



Pd(II)-catalyzed intramolecular aminopalladation/direct C–H arylation under aerobic conditions: synthesis of pyrrolo[1,2-*a*]indoles



Tiffany Piou^{a,b}, Luc Neuville^{b,*}, Jieping Zhu^{a,b,*}

^a Institute of Chemical Sciences and Engineering, Ecole Polytechnique Fédérale de Lausanne, EPFL-SB-ISIC-LSPN, CH-1015 Lausanne, Switzerland

^b Centre de Recherche de Gif, Institut de Chimie des Substances Naturelles, Laboratoire International Associé, CNRS, 91198 Gif-sur-Yvette Cedex, France

ARTICLE INFO

Article history:

Received 27 November 2012

Received in revised form 21 December 2012

Accepted 2 January 2013

Available online 9 January 2013

Keywords:

Palladium catalysis

Domino reaction

C–H activation

Indole

Aminopalladation

ABSTRACT

Heating a DMA/pivalic acid (v/v=4/1) solution of diversely substituted 6-(phenylamino)hex-2-ynoates in the presence of a catalytic amount of Pd(OAc)₂ under oxygen atmosphere afforded pyrrolo[1,2-*a*]indoles in moderate to good yields. A domino sequence involving intramolecular aminopalladation followed by C–H activation and reductive elimination was proposed to account for the observed bis-cyclization.

© 2013 Elsevier Ltd. All rights reserved.

1. Introduction

Pyrrolo[1,2-*a*]indole is an important structural motif found in many bioactive natural products and pharmaceuticals (Fig. 1).¹ Therefore, development of new synthetic methodologies allowing a rapid construction of this skeleton is highly desirable. Most of the existing methods required the pre-construction of functionalized indoles or pyrrolidines,^{2,3} which were in turn prepared by multi-step processes.⁴ Based on the Larock's indole synthesis,⁵ Lu and co-workers at Boehringer-Ingelheim pharmaceuticals developed recently a one-step synthesis of polycyclic indoles by way of a Pd(0)-catalyzed intramolecular cyclization of 2-haloanilides bearing a properly tethered alkynyl function (Eq. 1, Scheme 1).⁶ While the reaction is applicable to a wide range of substrates, it relied on the use of *ortho*-halogenated anilides as starting materials. In connection with our ongoing projects dealing with the development of palladium-catalyzed domino processes⁷ involving C–H functionalization⁸ as a key step,⁹ we became interested in the development of a Pd(II)-catalyzed cyclization of **1** for the synthesis of pyrrolo[1,2-*a*]indoles **2** (Eq. 2, Scheme 1). If realized, this reaction would be more atom-economic and sustainable relative to Boehringer-Ingelheim Pharmaceuticals' protocol. Recent elegant work from Glorius¹⁰ on the Pd(II)-catalyzed oxidative cyclization of *N*-aryl enaminones to indoles¹¹ and from Jiao¹² on the

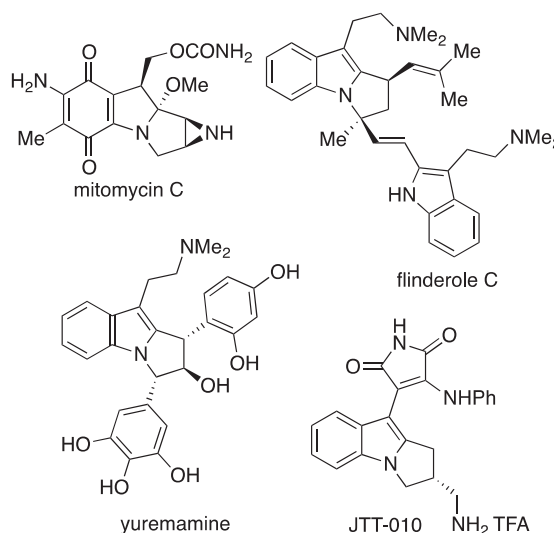
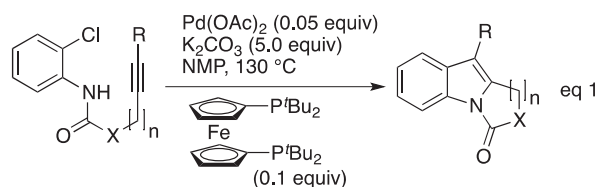


Fig. 1. Representative compounds containing pyrrolo[1,2-*a*]indole cores.

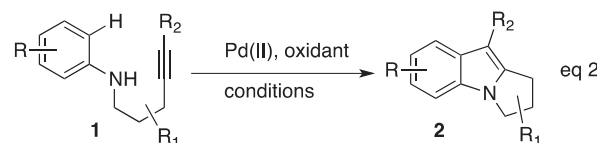
Pd(II)-catalyzed one pot synthesis of indoles from simple aniline and an activated alkyne provided hints to our reaction design.¹³ Although aminometallation/C–C bond formation via C–H functionalization has been widely exploited in rhodium catalysis,¹⁴ we noted that examples involving Pd(II)-catalyzed aminopalladation/C–H functionalization under oxidative conditions are scarce.^{15,16}

* Corresponding authors. E-mail addresses: luc.neuville@icsn.cnrs-gif.fr (L. Neuville), jieping.zhu@epfl.ch (J. Zhu).

Lu and co-workers



This work



Scheme 1. Palladium-catalyzed synthesis of pyrrolo[1,2-a]indoles.

2. Results and discussions

We began our studies using the easily accessible ethyl 6-(phenylamino)hex-2-ynoate **1a**¹⁷ as a model substrate for the survey of reaction conditions. As shown in Table 1, the double cyclization of **1a** proceeded in the presence of palladium acetate (0.1 equiv), pivalic acid¹⁸ (1.0 equiv) under oxygen atmosphere¹⁹ in DMA at 120 °C to afford the desired ethyl 2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (**2a**) in 34% yield (entry 1, Table 1). When the reaction was performed in a mixture of DMA/PivOH (v/v=4/1), we were pleased to isolate the expected pyrrolo[1,2-a]indole **2a** in 63% yield. Using palladium dichloride as catalyst instead of palladium acetate under the otherwise identical conditions led to complete degradation of the starting material (entry 10). Compound **2a** was isolated in lower yield when the reaction was performed in pure pivalic acid (entry 3) and other solvent mixtures (entries 4 and 5). Addition of a co-oxidant in the reaction in combination with molecular oxygen is detrimental to the reaction efficiency (entries 7–9). It should be mentioned that air was also a competent terminal oxidant to afford the desired product **2a** (entry 6).

Table 1

Pd-catalyzed bis-cyclization of ethyl 6-(phenylamino)hex-2-ynoate, survey of reaction conditions

Entry	Co-oxidant ^b	Solvent	Yield ^a (%)
1	—	DMA ^c	34
2	—	DMA/PivOH (4/1)	63
3	—	PivOH	57
4	—	Xylene/PivOH (4/1)	50
5	—	DMSO/PivOH (4/1)	10
6	— ^d	DMA/PivOH (4/1)	52
7	BQ	DMA/PivOH (4/1)	35
8	Cu(OAc) ₂	DMA/PivOH (4/1)	18
9	PhI(OAc) ₂	DMA/PivOH (4/1)	34
10 ^e	—	DMA/PivOH (4/1)	n. d.

^a Isolated yield.^b Co-oxidant (2.0 equiv).^c PivOH (1.0 equiv).^d Under air atmosphere instead of O₂.^e PdCl₂ (0.1 equiv).

Table 2

Scope of the reaction^a

Entry	Substrate 1	Product 2	Yield ^b (%)
1			2b , R=OMe, 41 2c , R=CO ₂ Et, 63 2d , R=CN, 46 2e , R=Me, 51 2f , R=Cl, 46
2			2g , R=6-Me and 2g' , R=4-Me, 83 (1.2/1)
3			2h , 18
4			2i , 72
5			2j / 2j' , 72 (3.2/1)
6			2k , R ₁ =R ₂ =R ₃ =H, 70 2l , R ₁ =R ₃ =H, R ₂ =CO ₂ Et, 69 2m , R ₁ =R ₃ =Me, R ₂ =H, 77
7			2n , R=CO ₂ Me, 51 2o , R=CO ₂ ⁱ Pr, 47 2p , R=Ph, trace

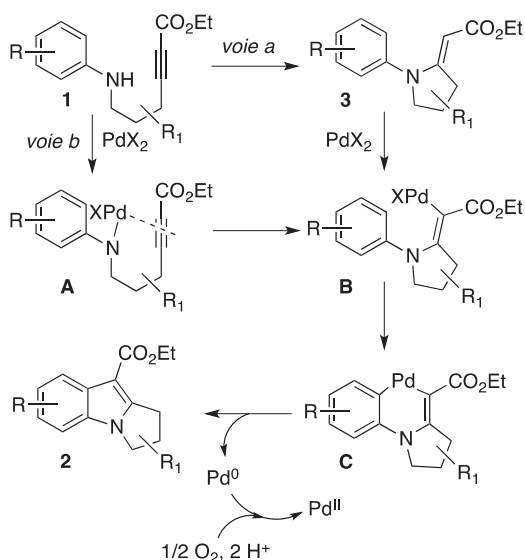
^a Reactions were carried out under oxygen atmosphere using **1a** (1.0 equiv), Pd(OAc)₂ (0.1 equiv), and DMA/PivOH (v/v=4/1) at 120 °C, 4–8 h.^b Isolated yield.

groups at *para*-position of aniline were tolerated furnishing the corresponding tricyclic indoles (**2b–f**) (entry 1, Table 2). When *N*-*meta*-tolyl-anilide was employed, a mixture of two separable regioisomers **2g** and **2g'** (1.2/1 ratio) was isolated in 83% overall yield in favor of the 6-substituted pyrrolo[1,2-a]indole **2g** (entry 2). However, the presence of a substituent *ortho* to anilide reduced dramatically the yield of the domino process (entry 3). Ethyl 6-((3,5-dimethylphenyl)amino)hex-2-ynoate **1i** underwent double cyclizations to produce the indole derivative **2i** in good yield (entry 4). Substrates containing *gem*-dimethyl group on the side chain reacted smoothly to provide the corresponding pyrrolo[1,2-a]indoles in good yields (entry 6). Cyclization of the β -naphthylaniline derivative **1j** furnished two separable regioisomers **2j** and **2j'** in 72% overall yield in favor of the 1,2-fused indole **2j** (ratio 3.2:1, entry 5). The presence of an electron-withdrawing group at the terminal position of alkyne is mandatory as the *N*-(5-phenylpent-4-yn-1-yl)

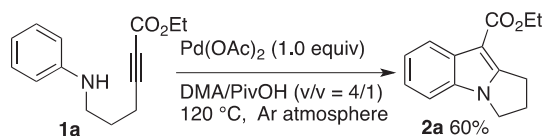
With the optimum conditions [Pd(OAc)₂ (0.1 equiv), O₂ (1.0 atm), DMA/PivOH=4/1, c 0.2 M at 120 °C] in hand, the scope of the domino process was next examined. The results are summarized in Table 2. Substrates with electron-withdrawing or -donating

aniline (**1p**) failed to produce the expected pyrrolo[1,2-*a*]indole **2p** (entry 7).

Treatment of **1a** in the presence of a stoichiometric amount of palladium acetate under an argon atmosphere afforded cyclization product **2a** in a similar yield (Scheme 3). This result is consistent with the Pd(II)/Pd(0) catalytic cycle. Mechanistically, either an intramolecular aza-Michael addition (*voie a*, Scheme 2) or an intramolecular aminopalladation (*voie b*, Scheme 2) could initiate the present domino process. To probe the reaction mechanism, (*E*)-ethyl-2-(1-phenylpyrrolidin-2-ylidene)acetate (**3a**) was synthesized.²⁰ Submitting **3a** to our standard conditions afforded **2a** in only 7% yield together with pyrrole (**5**, 8%)²¹ and recovered starting material (Scheme 4). The result of this control experiment indicated that the *voie a* might not be the main pathway responsible for the conversion of **1a** to **2a**. It is interesting to note that Glorius has observed that *N*-methyl *N*-phenyl enaminone **6** failed to undergo the intramolecular dehydrogenative coupling to form indole under the conditions that were effective for the cyclization of secondary enaminone **7**.^{10b} The deviation from the co-planarity between nitrogen atom and the double bond in tertiary enaminones **3a** and **6** could reduce the nucleophilicity of its β -carbon, impeding therefore the formation of the vinyl palladium intermediate of type B, hence the failure of the cyclization.

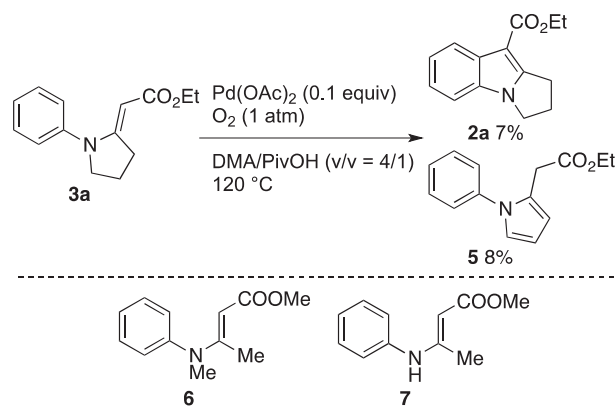


Scheme 2. Proposed mechanism for the domino process.



Scheme 3. Reaction of **1a** in the presence of a stoichiometric amount of palladium acetate under argon atmosphere.

On the basis of these preliminary results, we assumed that *voie b* (Scheme 2) would be operative to account for the formation of pyrrolo[1,2-*a*]indoles (**2**) from linear precursors **1**. Coordination of Pd(II) to the substrate **1** followed by *syn*-aminopalladation of alkyne via intermediate **A** would afford the vinyl-Pd(II) complex **B**.²² Activation of the neighboring aromatic C–H bond would lead to the formation of a six-membered palladacycle **C**, which upon reductive elimination would provide the pyrrolo[1,2-*a*]indole **2** and Pd(0). Oxidation of Pd(0) to Pd(II) by O₂ completed then the catalytic cycle.



Scheme 4. Control experiment: attempted cyclization of enaminone **3a** under optimized conditions.

3. Conclusion

In summary, we disclosed a palladium-catalyzed one-step synthesis of pyrrolo[1,2-*a*]indoles (**2**) from diversely substituted ethyl 6-(phenylamino)hex-2-ynoates. A domino sequence involving intramolecular aminopalladation followed by direct C–H arylation was proposed to account for the bis-cyclization process.

4. Experimental part

4.1. General information

Melting points (mp) were recorded using Büchi B-540 melting point apparatus and are uncorrected. Infrared spectra were recorded on a Perkin Elmer Spectrum BX FT-IR spectrometer. Proton NMR (¹H) spectra were recorded at 500 MHz on a Bruker AC-500 spectrometer or at 300 MHz on a Bruker AC-300 spectrometer. Carbon NMR (¹³C) spectra were similarly recorded at 125 or 75 MHz. Proton NMR (¹H NMR) and carbon NMR (¹³C NMR) chemical shifts (δ) are reported in parts per million (ppm) relative to residual proton signals in CDCl₃ (δ =7.26, 77.23). Coupling constants (*J*) are reported in Hertz (Hz) and refer to apparent multiplicities. The following abbreviations are used for the multiplicities: s: singlet, d: doublet, t: triplet, q: quartet, quint: quintet, sex: sextet, sept: septet, m: multiplet. Mass spectra were obtained either from an AEI MS-50 (EI) or an AEI MS-9 using positive or negative electron spray (ES⁺ or ES[−]) for the high resolution mass spectra (HRMS). Flash chromatography was performed using SDS silicagel 60 (35–70 μ m). Thin layer chromatography (TLC) was carried out on 5×20 cm plates with a layer thickness of 0.25 mm (SDS Silicagel 60 F254). Visualization was achieved under a UVP mineralight UVGL-58 lamp. All reagents were obtained from commercial suppliers unless otherwise stated. When necessary, organic solvents were routinely dried and/or distilled prior to use and stored over molecular sieves under argon. Procedures for the synthesis of substrates **1a–p** and **3a** are detailed in the [Supplementary data](#).

4.2. General procedure for the pyrrolo[1,2-*a*]indoles formations

To a solution of **1** (1.0 equiv) in a mixture of DMA/PivOH (*v/v*=4/1, c 0.16 M) was added Pd(OAc)₂ (0.1 equiv). The reaction mixture was stirred under oxygen atmosphere at 120 °C until the complete disappearance of the starting material (monitored by TLC, usually 4 h). The solution was then cooled to rt, diluted with ethyl acetate, and washed with water, saturated aqueous NaHCO₃, successfully. The organic phase was dried over Na₂SO₄ and filtered. The filtrate

was concentrated under vacuum. The residue was purified by flash column chromatography on silica gel.

4.3. Characterization data

4.3.1. Ethyl 2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2a). Yield=63%. Orange solid. Mp 87–89 °C. ^1H NMR (300 MHz, CDCl_3) δ 7.95 (dd, $J=2.1$, 6.0 Hz, 1H), 7.09–7.00 (m, 3H), 4.20 (q, $J=7.0$ Hz, 2H), 3.90 (t, $J=7.2$ Hz, 2H), 2.46 (quint, $J=7.0$ Hz, 2H), 1.26 (t, $J=7.0$ Hz, 3H). ^1H NMR spectroscopic data is identical to those reported in the literature.^{3b}

4.3.2. Ethyl 7-methoxy-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2b). Yield=41%. Orange solid. Mp 114–116 °C. R_f (heptane/EtOAc 9:1) 0.53. IR (neat, cm^{-1}) ν 2900, 1673, 1606, 1578, 1479, 1377, 1301, 1229, 1132, 1041, 805, 796, 713. ^1H NMR (300 MHz, CDCl_3) δ 7.64 (d, $J=2.4$ Hz, 1H), 7.12 (d, $J=8.6$ Hz, 1H), 6.83 (dd, $J=2.4$, 8.6 Hz, 1H), 4.35 (q, $J=7.2$ Hz, 2H), 4.07 (t, $J=7.0$ Hz, 2H), 3.89 (s, 3H), 3.27 (t, $J=7.0$ Hz, 2H), 2.63 (quint, $J=7.0$ Hz, 2H), 1.41 (t, $J=7.2$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.8, 155.9, 153.1, 132.1, 128.0, 111.7, 110.7, 103.7, 99.3, 59.4, 56.0, 44.9, 26.8, 26.6, 14.9. HRMS m/z (ES+) calcd for $\text{C}_{15}\text{H}_{18}\text{NO}_3$: 260.1287, found 260.1300.

4.3.3. Diethyl 2,3-dihydro-1H-pyrrolo[1,2-a]indole-7,9-dicarboxylate (2c). Yield=63%. Yellow solid. Mp 130–131 °C. R_f (toluene/EtOAc 9:1) 0.41. IR (neat, cm^{-1}) ν 2980, 1706, 1685, 1424, 1269, 1198, 1109, 1024, 923, 835, 767, 677. ^1H NMR (300 MHz, CDCl_3) δ 8.81 (d, $J=1.2$ Hz, 1H), 7.90 (dd, $J=1.2$, 8.3 Hz, 1H), 7.20 (d, $J=8.3$ Hz, 1H), 4.43–4.33 (m, 4H), 4.09 (t, $J=7.2$ Hz, 2H), 4.27 (t, $J=7.2$ Hz, 2H), 2.65 (quint, $J=7.2$ Hz, 2H), 1.42 (t, $J=7.1$ Hz, 3H), 1.41 (t, $J=7.1$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 167.8, 165.2, 154.3, 135.2, 130.6, 124.1, 124.0, 123.5, 109.6, 100.7, 60.8, 59.7, 44.7, 26.8, 26.3, 14.8, 14.6. HRMS m/z (ES+) calcd for $\text{C}_{17}\text{H}_{19}\text{NO}_4\text{Na}$: 324.1212, found 324.1227.

4.3.4. Ethyl 7-cyano-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2d). Yield=46%. Orange solid. Mp 190–193 °C. R_f (toluene/EtOAc 8:2) 0.30. IR (neat, cm^{-1}) ν 2986, 2215, 1675, 1614, 1421, 1224, 1156, 1111, 888, 832, 783, 744. ^1H NMR (300 MHz, CDCl_3) δ 8.40 (d, $J=1.5$ Hz, 1H), 7.40 (dd, $J=1.5$, 8.4 Hz, 1H), 7.26 (d, $J=8.4$ Hz, 1H), 4.38 (q, $J=7.4$ Hz, 2H), 4.15 (t, $J=7.4$ Hz, 2H), 3.31 (t, $J=7.2$ Hz, 2H), 2.71 (quint, $J=7.2$ Hz, 2H), 1.43 (t, $J=7.4$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 164.7, 155.0, 134.3, 130.7, 126.8, 125.0, 120.7, 110.8, 104.7, 100.5, 60.0, 44.9, 26.8, 26.2, 14.8. HRMS m/z (ES+) calcd for $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_2\text{Na}$: 277.0953, found 277.0951.

4.3.5. Ethyl 7-methyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2e). Yield=51%. Orange solid. Mp 77–78 °C. ^1H NMR (300 MHz, CDCl_3) δ 7.91 (d, $J=1.4$ Hz, 1H), 7.12 (d, $J=8.0$ Hz, 1H), 7.01 (dd, $J=1.4$, 8.0 Hz, 1H), 4.36 (q, $J=7.2$ Hz, 2H), 4.06 (t, $J=7.0$ Hz, 2H), 3.26 (t, $J=7.0$ Hz, 2H), 2.62 (quint, $J=7.0$ Hz, 2H), 2.48 (s, 3H), 1.41 (t, $J=7.2$ Hz, 3H). ^1H NMR spectroscopic data is identical to those reported in the literature.^{3b}

4.3.6. Ethyl 7-chloro-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2f). Yield=46%. Orange solid. Mp 141–142 °C. R_f (toluene/EtOAc 9:1) 0.65. IR (neat, cm^{-1}) ν 2978, 1682, 1544, 1448, 1425, 1376, 1317, 1202, 1103, 1037, 886, 778. ^1H NMR (300 MHz, CDCl_3) δ 8.07 (m, 1H), 7.13 (m, 2H), 4.36 (q, $J=7.0$ Hz, 2H), 4.09 (t, $J=7.4$ Hz, 2H), 3.28 (t, $J=7.4$ Hz, 2H), 2.66 (quint, $J=7.4$ Hz, 2H), 1.41 (t, $J=7.0$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.3, 154.1, 132.1, 131.3, 127.7, 122.2, 121.3, 110.9, 99.5, 59.7, 44.9, 26.8, 26.4, 14.9. HRMS m/z (ES+) calcd for $\text{C}_{14}\text{H}_{14}\text{NO}_2\text{NaCl}$: 286.0611, found 286.0617.

4.3.7. Ethyl 6-methyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2g). Yield=45%. Orange solid. Mp 94–97 °C. R_f (toluene/EtOAc 9:1) 0.77. IR (neat, cm^{-1}) ν 2924, 1684, 1528, 1443, 1421, 1342,

1299, 1190, 1138, 1089, 762, 735. ^1H NMR (300 MHz, CDCl_3) δ 7.52 (s, 1H), 7.30–7.23 (m, 1H), 7.16 (d, $J=7.6$ Hz, 1H), 4.50 (q, $J=7.2$ Hz, 2H), 4.28 (t, $J=7.0$ Hz, 2H), 3.48 (t, $J=7.0$ Hz, 2H), 3.04 (s, 3H), 2.80 (quint, $J=7.0$ Hz, 2H), 1.57 (t, $J=7.2$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.4, 153.3, 133.6, 132.6, 129.5, 124.1, 122.1, 107.7, 100.8, 59.6, 44.9, 27.6, 26.4, 22.9, 14.8. HRMS m/z (ES+) calcd for $\text{C}_{15}\text{H}_{18}\text{NO}_2$: 244.1338, found 244.1341.

4.3.8. Ethyl 8-methyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2g'). Yield=38%. Orange solid. Mp 98–100 °C. R_f (toluene/EtOAc 9:1) 0.56. IR (neat, cm^{-1}) ν 2978, 1679, 1542, 1427, 1372, 1209, 1128, 1106, 1038, 811, 781, 744. ^1H NMR (500 MHz, CDCl_3) δ 7.97 (dd, $J=0.9$, 7.78 Hz, 1H), 7.05 (m, 2H), 4.35 (q, $J=7.1$ Hz, 2H), 4.06 (t, $J=7.0$ Hz, 2H), 3.26 (t, $J=7.0$ Hz, 2H), 2.63 (quint, $J=7.0$ Hz, 2H), 2.47 (s, 3H), 1.41 (t, $J=7.1$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.8, 152.5, 133.2, 131.7, 128.9, 123.3, 121.3, 110.1, 99.4, 59.4, 44.5, 26.8, 26.3, 21.8, 14.9. HRMS m/z (ES+) calcd for $\text{C}_{15}\text{H}_{17}\text{NO}_2\text{Na}$: 266.1157, found 266.1161.

4.3.9. Ethyl 5-methyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2h). Yield=18%. Orange solid. Mp 117–118 °C. R_f (toluene/EtOAc 8:2) 0.55. IR (neat, cm^{-1}) ν 2996, 1668, 1618, 1551, 1469, 1383, 1231, 1161, 1047, 838, 742. ^1H NMR (500 MHz, CDCl_3) δ 7.98 (d, $J=8.1$ Hz, 1H), 7.11 (t, $J=8.1$ Hz, 1H), 6.94 (d, $J=8.1$ Hz, 1H), 4.42 (t, $J=7.2$ Hz, 2H), 4.37 (q, $J=7.0$ Hz, 2H), 3.26 (t, $J=7.2$ Hz, 2H), 2.66 (s, 3H), 2.63 (quint, $J=7.2$ Hz, 2H), 1.43 (t, $J=7.0$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.8, 153.3, 132.5, 131.4, 123.6, 121.9, 120.9, 119.4, 99.5, 59.4, 47.8, 27.0, 25.8, 18.1, 14.9. HRMS m/z (ES+) calcd for $\text{C}_{15}\text{H}_{18}\text{NO}_2$: 244.1338, found 244.1339.

4.3.10. Ethyl 6,8-dimethyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2i). Yield=72%. Yellow solid. Mp 155–158 °C. R_f (toluene/EtOAc 9:1) 0.67. IR (neat, cm^{-1}) ν 2978, 1681, 1525, 1420, 1370, 1202, 1176, 1093, 1034, 841, 783, 748. ^1H NMR (500 MHz, CDCl_3) δ 6.83 (d, $J=2.2$ Hz, 2H), 4.31 (q, $J=7.1$ Hz, 2H), 4.00 (t, $J=7.0$ Hz, 2H), 3.24 (t, $J=7.0$ Hz, 2H), 2.83 (s, 3H), 2.56 (quint, $J=7.0$ Hz, 2H), 2.42 (s, 3H), 1.38 (t, $J=7.1$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.4, 152.8, 133.9, 132.1, 131.8, 127.2, 125.7, 107.7, 100.4, 59.4, 44.6, 27.5, 26.2, 22.7, 21.5, 14.8. HRMS m/z (ES+) calcd for $\text{C}_{16}\text{H}_{19}\text{NO}_2\text{Na}$: 280.1313, found 280.1319.

4.3.11. Ethyl 9,10-dihydro-8H-benzo[e]pyrrolo[1,2-a]indole-11-carboxylate (2j). Yield=54%. Orange solid. Mp 169–172 °C. R_f (toluene/EtOAc 9:1) 0.65. IR (neat, cm^{-1}) ν 2987, 1706, 1682, 1453, 1379, 1264, 1139, 1027, 800, 768, 685. ^1H NMR (300 MHz, CDCl_3) δ 9.80 (d, $J=8.7$ Hz, 1H), 7.89 (d, $J=7.9$ Hz, 1H), 7.62–7.56 (m, 2H), 7.45 (t, $J=7.9$ Hz, 1H), 7.27 (d, $J=7.9$ Hz, 1H), 4.38 (q, $J=7.2$ Hz, 2H), 4.03 (t, $J=7.1$ Hz, 2H), 3.18 (t, $J=7.1$ Hz, 2H), 2.48 (quint, $J=7.1$ Hz, 2H), 1.43 (t, $J=7.2$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.8, 151.4, 130.6, 129.8, 129.1, 128.5, 127.2, 125.7, 125.2, 123.9, 123.8, 111.2, 102.5, 59.7, 45.0, 28.0, 26.0, 14.8. HRMS m/z (ES+) calcd for $\text{C}_{18}\text{H}_{17}\text{NO}_2\text{Na}$: 302.1157, found 302.1167.

4.3.12. Ethyl 2,3-dihydro-1H-benzo[f]pyrrolo[1,2-a]indole-11-carboxylate (2j'). Yield=18%. Red solid. Mp 160–164 °C. R_f (heptane/EtOAc 9:1) 0.51. IR (neat, cm^{-1}) ν ^1H NMR (500 MHz, CDCl_3) δ 8.59 (s, 1H), 8.02–7.99 (m, 1H), 7.91–7.88 (m, 1H), 7.64 (s, 1H), 7.40–7.37 (m, 2H), 4.42 (q, $J=7.2$ Hz, 2H), 4.19 (t, $J=6.9$ Hz, 2H), 3.38 (t, $J=6.9$ Hz, 2H), 2.72 (quint, $J=6.9$ Hz, 2H), 1.46 (t, $J=7.2$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 157.9, 133.7, 132.0, 130.1, 129.9, 128.8, 127.5, 124.3, 123.5, 119.4, 105.7, 59.6, 44.7, 29.9, 26.8, 15.0. HRMS m/z (ES+) calcd for $\text{C}_{18}\text{H}_{17}\text{NO}_2\text{Na}$: 302.1157, found 302.1167.

4.3.13. Ethyl 2,2-dimethyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2k). Yield=70%. White solid. Mp 83–84 °C. R_f (toluene/EtOAc 9:1) 0.58. IR (neat, cm^{-1}) ν 2954, 1674, 1543, 1431, 1208, 1102, 1042, 1012, 922, 784, 747, 736. ^1H NMR (500 MHz, CDCl_3)

δ 8.13 (d, $J=7.6$ Hz, 1H), 7.27–7.19 (m, 3H), 4.38 (q, $J=7.1$ Hz, 2H), 3.85 (s, 2H), 3.12 (s, 2H), 1.43 (t, $J=7.1$ Hz, 3H), 1.34 (s, 6H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.8, 152.3, 133.2, 130.6, 121.9, 109.9, 100.2, 59.5, 57.8, 43.7, 41.7, 28.2, 14.9. HRMS m/z (ES+) calcd for $\text{C}_{16}\text{H}_{19}\text{NO}_2\text{Na}$: 280.1313, found 280.1321.

4.3.14. Diethyl 2,2,6,8-tetramethyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-7,9-dicarboxylate (2l). Yield=69%. Yellow solid. Mp 100–102 °C. R_f (toluene/EtOAc 9:1) 0.80. IR (neat, cm^{-1}) ν 2960, 1698, 1536, 1415, 1339, 1168, 1086, 1038, 820, 782. ^1H NMR (500 MHz, CDCl_3) δ 6.84 (s, 2H), 4.32 (q, $J=7.0$ Hz, 2H), 3.78 (s, 2H), 3.09 (s, 2H), 2.84 (s, 3H), 2.43 (s, 3H), 1.39 (t, $J=7.0$ Hz, 3H), 1.32 (s, 6H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.4, 152.0, 134.2, 132.1, 131.8, 126.7, 125.6, 107.6, 101.1, 59.5, 57.8, 42.9, 42.9, 28.2, 22.7, 21.5, 14.8. HRMS m/z (ES+) calcd for $\text{C}_{18}\text{H}_{24}\text{NO}_2$: 286.1807, found 286.1795.

4.3.15. Ethyl 2,2,6,8-tetramethyl-2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2m). Yield=77%. Yellow solid. Mp 100–102 °C. R_f (toluene/EtOAc 9:1) 0.80. IR (neat, cm^{-1}) ν 2960, 1698, 1536, 1415, 1339, 1168, 1086, 1038, 820, 782. ^1H NMR (500 MHz, CDCl_3) δ 6.84 (s, 2H), 4.32 (q, $J=7.0$ Hz, 2H), 3.78 (s, 2H), 3.09 (s, 2H), 2.84 (s, 3H), 2.43 (s, 3H), 1.39 (t, $J=7.0$ Hz, 3H), 1.32 (s, 6H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.4, 152.0, 134.2, 132.1, 131.8, 126.7, 125.6, 107.6, 101.1, 59.5, 57.8, 42.9, 42.9, 28.2, 22.7, 21.5, 14.8. HRMS m/z (ES+) calcd for $\text{C}_{18}\text{H}_{24}\text{NO}_2$: 286.1807, found 286.1795.

4.3.16. Methyl 2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2n). Yield=51%. White solid. Mp 75–78 °C. ^1H NMR (300 MHz, CDCl_3) δ 8.12–8.09 (m, 1H), 7.26–7.19 (m, 3H), 4.07 (t, $J=7.2$ Hz, 2H), 3.90 (s, 3H), 3.27 (t, $J=7.2$ Hz, 2H), 2.63 (quint, $J=7.2$ Hz, 2H). ^1H NMR spectroscopic data is identical to those reported in the literature.^{3b}

4.3.17. Isopropyl 2,3-dihydro-1H-pyrrolo[1,2-a]indole-9-carboxylate (2o). Yield=47%. Orange solid. Mp 94–96 °C. R_f (toluene/EtOAc 9:1) 0.62. IR (neat, cm^{-1}) ν 2978, 1678, 1543, 1425, 1382, 1300, 1203, 1088, 1017, 783, 750. ^1H NMR (300 MHz, CDCl_3) δ 8.03 (dd, $J=1.9$, 8.3 Hz, 1H), 7.12–7.04 (m, 3H), 5.13 (sept, $J=6.2$ Hz, 1H), 3.94 (t, $J=7.1$ Hz, 2H), 3.14 (t, $J=7.1$ Hz, 2H), 2.50 (quint, $J=7.1$ Hz, 2H), 1.25 (d, $J=6.2$ Hz, 6H). ^{13}C NMR (75 MHz, CDCl_3) δ 165.3, 152.9, 132.9, 131.1, 121.8, 121.7, 121.6, 110.0, 99.9, 66.6, 44.6, 26.8, 26.3, 22.6. HRMS m/z (ES+) calcd for $\text{C}_5\text{H}_{17}\text{NO}_2\text{Na}$: 266.1157, found 266.1150.

4.3.18. Ethyl 2-(1-phenyl-1H-pyrrol-2-yl)acetate (5). Yield=8%. Orange oil. R_f (heptane/EtOAc 8:2) 0.51. IR (neat, cm^{-1}) ν 1732, 1598, 1460, 1368, 1324, 1324, 1257, 1178, 1141, 1031, 768, 697, 676. ^1H NMR (500 MHz, CDCl_3) δ 7.44 (t, $J=7.0$ Hz, 2H), 7.37 (t, $J=7.0$ Hz, 1H), 7.32 (d, $J=7.0$ Hz, 2H), 6.80 (m, 1H), 6.26 (t, $J=2.7$ Hz, 1H), 6.24 (m, 1H), 4.06 (q, $J=7.4$ Hz, 2H), 3.59 (s, 2H), 1.17 (t, $J=7.4$ Hz, 3H). ^{13}C NMR (75 MHz, CDCl_3) δ 170.9, 140.1, 129.4, 127.6, 126.5, 125.7, 122.7, 110.0, 108.6, 61.1, 33.0, 14.3. HRMS m/z (ES+) calcd for $\text{C}_{20}\text{H}_{12}\text{NO}_2$: 284.0837, found 284.0873.

Acknowledgements

Financial support from EPFL (Switzerland), Swiss National Science Foundation (SNF), Swiss National Centres of Competence in Research (NCCR) and ICSN, CNRS (France) are gratefully acknowledged. T.P. thanks ICSN and EPFL for a doctoral fellowship.

Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.tet.2013.01.003>.

References and notes

- (a) Orlemans, E. O. M.; Verboom, W.; Scheltinga, M. W.; Reinhoudt, D. N.; Lelieveld, P.; Fiebig, H. H.; Winterhalter, B. R.; Double, J. A.; Bibby, M. C. *J. Med. Chem.* **1989**, *32*, 1612–1620; (b) Galm, U.; Hager, M. H.; Van Lanen, S. G.; Ju, J.; Thorson, J. S.; Shen, B. *Chem. Rev.* **2005**, *105*, 739–758; (c) Vepsäläinen, J. J.; Auriola, S.; Tukiainen, M.; Ropponen, N.; Callaway, J. C. *Planta Med.* **2005**, *71*, 1053–1057; (d) Sasase, T.; Yamada, H.; Sakoda, K.; Imagawa, N.; Abe, T.; Ito, M.; Sagawa, S.; Tanaka, M.; Matsushita, M. *Diabetes, Obes. Metab.* **2005**, *7*, 586–594; (e) Fernandez, L. S.; Buchanan, M. S.; Carroll, A. R.; Feng, Y. J.; Quinn, R. J.; Avery, V. M. *Org. Lett.* **2009**, *11*, 329–332.
- (a) Padwa, A.; Fryxell, G. E.; Gasdaska, J. R.; Venkatramanan, M. K.; Wong, G. S. K. *J. Org. Chem.* **1989**, *54*, 644–653; (b) Chen, H. G.; Hoeschetter, C.; Knochel, P. *Tetrahedron Lett.* **1989**, *30*, 4795–4798; (c) Caddick, S.; Aboutayab, K.; West, R. I. *J. Chem. Soc., Chem. Commun.* **1995**, 1353–1354; (d) Caddick, S.; Aboutayab, K.; Jenkins, K.; West, R. I. *J. Chem. Soc., Perkin Trans. 1* **1996**, 675–682; (e) Gilchrist, T. L.; Kemmit, P. D.; Germain, A. L. *Tetrahedron* **1997**, *53*, 4447–4456; (f) Yavari, I.; Adib, M.; Sayahi, M. H. *J. Chem. Soc., Perkin Trans. 1* **2002**, 1517–1519; (g) Ferreira, E. M.; Stoltz, B. M. *J. Am. Chem. Soc.* **2003**, *125*, 9578–9579; (h) Borah, N.; Deb, M. L.; Boruah, R. C.; Bhuyan, P. J. *Tetrahedron Lett.* **2005**, *46*, 3391–3393; (i) Wilson, R. M.; Thalji, R. K.; Bergman, R. G.; Ellman, J. A. *Org. Lett.* **2006**, *8*, 1745–1747; (j) Manian, R. D. R. S.; Jayashankaran, J.; Raghunathan, R. *Synlett* **2007**, 874–880; (k) Johansen, M. B.; Kerr, M. A. *Org. Lett.* **2008**, *10*, 3497–3500; (l) Enders, D.; Wang, C.; Raabe, G. *Synthesis* **2009**, 4119–4124; (m) Bunce, R. A.; Nammalwar, B. *J. Heterocycl. Chem.* **2009**, *46*, 172–177; (n) Lu, S.-C.; Duan, X.-Y.; Shi, Z.-J.; Li, B.; Ren, Y.-W.; Zhang, W.; Zhang, Y.-H.; Tu, Z.-F. *Org. Lett.* **2009**, *11*, 3902–3905; (o) Hong, L.; Sun, W.; Liu, C.; Wang, L.; Wang, R. *Chem.—Eur. J.* **2010**, *16*, 440–444; (p) Patil, D. V.; Cavitt, M. A.; France, S. *Org. Lett.* **2011**, *13*, 5820–5823; (q) Lu, S.-C.; Wei, S.-C.; Wang, W.-X.; Zhang, W.; Tu, Z.-F. *Eur. J. Org. Chem.* **2011**, 5905–5910; (r) Mi, Z.; Wei, Z.; Shuli, Y. *Chin. J. Chem.* **2012**, *30*, 2615–2623.
- For one step synthesis of pyrrolo[1,2-a]indole cores, see: (a) Coates, R. M.; Hutchins, C. W. *J. Org. Chem.* **1979**, *44*, 4742–4744; (b) Michael, J. P.; Chang, S.-F.; Wilson, C. *Tetrahedron Lett.* **1993**, *34*, 8365–8368; (c) Siebeneicher, H.; Bytschkov, I.; Doye, S. *Angew. Chem., Int. Ed.* **2003**, *42*, 3042–3044; (d) Takaya, J.; Udagawa, S.; Kusama, H.; Iwasawa, N. *Angew. Chem., Int. Ed.* **2008**, *47*, 4906–4909; (e) Fuchibe, K.; Kaneko, T.; Mori, K.; Akiyama, T. *Angew. Chem., Int. Ed.* **2009**, *48*, 8070–8073; (f) Gao, D.; Parvez, M.; Back, T. G. *Chem.—Eur. J.* **2010**, *16*, 14281–14284; (g) Takaya, J.; Miyashita, Y.; Kusama, H.; Iwasawa, N. *Tetrahedron* **2011**, *67*, 4455–4466; (h) Kim, J. H.; Lee, S.-g. *Org. Lett.* **2011**, *13*, 1350–1353; (i) Liu, L.; Wang, Y.; Zhang, L. *Org. Lett.* **2012**, *14*, 3736–3739.
- For selected reviews on the synthesis of indoles, see: (a) Gribble, G. W. *J. Chem. Soc., Perkin Trans. 1* **2000**, 1045–1075; (b) Gilchrist, T. L. *J. Chem. Soc., Perkin Trans. 1* **2001**, 2491–2515; (c) Zeni, G.; Larock, R. C. *Chem. Rev.* **2004**, *104*, 2285–2310; (d) Cacchi, S.; Fabrizi, G. *Chem. Rev.* **2005**, *105*, 2873–2920; (e) Humphrey, G. R.; Kuethe, J. T. *Chem. Rev.* **2006**, *106*, 2875–2911; (f) Ackermann, L. *Synlett* **2007**, 507–526; (g) Krüger (née Alex), K.; Tillack, A.; Beller, M. *Adv. Synth. Catal.* **2008**, *350*, 2153–2167; (h) Patil, N. T.; Yamamoto, Y. *Chem. Rev.* **2008**, *108*, 3395–3442; (i) Barluenga, J.; Rodríguez, F.; Fañanás, F. *J. Chem.—Asian J.* **2009**, *4*, 1036–1048; (j) Palmisano, G.; Penoni, A.; Sisti, M.; Tibiletti, F.; Tollari, S.; Nicholas, K. M. *Curr. Org. Chem.* **2010**, *14*, 2409–2441; (k) Patil, S. A.; Patil, R.; Miller, D. D. *Curr. Med. Chem.* **2011**, *18*, 615–637; (l) Cacchi, S.; Fabrizi, G.; Goggiamani, G. *Org. Biomol. Chem.* **2011**, *9*, 641–652; (m) Cacchi, S.; Fabrizi, G. *Chem. Rev.* **2011**, *111*, PR215–PR283; (n) Taber, D. F.; Tirunahari, P. K. *Tetrahedron* **2011**, *67*, 7195–7210.
- (a) Larock, R. C.; Yum, E. K. *J. Am. Chem. Soc.* **1991**, *113*, 6689–6690; (b) Larock, R. C.; Yum, E. K.; Refvik, M. D. *J. Org. Chem.* **1998**, *63*, 7652–7662; (c) Roesch, K. R.; Larock, R. C. *J. Org. Chem.* **2001**, *66*, 412–420 From *ortho*-chloroaniline, see: (d) Shen, M.; Li, G.; Lu, B. Z.; Hossain, A.; Roschangar, F.; Farina, V.; Senanayake, C. H. *Org. Lett.* **2004**, *6*, 4129–4132.
- (a) Liu, J.; Shen, M.; Zhang, Y.; Li, G.; Khodabocus, A.; Rodríguez, S.; Qu, B.; Farina, V.; Senanayake, C. H.; Lu, B. Z. *Org. Lett.* **2006**, *8*, 3573–3575; (b) Liu, J.; Zhang, Y.; Li, G.; Roschangar, F.; Farina, V.; Senanayake, C. H.; Lu, B. Z. *Adv. Synth. Catal.* **2010**, *352*, 2667–2671.
- For a general review on the domino process, see: Tietze, L. F. *Chem. Rev.* **1996**, *96*, 115–136.
- For recent reviews on C–H activation, see: (a) Dick, A. R.; Sanford, M. S. *Tetrahedron* **2006**, *62*, 2439–2463; (b) Alberico, D.; Scott, M. E.; Lautens, M. *Chem. Rev.* **2007**, *107*, 174–238; (c) Li, B.-J.; Yang, S.-D.; Shi, Z.-J. *Synlett* **2008**, 949–957; (d) Giri, R.; Shi, B.-F.; Engle, K. M.; Maugel, N.; Yu, J.-Q. *Chem. Soc. Rev.* **2009**, *38*, 3242–3272; (e) Wencel-Delord, J.; Dröge, T.; Liu, F.; Glorius, F. *Chem. Soc. Rev.* **2011**, *40*, 4740–4761; (f) Baudoin, O. *Chem. Soc. Rev.* **2011**, *40*, 4902–4911 For C–H functionalization in natural product synthesis, see: (g) McMurray, L.; O'Hara, F.; Gaunt, M. J. *Chem. Soc. Rev.* **2011**, *40*, 1885–1898; (h) Gutekunst, W. R.; Baran, P. S. *Chem. Soc. Rev.* **2011**, *40*, 1976–1991.
- (a) Cuny, G.; Bois-Choussy, M.; Zhu, J. *Angew. Chem., Int. Ed.* **2003**, *42*, 4774–4777; (b) Pinto, A.; Neuville, L.; Zhu, J. *Angew. Chem., Int. Ed.* **2007**, *46*, 3291–3295; (c) Salcedo, A.; Neuville, L.; Rondot, C.; Retailleau, P.; Zhu, J. *Org. Lett.* **2008**, *10*, 857–860; (d) Gerfaut, T.; Neuville, L.; Zhu, J. *Angew. Chem., Int. Ed.* **2009**, *48*, 572–577; (e) Pinto, A.; Neuville, L.; Zhu, J. *Tetrahedron Lett.* **2009**, *50*, 3602–3605; (f) Yao, B.; Jaccoud, C.; Wang, Q.; Zhu, J. *Chem.—Eur. J.* **2012**, *18*, 5864–5868; (g) Piou, T.; Neuville, L.; Zhu, J. *Org. Lett.* **2012**, *14*, 3760–3763; (h) Piou, T.; Neuville, L.; Zhu, J. *Angew. Chem., Int. Ed.* **2012**, *51*, 11561–11565.
- (a) Würtz, S.; Rakshit, S.; Neumann, J. J.; Dröge, T.; Glorius, F. *Angew. Chem., Int. Ed.* **2008**, *47*, 7230–7233; (b) Neumann, J. J.; Souvik, R.; Dröge, T.; Würtz, S.; Glorius, F. *Chem.—Eur. J.* **2011**, *17*, 7298–7303.

11. (a) Wei, Y.; Deb, I.; Yoshikai, N. *J. Am. Chem. Soc.* **2012**, *134*, 9098–9101 For copper-catalyzed cyclization of *N*-aryl enaminone, see: (b) Bernini, R.; Fabrizi, G.; Sferrazza, A.; Cacchi, S. *Angew. Chem., Int. Ed.* **2009**, *48*, 8078–8081 For iron-catalyzed cyclization of *N*-aryl enaminone, see: (c) Guan, Z.-H.; Yan, Z.-Y.; Ren, Z.-H.; Liu, X.-Y.; Liang, Y.-M. *Chem. Commun.* **2010**, 2823–2825.
12. Shi, Z.; Zhang, C.; Li, S.; Pan, D.; Ding, S.; Cui, Y.; Jiao, N. *Angew. Chem., Int. Ed.* **2009**, *48*, 4572–4576.
13. (a) Satoh, T.; Miura, M. *Chem.—Eur. J.* **2010**, *16*, 11212–11222; (b) Song, G.; Wang, F.; Li, X. *Chem. Soc. Rev.* **2012**, *41*, 3651–3678.
14. For selected examples, see: (a) Stuart, D. R.; Bertrand-Laperle, M.; Burgess, K. M. N.; Fagnou, K. *J. Am. Chem. Soc.* **2008**, *130*, 16474–16475; (b) Guimond, N.; Fagnou, K. *J. Am. Chem. Soc.* **2009**, *131*, 12050–12051; (c) Fukutani, T.; Umeda, N.; Hirano, K.; Satoh, T.; Miura, M. *Chem. Commun.* **2009**, 5141–5143; (d) Guimond, N.; Gouliaras, C.; Fagnou, K. *J. Am. Chem. Soc.* **2010**, *132*, 6908–6909; (e) Mochida, S.; Umeda, N.; Hirano, K.; Satoh, T.; Miura, M. *Chem. Lett.* **2010**, *39*, 744–746; (f) Hyster, T. K.; Rovis, T. *J. Am. Chem. Soc.* **2010**, *132*, 10565–10569; (g) Song, G.; Chen, D.; Pan, C.-L.; Crabtree, R. H.; Li, X. *J. Org. Chem.* **2010**, *75*, 7487–7490; (h) Su, Y.; Zhao, M.; Han, K.; Song, G.; Li, X. *Org. Lett.* **2010**, *12*, 5462–5465; (i) Rakshit, S.; Patureau, F. W.; Glorius, F. *J. Am. Chem. Soc.* **2010**, *132*, 9585–9587; (j) Stuart, D. R.; Alsabeth, P.; Kuhn, M.; Fagnou, K. *J. Am. Chem. Soc.* **2010**, *132*, 18326–18339; (l) Ackermann, L.; Lygin, A. V. *Org. Lett.* **2012**, *14*, 764–767; (m) Pham, M. V.; Ye, B.; Cramer, N. *Angew. Chem., Int. Ed.* **2012**, *51*, 10610–10614.
15. (a) Yip, K.-T.; Yang, M.; Law, K.-L.; Zhu, N.-Y.; Yang, D. *J. Am. Chem. Soc.* **2006**, *128*, 3130–3131; (b) Tang, S.; Peng, P.; Pi, S.-F.; Liang, Y.; Wang, N.-X.; Li, J.-H. *Org. Lett.* **2008**, *10*, 1179–1182; (c) Shi, Z.; Cui, Y.; Jiao, N. *Org. Lett.* **2010**, *12*, 2908–2911; (d) Zhong, H.; Yang, D.; Wang, S.; Huang, J. *Chem. Commun.* **2012**, 3236–3238.
16. Oxidative aminopalladation using tertiary amines as nucleophiles: (a) Yao, B.; Wang, Q.; Zhu, J. *Angew. Chem., Int. Ed.* **2012**, *51*, 5170–5174; (b) Yao, B.; Wang, Q.; Zhu, J. *Angew. Chem., Int. Ed.* **2012**, *51*, 12311–12315; (c) Xia, X.-F.; Wang, N.; Zhang, L.-L.; Song, X.-R.; Liu, X.-Y.; Liang, Y.-M. *J. Org. Chem.* **2012**, *77*, 9163–9170.
17. Synthesized from pent-4-ynoic acid and aniline (see Supplementary data).
18. Lafrance, M.; Fagnou, K. *J. Am. Chem. Soc.* **2006**, *128*, 16496–16497.
19. For reviews, see: (a) Stahl, S. S. *Angew. Chem., Int. Ed.* **2004**, *43*, 3400–3420; (b) Punniyamurthy, T.; Velusamy, S.; Iqbal, J. *Chem. Rev.* **2005**, *105*, 2329–2364; (c) Beccalli, E. M.; Broggini, G.; Martinelli, M.; Sottocornola, S. *Chem. Rev.* **2007**, *107*, 5318–5365; (d) Stahl, S. S. *Science* **2005**, *309*, 1824–1826; (e) Stoltz, B. M. *Chem. Lett.* **2004**, *33*, 362–367; (f) Schultz, M. J.; Sigman, M. S. *Tetrahedron* **2006**, *62*, 8227–8241; (g) Muzart, J. *Chem.—Asian J.* **2006**, *1*, 508–515; (h) Sigman, M. S.; Jensen, D. R. *Acc. Chem. Res.* **2006**, *39*, 221–229; (i) Beccalli, E. M.; Broggini, G.; Martinelli, M.; Sottocornola, S. *Chem. Rev.* **2007**, *107*, 5318–5365; (j) Piera, J.; Bäckvall, J. E. *Angew. Chem., Int. Ed.* **2008**, *47*, 3506–3523; (k) Gligorich, K. M.; Sigman, M. S. *Chem. Commun.* **2009**, 3854–3867; (l) Wendlandt, A. E.; Suess, A. M.; Stahl, S. S. *Angew. Chem., Int. Ed.* **2011**, *50*, 11062–11087; (m) Shi, Z.; Zhang, C.; Tang, C.; Jiao, N. *Chem. Soc. Rev.* **2012**, *41*, 3381–3430.
20. Synthesized from 1-phenylpyrrolidin-2-one (see Supplementary data). Michael, J. P.; de Koning, C. B.; Hosken, G. D.; Stanbury, T. V. *Tetrahedron* **2001**, *57*, 9635–9648.
21. Utimoto, K. *Pure Appl. Chem.* **1983**, *55*, 1845–1852.
22. For selected examples of aminopalladation, see: (a) Overman, L. E.; Remarchuk, T. P. *J. Am. Chem. Soc.* **2002**, *124*, 12–13; (b) Ney, J. E.; Wolfe, J. P. *J. Am. Chem. Soc.* **2005**, *127*, 8644–8651; (c) Brice, J. L.; Harang, J. E.; Timokhin, V. I.; Anastasi, N. R.; Stahl, S. S. *J. Am. Chem. Soc.* **2005**, *127*, 2868–2869; (d) Jaegli, S.; Dufour, J.; Wei, H.-L.; Piou, T.; Duan, X.-H.; Vors, J.-P.; Neuville, L.; Zhu, J. *Org. Lett.* **2010**, *12*, 4498–4501; (e) Jaegli, S.; Erb, W.; Retailleau, P.; Vors, J. P.; Neuville, L.; Zhu, J. *Chem.—Eur. J.* **2010**, *16*, 5863–5867.