

View Article Online View Journal

ChemComm

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: G. Dias, B. Rodrigues, J. M. Resende, H. Calado, C. Simone, V. Silva, B. A. D. Neto, M.O.F. Goulart, F. Ferreira, A. Meira, C. Pessoa, J. Corrêa and E. Silva Junior, *Chem. Commun.*, 2015, DOI: 10.1039/C5CC02383A.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

Journal Name

RSCPublishing

View Article Online DOI: 10.1039/C5CC02383A

Cite this: DOI: 10.1039/x0xx00000x

Received 00th January 2014, Accepted 00th January 2014

DOI: 10.1039/x0xx00000x

www.rsc.org/

Gleiston G. Dias,^{*a*} Bernardo L. Rodrigues,^{*a*} Jarbas M. Resende,^{*a*} Hállen D. R. Calado,^{*a*} Carlos A. de Simone,^{*b*} Valter H. C. Silva,^{*c*} Brenno A. D. Neto,^{*d*} Marilia O. F. Goulart,^{*e*} Fabricia R. Ferreira,^{*e*} Assuero S. Meira,^{*f*} Claudia Pessoa,^{*f*,*g*} José R. Correa*^{*d*} and Eufrânio N. da Silva Júnior*^{*a*}

Selective Endocytic Trafficking in Live Cells with

Fluorescent Naphthoxazoles and their Boron

Fluorescent naphthoxazoles and their boron derivatives have been synthesized and applied as superior and selective probes for endocytic pathway tracking in live cancer cells. The best fluorophores were compared with the commercially available acridine orange (co-staining experiments), showing far better selectivity.

Complexes

Endocytosis, as the key regulator process for molecular internalization in eukaryotic cells, has a fundamental role in cell homeostasis.1 Signal transduction, neurotransmission, intercellular communication and immune response are known to be readily affected by endocytic pathway dysfunction.² The complexity of the cellular environment renders the understanding of endocytosis as an outstanding challenge. Cellular homeostasis, for instance, depends on the proper communication of the cellular organelles and all other components associated with correct signals throughout the highly complex machinery. Deregulation in any of these signalizing pathways could be the answer which triggers a specific pathological process. All homeostasis unbalances may also be associated with endocytosis dysfunctions. Fluorescence techniques, in this sense, rose up, shedding light onto many doubts over cellular endocytic mechanisms and pathways.³ Despite the great advance of using fluorescent bioprobes,⁴⁻⁷ many unanswered questions and challenges still remain to be tackled. For some, the capacity of a bioprobe to transpose the cell membrane is one of the most desirable features.⁸ Very recently, some have shown the great potential of fluorogenic probes for studying dynamic cellular uptake in live cells.⁹

With the knowledge that many quinonoid derivatives sustaining the basic carbon scaffold are capable of transposing the cell-membrane and can present luminescence, as we¹⁰⁻¹² and others¹³⁻¹⁵ have already demonstrated, we envisaged the synthesis of fluorescent bioprobes from an affordable naturally occurring naphthoquinone. In this sense, we aimed to use the cell membrane permeability feature of such compounds to synthesize fluorescent derivatives for dynamics and uptake in live cells using the endocytic pathway. Herein, we disclose the rational

design, synthesis, characterization, single crystal X-ra, photophysics, DFT calculations, bioimaging, cellular uptake and dynamics in living cells (cancer cells) of new fluorescent derivatives which have allowed for selectively visualizing the whole endocytic pathway.

The structures of the synthesized fluorescent compounds ... visualized in Figure 1 and the synthetic details are found in the supplementary material (Scheme S1).



Figure 1. Structures of the fluorescent compounds evaluated in this work. Inset: fluorescence images under UV light (365 nm) in CH_2Cl_2 (50 μ M).

All compounds afforded single crystals suitable for X-ra characterization (Figure 2 and Tables S5 and S6) with different packing depending on the size of the side ring.



Figure 2. Molecular structure of 1-4 (top) and packing (bottom) depicted fro single crystal X-ray analysis.

This journal is © The Royal Society of Chemistry 2014

Molecular structures of 1 and 2 are basically planar except for the side pyrane ring. Compound 4 is in turn perfectly planar with methyl groups and fluorine atoms out of the plane. Distinct crystal packing for all compounds are also noted when comparing the four structures (Figure 2).

The photophysical properties of these molecules were investigated and pointed to the stability of these compounds in the excited state, as shown in Table 1.

Table 1. UV-Vis and fluorescence emission data (in different solvents) for 1, 2, 3 and 4 (10 μ M of concentration).

(nm) (nm) (nm) Hexane 348 396 48 CH ₃ Ph 353 402 49 CH ₂ Cl ₂ 351 407 56 EtOAc 350 403 53 MeOH 348 470 122 MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 EtOAc 350 477 127 MeOH 348 461 113 DMSO 352 412 60 MeCN 348 461 113 DMSO 352 412 60 MeCN 348 461 113 DMSO 355 408 53 GH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414	Compound	Solvent	λ_{max} (abs)	λ_{max} (em)	Stokes shift
Hexane 348 396 48 CH ₃ Ph 353 402 49 CH ₂ Cl ₂ 351 407 56 EtOAc 350 403 53 MeOH 348 470 122 MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 416 62 DMSO <td></td> <td></td> <td>(nm)</td> <td>(nm)</td> <td>(nm)</td>			(nm)	(nm)	(nm)
CH ₃ Ph 353 402 49 CH ₂ Cl ₂ 351 407 56 EtOAc 350 403 53 MeOH 348 470 122 MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 416 62 DMSO 359 419 60 Hexane <td rowspan="7">1</td> <td>Hexane</td> <td>348</td> <td>396</td> <td>48</td>	1	Hexane	348	396	48
CH ₂ Cl ₂ 351 407 56 I EtOAc 350 403 53 MeOH 348 470 122 MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 2 EtOAc 350 477 127 MeOH 348 469 121 MeOH 348 461 113 DMSO 352 412 60 MeCN 348 461 113 DMSO 352 412 60 407 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeOH 354 416 62 DMSO 359 419		CH ₃ Ph	353	402	49
1 EtOAc 350 403 53 MeOH 348 470 122 MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 2 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeOH 354 416 62 DMSO 359 419 60 Hexane 373 427 54		CH_2Cl_2	351	407	56
MeOH 348 470 122 MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 2 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeON 354 416 62 DMSO 359 419 60 <td< td=""><td>EtOAc</td><td>350</td><td>403</td><td>53</td></td<>		EtOAc	350	403	53
MeCN 348 412 64 DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 2 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 MeCN 348 461 113 DMSO 352 412 60 4 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 G Hexane 354 410 56 MeOH 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 67 <		MeOH	348	470	122
DMSO 353 414 61 Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 GHOAc 355 408 53 MeOH 354 410 56 MeOH 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 67 CH ₃ Ph 375 411 56 MeOH 355 <td>MeCN</td> <td>348</td> <td>412</td> <td>64</td>		MeCN	348	412	64
Hexane 348 421 73 CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 BtOAc 355 408 53 MeOH 354 410 56 MeOH 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 67 CH ₃ Ph 377 442 65 4 EtOAc 355 411 56 MeOH <td>DMSO</td> <td>353</td> <td>414</td> <td>61</td>		DMSO	353	414	61
CH ₃ Ph 353 433 80 CH ₂ Cl ₂ 371 445 74 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 BtOAc 355 408 53 MeOH 354 410 56 MeOH 354 416 62 DMSO 359 419 60 MeON 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 67 CH ₃ Ph 375 411 56 MeOH 355	2	Hexane	348	421	73
CH2Cl2 371 445 74 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH3Ph 358 409 51 CH2Cl2 357 414 57 BtOAc 355 408 53 MeOH 354 410 56 MeOH 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH3Ph 378 445 67 CH3Ph 378 445 67 CH3Ph 375 411 56 MeOH 355 412 57 MeOH 355 412 57 MeOH 355 412 57		CH ₃ Ph	353	433	80
2 EtOAc 350 477 127 MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 BtOAc 355 408 53 MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 65 Hexane 355 411 56 MeOH 355 412 57 MeOH 355 412 57 MeOH 355 412 57		CH ₂ Cl ₂	371	445	74
MeOH 348 469 121 MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 GH ₂ Cl ₂ 357 414 57 GHOH 354 410 56 MeOH 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 65 Hexane 355 411 56 MeOH 355 412 57 MeOH 355 412 57		EtOAc	350	477	127
MeCN 348 461 113 DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₃ Ph 378 445 65 EtOAc 355 411 56 MeOH 355 412 57 MeOH 355 412 57		MeOH	348	469	121
DMSO 352 412 60 Hexane 354 398 44 CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57		MeCN	348	461	113
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		DMSO	352	412	60
CH ₃ Ph 358 409 51 CH ₂ Cl ₂ 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57	3	Hexane	354	398	44
CH2Cl2 357 414 57 3 EtOAc 355 408 53 MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH3Ph 378 445 67 CH2Cl2 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57		CH ₃ Ph	358	409	51
3 EtOAc 355 408 53 MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 MeOH 355 411 56 MeOH 355 412 57		CH_2Cl_2	357	414	57
MeOH 354 410 56 MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57		EtOAc	355	408	53
MeCN 354 416 62 DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57		MeOH	354	410	56
DMSO 359 419 60 Hexane 373 427 54 CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57		MeCN	354	416	62
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		DMSO	359	419	60
CH ₃ Ph 378 445 67 CH ₂ Cl ₂ 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57 MaCN 355 422 67	4	Hexane	373	427	54
CH2Cl2 377 442 65 4 EtOAc 355 411 56 MeOH 355 412 57 MaCN 355 422 67		CH ₃ Ph	378	445	67
4 EtOAc 355 411 56 MeOH 355 412 57 MaCN 355 422 67		CH_2Cl_2	377	442	65
MeOH 355 412 57 MaCN 355 422 67		EtOAc	355	411	56
MaCN 255 422 67		MeOH	355	412	57
WIECIN 555 422 07		MeCN	355	422	67
DMSO 359 417 58		DMSO	359	417	58

The quantum yields measured in triplicate with the respective standard deviations are 0.80 ± 0.07 , 0.58 ± 0.05 , 0.61 ± 0.05 and 0.42 ± 0.03 .

In a general way, the boron complexes showed larger Stokes shift values, most likely due to the higher planarity and more efficient π -conjugation. Bathochromic displacements are observed when structures **1** and **3** were compared with their correspondent boron complexes, showing that the planarity introduced with the formation of the complex favors more electronic conjugation in the structure. Beyond the red shift, **3** and **4** displayed a shoulder at 356 and 360 nm, respectively, indicating an slight effect over its conformation as a consequence of the solvatochromic effect. The Stokes shifts for the BODIPYlike compound **2** are larger than those observed for the other structures (Table 1).

Considering the solvent effect is important for investigating this sort of derivatives, especially in the presence of polar solvents, therefore some theoretical calculations were performed aiming to shed some light on their behaviour in the ground and excited state. In this sense, DMSO was used (implicit treatment) as the medium for the calculations at PBE1PBE/6-311+G(2d,p)//CAM-B3LYP/6-31+g(d) level of theory. HOMO,

LUMO (basically of π -type) and electron density difference were also calculated and are in accordance with the good stability observed in the excited state for compounds **1-4** (Figure 13). Online

Electrochemical (cyclic voltammetry, CV)¹⁰³ analyses we as also performed (Figures S45-S46). All the analysed compound were reduced in two reduction waves (Figure S45 and Table °9) The first electron uptake was associated with a relatively stable radical anion whereas the second one is, presumably, a dihydro derivative, in accordance with similar compounds.¹⁶ Difference in current were noticed comparing the boron-free and boron containing derivatives. As shown in Figures S45-46, comparison of the synthesized compounds allowed us to suggest the complexation with boron slightly facilitates the reduction process (Table S9), with E_{pc} values less negative than those for the boron-free compounds.



Figure 3. Optimized geometries, HOMO and LUMO plots, and electron density difference map. Diagram energy levels calculated with DMSO as the solver Calculations at PBE1PBE/6-311+G(2d,p)//CAM-B3LYP/6-31+g(d) level of theory Structures calculated with implicit DMSO as the solvent showed similar orbitals localizations.

Finally, live cell-imaging experiments (PC3 cell lineages human prostate cancer cells) were conducted to evaluate the potential of the fluorescent compounds as bioprobes. Dye showed no fluorescent signal when incubated with live cell (Figure S41) whereas compounds **2-4** displayed a bright blus emission signal. These three compounds proved to be capable or staining the canonical endocytic pathway, meaning they a capable of selectively staining early endosome, late endosom and lysosome (Figures 4-6) during the cellular uptake process.

Panels (A) and (C) in Figures 4-5 show all membranous compartments strongly stained; there are no fluorescent signals inside of the nucleus. Acridine orange (AO) is widely used in order to stain endocytic pathways, however, AO also stains nucleic acids as can be observed in Figures 4-6. Under the some experimental conditions, **2-4** did not stain nucleic acids and cell nuclei are visualized as black voids. Compounds **2**, **3** and **4** could be found at different localizations inside the cytoplasm and therefore are found associated with different membranous compartments. Vesicles found at the cell periphery close to the plasmatic membrane is correlated with early endosomes (Figure 4 and 5). Vesicles found between the plasmatic membrane an

2 | J. Name., 2014, 00, 1-3

the nucleus are correlated with late endosomes (Figures 5), whereas vesicles found near to the nucleus are correlated with lysosomes (Figure 6). The overlay images between the tested compounds and acridine orange (panel (C) in the cited Figures) provided the evidence that these compounds are staining the membranous compartments belonging to the endocytic pathway, but in this case selectively. AO is widely used to stain acidic organelles in studies regarding the endocytic and autophagic pathways applied to eukaryotic cells, however, as demonstrated, with high affinity to nucleic acid components, which produce a non-specific pattern in the cells nuclei. **2-4** were able to stain specifically the membranous compartments belonging to the endocytic pathway without nucleus cross staining.



Figure 4. Live PC3 cancer cells stained with **2** (blue emission). (A) Positive control with cells stained with acridine orange (green emission). (B) **2** as the bioprobe and (C) Overlay of (A) and (B). **2** was found accumulated inside of spherical membranous organelles associated with endosomes (white arrow heads). White arrows show the nucleus stained with acridine orange (non-specific staining). N = cell nucleus. Reference scale bar 20 μ m.



Figure 5. Live PC3 cancer cells stained with 3 (blue emission). (A) Positive control with cells stained with acridine orange (green emission). (B) 3 as the bioprobe and (C) Overlay of (A) and (B). 3 was found accumulated inside of spherical membranous organelle correlated with early endosomes (white arrow heads), late endosomes (yellow arrow heads) and lysosomes (red arrows). White arrows show the nucleus stained with acridine orange (non-specific staining). N = cell nucleus. Reference scale bar 20 μ m.



Figure 6. Live PC3 cancer cells stained with 4 (blue emission). (A) Positive control with cells stained with acridine orange (green emission). (B) 4 as the bioprobe and (C) Overlay of (A) and (B). 4 was localized inside of spherical membranous organelles belong to the endocytic pathway. Red arrows show the staining membranous compartment correlated with lysosomes. White arrows show the nucleus stained with acridine orange (non-specific staining). N = cell nucleus. Reference scale bar 20 μ m.

The possibility of caveolae-mediated endocytosis was also evaluated to confirm the selective endocytic trafficking. Cells through caveolae-mediated endocytosis may also internalize the tested compounds irrespective of any affinity with the caveolar structure. Caveolar vesicles formation is a stable structure due of the lipids composition in these regions.^{17,18} This feature keeps the vesicles open by a time period greater than that observed ³/₁ membrane regions associated with clathrin-mediated vesicle formation.^{17,19} Most likely the tested compounds accumulate in the open vesicles during their formation. As a consequence they are found trapped inside the vesicles during their bud off from the cells membrane therefore targeting the endoplasmineticulum and matching with our finds related to vesicles stainin, near to the nuclei. To be sure about this hypothesis, an assay using the commercially available anti-caveolin-1 antibody (referenter), which is the universal marker specific for caveolar vesicle formation, has been performed (Figures 7-9).



Figure 7. Live PC3 cancer cells stained with **2** and immunodetection of caveolin 1 (universal caveolar marker). (A) Shows the cell nucleus stained with commercia available DAPI (blue), and the cytoplasmic endocytic vesicles stained by each 2 (also in blue). (B) Shows the fluorescence distribution related to caveolin 1 immunodetection (red). (C) Shows the overlay of the two fluorescent signals (blue and red). Arrows show vesicles belong to canonical endocytic pathways (abserr of caveolin-1 detection) and arrowheads show the vesicles belong to caveola - mediated endocytosis (vesicle with double staining - blue and red). (D) Shows the cells normal morphological aspects. Reference scale bar 25 μm.



Figure 8. Live PC3 cancer cells stained with **3** and immunodetection of caveolin-.. (A) Shows the cell nucleus stained with DAPI (blue), and the cytoplasmic endocyt vesicles stained by each **3** (also in blue). (B) Shows the fluorescence distribution related to caveolin-1 immunodetection (red). (C) Shows the overlay of the two fluorescent signals (blue and red). Arrows show vesicles belong to canonic endocytic pathways (absence of caveolin-1 detection) and arrowheads show the vesicles belonging to caveolae-mediated endocytosis (vesicle with double staining g

- blue and red). (D) Shows the cells normal morphological aspects. Reference scale bar 25 $\mu\text{m}.$

Figure 7 shows few vesicles stained near to cytoplasmic membrane correlated with caveolar vesicle bud off and near to the nucleus of the cell. This feature indicates the vesicles target endoplasmic reticulum which is the main docking point for caveolar vesicles. In other words, the caveolar entering is one possibility as well as the canonical endocytic pathway. The same behaviour could be noted for all tested compounds indicating the new bioprobes are capable of probing selectively the endocytic trafficking in live cells irrespective of the preferred internalization path. Similar results from that in Figure 7 have been observed in Figures 8 and 9.



Figure 9. Live PC3 cancer cells stained with 4 and immunodetection of caveolin-1. (A) Shows the cell nucleus stained with DAPI (blue), and the cytoplasmic endocytic vesicles stained by each 4 (also in blue). (B) Shows the fluorescence distribution related to caveolin-1 immunodetection (red). (C) Shows the overlay of the two fluorescent signals (blue and red). Arrows show vesicles belong to canonical endocytic pathways (absence of caveolin-1 detection) and arrowheads show the vesicles belonging to caveolae-mediated endocytosis (vesicle with double staining - blue and red). (D) Shows the cells normal morphological aspects. Reference scale bar 25 μ m.

In summary, four new fluorescent derivatives have been synthesized, characterized and successfully used as live cellimaging probes, capable of selectively staining the endocytic pathway. The compounds also showed no cytotoxicity when evaluated by MTT assay. Compounds 2-4 were able to permeate intact live cells, thus being trapped inside of acidic compartments, turning them into suitable molecules for investigating the lysosome biogenesis, host pathogens interaction and autophagic processes, as will be disclosed in due course. The selective endocytic trafficking in live cells and its dynamics have been probed using the synthesized compounds, which in turn, were capable of selectively reveal the internalization endocytic process in live cell through both the caveolar vesicles or the canonical pathway. The new designed bioprobes showed by far better results than the commercially available AO. The authors would like to thank CAPES, CNPq, FAPDF, DPP-UnB, FINATEC and FAPEMIG.

Notes and references

^{*a*} Institute of Exact Sciences, Department of Chemistry, Federal University of Minas Gerais, 31270-901, Belo Horizonte, MG, Brazil. E-mail: eufranio@ufmg.br

^b Institute of Physics, Department of Physics and Informatics, University of São Paulo, 13560-160, São Carlos, SP, Brazil.

^c University Unit of Exact Sciences and Technology, State University 563A Goiás, 75024-010, Anápolis, GO, Brazil.

^d Laboratory of Medicinal & Technological Chemistry, Institute of Chemistry, University of Brasilia, 70904970, P.O.Box 4478, Brasilia, D^r, Brazil. E-mail: correa@unb.br

^e Institute of Chemistry and Biotechnology, Federal University of Alagoa 57072-970, Maceió, AL, Brazil.

^{*f*} Department of Physiology and Pharmacology, Federal University Ceará, 60430-270, Fortaleza, CE, Brazil.

^g Fiocruz-CE, 60180-900, Fortaleza, CE, Brazil.

Electronic Supplementary Information (ESI) available: [details of armsupplementary information available should be included here]. See DOI: 10.1039/c000000x/

- 1. M. N. Levine, T. T. Hoang and R. T. Raines, *Chem. Biol.*, 2013, 2 , 614-618.
- 2. S. D. Conner and S. L. Schmid, *Nature*, 2003, **422**, 37-44.
- A. A. Marti, S. Jockusch, N. Stevens, J. Ju and N. J. Turro, Acc. Cne. Res., 2007, 40, 402-409.
- 4. L. D. Lavis and R. T. Raines, ACS Chem. Biol., 2014, 9, 855-866.
- 5. L. D. Lavis and R. T. Raines, ACS Chem. Biol., 2008, 3, 142-155.
- 6. N. Johnsson and K. Johnsson, ACS Chem. Biol., 2007, 2, 31-38.
- (a) B. A. D. Neto, J. R. Correa and R. G. Silva, RSC Adv., 2013, 3, 529
 5301. (b) B. A. D. Neto, J. R. Corrêa, P. H. P. R. Carvalho, D. C. B. I. Santos, B. C. Guido, C. C. Gatto, H. C. B. de Oliveira, M. Fasciotti, M N. Eberlin and E. N. da Silva Júnior, J. Braz. Chem. Soc., 2012, 23, 77
 781. (c) P. H. P. R. Carvalho, J. R. Correa, B. C. Guido, C. C. Gatto, H. C. B. De Oliveira, T. A. Soares and B. A. D. Neto, Chem.-Eur. J., 201
 20, 15360-15374. (d) C. D'Angelis do E. S. Barbosa, J. R. Corrêa, G. A. Medeiros, G. Barreto, K. G. Magalhães, A. L. de Oliveira, J. Spence, M. O. Rodrigues and B. A. D. Neto, Chem. -Eur. J., 2015, 21, 505
- 8. A. Nadler and C. Schultz, Angew. Chem., Int. Ed., 2013, 52, 2408-241 .
- A. A. R. Mota, P. H. P. R. Carvalho, B. C. Guido, H. C. B. de Oliveira. T. A. Soares, J. R. Correa and B. A. D. Neto, *Chem. Sci.*, 2014, 5, 399⁵ 4003.
- E. H. G. da Cruz, P. H. P. R. Carvalho, J. R. Correa, D. A. C. Silva, E. B. T. Diogo, J. D. de Souza Filho, B. C. Cavalcanti, C. Pessoa, H. C. de Oliveira, B. C. Guido, D. A. da Silva Filho, B. A. D. Neto and E. I da Silva Júnior, *New J. Chem.*, 2014, **38**, 2569-2580.
- S. N. Sunassee, C. G. L. Veale, N. Shunmoogam-Gounden, O. Osopi-D. T. Hendricks, M. R. Caira, J.-A. de la Mare, A. L. Edkins, A. Pinto, E. N. da Silva Júnior and M. T. Davies-Coleman, *Eur. J. Med. Chem.*, 2013, **62**, 98-110.
- E. N. da Silva Júnior, C. F. de Deus, B. C. Cavalcanti, C. Pessoa, L. V. Costa-Lotufo, R. C. Montenegro, M. O. de Moraes, M. C. F. R. Pint, C. A. de Simone, V. F. Ferreira, M. O. F. Goulart, C. K. Z. Andrade a A. V. Pinto, *J. Med. Chem.*, 2010, **53**, 504-508.
- (a) R. Araya-Maturana, W. Cardona, B. K. Cassels, T. Delgado-Castro. J. Ferreira, D. Miranda, M. Pavani, H. Pessoa-Mahana, J. Soto-Delga o and B. Weiss-Lopez, *Bioorg. Med. Chem.*, 2006, **14**, 4664-4669. (b) (E. M. Carvalho, A. S. Silva, I. M. Brinn, A. V. Pinto, M. C. F. R. Pint, S. Lin, T. A. Moore, D. Gust and M. Maeder, *Phys. Chem. Chem. Phys.*, 2002, **4**, 3383-3389. (c) C. E. M. Carvalho, I. M. Brinna, A. V. Pint M. C. F. R. Pinto, *J. Photoch. Photobio. A*, 1999, **123**, 61-65.
- 14. V. P. Torchilin, Adv. Drug Deliv. Rev., 2005, 57, 95-109.
- 15. E. M. Perchellet, B. J. Sperfslage, G. Qabaja, G. A. Jones and P. Perchellet, *Anti-Cancer Drugs*, 2001, **12**, 401-417.
- Henning L., Ole H. Organic Electrochemistry. 4th Edition, Mac Dekker, Inc, New York, 2001.
- 17. L. Pelkmans and A. Helenius, *Traffic*, 2002, **3**, 311-320.
- A. Zloza, D. W. Kim, J. Broucek, J. M. Schenkel and H. L. Kaufma J. Interferon Cytokine Res., 2014, 34, 915-919.
- 19. G. Sowa, Front. Physiol., 2012, 2, 120.

4 | J. Name., 2014, 00, 1-3

This journal is $\ensuremath{\mathbb{C}}$ The Royal Society of Chemistry 2014