

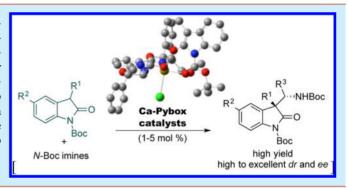
# Calcium-Catalyzed Asymmetric Synthesis of 3-Tetrasubstituted Oxindoles: Efficient Construction of Adjacent Quaternary and Tertiary Chiral Centers

Shota Shimizu, Tetsu Tsubogo, Pengyu Xu, and Shū Kobayashi\*

Department of Chemistry, School of Science, The University of Tokyo, Hongo, Bunkyo-ku, Tokyo 113-0033, Japan

Supporting Information

**ABSTRACT:** Chiral Ca-catalyzed asymmetric addition reactions of 3-substituted oxindoles with N-Boc-imines afford 3-tetrasubstituted oxindole derivatives bearing adjacent quaternary and tertiary chiral centers, which are key structures for biological activities. Ubiquitous and nontoxic Ca catalysts (1–10 mol %) work well in this reaction, and high yields (up to 99%) and selectivities (up to >99% ee) of the products with wide substrate scope have been attained. The structures of the chiral Ca catalysts and intermediary Ca enolates are also discussed.



xindole derivatives are an interesting class of compounds for several biological activities, such as anticancer, anti-HIV,<sup>2</sup> antimalarial,<sup>3</sup> antitubercular,<sup>4</sup> progesterone receptor agonist,<sup>5</sup> and MDM2 inhibitor,<sup>6</sup> etc.<sup>7</sup> For asymmetric synthesis of these compounds, construction of the quaternary chiral centers often observed in their structures is one of the most challenging tasks. While the reactions of 3-substituted oxindoles with imines potentially provide an efficient method for making adjacent quaternary and tertiary chiral centers, ocntrol of such stereogenic centers is known to be very difficult presumably because of steric congestion. Up to now, only four reports appear for this reaction, and yields, selectivities, catalyst efficiency, and/ or substrate generality are not satisfactory. 10 In 2008, Chen and co-workers reported the first example of the reaction of 3substituted oxindoles with N-Boc-imines using bifunctional thiourea-tertiary amine organocatalysts, and high enantioselectivities (up to 95% ee) were obtained for several 3-substituted oxindoles with aromatic and heteroaromatic N-Boc-imines. 10a However, a racemic compound was obtained in a reaction with an aliphatic N-Boc-imine. In 2009, Maruoka and co-workers reported phosphonium salts as chiral phase-transfer catalysts and demonstrated the reactions of 3-aryloxindoles with aromatic N-Boc-imines and good enantioselectivities (up to 88% ee) were obtained. 10b Liu, Chen, and co-workers reported modified cinchona alkaloid-catalyzed reactions of N-unprotected 3substituted oxindoles with aromatic N-Ts-imines, and good enantioselectivities (up to 89% ee) were obtained. 10c Finally, in 2013, Zhang and Peng reported the reactions of N-unprotected 3-bromooxindoles with aromatic N-Ts-imines catalyzed by bifunctional thiourea derivatives, and high enantioselectivities (up to 99% ee) were obtained. In those four reports, while high yields and high selectivities were obtained in some cases, substrates were limited to aromatic and heteroaromatic imines,

and no enantioselective reactions with aliphatic imines were reported. In addition, organocatalysts (nonmetal catalysts) are used in all of the previous examples, and relatively high loading amounts of catalysts (10 mol %) are used except for phosphonium salts (3 mol %). Here, we report the reactions of 3-substituted oxindoles with *N*-Boc-imines using chiral calcium catalysts. Not only aromatic imines but also aliphatic imines work well, and high yields and high diastereo- and enantioselectivities have been attained using 1–10 mol % loadings of the calcium catalysts with wide substrate scope.

Recently, we have been interested in calcium compounds as catalysts in organic transformations. Calcium is a ubiquitous element, nontoxic, and readily available. 11 Previously, we have investigated chiral calcium alkoxide-catalyzed asymmetric reactions, <sup>12</sup> such as asymmetric Michael reactions, <sup>12a-e,h,i</sup> asymmetric [3 + 2] cycloaddition reactions, <sup>12a,b,i</sup> and asymmetric protonation reactions. <sup>12f</sup> More recently, we have developed chiral calcium chloride <sup>12h,i</sup> and chiral calcium iodide (CaI<sub>2</sub>) <sup>13</sup> for catalytic asymmetric transformations. We chose 3-methylsubstituted oxindole 1a and N-Boc-imine 2a as models, and several reaction conditions were examined. It was found that a combination of CaI2 with Pybox L gave promising results. When Cal<sub>2</sub> and L1 were used, the reaction proceeded smoothly to afford the desired product 3aa in 85% yield with high diastereoselectivity and moderate enantioselectivity (Table 1, entry 1). Ligand screening showed that L5 gave the best enantioselectivity (entries 2-5). With the optimized ligand L5, several solvents were screened. Slightly higher enantioselectivities were observed when diethyl ether (Et<sub>2</sub>O) and dichloromethane (DCM) were employed (entries 6-8). By lowering the

Received: March 17, 2015



**Organic Letters** Letter

Table 1. Optimization of Reaction Conditions

entry	x	у	temp [°C]	Py- box	solvent	yield [%]å	3aa:4 aa <sup>b</sup>	ee [%]°
$1^{d}$	10	10	-40	L1	Tol	85	94:6	57
$2^{d}$	10	10	-40	L2	Tol	85	94:6	21
$3^{d}$	10	10	-40	L3	Tol	87	94:6	66
4 <sup>d</sup>	10	10	-40	L4	Tol	86	95:5	21
5 <sup>d</sup>	10	10	-40	L5	Tol	78	94:6	75
$6^{\rm d}$	10	10	-40	L5	Et <sub>2</sub> O	83	95:5	78
$7^{d}$	10	10	-40	L5	THF	80	95:5	31
8 <sup>d</sup>	10	10	-40	L5	DCM	83	94:6	78
$9^{d}$	10	10	-78	L5	DCM	88	94:6	93
$10^{d}$	5	5	-78	L5	DCM	85	94:6	76
$11^d$	5	7.5	-78	L5	DCM	85	93:7	99
$12^{e,f,g}$	5	7.5	-78	L5	DCM	87	95:5	98
13 <sup>e,g,h</sup>	3	4.5	-78	L5	DCM	91	95:5	97
$14^{\rm e,g,h}$	2	3	-78	L5	DCM	81	95:5	92
$15^{e_ig_ii,j}$	1	1.5	-78	L5	DCM	97	94:6	98
$16^{f,g,k}$	5	7.5	-78	L5	DCM	92	94:6	87
17 <sup>f,g,l</sup>	5	7.5	-78	L5	DCM	89	94:6	14

<sup>a</sup>Isolated yield. <sup>b</sup>Determined by <sup>1</sup>H NMR analysis of the crude reaction mixture. Determined by chiral HPLC analysis.  $^d$ Et<sub>3</sub>N was added at -78 °C.  $^e$ Et<sub>3</sub>N was added at rt.  $^f$ Slow addition of imine over 1 h; the reaction was further continued for 17 h. g1a (1.0 equiv), 2a (1.5 equiv). <sup>h</sup>Slow addition of imine over 10 h; the reaction was continued for 8 h. Slow addition of imine over 20 h; the reaction was continued for 22 h. <sup>j</sup>0.4 M. <sup>k</sup>CaCl<sub>2</sub> was used instead of CaI<sub>2</sub>. <sup>l</sup>Ca(O-i-Pr)<sub>2</sub> was used instead of CaI<sub>2</sub> without Et<sub>3</sub>N.

reaction temperature, the enantioselectivity was further improved to 93% ee (entry 9). To decrease the catalyst loading, several conditions were investigated (entries 10–15). When the ratio of CaI2 to L5 was 1:1.5, the desired product 3aa was obtained in 87% yield with 95:5 dr and 98% ee (entry 12). Further decreasing the catalyst loading showed promising results (entries 13-15). Finally, the reaction proceeded smoothly to afford the desired adduct in 97% yield with high diastereoselectivity (3aa/4aa = 94/6), and the enantioselectivity of 3aa was 97% using 1 mol % of the catalyst (entry 15). On the other hand, while the enantioselectivity decreased slightly using CaCl<sub>2</sub> instead of CaI<sub>2</sub> (entry 16), a significant drop in the enantioselectivity was observed when calcium isopropoxide  $(Ca(O-i-Pr)_2)$  was used as a calcium source (entry 17).

Next, we investigated the substrate scope of this asymmetric reaction (Table 2). When we applied tolyl imines 2b-d, high yields, high diastereoselectivities, and excellent enantioselectivities were observed in all cases (entries 2-4). In the reactions

Table 2. Substrate Scope<sup>a</sup>

		2	2	yield <sup>b</sup>	_	$ee^d$
entry	$\mathbb{R}^1$	$\mathbb{R}^2$	$\mathbb{R}^3$	(%)	3:4 <sup>c</sup>	(%)
1	Me (1a)	Н	Ph (2a)	87 ( <b>3aa</b> )	95:5	98
2	Me (1a)	Н	$o ext{-}MeC_6H_4$ (2b)	93 (3ab)	97:3	97
3 <sup>e</sup>	Me (1a)	Н	$m\text{-MeC}_6H_4$ (2c)	80 (3ac)	95:5	97
$4^f$	Me (1a)	Н	$p\text{-MeC}_6H_4$ (2d)	95 (3ad)	96:4	95
5	Me (1a)	Н	$   \begin{array}{c}     m\text{-}\mathrm{CF}_3\mathrm{C}_6\mathrm{H}_4\\ \mathbf{(2e)} \end{array} $	93 (3ae)	96:4	97
6	Me (1a)	Н	<i>p</i> -OMe C <sub>6</sub> H <sub>4</sub> (2f)	91 ( <b>3af</b> )	94:6	95
7	Me (1a)	Н	p-ClC <sub>6</sub> H <sub>4</sub> (2g)	92 (3ag)	97:3	96
8	Me (1a)	Н	2-furyl (2h)	89 (3ah)	97:3	96
9	Me (1a)	Н	2-thienyl (2i)	91 (3ai)	98:2	98
10	Me (1a)	Н	2-naphthyl (2j)	84 (3aj)	97:3	97
$11^{f,g}$	Me (1a)	Н	Et (2k)	96 (3ak)	97:3	90
$12^{f,g,h,i}$	Me (1a)	Н	$n-C_5H_{11}$ (21)	95 (3al)	93:7	72
$13^{e,h,i}$	Me (1a)	Н	<i>i</i> -Bu (2m)	99 (3am)	94:6	84
$14^e$	Bn (1b)	Н	Ph (2a)	81 (3ba)	94:6	95
15 <sup>e</sup>	Bn (1b)	Н	2-thienyl (2i)	84 (3bi)	96:4	98
16	n-Pr (1c)	Н	Ph (2a)	83 (3ca)	94:6	98
17	n-Pr (1c)	Н	Et (2k)	43 (3ck)	97:3	72
$18^{e,h,i}$	n-Pr (1c)	Н	<i>i</i> -Bu (2m)	93 (3cm)	97:3	74
19	i-Pr (1d)	Н	Ph (2a)	80 (3da)	95:5	>99
$20^{e,h,i}$	i-Pr (1d)	Н	<i>i</i> -Bu (2m)	77 (3dm)	95:5	73
21	$1e^{j}$	Н	Ph (2a)	80 (3ea)	95:5	93
$22^f$	$1e^{j}$	Н	$p\text{-MeC}_6H_4$ (2d)	82 (3ed)	94:6	95
$23^f$	$1e^{j}$	Н	p-ClC <sub>6</sub> H <sub>4</sub> (2g)	88 (3eg)	96:4	97
$24^e$	$1e^{j}$	Н	2-furyl (2h)	93 (3eh)	96:4	91
25	$\mathbf{1f}^k$	Me	Ph (2a)	93 (3fa)	94:6	95
$26^e$	$1g^l$	F	Ph (2a)	82 (3ga)	94:6	93

<sup>a</sup>2 was added over 1 h unless otherwise noted. <sup>b</sup>Isolated yield. <sup>c</sup>Determined by <sup>1</sup>H NMR analysis of the crude reaction mixture.  $^d$ Determined by chiral HPLC analysis.  $^e$ Slow addition of imine over 10 h; the reaction was further continued for 8 h. fSlow addition of imine over 5 h; the reaction was further continued for 13 h. gImine (3 equiv) was used. <sup>h</sup>10 mol % of the catalyst was used. <sup>i</sup>Pybox L3 was used. <sup>j</sup>1e:  $R^1 = CH_2CH(O(CH_2)_2O), R^2 = H.$  \*1f:  $R^1 = CH_2CH(O(CH_2)_2O),$  $R^2 = Me^{-l}1g$ :  $R^1 = CH_2CH(O(CH_2)_2O)$ ,  $R^2 = F$ .

with imines 2e, 2f, and 2g, bearing electron-withdrawing and electron-donating substituents at the benzene rings, the reactions also proceeded smoothly to afford the desired products 3ae, 3af, and 3ag in excellent yields with high diastereo- and enantioselectivities (entries 5-7). When heteroaromatic imines 2h and 2i were examined, excellent enantioselectivities were obtained (entries 8 and 9). The 2-naphthylaldehyde-derived imine 2j also gave excellent enantioselectivity (entry 10). For aliphatic imines, imine 2k derived from propionaldehyde reacted with oxindole 1a to afford the desired product 3ak in excellent yield with excellent diastereo- and enantioselectivities (entry 11).  $n-C_5H_{11}$  imine 21 and *i*-Bu imine 2m also reacted smoothly to provide the desired products 3al and 3am, respectively in excellent yields with good to high diastereo- and enantioseOrganic Letters Letter

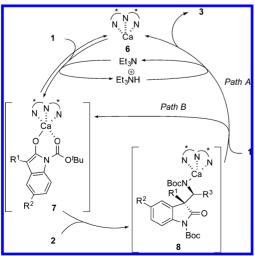
lectivities using Pybox L3 as the ligand (entries 12 and 13). It should be noted that aliphatic imines reacted smoothly to give the desired adducts in excellent yields with good to excellent diastereo- and enantioselectivies. In previous reports, as mentioned, there were no examples of enantioselective reactions using aliphatic imines. 10 The substrate scope of oxindoles 1 was also surveyed. 3-Benzyl-substituted oxindole 1b reacted with imines 2a and 2i smoothly to afford the desired adducts in high yields with excellent diastereo- and enantioselectivities (entries 14 and 15). The absolute configuration was determined by comparing the optical rotation of 3bi with that of the literature. <sup>10a</sup> 3-n-Propyl- and 3-isopropyl-substituted oxindoles 1c and 1d also reacted with imine 2a smoothly, and almost perfect enantioselectivities were obtained (entries 16 and 19). Aliphatic imines 2k and 2m also worked well (entries 17, 18, and 20). We also examined the reactions of substituted oxindoles 1e, 1f, and 1g (entries 21-26). <sup>14</sup> In all cases, the reactions proceeded smoothly to afford the desired adducts in high yields with excellent diastereo- and enantioselectivities. The adduct 3ea was readily converted to spirooxindole derivative  $\mathbf{5ea}$  in good yield without loss of any optical purity (Scheme 1). Thus, we

Scheme 1. Transformation to Spirooxindole

have succeeded in providing oxindole derivatives that have adjacent quaternary and tertiary chiral centers in satisfactory yields and stereoselectivities with wide substrate scope.

An assumed catalytic cycle of this asymmetric reaction is shown in Scheme 2. CaI<sub>2</sub> and ligand **L5** may form a 1:1 complex (6) and a 1:2 complex in solution; that is supported by NMR experiments. <sup>13,18</sup> Chiral Ca catalyst 6 may work as a monomeric form because a linear correlation between ligand ee's and product ee's was observed. <sup>18,19</sup> Ca catalyst 6 activates oxindole 1 to form Ca enolate 7, which was detected by ESI HRMS analysis. <sup>18</sup> One enantioface (the *Re* face) of 7 is effectively shielded by the chiral

Scheme 2. Assumed Catalytic Cycle<sup>a</sup>

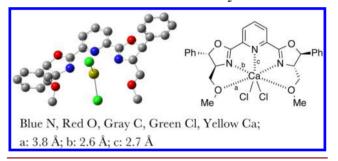


<sup>&</sup>lt;sup>a</sup>Iodide ions are omitted for clarification.

ligand (vide infra), and 7 attacks imine 2's *Re* face smoothly to give intermediate 8. There may be two possible ways to complete the catalytic cycle. In the first, the intermediate 8 is protonated by Et<sub>3</sub>NH<sup>+</sup>I<sup>-</sup> to give product 3 along with regeneration of the chiral Ca catalyst 6 (path A). In the second, the intermediate 8 deprotonates 1 to form the Ca enolate 7 (path B). Given the steric bulkiness of 1 and 8, path A may be more likely; however, path B cannot be ruled out at this moment. It is noteworthy that the Ca catalysts, which have both mild Lewis acidity and Brønsted basicity, <sup>11</sup> may play a key role in obtaining high yields and stereoselectivities and also efficient catalyst turnover.

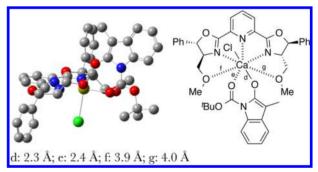
We further investigated stereochemical structures of the chiral Ca catalyst and the Ca enolate. While X-ray crystallographical analysis of  $Ca(ClO_4)_2$ -Pybox was conducted, we carried out DFT calculation of  $CaCl_2$ -L5 (Scheme 3). The optimized

Scheme 3. Stereochemical Model of CaCl<sub>2</sub>-L5



structure has  $C_2$ -symmetry, and the three nitrogen atoms of L5 coordinate to Ca. On the other hand, the length between Ca and O of the methoxy group of L5 is 3.8 Å, suggesting weak coordination of O to Ca. We also investigated the structure of the Ca enolate (Scheme 4). The DFT calculation showed that the

Scheme 4. Stereochemical Model of Ca Enolate



three nitrogen atoms and the carbonyl oxygen of the Boc group coordinated to Ca. The length between Ca and O of the methoxy groups of L5 is 3.9 and 4.0 Å, also suggesting weak interactions between O and Ca. The stereochemical model of the Ca enolate indicates that the plane of chiral ligand L5 and the plane of the enolate are almost perpendicular and that the Re face of the enolate is efficiently shielded by the CH<sub>2</sub>OMe arm of L5.

In conclusion, we have developed a new catalytic method for the synthesis of optically active 3-tetrasubstituted oxindoles using chiral Ca catalysts. Adjacent quaternary and tertiary chiral centers have been created efficiently by the Ca-catalyzed reactions of 3substituted oxindoles with imines. High diastereo- and enantioselectivities have been realized with wide substrate scope. This is the first example of a chiral metal-catalyzed reaction of a 3-substituted oxindole with an imine. The Organic Letters Letter

ubiquitous and nontoxic Ca catalyst worked well, and even 1 mol % of the catalyst gave a high yield and high selectivities of the product. This reaction may provide an efficient approach to the synthesis of biologically important oxindoles, including spirooxindole derivatives. Further studies on this topic are underway in our laboratories.

# ASSOCIATED CONTENT

# S Supporting Information

Experimental section and copies of NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

### AUTHOR INFORMATION

### **Corresponding Author**

\*E-mail: shu kobayashi@chem.s.u-tokyo.ac.jp.

### **Notes**

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

This work was partially supported by a Grant-in-Aid for Science Research from the Japan Society for the Promotion of Science (JSPS), Global COE Program, The University of Tokyo, MEXT, Japan, Japan Science Technology Agency (JST). We thank Mr. K. Takasugi (The University of Tokyo and JEOL Ltd.) for helpful discussions about NMR experiments.

### REFERENCES

- (1) Yu, B.; Yu, D.-Q.; Liu, H.-M. Eur. J. Med. Chem. 2014, DOI: 10.1016/j.ejmech.2014.06.056.
- (2) Kumari, G.; Nutan; Modi, M.; Gupta, S. K.; Singh, R. K. Eur. J. Med. Chem. 2011, 46, 1181.
- (3) (a) Rottmann, M.; McNamara, C.; Yeung, B. K. S.; Lee, M. C. S.; Zhou, B.; Russell, B.; Seitz, P.; Plouffe, D. M.; Dharia, N. V.; Tan, J.; Cohen, S. B.; Spencer, K. R.; Gonzalez-Paez, G. E.; Lakshminarayana, S. B.; Goh, A.; Suwanarusk, R.; Jegla, T.; Schmitt, E. K.; Beck, H.-P.; Brun, R.; Nosten, F.; Renia, L.; Dartois, V.; Keller, T. H.; Fidock, D. A.; Winzeler, E. A.; Diagana, T. T. Science 2010, 329, 1175. (b) Yeung, B. K. S.; Zou, B.; Rottmann, M.; Lakshminarayana, S. B.; Ang, S. H.; Leong, S. Y.; Tan, J.; Wong, J.; Keller-Maerki, S.; Fischli, C.; Goh, A.; Schmitt, E. K.; Krastel, P.; Francotte, E.; Kuhen, K.; Plouffe, D.; Henson, K.; Wagner, T.; Winzeler, E. A.; Petersen, F.; Brun, R.; Dartois, V.; Diagana, T. T.; Keller, T. H. J. Med. Chem. 2010, 53, 5155.
- (4) Vintonyak, V. V.; Warburg, K.; Kruse, H.; Grimme, S.; Hubel, K.; Rauth, D.; Waldmann, H. Angew. Chem., Int. Ed. 2010, 49, 5902.
- (5) Fensome, A.; Adams, W. R.; Adams, A. L.; Berrodin, T. J.; Cohen, J.; Huselton, C.; Illenberger, A.; Karen, J. C.; Hudak, M. A.; Marella, A. G.; Melenski, E. G.; McComas, C. C.; Mugford, C. A.; Slayeden, O. D.; Yudt, M.; Zhang, J.; Zhang, P.; Zhu, Y.; Winneker, R. C.; Wrobel, J. E. J. Med. Chem. 2008, 51, 1861.
- (6) (a) Ding, K.; Lu, Y.; Nikolovska-Coleska, Z.; Qui, S.; Ding, Y.; Gao, W.; Stuckey, J.; Krajewski, K.; Roller, P. P.; Tomita, Y.; Parrish, D. A.; Deschamps, J. R.; Wang, S. *J. Am. Chem. Soc.* 2005, 127, 10130. (b) Ding, K.; Lu, Y.; Nikolovska-Coleska, Z.; Wang, G.; Qiu, S.; Shangary, S.; Gao, W.; Qin, D.; Stukey, J.; Krajewski, K.; Roller, P. P.; Wang, S. *J. Med. Chem.* 2006, 49, 3432.
- (7) (a) Macpherson, L. J.; Dubin, A. E.; Evans, M. J.; Marr, F.; Schultz, P. G.; Cravatt, B. F.; Patapoutian, A. *Nature* **2007**, *445*, 541. (b) Volk, B.; Barkóczy, J.; Hegedus, E.; Udvari, S.; Gacsályi, I.; Mezei, T.; Pallagi, K.; Kompagne, H.; Lévay, G.; Egyed, A.; Hársing, L. G., Jr.; Spedding, M.; Simig, G. *J. Med. Chem.* **2008**, *51*, 2522.
- (8) For reviews of the catalytic enantioselective construction of quaternary chiral centers, see: (a) Fuji, K. Chem. Rev. 1993, 93, 2037. (b) Corey, E. J.; Guzman-Perez, A. Angew. Chem., Int. Ed. 1998, 37, 388.
- (c) Christoffers, J.; Mann, A. Angew. Chem., Int. Ed. 2001, 40, 4591.
- (d) Christoffers, J.; Baro, A. Angew. Chem., Int. Ed. 2003, 42, 1688.

- (e) Douglas, C. J.; Overman, L. E. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 5363. (f) Ramon, D. J.; Yus, M. Curr. Org. Chem. 2004, 8, 149. (g) Peterson, E. A.; Overman, L. E. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 11943. (h) Christoffers, J.; Baro, A. Adv. Synth. Catal. 2005, 347, 1473. (i) Trost, B. M.; Jiang, C. Synthesis 2006, 369. (j) Bella, M.; Gasperi, T. Synthesis 2009, 1583. (k) Hawner, C.; Alexakis, A. Chem. Commun. 2010, 46, 7295. (l) Das, J. P.; Marek, I. Chem. Commun. 2011, 47, 4593. (m) Vesely, J.; Ramon, R. Chem. Soc. Rev. 2014, 43, 611.
- (9) For reviews of syntheses of oxindole alkaloids, see: (a) Marti, C.; Carreira, E. M. Eur. J. Org. Chem. 2003, 2209. (b) Galliford, C. V.; Scheidt, K. A. Angew. Chem., Int. Ed. 2007, 46, 8748. (c) Trost, B. M.; Brennan, M. K. Synthesis 2009, 3003. (d) Singh, G. S.; Desta, Z. Y. Chem. Rev. 2012, 112, 6104.
- (10) (a) Tian, X.; Jiang, K.; Peng, J.; Du, W.; Chen, Y. C. Org. Lett. 2008, 10, 3583. (b) He, R.; Ding, C.; Maruoka, K. Angew. Chem., Int. Ed. 2009, 48, 4559. (c) Cheng, L.; Liu, L.; Jia, H.; Wang, D.; Chen, Y. J. J. Org. Chem. 2009, 74, 4650. (d) Li, J.; Zhang, D. G.; Peng, Y. Chem. Commun. 2013, 49, 1330.
- (11) (a) Harder, S. Chem. Rev. 2010, 110, 3852. (b) Yamashita, Y.; Kobayashi, S. Acc. Chem. Res. 2011, 44, 58. (c) Yamashita, Y.; Tsubogo, T.; Kobayashi, S. Chem. Sci. 2012, 3, 967. (d) Tsubogo, T.; Yamashita, Y.; Kobayashi, S. Top. Organomet. Chem. 2013, 45, 243.
- (12) (a) Saito, S.; Tsubogo, T.; Kobayashi, S. J. Am. Chem. Soc. 2007, 129, 5364. (b) Tsubogo, T.; Saito, S.; Seki, K.; Yamashita, Y.; Kobayashi, S. J. Am. Chem. Soc. 2008, 130, 13321. (c) Kobayashi, S.; Tsubogo, T.; Saito, S.; Yamashita, Y. Org. Lett. 2008, 10, 807. (d) Tsubogo, T.; Yamashita, Y.; Kobayashi, S. Angew. Chem., Int. Ed. 2009, 48, 9117. (e) Tsubogo, T.; Kano, Y.; Ikemoto, K.; Yamashita, Y.; Kobayashi, S. Tetrahedron: Asymmetry 2010, 21, 1221. (f) Poisson, T.; Yamashita, Y.; Kobayashi, S. J. Am. Chem. Soc. 2010, 132, 7890. (g) Poisson, T.; Tsubogo, T.; Yamashita, Y.; Kobayashi, S. J. Org. Chem. 2010, 75, 963. Our current contribution of chiral calcium salt-catalyzed asymmetric reactions: (h) Tsubogo, T.; Yamashita, Y.; Kobayashi, S. Chem.—Eur. J. 2012, 18, 13624. (i) Hut'ka, M.; Tsubogo, T.; Kobayashi, S. Adv. Synth. Catal. 2013, 355, 1561.
- (13) Tsubogo, T.; Shimizu, S.; Kobayashi, S. Chem.—Asian. J. 2013, 8, 872.
- (14) We synthesized acetal-protected oxindoles **1e**–**g** by a modified method. See: Hamashima, Y.; Suzuki, T.; Takano, H.; Shimura, Y.; Sodeoka, M. *J. Am. Chem. Soc.* **2005**, *127*, 10164. The detailed procedure is shown in the Supporting Information.
- (15) Reported methods to construct spirooxindole cores: (a) Hong, S.; Jung, M.; Park, Y.; Ha, M. W.; Park, C.; Lee, M.; Park, H. *Chem. —Eur. J.* **2013**, *19*, 9599. (b) Goff, R. L.; Lawson, A. M.; Daïch, A.; Comesse, S. *Org. Biomol. Chem.* **2013**, *11*, 1818 See also ref 3a,b.
- (16) Berrée, F.; Chang, K.; Cobas, A.; Rapoport, H. J. Org. Chem. 1996, 61, 715.
- (17) Racemization of elacomine and horsfiline via retro-Mannich reactions were suggested; (a) Pellegrin, C.; Weber, M.; Borschberg, H. J. Helv. Chim. Acta 1996, 79, 151. (b) Trost, B. M.; Brennen, M. K. Org. Lett. 2006, 8, 2027 In our case, however, the racemization was suppressed..
- (18) See the Supporting Information.
- (19) Girard, V.; Kagan, H. B. Angew. Chem., Int. Ed. 1998, 37, 2922.
- (20) Yamamura, M.; Miyake, J.; Imamura, Y.; Nabeshima, T. Chem. Commun. 2011, 47, 6801.
- (21) Because of the difficulty of calculation, we used CaCl<sub>2</sub> instead of CaI<sub>2</sub>. CaCl<sub>2</sub> also showed a high yield and high selectivities (see Table 1, entry 16).