



Tetrahedron Letters 44 (2003) 3745-3748

TETRAHEDRON LETTERS

Stereoselective synthesis of the C_7-C_{21} segment of epothilone A via asymmetric alkoxyallyl- and crotylboration^{\Leftrightarrow}

P. Veeraraghavan Ramachandran,* Bodhuri Prabhudas, Debarshi Pratihar, J. Subash Chandra and M. Venkat Ram Reddy

Herbert C. Brown Center for Borane Research, Department of Chemistry, 560 Oval Drive, Purdue University, West Lafayette, IN 47907-2084, USA

Received 6 February 2003; revised 21 March 2003; accepted 24 March 2003

Abstract—The synthesis of the C_7 – C_{21} fragment of epothilone A involving asymmetric alkoxyallyl- and crotylboration using α -pinene-derived reagents is described. © 2003 Elsevier Science Ltd. All rights reserved.

As part of our program on the applications of pinanebased versatile reagents¹ for organic syntheses, we have recently described the synthesis of several biologically active molecules via asymmetric allylboration–ring-closing metathesis pathway.² Herein we describe our approach towards the C_7 – C_{21} synthon (1) of potent anti-cancer agent epothilone $A^{3,4}$ via α -pinene-derived asymmetric alkoxyallyl- and crotylboration. Our retrosynthetic analysis of 1 is shown in Scheme 1.

We envisaged the synthesis of 1 by the convergence of 2, 3, and 4. 4-Chloromethyl-2-methylthiazole was pre-

pared from 1,3-dichloroacetone and thioacetamide using a literature procedure⁵ and converted to the Wittig salt **2**. We chose to prepare **3** starting with an alkoxyallylboration of acetaldehyde using B-(Z)- γ -(2-methoxyethoxy)methoxyallyldiisopinocampheylborane.⁶ Protection of **5** with benzyl bromide, followed by a hydroboration–oxidation provided the corresponding alcohol, which was oxidized using Dess–Martin periodinane (DMP)⁷ to the required aldehyde **3** (Scheme 2). Although one of the chiral centers derived from alkoxyallylboration will be converted to the ketone **13**, the protocol does not necessitate any additional steps to create this chiral center.



Scheme 1.

0040-4039/03/\$ - see front matter @ 2003 Elsevier Science Ltd. All rights reserved. doi:10.1016/S0040-4039(03)00748-2

Keywords: epothilone; alkoxyallylboration; crotylboration; α -pinene; Wittig reaction.

^{*} Contribution No. 23 from Herbert C. Brown Center for Borane Research.

^{*} Corresponding author. E-mail: chandran@purdue.edu



Scheme 2.

The preparation of 1 was initially attempted as shown in Scheme 3. TBS-protection of commercially available methyl (R)-3-hydroxy-2-methylpropionate 8, followed by reduction of the methyl ester and iodination provided 9. Nucleophilic substitution of the iodide with allyl magnesium bromide in the presence of catalytic amounts of lithium tetrachlorocuprate⁸ or stoichiometric cuprous iodide afforded variable yields (22-50%) of 10, which when subjected to hydroboration-oxidation, followed by iodination furnished the 1°-iodide 11. Wittig coupling of aldehyde 3 with 11 gave 58% of the olefin 12. Debenzylation under Birch reduction conditions, followed by DMP-oxidation provided the ketone 13. The coupling⁹ of Wittig salt 2 and 13 yielded 14. Silvl deprotection, followed by a DMP-oxidation furnished the required synthon 1 for epothilone A (Scheme 3).

The unreliable yields during the allyl-substitution of the 1°-iodide 9 compelled us to modify the synthetic scheme. We designed a protocol utilizing pinane-based crotylboration.¹⁰ A bonus from this procedure is the replacement of the relatively expensive chiral ester 8.

We began our synthesis with the Z-crotylboration of *tert*-butyldimethylsiloxyacetaldehyde **15** to provide the homoallylic alcohol **16**. Silyl protection of the hydroxyl group, followed by oxidative cleavage of the

olefin furnished the aldehyde 17. Wittig coupling with 3-benzyloxypropyl iodide provided the olefin 18. A simultaneous debenzylation and reduction of the double bond was achieved by Pd-catalyzed hydrogenation. The alcohol was converted to the corresponding iodide 19, which was utilized for a Wittig reaction with the aldehyde 3 to provide 20. It is noteworthy that the Z-olefin was formed selectively and in high yields from the substrate that is prone to β -elimination. Debenzylation using lithium in liquid ammonia achieved the formation of the 2°-alcohol without affecting the double bond. The alcohol was oxidized using DMP to the ketone 21, which was coupled⁹ with 2 to provide 22 in good yields. Selective deprotection of the two silvl groups using dilute HCl, followed by Pb(OAc)₄ cleavage provided the target synthon 1 (Scheme 4).

The structures of all of the intermediates were confirmed by ¹³C and ¹H NMR spectroscopy and mass spectrometry.

Thus, we have achieved a large-scale preparation of the C_7 - C_{21} synthon 1^{11} of epothilone A using Wittig reaction and pinane-based alkoxyallyl- and crotylboration as key steps. Studies towards completing the preparation of epothilone A and the synthesis of other epothilones, particularly certain fluorinated analogs are under way.





Scheme 4.

Acknowledgements

Financial support from the Herbert C. Brown Center for Borane Research and Aldrich Chemical Company is greatly acknowledged.

22References

- 1. Ramachandran, P. V. Aldrichim. Acta 2002, 35, 23.
- (a) Ramachandran, P. V.; Chandra, J. S.; Reddy, M. V. R. J. Org. Chem. 2002, 67, 7547; (b) Reddy, M. V. R.; Rearick, J. P.; Hoch, N.; Ramachandran, P. V. Org. Lett. 2001, 3, 19; (c) Reddy, M. V. R.; Yucel, A. J.; Ramachandran, P. V. J. Org. Chem. 2001, 66, 2512; (d) Ramachadran, P. V.; Reddy, M. V. R.; Brown, H. C. Tetrahedron Lett. 2000, 41, 583.
- For reviews on epothilones, see: (a) Stachel, S. J.; Biswas, K.; Danishefsky, S. J. Curr. Pharm. Des. 2001, 7, 1277; (b) Mulzer, J. Monatsh. Chem. 2000, 131, 205; (c) Nicolaou, K. C.; Roschanger, F.; Vourloumis, D. Angew. Chem., Int. Ed. 1998, 37, 2015; (d) Finlay, M. R. V. Chem. Ind. 1997, 991; (e) Wessjohann, L. Angew. Chem., Int. Ed. Engl. 1997, 36, 715.
- 4. For the syntheses of epothilone A, see: (a) Schinzer, D.; Limberg, A.; Bauer, A.; Bohm, O. M.; Cordes, M. Angew. Chem., Int. Ed. Engl. 1997, 36, 523; (b) Nicolaou, K. C.; He, Y.; Vourloumis, D.; Vallberg, H.; Roschangar, F.; Sarabia, F.; Ninkovic, S.; Yang, Z.; Trujillo, J. I.; J. Am. Chem. Soc. 1997, 119, 7960; (c) Nicolaou, K. C.; Ninkovic, S.; Sarabia, F.; Vourloumis, D.; He, Y.; Vallberg, H.; Finlay, M. R. V.; Yang, Z. J. Am. Chem. Soc. 1997, 119, 7974; (d) Meng, D. F.; Bertinato, P.; Balog, A.; Su, D. S.; Kamenecka, T.; Sorensen, E. J.; Danishefsky, S. J. J. Am. Chem. Soc. **1997**, *119*, 10073; (e) Schinzer, D.; Bauer, A.; Bohm, O. M.; Limberg, A.; Cordes, M. Chem. Eur. J. 1999, 5, 2483-2491; (f) Zhu, B.; Panek, J. S. Org. Lett. 2000, 2, 2575; (g) Sawada, D.; Kanai, M.; Shibasaki, M. J. Am. Chem. Soc. 2000, 122, 10521; (h) Furstner, A.; Mathes,

C.; Lehmann, C. W. *Chem. Eur. J.* **2001**, *7*, 5299; (i) Hindupur, R. M.; Panicker, B.; Valluri, M.; Avery, M. A. *Tetrahedron Lett.* **2001**, *42*, 7341; (j) Bode, J. W.; Carreira, E. M. *J. Am. Chem. Soc.* **2001**, *123*, 3611; (k) Liu, Z.-Y.; Chen, Z.-C.; Yu, C.-Z.; Wang, R.-F.; Zhang, R.-Z.; Huang, C.-S.; Yan, Z.; Cao, D.-R.; Sun, J.-B.; Li, G. *Chem. Eur. J.* **2002**, *8*, 3747.

- Hooper, F. E.; Johnson, T. B. J. Am. Chem. Soc. 1934, 56, 470.
- 6. (a) Brown, H. C.; Jadhav, P. K.; Bhat, K. S. J. Am Chem. Soc. **1988**, 110, 1535; (b) Smith, A. L.; Pitsinos, E. N.; Hwang, C. K.; Mizuno, Y.; Saimoto, H.; Scarlato, G. R.; Suzuki, T.; Nicolaou, K. C. J. Am. Chem. Soc. **1993**, 115, 7612–7624; (c) The reagent used was prepared from (–)- α -pinene.
- 7. Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. 1991, 13, 7277.
- For reviews on cuprate coupling, see: (a) Tamura, M.; Kochi, J. Synthesis 1971, 303; (b) Fouquet, G.; Schlosser, M. Angew. Chem., Int. Ed. Engl. 1974, 13, 82; (c) Schlosser, M. Angew. Chem., Int. Ed. Engl. 1974, 13, 701.
- Zhao, Z.; Scarlato, G. R.; Armstrong, R. W. Tetrahedron Lett. 1991, 32, 1609.
- (a) Brown, H. C.; Bhat, K. S. J. Am Chem. Soc. 1986, 108, 5919; (b) The reagent used was prepared from (-)-α-pinene.
- 11. Characterization of 1: ¹H NMR: δ (ppm) (CDCl₃): 9.58 (d, J=1.95 Hz, 1H), 6.94 (s, 1H), 6.47 (s, 1H), 5.39–5.43 (m, 2H), 4.71 (d, J=6.84 Hz, 1H), 4.63 (d, J=6.87 Hz, 1H), 4.12 (t, J=6.75 Hz, 1H), 3.78–3.85 (m, 1H), 3.51– 3.63 (m, 3H), 3.37 (s, 3H), 2.69 (s, 3H), 2.27–2.42 (m, 3H), 1.97–2.07 (m, 5H), 1.33–1.41 (m, 4H), 1.06 (d, J=6.99 Hz, 3H); ¹³C NMR δ (ppm) (CDCl₃): 205.14, 164.58, 152.76, 138.28, 131.11, 125.98, 121.54, 115.98, 92.78, 81.56, 71.82, 67.06, 59.06, 46.24, 31.97, 30.08, 27.38, 26.88, 19.27, 13.84, 13.34; EI-MS: *m*/*z* 256, 89 [(CH₃OCH₂CH₂OCH₂)⁺, 100%]; CI-MS: *m*/*z* 396 (M+ H)⁺, 290 [(M+H)–(HOCH₂OCH₂CH₂OCH₃), 100%], 256, 168, 89; HRMS-CI: (M+H) 396.2205 (observed), 396.2209 (calcd).

Characterization of **20**: ¹H NMR δ (ppm) (CDCl₃): 7.22–7.43 (m, 5H), 5.36–5.51 (m, 2H), 4.79 (s, 2H), 4.62 (d,

J=11.88 Hz, 1H), 4.51 (d, *J*=11.82 Hz, 1H), 3.55–3.73 (m, 5H), 3.46–3.53 (m, 4H), 3.38 (s, 3H), 2.24–2.45 (m, 2H), 1.97–2.08 (m, 2H), 1.61–1.64 (m, 1H), 1.26–1.42 (m, 3H), 1.19 (d, *J*=5.49 Hz, 3H), 0.90 (s, 9H), 0.89 (s, 9H), 0.82 (d, *J*=6.57 Hz, 3H), 0.05 (m, 12H); ¹³C NMR δ (ppm) (CDCl₃): 138.85, 131.95, 128.30, 127.65, 127.45, 125.69, 95.48, 79.88, 75.87, 71.78, 71.29, 67.04, 65.36, 59.03, 35.31, 33.46, 28.13, 27.73, 26.04, 25.97, 18.39, 18.23, 15.32, 13.37, –3.93, –4.77, –5.24, –5.33; ESI: *m/z* 675 (Na adduct); HRMS-ESI: 675.4460 (actual), 675.4452 (calcd).

Characterization of 22: ¹H NMR δ (ppm) (CDCl₃): 6.92

(s, 1H), 6.48 (s, 1H), 5.32–5.48 (m, 2H), 4.71 (d, J=6.87 Hz, 1H), 4.63 (d, J=6.87 Hz, 1H), 4.12 (t, J=6.86 Hz, 1H), 3.79–3.86 (m, 1H), 3.44–3.63 (m, 6H), 3.37 (s, 3H), 2.69 (s, 3H), 2.31–2.45 (m, 2H), 1.99–2.10 (m, 5H), 1.59–1.64 (m, 1H), 1.07–1.44 (m, 4H), 0.87 (s, 9H), 0.86 (s, 9H), 0.79 (d, J=6.87 Hz, 3H), 0.02 (m, 12H); ¹³C NMR δ (ppm) (CDCl₃): 164.48, 152.83, 138.41, 131.95, 125.23, 121.48, 115.88, 92.75, 81.67, 75.85, 71.83, 67.00, 65.34, 59.02, 35.24, 33.45, 31.91, 27.76, 27.69, 26.02, 25.95, 19.24, 18.36, 18.21, 13.79, 13.35, –3.95, –4.79, –5.26, –5.35; ESI: m/z 656, 678 (Na adduct); HRMS-ESI: 656.4202 (actual), 656.4200 (calcd).