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Article

Design and SAR of withangulatin A analogs that act as covalent TrxR inhibitors through Michael addition reaction showing potential in cancer treatment

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ABSTRACT

The thioredoxin system plays an important role in cancer cells. Inhibiting thioredoxin reductase (TrxR) has emerged as an effective strategy to selectively target cancer cells. Withangulatin A (WA), a natural product extracted from the whole herb of *Physalis angulata* L. (Solanaceae) exhibits potent anticancer activity and other diverse pharmacological activities. To improve activity and targeting, we designed and prepared 41 semisynthetic analogs of WA. Biological evaluation indicated that the most promising compound 13a displayed the most significant effect on HT-29 cells (human colon cancer cells) (IC₅₀ = $0.08 \,\mu$ M). A structure-activity relationship study indicated that α , β -unsaturated ketones and ester are necessary groups, allowing 13a to undergo Michael addition reactions with mercaptan and selenol. LC-MS analysis confirmed that 13a modified selenocysteine 498 (U) residues in the redox centers of TrxR, resulting in enzyme inhibition. Therefore, compound 13a acts a novel TrxR inhibitor and may be a promising candidate for cancer intervention.

1 INTRODUCTION

In the body and cells of humans, reactive oxygen species (ROS) are natural byproducts of oxidative respiration and can function in signal transduction and pathogen clearance during innate immune responses. Compared with noncancerous cells, cancer cells have been reported to harbor increased levels of ROS.¹ ROS usually maintain a high but appropriate level in various cancer cells, primarily due to their active metabolism in response to oncogenic signals.² In fact, cancer cells take advantage of this appropriate level of oxidative stress for several significant biological activities, such as proliferation, angiogenesis, and metastasis.³ However, sustained high levels of ROS can irreversibly damage DNA and lipids and ultimately cause cancer cell apoptosis.⁴ The alteration of the redox environment has been suggested to be the 'Achilles' heel' of cancer cells and could be exploited to target them for cancer therapy. Thus, there is increasing interest in modulating cellular redox signaling as a potential strategy for various cancer treatments.⁵

The major source of ROS is mitochondrial metabolism within the cells. The cascade of reversible redox reactions of ROS with the thiol in thiocysteine (Cys) side chains, the selenol in selenocysteine (Sec) side chains, or the sulfide in methionine (Met) side chains make ROS important messenger molecules at the physiological level that regulate diverse redox signaling pathways involved in biological processes.⁵ The thioredoxin (Trx) system, one of the key antioxidant systems, is composed of thioredoxin reductase (TrxR), Trx, and NADPH and regulates numerous cellular signaling pathways involved in cell proliferation and survival.⁶⁻⁸ Mammalian TrxRs are unique large selenoproteins (MW, 114000 Dalton or larger) containing a unique catalytically active selenolthiol/selenenylsulfide in the conserved C-terminal sequence -Gly-Cys-Sec-Gly.^{9,10} Trxs from mammalian cells are proteins with a MW of 12000 Dalton that contain a conserved pair of cysteine residues (-CGPC-) in the active site and act as the major disulfide

reductase responsible for maintaining cytosolic proteins in their reduced state.¹¹ TrxR is the only known physiological enzyme that catalyzes the reduction of oxidized Trx. The Trx system is often overexpressed in most tumors.^{12,13} The level of TrxR in cancer cells is often increased by 10-fold or even more compared with that in normal tissues, and cancer proliferation seems to be crucially dependent on an active Trx system, making it a potential target for anticancer drug discovery.¹⁴

Natural products and their derivatives are an invaluable source of therapeutic agents and have driven pharmaceutical discovery over the past century. The diversity of natural products continues to provide a unique source of bioactive lead compounds for drug development.^{15,16} Many compounds used in cancer chemotherapy are derived from natural products, such as paclitaxel, camptothecin, and etoposide. *Physalis angulata L* is a branched annual shrub that belongs to the Solanaceae family and is distributed worldwide throughout tropical and subtropical regions.¹⁷ It is a common traditional Chinese medicine and has been used for the treatment of conditions such as tumors, hepatitis, rheumatism, nephronia and ulcers.¹⁸ Withangulatin A (Figure 1, WA, 1), a group of C28 steroids, is a natural withanolide isolated from *P. angulata* L.¹⁹ WA can exert anticancer effect by changing cell morphology²⁰ and inhibiting topoisomerase.¹⁷ Its analogues physapubescin B have been reported to induce cell cycle G2/M arrest in combination with GSH.²¹ Previous studies have shown that WA analogues possesses potent anticancer activity by consuming GSH and generating ROS via the α , β -unsaturated ketones in ring A.^{21,22} The evidence suggests that WA can combine mercaptan and selenol through Michael addition reaction. Despite the well-documented anticancer activity of WA, its target specificity and efficacy are moderate. Hence, it is necessary to improve the potency and increase the targeting of **WA** for future development. In this study, we report the preparation and biological evaluation of 41 semisynthetic analogs of this fascinating class of cytotoxic agents. The cytotoxicity of many derivatives was higher than that of the parent,

and some of the compounds showed good TrxR inhibition. Compound 13a acts as a novel TrxR
 inhibitor and may be a promising candidate for cancer intervention.

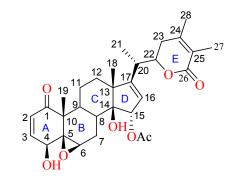
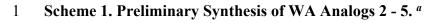


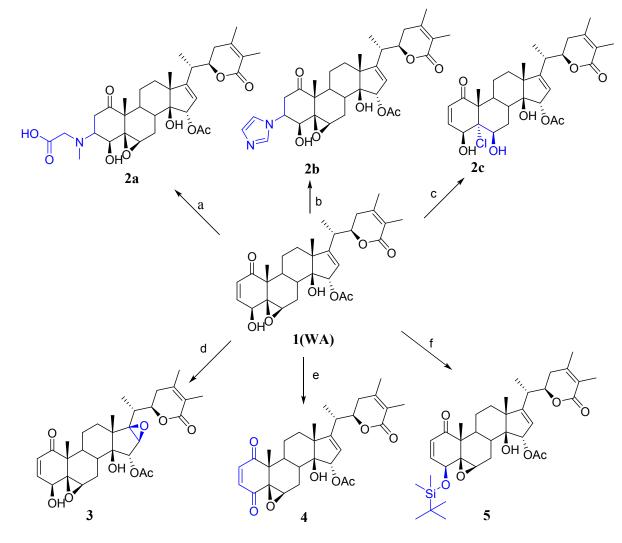
Figure 1. The natural product Withangulatin A (WA, 1).

RESULTS AND DISCUSSION

6 The initial modification and biological evaluation of WA analogs 2-5

Studies have asserted that a 2-ene-1-one and a 5 β , β -epoxide are essential for the anticancer activity of withanolides.²³⁻²⁵ Structure-activity relationship (SAR) analysis of the activity of withanolide E against renal carcinoma has been performed.²⁶ 2-Ene-1-one easily reacts with mercaptoyl through Michael addition, and it is an essential group to induce cells to produce ROS. We initiated (Scheme 1) our study by destroying the α , β -unsaturated ketone system (2a, 2b) and the epoxide (2c). The biological results (Table 1) asserted that the α , β -unsaturated ketones and the epoxide are the essential pharmacophores that should not be changed. When we kept the groups and modified the double bond C16=C17 to an epoxide (3, the crystal structure and the relatedcrystal data of **3** is shown in Figure S1 and Table S1, respectively) or changed the hydroxyl group to a carbonyl group (4), the cytotoxic activity was not improved. A slight change in the biological activity was observed when the only modification was on the hydroxy group (5). This finding prompted us to perform the next modification focused on adding different substituents to the hydroxy position.





^{*a*} Reagents and conditions: (a) NH(CH₃)CH₂COOH, anhydrous CH₂Cl₂, rt, 2h; (b) imidazole, anhydrous CH₂Cl₂, rt, 2h; (c) HCl solution in ether, anhydrous CH₂Cl₂, rt, 1h; (d) mCPBA anhydrous CH₂Cl₂, 0 °C, 12h; (e) MnO₂, EtOAc, CHCl₃, rt, 8h; (f) TBDMSCl, imidazole, anhydrous CH₂Cl₂, rt, 12h.

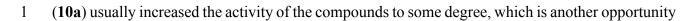
Compd	Yield (%)	HT-29	HCT-116	HepG2	MCF-7
1(WA)	NP ^b	2.48 ± 0.23	2.37 ± 0.22	2.85 ± 0.12	6.32 ± 0.74
2a	85	> 20.00	> 20.00	> 20.00	> 20.00
2 b	85	> 20.00	> 20.00	> 20.00	> 20.00
2c	60	> 20.00	> 20.00	> 20.00	> 20.00
3	78	4.55 ± 0.66	5.04 ± 1.02	4.66 ± 0.45	9.36 ± 1.09
4	81	3.50 ± 0.08	3.11 ± 0.10	6.05 ± 0.10	> 20.00
5	41	1.06 ± 0.04	1.77 ± 0.08	3.01 ± 0.21	6.14 ± 0.78

^{*a*} Cell viability was determined by the MTT assay after 24 h of treatment.

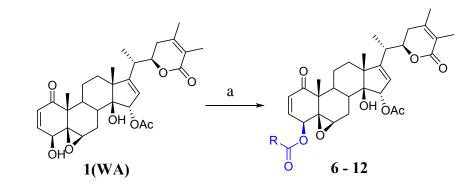
^b Natural product, 58 mg from 1.0 kg of the dried whole *P. angulata* L. var. *villosa* plant.

4 Esterification of WA and biological evaluation of analogs 6-12

The next step focused on modifications at position 4-OH. Compounds 6-12 were synthesized as outlined in Scheme 2. These compounds are all the esterification products of WA. After screening the cytotoxic activity against four cancer cell lines (Table 2), most compounds were found to be more active than their parent, and three compounds, 8a, 9a and 10a attracted our attention. 8a having an α , β -unsaturated carbonyl group, **9a** with a dimethylcarbamate group, and **10a** having a boc amine moiety were the most potential compounds in this series (Table 2). These results suggest that different substituents at position 4 significantly affected the activity of the compound in cancer cells. The analysis of specific structural segments revealed that the introduction of an α , β -unsaturated ketone at 4-OH (8a) greatly improved the bioactivity of the compound against the four cell lines. In addition to other factors, this result suggested that the Michael acceptor, an α , β -unsaturated ketone, in the chain more favorably formed covalent interactions than did the ketone in the skeleton of WA, presenting an opportunity for us to introduce new types of modifications. On the other hand, we noticed that the introduction of carbamates (9a) or boc amino acid fragment



- 2 to optimize our compounds.
- 3 Scheme 2. Synthesis of WA Analogs 6-12. *a*



(a) For 6: CDI, CH₃OH, anhydrous CH₂Cl₂, rt, 8h; For 7a: (CH₃CO)₂O, Et₃N, anhydrous CH₂Cl₂,
rt, 12h; For 7b: (CH₃CH₂CO)₂O, Et₃N, anhydrous CH₂Cl₂, rt, 16h; For 8a, 10, 11 and 12: RCOOH, DMAP, EDCI, Et₃N, anhydrous CH₂Cl₂, 0 °C, 25h; For 9: N(CH₃)₂COCl, Et₃N, anhydrous
CH₂Cl₂, 0 °C, 12h.

9 Table 2. Cytotoxic Activity (IC₅₀ Values in µM) of 6-12 against Four Cell Lines. ^a

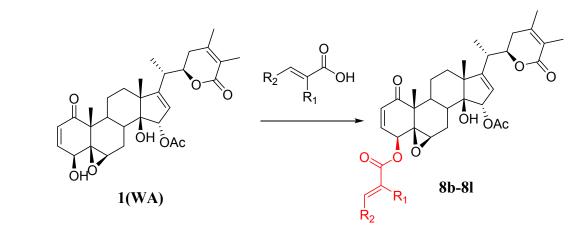
Compd	R	Yield (%)	HT-29	HCT-116	HepG2	MCF-7
6	OCH ₃	72	1.79 ± 0.04	1.95 ± 0.07	2.81 ± 0.11	4.33 ± 0.08
7a	CH_3	85	0.80 ± 0.07	0.60 ± 0.09	1.57 ± 0.15	3.05 ± 0.24
7b	CH ₂ CH ₃	76	1.08 ± 0.12	0.57 ± 0.11	1.66 ± 0.03	4.81 ± 0.11
8 a	CH=CH ₂	71	0.78 ± 0.15	0.33 ± 0.07	0.52 ± 0.09	1.06 ± 0.12
9a	$N(CH_3)_2$	75	1.01 ± 0.08	0.67 ± 0.06	1.06 ± 0.14	2.10 ± 0.13
10a	CHNHBoc	64	0.63 ± 0.04	0.47 ± 0.06	0.64 ± 0.09	2.16 ± 0.34
11	\mathbf{r}^{s}	37	2.39 ± 0.11	3.87 ± 0.45	2.91 ± 0.12	4.08 ± 0.19
12	CH ₂ CH ₂ Ph	32	4.05 ± 0.33	5.84 ± 0.41	5.76 ± 0.63	8.18 ± 0.34

a Cell viability was determined by the MTT assay after 24 h of treatment.

Esterification of WA with an α , β -unsaturated carboxylic acids and the biological evaluation of 8b-8l

Based on the above findings, we aimed to study the SAR by modifying the ketone to an α , β -unsaturated ketone. Compounds 8b-8l (Scheme 3) were prepared with different acyl chlorides catalyzed by DMAP and EDCI in anhydrous CH₂Cl₂. The inhibitory activities against four cell lines, namely, HT-29, HCT-116, HepG2, and MCF-7, were evaluated. As shown Table 3, the inhibitory activity of almost all the substituent groups at the β position of the ketone were better than that of the natural product **WA**. The activity of **8b** with a methyl group at the β position was better than that of 81 with a methyl group at the α position. Among these analogs, the compound with the most potential was 8e, which contained an ester at the end.

Scheme 3. Synthesis of WA Analogs 8b-8l.^a



^a Reagents and conditions: DMAP, EDCI, Et₃N, anhydrous CH₂Cl₂, 0 °C, 25h.

Table 3. Cytotoxic Activity (IC₅₀ Values in µM) of 8a-8l against Four Cell Lines.^a

Compd	R ₁	R ₂	Yield (%)	HT-29	HCT-116	HepG2	MCF-7
8 a	Н	Н	71	0.78 ± 0.15	0.33 ± 0.07	0.52 ± 0.09	1.06 ± 0.12
8b	Н	CH ₃	60	0.37 ± 0.06	0.23 ± 0.05	0.30 ± 0.08	0.53 ± 0.09
8c	Н	(CH ₃) ₂	68	0.46 ± 0.02	0.28 ± 0.02	0.26 ± 0.04	0.75 ± 0.11

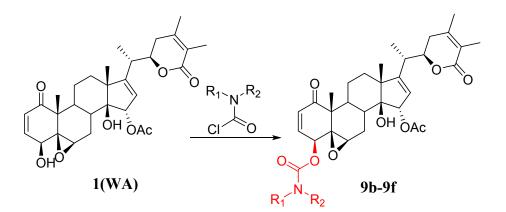
8d	Н	CH=CHCH ₃	74	1.42 ± 0.22	1.35 ± 0.09	1.52 ± 0.12	2.62 ± 0.13
8e	Н	COOCH ₂ CH ₃	79	0.23 ± 0.01	0.25 ± 0.08	0.24 ± 0.03	0.28 ± 0.04
8f	Н	OCH ₂ CH ₃	69	0.59 ± 0.07	0.67 ± 0.03	0.57 ± 0.05	1.21 ± 0.12
8g	Н	Ď	71	0.45 ± 0.06	0.43 ± 0.03	0.31 ± 0.04	0.76 ± 0.17
8h	Н	$\mathbb{L}^{s}\!$	59	0.43 ± 0.10	0.38 ± 0.04	0.53 ± 0.03	0.63 ± 0.10
8 i	Н	Ph	37	0.77 ± 0.10	1.23 ± 0.06	0.70 ± 0.21	1.71 ± 0.19
8j	Н	4-F-Ar	53	0.72 ± 0.02	1.03 ± 0.14	0.61 ± 0.08	1.58 ± 0.30
8k	Н	3-CF ₃ -Ar	67	0.81 ± 0.25	0.65 ± 0.05	0.42 ± 0.01	1.39 ± 0.21
81	CH_3	Н	47	1.12 ± 0.17	1.34 ± 0.13	1.45 ± 0.12	2.11 ± 0.29

"Cell viability was determined by the MTT assay after 24 h of treatment.

Carbamate modification of WA and biological evaluation of 9b-9f

However, the carbamate group is very important in drug design and discovery. By introducing this bioactive pharmacophore to WA, **9a - 9f** (Scheme 4) were synthesized. The introduction of carbamate groups, regardless of whether the group was added to an opening alkyl group (**9a - 9c**) or to an alkyl ring (**9d - 9f**), improved the cytotoxic activity (Table 4) of the designed compounds. Among these compounds, compound 9c, with the ethyl carbamate, showed the best activity.

8 Scheme 4. Synthesis of WA Analogs 9b-9f. ^a



^a Reagents and conditions: DMAP, Et₃N, N₂, anhydrous CH₂Cl₂, 0 °C, 12h.

Compd	R ₁	R ₂	Yield (%)	HT-29	HCT-116	HepG2	MCF-7
9a	CH ₃	CH ₃	75	1.01 ± 0.08	0.67 ± 0.06	1.06 ± 0.14	2.10 ± 0.13
9b	CH ₃	$\mathrm{CH}_{2}\mathrm{CH}_{3}$	70	1.78 ± 0.23	1.97 ± 0.15	2.37 ± 0.55	1.79 ± 0.14
9c	$\mathrm{CH}_2\mathrm{CH}_3$	CH ₂ CH ₃	73	0.75 ± 0.17	0.43 ± 0.08	0.69 ± 0.12	1.21 ± 0.11
9d		_/	52	1.12 ± 0.13	0.89 ± 0.09	1.23 ± 0.09	2.14 ± 0.23
9e	Ļ	J	39	1.38 ± 0.04	0.73 ± 0.01	0.41 ± 0.01	4.31 ± 0.70
9f			42	0.99 ± 0.22	0.78 ± 0.11	0.67 ± 0.09	3.89 ± 0.63

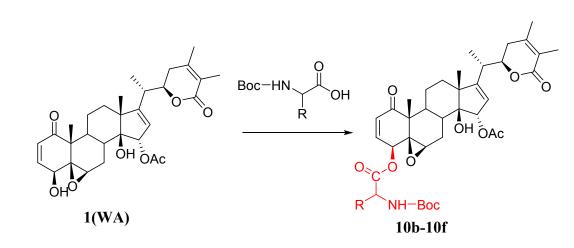
1 Table 4. Cytotoxic Activity (IC₅₀ Values in μ M) of 9a-9f against Four Cell Lines. ^{*a*}

^{*a*} Cell viability was determined by the MTT assay after 24 h of treatment.

Amino group modification of WA and biological evaluation of 10a-10f

Amino groups are important building blocks in drug development and introducing amino acids into anticancer molecules can improve their selectivity to cancer cells, enhance the lipid solubility of drugs, and alleviate the toxicity of drugs to cells.⁷ Compounds having amino groups, such as Docelaxel, Vancomycin, Polymyxin B, Daptomycin, 5-fluorouracil, etc., have superior anticancer activity compared with the original mother nucleus. Thus, amino acid groups were introduced in the modification of WA, and 10a - 10f (Scheme 5) were synthesized. The biological evaluation indicated that amino group substitutions improved the cytotoxic activity of WA. Among these compounds (Table 5), **10e**, the Boc-Leu-modified WA analog showed the greatest potential.

12 Scheme 5. Synthesis of WA Analogs 10b-10f. *a*



^a Reagents and conditions: DMAP, EDCI, Et₃N, anhydrous CH₂Cl₂, 0 °C, 25h.

Table 5. Cytotoxic Activity (IC₅₀ Values in μ M) of 10a-10f against Four Cell Lines. ^a

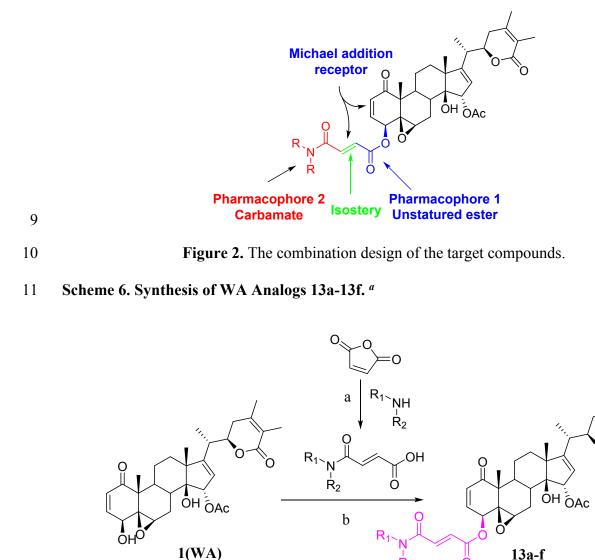
Compd	R	Yield (%)	HT-29	HCT-116	HepG2	MCF-7
10a	Н	64	0.63 ± 0.04	0.47 ± 0.06	0.64 ± 0.09	2.16 ± 0.34
10b	CH ₃	68	1.79 ± 0.04	1.95 ± 0.07	2.81 ± 0.11	4.33 ± 0.08
10c	CH(CH ₃) ₂	67	0.80 ± 0.07	0.60 ± 0.09	1.57 ± 0.15	3.05 ± 0.24
10d	CH(CH ₃) CH ₂ CH ₃	71	1.08 ± 0.12	0.57 ± 0.11	1.66 ± 0.03	4.81 ± 0.11
10e	CH ₂ CH (CH ₃) ₂	73	0.78 ± 0.15	0.33 ± 0.07	0.52 ± 0.09	1.06 ± 0.12
10f		41	1.01 ± 0.08	0.67 ± 0.06	1.06 ± 0.14	2.10 ± 0.13

^a Cell viability was determined by the MTT assay after 24 h of treatment.

Combined modifications and biological evaluation of 13a-13f

Given the above SAR studies, we concluded that the motif of an additional unsaturated ketone,
which may act as a Michael addition receptor, and the alkyl carbamate groups improved the
bioactivity of WA. Considering the bioisosterism principle of drug design in medicinal chemistry,
a double bond was used to link the ketone and the carbamate groups together (Figure 2, Scheme
In accordance with this rule, we found that the bioactivity of all the analogs (Table 6) was not
only stronger than that of WA but also stronger than that of the analogs with just one of the ketone

or the carbamate groups. Among **13a-13f**, **13a** displayed the highest potency against HT-29 cells with an IC₅₀ value of 0.08 μ M, which was more than a 30-fold increase compared to that of the starting compound WA. For the human normal cells, human normal hepatocytes L02 and human tubular epithelial cells HK-2, the experiment proved that WA is not so sensitive to the normal cells and the cancer cells as well in the same concentration (Table 7). But for 13a, it is safe enough to normal cells when it is effective to the cancer cells at the concentration of 0.08 µM for HT-29, which is 66 times low for that of L02, and 56 times low for HK-2 (Table 7). The results show that 13a contrast WA has better safety at effective concentration.



ACS Paragon Plus Environment

13a-f

^a Reagents and conditions: (a) Et₃N, anhydrous CH₂Cl₂, rt, 1h; (b) DMAP, EDCI, Et₃N,

2 anhydrous CH_2Cl_2 , 0 °C, 25h.

Table 6. Cytotoxic Activity (IC₅₀ Values in μ M) of 13a-13f against Four Cell Lines. ^{*a*}

Compd	R ₁	R ₂	Yield (%)	HT-29	HCT-116	HepG2	MCF-7
13 a	CH ₃	CH ₃	68	0.08 ± 0.02	0.13 ± 0.03	0.11 ± 0.01	0.33 ± 0.05
13b	CH ₂ CH ₃	CH ₂ CH ₃	56	0.30 ± 0.06	0.25 ± 0.02	0.31 ± 0.04	0.45 ± 0.06
13c	$CH(CH_3)_2$	$CH(CH_3)_2$	64	0.19 ± 0.04	0.35 ± 0.03	0.36 ± 0.05	0.62 ± 0.13
13d	\	_/	47	0.25 ± 0.06	0.20 ± 0.04	0.29 ± 0.03	0.38 ± 0.02
13e	Ĺ		61	0.37 ± 0.02	0.16 ± 0.04	0.31 ± 0.05	0.61 ± 0.07
13f	Lo	7	70	0.43 ± 0.07	0.46 ± 0.07	0.37 ± 0.05	0.76 ± 0.15

^{*a*} Cell viability was determined by the MTT assay after a 24 h treatment.

5 Table 7. Cytotoxic Activity (IC₅₀ Values in μ M) of WA, 13a and myricetin against six Cell 6 Lines. ^{*a*}

·	Lincy.						
	Compd	L02	НК-2	HT-29	HCT-116	HepG2	MCF-7
	WA	8.21 ± 0.76	6.41 ± 0.66	2.48 ± 0.23	2.37 ± 0.22	2.85 ± 0.12	6.32 ± 0.74
	13 a	5.25 ± 0.49	4.53 ± 0.37	0.08 ± 0.02	0.13 ± 0.03	0.11 ± 0.01	0.33 ± 0.05
	myricetin	> 20.00	> 20.00	> 20.00	> 20.00	> 20.00	> 20.00

^{*a*} Cell viability was determined by the MTT assay after a 24 h treatment.

8 Inhibition of TrxR in both cell-free and cell-based assays

9 After several SAR studies, 13a was shown to display the highest cytotoxicity against the four
10 cancer cell lines; thus, we addressed the possible cellular target of the compound. The expression
11 of TrxR in four cancer cells and two normal cells was detected using Western blot (Figures S2).
12 The results showed that the TrxR expression levels of HT-29, HCT-116 and HepG2 were much
13 higher than those of MCF-7, L02 and HK-2 cells. Meanwhile, the cytotoxic activity of WA and its

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derivatives against several kinds of cells was tested and suggested that the cells with high expression of TrxR were more active than those with low expression levels (Table 7). The core feature of **WA** is the presence of an α , β -unsaturated ketone moiety in ring A, which is the key structural motif of many reported TrxR inhibitors.²⁸⁻³⁰ We speculated that **13a** might be a novel inhibitor of TrxR after using PharmMapper to screen targets of the potential protein (Figures S3).³¹ A DTNB reduction assay was used to determine the activity of TrxR. First, compounds with good cytotoxic activity were selected for preliminary screening of TrxR inhibitory activity, as a result, 13a showed the best inhibitory activity on TrxR (Figure 3A). Then, in vitro experiments on enzymes were performed. As shown in Figure 3B, WA clearly inhibited TrxR with an IC₅₀ value of approximately 2.64 μ M (line with closed squares), and **13a** clearly inhibited TrxR with an IC₅₀ value of approximately 0.85 μ M (line with closed squares). Trx was weakly inhibited by both compounds (IC₅₀ > 2.50 μ M), suggesting that WA and 13a inhibited TrxR much more than Trx. We then examined the relationship of the compounds to some intracellular (HT-29) enzymes involved in oxidative stress. Glutathione reductase (GR) is a homolog of TrxR with a similar overall structure. Like TrxR, glutathione peroxidase (GPx) is a Sec-containing enzyme. Both GR and GPx are closely related to ROS. As shown in Figures 3C and 3D, the inhibition of TrxR induced by WA and 13a was much higher than that of the other enzymes in HT-29 cells. The TrxR inhibition activity was tested in 30 min, 60 min and 90 min, respectively (Figure 3E). Comparing the inhibitory activity of compound **8a**, which did not have dimethylformamide group, the TrxR inhibitory activity of **13a** is ten times stronger than that of **8a**. It means that the dimethylformade group is the necessary bioactivity group for inhibiting TrxR. When the double bond of the acrylamide group in 13a is reduced as in 13a' (Figures S4), the TrxR inhibition bioactivity dropped three to four times. It suggested that the double bond is the indispensable group contributing to the

TrxR inhibition activity. All the results indicated that 13a had stronger inhibitory activities than
 did WA inside and outside the HT-29 cells. Taken together, the results indicated that 13a
 selectively targeted and inhibited TrxR in both cell-free and cell-based assays.

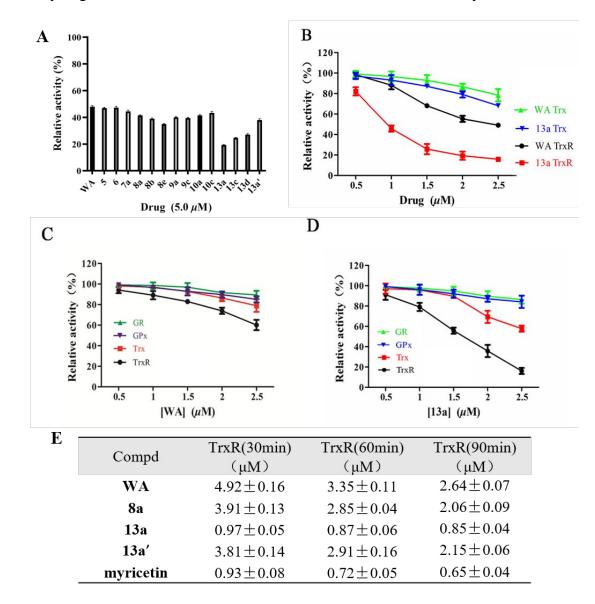


Figure 3. Enzyme and protein inhibition. All activity values are expressed as the percentage of the 6 control. (A) TrxR inhibition activity by the compounds at 5.0 μ M. (B) Inhibition of TrxR and Trx 7 induced by **WA** and **13a**. (C) Inhibition of TrxR, Trx, GR and GPx induced by WA in HT-29 cells.

1 (D) Inhibition of TrxR, Trx, GR and GPx induced by 13a in HT-29 cells. (E) Inhibition of TrxR
2 by different incubation time (30min, 60min, 90min).

5 Kinact and Ki

6 The reaction kinetic parameters including the covalent bond formation (K_i), the rate constant 7 for irreversible inactivation (k_{inact}) was determined and shown in Figure 4. Comparing with the K_i 8 of **WA**, **13a** has higher affinity, which is about 10 times of **WA**. For the K_{inact} of **13a** and **WA**, it 9 showed that **13a** had better covalent response capacity than **WA**. In general, the kinetic parameters 10 showed that **13a** is a better covalent bond inhibitor of TrxR than **WA**.

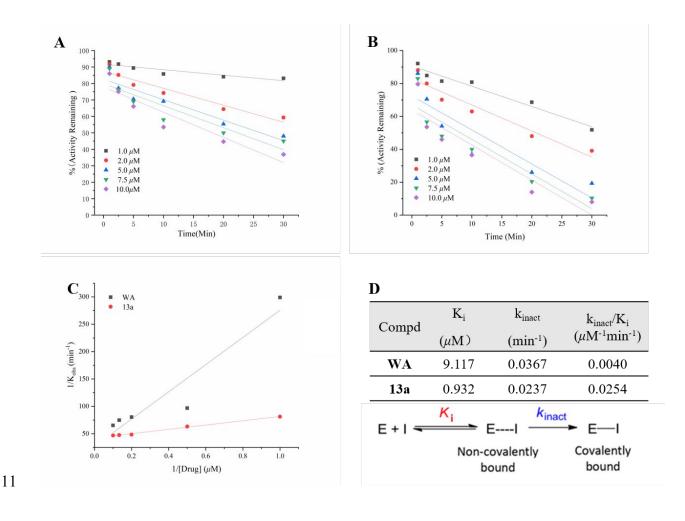


Figure 4. The kinetics of the inhibitors WA and 13a to TrxR. (A) The reactive velocity of WA on TrxR. (B) The reactive velocity of 13a on TrxR. (C) Double-reciprocal plot of K_{obs} and concentration of WA and 13a. (D) Ki and K_{inact} of WA and 13a on TrxR.

Molecular modeling. Since 13a showed relatively good TrxR inhibitory activity, molecular docking experiments of **13a** and the starting compound WA were conducted to study whether there was a direct interaction between the compound and TrxR. In addition, 8e and 9c, the compounds with the highest potential in their respective series, were included in a docking study to interpret the interaction mechanisms between the designed compounds and TrxR. A covalent docking study is performed using Schrödinger software. The crystal structure of rat TrxR1 was obtained from the Protein Data Bank (PDB ID: 3EAN). Given Michael reaction mechanics, Sec498 is believed to be the Michael donor.^{32,33} Therefore, we initially used Sec498 as the binding residue to perform covalent docking. However, a persistent error remained when Sec was used for docking because Sec is not a standard amino acid. Therefore, we mutated Sec to Cys on the basis that Sec is a cysteine analog with a selenium-containing selenol group in place of the sulfur-containing thiol group, which will have a minute influence on the docking result. The molecular docking results and the docking interaction methods are presented in Figure 5, Figure S5 and Table S2, respectively. Through the combined analysis of the docking score and the cytotoxic activity, the following two essential conclusions were reached. One conclusion is that the Michael addition can take place at the β -position of the α , β -unsaturated carbonyl in ring A of WA (position 3; Figures 5A, 5C, 5D and 5F), the β -position of the α , β -unsaturated carbonyl in the side chain of 4-OH (position 3', Figures 5B and 5E) and the α -position of the α , β -unsaturated carbonyl in the side chain of 4-OH (position 2', Figure 5G). The other conclusion is that the interaction with the epoxy group is important (Table S2); for example, in 1 from Figure 4A, A494 interacts with the epoxy,

and Cys498 interacts with 8e (Figure 5B). Moreover, Cys498 interacts with 13a (Figure 5E), which is in agreement with the classic SAR studies that epoxy at positions 5 and 6 is an essential group for the bioactivity of withanolides.²⁴ These basic conclusions indicate that the α , β -unsaturated chain at 4-OH from compound 13a is necessary. The Michael donor, residue Cys498, is preferred to react with the α , β -unsaturated chain at 4-OH assisted by the reaction between the epoxy and the amine from Cys498. Once the covalent bond formed, which pushed the whole 13a inward to the gap between chains A and B, the inner residue A472 pulled the 14-OH group via hydrogen bonding interactions (Figure 5E, Table S2). In contrast, when Michael addition took place at position 3 or 2', the corresponding interactions could not form, and the docking scores of the compounds were smaller, especially at position 2' (docking score: -3.816), than those of the compounds in which docking occurred at position 3' (docking score: -5.614). Thus, **13a** has a strong binding interaction with TrxR through a covalent bond, which is consistent with the cytotoxic activity studies.

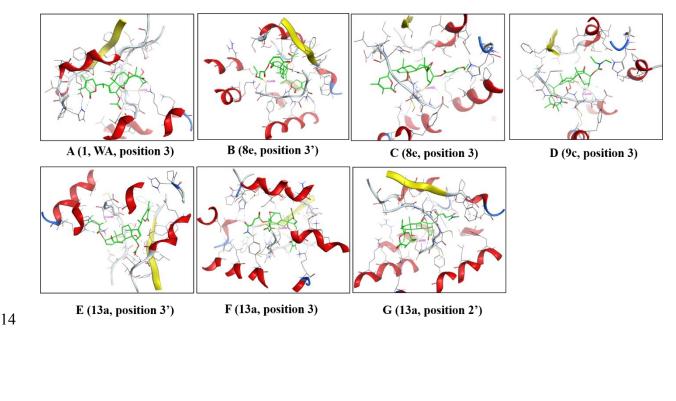
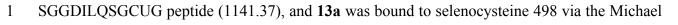


Figure 5. The covalent docking adducts of TrxR (PDB: 3EAN mutated Sec498 to Cys498) and 1(A), 8e (B and C), 9c (D) and 13a (E, F and G). The number in the brackets represent the position in Figures 2 and 4 on the compounds where Michael addition occurred.

Identification of the covalent adduct between 13a and TrxR. Molecular docking suggested that 13a might modify the reduced enzyme by forming a covalent adduct with the nascent selenol/thiol of the active site. To confirm this hypothesis, we preincubated the reduced enzyme with a large excess of **13a** (20-fold) and then digested the protein with trypsin. By analyzing the mass of the peptides, we identified the modified residue. Usually, TrxR will produce the classic reactive peptides WGLGGTCVNVGCIPK (53-67), RSGGDILQSGCUG (487-499) and SGGDILQSGCUG (488-498), and these peptides were identified by the protein mass of the control, as shown in Table 8 and Figure 6. Figure 6A shows the control tryptic ion at m/z 571.6876, with distinctive isotope patterns and confirmed by the critical b and y ions, which were identified as a doubly charged ion of the SGGDILQSGCUG peptide (residues 488–499 in TrxR). The MS¹ of the peptide SGGDILQSGCUG (residues 488-499 in TrxR) was identified through a single charged ion with m/z 1142.3667 (Figure 6B) and a doubly charged ion with m/z 571.6870 (Figure 6C). Figure 6D presents the MS spectrum of CID products from the control tryptic ion at m/z649.7377, which was identified as a doubly charged ion of the RSGGDILQSGCUG peptide (residues 487–499 in TrxR). In the reaction sample, we identified adducts of **13a** and the peptide RSGGDILQSGCUG or SGGDILQSGCUG. The tryptic ion at m/z 976.4026 in Figure 6E was identified as a doubly charged ion with one 13a (MW, 651.75) adducted to the RSGGDILQSGCUG peptide (MW, 1297.47, residues 487-499 in TrxR). In Figure 6F, the MS/MS spectrum of CID products from the tryptic ion at m/z 898.4107 is shown. This ion with distinctive isotope patterns was identified as a doubly charged ion with one **13a** adducted to the



2 addition reaction, which was undoubtedly confirmed by the critical *b* and *y* ions (Figure 6F).

Table 8. Modified Peptides in the Tryptic Digest of 13a-Treated TrxR

peptide sequence	<i>m / z</i>	charge	$t_{\rm R}({\rm min})$	$[M + H]^{+}$	sample
WGLGGTCVNVGCIPK	752.3722	2	38.76	1503.7371	control
SGGDILQSGCUG	571.6876	2	32.67	1142.3679	control
SGGDILQSGCU-13aG	898.4107	2	31.71	1795.8141	reaction
RSGGDILQSGCUG	649.7374	2	28.63	1298.4675	control
RSGGDILQSGCU-13aG	976.4026	2	33.89	1951.7979	reaction

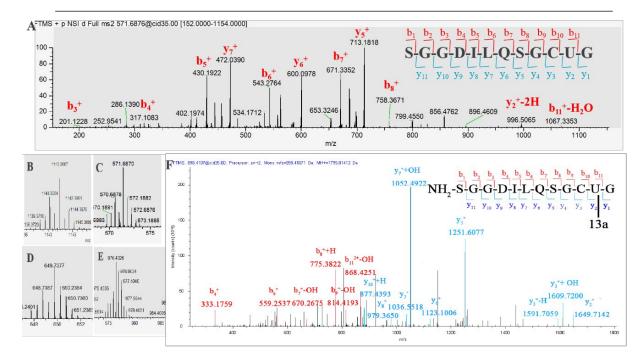


Figure 6. LC-MS/MS analysis of control peptides (A-D) and the **13a**-TrxR adducted tryptic peptide (E, F).

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The binding ability of 13a to sulfhydryl groups. Intracellular Cys residues are prone to Michael addition reactions with α , β -unsaturated ketones. To examine the ability of the compounds to bind to sulfhydryl groups, we conducted in vitro binding experiments (Figure S6).

The reaction products of 13a and NAC were analyzed using online NMR spectroscopy (Figure S6). The experimental results showed that the double bonds at positions 2 and 3 from the A ring were easier to react with SH on NAC than the double bonds at positions 2' and 3' from the side chain. Meanwhile, the reaction products of WA, 13a with NAC and DTT were also detected by LC-MS. The results showed that both WA and 13a readily react with DTT and NAC, the standard test reagents used for the detection of sulfhydryl groups (Figure S7). Next, we examined the effect of the compounds on the total amount of sulfhydryl groups in HT-29 cells. As shown in Figure S6D, with increasing compound concentrations, the total content of sulfhydryl groups in cells decreased gradually, and the effect of 13a was much stronger than that of WA. This finding suggests that the compounds can directly or indirectly reduce the total amount of sulfhydryl groups in HT-29 cells.

Effects of 13a on ASK-1 and P53 signaling. We determined the impact of 13a on the expression of TrxR and Trx in HT-29 cells by Western blot analysis. As shown in Figure 7A, 13a significantly reduced the relative levels of TrxR and Trx expression in HT-29 cells. TrxR inhibitory activity can destroy the redox function of some cells, leading to apoptosis signaling and regulating the activation of a kinase (ASK1). The reduced Trx binds to the Cys residues of ASK1, inhibiting its kinase activity.³⁴ As shown in Figure 7A, **13a** significantly reduced the ASK1 to p-ASK1 expression ratio in HT-29 cells. Trx can affect the oxidation/reduction state of key Cys residues in the DNA binding domain of the human cancer suppressor gene (P53), thereby affecting

the ability of P53 to bind to DNA. When cancer cells use the Trx system to fight apoptosis, this process may be accompanied by the inactivation of the P53 system, thus enhancing the viability of cells.³⁵ **13a** significantly increased the relative levels of P53 to GAPDH expression in HT-29 cells. Hence, the decrease in the levels of TrxR and Trx expression induced by **13a** may contribute to the TrxR inhibitory activity and anticancer activity of **13a** in cancer cells.

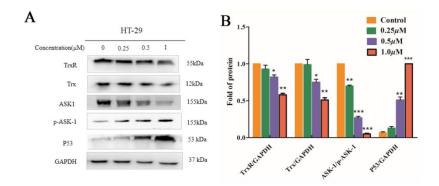


Figure 7. (A) Effect of 13a on the relative levels of TrxR, Trx, ASK-1, p-ASK-1 and P53
expression in HT-29 cells. Cells were treated with 13a or vehicle for 24 h, and the levels of TrxR,
Trx, ASK-1, p-ASK-1 and P53 relative to the expression of the control GAPDH were determined
by Western blot analysis. Data are representative of three independent experiments. *P < 0.05, **P
< 0.01 vs the control.

Effect of 13a on ROS accumulation in cells. The TrxR system is a vital component involved in the maintenance of intracellular redox homeostasis and in defense against oxidative stress. TrxR is the only known enzyme that maintains reduced Trx pools for ribonucleotide reductase in DNA synthesis³⁶ and the pools of many antioxidant enzymes, such as peroxiredoxins³⁷ and Met sulfoxide reductases,³⁸ under physiological conditions. The inhibition of TrxR can disturb the cellular redox balance and cause the accumulation of ROS in cells. DCFH-DA is a well-established probe used to detect the intracellular production of ROS. After being taken up by cells, DCFH-DA is hydrolyzed by cellular esterases into dichlorodihydrofluorescin (DCFH), which is trapped within

the cell. The nonfluorescent DCFH is then oxidized to fluorescent dichlorofluorescin by the action of cellular ROS. In our experiments (Figure 8), the level of ROS in HT-29 and HCT-116 cells was dose-dependently increased by **13a** treatment. Pretreatment with 5 mM GSH, an ROS scavenger, could weaken the compound-induced ROS production in two types of cells. In addition, pretreatment with 5 mM GSH rescued two types of cells from the cytotoxic effects of **13a**, suggesting that the antiproliferative effect of **13a** was ROS dependent.

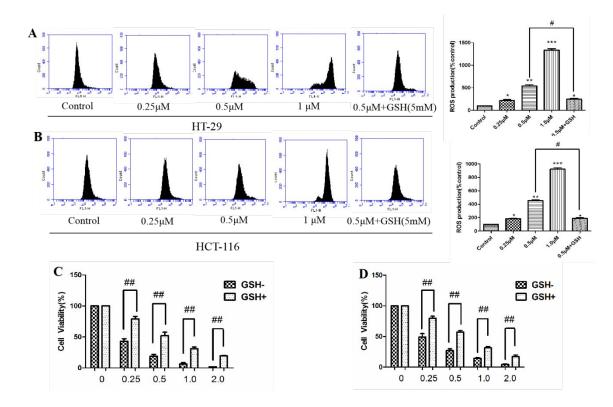


Figure 8. Effect of **13a** on ROS accumulation in cells. (A) HT-29 cells were incubated with the indicated concentrations of **13a** for 24 h. The ROS level was measured by flow cytometry. (B) HCT-116 cells were incubated with the indicated concentrations of **13a** for 24 h. The ROS level was measured by flow cytometry. (C) HT-29 cells were treated with the indicated concentrations of **13a** alone or in combination with the ROS scavenger GSH (5 mM), and cell viability was determined by MTT assay. (D) HCT-116 cells were treated with the indicated concentrations of

13a alone or in combination with the ROS scavenger GSH (5 mM), and cell viability was determined by MTT assay.

Monoclonal formation experiment. Cloning experiments are an effective method to determine the proliferative capacity of a single cell. To test the effect of 13a on the ability of HT-29 cells to form monoclonal cells, we treated cells with different concentrations of 13a for 24 h. After treatment administration, the normal culture medium was changed daily for 14 days. The results (Figure 9A) showed that the number of clones in the administration group was significantly reduced compared with that in the DMSO control group, showing the concentration-dependent inhibition of HT-29 cell monoclonal ability by 13a. The study demonstrated that 13a can inhibit the proliferation of colon cancer cells.

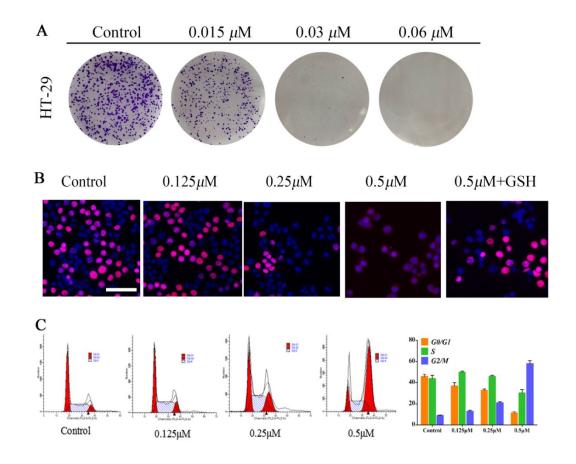


Figure 9. Effect of 13a on the proliferation of HT-29 cells. (A) The effect of 13a on the ability of HT-29 cells to form monoclonal cells. (B) HT-29 cells were incubated with various concentrations of 13a alone or in combination with the ROS scavenger GSH (5 mM) for 24 h. EdU staining was then performed, and the cells were observed by confocal microscopy. Blue cells were counted as Hoechst-positive cells; red cells were counted as EdU-positive cells. Scale bar: 20 μ m. (C) HT-29 cells were treated with the indicated concentrations of **13a** for 24 h, and the distribution of the cell cycle was monitored by flow cytometry. Columns showing the percentage of HT-29 cells in the G0/G1, S and G2/M phases of the cell cycle.

9 EdU infiltration experiment. Next, an EdU kit was used to further verify the effect of the drug 10 on cell proliferation. EdU is a thymine nucleoside analog that replaces T during cell proliferation 11 to infiltrate into the DNA that is being copied. We applied different concentrations of 13a to HT-12 29 cells, and the number of EdU-positive cells was significantly lower in the treatment group than 13 in the control group (Figure 9B), which indicates that 13a can effectively inhibit the proliferation 14 of cancer cells.

Induced G2/M arrest in HT-29 cells. To learn more about the antiproliferative mechanism of **13a**, the effect of **13a** on cell cycle progression was explored using flow cytometric analysis after propidium iodide (PI) staining (Figure 9C). The results showed that the number of cells in the G2/M fraction increased significantly and dose-dependently after 24 h of exposure to **13a**. The cell cycle distribution is summarized in the histograms shown in Figure 8C, and the results suggest that **13a** induced G2/M phase arrest in HT-29 cells, which was consistent with reports that **13a** induced G2/M arrest.

Induction of HT-29 cell apoptosis. To determine whether the inhibitory effects of 13a on colon
 cancer cell proliferation were accompanied by enhanced cancer cell apoptosis, Annexin V-FITC

and PI staining were carried out, and the percentages of apoptotic cells were tested using a flow cytometry assay. HT-29 cells were incubated with different concentrations of compound 13a for 24 h. The results (Figure 10A) show that 13a mainly induced apoptotic cell death in HT-29 cells. Furthermore, a Western blot assay (Figure 10C) revealed that treatment with 13a dramatically decreased the expression levels of antiapoptotic Bcl-2 but increased the expression levels of proapoptotic Bax in HT-29 cells. Quantitative analysis revealed that treatment with 13a significantly increased the ratio of Bax to Bcl-2. Caspase activation is a critical event in the initiation and execution of apoptosis in cells. Treatment with 13a also significantly increased the relative levels of cleaved-caspase-3 and poly ADP-ribose polymerase (PARP) in HT-29 cells.

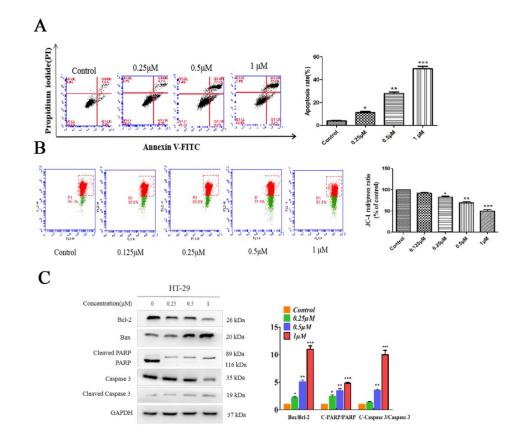




Figure 10. Effect of 13a on apoptosis induction in HT-29 cells. (A) HT-29 cells were incubated with the indicated different concentrations of 13a for 24 h and stained with Annexin V-APC and 7-AAD. (B) HT-29 cells were incubated with the indicated concentrations of 13a for 24 h and

stained with JC-1. (C) Western blot analysis was performed with antibodies specific for Bcl-2, Bax, caspase-3, cleaved-caspase-3, PARP, and cleaved-PARP, and GAPDH was used as the loading control.

Effect of 13a on the mitochondrial membrane potential of cells. Above, we demonstrated that **13a** significantly increased the ratio of Bax to Bcl-2. An increased ratio of Bax/Bcl-2 protein can promote mitochondrial membrane depolarization and induce apoptosis. Many studies have shown that mitochondrial dysfunction is closely related to apoptosis and plays an important role in the apoptosis pathway. Therefore, we examined whether the drug can cause mitochondrial damage. JC-1 is a cationic lipid fluorescent dye and is usually used to detect mitochondrial membrane potential. The changes in mitochondrial membrane potential after administration were detected by flow cytometry. The results showed that, with increasing 13a concentrations, the ratio of red fluorescence to green fluorescence relative to the blank control group decreased (Figure 10B). This finding indicated that **13a** can reduce the mitochondrial membrane potential of HT-29 cells.

15 CONCLUSIONS

In summary, we designed 41 semisynthetic analogs of WA. Iterative SAR studies of the target compounds were performed. Among all the compounds, **13a** displayed the highest cytotoxicity toward HT-29 cells ($IC_{50} = 0.08 \ \mu M$), which is approximately 30 times greater than the initial natural product WA ($IC_{50} = 2.48 \ \mu M$). Furthermore, **13a** suppressed TrxR expression through covalent bonding to 498 amino acid residues by Michael reaction, as demonstrated by protein MS/MS and molecular docking. As a result, **13a** promoted intracellular ROS accumulation and decreased the level of reduced Trx by changing the expression of ASK1 and P53; moreover, **13a**

ASK1

P53

proliferation

exerted an antiproliferation effect by inducing apoptosis and cell cycle arrest (Figure 11). 13a, as novel TrxR inhibitor, may be a promising candidate for cancer intervention. covalent TrxR 13a Trx Bax Bcl-2 ROS cell cycle arres Caspase3 Apoptosis PARP Figure 11. Summary of the physiological activity of 13a in HT-29 cells.

EXPERIMENTAL SECTION

Chemistry

All reagents were purchased from Sigma-Aldrich, Energy Chemical, Aladdin or Macklin, Thin layer chromatography (TLC) was performed using silica gel-coated aluminum plates with fluorescent indicators and was visualized with ultraviolet light at 254 nm. Melting points were measured on an XT-4 digital display micromelting point apparatus (Taike, Beijing, China), uncorrected. ¹H NMR spectra were recorded on a Bruker Ultrashield 500 Plus NMR spectrometer, and ¹³C NMR spectra were recorded on a Bruker Ultrashield 125 Plus NMR spectrometer. Chemical shift values (δ) were expressed as parts per million (ppm) relative to tetramethylsilane as the internal standard. High-resolution electrospray ionization (HRESI) mass spectra were carried out using an Agilent 6520B Q-TOF mass spectrometer (Agilent Technologies, Santa Clara,

1 CA, USA). High performance liquid chromatography (HPLC) was performed for purity checking 2 using Agilent 1100 series HPLC on a Shimadzu XDB-C18 (5 μ m, 4.6 × 150 mm²) column. All the 3 final compounds were found to be pure up to 95% or higher.

Isolation of Naturally Occurring Withangulatin A (1) Withangulatin A (1) was obtained from the whole herb of *Physalis angulata* L.¹⁹ White powder; m.p. 151-152 °C; ¹H NMR (500 MHz, $CDCl_3$ δ 6.95 (dd, J = 10.0 and 6.0 Hz, 1H, H-3), 6.18 (d, J = 10.0 Hz, 1H, H-2), 5.69 (d, J = 5.0Hz, 1H, H-16), 5.23 (d, J = 5.0 Hz, 1H, H-15), 4.23 (m, 1H, H-22), 3.82 (d, J = 2.5 Hz, 1H, H-4), 2.59 (dt, J = 14.0 and 3.0 Hz, 1H, H-7 α), 2.50 (m, 1H, H-20), 2.40 (t, J = 16.0 Hz, 1H, H-23 β), 2.17 (dd, J = 16.0 and 3.5 Hz, 1H, H-23 α), 1.95 (s, 3H, OAc-15), 1.93 (s, 3H, H-28), 1.86 (s, 3H, H-27), 1.82-1.50 (m, 7H, H-7 β , H-8, H-9, H-11, H-12), 1.43 (s, 3H, H-19), 1.12 (d, J = 6.0 Hz, 3H, H-21), 1.10 (s, 3H, H-18); ¹³C NMR (125 MHz, CDCl₃) δ 202.7 (C-1), 169.9 (C-15-OCCH₃), 166.2(C-26), 162.6(C-17), 148.6(C-24), 142.67(C-3), 131.6(C-2), 122.4(C-25), 121.1(C-16), 83.7(C-15), 81.6(C-14), 79.4(C-22), 69.7(C-4), 63.7(C-5), 63.3(C-6), 52.2(C-13), 47.7(C-10), 39.3(C-9), 37.6(C-12), 35.5(C-20), 34.7(C-8), 33.1(C-23), 24.8(C-7), 22.1(C-11), 21.5(C-15-OCCH₃), 20.6(C-28), 17.7(C-18), 17.6(C-21), 16.1(C-19), 12.6(C-27); HRMS: [M + Na]⁺ calcd for $C_{30}H_{38}O_8Na$ 549.2459, found 549.2463; purity 95.78% (MeOH/H₂O = 75/25, R_t = 3.029 min).

17 General Procedure for the Synthesis of 2–12.

(2a). To a solution of the 1 (26.3 mg) and sarcosine (8.9 mg) in anhydrous CH₂Cl₂, and the mixture
was stirred at 25 °C for 2h. 2a was obtained as a white powder, 26.1 mg, yield 85 %; m.p. 163164 °C; ¹H NMR (500 MHz, CDCl₃) δ 5.71 (d, J = 5.0 Hz, 1H), 5.24 (d, J = 5.0 Hz, 1H), 4.23 (m,
1H), 3.62 (s, 3H), 3.32 (s, 1H), 3.13 (dd, J = 12.0 Hz, 5.0 Hz, 1H), 2.86 (m, 1H), 2.69 (s, 1H),
2.59-2.45 (m, 7H), 2.40 (t, J = 16.0 Hz, 1H), 2.19 (dd, 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 3H), 1.93 (s,
3H), 1.86 (s, 3H), 1.82-1.50 (m, 8H), 1.37 (s, 3H), 1.13 (d, J = 6.0 Hz, 3H), 1.07 (s, 3H); ¹³C NMR

3 4	1	$(125 \text{ MHz}, \text{CDCl}_3) \delta 211.2, 169.8, 166.3, 162.6, 148.6, 122.4 121.0, 83.8, 81.6, 79.5, 73.7, 67.2,$
5 6	2	64.8, 63.7, 62.6, 52.3, 50.5, 50.5, 38.2, 37.4, 36.0, 35.5, 34.2, 33.1, 24.9, 21.8, 21.6, 20.6, 17.7,
7 8 9	3	16.8, 16.1, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{33}H_{45}NNaO_{10}$ 638.2935, found 638.2936; purity
10 11 12	4	96.16% (MeOH/H ₂ O = 40/60-80/20, R_t = 3.147 min).
13 14	5	(2b). To a solution of the 1 (26.3 mg), and imidazole (3.4 mg) anhydrous $CH_2Cl_{2,}$ and the mixture
15 16	6	was stirred at 25 °C for 2h. 2b was obtained as a faint yellow powder, 25.3 mg, yield 85 %; m.p.
17 18 19	7	157-159 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.54 (s, 1H),7.26 (s, 1H), 7.08 (s, 1H), 5.73 (d, J = 5.0
20 21	8	Hz, 1H) ,5.18 (d, <i>J</i> = 5.0 Hz, 1H) ,4.73 (m, 1H), 4.23 (m, 1H), 3.61 (s, 3H), 3.46 (dd, <i>J</i> = 12.0 Hz,
22 23	9	5.0 Hz, 1H), 3.10 (s, 1H), 2.87 (m, 1H), 2.59-2.45 (m, 7H), 2.09 (dd, 16.0 Hz, 3.5 Hz, 1H), 1.93
24 25 26	10	(s, 3H), 1.85(s, 3H), 1.82-1.50 (m, 8H), 1.46 (s, 3H), 1.12 (d, $J = 6.0$ Hz, 3H), 1.09 (s, 3H); ¹³ C
20 27 28	11	NMR (125 MHz, CDCl ₃) δ 209.8, 169.9, 166.5, 162.6, 148.7, 122.3, 121.0, 117.3, 84.0, 81.5, 79.7,
29 30	12	63.0, 61.6, 56.2, 53.6, 52.2, 50.8, 39.7, 38.7, 37.5, 35.5, 34.1, 33.2, 29.9, 24.5, 22.0, 21.6, 20.6,
31 32	13	17.9, 16.9, 16.1, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{33}H_{42}N_2O_8Na$ 617.2833, found 617.2836;
33 34 35	14	purity 95.15% (MeOH/H ₂ O = 40/60-80/20, R_t = 4.535 min).
36 37 38	15	(2c). To a solution of the 1 (26.3 mg), and HCl solution in ether anhydrous CH_2Cl_2 , and the mixture

е was stirred at 25 °C for 1h. 2c was obtained as a faint yellow powder, 16.7 mg, yield 60 %; m.p. 147-149 °C; ¹H NMR (500 MHz, CDCl₃) δ 6.59 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 5.93 (d, J = 10.0Hz, 1H), 5.69 (d, J = 5.0 Hz, 1H), 5.31 (d, J = 5.0 Hz, 1H), 5.17 (s, 1H), 4.72 (s, 1H), 4.23 (m, 1H), 2.59 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0Hz, 3.5 Hz, 1H) 2.02 (s, 3H), 1.95 (s, 3H), 1.86 (s, 3H), 1.82-1.50 (m, 5H), 1.63 (s, 3H), 1.14 (d, J = 6.0 Hz,3H), 1.13 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 209.8, 169.9, 166.5, 162.6, 148.7, 122.3, 121.0, 117.3, 84.0, 81.5, 79.7, 63.0, 61.6, 56.2, 53.6, 52.2, 50.8, 39.7, 38.7, 37.5, 35.5, 34.1,

2		
3 4	1	33.2, 29.9, 24.5, 22.0, 21.6, 20.6, 17.9, 16.9, 16.1, 12.6; HRMS: $[M + Na]^+$ calcd for
5 6 7	2	$C_{33}H_{45}NO_{10}Na 585.2226$, found 585.2229; purity 95.78% (MeOH/H ₂ O = 75/25, R _t = 3.686 min).
8 9	3	(3). To a solution of the 1 (26.3 mg) and 3-Chloroperbenzoic acid (8.7 mg) in anhydrous $CH_2Cl_{2,}$
10 11 12	4	and the mixture was stirred at 0 °C for 12h. 3 was obtained as a white powder, 21.1mg, yield 78
13 14	5	%; m.p. 131-133 °C; Crystal Data for 3 (CCDC-2008080): orthorhombic, space group
15 16	6	P2 ₁ 2 ₁ 2 ₁ (no. 19), $a = 7.4020(2)$ Å, $b = 13.4298(4)$ Å, $c = 26.9766(9)$ Å, $V = 2681.67(14)$ Å ³ , $Z = 12.4298(4)$ Å
17 18	7	4, $T = 130.0$ K, $\mu(CuK\alpha) = 0.814$ mm ⁻¹ , $D_{calc} = 1.344$ g/cm ³ , 20906 reflections measured (6.552°
19 20 21	8	$\leq 2\Theta \leq 149.01^{\circ}$), 5373 unique ($R_{int} = 0.0737$, $R_{sigma} = 0.0636$) which were used in all calculations.
22 23	9	The final R_1 was 0.0468 (I > 2 σ (I)) and wR_2 was 0.1108 (all data). ¹ H NMR (500 MHz, CDCl ₃) δ
24 25	10	6.94 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.17 (d, J = 10.0 Hz, 1H), 4.99 (s, 1H), 4.49 (m, 1H), 3.80 (dd,
26 27 28	11	<i>J</i> = 4.8 Hz, 1.8 Hz, 1H), 3.48 (s, 1H), 3.34 (s, 1H), 3.17 (brs, 1H), 2.57 (dt, <i>J</i> = 14.0 Hz, 3.0 Hz,
29 30	12	1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.11 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.05 (s, 3H), 1.98
31 32	13	(dd, J = 10.5 Hz, 2.5 Hz, 1H), 1.93 (s, 3H), 1.88 (s, 3H), 1.74-1.59 (m, 4H), 1.40 (s, 3H), 1.39 (m,
33 34 25	14	2H), 1.14 (s, 3H), 1.00 (d, $J = 6.0$ Hz, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ 202.8, 169.5, 166.4,
35 36 37	15	149.3, 142.8, 131.3, 122.1, 81.1, 77.6, 76.8, 76.1, 69.4, 63.8, 63.0, 59.0, 47.6, 46.4, 38.9, 34.6,
38 39	16	33.8, 32.5, 31.5, 25.8, 21.3, 21.1, 20.6, 17.8, 15.7, 13.4, 12.6; HRMS: [M + Na] ⁺ calcd for
40 41	17	$C_{30}H_{38}O_9Na$ 565.2408, found 565.2411; purity 97.91% (MeOH/H ₂ O = 60/40-100/0, R _t = 6.117)
42 43 44 45	18	min).
46 47	19	(4). To a solution of the 1 (26.3 mg) and MnO_2 (17.2 mg) in anhydrous $CH_2Cl_{2,}$ and the mixture
48 49	20	was stirred at 25 °C for 8h. 4 was obtained as a faint yellow powder, 21.2 mg, yield 81 %; m.p.
50 51 52	21	249-251 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 6.87 (q, 10.0 Hz, 2H), 5.73 (d, J = 2.8 Hz 1H), 5.22
52 53 54	22	(d, J = 2.8 Hz 1H), 4.24 (m, 1H), 3.57 (d, J = 2.5 Hz, 1H), 2.63 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.52
55	23	(d, J = 7.5 Hz, 1H), 2.40 - 2.33 (m, 2H), 2.18 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.99 (s, 3H), 1.94 (

- (d, J = 7.5 Hz, 1H), 2.40 -2.33 (m, 2H), 2.18 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.99 (s, 3H), 1.94 (s,

3H), 1.87 (s, 3H), 1.80-1.65 (m, 7H), 1.40 (s, 3H), 1.15 (d, J=7.0 Hz, 3H), 1.11 (s, 3H); ¹³C NMR
 (125 MHz, CDCl₃) δ 202.4, 193.8, 166.2, 162.3, 148.5, 141.3, 139.6, 122.4, 121.4, 83.8, 81.2,
 79.3, 64.2, 63.7, 52.2, 50.2, 38.4, 37.7, 35.5, 34.6, 33.0, 29.9, 24.7, 23.3, 21.6, 20.6, 19.4, 17.7,
 16.2, 12.6; HRMS: [M + NH₄]⁺ calcd for C₃₀H₄₀NO₈ 542.2751, found 542.2750; purity 98.20%
 (MeOH/H₂O = 60/40-100/0, R_t = 8.344 min).

(5). To a solution of the 1 (26.3 mg), TBDMSCl (17.2 mg) and imidazole(1.0 mg) anhydrous CH₂Cl₂ and the mixture was stirred at 25 °C for 12h. 5 was obtained as a faint yellow powder, 13.1 mg, yield 41 %; m.p. 194-196 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.13 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.88 (d, J = 10.5 Hz, 1H), 5.76 (d, J = 3.0 Hz, 1H), 5.18 (d, J = 3.0 Hz, 1H), 4.48 (m, 1H), 4.22 (m, 1H), 3.37 (brs, 1H), 2.59 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 14.0 Hz, 3.0 Hz, 1H), 3.50 (m, 1H), 3.50 16.0 Hz, 1H, 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.11 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.86 (m, 1H), 1.75 - 1.59 (m, 5H), 1.60 (s, 3H), 1.41 (m, 2H), 1.40 (s, 3H), 1.12 (d, J = 7.5 Hz, 3H), 1.06 (s, 3H), 1.75 - 1.59 (m, 5H), 1.60 (s, 3H), 1.41 (m, 2H), 1.40 (s, 3H), 1.12 (d, J = 7.5 Hz, 3H), 1.06 (s, 3H), 1.41 (m, 2H), 1.40 (s, 3H), 1.40 (s, 3 3H), 0.93 (s, 9H), 0.17 (s, 3H), 0.13 (s, 3H); 13 C NMR (125 MHz, CDCl₃), δ 169.7, 166.2, 162.3 148.5, 136.1, 130.0, 122.4, 121.5, 84.2, 81.4, 79.3, 79.0, 61.9, 60.1, 53.6, 52.3, 51.1, 39.2, 38.6, 37.3, 35.5, 34.4, 33.0, 32.1, 29.9, 27.4, 25.9, 24.7, 22.8, 21.7, 21.4, 20.6, 18.2, 17.7, 16.0, 14.3, 13.0; HRMS: $[M + Na]^+$ calcd for C₃₆H₅₂O₈SiNa 663.3324, found 663.3326; purity 97.52% $(MeOH/H_2O = 60/40-100/0, R_t = 14.313 min).$

18 (6). To a solution of the 1 (26.3 mg) in anhydrous pyridine (1.0 mL)was added *N*,*N*-19 Carbonyldiimidazole (8.1 mg), and the mixture was stirred at 25 °C for 8h, Then add 0.5mL 20 methanol, **6** was obtained as a faint yellow powder, 21.0 mg, yield 72 %; m.p. 118-120 °C; ¹H 21 NMR (500 MHz, CDCl₃) δ 7.05 (dd, *J* = 10.0 Hz, 6.0 Hz, 1H), 6.28 (d, *J* = 10.5 Hz, 1H), 5.69 (d, 22 *J* = 3.0 Hz, 1H), 5.22 (d, *J* = 3.0 Hz, 1H), 4.65 (d, *J* = 6.5 Hz, 1H), 4.22 (m, 1H), 3.77 (s, 3H), 23 3.37 (brs, 1H), 2.59 (dt, *J* = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, *J* = 16.0 Hz, 1H), 2.17

(dd,
$$J = 16.0$$
 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.60 (s, 3H), 1.41 (m, 2H), 1.39 (s, 3H), 1.11 (d, $J = 7.5$ Hz, 3H), 1.08 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 201.3, 169.7, 166.2, 162.5, 155.2,148.5, 139.7, 134.0, 122.4, 121.1, 83.7, 81.6, 79.4, 75.0, 61.4, 60.4, 55.3, 52.2, 48.0, 39.6, 37.4, 35.5, 34.6, 33.0, 24.6, 21.5, 21.2, 20.6, 17.7, 16.0, 15.9, 12.6; HRMS: [M + NH₄]⁺ calcd for C₃₂H₄₄NO₁₀ 602.2960, found 602.2965; purity 98.20% (MeOH/H₂O = 75/25, R_t = 4.337 min).

(7a). To a solution of the 1 (26.3 mg) in anhydrous pyridine (1.0 mL) was added Ac₂O (0.5 mL), and the mixture was stirred at 25 °C for 12h. 7a was obtained as a faint yellow powder, 24.2 mg, yield 86 %; m.p. 126-128 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.07 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.24 (d, J = 10.5 Hz, 1H), 5.69 (d, J = 3 Hz, 1H), 5.22 (d, J = 3 Hz, 1H), 4.70 (d, J = 6.5 Hz, 1H),4.23 (m, 1H), 3.34 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.06 (s, 3H), 1.97 (m, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 - 1.59 (m, 5H), 1.60 (s, 3H), 1.41 (m, 2H), 1.40 (s, 3H), 1.12 (d, J =7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125MHz, CDCl₃), δ 201.5, 170.2, 169.7, 166.2, 162.5, 148.5, 140.7, 133.3, 122.4, 121.1, 83.7, 81.6, 79.4, 72.1, 61.1, 60.3, 52.2, 48.1, 39.7, 37.4, 35.5, 34.6, 33.0, 24.6, 21.5, 21.0, 20.6, 17.7, 16.0, 15.7, 12.6; HRMS: [M + NH₄]⁺ calcd for C₃₂H₄₄NO₉ 586.3011, found 586.3013; purity 99.34% (MeOH/H₂O = 60/40-100/0, R_t = 9.679 min).

(7b). To a solution of the 1 (26.3 mg) in anhydrous pyridine (1.0 mL) was added propionic
anhydride (0.5 mL), and the mixture was stirred at 25 °C for 16h. 7b was obtained as a faint yellow
powder, 22.1 mg, yield 76 %; m.p. 105-107 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.07 (dd, J = 10.0
Hz, 6.0 Hz, 1H) , 6.24 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H),
4.71 (d, J = 5.5 Hz, 1H), 4.23 (m, 1H), 3.34 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m,
1H), 2.34 (m, 3H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 3H), 1.94 (s, 3H), 1.93 (s, 3H), 1.85

(s, 3H), 1.84 -1.67 (m, 6H), 1.40 (s, 3H), 1.12- 1.10 (m, 5H), 1.08 (s, 3H); ¹³C NMR (125 MHz, CDCl₃), δ 201.5, 173.6, 169.7, 166.3, 162.4, 148.4, 140.9, 133.2, 122.3, 121.1, 83.7, 81.6, 79.3, 71.9, 61.0, 52.2, 48.0, 39.6, 35.4, 34.6, 33.0, 27.5, 24.5, 21.5, 21.0, 20.6, 17.7, 16.0, 15.6, 12.6; HRMS: [M + NH₄]⁺ calcd for C₃₃H₄₆NO₉ 600.3167, found 600.3169; purity 96.86% (MeOH/H₂O = 60/40-100/0, R₁ = 11.048 min).

6 (8a). To a solution of the crylic acid (7.2 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous
7 CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and
8 the mixture was stirred at 25 °C for 24h. 8a was obtained as a yellow powder, 22.9 mg, yield 71
9 %; m.p. 99-101 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.49 (s, 1H), 7.38 (d, *J* = 13.0 Hz, 1H), 7.13
10 (dd, *J* = 10.0 Hz, 6.0 Hz, 1H), 6.63 (d, *J* = 3.0 Hz, 1H), 6.47 (q, *J* = 1.5 Hz, 1H), 6.27 (d, *J* = 3.5
11 Hz, 1H), 6.25 (d, *J* = 1.5 Hz, 1H) 5.69 (d, *J* = 2.5 Hz, 1H), 5.23 (d, *J* = 2.5 Hz, 1H), 4.82 (d, *J* =
10.0 Hz, 1H), 4.23 (m, 1H), 3.38 (brs, 1H), 2.57 (dt, *J* = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40
13 (t, *J* = 16.0 Hz, 1H), 2.18 (dd, *J* = 16.0 Hz, 3.5 Hz, 1H), 1.95 (s, 3H), 1.93(s, 3H), 1.86 (s, 3H),
14 1.82 (m, 1H), 1.74 -1.59 (m, 5H), 1.42 (s, 3H), 1.41 (m, 2H), 1.12 (d, *J* = 7.5 Hz, 3H), 1.09 (s,
3H); ¹³C NMR (125 MHz, CDCl₃), δ 201.4, 169.7, 166.2, 162.5, 148.5, 140.6, 133.5, 132.1, 127.8,
16 122.4, 121.1, 83.8, 81.7, 79.4, 72.2, 61.1, 60.4, 52.2, 48.1, 39.4, 37.4, 35.5, 34.6, 33.0, 29.9, 24.6,
17 21.5, 21.0, 20.6, 17.7, 16.0, 15.7, 12.6; HRMS: [M + NH₄]⁺ calcd for C₃₃H₄₄NO₉ 598.3011, found
18 598.3010; purity 98.36% (MeOH/H₂O = 75/25, R_t = 7.101 min).

19 (**8b**). To a solution of the crotonic acid (8.6 mg), DMAP (4mg) and EDCI (200 mg) in anhydrous 20 CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1(26.3 mg), and 21 the mixture was stirred at 25 °C for 24h. **8b** was obtained as a faint yellow powder, 17.8 mg, yield 22 60 %; m.p. 116-118 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.11 (dd, *J* = 10.0 Hz, 6.0 Hz, 1H), 6.95 23 (q, *J* = 6.5 Hz, 1H), 6.24 (d, *J* = 10.0 Hz, 1H), 5.81 (dd, *J* = 13.0 Hz, 1.5 Hz 1H), 5.69 (d, *J* = 2.5

Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.76 (d, J = 5.0 Hz, 1H), 4.23 (m, 1H), 3.36 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93(s, 3H), 1.88 (dd, J = 6.0 Hz, 1.5 Hz 3H), 1.86 (s, 3H), 1.82 (m, 1H), 1.74 -1.59 (m, 5H), 1.42 (s, 3H), 1.41 (m, 2H), 1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125) MHz, CDCl₃), δ 201.5, 169.7, 166.2, 165.5, 162.5, 148.5, 146.4, 141.0, 133.2, 122.4, 121.9, 121.1, 83.7, 81.7, 79.3, 71.8, 61.0, 60.4, 52.2, 48.1, 39.7, 37.3, 35.4, 34.6, 33.0, 24.6, 21.5, 21.0, 20.6, 18.3, 17.6, 16.0, 15.7, 12.6; HRMS: $[M + NH_4]^+$ calcd for $C_{34}H_{46}NO_9$ 612.3167, found 612.3164; purity 95.25% (MeOH/H₂O = 75/25, R_t = 7.645 min).

(8c). To a solution of the 3-methylbut-2-enoic acid (10.0 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 8c was obtained as a faint yellow powder, 20.7 mg, yield 68 %; m.p. 108-110 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.07 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.24 (d, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.90 (t, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 5.20 (d, J = 2.5 1.5 Hz, 1H), 4.83 (s, 1H), 4.72 (d, J = 10.0 Hz, 1H), 4.23 (m, 1H), 3.35 (brs, 1H), 3.04 (s, 2H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.76 (s, 3H), 1.83 -1.59 (m, 5H), 1.40 (m, 2H), 1.39 (s, 3H), 1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125 MHz, $CDCl_3$, δ 201.5, 170.6, 169.7, 166.2, 162.5, 148.6, 140.5, 137.9, 133.4, 122.4, 121.1, 115.4, 83.7, 81.6, 79.4, 72.2, 61.1, 60.3, 52.2, 48.1, 43.4, 39.7, 37.4, 35.5, 34.6, 33.0, 24.6, 22.5, 21.5, 21.0, 20.6, 17.7, 16.0, 15.6, 12.6; HRMS: $[M + NH_4]^+$ calcd for C₃₅H₄₈NO₉ 626.3324, found 626.3325; purity 96.96% (MeOH/H₂O = 75/25, R_t = 10.075 min).

(8d). To a solution of the sorbic acid (11.2 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous
CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and

1	the mixture was stirred at 25 °C for 24h. 8d was obtained as a faint yellow powder, 22.9 mg, yield
2	74 %; m.p. 91-93 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.20 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 7.11 (q,
3	<i>J</i> = 6.5 Hz, 1H), 6.24 (d, <i>J</i> = 10.0 Hz, 1H), 6.16 (m, 1H), 5.73 (d, <i>J</i> = 15.0 Hz, 1H), 5.69 (d, <i>J</i> =
4	2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.77 (d, J = 5.0 Hz, 1H), 4.23 (m, 1H), 3.36 (brs, 1H), 2.57
5	(dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5
6	Hz, 1H), 1.94 (s, 3H), 1.93(s, 3H), 1.86 (s, 3H), 1.85 (s, 3H), 1.82 (m, 1H), 1.74 -1.59 (m, 5H),
7	1.43 (s, 3H), 1.42 (m, 2H), 1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ
8	201.6, 169.7, 166.4, 166.2, 162.4, 148.5, 146.5, 141.1, 140.7, 133.2, 129.8, 122.4, 121.1, 117.9,
9	83.7, 81.7 79.3, 71.8, 61.0, 53.6, 52.2, 48.1, 39.7, 37.3, 35.4, 34.6, 32.9, 24.6, 21.5, 21.0, 20.6,
10	18.9, 17.6, 16.0, 15.7, 12.6; HRMS: $[M + NH_4]^+$ calcd for $C_{36}H_{48}NO_9$, 638.3324; found 638.3327;
11	purity 96.13% (MeOH/H ₂ O = $60/40-100/0$, R _t = 13.488 min).

(8e). To a solution of the monoethyl fumarate (14.4 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 8e was obtained as a yellow powder, 25.8 mg, yield 79 %; m.p. 93-95 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.07 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.81 (s, 2H), 6.28 (d, J = 10.0 Hz, 1H), 5.69 (d, J = 2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.83 (d, J = 10.0 Hz, 1H), 4.27 (q, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.38 (brs, 1H), 3.04 (s, 2H), 2.57 (dt, J) = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93(s, 3H), 1.86 (s, 3H), 1.83 -1.67 (m, 6H), 1.43 (s, 3H), 1.40 (m, 2H), 1.32 $(t, J = 6.0 \text{ Hz}, 3\text{H}), 1.12 (d, J = 7.5 \text{ Hz}, 3\text{H}), 1.09 (s, 3\text{H}); {}^{13}\text{C} \text{ NMR} (125 \text{ MHz}, \text{CDCl}_3), \delta 201.2,$ 169.7, 166.2, 164.8, 164.2, 162.5, 148.6, 139.8, 135.0, 133.9, 132.5, 122.4, 121.0, 83.7, 81.6, 79.4, 72.9, 61.7, 61.3, 60.2, 52.2, 48.0, 39.6, 37.4, 35.5, 34.5, 33.1, 24.5, 21.5, 21.0, 20.6, 17.8, 16.0,

15.9, 14.3, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{36}H_{44}O_{11}Na$ 675.2776, found 675.2778. Purity 95.43% (MeOH/H₂O = 75/25, R_t = 7.769 min).

(8f). To a solution of the 3-Ethoxyacrylic Acid (11.6 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 8f was obtained as a yellow powder, 21.5 mg, yield 69 %; m.p. 124-126 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.55 (d, J = 11.0 Hz, 1H), 7.11 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.23 (d, J = 10.5 Hz, 1H), 5.69 (d, J = 3 Hz, 1H), 5.22 (d, J = 3.0 Hz, 1H), 5.13 (d, J = 11.0 Hz, 1H), 4.76 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.90 (q, J = 6.0 Hz, 2H), 3.35 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 4H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H),1.75 - 1.59 (m, 5H), 1.41 (s, 3H), 1.41 (m, 1H), 1.38 (m, 1H), 1.33 (t, J = 6.0 Hz, 3H), 1.12 (d, J=7.5 Hz, 3H), 1.08 (s, 3H); 13 C NMR (125 MHz, CDCl₃), δ 201.6, 169.7, 167.0, 166.2, 163.5, 162.4, 148.6, 141.4, 133.0, 122.4, 121.1, 95.5, 83.7, 81.6, 79.3, 71.3, 67.0, 61.0, 60.5, 53.6, 52.2, 48.1, 39.7, 37.4, 35.4, 34.6, 33.0, 24.6, 21.5, 21.0, 20.6, 17.6, 16.0, 15.7, 14.5, 12.6; HRMS: [M+ Na]⁺ calcd for C₃₅H₄₄O₁₀Na 647.2827, found 647.2828; purity 95.08% (MeOH/H₂O = 75/25, R_t = 7.960 min).

17 (8g). To a solution of the 3- (2-Furyl) acrylic acid (13.7 mg), DMAP (4 mg) and EDCI (200 mg) 18 in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 19 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 8g was obtained as a yellow powder, 22.9 20 mg, yield 71 %; m.p. 120-122 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.49 (s, 1H), 7.38 (d, *J* = 13.0 21 Hz, 1H), 7.13 (dd, *J* = 10.0 Hz, 6.0 Hz, 1H), 6.63 (d, *J* = 3.25 Hz, 1H), 6.47 (q, 1H), 6.27 (d, *J* = 22 3.25 Hz, 1H), 6.25 (d, *J* = 1.75 Hz, 1H), 5.69 (d, *J* = 2.5 Hz, 1H), 5.23 (d, *J* = 2.5 Hz, 1H), 4.82 23 (d, *J* = 5.0 Hz, 1H), 4.23 (m, 1H), 3.38 (brs, 1H), 2.57 (dt, *J* = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H),

1	2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.95 (s, 3H), 1.93(s, 3H), 1.86 (s,
2	3H), 1.83 (m, 1H), 1.74 -1.59 (m, 5H), 1.46 (s, 3H), 1.40 (m, 2H), 1.12 (d, <i>J</i> = 7.5 Hz, 3H), 1.09
3	(s, 3H); ¹³ C NMR (125 MHz, CDCl ₃) δ 201.6, 169.7, 166.3, 166.2, 162.4, 150.8, 148.6, 145.2,
4	140.9, 133.4, 132.2, 122.3, 121.0, 115.7, 114.7, 112.6, 83.7, 81.6, 79.3, 72.0, 61.1, 60.5, 52.2, 48.1,
5	39.7, 37.4, 35.5, 34.6, 33.0, 24.6, 21.5, 21.0, 20.6, 17.7, 16.0, 15.8, 12.6; HRMS: [M+ NH ₄] ⁺ calcd
6	for $C_{37}H_{46}NO_{10}$, 664.3116, found 664.3121; purity 97.12% (MeOH/H ₂ O = 75/25, R _t = 10.616
7	min).

(8h). To a solution of the 3-(2-Thienyl)Acrylic Acid (15.4 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 8h was obtained as a yellow powder, 19.6 mg, yield 59 %; m.p. 83-85 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.75(d, J = 13.5 Hz, 1H), 7.40 (d, J = 4.5 Hz, 1H), 7.13 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 7.06 (dd, J = 13.5 Hz, 4.5 Hz, 1H), 6.27 $(d, J = 8.0 \text{ Hz}, 1\text{H}), 6.17 (d, J = 8.0 \text{ Hz}, 1\text{H}), 6.27 (d, J = 13.0 \text{ Hz}, 1\text{H}), 6.25 (d, J = 1.75 \text{ Hz}, 1\text{H}), 6.17 (d, J = 1.75 \text{ Hz}, 1\text{Hz}, 1\text{$ 5.69 (d, J = 2.5 Hz, 1H), 5.23 (d, J = 2.5 Hz, 1H), 4.83 (d, J = 5.0 Hz, 1H), 4.23 (m, 1H), 3.39 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J= 16.0 Hz, 3.5 Hz, 1H), 1.95 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.74 - 1.59 (m, 5H), 1.47 (s, 3H), 1.40 (m, 2H), 1.12 (d, J = 7.5 Hz, 3H), 1.10 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 201.5, 169.7, 166.2, 166.0, 162.5, 148.6, 140.9, 139.4, 138.5, 133.4, 131.7, 129.2, 128.3, 122.4, 121.1, 115.7, 83.7, 81.7, 79.3, 72.1, 61.1, 60.4, 53.6, 52.2, 48.1, 39.7, 37.4, 35.5, 34.6, 33.0, 24.6, 21.6, 21.0, 20.6, 17.7, 16.0, 15.9, 12.6; HRMS: $[M + NH_4]^+$ calcd for $C_{37}H_{46}NO_{10}SNa$ 680.2888, found 580.2891; purity 99.39% (MeOH/H₂O = 75/25, R_t = 10.115 min).

(8i). To a solution of the cinnamic acid (14.8 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous
CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and

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1	the mixture was stirred at 25 °C for 24h. 8i was obtained as a yellow powder, 12.4 mg, yield 37
2	%; m.p. 104-106 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.65 (d, <i>J</i> = 13.5 Hz, 1H), 7.51 (m, 2H), 7.38
3	(m, 2H), 7.15 (dd, <i>J</i> = 10.0 Hz, 6.0 Hz, 1H), 6.39 (d, <i>J</i> = 13.5 Hz, 1H), 6.28 (d, <i>J</i> = 8.25 Hz, 1H),
4	6.27 (d, <i>J</i> = 13.0 Hz, 1H), 5.69 (d, <i>J</i> = 2.5 Hz, 1H), 5.23 (d, <i>J</i> = 2.5 Hz, 1H), 4.84 (d, <i>J</i> = 5.5 Hz,
5	1H), 4.23 (m, 1H), 3.40 (brs, 1H), 2.57 (dt, <i>J</i> = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, <i>J</i> =
6	16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 1H), 1.95 (s, 3H), 1.93 (s, 3H), 1.86 (s,
7	3H), 1.83 (m, 1H), 1.74 -1.59 (m, 5H), 1.48 (s, 3H), 1.40 (m, 2H), 1.12 (d, <i>J</i> = 7.5 Hz, 3H), 1.10
8	(s, 3H); ¹³ C NMR (125 MHz, CDCl ₃) δ 201.5, 169.7, 166.2, 166.1, 162.5, 148.6, 140.9, 134.2,
9	133.4, 131.7, 130.8, 129.1, 128.4, 122.4, 121.1, 117.1, 83.7, 81.6, 79.3, 72.1, 61.1, 60.4, 53.6, 52.2,
10	48.1, 39.7, 37.4, 35.4, 34.6, 33.0, 24.6, 21.5, 21.0, 20.6, 17.7, 16.0, 15.8, 12.6; HRMS: [M + Na] ⁺
11	calcd for $C_{39}H_{44}O_9Na$ 679.2878, found 679.2880; purity 97.71% (MeOH/H ₂ O = 75/25, R _t = 13.904)
12	min).

13 (8). To a solution of the 4-Fluorocinnamic acid (16.6 mg), DMAP (4 mg) and EDCI (200 mg) in 14 anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 15 mg), and the mixture was stirred at 25 °C for 24h. 8j was obtained as a yellow powder, 17.9 mg, 16 yield 53.0 %; m.p. 127-129 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.61 (d, J = 13.5 Hz, 1H), 7.50 (m, 2H), 7.16 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 7.08 (m, 2H), 6.29 (m, 2H), 5.69 (d, J = 2.5 Hz, 1H), 5.23 17 (d, J = 2.5 Hz, 1H), 4.84 (d, J = 5.5 Hz, 1H), 4.23 (m, 1H), 3.40 (brs, 1H), 2.57 (dt, J = 14.0 Hz), 100 Hz18 19 3.0 Hz, 1H, 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, J = 16.0 Hz, 3.5 Hz, 100 Hz))} 20 1H), 1.95 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.74 -1.59 (m, 5H), 1.47 (s, 3H), 1.40 21 (m, 2H), 1.12 (d, J = 7.5 Hz, 3H), 1.10 (s, 3H); ¹³C NMR (125MHz, CDCl₃) δ 201.5, 169.7, 166.2, 22 166.0, 162.5, 148.5, 144.9, 140.8, 133.4, 130.3, 130.3, 122.4, 121.1, 116.9, 116.3, 116.2, 83.7, 81.7, 79.4, 72.2, 61.1, 60.5, 52.2, 48.1, 39.7, 37.4, 35.5, 34.6, 33.1, 24.6, 21.5, 21.0, 20.6, 17.7, 23

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16.0, 15.8, 12.6; HRMS: [M + Na]⁺ calcd for C₃₉H₄₃FO₉Na 697.2783, found 697.2781; purity
 2 95.36% (MeOH/H₂O = 75/25, R_t = 13.000 min).

3 (8k). To a solution of the 3- (Trifluoromethyl)cinnamic acid (21.6 mg), DMAP (4 mg) and EDCI 4 (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then 5 add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 8k was obtained as a yellow powder, 6 24.3 mg, yield 67 %; m.p. 129-131 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.74 (s, 1H), 7.67 (m, 3H), 7 7.52 (m, 1H), 7.14 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.45 (d, J = 13.5 Hz, 1H), 6.29 (d, J = 10.0 Hz, 8 1H), 5.69 (d, J = 2.5 Hz, 1H), 5.24 (d, J = 2.5 Hz, 1H), 4.87 (d, J = 5.5 Hz, 1H), 4.23 (m, 1H), 9 3.40 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 10 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 1H), 1.95 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H),11 1.74 -1.59 (m, 5H), 1.48 (s, 3H), 1.40 (m, 2H), 1.12 (d, J = 7.5 Hz, 3H), 1.10 (s, 3H); ¹³C NMR 12 $(125 \text{ MHz}, \text{CDCl}_3) \delta 201.4, 169.7, 166.2, 165.6, 162.5, 148.5, 144.3, 140.5, 135.0, 133.6, 131.3, 140.5, 1$ 13 129.7, 124.9, 122.4, 121.1, 119.2, 100.1, 83.7, 81.6, 79.4, 72.4, 61.2, 60.4, 53.6, 52.2, 48.1, 39.7, 14 37.4, 35.5, 34.6, 33.1, 29.9, 24.6, 21.5, 21.0, 20.6, 17.7, 16.0, 15.9, 12.6; HRMS: [M + Na]⁺ calcd for $C_{40}H_{43}F_{3}O_{9}Na$ 747.2751, found 747.2750; purity 96.89% (MeOH/H₂O = 75/25, R_t = 12.144) 15 16 min).

17 (81). To a solution of the methacrylic acid (8.6 mg), DMAP (4 mg) and EDCI (200 mg) in 18 anhydrous CH_2Cl_2 , add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 19 mg), and the mixture was stirred at 25 °C for 24h. 8I was obtained as a white powder, 14.0 mg, 20 yield 47 %; m.p. 105-107 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.12 (dd, J = 10.0 Hz, 6.0 Hz,1H), 21 6.26 (d, J = 10.0 Hz, 1H), 6.08 (s, 1H), 5.69 (d, J = 2.5 Hz, 1H), 5.60 (s, 1H), 5.23 (d, J = 2.5 Hz, 22 1H), 4.77 (d, J = 10.0 Hz, 1H), 4.23 (m, 1H), 3.37 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 23 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (

s, 3H), 1.92 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.74 -1.59 (m, 5H), 1.43 (s, 3H), 1.40 (m, 2H), 1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 201.4,169.7, 166.6, 166.2, 162.5, 148.5, 140.8, 135.6, 133.4, 127.0, 122.4, 121.1, 83.7, 81.7, 79.3, 72.3, 61.0, 60.41, 53.6, 52.2, 48.0, 39.7, 37.4, 35.4, 34.6, 33.0, 29.9, 24.6, 21.5, 21.0, 20.6, 18.5, 17.7, 16.0, 15.6, 12.6; HRMS: [M + Na]⁺ calcd for C₃₈H₄₇NO₁₀Na 617.2721, found 617.2725; purity 95.77% (MeOH/H₂O = 60/40-100/0, R_t = 11.96 min).

(9a). To a solution of the 1 (26.3 mg), DMAP (4 mg) in anhydrous CH₂Cl₂ was added dimethylcarbamyl chloride (0.2 mL) and TEA (0.5 mL), and the mixture was stirred at 0 °C for 12 h. 9a was obtained as a white powder, 22.4 mg, yield 75 %; m.p. 109-111 °C; ¹H NMR (500 MHz, $CDCl_3$) δ 7.07 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.23 (d, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.64 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.35 (brs, 1H), 2.89 (s, 3H), 2.87(s, 3H), 2.57 (dt, J = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 1H 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.60 (s, 3H), 1.41 (m, 2H), 1.40 (s, 3H), 1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125) MHz, CDCl₃) δ 201.7, 169.7, 166.2, 162.4, 155.6, 148.5, 142.1, 132.9, 122.4, 121.2, 83.7, 81.7, 79.3, 72.6, 61.0, 60.5, 52.2, 48.0, 39.6, 37.4, 36.8, 36.3, 33.0, 24.6, 21.5, 21.0, 20.6, 17.6, 16.0, 15.5, 12.6; HRMS: $[M + Na]^+$ calcd for C₃₃H₄₃NO₉Na 620.2830, found 620.2830; purity 95.26% $(MeOH/H_2O = 60/40-100/0, R_t = 8.098 min).$

19 (9b). To a solution of the 1 (26.3 mg), DMAP (4 mg) in anhydrous CH_2Cl_2 was added DL-2-20 Methylbutyryl chloride (0.2 mL) and TEA (0.5 mL), and the mixture was stirred at 0 °C for 12 h. 21 9b was obtained as a white powder, 21.4 mg, yield 70 %; m.p. 111-113 °C; ¹H NMR (500 MHz, 22 $CDCl_3$) δ 7.12 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.23 (d, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 23 5.22 (d, J = 2.5 Hz, 1H), 4.62 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.34 (brs, 1H), 2.84 (m, 4H), 2.57

1	(dt, J = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H)
2	Hz, 1H), 1.93 (s, 3H), 1.92 (s, 3H), 1.85 (s, 3H), 1.75 -1.59 (m, 5H), 1.38 (m, 6H), 1.10 (m, 12H);
3	$^{13}\mathrm{C}$ NMR (125 MHz, CDCl ₃) δ 201.7, 169.7, 166.4, 162.4, 155.3, 148.7, 142.6, 132.8, 130.0,
4	122.3, 121.0, 83.7, 81.6, 79.3, 72.5, 61.0, 60.5, 53.6, 52.2, 48.0, 39.6, 37.4, 36.8, 36.3, 34.6, 33.0,
5	29.8, 24.5, 21.5, 20.6, 17.7, 16.0, 15.3, 14.3, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{34}H_{45}NO_9Na$
6	634.2987, found 634.2988; purity 97.09% (MeOH/H ₂ O = $60/40-100/0$, R _t = 10.990 min).

(9c). To a solution of the 1 (26.3 mg), DMAP (4 mg) in anhydrous CH₂Cl₂ was added diethylcarbamyl chloride (0.2 mL) and TEA (0.5 mL), and the mixture was stirred at 0 °C for 12 h. 9c was obtained as a white powder, 22.4 mg, yield 73 %; m.p.: 224-225 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.15 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.23 (d, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.63 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.35 (brs, 1H), 3.04 (m, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.63 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.35 (brs, 1H), 3.04 (m, 1H), 5.22 (d, J = 2.5 Hz, 1H), 4.63 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 5.25 (brs, 1H), 5.24 (m, 1H), 5.25 (brs, 1H), 5.24 (m, 1H), 5.25 (brs, 1H) 4H), 2.57 (dt, J = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.85 (s, 3H), 1.83 (m, 1H), 1.75 - 1.59 (m, 5H), 1.48 (t, J = 7.5 Hz, 6H), 1.41 (m, 2H), 1.38 (s, 3H), 1.11 (d, J = 7.5 Hz, 3H), 1.08 (s, 3H); ¹³C NMR (125) MHz, CDCl₃) δ 201.77, 169.69, 166.32, 162.42, 154.95, 148.62, 142.36, 132.75, 122.36, 121.05, 83.72, 81.66, 79.34, 72.47, 61.03, 60.54, 52.20, 48.00, 42.35, 39.66, 37.46, 35.52, 34.70, 33.08, 32.07, 29.84, 27.36, 24.60, 22.83, 21.47, 20.99, 20.58, 17.72, 16.00, 15.27, 14.25, 12.58, 11.37; HRMS: $[M + H]^+$ calcd for C₃₅H₄₈NO₉ 626.3324, found 626.3322; purity 95.66% (MeOH/H₂O = 60/40-100/0, $R_t = 11.005$ min).

20 (9d). To a solution of the 1 (26.3 mg), DMAP (4 mg) in anhydrous CH_2Cl_2 was added 1-21 Pyrrolidinecarbonyl chloride (0.2 mL) and TEA (0.5 mL), and the mixture was stirred at 0 °C for 22 12 h. 9d was obtained as a white powder, 16.2 mg, yield 52 %; m.p. 98-100 °C; ¹H NMR (500 23 MHz, CDCl₃) δ 7.15 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.23 (d, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz,

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1	1H), 5.22 (d, <i>J</i> = 2.5 Hz, 1H), 4.63 (d, <i>J</i> = 6.0 Hz, 1H), 4.23 (m, 1H), 3.38-3.23 (m, 5H), 2.57 (dt,
2	<i>J</i> = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, <i>J</i> = 16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz, 3.5 Hz,
3	1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 5H), 1.75 -1.59 (m, 5H), 1.41 (m, 2H), 1.40
4	(s, 3H), 1.12 (d, <i>J</i> =7.5 Hz, 3H), 1.09 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃) δ 201.8, 169.7, 166.2,
5	162.5, 153.9, 148.5, 142.3, 132.8, 122.4, 121.1, 83.7, 81.7, 79.3, 72.3, 60.9, 60.6, 52.2, 48.0, 46.5,
6	46.2, 39.7, 37.4, 35.5, 34.7, 33.0, 32.1, 29.9, 25.8, 25.1, 24.6, 22.8, 21.5, 21.0, 20.6, 17.7, 16.0,
7	15.4, 14.3, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{35}H_{45}NO_9Na$ 646.2987, found 646.2989; purity
8	95.24% (MeOH/H ₂ O = 60/40-100/0, R_t = 12.44 min).

9 (9e). To a solution of the 1 (26.3 mg), DMAP (4 mg) in anhydrous CH₂Cl₂ was added piperidine-10 1-carbonyl chloride (0.2 mL) and TEA (0.5 mL), and the mixture was stirred at 0 °C for 12 h. 9e was obtained as a white powder, 12.4 mg, yield 39 %; m.p. 97-99 °C;¹H NMR (500 MHz, CDCl₃) 11 12 δ 7.12 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.23 (d, J = 10.0 Hz, 1H), 5.68 (d, J = 2.5 Hz, 1H), 5.22 (d, 13 J = 2.5 Hz, 1H), 4.68 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.39-3.35 (m, 5H), 2.57 (dt, J = 14.0 Hz, 14 3.5 Hz, 1H, 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, J = 16.0 Hz, 3.5 Hz, 100 Hz))} 15 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.75 -1.59 (m, 11H), 1.41 (m, 2H), 1.38 (s, 3H), 1.12 (d, J = 7.5 16 Hz, 3H), 1.08 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 201.8, 169.7, 166.2, 162.4, 154.5, 148.5, 17 142.2, 132.8, 130.8, 122.4, 121.1, 83.7, 81.7, 79.3, 72.4, 61.6, 60.6, 52.2, 48.0, 45.3, 39.7, 37.5, 35.5, 34.7, 33.0, 29.9, 27.4, 24.6, 24.4, 22.8, 21.5, 21.1, 20.6, 17.7, 16.0, 15.6, 14.3, 12.6; HRMS: 18 19 $[M + Na]^+$ calcd for C₃₆H₄₇NO₉Na 660.3143, found 660.3145; purity 97.48% (MeOH/H₂O = 20 60/40-100/0, R_t = 12.493 min).

(9f). To a solution of the 1 (26.3 mg), DMAP (4 mg) in anhydrous CH₂Cl₂ was added 4Morpholinecarbonyl chloride (0.2 mL) and TEA (0.5 mL), and the mixture was stirred at 0 °C for
12 h. 9f was obtained as a white powder, 13.4 mg, yield 42 %.; m.p. 102-104 °C; ¹H NMR (500

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3 4	1	MHz, CDCl ₃) δ 7.11 (dd, <i>J</i> = 10.0 Hz, 6.0 Hz, 1H), 6.25 (d, <i>J</i> = 10.0 Hz, 1H), 5.69 (d, <i>J</i> = 2.5 Hz,
5 6	2	1H), 5.22 (d, <i>J</i> = 2.5 Hz, 1H), 4.71 (d, <i>J</i> = 6.0 Hz, 1H), 4.23 (m, 1H), 3.67 (m, 4H), 3.43 (m, 4H),
7 8	3	3.36 (s, 1H), 2.57 (dt, <i>J</i> = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, <i>J</i> = 16.0 Hz, 1H), 2.17 (dd,
9 10 11	4	J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.75 -1.59 (m, 5H), 1.41 (m,
12 13	5	2H), 1.36 (s, 3H), 1.12 (d, J =7.5 Hz, 3H), 1.09 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃) δ 201.6,
14 15	6	169.7, 162.5, 154.5, 148.5, 141.5, 133.1, 130.8, 122.4, 121.1, 83.7, 81.7, 79.4, 72.8, 61.3, 60.5,
16 17 18	7	52.2, 48.0, 39.6, 37.5, 35.5, 34.6, 33.1, 32.1, 31.6, 29.9, 29.8, 29.5, 29.4, 27.4, 24.6, 22.9, 21.5,
19 20	8	21.1, 20.6, 17.8, 16.0, 15.7, 14.3, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{35}H_{45}NO_{10}Na$ 662.2936,
21 22	9	found 662.2939; purity 99.50% (MeOH/H ₂ O = 60/40-100/0, R_t = 16.715 min).
23 24		
25 26	10	(10a). To a solution of the BOC-Glycine (17.5 mg), DMAP (4 mg) and EDCI (200 mg) in
27 28	11	anhydrous $CH_2Cl_{2,}$ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3
29 30	12	mg), and the mixture was stirred at 25 °C for 24h. 10a was obtained as a white powder, 20.4 mg,
31 32	13	yield 64 %; m.p. 115-117 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.02 (dd, J = 10.0 Hz, 6.0 Hz, 1H),
33 34 35	14	6.26 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 3 Hz, 1H), 5.21 (d, J = 3 Hz, 1H), 4.94 (t, J = 5.0 Hz, 1H),
36 37	15	4.81 (d, <i>J</i> = 6.0 Hz, 1H), 4.23 (m, 1H), 3.92 (d, <i>J</i> = 5.0 Hz, 2H), 3.35 (brs, 1H), 2.57 (dt, <i>J</i> = 14.0
38 39	16	Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, <i>J</i> = 16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz, 3.5 Hz, 1H), 2.04
40 41 42	17	(s, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 4H), 1.41 (m, 2H),
42 43 44	18	1.43 (s, 9H), 1.40 (s, 3H), 1.12 (d, J =7.5 Hz, 3H), 1.08 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ
45 46	19	201.6, 169.7, 166.2, 162.5, 155.7, 148.6, 133.7, 122.3, 121.0, 83.7, 81.6, 79.4, 72.6, 61.4, 60.2,
47 48	20	52.1, 48.0, 39.6, 37.4, 35.4, 34.5, 33.0, 24.5, 21.5, 21.2, 20.6, 17.7, 16.0, 14.3, 12.6; HRMS: [M+
49 50 51	21	NH_4] ⁺ calcd for $C_{37}H_{53}N_2O_{11}$ 701.3644, found 701.3644; purity 96.63% (MeOH/H ₂ O = 75/25, R _t)
52 53	22	= 6.355 min).
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1	(10b). To a solution of the Boc-L-alanine (18.9 mg), DMAP (4 mg) and EDCI (200 mg) in
2	anhydrous CH_2Cl_2 , add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3
3	mg), and the mixture was stirred at 25 °C for 24h. 10b was obtained as a white powder, 23.7 mg,
4	yield 68 %; m.p. 96-98 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.02 (dd, <i>J</i> = 10.0 Hz, 6.0 Hz, 1H), 6.26
5	(d, J = 10.5 Hz, 1H), 5.68 (d, J = 3 Hz, 1H), 5.21 (d, J = 3 Hz, 1H), 5.03 (t, J = 5.0 Hz, 1H), 4.81
6	(d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.38 (m, 1H), 3.37 (brs, 1H), 2.60-2.50 (m, 5H), 2.40 (t, J =
7	16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s,
8	3H), 1.83 (m, 1H), 1.75 -1.59 (m, 4H), 1.43 (s, 9H), 1.41 (m, 2H), 1.40 (s, 3H), 1.12 (d, <i>J</i> =7.5 Hz,
9	3H), 1.08 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃), <i>δ</i> 201.5, 171.7, 169.7, 166.2, 162.5, 155.9, 148.6,
10	140.2, 133.3, 122.4, 121.0, 83.7, 81.6, 79.6, 79.4, 71.8, 61.8, 60.4, 52.2, 48.0, 39.5, 37.4, 36.3,
11	35.5, 34.8, 34.5, 33.1, 28.5, 24.6, 21.5, 21.3, 20.6, 17.7, 16.1, 16.0, 12.6; HRMS: [M + Na] ⁺ calcd
12	for $C_{38}H_{51}NO_{11}Na$ 720.3354, found 720.3358; purity 98.88% (MeOH/H ₂ O = 75/25, R _t = 6.389)
13	min).

(10c). To a solution of the Boc-L-valine (21.7 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 10c was obtained as a white powder, 24.3 mg, yield 67 %; m.p. 98-99 °C; ¹H NMR (500 MHz, CDCl₃) δ 6.99 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.25 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 3 Hz, 1H), 5.22 (d, J = 3 Hz, 1H), 4.84 (dd, J = 23.0 Hz, 1H)6.0 Hz, 1H), 4.29 (m, 1H), 4.23 (m, 1H), 3.35 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.93(s, 3H), 1.94 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.50 (m, 1H), 1.43 (s, 9H), 1.40 (s, 3H), 1.39 (m, 2H),1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H), 0.95 (m, 6H); ¹³C NMR (125 MHz, CDCl₃), δ 201.6, 172.8, 169.7, 166.2, 162.5, 148.5, 140.0, 133.4, 122.4, 121.1, 83.7, 81.6, 79.4, 72.0, 61.9, 60.3,

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3 4	1	53.6,52.4, 52.2, 47.9, 42.0, 39.7, 37.5, 35.5, 34.6, 33.1, 29.9, 29.5, 28.5, 24.9, 24.6, 23.1, 22.0,
5 6	2	21.6, 21.5, 20.6, 17.7, 16.3, 16.0, 12.6; HRMS: $[M + NH_4]^+$ calcd for $C_{40}H_{59}N_2O_{11}$ 743.4113,
7 8 9	3	found 743.4117; purity 97.18% (MeOH/H ₂ O = 75/25, R_t = 6.394 min).
10 11 12	4	(10d). To a solution of the Boc-L-Isoleucine (23.1 mg), DMAP (4 mg) and EDCI (200 mg) in
13 14	5	anhydrous CH ₂ Cl ₂ , add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3
15 16	6	mg), and the mixture was stirred at 25 °C for 24h. 10d was obtained as a white powder, 26.2 mg,
17 18 19	7	yield 71 %; m.p. 110-112 °C; ¹ H NMR (500 MHz, CHCl ₃) δ 6.99 (dd, J = 10.0 Hz, 6.0 Hz, 1H),
20 21	8	6.25 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 3 Hz, 1H), 5.22 (d, J = 3.0 Hz, 1H), 4.95 (d, J = 10.0 Hz,
22 23	9	1H), 4.84 (d, <i>J</i> = 5.5 Hz, 1H), 4.29 (m, 1H), 4.23 (m, 1H), 3.35 (brs, 1H), 2.57 (dt, <i>J</i> = 14.0 Hz,
24 25	10	3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s,
26 27 28	11	3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 2H), 1.75 -1.59 (m, 4H), 1.50 (m, 1H), 1.44 (s, 9H), 1.42
28 29 30	12	(m, 2H), 1.40 (s, 3H), 1.39 (m, 2H), 1.12 (d, <i>J</i> =7.5 Hz, 3H), 1.09 (s, 3H), 0.93 (m, 6H); ¹³ C NMR
31 32	13	(125 MHz, CDCl ₃), δ 201.6, 172.8, 169.7, 166.2, 162.5, 148.5, 139.9, 133.5, 122.4, 121.1, 83.7,
33 34	14	81.6, 79.4, 72.2, 61.8, 60.1, 58.2, 52.2, 47.9, 39.7, 38.2, 37.5, 35.5, 34.6, 33.1, 29.9, 28.5, 25.0,
35 36 37	15	24.6, 21.5, 20.6, 17.7, 16.2, 16.0, 15.7, 12.6, 11.9; HRMS: $[M + NH_4]^+$ calcd for $C_{41}H_{61}N_2O_{11}$
38 39	16	757.4270, found 757.4274; purity 96.03% (MeOH/H ₂ O = 75/25, R_t = 7.444 min).
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(10e). To a solution of the rac-Boc Norvaline (21.7 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 10e was obtained as a white powder, 22.8 mg, yield 73 %; m.p. 109-111 °C; ¹H NMR (500 MHz, CDCl₃) δ 6.99 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.25 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 3.0 Hz, 1H), 5.22 (d, J = 3.0 Hz, 1H), 4.94 (d, J = 10.0 Hz, 1H)1H), 4.88 (d, J = 5.5 Hz, 1H), 4.26 (m, 1H), 4.23 (m, 1H), 3.37 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.22 (d, J = 7.5 Hz, 2H), 2.17 (dd, J = 16.0

1	Hz, 3.5 Hz, 1H), 2.01 (m, 2H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59
2	(m, 5H), 1.50 (m, 1H), 1.44 (s, 9H), 1.42 (m, 1H), 1.39 (s, 3H), 1.12 (d, <i>J</i> =7.5 Hz, 3H), 1.09 (s,
3	3H), 0.88 (t, $J = 6.5$ Hz, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ 201.5, 172.5, 169.7, 166.2, 162.5,
4	148.5, 139.9, 133.5, 130.1, 130.0, 122.4, 121.1, 83.7, 81.6, 79.4, 72.3, 71.2, 61.9, 61.0, 60.3, 52.2,
5	47.9, 39.7, 37.5, 36.1, 35.5, 34.6, 33.1, 32.1, 31.5, 29.9, 29.9, 29.8, 29.5, 29.4, 28.5, 27.4, 25.7,
6	$24.6, 22.8, 21.5, 20.6, 19.2, 17.7, 17.6, 16.0, 14.3, 12.6; HRMS: [M + NH_4]^+ calcd for C_{40}H_{59}N_2O_{11}$
7	743.4113, found 743.4115; purity 95.13% (MeOH/H ₂ O = 75/25, R_t = 7.441 min).

(10f). To a solution of the N-Boc-proline (21.5 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂ add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 10f was obtained as a white powder, 14.8 mg, yield 41 %; m.p. 104-106 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.02 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.24 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 3.0 Hz, 1H), 5.22 (d, J = 3.0 Hz, 1H), 4.78 (d, J = 5.5 Hz, 1H), 4.23 (m, 1H), 3.37 (m, 4H), 2.57 (dt, J = 14.0 Hz, 3.5 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0Hz, 1H), 2.22 (d, J = 7.5 Hz, 2H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.01 (m, 2H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.50 (m, 1H), 1.44 (s, 9H), 1.42 (m, 1H), 1.40 (s, 9H), 1.38 (s, 3H), 1.12 (d, J = 7.5 Hz, 3H), 1.08 (s, 3H); ¹³C NMR (125 MHz, CDCl₃), δ 201.5, 172.5, 169.6, 166.2, 162.6, 155.7, 148.5, 140.5, 140.2, 133.5, 130.1, 130.0, 122.4, 121.1, 83.7, 81.6, 80.3, 79.4, 72.3, 71.2, 61.9, 61.0, 60.3, 59.2, 59.0, 52.2, 48.1, 46.7, 46.5, 39.6, 37.5, 35.5, 34.6, 33.2, 31.2, 29.9, 28.6, 28.5, 27.4, 25.7, 24.6, 23.7, 21.5, 20.6, 17.8, 16.0, 15.8, 14.3, 12.6; HRMS: $[M + Na]^+$ calcd for C₄₀H₅₃NO₁₁Na 746.3511, found 746.3513; purity 96.92% $(MeOH/H_2O = 60-100, R_t = 12.259 min).$

(11). To a solution of the 2-thienylacetic acid (14.2 mg), DMAP (4 mg) and EDCI (200 mg) in
anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3

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3 4	1	mg), and the mixture was stirred at 25 °C for 24h. 11 was obtained as a white powder, 12.0 mg,
5 6	2	yield 37 %; m.p. 86-88 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.11 (dd, <i>J</i> = 10.0 Hz, 6.0 Hz, 1H), 6.38
7 8 9	3	(d, J = 15.0 Hz, 1H), 6.26 (d, J = 10.5 Hz, 1H), 6.10 (m, 1H), 5.87 (d, J = 10.0Hz, 1H), 5.69 (d, J
10 11	4	= 3.0 Hz, 1H), 5.23 (d, J = 3.0 Hz, 1H), 4.79 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.37 (brs, 1H),
12 13	5	2.57 (dt, <i>J</i> = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, <i>J</i> = 16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz,
14 15 16	6	3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.60 (m, 5H), 1.59 (s,
17 18	7	2H), 1.42 (s, 3H), 1.39 (m, 2H), 1.12 (d, $J = 7.5$ Hz, 3H), 1.09 (s, 3H); ¹³ C NMR (125 MHz,
19 20	8	CDCl ₃), δ 201.4, 169.7, 166.2, 165.3, 162.5, 148.6, 140.6, 133.5, 132.1, 127.8, 122.4, 121.1, 83.7,
21 22	9	81.6, 79.4, 72.2, 61.1, 60.4, 52.2, 48.1, 39.7, 37.4, 35.5, 34.6, 33.0, 29.9, 29.5, 27.4, 24.6, 21.5,
23 24 25	10	21.0, 20.6, 17.7, 16.0, 15.7, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{36}H_{42}O_9Na$, 673.2442, found
26 27	11	673.2445; purity 95.17% (MeOH/H ₂ O = $60/40-100/0$, R _t = 15.069 min).
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29 30	12	(12). To a solution of the 3-phenylpropionic acid (15.0 mg), DMAP (4 mg) and EDCI (200 mg) in
31 32	13	anhydrous CH_2Cl_2 , add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3
33 34 35	14	mg), and the mixture was stirred at 25 °C for 24h. 12 was obtained as a white powder, 10.3 mg,
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10.3 mg, yield 32 %; m.p. 102-104 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.29 (m, 2H), 7.19 (m, 3H), 7.02 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.21 (d, J = 10.5 Hz, 1H), 5.69 (d, J = 3.0 Hz, 1H), 5.22 (d, J = 3.0 Hz, 1H), 4.72 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.34 (brs, 1H), 2.92 (d, J = 7.5 Hz, 2H), 2.57 (m, 3H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.41 (m, 1H), 1.39 (s, 3H), 1.38 (m, 1H), 1.12 $(d, J = 7.5 \text{ Hz}, 3\text{H}), 1.08 (s, 3\text{H}); {}^{13}\text{C} \text{ NMR} (500 \text{ MHz}, \text{CHCl}_3), \delta 201.4, 172.0, 169.7, 166.2, 162.5, 162.5)$ 148.5, 140.6, 140.1, 133.3, 128.7, 128.4, 126.5, 122.4, 121.1, 83.7, 81.7, 79.4, 72.1, 61.1, 60.3, 52.2, 48.0, 39.7, 37.4, 35.8, 34.6, 33.1, 30.9, 29.9, 24.6, 21.5, 21.0, 20.6, 17.7, 16.0, 15.7, 12.6;

1 HRMS: $[M + Na]^+$ calcd for C₃₉H₄₆O₉ 681.3034, found 681.3032; purity 95.67% (MeOH/H₂O = 2 60/40-100/0, R_t = 17.449 min).

3 General Procedure for the Synthesis of 13a-13f.

Maleic anhydride was used as the starting material, the corresponding amino unsaturated
carboxylic acids were obtained by reaction with different amines, and then react with 1 to get the
target compound.

(13a). To a solution of the 4-(dimethylamino)-4-oxobut-2-enoic acid (14.3 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 13a was obtained as a white powder, 22.1 mg, yield 68 %; m.p. 121-123 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.37 (d, J = 13.5 Hz, 1H), 7.07 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.74 (d, J = 13.5 Hz, 1H), 6.27 (d, J = 13.5 Hz, 1H), 5.69 (d, J = 3.0 Hz, 1H), 5.22 (d, J = 3 Hz, 1H), 4.82 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.38 (brs, 1H), 3.10 (s, 3H), 3.03 (s, 3H), 2.57 (m, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 - 1.59 (m, 1H), 1.95 (m, 1H), 1.95 (m, 1H), 1.75 - 1.59 (m, 1H), 1.595H), 1.42 (s, 3H), 1.38 (m, 1H), 1.12 (d, J=7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125 MHz, CDCl₃), 8 201.2, 169.7, 166.3, 164.9, 164.4, 162.6, 148.6, 140.0, 135.1, 133.8, 130.0, 122.4, 121.0, 83.7, 81.6, 79.4, 72.7, 61.2, 60.3, 53.6, 52.2, 48.1, 39.7, 37.6, 35.9, 35.5, 34.5, 33.1, 29.9, 24.5, 22.8, 21.5, 21.0, 20.6, 17.8, 16.0, 15.8, 14.3; HRMS: $[M + Na]^+$ calcd for $C_{36}H_{45}NO_{10}Na$ 674.2936, found 674.2937; purity 97.66% (MeOH/H₂O = 75/25, R_t = 3.169 min).

(13b). To a solution of the 4-(dimethylamino)-4-oxobut-2-enoic acid (14.3 mg), DMAP (4 mg)
and EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C
for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 13b was obtained as a

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53	4
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1	white powder, 19.0 mg, yield 56 %; m.p. 123-125 °C; ¹ H NMR (500 MHz, CDCl ₃) δ 7.21 (d, $J =$
2	12.5 Hz, 1H), 7.02 (dd, <i>J</i> = 10.0 Hz, 6.0 Hz, 1H), 6.82 (d, <i>J</i> = 12.5 Hz, 1H), 6.27 (d, <i>J</i> = 10.5 Hz,
3	1H), 5.69 (d, <i>J</i> = 3.0 Hz, 1H), 5.22 (d, <i>J</i> = 3.0 Hz, 1H), 4.82 (d, <i>J</i> = 6.0 Hz, 1H), 4.23 (m, 1H), 3.55
4	(q, J = 6.0 Hz, 4H), 3.38 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J =
5	16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz, 3.5 Hz, 1H), 2.01 (m, 4H), 1.94 (s, 3H), 1.93 (s, 3H), 1.90
6	(m, 2H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.42 (s, 3H), 1.41 (m, 1H), 1.38 (m, 1H),
7	1.12 (d, <i>J</i> =7.5 Hz, 3H), 1.09 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ 201.2, 169.7, 166.2, 165.0,
8	162.5, 162.3, 148.6, 140.0, 135.8, 133.8, 129.8, 122.4, 121.1, 83.7, 81.6, 79.4, 72.6, 61.1, 60.3,
9	52.2, 48.1, 46.9, 46.4, 39.7, 37.4, 35.5, 34.5, 33.1, 26.2, 24.5, 24.4, 21.5, 21.0, 20.6, 17.7, 16.0,
10	15.8, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{38}H_{47}NO_{10}Na$ 702.3249, found 702.3247; purity 97.14%
11	(MeOH/H ₂ O = 75/25, R_t = 3.295 min).

(13c). To a solution of the 4-(diisopropylamino)-4-oxobut-2-enoic acid (19.9 mg), DMAP (4 mg) 2 3 and EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C 4 for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 13c was obtained as a 5 white powder, 22.6 mg, yield 64 %; m.p. 124-126 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.34 (d, J = 6 13.5 Hz, 1H), 7.08 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.60 (d, J = 13.5 Hz, 1H), 6.27 (d, J = 10.5 Hz, 7 1H), 5.69 (d, J = 3.0 Hz, 1H), 5.22 (d, J = 3.0 Hz, 1H), 4.81 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.93 (m, 2H), 3.37 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 8 9 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -20 1.59 (m, 5H), 1.42 (s, 3H), 1.41 (m, 1H), 1.38 (s, 3H), 1.37 (s, 3H), 1.25 (s, 3H), 1.24 (s, 3H), 1.12 21 $(d, J=7.5 \text{ Hz}, 3\text{H}), 1.08 (s, 3\text{H}); {}^{13}\text{C} \text{ NMR} (125 \text{ MHz}, \text{CDCl}_3), \delta 201.2, 169.7, 166.2, 165.1, 164.0, 100 \text{ MHz}, 100 \text{$ 22 162.5, 148.6, 140.1, 138.7, 133.7, 128.1, 122.4, 121.0, 83.7, 81.6, 79.4, 72.6, 61.1, 60.3, 53.6, 52.2, 48.0, 39.7, 37.4, 35.5, 34.6, 33.1, 24.6, 21.5, 17.7, 17.8, 16.0, 15.9, 15.8, 12.6; HRMS: [M + Na]⁺ 23

1 calcd for $C_{40}H_{53}NO_{10}Na$ 730.3562, found 730.3560; purity 95.62% (MeOH/H₂O = 75/25, R_t = 2.990 min).

(13d). To a solution of the 4-oxo-4-(pyrrolidin-1-yl) but-2-enoic acid (16.9 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 13d was obtained as a white powder, 15.9 mg, yield 47 %; m.p. 126-128 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.38 (d, J = 10.5 Hz, 1H), 7.08 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.68 (d, J = 10.5 Hz, 1H), 6.28 (d, J = 7.5 Hz, 1H), 5.69 (d, J = 3.0 Hz, 1H), 5.23 (d, J = 3.0 Hz, 1H), 4.82 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.62(t, J = 5.0 Hz, 1H), 3.47(t, J = 5.0 Hz, 1H), 3.38 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H),2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 2H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.42 (s, 3H), 1.38-1.27 (m, 5H), 1.12 (d, J = 7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125 MHz, CDCl₃), δ 201.2, 169.7, 166.2, 165.0, 163.1, 162.5, 148.5, 140.0, 136.0, 133.8, 129.5, 122.4, 121.1, 83.7 81.6, 79.4, 72.6, 61.2, 60.3, 53.6, 52.2, 48.0, 47.4, 43.4, 39.7, 37.4, 35.5, 34.5, 33.1, 26.8, 25.6, 24.6, 22.9, 21.5, 21.02, 20.6, 17.7, 16.0, 15.8, 14.3, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{38}H_{47}NO_{10}Na$ 700.3092, found 700.3097; purity 95.21% (MeOH/H₂O = 75/25, R_t = 5.243 min).

(13e). To a solution of the 4-oxo-4-(piperidin-1-vl)but-2-enoic acid (18.3 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 13e was obtained as a white powder, 21.1 mg, yield 61 %; m.p. 114-116 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.06 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.60 (d, J = 10.5 Hz, 1H), 6.24 (d, J = 7.5 Hz, 1H), 5.98 (d, J = 10.5 Hz, 1H), 5.68 (d, J = 3 Hz, 1H), 5.21 (d, J = 3 Hz, 1H), 4.80 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.45 (m, 1H), 3.34 (brs, 1H), 3.25 (m, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J

1	= 16.0 Hz, 1H), 2.17 (dd, <i>J</i> = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 2H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86
2	(s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.39 (s, 3H), 1.38 (m, 1H), 1.25-1.18 (m, 6H), 1.12 (d,
3	J =7.5 Hz, 3H), 1.08 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ 201.4, 171.3, 169.7, 166.2, 165.9,
4	163.7, 162.5, 148.6, 140.3, 139.9, 133.6, 122.4, 121.9, 121.0, 83.7, 81.6, 79.4, 72.2, 61.2, 60.6,
5	60.2, 52.2, 48.0, 42.6, 39.7, 39.0, 37.4, 35.5, 34.5, 33.1, 24.6, 21.5, 21.2, 21.1, 20.6, 17.7, 16.0,
6	15.9, 14.4, 14.2, 12.7, 12.6; HRMS: $[M + Na]^+$ calcd for $C_{39}H_{49}NO_{10}Na$ 714.3249, found
7	714.3251; purity 95.49% (MeOH/H ₂ O = 75/25, R_t = 4.524 min).

(13f). To a solution of the 4-morpholino-4-oxobut-2-enoic acid (18.5 mg), DMAP (4 mg) and EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for 1h, Then add 1 (26.3 mg), and the mixture was stirred at 25 °C for 24h. 13f was obtained as a white powder, 24.3 mg, yield 70 %; m.p. 106-108 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.32 (d, J= 10.5 Hz, 1H), 7.08 (dd, J = 10.0 Hz, 6.0 Hz, 1H), 6.76 (d, J = 10.5 Hz, 1H), 6.28 (d, J = 7.5 Hz, 1H), 5.69 (d, J = 3 Hz, 1H), 5.23 (d, J = 3 Hz, 1H), 4.82 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.70 (m, 6H), 3.35(m, 2H), 3.38 (brs, 1H), 2.57 (dt, J = 14.0 Hz, 3.0 Hz, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 2.04 (s, 1H), 1.94 (s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 - 1.59 (m, 5H), 1.41 (s, 3H), 1.38 (m, 1H), 1.12 (d, J=7.5 Hz, 3H), 1.09 (s, 3H); ¹³C NMR (125 MHz, CDCl₃), δ 201.2, 169.6, 166.2, 164.7, 163.3, 162.6, 148.5, 139.8, 134.4, 133.9, 130.7, 122.4, 121.0, 83.7, 81.6, 79.5, 72.8, 66.9, 61.2, 60.3, 52.2, 48.1, 46.6, 42.6, 39.7, 37.4, 35.5, 34.5, 33.2, 24.6, 21.5, 21.1, 20.6, 17.8, 16.0, 15.9, 12.6; HRMS: [M + Na]⁺ calcd for $C_{38}H_{47}NO_{10}Na$ 716.3041, found 716.3043; purity 95.84% (MeOH/H₂O = 75/25, R_t = 4.160 min).

(13a'). To a solution of the 4-(dimethylamino)-4-oxobutanoic acid (14.5 mg), DMAP (4 mg) and
EDCI (200 mg) in anhydrous CH₂Cl₂, add 0.5 mL TEA, and the mixture was stirred at 0 °C for

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	1h, then, 1 (26.3 mg) was added, and the mixture was stirred at 25 °C for 24h. 13a' was obtained
	as a white powder, 24.5 mg, yield 75 %; m.p. 101-102 °C; ¹ H NMR (500 MHz, CDCl ₃) δ7.05 (dd,
	<i>J</i> = 10.0 Hz, 6.0 Hz, 1H), 6.24 (d, <i>J</i> = 13.5 Hz, 1H), 5.68 (d, <i>J</i> = 3.0 Hz, 1H), 5.22 (d, <i>J</i> = 3 Hz,
-	1H), 4.78 (d, J = 6.0 Hz, 1H), 4.23 (m, 1H), 3.38 (brs, 1H), 3.05 (s, 3H), 3.00 (s, 3H), 2.68 (m,
	4H), 2.57 (m, 1H), 2.50 (m, 1H), 2.40 (t, J = 16.0 Hz, 1H), 2.17 (dd, J = 16.0 Hz, 3.5 Hz, 1H), 1.94
)	(s, 3H), 1.93 (s, 3H), 1.86 (s, 3H), 1.83 (m, 1H), 1.75 -1.59 (m, 5H), 1.42 (s, 3H), 1.38 (m, 1H),
,	1.12 (d, <i>J</i> =7.5 Hz, 3H), 1.09 (s, 3H); ¹³ C NMR (125 MHz, CDCl ₃), δ 201.3, 172.1, 170.7, 169.6,
)	166.1, 162.4, 148.4, 140.5, 133.1, 122.3, 120.1, 83.6, 81.5, 79.2, 71.8, 61.2, 60.3, 52.1, 47.9, 39.6,
)	37.3, 37.0, 35.5, 35.3, 34.5, 33.0, 31.45, 29.4, 28.4, 28.1, 24.5, 21.3, 21.0, 20.4, 17.6, 15.9, 15.7,
)	12.5; HRMS: $[M + Na]^+$ calcd for $C_{36}H_{47}NO_{10}Na$ 676.3092, found 676.3096; purity 95.26%
	(MeOH/H ₂ O = 75/25, R_t = 3.496 min).

12 Cytotoxicity assays

HT-29, HCT-116, HepG2 MCF-7, L02 and HK-2 cells were obtained from the Shanghai
Institute of Biochemistry and Cell Biology, Chinese Academy of Sciences and were cultured in
DMEM or RPMI supplemented with 10% FBS, 2 mM glutamine, and 100 units/mL
penicillin/streptomycin and maintained in a humidified atmosphere of 5% CO₂ at 37°C.

17 MTT assay

18 HT-29, HCT-116, HepG2, MCF-7, L02 and HK-2 cells were seeded in 96-well plates and then 19 treated with vehicle alone or with the tested compounds for 24 h. Then, 20 μ L of MTT (5 mg/mL, 20 in PBS) was added to each well, and the samples were further incubated for 4 h. The MTT 21 formazan formed by viable cells was dissolved in DMSO (150 μ L), and absorbance was measured 22 using a microplate reader (570 nm).

23 Measurement of *in vitro* TrxR activity by DTNB Assay³⁹

TrxR activity was determined at room temperature using a microplate reader. NADPH-reduced TrxR (0.16 µM) was incubated with different concentrations of compounds for different time at room temperature (the final volume of the mixture was 50 μ L) in a 96-well plate. A master mixture in TE buffer (50 µL) containing DTNB and NADPH was added (final concentration of 2 mM and $200 \,\mu\text{M}$, respectively), and the linear increase in absorbance at 412 nm during the initial 3 min was recorded. The same amounts of DMSO (0.1%, v/v) were added to the control experiments, and the activity was expressed as the percentage of the control. The K_{inact} and K_i through time-course experiments TrxR activity was determined at room temperature using a microplate reader. Different concentrations of WA and 13a were incubated in TrxR for 0-30 min, residual enzyme activity was measured by DTNB Assay. Measurement of in vitro Trx activity by insulin reduction assay Trx activity was determined at room temperature using a microplate reader. E. coli Trx (5.0 µM) was incubated with different concentrations of WA or 13a for 1.5 h at room temperature (the final volume of the mixture was 50 μ L) in a 96-well plate. A master mixture in TE buffer (50 μ L) containing insulin and DTT (final concentrations of 0.17 mM and 2.0 mM, respectively) was added. The decrease in absorbance (A340) was recorded immediately after the initial 1-min analysis of a blank reference. Determination of TrxR, Trx, GR and GPx activity in HT-29 cells After HT-29 cells were treated with different concentrations of 13a for 24 h. The cells were harvested and washed twice with PBS. Total cellular proteins were extracted by RIPA buffer (50

22 mM Tris-HCl, pH 7.5, 2 mM EDTA, 0.5% deoxycholate, 150 mM NaCl, 1% Triton X-100, 0.1%

23 SDS, 1 mM Na₃VO₄, and 1 mM PMSF) for 30 min on ice. Total protein content was quantified

24 using the BCA procedure. TrxR activity in cell lysates was measured by the end point insulin

reduction assay. Briefly, the cell extract containing 50 μ g of total protein was incubated in a final reaction volume of 50 µL containing 100 mM Tris-HCl (pH 7.6), 0.3 mM insulin, 660 µM NADPH, 3 mM EDTA, and 15 µM E. coli Trx for 30 min at 37°C. The reaction was terminated by adding 200 µL of 1 mM DTNB in 6 M guanidine hydrochloride, pH 8.0. A blank sample, containing everything except Trx, was treated in the same manner. The absorbance at 412 nm was measured, and the blank value was subtracted from the corresponding absorbance value of the sample. The same amounts of DMSO were added to the control experiments, and the activity was expressed as the percentage of the control. Trx activity in cell lysates was measured by the end point insulin reduction assay. Cell extract containing 50 μ g of total protein was incubated in TE buffer (50 µL) containing insulin and DTT (final concentrations of 0.17 mM and 2.0 mM, respectively). The decrease in absorbance (A340) was recorded immediately after the initial 1-min analysis of a blank reference. GR assay. Cell extract containing 50 µg of total protein was incubated in TE buffer (50 mM Tris-HCl with 1 mM EDTA, pH 7.5), and oxidized glutathione (GSSG) and NADPH (50 μ L, final concentration of 1 mM and 400 μ M, respectively) was added. GR activity was determined by measuring the decrease in absorbance at 340 nm during the initial 3 min. The same amounts of DMSO were added to the control experiments, and the activity was expressed as the percentage of the control. GPx activity was measured indirectly by a coupled reaction with GR. GSSG, produced by the reduction of hydroperoxides by GPx, is recycled to its reduced state by GR and NADPH. The oxidation of NADPH to NADP⁺ is accompanied by a decrease in absorbance at 340 nm. The rate of decrease in A340 is directly proportional to GPx activity. To the wells of a 96-well microliter plate, 130 µL of TE buffer, 10 µL of a freshly prepared NADPH solution (4.0 mM in TE buffer), and 50 μ g of total protein were added. Then, 10 μ L of a baker's yeast GR solution and 10 μ L of freshly prepared GSH solution (5.0 mM in TE buffer) were

added. After the addition of 20 μ L of an H₂O₂ solution (5.0 mM in water), the final volume in each well was 200 μ L. The background of the GPx-independent NADPH oxidation rate (r1) was determined by replacing the GPx solution with TE buffer. The rate of decrease in the absorption of NADPH at 340 nm (r2) was measured for 4 min at intervals of 10 s at room temperature. The relative GPx activity was calculated by subtracting the r1 from r2 and was expressed as the percentage of the control.

7 Analysis of the mass spectrum of 13a-treated and untreated TrxR

8 Reduced-rat liver TrxR (1.0 μ M) was incubated with **13a** (10.0 μ M) at room temperature. After

9 1.5 h, the **13a**-modified protein was purified by SDS–PAGE and digested with trypsin in the gel.

10 The resulting peptides were analyzed using LC-MS/MS.

11 Mercapto binding experiments

NAC (10mg) were incubated with 13a (5mg) at room temperature, and the online NMR analysis
was developed. DTT (10mg) and NAC (10mg) were incubated with 13a (5mg) and WA(5mg) at
room temperature. After 1.5 h, the reaction liquid was analyzed using LC-MS/MS.

15 Determination of intracellular ROS

16 To assay the intracellular accumulation of ROS, HT-29 or HCT-116 cells $(3.0 \times 10^5 \text{ cells/well})$ 17 were seeded into 6-well plates. After 24 h, the cells were treated with different concentrations of 18 13a or GSH (5 mM) for an additional 24 h. The cells were collected, washed with PBS, and stained 19 with DCFH-DA. ROS were determined by flow cytometry.

20 Monoclonal formation experiment

HT-29 cells were incubated in six-well plates (5×10^2 /well) for 12 h and then treated with DMSO (1%) or various concentrations (0.0125, 0.03, 0.06 μ M) of 13a for 24 h. The added medium was removed and replaced with normal fresh medium, and the solution was changed every 2-3 days. The formation of clones was observed under an optical microscope. After 14 days, the culture was terminated when visible clones appeared in the 6-well plates. The culture solution was discarded, washed with PBS 2 times, and added to each hole 4% paraformaldehyde was fixed for 30 min. The fixing solution was discarded, 1 mL crystal violet staining solution was added to each hole for 20 min, and then PBS was used slowly wash away the stain and let dry in the air before taking photos.

7 EdU infiltration experiment

According to the manual for the EdU labeling/detection kit (Ribobio, Guangzhou, China), HT-29 cells were incubated in a final volume of 100 μ L of complete medium at 5.0 × 10³ cells per well in 96-well plates. Following incubation for 24 h, the cells were replaced with fresh medium containing 13a at different concentrations (0.125-0.500 μ M) or DMSO as a vehicle (< 0.1%). After 20 h, 100 μ L of fresh medium containing 50 mM EdU labeling agent was added to the cell culture to incubate for an additional 8 h at 37 °C under 5% CO². The cells were then fixed with 4% paraformaldehyde (pH 7.4) for 30 min and incubated with glycine for 5 min. After washing with PBS, the cells were stained with anti-EdU working solution at room temperature for 30 min. After washing with 0.5% Triton X-100 in PBS, the cells were incubated with 5 µg mL⁻¹ Hoechst 33258 at room temperature for 30 min and then observed under a High Content Imaging System (Image Xpress Micro Confocal, Molecular Devices).

19 Induced G2/M arrest in HT-29 cells

HT-29 cells were incubated in six-well plates $(3.0 \times 10^{5}/\text{ well})$ and treated with DMSO (1%) or various concentrations (0.125, 0.25, 0.5 μ M) of **13a** for 24 h. The cells were collected, washed with PBS, fixed in ice-chilled 70% EtOH for 12 h and incubated again for 30 min at 37 °C in a Page 59 of 68

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1 staining and stained with RNaseA and PI (Beyotime, Nantong, People's Republic of China).

2 Mitochondrial membrane potential was determined by flow cytometry.

3 Apoptosis analysis

HT-29 cells were incubated in six-well plates (3.0×10^5 /well) and treated with DMSO (1%) or

5 13a for 24 h. The cells were collected, washed with PBS, and stained with FITC-Annexin-V and

6 PI. Apoptosis was determined by flow cytometry.

7 Effect of 13a on the mitochondrial membrane potential of cells

8 The mitochondrial membrane potential (M.M.P) was determined by a JC-1 assay kit (Nanjing 9 KeyGen Biotech Co., Ltd., Nanjing, China). After collection, MG-63 cells were washed with PBS 10 twice, supplemented with 500 mL of JC-1 dye staining solution, and then incubated in the dark at 11 37 °C for 25 min. After incubation, the cells were centrifuged at 2000 rpm for 5 min and washed 12 twice with 1× incubation buffer. The fluorescence was then detected using flow cytometry (488 13 nm excitation and 525 nm emission filters) after the cells were resuspended in 500 μ L of 1× 14 incubation buffer.

15 Western blot analysis

HT-29 cells were incubated with various concentrations of 13a or 0.1% DMSO for 24 h. 16 17 Harvested after trypsinization, cells were treated with 1×RIPA lysis buffer (50 mM Tris-HCl, pH 7.4, 150 mM NaCl, 0.25% deoxycholic acid, 1% NP-40, 1 mM EDTA and protease inhibitors) 18 19 (Amresco, Solon, USA) to extract the total proteins. An aliquot of proteins from the total cell 20 lysates (30 to 40 μ g per lane) was separated by sodium dodecyl sulfate (8%, 12% or 15%) 21 polyacrylamide gel electrophoresis (SDS-PAGE, BioRad Laboratories, Hercules, CA), wet-22 transferred to PVDF membrane (BioRad Laboratories, Hercules, CA), blotted with primary 23 antibodies specific for TrxR, Trx, ASK1, p-ASK1, P53, Bcl-2, Bax, PARP, cleaved-PARP,

3 4	1	cleaved-caspase-3, caspase-3, and GAPDH, and probed with secondary isotype-specific antibodies
5 6	2	tagged with horseradish peroxidase (Cell Signaling Technology). Bound immunocomplexes were
7 8 9	3	detected using a ChemiDOC TM XRS + system (BioRad Laboratories, Hercules, CA).
10 11 12	4	Supporting Information
13 14 15	5	The Supporting Information is available free of charge on the ACS Publications website at xxxx.
16 17	6	Supplementary figures, NMR spectra, analytical HPLC traces and small molecule X-ray
18 19 20	7	crystal structure for 3 .
20 21 22	8	Molecular formula strings with pharmacological data (CSV)
23 24 25 26	9	AUTHOR INFORMATION
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9 ABBREVIATIONS

DCFH, dichlorodihydrofluorescin; DCFH-DA, 2',7'-dichlorofluorescin diacetate; DTT, DLdithiothreitol; FITC, fluorescein 5-isothiocyanate; GPx, glutathione peroxidase; GR, glutathione
reductase; GSH, reduced glutathione; GSSG, oxidized glutathione; MTT, 3-(4,5-dimethyl-2thiazolyl)-2,5-diphenyl-2Htetrazoliumbromide; NAC, N-acetylcysteine; PI, propidium iodide;
ROS, reactive oxygen species; Sec, selenocysteine; TE buffer, 50 mM Tris-HCl with 1 mM EDTA,
pH 7.5; Trx, thioredoxin; TrxR, thioredoxin reductase.

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