Tetrahedron Letters 55 (2014) 6500-6503

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Total synthesis of plagiochin G and derivatives as potential cancer chemopreventive agents



^a Department of Natural Products Chemistry, Key Lab of Chemical Biology of Ministry of Education, School of Pharmaceutical Science, Shandong University, Jinan 250012, PR China ^b Natural Products Research Laboratories, UNC Eshelman School of Pharmacy, University of North Carolina, Chapel Hill, NC 27599, United States

^c Department of Biochemistry, Kyoto Prefectural University of Medicine, Kamigyo-ku, Kyoto 602-0841, Japan

^d Chinese Medicine Research and Development Center, China Medical University and Hospital, Taichung, Taiwan

ARTICLE INFO

Article history: Received 11 June 2014 Revised 6 October 2014 Accepted 7 October 2014 Available online 13 October 2014

Keywords: Bisbibenzyls Plagiochin G Intramolecular S_NAr reaction Cancer chemopreventive agents Epstein–Barr virus early antigen (EBV-EA)

ABSTRACT

A new and efficient total synthesis has been developed to obtain plagiochin G (**22**), a macrocyclic bisbibenzyl, and four derivatives. The key 16-membered ring containing biphenyl ether and biaryl units was closed via an intramolecular S_NAr reaction. All synthesized macrocyclic bisbibenzyls inhibited Epstein–Barr virus early antigen (EBV-EA) activation induced by the tumor promoter 12-0-tetradecanoyl-phorbol-13-acetate (TPA) in Raji cells and, thus, are potential cancer chemopreventive agents.

© 2014 Elsevier Ltd. All rights reserved.

Cancer, the second leading cause of death in humans, is a group of illnesses resulting from abnormal growth of cells in the body. Many cancer therapies, especially various anticancer agents, have been developed since the beginning of the last century. However, several problems, such as adverse side effects and drug resistance, have also encouraged scientists to explore strategies to prevent premalignant cells from completing the process of carcinogenesis. This concept known as 'cancer chemoprevention' has been developed over the past few decades,¹ and the Epstein–Barr virus early antigen (EBV-EA) activation assay has been established to quickly evaluate chemopreventive activity in vitro.^{2–4}

Macrocyclic bisbibenzyls are phenolic natural products that occur mainly in liverworts and exhibit remarkable biological activities, such as 5-lipoxygenase, cyclooxygenase, and calmodulin inhibitory effects, as well as antifungal, anti-HIV, antimicrobial, and cytotoxic activities.⁵⁻⁷ These compounds are divided into four distinct structural types (I–IV, Fig. 1),⁸ each containing four aromatic rings (labeled A–D) and two ethylene bridges, and originate biosynthetically from bibenzyl lunularin or its precursor lunularin acid.^{7,9} The plagiochin family of type **II** macrocyclic bisbibenzyls

includes natural plagiochins A–D isolated from the liverwort *Plagiochila acanthophylla* by Hashimoto et al.¹⁰ and plagiochins E–H synthesized by Speicher et al.¹¹

The unusual structures and intriguing biological activities of macrocyclic bisbibenzyls have made them attractive synthetic targets. In 1992, Keseru et al.¹² synthesized plagiochins C and D by using a Wurtz-type coupling at position **a** to close the 16-membered ring (Scheme 1). In 1999, Fukuyama et al.^{13,14} used an intramolecular Still-Kelly reaction at position **c** to accomplish the macrocyclization in the syntheses of plagiochins A and D. 'Plagiochin E', initially reported as a natural product from the liverwort *Marchantia polymorpha*,¹⁵ was totally synthesized in 2009 by Speicher et al., who revised the structure of the isolated 'plagiochin E' to that of riccardin D.^{16,17} Subsequently, in 2010, Speicher et al. reported the syntheses of plagiochins E-H by employing an intramolecular McMurry reaction at position **a** as the key macrocyclization step (Scheme 1).¹¹ In 2011, Cortes Morales et al.¹⁸ used an intramolecular S_NAr reaction at position **d** to form the 16-membered macrocyclic ring of plagiochin D (Scheme 1). Recently, Jiang et al. prepared plagiochin E using an intramolecular McMurry reaction at position **a** for the macrocyclization (Scheme 1).¹⁹

In the course of our ongoing efforts to find bioactive macrocyclic bisbibenzyls, we developed a new route to synthesize plagiochin G and several ester derivatives. We initially synthesized two







^{*} Corresponding authors. Tel.: +86 531 88382012; fax: +86 531 88382019 (H.-X. L.); tel.: +1 (919) 962 0066; fax: +1 (919) 966 3893 (K.-H.L.).

E-mail addresses: louhongxiang@sdu.edu.cn (H.-X. Lou), khlee@unc.edu (K.-H. Lee).



Figure 1. Four distinct structural types of macrocyclic bisbibenzyls.



Scheme 1. Retrosynthetic analysis of plagiochin G.

lunularin precursors (**10** and **16**), then formed the aryl—aryl bond between rings *B* and *D* by using a Pd-catalyzed Suzuki–Miyaura coupling reaction, and finally applied an intramolecular S_NAr reaction²⁰ at position **d** to achieve the key 16-membered ring closure (Scheme 1 and Figure 2). Furthermore, we found that all macrocyclic bisbibenzyls produced by this new route exhibited potential cancer chemopreventive activity. Herein, we report the synthetic details for producing plagiochin G (**22**) and its derivatives (**19–21** and **22a–22d**) as well as the evaluation of their cancer chemopreventive activity.

Chemistry

The total synthesis of plagiochin G was achieved in 12.5% overall yield in 14 steps as shown in Scheme 2. Both rings *B* and *D* were produced from the commercially available 2-hydroxy-5-methoxybenzaldehyde (1). The phenol moiety of **1** was protected by reaction with benzyl bromide. Then, the aldehyde moiety was



Figure 2. Construction unit system for the synthesis of plagiochin G.

reduced to a benzyl alcohol with NaBH₄, and the resulting compound ($\mathbf{3}$) was treated with PPh₃·HBr to give the phosphonium salt $\mathbf{4}$.

Ring *C* was developed from commercially available 3-hydroxy-4-methoxybenzaldehyde (**5**). Protection of the phenol moiety with chloromethyl methyl ether yielded compound **6**.²¹ Units **4** and **6** were coupled with a Wittig reaction in the presence of K_2CO_3 and 18-crown-6 to give the *D*–*C* segment **7** (obtained as an *E*/*Z* mixture in a ratio 1:1),²² which was hydrogenated over Pd/C to give bibenzyl **8**. The deprotected hydroxy group in **8** was then converted to the corresponding triflate in **9**, which underwent a PdCl₂(dppf) mediated conversion to the pinacolboronate ester **10**, produced in 62% overall yield from **5**.²³

The synthesis of the second bibenzyl sub-unit (**16**) began with commercially available 4-fluorobenzaldehyde (**11**), corresponding to ring A. Compound **11** was nitrated to yield compound **12**, which was linked with phosphonium salt **4** by using a Wittig reaction to give *A*–*B* segment **13** (obtained as an *E*/*Z* mixture in a ratio 7:1).²² The double bond of each isomer was reduced with Wilkinson's catalyst to afford a single compound **14**; the benzyl ether was retained under the mild hydrogenation conditions.²⁴ Subsequently, O-debenzylation was accomplished using concentrated HCl in HOAc to yield compound **15**,¹⁸ and the free hydroxyl group was then converted to a triflate in **16**.²³

The triflate **16** and the pinacolboronate ester **10** were combined via an aryl-aryl bond between rings *B* and *D* by using the Suzuki–Miyaura coupling reaction to afford **17**, followed by deprotection of the phenol methoxymethyl ether using *p*-toluene-sulfonic acid. Macrocyclization to give **19** was achieved in 89% yield through an intramolecular S_NAr reaction using K_2CO_3 in DMF at room temperature. Next, the nitro group was removed in a two-step sequence of reduction and deamination¹⁸ to give plagiochin trimethyl ether (**21**). Plagiochin G (**22**) was finally obtained after cleavage of the methyl ethers.^{11,26}

The acetyl ester derivative **22a** was prepared by reaction of **22** with acetyl chloride and Et_3N in CH_2Cl_2 with DMAP as a catalyst. Derivatives **22b–22d** were synthesized from **22** by esterification with 3-(1*H*-imidazol-1-yl)propanoic acid, 3-(1*H*-1,2,4-triazol-1-yl)propanoic acid, and 3-(1*H*-tetrazol-1-yl)propanoic acid, respectively (Scheme 3).

Biological evaluation

To evaluate the cancer chemoprevention effects of compounds (**19–22** and **22a–22d**) in vitro, we assayed the eight compounds for inhibition of EBV-EA activation.²⁷ Glycyrrhetic acid was used as a positive control. In this assay, all tested compounds showed inhibitory effects toward EBV-EA activation without cytotoxicity to Raji cells. As shown in Table 1, plagiochin G (**22**) exhibited the highest potency with 88%, 45%, and 19% inhibition at 1×10^3 , 5×10^2 , 1×10^2 mol ratio/TPA, respectively, and IC₅₀ value of 481 µM, with highly preserved viability of Raji cells. The four ester



Scheme 2. Synthesis of plagiochin G. Reagents and conditions: (a) BnBr, K₂CO₃, acetone, reflux; (b) NaBH₄, MeOH, 0 °C, 20 min, then rt, 1 h; (c) PPh₃-HBr, MeCN, reflux, 1 h; (d) chloromethyl methyl ether, diisopropylethylamine, CH₂Cl₂, 0 °C, then rt, 6 h; (e) K₂CO₃, 18-crown-6, CH₂Cl₂, reflux, 8 h; (f) Pd/C (10%), 1 bar H₂, MeOH, rt, 24 h; (g) Et₃N, CH₂Cl₂, 0 °C, trifluoromethanesulfonic anhydride, then rt, 0.5 h; (h) PdCl₂(dppf), Et₃N, pinacolborane, dioxane, reflux; (i) H₂SO₄, HNO₃, -5 °C, then rt 1 h; (j) Wilkinson's catalyst, H₂, THF-*t*-BuOH (1:1), rt, 24 h; (k) conc. HCl, HOAc, 70 °C, 6 h; (l) EtOH, Na₂CO₃ (2 M), Pd(PPh₃)₄, toluene, reflux, 18 h; (m) *p*-toluenesulfonic acid, MeOH, 40 °C, 4 h; (n) K₂CO₃, DMF, rt, 24 h; (o) Pd/C (10%), 1 bar H₂, THF, rt, 4 h; (p) NaNO₃ (1.2 M in H₂O), NaHSO₃ (1.0 M in H₂O), EtOH–HOAc (5:4), rt, 3 h; (q) BBr₃ (1 M in CH₂Cl₂), CH₂Cl₂, -78 °C, then rt, 1 h.



Scheme 3. Syntheses of ester derivatives of plagiochin G. Reagents and conditions: (22a) acetyl chloride, Et₃N, DMAP, CH₂Cl₂, rt, 1 h; (22b/22c/22d) 3-(1*H*-imidazol-1-yl)propanoic acid/3-(1*H*-1,2,4-triazol-1-yl)propanoic acid/3-(1*H*-tetrazol-1-yl)propanoic acid, EDCI, DMAP, CH₂Cl₂, rt 4 h.

Relative ratio" of EBV-EA activation with respect to positive control (100%) in the presence of 22 and related compounds					
Compound	Compound concentration (mol ratio/TPA ^b)				IC ₅₀ ^c (μΝ
	1000	500	100	10	
19	$14.9 \pm 0.5 (70)^{d}$	58.2 ± 0.7	83.1 ± 2.4	100 ± 0.5	491
20	15.3 ± 0.4 (70)	59.6 ± 0.6	84.6 ± 2.3	100 ± 0.4	500
21	13.0 ± 0.5 (70)	56.8 ± 0.5	81.1 ± 2.5	100 ± 0.5	490
22	11.5 ± 0.6 (70)	54.3 ± 0.6	80.1 ± 2.3	100 ± 0.5	481
22a	13.9 ± 0.5 (70)	57.9 ± 0.5	82.4 ± 2.5	100 ± 0.6	495
22b	13.0 ± 0.4 (60)	55.4 ± 1.5	79.1 ± 2.3	100 ± 0.5	479
22c	13.8 ± 0.5 (60)	56.0 ± 1.6	80.0 ± 2.5	100 ± 0.3	482
22d	14.0 ± 0.5 (60)	57.6 ± 1.4	81.6 ± 2.4	100 ± 0.5	488
Glycyrrhetic acid ^e	7.4 ± 0.5 (60)	35.7 ± 0.8	83.2 ± 2.0	100 ± 0.3	413

^a Values represent percentages relative to the positive control value (100%).

^b TPA concentration is 20 ng/mL (32 pmol/mL).

The molar ratio of compound, relative to TPA, required to inhibit 50% of the positive control activated with 32 pmol TPA.

Values in parentheses are viability percentages of Raji cells. In all other experiments, viability was >80%.

Positive control.

Table 1

derivatives (22a-22d) showed similar inhibitory effects, while the three synthetic precursors (19-21) of 22 were comparably or slightly less potent.

Conclusions

In this study, a new and efficient total synthesis of **22** and four ester derivatives (22a-22d) was successfully accomplished in 12-16 steps. An intramolecular S_NAr reaction was used for the formation of the 16-membered ring. All tested synthetic macrocyclic bisbibenzyls exhibited potential cancer chemopreventive activity as evaluated by an EBV-EA activation assay. To the best of our knowledge, this is the first report of macrocyclic bisbibenzyls with cancer chemopreventive activity. The new synthetic route reported herein is an additional effective strategy to construct variously substituted macrocyclic bisbibenzyls and should greatly facilitate our further synthesis and SAR study of cancer-preventative derivatives of 22.

Acknowledgments

This work was financially supported by the National Natural Science Foundation of China (No. 81172956) and by NIH grant CA177584 from the National Cancer Institute awarded to K.H. Lee.

Supplementary data

Supplementary data (experimental section and spectroscopic data) associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2014.10.038.

References and notes

- 1. Gravitz, L. Nature 2011, 471, S5.
- 2. Ito, Y.; Yanase, S.; Fujita, J.; Harayama, T.; Takashima, M.; Imanaka, H. Cancer Lett. 1981, 13, 29.
- 3. Itoigawa, M.; Ito, C.; Tokuda, H.; Enjo, F.; Nishino, H.; Furukawa, H. Cancer Lett. 2004. 214. 165.
- Suzuki, M.; Nakagawa-Goto, K.; Nakamura, S.; Tokuda, H.; Morris-Natschke, S. L.; Kozuka, M.; Nishino, H.; Lee, K.-H. Pharm. Biol. 2006, 44, 178.
- 5 Asakawa, Y. Chemical Constituents of the Hepaticae; Springer, 1982.
- 6. Asakawa, Y.; Heidelberger, M.; Herz, W.; Zechmeister, L. Progress in the Chemistry of Organic Natural Products: Fortschritte der Chemie Organischer Naturstoffe; Springer-Verlag, 1982.

- 7. Asakawa, Y. Chemical Constituents of the Bryophytes: Springer, 1995.
- 8. Harrowven, D. C.; Kostiuk, S. L. Nat. Prod. Rep. 2012, 29. 223.
- Asakawa, Y.: Matsuda, R. Phytochemistry 1982, 21, 2143. Q
- Hashimoto, T.; Tori, M.; Asakawa, Y.; Fukazawa, Y. Tetrahedron Lett. 1987, 28, 10. 6295
- 11. Speicher, A.; Groh, M.; Hennrich, M.; Huynh, A.-M. Eur. J. Org. Chem. 2010, 2010, 6760
- Keseru, G. M.; Mezey-Vandor, G.; Nogradi, M.; Vermes, B.; Kajtar-Peredy, M. 12 Tetrahedron 1992, 48, 913.
- Fukuyama, Y.; Yaso, H.; Mori, T.; Takahashi, H.; Minami, H.; Kodama, M. 13. Heterocycles 2001, 54, 259.
- Fukuyama, Y.; Yaso, H.; Nakamura, K.; Kodama, M. Tetrahedron Lett. 1999, 40, 14. 105.
- 15. Niu, C.; Qu, J.-B.; Lou, H.-X. Chem. Biodivers. 2006, 3, 34.
- Speicher, A.; Groh, M.; Zapp, J.; Schaumlöffel, A.; Knauer, M.; Bringmann, G. 16. Synlett 2009, 1852.
- 17 Toyota, M.; Nagashima, F.; Asakawa, Y. Phytochemistry 1988, 27, 2603.
- Cortes Morales, J. C.; Guillen Torres, A.; Gonzalez-Zamora, E. Eur. J. Org. Chem. 18. 2011, 3165.
- Jiang, J.; Sun, B.; Wang, Y. Y.; Cui, M.; Zhang, L.; Cui, C. Z.; Wang, Y. F.; Liu, X. G.; 19. Lou, H. X. Bioorg. Med. Chem. 2012, 20, 2382.
- 20 Zhu, J. Synlett 1997, 133.
- 21. Oh, J. H.; Kim, S. J.; Kim, J. H.; Shin, D. H. US20100143843A1, 2010.
- 22 Boden, R. M. Synthesis 1975, 1975, 784.
- 23. Thompson, A. L.; Kabalka, G. W.; Akula, M. R.; Huffman, J. W. Synthesis 2005, 547.
- 24 Jourdant, A.; González-Zamora, E.; Zhu, J. J. Org. Chem. 2002, 67, 3163.
- Konoshima, T.; Konishi, T.; Takasaki, M.; Yamazoe, K.; Tokuda, H. Biol. Pharm. 25. Bull. 2001, 24, 1440.
- 26. Schaumlöffel, A.; Groh, M.; Knauer, M.; Speicher, A.; Bringmann, G. Eur. J. Org. Chem. 2012, 2012, 6878.
- 27. Inhibition of EBV-EA activation assay. Inhibition of EBV-EA activation was assayed using Raji cells (virus nonproducer type), an EBV genome-carrying human lymphoblastoid cell, which were cultivated in 10% fetal bovine serum (FBS) RPMI 1640 medium. The indicator cells (Raji, 1×10^6 /mL) were incubated at 37 °C for 48 h in 1 mL of medium containing n-butyric acid (4 mM as trigger), TPA (32 pM = 20 ng in 2 µL of DMSO as inducer), and various amounts of the test compounds dissolved in 5 µL of DMSO (ca. 0.7% DMSO). Smears were made from the cell suspension. The EBV-EA inducing cells were stained with high titer EBV-EA positive serum from NPC patients and detected by an indirect immunofluorescence technique.²⁵ In each assay, at least 500 cells were counted, and the number of stained cells (positive cells) was recorded. Triplicate assays were performed for each data point. The average EBV-EA induction of the test compound was expressed as IC₅₀ value and as a relative ratio to the positive control experiment (100%), which was carried out with n-butyric acid (4 mM) plus TPA (32 pM). In the experiments, the EBV-EA induction was normally around 35%, and this value was taken as the positive control (100%). n-Butyric acid (4 mM) alone induced 0.1% EA-positive cells. The viability of treated Raji cells was assayed by the trypan blue staining method. The cell viability of the TPA positive control was greater than 80%. Therefore, only the compounds that induced less than 80% (% of control) of the EBVactivated cells (those with a cell viability of more than 60%) were considered able to inhibit the activation caused by promoter substances. Student's t-test was used for all statistical analyses.