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Lysine-Derived Protein-bound Heyns Compounds in Bakery Products

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1 Abstract

2 Fructose and dicarbonyl compounds resulting from fructose in heated foods have been linked 3 to pathophysiological pathways of several metabolic disorders. Up to now, very little is known about the Maillard reaction of fructose in food. Heyns rearrangement compounds 4 5 (HRCs), the first stable intermediates of the Maillard reaction between amino components and 6 fructose, have not yet been quantitated as protein-bound products in food. Therefore, the 7 HRCs glucosyllysine and mannosyllysine were synthesized and characterized by NMR. 8 Protein-bound HRCs in cookies containing various sugars and in commercial bakery products 9 were quantitated after enzymatic hydrolysis by RP-HPLC-ESI-MS/MS in the multiple 10 reaction monitoring mode through application of the standard addition method. Protein-bound 11 HRCs were quantitated for the first time in model cookies and in commercial bakery products 12 containing honey, banana and invert sugar syrup. Concentrations of HRCs from 19 to 13 287 mg/kg were found, which were similar to or exceeded the content of other frequently 14 analyzed Maillard reaction products, such as N- ε -carboxymethyllysine (10–76 mg/kg), N- ε -15 carboxyethyllysine (2.5–53 mg/kg) and methylglyoxal-derived hydroimidazolone 1 (10– 16 218 mg/kg) in the analyzed cookies. These results show that substantial amounts of HRCs 17 form during food processing. Analysis of protein-bound HRCs in cookies is therefore useful 18 to evaluate the Maillard reaction of fructose.

19 Keywords

20

20 Maillard reaction; glycation; fructose; Heyns compound; bakery product; HPLC-MS/MS

22 Introduction

23 During storage and thermal processing of food, reducing sugars and amino components may 24 react to ultimately form aroma and color compounds due to the Maillard reaction. Aldoses, 25 such as glucose, react with amino groups to Amadori rearrangement compounds (ARCs) as 26 the first stable intermediates. The formation of ARCs is well described, and quantitation of 27 lysine derivatization is widely used to estimate the extent of early stage Maillard reactions in food and in physiological systems.¹ By contrast, the reaction of amino components and 28 29 ketoses in food has hardly been investigated. Analogous to ARCs, the epimeric Heyns 30 rearrangement compounds (HRCs, figure 1) N-E-glucosyllysine and N-E-mannosyllysine are the first stable intermediates of the Maillard reaction of fructose and lysine.² Both HRCs and 31 32 ARCs degrade to 1,2-dicarbonyl-compounds, which react with amino components under 33 formation of advanced glycation endproducts (AGEs). N- ε -carboxymethyllysine (CML) is a major AGE on lysine side chains.³ It is formed via oxidative degradation of ARCs and 34 proposedly of HRCs⁴ and during the reaction of lysine and glyoxal.⁵ The reaction of lysine 35 36 and methylglyoxal leads to the formation of N- ε -carboxyethyllysine (CEL). A typical AGE on 37 arginine side chains is methylglyoxal-derived hydroimidazolone 1 (MG-H1) which originates from the reaction of arginine and methylglyoxal.⁶ 38

Heyns was the first to describe the synthesis of the products named after him.² Free HRCs 39 were found in stored lyophilized apricots,⁷ tobacco,⁸ raw licorice,⁹ tomato powder¹⁰ and 40 garlic¹¹ via paper chromatography,^{7,8} thin layer chromatography,⁹ cation exchange 41 chromatography with refractive index detection¹⁰ and HPLC-MS/MS.¹¹ Reduced protein-42 bound HRCs were found in human ocular lens protein after acid reduction and hydrolysis 43 44 using RP-HPLC.¹² By adding N- α -hippuryllysine to a cookie dough, it was shown that peptide-bound HRCs form under baking conditions.¹³ However, the direct analysis of protein-45 46 bound HRCs in food has yet to be described. Since the use of high fructose corn syrup

(HFCS) has replaced sucrose in the United States by almost 50% in the last 40 years, free
fructose has become far more available in foods.¹⁴ Thus, the potential of fructose-induced
Maillard reaction in food has increased. To evaluate the extent of the early stage of the
Maillard reaction in foods containing honey, HFCS and fruits, analysis of protein-bound
HRCs is necessary.

The essential amino acid lysine is not available for nutrition, when bound in ARCs and AGEs, however, the human colonic microbiota is able to degrade fructosyllysine, CML and pyrraline.¹⁵ Dietary AGEs are discussed to have adverse metabolic effects through their accumulation in human tissue and the postulated interaction with AGE-receptors, which leads to inflammation.¹⁶ Thus, dietary HRCs and other fructose derived MRP could have pathophysiological effects due to the loss of lysine and because of the inflammatory effects of dietary AGEs.

It was therefore the purpose of this study to obtain information about the role of fructose inglycation, more precisely to quantitate protein-bound HRCs in food.

61

62 Materials and methods

63 Chemicals. Ethanol and HPLC gradient grade acetonitrile were purchased from VWR 64 Prolabo (Leuven, Belgium). Nonafluoropentanoic acid (NFPA), N.N-dimethylformamide 65 (DMF, <0.01% water), deuterium oxide, hydrochloric acid, ammonia solution (25%), 66 prolidase (138 U/mg protein), pepsin (3555 U/mg protein), and leucine aminopeptidase 67 (18 U/mg protein) from Sigma-Aldrich were employed (Steinheim, Germany) and acetic acid, 68 fructose, glucose, and sucrose from Roth (Karlsruhe, Germany). N-α-tert-69 butyloxycarbonyllysine (Boc-lysine), petroleum ether (boiling point range: 40-60 °C), and 70 tris(hydroxymethyl)aminomethane (TRIS) were obtained from Fluka (St Gallen, Switzerland) 71 and triphenyl tetrazolium chloride from Kallies (Sebnitz, Germany). Pronase E (4000 PU/mg 72 protein) from Merck (Darmstadt, Germany) was used and pyridine from Acros (Geel, 73 Belgium). Formic acid and sodium hydroxide were purchased from Grüssing (Filsum, 74 Germany) and ninhydrin from Serva (Heidelberg, Germany). The water used for the 75 preparation of solutions and buffers was prepared by a Purelab Plus purification system (USF 76 Elga, Ransbach-Baumbach, Germany).Water for HPLC-MS/MS was obtained by a Bi 18 E double distillation system (QCS, Maintal, Germany). Fructosyllysine,¹⁷ CML¹⁸, CEL¹⁸ and 77 MG-H1¹⁸ (Figure 1) were synthesized and characterized in our laboratory according to the 78 79 literature stated. The substances met the spectroscopic and chromatographic characteristics 80 published in the respective protocols.

81

82 Synthesis and Isolation of N-E-(2-Deoxy-D-glucosyl)-L-lysine (Glucosyllysine) and N-E-(2-**Deoxy-D-mannosyl)-L-lysine (Mannosyllysine).** The compounds were synthesized^{2,13} and 83 isolated 17,19 according to literature methods with minor modifications. Boc-lysine (0.62 g. 84 2.5 mmol) and anhydrous fructose (4.5 g, 14 mmol) were refluxed in 37 mL of dry 85 86 dimethylformamide for one hour. After evaporation of the solvent, the residue was dissolved 87 in 100 mL of water. Black precipitate was removed by filtration. The pH value was adjusted 88 to 2.0 with 6 M hydrochloric acid. For removal of the Boc protecting group and uncharged 89 by-products, the solution was applied to a 200x25 mm glass column filled with 80 mL of cation exchange resin Lewatit S 100 (20–50 mesh) in the H⁺ form, previously equilibrated 90 91 with 250 mL of 6 M hydrochloric acid and 250 mL of water. The sugar was eluted with 92 250 mL of water and after one night of incubation, material bound to the exchanger was 93 eluted with 250 mL of 2 M ammonia. The eluate was evaporated to dryness and dissolved in 94 20 mL of 0.1 M pyridinium formate buffer, pH 3.0. The pH value was adjusted to 3.0 with 95 formic acid. The solution was applied to a 750x15 mm semipreparative ion-exchange column 96 filled with DOWEX 50 W-X8 (100-200 mesh) in the pyridinium form, previously

97 equilibrated with 360 mL of 6 M hydrochloric acid, 360 mL of water, 360 mL of 2 M aqueous 98 pyridine, 360 mL of water and 360 mL of 0.1 M pyridinium formate buffer (pH 3.00). 99 Separation was achieved with elution by gravity of 360 mL of 0.25 M pyridinium acetate 100 buffer (pH 4.05) and 360 mL of 0.3 M pyridinium acetate buffer (pH 4.05) at a flow rate of 101 0.4 mL/min. A fraction collector (RediFrac, Pharmacia Biotech, Sweden) was used and 102 fractions of 10 mL were collected. Of each fraction, 1 μ L was spotted on two TLC plates. 103 Spraying with 1% triphenyl tetrazolium chloride in 1 M sodium hydroxide and 0.1% 104 ninhydrin in ethanol, respectively, indicated that the HRCs eluted with 70–220 mL of the last 105 buffer. An aliquot of those fractions (5 μ L) was diluted in 495 μ L of the loading buffer and 106 subjected to amino acid analysis. Fractions eluting with 70-190 mL of 0.3 M pyridinium 107 acetate buffer (pH 4.05) contained glucosyllysine, and mannosyllysine eluted with 130-108 220 mL of this buffer. The fractions that contained only one HRC were pooled, dried in 109 vacuo, repeatedly dissolved in water and evaporated, lyophilized and characterized.

110 N- ε -(2-deoxy-D-glucosyl)-L-lysine (glucosyllysine): HPLC-MS/MS: see Figure 2 A; 111 elemental analysis: C₁₂H₂₄N₂O₇ (MW = 308.32), calculated: C 46.75%, H 7.85%, N 9.09%; 112 found: C 39.59%, H 6.84%, N 7.75%; content = 76.3%, based on nitrogen and 113 chromatographic purity from LC-MS/MS (ratio of glucosyllysine to mannosyllysine and 114 fructosyllysine). Yield = 281 mg (molar yield = 27.8%). NMR data: see Table 1 and 115 supporting information Figure S1–S5.

116 *N*-ε-(2-deoxy-D-mannosyl)-L-lysine (mannosyllysine): HPLC-MS/MS: see Figure 2 B; 117 elemental analysis: $C_{12}H_{24}N_2O_7$ (MW = 308.32), calculated: C 46.75%, H 7.85%, N 9.09%; 118 found: C 41.19%, H 7.44%, N 6.91%; content = 64.5%, based on nitrogen and 119 chromatographic purity from LC-MS/MS (ratio of mannosyllysine to glucosyllysine and 120 fructosyllysine). Yield = 23 mg (molar yield = 2.1%). NMR data: see Table 2 and supporting 121 information Figure S6–S10.

122

Characterization of Glucosyllysine and Mannosyllysine. ¹H and ¹³C NMR spectra were 123 124 recorded on an Avance III HDX 500 MHz Ascend instrument from Bruker (Rheinstetten, 125 Germany) at 500.13 MHz and 125.75 MHz, respectively. Glucosyllysine (7.1 mg) and 126 mannosyllysine (5.4 mg) were each dissolved in 770 μ L of deuterium oxide and stored at 127 room temperature overnight to allow mutarotation. All chemical shifts are given in parts per 128 million (ppm), those of protons relative to the internal HOD signal (4.70 ppm), those of carbon atoms relative to external standard tetramethylsilane. Assignments of ¹H and ¹³C 129 signals are based on ¹H-¹H correlation spectroscopy (COSY), ¹H-¹³C heteronuclear single 130 quantum coherence (HSQC); ¹H-¹³C heteronuclear multiple bond correlation (HMBC) and 131 132 ¹³C distortionless enhancement by polarization transfer (DEPT) experiments. Elemental 133 analysis was performed on a vario MICRO cube CHNS elemental analyzer (Elementar, 134 Hanau, Germany).

135

136 Bakery Products. All bakery products, wheat flour, margarine, honey, bananas and baking 137 powder were purchased in local retail stores. Cookie doughs were prepared from 10.00 g of 138 wheat flour, 5.90 g of margarine, 0.30 g of baking powder, 7.5 mL of water and 5.00 g of 139 either fructose, glucose, or sucrose. For a honey cookie dough, 6.60 g of honey and 5.5 mL of 140 water and for a banana cookie dough, 10.00 g of banana and 3.20 g of sucrose were added to 141 wheat flour, margarine and baking powder. Cookies were formed manually to a thickness of 142 about 5 mm and a diameter of about 5 cm. The cookies were baked in a laboratory drying 143 oven for 10 min at 175 °C. After cooling, the cookies were homogenized by mortar and 144 pestle. A sample of 160 mg of a baked or purchased cookie was weighed into a 2 mL plastic tube and defatted using 1 mL of petroleum ether as described previously.¹³ The sample was 145 146 mixed vigorously by vortex, centrifuged (10,600 x g, 10 min) and the petroleum ether was 147 discarded. After repeating this procedure, 500 μ L of water was added to the precipitate. The suspension was mixed by vortex and dialyzed against water at 4 °C for 2 days (cellulose 148

tubings, MWCO, 14 kDa, Sigma-Aldrich). The water was changed twice per day. Theretentates were lyophilized.

151

152 Enzymatic Hydrolysis. Samples containing 2–3 mg of protein (8–40 mg of lyophilized cookie, depending on protein content) were used according to the literature.²⁰ Hvdrochloric 153 154 acid (1.05 mL, 0.02 M) containing 356 U of pepsin was added. After 24 h incubation in a 155 drying chamber at 37 °C, 300 µL of 2 M TRIS buffer, pH 8.2 containing 400 PU of pronase E 156 was added. Following 24 h of further incubation, 22 μ L of 2 M TRIS buffer containing 0.3 U 157 of leucine-aminopeptidase and 2 U of prolidase was added. After 48 h, the samples were 158 frozen. Centrifugation (10,600 x g, 10 min) and membrane filtration (0.45 μ m) of thawed 159 samples were performed prior to HPLC analysis. An aliquot of the hydrolysate (150 μ L) was 160 suspended in 450 μ L of the loading buffer for amino acid analysis (0.12 N lithium citrate, pH 161 2.20). After membrane filtration (0.45 μ m), samples were subjected to amino acid analysis.

162

163 Amino Acid Analysis. Amino acids separation was performed on a PEEK column filled with 164 the cation exchange resin LCA K07/Li (150 mm \times 4.6 mm, 7 μ m) using the amino acid 165 analyzer S 433 (Sykam, Fürstenfeldbruck, Germany) with a gradient program utilized previously.²¹ Loading and running buffers for this lithium system were purchased from 166 167 Sykam. After post-column derivatization with ninhydrin, the absorbance of the effluent was 168 recorded with a two-channel photometer simultaneously working at 440 nm and 570 nm, respectively. A commercial amino acid mixture was obtained (Sigma-Aldrich, Steinheim, 169 170 Germany) and used for external calibration.

171

High-Pressure Liquid Chromatography (HPLC) with Tandem Mass Spectrometric
 Detection. Protein-bound MRPs were analyzed as described before^{15,21,22} on a high pressure
 gradient system 1200 Series (Agilent Technologies, Böblingen, Germany), consisting of a

175 binary pump, an online degasser, an autosampler, a column oven, and a diode array detector. 176 Chromatographic separation was performed on a Zorbax SB C-18 column (50 mm x 2.1 mm, 177 $3.5 \,\mu\text{m}$, Agilent), with 10 mM NFPA in double-distilled water (solvent A) and 10 mM NFPA 178 in acetonitrile (solvent B) as the mobile phases. The injection volume was 2 μ L. Analysis was 179 conducted at 35 °C using a flow rate of 0.25 mL/min. A gradient of 5% B (0 min) to 32% B 180 (15 min), then to 85% B (16–19 min), and finally to 5% B (20 min) was used. An 181 equilibration time of 8 min before the next run was implemented. The HPLC was coupled to 182 the mass spectrometer 6410 Triple Quad (Agilent) working in the positive mode with a source 183 temperature of 350 °C and a capillary voltage of 4000 V. Nitrogen was used as the nebulizing 184 gas (nitrogen generator 5183-2003, Agilent) with a flow rate of 11 L/min and a pressure of 185 35 psi. The multiple-reaction monitoring mode (MRM) was used for mass spectrometric 186 detection with the conditions listed in table 3. These conditions were optimized by scan and 187 product ion scan measurements of the respective standards (Figure 2). Data acquisition and 188 evaluation was conducted with the software Mass Hunter B.02.00 (Agilent).

189 The MRPs were quantitated by the standard addition method with three runs. For each run, 190 100 μ L of enzymatic hydrolysate was mixed either with 20 μ L of water, or with 10 μ L of 191 water and 10 μ L of a standard solution, or with 20 μ L of standard solution. The standard 192 solution contained N- ϵ -glucosyllysine (19.0 μ g/mL), N- ϵ -fructosyllysine (70.1 μ g/mL), CML 193 (5.0 μ g/mL), CEL (5.0 μ g/mL), and MG-H1 (15.0 μ g/mL) dissolved in water. Limits of 194 detection (LOD) and quantitation (LOQ) for the compounds are shown in table 4. For a 195 matrix calibration, the standard solution was added to an enzymatic hydrolysate of water. 196 Concentration ranges of the calibration are shown in table 4.

197

198 Statistical Treatment. All samples were analyzed at least in duplicate. Baking experiments 199 were done one time. Limits of detection (LOD) and quantitation (LOQ) were calculated as the 200 concentrations of MRPs necessary to show peaks at signal-to-noise ratios of 3 and 10, 9 201 respectively. Coefficients of determination were calculated for standard addition regression 202 lines. The intercept accuracy was calculated as the percentage difference between the area of 203 the sample peak without addition of standards and the intercept of the regression line. The 204 linearity of the matrix calibration was determined with the Mandel's fitting test for $\alpha = 0.05$.

205

206 **Results and discussion**

207 Synthesis of Heyns Rearrangement Compounds (HRCs). The aim of this study was to 208 quantitate protein-bound HRCs in bakery products containing fructose in order to evaluate the 209 early stage of the Maillard reaction with fructose. Therefore, the epimeric HRCs 210 glucosyllysine and mannosyllysine (Figure 1) needed to be synthesized and characterized as 211 reference material for chromatographic analysis. Protocols for synthesis of HRCs were adapted^{2,13} with some modifications, whereas isolation and separation were performed as 212 described for fructosyllysine¹⁷ and formyline.¹⁹ The peaks of glucosyllysine and 213 214 mannosyllysine overlapped during semi-preparative ion-exchange chromatography as 215 determined by amino acid analysis. Only fractions containing one product were pooled and 216 processed further. One- and two-dimensional NMR spectroscopy (Tables 1 and 2, supporting 217 information Figures S1–S10) confirmed the structural identity of the compounds. Residual 218 acetic acid from the chromatography eluent was identified as an impurity. Chemical shifts of glucosyllysine were in accordance with those published in the literature.^{13,23,24} In pyranose 219 220 sugars, coupling constants $J_{1,2}$ of two vicinal axial protons are higher than 7 Hz, while the 221 coupling constants of vicinal protons of which at least one is in equatorial position are smaller than 5 Hz.25 Mannosyllysine had similar chemical shifts and coupling constants as 222 223 glucosyllysine. However, the pyranosyl form of glucosyllysine has an axial proton at C₂ and 224 therefore the coupling constant of the protons at C₂ and C₃ is 10.5 Hz and that of the protons at C_2 and C_1 of the β -pyranosyl moiety is 8.5 Hz. In mannosyllysine those coupling constants 225

were 4.6 and 1.7 Hz, indicating a proton in equatorial position at C₂. Based on this findings 226 and by comparison to similar structures,²⁶ mannosyllysine and glucosyllysine were identified. 227 228 Glucosyllysine was found in α - and β -pyranose isomers (ca. 75% and 25%). In addition to α -229 and β -pyranose isomers (ca. 46% and 25%), mannosyllysine also formed considerable amounts of α - and β -furance isomers (ca. 18% and 9%). The H-1 ¹H NMR signal of the α -230 231 furanose form of mannosyllysine overlapped with the H-1 signal of emerging glucosyllysine 232 which may have been formed during the period of time between dissolution of the compound and the NMR measurement.²⁷ Therefore, the percentage of α -furanose might be 233 234 overestimated.

235 The reaction mixture before ion exchange chromatography contained the HRCs 236 glucosyllysine and mannosyllysine, but also the ARC fructosyllysine. After isolation, 76% of 237 glucosyllysine, 3% of mannosyllysine and 6% of fructosyllysine were present in the 238 glucosyllysine isolate as deduced from amino acid analysis, HPLC-MS/MS and elemental 239 analysis. The mannosyllysine isolate contained 65% of mannosyllysine, 6% of glucosyllysine 240 and 6% of fructosyllysine. Though concentrations of mannosyllysine were higher than those 241 of glucosyllysine in the synthesis mixture after heating (molar yield glucosyllysine: ca. 30%, 242 mannosyllysine: ca. 50%), concentrations of mannosyllysine in the eluted fractions and the 243 vield of isolated mannosyllysine were more than 10 times smaller than those of glucosyllysine 244 (molar yield glucosyllysine: 27.8%, mannosyllysine: 2.1%). Sugars with lower shares of the 245 pyranose forms, such as mannosyllysine, show higher concentrations of the open-chain form 246 and are therefore more reactive via the Maillard reaction and the Lobry de Bruyn-Alberda van Ekenstein transformation.²⁸ Mannosyllysine, therefore, is less stable than glucosyllysine 247 248 during the isolation process. Yields and purity of both substances were high enough for 249 chromatographic analysis.

250

251 Determination of HRCs in Bakery Products. A method to quantitate the HRCs 252 glucosyllysine and mannosyllysine, as well as the ARC fructosyllysine and the AGEs CML, 253 CEL, and MG-H1, in bakery products was established using enzymatic hydrolysis and LC-254 MS/MS. Enzymatic hydrolysis has already been used for the quantitation of different MRPs 255 such as fructosyllysine, pyrraline and CML.^{21,29} The rate of release after enzymatic digestion 256 was at least 80% of the rate after acid hydrolysis for aliphatic amino acids in roasted peanuts and dried pasta.^{29,30} In eye lens protein, the rate of release after enzymatic digestion compared 257 to acid hydrolysis was 52–91% for CML.³¹ The quantitative data presented in this study might 258 259 therefore be slightly underestimated. LC-MS/MS analysis using the standard addition method 260 of enzymatically hydrolyzed standard solutions of glucosyllysine and mannosyllysine showed 261 that mannosyllysine is prone to interconversion during enzymatic hydrolysis. Of 262 mannosyllysine, only $29 \pm 8\%$ (SD, n=3) could be recovered unchanged. The larger part of 263 $63 \pm 7\%$ had been transformed to glucosyllysine and $9 \pm 1\%$ were missing. On the other hand 264 glucosyllysine is stable with a recovery of $88 \pm 3\%$. The missing 12 % of glucosyllysine had 265 been transformed to mannosyllysine. Traces of fructosyllysine were detected, when the 266 glucosyllysine standard was subjected to enzymatic hydrolysis, but CML was not detectable 267 in enzymatic hydrolysates of both HRCs. Thus, the HRCs isomerize during enzymatic 268 hydrolysis, with glucosyllysine as the favored product. As explained above, mannosyllysine is 269 more prone to Lobry de Bruyn-Alberda van Ekenstein transformation, leading to the 270 formation of the more stable glucosyllysine. In the present study, mannosyllysine was only 271 found in traces in bakery products, either due to isomerization to glucosyllysine or due to its 272 absence in the products. Nonetheless, as the quantification of glucosyllysine probably 273 encloses mannosyllysine present in the food product, the results will be given as the sum of 274 both compounds and designated glucosyl/mannosyllysine (LysGlc/Man) this study. Because 275 only trace amounts of mannosyllysine were found in bakery products the 9% loss of 276 mannosyllysine during enzymatic hydrolysis can also be assumed to be negligible.

277 LysGlc/Man analysis was performed via an HPLC-MS/MS method used previously for ARCs and AGEs.^{15,21,22} Product ion spectra of cookie samples were compared to those of standard 278 279 solutions (Figure 2 A, D). Transitions that were both selective and abundant were chosen for 280 quantitation (Table 3). With this HPLC-MS/MS method, simultaneous quantitation of the two 281 lysine HRCs and the ARC fructosyllysine was enabled (Figure 2 E, F). Fructosyllysine and 282 glucosyllysine also had minor peaks at the transitions of the other compound at their peak 283 maxima, but were separated via their retention time (Figure 2 A, C). This led to ca. 15 μ g falsely detected glucosyllysine per mg fructosyllysine and ca. $70 \,\mu g$ falsely detected 284 285 fructosyllysine per mg glucosyllysine. Consequently, for the ratios glucosyllysine to 286 fructosyllysine found in this study, up to 12% of the quantitated amounts might be 287 overestimated. In addition to HRCs and fructosyllysine, the AGEs CML and CEL as major lysine derivatives and MG-H1 as a major arginine derivative⁶ were quantitated. The standard 288 289 addition method was used because stable isotope labeled standards were not available for 290 HRCs. Linearity for the concentration ranges of the standard additions was verified with 291 matrix calibrations of the analytes (concentrations see Table 4). Coefficients of determination 292 of the standard addition regression lines were $r^2 > 0.99$ for 96% of the samples and $r^2 > 0.98$ 293 for all samples. The intercept accuracy was between 85 and 115% for 96% of all samples and 294 between 75 and 125% for all samples. Such intercept accuracy may be caused by overlapping 295 peaks of the fructosyllysine and glucosyllysine transitions. LODs and LOQs were sufficiently 296 low to enable quantitation of all analytes in most samples (Table 4).

297

298 The Maillard Reaction of Fructose in Cookies Baked with Fructose, Glucose, Honey and 299 Banana. In order to simulate the formation of ARC and HRC under baking conditions, model 300 cookies containing either fructose or glucose were baked. In both cookies, protein-bound 301 LysGlc/Man and fructosyllysine were quantitated (Table 5). Of total lysine, 6.8% were 302 modified to fructosyllysine in glucose-containing cookies, while only 1.6% of total lysine 13

303 were modified to LysGlc/Man in fructose-containing cookies. In cookies baked in the 304 presence of hippuryllysine, 33% of hippuryllysine were derivatized to hippuryl-glucosyllysine 305 in a fructose-containing cookie and 63% of hippuryllysine were derivatized to hippurylfructosyllysine in a cookie containing glucose.¹³ While the ratio between ARCs and HRCs is 306 307 similar, the yields are 5 times higher in the hippuryllysine cookies. When hippuryllysine is 308 used as a model amine, all lysine amino groups can be glycated, whereas in protein, some 309 lysine side chains are not or less available for derivatization with fructose, causing lower vields of HRCs.⁴ In raw cookie dough and in sucrose-containing cookies, fructosyllysine was 310 311 found in traces and LysGlc/Man was not detectable or found only in traces. Thus, protein-312 bound HRC form under baking conditions in cookies from fructose in significant 313 concentrations.

314 Fructosyllysine was also found in fructose-containing cookies and LysGlc/Man in cookies 315 containing glucose with yields of 0.8% and 0.2%, respectively. The formation of ARCs in the 316 presence of fructose and of HRCs in the presence of glucose was described before and was 317 explained via the addition of a second amino moiety to the product followed by elimination of 318 the first.^{27,32} These subsequent Amadori and Heyns rearrangement reactions were also observed, when ARCs and HRCs were incubated with additional amino compounds.²⁷ On the 319 320 other hand, the isomerization of glucose to fructose and vice versa through the Lobry de 321 Bruyn-Alberda van Ekenstein transformation could lead to this observation (Figure 3). This 322 transformation is described to take place not only in basic solutions, but also in neutral aqueous glucose solutions^{33,34} and cookies.³⁵ Fructose concentrations in heated glucose 323 solutions rose up to 38% of the initial glucose concentrations.³⁶ This means that HRCs also 324 325 form from glucose and that they play a minor role in the Maillard reaction with glucose as 326 well. Additionally, the extent of Maillard reaction with fructose can be overestimated through 327 this pathway.

328 To obtain information about the formation of HRCs from fructose in the presence of glucose 329 and to simulate realistic baking conditions, additional cookies containing honey and banana 330 were baked under the same conditions and the same water content as the cookies with 331 fructose and glucose. While in honey cookies the sugar content of the fructose and glucose 332 cookie was completely replaced by honey saccharides, in banana cookies only 36% of the 333 sugar was replaced by banana saccharides, since otherwise the dough would have been too 334 fluid. The remaining sugar was replaced by sucrose. In honey and banana, fructose and 335 glucose occur nearly in the ratio 1:1. Differences between reactivity of glucose and fructose 336 were also found when the molar conversion of the parent saccharides was compared: Per mol 337 of added fructose, the fructose cookie contained 0.21 mmol LysGlc/Man, while the honey 338 cookie had 0.15 mmol and the banana cookie had 0.36 mmol LysGlc/Man. In comparison to 339 that, the glucose cookie contained 0.83 mmol fructosyllysine per mol glucose, the honey 340 cookie 1.36 mmol and the banana cookie 1.42 mmol fructosyllysine per mol glucose. 341 Apparently, fructose has a lower potential to form early stage MRPs. Additionally, the HRC 342 formation from fructose occurs in systems containing both fructose and glucose like in 343 cookies with honey or banana and is similar to the HRC formation in systems without 344 glucose.

345 The AGEs CML, CEL and MG-H1 as important lysine and arginine AGEs were analyzed in 346 all cookies to study differences in the advanced Maillard reaction with fructose and glucose. 347 The CML contents were similar in the cookies containing fructose, glucose, honey and 348 banana (Table 5). This indicates that glucose and fructose do not differ significantly in the 349 formation of CML under baking conditions. An explanation might be that fructosyllysine and glucosyllysine similarly degrade to CML.⁴ By contrast, the highest MG-H1 and CEL 350 351 concentrations were found in the cookies with the highest fructose content. In aqueous sugar-352 lysine model systems at above 100 °C, the formation of methylglyoxal from fructose was up to three times higher than the formation from glucose.^{37,38} Higher concentrations of 353 15

354 methylglyoxal explain the increased levels of the methylglyoxal-arginine adduct MG-H1 and 355 the methylglyoxal-lysine adduct CEL. Preliminary experiments have shown that the 356 percentage of arginine modified to MG-H1 and lysine modified to CEL correlates with the 357 concentrations of fructose or HRCs (data not shown). We are currently working on a further 358 exploration of the formation of MG-H1 and CEL from fructose which could be an indicator 359 for the "fructose pathway" of the Maillard reaction. The total loss of detectable lysine was ca. 360 15% in the glucose-containing cookie and ca. 60% in the fructose-containing cookie. While 361 the formation of fructosyllysine, LysGlc/Man, CML, and CEL in the glucose cookie accounts 362 for ca. 48% of the total lysine loss, the formation of the same products accounts for not more 363 than 4% of the total lysine loss in the fructose cookie. Therefore other derivatization or 364 fragmentation pathways have to be of major importance in the Maillard reaction with 365 fructose. Additionally, the nutritional loss of the essential amino acid lysine is much higher in 366 the fructose cookies than in the glucose cookies.

367

368 The Maillard Reaction of Fructose in Commercial Bakery Products. To evaluate HRC 369 formation and the degree of fructose-induced Maillard reaction in commonly consumed food, 370 commercial bakery products containing fructose were analyzed. The different cookies, cakes 371 and gingerbreads contained honey, invert sugar syrup and fruits as sweeteners. Protein-bound 372 HRCs could be quantitated in all fructose containing bakery products in the range of 19 to 373 287 mg/kg. As expected, the lowest concentration of HRCs was found in the banana bakery 374 products which had the lowest amount of fructose (Table 5). The highest HRC level was 375 found in a cookie containing invert sugar syrup and honey. The amounts measured lead to a 376 consumption of 0.6 to 9 mg LysGlc/Man per 30 mg serving of the bakery product. This equals 377 16–90% of the ARC consumption with those bakery products.

Fructosyllysine, CML, CEL, and MG-H1 were also analyzed in the commercial bakery
 products. Contents of fructosyllysine (1500–6400 mg/kg protein), CML, CEL and MG-H1
 16

380 were in similar ranges as stated in the literature for bakery products (fructosyllysine in bakery products: 400-13000 mg/kg protein, calculated from furosine;^{17,39} CML in biscuits: 10-381 26 mg/kg; CEL in biscuits: 4.9–34; MG-H1 in biscuits: 40–369 mg/kg).⁴⁰ Contents of HRC 382 383 exceeded the contents of CML and CEL in all and those of MG-H1 in most analyzed cookies. 384 Therefore, substances of the early stage of Maillard reaction are of major importance in 385 cookies, and ARCs and HRCs are useful indicators for Maillard reaction with glucose and 386 fructose, respectively. To evaluate the relevance of the Maillard reaction with fructose in 387 other fructose containing foods and in pathophysiological pathways, further analysis of HRCs 388 is important. Furthermore, indicators to differentiate the advanced stage of the Maillard 389 reaction with fructose and glucose need to be found.

Fructose, either of dietary origin or formed in the polyol pathway, is discussed as a causative factor for diabetic complications, nonalcoholic fatty liver disease and the metabolic syndrome.^{41,42} Because the formation of HRCs under physiological conditions could be observed,^{4,12} the Maillard reaction with fructose and fructose metabolites is discussed to play a role in the pathophysiological pathways of the diseases mentioned above.^{41,43} The established protocol for analysis of HRCs adapted for pathophysiological material could bring insight into the role of the fructose pathway of the Maillard reaction in those diseases.

397

With the established method and the synthesized HRCs, protein bound HRCs could be quantitated. Analysis of protein-bound HRCs was used to evaluate the early stage of the Maillard reaction with fructose in bakery products. We could show that HRCs form under baking conditions not only in cookies baked with fructose but also in cookies with glucose or both sugars and in commercial bakery products containing honey, banana and invert sugar syrup. Protein-bound HRCs were quantitated in similar or higher levels (19–287 mg/kg) than the AGEs CML (10–76 mg/kg), CEL (2.5–53 mg/kg) and MG-H1 (10–218 mg/kg). Analysis

405 of protein-bound HRCs is therefore suitable to evaluate the Maillard reaction of fructose in406 cookies.

407

408 **Supporting information**

409 1D/2D NMR spectra of glucosyllysine and mannosyllysine. Figure S1: ¹H NMR spectrum,

410 Figure S2: ¹³C NMR spectrum, Figure S3: COSY correlations, Figure S4: HSQC correlations,

411 Figure S5: HMBC correlations of glucosyllysine. Figure S6: ¹H NMR spectrum, Figure S7:

412 ¹³C NMR spectrum, Figure S8: COSY correlations, Figure S9: HSQC correlations, Figure

413 S10: HMBC correlations of mannosyllysine. This material is available free of charge via the

414 Internet at http://pubs.acs.org.

415

416 Abbreviations used

AGE, advanced glycation end product; ARC, Amadori rearrangement compound; CEL, *N*-εcarboxyethyllysine; CML, *N*-ε-carboxymethyllysine; DMF, *N*,*N*-dimethylformamide; HFCS,
high fructose corn syrup; HPLC, high pressure liquid chromatography; HRC, Heyns
rearrangement compound; LOD, limit of detection; LOQ, limit of quantitation; LysGlc/Man,
glucosyl/mannosyllysine; MG-H1, methylglyoxal-derived hydroimidazolone, MRM, multiple
reaction monitoring; MRP, Maillard reaction product; MS, mass spectrometry; NFPA,
nonafluoropentanoic acid; RP, reversed-phase; TRIS, tris(hydroxymethyl)aminomethane

424

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556		22

Figure captions

Figure 1. Structures of the Maillard reaction products analyzed in this study.

Figure 2. Product ion spectra of glucosyllysine (A), mannosyllysine (B), and fructosyllysine (C) obtained during HPLC-MS/MS analysis of standard solutions, and of glucosyllysine in a hydrolyzed cookie sample (D). Operating conditions were those for the transition m/z 309 \rightarrow 273 (Table 3) For (D) 5 µL of NFPA were added to 100 µL of the sample for better separation of glucosyllysine and fructosyllysine. Precursor (italic numbers), quantifier (bold numbers) and qualifier ions are denoted. HPLC-MS/MS chromatograms of a mixture of glucosyllysine and fructosyllysine (1) and a hydrolyzed cookie sample (2) with the transitions m/z 309 \rightarrow 210 (E, glucosyllysine) and m/z 309 \rightarrow 273 (F, fructosyllysine).

Figure 3. Reactions of glucose and fructose with amino components: 1 Lobry de Bruyn-Alberda van Ekenstein transformation, 2 formation of Schiff base and Amadori rearrangement compound, 3 formation of Schiff base and Heyns rearrangement compound, 4 formation of Schiff base and subsequent Heyns and Amadori rearrangement compounds

C Atom	$\delta_{\rm C}$ [ppm]	$\delta_{\rm H}$ [ppm]	$J(\mathbf{x},\mathbf{y})$ [Hz]	¹ H- ¹ H	$^{1}\text{H}-^{13}\text{C}$			
(figure 1)				COSY	HMBC			
				coupling	coupling			
glucose moiety								
1	88.00 ^[a] /93.26	5.46 ^[a] /4.87 ^[b] d	3.6 ^[a] /8.5 ^[b] (1,2)	2 ^[a]	3; 5			
	^[b] (CH)							
2	59.97 ^[a] /2.48 ^{[b}	3.16 ^[a] /2.75 ^[b] dd	3.6 ^[a] /8.5 ^[b] (1,2);	1; 3 ^[a]	1; 4; 12			
	[]] (CH)		10,6 (2,3)					
3	69.99 (CH)	3.83 dd	10.6 (2,3); 9.1 (3,4)	2; 4	2; 5			
4	69.68 (CH)	3.39 dd	9.1 (3,4); 8.6 (4,5)	3	2; 3; 5; 6			
5	71.26 (CH)	3.80 m	-	6	3; 4; 6			
6	60.28 (CH ₂)	3.70 d	5.4 (5,6)	5	5			
lysyl moie	lysyl moiety							
7	174.55 (Ci)	-	-	-	-			
8	54.46 (CH)	3.66 t	6.2	9	7; 9; 10			
9	29.84 (CH ₂)	1.82 m	-	8; 10	7; 8; 10;			
					11			
10	21.57 (CH ₂)	1.40 m	-	9; 11	8; 9; 11;			
					12			
11	25.33 (CH ₂)	1.68 m	-	10; 12	9; 10; 12			
12	45.35 (CH ₂)	3.09 t	7.5	11	2; 10; 11			

Table 1.¹³C and ¹H NMR Spectroscopic Data of N^{ε} -(2-deoxy-D-glucosyl)-L-lysine

Data of [a] α -pyranosyl-conformation (75%) and [b] β -pyranosyl-conformation (25%) of

glucosyllysine; percentage of conformation calculated from the peak-area of the proton at C_1

$\delta_{C}[ppm]$	δ _H [ppm]	J(x,y) [Hz]	¹ H- ¹ H	¹ H- ¹³ C				
			COSY	HMBC				
			coupling	coupling				
mannose moiety								
89.58 ^[a] /91.95 ^[b] /	5.36 ^[a] /5.12 ^[b] /5.47	1.7 ^[a] /1.7 ^[b] /5.	2 ^[a]	3 ^[a] ; 5 ^[a]				
98.75 ^[c] /	$^{[c]}/5.43^{[d]}d$	3 ^[c] /5.5 ^[d]						
93.88 ^[d] / (CH)		(1,2)						
61.09(CH)	3.30 m		1; 3	3				
67.38 (CH)	4.06 dd	4.6 (2,3); 9.5	2;4	4				
		(3,4)						
66.41 (CH)	3.53 dd	9.5 (3,4); 9.5	3; 5	3; 5; 6				
		(4,5)						
71.90 (CH)	3.80 m	-	4; 6	4				
60.27 (CH ₂)	3.76 d	3.9 (5,6)	5	4; 5				
ty				<u> </u>				
174.62 (Ci)	-	-	-	-				
54.53 (CH)	3.68 t	6.2	9	7; 9; 10				
29.94 (CH ₂)	1.82 m	-	8; 10	7; 8; 10; 11				
21.69 (CH ₂)	1.40 m	-	9; 11	8; 9; 11; 12				
26.37 (CH ₂)	1.68 m	-	10; 12	9: 10: 12				
47.21 (CH ₂)	2.95 t	7.7	11	10: 11				
	noiety 89.58 ^[a] /91.95 ^[b] / 98.75 ^[c] / 93.88 ^[d] / (CH) 61.09(CH) 67.38 (CH) 67.38 (CH) 71.90 (CH) 60.27 (CH ₂) ty 174.62 (Ci) 54.53 (CH) 29.94 (CH ₂) 21.69 (CH ₂) 26.37 (CH ₂)	noiety $89.58^{[a]/91.95^{[b]/}}$ $5.36^{[a]/5.12^{[b]/5.47}}$ $98.75^{[c]/}$ $^{[c]/5.43^{[d]} d$ $93.88^{[d]/} (CH)$ $^{[c]/5.43^{[d]} d$ $61.09(CH)$ $3.30 m$ $67.38 (CH)$ $4.06 dd$ $66.41 (CH)$ $3.53 dd$ $71.90 (CH)$ $3.80 m$ $60.27 (CH_2)$ $3.76 d$ ty $174.62 (Ci)$ $71.90 (CH)$ $3.68 t$ $29.94 (CH_2)$ $1.82 m$ $21.69 (CH_2)$ $1.40 m$ $26.37 (CH_2)$ $1.68 m$	Inviety Somethy Somethy <t< td=""><td>Indiana Indiana Indiana</td></t<>	Indiana Indiana				

Table 2.¹³C and ¹H NMR Spectroscopic Data of N^{ε} -(2-deoxy-D-mannosyl)-L-lysine

Data of [a] α -pyranosyl-conformation (46%), [b] β -pyranosyl-conformation (25%), [c] α -

furanosyl-conformation (18%) and [d] β -furanosyl-conformation (9%) of mannosyllysine;

percentage of conformation calculated from the peak-area of the proton at C_1

Time	Compound	Precursor	Product	Fragmentor	Collision	Dwell	$Q/q^{[a]}$
frame		ion [m/z]	ion	voltage [V]	energy	time	
			[m/z]		[eV]	[ms]	
4–8.3 min	CML	205	84	100	20	70	Q
		205	130	100	10	70	q
	CEL	219	84	100	20	70	Q
		219	130	100	10	70	q
	Glucosyllysine	309	180	100	20	100	q
		309	210	100	10	100	Q
	Fructosyllysine	309	110	100	20	100	q
		309	273	100	10	100	Q
8.5–19.5	MG-H1	229	114	90	20	200	Q
min		229	166	90	20	200	q

Table 3. Transitions Recorded during MRM Measurement of MRPs in bakery products.

[a] Q, transition used for quantitation; q, transition used for the confirmation of the presence

of the analyte.

Table 4. Performance Parameters of the HPLC-MS/MS Method used for the Analysis ofMRPs in bakery products.

MRP	LOD ^[a]	LOQ	c _V	calibration
				range
	[mg/kg cookie]	[mg/kg cookie]	[%]	[mg/L]
Fructosyllysine	2.1	5.2	0.4–20.1	0.08–38.9
Glucosyllysine ^[b]	1.3	3.7	2.0-20.4	0.02–10.4
CML	0.6	1.1	0.5–19.7	0.005-1.4
CEL	0.3	0.7	0.8–19.9	0.005-2.7
MG-H1	1.0	2.6	1.2–20.3	0.02-8.2

[a] Limits of detection (LOD) and limits of quantitation (LOQ) are given in mg/kg cookie for protein-bound MRPs based on the average protein content of all analyzed cookies. Coefficients of variation (c_V) were determined by triplicate analysis of different cookies.

Table 5. Concentrations of protein-bound MRPs in cookie dough, baked cookies and

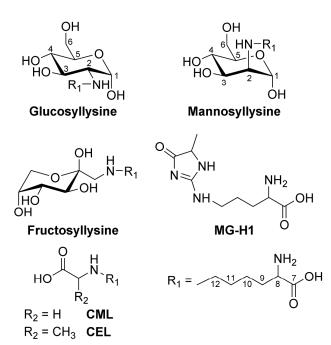
purchased baked products in mg/kg cookie

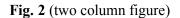
	sweetener	Fructosyllysine	LysGlc/Man	CML	CEL	MG-H1
dough		tr	nd	nd-tr	nd	nd-tr
baked	sucrose ^[a]	tr	tr	1.71 ±	nd	tr
cookies				0.34		
	glucose	459 ± 37	11.28 ± 0.61	7.9 ± 1.5	1.541 ±	8.8 ± 1.7
					0.013	
	fructose	57.7 ± 1.1	118 ± 24	6.22 ±	4.76 ±	137.5 ±
				0.71	0.87	7.3
	honey	312 ± 63	40.4 ± 1.3	6.940 ±	2.97 ±	27.7 ±
				0.057	0.15	2.5
	banana	63.65 ± 0.29	15.7 ± 1.7	7.67 ±	2.45 ±	14.7 ±
				0.18	0.12	3.0
purchased	honey (7) ^[c]	108–469	19–208	13–76	2.5-32	10–93
bakery	banana (1)	61.3 ± 4.4	24.9 ± 1.5	10.5 ±	23.5 ±	60.7 ±
products ^[b]				1.5	4.2	3.4
	invert sugar	116–717	82–287	32–42	12–53	45–218
	syrup (4) ^[d]					

[a] data are mean \pm SD ($n \ge 2$) for one kind of product, data are given as ranges for groups of different bakery products, the number of different bakery products is stated in brackets. [c] 4 products also contained glucose syrup [d] 2 products also contained small amounts of honey; nd, not detected (below LOD); tr, traces (between LOD and LOQ)

Figures

Fig. 1 (one-column figure)





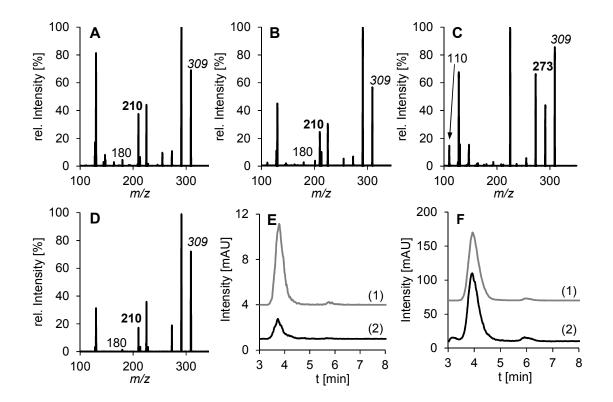
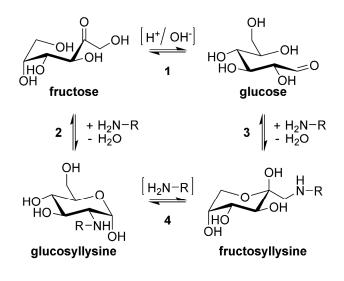


Fig 3. (one column figure)



TOC graphic

