

Preparation of Tri- and Tetrasubstituted Allenes via Regioselective Lateral Metalation of Benzylic (Trimethylsilyl)alkynes Using TMPZnCl·LiCl

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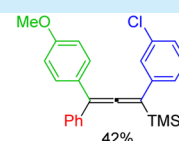
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Supporting Information

ABSTRACT: The zincation of various 1-(trimethylsilyl)-3-aryl-1-propynes with TMPZnCl·LiCl followed by a Pd-catalyzed coupling with aryl halides provides arylated allenes in 52–92% yield. Subsequent metalation with TMPZnCl·LiCl and cross-coupling with a second different aryl halide provides regioselectively tetrasubstituted allenes in 42–70% yield. This sequence can be performed in a one-pot procedure. DFT calculations and NMR studies support the formation of allenylzinc and propargyllithium intermediates starting from 1-(trimethylsilyl)-3-phenyl-1-propyne.

one-pot procedure

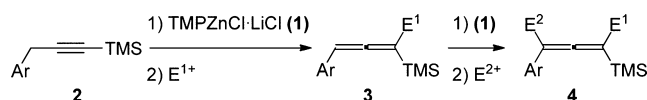
- 1) TMPZnCl·LiCl, 25 °C, 1 h
- 2) 1-bromo-3-chlorobenzene
Pd cat., 50 °C, 4 h
- 3) TMPZnCl·LiCl, 25 °C, 1 h
- 4) 4-iodoanisole
Pd cat., 50 °C, 12 h



Allenic structures are found in a number of useful organic molecules such as natural products¹ or materials.² They are also important intermediates³ for carbo- and heterocycle synthesis,⁴ making the preparation of substituted allenes a valuable synthetic target.⁵ Various transition-metal-catalyzed functionalizations of the allenic moiety have been reported.⁶ For example, Ma reported the preparation of trisubstituted allenes using a Pd-catalyzed arylation of zincated allenes obtained by LDA deprotonation of the corresponding allenes.⁷

Recently, we have reported that TMPZnCl·LiCl (**1**, TMP = 2,2,6,6-tetramethylpiperidyl) is an excellent base for the selective metalation of various substrates.⁸ The high kinetic basicity of **1** allows the efficient metalation of a range of organic molecules including nitriles, esters, and various functionalized unsaturated molecules.⁹ We have envisioned that lateral metalations¹⁰ of benzylic alkynes of type **2** with TMPZnCl·LiCl (**1**) will produce intermediate allenylzinc reagents, which after quenching with an electrophile E¹⁺ will produce trisubstituted allenes of type **3** (Scheme 1).⁷

Scheme 1. Successive Lateral Metalation of Benzylic Alkynes of Type 2 with TMPZnCl·LiCl (**1**)

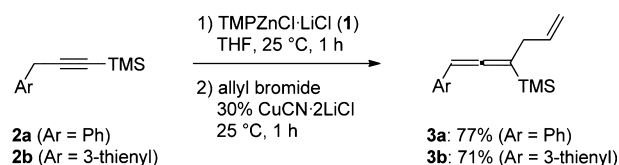


A subsequent metalation with TMPZnCl·LiCl (**1**) will give a new zincated intermediate, which after trapping with a second electrophile E²⁺ will furnish tetrasubstituted allenes of type **4**. Herein, we report the successful realization of this reaction sequence using a palladium¹¹ or a copper-catalyzed¹² reaction of 1-(trimethylsilyl)-3-aryl-1-propynes with electrophiles.

In preliminary experiments, we have treated the TMS-protected alkyne (**2a**) with TMPZnCl·LiCl (**1**; 1.2 equiv) in

THF at 25 °C for 1 h. Addition of CuCN·2LiCl (30 mol %) and allyl bromide led to the allylated allene **3a** in 77% yield, indicating that TMPZnCl·LiCl (**1**) achieved a smooth deprotonation of the benzylic hydrogen.¹³ This result was confirmed by extending the metalation to the heterobenzylic derivative (**2b**), which after allylation, similarly produced the corresponding allene (**3b**) in 71% yield (Scheme 2). With these results in hand, we examined

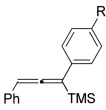
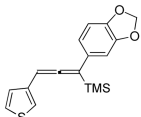
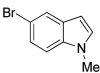
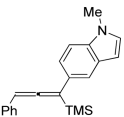
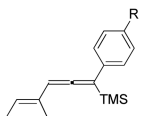
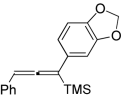
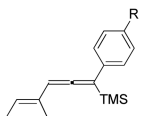
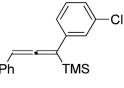
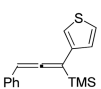
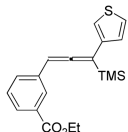
Scheme 2. In Situ Trapping of Zinc Reagents with CuCN·2LiCl and Subsequent Allylation Reaction



the direct Pd-catalyzed arylation of the TMS-protected alkynes of type **2**. Thus, the reaction of the alkynes **2a–c** with TMPZnCl·LiCl (**1**; 1.2 equiv) in THF at 25 °C for 1 h, followed by the addition of an aryl or heteroaryl bromide or iodide (**5a–j**), provides the arylated allenes **3a–p** in 52–92% yield (Table 1). As Pd catalyst, we have found after an extensive screening that three catalytic systems gave the best results: (a) 2% Pd(OAc)₂/2% DPE-Phos;¹⁴ (b) 2% Pd(OAc)₂/4% S-Phos;¹⁵ (c) 2% PEPPSI-*i*-Pr.¹⁶ Both donor- and acceptor-substituted aryl halides afforded the trisubstituted allenes **3c–k** in 60–92% yield (entries 1–9). The thienyl-substituted alkyne (**2b**) behaves similarly and produced the allenes (**3l–m**) in 65–78% yield (entries 10–11). Remarkably, an ester substituent is perfectly tolerated, and the alkyne (**2c**) leads to the corresponding trisubstituted allenes

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Table 1. Zincation with TMPZnCl·LiCl (1) Followed by a Negishi Cross-Coupling Reaction with Various Electrophiles

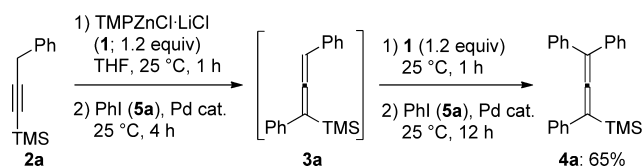
$\text{Ar}-\text{C}\equiv\text{C}-\text{TMS} \xrightarrow[\text{THF, 25 } ^\circ\text{C, 1 h}]{\text{1) TMPZnCl}\cdot\text{LiCl (1, 1.2 equiv)}} \text{Ar}-\text{C}(\text{Ar}^1)=\text{C}(\text{TMS})-\text{Ar}^2$ 2) Ar^1X (5a-j, 1.0 equiv) Pd cat., 25–50 $^\circ\text{C}$, 2–12 h 2a (Ar = Ph) 2b (Ar = 3-thienyl) 2c (Ar = 3-carboethoxyphenyl) 3c-p: 52–92%					
entry	electrophile ^a ($^\circ\text{C}$, h) ^b	product yield (%) ^c	entry	electrophile ^a ($^\circ\text{C}$, h) ^b	product yield (%) ^c
1	 5a R = H, X = I (25, 4)	 3c : 68 ^d	10	 5j (50, 12)	 3l : 65 ^f
2	5b R = OMe, X = I (25, 5)	3d : 76 ^d			
3	5c R = NMe ₂ , X = Br (50, 4)	3e : 66 ^e			
4	5d R = OPiv, X = Br (50, 6)	3f : 71 ^f			
5	5e R = SMe, X = Br (50, 6)	3g : 75 ^f			
6	 5f (50, 3)	 3h : 64 ^f	11	 5a (50, 2)	 3m : 78 ^f
7	 5g (25, 4)	 3i : 92 ^f	12	5a R = H, X = I (25, 2)	 3n : 52 ^d
			13	5b R = OMe, X = I (25, 2)	3o : 59 ^d
8	 5h (50, 4)	 3j : 60 ^f			
9	 5i (25, 2)	 3k : 73 ^g	14	 5i (25, 3)	 3p : 74 ^g

^a1.0 equiv of electrophile was used. ^bReaction time at 25 or 50 $^\circ\text{C}$ for full conversion. ^cIsolated yield of analytically pure product. ^dPd catalyst: 2% Pd(OAc)₂/4% S-Phos. ^ePd catalyst: 4% Pd(OAc)₂/4% DPE-Phos. ^fPd catalyst: 2% Pd(OAc)₂/2% DPE-Phos. ^gPd catalyst: 2% PEPPSI-*i*Pr.

(**3n–p**) in 52–74% yield (entries 12–14). In all cases, the arylation is regioselective (only allenic derivatives are obtained and no arylated propargylic compounds could be detected).

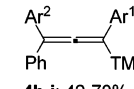
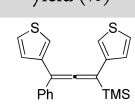
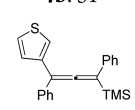
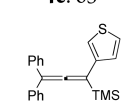
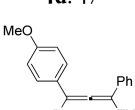
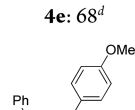
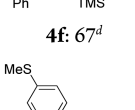
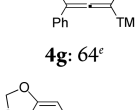
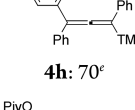
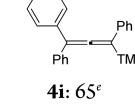
Then, we developed a one-pot procedure allowing a direct conversion of alkyne **2a** to the tetrasubstituted allenes of type **4** (Scheme 3 and Table 2). Thus, alkyne **2a** was treated as before with TMPZnCl·LiCl (1, 1.2 equiv) and iodobenzene (**5a**, 1.0 equiv) leading to **3c**, which was not isolated but directly zincated with TMPZnCl·LiCl (1, 1.2 equiv) and iodobenzene (**5a**, 1.0 equiv), affording the tetrasubstituted allene **4a** in 65% isolated yield.

Scheme 3. Successive Zincation with TMPZnCl·LiCl (1) and Subsequent Negishi Cross-Coupling Reactions



Replacing the aryl iodide (**5a**) by 3-bromothiophene (**5i**) similarly produced the tetrasubstituted allene **4b** in 51% yield (Table 2, entry 1). The use of two different aryl or heteroaryl halides was also possible. Thus, treatment of the alkyne (**2a**) with TMPZnCl·LiCl (1, 1.2 equiv), a Pd catalyst, and iodobenzene (**5a**, 1.0 equiv) for 4 h at 25 $^\circ\text{C}$ led to the intermediate allene **3c**, which was directly metalated with TMPZnCl·LiCl (1, 1.2 equiv), Pd catalyst, and 3-bromothiophene (**5i**, 1.0 equiv), affording the tetrasubstituted allene **4c** in 63% overall yield (entry 2). By inverting the addition order of the two electrophiles E^{1+} and E^{2+} using first the heterocyclic bromide (**5i**) and then iodobenzene (**5a**), we have obtained the regioisomeric tetrasubstituted allene **4d** in 47% overall yield (entry 3). This approach was used for the preparation of the tetrasubstituted allenes (**4e–j**) in 42–70% overall yield (entries 4–9). Remarkably, this method allows the synthesis of the tris-3-thienylallene **4k** in 65% yield (Scheme 4). The structure of several allenes (**3c**, **4c**, **4f**, **4k**; Tables 1 and 2) has been confirmed by X-ray diffraction analysis.¹⁷ We have briefly studied the scope of these new metalations of alkynes and allenes and found that the related diyne **6** was similarly zincated

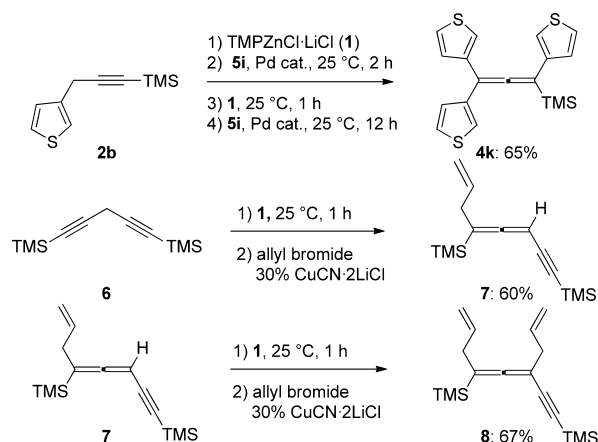
Table 2. One-Pot Tetrafunctionalization of Allenes via Successive Zincation Using TMPZnCl·LiCl (1) and Negishi Cross-Coupling Reactions with Various Electrophiles

		1) 1, THF, 25 °C, 1 h 2) Ar ¹ X (5, 1.0 equiv), Pd cat. 3) 1 (1.2 equiv), 25 °C, 1 h 4) Ar ² X (5, 1.0 equiv), Pd cat.		4b-j: 42–70%
entry	electrophile ^a (°C, h) ^b	product yield (%) ^c		
1	1) 5i (25, 2) 2) 5i (25, 12)	 4b: 51^d		
2	1) 5a (25, 4) 2) 5i (25, 12)	 4c: 63^d		
3	1) 5i (25, 2) 2) 5a (25, 12)	 4d: 47^d		
4	1) 5a (25, 4) 2) 5b (25, 12)	 4e: 68^d		
5	1) 5b (25, 5) 2) 5a (25, 12)	 4f: 67^d		
6	1) 5a (50, 2) 2) 5e (50, 2)	 4g: 64^e		
7	1) 5a (50, 2) 2) 5g (50, 3)	 4h: 70^e		
8	1) 5a (50, 2) 2) 5d (50, 12)	 4i: 65^e		
9	1) 5h (50, 4) 2) 5b (50, 12)	 4j: 42^e		

^a1.0 equiv of electrophile was used. ^bReaction time at 25 or 50 °C for full conversion. ^cIsolated yield of analytically pure product. ^dPd catalyst: 2% Pd(OAc)₂/4% S-Phos. ^ePd catalyst: 2% Pd(OAc)₂/2% DPE-Phos.

with TMPZnCl·LiCl (**1**) within 1 h at 25 °C (Scheme 4).¹⁸ After a copper-catalyzed allylation with allyl bromide, the trisubstituted allene **7** was obtained in 60% yield.¹⁹ The unsaturated

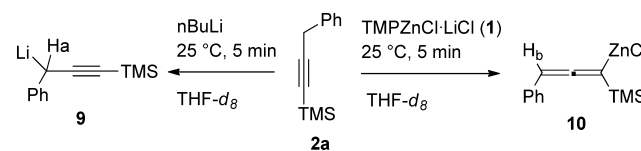
Scheme 4. Preparation of a Tri-3-thienylallene (4k) and Lateral Metalation of Alkynes 6 and 7 with TMPZnCl·LiCl (1)



alkynylallene **7** was cleanly allylated affording the highly unsaturated product **8** in 67% yield.

In order to establish the nature of the zincated intermediates (allenyl or propargylic) occurring during the metalation of **2a**, we performed an NMR study that showed that the lithiation of **2a** produces only the propargylic lithium species **9** as seen by the chemical shift of the propargylic proton H_a at 3.33 ppm in the ¹H NMR spectra (Scheme 5).¹⁷

Scheme 5. NMR Experiments Showed Direct Formation of Allenylzinc Reagent



The zincation of **2a** using TMPZnCl·LiCl (**1**) produces an allenylzinc species **10** (as seen by the allenyl ¹³C signal at 202.8 ppm, no propargyl isomer was observed and the allenyl proton (H_b) has a chemical shift of 4.99 ppm in the ¹H NMR spectra).¹⁷ From these studies, it becomes clear that the propargyl isomer **9** is the most stable organometallic species in the case of the lithium cation, whereas the allenylzinc structure **10** is the most stable (zinc cation).²⁰ Theoretical calculations at the MP2 level of theory confirmed a major difference in stability order for the propargyl allenyl isomers of the respective organometallics. An endothermic enthalpy of 1.4 kJ/mol for the two organolithium isomers (**9** and **9a**; Figure 1) is opposed to 7.8 kJ/mol for the zinc species **10** and **10a**.¹⁷

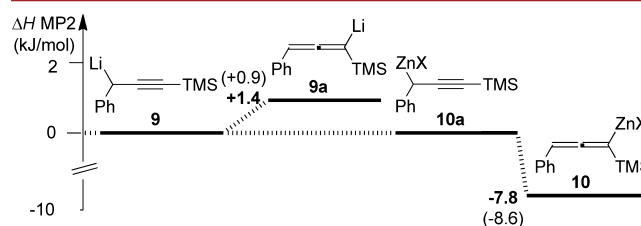


Figure 1. Propargyl allenyl isomerization in THF solution (SMD/B3LYP/6-31G(d)) at the MP2(FC)/6-31+G(d,p) level (Li = Li(THF)₃, ZnX = ZnCl(THF)₂, gas-phase values in parentheses).

In conclusion, we have reported an efficient Pd-catalyzed arylation of some 1-(trimethylsilyl)-3-aryl-1-propynes using $\text{TMPZnCl}\cdot\text{LiCl}$ leading to trisubstituted allenes. Interestingly, we have performed a one-pot bis-arylation of 1-(trimethylsilyl)-3-phenyl-1-propyne affording regioselectively tetrasubstituted allenes. Quantum chemical calculations and NMR studies support the formation of allenylzinc and propargyllithium intermediates.

■ ASSOCIATED CONTENT

■ Supporting Information

Full experimental details and NMR data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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