Subscriber access provided by UNIV OF DURHAM

Exploration of Biaryl Carboxylic Acids as Proton Shuttles for the Selective Functionalization of Indole C–H Bonds

Jing-Jing Pi, Xiao-Yu Lu, Jing-Hui Liu, Xi Lu, Bin Xiao, Yao Fu, and Nicolas Guimond J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.8b00417 • Publication Date (Web): 17 Apr 2018 Downloaded from http://pubs.acs.org on April 17, 2018

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

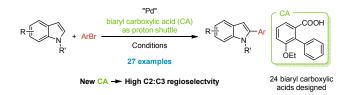
Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Exploration of Biaryl Carboxylic Acids as Proton Shuttles for the Selective Functionalization of Indole C–H Bonds

Jing-Jing Pi,[†] Xiao-Yu Lu,[†] Jing-Hui Liu,[†] Xi Lu,[†] Bin Xiao,[†] Yao Fu*[†] and Nicolas Guimond*[‡]

⁺Hefei National Laboratory for Physical Sciences at the Microscale, CAS Key Laboratory of Urban Pollutant Conversion, Anhui Province Key Laboratory of Biomass Clean Energy, iChEM, University of Science and Technology of China, Hefei 230026, China

‡Chemical Development Department, Bayer Pharma AG, Friedrich-Ebert-Str. 217-333, 42096 Wuppertal, Germany. Supporting Information Placeholder



ABSTRACT: A survey of diversely substituted 2-arylbenzoic acid were synthesized and tested for use as proton shuttle in the direct arylation of indoles with bromobenzenes. It was found that 3-ethoxy-2-phenylbenzoic acid gives superior yield and selectivity for this class of substrates.

The direct functionalization of C-H bonds leading to value-added chemicals is a very intense line of research in transition metal catalysis.¹ The possibility to access a variety of functionalized substrates without the requirement for prefunctionalization can contribute to decrease step count, waste and the time necessary to prepare a given target. Over the past couple decades, it has been demonstrated in numerous catalytic systems, (with various metals) that carboxylic acids additives can have a dramatic impact on reactivity when it comes to C-H bond functionalization.² The rationale behind most cases is that the acid would bind to the transition metal and mediate the arvl deprotonation in the so-called concertedmetallation-deprotonation (CMD) event.³ The protonated carboxylic acid is then released in the solution and transfers its proton to a stronger, typically inorganic, base. With regards to palladium, perhaps the most versatile carboxylic acid to perform this task so far is pivalic acid.⁴ It is however not the only potent additive discovered for this purpose. In addition to Yu's new system using a modified pyridone⁵ and some protected amino acids derivatives,⁶ several reports found that the use of 2-substituted benzoic acids as well as bulky aliphatic carboxylic acids can also perform very well in some instances.⁷

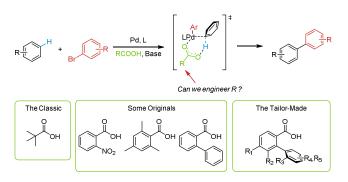
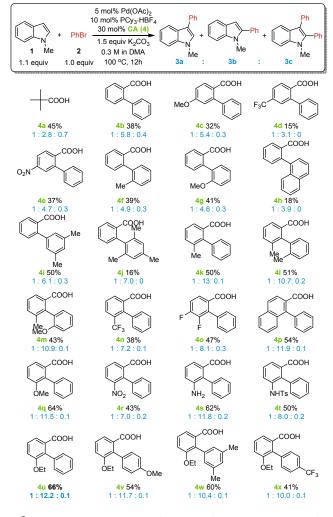


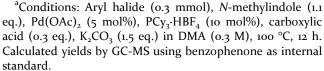
Figure 1. Role of carboxylic acids in C-H functionalization reactions

Given that most carboxylic acids that were employed for such purpose have been simple, commercially available structures, we were curious to see if one could possibly engineer a motif that would perform better, or differently than the ones most commonly employed (Figure 1).⁸ With this mindset, we set out to focus our efforts on the biaryl carboxylic acid motif. As an analogy to the Buchwald ligands, which were presumably initially designed on the premise that they would be even bulkier than the very versatile P(t-Bu)₃ ligand while being electronically tunable, we were curious to explore this motif for similar reasons. Herein, we report the synthesis and finding of a biaryl carboxylate proton shuttle that allows higher yield and selectivity for the direct arylation of indoles using aryl bromide electrophiles.⁹

In order to find a carboxylic acid that could be more efficient than pivalic acid, a system where the reaction with pivalic acid could be improved upon was necessary. For this purpose, our choice landed on the reaction of Nmethylindoles with bromobenzene.¹⁰ Given that the use of pivalic acid for this coupling gives a modest 45% yield and a 2.8:1 C2:C3 arylation regioselectivity, we thought it would be a good starting point to evaluate the performance of various biaryl carboxylic acids. Scheme 1 shows a selection of carboxylic acid we designed, synthesized and tested. We first started our investigation probing the electronic properties of the aromatic ring bearing the carboxylic acid. From this survey, it was impossible to observe any consistent trend relating to the electronic nature of the benzoic acid ring itself. It seems like both electron-rich (4c) and electron poor (4d, 4e) carboxylic acids performed worse than the naked 2-phenylbenzoic acid (4b). While trying successively various different set of substitution, we noticed that a substituent at the 3position of the benzoic acid had a pronounced positive impact on both the regioselectivity of arylation as well as the yield. Perhaps this substitution twists the second aromatic ring out of conjugation, so that it sits perpendicular to the benzoic acid. Interestingly, placing an orthosubstituent on this 2-aryl moiety to force this same type of conformation only resulted in diminished reactivity and regioselectivity.

Scheme 1. Evaluation of Biarylcarboxylic Acids^a



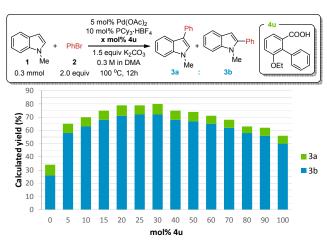


Among the substituents tested at the 3-position, both the methoxy and the free amine gave significantly higher yields and regioselectivities than what we obtained with the pivalic acid system. In this case, a substituent with an electron donating group is beneficial. We also observed that an ethoxy group (**4u**) performed slightly better than the corresponding methoxy (**4q**). Thus, **4u** gave us a yield of 66%, and a 12.2:1 regioselectivity. This carboxylic acid performing significantly better than pivalic acid for this cross-coupling, we revisited some reaction parameters to improve the yield of the direct arylation. We thus established that increasing the amount of aryl bromide equivalents to 2 allowed the calculated yield to go from 66% to 81%.

The impact of the carboxylic acid loading on the reaction outcome was also studied. Scheme 2 shows the results obtained for the reaction of *N*-methylindole with bromobenzene. It is interesting to note that going from no carboxylic acid to 5 mol%, the calculated yield almost doubles. Adding more $4\mathbf{u}$ helps improving the yield of the

major regioisomer up to 72%. Amounts ranging from 20 mol% to 30 mol% provided the best results. Of note, regardless of the amount of the carboxylic acid used, roughly 7% of the regioisomer **3a** was obtained in all cases. This is perhaps due to a background reaction occurring either with the 10 mol% acetate contained in the reaction due to the presence of Pd(OAc)2 or with the carbonate base in solution.

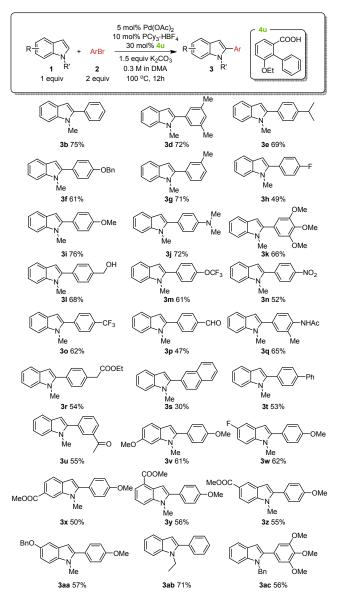
Scheme 2. Influence of Carboxylic Acid Loading on Reaction Outcome



^aConditions: *N*-methylindole (0.3 mmol), Aryl halide (2.0 eq.), $Pd(OAc)_2$ (5 mol%), $PCy_3 \cdot HBF_4$ (10 mol%), carboxylic acid **4u** (x eq.), K_2CO_3 (1.5 eq.) in DMA (0.3 M), 100 °C, 12 h. Calculated yields by GC-MS using benzophenone as internal standard.

With these conditions in hand, we then explored the scope of this regioselective indole direct arylation (scheme 3). While keeping the indole coupling partner constant, the reaction proved to be quite reliable with a variety of para- or meta-substituted aryl bromides. Both electron-donating as well as electron-withdrawing group were generally well tolerated. Functional groups such as methoxy (**3i**, **3k**), free alcohol (**3l**), nitro (**3n**), amine (**3j**), aldehyde (3p), ester (3r) and ketone (3u) were also suitable for the reaction. As exceptions however, the use of 3 or 4-bromo-benzonitrile was not tolerated.¹¹ With regards to the indole coupling partner, C4-, C5- and C-6substitutions were also tolerated. The presence of electron-withdrawing and -donating groups had little impact on the positive outcome, providing yields ranging from 50-62%. Some other alkyl substitutions are also possible on the indole's nitrogen. For example, an ethyl (3ab) and a benzyl (3ac) group were compatible. The presence of protecting groups such as acyl or tosyl however suppressed the reactivity towards C-H functionalization.

Scheme 3. Palladium-Catalyzed Regioselective C-2 Indole Arylation



^aConditions: *N*-methylindole derivative (0.3 mmol), Aryl halide (2.0 eq.), $Pd(OAc)_2$ (5 mol%), $PCy_3 \cdot HBF_4$ (10 mol%), carboxylic acid **4u** (0.3 eq.), K_2CO_3 (1.5 eq.) in DMA (0.3 M), 100 °C, 12 h. Isolated yields of the major regioisomer are shown.

It was demonstrated by Hartwig in 2011¹² that there are some cases of C-H functionalization reactions proceeding via CMD where the phosphine ligand is actually superfluous. In such scenarios, it is likely that the carboxylate acts as main ligand. It is thus conceivable that the use of a bulky carboxylic acid with a biaryl motif like the ones prepared in this study could act both as ligand and proton shuttle. With this mindset, a reaction was attempted without additional ligand. The results are shown in Scheme 4. A yield of 40% was obtained with a regioselectivity similar to the one we obtain in the presence of a phosphine ligand. These results correlate well with the study from Hartwig, suggesting that the role of the phosphine ligand in this case might be simply to provide some stability to the palladium in off-catalytic cycle intermediates.

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18 19

20

21 22

23

24

25 26

27

28

29

30

31

32

33

34

35

36 37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

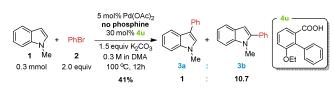
55

56

57 58 59

60

Scheme 4. Phosphine-Free Direct Arylation



In summary, we designed a library of biarylcarboxylic acids proton shuttles in the hope of finding a structural motif that would allow better yield and selectivity in the C2-direct arylation of indole. We found that an electron-donating substituent at the 3-position of the 2-arylbenzoic acid additive increased both the parameters significantly. The generality of this approach was then demonstrated with a variety of indole and aryl bromides. We hope that this report will trigger further interest in the development of tunable proton shuttles for application in diverse C–H functionalization reactions.

Experimental Section

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58 59

60

General information. All the reactions were carried out in oven-dried Schlenk tubes under argon atmosphere. The following chemicals were purchased and used as received: Palladium diacetate (Sigma-Aldrich), Tricyclohexylphosphonium tetrafluoroborate (Sigma-Aldrich), Potassium carbonate Chemical Reagent (Sinopharm Со., Ltd), N,N-Dimethylacetamide (J&K), 1-Methylindole (Adamas-beta,), Bromobenzene (Sinopharm Chemical Reagent Co., Ltd,), Pivalic acid (Ding Chemistry), 3-Nitro-2-phenylbenzoic acid (Hopschem), Bromoethane (Energy Chemical), 2-Bromo-3hydroxybenzaldehyde (Bide Pharmatech Ltd), 6-Methoxyindole (Meryer), 5-Fluoroindole (Adamas-beta), Methyl 1-Methylindole-6-carboxylate (Accela), Methyl 1methyl-1H-indole-4-carboxylate (Bide Pharmatech Ltd), Methyl indole-5-carboxylate (J&K), 5-(Benzyloxy)indole (Adamas-beta). All other commercial reagents were purchased from commercial vendors and used without further purification unless otherwise noted. ¹H NMR, ¹³C NMR and ¹⁹F NMR spectra were recorded on a Bruker 400 MHz spectrometer at 295 K in CDCl₃ unless otherwise noted. ¹H NMR and ¹³C NMR spectra of compound 4s were recorded on a Bruker 400 MHz spectrometer at 295 K in DMSO-d₆. ¹⁹F NMR were reported as 19F exp. comp. pulse decoupling (F19CPD) unless otherwise noted. HRMS analysis was performed on Finnigan LCQ advantage Max Series MS System. Gas chromatographic (GC) analysis was acquired on a Shimadzu GC-2010 plus Series GC system equipped with a flame-ionization detector. Infrared spectra were recorded on a Nicolet 8700 (Thermo Scientific Instrument Co. U.S.A) Fourier Transform Infrared Spectrometer (FT-IR), and were reported in wave numbers (cm⁻¹). Melting point was performed on X-4 Melting-point Apparatus with Microscope. HRMS were recorded with an Orbitrap analyser. Organic solutions were concentrated under reduced pressure on Buchi rotary evaporator. Column chromatographic purification of products was accomplished using forced-flow chromatography on Silica Gel (300-400 mesh).

General procedure for palladium-catalyzed indole arylation using various carboxylic acids (Scheme 1): $Pd(OAc)_2$ (5 mol%), PCy_3 'HBF₄ (10 mol%), carboxylic acid 4 (30 mol%) and K₂CO₃ (1.5 equiv.) were added to a Schlenk tube equipped with a stir bar. The Schlenk tube was evacuated and filled with argon (three cycles). To these solids, DMA (0.3 M), bromobenzene (1.0 equiv, 0.3 mmol) and 1-methylindole (1.1 equiv, 0.33 mmol) were added consecutively under a positive flow of argon. The reaction mixture was stirred at 100°C for 12 hours. The yield was determined by GC with benzophenone as internal standard.

General procedure for the evaluation of different 4u loading on the reaction outcome (Scheme 2) $Pd(OAc)_2$ (5 mol%), PCy_3 ·HBF₄ (10 mol%), carboxylic acid 4u (x mol%) and K_2CO_3 (1.5 equiv.) were added to a Schlenk tube equipped with a stir bar. The Schlenk tube was evacuated and filled with argon (three cycles). To these solids, DMA (0.3 M), bromobenzene (2.0 equiv, 0.6 mmol) and 1-methylindole (1.0 equiv, 0.3 mmol) were added consecutively under a positive flow of argon. The reaction mixture was stirred at 100°C for 12 hours. The yield was determined by GC using benzophenone as internal standard.

General procedure for palladium-catalyzed regioselective indole arylation (Scheme 3): $Pd(OAc)_2$ (5 mol%), PCy_3 .HBF₄ (10 mol%), carboxylic acid **4u** (30 mol%) and K_2CO_3 (1.5 equiv.) were added to a Schlenk tube equipped with a stir bar. The Schlenk tube was evacuated and backfilled with argon (three cycles). To these solids, DMA (0.3 M), aryl bromide (2.0 equiv.) and indole (0.3 mmol) (if the indole is a solid, it was added together with $Pd(OAc)_2$) were added consecutively under a positive flow of argon. The reaction mixture was stirred at 100 °C for 12 hours. In order to remove DMA, the reaction mixture was poured into 50 mL of ice water and the resulting mixture was extracted with ethyl acctate (4 x 30 mL). The combined organic layer was dried over Na_2SO_4 , filtered, concentrated in vacuo and purified by column chromatography.

2-phenyl-1-methylindole (**3b**). The compound was prepared following the general procedure using 1-methylindole and bromobenzene as reactants. The product was isolated by column chromatography (PE:EA = 100:1) as a light yellow solid (46.6 mg, 75%). Spectral data is consistent with previous reports.¹³ **H NMR (400 MHz, CDCl**₃) δ 7.63 (d, *J* = 7.8 Hz, 1H), 7.53 – 7.48 (m, 2H), 7.48 – 7.42 (m, 2H), 7.42 – 7.33 (m, 2H), 7.28 – 7.21 (m, 1H), 7.17 – 7.11 (m, 1H), 6.56 (s, 1H), 3.72 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 141.7, 138.5, 133.0, 129.5, 128.6, 128.1, 128.0, 121.8, 120.6, 120.0, 109.7, 101.8, 31.3. **HRMS (APCI)** calcd for C₁₅H₁₄N⁺ [(M+H)⁺]: 208.1121; found: 208.1120.

2-(3,5-dimethylphenyl)-1-methylindole (3d). The compound was prepared following the general procedure using 1-methylindole and 1-bromo-3,5-dimethylbenzene as reactants. The product was isolated by column chromatography (PE:EA = 100:1) as a light yellow solid (50.8 mg, 72%). Spectral data is consistent with previous reports.¹⁴ **1H NMR (400 MHz, CDCl**₃) δ 7.62 (d, J = 7.8 Hz, 1H), 7.34 (d, J = 8.2 Hz, 1H), 7.22 (dd, 1H), 7.13 (d, J = 10.2 Hz, 3H), 7.04 (s, 1H), 6.52 (s, 1H), 3.73 (s, 3H), 2.38 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 142.0, 138.4, 138.1, 132.9, 129.7, 128.2, 127.3, 121.6, 120.5, 119.9, 109.7, 101.6, 31.3, 21.5. HRMS (APCI) calcd for C₁₇H₁₈N⁺ [(M+H)⁺]: 236.1434; found: 236.1436.

2-(4-isopropylphenyl)-1-methylindole (3e). The compound was prepared following the general procedure using 1-methylindole and 1-bromo-4-isopropylbenzene as reactants. The product was isolated by column chromatography (PE:EA

2

3

4

5

6

7

58 59

60

= 100:1) as a light yellow solid (51.6 mg, 69%). Spectral data is consistent with previous reports.¹⁵ ¹H NMR (400 MHz, CDCl₃) δ 7.62 (d, *J* = 7.8 Hz, 1H), 7.45 – 7.40 (m, 2H), 7.32 (dd, *J* = 10.7, 8.3 Hz, 3H), 7.25 – 7.20 (m, 1H), 7.16 – 7.10 (m, 1H), 6.53 (s, 1H), 3.72 (s, 3H), 2.96 (hept, *J* = 6.9 Hz, 1H), 1.30 (d, *J* = 6.9 Hz, 6H ¹³C NMR (101 MHz, CDCl₃) δ 148.8, 141.8, 138.4, 130.4, 129.4, 128.1, 126.7, 121.6, 120.5, 119.9, 109.7, 101.4, 34.1, 31.3, 24.1. HRMS (APCI) calcd for C₁₈H₂₀N⁺ [(M+H)⁺]: 250.1590; found: 250.1595.

8 2-(4-(benzyloxy)phenyl)-1-methylindole (3f). The com-9 pound was prepared following the general procedure using 1-10 methylindole and 1-(benzyloxy)-4-bromobenzene as reac-11 tants. The product was isolated by column chromatography 12 (PE:EA = 50:1) as a light yellow solid (57.4 mg, 61%). Spectral data is consistent with previous reports.¹⁶ ¹H NMR (400 13 **MHz**, **CDCl**₃) δ 7.61 (d, J = 7.8 Hz, 1H), 7.48 - 7.37 (m, 6H), 14 7.33 (dd, J = 7.6, 5.0 Hz, 2H), 7.25 - 7.18 (m, 1H), 7.15 - 7.10 15 (m, 1H), 7.07 - 7.03 (m, 2H), 6.49 (s, 1H), 5.09 (s, 2H), 3.69 (s, 16 3H). ¹³C NMR (101 MHz, CDCl₃) δ 158.8, 141.5, 138.3, 136.9, 17 130.7, 128.8, 128.2, 128.1, 127.6, 125.6, 121.5, 120.4, 119.9, 115.0, 18 109.6, 101.2, 70.2, 31.2. HRMS (APCI) calcd for C22H20NO⁺ 19 $[(M+H)^+]$: 314.1539; found: 314.1543. 20

1-methyl-2-(*m*-tolyl)-indole (3g). The compound was pre-21 pared following the general procedure using 1-methylindole 22 and 1-(benzyloxy)-4-bromobenzene as reactants. The prod-23 uct was isolated by column chromatography (PE:EA = 100:1) 24 as a light yellow solid (47.1 mg, 71%). Spectral data is consistent with previous reports.¹⁷¹H NMR (400 MHz, CDCl₃) δ 25 7.63 (d, J = 7.8 Hz, 1H), 7.33 (ddd, J = 17.0, 11.0, 6.4 Hz, 4H), 26 7.25 - 7.19 (m, 2H), 7.14 (t, J = 7.4 Hz, 1H), 6.55 (s, 1H), 3.74 (s, 27 3H), 2.43 (s, 3H). ¹³C NMR (101 MHz, CDCl₂) δ 141.9, 138.4, 28 138.3, 132.9, 130.2, 128.8, 128.5, 128.1, 126.6, 121.7, 120.6, 119.9, 29 109.7, 101.6, 31.3, 21.7. HRMS (APCI) calcd for $C_{16}H_{16}N^{+}$ 30 $[(M+H)^+]$: 222.1277; found: 222.1279.

31 2-(4-fluorophenyl)-1-methylindole (3h). The compound 32 was prepared following the general procedure using 1-33 methylindole and 1-bromo-4-fluorobenzene as reactants. The 34 product was isolated by column chromatography (PE:EA = 100:1) as a light yellow solid (33.1 mg, 49%). Spectral data is consistent with previous reports.¹⁸ ¹H NMR (400 MHz, 35 36 **CDCl**₃) δ 7.62 (d, *J* = 7.8 Hz, 1H), 7.49 – 7.41 (m, 2H), 7.34 (d, *J* 37 = 8.2 Hz, 1H), 7.27 - 7.21 (m, 1H), 7.18 - 7.10 (m, 3H), 6.52 (s, 38 1H), 3.70 (s, 3H) ¹³C NMR (101 MHz, CDCl₃) δ 162.7 (d, J = 39 247.7 Hz), 140.6, 138.4, 131.2 (d, J = 8.1 Hz), 129.0 (d, J = 3.3 40 Hz), 128.0, 121.9, 120.6, 120.1, 115.7 (d, J = 21.6 Hz), 109.8, 101.8, 41 31.2. ¹⁹F NMR (376 MHz, CDCl₃) δ -113.80 (s). HRMS (APCI) 42 calcd for $C_{15}H_{13}FN^+$ [(M+H)⁺]: 226.1027; found: 226.1029.

43 2-(4-methoxyphenyl)-1-methylindole (3i). The compound 44 was prepared following the general procedure using 1-45 methylindole and 1-bromo-4-methoxybenzene as reactants. 46 The product was isolated by column chromatography (PE:EA = 50:1) as a light yellow solid (54.1 mg, 76%). Spectral data is consistent with previous reports.¹⁹ ¹H NMR (400 MHz, 47 48 **CDCl**₂) δ 7.63 (d, J = 7.8 Hz, 1H), 7.48 – 7.41 (m, 2H), 7.36 (d, J49 = 8.2 Hz, 1H), 7.23 (dd, J = 8.1, 1.0 Hz, 1H), 7.17 – 7.11 (m, 1H), 50 7.04 - 6.99 (m, 2H), 6.51 (s, 1H), 3.88 (s, 3H), 3.73 (s, 3H) ¹³C 51 NMR (101 MHz, CDCl₂) δ 159.6, 141.6, 138.3, 130.8, 128.1, 125.4, 52 121.5, 120.4, 119.9, 114.1, 109.6, 101.1, 55.5, 31.2. HRMS (APCI) 53 calcd for $C_{16}H_{16}NO^+$ [(M+H)⁺]: 238.1226; found: 238.1228. 54

N,N-dimethyl-4-(1-methylindol-2-yl)aniline (3j). The
compound was prepared following the general procedure
using 1-methylindole and 4-bromo-*N,N-*dimethylaniline as
reactants. The product was isolated by column chromatog-

raphy (PE:EA = 10:1) as a light yellow solid (54.1 mg, 72%). Spectral data is consistent with previous reports.²⁰ ¹H NMR (**400 MHz, CDCl**₃) δ 7.59 (d, *J* = 7.8 Hz, 1H), 7.37 (dd, *J* = 8.8, 1.9 Hz, 2H), 7.31 (d, *J* = 8.1 Hz, 1H), 7.23 – 7.16 (m, 1H), 7.14 – 7.08 (m, 1H), 6.78 (dd, *J* = 8.7, 1.5 Hz, 2H), 6.47 (d, *J* = 2.3 Hz, 1H), 3.70 (s, 3H), 2.98 (s, 6H) ¹³C NMR (101 MHz, CDCl₃) δ 150.2, 142.4, 138.2, 130.4, 128.3, 121.1, 120.7, 120.1, 119.7, 112.2, 109.5, 100.4, 40.6, 31.2. HRMS (APCI) calcd for C₁₇H₁₉N₂⁺ [(M+H)⁺]: 251.1543; found: 251.1546.

1-methyl-2-(3,4,5-trimethoxyphenyl)-indole (3**k**). The compound was prepared following the general procedure using 1-methylindole and 5-bromo-1,2,3-trimethoxybenzene as reactants. The product was isolated by column chromatography (PE:EA = 20:1) as a light yellow solid (58.9 mg, 66%). Spectral data is consistent with previous reports.²¹ ¹**H NMR (400 MHz, CDCl**₃) δ 7.65 (d, *J* = 7.8 Hz, 1H), 7.38 (d, *J* = 8.2 Hz, 1H), 7.27 (ddd, *J* = 8.2, 5.7, 1.2 Hz, 1H), 7.19 – 7.13 (m, 1H), 6.73 (s, 2H), 6.57 (s, 1H), 3.94 (s, 3H), 3.92 (s, 6H), 3.77 (s, 3H) ¹³**C NMR (101 MHz, CDCl**₃) δ 153.3, 141.6, 138.4, 138.1, 128.5, 127.9, 121.8, 120.5, 120.1, 109.7, 106.8, 101.5, 61.1, 56.3, 31.3. **HRMS (APCI)** calcd for C₁₈H₂₀NO₃⁺ [(M+H)⁺]: 298.1438; found: 298.1439.

(4-(1-methylindol-2-yl)phenyl)methanol (3l). The compound was prepared following the general procedure using 1methylindole and (4-bromophenyl)methanol as reactants. The product was isolated by column chromatography (PE:EA = 5:1) as a light yellow solid (48.4 mg, 68%). Spectral data is consistent with previous reports.²² ¹H NMR (400 MHz, CDCl₃) δ 7.63 (d, *J* = 7.8 Hz, 1H), 7.52 – 7.48 (m, 2H), 7.45 (d, *J* = 8.3 Hz, 2H), 7.35 (dd, *J* = 8.2, 0.5 Hz, 1H), 7.27 – 7.22 (m, 1H), 7.17 – 7.11 (m, 1H), 6.56 (d, *J* = 0.5 Hz, 1H), 4.74 (s, 2H), 3.73 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 141.34, 140.60, 138.48, 132.27, 129.62, 128.03, 127.22, 121.84, 120.58, 120.01, 109.74, 101.81, 65.11, 31.30. IR (neat/cm⁻¹) 3345, 3053, 3026, 2939, 2916, 1611, 1498, 1465, 1431, 1412, 1382, 1359, 1340, 1316, 1043, 1020, 1006, 825, 783, 751, 734. HRMS (APCI) calcd for C₁₆H₁₆NO⁺ [(M+H)⁺]: 238.1226; found: 238.1230.

1-methyl-2-(4-(trifluoromethoxy)phenyl)-indole (3m). The compound was prepared following the general proce-1-methylindole dure using and 1-bromo-4-(trifluoromethoxy)benzene as reactants. The product was isolated by column chromatography (PE:EA = 50:1) as a light yellow solid (53.3 mg, 61%) M. P. 76.5-77.8 °C 1H NMR (400 **MHz**, **CDCl**₃) δ 7.68 (d, J = 7.9 Hz, 1H), 7.59 – 7.52 (m, 2H), 7.40 (d, J = 8.2 Hz, 1H), 7.35 (d, J = 8.0 Hz, 2H), 7.33 - 7.28 (m, J = 8.0 Hz, 2H), 7.34 (m, J = 8.0 Hz), 7.34 (m, J1H), 7.22 - 7.18 (m, 1H), 6.60 (s, 1H), 3.76 (s, 3H). ¹³C NMR (101 MHz, CDCl₂) δ 149.0, 140.1, 138.6, 131.7, 130.8, 128.0, 122.2, 121.2, 120.8, 120.7 (q, J = 257.5 Hz), 120.2, 109.8, 102.3, 31.3. ¹⁹F NMR (376 MHz, CDCl₃) δ -57.68 (s). IR (neat/cm⁻¹) 3054, 2945, 1541, 1497, 1478, 1466, 1434, 1341, 1317, 1306, 1260, 1224, 1204, 1172, 1166, 1101, 860, 784, 751, 744, 736. HRMS (APCI) calcd for $C_{16}H_{12}F_3NO^+$ [(M+H)⁺] 292.0944, found 292.0948.

1-methyl-2-(4-nitrophenyl)-indole (3n). The compound was prepared following the general procedure using 1methylindole and 1-bromo-4-nitrobenzene as reactants. The product was isolated by column chromatography (PE:EA = 20:1) as a light yellow solid (39.4 mg, 52 %). Spectral data is consistent with previous reports.²³ ¹H NMR (400 MHz, **CDCl**₃) δ 8.39 – 8.27 (m, 2H), 7.73 – 7.64 (m, 3H), 7.41 (dd, *J* = 8.3, o.6 Hz, 1H), 7.33 (ddd, *J* = 8.2, 7.0, 1.1 Hz, 1H), 7.22 – 7.15 (m, 1H), 6.72 (d, *J* = o.6 Hz, 1H), 3.81 (s, 3H). ¹³C NMR (101 MHz, **CDCl**₃) δ 147.10, 139.36, 139.04, 129.62, 127.82, 124.03, 123.10, 121.16, 120.60, 110.03, 104.20, 31.71. **IR (neat/cm⁻¹)** 3057, 2920, 2849, 1560, 1533, 1510, 1465, 1454, 1343, 1318, 1107, 866, 854, 791, 750, 704, 696. **HRMS (APCI)** calcd for $C_{15}H_{13}N_2O_2^+$ [(M+H)⁺]: 253.0972; found: 253.0980.

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

58 59

60

1-methyl-2-(4-(trifluoromethyl)phenyl)-indole (30). The compound was prepared following the general procedure 1-methylindole using and 1-bromo-4-(trifluoromethyl)benzene as reactants. The product was isolated by column chromatography (PE:EA = 20:1) as a light yellow solid (51.2 mg, 62 %). Spectral data is consistent with previous reports.¹⁸ ¹H NMR (400 MHz, CDCl₃) δ 7.74 (d, J = 8.1 Hz, 2H), 7.65 (t, J = 8.7 Hz, 3H), 7.40 (d, J = 8.2 Hz, 1H), 7.33 - 7.27 (m, 1H), 7.22 - 7.15 (m, 1H), 6.64 (s, 1H), 3.78 (s, 3H) ¹³C NMR (101 MHz, CDCl₃) δ 139.7, 138.7, 136.4, 129.8 (q, *J* = 32.5 Hz), 129.4, 127.8, 125.5 (q, *J* = 3.7 Hz), 122.4, 121.8 (q, *J* = 205.8 Hz), 120.8, 120.2, 109.8, 102.8, 31.3. ¹⁹F NMR (376 MHz, CDCl₃) δ -62.51 (s). HRMS (APCI) calcd for C₁₆H₁₂F₂N⁴ [(M+H)⁺]: 276.0995; found: 276.1000.

4-(1-methylindol-2-yl)benzaldehyde (3p). The compound 16 was prepared following the general procedure using 1-17 methylindole and 4-bromobenzaldehyde as reactants. The 18 product was isolated by column chromatography (PE:EA = 19 20:1) as a light yellow solid (33.2 mg, 47 %). Spectral data is 20 consistent with previous reports.²⁴ ¹H NMR (400 MHz, 21 **CDCl**₃) δ 10.08 (s, 1H), 8.08 – 7.91 (m, 2H), 7.73 – 7.61 (m, 3H), 22 7.40 (dd, *J* = 8.2, 0.5 Hz, 1H), 7.31 (ddd, *J* = 8.2, 7.1, 1.1 Hz, 1H), 23 7.22 – 7.13 (m, 1H), 6.70 (d, J = 0.5 Hz, 1H), 3.80 (s, 3H) ¹³C 24 NMR (101 MHz, CDCl₂) δ 190.9, 139.1, 138.2, 137.9, 134.4, 129.1, 128.6, 126.9, 121.7, 120.0, 119.4, 109.0, 102.6, 30.7. IR (neat/cm⁻ 25 ¹) 3052, 2922, 2852, 2735, 1702, 1604, 1567, 1466, 1389, 1359, 26 1341, 1318, 1305, 1207, 1168, 843, 786, 750. HRMS (APCI) calcd 27 for $C_{16}H_{14}NO^+$ [(M+H)⁺]: 236.1070; found: 236.1074. 28

N-(2-methyl-4-(1-methylindol-2-yl)phenyl)acetamide 29 (3q). The compound was prepared following the general 30 procedure using 1-methylindole and N-(4-bromo-2-31 methylphenyl)acetamide as reactants. The product was iso-32 lated by column chromatography (PE:EA = 2:1) as a light yel-33 low solid (54.3 mg, 65 %). M. P. 162.0-163.5 °C ¹H NMR (400 34 **MHz**, **CDCl**₃) δ 7.92 – 7.83 (m, 1H), 7.61 (d, J = 7.8 Hz, 1H), 35 7.37 - 7.30 (m, 3H), 7.27 - 7.21 (m, 1H), 7.13 (t, J = 7.4 Hz, 2H), 6.51 (s, 1H), 3.72 (s, 3H), 2.31 (s, 3H), 2.23 (s, 3H) ¹³C NMR (101 36 MHz, CDCl₃) δ 168.6, 141.2, 138.4, 135.5, 131.5, 129.8, 129.4, 37 128.0, 127.8, 123.4, 121.7, 120.5, 119.9, 109.7, 101.6, 31.3, 24.5, 18.0. 38 IR (neat/cm⁻¹) 3274, 3050, 2922, 1666, 1612, 1587, 1518, 1465, 39 1433, 1369, 1339, 1301, 1266, 1242, 1009, 837, 784, 751, 736, 702. 40 **HRMS (APCI)** calcd for $C_{18}H_{10}N_2O^+$ [(M+H)⁺]: 279.1492; 41 found: 279.1496.

42 ethyl 2-(4-(1-methylindol-2-yl)phenyl)acetate (3r). The 43 compound was prepared following the general procedure 44 using 1-methylindole and ethyl 2-(4-bromophenyl)acetate as 45 reactants. The product was isolated by column chromatog-46 raphy (PE:EA = 10:1) as a light yellow solid (47.5 mg, 54%) M. **P.** 64.3-66.2 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.62 (d, J = 7.8 47 Hz, 1H), 7.46 (d, J = 7.8 Hz, 2H), 7.41 – 7.31 (m, 3H), 7.23 (dd, J 48 = 14.2, 6.1 Hz, 1H), 7.13 (t, J = 7.4 Hz, 1H), 6.55 (s, 1H), 4.18 (q, J 49 = 7.1 Hz, 2H), 3.72 (s, 3H), 3.66 (s, 2H), 1.28 (t, J = 7.1 Hz, 3H) 50 ¹³C NMR (101 MHz, CDCl₃) δ 171.5, 141.3, 138.5, 134.0, 131.7, 51 129.6, 129.5, 128.1, 121.8, 120.6, 120.0, 109.7, 101.8, 61.1, 41.2, 31.3, 52 14.3. IR (neat/cm⁻¹) 3053, 2979, 2936, 1734, 1611, 1502, 1465, 53 1431, 1415, 1385, 1367, 1339, 1315, 1251, 1223, 1157, 1031, 826, 781, 54 750, 736. HRMS (APCI) calcd for $C_{10}H_{20}NO_2^+$ [(M+H)⁺]: 55 294.1489; found: 294.1490.

1-methyl-2-(naphthalen-2-yl)indole (3s). The compound
was prepared following the general procedure using 1-

methylindole and 2-bromonaphthalene as reactants. The product was isolated by column chromatography (PE:EA = 100:1) as a light yellow solid (23.2 mg, 30 %) Spectral data is consistent with previous reports. ²⁵ ¹H NMR (400 MHz, CDCl₃) δ 8.03 – 7.90 (m, 4H), 7.69 (dd, *J* = 15.3, 8.1 Hz, 2H), 7.60 – 7.53 (m, 2H), 7.43 (d, *J* = 8.2 Hz, 1H), 7.32 (t, *J* = 7.4 Hz, 1H), 7.21 (t, *J* = 7.3 Hz, 1H), 6.71 (s, 1H), 3.84 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 141.7, 138.6, 133.4, 132.8, 130.3, 128.4, 128.3, 128.2, 128.2, 127.9, 127.3, 126.7, 126.5, 121.9, 120.6, 120.1, 109.8, 102.2, 31.5. HRMS (APCI) calcd for C₁₉H₁₆N⁺ [(M+H)⁺]: 258.1277; found: 258.1279.

2-([1,1'-biphenyl]-4-yl)-1-methylindole (3t). The compound was prepared following the general procedure using 1methylindole and 4-bromo-1,1'-biphenyl as reactants. The product was isolated by column chromatography (PE:EA = 100:1) as a light yellow solid (45.1 mg, 53 %) Spectral data is consistent with previous reports.²⁶ ¹H NMR (400 MHz, **CDCl**₃) δ 7.71 – 7.67 (m, 2H), 7.65 (dd, *J* = 5.2, 3.3 Hz, 3H), 7.61 – 7.55 (m, 2H), 7.47 (dd, *J* = 10.4, 4.8 Hz, 2H), 7.41 – 7.34 (m, 2H), 7.29 – 7.23 (m, 1H), 7.19 – 7.11 (m, 1H), 6.61 (s, 1H), 3.78 (s, 3H) ¹³C NMR (101 MHz, CDCl₃) δ 141.3, 140.8, 140.6, 138.6, 131.8, 129.8, 129.0, 128.1, 127.7, 127.3, 127.2, 121.9, 120.6, 120.0, 109.8, 101.9, 31.4. HRMS (APCI) calcd for C₂₁H₁₈N⁺ [(M+H)⁺]: 284.1434; found: 284.1438.

1-(3-(1-methylindol-2-yl)phenyl)ethan-1-one (3**u**). The compound was prepared following the general procedure using 1-methylindole and 1-(3-bromophenyl)ethan-1-one as reactants. The product was isolated by column chromatography (PE:EA = 40:1) as a light yellow oil (41.1 mg, 55 %) ¹H **NMR (400 MHz, CDCl**₃) δ 8.11 (t, *J* = 1.6 Hz, 1H), 8.03 – 7.94 (m, 1H), 7.75 – 7.69 (m, 1H), 7.65 (d, *J* = 7.9 Hz, 1H), 7.58 (t, *J* = 7.7 Hz, 1H), 7.38 (d, *J* = 8.2 Hz, 1H), 7.31 – 7.25 (m, 1H), 7.20 – 7.12 (m, 1H), 6.62 (s, 1H), 3.76 (s, 3H), 2.66 (s, 3H). ¹³C **NMR (101 MHz, CDCl**₃) δ 198.0, 140.4, 138.6, 137.5, 133.9, 133.5, 129.2, 129.0, 128.0, 127.8, 122.2, 120.8, 120.2, 109.8, 102.4, 31.4, 26.9. **IR (neat/cm⁻¹)** 3055, 2923, 2852, 1686, 1602, 1580, 1538, 1464, 1418, 1385, 1356, 1341, 1315, 1279, 1266, 1237, 1201, 783, 750, 736, 696, 589. **HRMS (APCI)** calcd for C₁₇H₁₆NO⁺ [(M+H)⁺]: 250.1226; found: 250.1230.

6-methoxy-2-(4-methoxyphenyl)-1-methylindole (3v). The compound was prepared following the general procedure using 6-methoxy-1-methylindole and 1-bromo-4-methoxybenzene as reactants. The product was isolated by column chromatography (PE:EA = 50:1) as a light yellow solid (48.9 mg, 61 %) Spectral data is consistent with previous reports.²⁷ ¹H NMR (400 MHz, CDCl₃) δ 7.53 - 7.48 (m, 1H), 7.45 - 7.41 (m, 2H), 7.03 - 6.99 (m, 2H), 6.85 - 6.82 (m, 2H), 6.44 (s, 1H), 3.92 (s, 3H), 3.88 (s, 3H), 3.69 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.4, 156.3, 140.6, 139.0, 130.6, 125.6, 122.4, 121.0, 114.1, 109.6, 100.9, 93.5, 55.9, 55.5, 31.3.

5-fluoro-2-(4-methoxyphenyl)-1-methylindole (3w). The compound was prepared following the general procedure using 5-fluoro-1-methylindole and 1-bromo-4methoxybenzene as reactants. The product was isolated by column chromatography (PE:EA = 50:1) as a light yellow solid (47.5 mg, 62 %) M. P. 107.2-108.9 °C. ¹H NMR (400 **MHz**, **CDCl**,) δ 7.54 (dd, J = 8.5, 5.4 Hz, 1H), 7.43 (d, J = 8.5 Hz, 2H), 7.04 (dd, I = 12.0, 4.8 Hz, 3H), 6.92 (t, I = 9.2 Hz, 1H), 6.49 (s, 1H), 3.89 (s, 3H), 3.68 (s, 3H). ¹³C NMR (101 MHz, CDCl,) δ 159.8 (d, J = 237.1 Hz), 159.6, 142.1 (d, J = 3.8Hz), 138.3 (d, J = 12.0 Hz), 130.6, 125.1, 124.5, 121.0 (d, J = 10.0 Hz), 114.1, 108.4 (d, J = 24.4 Hz), 101.1, 96.1 (d, J = 26.2 Hz), 55.5, 31.3. ¹⁹F NMR (376 MHz, CDCl₃) δ -120.94 (s). IR

2

3

4

5

7

 $(neat/cm^{-1})$ 3035, 3001, 2942, 2908, 2836, 1612, 1584, 1575, 1548, 1498, 1467, 1426, 1357, 1294, 1281, 1250, 1176, 1081, 1036, 958, 832, 808, 612, 568. HRMS (APCI) calcd for C₁₆H₁₅FNO⁺ $[(M+H)^+]$ 256.1132, found 256.1152.

methvl 2-(4-methoxyphenyl)-1-methylindole-6**carboxylate** (3**x**). The compound was prepared following the general procedure using methyl 1-methylindole-6-6 carboxylate and 1-bromo-4-methoxybenzene as reactants. The product was isolated by column chromatography (PE:EA 8 = 20:1) as a light yellow solid (44.3 mg, 50 %) M. P. 153.9-155.3 9 ^oC ¹H NMR (**400** MHz, CDCl₃) δ 8.13 (s, 1H), 7.84 (dd, *J* = 8.3, 10 1.3 Hz, 1H), 7.62 (d, J = 8.3 Hz, 1H), 7.50 - 7.40 (m, 2H), 7.07 -11 6.98 (m, 2H), 6.54 (s, 1H), 3.96 (s, 3H), 3.88 (s, 3H), 3.79 (s, 12 3H). ¹³C NMR (101 MHz, CDCl₂) δ 168.4, 159.9, 144.9, 137.6, 13 131.8, 130.8, 124.6, 122.8, 121.1, 119.8, 114.2, 112.1, 101.6, 55.5, 52.0, 31.4. IR (neat/cm⁻¹) 3009, 2947, 2841, 1712, 1610, 1493, 1464, 14 1437, 1337, 1306, 1289, 1251, 1231, 1177, 1109, 1032, 873, 838, 825, 15 788, 743. HRMS (APCI) calcd for $C_{18}H_{18}NO_3^+$ [(M+H)⁺]: 16 296.1281; found: 296.1287. 17

methvl 2-(4-methoxyphenyl)-1-methylindole-4-18 **carboxylate** (3**y**). The compound was prepared following the 19 general procedure using methyl 1-methylindole-4-20 carboxylate and 1-bromo-4-methoxybenzene as reactants. 21 The product was isolated by column chromatography (PE:EA 22 = 20:1) as a light yellow solid (49.6 mg, 56 %) Spectral data is 23 consistent with previous reports.15 H NMR (400 MHz, **CDCl**₂) δ 7.93 (d, J = 7.5 Hz, 1H), 7.54 (d, J = 8.1 Hz, 1H), 7.47 24 (d, J = 8.5 Hz, 2H), 7.26 (dd, J = 9.0, 6.6 Hz, 1H), 7.14 (s, 1H),25 7.02 (d, J = 8.5 Hz, 2H), 3.98 (s, 3H), 3.87 (s, 3H), 3.76 (s, 3H). 26 ¹³C NMR (101 MHz, CDCl₃) δ 168.3, 159.8, 143.8, 139.2, 130.8, 27 127.8, 124.8, 123.5, 121.0, 120.6, 114.4, 114.2, 102.6, 55.5, 51.8, 31.4. 28 IR (neat/cm⁻¹) 3065, 2997, 2947, 2837, 1709, 1611, 1575, 1496, 29 1451, 1433, 1385, 1363, 1347, 1262, 1251, 1200, 1178, 1143, 1121, 1036, 30 837, 791, 752.

31 2-(4-methoxyphenyl)-1-methylindole-5methyl 32 carboxylate (3z). The compound was prepared following the 33 general procedure using methyl 1-methylindole-5-carboxylate 34 and 1-bromo-4-methoxybenzene as reactants. The product 35 was isolated by column chromatography (PE:EA = 20:1) as a light yellow solid (48.7 mg, 55%). M. P. 98.0-100.2 °C. 1H 36 NMR (400 MHz, CDCl₃) δ 7.47 - 7.40 (m, 2H), 7.24 (d, J = 37 2.6 Hz, 1H), 7.11 (d, J = 2.4 Hz, 1H), 7.04 - 6.98 (m, 2H), 6.91 38 (dd, J = 8.8, 2.5 Hz, 1H), 6.45 (s, 1H), 3.88 (s, 3H), 3.87 (s, 3H), 39 3.70 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.5, 154.4, 142.2, 40 133.7, 130.7, 128.4, 125.5, 114.1, 111.7, 110.4, 102.2, 100.8, 56.1, 55.5, 41 31.3. IR (neat/cm⁻¹) 3032, 2995, 2943, 2834, 1735, 1613, 1575, 42 1544, 1498, 1477, 1453, 1430, 1386, 1345, 1287, 1249, 1218, 1204, 43 1176, 1141, 1035, 943, 836, 790, 739. HRMS (APCI) calcd for 44 $C_{18}H_{18}NO_3^+$ [(M+H)⁺] 296.1281, found 296.1287.

45 5-(benzyloxy)-2-(4-methoxyphenyl)-1-methylindole

46 (3aa). The compound was prepared following the general procedure using 5-(benzyloxy)-1-methylindole and 1-bromo-47 4-methoxybenzene as reactants. The product was isolated by 48 column chromatography (PE:EA = 30:1) as a light yellow solid 49 (58.7 mg, 57%) M. P. 161.1-163.0 °C. ¹H NMR (400 MHz, 50 **CDCl**₃) δ 7.49 (d, J = 7.5 Hz, 2H), 7.40 (dd, J = 15.8, 8.0 Hz, 51 4H), 7.31 (t, J = 7.3 Hz, 1H), 7.24 (d, J = 7.7 Hz, 2H), 7.17 (s, 52 1H), 6.99 (t, I = 8.6 Hz, 3H), 5.13 (s, 2H), 3.87 (s, 3H), 3.69 (s, 53 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.6, 153.6, 142.2, 137.9, 54 133.9, 130.7, 128.6, 128.4, 127.8, 127.7, 125.4, 114.1, 112.4, 110.4, 55 103.9, 100.8, 71.1, 55.5, 31.3. IR (neat/cm⁻¹) 3063, 3033, 3006, 56 2917, 2835, 1609, 1579, 1574, 1545, 1466, 1451, 1431, 1375, 1342, 57 1291, 1243, 1189, 1174, 1134, 1099, 1022, 829, 801, 795, 783, 743, 58

705, 692. HRMS (APCI) calcd for $C_{23}H_{22}NO_2^+$ [(M+H)⁺] 344.1645, found 344.1671.

1-ethyl-2-phenylindole (3ab). The compound was prepared following the general procedure using 1-ethylindole and bromobenzene as reactants. The product was isolated by column chromatography (PE:EA = 50:1) as a light yellow solid (47.1 mg, 71 %) Spectral data is consistent with previous reports.²⁸ ¹H NMR (400 MHz, CDCl₂) δ 7.70 (d, I = 7.8 Hz, 1H), 7.59 - 7.48 (m, 4H), 7.46 (dd, J = 9.2, 4.7 Hz, 2H), 7.33 - 7.26 (m, 1H), 7.20 (dd, I = 10.5, 4.4 Hz, 1H), 6.59 (s, 1H), 4.25 (q, I =7.2 Hz, 2H), 1.37 (t, J = 7.2 Hz, 3H) ¹³C NMR (101 MHz, $CDCl_3$) δ 141.2, 137.2, 133.3, 129.5, 128.6, 128.4, 128.0, 121.6, 120.7, 119.9, 110.0, 102.2, 38.9, 15.5. HRMS (APCI) calcd for $C_{16}H_{16}N^{+}$ [(M+H)⁺]: 222.1277; found: 222.1279.

1-benzyl-2-(3,4,5-trimethoxyphenyl)-indole (3ac). The compound was prepared following the general procedure using 1-benzylindole and 5-bromo-1,2,3-trimethoxybenzene as reactants. The product was isolated by column chromatography (PE:EA = 20:1) as a light yellow oil (62.7 mg, 56 %) ¹H NMR (400 MHz, CDCl₂) δ 7.71 – 7.65 (m, 1H), 7.33 – 7.27 (m, 2H), 7.23 (t, J = 3.7 Hz, 2H), 7.21 – 7.14 (m, 2H), 7.09 (d, J= 7.4 Hz, 2H), 6.65 (d, *J* = 2.8 Hz, 1H), 6.60 (d, *J* = 3.5 Hz, 2H), 5.37 (s, 2H), 3.86 (s, 3H), 3.60 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) § 153.2, 141.8, 138.7, 138.3, 137.9, 128.9, 128.2, 128.1, 127.2, 125.8, 122.1, 120.6, 120.4, 110.2, 106.2, 101.9, 61.0, 55.8, 47.9. IR (neat/cm⁻¹) 3058, 2999, 2935, 2830, 1601, 1582, 1497, 1462, 1453, 1415, 1347, 1305, 1238, 1127, 1006, 844, 786, 750, 732, 696. HRMS (APCI) calcd for $C_{24}H_{24}NO_3^+$ [(M+H)⁺]: 374.1751; found: 374.1756.

Reaction with no phosphine (Scheme 4). $Pd(OAc)_2$ (5 mol%), carboxylic acid **4u** (30 mol%) and K₂CO₂ (1.5 equiv.) were added to a Schlenk tube equipped with a stir bar. The Schlenk tube was evacuated and filled with argon (three cycles). To these solids, DMA (0.3 M), bromobenzene (2.0 equiv, 0.6 mmol) and 1-methylindole (1.0 equiv, 0.3 mmol) were added consecutively under a positive flow of argon. The reaction mixture was stirred at 100°C for 12 hours. The yield was determined by GC with benzophenone as internal standard (41% total yield, 3a : 3b = 1 : 10.7).

Preparation of biaryl carboxylic acids 4 **General procedure A:**

1) Ester formation : Thionyl chloride (3.6 equiv.) was added dropwise to the solution of ortho-bromobenzoic acid (1.0 equiv.) in DCM (2 M) at 0 °C. The mixture was stirred at rt for 1h before methanol (50 mL) was added dropwise. The solution was stirred at rt for 12 h (or at reflux temperature for 2 h). Excess SOCl2 and methanol were removed under reduced pressure. The residue was dissolved in ethyl acetate, washed with water and 5% sodium hydroxide solution, dried over MgSO₄, filtered, and concentrated in vacuo to give crude methyl 2-bromobenzoate, which was used in the next step without further purification

2) Suzuki-Miyaura reaction : ²⁹ An oven-dried resealable Schlenk tube was charged with Pd(OAc)2 (5.0 mol%), dppf (6.0 mol%), methyl 2-bromobenzoate (1.0 equiv.), arylboronic acid (2.0 equiv.), and K3PO4 (3.0 equiv.). The vessel was evacuated and filled with argon (three cycles) and then DME (0.4 M) was injected into the Schlenk tube under argon atmosphere. The mixture was stirred vigorously at 80 °C for 12h. The reaction mixture was then cooled to room temperature, diluted with ethyl acetate and washed with water. The organic layer was dried over anhydrous Na2SO4 and concentrated. The crude methyl 2-arylbenzoate was purified by column chromatography.

3) Ester saponification: the methyl 2-arylbenzoate and 50% potassium hydroxide (0.5 mL/mmol substrate) were stirred overnight in methanol (0.3 M). The solvent was removed under reduced pressure and the residue was dissolved in water. The solution was washed with ether and acidified to pH 2. The aqueous phase was extracted with ethyl acetate. The combined organic phase was dried over MgSO4, filtered, and concentrated under reduced pressure. The crude material was purified by column chromatography.⁵

General procedure B:

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

58 59

60

 Phenol alkylation: A suspension of 2-bromo-3hydroxybenzaldehyde (1 equiv.), K2CO3 (3 equiv.) and alkyl halide (3.0 equiv.) in DMF (0.5 M) was stirred at 60°C for 3 h. The reaction was then quenched with water and the resulting mixture was extracted with EtOAc. The combined organic layers were washed with brine, dried over anhydrous Na2SO4 and concentrated. The crude 2-bromo-3-alkoxybenzaldehyde was purified by column chromatography.

18 2) Suzuki-Miyaura reaction:²⁹ An oven-dried resealable 19 Schlenk tube was charged with Pd(OAc)₂ (5.0 mol%), dppf 20 (6.0 mol%), 2-bromo-3-alkoxybenzaldehyde (1.0 equiv.), ar-21 ylboronic acid (2.0 equiv.), and K3PO4 (3.0 equiv.). The ves-22 sel was evacuated and filled with argon (three cycles) and 23 then DME (0.4 M) was injected into the Schlenk tube under 24 argon atmosphere. The mixture was stirred vigorously at 80 °C for 12h. The reaction mixture was then cooled to room 25 temperature, diluted with ethyl acetate and washed with 26 water. The organic layer was dried over anhydrous Na2SO4 27 and concentrated. The crude 2-aryl-3-alkoxybenzaldehyde 28 was purified by column chromatography. 29

3) Aldehyde oxidation: The 2-aryl-3-alkoxybenzaldehyde was 30 dissolved in methanol (0.5 M) and 50% KOH (0.35 ml/mmol 31 substrate) was added. A 30% hydrogen peroxide solution 32 (0.96 ml/mmol substrate) was then slowly added dropwise. 33 The mixture was stirred for half an hour at 65 °C and cooled 34 to room temperature. After removing methanol under re-35 duced pressure, the resulting mixture was acidified to pH = 2, diluted with water and extracted with ethyl acetate. the 36 combined organic layers was dried over anhydrous Na2SO4 37 and concentrated. The crude material was purified by col-38 umn chromatography. 39

General Procedure C: According to the reported litera-40 ture:³⁰6 In a glove box, a 25 mL of the Schlenk tube equipped 41 with a stir bar was charged with Pd(OAc)₂ (8 mol%), ligand 42 (8 mol%), benzoic acid (0.2 mmol), iodobenzene (2.0 equiv), 43 Ag2CO3 (0.5 equiv), Cs2CO3 (0.5 equiv). The tube was fitted 44 with a rubber septum and taken out of the glove box. HFIP (1 45 mL) was added to the Schlenk tube though the rubber sep-46 tum using syringe and then the septum was replaced by a Teflon screwcap under argon flow. The reaction mixture was 47 stirred at 30°C for 36 h. Upon completion, diluted HCl (15 48 mL) was added to the reaction mixture with stirring followed 49 by water (15 mL). The resulting reaction mixture was extract-50 ed twice with ether (50 mL x 2) and the combined organic 51 fractions were dried over Na2SO4, filtered and concentrated 52 under reduced pressure. The obtained residue was purified 53 by column chromatography.

54 [1,1'-biphenyl]-2-carboxylic acid (4b). Following general
55 procedure A, 2-bromobenzoic acid (5 mmol) and phenyl56 boronic acid were used. The methyl ester intermediate was
57 purified using petroleum ether/ethyl acetate (PE:EA = 20:1).

The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to get white solid, yield 80%, 792mg. Spectral data is consistent with previous reports.³¹ **1H NMR (400 MHz, CDCl3)** δ 10.96 (s, 1H), 7.96 (dd, *J* = 7.8, 1.3 Hz, 1H), 7.57 (td, *J* = 7.6, 1.4 Hz, 1H), 7.46 - 7.32 (m, 7H). **13C NMR (101 MHz, CDCl3)** δ 173.7, 143.6, 141.2, 132.2, 131.4, 130.8, 129.5, 128.6, 128.2, 127.5, 127.3.

5-methoxy-[1,1'-biphenyl]-2-carboxylic acid (4c). The compound was prepared following general procedure C using 4-methoxybenzoic acid (5 mmol) and iodobenzene as reactants. The product was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 4:1) as the eluent to get a white solid (582mg 51%). Spectral data is consistent with previous reports.³² **1H NMR (400 MHz, CDCl3)** δ 7.99 (d, *J* = 8.7 Hz, 1H), 7.41 – 7.34 (m, 3H), 7.33 – 7.28 (m, 2H), 6.91 (dd, *J* = 8.7, 2.6 Hz, 1H), 6.82 (d, *J* = 2.6 Hz, 1H), 3.86 (s, 3H). **13C NMR (101 MHz, CDCl3)** δ 172.0, 162.4, 146.3, 141.3, 133.4, 128.4, 127.9, 127.3, 121.0, 116.7, 112.6, 55.5.

5-(trifluoromethyl)-[1,1'-biphenyl]-2-carboxylic acid (4d). The compound was prepared following general procedure C using 4-(trifluoromethyl)benzoic acid (5 mmol) and iodobenzene as reactants. The product was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 2:1) as the eluent to get a white solid (559 mg, 42%). **1H NMR (400 MHz, CDCl3)** δ 8.04 (d, *J* = 8.1 Hz, 1H), 7.69 (d, *J* = 8.3 Hz, 1H), 7.65 (s, 1H), 7.45 – 7.39 (m, 3H), 7.37 – 7.32 (m, 2H). **13C NMR (101 MHz, CDCl3)** δ 172.2, 144.1, 139.7, 133.9 (q, *J* = 32.8 Hz), 132.7, 131.2, 128.5, 128.3, 128.2 (q, *J* = 3.9 Hz), 124.2 (q, *J* = 3.7 Hz), 123.6 (q, *J* = 272.9 Hz). **19F NMR (376 MHz, CDCl3)** δ -63.08 (s). **HRMS (ESI)** calcd for C14H9O2F3Na+ [(M+Na)+] 289.0447, found 289.0439.

5-nitro-[1,1'-biphenyl]-2-carboxylic acid (4e). The compound was prepared following general procedure C using 4-nitrobenzoic acid (5 mmol) and iodobenzene as reactants. The product was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 1:1) as the eluent to get a white solid (352 mg, 29%). Spectral data is consistent with previous reports.³³ **1H NMR (400 MHz, CDCl3)** δ 8.26 (d, *J* = 7.1 Hz, 2H), 8.07 (d, *J* = 9.2 Hz, 1H), 7.44 (d, *J* = 2.2 Hz, 3H), 7.40 – 7.31 (m, 2H). **13C NMR (101 MHz, CDCl3)** δ 171.9, 149.7, 144.9, 138.7, 134.9, 131.8, 128.7, 128.4, 126.0, 122.0.

2'-methyl-[1,1'-biphenyl]-2-carboxylic acid (4f). The compound was prepared following general procedure A using 2-bromobenzoic acid (5 mmol) and o-tolylboronic acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to get a white solid (711 mg 67%). Spectral data is consistent with previous reports.³⁴ ¹H NMR (400 MHz, CDCl₃) δ 8.05 - 8.02 (m, 1H), 7.57 (td, *J* = 7.5, 1.4 Hz, 1H), 7.46 - 7.40 (m, 1H), 7.28 - 7.17 (m, 4H), 7.10 - 7.06 (m, 1H), 2.07 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 172.8, 143.7, 141.3, 135.4, 132.5, 131.4, 130.9, 129.7, 129.3, 128.6, 127.4, 127.3, 125.4, 20.1.

2'-methoxy-[1,1'-biphenyl]-2-carboxylic acid (4g). The compound was prepared following general procedure A using 2-bromobenzoic acid (5 mmol) and (2-methoxyphenyl)boronic acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 4:1) as the eluent to get a white solid (798 mg 70%).

2

3

4

56

57

58 59

60

Spectral data is consistent with previous reports.³⁵ ¹H NMR (400 MHz, CDCl₃) δ 7.93 (dd, *J* = 7.8, 1.0 Hz, 1H), 7.58 (td, *J* = 7.6, 1.4 Hz, 1H), 7.41 (td, *J* = 7.7, 1.2 Hz, 1H), 7.38 – 7.27 (m, 3H), 7.04 (td, *J* = 7.4, 0.9 Hz, 1H), 6.88 (d, *J* = 8.2 Hz, 1H), 3.70 (s, 3H).³C NMR (101 MHz, CDCl₃) δ 174.0, 156.2, 139.2, 132.4, 131.7, 130.8, 130.3, 129.9, 129.2, 127.2, 121.0, 110.7, 55.1.

5 2-(naphthalen-1-yl)benzoic acid (4h). The compound was 6 prepared following general procedure A using 2-7 bromobenzoic acid (5 mmol) and 1-naphtylboronic acid as 8 reactants. The methyl ester intermediate was purified using 9 petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic 10 acid was isolated by column chromatography, using petrole-11 um ether/ethyl acetate (PE:EA = 5:1) as the eluent to get a 12 white solid (732 mg 59%). Spectral data is consistent with previous reports.³⁶ H NMR (400 MHz, CDCl₃) δ 8.04 (dd, J 13 = 7.9, 1.4 Hz, 1H), 7.88 (dd, J = 16.1, 8.2 Hz, 2H), 7.61 (td, J = 14 7.5, 1.3 Hz, 1H), 7.52 - 7.42 (m, 4H), 7.39 - 7.32 (m, 2H), 7.28 15 (dd, J = 7.0, 1.2 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 172.4, 16 142.1, 139.3, 133.4, 132.4, 132.3, 132.1, 131.0, 130.2, 128.3, 127.8, 17 127.7, 126.1, 125.7, 125.6, 125.2. 18

3',5'-dimethyl-[1,1'-biphenyl]-2-carboxylic acid (4i). The 19 compound was prepared following general procedure A us-20 2-bromobenzoic acid (5 mmol) ing and 3,5-21 dimethylphenylboronic acid as reactants. The methyl ester 22 intermediate was purified using petroleum ether/ethyl ace-23 tate (PE:EA = 20:1). The carboxylic acid was isolated by col-24 umn chromatography, using petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to get a white solid (803 mg 71%). 25 Spectral data is consistent with previous reports.³⁷ ¹H NMR 26 (400 MHz, CDCl₃) δ 7.94 – 7.91 (m, 1H), 7.54 (td, J = 7.5, 1.4 27 Hz, 1H), 7.44 - 7.34 (m, 2H), 6.98 (dd, J = 10.2, 0.6 Hz, 3H), 28 2.34 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 173.8, 143.6, 141.0, 29 137.7, 132.1, 131.3, 130.7, 129.5, 129.2, 127.1, 126.5, 21.4. 30

2',4',6'-trimethyl-[1,1'-biphenyl]-2-carboxylic acid (4j). 31 The compound was prepared following general procedure A 32 using 2-bromobenzoic acid (5 mmol) and 2,4,6-33 trimethylphenylboronic acid as reactants. The methyl ester 34 intermediate was purified using petroleum ether/ethyl ace-35 tate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate 36 (PE:EA = 5:1) as the eluent to get a white solid (240 mg 20%). 37 Spectral data is consistent with previous reports.³⁸ ¹H NMR 38 (**400** MHz, CDCl₃) δ 8.10 (dd, *J* = 7.9, 1.2 Hz, 1H), 7.60 (td, *J* = 39 7.5, 1.4 Hz, 1H), 7.44 (td, J = 7.7, 1.3 Hz, 1H), 7.14 (dd, J = 7.6, 40 1.0 Hz, 1H), 6.91 (s, 2H), 2.34 (s, 3H), 1.92 (s, 6H). ¹³C NMR 41 (101 MHz, CDCl₃) δ 170.4, 142.7, 137.7, 136.6, 135.2, 133.0, 131.3, 42 131.2, 127.9, 127.1, 21.1, 20.5.

43 6-methyl-[1,1'-biphenyl]-2-carboxylic acid (4k). The com-44 pound was prepared following general procedure A using 2-45 bromo-3-methyl-benzoic acid (5 mmol) and phenylboronic 46 acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The car-47 boxylic acid was isolated by column chromatography, using 48 petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to 49 get a white solid (657 mg 62%). Spectral data is consistent 50 with previous reports.³⁷ H NMR (400 MHz, CDCl₃) δ 7.78 51 (d, J = 7.8 Hz, 1H), 7.44 (d, J = 7.6 Hz, 1H), 7.41 - 7.29 (m, 4H),52 7.16 - 7.13 (m, 2H), 2.09 (s, 3H). ¹³C NMR (101 MHz, CDCl₂) δ 53 173.3, 142.7, 140.1, 137.7, 134.1, 130.2, 128.6, 128.1, 128.1, 127.2, 54 127.1, 20.9. 55

2',6-dimethyl-[1,1'-biphenyl]-2-carboxylic acid (4l). The compound was prepared following general procedure A using 2-bromo-3-methylbenzoic acid (5 mmol) and o-

tolylboronic acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to get a white solid (543 mg 48%). Spectral data is consistent with previous reports.³⁹ ¹H **NMR** (400 MHz, **CDCl**₃) δ 7.78 (d, *J* = 7.8 Hz, 1H), 7.44 (d, *J* = 7.6 Hz, 1H), 7.41 – 7.29 (m, 4H), 7.16 – 7.13 (m, 2H), 2.09 (s, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 173.3, 142.7, 140.1, 137.7, 134.1, 130.2, 128.6, 128.1, 128.1, 127.2, 127.1, 20.9.

2'-methoxy-6-methyl-[1,1'-biphenyl]-2-carboxylic acid (4m). The compound was prepared following general procedure A using 2-bromo-3-methylbenzoic acid (5 mmol) and 2methoxyphenylboronic acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 3:1) as the eluent to get a white solid (605 mg 55%). M. P. 129.8-130.9°C. ¹H NMR (400 MHz, CDCl₃) δ 7.84 (d, J = 7.8 Hz, 1H), 7.47 (d, I = 7.6 Hz, 1H), 7.39 - 7.30 (m, 2H), 7.05 - 7.30 (m, 2H), 6.98 (m, 2H), 6.95 (d, J = 8.4 Hz, 1H), 3.71 (s, 3H), 2.11 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 173.5, 156.5, 139.3, 138.2, 134.1, 130.5, 129.9, 128.9, 128.8, 128.0, 127.1, 120.6, 110.9, 55.6, 20.5. **HRMS (ESI)** calcd for $C_{15}H_{14}O_3Na^+$ [(M+Na)⁺] 265.0835, found 265.0830.

6-(trifluoromethyl)-[1,1'-biphenyl]-2-carboxylic acid (4n). The compound was prepared following general procedure A using 2-bromo-3-(trifluoromethyl)benzoic acid (5 mmol) and phenylboronic acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 4:1) as the eluent to get a white solid (678 mg 51%). ¹H NMR (400 MHz, CDCl₃) δ 8.05 (dd, J = 7.8, 0.7 Hz, 1H), 7.93 (dd, J = 7.9, 0.7 Hz, 1H), 7.57 (td, J = 7.9, 0.6 Hz, 1H), 7.41 - 7.33 (m, 3H), 7.25 - 7.20 (m, 2H). ¹³C NMR (101 MHz, **CDCl**₂) δ 171.6, 141.8, 136.4, 133.1, 132.8, 130.5 (q, *J* = 29.6 Hz), 129.6 (q, J = 5.4 Hz), 129.2 (q, J = 1.2 Hz), 128.1, 127.7, 127.4, 123.6 (q, J = 549.2, 274.6 Hz). ¹⁹F NMR (376 MHz, CDCl₂) δ -57.09 (s). HRMS (ESI) calcd for $C_{14}H_0O_2F_3Na^+$ [(M+Na)⁺] 289.0447, found 289.0444.

5,6-difluoro-[1,1'-biphenyl]-2-carboxylic acid (40). The compound was prepared following general procedure A using 2-bromo-3,4-difluorobenzoic acid (5 mmol) and phenylboronic acid as reactants. The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to get a white solid (550 mg 47%). M. P. 152.1-154.0°C. ¹H **NMR** (400 MHz, CDCl₃) δ 7.79 (ddd, *J* = 8.8, 5.0, 1.9 Hz, 1H), 7.45 - 7.39 (m, 3H), 7.30 - 7.26 (m, 2H), 7.25 - 7.20 (m, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 170.9, 153.5 (dd, *J* = 257.2, 14.2) Hz), 148.2 (dd, *J* = 247.8, 12.7 Hz), 134.0 (dd, *J* = 14.4, 1.0 Hz), 132.7 (d, J = 2.2 Hz), 129.2 (d, J = 1.2 Hz), 128.5, 128.3, 127.4 (dd, *J* = 7.9, 4.5 Hz), 126.4 (dd, *J* = 3.6, 0.9 Hz), 116.1 (d, *J* = 17.7 Hz). ¹⁹F NMR (376 MHz, CDCl₃) δ -129.23 (d, J = 22.0 Hz), -137.74 (d, J = 22.0 Hz). IR (neat/cm⁻¹) 3436, 3056, 2928, 2874, 1709, 1618, 1587, 1577, 1493, 1438, 1423, 1407, 1295, 1281, 1105, 953, 765, 756, 697. HRMS (ESI) calcd for $C_{13}H_8O_2F_2Na^+$ [(M+Na)⁺] 257.0385, found 257.0377.

1-phenyl-2-naphthoic acid (4p). The compound was prepared following general procedure A using 1-bromo-2naphtoic acid (5 mmol) and phenylboronic acid as reactants.

The methyl ester intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 5:1) as the eluent to get a white solid (931 mg 75%). Spectral data is consistent with previous reports.⁴⁰ ¹H NMR (400 MHz, CDCl₃) δ 8.01 (d, *J* = 8.7 Hz, 1H), 7.92 (d, J = 9.2 Hz, 2H), 7.61 – 7.53 (m, 2H), 7.51 – 7.39 (m, 4H), 7.35 – 7.28 (m, 2H). ¹³C NMR (101 MHz, CDCl₂) δ 173.6, 142.9, 138.7, 135.3, 132.9, 129.7, 128.2, 128.0, 128.0, 128.0, 127.9, 127.6, 126.8, 126.7, 126.0.

1

2

3

4

5

6

7

8

9

58 59

60

6-methoxy-[1,1'-biphenyl]-2-carboxylic acid (4q). The 10 compound was prepared following general procedure B using 11 2-bromo-3-hydroxybenzaldehyde (5 mmol), iodomethane 12 and phenylboronic acid as reactants. The 2-arylbenzalhedyde 13 intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 20:1). The carboxylic acid was isolated by col-14 umn chromatography, using petroleum ether/ethyl acetate 15 (PE:EA = 3:1) as the eluent to get a white solid (798 mg 70%). 16 Spectral data is consistent with previous reports.⁴¹ H NMR 17 (400 MHz, CDCl₂) δ 7.50 (dd, J = 7.8, 1.1 Hz, 1H), 7.42 - 7.33 18 (m, 4H), 7.27 (d, J = 1.7 Hz, 1H), 7.26 – 7.24 (m, 1H), 7.13 (dd, J 19 = 8.3, 0.9 Hz, 1H), 3.75 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 20 172.9, 157.2, 136.4, 131.9, 131.8, 129.6, 128.5, 127.9, 127.3, 122.3, 21 114.8, 56.2. HRMS (ESI) calcd for $C_{14}H_{12}O_3Na^+$ [(M+Na)⁺] 22 251.0679, found 251.0673.

23 6-amino-[1,1'-biphenyl]-2-carboxylic acid (4s). The com-24 pound was prepared following general procedure A using 3amino-2-bromobenzoic acid (10 mmol) and phenylboronic 25 acid as reactants. The methyl ester intermediate was purified 26 using petroleum ether/ethyl acetate (PE:EA = 4:1). The car-27 boxylic acid was isolated by column chromatography, using 28 petroleum ether/ethyl acetate/acetic acid (PE:EA:AcOH = 29 1:1:0.02) as the eluent to get a red solid (852 mg 40%). ¹H 30 NMR (400 MHz, DMSO) δ 7.47 - 7.37 (m, 6H), 7.30 - 7.25 31 (m, 2H). ¹³C NMR (101 MHz, DMSO) δ 168.6, 135.0, 134.0, 32 133.9, 132.8, 129.4, 128.5, 128.3, 127.9, 125.5, 124.0. HRMS (ESI) 33 calcd for $C_{12}H_{11}O_2NNa^+$ [(M+Na)⁺] 236.0682, found 236.0679.

34 6-((4-methylphenyl)sulfonamido)-[1,1'-biphenyl]-2**carboxylic acid (4t).** Following a literature procedure :⁴² To 35 a mixture of TsCl (715 mg, 3.75 mmol) in pyridine (7.5 mL) at 36 o^oC was added 6-amino-[1,1'-biphenyl]-2-carboxylic acid (4s) 37 (533 mg, 2.5 mmol) and the resulting reaction mixture stirred 38 for 20 h at room temperature. Removal of the solvent under 39 vacuum yielded a residue which was purified by column 40 chromatography (PE:EA = 1:1) to afford the product 4t as a 41 light yellow solid (90%, 826mg). M. P. 146.8-148.0 °C. 'H 42 **NMR** (400 MHz, CDCl₂) δ 7.98 (dd, J = 8.2, 1.2 Hz, 1H), 7.70 43 (dd, J = 7.9, 1.2 Hz, 1H), 7.42 - 7.39 (m, 3H), 7.35 (dt, J = 2.7, 44 1.9 Hz, 1H), 7.31 - 7.28 (m, 1H), 7.20 (dd, J = 8.6, 0.6 Hz, 2H), 45 6.63 - 6.60 (m, 2H), 6.29 (s, 1H), 2.41 (s, 3H). ¹³C NMR (101 46 MHz, CDCl₂) δ 171.0, 144.3, 135.9, 135.6, 134.9, 134.3, 130.3, 129.8, 129.1, 128.8, 128.6, 128.5, 127.4, 127.0, 125.2, 21.7. IR 47 (neat/cm⁻¹) 3537, 3459, 3275, 3065, 3028, 2923, 1712, 1698, 48 1636, 1597, 1578, 1499, 1463, 1390, 1324, 1290, 1167, 1090, 975, 49 854, 816, 761, 747, 697, 665, 556. HRMS (ESI) calcd for 50 $C_{20}H_{17}O_4NNaS^+$ [(M+Na)⁺] 390.0770, found 390.0768. 51

6-ethoxy-[1,1'-biphenyl]-2-carboxylic acid (4u). The com-52 pound was prepared following general procedure B using 2-53 bromo-3-hydroxybenzaldehyde (10 mmol), bromoethane and 54 phenylboronic acid as reactants. The 2-arylbenzalhedyde 55 intermediate was purified using petroleum ether/ethyl ace-56 tate (PE:EA = 50:1). The carboxylic acid was isolated by col-57 umn chromatography, using petroleum ether/ethyl acetate

(PE:EA = 5:1) as the eluent to get a yellow solid (1.80 g 76%). M. P. 122.9-125.1 °C. ¹H NMR (400 MHz, CDCl₂) δ 7.48 (dd, J = 7.8, 1.1 Hz, 1H), 7.39 – 7.31 (m, 4H), 7.27 (d, J = 1.7 Hz, 1H), 7.25 (d, J = 1.4 Hz, 1H), 7.12 (dd, J = 8.2, 0.9 Hz, 1H), 3.97 (q, J = 7.0 Hz, 2H), 1.22 (t, J = 7.0 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) & 172.9, 156.5, 136.3, 132.2, 131.7, 129.6, 128.3, 127.6, 127.0, 122.2, 116.4, 64.7, 14.6. HRMS (ESI) calcd for $C_{15}H_{14}O_{2}Na^{+}$ [(M+Na)⁺] 265.0835, found 265.0837.

6-ethoxy-4'-methoxy-[1,1'-biphenyl]-2-carboxylic acid (4v). The compound was prepared following general procedure B using 2-bromo-3-hydroxybenzaldehyde (5 mmol), bromoethane and 4-methoxyphenylboronic acid as reactants. The 2-arylbenzalhedyde intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 40:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 3:1) as the eluent to get a light yellow solid (884 mg 65%). M. P. 119.6-121.7 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.47 (dd, J = 7.8, 1.1 Hz, 1H), 7.33 (t, J = 8.0 Hz, 1H), 7.21 (dt, 2H), 7.10 (dd, J = 8.2, 1.0 Hz, 1H), 6.91 (dt, 2H), 3.98 (q, J = 7.0 Hz, 2H), 3.84 (s, 3H), 1.25 (t, J = 7.0 Hz, 2H), 3.84 (s, 3H), 1.25 (t, J = 7.0 Hz, 2H), 3.84 (s, 3H), 1.25 (t, J = 7.0 Hz, 2H), 3.84 (s, 3H), 3.84 (s, 3H)Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 173.6, 158.8, 156.7, 132.1, 131.9, 130.9, 128.6, 128.1, 122.4, 116.4, 113.3, 64.8, 55.3, 14.7. IR (neat/cm⁻¹) 3437, 3065, 2978, 2959, 2931, 2882, 2838, 2649, 2571, 1698, 1611, 1591, 1577, 1515, 1465, 1452, 1443, 1412, 1395, 1302, 1257, 1239, 1175, 1112, 1054, 828, 770, 758. HRMS (ESI) calcd for $C_{16}H_{16}O_4Na^+$ [(M+Na)⁺] 295.0941, found 295.0936.

6-ethoxy-3',5'-dimethyl-[1,1'-biphenyl]-2-carboxylic acid (4w). The compound was prepared following general procedure B using 2-bromo-3-hydroxybenzaldehyde (5 mmol), bromoethane and 3,5-dimethylphenylboronic acid as reactants. The 2-arylbenzalhedyde intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 50:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 4:1) as the eluent to get a light yellow solid (946 mg 70%). M. P. 143.7-145.4 °C. ¹H **NMR** (400 MHz, CDCl₃) δ 7.47 (dd, J = 7.8, 1.0 Hz, 1H), 7.34 (t, J = 8.0 Hz, 1H), 7.10 (dd, J = 8.3, 1.0 Hz, 1H), 6.96 (s, 1H), 6.91 (s, 2H), 3.99 (q, J = 7.0 Hz, 2H), 2.32 (s, 6H), 1.25 (t, J =7.0 Hz, 3H). ¹³C NMR (101 MHz, CDCl₂) δ 172.5, 156.6, 137.1, 136.0, 132.3, 132.0, 129.1, 128.2, 127.6, 122.3, 116.5, 64.8, 21.5, 14.7. **HRMS (ESI)** calcd for $C_{17}H_{18}O_3Na^+$ [(M+Na)⁺] 293.1148, found 293.1140.

6-ethoxy-4'-(trifluoromethyl)-[1,1'-biphenyl]-2-carboxylic acid (4x). The compound was prepared following general procedure B using 2-bromo-3-hydroxybenzaldehyde (5 mmol), bromoethane and 4-(trifluoromethyl)phenylboronic acid as reactants. The 2-arylbenzalhedyde intermediate was purified using petroleum ether/ethyl acetate (PE:EA = 50:1). The carboxylic acid was isolated by column chromatography, using petroleum ether/ethyl acetate (PE:EA = 3:1) as the eluent to get a white solid (868 mg 56%). M. P. 168.0-169.2 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.61 (d, J = 8.0 Hz, 2H), 7.55 (dd, J = 7.8, 1.1 Hz, 1H), 7.40 (t, J = 8.1 Hz, 1H), 7.35 (d, J = 7.9)Hz, 2H), 7.14 (dd, J = 8.3, 0.9 Hz, 1H), 3.98 (q, J = 7.0 Hz, 2H), 1.22 (t, J = 7.0 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 172.7, 156.5, 140.6, 131.2, 131.1, 130.1, 129.2 (q, J = 32.3 Hz), 129.1, 124.6 (q, J = 271.9 Hz), 124.6 (q, J = 3.7 Hz), 122.7, 116.6, 64.8, 14.6.¹⁹F NMR (376 MHz, CDCl₂) δ -62.30 (s). IR (neat/cm⁻¹) 3424, 3072, 2988, 2914, 2880, 2676, 2643, 2571, 1705, 1616, 1594, 1580, 1454, 1405, 1325, 1303, 1262, 1158, 1119, 1066, 868, 836, 760. **HRMS (APCI)** calcd for $C_{16}H_{14}O_3F_3^+$ [(M+H)⁺] 311.0890, found 311.0890.

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

58 59

60

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Copies of ¹H, ¹³C and ¹⁹F (if applicable) NMR spectra.

AUTHOR INFORMATION

Corresponding Authors

* E-Mail: nicolas.guimond@bayer.com

* E-Mail: fuyao@ustc.edu.cn

Present Addresses

Author Contributions

The manuscript was written through contributions of all authors. / All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

We thank Prof. Lei Liu and Tom Kinzel for helpful discussions.

REFERENCES

¹ For selected reviews on direct arylation, see: a) Baudoin, O. Transition Metal-Catalyzed Arylation of Unactivated C(sp³)-H Bonds *Chem. Soc. Rev.* 2011, 40, 4902-4911. b) Lyons, T. W.; Sandford, M. S. Palladium-Catalyzed Ligand-Directed C-H Functionalization Reactions *Chem. Rev.* 2010, 110, 1147-1169. c) Ackermann, L.; Vicente, R. Kapdi, A. R. Transition-Metal-Catalyzed Direct Arylation of (Hetero)Arenes by C-H Bond Cleavage *Angew. Chem. Int. Ed.* 2009, 48, 9792-9826.

² For a review, see: Ackermann, L. Carboxylate-Assisted Transition-Metal-Catalyzed C–H Bond Functionalizations: Mechanism and Scope *Chem. Rev.* **2011**, *111*, 1315-1345.

44 ³ a) Davies, D. L.; Macgregor, S. A.; McMullin, C. L. 45 Computational Studies of Carboxylate-Assited C-H Acti-46 vation and Functionalization at Group 8-10 Transition Met-47 al Centers Chem. Rev. 2017, 117, 8649-8709 b) Gorelsky, 48 S. I.; Lapointe, D.; Fagnou, K. Analysis of the Palladium-49 Catalyzed (Aromatic)C-H Bond Metalation-Deprotonation 50 Mechanism Spanning the Entire Spectrum of Arenes J. 51 Org. Chem. 2012, 77, 658-668. c) Lapointe, D.; Fagnou, K. 52 Overview of the Mechanistic Work on the Concerted 53 Metallation-Deprotonation Pathway Chem. Lett. 2010, 39, 54 1118-1126. and references cited therein.

 ⁴ For seminal report, see: Lafrance, M.; Fagnou, K. Palladium-Catalyzed Benzene Arylation: Incorporation of Catalytic Pivalic Acid as Proton Shuttle and a Key Element in Catalyst Design J. Am. Chem. Soc. 2006, 128, 16496-16497.

⁵ Wang, P.; Verma, P.; Xia, G.; Shi, J.; Qiao, J. X.; Tao, S.; Cheng, P. T. W.; Poss, M. A.; Farmer, M. E.; Yeung, K.-S.; Yu, J.-Q. Ligand-accelerated non-directed C–H functionalization of arenes *Nature* **2017**, *551*, 489-494.

⁶ For a review, see: Engle, K. M.; Yu, J.-Q. Developing Ligands for Palladium(II)-Catalyzed C–H Functionalization: Intimate Dialogue between Ligand and Substrate *J. Org. Chem.* **2013**, *78*, 8927-8955.

⁷ For selected examples, see: a) Viart, H. M.-F.; Bachmann, A.; Kavitare, W.; Sarpong, R. β-Carboline Amides as Intrinsic Directing Groups for $C(sp^2)$ -H Functionalization J. Am. Chem. Soc. 2017, 139, 1325-1329. b) Fujihara, T.; Yoshida, A.; Satou, M.; Tanji, Y.; Terao, J.; Tsuji, Y. Steric Effect of Carboxylic Acid Ligands on Pd-Catalyzed C-H Activation Reactions Catal. Commun. 2016, 84, 71-74. c) Feng, Y.; Wang, Y.; Landgraf, B.; Liu, S.; Chen, G. Facile Benzo-Ring Construction via Palladium-Catalyzed Functionalization of Unactivated sp³ C-H Bonds under Mild Reaction Conditions Org. Lett. 2010, 12, 3414-3417. d) Ackermann, L.; Vicente, R. Catalytic Direct Arylations in Polyethylene Glycol (PEG): Recyclable Palladium(0) Catalyst for C-H Bond Cleavages in the Presence of Air Org. Lett. 2009, 11, 4922-4925. e) Lebrasseur, N.; Larrosa, I. Room Temperature and Phosphine Free Palladium Catalyzed Direct C-2 Arylation of Indoles J. Am. Chem. Soc. 2008, 130, 2926-2927.

⁸ For examples applied to enantioselective C–H bond functionalization, see: a) Yang, L.; Neuburger, M.; Baudoin, O. Chiral Bifunctional Phosphine-Carboxylate Ligands for Palladium(0)-Catalyzed Enantioselective C–H Arylation *Angew. Chem. Int. Ed.* **2018**, *57*, 1394-1398. b) Saget, T; Lemouzy, S. J.; Cramer, N. Chiral Monodentate Phosphines and Bulky Carboxylic Acids: Cooperative Effects in Palladium-Catalyzed Enantioselective C(sp³)–H Functionalization *Angew. Chem. Int. Ed.* **2012**, *51*, 2238-2242.

⁹ For examples of direct arylation of indoles, see: a) Eom, M. S.; Noh, J.; Kim, H.-S.; Yoo, S.; Han, M. S.; Lee, S. High-Throughput Screening Protocol for the Coupling Reactions of Aryl Halides Using a Colorimetric Chemosensor for Halide Ions *Org. Lett.* **2016**, *18*, 1720-1723. b) Liégault, B.; Petrov, I.; Gorelsky, S. I.; Fagnou, K. Modulating Reactivity and Diverting Selectivity in Palladium-Catalyzed Heteroaromatic Direct Arylation Through the Use of a Chloride Activating/Blocking Group *J. Org. Chem.* **2010**, *75*, 1047-1060. c) Lane, B. S.; Brown, M. A.; Sames, D. Direct Palladium-Catalyzed C-2 and C-3 Arylation of Indoles: A Mechanistic Rationale for Regioselectivity *J. Am. Chem. Soc.* **2005**, *127*, 8050-8057. and references therein.

¹⁰ For reviews on the direct arylation of indoles, see: a) Sandtorv, A. H. Transition Metal-Catalyzed C–H Activation of Indoles. *Adv. Synth. Catal.* **2015**, *357*, 2403-2435. b) Bellina, F.; Rossi, R. Recent Advances in the Synthesis of (Hetero)Aryl-Substituted Heteroarenes via Transition Metal-Catalysed Direct (Hetero)Arylation of Heteroarene C-H Bonds with Aryl Halides or Pseudohalides, Diaryliodonium Salts, and Potassium Aryltrifluoroborates. *Tetrahedron* 2009, 65, 10269-10310. c) Joucla, L.; Djakovitch, L. Transition Metal-Catalysed, Direct and Site-Selective N1-, C2- or C3-Arylation of the Indole Nucleus: 20 Years of Improvements. *Adv. Synth. Catal.* 2009, 351, 673-714.

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

58 59

60

¹¹ The addition of 1 eq. benzonitrile to a working reaction mixture didn't impact the outcome of the reaction.

¹² Tan, Y.; Hartwig, J. F. Assessment of the Intermediacy of Arylpalladium Carboxylate Complexes in the Direct Arylation of Benzene: Evidence for C–H Bond Cleavage by "Ligandless" Species *J. Am. Chem. Soc.* **2011**, *133*, 3308-3311.

¹³ Fairlamb, I.; Reay, A.; Neumann, L. Catalyst Efficacy of Homogeneous and Heterogeneous Palladium Catalysts in the Direct Arylation of Common Heterocycles. *Synlett* **2016**, *27*, 1211-1216.

¹⁴ Arroniz, C.; Denis, J. G.; Ironmonger, A.; Rassias, G.; Larrosa, I. An Organic Cation as a Silver(I) Analogue for the Arylation of sp² and sp³ C-H Bonds with Iodoarenes. *Chem. Sci.* **2014**, *5*, 3509-3514.

¹⁵ Liu, C.; Miao, T.; Zhang, L.; Li, P.; Zhang, Y.; Wang, L. Palladium-Catalyzed Direct C2 Arylation of N-Substituted Indoles with 1-Aryltriazenes. *Chem. Asian J.* 2014, *9*, 2584-2589.

28 2014, 9, 2584-2589.
29 ¹⁶ Lebrasseur, N.; Larrosa, I. Room Temperature and Phosphine Free Palladium Catalyzed Direct C-2
31 Arylation of Indoles. J. Am. Chem. Soc. 2008, 130, 2926-2927.

¹⁷ Cai, C.; Lu, G.-p. Palladium-Catalyzed Direct C-2
Arylation of Indoles with Aryl Halides in Aqueous Medium. *Synlett* 2012, 23, 2992-2996.

¹⁸ Liang, Z.; Yao, B.; Zhang, Y. Pd(OAc)₂-Catalyzed
Regioselective Arylation of Indoles with Arylsiloxane in
Acidic Medium. *Org. Lett.* **2010**, *12*, 3185-3187.

¹⁹ Ackermann, L.; Dell'Acqua, M.; Fenner, S.; Vicente,
R.; Sandmann, R. Metal-Free Direct Arylations of
Indoles and Pyrroles with Diaryliodonium Salts. *Org. Lett.* 2011, *13*, 2358-2360.

²⁰ Gale, D.; Lin, J.; Wilshire, J. The Cyclization of Some
Methylphenylhydrazones to Indoles. The Preparation of
Some Polymethine Dyes Related to Astrazon Orange R. *Aust. J. Chem.* 1976, *29*, 2747-2751.

²¹ Nandi, D.; Jhou, Y.-M.; Lee, J.-Y.; Kuo, B.-C.; Liu,
C.-Y.; Huang, P.-W.; Lee, H. M. Pd(0)-Catalyzed Decarboxylative Coupling and Tandem C-H Arylation/Decarboxylation for the Synthesis of Heteroaromatic Biaryls. *J. Org. Chem.* **2012**, *77*, 9384-9390.

²² Dolla, N. K.; Chen, C.; Larkins-Ford, J.; Rajamuthiah,
R.; Jagadeesan, S.; Conery, A. L.; Ausubel, F. M.; Mylonakis, E.; Bremner, J. B.; Lewis, K.; Kelso, M. J. On
the Mechanism of Berberine-INF55 (5-Nitro-2phenylindole) Hybrid Antibacterials. *Aust. J. Chem.* **2015**, *67*, 1471-1480.

²³ Denmark, S. E.; Baird, J. D.; Regens, C. S. Palladium-Catalyzed Cross-Coupling of Five-Membered Heterocyclic Silanolates. *J. Org. Chem.* **2008**, *73*, 1440-1455.

²⁴ Ahmed, J.; Sau, S. C.; P, S.; Hota, P. K.; Vardhanapu, P. K.; Vijaykumar, G.; Mandal, S. K. Direct C-H Arylation of Heteroarenes with Aryl Chlorides by Using an Abnormal N-Heterocyclic-Carbene-Palladium Catalyst. *Eur. J. Org. Chem.* **2017**, *2017*, 1004-1011.

²⁵ Liu, C.; Ding, L.; Guo, G.; Liu, W.; Yang, F.-L. Palladium-Catalyzed Direct Arylation of Indoles with Arylsulfonyl Hydrazides. *Org. Biomol. Chem.* **2016**, *14*, 2824-2827.

²⁶ Zhang, L.; Li, P.; Liu, C.; Yang, J.; Wang, M.; Wang, L. A Highly Efficient and Recyclable Fe₃O₄ Magnetic Nanoparticle Immobilized Palladium Catalyst for the Direct C-2 Arylation of Indoles with Arylboronic Acids. *Catal. Sci. Technol.* **2014**, *4*, 1979-1988.

²⁷ Von Angerer, E.; Prekajac, J.; Strohmeier, J. 2-Phenylindoles. Relationship between Structure, Estrogen Receptor Affinity, and Mammary Tumor Inhibiting Activity in the Rat. *J. Med. Chem.* **1984**, *27*, 1439-1447.

²⁸ Liu, B.; Song, C.; Sun, C.; Zhou, S.; Zhu, J. Rhodium(III)-Catalyzed Indole Synthesis Using N-N Bond as an Internal Oxidant. *J. Am. Chem. Soc.* **2013**, *135*, 16625-16631.

²⁹ a) Dahl, B. J.; Branchaud, B. P. 180° Unidirectional Bond Rotation in a Biaryl Lactone Artificial Molecular Motor Prototype. *Org. Lett.* **2006**, *8*, 5841-5844. b) Shen, X.; Jones, G. O.; Watson, D. A.; Bhayana, B.; Buchwald, S. L. Enantioselective Synthesis of Axially Chiral Biaryls by the Pd-Catalyzed Suzuki-Miyaura Reaction: Substrate Scope and Quantum Mechanical Investigations. *J. Am. Chem. Soc.* **2010**, *132*, 11278-11287. c) Tarus, B.; Bertrand, H.; Zedda, G.; Di Primo, C.; Quideau, S.; Slama-Schwok, A. Structure-Based Design of Novel Naproxen Derivatives Targeting Monomeric Nucleoprotein of Influenza A virus. *J. Biomol. Struct. Dyn.* **2015**, *33*, 1899-1912.

³⁰ Zhu, C.; Zhang, Y.; Kan, J.; Zhao, H.; Su, W. Ambient-Temperature Ortho C-H Arylation of Benzoic Acids with Aryl Iodides with Ligand-Supported Palladium Catalyst. *Org Lett* **2015**, *17*, 3418-3421.

³¹ Fodor, A.; Magyar, Á.; Barczikai, D.; Pirault-Roy, L.; Hell, Z. Study of the Structure-Activity Relationship in a Heterogeneous Copper–Palladium Catalysed Suzuki-Miyaura Coupling. *Catal. Lett.* **2015**, *145*, 834-839.

³² Liu, K.-M.; Zhang, R.; Duan, X.-F. Room-Temperature Cobalt-Catalyzed Arylation of Aromatic Acids: Overriding the Ortho-Selectivity via the Oxidative Assembly of Carboxylate and Aryl Titanate Reagents using Oxygen. *Org. Biomol. Chem.* **2016**, *14*, 1593-1598.

1
2
3
4
5
6
7
7 8
8
9
10 11
11
12
13
14
15 16
16
17
17 18
10
19 20
20
21 22
22
23
23 24 25
27
25 26
26
27 28
28
29 30 31 32
30
21
51
32
33
34
35
 33 34 35 36 37 38
20
3/
39
40
41
42
42
44
45
46
47
48
49
50
51
52
53
54
54 55
56
57
58

60

³³ Wang, Y.; Gulevich, A. V.; Gevorgvan, V. General and Practical Carboxyl-Group-Directed Remote C-H Oxygenation Reactions of Arenes. Chem. Eur. J. 2013, 19, 15836-15840. ³⁴ Fukuyama, T.; Maetani, S.; Miyagawa, K.; Ryu, I. Synthesis of Fluorenones through Rhodium-Catalyzed Intramolecular Acylation of Biarylcarboxylic Acids. Org. Lett. 2014, 16, 3216-3219. ³⁵ Zeng, J.; Liu, K. M.; Duan, X. F. Selective Co/Ti Cooperatively Catalyzed Biaryl Couplings of Aryl Halides with Aryl Metal Reagents. Org. Lett. 2013, 15, 5342-5345. ³⁶ Papaianina, O.; Amsharov, K. Y. Aluminum Oxide Mediated C-F Bond Activation in Trifluoromethylated Arenes. Chem. Commun. 2016, 52, 1505-1508. ³⁷ Dai, J.-J.; Xu, W.-T.; Wu, Y.-D.; Zhang, W.-M.; Gong, Y.; He, X.-P.; Zhang, X.-Q.; Xu, H.-J. Silver-Catalyzed $C(sp^2)$ -H Functionalization/C-O Cyclization Reaction at Room Temperature. J. Org. Chem. 2015, 80, 911-919. ³⁸ Korolev, D. N.; Bumagin, N. A. An Improved Protocol for Ligandless Suzuki-Miyaura Coupling in Water. Tetrahedron Lett. 2006, 47, 4225-4229. ³⁹ Li, Y.; Ding, Y.-J.; Wang, J.-Y.; Su, Y.-M.; Wang, X.-S. Pd-Catalyzed C-H Lactonization for Expedient Synthesis of Biaryl Lactones and Total Synthesis of Cannabinol. Org. Lett. 2013, 15, 2574-2577. ⁴⁰ Aissaoui, R.; Nourry, A.; Coquel, A.; Dao, T. T. H.; Derdour, A.; Helesbeux, J.-J.; Duval, O.; Castanet, A.-S.; Mortier, J. ortho-Lithium/Magnesium Carboxylate-Driven Aromatic Nucleophilic Substitution Reactions on Unprotected Naphthoic Acids. J. Org. Chem. 2012, 77, 718-724. ⁴¹ Meyers, A. I.; Gabel, R.; Mihelich, E. D. Nucleophilic Aromatic Substitution on o-(Methoxy)aryloxazolines. A Convenient Synthesis of o-Alkyl-, o-Alkylidene-, and o-Arylbenzoic Acids. J. Org. Chem. 1978, 43, 1372-1379. ⁴² Aderibigbe, B. A.; Green, I. R.; Mabank, T.; Janse van Rensburg, M.; Morgans, G. L.; Fernandes, M. A.; Michael, J. P.; van Otterlo, W. A. L. Observations Concerning the Synthesis of Heteroatom-Containing 9-Membered Benzo-Fused Rings by Ring-Closing Metathesis. Tetrahedron 2017, 73, 4671-4683.