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N-Trichloroethoxycarbonyl-glucosamine derivatives as glycosyl donors¹

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Abstract

D-Glucosamine can be readily transformed into 1,3,4,6-tetra-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)-D-glucopyranose (2). From this intermediate valuable glycosyl donors can be obtained; reaction with ethanethiol in the presence of boron trifluoride etherate afforded ethyl 3,4,6-tri-O-acetyl-2-deoxy-1-thio-2-(2,2,2-trichloroethoxycarbonylamino)- β -D-glucopyranoside (4) which gave, upon N-acetylation, the N-acetyl-N-trichloroethoxycarbonyl derivative (5). Selective removal of the 1-O-acetyl group in 2 followed by treatment with trichloroacetonitrile in the presence of base afforded 3,4,6-tri-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)- α -D-glucopyranosyl trichloroacetimidate (6). Reaction of 5 with five selectively protected glycosides as glycosyl acceptors in the presence of N-iodosuccinimide/trifluoromethanesulfonic acid as the promoter system furnished the corresponding β -glycosides in good yields, thus exhibiting the valuable glycosyl donor properties of 5. Reductive removal of the trichloroethoxycarbonyl (Teoc) group afforded the corresponding N-acetyl-protected saccharides in high yields. The imidate 6 reacted with three of the above acceptors in the presence of catalytic amounts of trimethylsilyl trifluoromethanesulfonate to give the β -linked disaccharides in even better yields. The direct replacement of the N-Teoc group by the N-acetyl group using zinc/acetic anhydride, via the free amines as transient intermediates, adds to the high efficiency and convenience of this methodology. © 1996 Elsevier Science Ltd.

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¹ Glycosyl imidates, part 77. For part 76, see Ref. [1].

1. Introduction

D-Glucosamine is an important constituent of various glycoconjugates [2]. Most frequently it is N-acetylated and found in β -glycosidic linkage. Glycoside bond formation with donors derived from N-acetyl-D-glucosamine occurs generally via neighboring group participation to give an oxazoline intermediate [3]; these oxazolines do not exert strong glycosyl donor properties because the methyl-substituted N-protonated oxazolinium system is rather stable. Therefore, various alternative glucosamine donors have been investigated, amongst which the phthalimido group [2,3] and the azido group [2-5] have gained wide use. The azido group serves as an excellent latent functionality for the amino group [2,5]; however, the preparation of the required azidodeoxyglucose is still not very economical [5]. Phthalimido sugars can be readily obtained from amino sugars and, for instance, in combination with trichloroacetimidate activation, good glycosyl donors are available for β -glycoside bond formation via reactive N-acylated oxazolinium intermediates [2,5]. However, cleavage of the phthalimido group requires basic conditions which often result in partial product decomposition [6] (Improvements have recently been reported [7].) Therefore, tetrachlorophthalimido(TCP)-protected glycosyl donors were recently employed [8,9], which exhibit high glycosyl donor properties and permit removal of the TCP group under very mild nonbasic conditions [1,9].

For various applications the intermediate generation of a free amino group — especially under basic conditions — leads to undesired side reactions. Therefore, methods retaining the *N*-acetyl functionality (or an equivalent) in the glycosyl donor are in great demand. A convenient solution of this problem is the formation of N, N-diacetyl



Scheme 1.

derivatives [10]. Thus, thioglycosides of N,N-diacetyl-D-glucosamine (Scheme 1, A: $R^{N} = Ac$, $R^{C} = Me$, X = SMe) turned out to be excellent glycosyl donors for very reactive acceptors (HOR), giving β -glycosides **B** (presumably via **C**) in high yields. However, for less reactive acceptors acetyl transfer from the donor to the acceptor $(\rightarrow \mathbf{D} + \mathbf{E})$ turned out to be a competing side reaction [10]. Therefore, we initiated a program to replace one of the N-acetyl groups by a readily removable alkoxycarbonyl moiety, thus lowering the N-acetyl transfer character of the glycosyl donor [11]. Because alkoxycarbonyl residues participating in the formation of oxazolinium systems of type C can undergo cleavage to an alkylcarbonium ion, the inductively electron-withdrawing trichloroethoxycarbonyl (Teoc) moiety was chosen (A: $R^{N} = Ac$, $R^{C} = OCH_{2}CCl_{2}$), thus avoiding this undesired side reaction and increasing the glycosyl donor properties of intermediate \mathbb{C}^{2} . For comparison reasons the corresponding N-deacetyl compound (A: $R^{N} = H$, $R^{C} = OCH_{2}CCl_{3}$) was investigated, which, as has already been shown [11,13,14], can be obtained with trichloroacetimidate as leaving group; this compound exhibited, in the few cases thus far applied, good glycosyl donor properties due to the strong electron-withdrawing effect of the R^{C} substituent in the oxazolinium intermediate **F** (or in **G**).

2. Results and discussion

For the synthesis of the known N-Teoc-protected O-acetyl-glucosamine 2β [15] (Scheme 2) the free amine 1 was prepared [16] which was then transformed with trichloroethoxycarbonyl chloride (Teoc-Cl) into 2β ; reaction with ethanethiol in the presence of boron trifluoride etherate gave, as described by Kunz et al. [17], thioglycoside 4 in good yield which has also been investigated as a glycosyl donor [17,18]. Reaction of 4 with acetic anhydride in the presence of 4-dimethylaminopyridine (DMAP) and Hünig base afforded the desired N-acetyl-N-Teoc-protected glucosamine donor 5. For the synthesis of trichloroacetimidate 6 transformation of 4 with N-iodosuccinimide (NIS)/trifluoromethanesulfonic acid (TfOH) in the presence of water provides the 1-O-unprotected glucosamine derivative 3 which can be transformed into 6 in very high yield either by treatment with trichloroacetonitrile in the presence of potassium carbonate [13] or in the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) as base. The synthesis of 3 can be performed much more efficiently by direct reaction of D-glucosamine with Teoc-Cl in the presence of aqueous sodium hydrogen carbonate and then per-O-acetylation, furnishing $2\alpha,\beta$ in almost quantitative yield. Regioselective hydrazinolysis of the anomeric O-acetyl group could be carried out even in the presence of the quite reactive Teoc group, thus leading to 3 in three steps in high overall yield. Because transformation of 2 into 6 can also be performed in a one-pot procedure in high yield, the synthesis of glycosyl donor 6 is very straightforward and efficient.

The investigation of the glycosylation potential of donors 5 and 6 was performed with glycosides 7a-e (Scheme 3) which represent a broad scale of different acceptor

² In equilibrium reactions with the acceptor, ortho-acid derivatives could also be formed [12].





Scheme 3.

reactivities. Reaction of 5 with the highly reactive primary hydroxy group of 7a [19] under typical conditions for the activation of thioglycosides (1-2 equivalents of NIS and)0.1-0.2 equivalent of TfOH at 0 °C [20]) afforded the β -linked disaccharide 8a in practically quantitative yield (Scheme 4). Immediate reductive removal of the N-Teoc group with zinc in acetic acid [21] to afford the N-acetyl derivative 9a could be performed in 93% yield. The 3',4'-O-unprotected lactose derivative 7b is also an excellent glycosyl acceptor [22]. Because of the high overall acceptor reactivity, high regioselectivity for 3'-O attack could only be achieved at -30 °C, thus affording 8b in 70% yield, which was again immediately transformed into 9b. Axial hydroxy groups generally exhibit lower acceptor reactivity. Therefore, the 4-O-unprotected galactose derivative 7c [23] was treated with 5; under standard conditions disaccharide 8c and hence 9c were obtained in very good yields. Also the 2-O-unprotected mannose derivative 7d [24] could be transformed into disaccharide 8d, though in lower yield; 9d formation was again high-yielding. Finally, the 3',4'-O-unprotected O-benzoyl-protected lactose derivative 7e [25] was employed as acceptor. Obviously, the O-benzoyl groups not only lower the glycosyl acceptor properties but also the regioselectivity in the reaction with 5, thus furnishing a 2:1 mixture of 8eA,B in 65% overall yield. For the structural assignment, removal of the N-Teoc group under standard conditions (\rightarrow 9eA, **B**), then O-acetylation, and separation provided 10eA and 10eB (¹H NMR; 10eA: δ 5.28, H-4'; **10eB**: δ 3.99, H-4'; 4.93, H-3').

We then turned our attention to glycosyl donor **6**, because transformation of *N*-Teoc-protected compounds into the corresponding *N*-acetyl derivatives could be performed directly with zinc and acetic anhydride in high yield [26], thus generating free amines as intermediates only in low concentrations. Reaction of **6** with **7a** in the presence of catalytic amounts of trimethylsilyl trifluoromethanesulfonate (TMSOTf) (0.01 equivalent), applying the inverse procedure [27], furnished the β -linked disaccha-







ride **11a** in practically quantitative yield (Scheme 5); thus the excellent properties of **6** as a glycosyl donor are exhibited [17,18,28]. Subsequent replacement of the *N*-Teoc group in **11a** by the *N*-acetyl group with zinc and acetic anhydride furnished **9a** in 84% yield. The less reactive acceptors **7c** and, as already reported [13], **7d** gave with **6** under the same reaction conditions excellent yields of **11c** and **11d**; replacement of the *N*-Teoc group using zinc and acetic anhydride readily afforded the desired *N*-acetyl derivatives **9c** and **9d**.

In conclusion, the N-acetyl-N-Teoc-protected thioglycoside 5 is a good glycosyl donor for acceptors of very different reactivity; the convenient and selective removal of the N-Teoc group in the products in the presence of various protecting groups by zinc and acetic acid provides a generally applicable method for the synthesis of β -glycosides of N-acetyl-D-glucosamine. The ease of formation of the N-Teoc-protected trichloroace-timidate 6, the simple activation by catalytic amounts of TMSOTf, the very high glycosylation yields, and the high-yielding direct replacement of the N-Teoc group by the N-acetyl group with zinc and acetic anhydride make this method even more efficient. It is therefore a valuable addition to the available repertoire of glycoside bond-forming reactions in glycoconjugate synthesis.

3. Experimental

Solvents were purified in the usual way; boiling range of the petroleum ether (PE) used: 35-60 °C. ¹H NMR spectra: Bruker AC 250 (250 MHz) and Bruker DRX 600 (600 MHz); solvent, CDCl₃; internal standard, Me₄Si; chemical shifts and coupling constants were partially obtained from COSY spectra. Flash chromatography: silica gel (Baker: particle size 40 μ m). Thin-layer chromatography (TLC): foil plates, Silica Gel 60 F₂₅₄ (Merck; layer thickness, 0.2 nm). Optical rotations: Perkin–Elmer polarimeter 241 MC; 1-dm cell; temperature, 20 °C. Elemental analyses: Heraeus CHN-O-rapid.

1,3,4,6-Tetra-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)- α , β -D-glucopyranose (2α , β).—(a) From 1. Transformation of 1β [16] into 2β was performed as previously described [15].

(b) Directly from D-glucosamine yielding $2\alpha,\beta$. Trichloroethoxycarbonyl chloride (8.1 mL, 60 mmol) was added dropwise at room temperature to a vigorously stirred solution of D-glucosamine hydrochloride (10.8 g, 50 mmol) and NaHCO₃ (12.6 g, 150 mmol) in water (100 mL). The mixture was stirred for 1 h, then neutralized with 1 M HCl, concentrated, and dried in vacuo. A solution of the residue in pyridine (50 mL) and Ac₂O (25 mL) was stirred overnight at room temperature and then concentrated. The residue was eluted from a column of silica gel with 4:1 hexane–EtOAc to give $2\alpha,\beta$ (21.2 g, 92%) as a colorless oil which was identical to previously obtained material [15].

3,4,6-Tri-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)- α , β -D-glucopyranose (3).—(a) From 2. A solution of 2 (2.0 g, 3.82 mmol) and hydrazinium acetate (458 mg, 4.97 mmol) in DMF (20 mL) was stirred for 20 min at room temperature, then diluted with EtOAc (80 mL), washed with satd. aq NaCl (2 × 30 mL) and water (2 × 30 mL), dried (MgSO₄), and concentrated. The residue was eluted with 2:1 toluene–EtOAc to give 3 as a white foam (1.55 g, 83%); ¹H NMR (250 MHz, CDCl₃): δ 2.01–2.21 (m, 9 H, 3 × Ac), 3.65 (s, 1 H, OH), 4.01–4.29 (m, 4 H, H-2, H-5, H-6a, H-6b), 4.61–4.83 (m, 2 H, CH₂CCl₃), 5.09–5.48 (m, 4 H, H-1, H-3, H-4, NH).

(b) From 4 [17,28]. To a mixture of 4 (1.66 g, 3.17 mmol), water (114 μ L, 6.34 mmol), and NIS (0.79 g, 3.49 mmol) in CH₂Cl₂ (15 mL) was added TfOH (20 μ L, 0.22 mmol). After 30 min the reaction mixture was diluted with CH₂Cl₂ (15 mL), washed successively with satd. aq NaHCO₃, satd. aq Na₂S₂O₃, and water, dried (MgSO₄), filtered, and concentrated. The residue was eluted from a column of silica gel with 2:1 toluene–EtOAc to give 3 as a white foam (1.25 g, 81%).

Ethyl 3,4,6-tri-O-acetyl-2-[N-acetyl-N-(2,2,2-trichloroethoxycarbonylamino)]-2-deoxy-1-thio-β-D-glucopyranoside (5).—A solution of ethyl 3,4,6-tri-O-acetyl-2-deoxy-1thio-2-(2,2,2-trichloroethoxycarbonylamino)-β-D-glucopyranoside (4) [17,28] (1.74 g, 3.31 mmol) in dry CH₂Cl₂ (15 mL) was stirred for 72 h at room temperature and then concentrated. The residue was eluted from a column of silica gel with 6:1 toluene–EtOAc to give syrupy 5 (1.71 g, 93%); $[\alpha]_D - 7.7^\circ$ (c 1, CHCl₃); ¹H NMR (600 MHz, CDCl₃; 2 rotamers, 2:1 a–b): (a) δ 1.20–1.23 (m, 3 H, SCH₂CH₃), 1.93–2.05 (m, 9 H, 3 × Ac), 2.51 (s, 3 H, NHAc), 2.61–2.66 (m, 2 H, SCH₂CH₃), 3.68–3.70 (m, 1 H, H-5), 4.07–4.10 (m, 1 H, H-6a), 4.20–4.24 (m, 1 H, H-6b), 4.82 (d, 1 H, J_{gem} 7.4 Hz, CH₂CCl₃), 4.96 (d, 1 H, J_{gem} 7.4 Hz, CH₂CCl₃), 4.98 (dd, 1 H, J_{1,2} 10.3, J_{2,3} 9.5 Hz, H-2), 5.14 (dd, 1 H, J_{3,4} = J_{4,5} = 9.5 Hz, H-4), 5.32 (d, 1 H, J_{1,2} 10.3 Hz, H-1), 5.74 (dd, 1 H, $J_{2,3} = J_{3,4} = 9.5$ Hz, H-3); (b) δ 1.20–1.23 (m, 3 H, SCH₂CH₃), 1.93–2.05 (m, 9 H, 3 × Ac), 2.43 (s, 3 H, NHAc), 2.61–2.66 (m, 2 H, SCH₂CH₃), 3.75–3.78 (m, 1 H, H-5), 4.07–4.10 (m, 1 H, H-6a), 4.20–4.24 (m, 1 H, H-6b), 4.39 (dd, 1 H, $J_{1,2} = J_{2,3} = 9.8$ Hz, H-2), 4.80 (d, 1 H, J_{gem} 11.9 Hz, CH₂CCl₃), 4.88 (d, 1 H, J_{gem} 11.9 Hz, CH₂CCl₃), 5.02 (dd, 1 H, $J_{3,4} = J_{4,5} = 9.5$ Hz, H-4), 5.50 (d, 1 H, $J_{1,2}$ 9.8 Hz, H-1), 5.78 (dd, 1 H, $J_{2,3}$ 9.8, $J_{3,4}$ 9.5 Hz, H-3). Anal. Calcd for C₁₃H₂₆Cl₃NO₁₀S (566.8): C, 40.25; H, 4.26; N, 2.47. Found: C, 40.16; H, 4.78; N, 2.94.

3,4,6-Tri-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)- α -D-glucopyranosyl trichloroacetimidate (6).—(a) From 3. A solution of 3 (2.01 g, 4.01 mmol), trichloroacetonitrile (4 mL, 40 mmol), and DBU (0.15 mL, 1 mmol) in CH₂Cl₂ (12 mL) was stirred at room temperature for 20 min and then concentrated. The residue was eluted from a column of silica gel with 2:1 hexane-EtOAc containing 0.1% of triethylamine to afford 6 as a white solid (1.98 g, 73%); it was identical to material previously obtained from 3 with K₂CO₃ as base [13].

(b) From 2 in a one-pot procedure. A solution of 2 (1 g, 1.9 mmol) and hydrazinium acetate (0.19 g, 2.1 mmol) in DMF (8 mL) was stirred for 20 min at room temperature, then diluted with EtOAc (50 mL), washed with water, then with satd. aq NaHCO₃, and again with water, dried (MgSO₄), and concentrated. A solution of the residue, trichloroacetonitrile (1 mL, 10 mmol), and DBU (0.15 mL, 1 mmol) in CH₂Cl₂ (10 mL) was stirred for 30 min at room temperature and then concentrated. Workup was performed as described above to yield **6** (0.90 g, 76%).

Compounds **7a-e** were obtained following literature procedures: **7a** [19], **7b** [22], **7c** [23], **7d** [24], **7e** [25].

Methyl O-(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 6)-2,3,4-tri-O-benzyl- α -D-glucopyranoside (9a).—(a) Synthesis of 8a. TfOH (34 μ L of a 0.1 M solution of TfOH in dry CH₂Cl₂, 3.4 μ mol) was added to a stirred solution of 5 (200 mg, 0.343 mmol), methyl 2,3,4-tri-O-benzyl- α -D-glucopyranoside (7a) [19] (122 mg, 0.263 mmol), and NIS (89 mg, 0.395 mmol) in dry CH₂Cl₂ (2 mL). The mixture was stirred at room temperature under dry Ar for 2 h, then diluted with CH₂Cl₂ (20 mL), washed with satd. aq NaHCO₃, aq Na₂S₂O₃ (0.5 M), and satd. aq NaCl, dried (MgSO₄), and concentrated in vacuo. The residue was eluted from a column of silica gel with 9:1 toluene–EtOAc to give syrupy 8a (236 mg, 93%).

(b) **9a** from **8a**. A solution of **8a** (303 mg, 0.319 mmol) and freshly activated Zn-powder (300 mg) was stirred for 5 h in AcOH (5 mL), diluted with toluene, filtered, and concentrated in vacuo. The residue was eluted from a column of silica gel with 1:1 toluene–EtOAc to give syrupy **9a** (239 mg, 93%); $[\alpha]_D + 5.2^\circ$ (c 1, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 1.79–2.00 (4 s, each 3, 4 × Ac), 3.33 (s, 3 H, OCH₃) 3.45 (dd, 1 H, $J_{3,4} = J_{4,5} = 9.6$ Hz, H-4), 3.50 (dd, 1 H, $J_{1,2}$ 3.5, $J_{2,3}$ 9.6 Hz, H-2), 3.62–3.66 (m, 1 H, H-5'), 3.69 (dd, 1 H, $J_{5,6a}$ 4.2, $J_{6a,6b}$ 10.6 Hz, H-6a), 3.71–3.75 (m, 1 H, H-5), 3.82 (ddd, 1 H, $J_{1',2'}$ 8.4, $J_{2',NH}$ 9.0, $J_{2',3'}$ 9.5 Hz, H-2'), 3.95 (dd, 1 H, $J_{2,3} = J_{3,4} = 9.6$ Hz, H-3), 4.02 (dd, 1 H, $J_{5,6b}$ 1.4 Hz, H-6b), 4.07 (dd, 1 H, $J_{5',6'a}$ 2.4, $J_{6'a,6'b}$ 12.6 Hz, H-6'a), 4.19 (dd, 1 H, $J_{5',6'b}$ 4.7 Hz, H-6'b), 4.57 (d, 1 H, H-1), 4.65 (d, 1 H, H-1'), 4.53–4.96 (m, 6 H, $3 \times CH_2$ Ph), 5.01 (t, 1 H, $J_{3',4'} = J_{4',5'} = 9.5$ Hz, H-4'), 5.24 (t, 1 H, $J_{2',3'} = J_{3',4'} = 9.5$ Hz, H-3'), 5.37 (d, 1 H, $J_{2',NH}$ 9.0 Hz, NH), 7.25–7.32 (m, 15 H, Ph).

Anal. Calcd for C₄₂H₅₁NO₁₄ (793.8): C, 63.54; H, 6.47; N, 1.76. Found: C, 63.03; H, 6.54; N, 2.32.

(c) 9a from 11a. To a solution of 11a (170 mg, 0.181 mmol) in Ac_2O (5 mL) was added freshly activated Zn-powder (300 mg); the mixture was stirred at room temperature for 5 h, then diluted with toluene, filtered, and concentrated in vacuo. The residue was eluted from a column of silica gel with 1:1 toluene–EtOAc to give 9a as a syrup (121 mg, 84%) which was identical with the material obtained from 8a.

Benzyl O-(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 3)-O-(2,6-di-O-benzyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-2,3,6-tri-O-benzyl- β -D-glucopyranoside (9b).—(a) Synthesis of 8b. A solution of benzyl O-(2,6-di-O-benzyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-2,3,6-tri-O-benzyl- β -D-glucopyranoside (7b) [22] (200 mg, 0.226 mmol), 5 (154 mg, 0.271 mmol), NIS (66 mg, 0.283 mmol), and molecular sieves 4 Å in dry CH₂Cl₂ (3 mL) was stirred for 1 h under dry Ar and then cooled to -30 °C. TMSOTf (10 μ L, 0.056 mmol) was added and the mixture was stirred for 3 h at -30 °C, then diluted with CH₂Cl₂ (20 mL), washed with satd. aq NaHCO₃ (10 mL), aq Na₂S₂O₃ (10 mL, 0.5 M), and satd. aq NaCl, dried (MgSO₄), and concentrated. The residue was eluted from a column of silica gel with 6:1 toluene–EtOAc to give syrupy 8b (220 mg, 70%).

(b) **9b** from **8b**. A solution of **8b** (180 mg, 0.129 mmol) in AcOH (4 mL) was treated as described for the preparation of **9a**. The residue was eluted from a column of silica gel with 1:1 toluene–EtOAc to give amorphous **9b** (145 mg, 91%); $[\alpha]_D - 3.6^{\circ}$ (*c* 1, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 1.50, 1.95–2.03 (4 s, each 3 H, 4 × Ac), 2.59 (b, 1 H, OH), 3.30–3.32 (m, 1 H, H-5'), 3.42–3.59 (m, 6 H, H-3, H-5, H-6a, H-2', H-3', H-2''), 3.58 (dd, 1 H, $J_{1,2}$ 8.3, $J_{2,3}$ 7.8 Hz, H-2), 3.68–3.76 (m, 5 H, H-6b, H-6'a, H-6'b, H-2'', H-5''), 3.90–4.00 (m, 2 H, H-4, H-4'), 4.13 (dd, 1 H, $J_{5'',6''a}$ 2.16, $J_{6''a,6''b}$ 11.8 Hz, H-6''a), 4.19 (dd, 1 H, $J_{5'',6''b}$ 5.4 Hz, H-6''b), 4.31–5.02 [m, 17 H, H-1, H-1', H-1'', H-4'', NH, $6 \times CH_2$ Ph (d, 1 H, $J_{1'',2''}$ 8.4 Hz, H-1'')], 5.12 (dd, 1 H, $J_{2'',3''} = J_{3'',4''} = 10.4$ Hz, H-3''), 7.21–7.25 (m, 30 H, Ph). Anal. Calcd for C₆₈H₇₇NO₁₉ (1212.4): C, 67.36; H, 6.40; N, 1.15. Found: C, 67.38; H, 6.45; N, 1.38.

Methyl O-(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 4)-6-Obenzoyl-2,3-di-O-benzyl- α -D-galactopyranoside (9c).—(a) A mixture of methyl 6-Obenzoyl-2,3-di-O-benzyl- α -D-galactopyranoside (7c) [23] (159 mg, 0.332 mmol), 5 (443 mg, 0.781 mmol), and NIS (177 mg, 0.797 mmol) was treated as described for the preparation of 9a. The residue was eluted from a column of silica gel with 6:1 toluene-EtOAc to give syrupy 8d (218 mg, 67%).

(b) 9c from 8c. A solution of 9a (165 mg, 0.168 mmol) in AcOH (4 mL) was treated as described for the preparation of 9a. The residue was eluted from a column of silica gel with 1:1 toluene–EtOAc to give syrupy 9c (124 mg, 91%); $[\alpha]_D$ +15.2° (*c* 1, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 1.51, 1.93, 1.96, 2.00 (4 s, each 3 H, 4 × Ac), 3.26 (s, 3 H, OCH₃), 3.53–3.55 (m, 1 H, H-5'), 3.78 (dd, 1 H, $J_{1,2}$ 3.3, $J_{2,3}$ 9.9 Hz, H-2), 3.92–4.09 (m, 6 H, H-3, H-4, H-5, H-2', H-6'a, H-6'b), 4.28 (dd, 1 H, $J_{5,6a}$ 11.8, $J_{6a,6b}$ 7.1 Hz, H-6a), 4.55–4.69 [m, 8 H, H-1, H-6b, H-1', H-3', 2 × CH₂Ph (d, 1 H, $J_{1',2'}$ 9.0 Hz, H-1')], 4.99 (dd, 1 H, $J_{3',4'} = J_{4',5'} = 9.7$ Hz, H-4'), 5.59 (dd, 1 H, $J_{2',NH}$ 9.6 Hz, NH), 7.24–7.97 (m, 15 H, Ph). Anal. Calcd for C₄₂H₄₉NO₁₅ (807.8): C, 62.44; H, 6.11; N, 1.73. Found: C, 62.07; H, 6.12; N, 1.72.

(c) 9c from 11c. To a solution of 11c (80 mg, 0.081 mmol) in Ac_2O (5 mL) was added freshly activated Zn-powder (300 mg); the mixture was stirred at room temperature for 5 h and then worked up as described for 9a to yield 9c as syrup (55 mg, 84%).

Allyl O-(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)-($1 \rightarrow 2$)-3,4,6-tri-O-benzyl- α -D-mannopyranoside (9d).—(a) Synthesis of 8d. A solution of allyl 3,4,6-tri-O-benzyl- α -D-mannopyranoside (7d) [24] (175 mg, 0.377 mmol), 5 (320 mg, 0.564 mmol), and NIS (135 mg, 0.603 mmol) was treated as described for the preparation of 9a. The residue was eluted from a column of silica gel with 6:1 toluene-EtOAc to give syrupy 8d (197 mg, 54%).

(b) 9d from 8d. A solution of 8d (197 mg, 0.204 mmol) in AcOH (4 mL) was treated as described for the preparation of 9a. The residue was eluted from a column of silica gel with 1:1 toluene–EtOAc to give syrupy 9d (139 mg, 86%); $[\alpha]_D + 14.6^{\circ}$ (*c* 1, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 1.73, 1.98, 1.99, 2.01 (4 s, each 3 H, 4 × Ac), 3.27 (ddd, 1 H, $J_{1'.2'}$ 8.2, $J_{2'.NH}$ 7.3, $J_{2'.3'}$ 9.5 Hz, H-2'), 3.63–4.14 (m, 10 H, H-2, H-3, H-4, H-5, H-6a, H-6b, H-5', H-6'a, $CH_2CH = CH_2$), 4.24 (dd, 1 H, $J_{5'.6'b}$ 5.2, $J_{6'a,6'b}$ 12.2 Hz, H-6'b), 4.45–4.91 (m, 6 H, $3 \times CH_2Ph$), 4.77 (d, 1 H, $J_{1,2}$ 1.2 Hz, H-1), 5.00–5.22 [m, 3 H, H-1', $CH_2CH = CH_2$ (d, 1 H, H-1')], 5.44 (b, 1 H, NH), 5.73 (dd, 1 H, $J_{2'.3'} = J_{3'.4'} = 9.5$ Hz, H-3'), 5.81–5.85 (m, 1 H, $CH_2CH = CH_2$), 7.21–7.34 (m, 15 H, Ph). Anal. Calcd for $C_{44}H_{53}NO_{14}$ (828.9): C, 63.75; H, 6.44; N, 1.68. Found: C, 63.81; H, 6.48; N, 1.55.

(c) 9d from 11d. To a solution of 10d (310 mg, 0.297 mmol) in Ac_2O (5 mL) was added freshly activated Zn-powder (300 mg); the mixture was stirred at room temperature for 6 h and then worked up as described for 9a to yield 9d as syrup (209 mg, 85%).

8-(Methoxycarbonyl)octyl O-(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 3)-(4-O-acetyl-2,6-di-O-benzoyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-2,3,6tri-O-benzoyl- β -D-glucopyranoside (10eA) and 8-(methoxycarbonyl)octyl O-(2acetamido-3,4,6-O-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)-(1 \rightarrow 4)-(3-O-acetyl-2,6di-O-benzoyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-2,3,6-tri-O-benzoyl- β -D-glucopyranoside (10eB).—(a) Synthesis of 8e. A mixture of 8-(methoxycarbonyl)octyl O-(2,6-di-O-benzoyl- β -D-galactopyranosyl)-(1 \rightarrow 4)-2,3,6-tri-O-benzoyl- β -D-glucopyranoside (7e) [25] (160 mg, 0.157 mmol), 5 (142 mg, 0.251 mmol), and NIS (58 mg, 0.264 mmol) was treated as described for the preparation of 9a. The residue was eluted from a column of silica gel with 3:1 toluene-EtOAc to give syrupy 8eA and 8eB (164 mg, 65%).

(b) 9eA,B from 8eA,B. A solution of 8eA,B (164 mg, 0.101 mmol) in AcOH (5 mL) was treated as described for the preparation of 9a. The residue was eluted from a column of silica gel with 2:3 toluene-EtOAc to give amorphous 9eA,B (134 mg, 92%).

 $J_{5,6b}$ 1.2, $J_{6a,6b}$ 12.0 Hz, H-6b), 4.57 (d, 1 H, $J_{1',2'}$ 8.0 Hz, H-1'), 4.58 (d, 1 H, $J_{1,2}$ 7.9 Hz, H-1), 4.71 (d, 1 H, $J_{1'',2''}$ 8.2 Hz, H-1"), 4.91 (dd, 1 H, $J_{3'',4''} = J_{4'',5''} = 9.6$ Hz, H-4"), 4.95 (d, 1 H, $J_{2",NH}$ 8.3 Hz, NH), 5.13 (t, 1 H, $J_{2",3"} = J_{3",4"} = 9.5$ Hz, H-3"), 5.28 (t, 1 H, $J_{3',4'} = J_{4',5'} = 3.4$ Hz, H-4'), 5.32 (dd, 1 H, $J_{1',2'}$ 8.0, $J_{2',3'}$ 9.6 Hz, H-2'), 5.37 (dd, 1 H, $J_{1,2}$ 7.9, $J_{2,3}$ 9.5 Hz, H-2), 5.69 (dd, 1 H, $J_{2,3} = J_{3,4} = 9.5$ Hz, H-3), 7.25-8.06 (m, 25 H, Ph); **10eB:** δ 0.97-2.20 [m, 29 H, (CH₂)₇CH₂COOCH₃, 5 × Ac], 3.53 (m, 1 H, H-2"), 3.54-3.78 [m, 10 H, H-5, H-5', H-6'a, H-5", H-6"a, (CH₂)₇CH₂COOCH₃, OCH₃), 3.91 (dd, 1 H, J_{5",6"b} 4.4, J_{6"a,6"b} 12.1 Hz, H-6"b), 3.99 (dd, 1 H, $J_{3',4'} = J_{4',5'} = 2.2$ Hz, H-4'), 4.12 (dd, 1 H, $J_{5',6'b}$ 6.0, $J_{6'a,6'b}$ 11.1 Hz, H-6'b), 4.21 (t, 1 H, $J_{3,4} = J_{4,5} = 9.5$ Hz, H-4), 4.35 (dd, 1 H, $J_{5,6a}$ 4.4, $J_{6a,6b}$ 12.0 Hz, H-6a), 4.56 (dd, 1 H, J_{5,6b} 1.5, J_{6a,6b} 12.0 Hz, H-6b), 4.60 (d, 1 H, J_{1,2} 8.2 Hz, H-1), 4.65 (d, 1 H, $J_{1',2'}$ 7.8 Hz, H-1'), 4.81 (dd, 1 H, $J_{3'',4''} = J_{4'',5''} = 9.7$ Hz, H-4"), 4.93 (dd, 1 H, $J_{2',3'}$ 10.4, $J_{3',4'}$ 2.5 Hz, H-3'), 5.10 (d, 1 H, $J_{1'',2''}$ 8.0 Hz, H-1"), 5.38–5.39 (m, 2 H, H-2, H-2'), 5.68 (dd, 1 H, $J_{2,3} = J_{3,4} = 9.5$ Hz, H-3), 5.94 (dd, 1 H, $J_{2'',3''} = J_{3'',4''} = 10.5$ Hz, H-3"), 6.51 (d, 1 H, $J_{2",NH}$ 6.5 Hz, NH), 7.34–8.03 (m, 25 H, Ph). **10eA**: $[\alpha]_D$ +25.5° (c 1, CHCl₃). Anal. Calcd for C₇₃H₈₁NO₂₇ (1404.3): C, 62.43; H, 5.81; N, 0.95. Found: C, 62.90; H, 5.87; N, 1.12. **10eB**: $[\alpha]_{D}$ + 32.9° (c 1, CHCl₃). Anal. Calcd for C₇₃H₃₁NO₂₇ (1404.3): C, 62.43; H, 5.81; N, 0.95. Found: C, 62.29; H, 6.00; N, 1.10.

Methyl $O_{13}^{A,6-tri-O-acetyl-2-deoxy-(2,2,2-trichloroethoxycarbonylamino)-\beta-D-glucopyranosyl]-(1 <math>\rightarrow$ 6)-2,3,4-tri-O-benzyl- α -D-glucopyranoside (11a). — TMSOTf (47 μ L of a 0.1 M solution of TMSOTf in dry CH₂Cl₂, 4.7 μ mol) was added to a solution of **7a** [19] (170 mg, 0.366 mmol) and **6** (300 mg, 0.476 mmol) in dry CH₂Cl₂ (3 mL) under dry Ar at room temperature. The mixture was stirred for 15 min, neutralized with triethylamine, and concentrated in vacuo. The residue was eluted from a column of silica gel with 2:1 PE-EtOAc to give syrupy **11a** (329 mg, 97%); [α]_D + 15.0° (*c* 1, CHCl₃); ¹H NMR (250 MHz, CDCl₃): δ 1.99–2.04 (3 s, each 3 H, 3 × OAc), 3.36 (s, 3 H, OCH₃), 3.43–3.76 (m, 5 H, H-3, H-4, H-5, H-2', H-5'), 3.94–4.20 (m, 5 H, H-2, H-6a, H-6b, H-6'a, H-6'b), 4.39 (d, 1 H, $J_{1',2'}$ 8.3 Hz, H-1'), 4.52–5.02 (m, 11 H, H-1, H-4', CH₂CCl₃, 3 × CH₂Ph, NH), 5.19 (dd, 1 H, $J_{2',3'} = J_{3',4'} = 9.3$ Hz, H-3'), 7.30–7.39 (m, 15 H, Ph). Anal. Calcd for C₄₃H₅₀Cl₃NO₁₅ (927.2): C, 54.63; H, 5.33; N, 1.48. Found: C, 54.76; H, 5.41; N, 1.58.

Methyl O-[3,4,6-tri-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)-β-Dglucopyranosyl]-(1 → 4)-2,3-di-O-benzyl-6-O-benzoyl-α-D-galactopyranoside (11c).—A mixture of 7c [21] (94 mg, 0.196 mmol) and 6 (155 mg, 0.255 mmol) was treated as described for the preparation of 11a. The residue was eluted from a column of silica gel with 2:1 PE-EtOAc to give syrupy 11c (141 mg, 76%); $[\alpha]_D + 12^\circ$ (c 1, CHCl₃); ¹H NMR (250 MHz, CDCl₃): δ 1.97, 2.00, 2.03 (3 s, each 3 H, 3 × OAc), 3.31 (s, 3 H, OCH₃), 3.49-3.52 (m, 1 H, H-5'), 3.71 (ddd, 1 H, J_{1',2'} 8.7, J_{2',3'} 9.5, J_{2',NH} 7.5 Hz, H-2'), 3.86-5.01 [m, 18 H, H-1, H-2, H-3, H-4, H-5, H-6a, H-6b, H-1', H-3', H-4', H-6'a, H-6'b, CH₂CCl₃, 2 × CH₂Ph, (d, 1 H, J_{1',2'} 8.7 Hz, H-1')], 5.57 (d, 1 H, J_{2',NH} 7.5 Hz, NH), 7.27-8.00 (m, 15 H, Ph). Anal. Calcd for C₄₃H₄₈Cl₃NO₁₆ (941.2): C, 53.91; H, 5.04; N, 1.48. Found: C, 53.95; H, 5.02; N, 1.72.

Allyl O-[3,4,6-tri-O-acetyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)- β -Dglucopyranosyl]-(1 \rightarrow 2)-3,4,6-tri-O-benzyl- α -D-mannopyranoside (11d).—This compound was obtained in the same yield as previously described [13].

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