

Synthesis and biological evaluation of thiazolo-triazole derivatives*

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Summary — Two series of isomeric thiazolo[3,2-*b*][1,2,4]triazole and thiazolo[2,3-*c*][1,2,4]triazole derivatives were prepared following multiple synthetic pathways. The obtained compounds were submitted to preliminar pharmacological assays to evaluate their anti-inflammatory, analgesic and antipyretic activity. Suggestions about structure–activity relationships between the two classes of isomers were delineated. Moreover, some of the starting molecules, phenacylthio[1,2,4]triazoles were submitted to microbiological analysis to test their antibacterial and antimycotic activity.

Résumé — **Synthèse et évaluation biologique de thiazolo-triazoles.** Deux séries de thiazolo[3,2-*b*][1,2,4]triazoles et de thiazolo[2,3-*c*][1,2,4]triazoles isomères ont été préparés en utilisant de nombreuses voies synthétiques. Les composés obtenus ont été soumis à des essais pharmacologiques préliminaires, en vue d'évaluer leur activité anti-inflammatoire, analgésique et antipyrétique. Des suggestions quant à des relations structure–activité entre les deux classes d'isomères ont été émises. De plus, certaines des matières premières phénacylthio[1,2,4]triazoles ont été soumises à une analyse microbiologique dans le but d'évaluer leur activité antibactérienne et antimycotique.

thiazolo[3,2-*b*][1,2,4]triazoles / thiazolo[2,3-*c*][1,2,4]triazoles / anti-inflammatory activity / analgesic activity / antipyretic activity / microbiological tests

Introduction

Thiazole nucleus is known to be present in various molecules having a biological activity [1–13], as well as 4H-1,2,4-triazole moiety takes a part of antiflogistic and antipyretic compounds [14–20]. Such activities have been also found in molecules containing the fused thiazolo-triazole bicyclic system [14, 16]. To the best of our knowledge, however, literature reports only few data regarding pharmacological aspects in the several papers describing the synthesis of thiazolo-triazole derivatives [21–33].

From these considerations, our aim in the present work was – which is a part of a research program directed to obtain compounds of pharmaceutical interest – through chemical and structural variations on molecules whose activity has been already shown, to

make a contribution to the structure–activity studies on the thiazolo-triazole nucleus, which can exist in both the isomer forms: thiazolo[3,2-*b*][1,2,4]triazole and thiazolo[2,3-*c*][1,2,4]triazole.

Therefore, starting from some 3-mercapto-5-alkoxyphenyl-4H-1,2,4-triazoles **Ia–c**, described and tested in a previous work [34], we prepared the two series of isomeric bicyclic derivatives: 2,6-disubstituted thiazolo[3,2-*b*][1,2,4]triazole **1–27** and 3,5-disubstituted thiazolo[2,3-*c*][1,2,4]triazole **28–51** (schemes 1 and 2).

Biological significance of alkoxyphenyl radicals present in the synthesized compounds, in 2- or 3-position of the two series, respectively, has been well established in precedent papers [35–37]. Substitution in –5 or –6 positions by phenyl or substituted phenyl moieties was also developed to evaluate the influence of groups with different sterical, electronic and hydro/lipophilicity properties on the pharmacological activity of the resulting compounds.

Furthermore, by considering that several S-substituted thio-1,2,4-triazoles have shown antibacterial, antiviral

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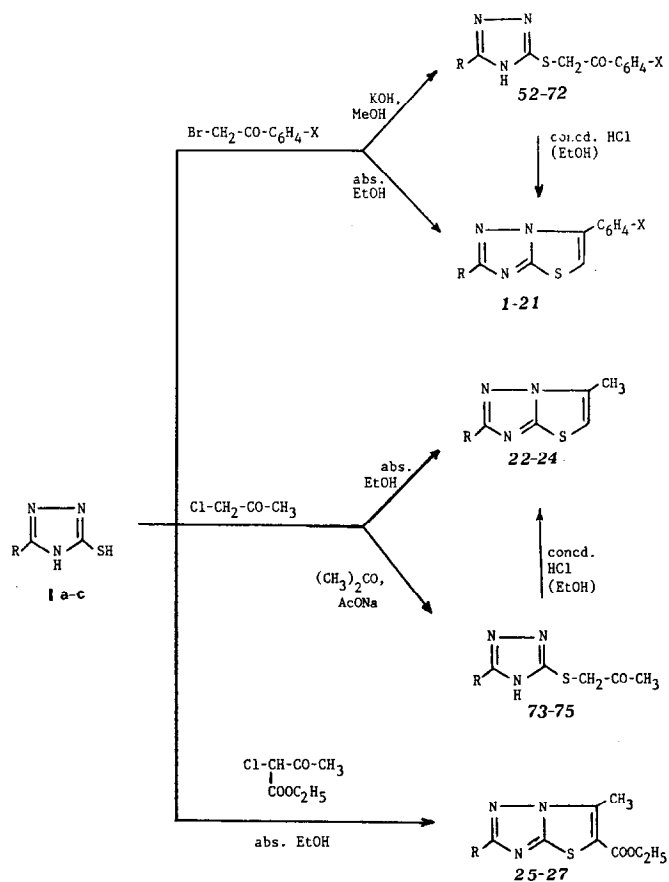
as well as antifungal activities [38], some of the starting products, 5-alkoxyphenyl-3-phenacylthio-s-triazoles, were also submitted to microbiological tests.

Chemistry

Schemes 1 and 2 report the reactions which led to thiazolo-triazole derivatives.

Derivatives of thiazolo[3,2-*b*][1,2,4]triazole series **1–27** were obtained either by a direct reaction between mercaptotriazoles **I** [34] and the appropriate α -halogenoketone (phenacyl bromides, chloroacetone or ethyl 2-chloroacetoacetate), or through an alternative pathway involving the formation of 3-phenacyl-(acetyl)thiotriazoles **52–75**, which were then cyclized by an acid-catalyzed reaction (scheme 1). In the first case, compounds were obtained in a 6–8 h period with good yields, whereas in the alternative reaction, after the quite instantaneous step leading phenacylthio-derivatives **52–75**, their cyclization into compounds **1–27** requires a very long time, showing a clear relation with the substituent present in the phenacyl bromide. In fact, in the presence of an electron-donor group an 8-h period is sufficient, whereas with an electron-withdrawing substituent 24 h were necessary for the reaction to be completed.

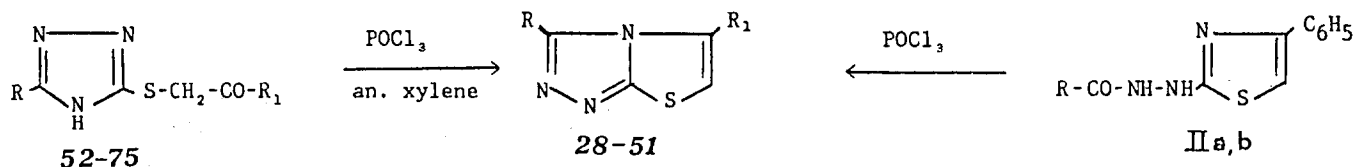
Thiazolo[2,3-*c*][1,2,4]triazole derivatives **28–51**, as described in [21], were prepared from the same ketones **52–75**, by cyclization with POCl_3 in anhydrous xylene (scheme 2): compounds so obtained showed different mp and solubility properties than the corresponding thiazolo[3,2-*b*][1,2,4]triazole isomers **1–27**. However, to have a confirmation of the occurred cyclization into thiazolo[2,3-*c*][1,2,4]triazole form, we applied, for some terms, an alternative and more specific synthesis: 1-alkoxyaroyl-2-[4-phenyl-thiazol-2-yl]-hydrazines **II**, described in [34], were cyclized in POCl_3 , without any solvent and at 120–130°C for some hours, to give thiazolo-



Scheme 1.

I a: $\text{R} = 3,4,5\text{-(OCH}_3)_3\text{-C}_6\text{H}_2$;
 b: $\text{R} = 3,4\text{-(O-CH}_2\text{-O)-C}_6\text{H}_3$;
 c: $\text{R} = 3,5\text{-(OCH}_3)_2\text{-4-OC}_2\text{H}_5\text{-C}_6\text{H}_2$.

X = H; p-Cl; p-Br; p-CH₃;
 p-OCH₃; p-C₆H₅; p-NO₂.



Scheme 2.

Compds **28–34**, **49**, **52–58**, **73**, **IIa**: $\text{R} = 3,4,5\text{-(OCH}_3)_3\text{-C}_6\text{H}_2$
 Compds **35–41**, **50**, **59–65**, **74**, **IIb**: $\text{R} = 3,4\text{-(O-CH}_2\text{-O)-C}_6\text{H}_3$
 Compds **42–48**, **51**, **66–72**, **75**: $\text{R} = 3,5\text{-(OCH}_3)_2\text{-4-OC}_2\text{H}_5\text{-C}_6\text{H}_2$

$\text{R}_1 = \text{C}_6\text{H}_5$; $\text{C}_6\text{H}_4\text{-Cl(p)}$; $\text{C}_6\text{H}_4\text{-Br(p)}$; $\text{C}_6\text{H}_4\text{-CH}_3\text{(p)}$; $\text{C}_6\text{H}_4\text{-OCH}_3\text{(p)}$; $\text{C}_6\text{H}_4\text{-C}_6\text{H}_5\text{(p)}$; $\text{C}_6\text{H}_4\text{-NO}_2\text{(p)}$; CH_3

[2,3-*c*][1,2,4]triazole derivatives (compound **28** and **35**, respectively) (scheme 2) [21]. Compounds obtained by this way are identical (mixed mp, superimposable IR spectrum) to those prepared following the previous method. It should be noted that the last synthesis had only a speculative value in our study, as it gives very low yields (20–25%), with respect to the first procedure. The structure assigned to the prepared compounds is in agreement with elemental analysis and IR, ¹H-NMR and mass spectral data. In particular, it could be pointed out that:

A) Thiazolo[3,2-*b*][1,2,4]triazole derivatives **1–27** show, in their IR spectra, the disappearance of N–H and C = O vibration bands, present in the starting materials **1 a–c** or **52–75**. The only relevant signals refer to C = N and C–C bonds (1610–1550 cm⁻¹) and to thiazole C–S–C (around 1125 cm⁻¹). In the ¹H-NMR spectrum, thiazolic H₆ proton appears as a singlet at δ 7.90–7.70, in agreement with data reported for analogous compounds [39]. In the mass analysis, molecular ion represents the base peak, while the remaining fragmentation scheme agrees with data scheduled for the fused thiazolo-triazole nucleus [39, 40].

B) Thiazolo[2,3-*c*][1,2,4]triazole derivatives **28–51** show IR and NMR spectra analogous to those of the isomers **1–27**; nevertheless, thiazole H₆ appears as a singlet at δ 7.90–7.45, lightly shifted to lower fields than the corresponding isomers, differently from what is reported by some authors for similar molecules [39]; mass spectrometry also gives the same results as for thiazolo[3,2-*b*][1,2,4]triazole series, with the molecular ion as the base peak. However, in this case it is evident the signal deriving from the loss of a N₂ molecule [M⁺–28], whose peak is rilevable for the beginning of the fragmentation. This finding is in agreement with previous reports [40] that indicate this kind of cleavage to be easier in heterocyclic derivatives fused on 'c' side.

C) In phenacyl(acetonyl)thiotriazoles **52–75**, carbonyl group gives a strong band at 1705–1675 cm⁻¹ in the IR spectra, while NMR analysis shows, in particular, the singlet due to the methylene group (δ 4.90–4.85). In the mass spectra (compounds **52** and **60**), the fragmentation pattern follows what is expected for these compounds: C₆H₅–CO⁺ ion (*m/z* 105) is the base peak, with an intense M⁺ ion (50–60%), in analogy with literature data [41].

Results and discussion

Results of the pharmacological assays relative to the tested thiazolo-triazole derivatives are gathered in tables IV and V.

The orientative acute toxicity assay (LD₅₀) showed that none of the evaluated compounds produced lethal effects up to the maximum dosage administered (1200 mg/kg *po* and 500 mg/kg *sc*), both in rats and mice. Compounds did not induce any significant behavioural modification at the employed dose.

Carrageenan induced oedema test indicates that the examined compounds showed, at the dose of 100 mg/kg *po*, an anti-inflammatory activity, in some cases (compounds **5** and **12**) comparable to that of ASA, given at the same dose (table IV). No activity was instead found in derivatives supporting an unsubstituted phenyl in –6 position of thiazolo[3,2-*b*]-[1,2,4]triazole series (compounds **1** and **8**). Similar results are also presented by thiazolo[2,3-*c*][1,2,4]triazole derivatives, except for compounds **28**, **32** and **39**, which show a slight activity (table IV).

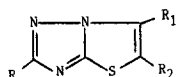
Analgesic activity (phenylquinone writhing test) is present in most of the tested compounds; in particular, while derivatives of [3,2-*b*] series show an inhibition value ranging from 13.3 to 29%, in the isomeric [2,3-*c*] series the equivalent compounds appeared to be more effective, with a percent inhibition of 20–60%, very close, for many terms, to that displayed by PBZ (table IV).

With respect to the antipyretic activity test (yeast induced pyrexia, table V), the examined compounds showed an interesting activity, often near to that of ASA; compound **3** even induced an hypothermal effect that appeared also in some 3-phenacylthio[1,2,4]triazoles derivatives (compounds **61** and **62**) (table V), which showed, in the meantime, also a significative analgesic activity (table IV).

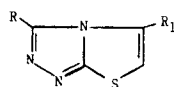
In the microbiological assay, the tested 3-phenacylthio-triazoles exhibited only a slight antimycotic activity (MICs = 25–100 µg/ml) against the used strains (see table VI).

Derivatives containing a *p*-bromophenyl substituent (compounds **54** and **61**) appeared to be as the more effective terms, with MICs ranging from 25 to 50 µg/ml. No significant activity was instead observed on the bacterial species used (MICs ≥ 100 µg/ml).

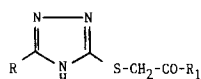
Findings of our preliminar pharmacological tests allow to delineate a relationship between the structure of the synthesized and tested thiazolo-triazole derivatives and their anti-inflammatory activity: i) compounds of [3,2-*b*] series show a good anti-inflammatory activity, whose higher expression appears in derivatives with a *p*-OCH₃ substituent in the phenyl ring in –6 (compounds **5** and **12**); such an activity is lacking when an unsubstituted phenyl is present; ii) the antipyretic activity is also significantly more relevant in the [3,2-*b*] series, and reaches the maximum level in the 6-(*p*-bromophenyl) derivative with a trimethoxyphenyl group in 2-position (compound **3**); in phenacylthio[1,2,4]triazoles series, the same activity is present in derivatives with a pipe-

Table I. 2-Alkoxyphenylthiazolo[3,2-*b*][1,2,4]triazoles 6- and 5,6-substituted 1–27.

Compd no	R	R ₁	R ₂	mp (°C)	Cryst solvent	Yield %	Analysis (C, H, N)
1	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₅	H	159–60	EtOH	82	C ₁₉ H ₁₇ N ₃ O ₃ S
2	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -Cl(p)	H	164–65	EtOH	78	C ₁₉ H ₁₆ N ₃ O ₃ SCl
3	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -Br(p)	H	159–60	EtOH	70	C ₁₉ H ₁₆ N ₃ O ₃ SBr
4	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -CH ₃ (p)	H	166–67	EtOH	80	C ₂₀ H ₁₉ N ₃ O ₃ S
5	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -OCH ₃ (p)	H	166–67	EtOH	65	C ₂₀ H ₁₉ N ₃ O ₄ S
6	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -C ₆ H ₅ (p)	H	177–78	DMF-EtOH	75	C ₂₅ H ₂₁ N ₃ O ₃ S
7	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -NO ₂ (p)	H	236–37	DMF-EtOH	75	C ₁₉ H ₁₆ N ₄ O ₅ S
8	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₅	H	189–90	DMF-EtOH	85	C ₁₇ H ₁₁ N ₃ O ₂ S
9	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -Cl(p)	H	182–83	DMF	80	C ₁₇ H ₁₀ N ₃ O ₂ SCl
10	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -Br(p)	H	203–204	DMF-EtOH	90	C ₁₇ H ₁₀ N ₃ O ₂ SBr
11	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -CH ₃ (p)	H	174	DMF-EtOH	82	C ₁₈ H ₁₃ N ₃ O ₂ S
12	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -OCH ₃ (p)	H	167–68	DMF-EtOH	90	C ₁₈ H ₁₃ N ₃ O ₃ S
13	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -C ₆ H ₅ (p)	H	225–26	DMF-EtOH	96	C ₂₃ H ₁₅ N ₃ O ₂ S
14	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -NO ₂ (p)	H	227–28	DMF-EtOH	86	C ₁₇ H ₁₀ N ₄ O ₄ S
15	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₅	H	149–50	EtOH	80	C ₂₀ H ₁₉ N ₃ O ₃ S
16	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -Cl(p)	H	126–27	EtOH	70	C ₂₀ H ₁₈ N ₃ O ₃ SCl
17	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -Br(p)	H	137–38	EtOH	70	C ₂₀ H ₁₈ N ₃ O ₃ SBr
18	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -CH ₃ (p)	H	139–40	EtOH	88	C ₂₁ H ₂₁ N ₃ O ₃ S
19	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -OCH ₃ (p)	H	178–80	EtOH	73	C ₂₁ H ₂₁ N ₃ O ₄ S
20	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -C ₆ H ₅ (p)	H	184–85	DMF-EtOH	85	C ₂₆ H ₂₃ N ₃ O ₃ S
21	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -NO ₂ (p)	H	184	DMF-EtOH	77	C ₂₀ H ₁₈ N ₄ O ₅ S
22	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	CH ₃	H	154–55	EtOH	78	C ₁₄ H ₁₅ N ₃ O ₃ S
23	3,4-(O-CH ₂ -O)-C ₆ H ₃	CH ₃	H	156–57	DMF-EtOH	80	C ₁₂ H ₉ N ₃ O ₂ S
24	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	CH ₃	H	132–33	EtOH	70	C ₁₅ H ₁₇ N ₃ O ₃ S
25	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	CH ₃	COOC ₂ H ₅	149–50	EtOH	75	C ₁₇ H ₁₉ N ₃ O ₅ S
26	3,4-(O-CH ₂ -O) ₂ -C ₆ H ₃	CH ₃	COOC ₂ H ₅	182–83	DMF-EtOH	80	C ₁₅ H ₁₃ N ₃ O ₄ S
27	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	CH ₃	COOC ₂ H ₅	177–78	EtOH	67	C ₁₈ H ₂₁ N ₃ O ₅ S

Table II. 3-Alkoxyphenyl-5-aryl(methyl)thiazolo[2,3-*c*][1,2,4]triazoles **28–51**.

<i>Compd no</i>	<i>R</i>	<i>R</i> ₁	<i>mp</i> (°C)	<i>Cryst solvent</i>	<i>Yield</i> %	<i>Analysis</i> (C, H, N)
28	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₅	168–69	EtOH	78	C ₁₉ H ₁₇ N ₃ O ₃ S
29	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -Cl(p)	171–72	EtOH	60	C ₁₉ H ₁₆ N ₃ O ₃ SCl
30	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -Br(p)	185–86	EtOH	55	C ₁₉ H ₁₆ N ₃ O ₃ SBr
31	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -CH ₃ (p)	165–66	EtOH	60	C ₂₀ H ₁₉ N ₃ O ₃ S
32	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -OCH ₃ (p)	162–63	DMF-EtOH	70	C ₂₀ H ₁₉ N ₃ O ₄ S
33	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -C ₆ H ₅ (p)	139–40	DMF-EtOH	61	C ₂₅ H ₂₁ N ₃ O ₃ S
34	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -NO ₂ (p)	239–40	EtOH	50	C ₁₉ H ₁₆ N ₄ O ₅ S
35	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₅	198–200	DMF-EtOH	85	C ₁₇ H ₁₁ N ₃ O ₂ S
36	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -Cl(p)	188–89	DMF-EtOH	70	C ₁₇ H ₁₀ N ₃ O ₂ SCl
37	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -Br(p)	208–209	DMF-EtOH	73	C ₁₇ H ₁₀ N ₃ O ₂ SBr
38	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -CH ₃ (p)	183–84	DMF-EtOH	70	C ₁₈ H ₁₃ N ₃ O ₂ S
39	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -OCH ₃ (p)	188–89	DMF-EtOH	65	C ₁₈ H ₁₃ N ₃ O ₃ S
40	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -C ₆ H ₅ (p)	237–38	DMF-EtOH	80	C ₂₃ H ₁₅ N ₃ O ₂ S
41	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -NO ₂ (p)	198–99	DMF-EtOH	65	C ₁₇ H ₁₀ N ₄ O ₄ S
42	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₅	137–38	EtOH	55	C ₂₀ H ₁₉ N ₃ O ₃ S
43	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -Cl(p)	184–85	EtOH	70	C ₂₀ H ₁₈ N ₃ O ₃ SCl
44	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -Br(p)	179–80	EtOH	65	C ₂₀ H ₁₈ N ₃ O ₃ SBr
45	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -CH ₃ (p)	192–93	EtOH	60	C ₂₁ H ₂₁ N ₃ O ₃ S
46	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -OCH ₃ (p)	175–77	EtOH	55	C ₂₁ H ₂₁ N ₃ O ₄ S
47	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -C ₆ H ₅ (p)	193–94	DMF-EtOH	60	C ₂₆ H ₂₃ N ₃ O ₃ S
48	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -NO ₂ (p)	222–23	EtOH	54	C ₂₀ H ₁₈ N ₄ O ₅ S
49	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	CH ₃	123–24	EtOH	45	C ₁₄ H ₁₅ N ₃ O ₃ S
50	3,4-(O-CH ₂ -O)-C ₆ H ₃	CH ₃	136–38	EtOH	60	C ₁₂ H ₉ N ₃ O ₂ S
51	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	CH ₃	140–41	EtOH	60	C ₁₅ H ₁₇ N ₃ O ₃ S

Table III. 5-Alkoxyphenyl-3-phenacyl(acetonyl)thio-4H-1,2,4-triazoles **52–75**.

Compd no	R	R ₁	mp (°C)	Cryst solvent	Yield %	Analysis (C, H, N)	IR (KBr) –NH	cm ⁻¹ C=O
52	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₅	118–19	EtOH	85	C ₁₉ H ₁₉ N ₃ O ₄ S	3620	1685
53	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -Cl(p)	198–99	EtOH	90	C ₁₉ H ₁₈ N ₃ O ₄ SCl	3275	1685
54	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -Br(p)	212–13	EtOH	88	C ₁₉ H ₁₈ N ₃ O ₄ SBr	3280	1685
55	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -CH ₃ (p)	180–81	EtOH	85	C ₂₀ H ₂₁ N ₃ O ₄ S	3280	1680
56	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -OCH ₃ (p)	140–41	EtOH	85	C ₂₀ H ₂₁ N ₃ O ₅ S	3260	1680
57	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -C ₆ H ₅ (p)	119–20	EtOH	60	C ₂₅ H ₂₃ N ₃ O ₄ S	3230	1670
58	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	C ₆ H ₄ -NO ₂ (p)	212–13	EtOH	78	C ₁₉ H ₁₈ N ₃ O ₆ S	3270	1696
59	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₅	169–71	EtOH	95	C ₁₇ H ₁₃ N ₃ O ₃ S	3440 (br)	1702
60	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -Cl(p)	184–85	EtOH	88	C ₁₇ H ₁₂ N ₃ O ₃ SCl	3420 (br)	1690
61	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ Br(p)	183–84	DMF-EtOH	84	C ₁₇ H ₁₂ N ₃ O ₃ SBr	3400 (br)	1686
62	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -CH ₃ (p)	168–70	EtOH	85	C ₁₈ H ₁₅ N ₃ O ₃ S	3430 (br)	1685
63	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -OCH ₃ (p)	149–50	EtOH	80	C ₁₈ H ₁₅ N ₃ O ₄ S	3400 (br)	1675
64	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -C ₆ H ₅ (p)	167–68	DMF-EtOH	80	C ₂₃ H ₁₇ N ₃ O ₃ S	3370 (br)	1680
65	3,4-(O-CH ₂ -O)-C ₆ H ₃	C ₆ H ₄ -NO ₂ (p)	202–203	DMF-EtOH	86	C ₁₇ H ₁₂ N ₃ O ₅ S	3420 (br)	1695
66	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₅	116–17	EtOH	80	C ₂₀ H ₂₁ N ₃ O ₄ S	3560	1678
67	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -Cl(p)	166–67	DMF-EtOH	80	C ₂₀ H ₂₀ N ₃ O ₄ SCl	3510 (br)	1680
68	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -Br(p)	175	EtOH	75	C ₂₀ H ₂₀ N ₃ O ₄ SBr	3260	1692
69	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -CH ₃ (p)	132–33	EtOH	80	C ₂₁ H ₂₃ N ₃ O ₄ S	3570	1680
70	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -OCH ₃ (p)	123–24	EtOH	75	C ₂₁ H ₂₃ N ₃ O ₅ S	3470 (br)	1675
71	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -C ₆ H ₅ (p)	201–203	DMF-EtOH	74	C ₂₆ H ₂₅ N ₃ O ₄ S	3400 (br)	1680
72	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	C ₆ H ₄ -NO ₂ (p)	177–78	EtOH	75	C ₂₀ H ₂₀ N ₃ O ₆ S	3520 (br)	1695
73	3,4,5-(OCH ₃) ₃ -C ₆ H ₂	CH ₃	125–26	EtOH	75	C ₁₄ H ₁₇ N ₃ O ₄ S	3300	1712
74	3,4-(O-CH ₂ -O)-C ₆ H ₃	CH ₃	160–61	EtOH	83	C ₁₂ H ₁₁ N ₃ O ₃ S	3335	1700
75	3,5-(OCH ₃) ₂ -4-OC ₂ H ₅ -C ₆ H ₂	CH ₃	128–29	EtOH	75	C ₁₅ H ₁₉ N ₃ O ₄ S	3280	1710

ronyl radical in –5 (compounds **61** and **62**); iii) it would seem that the [2,3-*c*]-type structure drives the activity of compounds toward the analgesic one; in fact, most derivatives of this series, whilst they did not display any antipyretic activity, show an analgesic activity higher than that of the analogues derivatives owing to [3,2-*b*] series and comparable to that of PBZ, used as a reference drug.

The exhaustive study on all the compounds of the three series, with complete data and, in particular, with the results of tests directed to verify an eventual central component of the analgesic activity, by assaying naloxone-induced antagonism and the potentiating effect of morphine, will be the object of a separate work. Furthermore, *in vitro* experiments will

be performed to better clarify if the anti-inflammatory activity of these substances is directed to the inhibition of prostaglandins synthesis or to a scavenging effect on oxidative species.

Experimental protocols

Chemicals

Melting points were determined on a Büchi model 530 apparatus and are uncorrected. IR spectra were recorded with a Perkin-Elmer model 281 spectrometer in KBr disks; ¹H-NMR spectra on a Bruker AC 250 instrument, at 250 Hz, in DMSO-*d*₆ and using TMS as IS; mass spectra on a Carlo Erba/Kratos model MS 25 RFA instrument, at 70 eV. Elemental analyses were performed on a Carlo Erba model 1106 Analyzer; results are within ± 0.4% of theoretical values.

Table IV. Anti-inflammatory (carrageenan edema test) and analgesic activity (phenylquinone writhing test) of thiazolo-triazole and 3-phenacylthio-triazole derivatives. Values are expressed as the percentage of inhibition. Unreported values (–) indicate an inhibition lower than 10%.

Thiazolo[3,2- <i>b</i>][1,2,4]triazoles				Thiazolo[2,3- <i>c</i>][1,2,4]triazoles				3-Phenacylthio- <i>s</i> -triazoles			
Compound	Anti-inflammatory ^a activity		Analgesic ^b activity	Compound	Anti-inflammatory ^a activity		Analgesic ^b activity	Compound	Anti-inflammatory ^a activity		Analgesic ^b activity
	2 <i>h</i>	4 <i>h</i>			2 <i>h</i>	4 <i>h</i>			2 <i>h</i>	4 <i>h</i>	
1	–	–	28.8*	28	13.0	22.2	55.6**	52	–	–	19.0*
3	35.7*	24.6	26.0*	30	–	–	58.0**	54	–	–	–
4	39.2*	14.2	20.0	31	–	–	48.1*	55	–	–	–
5	26.6	50.0**	23.7**	32	–	14.4	–				
8	–	–	24.7**	35	–	–	19.5*	59	11.8	27.9	39.0**
10	33.9*	7.1	13.0	37	–	–	26.0*	61	–	–	17.0**
11	24.1	15.0	19.0	38	–	–	–	62	–	–	71.0**
12	40.0*	50.0**	15.4	39	–	20.0	59.8**				
			Acetylsalicylic acid		60.0**	78.7**					
			Phenylbutazone				61.0**				

^aCompounds were administered at a dose of 100 mg/kg *po* or ^bsc. *The significance of variation vs controls was $P < 0.05$ and ** $P < 0.01$.

The purity of the synthesized compounds was checked by means of TLC on silica gel aluminium sheets (Merck 60 F₂₅₄).

All the starting reactants were commercially available. Phenacyl bromides were recrystallized from methanol before use.

2-Alkoxyphenyl-thiazolo[3,2-*b*][1,2,4]triazoles 6- and 5,6-substituted 1–27

Method A. From mercaptotriazoles and α -halogeno-ketones condensation: 0.01 mol of 3-mercaptotriazoles **1a–c** [34] were suspended in 150 ml of refluxing absolute ethanol and 0.01 mol of the chosen α -halogenoketone was added. The mixture was refluxed under stirring for 5–24 h, until a clear solution was reached; in some cases small amounts of solvent were added to achieve a complete solubilization. At the end, the solvent was evaporated *in vacuo* and the residue treated with a 20% NaHCO₃ aqueous solution, at 50°C for 30 min; after cooling, the solid was filtered off, washed with water and suitably crystallized. In table I, physico-chemical data of the obtained compounds are reported.

Method B. From phenacyl(acetonyl)-thiotriazoles: an ethanolic solution or suspension of phenacyl(acetonyl)thio-triazole derivatives **52–75** (1 g) was refluxed with 5 ml 10 N HCl for 5 h. After concentration, the obtained solid was collected and treated, for 30 min at 50°C, with a 20% NaHCO₃ aqueous solution. The product was then filtered, washed with water and crystallized. The compounds so prepared resulted identical

(TLC, mixed mp, superimposable IR) to the corresponding derivatives obtained by method A (table I).

3-Alkoxyphenyl-5-aryl(methyl)-thiazolo[2,3-*c*][1,2,4]triazoles 28–51

Method A. From phenacyl(acetonyl)thio-triazoles: 0.01 mol of 3-phenacyl(acetonyl)-thio[1,2,4]triazoles **52–75** were solubilized with heating in anhydrous xylene (60 ml) and 10 ml POCl₃ were added. The reaction was refluxed for 10–15 h and the solvent removed *in vacuo*. After washing with a 20% NaHCO₃ aqueous solution at 50°C, the solid was collected and crystallized from a suitable solvent. Table II lists mp, yield and crystallization solvent of the obtained compounds.

Method B. From alkoxybenzoylthiazol-2-yl-hydrazines: 1 g of 1-alkoxybenzoyl-2-(4-phenylthiazol-2-yl)-hydrazines (compounds **IIa** or **IIb** [34]) was added with 8 ml of POCl₃ and heated at 120–130°C on a silicon oil bath for 3 h. After cooling, the solution was dropped in ice water, obtaining an amorphous solid which, after washing with 10% aqueous Na₂CO₃, diluted HCl and then water, was crystallized from the suitable solvent. The obtained compounds (**28** and **35**) appeared identical (TLC, mixed mp, IR) to those synthesized following the previous method A (table II).

5-Alkoxyphenyl-3-phenacyl(acetonyl)thio-4H-1,2,4-triazoles 52–75

0.01 mol of mercaptotriazoles **1a–c** [34] were dissolved or suspended in methanol (25 ml) containing 0.01 mol of KOH.

Table V. Antipyretic activity (brewer's yeast induced pyrexia) on rats of thiazolo-triazole and phenacylthio-triazole derivatives. Percent of inhibition 1 to 5 h after the pyrogen. Compounds were given orally at 100 mg/kg. Inhibition values lower than 10% have been indicated as —.

Thiazolo[3,2- <i>b</i>][1,2,4]triazoles						Thiazolo[2,3- <i>c</i>][1,2,4]triazoles						3-Phenacylthio- <i>s</i> -triazoles					
Compd	1 h	2 h	3 h	4 h	5 h	Compd	1 h	2 h	3 h	4 h	5 h	Compd	1 h	2 h	3 h	4 h	5 h
1	59*	20	30	71*	72*	28	16	28	45*	35	30	52	—	—	—	—	—
3	100*	> 100*	> 100*	100*	> 100*	30	82*	69*	60*	48*	31*	54	—	—	—	—	—
4	72*	55*	51*	49*	43*	31	67*	51*	53*	55**	51**	55	> 100**	85*	45*	—	—
5	94*	79**	46**	65**	59**	32	—	34	53*	62*	65*						
8	18	34*	54**	50**	46*	35	56	46*	65*	64**	45**	59	—	—	—	—	—
10	70*	93**	71**	61**	40**	37	53	31	30*	14	13	61	> 100**	> 100**	100**	85**	47**
11	86*	92**	74**	61**	50**	38	51**	50*	56**	63**	63*	62	> 100**	> 100**	100**	92**	59*
12	21	34*	57**	61**	69**	39	51	30*	51**	52**	37**						
Acetylsalicylic acid							74*	97*	92*	91*	66*						
Phenylbutazone							96*	94**	97*	100**	100**						

Significance vs controls: * $P < 0.05$; and ** $P < 0.01$.

Table VI. Antimycotic and antibacterial activity relative to phenacylthiotriazoles (MICs: $\mu\text{g/ml}$).

Compound no	Mycetes					Bacteria			
	<i>C albicans</i> 3153	<i>C krusei</i>	<i>C parapsilosis</i>	<i>C guilliermondii</i>	<i>T glabrata</i>	<i>E coli</i> ATCC 25933	<i>E coli</i>	<i>S aureus</i> ATCC 25923	<i>S aureus</i> Oxford
52	100	100	100	50	50	> 100	> 100	> 100	> 100
54	50	25	100	50	50	50	100	100	100
56	50	50	100	25	25	100	> 100	100	100
59	100	100	100	50	50	> 100	> 100	> 100	> 100
61	50	25	50	25	25	50	100	100	100
63	50	50	100	25	25	100	100	100	100
Miconazole	1.56	0.39	0.39	1.56	3.12	12.5	12.5	12.5	12.5

The mixture was heated and, after filtration of the insoluble part, added dropwise and under continuous stirring of 0.01 mol of the wished phenacyl bromide in a 5-ml methanol solution. The reaction was completed on a boiling water bath for 30 min. For acetyl derivatives (compounds 73–75), reactions were carried out in acetone (40 ml), in the presence of 0.015 mol of anhydrous sodium acetate and 0.012 mol of chloroacetone.

In both the cases, after cooling, the formed precipitate was collected, washed with 20% Na_2CO_3 and then with water, and finally crystallized from the suitable solvent. Characteristics of the prepared compounds are reported in table III.

Pharmacology

Representative terms of both thiazolo[3,2-*b*]triazole and thiazolo[2,3-*c*]triazole, along with some phenacylthio[1,2,4]triazole derivatives were tested for anti-inflammatory, analgesic and antipyretic activity (see tables IV and V); their acute toxicity was also verified.

Tests were performed on groups of five male adult animals, using Wistar rats (body weight 250 ± 12 g) and Swiss mice (body weight 28 ± 2 g), kept in controlled environmental and nutritional conditions. All the examined compounds were

suspended in a 0.5% aqueous carboxymethylcellulose solution and administered, orally or subcutaneously, at a fixed dose of 100 mg/kg. Acetylsalicylic acid (ASA) and phenylbutazone (PBZ) were administered at the same dose, under identical experimental conditions, as reference drugs.

Acute toxicity and behavioural effects

Analysis was carried out on both rats and mice. Compounds were administered orally and subcutaneously in different doses (100 to 1200 mg/kg *po*, and 100 to 500 mg/kg *sc*). Animals were kept under observation during the 8 days following the treatment.

Anti-inflammatory activity

Paw oedema inhibition test [42] was used on rats. Carrageenan (0.1 ml of a 1% suspension in saline) was injected into the sub-plantar tissue of the right hindpaw, 30 min after the oral administration (100 mg/kg) of the test compounds. Volume was measured using a mercury plethysmometer (U Basile, Italy) before, and 2 and 4 h after the injection of the irritant.

The increase in volume of the paw was adopted as a measure of the oedema. Swelling in treated animals was calculated as a percentage of inhibition in comparison to controls (table IV).

Analgesic activity

The phenylquinone writhing test was used on mice, according to the method of Siegmund modified by Hendershot and Forsaith [43]. 0.1 ml/10 g bw of a hydroalcoholic solution of 0.02% phenylquinone were injected *ip* 20 min after administration of the test compounds (100 mg/kg *sc*). The analgesic activity was evaluated by the number of writhes between the 5th and the 15th min following the injection of phenylquinone. Results were expressed as a percentage of protection with respect to the control group (table IV).

Antipyretic activity

The method of Bianchi *et al* [44] was used on rats. Rectal temperature was taken (Termist LSI numeric thermometer) during the 3 days before the experiments, excluding animals for which thermal oscillations were higher than 0.5°C. On the day of the experiment, after the temperature control, the animals were dosed *sc* with 2 ml/100 g bw of brewer's dried yeast suspended at 20% w/v in a mixture of 20% maize oil and 1% Tween-20, in 0.5% aqueous carboxymethylcellulose. After 4 h and 30 min, the temperature was measured again (time 0) and the test compounds administered orally. Rectal temperature changes were then registered every hour for the following 5 h.

Table V reports the results of the test as the percentage of reduction of pyresis in treated animals as compared to controls.

Microbiology

Some phenacylthiotriazole derivatives (compounds **52**, **54**, **56**, **59**, **61** and **63**) were tested *in vitro* for antimicrobial activity, against Gram-negative and Gram-positive bacteria (*E. coli* ATCC 25922 and one strain recently isolated from a clinical sample; *S. aureus* ATCC 25923 and Oxford). Antimycotic activity, against a number of species of *Candida* (*C. albicans*, *C. krusei*, *C. parapsilosi*, *C. guilliermondii*), *C. albicans* NCPF 3153, and *Torulopsis glabrata* was also assessed.

Compounds, in dimethylsulfoxide solutions, were tested within a concentration range of 6.25 to 200 µg/ml, using the method of the minimal inhibiting concentration (MIC). Isosensitest Agar (Oxoid) and Sabouraud dextrose broth were used as growth media, respectively, for bacteria and mycetes.

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