{O}_5$ calcd. C, 67.96; H, 5.48; N,
46 9.14, found C, 68.47; H, 5.21; N, 10.42%.

53 ASSOCIATED CONTENT

54 Supporting Information

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3 Experimental details and Spectroscopic characterization of 4a-j, 5a-l and 9a-c and X-ray
4 crystallographic information for compound 4{1,12} are available in supporting information.
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32 NOTES

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Graphical abstract

