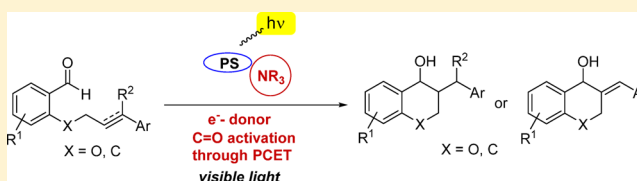


Photoredox-Catalyzed Ketyl–Olefin Coupling for the Synthesis of Substituted Chromanols

Eleonora Fava,^{†,‡} Masaki Nakajima,[†] Anh L. P. Nguyen,[†] and Magnus Rueping^{*,†,‡}[†]RWTH Aachen University, Institute of Organic Chemistry, Landoltweg 1, 52074, Aachen, Germany[‡]King Abdullah University of Science and Technology (KAUST), KAUST Catalysis Center (KCC), Thuwal, 23955-6900, Saudi Arabia

S Supporting Information

ABSTRACT: A visible light photoredox-catalyzed aldehyde olefin cyclization is reported. The method represents a formal hydroacylation of alkenes and alkynes and provides chromanol derivatives in good yields. The protocol takes advantage of the double role played by trialkylamines (NR₃) which act as (i) electron donors for reducing the catalyst and (ii) proton donors to activate the substrate via a proton-coupled electron transfer.



Carbon–carbon and carbon–heteroatom bond-forming reactions represent powerful transformations useful for the assembly of simple as well as sophisticated molecular architectures. Among the variety of established methods reported so far, the generation of ketyl radical species by the reductive umpolung reaction of carbonyl derivatives and their application in carbon–carbon bond-forming reactions represents an appealing route for the preparation of complex molecules. The ketyl–olefin coupling, as first described by Molander and Kenny,¹ was promoted by samarium diiodide as a stoichiometric single-electron reducing agent and enhanced by its large reduction potential (up to -2.05 V in the presence of HMPA).² Alternatively, organotin compounds as reducing agents have been used in either stoichiometric^{3,4} or catalytic amounts,⁵ in this case requiring an additional metal hydride. Other relevant procedures involve the use of zinc/trimethylchlorosilane as reagent,⁶ or titanocene complexes,⁷ well-known as pinacol-coupling promoters. Among the ketyl–olefin couplings reported, the major protocols imply the use of stoichiometric amounts of (transition) metals.⁸

Recently, the possibility of generating ketyl radicals by visible light activation^{9–13} was shown to be a feasible alternative to the common methods, avoiding the use of toxic and/or environmentally unfriendly reducing reagents and yet enabling access to a wide range of complex alcohols. However, the reduction of aldehydes ($E_{1/2}^{\text{red}} = -2.17$ V vs Ag/AgNO₃ for benzaldehydes),¹⁴ possessing an exceptionally negative redox potential, represents an obstacle for the most commonly employed visible light photocatalysts. In this context, Pandey reported the visible-light-promoted one-electron reductive activation of aldehydes and ketones and their intramolecular addition to activated olefins (five examples) with a system consisting of DCA (9,10-dicyanoanthracene) and PPh₃ or DCA, DMN (1,5-dimethoxynaphthalene) and ascorbic acid.^{11a}

Knowles and co-workers reported a catalytic protocol in which ketones are successfully reduced to ketyl radicals by the concerted transfer of both a proton and an electron.^{11b} This phenomenon is named proton-coupled electron transfer (PCET).¹⁵

In an attempt to demonstrate the active role of proton donors in the reduction step, Knowles and co-workers also reported an asymmetric aza-pinacol cyclization of ketones and hydrazones where the binding effect of the chiral phosphoric acids led to the formation of enantioenriched vicinal amino alcohols.^{11c}

Recently, our group described the photoredox-catalyzed pinacol coupling of aldehydes, ketones, and aldimines through hydrogen-bonding activation.^{12a} With this work, we disclose the possibility of generating a long-lived ketyl radical using tertiary amines as the electron and proton source, replacing the combination of Hantzsch ester and diphenyl phosphate (DPP).

Based on our previous work,^{12a} we envisioned the possibility of performing the ketyl radical addition on unsaturated bonds triggered by visible light, which would initiate the single-electron reduction from a photoexcited polypyridyl iridium complex to a carbonyl group (aldehyde or ketone). We postulated that the so-generated radical intermediate would be able to undergo subsequent intramolecular addition to a pendant alkene or alkyne moiety (Scheme 1).

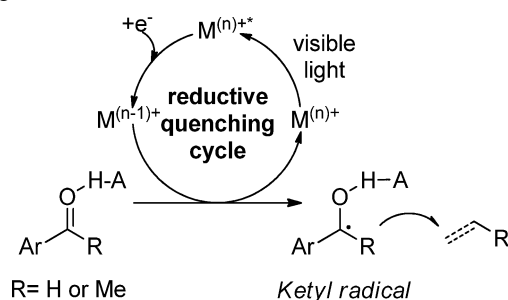
Herein, we report the realization of this activation mode in the context of a catalytic intramolecular reductive coupling of aldehydes with alkenes and alkynes. This transformation allows the synthesis of substituted 3-benzylchroman-4-ols (**5**), containing a chroman unit, that exhibit anti picornavirus

Special Issue: Photocatalysis

Received: April 30, 2016



Scheme 1. Photoredox-Catalyzed Reduction of Aldehydes through PCET



activity, commonly responsible for the upper respiratory tract infections in humans.¹⁶

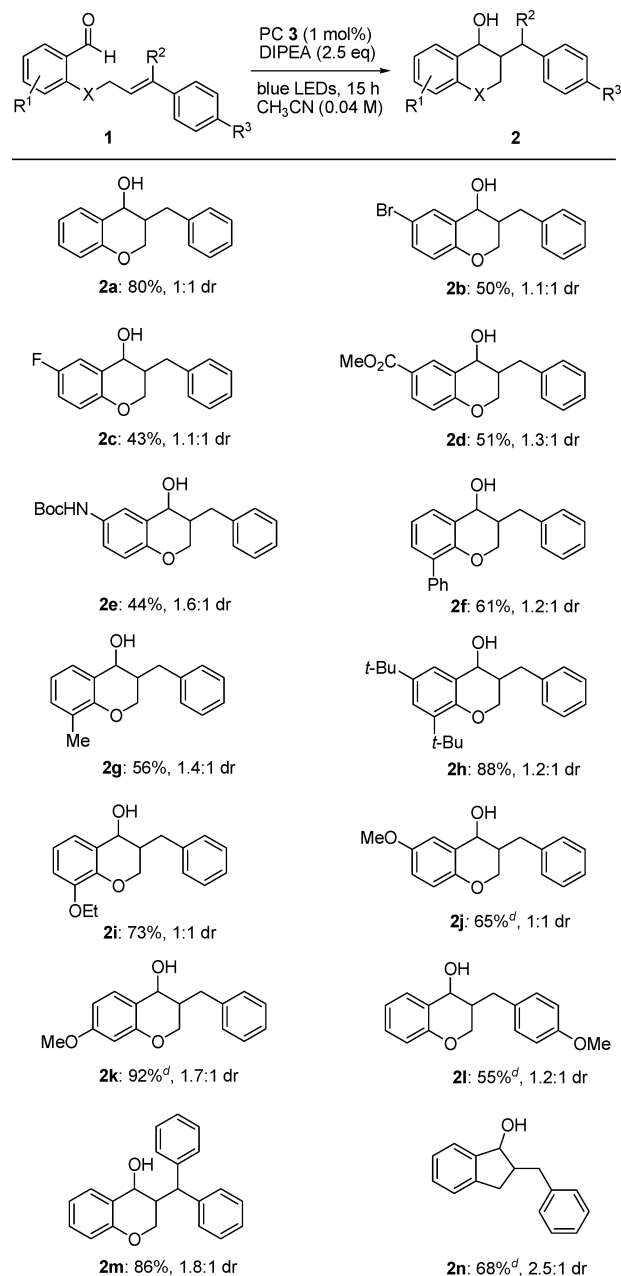
Remarkably, we observed that tertiary amines not only provided the desired products in higher yields but also proved to be crucial for the reaction to proceed.

We began our studies by evaluating conditions for the cyclization of 2-(cinnamyloxy)benzaldehyde (**1a**) to chromanol **2a** in the presence of Ir(ppy)₂(dtb-bpy)PF₆ (PC **3**, 1 mol %), trialkylamines **4a–c** as reductive quencher (RQ), and irradiation with a blue LED light source (11W, $\lambda_{\text{max}} = 450$ nm). As shown in Table S1 (Supporting Information), the reaction proceeded with good yield in the presence of 2.5 equiv of tributylamine (**4b**, entry 5). Solvent evaluation confirmed acetonitrile to be the best solvent, whereas aprotic (entries 1 and 2) and protic (entry 3) polar solvents were shown to be unsuccessful as reaction media. Further screening of electron donors in acetonitrile confirmed Hünig base (**4c**) to be the best among the trialkylamines tested, providing the desired product in 80% yield (entry 6). As previously mentioned, the use of Hantzsch esters **5a** and **5b** alone or in combination with 10 mol % of DPP did not lead to the expected product (entries 7–10). A series of control experiments demonstrated that the reaction did not take place in the absence of light, catalyst, or **3**. With the optimized conditions in hand (acetonitrile, rt, 1 mol % PC **3**), we began to evaluate a variety of substrates (Table 1).

As shown in Table 1, the reaction tolerates a wide variety of substituents, confirming the ability of the ketyl radical to couple to the olefin in the presence of both electron-withdrawing and -donating groups on the aromatic aldehyde moiety.

Following the development of a successful formal hydroacylation we investigated the reaction in which the olefin was replaced by an alkyne group (Table 2). Also in this case, the reaction gave the expected products (**2o** and **2p**) in good yields.

Regarding the reaction mechanism, we propose the following: Irradiation with visible light results in the formation of the reductive species Ir²⁺. Hünig base (*i*-Pr₂NEt) acts as a sacrificial electron donor; therefore, the single-electron oxidation weakens the adjacent C–H bond¹⁷ and the 1,2-H-shift becomes energetically favored. The following deprotonation through the tertiary amine provides the ammonium derivative, which is now able to engage in a proton-coupled electron transfer with the substrate, lowering its potential for Ir²⁺ reduction and generation of a neutral ketyl intermediate. Alternatively, the ammonium radical could also activate the carbonyl functionality. The ketyl radical is able to add to the unsaturated bond, forming the chroman ring and a benzyl radical which upon hydrogen atom transfer leads to the final product (Scheme 2).

Table 1. Scope of the Reaction^{a–c}

^aAll reactions of **1** (0.12 mmol) with DIPEA (0.3 mmol) were carried out in the presence of PC **3** (1 mol %) in acetonitrile (3.0 mL) under irradiation with blue LEDs (11 W, 450 nm) for 15 h at 25 °C.

^bIsolated yields are reported. ^cThe dr is given for *anti/syn* ratio.

^dReaction performed using 2.5 mol % of photocatalyst.

CONCLUSIONS

In conclusion, we report a ketyl–olefin coupling for the preparation of substituted 3-benzylchroman-4-ols, promoted by visible light photoredox catalysis. Importantly, we uncovered trialkylamines as a cheap and readily available alternative to the previously reported electron–proton donor system consisting of Hantzsch ester/Brønsted acid. By employing the trialkylamine/photocatalyst system, we have been able to develop an efficient intramolecular ketyl–olefin and ketyl–alkyne coupling employing readily prepared substrates. The formal hydroacylation protocol provides the chromanol derivatives in good yield under mild reaction conditions and with low catalyst

Table 2. Scope of the Intramolecular Ketyl–Alkyne Cyclization^a

entry	product	yield ^b (%)	E/Z
1	Ar = <i>m</i> -MePh (2a)	57	1:1
2	Ar = <i>p</i> -OMePh (2p)	59	1:1

^aAll reactions of **1** (0.12 mmol) with DIPEA (0.3 mmol) were carried out in the presence of PC **3** (2.5 mol %) in acetonitrile (3.0 mL) under irradiation with blue LEDs (11 W, 450 nm) for 18 h at 25 °C.

^bIsolated yields are reported.

loadings. Efforts are currently underway to expand this concept to further transformations.

EXPERIMENTAL SECTION

General Information. All reactions were performed with oven-dried glassware and under an inert atmosphere (argon) unless otherwise stated. Acetonitrile was distilled from calcium hydride and stored over 4 Å molecular sieves under nitrogen/argon atmosphere. Tetrahydrofuran was distilled from Solvona/benzophenone. Pentane was distilled in a standard distillation apparatus. Other solvents were used as purchased unless otherwise stated. Commercial reagents were used as purchased without further purification. Organic solutions were concentrated under reduced pressure on a rotary evaporator. Chromatographic purification of products was carried out using silica gel (230–400 mesh). Thin-layer chromatography was carried out using aluminum plates coated with silica (230–400 mesh) which were visualized under UV light (254 nm) or by staining with aqueous potassium permanganate solutions or vanillin alcoholic solution. ¹H NMR spectra were recorded in deuterated solvents on spectrometers at 300, 400, or 600 MHz with residual protic solvent as the internal standard. ¹³C NMR spectra were recorded in deuterated solvents on spectrometers at 75, 101, or 151 MHz with the central peak of the deuterated solvent as the internal standard. Chemical shifts (δ) are given in parts per million (ppm), and coupling constants (*J*) are given in hertz (Hz) rounded to the nearest 0.1 Hz. The ¹H NMR spectra are reported as δ downfield from tetramethylsilane (multiplicity, coupling constant *J*, number of protons). The ¹³C NMR spectra are reported as

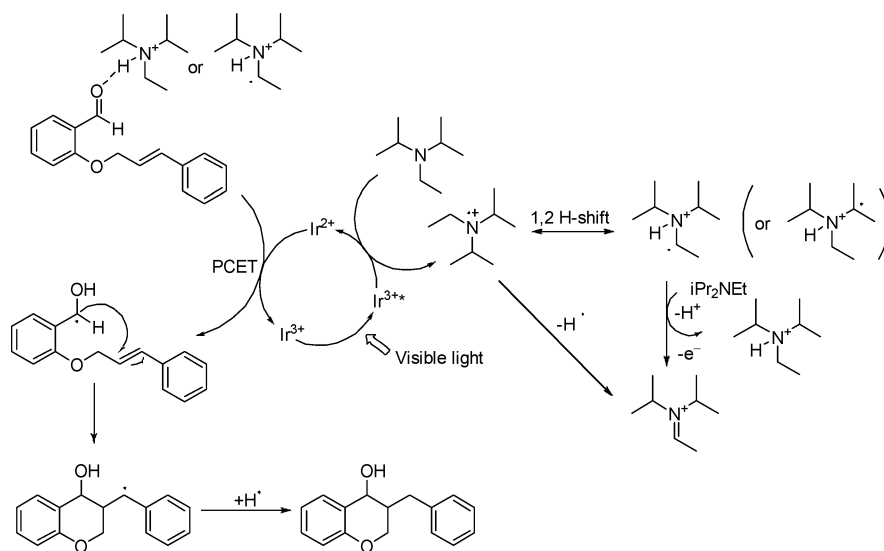
δ. For the F-containing compound (**2c**), a simultaneously proton and fluorine decoupled ¹³C NMR spectrum was recorded. Assignments are aided by the use of DEPT-135, COSY and HMQC spectra where necessary. IR spectra were recorded on a FTIR spectrometer. Low resolution mass spectra (EI/CI) were recorded on a mass spectrometer with (EI) or (CI) detectors. High-resolution mass spectrometry (HRMS) was performed by EI using a double-focusing mass spectrometer and by ESI using a hybrid ion-trap mass spectrometer. Melting points were recorded at ambient pressure and are uncorrected.

General Procedure for the Synthesis of Chromanols. A Schlenk tube was charged with aldehyde **1** (1 equiv), catalyst PC **3** (1 or 2.5 mol %), and degassed DIPEA (2.5 equiv). It was capped, evacuated, and backfilled with argon. Subsequently, degassed acetonitrile (3 mL) was added via syringe. The vial was placed in a 100 mL beaker containing a blue LED strip glued on the inner wall, and the reaction mixture was stirred for 15 h. After this time, the solvent was evaporated, and the reaction mixture was purified on silica gel (gradient pentane/EtOAc 95:5).

3-Benzylchroman-4-ol (2a**).**¹⁸ The title compound was synthesized according to the general procedure employing **1a** (0.12 mmol, 28.6 mg), 1 mol % (1.1 mg) of PC **3**, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 80% (23 mg); *anti/syn* 1:1; ¹H NMR (600 MHz, CDCl₃) δ 7.34–7.18 (m, 14H), 6.96 (td, *J* = 7.5, 0.9 Hz, 1H), 6.90 (dd, *J* = 10.6, 4.2 Hz, 2H), 6.85 (d, *J* = 8.5 Hz, 1H), 4.51 (dd, *J* = 9.4, 3.1 Hz, 2H), 4.23 (dd, *J* = 11.1, 2.6 Hz, 1H), 4.14–4.05 (m, 2H), 3.98 (dd, *J* = 11.1, 4.2 Hz, 1H), 2.89 (dd, *J* = 13.8, 8.4 Hz, 1H), 2.75–2.64 (m, 2H), 2.54 (dd, *J* = 13.8, 9.3 Hz, 1H), 2.40–2.29 (m, 1H), 2.28–2.17 (m, 1H), 1.92 (bs, 1H), 1.74 (bs, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 154.3, 154.2, 139.2, 139.1, 130.2, 130.1, 129.9, 129.8, 129.1 (2C), 128.6, 128.5, 126.3 (2C), 124.1, 123.2, 120.9, 120.5, 116.9 (2C), 67.6, 64.9 (2C), 64.6, 41.5, 40.0, 34.6, 32.8; FT-IR ν_{max} (ATR) cm⁻¹ 3372, 2922, 1582, 1485, 1450, 1302, 1264, 1222, 1119, 1074, 1039, 909, 748, 698; *m/z* (EI) 240 ([M]⁺, 100), 91 ([PhCH₂]⁺, 28); HRMS (ESI) calcd for [C₁₆H₁₆O₂ + Na]⁺ 263.10425, found 263.10419.

3-Benzyl-6-bromochroman-4-ol (2b**).** The title compound was synthesized according to the general procedure employing **1b** (0.12 mmol, 38.1 mg), 2.5 mol % (2.74 mg) of PC **3**, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 50% (19.1 mg); *anti/syn* 1.1:1; ¹H NMR (400 MHz, CD₃OD) δ 7.45 (d, *J* = 2.4 Hz, 1H, *syn*), 7.34 (d, *J* = 3.1 Hz, 1.1 × 1H, *anti*), 7.31–7.13 (m, 12.6H), 6.73–6.65 (m, 2.1H), 4.47 (d, *J* = 3.4 Hz, 1H, *syn*), 4.36 (d, *J* = 5.0 Hz, 1.1 × 1H, *anti*), 4.13 (dd, *J* = 11.2, 2.8 Hz, 1.1 × 1H, *anti*), 4.03–3.94 (m, 2H, *syn*), 3.89 (dd, *J* =

Scheme 2. Plausible Reaction Mechanism



11.1, 5.5 Hz, 1.1 × 1H, *anti*), 2.87 (dd, *J* = 13.6, 7.1 Hz, 1H, *syn*), 2.76 (dd, *J* = 13.8, 6.1 Hz, 1H, *syn*), 2.55 (dd, *J* = 13.6, 8.3 Hz, 1.1 × 1H, *anti*), 2.45 (dd, *J* = 13.7, 9.3 Hz, 1.1 × 1H, *anti*), 2.25–2.17 (m, 1H, *syn*), 2.16–2.08 (m, 1.1 × 1H, *anti*); ¹³C NMR (101 MHz, CD₃OD) δ 153.5 (2C), 139.4, 139.2, 132.4 (2C), 131.60 (2C), 128.8, 128.7, 128.1 (2C), 127.0, 126.1, 126.0, 125.8, 118.1 (2C), 111.9, 111.5, 66.4, 65.0, 64.8, 64.2, 41.5, 40.2, 34.2, 32.1; FT-IR ν_{\max} (ATR) cm⁻¹ 2926, 1479, 1407, 1260, 1224, 1191, 1085, 1055, 1021, 819, 748, 699; *m/z* (EI) 319 ([M]⁺, 9), 91 ([PhCH₂]⁺, 100); HRMS (EI) calcd for [C₁₆H₁₅O₂⁷⁹Br]⁺ 318.02499, found 318.02492.

3-Benzyl-6-fluorochroman-4-ol (2c). The title compound was synthesized according to the general procedure employing **1c** (0.12 mmol, 30.8 mg), 1 mol % (1.1 mg) of PC 3 and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 43% (13.2 mg); mp = 102–104 °C; *anti/syn* 1.1:1; ¹H NMR (600 MHz, CD₃OD) δ 7.31–7.23 (m, 6.5 H), 7.21–7.15 (m, 4.3 H), 7.06 (dd, *J* = 9.0, 3.1 Hz, 1.1 × 1H, *anti*), 6.96 (dd, *J* = 8.9, 3.1 Hz, 1H, *syn*), 6.89 (dtd, *J* = 11.6, 8.5, 3.1 Hz, 2H), 6.78–6.71 (m, 2H), 4.48 (d, *J* = 3.5 Hz, 1H, *syn*), 4.37 (d, *J* = 5.1 Hz, 1.1 × 1H, *anti*), 4.12 (dd, *J* = 11.1, 2.8 Hz, 1.1 × 1H, *anti*), 4.01–3.94 (m, 2H, *syn*), 3.87 (dd, *J* = 11.1, 5.6 Hz, 1.1 × 1H, *anti*), 2.88 (dd, *J* = 13.6, 7.2 Hz, 1H, *syn*), 2.79 (dd, *J* = 13.8, 6.0 Hz, 1.1 × 1H, *anti*), 2.56 (dd, *J* = 13.6, 8.3 Hz, 1H, *syn*), 2.46 (dd, *J* = 13.8, 9.4 Hz, 1.1 × 1H, *anti*), 2.24–2.17 (m, 1H, *syn*), 2.16–2.11 (m, 1.1 × 1H, *anti*); ¹⁹F NMR (564 MHz, CD₃OD) δ -125.8, -126.4; ¹³C NMR (101 MHz, CD₃OD) δ 156.9, 156.6, 150.4, 150.3, 139.5, 139.2, 128.8, 128.7, 128.1, 128.0, 125.9 (2C), 125.8, 125.0, 117.2, 117.1, 115.5, 115.4, 115.3 (2C), 66.8, 65.0, 64.7, 64.4, 41.6, 40.3, 34.2, 32.1; FT-IR ν_{\max} (ATR) cm⁻¹ 3370, 2926, 2493, 1488, 1436, 1251, 1201, 1027, 736, 698; *m/z* (EI) 258 ([M]⁺, 48), 91 ([PhCH₂]⁺, 100); HRMS (EI) calcd for [C₁₆H₁₅O₂F]⁺ 258.10506, found 258.10537.

Methyl 3-Benzyl-4-hydroxychroman-6-carboxylate (2d). The title compound was synthesized according to the general procedure employing **1d** (0.12 mmol, 35.6 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 51% (18.2 mg); *anti/syn* 1.3:1; ¹H NMR (400 MHz, CD₃OD) δ 8.05 (d, *J* = 2.1 Hz, 1.3 × 1H, *anti*), 7.92 (d, *J* = 2.2 Hz, 1H, *syn*), 7.80 (ddd, *J* = 14.4, 8.6, 2.2 Hz, 2.3H), 7.32–7.11 (m, 11.5H), 6.81 (dd, *J* = 14.4, 8.6 Hz, 2.3H), 4.52 (d, *J* = 3.2 Hz, 1H, *syn*), 4.42 (d, *J* = 4.9 Hz, 1.3 × 1H, *anti*), 4.21 (dd, *J* = 11.2, 2.9 Hz, 1.3 × 1H, *anti*), 4.11–3.99 (m, 2H, *syn*), 3.96 (dd, *J* = 11.1, 5.3 Hz, 1.3 × 1H, *anti*), 3.84 (s, 1.3 × 3H, *anti*), 3.81 (s, 3H, *syn*), 2.88 (dd, *J* = 13.6, 7.1 Hz, 1H, *syn*), 2.75 (dd, *J* = 13.7, 6.1 Hz, 1.3 × 1H, *anti*), 2.56 (dd, *J* = 13.6, 8.3 Hz, 1H, *syn*), 2.43 (dd, *J* = 13.7, 9.3 Hz, 1.3 × 1H, *anti*), 2.26–2.13 (m, 2.3H); ¹³C NMR (101 MHz, CD₃OD) δ 167.0 (2C), 158.6 (2C), 139.4, 139.1, 132.3, 132.2, 130.4, 130.3, 128.8, 128.7, 128.1 (2C), 126.0, 125.9, 124.8, 123.9, 122.1, 121.7, 116.3 (2C), 66.4, 65.4, 65.2, 64.2, 51.0, 50.9, 41.4, 40.2, 34.2, 32.2; FT-IR ν_{\max} (ATR) cm⁻¹ 2949, 1711, 1611, 1496, 1436, 1253, 1192, 1125, 1012, 767, 700; *m/z* (EI) 298 ([M]⁺, 12), 91 ([PhCH₂]⁺, 40); HRMS (EI) calcd for [C₁₈H₁₈O₄]⁺ 298.11996, found 298.12009.

tert-Butyl (3-Benzyl-4-hydroxychroman-6-yl)carbamate (2e). The title compound was synthesized according to the general procedure employing **1e** (0.12 mmol, 42.4 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 44% (18.7 mg); *anti/syn* 1.6:1; ¹H NMR (600 MHz, CDCl₃) δ 7.49–7.15 (m, 15.6H), 7.08 (dd, *J* = 8.8, 2.6 Hz, 1.6 × 1H, *anti*), 6.99 (dd, *J* = 8.8, 2.6 Hz, 1H, *syn*), 6.80 (d, *J* = 8.8 Hz, 1.6 × 1H, *anti*), 6.76 (d, *J* = 8.8 Hz, 1H, *syn*), 6.45–6.24 (m, 2.6H), 4.52–4.41 (m, 2.6H), 4.19 (dd, *J* = 11.0, 1.9 Hz, 1.6 × 1H, *anti*), 4.12–4.01 (m, 2H, *syn*), 3.92 (dd, *J* = 11.0, 3.9 Hz, 1.6 × 1H, *anti*), 2.86 (dd, *J* = 13.5, 8.6 Hz, 1H, *syn*), 2.73–2.61 (m, 2.6H), 2.51 (dd, *J* = 13.7, 9.5 Hz, 1.6 × 1H, *anti*), 2.31–2.26 (m, 1H, *syn*), 2.21–2.17 (m, 1.6 × 1H, *anti*), 1.91–1.62 (overlapping of two bs, 2.6H), 1.51 (s, 1.6 × 9H, *anti*), 1.48 (s, 9H, *syn*); ¹³C NMR (151 MHz, CDCl₃) δ 153.2, 153.1, 150.3, 150.2, 139.2, 139.1, 131.4, 131.3, 131.0 (2C), 129.1 (2C), 128.5 (2C), 126.3 (2C), 124.4, 123.5, 121.2, 120.7, 117.2, 117.1, 80.4 (2C), 67.6, 65.1, 64.8, 64.6, 41.3, 40.0, 34.6, 32.7, 28.4, 28.3; FT-IR ν_{\max} (ATR) cm⁻¹: 3350, 2981, 2932, 1696, 1499, 1454, 1367, 1309, 1241, 1218, 1163, 1023, 822, 733, 699; *m/z* (EI)

298 ([M – (CH₃)₃C]⁺, 100), 91 ([PhCH₂]⁺, 67); HRMS (EI) calcd for [C₁₆H₁₆O₂N₁]⁺ ([M – Boc]⁺) 254.11756, found 254.11706.

3-Benzyl-8-phenylchroman-4-ol (2f). The title compound was synthesized according to the general procedure employing **1f** (0.12 mmol, 37.7 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 61% (23 mg); *anti/syn* 1.2:1; ¹H NMR (600 MHz, CD₃OD) δ 7.48–7.44 (m, 1.2 × 1H, *anti*), 7.44–7.40 (m, 1H, *syn*), 7.37–7.05 (m, 24.2H), 7.00–6.94 (m, 1.2 × 1H, *anti*), 6.92–6.86 (m, 1H, *syn*), 4.49 (d, *J* = 2.7 Hz, 1H, *syn*), 4.44 (d, *J* = 4.1 Hz, 1.2 × 1H, *anti*), 4.15 (dd, *J* = 11.0, 2.6 Hz, 1.2 × 1H, *anti*), 4.00 (t, *J* = 10.9 Hz, 1H, *syn*), 3.92 (dd, *J* = 10.6, 3.5 Hz, 1H, *syn*), 3.86 (dd, *J* = 11.0, 4.5 Hz, 1.2 × 1H, *anti*), 2.85 (dd, *J* = 13.5, 7.5 Hz, 1H, *syn*), 2.70 (dd, *J* = 13.7, 6.4 Hz, 1.2 × 1H, *anti*), 2.55 (dt, *J* = 18.0, 9.0 Hz, 1H, *syn*), 2.51–2.44 (m, 1.2 × 1H, *anti*), 2.25–2.18 (m, 1H, *syn*), 2.17–2.12 (m, 1.2 × 1H, *anti*); ¹³C NMR (151 MHz, CD₃OD) δ 151.1 (2C), 139.6, 139.4, 138.5, 138.4, 130.2, 130.1, 129.8 (2C), 129.7 (2C), 129.2 (2C), 128.8, 128.7, 128.1 (2C), 127.5 (2C), 126.4 (2C), 125.9, 125.8, 125.1, 123.9, 120.2, 119.9, 67.1, 64.8, 64.6, 64.5, 41.5, 40.3, 34.3, 32.5; FT-IR ν_{\max} (ATR) cm⁻¹ 1594, 1496, 1469, 1430, 1219, 1017, 754, 697; *m/z* (EI) 316 ([M]⁺, 63%), 91 ([PhCH₂]⁺, 27%); HRMS (ESI) calcd for [C₂₂H₂₀O₂ + Na]⁺ 339.13555, found 339.13525.

3-Benzyl-8-methylchroman-4-ol (2g). The title compound was synthesized according to the general procedure employing **1g** (0.12 mmol, 30.3 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 56% (17.1 mg); *anti/syn* 1.4:1; ¹H NMR (600 MHz, CD₃OD) δ 7.30–7.21 (m, 7.5H), 7.20–7.12 (m, 5.5H), 7.05–7.02 (m, 2.4H), 6.99 (d, *J* = 7.3 Hz, 1.4 × 1H, *anti*), 6.80 (t, *J* = 7.5 Hz, 1H, *syn*), 6.73 (t, *J* = 7.5 Hz, 1.4 × 1H, *anti*), 4.45 (d, *J* = 3.2 Hz, 1.4 × 1H, *anti*), 4.38 (d, *J* = 4.2 Hz, 1H, *syn*), 4.19 (dd, *J* = 11.0, 2.6 Hz, 1H, *syn*), 4.06–4.02 (m, 1.4 × 2H, *anti*), 3.94 (dd, *J* = 11.0, 4.6 Hz, 1H, *syn*), 2.87 (dd, *J* = 13.7, 7.4 Hz, 1.4 × 1H, *anti*), 2.68 (dd, *J* = 13.7, 6.5 Hz, 1H, *syn*), 2.58 (dd, *J* = 13.6, 8.1 Hz, 1.4 × 1H, *anti*), 2.45 (dd, *J* = 13.7, 9.1 Hz, 1H, *syn*), 2.16 (s, 3H, *syn*), 2.12 (s, 1.4 × 3H, *anti*) OH signals not observed; ¹³C NMR (151 MHz, CD₃OD) δ 152.3, 152.2, 139.7, 139.5, 129.9, 129.8, 128.8, 128.7, 128.1 (2C), 127.9, 127.8, 125.8 (2C), 125.2 (2C), 123.9, 122.7, 119.6, 119.3, 66.9, 64.7, 64.6, 64.4, 41.8, 40.6, 34.3, 32.6, 14.8 (2C); FT-IR ν_{\max} (ATR) cm⁻¹ 3352, 2919, 1598, 1472, 1258, 1210, 1092, 1023, 747, 698; *m/z* (EI) 254 ([M]⁺, 59), 253 ([M – 1], 100), 91 ([PhCH₂]⁺, 39); HRMS (EI) calcd for [C₁₇H₁₈O₂]⁺ 254.13013, found 254.13046.

3-Benzyl-6,8-di-tert-butylchroman-4-ol (2h). The title compound was synthesized according to the general procedure employing **1h** (0.12 mmol, 42.1 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 88% (37.5 mg); *anti/syn* 1.2:1; ¹H NMR (400 MHz, CD₃OD) δ 7.33–7.07 (m, 15.4H), 4.48 (d, *J* = 3.5 Hz, 1H, *syn*), 4.39 (d, *J* = 3.3 Hz, 1.2 × 1H, *anti*), 4.13 (dd, *J* = 10.9, 2.5 Hz, 1.2 × 1H, *anti*), 4.03–3.94 (m, 2H, *syn*), 3.90 (ddd, *J* = 10.9, 4.0, 1.1 Hz, 1.2 × 1H, *anti*), 2.88 (dd, *J* = 13.5, 7.2 Hz, 1H, *syn*), 2.67 (dd, *J* = 13.6, 6.0 Hz, 1.2 × 1H, *anti*), 2.57 (dd, *J* = 13.5, 8.3 Hz, 1H, *syn*), 2.42 (dd, *J* = 13.6, 9.8 Hz, 1.2 × 1H, *anti*), 2.23–2.15 (m, 1H, *syn*), 2.14–2.08 (m, 1.2 × 1H, *anti*), 1.37 (s, 1.2 × 9H, *anti*), 1.33 (s, 9H, *syn*), 1.29 (s, 1.2 × 9H, *anti*), 1.25 (s, 9H, *syn*); ¹³C NMR (101 MHz, CD₃OD) δ 150.6 (2C), 141.8, 141.5, 139.8, 139.6, 136.3, 136.1, 128.8 (2C), 128.0 (2C), 125.8, 125.7, 125.2, 124.7, 124.0, 123.1, 123.0, 122.4, 67.7, 65.5, 63.9, 63.0, 41.7, 40.6, 34.5, 34.5, 34.3, 33.7, 33.7, 32.5, 30.7 (2C), 28.9 (2C); FT-IR ν_{\max} (ATR) cm⁻¹ 2956, 2870, 1479, 1450, 1361, 1229, 1171, 1130, 1028, 882, 749, 701, 532; *m/z* (EI) 352 ([M]⁺, 65), 91 ([PhCH₂]⁺, 15); HRMS (EI) calcd for [C₂₄H₃₂O₂]⁺ 352.23968, found 352.23972.

3-Benzyl-8-ethoxychroman-4-ol (2i). The title compound was synthesized according to the general procedure employing **1i** (0.12 mmol, 33.8 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 73% (25 mg); *anti/syn* 1:1; ¹H NMR (600 MHz, CDCl₃) δ 7.36–7.14 (m, 10H), 6.95–6.77 (m, 6H), 4.48–4.47 (m, 2H), 4.29 (dd, *J* = 11.1, 2.7 Hz, 1H), 4.22 (ddd, *J* = 10.7, 3.7, 1.2 Hz, 1H), 4.15–4.04 (m, 6H), 2.88 (dd, *J* = 13.7, 8.4 Hz, 1H), 2.70 (dd, *J* = 14.0, 7.0

Hz, 1H), 2.68 (dd, $J = 13.7, 7.2$ Hz, 1H), 2.59 (dd, $J = 13.9, 8.5$ Hz, 1H), 2.35–2.29 (m, 1H), 2.22 (dq, $J = 12.9, 4.3, 2.8$ Hz, 1H), 1.97 (bs, 2H), 1.47–1.45 (m, 6H); ^{13}C NMR (151 MHz, CDCl_3) δ 147.6, 147.5, 144.1, 144.0, 139.2, 139.1, 129.1 (2C), 128.5 (2C), 126.3 (2C), 125.0, 124.0, 121.8, 121.6, 120.4, 120.1, 112.6 (2C), 67.4, 65.4, 65.2, 64.7, 64.4, 64.3, 41.3, 39.8, 34.6, 32.9, 14.8 (2C); FT-IR ν_{max} (ATR) cm^{-1} 3506, 2921, 1585, 1478, 1391, 1315, 1255, 1214, 1080, 1030, 964, 900, 737; m/z (EI) 284 ($[\text{M}]^+$, 23), 91 ($[\text{PhCH}_2]^+$, 100); HRMS (EI) calcd for $[\text{C}_{18}\text{H}_{20}\text{O}_3]^+$ 284.14070, found 284.14077.

3-Benzyl-6-methoxychroman-4-ol (2j). The title compound was synthesized according to the general procedure employing **1j** (0.12 mmol, 32.2 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 65% (21 mg); mp = 112–113 °C; *anti/syn* 1:1; ^1H NMR (400 MHz, CD_3OD) δ 7.30–7.13 (m, 10H), 6.88 (d, $J = 3.0$ Hz, 1H), 6.80–6.63 (m, 5H), 4.46 (d, $J = 3.5$ Hz, 1H), 4.36 (d, $J = 4.6$ Hz, 1H), 4.09 (dd, $J = 11.0, 2.7$ Hz, 1H), 3.99–3.89 (m, 2H), 3.84 (ddd, $J = 11.0, 5.1, 0.7$ Hz, 1H), 3.73 (s, 3H), 3.70 (s, 3H), 2.87 (dd, $J = 13.6, 7.2$ Hz, 1H), 2.74 (dd, $J = 13.7, 6.2$ Hz, 1H), 2.56 (dd, $J = 13.6, 8.3$ Hz, 1H), 2.47 (dd, $J = 13.7, 9.3$ Hz, 1H), 2.23–2.15 (m, 1H), 2.14–2.07 (m, 1H); ^{13}C NMR (101 MHz, CD_3OD) δ 153.7, 153.4, 148.2, 148.1, 139.7, 139.5, 128.8, 128.7, 128.0 (2C), 125.8, 125.7, 124.9, 123.9, 116.7 (2C), 115.5 (2C), 113.9, 113.8, 67.1, 64.8, 64.5 (2C), 54.7 (2C), 42.0, 40.7, 34.3, 32.3; FT-IR ν_{max} (ATR) cm^{-1} 2923, 2471, 1490, 1430, 1265, 1205, 1161, 1037, 807, 703; m/z (EI) 270 ($[\text{M}]^+$, 71), 91 ($[\text{PhCH}_2]^+$, 33); HRMS (EI) calcd for $[\text{C}_{17}\text{H}_{18}\text{O}_3]^+$ 270.12505, found 270.12551.

3-Benzyl-7-methoxychroman-4-ol (2k).¹⁹ The title compound was synthesized according to the general procedure employing **1k** (0.12 mmol, 32.2 mg), 2.5 mol % (2.74 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 92% (29.8 mg); mp = 100–103 °C; *anti/syn* 1.7:1; ^1H NMR (400 MHz, CD_3OD) δ 7.42–7.21 (m, 7.2H), 7.20–7.12 (m, 8H), 7.09 (d, $J = 8.5$ Hz, 1H), 6.51 (dd, $J = 8.5, 2.5$ Hz, 1.7 \times 1H, *anti*), 6.44 (dd, $J = 8.5, 2.5$ Hz, 1H, *syn*), 6.35 (d, $J = 2.5$ Hz, 1.7 \times 1H, *anti*), 6.30 (d, $J = 2.5$ Hz, 1H, *syn*), 4.41 (d, $J = 3.1$ Hz, 1H, *syn*), 4.34 (d, $J = 3.9$ Hz, 1.7 \times 1H, *anti*), 4.15 (dd, $J = 11.0, 2.6$ Hz, 1.7 \times 1H, *anti*), 4.11–3.93 (m, 2H, *syn*), 3.89 (ddd, $J = 11.0, 4.2, 0.9$ Hz, 1.7 \times 1H, *anti*), 3.72 (s, 3H, *anti*), 3.69 (s, 3H, *syn*), 2.85 (dd, $J = 13.6, 7.4$ Hz, 1H, *syn*), 2.66 (dd, $J = 13.7, 6.6$ Hz, 1.7 \times 1H, *anti*), 2.57 (dd, $J = 13.6, 8.0$ Hz, 1H, *syn*), 2.46 (dd, $J = 13.7, 9.1$ Hz, 1.7 \times 1H, *anti*), 2.23–2.14 (m, 1H, *syn*), 2.14–2.03 (m, 1.7 \times 1H, *anti*); ^{13}C NMR (101 MHz, CD_3OD) δ 160.7 (2C), 155.2 (2C), 139.6, 139.5, 131.1, 130.9, 128.8, 128.7, 128.0 (2C), 125.8, 125.7, 117.1, 115.6, 107.3, 106.8, 100.5 (2C), 66.3, 64.5, 64.2, 64.1, 54.3, 54.2, 42.0, 40.7, 34.2, 32.6; FT-IR ν_{max} (ATR) cm^{-1} : 3215, 2960, 1737, 1667, 1494, 1366, 1221, 1092, 1032, 734; m/z (EI) 270 ($[\text{M}]^+$, 15), 91 ($[\text{PhCH}_2]^+$, 89); HRMS (EI) calcd for $[\text{C}_{17}\text{H}_{18}\text{O}_3]^+$ 270.12505, found 270.12534.

3-(4-methoxybenzyl)chroman-4-ol (2l).²⁰ The title compound was synthesized according to the general procedure employing **1l** (0.12 mmol, 32.2 mg), 2.5 mol % (2.74 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 55% (17.7 mg); mp = 121–126 °C; *anti/syn* 1.2:1; ^1H NMR (600 MHz, CDCl_3) δ 7.32 (dd, $J = 7.6, 1.4$ Hz, 1.2 \times 1H, *anti*), 7.25–7.17 (m, 5.4H), 7.11–7.08 (m, 1.2 \times 1H, *anti*), 6.95 (td, $J = 7.5, 0.9$ Hz, 1.2 \times 1H, *anti*), 6.91–6.82 (m, 7.4H), 4.50 (t, $J = 3.5$ Hz, 1H, *syn*), 4.48 (t, $J = 4.1$ Hz, 1.2 \times 1H, *anti*), 4.21 (dd, $J = 9.2, 4.6$ Hz, 1.2 \times 1H, *anti*), 4.11–4.06 (m, 2H, *syn*), 3.97 (dd, $J = 11.0, 4.0$ Hz, 1.2 \times 1H, *anti*), 3.80 (s, 3H, *syn*), 3.79 (s, 1.2 \times 3H, *anti*), 2.82 (dd, $J = 13.8, 8.5$ Hz, 1H, *syn*), 2.68–2.58 (m, 2.2H), 2.48 (dd, $J = 14.0, 9.3$ Hz, 1.2 \times 1H, *anti*), 2.32–2.24 (m, 1H, *syn*), 2.21–2.14 (m, 1.2 \times 1H, *anti*), 1.91 (bs, 1.2 \times 1H, *anti*), 1.72 (bs, 1H, *syn*); ^{13}C NMR (151 MHz, CDCl_3) δ 158.1 (2C), 154.3, 154.2, 131.1, 131.0, 130.1 (2C), 130.0 (2C), 129.9, 129.7, 124.2, 123.2, 120.9, 120.5, 116.9 (2C), 114.0, 113.9, 67.6, 65.0, 64.9, 64.6, 55.3 (2C), 41.6, 40.1, 33.7, 31.9; FT-IR ν_{max} (ATR) cm^{-1} : 3392, 2928, 1609, 1582, 1510, 1486, 1450, 1298, 1246, 1176, 1031, 850, 813, 753; m/z (EI) 270 ($[\text{M}]^+$, 27), 121 ($[\text{4-MeOPhCH}_2]^+$, 100); HRMS (EI) calcd for $[\text{C}_{17}\text{H}_{18}\text{O}_3]^+$ 270.12505, found 270.12548.

3-Benzhydrylchroman-4-ol (2m). The title compound was synthesized according to the general procedure employing **1m** (0.12 mmol, 37.7 mg), 1 mol % (1.1 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 86% (32.7 mg); *anti/syn* 1.8:1; ^1H NMR (600 MHz, CDCl_3) δ 7.43 (d, $J = 7.2$ Hz, 2H), 7.39 (d, $J = 7.2$ Hz, 2H), 7.35–7.18 (m, 26.7H), 7.12–7.09 (m, 3H), 6.97–6.85 (m, 5.5H), 4.44 (bs, 1H, *syn*), 4.41 (bs, 1.8 \times 1H, *anti*), 4.27 (dd, $J = 11.2, 2.3$ Hz, 1.8 \times 1H, *anti*), 4.14–4.06 (m, 2H, *syn*), 4.00 (ddd, $J = 10.8, 3.5, 1.5$ Hz, 1H, *syn*), 3.96–3.93 (m, 1.8 \times 1H, *anti*), 3.70 (d, $J = 12.4$ Hz, 1.8 \times 1H, *anti*), 2.91 (ddd, $J = 11.9, 7.5, 3.2$ Hz, 1H, *syn*), 2.79 (dq, $J = 12.4, 2.3$ Hz, 1.8 \times 1H, *anti*), 2.00 (bs, 1.8 \times 1H, *anti*), 1.80 (bs, 1H, *syn*); ^{13}C NMR (151 MHz, CDCl_3) δ 154.4 (2C), 142.6, 142.4, 142.3 (2C), 130.9, 130.5, 130.0, 129.9, 128.9 (2C), 128.8 (2C), 128.1 (2C), 127.9, 127.8, 126.8, 126.7, 126.6, 126.6, 124.0, 122.5, 121.0, 120.5, 117.1, 116.9, 65.8, 64.1, 64.0, 63.1, 49.6, 49.2, 43.0, 41.9; FT-IR ν_{max} (ATR) cm^{-1} : 3036, 2927, 1702, 1619, 1516, 1491, 1446, 1318, 1226, 1117, 1014, 972, 755; m/z (EI) 167 ($[\text{CHPh}_2]^+$, 59), 151 ($[\text{M} - \text{CPh}_2]^+$, 28); HRMS (EI) calcd for $[\text{C}_{22}\text{H}_{20}\text{O}_2]^+$ 316.14578, found 316.14636.

2-Benzyl-2,3-dihydro-1H-inden-1-ol (2n). The title compound was synthesized according to the general procedure employing **1n** (0.12 mmol, 26.7 mg), 2.5 mol % (2.74 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 68% (14.6 mg); *anti/syn* 2.5:1; ^1H NMR (600 MHz, CDCl_3) δ 7.41–7.35 (m, 4H), 7.35–7.29 (m, 10H), 7.28–7.21 (m, 14.5H), 7.19–7.15 (m, 3H), 5.00 (d, $J = 5.4$ Hz, 1H, *syn*), 4.95 (d, $J = 6.6$ Hz, 2.5 \times 1H, *anti*), 3.14–3.07 (m, 4.5H), 3.01 (dd, $J = 15.3, 7.3$ Hz, 3H), 2.87–2.83 (m, 2.5H), 2.81–2.75 (m, 4H), 2.68 (dq, $J = 14.0, 7.9$ Hz, 1H, *syn*), 2.62–2.48 (m, 6H), 1.62 (bs, 3.5H); ^{13}C NMR (151 MHz, CDCl_3) δ 144.7, 144.5, 143.5, 141.5, 141.2, 140.6, 128.9 (2C), 128.7, 128.5, 128.4, 128.2, 126.8 (2C), 126.2, 125.9, 125.0, 124.8, 124.7, 123.9, 80.8, 76.3, 52.3, 46.9, 39.3, 35.9, 35.8, 35.0; FT-IR ν_{max} (ATR) cm^{-1} 3363, 2932, 1487, 1454, 1257, 1102, 1026, 804, 745, 697; m/z (EI) 223 ($[\text{M} - 1]^+$, 21), 132 ($[\text{M} - [\text{PhCH}_2]^+$, 100) 91 ($[\text{PhCH}_2]^+$, 80); HRMS (EI) calcd for $[\text{C}_{16}\text{H}_{16}\text{O}]^+$ 224.11957, found 224.11989.

3-(3-Methylbenzylidene)chroman-4-ol (2o). The title compound was synthesized according to the general procedure employing **1o** (0.12 mmol, 30 mg), 2.5 mol % (2.74 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 59% (17.7 mg); *E/Z* 1:1; ^1H NMR (400 MHz, CDCl_3) δ 7.41 (dd, $J = 7.7, 1.6$ Hz, 1H), 7.31–7.18 (m, 6H), 7.16–7.07 (m, 2H), 7.06–7.00 (m, 2H), 7.03–6.8 (m, 6H), 6.77 (s, 1H), 5.32 (s, 1H), 5.19 (s, 1H), 4.97–4.86 (m, 3H), 4.57 (dd, $J = 12.0, 0.9$ Hz, 1H), 2.37 (s, 3H), 2.35 (s, 3H), 2.14 (bs, 1H), 2.05 (bs, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 154.4, 154.3, 138.1 (2C), 135.5, 135.4, 134.4, 133.4, 131.6, 130.3, 130.0, 129.8, 129.5 (2C), 129.2, 129.0, 128.7, 128.4, 128.3 (2C), 125.9, 125.8, 124.9, 124.0, 121.1 (2C), 117.2, 116.9, 69.3, 68.1, 63.0, 62.8, 21.4 (2C); FT-IR ν_{max} (ATR) cm^{-1} 3391, 2922, 1737, 1590, 1479, 1370, 1224, 993, 921, 756, 702; m/z (EI) 252 ($[\text{M}]^+$, 17), 104 ($[\text{3-MePh}]^+$, 25), 91 ($[\text{3-MePhCH}]^+$, 32); HRMS calcd for $[\text{C}_{17}\text{H}_{16}\text{O}_2]^+$ 252.11448, found 252.11402.

3-(4-Methoxybenzylidene)chroman-4-ol (2p).²¹ The title compound was synthesized according to the general procedure employing **1p** (0.12 mmol, 32 mg), 2.5 mol % (2.74 mg) of PC 3, and 2.5 equiv of DIPEA (38.8 mg). The product was purified by flash column chromatography: yield 57% (18.3 mg); *E/Z* 1:1; ^1H NMR (600 MHz, CDCl_3) δ 7.46–7.43 (m, 2H), 7.42–7.39 (m, 1H), 7.30–7.28 (m, 1H), 7.24–7.20 (m, 2H), 7.20–7.16 (m, 2H), 6.99–6.84 (m, 9H), 6.76 (s, 1H), 5.33 (s, 1H), 5.18 (s, 1H), 4.99–4.89 (m, 2H), 4.87 (dd, $J = 12.0, 1.2$ Hz, 1H), 4.59–4.55 (m, 1H), 3.83 (s, 3H), 3.82 (s, 3H), 2.20 (bs, 1H), 2.07 (bs, 1H); ^{13}C NMR (151 MHz, CDCl_3) δ 159.3, 159.1, 154.4, 154.3, 133.3, 133.2, 131.9, 131.3, 130.3, 130.2, 130.0, 129.8, 129.1, 128.8, 128.1, 128.0, 125.0, 124.1, 121.1 (2C), 117.3, 116.9, 113.9 (2C), 69.4, 68.2, 63.0, 62.8, 55.3 (2C); FT-IR ν_{max} (ATR) cm^{-1} 3393, 2958, 2929, 1606, 1509, 1485, 1458, 1246, 1177, 1031, 995, 828, 755; m/z (EI) 268 ($[\text{M}]^+$, 25), 120 ($[\text{3-MeOPhCH}]^+$, 100).

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b01006.

¹H and ¹³C NMR spectra of products (PDF)

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: magnus.rueping@rwth-aachen.de.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The research leading to these results has received funding from the European Research Council under the European Union's Seventh Framework Programme (FP/2007-2013)/ERC Grant Agreement No. 617044 (SunCatChem).

■ REFERENCES

- (1) (a) Molander, G. A.; Kenny, C. *Tetrahedron Lett.* **1987**, 28, 4367. (b) Molander, G. A.; Kenny, C. *J. Am. Chem. Soc.* **1989**, 111, 8236.
- (2) Values calculated vs Ag/AgNO₃ as reference electrode in THF: (a) Shabangi, M.; Flowers, R. A., II *Tetrahedron Lett.* **1997**, 38, 1137. (b) Flowers, R. A., II. *Synlett* **2008**, 2008, 1427.
- (3) Selected examples using stoichiometric amount of tributyltin: (a) Beckwith, A. L. J.; Roberts, D. H. *J. Am. Chem. Soc.* **1986**, 108, 5893. (b) Ardisson, J.; Ferezou, J. P.; Julia, M.; Pancrazi, A. *Tetrahedron Lett.* **1987**, 28, 2001. (c) Sugawara, T.; Otter, B. A.; Ueda, T. *Tetrahedron Lett.* **1988**, 29, 75.
- (4) Enholm was the first to systematically explore the scope of this reaction: (a) Enholm, E. J.; Prasad, G. *Tetrahedron Lett.* **1989**, 30, 4939. (b) Enholm, E. J.; Burroff, J. A. *Tetrahedron Lett.* **1992**, 33, 1835.
- (5) Hays, D. S.; Fu, G. C. *J. Org. Chem.* **1996**, 61, 4.
- (6) Corey, E. J.; Pyne, S. G. *Tetrahedron Lett.* **1983**, 24, 2821.
- (7) Estévez, R. E.; Oller-López, J. L.; Robles, R.; Melgarejo, C. R.; Gansäuer, A.; Cuerva, J. M.; Oltra, J. E. *Org. Lett.* **2006**, 8, 5433.
- (8) (a) Molander, G. A.; Harris, C. R. *J. Org. Chem.* **1998**, 63, 812. (b) Corey, E. J.; Pyne, S. G. *Tetrahedron Lett.* **1983**, 24, 2821. (c) Streuff, J. *Synthesis* **2013**, 45, 281.
- (9) For selected reviews on photocatalysis, see: (a) Hoffmann, N. *Chem. Rev.* **2008**, 108, 1052. (b) Balzani, V.; Credi, A.; Venturi, M. *ChemSusChem* **2008**, 1, 26. (c) Ravelli, D.; Dondi, D.; Fagnoni, M.; Albini, A. *Chem. Soc. Rev.* **2009**, 38, 1999. (d) Ravelli, D.; Fagnoni, M.; Albini, A. *Chem. Soc. Rev.* **2013**, 42, 97.
- (10) For selected reviews on photoredox catalysis, see: (a) Zeitler, K. *Angew. Chem., Int. Ed.* **2009**, 48, 9785. (b) Yoon, T. P.; Ischay, M. A.; Du, J. *Nat. Chem.* **2010**, 2, 527. (c) Narayanam, J. M. R.; Stephenson, C. R. J. *Chem. Soc. Rev.* **2011**, 40, 102. (d) Ravelli, D.; Fagnoni, M. *ChemCatChem* **2012**, 4, 169. (e) Prier, C. K.; Rankic, D. A.; MacMillan, D. W. C. *Chem. Rev.* **2013**, 113, 5322.
- (11) (a) Pandey, G.; Hajra, S.; Ghorai, M. K.; Kumar, K. R. *J. Org. Chem.* **1997**, 62, 5966. (b) Tarantino, K. T.; Liu, P.; Knowles, R. R. *J. Am. Chem. Soc.* **2013**, 135, 10022. (c) Rono, L. J.; Yayla, H. G.; Wang, D. Y.; Armstrong, M. F.; Knowles, R. R. *J. Am. Chem. Soc.* **2013**, 135, 17735. (d) Petronijević, F. R.; Nappi, M.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2013**, 135, 18323. (e) Griesbeck, A. G.; Reckenthäler, M. *Beilstein J. Org. Chem.* **2014**, 10, 1143. (f) Wang, C.; Qin, J.; Shen, X.; Riedel, R.; Harms, K.; Meggers, E. *Angew. Chem., Int. Ed.* **2016**, 55, 685.
- (12) (a) Nakajima, M.; Fava, E.; Loescher, S.; Jiang, Z.; Rueping, M. *Angew. Chem., Int. Ed.* **2015**, 54, 8828. (b) Zhu, Z.; Das, A.; Bui, L.; Zhou, H. Z.; Curran, D. P.; Rueping, M. *J. Am. Chem. Soc.* **2013**, 135, 1823. (c) Nakajima, M.; Lefebvre, Q.; Rueping, M. *Chem. Commun.* **2014**, 50, 3619. (d) Lefebvre, Q.; Hoffmann, N.; Rueping, M. *Chem. Commun.* **2016**, 52, 2493.
- (13) For early examples of photochemical generation of ketyl radicals and their intramolecular addition on carbon-carbon double bonds, see: (a) Belotti, D.; Cossy, J.; Pete, J.-P.; Portella, C. *Tetrahedron Lett.* **1985**, 26, 4591. (b) Belotti, D.; Cossy, J.; Pete, J.-P.; Portella, C. *J. Org. Chem.* **1986**, 51, 4196. For an overview for the photochemical generation of ketyl radicals and their applications, see: (c) Cossy, J.; Belotti, D. *Tetrahedron* **2006**, 62, 6459.
- (14) Ishitani, O.; Yanagida, S.; Takamuku, S.; Pac, C. *J. Org. Chem.* **1987**, 52, 2790.
- (15) For a review on PCET, see: Huynh, M. H. V.; Meyer, T. J. *Chem. Rev.* **2007**, 107, 5004.
- (16) Conti, C.; Desideri, N. *Bioorg. Med. Chem.* **2009**, 17, 3720.
- (17) Dinnocenzo, J. P.; Banach, T. E. *J. Am. Chem. Soc.* **1989**, 111, 8646.
- (18) Bentley, J.; Nilsson, P. A.; Parsons, A. F. *J. Chem. Soc. Perkin Trans. 1* **2002**, 12, 1461.
- (19) Sangwan, N. K.; Rastogi, S. N. *Indian J. Chem. B* **1980**, 19, 500.
- (20) Gomis, M.; Kirkiacharian, B. S.; Likforman, J.; Mahuteau, J. *Bull. Soc. Chim. Fr.* **1988**, 3, 585.
- (21) Pivnenko, N. S.; Orlov, V. D.; Nodel'man, O. A.; Mikhed'kina, E. I.; Lavrushin, V. F. *J. Org. Chem. USSR (Engl. Trans.)* **1980**, 16, 1657.