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A one-pot three-component reaction providing tricyclic 1,4-benzoxazepine derivatives

Mehdi Ghandi^{a,*}, Tayebeh Momeni^a, Mohammad Taghi Nazeri^a, Nahid Zarezadeh^a, Maciej Kubicki^b

^a School of Chemistry, College of Science, University of Tehran, PO Box 14155 6455, Tehran, Iran
^b Faculty of Chemistry, Adam Mickiewicz University in Poznan, Umultowska 89b, 61-614 Poznan, Poland

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ABSTRACT

A new one-pot, three-component reaction of aromatic aldehydes, 2-aminophenol, and dimedone for the synthesis of tricyclic 1,4-benzoxazepine derivatives in moderate to good yields is described. © 2013 Elsevier Ltd. All rights reserved.

Multicomponent reactions (MCRs) are processes in which at least three different simple substrates react in one-pot to give the target materials.¹ These reactions, which have gained significant attention during the past years, do not occur through a single-step procedure, but rather via several sequential steps involving cascades or domino reactions.² Simplicity, greater efficiency, and atom economy with generation of molecular complexity and diversity in one-pot transformations are some advantages of these reactions.³

Benzo-fused seven-membered heterocycles containing two heteroatoms at positions 1 and 4 are used in medicinal chemistry due to their wide spectrum of biological activities.⁴ Examples include 1,4-benzodioxepine, 1,4-benzodithiepine, 1,4-benzoxazepine, 1,4-benzoxathiepine, 1,4-benzothiazepine, and 1,4-benzodiazepine.⁵ Benzodiazepines and benzoxazepines are known as non-peptide vasopressin V2 receptor agonists.⁵ Benzoxazepine derivatives form an important class of compounds with various biological activities.⁶ Among compounds containing this fragment are psychotropic and neurotropic agents,⁷ a non-nucleoside HIV-1 reverse transcriptase inhibitor,⁸ calcium antagonists,⁹ antidepressants,¹⁰ and analgesics.¹¹

Herein, we report a one-pot, three-component synthesis of 1,4-benzoxazepine derivatives via the multicomponent reaction of commercially available 2-aminophenol, dimedone, and various aromatic aldehydes. To our delight, the reaction of preheated 2-aminophenol and dimedone in DCE with benzaldehyde

proceeded smoothly to completion within 24 h in the presence of trifluoroacetic acid (TFA) (20 mol %), affording the tricyclic 1,4benzoxazepine **2a** in 68% yield (Scheme 1).¹² The analytical and spectroscopic data of **2a** were in agreement with the proposed structure.¹³ For example, the ¹H NMR spectrum of **2a** contained characteristic singlets at δ 1.06, 1.09, 2.65, 6.53, and 9.21 due to the two methyl groups, CH₂CO, CHPh, and NH protons, respectively, together with an AB quartet at δ 2.18 for the CH₂C=C group. The ¹³C NMR spectrum of **2a** exhibited 21 distinct signals including one at δ 192.6 due to the C=O group.

We next examined the substrate scope by reacting 2-aminophenol and dimedone with different aldehydes in DCE (Scheme 2). As indicated in Table 1, reactions of various aldehydes **1a–l** afforded the corresponding 1,4-benzoxazepine derivatives **2a–l**. Gratifyingly, aldehydes with electron-donating or electron-withdrawing groups underwent this multicomponent sequence (Table 1).¹⁴



Scheme 1. Synthesis of 1,4-benzoxazepine 2a.

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^{*} Corresponding author. Tel.: +98 21 61112250; fax: +98 21 66495291. *E-mail address:* ghandi@khayam.ut.ac.ir (M. Ghandi).

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M. Ghandi et al./Tetrahedron Letters xxx (2013) xxx-xxx



Scheme 2. Synthesis of 1,4-benzoxazepine derivatives 2a-l.

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Figure 1. X-ray crystal structure of compound 2a.

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M. Ghandi et al./Tetrahedron Letters xxx (2013) xxx-xxx



Scheme 3. Suggested mechanism for the formation of 2a.

Unambiguous evidence for the structure of **2a** was obtained by single-crystal X-ray-diffraction analysis (Fig. 1).¹⁵

Mechanistically, the reaction presumably proceeds via the initial formation of the enamine **I**, produced from condensation of 2-aminophenol with dimedone. Subsequent trapping with aldehyde **1a** via either the intermediate imine **II**¹⁶ or oxonium ion **III**,¹⁷ and intramolecular cyclization finally afford **2a** (Scheme 3).

Inspection of the results revealed that whereas an unsubstituted aldehyde afforded **2a** in moderate yield (entry 1 Table 1), those bearing electron-withdrawing groups produced the corresponding products in higher yields (entries 2, 3, 5, 7, and 10, Table 1). On the other hand, lower yields of 1,4-benzoxazepines were obtained by introducing electron-donating groups on the aromatic aldehyde (entries 4, 6, and 8, Table 1). This behavior supports the suggested mechanism since the introduction of an electron-withdrawing group on the phenyl ring accelerates the rate of formation or consumption of either intermediates **II** or **III**.

In conclusion, a range of 1,4-benzoxazepines **2a–I** has been synthesized in moderate to high yields via the one-pot, three-component reaction of 2-aminophenol, dimedone, and aldehydes **1a–I**. These new compounds broaden the scope of MCRs and may be of potential interest in drug discovery.

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- 12. General procedure for the synthesis of 1,4-benzoxazepine **2a**. A stirred solution of 2-aminophenol (0.109 g, 1 mmol) and dimedone (0.168 g, 1.2 mmol) in DCE (5 mL) was heated at reflux for 4 h. After completion of this step as indicated by TLC, benzaldehyde (0.106 g, 1 mmol) and TFA (0.023 g, 0.2 mmol) were added and heating at reflux was continued for 24 h. After completion, aqueous NaHCO₃ (10 mL, 20%) was added and the organic phase was separated, washed with H₂O (10 mL), and dried over anhydrous Na₂SO₄. The solvent was evaporated under reduced pressure and the residue was recrystallized from 30% EtOAc in hexane.
- 13. 3,3-Dimethyl-11-phenyl-3,4,5,11-tetrahydrodibenzo[b,e][1,4]oxazepin-1(2H)-one (**2a**). White solid: (398 mg, 68%); mp: 269–271 °C; IR (KBr) v: 3344 (NH), 1739 (CO) cm⁻¹; ¹H NMR (300 MHz, DMSO-d₆) δ 1.06 (s, 3H, Me), 1.09 (s, 3H, Me), 2.18 (AB quartet, 2H, *J* = 16.0 Hz, C=CCH₂), 2.65 (s, 2H, CH₂CO), 6.53 (s, 1H, CHPh), 6.61 (d, 1H, *J* = 7.3 Hz, Ar), 6.70 (t, 1H, *J* = 7.3 Hz, Ar), 6.82 (dd, 1H, *J* = 7.5 Hz, Ar), 7.03 (d, 2H, *J* = 8.5 Hz, Ar), 7.42–7.47 (m, 3H, Ar), 7.57 (d, 1H, *J* = 6.3 Hz, Ar), 9.21 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO-d₆) δ 27.6, 28.2, 31.8, 43,7, 49.3, 78.5, 104.4, 145.8, 116.3, 118.0, 119.0, 123.4, 123.9, 128.0, 130.1, 132.2, 134.0, 138.7, 149.0, 154.9, 192.6 (CO); MS (EI) *m/z*: 319 (92, M⁺), 318 (100), 242 (30), 234 (14), 77 (23); Anal. Calcd for C₂₁H₂₁NO₂: C, 78.97; H, 6.63; N, 4.39. Found: C, 78.54; H, 6.98; N, 4.45.
- 14. 11-([1.1'-Binhenvl]-4-vl)-3.3-dimethyl-3.4.5.11-
- tetrahydrodibenzo[b,e][1,4]oxazepin-1(2H)-one (**2f**). White solid: (395 mg, 65%); mp 277–279 °C; IR (KBr) v: 3324 (NH), 1741 (CO) cm⁻¹; ¹H NMR (300 MHz, DMSO-d₆) δ 1.10 (s, 3H, Me), 1.11 (s, 3H, Me), 2.21 (AB quartet, 2H, *J* = 15.9 Hz, C=CCH₂), 2.68 (s, 2H, CH₂CO), 6.62 (s, 1H, CHPh), 6.67 (d, 1H, *J* = 7.8 Hz, Ar), 6.76–6.83 (m, 2H, Ar), 7.04 (d, 1H, *J* = 7.7 Hz, Ar), 7.18 (d, 2H, *J* = 8.1 Hz, Ar), 7.29 (t, 1H, *J* = 7.1 Hz, Ar), 7.38 (t, 2H, *J* = 7.1 Hz, Ar), 7.47 (d, 2H, *J* = 8.1 Hz, Ar), 7.55 (d, 2H, *J* = 7.29 Hz, Ar), 9.19 (s, 1H, NH); ¹³C NMR (75 MHz, DMSO-d₆) δ 27.5, 28.4, 31.9, 43.6, 49.4, 78.8, 109.4, 120.2, 122.8, 123.2, 123.9, 126.2, 126.5, 126.7, 127.2, 127.4, 128.8, 128.9, 129.0, 129.2, 134.0, 138.9, 139.1, 139.3, 146.0, 154.8, 192.6 (CO); MS (EI) *m*/*z*: 395 (81, M⁺), 394 (100), 318 (41), 242 (40), 234 (7); Anal. Calcd for C₂₇H₂₅NO₂: C, 82.00; H, 6.37; N, 3.54. Found: C, 82.27; H, 6.44: N, 3.41.
- CCDC-921136. Copies of these data can be obtained free of charge via http:// www.ccdc.cam.ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336 033; or e-mail: deposit@ccdc.cam.ac.uk).
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