



Enantiomerically Pure Guanidine-Catalysed Asymmetric Nitroaldol Reaction

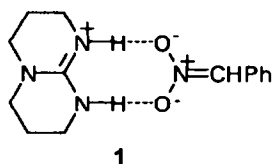
Rafael Chinchilla, Carmen Nájera* and Pablo Sánchez-Agulló

Departamento de Química Orgánica, Facultad de Ciencias, Universidad de Alicante, Apdo. 99, 03080 Alicante, Spain

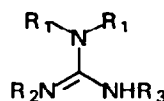
Abstract: Enantiomerically pure guanidines with and without C₂ symmetry have been prepared and used as catalysts in the addition of nitromethane to aldehydes (Henry reaction). The influence of the structure and amount of the guanidine, the solvent and the temperature in the enantioselectivity of the nitroaldol reaction has also been studied.

INTRODUCTION

The Henry (nitroaldol) reaction is one of the oldest carbon-carbon bond formation reactions in organic synthesis¹. However, only recently the first efficient² asymmetric nitroaldol reaction catalysed by rare earth metal complexes has been reported³. Tetramethylguanidine (TMG) has been used as basic catalyst in Michael additions of nitroalkanes and in the Henry reaction of nitromethane with levoglucosenone⁴. Guanidines have been used for molecular recognition of carboxylate⁵, phosphate⁶ and nitrate⁷ anions because of its ability to form strong hydrogen bonds between the guanidinium cation and the dioxoanion⁸. Moreover, the isolation of complex 12,9 formed between the bicyclic guanidine 1,5,7-triazabicyclo[4.4.0]dec-5-ene (TBD) and α -nitrotoluene suggested that these type of intermediates could be good models for an enantioselective guanidine-catalysed nitroaldol reaction^{9,10}. All these antecedents prompted us to study the use of chiral guanidines of type 2, derived from enantiomerically pure amines, as catalyst in the asymmetric Henry reaction.



1

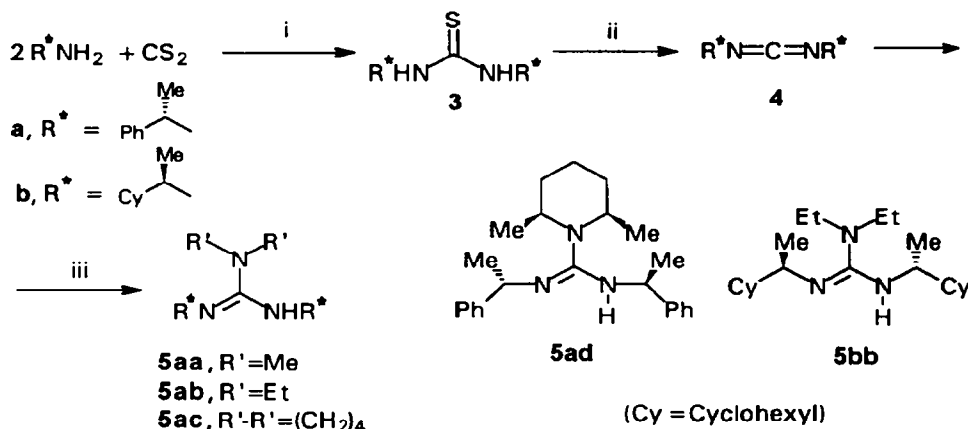


2

RESULTS AND DISCUSSION

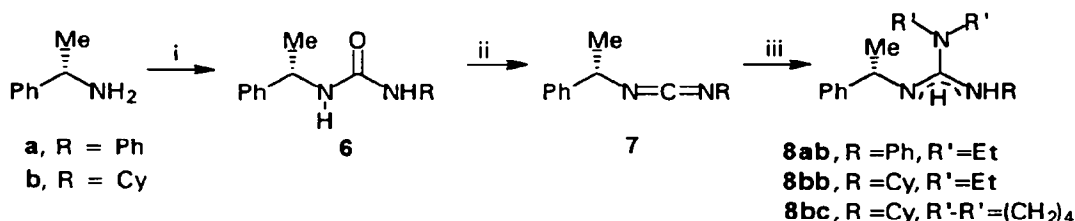
Guanidines **5** with C₂ symmetry were prepared starting from thioureas **3a** or **3b**, which were obtained by reaction of (*S*)-1-phenylethylamine or (*R*)-1-cyclohexylethylamine with carbon disulfide, respectively. Further desulfidryzation with diisopropyl azodicarboxylate (DIAD) and triphenylphosphine¹¹ afforded homochiral

carbodiimides **4a** and **4b** which were transformed into guanidines **5aa-be** by treatment with different secondary amines (Scheme 1).



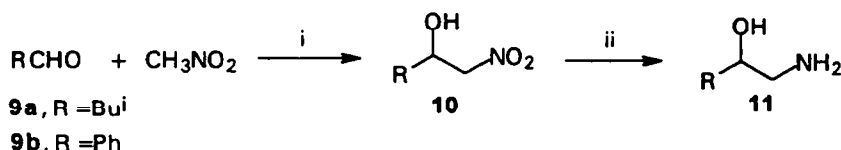
Scheme 1. Reagents and conditions: i, EtOH, reflux, 24 h; ii, DIAD, Ph_3P , THF, rt, 20 h; iii, Me_2NH , Et_2NH , pyrrolidine or *cis*-2,6-dimethylpiperidine, THF, rt, 24 h.

Homochiral guanidines **8** were obtained according to the synthetic sequence showed in Scheme 2. Reaction of (*S*)-1-phenylethylamine with phenyl or cyclohexyl isocyanate yielded the corresponding ureas **7**, which were dehydrated by treatment with bromotriphenylphosphonium bromide and triethylamine¹². After reaction with diethylamine or pyrrolidine, guanidines **8ab-bc** were obtained. Guanidines **5** and **8** were purified by extractive work-up followed by vacuum distillation.



Scheme 2. Reagents and conditions: i, PhNCO or CyNCO , AcOEt , rt, 5 min; ii, $\text{Br}_2\text{-PPh}_3$, NEt_3 , PhH , reflux, 90 min; iii, Et_2NH or pyrrolidine, THF, rt, 24 h.

The reaction of isopentanal (**9a**) or benzaldehyde (**9b**) with nitromethane in THF as solvent at room temperature in the presence of a catalytic amount of different chiral guanidines **5** or **8** (0.1 equiv) gave 1-nitro-4-methyl-2-pentanol (**10a**) or 2-nitro-1-phenyl-1-ethanol (**10b**) enantioselectively enriched (Scheme 3 and Table 1).



Scheme 3. *Reagents and conditions:* i, Guanidine **5** or **8**, THF; ii, H₂, 10% Pd-C, AcOEt, 24 h.

Table 1. Nitroaldol Reaction Catalysed by Homochiral Guanidines

entry	RCHO	guanidine	reaction time (h)	T(°C)	nitroalcohol			
					no.	yield(%) ^a	[α] _D ²⁵ ^b	e.e.(%)
1	9a	5aa	24	rt	10a	72	- 3.3	17 ^c
2	9a	5ab	10	rt	10a	85	-5.8	26 ^c
3	9a	5ab	7	-45	10a	40	-7.5	34 ^d
4	9a	5ab	9	-65	10a	33	-11.9	54 ^d
5	9a	5ac	16	rt	10a	77	-1.4	6 ^c
6	9a	5ad	72	rt	10a	8	-	-
7	9a	5bb	15	rt	10a	75	-3.9	18 ^c
8	9a	8ab	24	rt	10a	6	-	-
9	9a	8bb	24	rt	10a	60	-1.2	5 ^c
10	9a	8bc	24	rt	10a	62	0	0
11	9b	5ab	9	rt	10b	31	+4.9	13 ^c
12	9b	5ab	9	-65	10b	31	+11.1	33 ^c
13	9b	5ac	24	rt	10b	27	+1.7	5 ^c

^a Isolated pure compound, based on starting aldehyde. ^b Ethanol. ^c Deduced from [α] value. ^d Deduced from 1-amino-4-methyl-2-pentanol (**10a**) (see text). ^e Deduced from lit.¹³ [α] value for (*S*)-2-amino-1-phenyl-1-ethanol (**11b**) [α]_D¹⁸ +44.6 (EtOH).

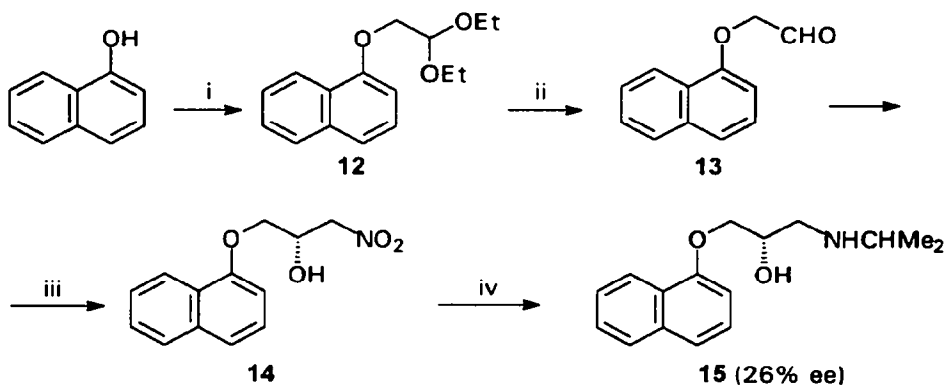
In general, the nitroaldol reaction took place with good chemical yields specially with isopentanal (**9a**). In the case of guanidines **5ad** and **8aa** (Table 1, entries 6 and 8) chemical yields were low, probably due to their lower basicity and steric hindrance respectively. The measured [α]_D value for nitroalcohols **10a** and **10b** showed that the best chiral discrimination was achieved with guanidines **5** with C₂ symmetry, specially when guanidine **5ab** was used.

The determination of the optical yield of nitroalcohols **10** could not be carried out by esterification with Mosher acid-DCC¹⁴ or with its chloride, because they suffer dehydration to the corresponding nitroalkenes. In order to determine the corresponding e.e.'s they were quantitatively transformed into aminoalcohols **11** by catalytic hydrogenation with palladium on carbon in ethyl acetate at room temperature (see Scheme 3). In the

case of 2-amino-1-phenyl-1-ethanol (**11b**) optical yields were deduced from the $[\alpha]_D$ values in comparison with the literature¹³ (see Table 1, footnote e). For 1-amino-4-methyl-2-pentanol (**11a**) the corresponding e.e.'s were deduced from ¹H NMR integral of the methine protons bonded to the OH group in the diastereomeric salts¹⁵ formed between the aminoalcohol and the (*R*)-enantiomer of the *O*-(4-chloro-2-methylphenyl)lactic acid¹⁷ in CDCl₃.

The influence of the amount of guanidine, the solvent and the temperature in the enantioselectivity of the Henry reaction was studied for the addition of nitromethane to isopentanal (**9a**) in the presence of guanidine **5ab**. When the amount of guanidine was reduced to 0.05 equiv the e.e. decreased to 22% and when 0.2 equiv were used the e.e. also decreased to 19%. The homochiral guanidine could be recuperated in all cases by extractive work-up and without any loss of activity. The reaction was carried out in different solvents such as: ethyl ether, hexane, toluene, dimethoxyethane, DMSO, nitromethane, and dioxane. In all cases chemical yields were rather similar (75-95%) except in dioxane where the reaction failed, whereas optical yields were in the range 18-29% for ethereal solvents and lower than 4% for the others. When the reaction was performed at low temperatures e.e.'s increased till 54% for compound **10a** and till 33% for compound **10b** (see Table 1, entries 3, 4 and 12).

This asymmetric reaction was applied to the synthesis of enantiomerically enriched propanolol **15**²⁰ according with the synthetic sequence outlined in Scheme 4. Reaction of the sodium alkoxide of 1-naphtol with 2-bromoacetaldehyde diethyl acetal afforded acetal **12** which were hydrolyzed in acidic medium to aldehyde **13**. Further guanidine **5ab**-catalysed nitroaldol reaction of nitromethane with aldehyde **13** at low temperature yielded enantiomerically enriched nitroalcohol **14**. *R*-Propanolol was then obtained by catalytic hydrogenation of **14** in the presence of acetone (26 % e.e.).



Scheme 4. Reagents and conditions: i, BrCH₂CH(OC₂H₅)₂, NaOH, DMSO, 90° C, 72 h, 63%; ii, 2N HCl, THF, rt, 24 h, 68%; iii, CH₃NO₂, **5ab** (0.1 eq), THF, -40° C, 8 h, 21%; iv, a) H₂, 10% Pd-C, MeOH, rt, 18 h; b) CH₃COCH₃, H₂, 10% Pd-C, MeOH, 50° C, 72 h, 30%.

EXPERIMENTAL

General.— Melting points were obtained with a Reichert ThermoVar apparatus and are uncorrected. Optical rotations were measured with an Optical Activity AA-100. IR spectra were obtained on a Pye Unicam SP3-200 spectrophotometer. NMR spectra were determined on a Bruker AC-300 (300 MHz for ^1H and 75 MHz for ^{13}C) using CDCl_3 as solvent and TMS as internal standard unless otherwise stated; chemical shifts are given in δ (ppm) and the coupling constants (J) are measured in Hz. Mass spectra (EI) were obtained on a Hewlett-Packard EM/GC HP-5988A spectrometer. Microanalyses were performed by the Microanalyses Service of the University of Alicante. High resolution mass spectra (EI) were determined by the corresponding Service of the University of Zaragoza.

Synthesis of Thioureas 3.—A solution of carbon disulfide (0.7 mL, 1.66 mmol) and the corresponding amine [(*S*)-1-phenylethylamine or (*R*)-1-cyclohexylethylamine] (3.32 mmol) in ethanol (6 mL) was refluxed during 20 h. The obtained precipitate was filtered off, washed with hexane and ether, and allowed to dry in air affording pure thioureas **3a** and **3b**.

(*S,S*)-*N,N'*-Bis(1-phenylethyl)thiourea (**3a**): yield 79%, mp 202–203°C; $[\alpha]_{\text{D}}^{25} +105.4$ (*c* 1.3; CHCl_3); ν_{max} (KBr) 3200 (NH), 1525, 1320 (C=S), 1065, 740 and 680 cm^{-1} ; δ_{H} 1.45 (d, $J=6.8$, 6H, 2x CH_3), 5.07 (br s, 2H, 2xCH), 6.24 (br s, 2H, 2xNH) and 7.01–7.21 (m, 10H, ArH); δ_{C} 22.96, 53.97, 125.56, 127.43, 128.78, 142.17 and 179.87; m/z 284 (M^+ , 47%), 120 (100), 105 (40), 103 (12), 79 (15), 77 (22) and 42 (14). Anal calcd for $\text{C}_{17}\text{H}_{20}\text{N}_2\text{S}$: C, 71.79; H, 7.09; N, 9.85; S, 11.27. Found: C, 71.84; H, 7.33; N, 9.93; S, 10.85.

(*R,R*)-*N,N'*-Bis(1-cyclohexylethyl)thiourea (**3b**): yield 63%; mp 182–183°C; $[\alpha]_{\text{D}}^{25} -24.8$ (*c* 2; CHCl_3); ν_{max} (KBr) 3180 (NH), 1540, 1430, 1325 (C=S), 1275 and 1135 cm^{-1} ; δ_{H} 0.88–1.82 (m, 22H, Cy), 1.08 (d, $J=6.7$, 6H, 2x CH_3), 3.93 (br s, 2H, 2xCH) and 5.72 (br s, 2H, 2xNH); δ_{C} 17.39, 26.00, 26.08, 26.22, 28.70, 29.32, 43.06, 54.46 and 179.91; m/z 296 (M^+ , 26%), 295 (15), 187 (100), 77 (61), 69 (39), 67 (16), 55 (54), 44 (44) and 41 (34). Anal calcd for $\text{C}_{17}\text{H}_{32}\text{N}_2\text{S}$: C, 68.86; H, 10.88; N, 9.45; S, 10.81. Found: C, 68.86; H, 11.50; N, 9.51; S, 9.83.

Synthesis of Carbodiimides 4.—A solution of triphenylphosphine (1.65 g, 6.27 mmol) in THF (5 mL) was added to a mixture of thiourea **3a** or **3b** (5.70 mmol) and diisopropyl azodicarboxylate (1.24 mL, 6.27 mmol) in THF (15 mL). The solution was stirred at room temperature for 15 h. The solvent was evaporated at reduced pressure (15 Torr) and the residue was extracted with hexane (3x20 mL). Filtration and evaporation of the hexane solution, and Kügelrohr distillation of the residue afforded carbodiimides **4a** or **4b**.

(*S,S*)-Bis(1-phenylethyl)carbodiimide (**4a**): yield 60%; bp 190°C/0.1 Torr, $[\alpha]_{\text{D}}^{23} +1.8$ (*c* 1.6; CH_2Cl_2); ν_{max} (film) 3010, 2060 (NCN), 1570, 1530, 740 and 670 cm^{-1} ; δ_{H} 1.46 (d, $J=6.8$, 6H, 2x CH_3), 4.55 (q, $J=6.8$, 2H, 2xCHN) and 7.21–7.34 (m, 10H, ArH); δ_{C} 24.52, 56.52, 125.83, 127.19, 128.35, 140.23 and 143.60; m/z 250 (M^+ , 4%), 105 (100), 103 (11), 79 (14) and (77 (22). HRMS m/z calcd for $\text{C}_{17}\text{H}_{18}\text{N}_2$: 250.1470. Found: 250.1472.

(*R,R*)-Bis(1-cyclohexylethyl)carbodiimide (**4b**): yield 42%; bp 165°C/0.1 Torr; $[\alpha]_{\text{D}}^{23} +79.7$ (*c* 1.9; CH_2Cl_2); ν_{max} (film) 2880, 2080 (NCN), 1430, 1240 and 1080 cm^{-1} ; δ_{H} 0.95–1.83 (m, 28H, Cy and 2x CH_3) and 3.21 (q, $J=6.5$, 2H, 2xCH); δ_{C} 21.84, 26.14, 28.83, 29.29, 44.36, 58.26 and 139.70; m/z 247 (M^+ -Me, 3%), 179 (17), 111 (64), 69 (100), 67 (14), 55 (71), 54 (11), 43 (11) and 41 (66). HRMS m/z calcd for $\text{C}_{16}\text{H}_{27}\text{N}_2$ (M^+ -Me): 247.2174. Found: 247.2166.

Synthesis of Ureas 6.—A solution of phenyl or cyclohexyl isocyanate [(2.17 mL, 20 mmol) or (2.55 mL, 20 mmol)] and (*S*)-1-phenylethylamine (2.83 mL, 22 mmol) in ethyl acetate (20 mL) was stirred for 5 min at

room temperature. The precipitate was filtered off, washed with hexane and ether and allowed to dry in air affording pure ureas **6a** or **6b**.

(S)-N-Phenyl-N'-(1-phenylethyl)urea (**6a**): yield 85%; mp 152-153°C; $[\alpha]_{\text{D}}^{25}$ -25.2 (*c* 2.2; CHCl₃); ν_{max} (KBr) 3290 (NH), 3000, 1620 (C=O), 1580, 1535, 1220, 740, 720 and 680 cm⁻¹; δ_{H} 1.23 (d, *J*=6.9, 3H, CH₃), 4.82 (m, 1H, CH), 6.21 (d, *J*=7.5, 1H, (CH₃CHNH)), 6.93 (m, 1H, ArH), 7.10-7.22 (m, 9H, ArH) and 7.83 (s, 1H, PhNH); δ_{C} 22.96, 49.61, 119.66, 122.65, 125.67, 126.90, 128.47, 128.84, 139.05, 144.31 and 155.96; *m/z* 240 (*M*⁺, 17%), 105 (33), 104 (6), 103 (7), 94 (7), 93(100), 92 (6), 79 (10), 77 (23), 65 (9) and 51 (8). Anal calcd for C₁₅H₁₆N₂O: C, 74.97; H, 6.71; N, 11.66. Found: C, 74.24; H, 6.74; N, 11.73.

(S)-N-Cyclohexyl-N'-(1-phenylethyl)urea (**6b**): yield 83%; mp 144-145°C; $[\alpha]_{\text{D}}^{25}$ -19.1 (*c* 2.2; CHCl₃); ν_{max} (KBr) 3300 (NH), 3040, 3010, 1615 (C=O), 1565, 1240, 740 and 690 cm⁻¹; δ_{H} 0.97-1.79 (m, 10H, Cy), 1.36 (d, *J*=6.9, 3H, CH₃), 3.48 (br s, 1H, NCHCH₂), 4.79 (m, 1H, CHN), 5.04 and 5.48 (2 br s, 2H, 2xNH) and 7.18-7.34 (m, 5H, ArH); δ_{C} 23.33, 24.73, 25.51, 33.63, 33.66, 48.52, 49.69, 125.82, 126.85, 128.45, 144.69 and 157.44; *m/z* 246 (*M*⁺, 68%), 120 (68), 106 (100), 105 (89), 104 (25), 98 (46), 79 (39), 77 (43), 56 (76), 55 (26), 43 (23), 42 (21) and 41 (30). Anal calcd for C₁₅H₂₂N₂O: C, 73.13; H, 9.00; N, 11.37. Found: C, 73.00; H, 9.47; N, 11.30.

Synthesis of Carbodiimides 7. - To a solution of bromotriphenylphosphonium bromide (6.4 g, 5.2 mmol) prepared *in situ* mixing triphenylphosphine (3.99 g, 15.2 mmol) and bromine (0.78 mL, 15.2 mmol) in benzene (20 mL) was added triethylamine (4.3 mL, 30.8 mmol) at room temperature. To this suspension was added the corresponding urea **6** (12.2 mmol) in small portions for 15 min, and the resulting mixture was refluxed for 90 min and then cooled to 10°C. The solution was filtered off and the solvent was evaporated at reduced pressure (15 Torr). The residue was extracted with hexane (3x20 mL) and the combined organic layers were filtered and evaporated yielding the corresponding carbodiimides which were purified by vacuum distillation.

(S)-N-Phenyl-N'-(1-phenylethyl)carbodiimide (**7a**): yield 54%; bp 185°C/0.1 Torr; $[\alpha]_{\text{D}}^{23}$ +28.1 (*c* 1.6; CH₂Cl₂); ν_{max} (film) 3000, 2940, 2080 (NCN), 1570, 1240, 720 and 680 cm⁻¹; δ_{H} 1.64 (d, *J*=6.7, 3H, CH₃), 4.79 (q, *J*=6.7, 1H, CH) and 7.01-7.27 (m, 10H, ArH); δ_{C} 25.19, 57.40, 123.49, 124.75, 125.73, 127.58, 128.63, 129.28, 129.45, 138.37, 140.15 and 143.05; *m/z* 222 (*M*⁺, 3%), 118 (50), 106 (11), 105 (100), 103 (18), 79 (18) and 77 (35). HRMS *m/z* calcd for C₁₅H₁₄N₂: 222.1157. Found: 222.1165.

(S)-N-Cyclohexyl-N'-(1-phenylethyl)carbodiimide (**7b**): yield 55%; bp 180°C/0.1 Torr; $[\alpha]_{\text{D}}^{23}$ -8.7 (*c* 1.7; CH₂Cl₂); ν_{max} (film) 3050, 3020, 2100 (NCN), 1670, 1550, 1440 y 695 cm⁻¹; δ_{H} 1.12-1.82 (m, 10H, Cy), 1.53 (d, *J*=6.7, 3H, CH₃), 3.16 (m, 1H, NCHCH₂), 4.58 (q, *J*=6.7, 1H, CHN) and 7.20-7.68 (m, 5H, ArH); δ_{C} 24.35, 24.56, 25.22, 34.61, 55.54, 56.46, 125.80, 127.07, 128.26, 139.79 and 143.81; *m/z* 228 (*M*⁺, 2%), 106 (13), 105 (199), 104 (15), 103 (12), 79 (15), 77 (39), 55 (28), 51 (12) and 41 (35). HRMS *m/z* calcd for C₁₅H₂₀N₂: 228.1626. Found: 228.1630.

Synthesis of Guanidines 5 and 8. General Procedure. - A solution of the corresponding carbodiimide (1 mmol) and amine (10 mmol) in THF (10 mL) was stirred at room temperature for 24 h. The solvent was evaporated at reduced pressure (15 Torr) and the remaining amine was removed under vacuum (0.1 Torr). The obtained crude guanidine was dissolved in ether (20 mL) and extracted with 2N HCl (3x10 mL). The combined aqueous layers were neutralised with NaOH and extracted with ether (3x15 mL) and the organic layers were dried over anhydrous sodium sulfate. Evaporation at reduced pressure (15 Torr) yielded guanidines **5aa-bb** and **8ab-bc** which were purified by Kügelrohr distillation.

(S,S)-N,N-Dimethyl-N',N''-bis(1-phenylethyl)guanidine (**5aa**): yield 60%; bp 160°C/0.1 Torr; $[\alpha]_{\text{D}}^{23}$ +52.1 (*c* 1; EtOH); ν_{max} (film) 3350, 3010, 2980, 2880, 1600, 735 and 675 cm⁻¹; δ_{H} (*d*₆-DMSO, 90°C) 1.36 (d, *J*=6.7,

6H, 2xCH₃CH), 2.62 (s, 6H, 2xCH₃N), 4.50 (q, *J*=6.7, 2H, 2xCHN) and 7.11-7.26 (m, 10H, ArH); δ_{C} (*d*₆-DMSO, 90°C) 24.21, 39.05, 53.37, 125.43, 125.49, 127.33, 146.42 and 154.37; *m/z* 295 (*M*⁺, 10%), 175 (20), 120 (51), 106 (14), 105 (100), 104 (14), 103 (16), 79 (25), 78 (13), 77 (31), 71 (44), 44 (29) and 42 (18). HRMS *m/z* calcd for C₁₉H₂₅N₃: 295.2048. Found: 295.2042.

(*S,S*)-*N,N*-Diethyl-*N'*,*N'*-bis(*l*-phenylethyl)guanidine (**5ab**): yield 62%; bp 170°C /0.1 Torr; [α]_D²⁵ +50.3 (c 1; EtOH); ν_{max} (film) 3380 (NH), 3040, 3000, 1610, 1440, 1080, 750 y 690 cm⁻¹; δ_{H} (*d*₆-DMSO, 90°C) 0.93 (t, *J*=7.0, 6H, 2xCH₃CH₂), 1.35 (d, *J*=7.0, 6H, 2xCH₃CH), 3.01 (q, *J*=7.0, 4H, 2xCH₂N), 4.57 (q, *J*=7.0, 1H, CHN), and 7.11-7.26 (m, 10H, ArH); δ_{C} (*d*₆-DMSO, 90°C) 12.16, 24.16, 42.44, 53.37, 125.38, 125.49, 127.29, 146.65 and 152.72; *m/z* 323 (*M*⁺, 10%), 218 (19), 203 (17), 120 (39), 106 (11), 104 (23), 103 (22), 99(17), 79 (30), 77 (29), 72 (28) and 43 (12); HRMS *m/z* calcd for C₂₁H₂₉N₃: 323.2361. Found: 323.2360.

(*S,S*)-*N,N'*-Bis(*l*-phenylethyl)-*N''*,*N''*-tetramethyleneguanidine (**5ac**): yield 56%; bp 170°C/0.1 Torr; [α]_D²⁹ +15.8 (c 1; EtOH); ν_{max} (film) 3380 (NH), 3030, 3000, 1600, 1440, 1380, 750 and 690 cm⁻¹; δ_{H} 1.45 (d, *J*=6.6, 6H, 2xCH₃), 1.74 (m, 4H, 2xCH₂CH₂N), 3.24 (br s, 4H, 2xCH₂N), 4.40 (d, *J*=6.6, 2H, 2xCHN) and 7.14-7.24 (m, 11H, ArH and NH); δ_{C} 25.14, 25.50, 48.11, 55.00, 125.87, 126.39, 128.22, 146.00 and 154.00; *m/z* 321 (*M*⁺, 2%), 217 (16), 202 (25), 120 (31), 107 (21), 105 (54), 103 (16), 79 (47), 78 (15), 77 (51), 70 (100), 55 (24), 51 (19) and 43 (28); HRMS *m/z* calcd for C₂₁H₂₇N₃: 321.2205. Found: 321.2207.

(*S,S*)-*cis*-*N,N'*-Bis(*l*-phenylethyl)-2,6-dimethylpiperidylamidine (**5ad**): yield 45%; bp 190°C/0.1 Torr; [α]_D²³ +13.7 (c 1; EtOH); ν_{max} (film) 3220, 3010, 2980, 1620, 740 and 680 cm⁻¹; δ_{H} 0.43 (d, *J*=5.6, 3H, CH₃), 1.05 (d, *J*=5.9, 3H, CH₃), 1.23-1.66 (m, 6H, CH₂), 1.29 (d, *J*=6.7, 3H, CH₃), 1.50 (d, *J*=5.5, 3H, CH₃), 2.50 (m, 2H, 2xCH₂CHN), 5.05 (m, 1H, PhCHN), 5.22 (m, 1H, PhCHN) and 7.12-7.39 (m, 10H, ArH); δ_{C} 20.56, 21.00, 22.00, 24.62, 26.88, 34.54, 50.22, 54.56, 55.16, 55.70, 126.49, 127.45, 128.13, 145.73, 149.17 and 150.00; *m/z* 363 (*M*⁺, 4%), 120 (23), 114 (15), 112 (199), 105 (88), 98 (22), 79 (27), 77 (27), 55 (28), 43 (19), 42 (25) and 41 (18). HRMS *m/z* calcd for C₂₄H₃₃N₃: 363.2674. Found: 363.2670.

(*R,R*)-*N,N'*-Bis(*l*-cyclohexylethyl)-*N''*,*N''*-diethylguanidine (**5bb**): yield 30%; bp 150°C/0.1 Torr; [α]_D²⁹ -119.3 (c 1; EtOH); ν_{max} (film) 2960, 2910, 1620, 1440 and 1365 cm⁻¹; δ_{H} (*d*₆-DMSO, 90°C) 0.94-1.45 (m, 22H, Cy), 1.04 (t, *J*=7.3, 6H, 2xCH₃CH₂), 1.69 (m, 6H, 2xCH₃CH) and 2.90-3.25 (m, 6H, 2xCH₂CH₃ and 2xCH₃CHN); δ_{C} (*d*₆-DMSO, 90°C) 12.09, 25.38, 25.45, 25.64, 28.31, 28.73, 42.52, 43.57 and 154.78; *m/z* 335 (*M*⁺, 3%), 127 (61), 99 (23), 83 (23), 72 (43), 71 (22), 69 (42), 55 (100), 43 (22) and 41 (54); HRMS *m/z* calcd for C₂₁H₄₁N₃: 335.3300. Found: 335.3310.

(*S*)-*N,N*-Diethyl-*N'*-phenyl-*N''*-(*l*-phenylethyl)guanidine (**8ab**): yield 30%; bp 155°C/0.1 Torr; [α]_D²⁵ +116 (c 3; EtOH); ν_{max} (film) 3340 (NH), 3020, 3000, 1600, 1565, 1470, 735, 710 and 680 cm⁻¹; δ_{H} 1.09 (t, *J*=7.1, 6H, 2xCH₃CH₂), 1.33 (d, *J*=6.8, 3H, CH₃CH), 3.15-3.33 (m, 4H, 2xCH₂N), 4.47 (q, *J*=6.8, 1H, CHN), 6.64 (d, *J*=8.2, 1H, ArH), 6.88 (m, 1H, ArH), 7.13-7.32 (m, 5H, ArH), 7.51 (m, 2H, ArH) and 7.66 (m, 1H, ArH); δ_{C} 12.96, 22.63, 42.75, 53.53, 121.21, 128.34, 128.38, 128.50, 128.89, 131.91, 144.03, 150.10 and 154.08; *m/z* 295 (*M*⁺, 14%), 175 (22), 120 (35), 119 (32), 105 (100), 99 (25), 93 (29), 79 (31), 77 (80), 72 (63), 51 (28) and 44 (41); HRMS *m/z* calcd for C₁₉H₂₅N₃: 295.2048. Found: 295.2040.

(*S*)-*N*-Cyclohexyl-*N'*,*N'*-diethyl-*N''*-(*l*-phenylethyl)guanidine (**8bb**): yield 78%; bp 160°C/0.1 Torr; [α]_D²⁵ -3.6 (c 1; EtOH); ν_{max} (film) 3040, 3010, 1610, 1440, 750 and 695 cm⁻¹; δ_{H} (*d*₆-DMSO, 90°C) 0.99 (t, *J*=6.7, 6H, 2xCH₃CH₂), 1.07-1.21 (m, 6H, Cy), 1.30 (d, *J*=6.1, 3H, CH₃CH), 1.50-1.80 (m, 5H, Cy) 3.06 (q, *J*=6.7, 4H, 2xCH₃CH₂), 3.15 (br s, 1H, NH), 4.50 (q, *J*=6.1, 1H, CH₃CH) and 7.13-7.36 (m, 5H, ArH); δ_{C} (*d*₆-DMSO, 90°C) 12.23, 24.33, 24.91, 33.33, 33.47, 42.25, 52.47, 54.09, 125.32, 125.64, 127.35, 147.24 and 152.98; *m/z* 301 (*M*⁺, 7%), 105 (100), 99 (30), 79 (26), 77 (27), 72 (77), 55 (42), 44 (27), 43 (65) and 41 (39); HRMS *m/z* calcd for C₁₉H₃₁N₃: 301.2518. Found: 301.2515.

(*S*)-*N*-Cyclohexyl-*N*',*N*'-tetramethylene-*N*"-(1-phenylethyl)guanidine (**8bc**): yield 83%; bp 165°C/0.1 Torr; $[\alpha]_D^{29} +10.5$ (c 1; EtOH); ν_{\max} (film) 3070, 1645, 1600, 1440, 750 and 695 cm^{-1} ; δ_{H} (d_6 -DMSO, 90°C) 1.00-1.20 (m, 6H, Cy), 1.30 (d, $J=6.7$, 3H, CH_3), 1.55 (m, 5H, Cy), 1.74 (m, 4H, $2\times\text{CH}_2\text{CH}_2\text{N}$), 3.22 (m, 4H, $2\times\text{CH}_2\text{N}$), 3.30 (br s, 1H, NH), 4.46 (q, $J=6.7$, 1H, CH_3CH) and 7.10-7.36 (m, 5H, ArH) δ_{C} (d_6 -DMSO, 90°C) 24.17, 24.31, 24.34, 33.60, 33.75, 47.29, 52.48, 54.12, 125.25, 125.66, 127.36, 147.73 and 152.26; m/z 299 (M^+ , 5%), 105 (83), 98 (33), 91 (26), 79 (26), 77 (33), 70 (100), 55 (89), 43 (47), 42 (26) and 41 (63); HRMS m/z calcd for $\text{C}_{19}\text{H}_{29}\text{N}_3$: 299.2361. Found: 299.2357.

Enantioselective Nitroaldol Reaction. Synthesis of Nitroalcohols 10. General Procedure.— A solution of nitromethane (81 μL , 1.5 mmol), the corresponding aldehyde (1 mmol) and the guanidine **5** or **8** (0.1 mmol) in THF (3 mL) was stirred during the time and at the temperature shown in Table 1. The solvent was removed at reduced pressure (15 Torr) and the resulting residue was dissolved in ether (20 mL) and washed with 2N HCl (2 \times 5 mL) and water (5 mL). The organic layer was dried (Na_2SO_4) and evaporated (15 Torr and 0.1 Torr in order to remove the remaining aldehyde), affording nitroalcohol **10**. The guanidine could be recuperated after neutralisation of the aqueous layer with NaOH and further extraction with ether.

1-Nitro-4-methyl-2-pentanol (**10a**). ν_{\max} (film) 3380 (OH), 1530 (NO_2), 1450, 1350, 1190 and 1130 cm^{-1} ; δ_{H} 0.95 (d, $J=4.4$, 3H, CH_3), 0.97 (d, $J=4.4$, 6H, $2\times\text{CH}_3$), 1.24 [m, 1H, $(\text{CH}_3)_2\text{CH}$], 1.50 (m, 1H, CHHCHOH), 1.81 (m, 1H, CHHCHOH), 2.78 (br s, 1H, OH) and 4.31-4.40 (m, 3H, CHOH and CH_2NO_2); δ_{C} 21.64, 23.04, 24.21, 42.39, 66.95 and 80.99; m/z 147 (M^+ , 2%), 97 (33), 83 (20), 81 (20), 69 (26), 67 (21), 55 (100), 43 (60) and 41 (87).

2-Nitro-1-phenyl-1-ethanol (**10b**). ν_{\max} (film) 3440 (OH), 3030, 3000, 1530 (NO_2), 1360, 1250, 1050, 720 and 680 cm^{-1} ; δ_{H} 3.39 (br s, 1H, OH), 4.38-4.55 (m, 2H, CH_2NO_2), 5.35 (dd, $J=3.3$ and 9.4, 1H, CHOH) and 7.32 (s, 5H, ArH); δ_{C} 70.78, 81.00, 125.82, 128.70, 128.79 and 138.10; m/z 167 (M^+ , 3%), 120 (100), 107 (42), 105 (60), 103 (52), 91 (36), 79 (51), 78 (43), 77 (97), 51 (43) and 43 (30).

Catalytic Hydrogenation of Nitroalcohols 10 to Aminoalcohols 11. General Procedure.— To a solution of the corresponding nitroalcohol **10** (2 mmol) in ethyl acetate (25 mL) was added 10% palladium on carbon (100 mg) and the resulting suspension was stirred under hydrogen atmosphere at normal pressure for 24 h. The mixture was filtered through celite and the solvent was evaporated (15 Torr) yielding quantitatively the aminoalcohols **11**.

1-Amino-4-methyl-2-pentanol (**11a**). ν_{\max} (film) 3260 (OH and NH), 1635, 1350, 1130 and 1070 cm^{-1} ; δ_{H} 0.92 (d, $J=4.6$, 3H, CH_3), 0.94 (d, $J=4.6$, 3H, CH_3), 1.16 [m, 1H, $(\text{CH}_3)_2\text{CH}$], 1.37 (m, 1H, CHHCHOH), 1.78 (m, 1H, CHHCHOH), 2.20 (br s, 3H, OH and NH_2), 2.50 (dd, $J=8.3$ and 12.7 Hz, 1H, CHHNH_2), 2.80 (dd, $J=3.2$ and 12.7 Hz, 1H, CHHNH_2) and 3.60 (m, 1H, CHOH); δ_{C} 21.95, 23.22, 24.30, 43.81, 47.84 and 69.81; m/z 117 (M^+ , 1%), 74 (23), 73 (100), 60 (27), 59 (47), 45 (26), 43 (80) and 41 (47).

2-Amino-1-phenyl-1-ethanol (**11b**). ν_{\max} (film) 3140 (OH and NH), 1465, 1350, 1225, 960 and 920 cm^{-1} ; δ_{H} 2.50 (br s, 3H, OH and NH_2), 2.76 (dd, $J=7.7$ and 12.9, 1H, CHHNH_2), 2.87 (dd, $J=4.1$ and 12.9, 1H, CHHNH_2), 4.58 (dd, $J=4.1$ and 7.7, 1H, CHOH) and 7.22-7.35 (m, 5H, ArH); δ_{C} 49.19, 74.22, 125.81, 127.37, 128.28 and 142.69; m/z 137 (M^+ , 2%), 107 (35), 105 (13), 91 (18), 79 (92), 78 (21), 77 (100), 51 (30) and 50 (16).

Synthesis of (1-naphthyloxy)acetaldehyde diethyl acetal (12**).** To a solution of 1-naphtol (0.97 g, 6.7 mmol) and sodium hydroxide (1.05 g, 26.2 mmol) in dimethylsulfoxide (20 mL) was added α -bromoacetaldehyde diethyl acetal (1.01 mL, 6.7 mmol). The resulting mixture was stirred at 90°C for 72 h

and the solvent was removed at reduced pressure (15 Torr). Water (15 mL) was added and the solution was extracted with ether (3x15 mL). The organic extracts were washed with water (2x10 mL), dried (Na_2SO_4) and evaporated (15 Torr) affording acetal **12** (1.1 g, 63%); ν_{max} (film) 3020, 1560, 1255, 1225, 1120, 1090, 1060, 780 and 760 cm^{-1} ; δ_{H} 1.23 (t, $J=7.1$, 6H, $2\times\text{CH}_3$), 3.57–3.80 (m, 4H, $2\times\text{OCH}_2\text{CH}_3$), 4.13 (d, $J=5.3$, 2H, ArOCH_2), 4.92 [t, $J=5.3$, 1H, $\text{CH}(\text{OC}_2\text{H}_5)$], 6.75 (d, $J=7.7$, 1H, ArH), 7.26–7.46 (m, 4H, ArH), 7.73 (m, 1H, ArH) and 8.29 (m, 1H, ArH); δ_{C} 15.23, 62.50, 68.74, 100.51, 104.77, 120.41, 121.88, 125.00, 125.53, 125.65, 126.21, 127.27, 134.37 and 154.18; m/z 260 (M^+ , 25%), 169 (21), 127 (25), 115 (64), 103 (100), 75 (59) and 47 (37).

Synthesis of (1-naphthylthio)acetaldehyde (13). To a solution of (1-naphthylthio)acetaldehyde diethyl acetal (**12**) (870 mg, 3.4 mmol) in THF (10 mL) was added 2N HCl (10 mL) and the mixture was stirred at room temperature for 24 h. Ether (25 mL) and water (25 mL) was added and the ethereal layer was separated. The aqueous layer was extracted with ether (3x10 mL) and all the combined organic layers were washed with water (2x10 mL) and dried over Na_2SO_4 . Evaporation of the solvent (15 Torr) yielded pure aldehyde **13** (425 mg, 68%); ν_{max} (film) 3020, 1715 (CO), 1560, 1250, 1225, 1090, 780 and 755 cm^{-1} ; δ_{H} 4.58 (s, 2H, ArOCH_2), 6.54 (d, $J=7.7$, 1H, ArH), 7.18–7.45 (m, 4H, ArH), 7.70 (m, 1H, ArH), 8.23 (m, 1H, ArH), and 9.83 (s, 1H, CHO); δ_{C} 72.66, 104.96, 121.59, 121.77, 125.33, 125.47, 125.63, 126.75, 127.52, 134.59, 153.29 and 199.43; m/z 186 (M^+ , 65%), 157 (27), 144 (21), 143 (35), 129 (21), 128 (29), 127 (56), 115 (100) and 63 (14).

Synthesis of 1-(1-naphthylthio)-3-nitro-2-propanol (14). A solution of 2-(1-naphthylthio)acetaldehyde (**13**) (360 mg, 1.9 mmol) and nitromethane (309 μL , 5.7 mmol) in THF (8 mL) was cooled to -40°C and to this mixture was added a solution of guanidine **5a** (62 mg, 0.2 mmol) in THF (2 mL). The resulting mixture was stirred at -40°C for 8 h and then water (2 mL) was added. The THF was evaporated at reduced pressure (15 Torr), ether was added (15 mL) and the organic layer was washed with 2N HCl (2x10 mL), water (10 mL) and dried (Na_2SO_4). After removal of the solvent *in vacuo* (15 Torr), flash chromatography (silica gel, CH_2Cl_2) gave the title compound (100 mg, 21%); $[\alpha]_{\text{D}}^{25} +4$ (c 1; EtOH); ν_{max} (film) 3420 (OH), 3020, 1535 (NO_2), 1250, 1225, 1085, 1055, 780 and 760 cm^{-1} ; δ_{H} 3.01 (br s, 1H, OH), 4.16 (m, 2H, ArOCH_2), 4.67–4.83 (m, 3H, CHOH and CH_2NO_2), 6.74 (d, $J=7.6$, 1H, ArH), 7.32–7.52 (m, 4H, ArH), 7.79 (m, 1H, ArH) and 8.14 (m, 1H, ArH); δ_{C} 67.40, 68.57, 78.04, 105.02, 121.35, 121.38, 125.17, 125.61, 126.65, 127.63, 134.45 and 153.39; m/z 247 (M^+ , 30%), 145 (11), 144 (100), 129 (12), 128 (12), 127 (33), 116 (21), 115 (76), 104 (24) and 60 (13).

Synthesis of 3-(isopropylamino)-1-(1-naphthylthio)-2-propanol (propanolol) (15). To a solution of 1-(1-naphthylthio)-3-nitro-2-propanol (**14**) (49 mg, 0.20 mmol) in methanol (8 mL) was added 10% palladium on carbon (30 mg) and the resulting suspension was stirred at room temperature under hydrogen atmosphere at normal pressure for 18 h. Acetone was then added (100 μL , 1.4 mmol) and the mixture was stirred for another 72 h at 50°C . The suspension was filtered through celite and the solvent evaporated (15 Torr) affording propanolol **15** which was purified by crystallization in hexane (15 mg, 30%); mp $88\text{--}89^\circ\text{C}$ (lit¹³, mp 96°C racemic mixture; mp 73°C pure enantiomer); $[\alpha]_{\text{D}}^{25} +2.8$ (c 0.8; EtOH) {lit¹³, $[\alpha]_{\text{D}}^{21} +10.6$ (c 1.02; EtOH) *R* enantiomer}; ν_{max} (KBr) 3260 (OH and NH), 3030, 1570, 1445, 1260, 1230, 1090, 1060, 780 and 760 cm^{-1} ; δ_{H} 1.09 (d, $J=6.3$, 6H, $2\times\text{CH}_3$), 2.83 (m, 1H, CHHNNH), 2.97 (dd, $J=3.1$ and 12.1, 2H, CHHNNH), 3.10 (br s, 2H, OH and NH), 4.03–4.21 (m, 3H, CHOH and ArOCH_2) and 6.79 (d, $J=7.5$, 1H, ArH), 7.31–7.48 (m, 4H, ArH), 7.78 (m, 1H, ArH) and 8.24 (m, 1H, ArH); δ_{C} 22.89, 22.96, 48.90, 49.57, 68.47, 70.73, 104.87, 120.54, 121.79, 125.18, 125.51, 125.78, 126.36, 127.47, 134.45 and 154.30; m/z 259 (M^+ , 5%), 144 (9), 127 (5), 116 (8), 115 (25), 72 (100), 57 (5), 56 (7), 43 (7) and 41 (5).

ACKNOWLEDGEMENTS

This work was supported by the DGICYT of the Spanish Ministerio de Educación y Ciencia (MEC) (Project no. PB91-0751). R. C. and P. S.-A. thank the MEC and Generalitat Valenciana, respectively, for grants.

REFERENCES AND NOTES

1. Rosini, G. In *Comprehensive Organic Synthesis*, Trost, B. M.; Fleming, I. Eds.; Pergamon Press: Oxford 1991, Vol. 2, pp. 321-340.
2. One example of enantioselective Henry reaction catalysed by quinidine has been described: van Aken, E.; Wynberg, H.; van Bolhuis, F. *Acta Chem. Scand.* **1993**, *47*, 122-124.
3. (a) Sasai, H.; Suzuki, T.; Arai, S.; Arai, T.; Shibasaki, M. *J. Am. Chem. Soc.* **1992**, *114*, 4418-4420. (b) Sasai, H.; Suzuki, T.; Itoh, N.; Shibasaki, M. *Tetrahedron Lett.* **1993**, *34*, 851-854. (c) Sasai, H.; Itoh, N.; Suzuki, T.; Shibasaki, M. *Tetrahedron Lett.* **1993**, *34*, 855-858. (d) Sasai, H.; Suzuki, T.; Itoh, N.; Arai, S.; Shibasaki, M. *Tetrahedron Lett.* **1993**, *34*, 2657-2660. (e) Sasai, H.; Suzuki, T.; Itoh, N.; Tanaka, K.; Date, T.; Okamura, K.; Shibasaki, M. *J. Am. Chem. Soc.* **1993**, *115*, 10372-10373. (f) Shibasaki, M.; Sasai, H. *Yuki Gosei Kagaku Kyokaiishi* **1993**, *51*, 972-984. *Chem. Abstr.* **1994**, *120*, 106039f.
4. (a) Forsyth, A. C.; Paton, R. M.; Watt, I. *Tetrahedron Lett.* **1989**, *30*, 993-996. (b) Forsyth, A. C.; Gould, R. O.; Paton, R. M.; Sadler, I. H.; Watt, I. *J. Chem. Soc. Perkin Trans. I* **1993**, 2737-2741.
5. (a) Müller, G.; Riede, J.; Schmidtchen, F. P. *Angew. Chem. Int. Ed. Engl.* **1988**, *27*, 1516-1517. (b) Echavarren, A.; Galán, A.; Lehn, J. M.; de Mendoza, J. *J. Am. Chem. Soc.* **1989**, *111*, 4994-4995.
6. Schmidtchen, F. P. *Tetrahedron Lett.* **1989**, *30*, 4493-4496.
7. Gleich, A.; Schmidtchen, F. P.; Mikulcic, P.; Müller, G. *J. Chem. Soc., Chem. Commun.* **1990**, 55-57.
8. The guanidine moiety plays an important role in the stabilization of the tertiary structure of some proteins using these type of bonds; see, for instance: Fink, M. L.; Bodanszky, M. *J. Am. Chem. Soc.* **1976**, *98*, 974-977.
9. Van Haken, E.; Wynberg, H.; van Bolhuis, F. *J. Chem. Soc., Chem Commun.* **1992**, 629-630.
10. Boyle, P. H.; Convery, M. A.; Davis, A. P.; Hosken, G. D.; Murray, B. A. *J. Chem. Soc., Chem. Commun.* **1992**, 239-242.
11. (a) Mitsunobu, O.; Kato, K.; Kakese, F. *Tetrahedron Lett.* **1969**, 2473-2475. (b) Mitsunobu, O.; Kato, K.; Tomari, M. *Tetrahedron* **1970**, *26*, 5731-5736.
12. Palomo, C.; Mestres, R. *Synthesis* **1981**, 373-374.
13. *Dictionary of Organic Compounds*; 5th Ed.; Chapman & Hall: New York, 1982.
14. Chinchilla, R.; Nájera, C.; Yus, M.; Heumann, A. *Tetrahedron: Asymmetry* **1990**, *1*, 851-854.
15. (*R*)-*O*-Acetyl mandelic acid has been used for the direct ¹H NMR analysis of the enantiomeric composition of amines and 2-aminoalcohols¹⁶.
16. Parker, D.; Taylor, R. J. *Tetrahedron* **1987**, *43*, 5451-5456.
17. (*R*)-*O*-Aryl lactic acids have been used in the kinetic resolution^{14,18} of racemic alcohols and carboxylic acids and for ¹H NMR analysis¹⁹ of chiral compounds.
18. (a) Chinchilla, R.; Nájera, C.; Yus, M.; Heumann, A. *Tetrahedron: Asymmetry* **1991**, *2*, 101-104. (b) Mazón, A.; Nájera, C.; Yus, M.; Heumann, A. *Tetrahedron: Asymmetry* **1992**, *3*, 1455-1466.
19. (a) Heumann, A. *J. Chem. Soc., Chem. Commun.* **1993**, 1113-1115. (b) Heumann, A.; Faure, R. *J. Org. Chem.* **1993**, *58*, 1276-1279.
20. Propanolol has properties as β -adrenergic blocker (*S* isomer) and contraceptive (*R* isomer)^{3c,16}.