

Communications to the Editor

Discovery of (2*S*)-1-(4-Amino-2,3,5-trimethylphenoxy)-3-{4-[4-(4-fluorobenzyl)phenyl]-1-piperazinyl}-2-propanol Dimethanesulfonate (SUN N8075): A Dual Na⁺ and Ca²⁺ Channel Blocker with Antioxidant Activity

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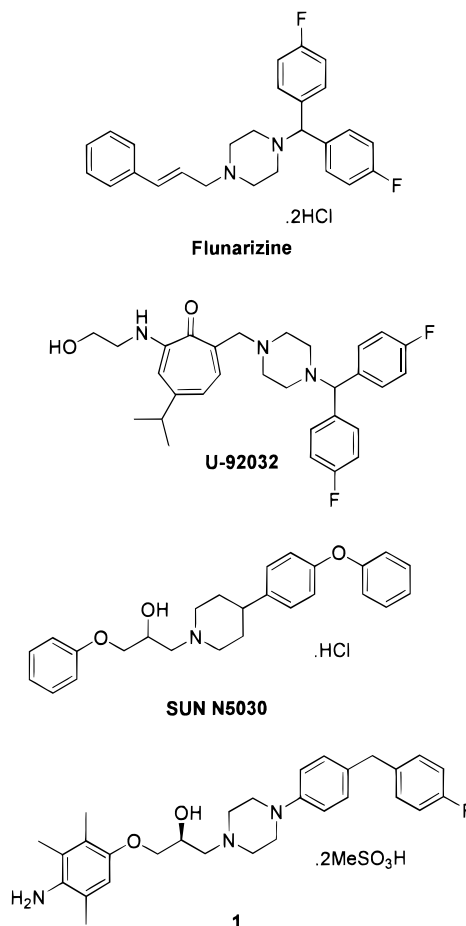
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Introduction. Stroke is a substantial public health problem and one of the leading causes of death in major industrialized countries. An effective and widely applicable treatment of cerebral ischemia would have an enormous public health impact; however, no such treatment has been found as yet.¹ Pathogenesis of progressive and delayed death of nerve cells that occurs in cerebral injury and cerebrovascular diseases, such as stroke and trauma, is considered to be caused by a rise in intracellular Ca²⁺ concentration to a pathological level due to ATP depletion following the failure of intracellular energy-dependent ion homeostasis.² This phenomenon is known as Ca²⁺ overload which induces functional disorders in the mitochondria, randomly activates various Ca²⁺-dependent enzymatic reactions, and invites further Ca²⁺ overload. It has been demonstrated that aberrant activation of Na⁺ and Ca²⁺ channels is extensively involved in the Ca²⁺ overload pathway and the accumulation of intracellular Na⁺ ions results in rapid Ca²⁺ overload by the reverse operation of Na⁺/Ca²⁺ exchange mechanism.^{2,3} In addition, the Ca²⁺-dependent activation of nitric oxide (NO) synthase, phospholipase A₂, and xanthine oxidase causes a marked increase in the generation of reactive oxygen species such as superoxide anion radical (O₂^{•-}), hydrogen peroxide (H₂O₂), hydroxy radical (OH[•]), and peroxy-nitrite (ONOO⁻) on reperfusion of ischemia, imparting irreparable damage to the cell membrane through extensive lipid peroxidation.⁴ These complex biochemical events mutually act as aggravating factors and repeat in a vicious amplifying cycle that ultimately leads to cell death.

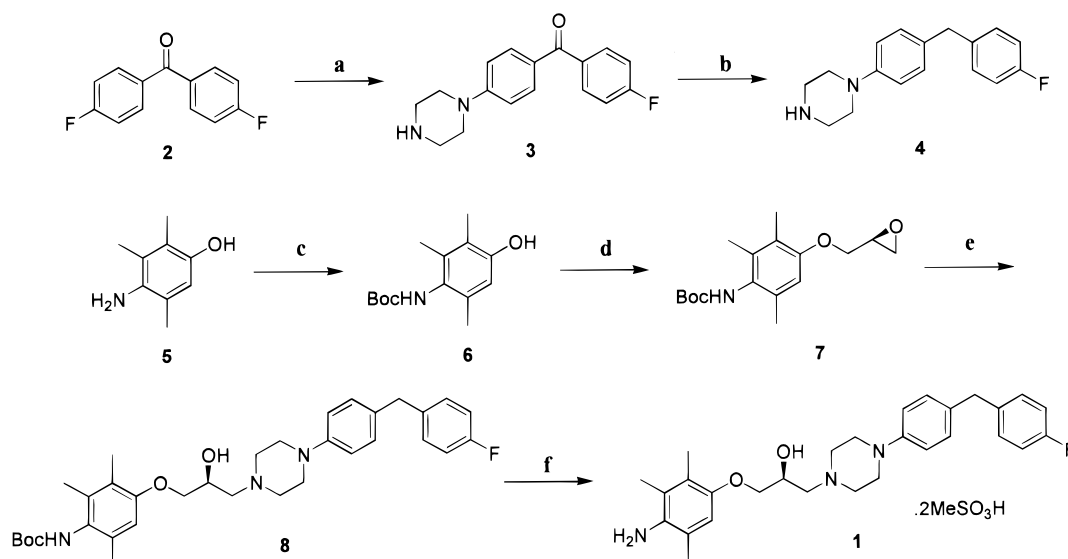
Although clinical trials of quite a few neuroprotectants, including *N*-methyl-D-aspartate (NMDA) receptor antagonists,⁵ AMPA receptor antagonists,⁶ Na⁺ and/or Ca²⁺ channel blockers,^{3c,7} and antioxidants,⁸ are currently underway, most of them interfere with only one of the pathways in the ischemic cascade and the benefits of such therapies have not yet been established.⁹ It has recently been demonstrated that a

Chart 1



combination of compounds, each possessing a different mechanism of action in preventing cell death, exhibits synergistic effects in ischemic animal models.¹⁰ Consequently, a novel neuroprotectant with multiple mechanisms of action may provide clinically significant efficacy. We previously reported a structurally novel class of arylpiperidines, represented by SUN N5030, as Na⁺ and Ca²⁺ channel blockers with reduced affinity for dopamine D₂ receptors¹¹ (Chart 1). In this communication, we report that our continuous endeavor for the development of agents for acute ischemic stroke based on molecular modification of flunarizine^{3b,c,12} as a prototype has resulted in the discovery of a candidate: (2*S*)-1-(4-amino-2,3,5-trimethylphenoxy)-3-{4-[4-(4-fluorobenzyl)phenyl]-1-piperazinyl}-2-propanol dimethanesulfonate (**1**; SUN N8075). This compound possesses not only neuronal Na⁺ and T-type¹² Ca²⁺-channel-blocking effects but also antioxidant activity, which governs potent neuroprotective activity in an in vivo transient middle cerebral artery occlusion (MCAO) model. A survey of literature has revealed that U-92032 endowed with similar multiple mechanisms of action against ischemia was reported by the Upjohn group.^{7b} The pharmacophore for the blockade of ion channels and

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Scheme 1^a

^a Reagents and conditions: (a) piperazine (3 equiv), Et₃N (2 equiv), MeCN, reflux, 12 h, 66%; (b) Et₃SiH (2.3 equiv), concd H₂SO₄ (2.2 equiv), TFA, rt, 2 h, 79%; (c) (Boc)₂O, THF, 80 °C, 2 h, 96%; (d) NaH, (S)-glycidyl 3-nitrobenzenesulfonate, DMF, rt, 2 h, 90%; (e) **4**, 2-propanol, reflux, 6 h, 90%; (f) TFA, CH₂Cl₂, rt, 2 h, then MeSO₃H (2.05 equiv), MeOH, rt, 65%.

Table 1. Biological Activity of **1**^a

entry	IC ₅₀ (μM)				anticonvulsant effects on audiogenic seizures in DBA/2 mice (n = 6) ⁱ ED ₅₀ (mg/kg; ip)	
	anti-veratridine ^b	T-type Ca ²⁺ currents ^c	lipid peroxidation ^d	D ₂ ^h		
1	0.42	2.0	0.31	> 10		6.5
flunarizine	0.29	2.2	> 50 ^e	0.228		11.4
U-92032	0.36	0.3	20.0 ^f	1.69		14.5
α-tocopherol	NT ^j	NT ^j	> 100 ^g	NT ^j		NT ^j

^a Each value represents multiple determinations (≥ 2) with a deviation of less than 20%. ^b Determined as inhibitory effects upon the veratridine-induced depolarization in rat cerebrocortical synaptosomes using a membrane potential sensitive fluorescent dye, rhodamine 6G.¹⁴ ^c Determined as inhibitory effects on low-threshold (T-type) Ca²⁺ currents in primary cultured rat cerebrocortical neurons using whole-cell voltage-clamp recording technique.^{12a} ^d Determined as inhibitory effects on automatic oxidation in plasma membrane of rat brain homogenates by measuring malondialdehyde and 4-hydroxyalkenals using peroxidized lipid colorimetric assay kit.¹⁷ ^e Results reported in refs 7c and 17b are 64.1 and 85 μM, respectively. ^f Result reported in ref 7c is 13.5 μM. ^g Result reported in ref 17b is > 100 μM. ^h Determined in competition experiments with [³H]raclopride.²⁰ ⁱ Compounds were administered intraperitoneally to DBA/2 mice (n = 6) 20 min prior to auditory stimulation of at least 90 dB for 1 min.^{22a} ^j NT, not tested.

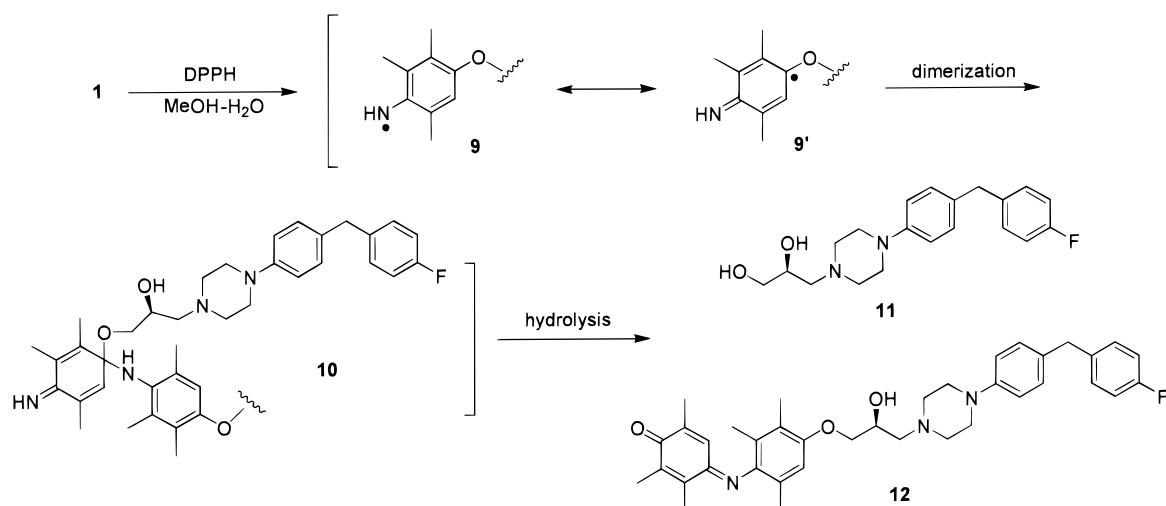
the mechanism of action for the antioxidant activity of **1** are also discussed.

Chemistry. Compound **1** was synthesized using the pathway shown in Scheme 1. Treatment of commercially available 4,4'-difluorobenzophenone with piperazine (3 equiv) in the presence of triethylamine in acetonitrile gave **3** in 66% yield. Reduction of **3** with Et₃SiH (2.3 equiv) in the presence of sulfuric acid (2.2 equiv) and trifluoroacetic acid (TFA) produced the requisite 1-[4-(4-fluorobenzyl)phenyl]piperazine (**4**) in 79% yield. Phenoxo-2,3-epoxypropane (**7**), the coupling partner of **4**, was synthesized from 2,3,5-trimethyl-4-aminophenol (**5**).¹³ Thus, chemoselective protection of the amino group in **5** with di-*tert*-butyl dicarbonate gave **6** as a single product in 96% yield. None of the corresponding *O*-butoxycarbonylated isomer was detected during this reaction. Treatment of **6** with sodium hydride produced the phenolate anion, which was quenched with (S)-glycidyl 3-nitrobenzenesulfonate to give **7** in 90% yield. Coupling reaction between **7** and **4** in 2-propanol proceeded smoothly to give **8** in 90% yield, without concomitant loss of optical purity. Deprotection of the Boc group in **8** by exposure to trifluoroacetic acid followed by treatment with methanesulfonic acid (2.05 equiv) furnished **1** as a dimethanesulfonate in 65% yield with > 98.5% ee after recrystallization from 2-propanol/

water (20:1). Its optical purity was determined by normal phase HPLC using a Daicel Chiralpak OD column (*n*-hexane:2-propanol:diethylamine = 600:400:1) and was found to increase to > 99% ee upon repeated recrystallization. Judging from the ¹⁹F NMR (500 MHz, DMSO-*d*₆), the signal attributable to trifluoroacetic acid could not be detected. Therefore, complete salt exchange occurred from the initially formed trifluoroacetate into the methanesulfonate during the final deprotecting step, presumably due to the higher acid dissociation constant of methanesulfonic acid (pK_a = -1.2) as compared with that of trifluoroacetic acid (pK_a = 0.2).

Results and Discussion. The inhibitory effect of compound **1** on Na⁺ channels stimulated by the neurotoxin veratridine was evaluated in rat cerebrocortical synaptosomes using the voltage-sensitive fluorescent dye rhodamine 6G.¹⁴ The effect of **1** on low-threshold (T-type) Ca²⁺ currents in primary cultured rat cerebrocortical neurons was examined using a whole-cell voltage-clamp recording technique.^{12a} The results are given as IC₅₀ values as shown in Table 1. Compound **1** blocks both Na⁺ and T-type Ca²⁺ channels with potency equal to that of the reference standards flunarizine and U-92032.^{7b} Compound **1** showed concentration-dependent block of T-type Ca²⁺ currents induced by a depolarizing pulse to -40 mV from a holding potential (V_H) of

Scheme 2



−100 mV. We also investigated the effect of **1** on neurotoxin binding sites of Na⁺ channels. Compound **1** displaced the binding of the site 2 ligand, [³H]batrachotoxin,^{15b} with high affinity (IC₅₀ = 0.75 μM) but had no effect on the site 1 ligand, [³H]saxitoxin, binding (IC₅₀ > 10 μM).¹⁶ [Each value represents replicate determinations, and tetrodotoxin was used as a reference compound: IC₅₀ = 0.0063 μM (site 1) and >10 μM (site 2).] The neurotoxin receptor site 2 of Na⁺ channels is believed to be localized on a region involved in voltage-dependent activation and inactivation and is allosterically linked to the transmembrane pore of the channels.^{15a,c} The Na⁺ channel site 2 blockers work in a voltage-dependent manner, implying event- and site-specific block of Na⁺ channels without primary hemodynamic adverse effects in ischemic diseases.^{3b,c}

Antioxidant activity was measured by lipid peroxidation-suppressing effect in rat cerebrocortical membranes using a BIOXYTECH/LPO-586 peroxidized colorimetric assay kit.¹⁷ Surprisingly, compound **1** exhibits lipid peroxidation-suppressing effects with a potency greater than that of flunarizine, U-92032, or α-tocopherol. These results prompted us to investigate the mechanism of action for antioxidant activity at the molecular level by reacting **1** with 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical.^{18a,b} Compound **1** dissolved in methanol/pH = 6.86 standard buffer solution (1:1, v/v) and when subjected to DPPH radical unexpectedly produced propanediol **11** and quinone monimine **12**. Both structures, **11** and **12**, were elucidated by ¹H NMR, IR, and HRMS spectra. In the ¹H NMR (400 MHz, CDCl₃) of **12**, the signal assignable to the hydrogen on the quinone monimine moiety appeared at δ 6.38 (singlet). In addition, the signal attributable to the hydrogen of the trimethylphenoxy ring was at δ 6.64 (singlet), showing a shift to lower field than that of the parent compound **1** (δ 6.57) due to the electron-withdrawing effect of the quinone monimine moiety. As shown in Scheme 2, the degradation pathway is proposed to be triggered by abstraction of the hydrogen atom from the amino group in the 2,3,5-trimethyl-4-aminophenoxy moiety followed by disproportionation of the generated anilino radical **9** and cyclohexadienyl radical **9'** to give the dimerization product **10**, which is subsequently hydrolyzed to produce compounds **11** and

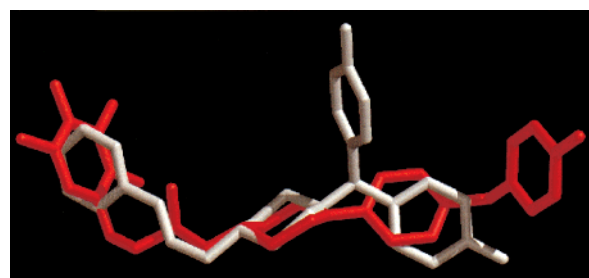


Figure 1. Overlay of energetically allowed conformers of **1** (red) and flunarizine (white).

12. These results suggest that the aminophenoxy moiety in **1** is the essential pharmacophore for antioxidant activity and that the mechanism of action is similar to that of α-tocopherol.^{18c} The direct evidence for radical-scavenging action was obtained on the ESR spectrum where the intensity of the signal of DPPH radical decreased dose-dependently upon the addition of **1**. With increasing evidence that the excess production of reactive oxygen species affects ion homeostasis and is a major causative factor in cell death,⁴ it is tempting to speculate that the ion channel-blocking and antioxidant properties of **1** may act synergistically against cerebral ischemia.¹⁰ The dopamine D₂ receptor binding affinity was assessed in a competitive binding assay using [³H]-raclopride in rat striatum membranes.¹⁹ In sharp contrast to flunarizine, compound **1** has no binding affinity for dopamine D₂ receptors (IC₅₀ > 10 μM). These differences clearly demonstrate that **1** not only has structural features distinctly different from those of flunarizine and related compounds but also may have a decreased clinical risk for extrapyramidal side effects.²⁰

To elucidate the common pharmacophore and/or the special arrangement of essential parts for inhibition of ion channels and influence for binding affinity at dopamine D₂ receptors, a systematic conformation search and a similar structure search by root-mean-square (rms) fitting were carried out for **1** and flunarizine. Structures were minimized with MM2 parameters implemented by MacroModel.²¹ As depicted in Figure 1, the corresponding pharmacophoric groups between **1** (red) and flunarizine (white) closely overlap, with the exception of each distal benzene ring. This

implies that the benzene ring directly connected to the piperazine ring in **1** likely occupies the same binding site of the ion channels as flunarizine, whereas the other benzene ring of the diphenylmethane moiety likely interferes with dopamine D₂ receptor binding.

Next, we investigated the effect of **1** on audiogenic seizures in DBA/2 mice to confirm in vivo activity and brain permeability.^{22a,b} Anticonvulsant and neuroprotective activities are mutually related in several voltage-dependent Na⁺ channel blockers.^{22c} Compound **1** exhibits potent anticonvulsant effects following systemic (ip) administration with an ED₅₀ of 6.5 mg/kg. The neuroprotective activity of **1** was assessed in a transient MCAO model.²³ The middle cerebral artery (MCA) occluded for 60 min and 24 h after reperfusion neuronal damage was quantified by staining brains with 2,3,5-triphenyltetrazolium chloride²⁴ (TTC). The relative area of infarction to the whole area of the corresponding cerebrum is calculated using a computerized image analysis system and was shown in percent (vide infra). Compound **1** was administered intravenously immediately after both MCAO and reperfusion (each 3 mg/kg). Consequently, **1** significantly reduced infarct size in all treated groups compared to vehicle-treated rats [$4.77 \pm 0.84\%$ ($n = 14$) vs $14.87 \pm 1.24\%$ ($n = 15$) (mean \pm SEM), $**p < 0.01$ (Student's t -tests)], amounting to 67.9% reduction in mean infarct volume, whereas U-92032 effected only minor reduction (<15%). In this model, rectal temperature was found to increase during transient MCAO to above 38.5 °C and **1** slightly reduced the ischemic hyperthermia within 1 °C. Mild hypothermia immediately after reperfusion reduces infarct size and improves cerebral outcome.²⁵ These results indicate that **1** has a pronounced neuroprotective efficacy against neuronal damage induced by transient focal ischemia in rats. Interestingly, **1** at the effective doses had no effects on systemic blood pressure and heart rate in anesthetized rats. These studies will be described in more detail in subsequent papers.

In conclusion, we describe a novel neuroprotectant, **1**, that blocks both neuronal Na⁺ and T-type Ca²⁺ channels but which does not affect dopamine D₂ receptors. Compound **1** is also a powerful antioxidant, significantly more potent than the standard. Since compound **1** (SUN N8075) has complementarily multiple beneficial effects against ischemia and remarkable neuroprotective activity in the transient MCAO model, this compound may have greater clinical efficacy for the treatment of acute ischemic stroke compared to candidates with a single mechanism of action in preventing neuronal cell death or damage.^{9,10} The structure–activity relationships (SAR) of this series of compounds will be reported in due course.²⁶

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Supporting Information Available: Experimental details for the synthesis of compounds **1**, **3**, **4**, **6–8**, **11**, and **12**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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